

Supporting Information

Computational Studies on the Possible Formation of Glycine via Open Shell Gas-Phase Chemistry in the Interstellar Medium

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In this work, We do not propose a three/four body collision. Instead, We are proposing a sequence of two body collisions as illustrated below with an example:

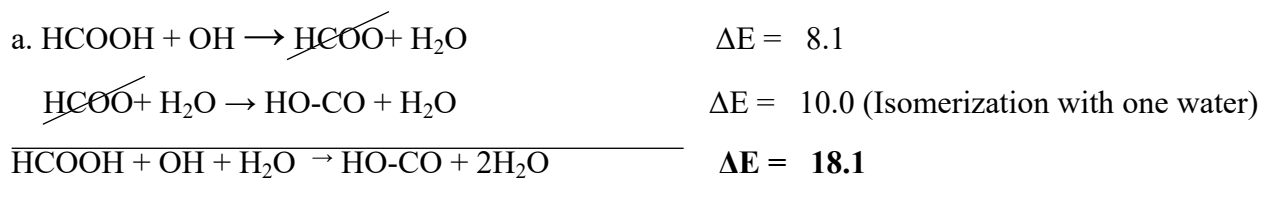
We deal with three bodies as part of this work. Yet, clearly three body collisions are highly improbable in low density gas-phase interstellar chemistry. In this context, we state explicitly that, we do not propose that the pre-reactant complexes are formed via a three-body collision. Instead, they are formed in a sequential order involving two body collisions. The exact sequence is insignificant because all of them lead to the same pre-reactant complex, and no barrier is encountered in the process as shown below. Once this pre-reactant complex is formed, further molecular reorganization occurring lead to the TS, which results in the product complexes. However, regardless of which sequence of two-body collision happen, the pre-reactant complex is always stabilized significantly.

We demonstrate details of the energetics for all possible sequences of two-body collisions resulting in the same stabilized pre-reactant complex. The same level of theory used in the figures corresponding to Path 3 (Figures 9 and 11 in the main text) are used here too. All energies are provided in kcal/mol.

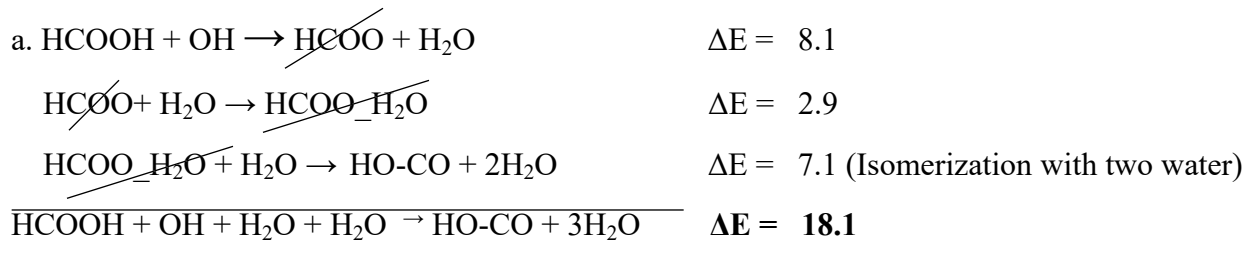
Points to note are that: the barrierless formation of the pre-reactant complex is always exothermic via this sequential 2-body reaction, and the sequential 2-body reaction paves the way for what looks like a three/four body reaction.

Such a sequential arrangement of 2 body reactions effectively leading to a 3 (or even 4 body in some cases) has been espoused previously by our research group (and others) in the context of metal-ion mediated interstellar chemistry. A prominent reference includes (kindly see the SI of this paper too): Thripati and Ramabhadran, *J. Phys. Chem. A*. 2021, 125, 3457-3472.

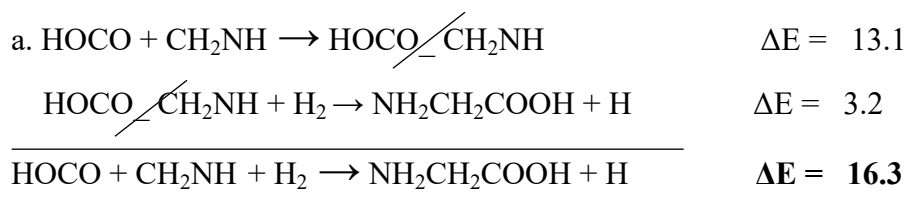
1a. $HCOOH + OH + H_2O \rightarrow HO-CO + 2H_2O$

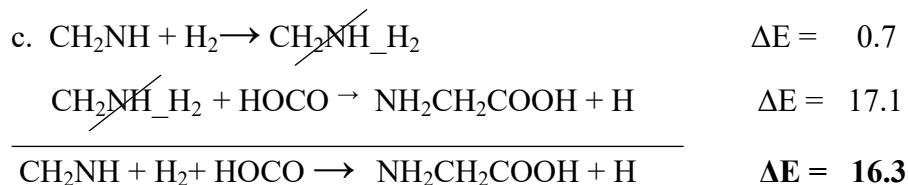
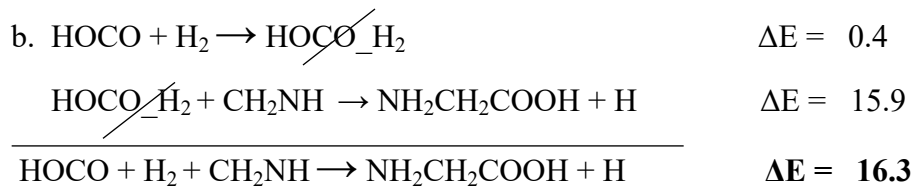


1b. $HCOOH + OH + H_2O + H_2O \rightarrow HO-CO + 3H_2O$

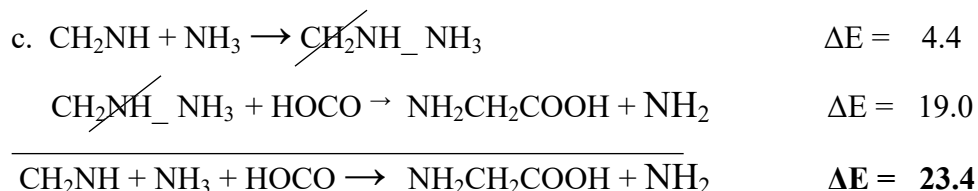
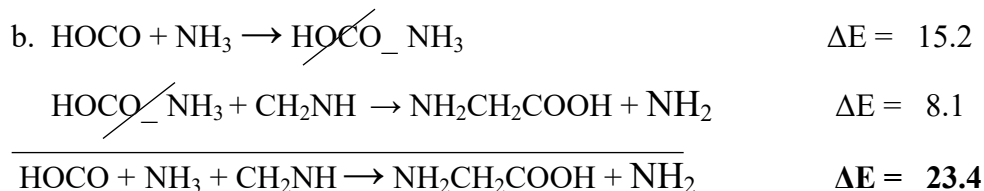
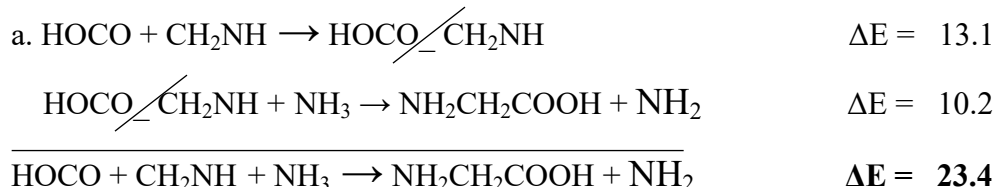


2. $HOCO$ radical + $CH_2NH + H_2 \rightarrow NH_2CH_2COOH + H$ radical





3. *HOCO radical* + *CH₂NH* + *NH₃* → *NH₂CH₂COOH* + *NH₂ radical*



Isomerization of HCOO/HO-CO via one-water molecule

Isomerization of HCOO/HO-CO via one-water molecule

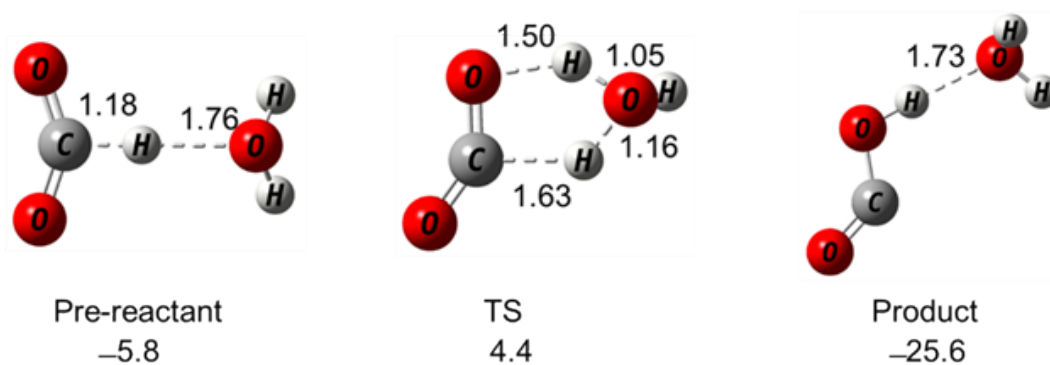


Figure SI_1: This a valid alternative mechanism for Figure 7(iii), reaction with one water molecule but the energy barrier is not nil.

Energy barrier calculations method used in this work:

Table_SI_1: Energy barrier = Transition State – (Infinity Separated Reactants)

The following method has been used for the main text.

Transition States	Infinity Separated Reactants
COCH ₂ NH ₂ _H ₂ O_ TS-III (Figure 3)	COCH ₂ NH ₂ and H ₂ O
CH ₂ COOH_ NH ₃ _ TS-III (Figure 6)	CH ₂ COOH and NH ₃
NHCH ₂ COOH_ H ₂ _ TS-V (Figure 11)	NHCH ₂ COOH and H ₂
NHCH ₂ COOH_ NH ₃ _ TS-VI (Figure 11)	NHCH ₂ COOH and NH ₃
NH ₂ CHCOOH_ H ₂ _ VII (Figure 12)	NH ₂ CHCOOH and H ₂
NH ₂ CHCOOH_ NH ₃ _ VIII (Figure 12)	NH ₂ CHCOOH and NH ₃
HCO_ CH ₂ NH_ TS-I	HCO and CH ₂ NH
NHCH ₂ CHO_ NH ₂ CH ₂ CO_ TS-II	HCO and CH ₂ NH
NHCH ₂ CHO_ NH ₂ CH ₂ CO_ PR_ TS	HCO, CH ₂ NH and 2H ₂ O

CH ₃ _CO ₂ _TS-I	CH ₃ and CO ₂
CH ₃ COO _CH ₂ COOH _TS-II	CH ₃ and CO ₂
CH ₃ COO _CH ₂ COOH _PR _TS	CH ₃ and CO ₂ and 2H ₂ O
HCOOH _OH _TS-I	HCOOH and OH
HCOO _HOCO _TS-II	HCOO
HCOO _HOCO-PR-TS-I	HCOO, 2H ₂ O
HOCO _CH ₂ NH _TS-III	HOCO and CH ₂ NH
NHCH ₂ COOH _NH ₂ CHCOOH _TS-IV	HOCO and CH ₂ NH
NHCH ₂ COOH _NH ₂ CHCOOH- PR-TS-II	HOCO and CH ₂ NH, and 2H ₂ O

Table _SI _2: Energy barrier = Transition State - Infinity Separated Reactants

The following method used for the supporting information.

Transition States	Infinity Separated Reactants
COCH ₂ NH ₂ _H ₂ O _TS-III	HCO and CH ₂ NH and H ₂ O
CH ₂ COOH _NH ₃ _TS-III	CH ₃ and CO ₂ and NH ₃
NHCH ₂ COOH _H ₂ _TS-V	HOCO and CH ₂ NH and H ₂
NHCH ₂ COOH _NH ₃ _TS-VI	HOCO and CH ₂ NH and NH ₃
NH ₂ CHCOOH _H ₂ _VII	HOCO and CH ₂ NH and H ₂
NH ₂ CHCOOH _NH ₃ _VIII	HOCO and CH ₂ NH and NH ₃

Comparing the energy barriers with different DFT methods on representative TSs.

