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Supporting Information

Metal-free PhI(OAc)₂-oxidized decarboxylation of propiolic acids towards synthesis of α-acetoxy ketones and insights into general decarboxylation with DFT calculations

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Content

1. Experimental Section	
2. Standard orientations of A, TS, B, C and CO ₂	
3. ¹ H and ¹³ C -NMR Analytical Date	
4. Copies of ¹ H- and ¹³ C-NMR spectra of products	
5. Mass spectrometry for mechanism study	
6. References	

1. Experimental Section

General Information.

The product was characterized via ¹H and ¹³C NMR using 500/126 MHz NMR spectrometer and 400/101 MHz NMR spectrometer at 20-25 °C, and CDCl₃ was used as the solvent. ¹H NMR spectra was reported in parts per million using tetramethyl silane ($\delta = 0.000$ ppm) or DMSO- d_6 (δ = 2.500 ppm) as an internal standard. The data of ¹H NMR are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, quin = quintet, and m = multiplet), coupling constants (J, Hz), and integration. ¹³C NMR spectra was reported in parts per million using solvent CDCl₃ (δ = 77.20 ppm) or DMSO- d_{δ} (δ = 39.50 ppm) as an internal standard. The data of ¹³C NMR are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet), and coupling constants (J, Hz). High-resolution mass spectroscopy (HRMS) spectra were obtained using a quadrupole time-of-flight (Q-TOF) MS spectrometer. Thin layer chromatography was performed on a glass plate coated with GF-254 silica gel and observed under 254 nm UV light. Reactions were monitored by TLC and column chromatography was performed using silica gel. Commercially available reagents are analytically pure and used without further purification unless otherwise specified. Acetic acid was HPLC grade and added with molecular sieves for more than two weeks. All solvents were dried with molecular sieves over two weeks. All reactions were performed at atmospheric pressure, and all reagents were weighing at room temperature in air.

Typical Procedure for the Synthesis of 1b-1x.¹

(i) To a solution of boronic acid (2.4 mmol, 1.2 equivalents), $Pd(OAc)_2$ (0.1 mmol, 5 mmol %), $Ag_2O(3.0 \text{ mmol}, 1.5 \text{ equivalents})$ and $K_2CO_3(4.0 \text{ mmol}, 2.0 \text{ equivalents})$ in ACN (6 mL) was added methyl propiolate (2.0 mmol) under argon. After stirring and reacting at 70°C for 12 hours, the solvent was removed by vacuum and the boronic acid was removed by the silica-gel column chromatography (petroleum ether / ethyl acetate = 10:1) to give a liquid or solid mixture. (ii) To the solution of mixture in MeOH (10 mL) was added 1 N NaOH (10 mL) and stirred overnight at room temperature. The MeOH was removed by vacuum, the aqueous phase was extracted with ethyl acetate (20 mL \times 3). Then added 2 N HCl until the pH reached 2, and extracted with ethyl acetate (20 mL \times 3). The organic layers were dried over anhydrous MgSO₄ and the solvents were removed in vacuo to give the pure solid propiolic acids in 34% to 73% yield.

General Procedure for the Synthesis of α-acyloxy ketones 2a-2za.

Propiolic acid 1 (0.50 mmol, 1.0 equivalents), H₂O (22 μ L, 1.25 mmol, 2.5 equivalents), 8 mL AcOH and PhI(OAc)₂ (483 mg, 1.5 mmol, 3.0 equivalents) were added in a round-bottom flask. The mixture was allowed to stir at 80 °C (oil bath temperature) under air for 1 h. After cooling to room temperature, the mixture was concentrated by rotary evaporation and the resulting residue was purified using a column chromatography (petroleum ether / ethyl acetate = 10:1) to give product **2**.

Investigation on hypervalent iodine reagents ^[a]

Ph	COOH $\frac{\text{hypervalent iodine reagent}}{\text{H}_2\text{O}, \text{AcOH}, T^{\circ}\text{C}, t \text{ h}}$	Ph O CH ₃ + 0 2a	Ph Ph O 3a
AcO-	-I-OAc F ₃ CCOO-I-OOCCF ₃		O O O H
Γ	DIB PIFA	PhIO	ABX
entry	hypervalent iodine reagent	yield of 2a (%) ^[b]	yield of 3a (%) ^[b]
1	PIFA (3.0 equiv)	3	-
2	PhIO (3.0 equiv)	23	trace
3	ABX (3.0 equiv)	18	2
4	DIB (3.0 equiv)	81	trace
5	DIB (2.0 equiv)	76	trace
6	DIB (2.5 equiv)	77	trace
7	DIB (2.8 equiv)	78	trace
8	DIB (3.5 equiv)	72	trace
9	DIB (3.3 equiv)	79	trace

[a] Reaction conditions: **1a** (0.5 mmol, 1.0 equivalent), hypervalent iodine reagents, H_2O (1.0 mmol, 2.0 equivalents) and acetic acid (2 mL) were refluxed at 80°C for 3 hours. [b] Yields were determined by ¹H NMR using mesitylene as a standard.

Investigation on water and solvent, temperature and time ^{[a],[b]}

	PhCOOH-	DIB	> r		CH ₃
	Fii — 000H	H ₂ O, AcOH, 1	<i>T</i> °C, <i>t</i> h	Ph [∕] 0	
	1a			2a	
entry	H ₂ O (equiv)	AcOH (mL)	<i>T</i> (°C)	<i>t</i> (h)	yield (%) [c]
1	2.0	2.0	80	3	81
2	2.0	2.0	70	3	73
3	2.0	2.0	75	3	77
4	2.0	2.0	85	3	79
5	2.0	2.0	90	3	76
6	1.5	2.0	80	3	78
7	2.5	2.0	80	3	83
8	3.0	2.0	80	3	80
9	2.5	1.0	80	3	76
10	2.5	4.0	80	3	84
11	2.5	6.0	80	3	87
12	2.5	8.0	80	3	96
13	2.5	9.0	80	3	95
14	2.5	8.0	80	5	96

entry	H ₂ O (equiv)	AcOH (mL)	<i>T</i> (°C)	<i>t</i> (h)	yield (%) ^[c]
15	2.5	8.0	80	4	95
16	2.5	8.0	80	2	96
17	2.5	8.0	80	1	97
18	2.5	8.0	80	0.5	79
19 ^[d]	2.5	8.0	80	1	96

[a] Reaction conditions: **1a** (0.5 mmol, 1.0 equivalent), DIB (1.5 mmol, 3.0 equivalents), H₂O (specified) and acetic acid (specified) were refluxed at $T \,^{\circ}C$ for *t* hours. [b] Using mesitylene as the standard, the yield of the by-product **3a** was determined by ¹H NMR as **trace**. [c] Yield of **2a** by ¹H NMR using mesitylene as an intnernal standard. [d] Argon atmosphere.

