

# Piezoelectrically Enhanced Photocatalysis of $K_xNa_{1-x}NbO_3$ (KNN) Microstructures for Efficient Water Purification

Runjiang Guo<sup>†a</sup>, Mengqian Liu<sup>†a</sup>, Yurui Xing<sup>a</sup>, Tanglong Bai<sup>a</sup>, Chenglong Zhao<sup>a</sup>, Haolin Huang<sup>a</sup>,  
and Hongti Zhang<sup>\*ab</sup>

<sup>a</sup>School of Physical Science and Technology, ShanghaiTech University, Shanghai, 201210, P.R. China;

<sup>b</sup>Shanghai Key Laboratory of High-resolution Electron Microscopy, ShanghaiTech University, Shanghai, 201210, P. R. China

\* Correspondence: [zhanght3@shanghaitech.edu.cn](mailto:zhanght3@shanghaitech.edu.cn)

† These authors contributed equally to this work.

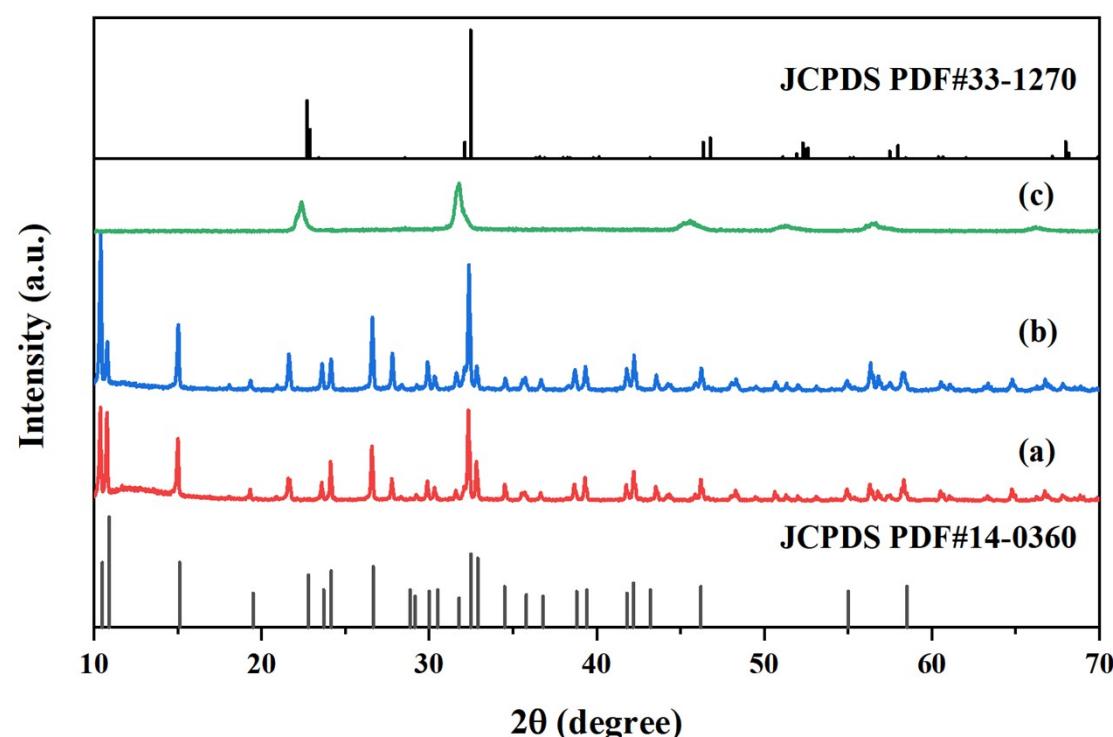


Fig.S1 XRD patterns of KNN powders synthesized under different temperatures: (a) 160 °C, (b) 180 °C, (c) 200 °C.

