

## Electronic Supporting Information

### Colloidal Synthesis of 1-D van der Waals material Nb<sub>2</sub>Se<sub>9</sub>: Study of Synergism of Coordinating agent in Co-solvent system

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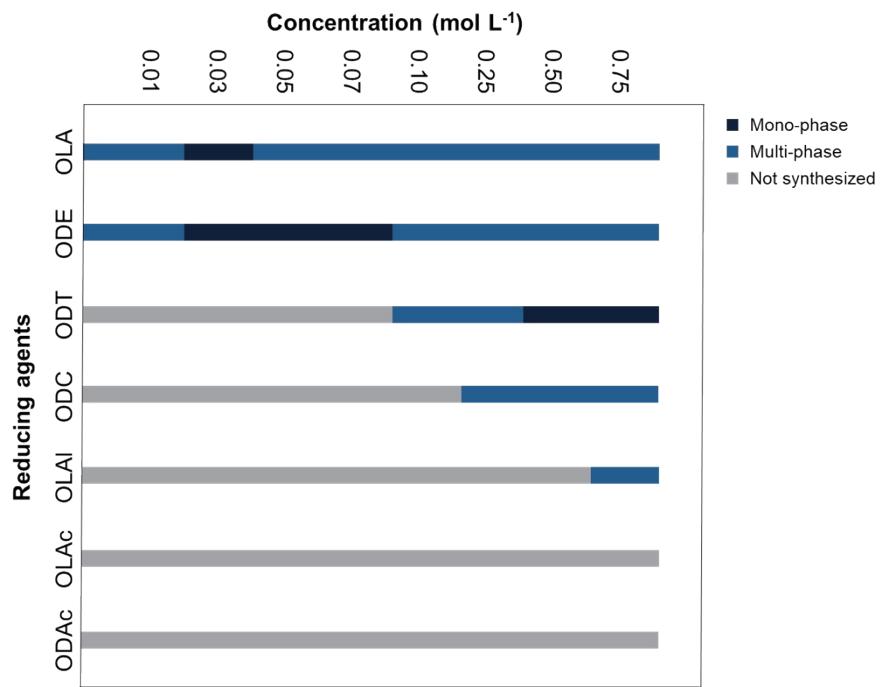
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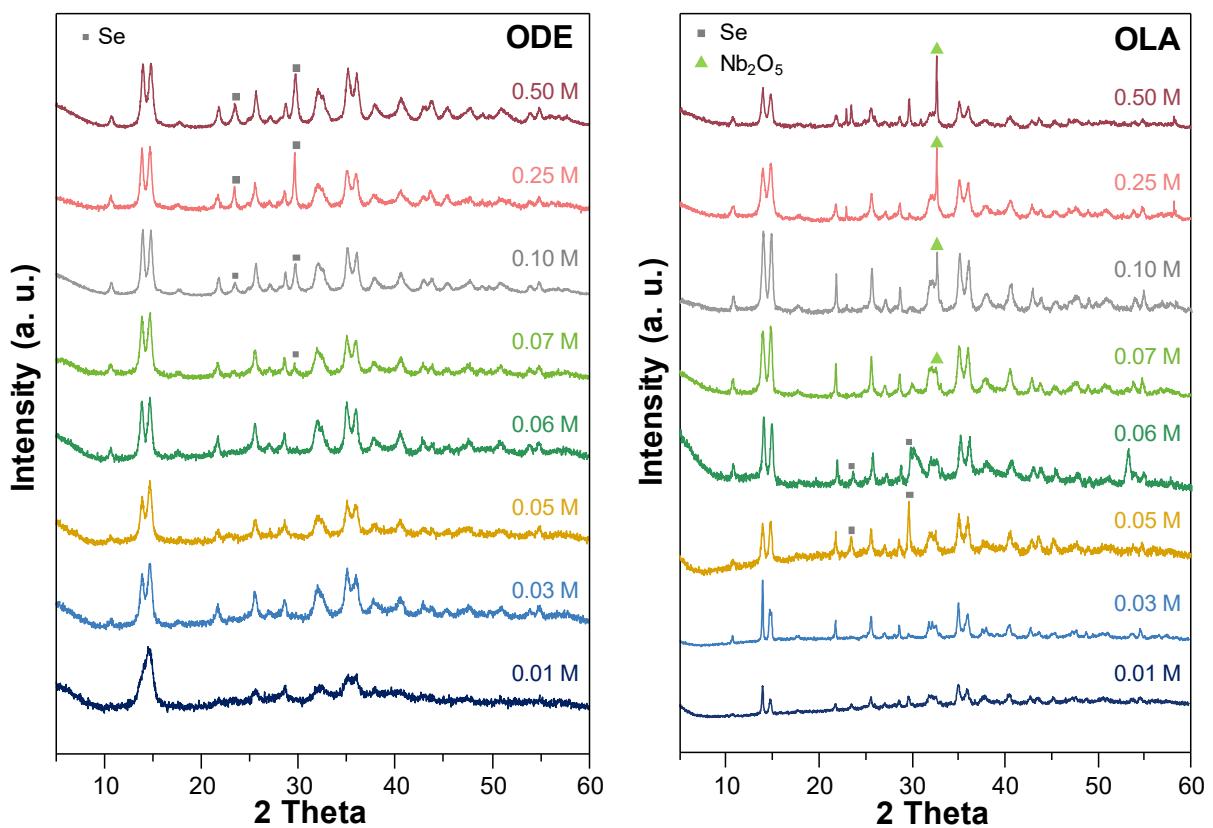
