

Supporting Information

Valid Design and Evaluation of Cathode and Anode Materials of Aqueous Zinc Ion Batteries with High-Rate Capability and Cycle Stability

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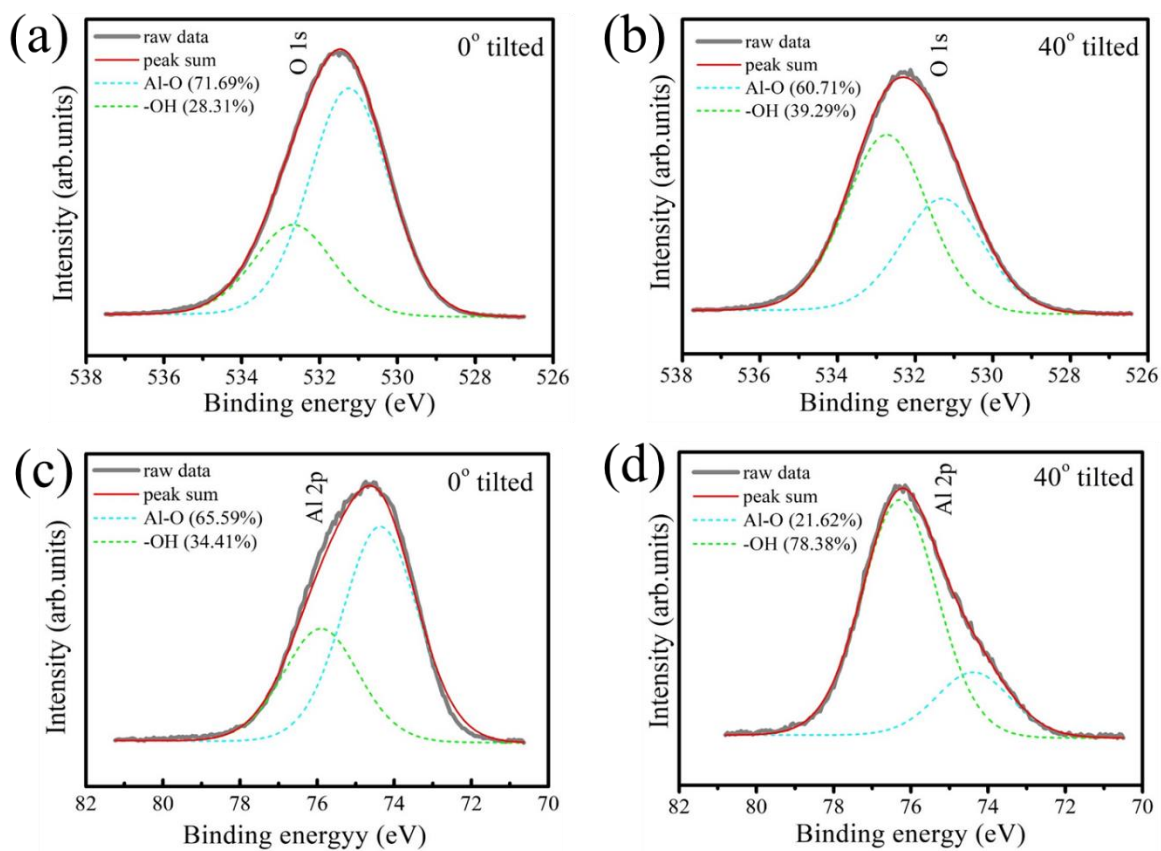


Figure S1. The narrow scan data of XPS spectra for the $\text{Al}_2\text{O}_3@\text{Zn}$. The normal XPS data for Al 2p and O 1s are shown in (a) and (b), respectively. Angle resolved XPS (ARXPS) data is shown in (c) and (d).