To validate our computational technique for obtaining the energy barriers, we have carried out single-point-energy calculations on representative TSs, and the energy values are shown in Table 1. This is in addition to the validation of the TS geometries themselves (presented in the SI). Overall, we observe, the same qualitative trend (UB2-PLYP(D3-BJ), UM06-2x, and U ω B97X-D) in the electronic energy barrier heights was noticed with respect to the B3-LYP(D3-BJ) method, and the energy values can be seen in Table 1. The analysis of energy values in Table 1 clearly indicates that our method (B3-LYP(D3-BJ)) is acceptable with respect to the other three DFT methods used in the paper.

Table SI_3. Comparing the energy barriers with different DFT methods on representative TSs at the UCCSD(T)(FC)/aug-cc-pVTZ//UB3-LYP(D3-BJ)/aug-cc-pVTZ, UCCSD(T)(FC)/aug-cc-pVTZ//B2-PLYP(D3-BJ)/aug-cc-pVTZ, UCCSD(T)(FC)/aug-cc-pVTZ//UM06-2x/aug-cc-pVTZ, and UCCSD(T)(FC)/aug-cc-pVTZ//U ω B97X-D/aug-cc-pVTZ levels of theories. Where $\Delta E^\ddagger = E_{TS} - E_{\text{Infinity separated reactants}}$. The energy values (included zero-point corrections) are provided in kcal/mol.

	B3-LYP $\Delta E^\ddagger$	B2-PLYP $\Delta E^\ddagger$	M06-2x $\Delta E^\ddagger$	ω B97X-D $\Delta E^\ddagger$
Glycine Precursor 1				
HCO ₂ CH ₂ NH ₂ _TS-I	7.5	7.7	7.7	7.4
NHCH ₂ CHO_NH ₂ CH ₂ CO_TS-II	22.5	22.6	22.4	22.6
NHCH ₂ CHO_NH ₂ CH ₂ CO_PR_TS	1.3	1.3	0.59	1.3
COCH ₂ NH ₂ _H ₂ O_TS-III	16.0	16.1	16.3	16.2
Glycine Precursor 2				
CH ₃ _CO ₂ _TS-I	21.2	20.6	20.6	21.4
CH ₃ COO ₂ CH ₂ COOH_TS-II	46.5	45.8	46.1	46.9
CH ₃ COO ₂ CH ₂ COOH_PR_TS	35.2	35.2	34.9	35.6
CH ₂ COOH_NH ₃ _TS-III	61.9	61.4	61.5	62.7
Glycine Path-3				
HCOOH_OH_TS-I	1.5	1.4	1.4	1.6
HCOO ₂ _HOCO_TS-II	20.6	18.9	18.6	18.8
HCOO ₂ _HOCO-PR-TS-I	11.2	12.7	13.0	12.9

HOCO_CH ₂ NH_TS-III	0.14	0.52	0.35	0.42
NHCH ₂ COOH_NH ₂ CHCOOH_TS-IV	9.0	9.3	9.3	9.3
NHCH ₂ COOH_NH ₂ CHCOOH- PR-TS-II	12.8	12.4	12.2	12.6
NHCH ₂ COOH_H ₂ _TS-V	19.7	19.4	19.7	19.4
NHCH ₂ COOH_NH ₃ _TS-VI	9.7	9.3	9.3	9.3
NH ₂ CHCOOH_H ₂ _VII	24.7	25.0	25.3	24.7
NH ₂ CHCOOH_NH ₃ _VIII	11.7	12.0	12.2	12.0

Effect of temperature on the transition states energy barriers:

To check if there is a prominent temperature effect on TSs, we have performed calculations at different temperatures, using the UB3-LYP(D3-BJ)/aug-cc-pVTZ method to optimize. The thermal corrections have been taken from optimized TSs geometries with various temperature conditions and added to the electronic energy values. The TSs energy barriers are shown in Table 2 with various temperatures.

Upon glancing at Table 2, it becomes clear that for all the pathways, the temperature effect is minimal. What is especially satisfying is that pathway 3 leading to glycine is feasible in the ISM, regardless of the temperature range. With regards to the first pathway, a barrier of ~ 7 kcal/mol is involved in the first step. We reckon this should be feasible only at temperatures higher than 200 K. The second pathway is definitely not feasible for the generation of interstellar glycine. Moreover, once glycine is formed, we also reckon that alanine and serine should be readily formed at all the temperature ranges studied.

Table_SI_4: Relative electronic energies computed at $T = 20\text{K}, 50\text{K}, 100\text{K}, 200\text{K}, 298.15\text{K}, 400\text{K},$ and 500K ($\Delta_{\text{rel}}E_{20}$, $\Delta_{\text{rel}}E_{50}$, $\Delta_{\text{rel}}E_{100}$, $\Delta_{\text{rel}}E_{200}$, $\Delta_{\text{rel}}E_{298.15}$, $\Delta_{\text{rel}}E_{400}$, and $\Delta_{\text{rel}}E_{500}$, respectively) for the TSs shown in the energy profile of Figure 1-15. The UB3-LYP(D3-BJ)/aug-cc-pVTZ level of theory used. Where $\Delta E^\ddagger = E_{\text{TS}} - E_{\text{Infinity separated reactants}}$.

	$\Delta E^\ddagger$						
	20K	50K	100K	200K	298.15K	400K	500K
Glycine Precursor 1							

HCO ₂ CH ₂ NH ₂ _TS-I	7.4	7.2	7.1	6.9	7.0	7.3	7.6
NHCH ₂ CHO_NH ₂ CH ₂ CO_TS-II	22.4	22.2	22.0	21.7	21.6	21.7	22.0
NHCH ₂ CHO_NH ₂ CH ₂ CO_PR_TS	1.0	0.5	0.06	0.62	0.45	0.22	1.22
COCH ₂ NH ₂ _H ₂ O_TS-III	15.8	15.4	15.0	14.4	14.3	14.6	15.1
Glycine Precursor 2							
CH ₃ _CO ₂ _TS-I	21.1	20.9	20.7	20.4	19.5	20.3	20.4
CH ₃ COO ₂ _CH ₂ COOH_TS-II	46.4	46.2	45.9	45.4	44.4	45.1	45.2
CH ₃ COO ₂ _CH ₂ COOH_PR_TS	34.9	34.4	33.8	33.0	32.1	33.2	33.9
CH ₂ COOH_NH ₃ _TS-III	61.7	61.3	60.8	60.1	59.0	59.8	60.1
Glycine Path-3							
HCOOH_OH_TS-I	1.8	1.9	2.2	2.4	2.4	2.3	2.1
HCOO ₂ _HOCO_TS-II	18.2	18.2	18.2	18.3	18.3	18.4	18.3
HCOO ₂ _HOCO-PR-TS-I	13.8	14.1	14.6	15.0	15.0	14.7	14.2
HOCO_CH ₂ NH_TS-III	0.30	0.13	0.07	0.18	0.09	0.11	0.38
NHCH ₂ COOH_NH ₂ CHCOOH_TS-IV	9.2	9.0	8.8	8.5	8.6	8.8	9.2
NHCH ₂ COOH_NH ₂ CHCOOH- PR-TS-II	12.9	13.4	14.0	14.6	14.5	13.9	12.9
NHCH ₂ COOH_H ₂ _TS-V	19.6	19.9	20.3	20.8	20.9	20.7	20.3
NHCH ₂ COOH_NH ₃ _TS-VI	9.6	9.9	10.2	10.4	10.2	9.6	8.9
NH ₂ CHCOOH_H ₂ _TS_VII	24.6	24.9	25.3	25.6	25.6	25.3	24.9
NH ₂ CHCOOH_NH ₃ _TS_VIII	11.7	12.0	12.2	12.3	12.1	11.5	10.7

Comparing the TSs energy barriers in the gas-phase and implicit CPCM model.

Whereas the liquids do not exist in the ISM, it shall be an instructive idea to compare the barriers in the gas-phase so that one could make a rough comparison of the interstellar reactions vis-à-vis, how these reactions could happen terrestrially. To calculate the electronic energy barrier in the gas-phase, UB3-LYP(D3-BJ)/aug-cc-pVTZ level of theory used and in the solution phase (UB3-LYP(D3-BJ)(CPCM))/aug-cc-pVTZ level of theory used. Table 3 clearly indicates that all the paths the TSs electronic energy barriers are of similar magnitude in the gas-phase and in the

solution phase with the CPCM⁷⁴ model. Hence it is interesting that, our conclusions that we draw for interstellar reactions may be extended to terrestrial chemistry too, insofar as the implicit solvent model is reliable, noting the limitation that the continuum modeling could miss out on capturing some weak local interactions.

Table_SI_5. Comparing the energy barriers in the gas-phase and CPCM model. The TSs optimized by using UB3-LYP(D3-BJ)/aug-cc-pVTZ level of theory for the gas-phase and (UB3-LYP(D3-BJ)(CPCM))/aug-cc-pVTZ level of theory for the solution phase. Where $\Delta E^\ddagger = E_{TS} - E_{\text{Infinity separated reactants}}$, the energy values (included zero-point corrections) are provided in kcal/mol.

	Gas-Phase ($\Delta E^\ddagger$)	CPCM($\Delta E^\ddagger$)
Glycine Precursor 1		
HCO_CH ₂ NH_TS-I	4.0	4.6
NHCH ₂ CHO_NH ₂ CH ₂ CO_TS-II	20.7	21.4
NHCH ₂ CHO_NH ₂ CH ₂ CO_PR_TS	8.7	4.4
COCH ₂ NH ₂ _H ₂ O_TS-III	15.5	17.9
Glycine Precursor 2	Gas-Phase($\Delta E^\ddagger$)	CPCM($\Delta E^\ddagger$)
CH ₃ _CO ₂ _TS-I	16.1	14.0
CH ₃ COO_CH ₂ COOH_TS-II	43.6	39.7
CH ₃ COO_CH ₂ COOH_PR_TS	20.4	19.8
CH ₂ COOH_NH ₃ _TS-III	38.2	27.6
Glycine Path-3	Gas-Phase($\Delta E^\ddagger$)	CPCM($\Delta E^\ddagger$)
HCOOH_OH_TS-I	6.6	4.1
HCOO_HOCO_TS-II	21.1	22.0
HCOO_HOCO-PR-TS-I	15.3	Unable to locate TS
HOCO_CH ₂ NH_TS-III	4.8	2.3
NHCH ₂ COOH_NH ₂ CHCOOH_TS-IV	7.2	8.3
NHCH ₂ COOH_NH ₂ CHCOOH-PR-TS-II	20.0	15.1
NHCH ₂ COOH_H ₂ _TS-V	12.0	10.2
NHCH ₂ COOH_NH ₃ _TS-VI	14.3	11.5
NH ₂ CHCOOH_H ₂ _VII	15.1	13.7
NH ₂ CHCOOH_NH ₃ _VIII	16.0	13.8

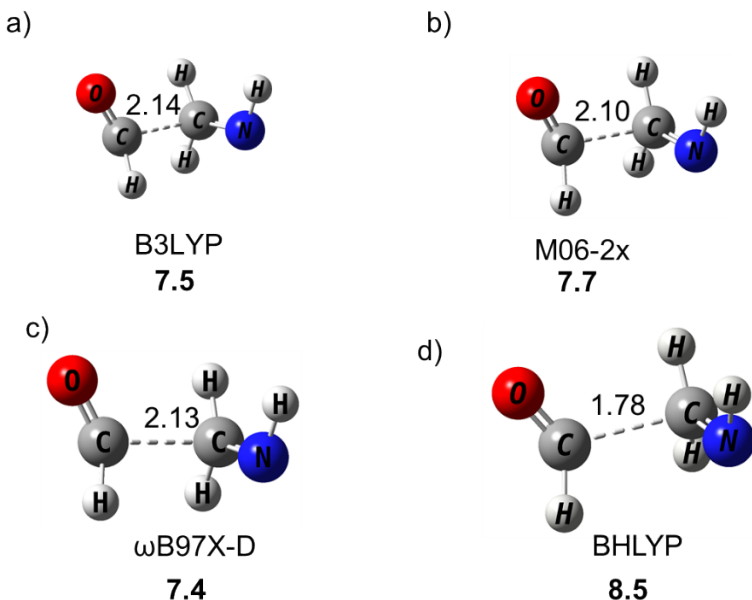


Figure SI_2: Optimized TSs of HCO/CH₂NH (UB3LYP). The geometry being calibrated with two different DFT methods. Throughout, the aug-cc-pVTZ basis set was used for the optimization. All the resulting optimized geometries are similar. The UCCSD(T)(FC)/aug-cc-pVTZ level of theory is used for the single-point energy calculation. The electronic energy barriers (including zero-point corrections) are given below the structure and the values given in kcal/mol.