2. Standard orientations of A, TS, B, C and CO₂

Intermedia	ate A	,	, _ ~ , _ ,	
Zero-point	correctio	on=	0.255354 (Hartree/Particle)	
Thermal co	orrection	to Energy=		0.277572
Thermal co	orrection	to Enthalpy=		0.278516
Thermal co	orrection	to Gibbs Free Ene	ergy=	0.198222
Sum of ele	ctronic ai	nd zero-point Ene	rgies=	-967.603222
Sum of ele	ctronic ai	nd thermal Energi	es=	-967.581004
Sum of ele	ctronic ai	nd thermal Enthal	pies=	-967.580059
Sum of ele	ctronic ai	nd thermal Free E	nergies=	-967.660353
6	0	1.577476	-2.012478	-0.067759
6	0	2.618978	-1.380897	-0.057164
6	0	0.357031	-2.801554	-0.073283
8	0	-0.793776	-2.125384	-0.044433
8	0	0.369880	-4.020968	-0.098962
53	0	-0.824409	0.083323	0.013840
8	0	-1.377667	2.255379	0.081974
6	0	-0.276370	2.974506	0.142300
8	0	0.850242	2.474758	0.147347
6	0	-0.523574	4.469210	0.199565
1	0	-1.198407	4.705258	1.027061
1	0	0.424632	4.992840	0.324202
1	0	-1.008188	4.796455	-0.725591
6	0	-2.979275	-0.210537	-0.013775
6	0	-3.496781	-1.304491	0.670000
6	0	-3.764682	0.692161	-0.721775
6	0	-4.882667	-1.496534	0.635757
1	0	-2.850129	-2.006154	1.181779
6	0	-5.148122	0.481818	-0.734477
1	0	-3.327729	1.542521	-1.229561
6	0	-5.705277	-0.607566	-0.060082
1	0	-5.309340	-2.348100	1.157025
1	0	-5.782896	1.173996	-1.279603

1	0	-6.779459	-0.765153	-0.079461
6	0	3.855975	-0.668784	-0.045055
6	0	5.072526	-1.381087	-0.099198
6	0	3.875761	0.740102	0.021817
6	0	6.284851	-0.694220	-0.086885
1	0	5.051071	-2.465010	-0.150156
6	0	5.096314	1.414541	0.033919
1	0	2.943148	1.293591	0.064496
6	0	6.299724	0.703538	-0.020190
1	0	7.217921	-1.248442	-0.129010
1	0	5.105789	2.499292	0.085904
1	0	7.246485	1.236168	-0.010325

TS Frec

15				
Frequenc	ies16	57.1923		
Zero-poir	nt correction	1=	0.253040 (Hartree/Particle)	
Thermal	correction to	o Energy=		0.275263
Thermal	correction to	o Enthalpy=		0.276207
Thermal	correction to	o Gibbs Free En	ergy=	0.194051
Sum of el	lectronic an	d zero-point Ene	ergies=	-967.562493
Sum of el	lectronic an	d thermal Energi	ies=	-967.540270
Sum of el	lectronic an	d thermal Enthal	pies=	-967.539326
Sum of el	lectronic an	d thermal Free E	inergies=	-967.621482
6	0	-1.165037	-1.317159	-1.351982
6	0	-2.123311	-0.887115	-0.698409
6	0	-0.293646	-2.146820	-2.319228
8	0	0.899345	-1.746410	-2.390245
8	0	-0.893382	-3.073719	-2.865994
53	0	0.683458	0.533371	-0.516739
8	0	2.128062	1.998478	0.280322
6	0	1.807137	3.192498	-0.191872
8	0	0.839801	3.383415	-0.921805
6	0	2.770797	4.281532	0.239696
1	0	3.655683	4.235837	-0.403615
1	0	2.289323	5.252364	0.117427
1	0	3.093917	4.133417	1.271828
6	0	1.749781	-0.872535	0.718831
6	0	2.504371	-1.870349	0.110326
6	0	1.647332	-0.715884	2.100636
6	0	3.201394	-2.749753	0.946989
1	0	2.501744	-1.985341	-0.968422
6	0	2.359538	-1.602656	2.912244
1	0	1.046497	0.073032	2.538555
6	0	3.132989	-2.615892	2.336128
1	0	3.792094	-3.543656	0.500307

1	0	2.301871	-1.500793	3.991771
1	0	3.677258	-3.306712	2.973261
6	0	-3.182209	-0.353850	0.074665
6	0	-3.502412	-0.918764	1.331563
6	0	-3.943166	0.736656	-0.407658
6	0	-4.552856	-0.399362	2.082542
1	0	-2.926092	-1.763866	1.693989
6	0	-4.990133	1.247721	0.354031
1	0	-3.702590	1.163849	-1.375782
6	0	-5.297227	0.683379	1.598123
1	0	-4.796647	-0.840041	3.044598
1	0	-5.570244	2.084927	-0.022092
1	0	-6.116732	1.084317	2.187354