## Supplementary Figures

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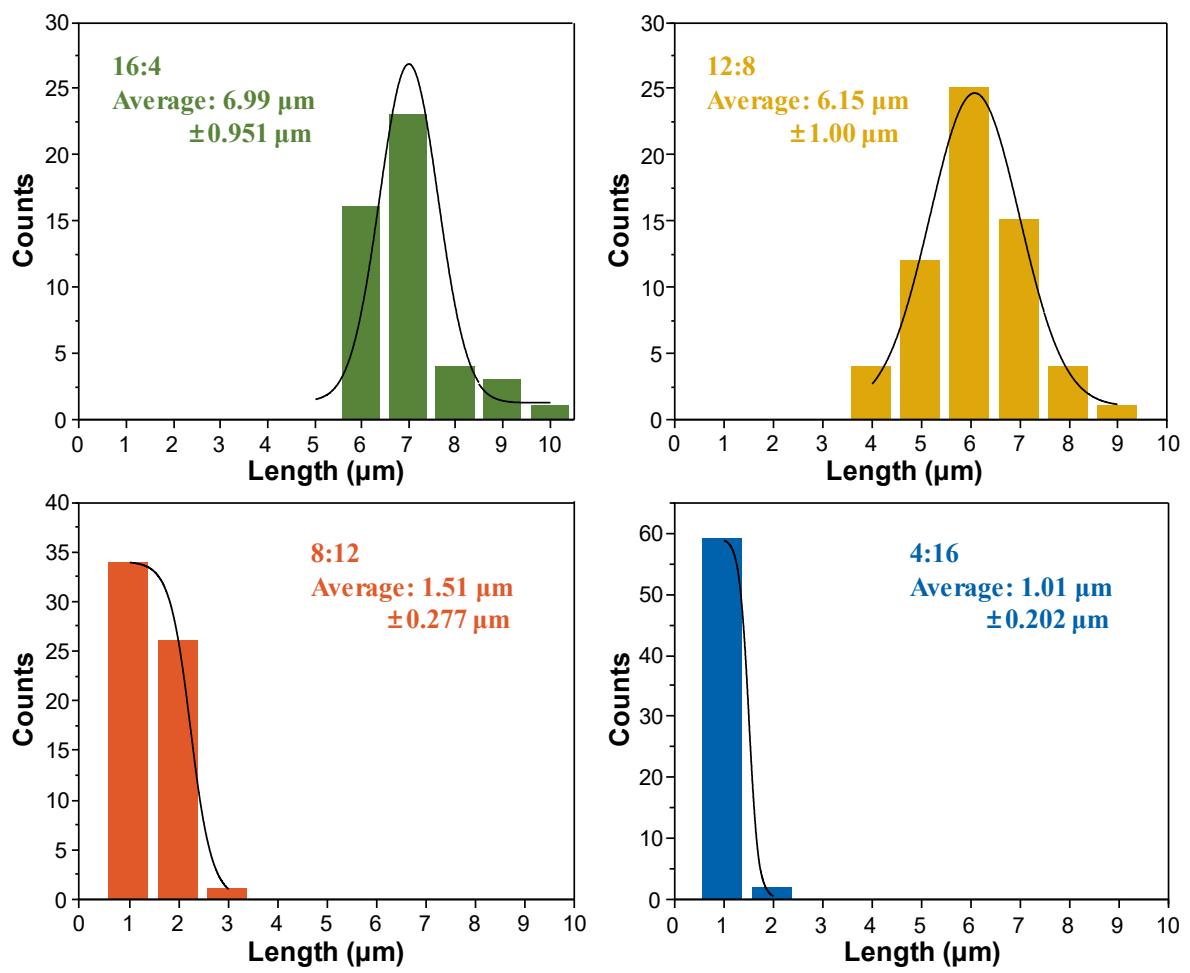
Solvent	Molecular structure	Functional group	Redox potential
Octadecanoic acid (ODAc)		COOH	● ODAc
Oleic acid (OLAc)		COOH	● OLAc
Oleylalcohol (OLAl)		OH	● OLAl
Octadecane (ODC)		alkane	● ODC
Octadecanethiol (ODT)		SH	● ODT
Octadecene (ODE)		alkene	● ODE
Oleylamine (OLA)		NH2	● OLA



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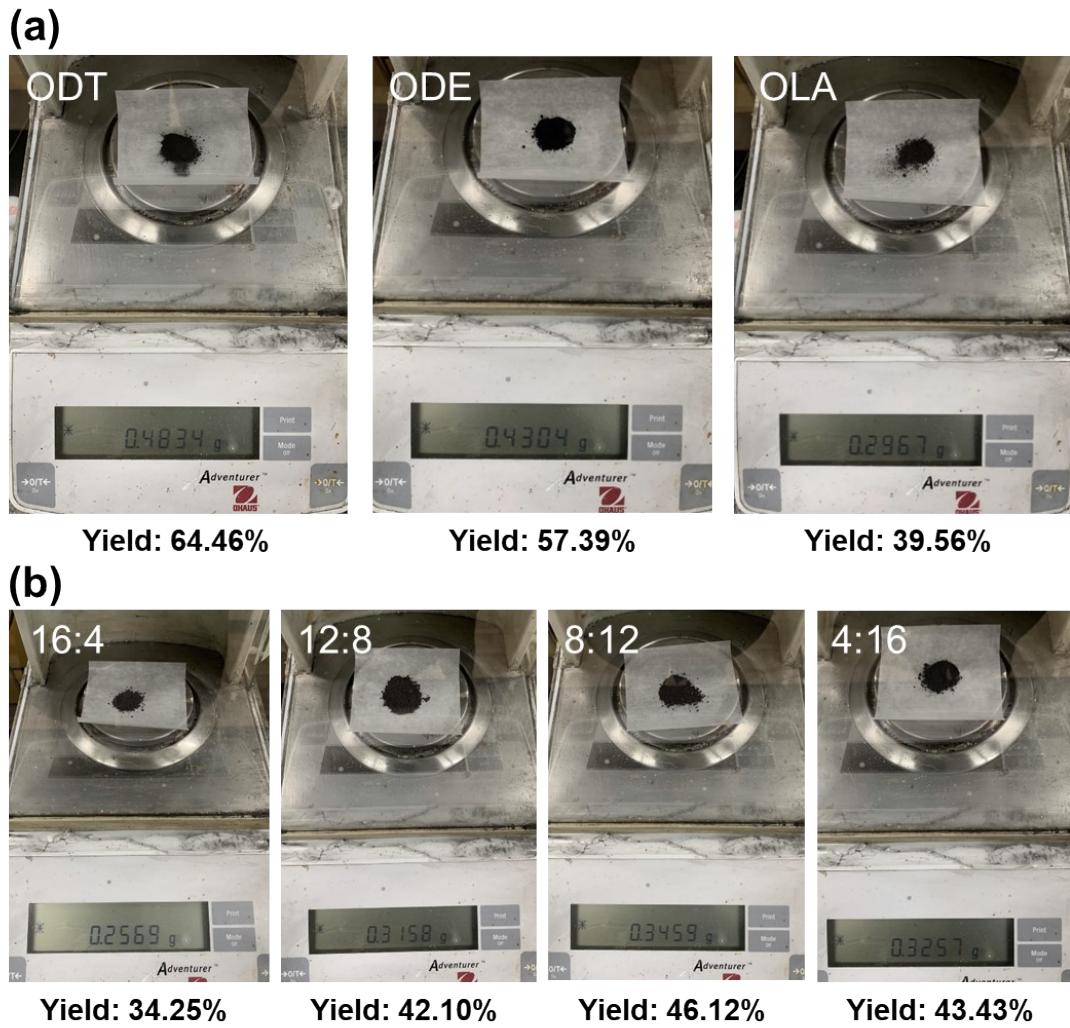
