Supplementary Information

Growth vector elaboration of fragments: regioselective functionalization of 5-hydroxy-6-azaindazole and 3-hydroxy-2,6-naphtyridine

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1. SOLUTION GIBBS ENERGIES (G_{sol})

pKa values were computed using the following isodesmic reaction:

$$M-H + \left(\begin{array}{c} \bigcirc \\ N \end{array} \right) \longrightarrow M^{\bigcirc} + \left(\begin{array}{c} \bigcirc \\ N \end{array} \right)$$

pKa values were obtained in DMSO solvent system, performed by PCM model using singlepoint calculations with B3LYP/6-311++G(d,p)//B3LYP/6-31+G(d,p) model. The solvation Gibbs energies were obtained to deprotonation reaction using pyridine as reference compound.

Gibbs energies in solution for neutral and deprotonated species. All the values were obtained in B3LYP/6-311++G(d,p) using DMSO in PCM. Values in Hartree.

In order to predict the regioselectivity, we employed density-functional theory (DFT) calculations as previously done in several studies to describe radical functionalization in heterocycle compounds.¹ To estimate the reactivity of the hydroxypyridine moiety, pKa values were calculated using isodesmic reactions²⁻⁷ in dimethyl sulfoxide (DMSO) using the proteochemometric (PCM) model and single-point calculations at the B3LYP/6-311++G(d,p) level using Gaussian 03.⁸⁻¹⁶ The solvation Gibbs energies were obtained for deprotonation reactions using pyridine as a reference compound (see the Supporting Information). The most acidic C–H was found to be next to the pyridone oxygen and pyrazole nitrogen (Figure 2).



Figure 1. Calculated pKa values in DMSO for heterocycles **1** and **2** using DFT (PCM/B3LYP/6-311++G(d,p)/B3LYP/6-31+G(d,p)). Arrows indicate the predicted reactive sites for C–H activation.

References:

- A)Cernak, T.; Dykstra, K. D.; Tyagarajan, S.; Vachal, P.; Krska, S. W. The Medicinal Chemist's Toolbox for Late Stage Functionalization of Drug-like Molecules. *Chem. Soc. Rev.* 2016, 45 (3), 546–576. B) Dong, Y.; Zhang, X.; Chen, J.; Zou, W.; Lin, S.; Xu, H. Switching the Site-Selectivity of C–H Activation in Aryl Sulfonamides Containing Strongly Coordinating N-Heterocycles. *Chem. Sci.* 2019, 10 (38), 8744–8751.
- 2. Shen, K.; Fu, Y.; Li, J.-N.; Liu, L.; Guo, Q.-X. What are the pKa values of C–H bonds in aromatic heterocyclic compounds in DMSO? *Tetrahedron* 2007, *63* (7), 1568-1576.
- 3. da Silva, G.; Chen, C. C.; Bozzelli, J. W. Bond dissociation energy of the phenol OH bond from ab initio calculations. *Chem. Phys. Lett.* 2006, **424** (1-3), 42-45.
- 4. da Silva, G.; Kim, C. H.; Bozzelli, J. W. Thermodynamic Properties (Enthalpy, Bond Energy, Entropy, and Heat Capacity) and Internal Rotor Potentials of Vinyl Alcohol, Methyl Vinyl Ether, and Their Corresponding Radicals. J. Phys. Chem. A 2006, **110** (25), 7925-7934.
- 5. Casasnovas, R.; Fernandez, D.; Ortega-Castro, J.; Frau, J.; Donoso, J.; Munoz, F. Avoiding gas-phase calculations in theoretical pKa predictions. *Theor. Chem. Acc.* 2011, **130** (1), 1-13.
- 6. Sastre, S.; Casasnovas, R.; Munoz, F.; Frau, J. Isodesmic reaction for pKa calculations of common organic molecules. *Theor. Chem. Acc.* 2013, **132** (2), 1310-1317.
- Chevallier, F.; Blin, T.; Nagaradja, E.; Lassagne, F.; Roisnel, T.; Halauko, Y. S.; Matulis, V. E.; Ivashkevich, O. A.; Mongin, F. Deproto-metallation and computed CH acidity of 2-aryl-1,2,3-triazoles. *Org. Biomol. Chem.* 2012, 10, 4878-4885.