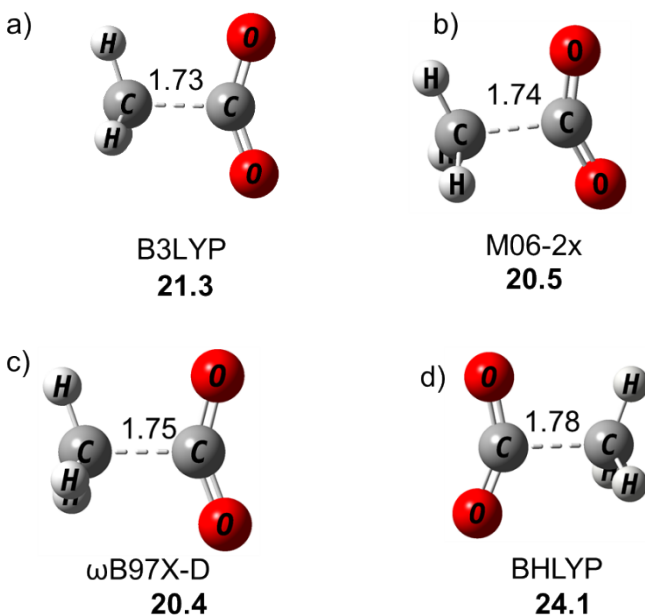


Figure SI_3: Optimized TSs of CH₃/CO₂ (UB3LYP). The geometry being calibrated with two different DFT methods. Throughout, the aug-cc-pVTZ basis set was used for the optimization. All the resulting optimized geometries are similar. The UCCSD(T)(FC)/aug-cc-pVTZ level of

theory is used for the single-point energy calculation. The electronic energy barriers (including zero-point corrections) are given below the structure and the values given in kcal/mol.

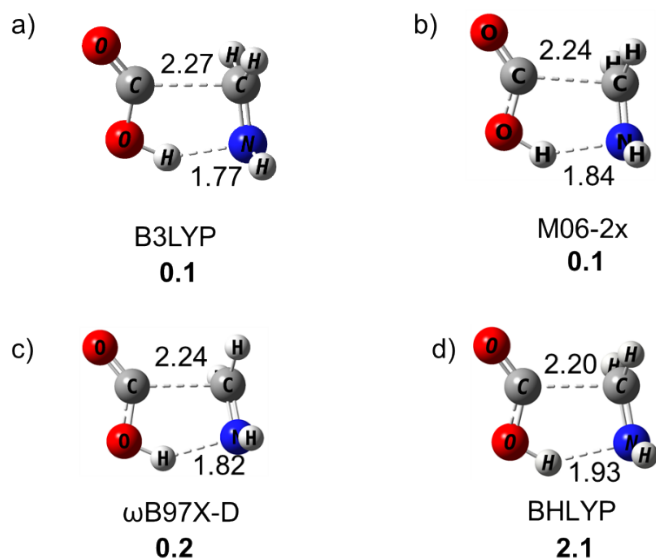


Figure SI_4: Optimized TSs of COOH radical/CH₂NH (UB3LYP). The geometry being calibrated with two different DFT methods. Throughout, the aug-cc-pVTZ basis set was used for the optimization. All the resulting optimized geometries are similar. The UCCSD(T)(FC)/aug-cc-pVTZ level of theory is used for the single point energy calculation. The electronic energy barriers (including zero-point corrections) are given below the structure and the values given in kcal/mol.

Table_SI_6. The computed TSs electronic energies, Gibbs free energies and enthalpies.

	$\Delta E + ZPE$	$\Delta E + \Delta G$	$\Delta E + \Delta H$
Glycine Precursor 1			
HCO-CH ₂ NH-TS-I	7.5	17.1	6.5
NHCH ₂ CHO-NH ₂ CH ₂ CO-TS-II	22.5	32.8	21.0
NHCH ₂ CHO-NH ₂ CH ₂ CO-PR-TS	13.4	29.6	- 2.2
COCH ₂ NH ₂ -H ₂ O-TS-III	16.0	35.9	13.1
Glycine Precursor 2			
CH ₃ -CO ₂ -TS-I	21.2	25.5	19.7
CH ₃ COO-CH ₂ COOH-TS-II	46.5	51.8	44.6
CH ₃ COO-CH ₂ COOH-PR-TS	19.2	58.4	31.1
CH ₂ COOH-NH ₃ -TS-III	61.9	76.9	58.6

Glycine Path-3	$\Delta E + ZPE$	$\Delta E + \Delta G$	$\Delta E + \Delta H$
HCOOH OH TS-I	- 1.5	7.0	- 2.8
HCOO HOCO TS-II	20.6	18.4	18.3
HCOO HOCO-PR-TS-I	- 11.2	4.6	- 16.2
HOCO CH ₂ NH TS-III	0.1	10.9	- 0.9
NHCH ₂ COOH NH ₂ CHCOOH TS-IV	9.0	20.1	7.7
NHCH ₂ COOH NH ₂ CHCOOH- PR-TS-II	- 12.8	16.8	- 16.6
NHCH ₂ COOH H ₂ TS-V	- 19.7	- 2.4	- 22.3
NHCH ₂ COOH NH ₃ TS-VI	- 9.7	9.8	- 11.7
NH ₂ CHCOOH H ₂ VII	- 24.7	- 7.4	- 27.1
NH ₂ CHCOOH NH ₃ VIII	- 11.7	7.3	- 13.5

Table_SI_7. TSs energy barriers without ZPE and ZPE

	Without ZPE	With ZPE
Glycine Precursor 1		
HCO CH ₂ NH Pre-reactant	3.8	2.6
HCO CH ₂ NH TS-I	5.3	7.5
HCOCH ₂ NH Int-I	15.4	12.1
NHCH ₂ CHO NH ₂ CH ₂ CO TS-II	21.0	22.5
COCH ₂ NH ₂ Int-II	26.9	21.9
COCH ₂ NH ₂ H ₂ O Int-III	33.9	26.8
COCH ₂ NH ₂ H ₂ O TS-III	9.9	16.0
Product (O-Protonated Glycine)	35.8	25.9
NHCH ₂ CHO NH ₂ CH ₂ CO PR Reactant	30.9	23.0
NHCH ₂ CHO NH ₂ CH ₂ CO PR TS	2.6	1.3
COCH ₂ NH ₂ PR Product	45.0	35.6
Glycine Precursor 2		
CH ₃ CO ₂ Pre-reactant	1.1	0.2
CH ₃ CO ₂ TS-I	17.7	21.2
CH ₃ CO ₂ Int-I	11.2	15.9
CH ₃ COO CH ₂ COOH TS-II	44.8	46.5
CH ₂ COOH Int-II	11.0	15.2
CH ₂ COOH NH ₃ Int-III	5.0	1.7
CH ₂ COOH NH ₃ TS-III	53.6	61.9
Product (O-Protonated Glycine)	9.8	19.5
CH ₃ COO CH ₂ COOH Pre-Reactant	12.3	3.6

CH ₃ COO_CH ₂ COOH_PR_TS	30.5	35.2
CH ₃ COO_CH ₂ COOH_PR_Product	13.5	4.1

Table_SI_8. TSs energy barriers without ZPE and ZPE

	Without ZPE	With ZPE
Glycine Path-3		
HCOOH_OH Pre-Reactant	10.4	8.2
HCOOH_OH TS-I	1.6	1.5
HCOO_H ₂ O_Int-I	10.9	9.0
HCOO_TS-II	21.5	0.9
HOCO_Int-III	16.5	3.1
HCOO_PR-PR-I	18.1	12.0
HCOO_PR_TS-I	15.7	11.2
HOCO_PR Product-I	35.7	28.0
HOCO_CH ₂ NH_Int-IV	12.8	11.2

HOCO_CH ₂ NH TS-III	1.4	0.14
NHCH ₂ COOH Int-V	24.1	21.5
NHCH ₂ COOH_NH ₂ CHCOOH TS-IV	8.4	9.0
NH ₂ CHCOOH Int-VI	44.8	41.2
NHCH ₂ COOH PR_Reactant-II	59.1	52.3
NHCH ₂ COOH_NH ₂ CHCOOH PR_TS-II	16.7	12.8
NH ₂ CHCOOH_PR_Pd-II	40.9	33.2
NHCH ₂ COOH_H ₂ Int-VII	37.7	34.1
NHCH ₂ COOH_H ₂ TS-V	23.6	19.7
NH ₂ CH ₂ COOH_H Int-VIII	35.8	29.9
NHCH ₂ COOH_NH ₃ Int-IX	29.3	25.6
NHCH ₂ COOH_NH ₃ TS-VI	11.4	9.7
NH ₂ CH ₂ COOH_NH ₂ Int-X	23.9	20.1
NH ₂ CHCOOH_H ₂ Int-XI	57.5	53.1
NH ₂ CHCOOH_H ₂ TS-VII	29.2	24.7
NH ₂ CH ₂ COOH_H Int-XII	35.8	29.9
NH ₂ CHCOOH_NH ₃ Int-XIII	51.9	46.8
NH ₂ CHCOOH_NH ₃ TS-VIII	14.2	11.7
NH ₂ CH ₂ COOH_NH ₂ Int-XIV	21.9	18.5

Table_SI_9. The imaginary frequencies of the TSs in different levels of theory in computed aug-cc-pVTZ basis-set using harmonic analysis.

	UB3-LYP	UB2-PLYP	UM06-2x	UωB97X-D
Glycine Precursor 1				
HCO_CH ₂ NH_TS-I	<i>i</i> 385.08 cm ⁻¹	<i>i</i> 521.11 cm ⁻¹	<i>i</i> 429.89 cm ⁻¹	<i>i</i> 471.96 cm ⁻¹
NHCH ₂ CHO_NH ₂ CH ₂ CO_TS-II	<i>i</i> 1968.32 cm ⁻¹	<i>i</i> 2175.92 cm ⁻¹	<i>i</i> 2119.86 cm ⁻¹	<i>i</i> 2123.79 cm ⁻¹
NHCH ₂ CHO_NH ₂ CH ₂ CO_PR_TS	<i>i</i> 1170.66 cm ⁻¹	<i>i</i> 1121.39 cm ⁻¹	<i>i</i> 1401.44 cm ⁻¹	<i>i</i> 1228.74 cm ⁻¹
COCH ₂ NH ₂ _H ₂ O_TS-III	<i>i</i> 1695.55 cm ⁻¹	<i>i</i> 1701.28 cm ⁻¹	<i>i</i> 1544.76 cm ⁻¹	<i>i</i> 1723.92 cm ⁻¹
Glycine Precursor 2				
CH ₃ _CO ₂ _TS-I	<i>i</i> 108.70 cm ⁻¹	<i>i</i> 649.06 cm ⁻¹	<i>i</i> 312.07 cm ⁻¹	<i>i</i> 253.82 cm ⁻¹
CH ₃ COO_CH ₂ COOH_TS-II	<i>i</i> 2022.58cm ⁻¹	<i>i</i> 2022.87 cm ⁻¹	<i>i</i> 1959.54 cm ⁻¹	<i>i</i> 2025.76 cm ⁻¹
CH ₃ COO_CH ₂ COOH_PR_TS	<i>i</i> 1349.55 cm ⁻¹	<i>i</i> 520.47cm ⁻¹	<i>i</i> 1291.2 cm ⁻¹	<i>i</i> 1315.06 cm ⁻¹
CH ₂ COOH_NH ₃ _TS-III	<i>i</i> 531.97 cm ⁻¹	<i>i</i> 587.60cm ⁻¹	<i>i</i> 627.54 cm ⁻¹	<i>i</i> 602.57 cm ⁻¹
Glycine Path-3				
Path-3	B3-LYP	B2-PLYP	M06-2x	ωB97X-D
HCOOH_OH_TS-I	<i>i</i> 1383.44 cm ⁻¹	<i>i</i> 1547.9258 cm ⁻¹	<i>i</i> 1445.15 cm ⁻¹	<i>i</i> 1496.16 cm ⁻¹
HCOO_HOCO_TS-II	<i>i</i> 1833.63 cm ⁻¹	<i>i</i> 1798.2241 cm ⁻¹	<i>i</i> 1722.05 cm ⁻¹	<i>i</i> 2123.79 cm ⁻¹
HCOO_HOCO-PR-TS-I	<i>i</i> 641.05 cm ⁻¹	<i>i</i> 613.4172 cm ⁻¹	<i>i</i> 346.08 cm ⁻¹	<i>i</i> 634.056 cm ⁻¹
HOCO_CH ₂ NH_TS-III	<i>i</i> 215.89 cm ⁻¹	<i>i</i> 267.5005 cm ⁻¹	<i>i</i> 259.213 cm ⁻¹	<i>i</i> 250.671 cm ⁻¹
NHCH ₂ COOH_NH ₂ CHCOOH_TS-IV	<i>i</i> 1925.00 cm ⁻¹	<i>i</i> 1957.365 cm ⁻¹	<i>i</i> 1872.09 cm ⁻¹	<i>i</i> 1941.09 cm ⁻¹
NHCH ₂ COOH_NH ₂ CHCOOH-PR-TS-II	<i>i</i> 1146.10 cm ⁻¹	<i>i</i> 1138.2513 cm ⁻¹	<i>i</i> 1140.47 cm ⁻¹	<i>i</i> 1171.28 cm ⁻¹
NHCH ₂ COOH_H ₂ _TS-V	<i>i</i> 1336.06 cm ⁻¹	<i>i</i> 1533.2631 cm ⁻¹	<i>i</i> 1830.31 cm ⁻¹	<i>i</i> 1511.64 cm ⁻¹
NHCH ₂ COOH_NH ₃ _TS-VI	<i>i</i> 1693.51 cm ⁻¹	<i>i</i> 1659.2477 cm ⁻¹	<i>i</i> 1704.51 cm ⁻¹	<i>i</i> 1842.82 cm ⁻¹
NH ₂ CHCOOH_H ₂ _VII	<i>i</i> 617.60 cm ⁻¹	<i>i</i> 1213.8028 cm ⁻¹	<i>i</i> 1577.15 cm ⁻¹	<i>i</i> 1080.53 cm ⁻¹
NH ₂ CHCOOH_NH ₃ _VIII	<i>i</i> 357.73 cm ⁻¹	<i>i</i> 897.3583 cm ⁻¹	<i>i</i> 1160.19 cm ⁻¹	<i>i</i> 918.16 cm ⁻¹