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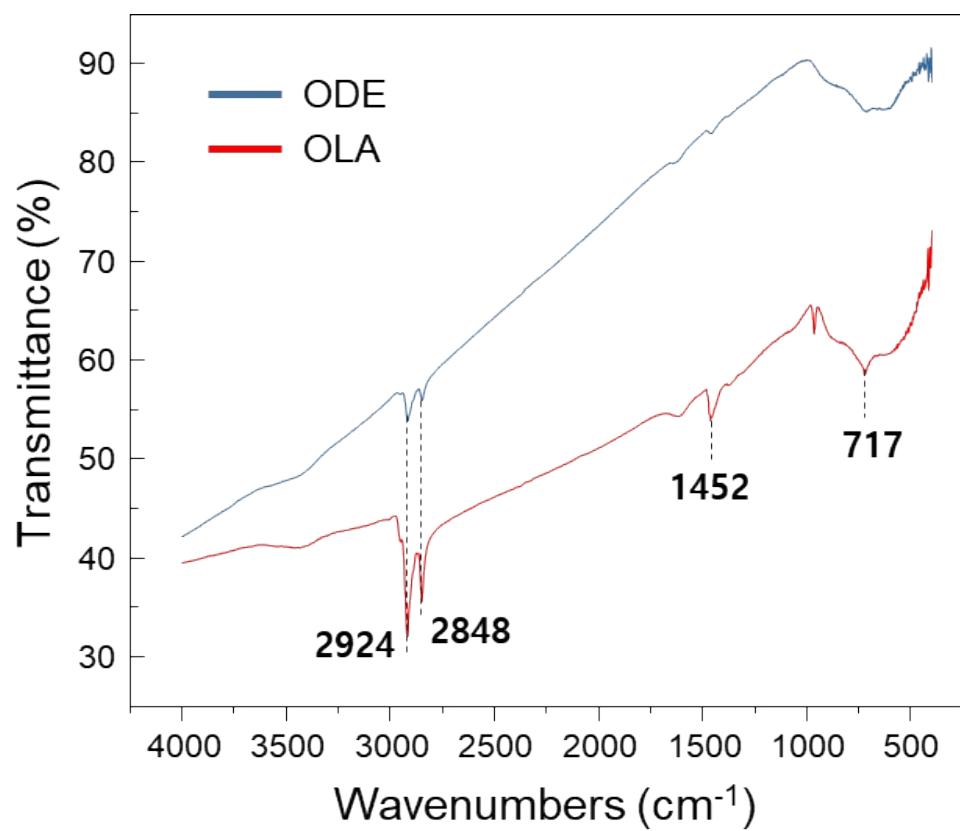
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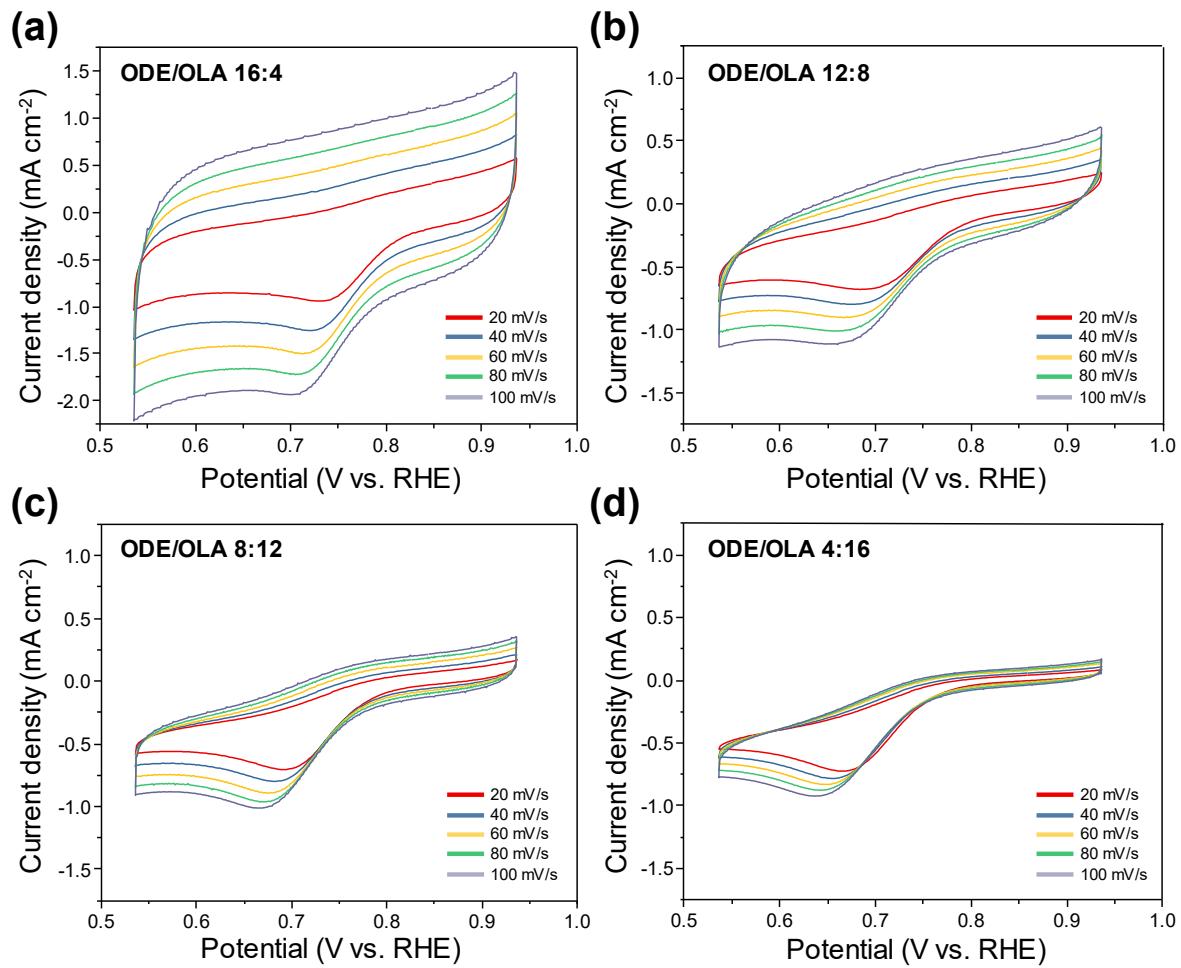
ODE:OLA	Morphology	Intensity ratio (010) / (100)	FWHM (010)	FWHM (100)
20:0	Rod	0.921	0.427	0.483
16:4	Wire	1.432	0.210	0.394
12:8	Wire	2.042	0.150	0.366
8:12	Sheet-like	2.752	0.114	0.165