- 8. Shen, K.; Fu, Y.; Li, J.-N.; Liu, L.; Guo, Q.-X. What are the pKa values of C–H bonds in aromatic heterocyclic compounds in DMSO? *Tetrahedron* 2007, *63* (7), 1568-1576.
- 9. da Silva, G.; Chen, C. C.; Bozzelli, J. W. Bond dissociation energy of the phenol OH bond from ab initio calculations. *Chem. Phys. Lett.* 2006, **424** (1-3), 42-45.
- da Silva, G.; Kim, C. H.; Bozzelli, J. W. Thermodynamic Properties (Enthalpy, Bond Energy, Entropy, and Heat Capacity) and Internal Rotor Potentials of Vinyl Alcohol, Methyl Vinyl Ether, and Their Corresponding Radicals. J. Phys. Chem. A 2006, **110** (25), 7925-7934.
- 11. Casasnovas, R.; Fernandez, D.; Ortega-Castro, J.; Frau, J.; Donoso, J.; Munoz, F. Avoiding gas-phase calculations in theoretical pKa predictions. *Theor. Chem. Acc.* 2011, **130** (1), 1-13.
- 12. Sastre, S.; Casasnovas, R.; Munoz, F.; Frau, J. Isodesmic reaction for pKa calculations of common organic molecules. *Theor. Chem. Acc.* 2013, **132** (2), 1310-1317.
- Chevallier, F.; Blin, T.; Nagaradja, E.; Lassagne, F.; Roisnel, T.; Halauko, Y. S.; Matulis, V. E.; Ivashkevich, O. A.; Mongin, F. Deproto-metallation and computed CH acidity of 2-aryl-1,2,3-triazoles. *Org. Biomol. Chem.* 2012, *10*, 4878-4885.
- 14. Tomasi, J.; Mennucci, B.; Cammi, R. Quantum Mechanical Continuum Solvation Models. *Chem. Rev.* 2005, **105** (8), 2999-3094.
- 15. Tomasi, J.; Persico, M. Molecular Interactions in Solution: An Overview of Methods Based on Continuous Distributions of the Solvent. *Chem. Rev.* 1994, *94* (7), 2027-2094.
- Gaussian 03, R. C., M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, J. A. Montgomery, Jr., T. Vreven, K. N. Kudin, J. C. Burant, J. M. Millam, S. S. Iyengar, J. Tomasi, V. Barone, B. Mennucci, M. Cossi, G. Scalmani, N. Rega, G. A. Petersson, H. Nakatsuji, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, M. Klene, X. Li, J. E. Knox, H. P. Hratchian, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, P. Y. Ayala, K. Morokuma, G. A. Voth, P. Salvador, J. J. Dannenberg, V. G. Zakrzewski, S. Dapprich, A. D. Daniels, M. C. Strain, O. Farkas, D. K. Malick, A. D. Rabuck, K. Raghavachari, J. B. Foresman, J. V. Ortiz, Q. Cui, A. G. Baboul, S. Clifford, J. Cioslowski, B. B. Stefanov, G. Liu, A. Liashenko, P. Piskorz, I. Komaromi, R. L. Martin, D. J. Fox, T. Keith, M. A. Al-Laham, C. Y. Peng, A. Nanayakkara, M. Challacombe, P. M. W. Gill, B. Johnson, W. Chen, M. W. Wong, C. Gonzalez, and J. A. Pople, Gaussian, Inc., Wallingford CT, 2004.



2. SYNTHETIC PROCEDURES - for compounds already published in the literature

All solvents used for extraction and chromatography procedures were used as received from commercial suppliers without further purification. All reagents were purchased and used as received unless otherwise noted. The physicochemical properties were calculated using Vortex (2015.07.42634, Dotmatics Ltd.) and MarvinSketch (15.7.20, 2015, ChemAxon).