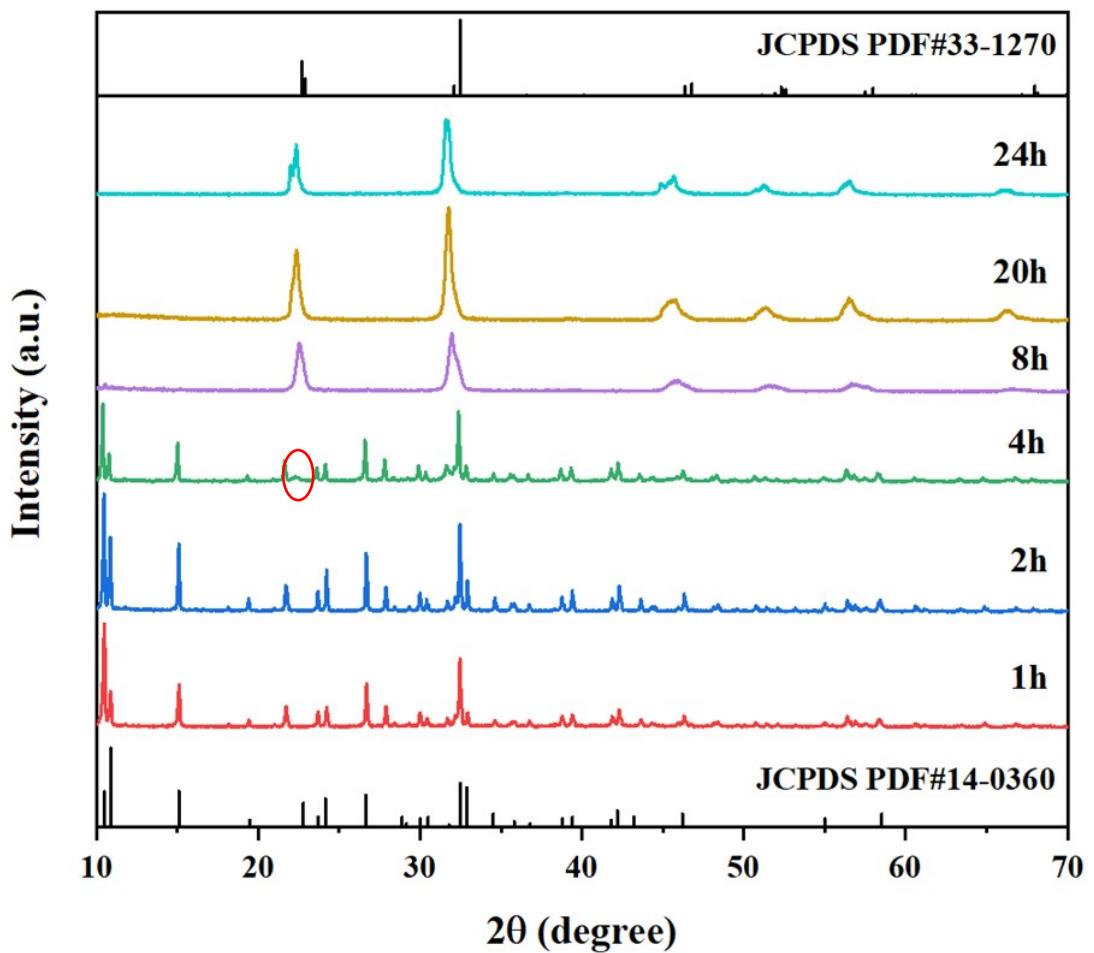


Fig.S2 The XRD patterns of KNN-6 powders synthesized under different reaction time.