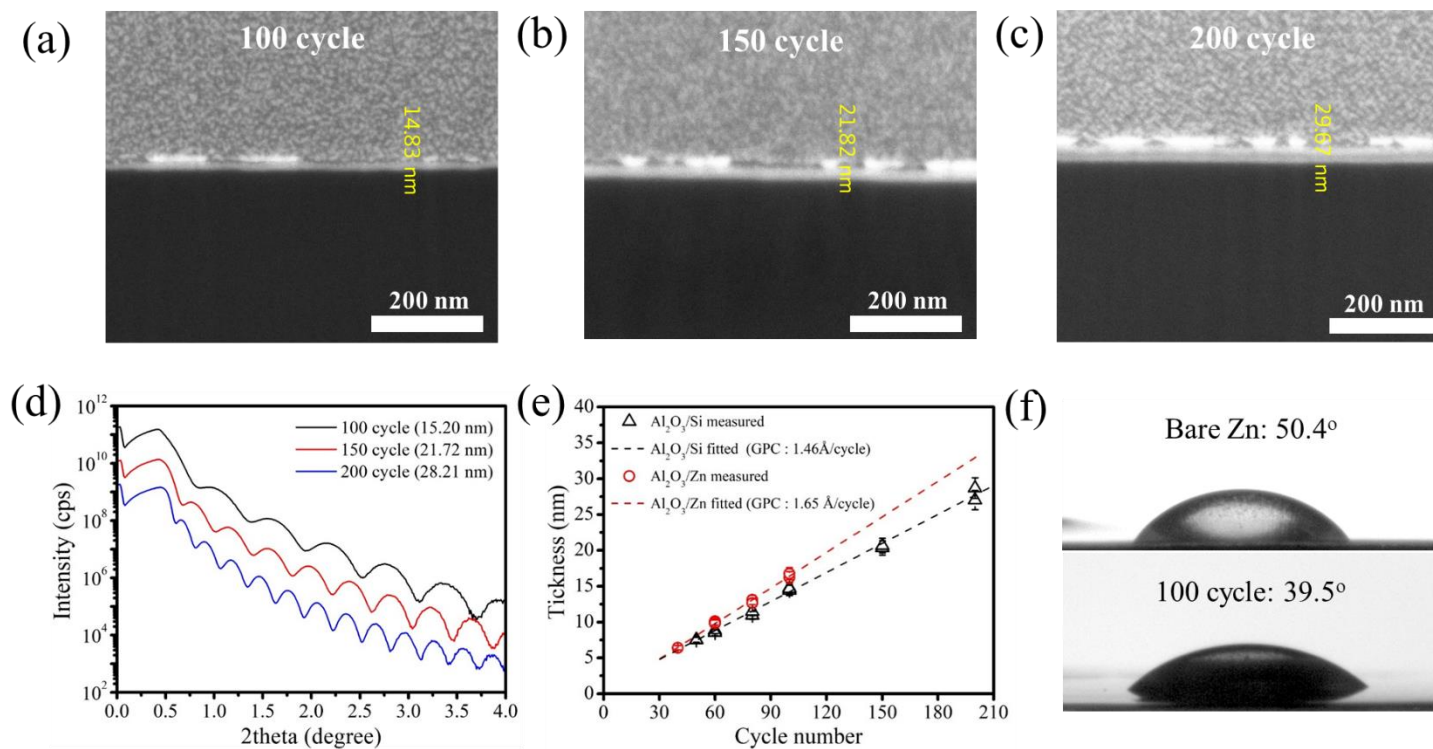


Figure S2. A thickness of the Al₂O₃@Si was analysed by SEM and XRR for the correction of complex refractive index values, which is compatible for the deposited Al₂O₃. The cross-section images of 100, 150, and 200Al₂O₃@Si are shown in (a), (b), and (c), measured by SEM. (d) XRR data of the Al₂O₃@Si (e) The growth rate of the Al₂O₃@Si and Al₂O₃@Zn indicated by growth per cycle (GPC). (f) The contact angles of electrolyte on bare Zn and 100Al₂O₃@Z