Intermediate B

Intermediate	B			
Zero-point con	rrection=	0.253783 (Hartree/Particle)		
Thermal corre	ection to E	Energy=		0.277609
Thermal corre	ection to E	enthalpy=		0.278554
Thermal corre	ection to C	Jibbs Free Ene	rgy=	0.189822
Sum of electro	onic and z	ero-point Ener	gies=	-967.612957
Sum of electro	onic and t	hermal Energie	es=	-967.589130
Sum of electro	onic and t	hermal Enthalp	oies=	-967.588186
Sum of electro	onic and t	hermal Free Er	nergies=	-967.676917
6	0	-0.979256	-0.628373	-0.210041
6	0	-2.198579	-0.662239	-0.251950
6	0	-1.105566	1.780524	2.793800
8	0	0.042327	1.559753	2.788802
8	0	-2.253600	2.007281	2.805641
53	0	1.149646	-0.865255	-0.093063
8	0	3.503475	-0.763449	-0.066937
6	0	3.935643	-1.941683	0.294887
8	0	3.185974	-2.892740	0.558695
6	0	5.448867	-2.068097	0.376934
1	0	5.726116	-3.097670	0.607881
1	0	5.902348	-1.758762	-0.569627
1	0	5.831338	-1.400836	1.156551
6	0	1.399996	1.216891	-0.734240
6	0	0.480096	1.747422	-1.629378
6	0	2.472844	1.937838	-0.225254
6	0	0.646562	3.078544	-2.029826
1	0	-0.352038	1.160506	-2.001034
6	0	2.617023	3.268025	-0.637895
1	0	3.185120	1.478375	0.447228
6	0	1.710141	3.837310	-1.535185
1	0	-0.060031	3.512279	-2.731359

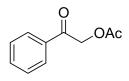
1	0	3.446502	3.852426	-0.250773
1	0	1.832658	4.869395	-1.849946
6	0	-3.626542	-0.710761	-0.315960
6	0	-4.265983	-1.570325	-1.232676
6	0	-4.412944	0.094440	0.532822
6	0	-5.657722	-1.621444	-1.294951
1	0	-3.662737	-2.192892	-1.885836
6	0	-5.804400	0.038891	0.460377
1	0	-3.926049	0.756202	1.241242
6	0	-6.430970	-0.817452	-0.450675
1	0	-6.139512	-2.289443	-2.003026
1	0	-6.400490	0.664111	1.118805
1	0	-7.515123	-0.859022	-0.502080

Intermediate C

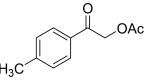
Zero-point correction= 0.241680 (Hartree/Particle) 0.259114 Thermal correction to Energy= Thermal correction to Enthalpy= 0.260058 Thermal correction to Gibbs Free Energy= 0.191860 Sum of electronic and zero-point Energies= -779.035768 Sum of electronic and thermal Energies= -779.018333 Sum of electronic and thermal Enthalpies= -779.017389 Sum of electronic and thermal Free Energies= -779.0855886 0 1.124781 -0.353442 -0.000022 6 0 2.345019 -0.362433 -0.000003 53 0 -0.985693 -0.668191 -0.000100 8 0 -0.801004 -0.000316 -3.353081 6 0 -3.666026 -2.069580 0.000149 0 8 -2.981436 -2.827281 0.000631 6 0 -5.159833 -2.356486 0.000060 0 1 -5.624847 -1.904330 0.882006 0 1 -5.624581 -1.905120-0.882432 1 0 -5.333696 -3.433507 0.000512 6 0 -1.346306 1.523910 -0.000052 6 0 -0.247079 2.371267 -0.000204 6 0 -2.660845 1.976129 0.000131 6 0 -0.483264 3.751761 -0.0001741 0 0.765515 1.984898 -0.000335 6 0 3.361877 0.000163 -2.867838 0 1 -3.491727 1.283186 0.000225 6 0 -1.788418 4.247616 0.000012 1 0 0.364719 4.430481 -0.000295 1 0 -3.886859 3.737749 0.000306 0 1 -1.962940 5.319445 0.000038 6 0 3.773945 -0.378446 0.000026

6	0	4.488163	-0.388485	1.215505
6	0	4.488206	-0.389200	-1.215421
6	0	5.882098	-0.407950	1.210871
1	0	3.940284	-0.383633	2.152417
6	0	5.882141	-0.408661	-1.210726
1	0	3.940361	-0.384900	-2.152355
6	0	6.582649	-0.418003	0.000088
1	0	6.422647	-0.417198	2.152847
1	0	6.422723	-0.418461	-2.152678
1	0	7.668661	-0.434223	0.000112
CO ₂				
Zero-poi	int correction	=	0.011565 (Hartree/Particle)	
Thermal	correction to	Energy=	0.014202	
Thermal	correction to	0.015146		
Thermal	correction to	Gibbs Free Ene	ergy=	-0.009149
Sum of e	electronic and	l zero-point Ener	rgies=	-188.578828
Sum of e	electronic and	l thermal Energi	es=	-188.576191
Sum of e	electronic and	l thermal Enthal	-188.575246	
Sum of e	electronic and	l thermal Free E	-188.599542	
6	0	0.000000	0.000000	0.000000
8	0	0.000000	0.000000	1.169356
8	0	0.000000	0.000000	-1.169356
	Ū	0.000000		

3. ¹H and ¹³C -NMR Analytical Date

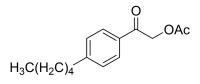


2-Oxo-2-phenylethyl Acetate (2a). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 84 mg (94%) of 2a. Colorless oil. ¹H NMR (500 MHz, CDCl₃) δ 7.90 (d, J = 7.5 Hz, 2H), 7.59 (t, J = 7.5 Hz, 1H), 7.47 (t, J = 7.0 Hz, 2H), 5.33 (s, 2H), 2.21 (s, 3H). ¹³C {H} NMR (126 MHz, CDCl₃) δ 192.3, 170.5, 134.3, 133.9, 128.9, 127.8, 66.1, 20.6. For the high-resolution mass spectrometry data please refer to [2].

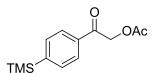


2-Oxo-2-(*p*-tolyl)ethyl Acetate (**2b**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 44 mg (46%) of **2b**.

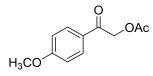
Colorless oil. ¹H NMR (500 MHz, CDCl₃) δ 7.81 (d, J = 7.5 Hz, 2H), 7.28 (d, J = 8.0 Hz, 2H), 5.32 (s, 2H), 2.42 (s, 3H), 2.23 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 191.9, 170.6, 145.0, 131.9, 129.7, 128.0, 66.1, 21.9, 20.7. For the high-resolution mass spectrometry data please refer to [2].



