

## Supporting Information

### **An engineered cascade-sensitized red-emitting upconversion nanoplatform with a tandem hydrophobic hydration-shell and metal-phenolic network decoration for single 808 nm triggered simultaneous tumor PDT and PTT enhanced CDT**

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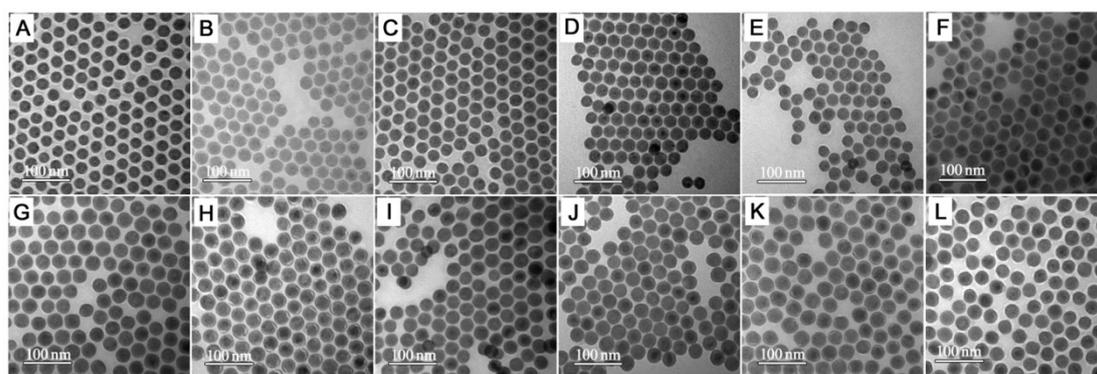
### **Preparation of NaErF<sub>4</sub>: Yb/Tm(69/1%):**

The NaErF<sub>4</sub>:Yb/Tm(69/1%) core was synthesized through binary solvothermal method. In brief, 0.6 mmol of corresponding rare-earth acetates (30 mol% Er(CH<sub>3</sub>COO)<sub>3</sub>, 69 mol% Yb(CH<sub>3</sub>COO)<sub>3</sub>, 1 mol% Tm(CH<sub>3</sub>COO)<sub>3</sub>), 3.6 mL OA and 9 mL ODE were mixed in a 50 mL three-necked bottle. The system was heated to 160 °C and reacted at this temperature for 30 min under argon protection. After the reaction solution cooled down, 1.30 mmol of NaOA was added and stirred for 10 min before adding 6 mL of methanol solution containing 0.2 mmol of NaOH and 2.4 mmol of NH<sub>4</sub>F. The reaction solution was stirred for 30 min, then heated to 100 °C to evaporate the low-boiling point solvents. Finally, the reaction solution was heated to 300 °C and reacted at this temperature for 60 min. Upon cooling to room temperature, the as-formed core nanoparticles of NaErF<sub>4</sub>:Yb/Tm(69/1%) were purified and re-dispersed in 6 mL of cyclohexane.

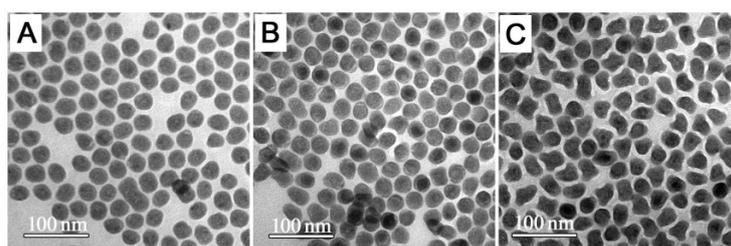
Synthesis of NaErF<sub>4</sub>:Yb/Tm(69/1%)@NaLuF<sub>4</sub>:Yb(x%). Typically, 0.3 mmol of corresponding rare-earth acetates (x% Yb(CH<sub>3</sub>COO)<sub>3</sub>, (100-x)% Lu(CH<sub>3</sub>COO)<sub>3</sub>), 3 mL of OA and 6 mL of ODE were mixed in a 50 mL three-necked bottle. Then, the resulting mixture was heated to 150 °C and reacted for 60 min under argon protection. After the mixture was cooled down to 50 °C, 3 mL of the as-prepared NaErF<sub>4</sub>:Yb/Tm(69/1%) core nanoparticles in cyclohexane were added. Thereafter, 5 mL methanol solution containing NaOH (0.75 mmol) and NH<sub>4</sub>F (1.2 mmol) was added dropwisely. After vigorous stirring at 50 °C for 30 min, the low-boiling point solvents in the reaction system were evaporated by heating to 100 °C. Subsequently, the reaction system heated to 300 °C and reacted at this temperature for 1 h under argon protection. Upon cooling to room temperature, the as-formed NaErF<sub>4</sub>:Yb/Tm(69/1%)@NaLuF<sub>4</sub>:Yb(x%) nanoparticles were purified and re-

dispersed in 3 mL of cyclohexane.