4:16	Sheet-like	2.858	0.106	1.847
0:20	Sheet	2.071	0.131	0.328



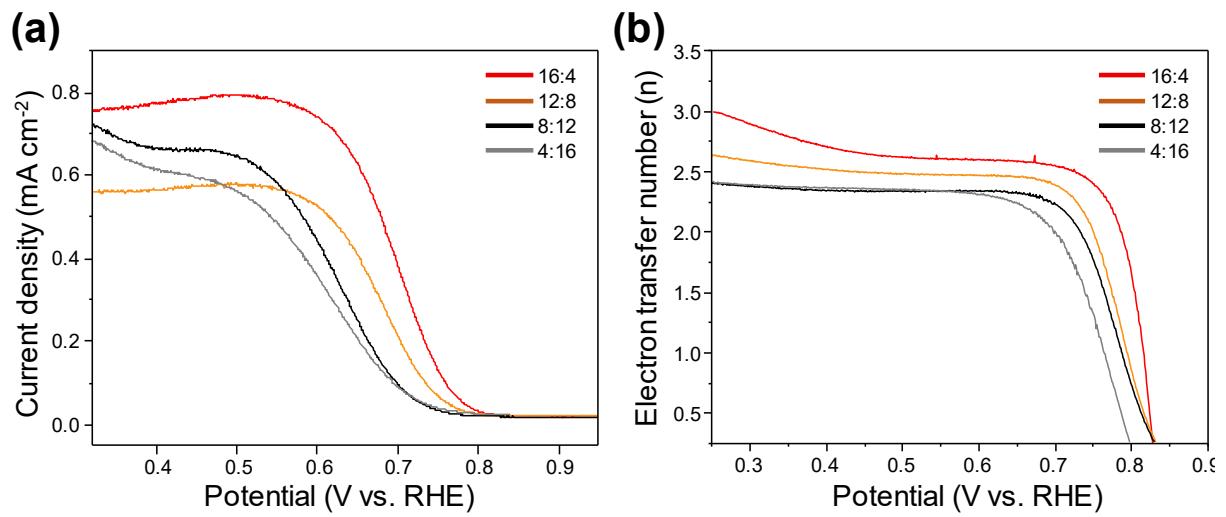
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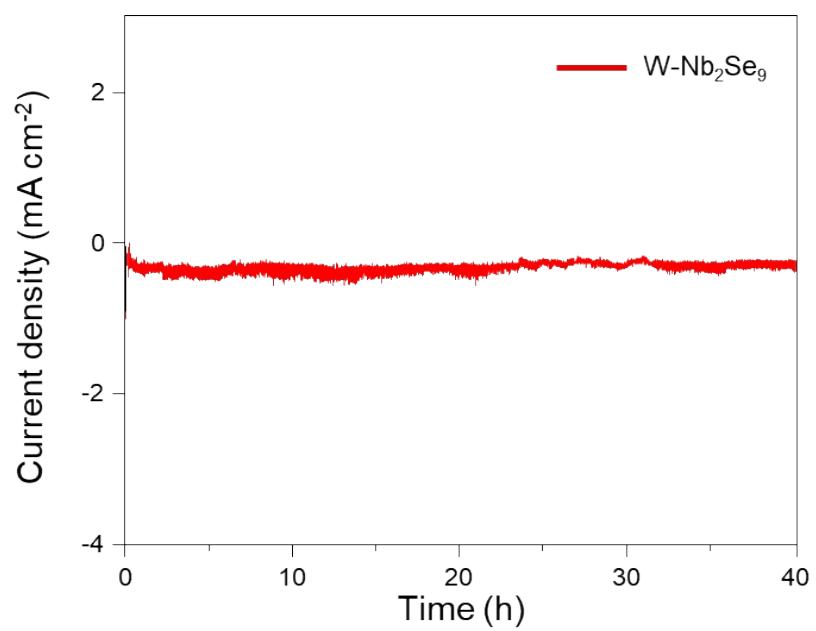
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**Fig. S8.** Chronoamperometry stability test of W-Nb<sub>2</sub>Se<sub>9</sub> from 16:4 condition.