Note: NP- Not Performed Calculation

Table_SI_10. The <S**2> values of the TSs in different levels of theory in computed aug-cc-pVTZ basis-set.

	UB3-LYP <S**2>	UB2-PLYP <S**2>	UM06-2x <S**2>	UωB97X-D <S**2>
Glycine Precursor 1				
HCO_CH ₂ NH_TS-I	0.7666	0.8181	0.7695	0.7729
NHCH ₂ CHO_NH ₂ CH ₂ CO_TS-II	0.7643	0.784	0.7661	0.7664
NHCH ₂ CHO_NH ₂ CH ₂ CO_PR_TS	0.7578	0.7723	0.7603	0.7599
COCH ₂ NH ₂ _H ₂ O_TS-III	0.7534	0.7566	0.7543	0.7537

Glycine Precursor 2				
CH ₃ _CO ₂ _TS-I	0.7559	0.7601	0.7581	0.7573
CH ₃ COO_CH ₂ COOH_TS-II	0.7592	0.773	0.7608	0.7601
CH ₃ COO_CH ₂ COOH_PR_TS	0.758	0.7668	0.7587	0.7578
CH ₂ COOH_NH ₃ _TS-III	0.7539	0.7576	0.7542	0.7542
Glycine Path-3				
HCOOH_OH_TS-I	0.7568	0.7629	0.7598	0.7573
HCOO_HOCO_TS-II	0.754	0.7622	0.7556	0.7549
HCOO_HOCO-PR-TS-I	0.7528	0.7571	0.7539	0.7533
HOCO_CH ₂ NH_TS-III	0.7644	0.8037	0.7669	0.7697
NHCH ₂ COOH_NH ₂ CHCOOH_TS-IV	0.7585	0.7715	0.7588	0.7591
NHCH ₂ COOH_NH ₂ CHCOOH- PR-TS-II	0.7544	0.7603	0.7549	0.7551
NHCH ₂ COOH_H ₂ _TS-V	0.7586	0.7695	0.76	0.7588
NHCH ₂ COOH_NH ₃ _TS-VI	0.7582	0.7705	0.7606	0.7595
NH ₂ CHCOOH_H ₂ _VII	0.7555	0.7634	0.759	0.7571
NH ₂ CHCOOH_NH ₃ _VIII	0.7569	0.7648	0.7606	0.7591

Table_SI_11. The electronic energies for all investigated compounds (in Hartrees)

	Zero-point correction (B3-LYP(D3-BJ))	UCCSD(T)	UCCSD(T)+ Zero-point correction
Glycine Precursor 1			
HCO_CH ₂ NH_Pre-reactant	0.054676	-208.1787908	-208.1241148
HCO_CH ₂ NH_TS-I	0.056308	-208.1641038	-208.1077958
HCOCH ₂ NH_Int-I	0.058084	-208.1972577	-208.1391737
NHCH ₂ CHO_NH ₂ CH ₂ CO_TS-II	0.055094	-208.1389887	-208.0838947
COCH ₂ NH ₂ _Int-II	0.060667	-208.2155074	-208.1548404
COCH ₂ NH ₂ _H ₂ O_Int-III	0.08519	-284.5689343	-284.4837443
COCH ₂ NH ₂ _H ₂ O_TS-III	0.083821	-284.4991237	-284.4153027
Product (O-Protonated Glycine)	0.089726	-284.5719939	-284.4822679
NHCH ₂ CHO_NH ₂ CH ₂ CO_PR_Reactant	0.107775	-360.9064967	-360.7987217
NHCH ₂ CHO_NH ₂ CH ₂ CO_PR_TS	0.101668	-360.8614704	-360.7598024
COCH ₂ NH ₂ _PR_Product	0.110188	-360.9289479	-360.8187599
Glycine Precursor 2			
CH ₃ _CO ₂ _Pre-reactant	0.042078	-228.1059817	-228.0639037
CH ₃ _CO ₂ _TS-I	0.045304	-228.0757276	-228.0304236
CH ₃ _CO ₂ _Int-I	0.047427	-228.0862	-228.038773
CH ₃ COO_CH ₂ COOH_TS-II	0.042525	-228.0326334	-227.9901084
CH ₂ COOH_Int-II	0.046549	-228.0864435	-228.0398945

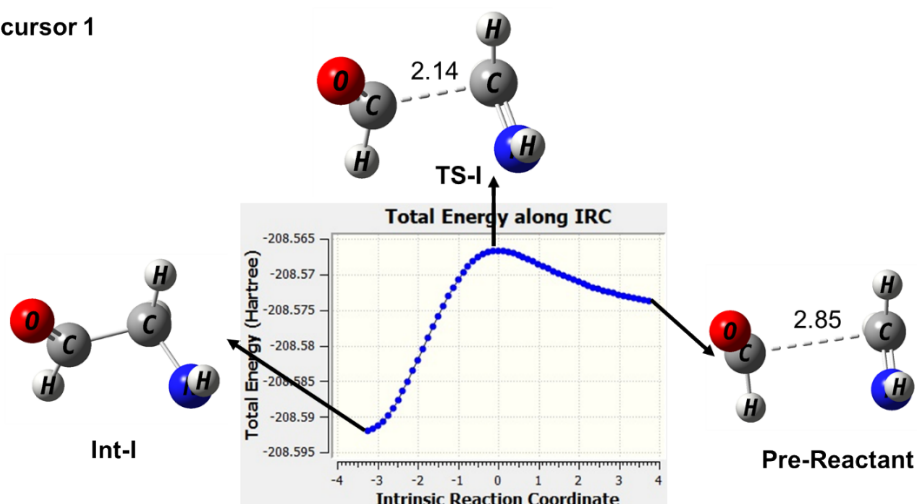
CH ₂ COOH NH ₃ Int-III	0.084773	-284.5926142	-284.5078412
CH ₂ COOH NH ₃ TS-III	0.087222	-284.4991237	-284.4119017
Product (O-Protonated Glycine)	0.089483	-284.5689671	-284.4794841
CH ₃ COO CH ₂ COOH Pre-Reactant	0.096148	-380.8083558	-380.7122078
CH ₃ COO CH ₂ COOH PR TS	0.08991	-380.7400885	-380.6501785
CH ₃ COO CH ₂ COOH PR Product	0.097336	-380.8103664	-380.7130304

Table_SI_12. The electronic energies for all investigated compounds (in Hartrees)

	Zero-point correction (B3-LYP(D3-BJ))	UCCSD(T)	UCCSD(T)+ Zero-point correction
Glycine Path-3			
HCOOH OH Pre-Reactant	0.045219	-265.1731921	-265.1279731
HCOOH OH TS-I	0.041888	-265.1591386	-265.1172506
HCOO H ₂ O Int-I	0.044766	-265.1739461	-265.1291801
HCOO Int-II	0.015809	-188.8254476	-188.8096386
HCOO TS-II	0.014357	-188.7911252	-188.7767682
HOCO Int-III	0.020899	-188.8517475	-188.8308485
HCOO PR-PR-I	0.068053	-341.5389623	-341.4709093
HCOO PR TS-I	0.065457	-341.535172	-341.469715
HOCO PR Product-I	0.070605	-341.5670242	-341.4964192
HOCO CH ₂ NH Int-IV	0.063187	-283.3524557	-283.2892687
HOCO CH ₂ NH TS-III	0.063263	-283.3343689	-283.2711059
NHCH ₂ COOH Int-V	0.064804	-283.3704589	-283.3056549
NHCH ₂ COOH NH ₂ CHCOOH TS-IV	0.061578	-283.3184916	-283.2569136
NH ₂ CHCOOH Int-VI	0.066409	-283.4035227	-283.3371137
NHCH ₂ COOH PR Reactant-II	0.115317	-436.1109081	-435.9955911
NHCH ₂ COOH NH ₂ CHCOOH PR TS-II	0.109269	-436.0432461	-435.9339771
NH ₂ CHCOOH PR Pd-II	0.113951	-436.0818359	-435.9678849
NHCH ₂ COOH H ₂ Int-VII	0.076438	-284.5458601	-284.4694221
NHCH ₂ COOH H ₂ TS-V	0.076896	-284.5234278	-284.4465318
NH ₂ CH ₂ COOH H Int-VIII	0.080132	-284.5429577	-284.4628257
NHCH ₂ COOH NH ₃ Int-IX	0.100608	-339.8592269	-339.7586189
NHCH ₂ COOH NH ₃ TS-VI	0.09765	-339.8308222	-339.7331722
NH ₂ CH ₂ COOH NH ₂ Int-X	0.100906	-339.850688	-339.749782
NH ₂ CHCOOH H ₂ Int-XI	0.077654	-284.5774216	-284.4997676
NH ₂ CHCOOH H ₂ TS-VII	0.077918	-284.5323601	-284.4544421
NH ₂ CH ₂ COOH H Int-XII	0.080132	-284.5429577	-284.4628257
NH ₂ CHCOOH NH ₃ Int-XIII	0.103028	-339.8953652	-339.7923372
NH ₂ CHCOOH NH ₃ TS-VIII	0.098772	-339.8352572	-339.7364852
NH ₂ CH ₂ COOH NH ₂ Int-XIV	0.100183	-339.8474789	-339.7472959

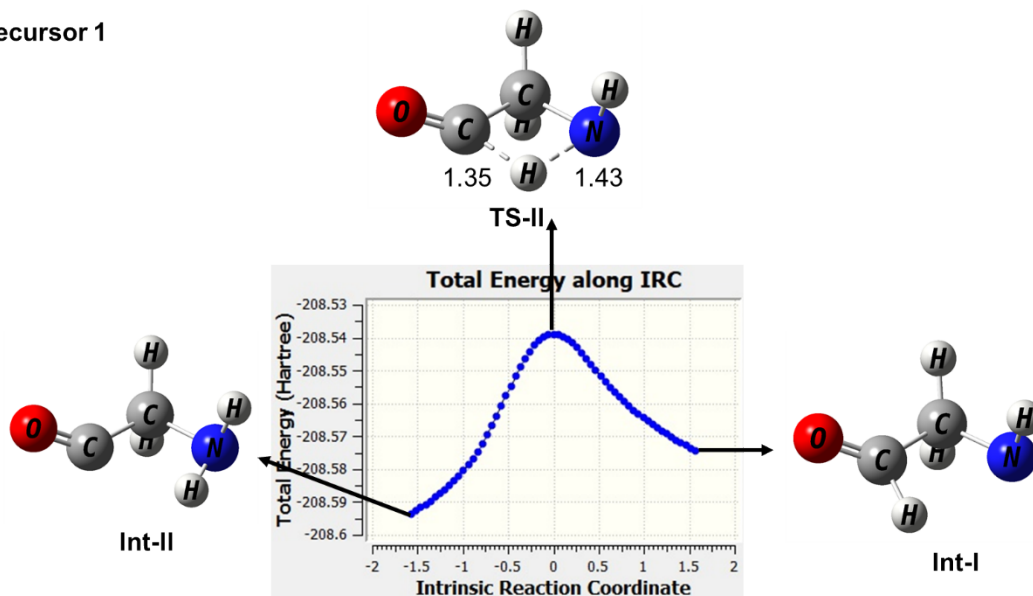
IRCs for the glycine precursors 1, glycine precursor 2, glycine path 3.
 The ub3lyp/aug-cc-pvtz irc=(calcfc,maxpoints=30, stepsize=5) level of theory used.