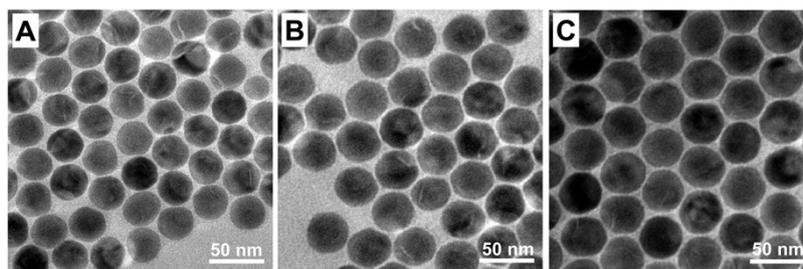
Preparation of core/multi-shell structured  $\text{NaErF}_4:\text{Yb}/\text{Tm}(69/1\%)\text{@NaLuF}_4:\text{Yb}(x\%)\text{@NaLuF}_4:\text{Nd}/\text{Yb}(20/10\%)$ ,  $\text{NaErF}_4:\text{Yb}/\text{Tm}(69/1\%)\text{@NaLuF}_4:\text{Yb}(15\%)\text{@NaLuF}_4:\text{Nd}/\text{Yb}(x/10\%)$  and  $\text{NaErF}_4:\text{Yb}/\text{Tm}(69/1\%)\text{@NaLuF}_4:\text{Yb}(15\%)\text{@NaLuF}_4:\text{Nd}/\text{Yb}(30/10\%)\text{@NaLuF}_4$ . The synthesis procedures are identical to those of in preparing core/shell structured  $\text{NaErF}_4:\text{Yb}/\text{Tm}(69/1\%)\text{@NaLuF}_4:\text{Yb}(x\%)$  nanoparticles. Particularly, thicker sensitization layer ( $\text{NaLuF}_4:\text{Yb}(15\%)$ ) and inert shell ( $\text{NaLuF}_4$ ) were achieved by increasing the rare-earth amounts in preparing the corresponding shell growth stock solutions.



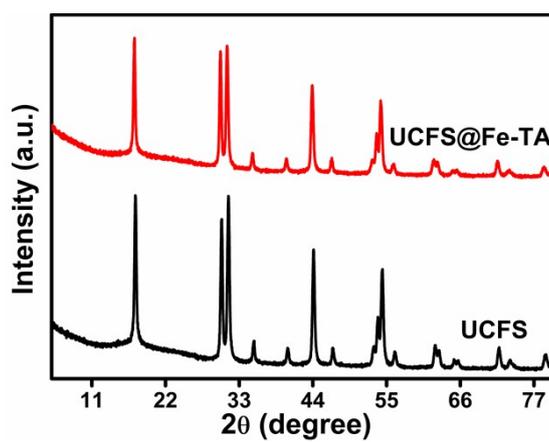
**Fig. S1** TEM images of the  $\text{NaErF}_4:\text{Yb}/\text{Tm}(69/1\%)\text{@NaLuF}_4:\text{Yb}(x\%)$  (A-F) and corresponding energy sensitization layer ( $\text{NaLuF}_4:\text{Nd}/\text{Yb}(20/10\%)$ ) coated  $\text{NaErF}_4:\text{Yb}/\text{Tm}(69/1\%)\text{@NaLuF}_4:\text{Yb}(x\%)$  (G-L), respectively. The  $\text{Yb}^{3+}$  doping ratios ( $x\%$ ) are 2.5%, 5%, 10%, 15%, 20% and 30% for the samples used in (A, G), (B, H), (C, I), (D, J), (E, K) and (F, L), respectively.