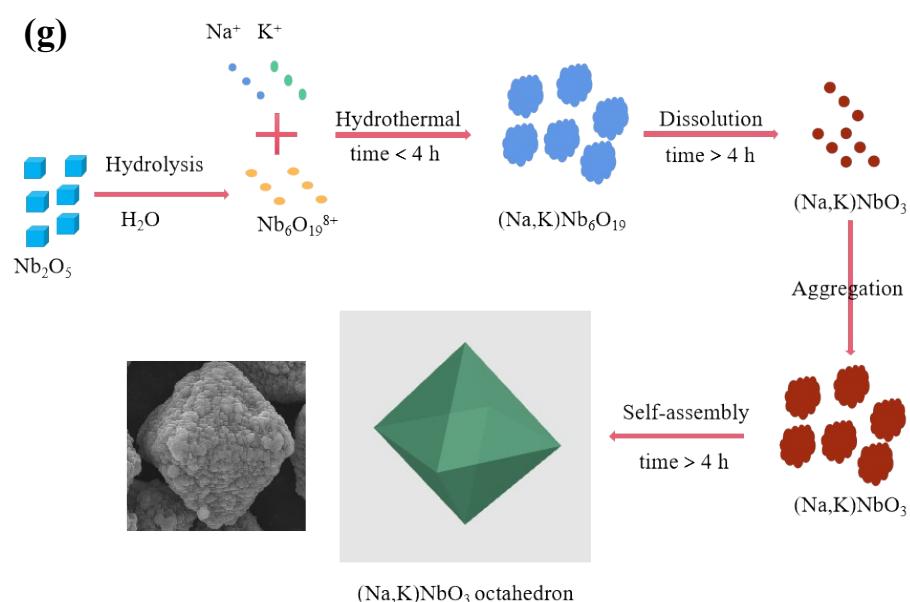
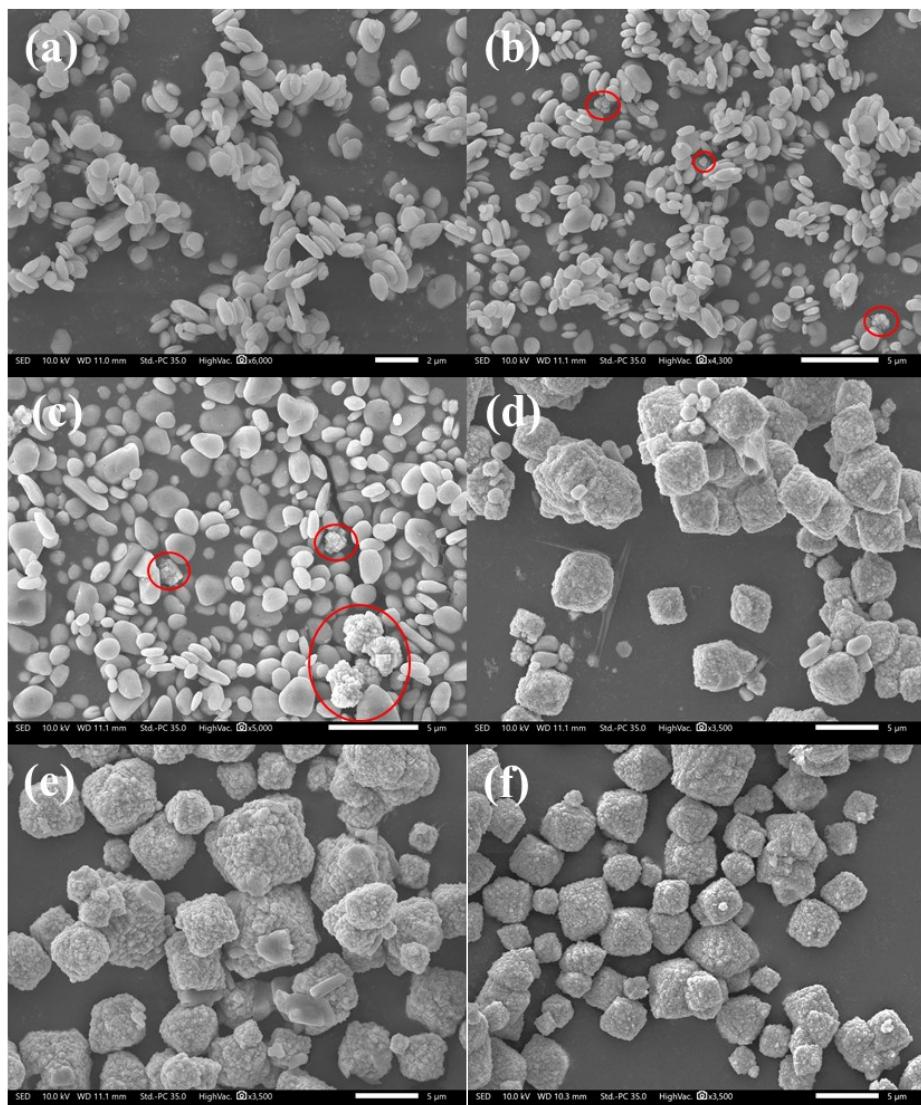


Fig.S3 The SEM images of KNN powders synthesized under different time (a) 1 h, (b) 2 h, (c) 4 h, (d) 8 h, (e) 20h, (f) 24 h; (g) schematic illustration of the KNN-6 formation mechanism.

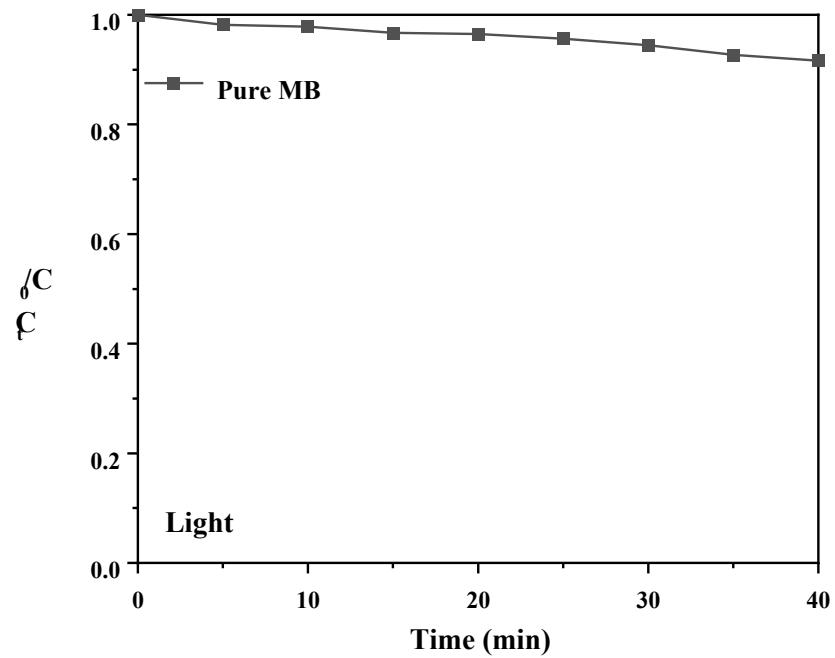


Fig.S4 The degradation of MB under light irradiation without adding any catalyst.