1H-Pyrazolo[3,4-c]pyridin-5-ol (1)



To a solution of 4-methyl-5-nitropyridin-2-ol 3 (3.52 g, 22.83 mmol) in methanol (500 mL) was added Raney[®] 2800 nickel (excess, 15 mL of the slurry in water) at room temperature. This suspension was purged with hydrogen and stirred for 5 hours. The solid material was filtered off and the filtrate was concentrated. Then, acetic anhydride (80 mL, 846 mmol) was added to the crude solid, the reaction was stirred for 17 hours at room temperature under nitrogen atmosphere. Then, NaNO₂ (7.879 g, 114.2 mmol) was added to the mixture at vigorous stirring at room temperature for 3 hours and after this time, the mixture was heated to 90°C for 3 hours. The insoluble material was filtered off and washed with EtOAc (3x20 mL), the solvent was removed under reduced pressure to give brown solid as the crude product. 1 N HCl (100 mL) aqueous solution was added to crude product and it was stirred for 17 hours at room temperature. The solvent was distilled off under reduced pressure resulting in the crude hydrochloride form. The product was purified by flash chromatography (dichloromethane to remove impurities, and then methanol – dichloromethane, 4:1), to yield and orange solid, 3.23 g (82%). ¹H NMR (400 MHz, XXXyridin- d_4) δ 8.89 (s, 1H), 8.30 (s, 1H), 7.44 (s, 1H).¹³C NMR (101 MHz, Methanol- d_4) δ 152.6, 135.1, 134.2, 133.0, 127.5, 99.5. IR (solid, v/cm⁻¹): 3154, 1656, 1537, 1402, 607. HRMS (EI) m/z [M+H]⁺ calcd for C₆H₆N₃O⁺: 136.0505; found: 136.0506. The obtained spectroscopic data are in accordance with those reported in the literature.¹

3-iodopyridine (7)



To a solution of p-TsOH.H₂O (244.5 g, 1.28 mol) in MeCN (2 L) was added **6** (40.1 g, 0.42 mol). The resulting suspension was cooled to 10-15 °C, then a solution of NaNO₂ (58.6 g, 0.85 mmol) and KI (176.3 g, 1.06 mol) in H₂O (200 mL) was added. The reaction mixture was stirred for 10 min, then allowed to stir at 20 °C for 1 hour. To the reaction mixture was then added a solution of NaHCO₃ aq. (1M, until pH = 9-10) and Na₂S₂O₃ (2M, 100 mL). The reaction was extracted with EtOAc (3 x 250 mL) and purified by flash chromatography (ethyl acetate – dichloromethane, 1:4), resulting in 55g of **7** (64%) as a pale solid. The obtained spectroscopic data are in accordance with those reported in the literature.²

Ethyl-2-(pyridinyl)-3-acetate (8)



A mixture of **7** (57.2 g, 0.278 mol), ethyl acetoacetate (72 mL, 0.834 mmol), Cul (5.2 g, 10 mol%), K_3PO_4 (175.4 g, 0.834 mol) and ethanol (48 mL, 0.834 mol) in DMSO (200 mL) was stirred under N_2 atmosphere at 80 °C for 17 h. After completion of the reaction, the mixture was diluted with water (200 mL) and extracted with ethyl acetate (3 x 100 mL). The pure product was obtained by flash column chromatography (methanol – dichloromethane, 1:19), resulting in 45.9 g of **8** (78%) as an orange oil. The obtained spectroscopic data are in accordance with those reported in the literature.³

3-(2-ethoxy-2-oxoethyl)pyridine 1-oxide (8.1)



A solution of mCPBA (61 mmol, 10.5 g) in chloroform (100 mL) was added dropwise to a stirring solution of **8** (41 mmol, 6.82g) in methanol (50 ml), at room temperature. The reaction was followed by LCMS. After 4 hours the reaction was completed, the solvent was removed under reduced pressure and then dissolved in 2M Na₂CO₃ solution and extracted with chloroform (3x 20 mL). The organic layer was dried with MgSO₄ and the solvent removed under reduced pressure to give 6,59 g of **8** (88%) as a pale crystal. M.p. 103 °C. The obtained spectroscopic data are in accordance with those reported in the literature.¹