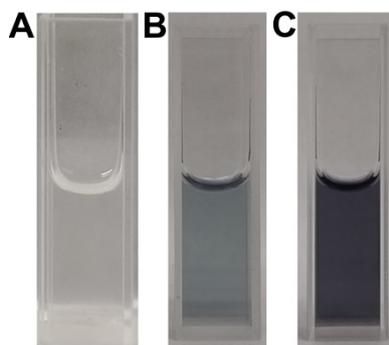
**Fig. S2** TEM images of  $\text{NaErF}_4:\text{Yb}/\text{Tm}(69/1\%)\text{@NaLuF}_4:\text{Yb}(15\%)\text{@NaLuF}_4:\text{Nd}/\text{Yb}(x/10\%)$ . The  $\text{Nd}^{3+}$  doping ratios ( $x\%$ ) are 30%, 40% and 50% for the samples used in (A), (B) and (C), respectively. The sensitization layer thickness in the  $\text{NaErF}_4:\text{Yb}/\text{Tm}(69/1\%)\text{@NaLuF}_4:\text{Yb}(15\%)\text{@NaLuF}_4:\text{Nd}/\text{Yb}(30/10\%)$  is  $\sim 1.1$  nm.



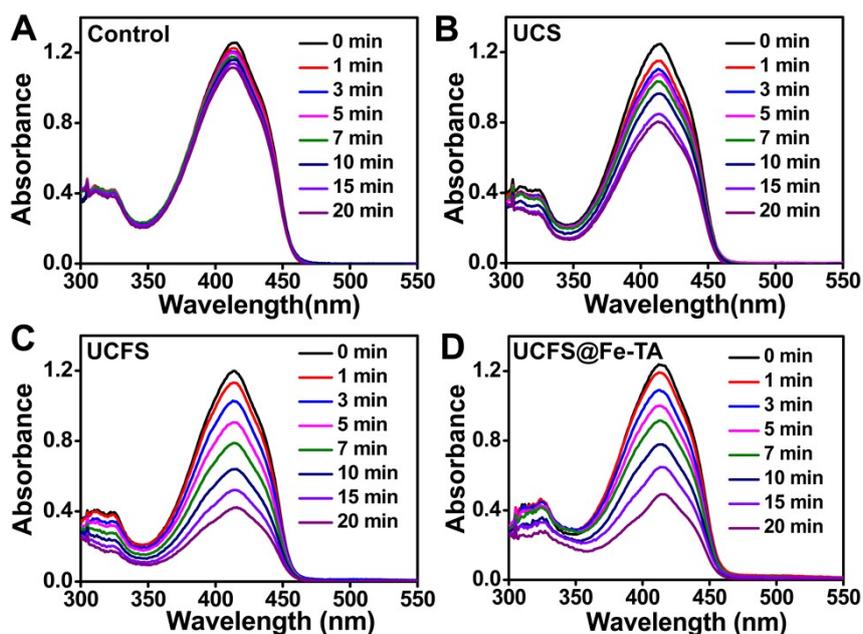
**Fig. S3** TEM images of  $\text{NaErF}_4:\text{Yb}/\text{Tm}(69/1\%)\text{@NaLuF}_4:\text{Yb}(15\%)\text{@NaLuF}_4:\text{Nd}/\text{Yb}(30/10\%)\text{@NaLuF}_4$  with different inter shell ( $\text{NaLuF}_4$ ) thickness: (A) 1.5 nm, (B) 3.0 nm, (C) 5.0 nm.



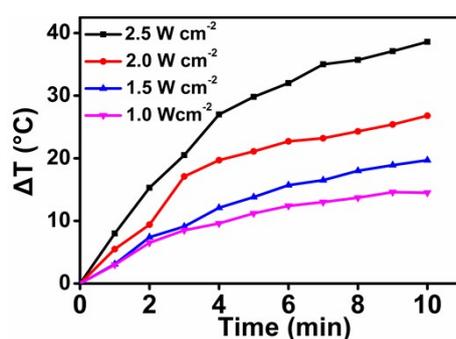
**Fig. S4** X-ray diffraction patterns of the UCFS and UCFS@Fe-TA.