Glycine Precursor 1



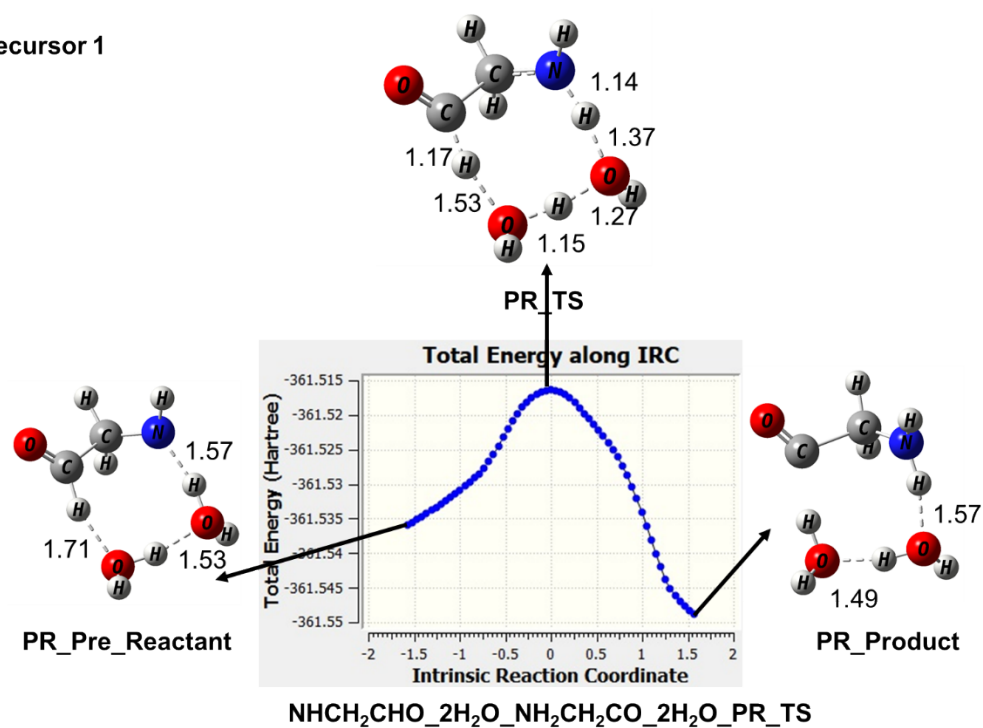
HCO₂CH₂NH₂_TS-I

Glycine Precursor 1

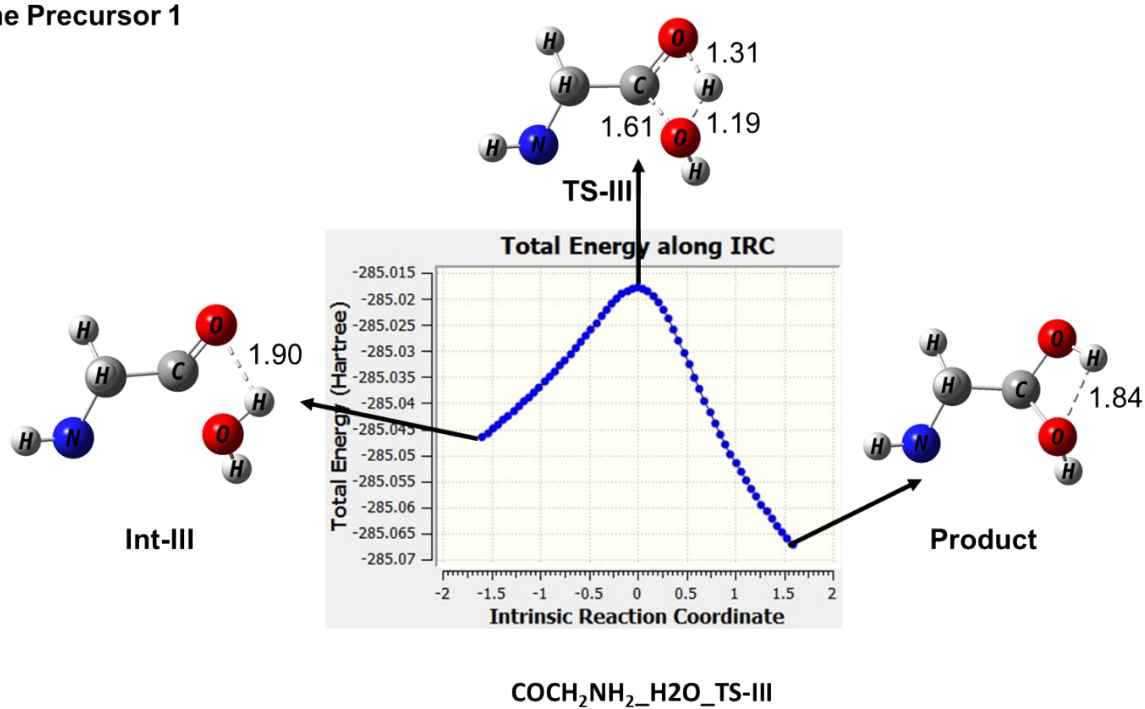


NHCH₂CHO_NH₂CH₂CO_TS-II

Glycine Precursor 1

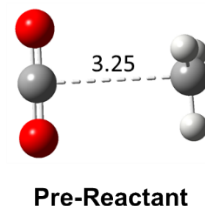
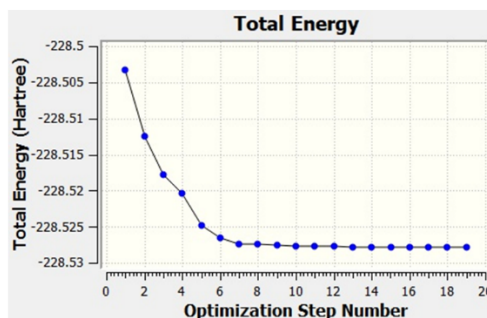
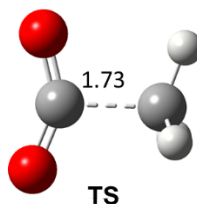
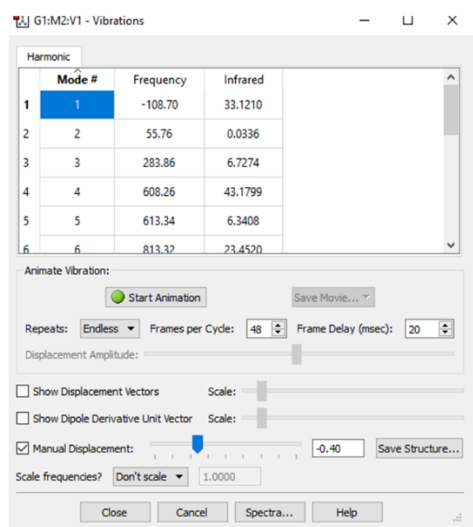


Glycine Precursor 1



Glycine Precursor 2

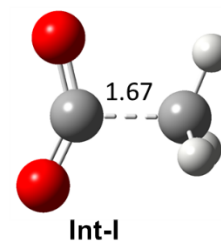
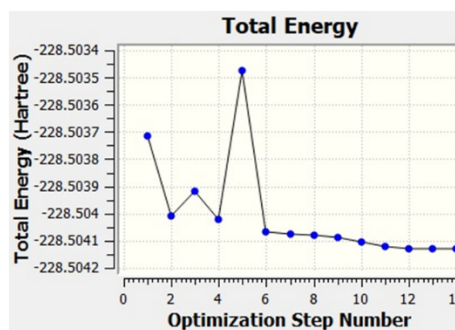
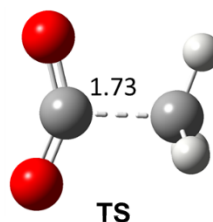
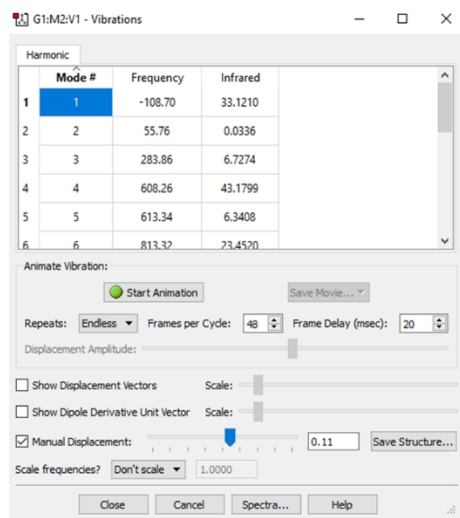
CH₃_CO₂_TS-I



The IRC calculation doesn't work,
So, slightly moved the structure
in the direction of imaginary vibration
and optimized