2-Oxo-2-(4-pentylphenyl)ethyl Acetate (**2c**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 104 mg (84%) of **2c**. Colorless oil. ¹H NMR (500 MHz, CDCl₃) δ 7.83 (d, *J* = 7.5 Hz, 2H), 7.28 (d, *J* = 7.5 Hz, 2H), 5.32 (s, 2H), 2.66 (t, *J* = 7.5 Hz, 2H), 2.22 (s, 3H), 1.66 – 1.60 (m, 2H), 1.36 – 1.30 (m, 4H), 0.89 (t, *J* = 6.5 Hz, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 191.9, 170.5, 149.9, 132.1, 129.0, 128.0, 66.1, 36.1, 31.5, 30.8, 22.6, 20.7, 14.1. HRMS (ESI-TOF) *m/z*: calcd for C₁₅H₂₀O₃ [M + H]⁺ 249.1486, found 249.1491.

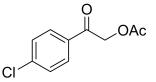


2-Oxo-2-(4-(trimethylsilyl)phenyl)ethyl Acetate (**2d**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 59 mg (47%) of **2d**. Colorless oil. ¹H NMR (500 MHz, CDCl₃) δ 7.86 (d, *J* = 7.5 Hz, 2H), 7.64 (d, *J* = 7.0 Hz, 2H), 5.33 (s, 2H), 2.22 (s, 3H), 0.29 (s, 9H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 192.5, 170.5, 148.6, 134.4, 133.9, 126.8, 66.2, 20.7, -1.3. HRMS (ESI-TOF) *m/z*: calcd for C₁₃H₁₈O₃Si [M + H]⁺251.1098, found 251.1104.

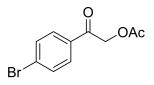


2-(4-Methoxyphenyl)-2-oxoethyl Acetate (2e). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 58 mg (56%) of 2e. Lightyellow oil. ¹H NMR (500 MHz, CDCl₃) δ 7.88 (d, *J* = 7.5 Hz, 2H), 6.94 (d, *J* = 8.0 Hz, 2H), 5.29 (s, 2H), 3.86 (s, 3H), 2.21 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 190.7, 170.5, 164.1, 130.1, 127.3, 114.1, 65.8, 55.6, 20.6. For the

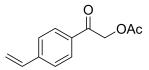
high-resolution mass spectrometry data please refer to [2].



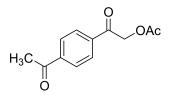
2-(4-Chlorophenyl)-2-oxoethyl Acetate (**2f**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 51 mg (48%) of **2f**. White solid, mp: 57–60 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.85 (d, *J* = 7.5 Hz, 2H), 7.46 (d, *J* = 7.5 Hz, 2H), 5.29 (s, 2H), 2.22 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 191.3, 170.5, 140.6, 132.7, 129.4, 129.3, 66.0, 20.7. HRMS (ESI-TOF) *m/z*: calcd for C₁₀H₉ClO₃ [M + H]⁺ 213.0313, found 213.0308.



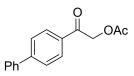
2-(4-Bromophenyl)-2-oxoethyl Acetate (**2g**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 39 mg (30%) of **2g**. White solid, mp: 77–79 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.78 (d, *J* = 7.5 Hz, 2H), 7.64 (d, *J* = 7.5 Hz, 2H), 5.29 (s, 2H), 2.23 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 191.5, 170.5, 133.1, 132.4, 129.4, 129.3, 66.0, 20.7. HRMS (ESI-TOF) *m/z*: calcd for C₁₀H₉BrO₃ [M + H]⁺ 256.9808, found 256.9802.



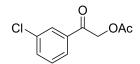
2-Oxo-2-(4-vinylphenyl)ethyl Acetate (**2h**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 85 mg (83%) of **2h**. Colorless oil. ¹H NMR (500 MHz, CDCl₃) δ 7.88 (d, *J* = 8.0 Hz, 2H), 7.50 (d, *J* = 8.5 Hz, 2H), 6.75 (dd, *J* = 17.5, 10.5 Hz, 1H), 5.89 (d, *J* = 17.5 Hz, 1H), 5.43 (d, *J* = 11.0 Hz, 1H), 5.33 (s, 2H), 2.24 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 191.7, 170.6, 143.0, 135.9, 133.4, 128.3, 126.7, 117.5, 66.2, 20.8. HRMS (ESI-TOF) *m/z*: calcd for C₁₂H₁₂O₃ [M + H]⁺ 205.0860, found 205.0854.



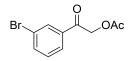
2-(4-Acetylphenyl)-2-oxoethyl Acetate (**2i**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 50 mg (45%) of **2i**. White solid, mp: 65–68 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.05 (d, *J* = 7.5 Hz, 2H), 7.99 (d, *J* = 7.5 Hz, 2H), 5.34 (s, 2H), 2.65 (s, 3H), 2.34 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 197.4, 192.0, 170.5, 141.0, 137.5, 128.9, 128.2, 66.3, 27.0, 20.7. HRMS (ESI-TOF) *m/z*: calcd for C₁₃H₁₄O₃ [M + H]⁺221.0809, found 221.0801.



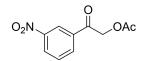
2-([1,1'-Biphenyl]-4-yl)-2-oxoethyl Acetate (**2j**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 103 mg (81%) of **2j**. White solid, mp: 102–105 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.99 (d, *J* = 7.5 Hz, 2H), 7.71 (d, *J* = 7.5 Hz, 2H), 7.62 (d, *J* = 7.5 Hz, 2H), 7.48 (t, *J* = 7.0 Hz, 2H), 7.41 (t, *J* = 7.5 Hz, 1H), 5.337 (s, 2H), 2.25 (s, 3H). ¹³C {H} NMR (126 MHz, CDCl₃) δ 192.0, 170.6, 146.8, 139.8, 133.1, 129.2, 128.63, 128.55, 127.7, 127.5, 66.2, 20.8. HRMS (ESI-TOF) *m/z*: calcd for C₁₆H₁₄O₃ [M + H]⁺ 255.1016, found 255.1013.