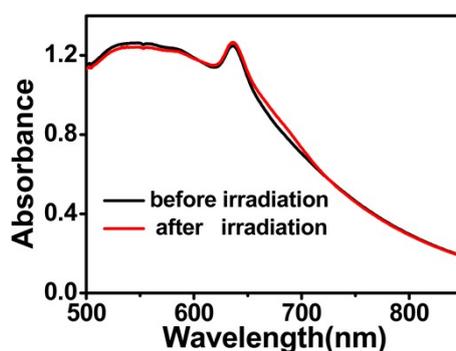
**Fig. S5** Photographs of the US (A), UCFS (B) and UCFS@Fe-TA (C).



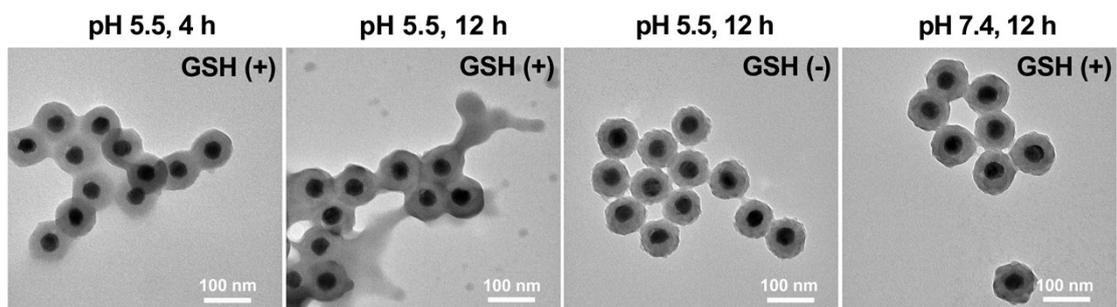
**Fig. S6** Time-dependent DPBF depletion capacity of the UCS, UCFS and UCFS@Fe-TA under 808 nm irradiation ( $2 \text{ W cm}^{-2}$ ).



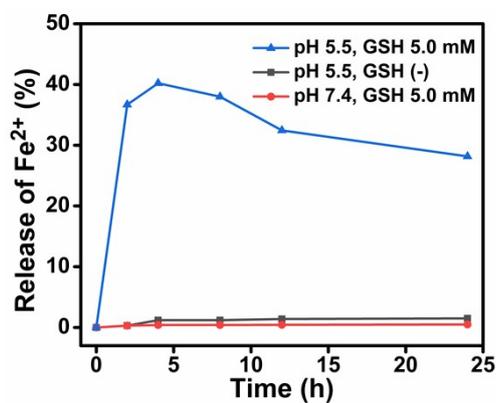
**Fig. S7** NIR irradiation power density-dependent temperature rise characteristic of the aqueous UCFS@Fe-TA ( $0.25 \text{ mg mL}^{-1}$ ).



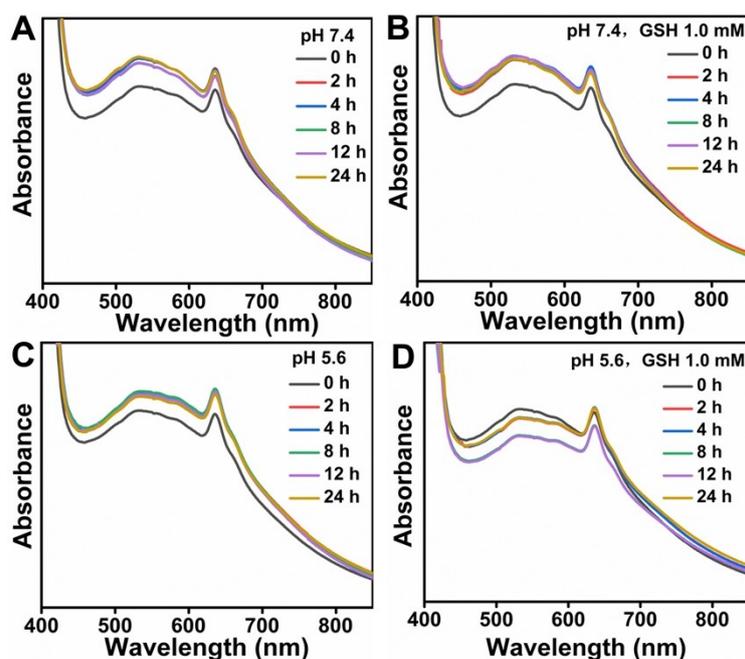
**Fig. S8** The absorbance change of UCFS@Fe-TA after continuous five circles of 808 nm irradiation.



