## **Supporting Information**

## **Diverse CsPbI<sub>3</sub> Assembly Structures: The Role of Surface Acids**

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## **Experimental part**

*Materials:* Lead (II) bromide (PbI<sub>2</sub>, 99%), Cesium carbonate (Cs<sub>2</sub>CO<sub>3</sub>, 99.9%), oleylamine (OAm, 80-90%), oleic acid (OA, 90%), and 1,4-Dodecyl benzene sulfonic acid (1,4-DBSA, 95%) were purchased from Aladdin. 1-octadecene (ODE, 90%), Ethyl acetate (EA, AR, 97%), and Hexane (anhydrous, 99.5%) were bought from Macklin. All chemicals were used without any further purification.

**Preparation of Cesium oleate precursors:**  $Cs_2CO_3$  (0.36 g, 1.1 mmoL), octadecene (15 mL), and oleic acid (1.5 mL) were added into 100 mL 3-neck flask, exhausted for half an hour at 120°C, and then heated to 150°C under Ar atmosphere until all  $Cs_2CO_3$  reacted with OA. The solution was kept at 120°C to avoid solidification before injection.

Synthesis of  $Cs_4PbI_6$  NCs: PbI<sub>2</sub> (0.54 mmoL, 0.1242 g), ODE (7.5 mL), OAm (1.5 mL), OA (1.5 mL) were loaded into a 100 mL 3-neck flask and degassed for half an hour at 120°C under Ar flow. The temperature was increased to 160°C under Ar atmosphere. The preheated Cs-oleate solution (0.75 mL, 0.033 mmoL) was swiftly injected into the transparent precursor solution. After 5 seconds, the reaction mixture was cooled down using an ice bath.

*Synthesis of nanowires-like CsPbI*<sub>3</sub> *NCs (CsPbI*<sub>3</sub>-*S*<sub>1</sub> *NCs):* PbI<sub>2</sub> (0.54 mmoL, 0.1242 g), ODE (7.5 mL), the molar ratio of 4-DBSA and OAm (2.6, 2.37 moL:0.91 moL) were loaded into a 100 mL 3-neck flask and degassed for half an hour at 120°C under Ar flow. The temperature was increased to 160°C and OA (1.5 mL) was injected into the 3-neck flask, and keeps heating up to 180°C under Ar atmosphere. When the temperature remains at 180°C, OAm (3.6 moL) was added into the reaction solution. The preheated Cs-oleate solution (0.75 mL, 0.033 mmoL) was injected into the transparent precursor solution swiftly. After 5 seconds, the reaction mixture was cooled down using an ice bath.

**Synthesis of nanospheres-like CsPbI<sub>3</sub> NCs (CsPbI<sub>3</sub>-S<sub>2</sub> NCs):** PbI<sub>2</sub> (0.54 mmoL, 0.1242 g), ODE (7.5 mL), the molar ratio of 4-DBSA and OAm (1.5, 2.37 moL:1.5 moL) were loaded into a 100 mL 3-neck flask and degassed for half an hour at 120°C under Ar flow. The temperature was increased to 160°C and OA (1.5 mL) was injected into the 3-neck flask, and keeps heating up to 180°C under Ar atmosphere. When the temperature remains at 180°C, OAm (3 moL) was added into the reaction solution. The preheated Cs-oleate solution (0.75 mL, 0.033 mmoL) was injected into the transparent precursor solution swiftly. After 5 seconds, the reaction mixture was cooled down using an ice bath.

**Synthesis of four-leaf clover-like CsPbI**<sub>3</sub> **NCs (CsPbI**<sub>3</sub>-**S**<sub>3</sub> **NCs):** PbI<sub>2</sub> (0.54 mmoL, 0.1242 g), ODE (7.5 mL), the molar ratio of 4-DBSA and OAm (1, 2.37 moL: 2.3 moL) were loaded into a 100 mL 3-neck flask and degassed for half an hour at 120°C under Ar flow. The temperature was increased to 160°C and OA (1.5 mL) was injected into the 3-neck flask, and keeps heating up to 180°C under Ar atmosphere. When the temperature remains at 180°C, OAm (2.3 moL) was added into the reaction solution. The preheated Cs-oleate solution (0.75 mL, 0.033 mmoL) was injected into the transparent precursor solution swiftly. After 5 seconds, the reaction mixture was cooled down using an ice bath.