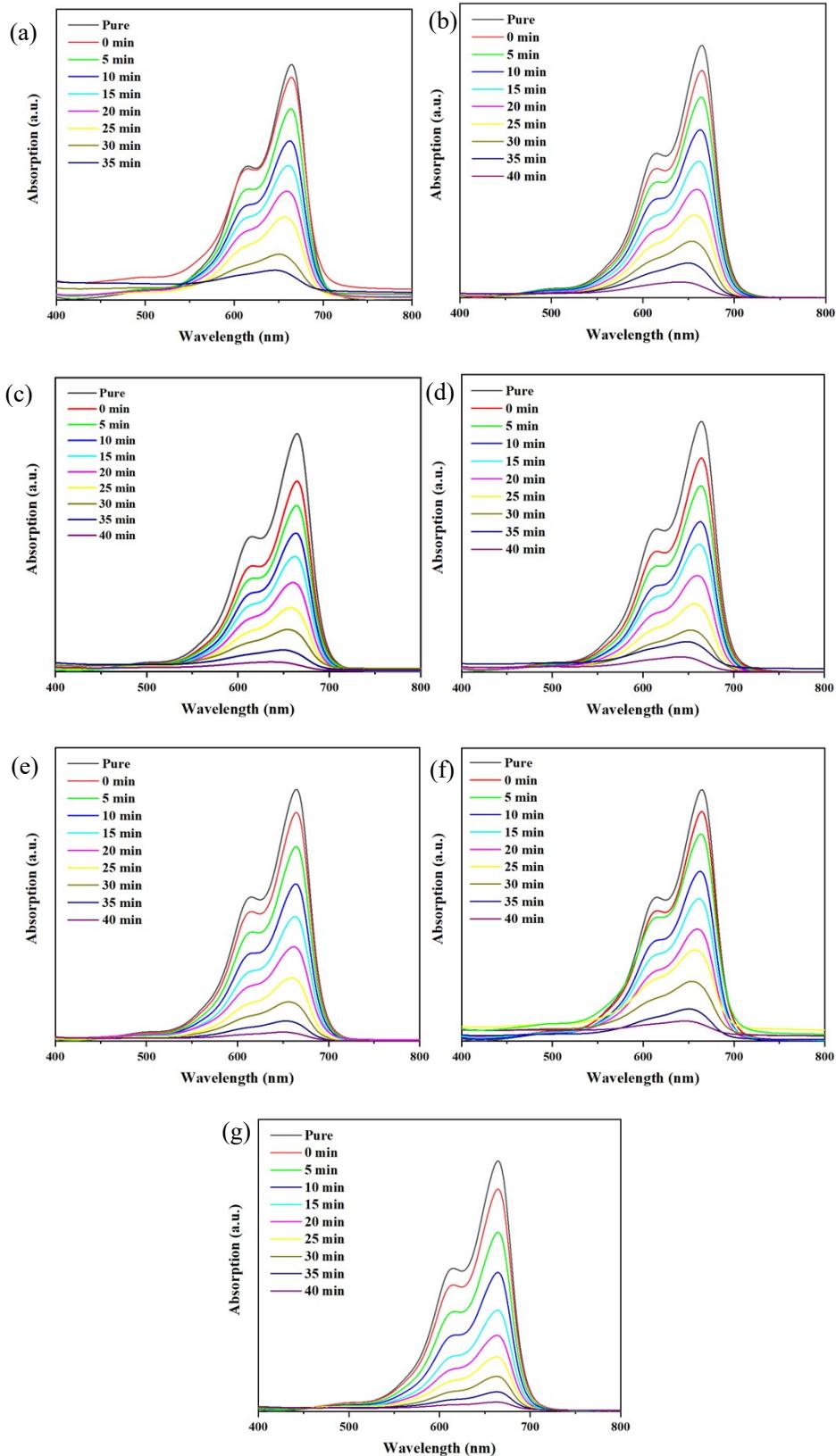


Fig.S5 The UV-vis absorption curves of MB solutions degraded by different KNN samples under condition of light irradiation together with ultrasonic vibration. (a) KNN-0, (b) KNN-5, (c) KNN-6, (d) KNN-7, (e) KNN-8, (f) KNN-9 and (g) KNN-10.