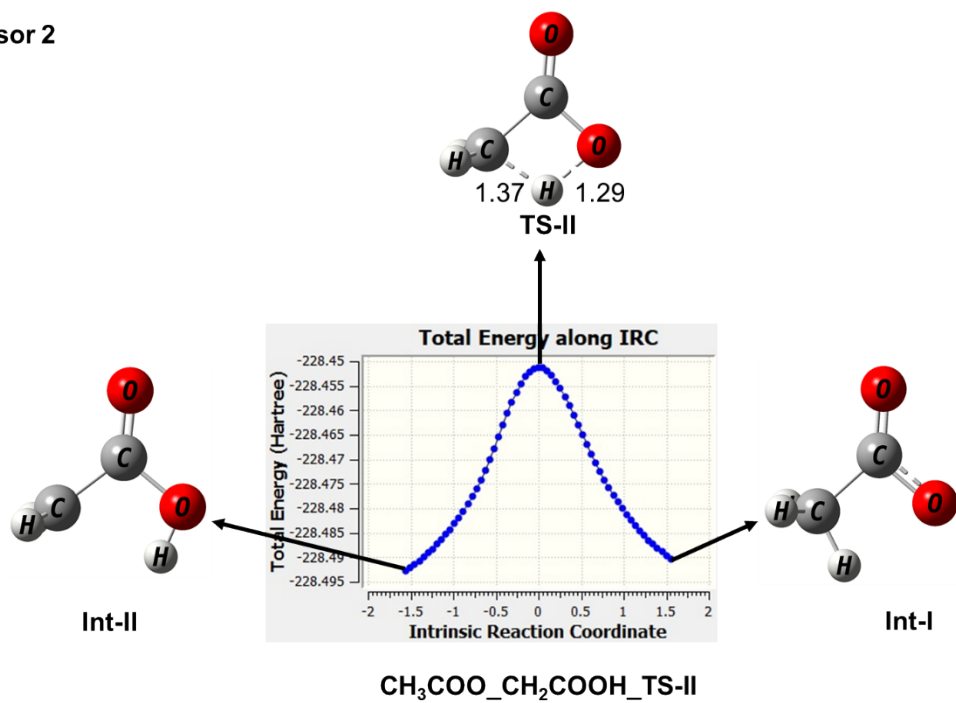
Glycine Precursor 2

CH₃_CO₂_TS-I

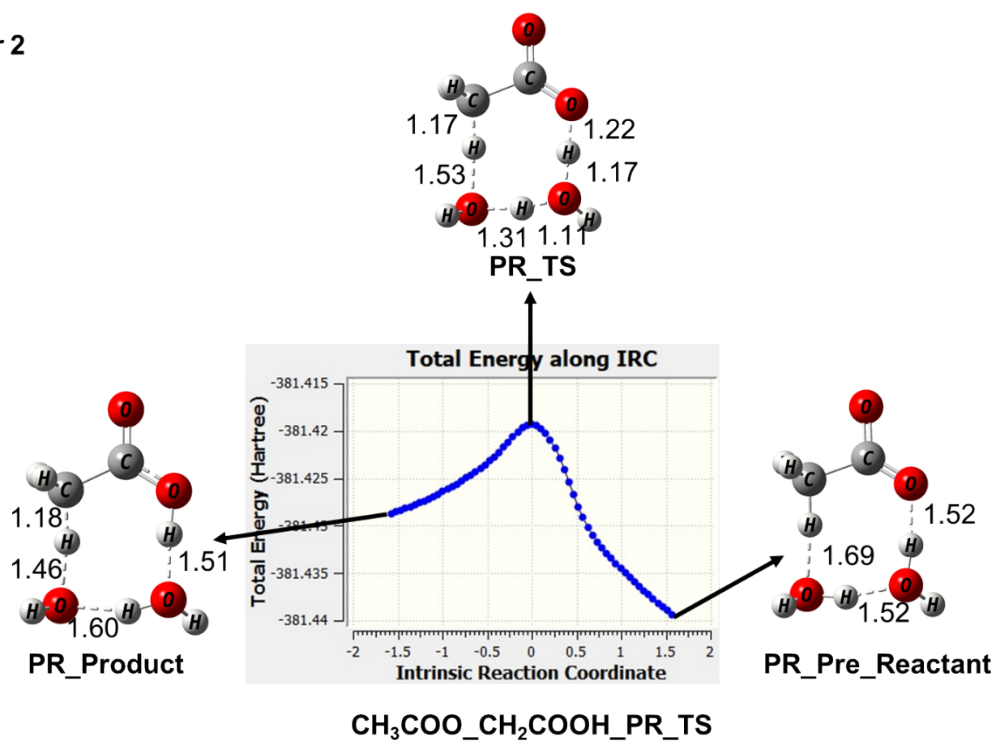


The IRC calculation doesn't work,
So, slightly moved the structure
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and optimized

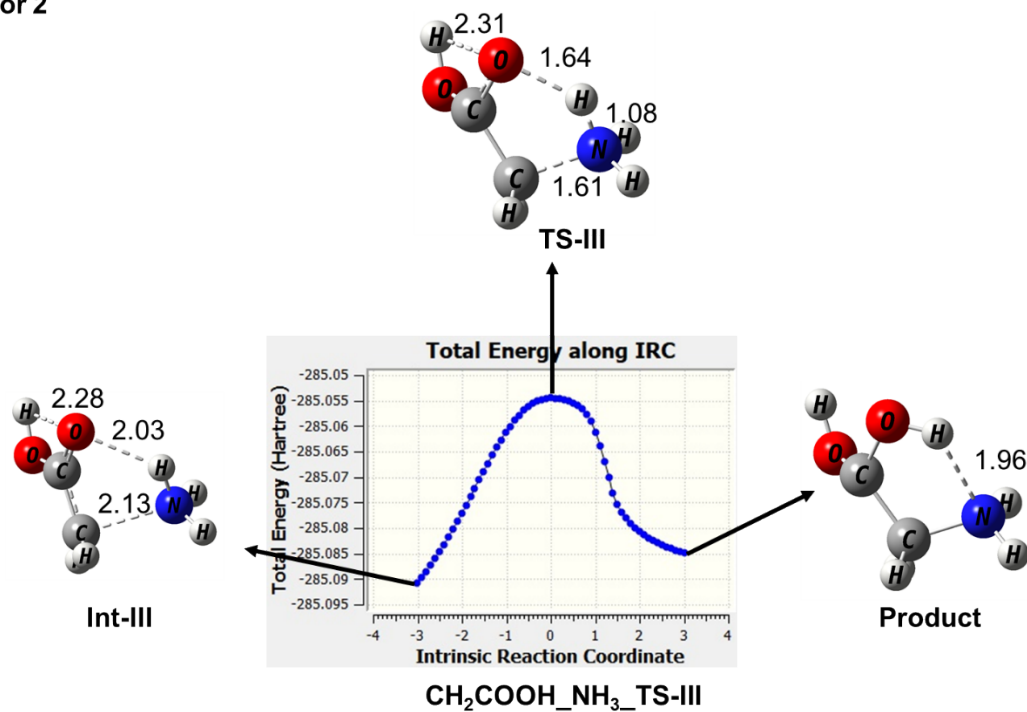
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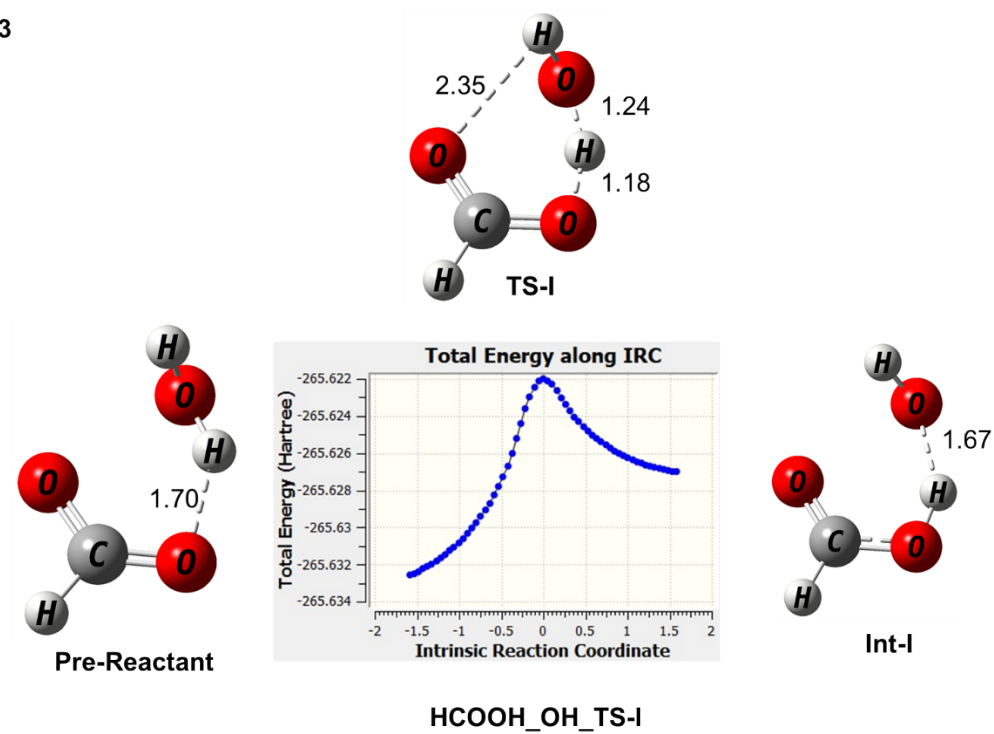
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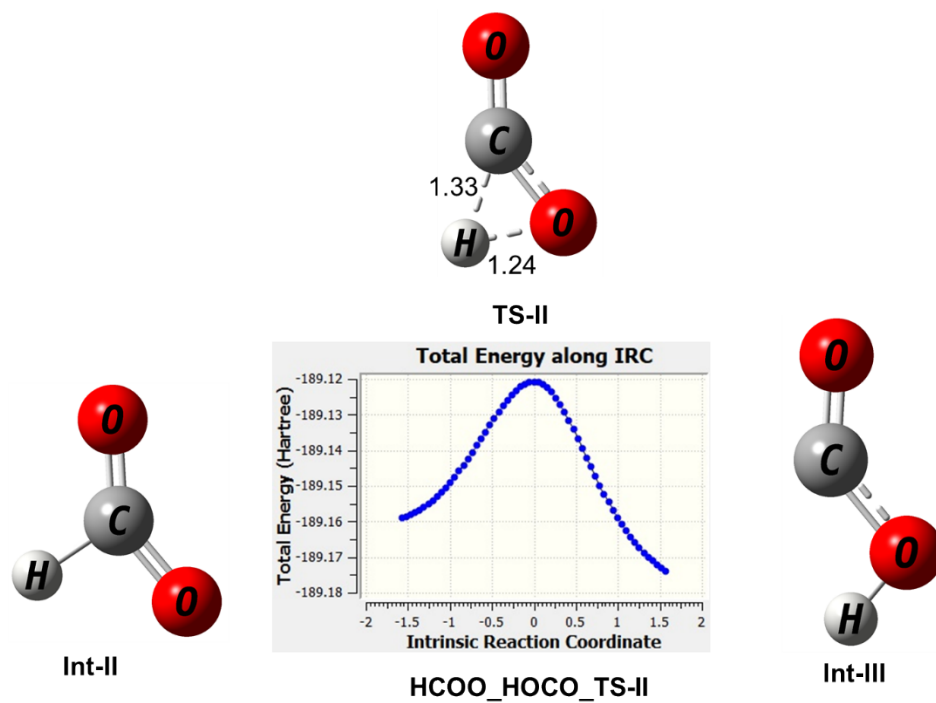
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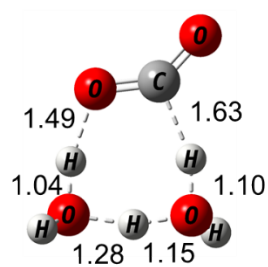
Glycine Path 3



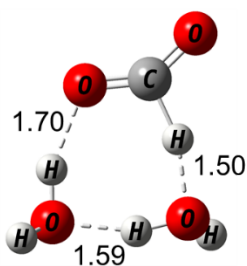
Glycine Path 3



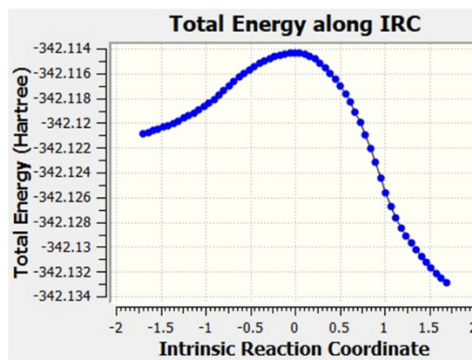
Glycine Path 3



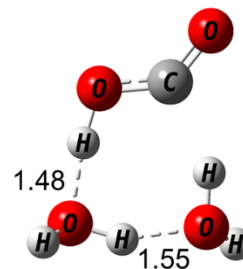
PR_TS-I



PR-PR-I

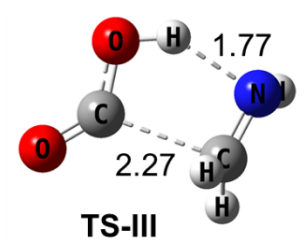


HCOO_HOCO-PR-TS-I

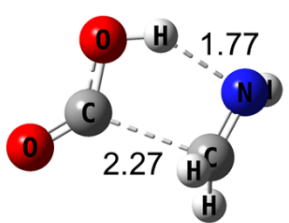


PR Product-I

Glycine Path 3

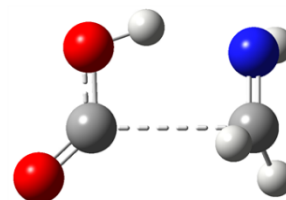
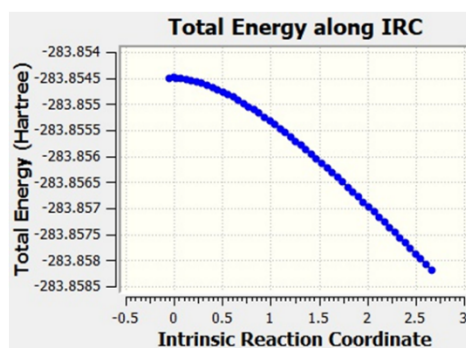


Optimized TS as an Initial Step

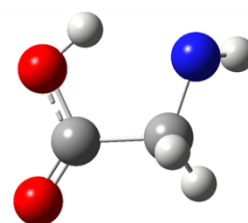
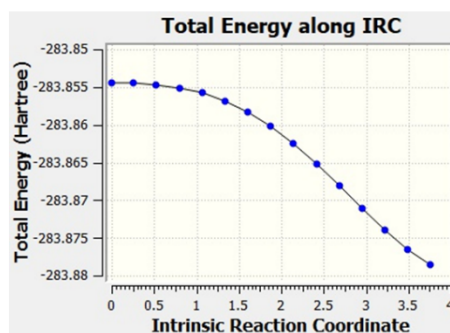