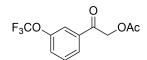
2-(3-Chlorophenyl)-2-oxoethyl Acetate (**2k**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 84 mg (79%) of **2k**. Khaki oil. ¹H NMR (500 MHz, CDCl₃) δ 7.89 (s, 1H), 7.79 (d, *J* = 7.5 Hz, 1H), 7.58 (d, *J* = 7.5 Hz, 1H), 7.44 (t, *J* = 7.5 Hz, 1H), 5.30 (s, 2H), 2.23 (s, 3H). ¹³C {H} NMR (126 MHz, CDCl₃) δ 191.3, 170.5, 135.8, 135.4, 134.0, 130.4, 128.1, 126.0, 66.1, 20.7. HRMS (ESI-TOF) *m/z*: calcd for C₁₀H₉ClO₃ [M + H]⁺ 213.0313, found 213.0315.



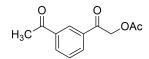
2-(3-Bromophenyl)-2-oxoethyl Acetate (**2l**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 75 mg (58%) of **2l**. Khaki oil. ¹H NMR (500 MHz, CDCl₃) δ 8.05 (s, 1H), 7.83 (d, *J* = 7.5 Hz, 1H), 7.74 (d, *J* = 8.0 Hz, 1H), 7.38 (t, *J* = 7.5 Hz, 1H), 5.30 (s, 2H), 2.24 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 191.2, 170.5, 136.9, 136.0, 131.0, 130.6, 126.4, 123.4, 66.1, 20.7. For the high-resolution mass spectrometry data please refer to [2].



2-(3-Nitrophenyl)-2-oxoethyl Acetate (**2m**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 26 mg (23%) of **2m**. Gold oil. ¹H NMR (500 MHz, CDCl₃) δ 8.75 (s, 1H), 8.48 (d, *J* = 8.5 Hz, 1H), 8.26 (d, *J* = 7.5 Hz, 1H), 7.74 (t, *J* = 8.5 Hz, 1H), 5.37 (s, 2H), 2.25 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 190.6, 170.5, 148.6, 135.5, 133.5, 130.5, 128.3, 122.9, 66.1, 20.7. For the mass spectrometry data please refer to [3].

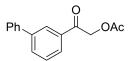


2-Oxo-2-(3-(trifluoromethoxy)phenyl)ethyl Acetate (**2n**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 52 mg (40%) of **2n**. Lightgoldenrodyellow solid, mp: 43–45 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.84 (d, *J* = 7.5 Hz, 1H), 7.77 (s, 1H), 7.55 (t, *J* = 7.5 Hz, 1H), 7.47 (d, *J* = 8.0 Hz, 1H), 5.31 (s, 2H), 2.23 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 191.1, 170.5, 149.8, 136.1, 130.7, 126.4, 126.2, 120.5 (d, *J* = 259.6 Hz), 120.4, 66.1, 20.6. HRMS (ESI-TOF) *m/z*: calcd for C₁₁H₉F₃O₄ [M + Na]⁺285.0346, found 285.0345.

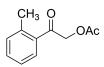


2-(3-Acetylphenyl)-2-oxoethyl Acetate (**20**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 74 mg (67%) of **20**. Colorless oil. ¹H NMR (500 MHz, CDCl₃) δ 8.48 (s, 1H), 8.20 (d, *J* = 7.5 Hz, 1H),

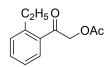
8.12 (d, J = 8.0 Hz, 1H), 7.63 (t, J = 8.0 Hz, 1H), 5.38 (s, 2H), 2.67 (s, 3H), 2.25 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 197.1, 191.9, 170.6, 137.7, 134.7, 133.4, 132.1, 129.6, 127.6, 66.2, 26.9, 20.7. HRMS (ESI-TOF) *m/z*: calcd for C₁₃H₁₄O₃ [M + H]⁺221.0809, found 221.0812.



2-([1,1'-Biphenyl]-3-yl)-2-oxoethyl Acetate (**2p**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 95 mg (75%) of **2p**. White solid, mp: 88–91 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.13 (s, 1H), 7.88 (d, *J* = 7.5 Hz, 1H), 7.83 (d, *J* = 7.5 Hz, 1H), 7.60 (d, *J* = 7.0 Hz, 2H), 7.56 (t, *J* = 8.0 Hz, 1H), 7.48 (t, *J* = 7.5 Hz, 2H), 7.40 (t, *J* = 7.5 Hz, 1H), 5.39 (s, 2H), 2.25 (s, 3H). ¹³C {H} NMR (126 MHz, CDCl₃) δ 192.3, 170.6, 142.3, 140.0, 134.9, 132.7, 129.5, 129.2, 128.2, 127.4, 126.7, 126.6, 66.3, 20.8. HRMS (ESI-TOF) *m/z*: calcd for C₁₆H₁₄O₃ [M + H]⁺255.1016, found 255.1008.

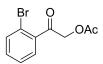


2-Oxo-2-(*o*-tolyl)ethyl Acetate (**2q**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 40 mg (41%) of **2q**. Colorless oil. ¹H NMR (500 MHz, CDCl₃) δ 7.61 (d, *J* = 8.0 Hz, 1H), 7.42 (t, *J* = 7.5 Hz, 1H), 7.28 (d, *J* = 7.5 Hz, 2H), 5.18 (s, 2H), 2.52 (s, 3H), 2.21 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 195.9, 170.6, 139.2, 134.4, 132.4, 132.3, 128.2, 125.9, 67.4, 21.3, 20.7. For the mass spectrometry data please refer to [4].