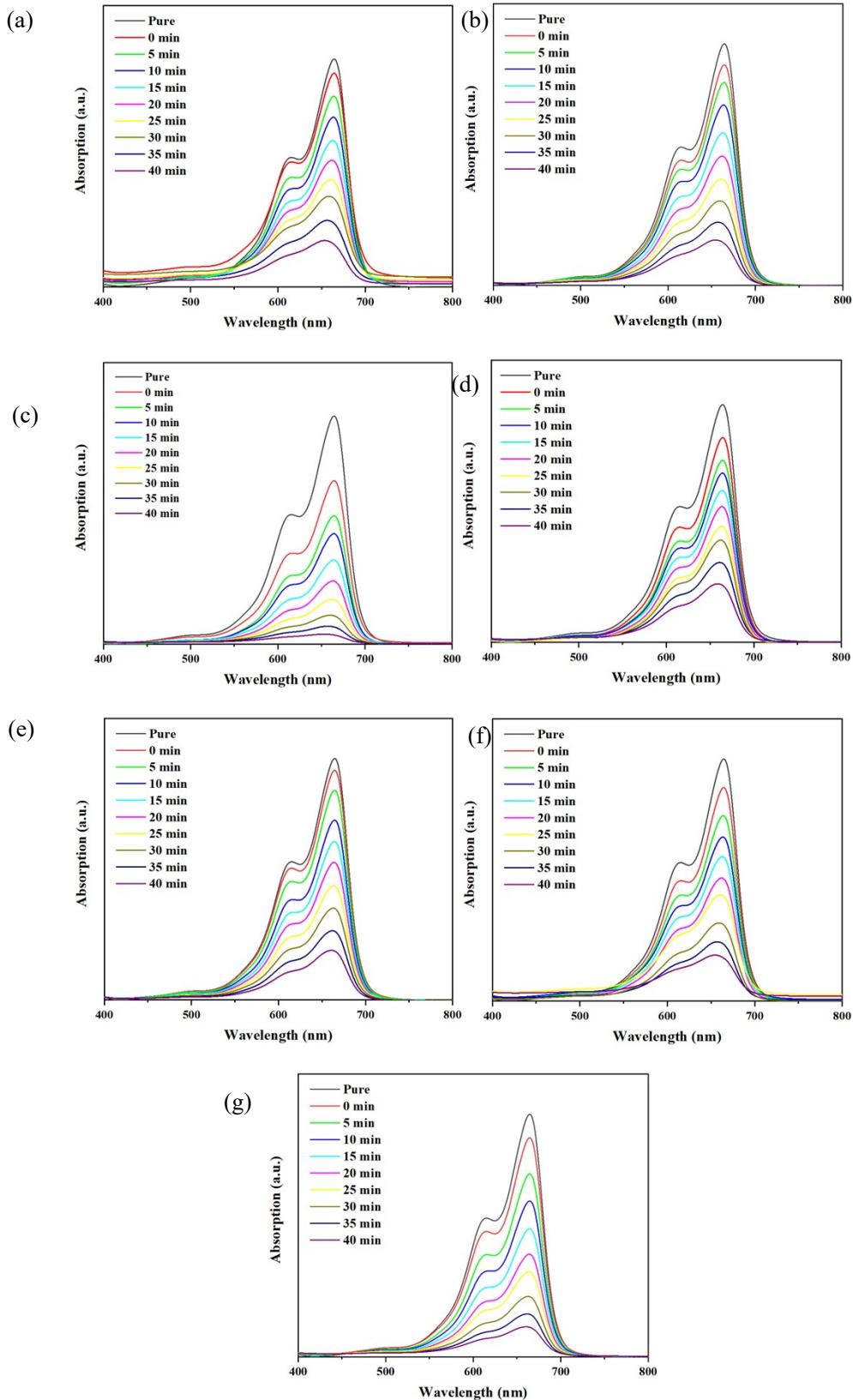


Fig.S6 The UV-vis absorption curves of MB solutions degraded by different KNN samples under condition of light irradiation. (a) KNN-0, (b) KNN-5, (c) KNN-6, (d) KNN-7, (e) KNN-8, (f) KNN-9 and (g) KNN-10.