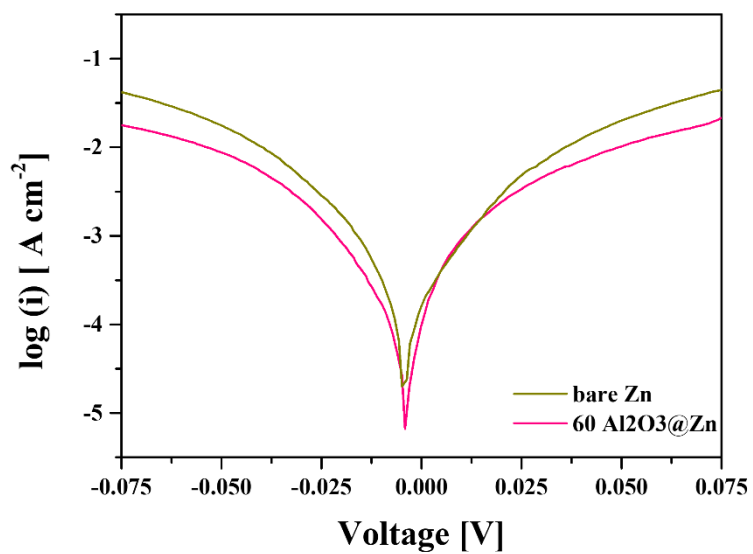


Figure S3. Tafel plot of bare Zn and 60 Al₂O₃@Zn. Linear sweep voltammetry (LSV) was conducted to calculate the corrosion current of the Zn/Zn symmetric cell. The voltage was swept from -150 to 150 mV at a scan rate of 1 mV s⁻¹.

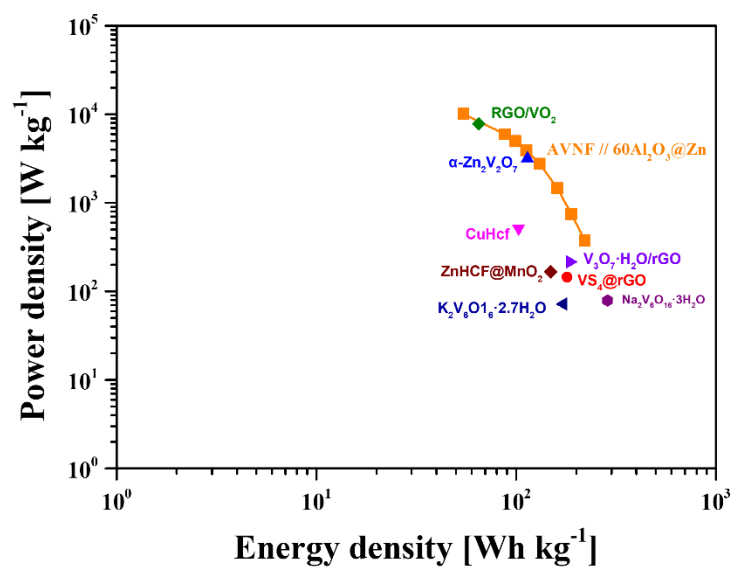


Figure S4. The ragone plot of AVNF//60Al₂O₃@Zn in comparison with other aqueous ZIBs.⁵⁹⁻⁶⁶