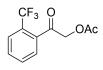


2-(2-Ethylphenyl)-2-oxoethyl Acetate (**2r**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 64 mg (62%) of **2r**. Lightgoldenrodyellow oil. ¹H NMR (500 MHz, CDCl₃) δ 7.56 (d, *J* = 7.5 Hz, 1H), 7.45 (t, *J* = 14.5, 7.5 Hz, 1H), 7.32 (d, *J* = 8.0 Hz, 1H), 7.27 (t, *J* = 7.5 Hz, 1H), 5.15 (s, 2H), 2.84 (q, *J* = 7.5 Hz, 2H), 2.21 (s, 3H), 1.22 (t, *J* = 7.5 Hz, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 196.4, 170.5, 145.0, 134.6, 132.3, 130.8, 128.0, 125.8, 67.6, 27.0,

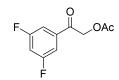
20.7, 16.0. HRMS (ESI-TOF) m/z: calcd for C₁₂H₁₄O₃ [M + H]⁺ 207.1016, found 207.1019.



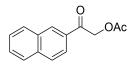
2-(2-Bromophenyl)-2-oxoethyl Acetate (**2s**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 61 mg (47%) of **2s**. Lightyellow oil. ¹H NMR (500 MHz, CDCl₃) δ 7.63 (d, *J* = 8.0 Hz, 1H), 7.50 (d, *J* = 8.0 Hz, 1H), 7.41 (t, *J* = 7.5 Hz, 1H), 7.36 (t, *J* = 7.5 Hz, 1H), 5.16 (s, 2H), 2.18 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 196.6, 170.5, 138.4, 134.0, 132.7, 129.5, 127.7, 119.3, 67.8, 20.6. HRMS (ESI-TOF) *m/z*: calcd for C₁₀H₉BrO₃ [M + H]⁺ 256.9808, found 256.9808.



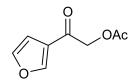
2-Oxo-2-(2-(trifluoromethyl)phenyl)ethyl Acetate (**2t**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 86 mg (70%) of **2t**. Lightyellow oil. ¹H NMR (400 MHz, CDCl₃) δ 7.75 (dd, *J* = 7.2, 2.0 Hz, 1H), 7.64 (td, *J* = 6.8, 2.4, 2H), 7.57 – 7.55(m, 1H), 5.04 (s, 2H), 2.18 (s, 3H). ¹³C {H} NMR (101 MHz, CDCl₃) δ 197.3, 170.4, 136.6, 132.1, 131.1, 127.8, 127.0 (q, *J* = 4.6 Hz), 123.5 (q, *J* = 275.7), 67.9, 20.5. For the high-resolution mass spectrometry data please refer to [4].



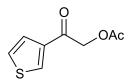
2-(3,5-Difluorophenyl)-2-oxoethyl Acetate (**2u**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 23 mg (21%) of **2u**. White solid, mp: 79–81 °C. ¹H NMR (500 MHz, CDCl₃) δ 7.42 (d, *J* = 4.0 Hz, 2H), 7.07 (t, *J* = 8.0 Hz, 1H), 5.25 (s, 2H), 2.23 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 190.2, 170.4, 163.4 (dd, *J* = 252, 11.3 Hz), 137.1 (t, *J* = 7.6 Hz), 111.1 (dd, *J* = 20.2, 6.3 Hz), 109.4 (t, *J* = 25.2 Hz), 66.0, 20.6. HRMS (ESI-TOF) *m/z*: calcd for C₁₀H₈F₂O₃ [M + H]⁺215.0515, found 215.0510.



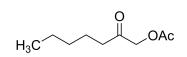
2-(Naphthalen-2-yl)-2-oxoethyl Acetate (**2v**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 74 mg (65%) of **2v**. Lightyellow solid, mp: 77–79 °C. ¹H NMR (500 MHz, CDCl₃) δ 8.41 (s, 1H), 7.95 (t, *J* = 8.5 Hz, 2H), 7.88 (dd, *J* = 14.5, 8.5 Hz, 2H), 7.62 (t, *J* = 7.5 Hz, 1H), 7.56 (t, *J* = 7.0 Hz, 1H), 5.47 (s, 2H), 2.26 (s, 3H). ¹³C {H} NMR (126 MHz, CDCl₃) δ 192.2, 170.7, 136.0, 132.5, 131.6, 129.7, 129.6, 129.02, 128.96, 128.0, 127.2, 123.4, 66.2, 20.8. For the high-resolution mass spectrometry data please refer to [2].



2-(Furan-3-yl)-2-oxoethyl Acetate (**2w**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 56 mg (66%) of **2w**. Lightyellow oil. ¹H NMR (500 MHz, CDCl₃) δ 8.12 (s, 1H), 7.49 (s, 1H), 6.79 (s, 1H), 5.04 (s, 2H), 2.21 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 187.7, 170.5, 147.3, 144.5, 124.5, 108.5, 66.6, 20.7. HRMS (ESI-TOF) *m/z*: calcd for C₈H₈O₄ [M + H]⁺ 169.0496, found 169.0501.



2-Oxo-2-(thiophen-3-yl)ethyl Acetate (**2x**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 10 / 1) to afford 58 mg (63%) of **2x**. Goldenrod oil. ¹H NMR (500 MHz, CDCl₃) δ 8.12 (s, 1H), 7.59 (d, *J* = 4.0 Hz, 1H), 7.37 (t, *J* = 3.0 Hz, 1H), 5.21 (s, 2H), 2.22 (s, 3H). ¹³C{H} NMR (126 MHz, CDCl₃) δ 186.9, 170.6, 138.8, 132.4, 127.0, 126.6, 66.4, 20.7. HRMS (ESI-TOF) *m/z*: calcd for C₈H₈O₃S [M + H]⁺ 185.0267, found 185.0270.



2-Oxoheptyl Acetate (2y). Purification was performed by column chromatography

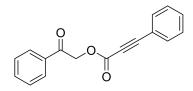
(petroleum ether / ethyl acetate = 4 / 1) to afford 34 mg (40%) of **2y**. White solid, mp: 17– 19 °C. ¹H NMR (400 MHz, DMSO-*d*₆) δ 4.73 (s, 2H), 2.40 (t, *J* = 7.2 Hz, 2H), 2.08 (s, 3H), 1.51 – 1.43 (m, 2H), 1.29 – 1.19 (m, 4H), 0.85 (t, *J* = 7.2 Hz, 3H). ¹³C {H} NMR (101 MHz, DMSO-*d*₆) δ 202.4, 172.6, 67.8, 37.7, 30.7, 22.4, 21.4, 20.2, 13.8. HRMS (ESI-TOF) *m/z*: calcd for C₉H₁₆O₃ [M + Na]⁺ 195.0992, found 195.0992.