1-ethoxy-3-(2-ethoxy-2-oxoethyl)XXXyridine-1-ium iodide (8.2)



Iodoethane (10 mL, 134 mmol) was added to **8.1** (6.48 g, 35 mmol), the mixture was stirred for 6 hours at 40 °C under nitrogen atmosphere. The mixture was concentred under reduced pressure, resulting in 9.79 g (83%) of crude product **8.2** as a dark orange oil. ¹H NMR (400 MHz, CDCl₃) δ 1.24 (t, *J* = 5 Hz, 3H,), 3,57 (s, 2H), 4.36 (q, *J* = 5Hz, 2H), 7.26 (s, 1H), 7.26 (s, 1H), 8.19 (s, 1H), 8.2 (s, 1H). The obtained spectroscopic data are in accordance with those reported in the literature.¹

ethyl 2-(4-cyanopyridin-3-yl)acetate (8.3)



KCN solution (3.55 g, 54.5 mmol) in water (20 mL) was added dropwise for 20 minutes to a solution of **8.2** (18.37, 54.5 mmol) in EtOH:H₂O (7:3, 30 mL) at 50 °C. After 1h, the reaction mixture was poured in ice and extracted with DCM (3x30), dried over MgSO₄ and concentrated under reduced pressure. The product was purified by Flash chromatography (Hexane – ethyl acetate, 6:4) to give 7.14 g of **8.3** (69%) of a dark orange oil. ¹H NMR (300 MHz, CDCl₃) δ 1.24 (t, *J* = 6 Hz, 3H,), 3.82 (s, 2H), 4.45 (q, *J* = 6Hz, 2H), 7.5 (d, *J* = 2Hz, 1H), 8.65 (d, *J*= 6Hz, 1H), 8.68 (s, 1H). IR (cm⁻¹): 2355, 1732, 1166, 1024, 773, 769, 578. HRMS (EI) m/z [M+] calcd for C₁₀H₁₁N₂O₂⁺: 191.0815; found: 191.0831. The obtained spectroscopic data are in accordance with those reported In the literature.¹

1,4-dihydro-2,6-naphthyridin-3(2H)-one (8.4)



A solution of **8.3** (2.79 g, 14.63 mmol) in acetic acid : ethanol (150 mL, 1: 4) was hydrogenated by 10% Pd/C (300 mg) and H₂ balloons. The reaction was stirred for 24 h at 50 °C. The reaction was filtered through celite, and the residue washed with hot acetic acid (3 x 10 mL). The solution was evaporated under reduced pressure, the residue was poured into a 10% aqueous K₂CO₃ solution and extracted with ethyl acetate (6x20mL). The crude product was purified by flash column chromatography (ethyl acetate), 1.77 g of **8.4** (82%) as a light brown solid. ¹H NMR (400 MHz, DMSO d_6) δ 1.24 (t, *J* = 6 Hz, 3H,), 3.82 (s, 2H), 4.45 (q, *J* = 6Hz, 2H), 7.5 (d, *J* = 2Hz, 1H), 8.65 (d, *J* = 6Hz, 1H), 8.68 (s, 1H). ¹³C NMR (100 MHz, DMSO- d_6) δ 32.4, 44.5, 120.5, 127.6, 140.3, 146.8, 147.8, 176.2. IV (cm⁻¹): 3398, 2357, 1647, 1217, 1020, 771, 667, 576. HRMS (EI) m/z [M+] calcd for C₈H₉N₂O⁺:149,0709; found: 149,0711. The obtained spectroscopic data are in accordance with those reported in the literature.¹

2,6-naphthyridin-3-ol (2)