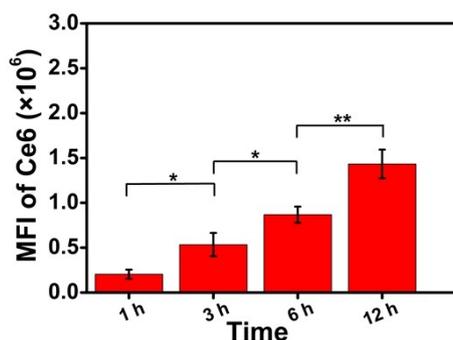
**Fig. S9** TEM observations of the Fe-TA dissociation on the surface of UCFS at different time and under different conditions. Results suggest Fe-TA will dissociate under acidic conditions in the presence of GSH.



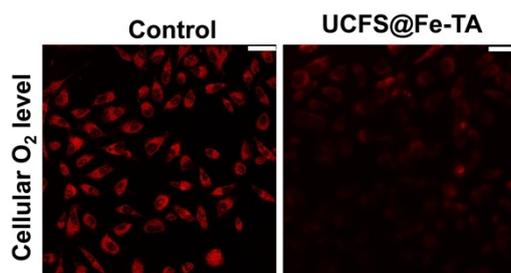
**Fig. S10** Evaluation of the Fe-TA dissociation on the surface of UCFS at different time and under different conditions by using *o*-phenanthroline method. Results suggest Fe-TA will dissociate under acidic conditions in the presence of GSH.



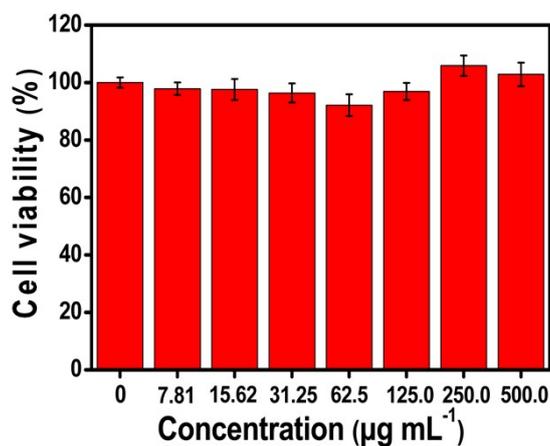
**Fig. S11** UV-vis spectra of UCFS@Fe-TA after incubation in acidic conditions with or without GSH addition. Results suggested that the absorbance of UCFS@Fe-TA showed significant increase in neutral conditions even with GSH addition (Fig. S11A-B), and also significant increase in acidic conditions without GSH addition (Fig. S11C). The absorbance of UCFS@Fe-TA decreased in acidic conditions with GSH addition at 8~12 h, however, increase back up to the original level at 24 h (Fig. S11D). Upon GSH was consumed completely, the  $\text{Fe}^{2+}$  might convert back to  $\text{Fe}^{3+}$  due to the oxidation effect of dissolved  $\text{O}_2$  in water (Fig. S11D). The NIR absorption capacity loss of UCFS@Fe-TA in acidic conditions with GSH addition is limited, indicating the Fe-TA dissociation and reliable PTT of Fe-TA are not in conflict.



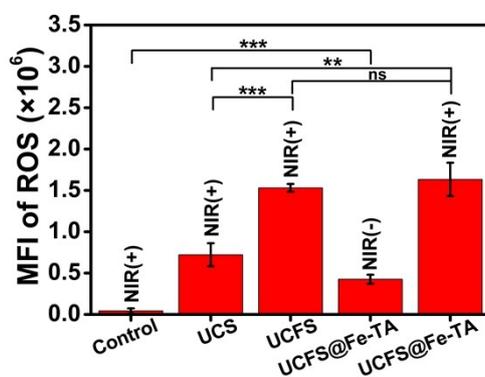
**Fig. S12** Mean fluorescence intensity (MFI) of Ce6 in MCF-7 cells after incubating with UCFS@Fe-TA for different times. \* $p < 0.05$ , \*\* $p < 0.01$  determined by Student's  $t$  test.