2-Oxopropyl Acetate (**2z**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 20 / 1) to afford 52 mg (45%) of **2z**. Amber liquid. ¹H NMR (400 MHz, DMSO-*d*₆) δ 4.71 (s, 2H), 2.18 (s, 3H), 2.03 (s, 3H). For the ¹³C{H} NMR data please refer to [4], and for the high-resolution mass spectrometry data please refer to [8].



2-Oxobutyl Acetate (**2za**). Purification was performed by column chromatography (petroleum ether / ethyl acetate = 20 / 1) to afford 17 mg (26%) of **2za**. Amber liquid. ¹H NMR (400 MHz, CDCl₃) δ 4.66 (s, 2H), 2.48-2.42 (q, J = 7.2 Hz, 2H), 2.17 (s, 3H), 1.10 (t, J = 7.4 Hz, 3H).



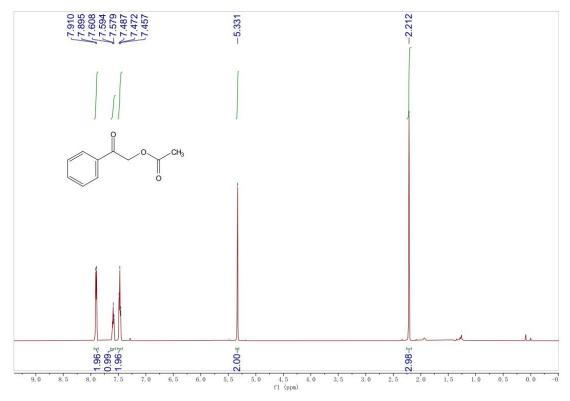
2-Oxo-2-phenylethyl 3-phenylpropiolate (**3a**). Purification was performed by column chromatography (petroleum ether/ ethyl acetate = 10/1) to afford 17 mg (35%) of **3a**. White solid, mp: 83–85 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.94 (d, *J* = 7.2 Hz, 2H), 7.65 – 7.61 (m, 3H), 7.53 – 7.45 (m, 3H), 7.40 (t, *J* = 7.6 Hz, 2H), 5.50 (s, 2H). ¹³C{H} NMR (101 MHz, CDCl₃) δ 191.1, 153.5, 134.3, 134.1, 133.3, 131.1, 129.1, 128.8, 128.0, 119.6, 88.2, 80.1, 67.2. For the high-resolution mass spectrometry data please refer to [5].



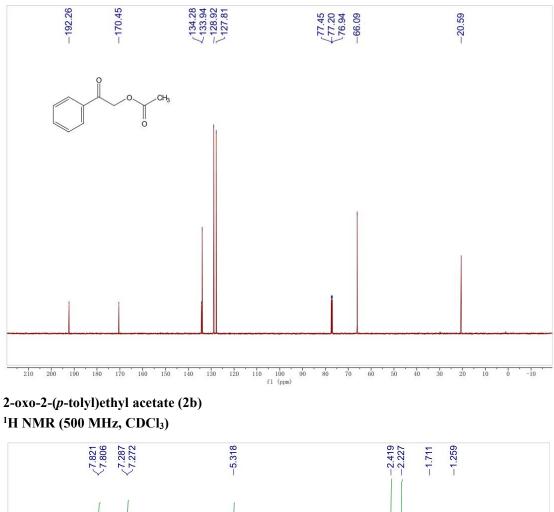
2-Hydroxy-1-phenylethan-1-one (4). Purification was performed by column S16

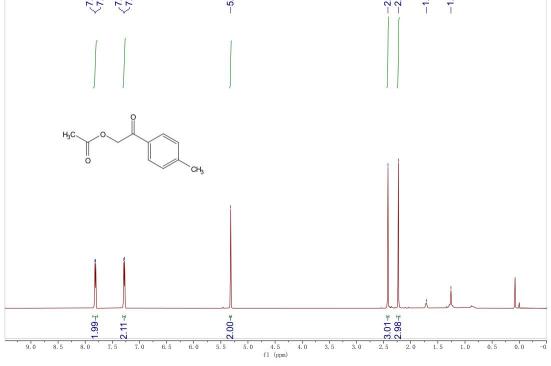
chromatography (petroleum ether/ ethyl acetate = 10/1) to afford 7 mg (10%) of 4. White solid, mp: 79–81 °C. ¹H NMR (400 MHz, CDCl₃) δ 7.93 (d, *J* = 7.6 Hz, 2H), 7.64 (t, *J* = 7.2 Hz, 1H), 7.51 (t, *J* = 8.0 Hz, 2H), 4.89 (s, 2H), 3.52 (s, 1H). ¹³C{H} NMR (101 MHz, CDCl₃) δ 198.6, 134.5, 133.5, 129.2, 127.9, 65.6. For the high-resolution mass spectrometry data please refer to [6].

4. Copies of ¹H- and ¹³C-NMR spectra of products 2-oxo-2-phenylethyl acetate (2a) ¹H NMR (500 MHz, CDCl₃)

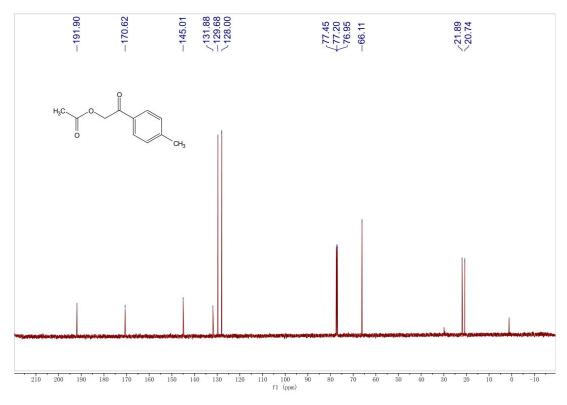


¹³C {¹H} NMR (126 MHz, CDCl₃)