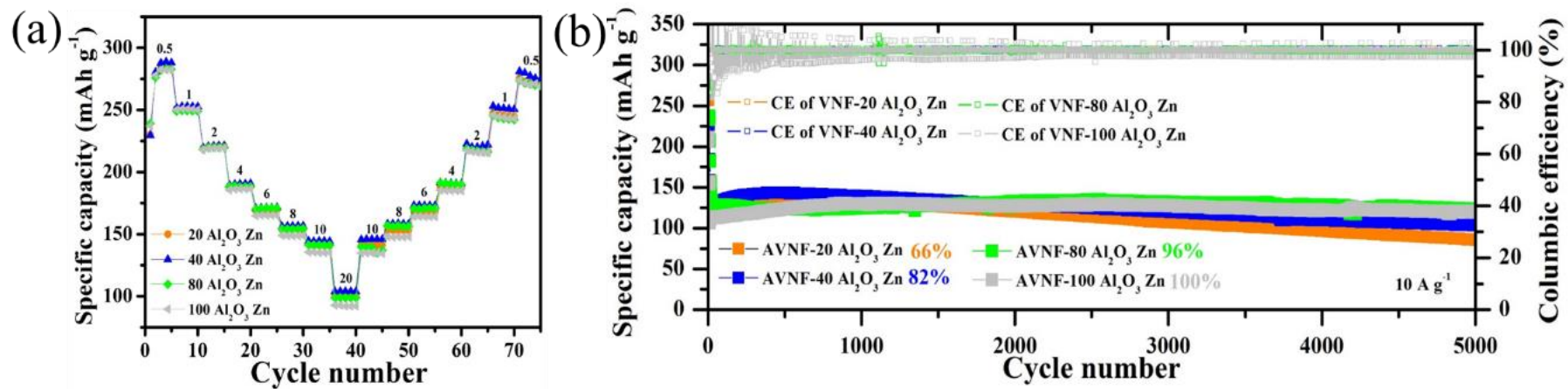


Figure S5. (a) A comparison of rate capabilities and (b) cycling performance of AVNF//Al₂O₃@Zn depending on the ALD cycle, which varied to 20, 40, 80, 100.