Activated charcoal (640 mg) was added to a solution of **6.4** (6.4 g, 44 mmol) in xylene (200 mL) at 120 °C under an oxygen saturated atmosphere. The reaction was kept under stirring for 48 h. The reaction was filtered through celite and washed with MeOH. The solvent was removed under reduced pressure and the product was purified by flash chromatography (ethyl acetate— methanol, 9: 1) resulting in 2.89 g (45%) of a yellow solid. The obtained spectroscopic data are in accordance with those reported in the literature.¹

5-methoxy-1H-pyrazolo [3,4-c] pyridine (5)



4 (2 g, 17.3 mmol) was added to a round bottom flask equipped with a rubber septum. The flask was purged with N₂ and dry DCM (20 mL) was added. Acetic anhydride (1.7 mL, 19.4 mmol) was added and the reaction was stirred at room temperature. After 2 hours, NaNO₂ (7 g, 101 mmol) and acetic anhydride (6.5 mL, 69.4 mmol) were added to the reaction mixture. The reaction was kept under vigorous stirring at room temperature for 5 hours. Thereafter, DCM was removed under reduced pressure and the residue was suspended in toluene (50 mL), the reaction mixture was stirred for 1h at 110 °C. The solvent was removed under reduced pressure, and the residue was dissolved in 1 mL of concentrated aqueous HCl solution (20 mL) and stirred for 1h at 50 °C. The mixture was cooled to room temperature and basified with 1M NaOH_{aq} to pH 8-9, extracted with chloroform (4 x 40 mL), the organic phase was dried over MgSO₄ and the solvent was removed under reduced pressure. The product was purified by flash chromatography (Hexane-- ethyl acetate, 2 : 1) yielding 600 mg of **5** (48%) as an orange solid. The obtained spectroscopic data are in accordance with those reported in the literature.¹

General procedure for one-pot to 1H-indazole



o-toluidine (400µL, 3.72 mmol) was added to a round-bottom flask via syringe and fitted with a rubber septum. The flask was purged with nitrogen and dry DCM (10 mL) was added. Acetic anhydride (2 mL, 18.2 mmol) was added and the reaction was stirred at room temperature for 1 hour and monitored by LCMS. After the total consumption of the amine, NaNO₂ (136 mg, 26 mmol) and acetic anhydride (2 mL, 18.2 mmol) were added to the reaction mixture. The reaction was left under vigorous stirring at room temperature for 5 hours. After that, the DCM was removed under reduced pressure and the residue was suspended in toluene (20 mL), the reaction mixture was stirred for 1 h at 110 °C. The solvent was removed under reduced pressure, and the residue was solved in 1M HCl_{aq.} (20 mL) and stirred for 2h at 50 °C. The mixture was cooled to room temperature and basified with saturated aqueous solution NaHCO₃ to pH 8-9, then the mixture was extracted with chloroform (3x50mL). The organic phase was dried over MgSO₄ and the solvent removed under reduced pressure. The product was purified by Flash chromatography (Hexane - ethyl acetate, 2:1) to give 179 mg of indazole (81%, orange powder). The spectral data are consistent with those reported in the literature.⁶

1-tosyl-1H-pyrazole (17)



202 mg, yield 89% as yellowish solid. ¹H NMR (300 MHz, CDCl₃) δ 8.09 (d, *J* = 2.7 Hz, 1H), 7.87 (d, *J* = 8.4 Hz, 2H), 7.69 (d, *J* = 0.9 Hz, 1H), 7.30 (d, *J* = 8.0 Hz, 2H), 6.37 (dd, *J* = 2.7, 1.6 Hz, 1H), 2.38 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 145.9, 145.2, 133.9, 131.1, 130.1, 128.1, 108.8, 77.1, 21.7. The spectral data are consistent with those reported in the literature.⁷

1-tosyl-1H-indole (18)



178 mg, yield 81% as white solid. m.p. 94 °C. ¹H NMR (400 MHz, CDCl₃) δ 2.40 (s, 3H), 6.65 (d, J = 3.2 Hz, 1H), 7.33 – 7.28 (m, 2H), 7.21 (t, J = 6.8 Hz, 1H), 7.30 (t, J = 7.6 Hz, 1H), 7.51 (d, J = 7.6 Hz, 1H), 7.56 (d, J = 3.6 Hz, 1H), 7.88 (d, J = 8.5 Hz, 2H), 7.97 (d, J = 8.0 Hz, 1H). The spectral data are consistent with those reported in the literature.⁸

