

Cationic vacancy engineering of p-TiO₂ for enhanced photocatalytic nitrogen fixation

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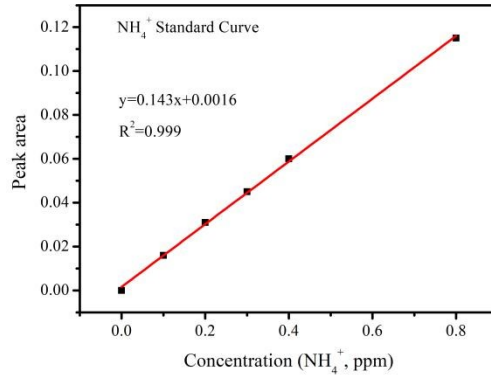


Fig. S1. Standard curve for NH₄⁺ concentration.

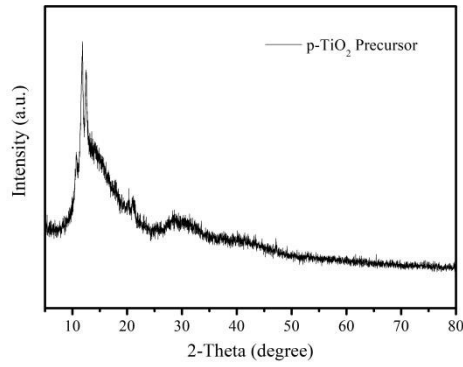


Fig. S2. XRD patterns of p-TiO₂ precursor.

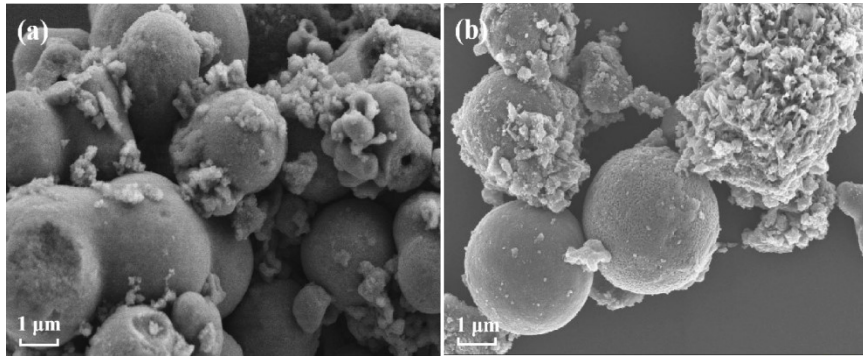


Fig. S3. SEM images of (a) n-TiO₂ and (b) p-TiO₂ (II).

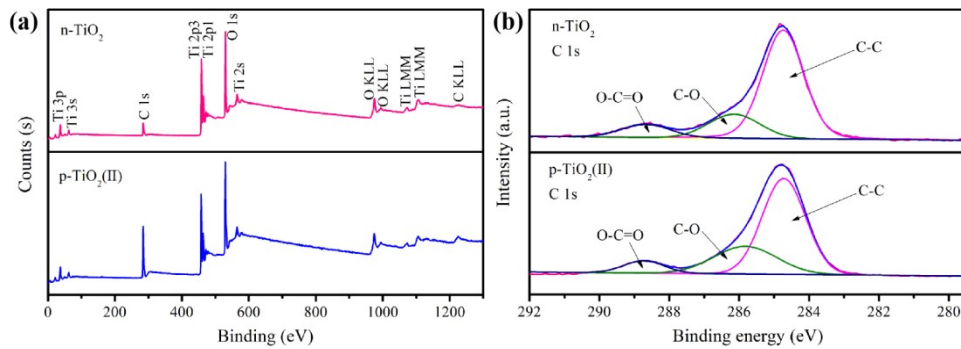


Fig. S4. (a) total XPS spectrum, (b) XPS spectra for C 1s of n-TiO₂ and p-TiO₂ (II).

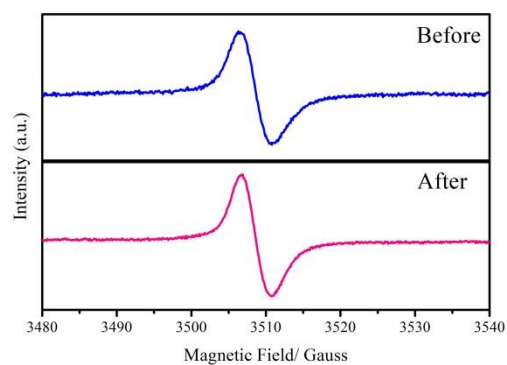


Fig. S5. The EPR of p-TiO₂ (II) before and after cycle reaction.

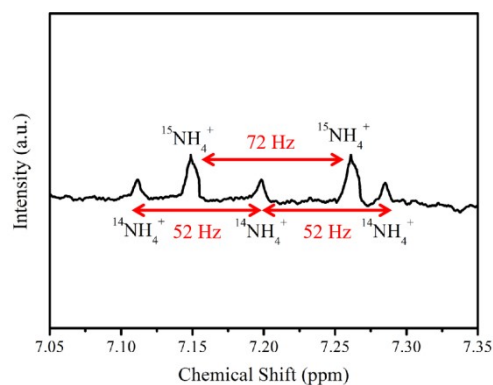


Fig. S6. ¹H NMR spectrum of the solution obtained after NRR on TiO₂ (II) using ¹⁵N₂ as the reaction

gas

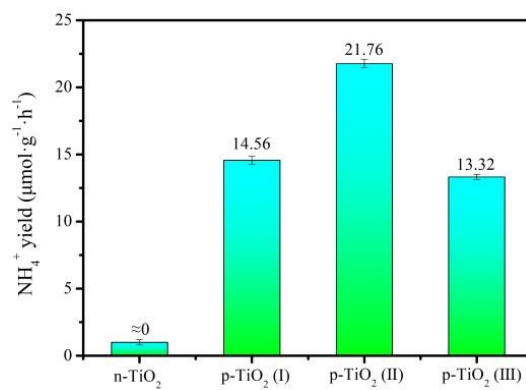


Fig. S7. The NH₄⁺ production rate of TiO₂ materials under visible light in Air.