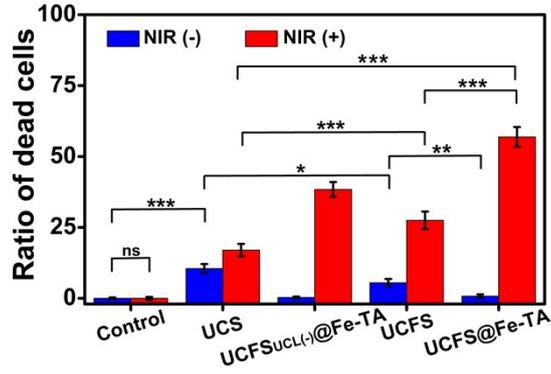
**Fig. S13** O<sub>2</sub> content evaluation in UCFS@Fe-TA treated MCF-7 cells by the intracellular O<sub>2</sub> level indicator [Ru(dpp)<sub>3</sub>]Cl<sub>2</sub>. The hypoxia nature of MCF-7 cells was relieved after incubating with UCFS@Fe-TA (12 h).



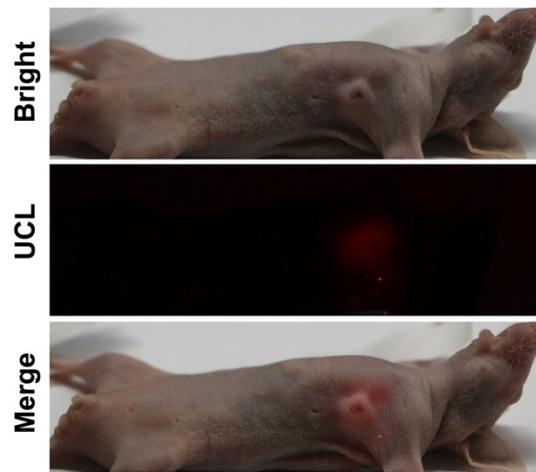
**Fig. S14** The viability of L929 cells after incubation with UCFS@Fe-TA for 24 h.



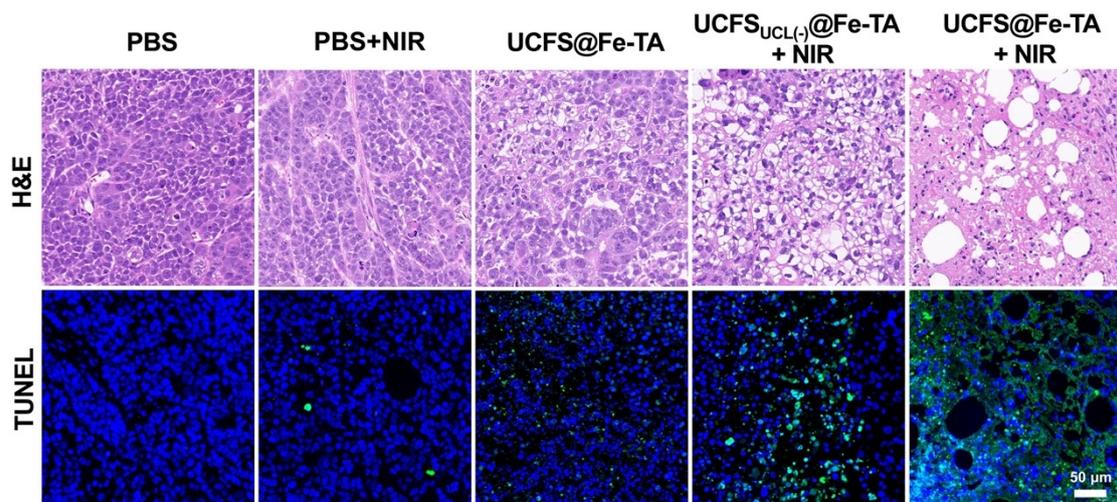
**Fig. S15** MFI of ROS in MCF-7 cells after treating by the as-involved nanoprob. **\*\*** $p < 0.01$ , **\*\*\*** $p < 0.001$  determined by Student's  $t$  test. ns, not significant ( $p > 0.05$ ).



**Fig. S16** Ratio of dead cells calculated from the calcein-AM/PI cell double staining results. \* $p < 0.05$ , \*\* $p < 0.01$ , \*\*\* $p < 0.001$  determined by Student's  $t$  test. ns, not significant ( $p > 0.05$ ).



**Fig. S17** The UCL based in vivo imaging of tumor bearing mouse after iv injecting with the UCFS@Fe-TA for 12 h, demonstrating tumor accumulation of UCFS@Fe-TA.



**Fig. S18** H&E and TUNEL staining images of tumor tissues after 14 days of treatments.