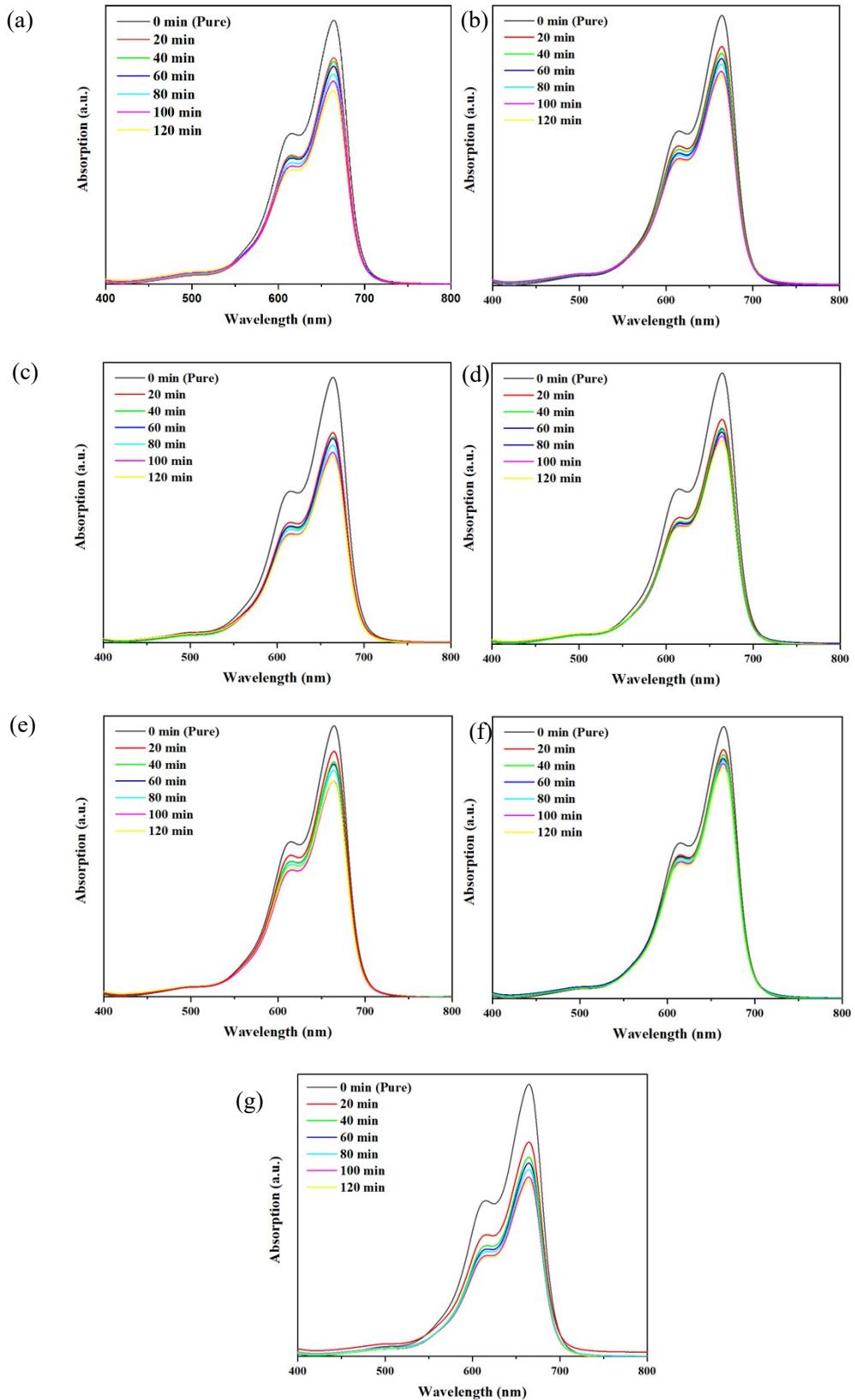


Fig.S7 The UV-vis absorption curves of MB solutions degraded by different KNN samples under condition of ultrasonic vibration. (a) KNN-0, (b) KNN-5, (c) KNN-6, (d) KNN-7, (e) KNN-8, (f) KNN-9 and (g) KNN-10.