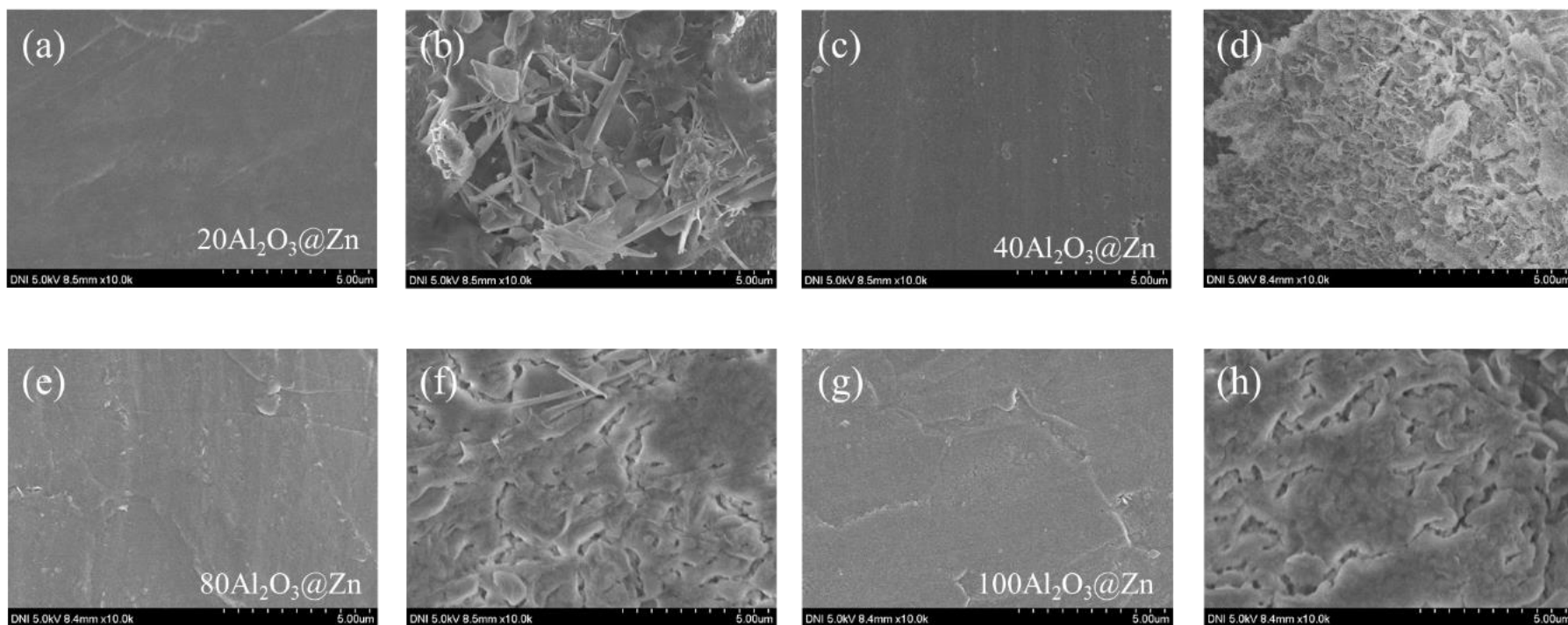


Figure S6. A series of the SEM images (top view) for 20, 40, 80 and 100 $\text{Al}_2\text{O}_3@\text{Zn}$ depending on the electrochemical cycling.

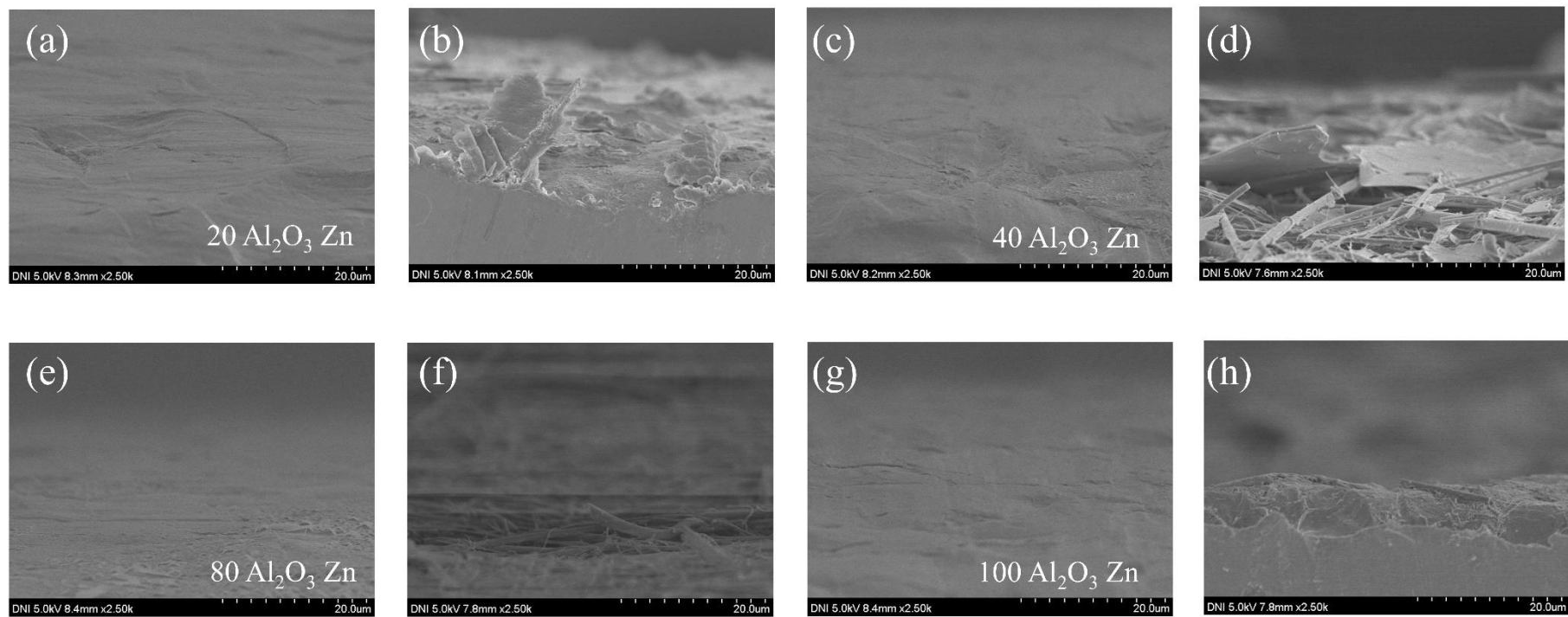


Figure S7. A series of the SEM images (cross-sectional view) for 20, 40, 80 and 100Al₂O₃@Zn depending on the electrochemical cycling.

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