1-tosyl-1H-indazole (19)



242 mg, yield 89% as yellowish solid. m.p. 102 °C. ¹H NMR (300 MHz, CDCl₃) δ 8.23 (dt, *J* = 8.5, 0.9 Hz, 1H), 8.20 (d, *J* = 0.9 Hz, 1H), 7.92 – 7.86 (m, 2H), 7.71 (dt, *J* = 8.0, 1.1 Hz, 1H), 7.58 (ddd, *J* = 8.4, 7.1, 1.2 Hz, 1H), 7.34 (ddd, *J* = 8.0, 7.1, 0.9 Hz, 1H), 7.30 – 7.20 (m, 2H), 2.37 (s, 3H). ¹³C NMR (75 MHz, CDCl₃) δ 145.4, 141.3, 140.3, 134.6, 129.8, 129.2, 127.5, 125.8, 124.2, 121.3, 113.2, 21.6. The spectral data are consistent with those reported in the literature.⁹

REFERENCES:

1. Silva Junior, P. E.; Rezende, L. C. D.; Gimenes, J. P.; Maltarollo, V. G.; Dale, J.; Trossini, G. H. G.; Emery, F. S.; Ganesan, A., *Rsc Advances* **2016**, *6* (27), 22777-22780.

2. Krasnokutskaya, E. A.; Semenischeva, N. I.; Filimonov, V. D.; Knochel, P., *Synthesis-Stuttgart* **2007**, (1), 81-84.

3. Ke, J.; He, C.; Liu, H. Y.; Xu, H.; Lei, A. W. *Chemical Communications* **2013**, *49* (60), 6767-6769.

4. Davis, O. A.; Croft, R. A.; Bull, J. A., Chemical Communications 2015, 51 (84), 15446-15449.

5. Fumagalli, F.; Emery, F. D., *Journal of Organic Chemistry* **2016**, *81* (21), 10339-10347.

6. Kumar, M. R.; Park, A.; Park, N.; Lee, S., Organic Letters **2011**, *13* (13), 3542-3545.

7. Yotphan, S.; Sumunnee, L.; Beukeaw, D.; Buathongjan, C.; Reutrakul, V., Organic & Biomolecular Chemistry **2016**, *14* (2), 590-597.

8. Youn, S. W.; Lee, S. R.; Kim, Y. A.; Kang, D. Y.; Jang, M. J., *Chemistryselect* **2016**, *1* (18), 5749-5757.

9. Tang, M.; Kong, Y. F.; Chu, B. J.; Feng, D., *Advanced Synthesis & Catalysis* **2016**, *358* (6), 926-939.

3. NMR-SPECTRA





 $^{13}\mbox{C-NMR}$ – 101 MHz in \mbox{CDCl}_3 : 2,6-naphthyridin-3-yl trifluoromethanesulfonate (9)





¹H-NMR – 300 MHz in CDCl₃: 3 - ((3-methylbut-2-en-1-yl) oxy) -2,6-naphthyridine (**10**)



¹³C-NMR – 75 MHz in CDCl₃: 3-((3-methylbut-2-en-1-yl) oxy) -2,6-naphthyridine (10)

¹H NMR (300 MHz, CDCl₃) - 5-methoxy-1H-pyrazolo [3,4-c] pyridine (5)



¹³C NMR (75 MHz, CDCl₃) - 5-methoxy-1H-pyrazolo [3,4-c] pyridine (5)









¹H-NMR – 400 MHz in CDCl₃: 5-methoxy-1-tosyl-1H-pyrazolo[3,4-c]pyridine (**11**)



¹H-NMR – 400 MHz in DMSO-d₆: 2,6-naphthyridin-3-ol (2)



¹³C NMR - 101 MHz in DMSO-d₆: 2,6-naphthyridin-3-ol (2)