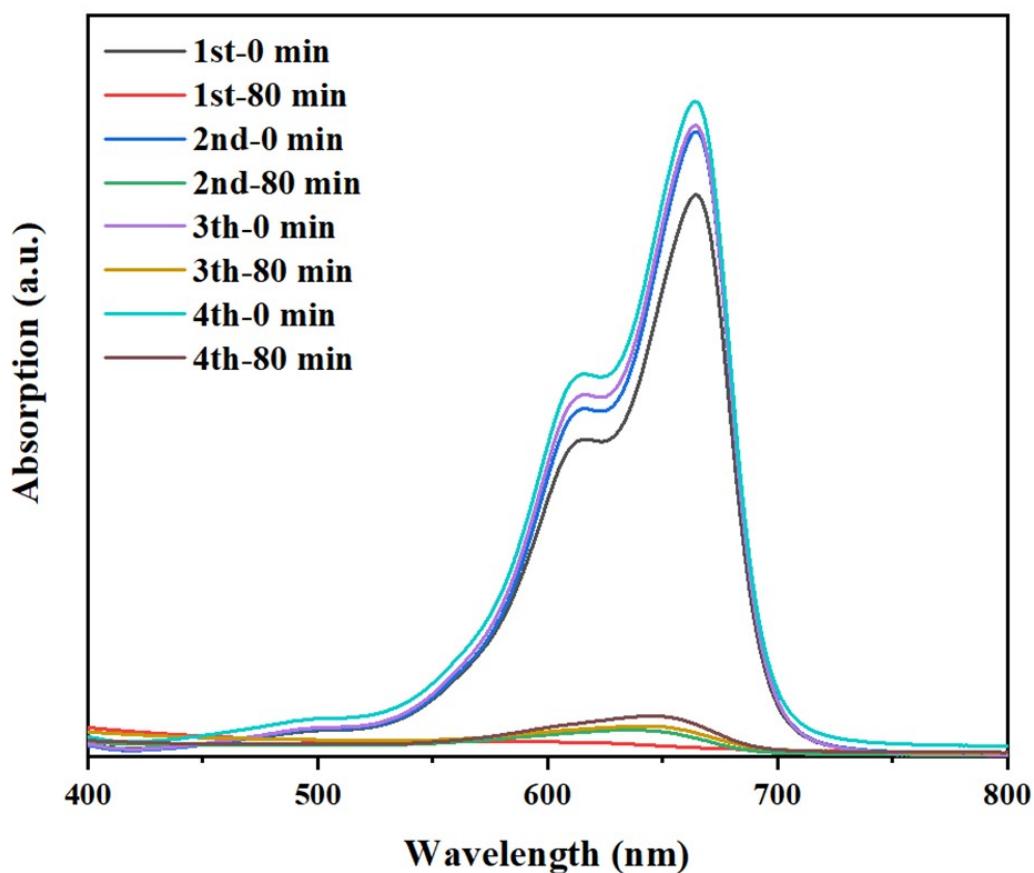


Fig.S8 The UV-vis absorption curves of MB solutions under four cycles of stability tests of the KNN-6 powder.

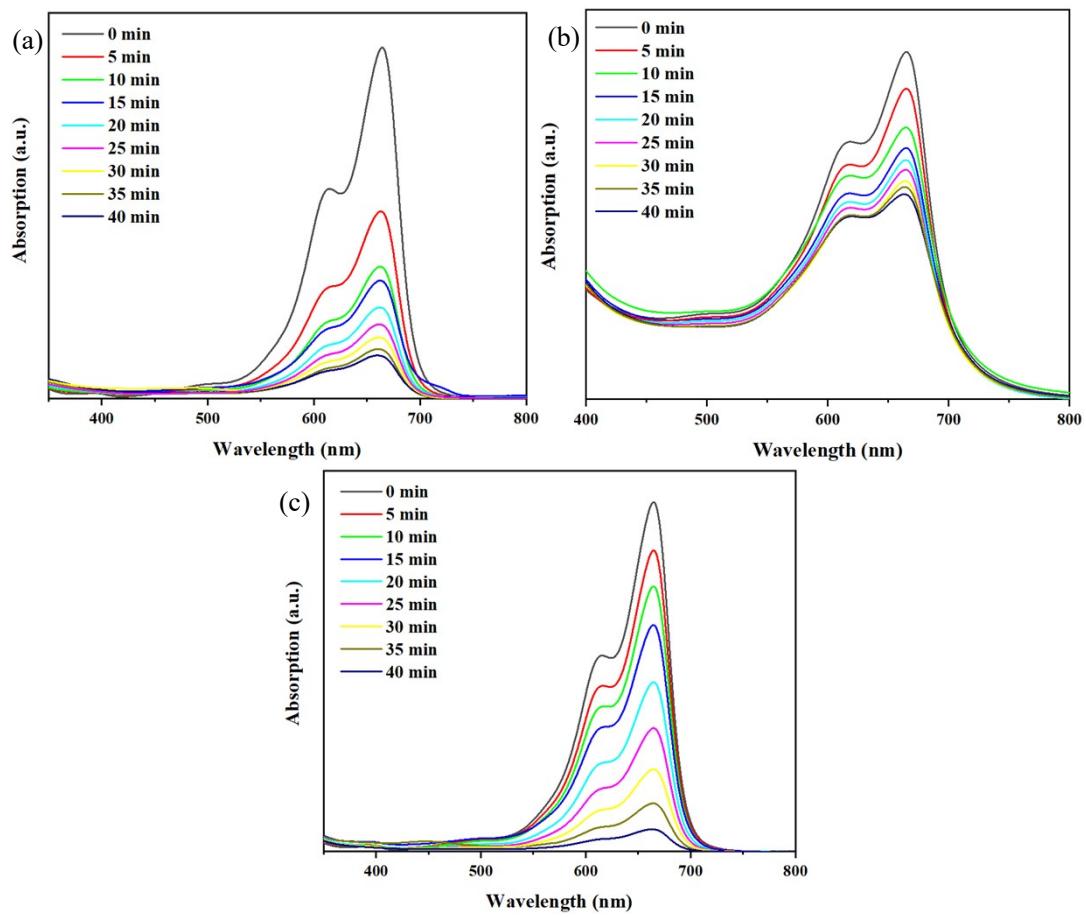


Fig.S9 The degradation efficiency of MB under different scavenger (a) EDTA-2Na. (b) BQ, (c) TBA.