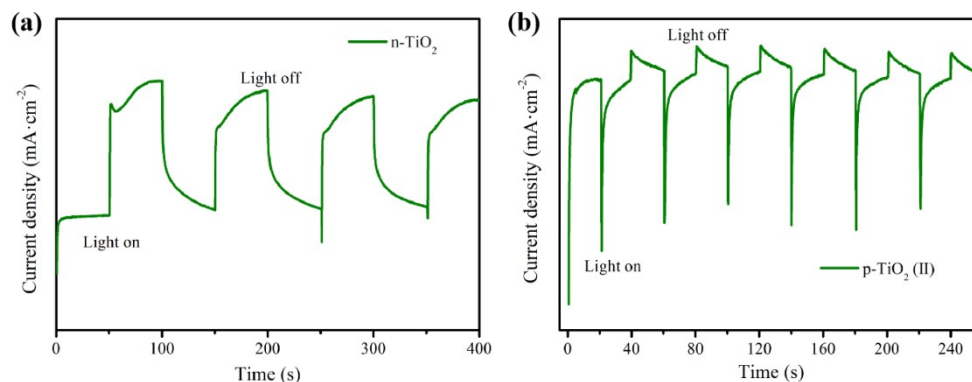


Fig. S8. The transient photocurrent curve of (a) n-TiO₂ and (b) p-TiO₂ (II).

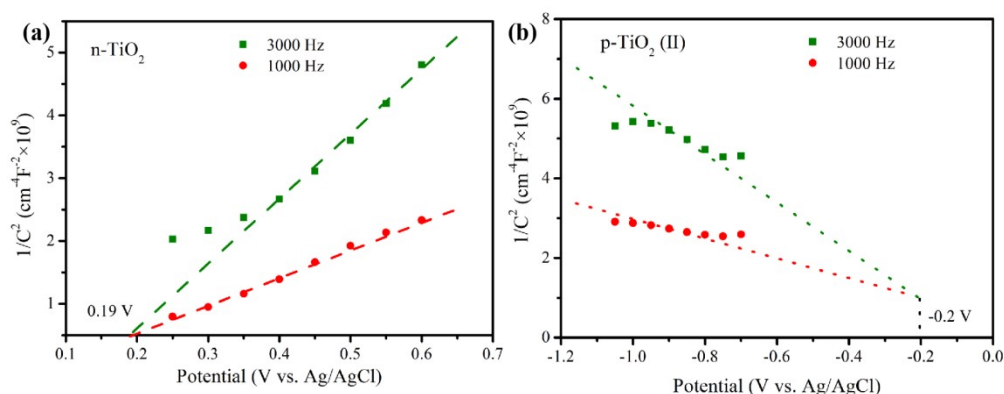


Fig. S9. Mott-Schottky plots of (a) n-TiO₂ and (b) p-TiO₂ (II).

Table S1. The result for photocatalytic N₂ fixation in previous literature about other photocatalysts and our work.

Catalyst	Reaction medium	Scavenger	Light Source	Ammonia generation rate	Reference
V _{Ti} -p-TiO ₂	H ₂ O (l)	No	300 W Xenon lamp, Full Spectrum	64.8 μmol·g ⁻¹ ·h ⁻¹	This work
Fe-doped TiO ₂	H ₂ O (g) 40 °C	No	360 W Hg-Arc Lamp, Full Spectrum	11.5 μmol·g ⁻¹ ·h ⁻¹	S1
Ru/TiO ₂	H ₂ O (l)	No	150 W Xe arc lamp, Full Spectrum	29.4 μmol·g ⁻¹ ·h ⁻¹	S2
Ru/TiO ₂	H ₂ O (l), alkaline	Ascorbic acid	250 W Xenon lamp, Full Spectrum	13.6 μmol·g ⁻¹ ·h ⁻¹	S3
Cu-doped TiO ₂	H ₂ O (l)	No	300 W Xenon lamp,	78.9 μmol·g ⁻¹ ·h ⁻¹	S4

			Full spectrum		
			300 W		
CN-OvTiO ₂	H ₂ O (l)	No	Xenon lamp, Full spectrum	48.7 μmol·g ⁻¹ ·h ⁻¹	S5
			300 W		
TiO ₂ nanotube	H ₂ O (l) Methanol	Methanol	Xenon lamp, Full spectrum	318 μmol·g ⁻¹ ·h ⁻¹	S6
			300 W		
N-doping TiO ₂	H ₂ O (l)	No	Xenon lamp, λ > 400 nm	80.09 μmol·g ⁻¹ ·h ⁻¹	S7
			300 W		
Ov-TiO ₂	H ₂ O (l) Methanol	Methanol	Xenon lamp, Full spectrum	324.8 μmol·g ⁻¹ ·h ⁻¹	S8

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