¹H-NMR – 300 MHz in DMSO-d₆: 4-(4-methoxyphenyl)-2,6-naphthyridin-3-ol (**12a**)





¹³C NMR - 101 MHz in DMSO-d6: 4-(4-methoxyphenyl)-2,6-naphthyridin-3-ol (12a)

¹H-NMR – 300 MHz in DMSO-d₆: 4-(4-(trifluoromethyl)phenyl)-2,6-naphthyridin-3-ol (**12b**)



110 100 f1 (ppm)



HSQC- 600 MHz in DMSO-d₆: 4-(4-(trifluoromethyl)phenyl)-2,6-naphthyridin-3-ol (**12b**)

HMBC- 600 MHz in DMSO-d₆: 4-(4-(trifluoromethyl)phenyl)-2,6-naphthyridin-3-ol (**12b**)



9.9 9.8 9.7 9.6 9.5 9.4 9.3 9.2 9.1 9.0 8.9 8.8 8.7 8.6 8.5 8.4 8.3 8.2 8.1 8.0 7.9 7.8 7.7 7.6 7.5 7.4 7.3 7.2 7.1 7.0 6.9 6.8 f2 (ppm)

¹H-NMR – 300 MHz in DMSO-d₆: 4-(3-hydroxy-2,6-naphthyridin-4-yl)benzonitrile (**12c**)



101 MHz in DMSO-d₆: 4-(3-hydroxy-2,6-naphthyridin-4-yl)benzonitrile (**12c**)



¹H-NMR – 300 MHz in DMSO-d₆: 4-(4-chlorophenyl)-2,6-naphthyridin-3-ol (**12d**)



¹³C NMR - 101 MHz in DMSO-d₆: 4-(4-chlorophenyl)-2,6-naphthyridin-3-ol (**12d**)





¹H-NMR – 600 MHz in DMSO-d₆: 4-(2,4-dichlorophenyl)-2,6-naphthyridin-3-ol (**12e**)



¹³C NMR - 150 MHz in DMSO-d₆: 4-(2,4-dichlorophenyl)-2,6-naphthyridin-3-ol (**12e**)



HSQC - 600 MHz in DMSO-d₆: 4-(2,4-dichlorophenyl)-2,6-naphthyridin-3-ol (**12e**)

HMBC - 600 MHz in DMSO-d₆: 4-(2,4-dichlorophenyl)-2,6-naphthyridin-3-ol (**12e**)





 1 H-NMR – 300 MHz in $MeO-d_{4}$: 5-methoxy-4-phenyl-1H-pyrazolo[3,4-c]pyridine (13)



$^1\text{H-NMR}$ – 75 MHz in $MeO\text{-}d_4\text{:}$ 5-methoxy-4-phenyl-1H-pyrazolo[3,4-c]pyridine (13)

¹H-NMR – 400 MHz in CDCl₃: 5-methoxy-1-(methylsulfonyl)-1H-pyrazolo[3,4-c]pyridine (15a)



¹³C NMR - 101 MHz in CDCl_{3:} 5-methoxy-1-(methylsulfonyl)-1H-pyrazolo[3,4-c]pyridine (15a)



¹H-NMR – 400 MHz in CDCl₃: 5-methoxy-1-(naphthalen-2-ylsulfonyl)-1H-pyrazolo[3,4-c]pyridine (**15b**)



¹³C NMR - 101 MHz in CDCl_{3:} 5-methoxy-1-(naphthalen-2-ylsulfonyl)-1H-pyrazolo[3,4-c]pyridine (**15b**)



¹H-NMR – 400 MHz in CDCl₃: 5-methoxy-1-(phenylsulfonyl)-1H-pyrazolo[3,4-c]pyridine (**15c**)



¹³C NMR - 101 MHz in CDCl_{3:} 5-methoxy-1-(phenylsulfonyl)-1H-pyrazolo[3,4-c]pyridine (15c)



¹H-NMR – 400 MHz in CDCl₃: 1-((4-bromophenyl)sulfonyl)-5-methoxy-1H-pyrazolo[3,4-c]pyridine (**15d**)