Optimized TS as an Initial Step

IRC Forward

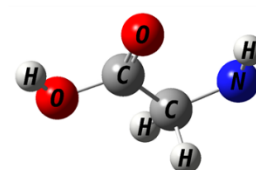
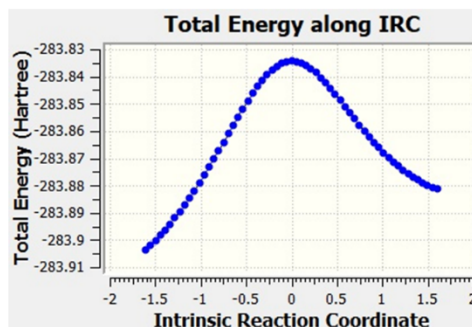
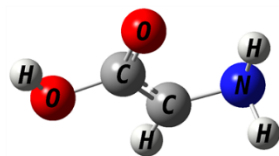
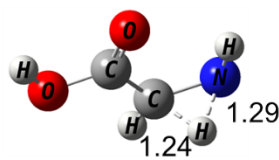


IRC Reverse



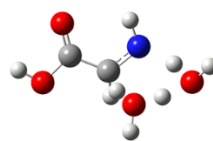
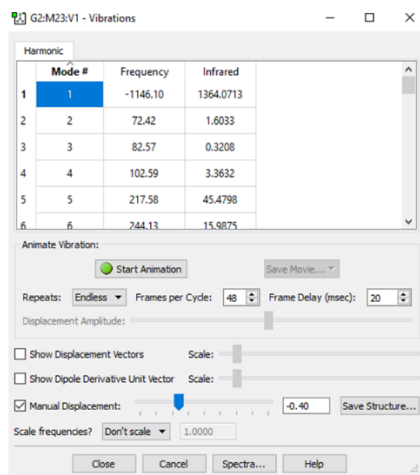
HOCO_CH₂NH_TS-III

Glycine Path 3

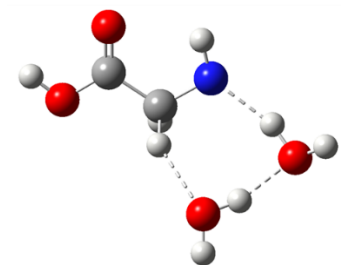
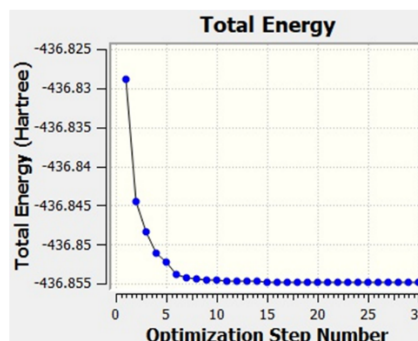


NHCH₂COOH_NH₂CHCOOH_TS-IV

Glycine Path 3



PR_TS-II

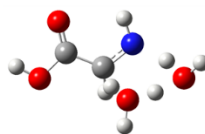
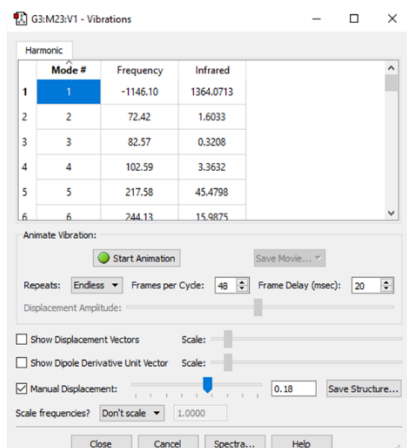


PR_Reactant-II

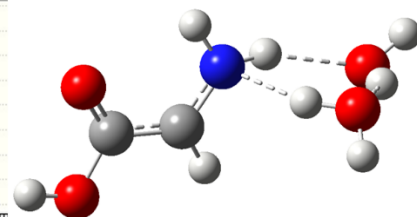
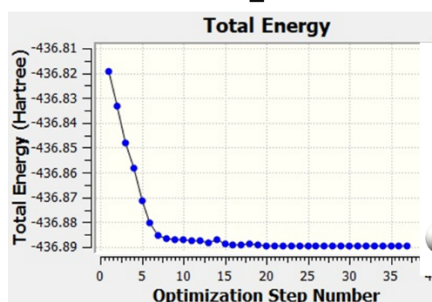
The IRC calculation doesn't work,
So, slightly moved the structure
in the direction of imaginary vibration
and optimized

NHCH₂COOH_NH₂CHCOOH-PR-TS-II

Glycine Path 3



PR_TS-II

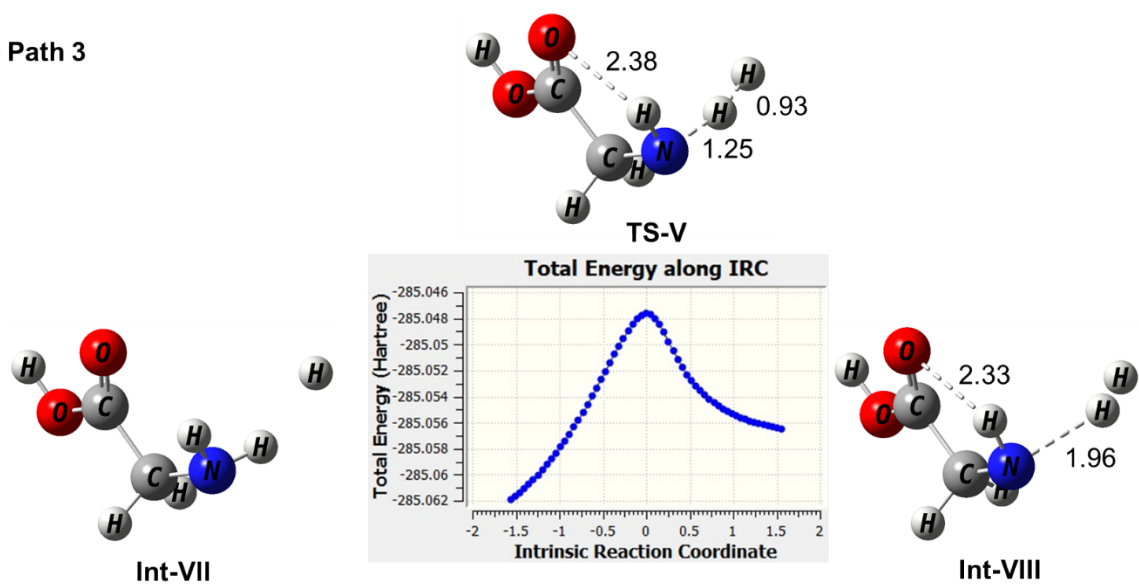


PR_Pd-II

The IRC calculation doesn't work,
So, slightly moved the structure
in the direction of imaginary vibration
and optimized

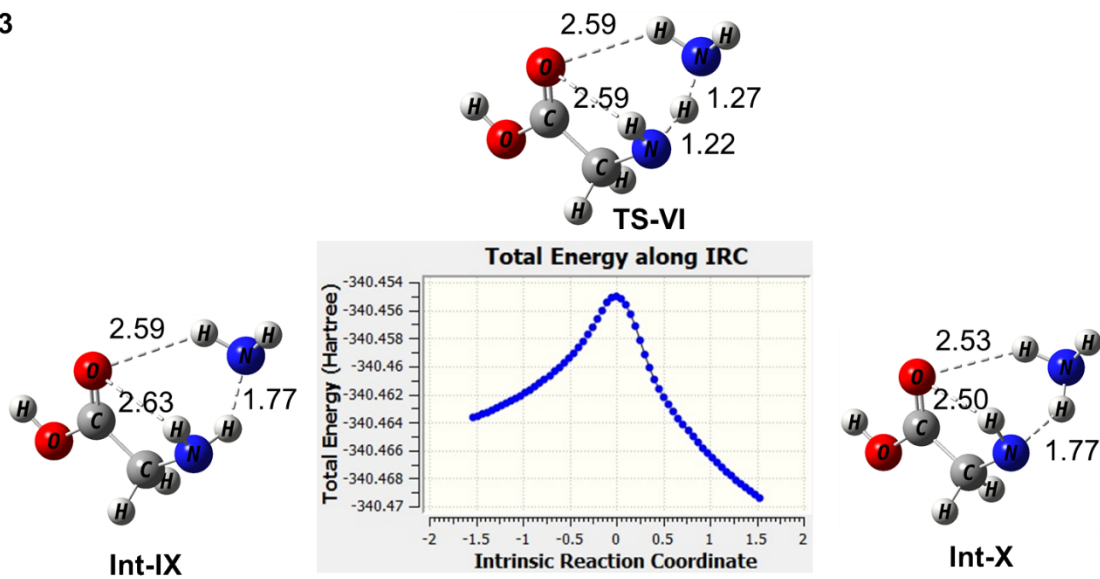
NHCH₂COOH_NH₂CHCOOH-PR-TS-II

Glycine Path 3



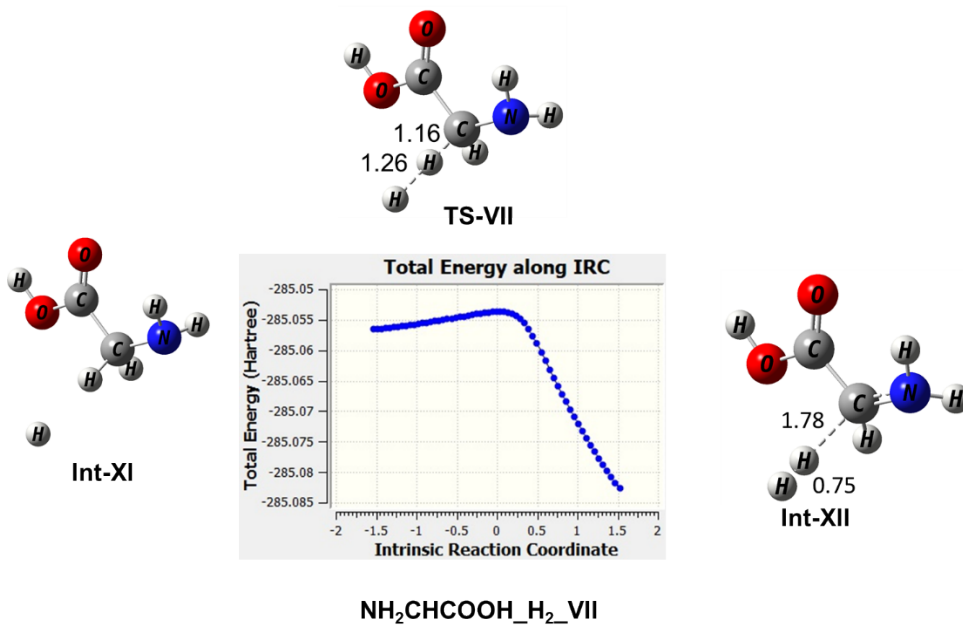
NHCH₂COOH_H₂_TS-V

Glycine Path 3

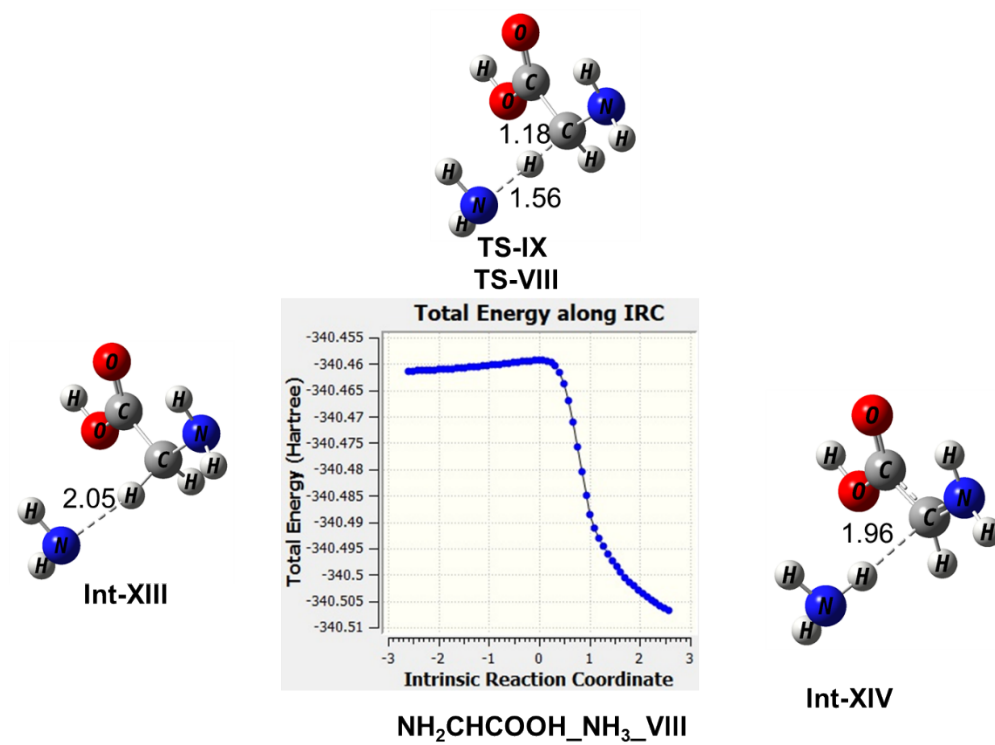


NHCH₂COOH_NH₃_TS-VI

Glycine Path 3



Glycine Path 3



Cartesian Coordinates of TSs and Magnitude of Imaginary Frequencies

The order of paths in which the coordinates are presented here is the same as that given in the main text. The same level of theory as described in the main text is used.