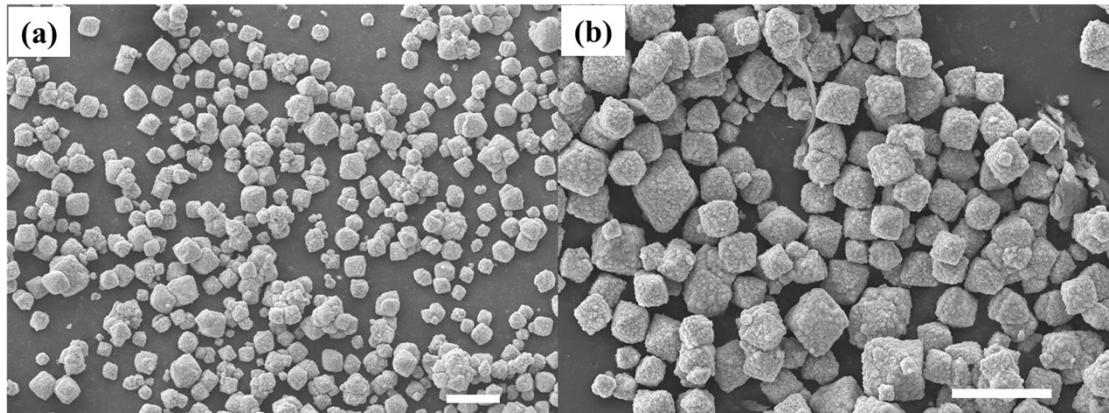


Fig.S10 SEM images of as-synthesized KNN-6 with low magnifications to show their overall octahedron morphologies. (scale bar: 10  $\mu\text{m}$ )

Tab.S1 The MB degradation efficiency in previous reports.

<b>Photocatalysts</b>	<b>Degradation Efficiency</b>	<b>Reaction conditions</b>	<b>Ref.</b>
Polymer/TiO <sub>2</sub> Nanofiber	180min 67%	Sun Light	S1
NaNbO <sub>3</sub>	180min 73%	Sun Light + Ultrasonic Vibrations	S2
CNTs/TiO <sub>2</sub> /AgNPs/Surfactant	120min 99%	Visible Light	S3
ZnO-Yb <sub>2</sub> O <sub>3</sub> -Pr <sub>2</sub> O <sub>3</sub>	60min 99.8%	Sun Light	S4
ZnO NPs-PWAC	60min 99%	Sun Light	S5
Ag-NaNbO <sub>3</sub>	180min 82%	SunLight+Ultrasonic Vibrations	S6
(Bi <sub>1/2</sub> Na <sub>1/2</sub> )TiO <sub>3</sub>	150min 54.2%	Ultrasonic Vibrations	S7
BiVO <sub>4</sub> -ZIF 8	130min 80%	Visible Light	S8
K <sub>0.4</sub> Na <sub>0.6</sub> NbO <sub>3</sub>	40min 98%	Sun Light+Ultrasonic Vibrations	This work

Tab.S2 The catalytic degradation efficiency of other niobate materials reported in works.

<b>Photocatalysts</b>	<b>Degraded Dye</b>	<b>Degradation Efficiency</b>	<b>Reaction conditions</b>	<b>Ref.</b>
NaNbO <sub>3</sub>	RhB	120 min 80%	Sunlight	S9
NaNbO <sub>3</sub>	MB	180 min 80%	Sun Light + Ultrasonic Vibrations	S2
Ag-doped NaNbO <sub>3</sub>	MB	180 min 90%	Sun Light + Ultrasonic Vibrations	S6
NaNbO <sub>3</sub>	RhB	80 min 90%	Sun Light + Ultrasonic Vibrations+heating/cooling	S10
KNbO <sub>3</sub>	RhB	180 min 96%	Sunlight	S11
N-doped KNbO <sub>3</sub>	RhB	18h 64%	Visible Light	S12
KNbO <sub>3</sub>	RhB	120 min 90%	Sun Light + Ultrasonic Vibrations	S13
K <sub>0.5</sub> Na <sub>0.5</sub> NbO <sub>3</sub>	BB41	90 min 90%	Sunlight	S14
K <sub>0.5</sub> Na <sub>0.5</sub> NbO <sub>3</sub>	RhB	100 min 90%	Sun Light + Ultrasonic Vibrations	S15
K <sub>0.4</sub> Na <sub>0.6</sub> NbO <sub>3</sub>	MB	40min 98%	Sun Light+Ultrasonic Vibrations	This work

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