¹H-NMR – 400 MHz in CDCl₃: 5-methoxy-1-((2,4,6-triisopropylphenyl)sulfonyl)-1H-pyrazolo[3,4c]pyridine (**15e**)



¹³C NMR - 101 MHz in CDCl_{3:} 5-methoxy-1-((2,4,6-triisopropylphenyl)sulfonyl)-1H-pyrazolo[3,4c]pyridine (**15e**)







¹³C NMR - 101 MHz in CDCl_{3:} 5-methoxy-3-tosyl-1H-pyrazolo[3,4-c]pyridine (14)



¹H-NMR – 300 MHz in CDCl₃: 5-methoxy-3-(naphthalen-2-ylsulfonyl)-1H-pyrazolo[3,4-c]pyridine (16b)





¹³C NMR - 75 MHz in CDCl_{3:} 5-methoxy-3-(naphthalen-2-ylsulfonyl)-1H-pyrazolo[3,4-c]pyridine (16b)

¹H-NMR – 400 MHz in CDCl₃: 5-methoxy-3-(phenylsulfonyl)-1H-pyrazolo[3,4-c]pyridine (**16c**)



¹³C NMR - 101 MHz in CDCl_{3:} 5-methoxy-3-(phenylsulfonyl)-1H-pyrazolo[3,4-c]pyridine (16c)



¹H-NMR – 400 MHz in CDCl₃: 3-((4-bromophenyl)sulfonyl)-5-methoxy-1H-pyrazolo[3,4-c]pyridine (16d)



¹³C NMR - 101 MHz in CDCl_{3:} 3-((4-bromophenyl)sulfonyl)-5-methoxy-1H-pyrazolo[3,4-c]pyridine (16d)



¹H-NMR – 400 MHz in CDCl₃: 5-methoxy-3-((2,4,6-triisopropylphenyl)sulfonyl)-1H-pyrazolo[3,4c]pyridine (**16e**)



¹³C NMR - 101 MHz in CDCl_{3:} 5-methoxy-3-((2,4,6-triisopropylphenyl)sulfonyl)-1H-pyrazolo[3,4c]pyridine (**16e**)





^{""1}H-NMR – 400 MHz in CDCl₃: 1-tosyl-1H-pyrazole (17)



^{13}C NMR - 101 MHz in $\text{CDCl}_{3:}$ 1-tosyl-1H-pyrazole (17)

¹H-NMR – 400 MHz in CDCl₃: 1-tosyl-1H-indazole (19)



 $^{\rm 13}C$ NMR - 101 MHz in $\ \rm CDCl_{\rm 3:}$ 1-tosyl-1H-indazole (19)





¹H-NMR – 300 MHz in CDCl₃: 3-tosyl-1H-indazole (20)

^{13}C NMR - 75 MHz in $\text{CDCl}_{3:}$ 3-tosyl-1H-indazole (20)



¹H-NMR – 400 MHz in CD₃OD: 3-tosyl-1H-pyrazolo[3,4-c]pyridin-5-ol (**21**)



¹³C NMR - 101 MHz in CD₃OD: 3-tosyl-1H-pyrazolo[3,4-c]pyridin-5-ol (**21**)





HMBC- 400 MHz in CD₃OD: 3-tosyl-1H-pyrazolo[3,4-c]pyridin-5-ol (**21**)



1-ethyl-5-methoxy-3-(phenylsulfonyl)-1H-pyrazolo[3,4-c]pyridine ¹H NMR (300 MHz, CDCl₃) (22)





2-(5-methoxy-3-(phenylsulfonyl)-1H-pyrazolo[3,4-c]pyridin-1-yl)acetonitrile. ¹H NMR (300 MHz, CDCl₃) (**23**)



2-(5-methoxy-3-(phenylsulfonyl)-1H-pyrazolo[3,4-c]pyridin-1-yl)acetonitrile. ¹H ¹³C NMR (75 MHz, CDCl₃) (**23**)