Glycine Precursors 1

HCO₂CH₂NH₂_TS-I

Imaginary Frequency = 385.0862i

0 2

C	0.94989500	0.13064700	0.45201400
O	1.76164200	-0.16392100	-0.34440600
C	-1.05323100	0.60124400	-0.15596800
H	-0.75872100	0.92960100	-1.15344900
N	-1.55628000	-0.56657800	0.10921900
H	-1.48238800	-1.18406200	-0.69923500
H	0.84467500	-0.25068600	1.48708000
H	-1.18271700	1.39121100	0.58003400

HCOCH₂NH₂_TS-II

Imaginary Frequency = 1968.3246i

0 2

C	-0.59082000	-0.09490500	-0.21501800
O	-1.72812100	-0.12883800	0.11436700
C	0.62007200	0.73526200	0.01136500
H	0.61277200	1.29099000	0.94837600
N	1.48370200	-0.50295100	-0.03293400
H	1.51229100	-0.88875500	0.91467400
H	0.26897400	-1.09325700	-0.51092200
H	0.86951000	1.40024300	-0.81460000

HCOCH₂NH₂_PR_TS

Imaginary Frequency = 1170.6700i

0 2

C	-1.29588800	0.50116100	0.04783400
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C	-0.85697600	-1.05050400	0.57952600
H	-1.80322600	-1.58404900	0.62975700
N	0.08010600	-1.56843600	-0.27948000
H	-0.26953400	-2.02989400	-1.10701100
H	-0.44234000	-0.82995800	1.56333500
O	1.10815300	1.69501500	0.20159200
H	1.33614200	2.36959900	-0.44507100
H	-0.33676900	1.18469300	0.08492700
O	-2.38738200	0.70540700	-0.34162900
H	1.75332900	0.75815000	0.01156900
O	2.28435100	-0.38140600	-0.20569700
H	1.14577200	-1.14299000	-0.24664200
H	2.93208800	-0.60257500	0.46722000

COCH₂NH₂_H₂O_TS-III

Imaginary Frequency = 1695.5538i

0 2

C	0.89438500	-0.55256000	0.32596500
H	1.06156100	-1.62076600	0.13845900
N	1.94061800	0.33540200	-0.17098300
H	2.12859000	0.15381700	-1.14910500
C	-0.42386800	-0.24071600	-0.30528700
O	-1.50670300	-0.91548700	-0.12204300
H	2.80252900	0.19654700	0.33963700
H	0.81029800	-0.42174700	1.41057000
O	-1.03157900	1.16323900	0.21043900
H	-1.87455600	0.31901600	0.13982300
H	-1.02958600	1.80295500	-0.51374100

Glycine Precursors 2

CH₃_CO₂_TS-I

Imaginary Frequency = 108.7065i

0 2

C	-0.33852400	0.02810600	0.00001200
O	-0.51809700	1.23588900	0.00000200
O	-0.88860600	-1.04346600	0.00000200
C	1.38803700	-0.17086800	-0.00000200
H	1.85461200	0.80287100	0.00009900

H	1.55086500	-0.74271900	-0.90583200
H	1.55106800	-0.74296900	0.90563600

CH₃CO₂_TS-II

Imaginary Frequency = 2022.5880i

0 2

C	0.24408800	-0.06804700	-0.00000100
O	-0.17232400	1.21818100	0.00000100
O	1.36759100	-0.46877200	0.00000000
C	-1.11753200	-0.67730300	0.00000000
H	-1.48895800	-1.09980600	-0.92598300
H	-1.34357200	0.67645400	0.00000800
H	-1.48895000	-1.09983000	0.92597500

CH₃CO₂_PR_TS-III

Imaginary Frequency = 1349.5589i

0 2

C	1.26775400	-0.10319200	-0.00915000
O	0.65214000	-1.20572800	-0.00535000
O	2.49595700	0.02834900	-0.10532200
C	0.60736800	1.25460500	0.14544000
H	0.75630700	1.62202800	1.16002900
H	-2.02028800	-0.15449700	-0.00889000
H	1.00909900	1.97343000	-0.56466500
O	-1.75221600	-1.22878200	0.15543900
H	-0.57490200	-1.22508200	0.07694900
H	-2.13335200	-1.78135100	-0.53412200
O	-2.06527900	1.13910900	-0.22461400
H	-0.54413100	1.17658800	-0.05857300
H	-2.38828500	1.61682600	0.55031400

CH₂COOH_NH₃_TS-III

Imaginary Frequency = 531.9789i

0 2

C	0.61546600	-0.75020400	0.52693600
H	0.55568900	-1.79108700	0.22523100
C	-0.61995300	0.04508100	0.37459600

O	-0.42341200	1.30396000	0.18465800
O	-1.61367900	-0.61407600	-0.37168000
H	-2.18458100	0.08913200	-0.70541700
H	1.08219300	-0.63753300	1.50169900
N	1.64843200	0.03931900	-0.43918600
H	2.60188100	0.12048100	-0.10081700
H	1.08627200	0.96370500	-0.36743200
H	1.64317800	-0.30826500	-1.39198000

Glycine Path-3

HCOOH_OH_TS-I

Imaginary Frequency = 1383.4455i

0 2

C	-1.01075900	0.10573400	-0.00125800
O	-0.29551500	1.12786400	-0.01334500
O	-0.56887100	-1.08111300	0.02204800
H	0.59174100	-0.83133100	0.01236600
O	1.58714800	-0.09437500	-0.10156600
H	1.78960300	0.33193400	0.74389100
H	-2.09888600	0.24598000	-0.00580500

HCOO_TS-II

Imaginary Frequency = 1833.6259i

0 2

C	0.07887300	0.23756000	0.00003800
O	1.20281400	-0.10025300	-0.00003700
O	-1.13724900	-0.20640200	-0.00001500
H	-0.99776400	1.02788300	0.00018600

HCOO_PR_TS

Imaginary Frequency = 641.0529i

0 2

C	1.15579400	-0.08214900	-0.01129600
O	2.28399800	0.33514800	-0.02129100
O	0.55887300	-1.17897500	0.04035100
H	-0.93166100	-1.09533500	0.06354200
O	-1.91314800	-0.74931900	0.08194200

H	-0.01823300	1.06094500	-0.07376700
H	-2.38101300	-1.10990200	-0.67838200
O	-1.03476100	1.48242600	-0.09194500
H	-1.14583500	2.02982700	0.69719100
H	-1.61772500	0.49312700	-0.01326200

COOH_CH₂NH_TS-III

Imaginary Frequency = 215.8982i

0 2

C	-0.87312600	-0.05902600	-0.02434000
O	-1.88531600	-0.65953700	-0.11157900
C	1.25185300	-0.83461600	0.16487500
H	1.22955600	-1.67021000	-0.53237900
N	1.85099900	0.29473700	-0.06765000
H	2.20547600	0.33913500	-1.02120400
H	0.96130900	-1.06419600	1.18655000
O	-0.61610700	1.21146500	0.09349900
H	0.38568600	1.27853700	0.14200900

NHCH₂COOH_NH₂CHCOOH_TS-IV

Imaginary Frequency = 1925.00i

0 2

C	0.75234300	-0.66371300	0.01831900
H	0.73928100	-1.72414300	-0.19069000
N	2.01051000	-0.00048000	-0.11316600
H	1.51210200	-0.46988200	0.99036000
C	-0.50214700	0.09421400	-0.00184000
O	-0.56575800	1.30436500	0.01296700
O	-1.59160100	-0.71066100	-0.01997000
H	-2.37058400	-0.13591700	-0.01032400
H	1.80332800	1.00066000	-0.04003300

NHCH₂COOH_NH₂CHCOOH- PR-TS-II

Imaginary Frequency = 1146.10i

0 2

C	0.24115500	-0.19961500	-0.57600800
H	0.11712500	0.25057400	-1.55918800

N	-0.55376700	-1.27731500	-0.29326000
H	-0.55132200	0.86784700	0.12844200
C	1.58281000	-0.12055300	0.00187400
O	2.00849700	-0.82595800	0.89307100
O	2.32425600	0.89221100	-0.53460300
H	3.17028100	0.88891300	-0.06548800
H	-0.14084500	-1.81234400	0.46990200
O	-1.46582000	1.46071300	0.57523600
H	-2.25287700	0.75118100	0.30297800
H	-1.57604300	2.28436600	0.08455500
O	-2.90790200	-0.35470300	-0.18308300
H	-3.49592400	-0.80727700	0.42860000
H	-2.01006000	-0.93915200	-0.29715700

NHCH₂COOH_H₂_TS-V

Imaginary Frequency = 1336.06i

0 2

C	-0.64875800	-0.75050900	-0.18376600
H	-0.65029200	-1.52436800	0.58427200
N	-1.90143300	-0.02739400	-0.18086000
H	-1.71073500	0.92896500	-0.49026400
H	-0.55318900	-1.27989400	-1.13986900
C	0.59162400	0.10795800	-0.03389800
O	0.61056000	1.30855300	-0.10902700
O	1.69533900	-0.64225200	0.16558400
H	2.45235300	-0.04019200	0.22218700
H	-2.19544900	0.18424400	1.02514800
H	-2.13705100	0.44790200	1.91807200

NHCH₂COOH_NH₃_TS-VI

Imaginary Frequency = 1693.51i

0 2

C	0.12835800	1.10277800	-0.15089600
H	0.04035300	1.24019500	-1.23021100
N	-1.18052600	1.11360100	0.45342300
H	-1.11444200	0.72975500	1.39479600
H	0.70964200	1.95910100	0.21211800
C	0.96860900	-0.13696700	0.09931500

O	0.64593200	-1.07674000	0.77979400
O	2.15550800	-0.06264500	-0.53983200
H	2.64545800	-0.87475800	-0.34244300
N	-2.33750000	-0.77492800	-0.61320100
H	-1.87357400	0.32874300	-0.18055600
H	-1.64735300	-1.48223300	-0.35635200
H	-3.12721700	-0.95129600	0.01088900

NH₂CHCOOH_H₂_VII

Imaginary Frequency = 617.60i

0 2

C	0.72856300	-0.63464600	-0.09752900
N	1.91768500	0.15175300	-0.05332400
H	0.67051900	-1.34532300	0.82724100
C	-0.55636900	0.14724600	-0.02276800
O	-0.63280300	1.32638700	0.21399800
O	-1.63463600	-0.64542400	-0.20066900
H	-2.42296900	-0.09511000	-0.08359000
H	1.73872800	1.07153200	0.32831600
H	2.36616900	0.24290900	-0.95199200
H	0.61855700	-2.13562600	1.80778800
H	0.71154400	-1.32396100	-0.93934800

NH₂CHCOOH_NH₃_VIII

Imaginary Frequency = 357.73i

0 2

C	0.31375900	0.50395000	-0.59760000
H	0.33468400	0.35816600	-1.67702500
N	0.43179900	1.87502100	-0.22883300
C	-0.88962500	-0.13490700	0.01741700
O	-1.46043600	0.27028900	0.99923300
O	-1.23775900	-1.27617800	-0.61948900
H	-1.98131300	-1.65959000	-0.13171900
H	0.15735800	2.03298500	0.73180200
H	1.35048300	2.25376300	-0.40563300
N	2.39082800	-1.00933700	0.36970800
H	1.98656000	-1.18993400	1.29440100
H	2.21196800	-1.87828400	-0.14359000

H	1.22262600	-0.14403700	-0.21122100
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