**Table S3.** ORR performance of catalysts obtained from RDE.

Metal	Configuration	E <sub>onset</sub> vs. RHE	Tafel slope	Reference
Pt	Pt (20 wt%)/C	1.07 V in 0.1 M KOH	57 mV dec <sup>-1</sup>	1
Ag	Ag (110)	0.91 V in 0.1 M KOH	80 mV dec <sup>-1</sup>	2
Au	Au (111)	0.84 V in 0.1 M KOH		3
Mo	MoS	0.78 V in 0.1 M KOH,	54.7 mV dec <sup>-1</sup>	4
Nb	W-Nb <sub>2</sub> Se <sub>9</sub>	0.84 V in 0.1 M KOH	69.4 mV dec <sup>-1</sup>	This work

## References

1. J. Perez, E. R. Gonzalez, and E. A. Ticianelli, *Electrochim. Acta*, 1998, **44**, 1329-1339.
2. B. B. Blizanac, P. N. Ross, and N. M. Markovic, *J. Phys. Chem. B*, 2006, **110**, 4735-4741.
3. R. R. Adzic, S. Strbac, and N. Anastasijevic, *Materials Chemistry and Physics*, 1898, **22**, 349-375.
4. C. Tang, Y. Jiao, B. Shi, J. N. Liu, Z. Xie, X. Chen, Q. Zhang, and S. Z. Qiao, *Angew. Chem., Int. Ed.*, 2020, **59**, 2-8.