Synthesis of NPLs-like CsPbl<sub>3</sub> NCs (CsPbl<sub>3</sub>-S<sub>4</sub> NCs):  $Pbl_2$  (0.54 mmoL, 0.1242 g), ODE (7.5 mL), the molar ratio of 4-DBSA and OAm (0.79, 2.37 moL: 3 moL) were loaded into a 100 mL 3-neck flask and degassed for half an hour at 120°C under Ar flow. The temperature was increased to 160°C and OA (1.5 mL) was

injected into the 3-neck flask, and keeps heating up to 180°C under Ar atmosphere. When the temperature remains at 180°C, OAm (1.5 moL) was added into the reaction solution. The preheated Cs-oleate solution (0.75 mL, 0.033 mmoL) was injected into the transparent precursor solution swiftly. After 5 seconds, the reaction mixture was cooled down using an ice bath.

Synthesis of CsPbI<sub>3</sub> NCs nanorods (CsPbI<sub>3</sub>-S<sub>6</sub> NCs): PbI<sub>2</sub> (0.54 mmoL, 0.1242 g), ODE (7.5 mL), the molar ratio of 4-DBSA and OAm (0.5, 2.37 moL: 4.6 moL) were loaded into a 100 mL 3-neck flask and degassed for half an hour at 120°C under Ar flow. The temperature was increased to 160°C and OA (1.5 mL) was injected into the 3-neck flask, and keeps heating up to 180°C under Ar atmosphere. The preheated Cs-oleate solution (0.75 mL, 0.033 mmoL) was injected into the transparent precursor solution swiftly. After 5 seconds, the reaction mixture was cooled down using an ice bath.

**Characterizations:** X-ray diffraction data of powder was recorded on Bruker D8 Advance using Nifiltered Cu K $\alpha$  radiation ( $\lambda$  = 1.542 Å). Fluorescence and absorption spectra were measured on Varian Cary Eclipse instrument and Shimadzu UV-3600, respectively. TEM measurements were carried out on a FEI Tecnai G20 with a Cu grid. X-ray photoelectron spectroscopy (XPS) measurements were performed using an achromatic Al K $\alpha$  source (1486.6 eV) and a double pass cylindrical mirror analyzer (ULVAC-PHI 5000 VersaProbe). FTIR results were measured with a Tensor-27 spectrometer. The absolute PLQY of NCs solution was determined using a Quantaurus-QY absolute photoluminescence quantum yield spectrometer (C11347-11, Hamamatsu Photonics, Japan).



Figure S1. The absorption data of Cs<sub>4</sub>PbI<sub>6</sub> NCs, OA-CsPbI<sub>3</sub> and Dumbbell-shaped CsPbI<sub>3</sub> NCs.



Figure S2. The XRD patterns of OA-CsPbI<sub>3</sub> and OA-CsPbI<sub>3</sub> (more OA) NCs.



Figure S3. The FTIR curves of OA-CsPbI<sub>3</sub> and OA-CsPbI<sub>3</sub> (more OA) NCs.



Figure S4. The size distribution of rod and heads of dumbbell-shaped CsPbl<sub>3</sub> NCs.



Figure S5. XPS refined curves of (a) O 1s of OA-CsPbI<sub>3</sub> and (b) dumbbell-shaped CsPbI<sub>3</sub> NCs.



Figure S6. FTIR curve of Cs<sub>4</sub>PbI<sub>6</sub> NCs.



Figure S7. FTIR curve of 4-DBSA.



**Figure S8.** Size distribution of (a)  $CsPbI_3-S_1$ , (b)  $CsPbI_3-S_2$ , (c)  $CsPbI_3-S_3$ , and (d)  $CsPbI_3-S_4$  NCs by changing the molar ratio of 4-DBSA and OAm.



Figure S9. TEM image (left) and size distribution (right) of CsPbl<sub>3</sub> -S<sub>6</sub> NCs.



Figure S10. PLQY data of OA-CsPbl<sub>3</sub> and CsPbl<sub>3</sub>-S<sub>n</sub> NCs.



Figure S11. (a) XRD pattern and (b) PL spectrum of CsPbI<sub>3</sub>-S<sub>6</sub> NCs.



**Figure S12.** TEM images of  $CsPbI_3$  NCs by adding 4-DBSA ligands in the second nucleation stage (a) and only adding 4-DBSA in the first nucleation (b).



Figure S13. The photos of OA-CsPbI<sub>3</sub> and dumbbell-shaped CsPbI<sub>3</sub> NCs in natural light.



Figure S14 Time-dependent XRD patterns of (a) OA-CsPbI<sub>3</sub> NCs and (b) Dumbbell-shaped-CsPbI<sub>3</sub>.