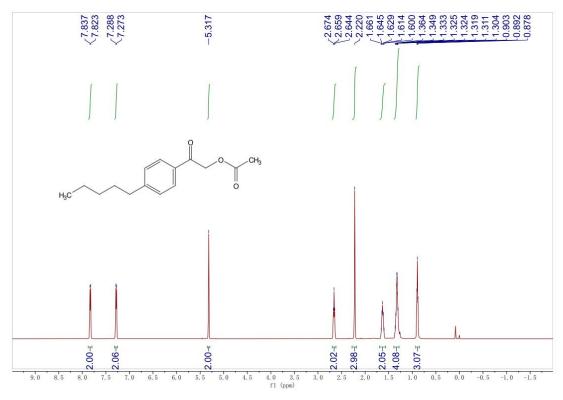




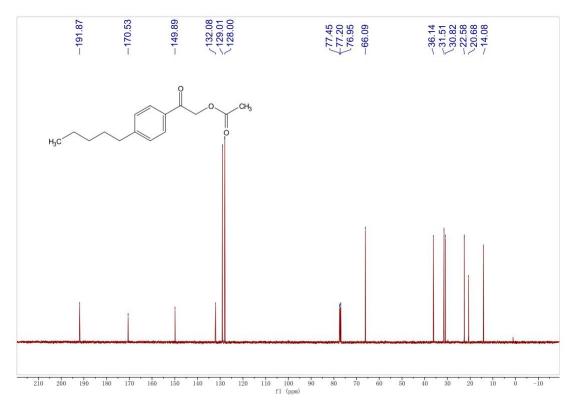
¹³C {¹H} NMR (126 MHz, CDCl₃)



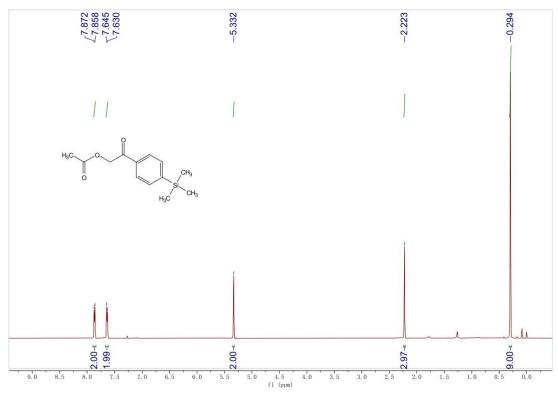
2-oxo-2-(4-pentylphenyl)ethyl acetate (2c) ¹H NMR (500 MHz, CDCl₃)



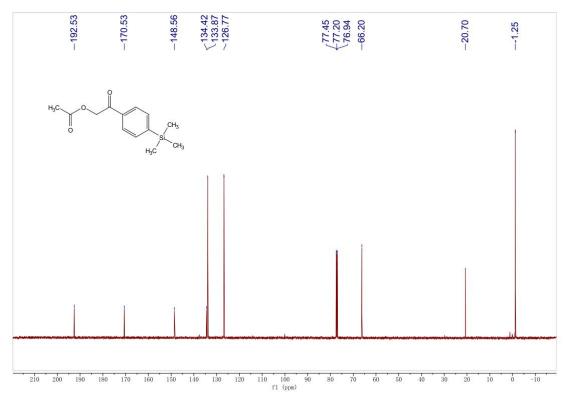
¹³C {¹H} NMR (126 MHz, CDCl₃)



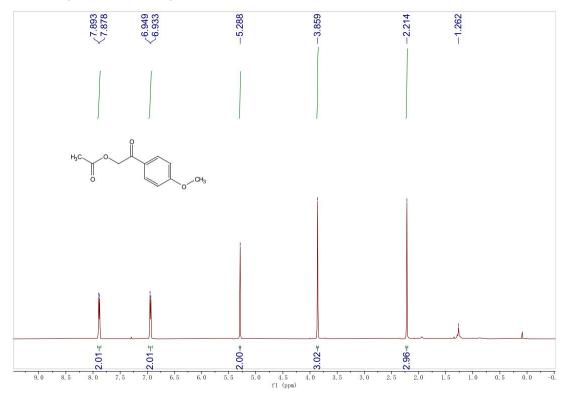
2-oxo-2-(4-(trimethylsilyl)phenyl)ethyl acetate (2d) ¹H NMR (500 MHz, CDCl₃)



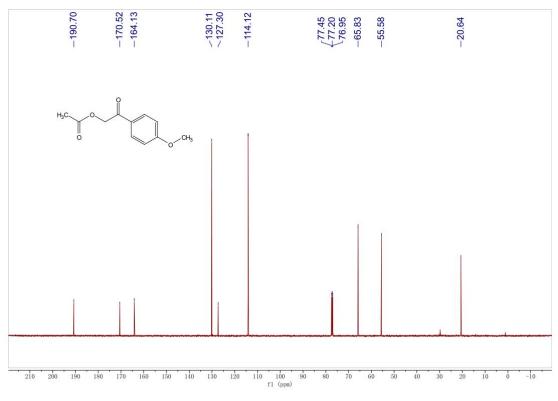
¹³C {¹H} NMR (126 MHz, CDCl₃)



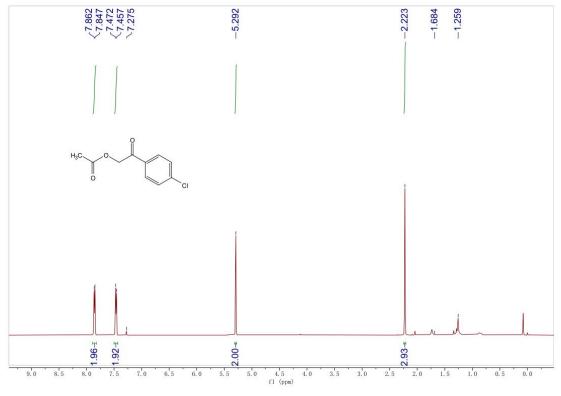
2-(4-methoxyphenyl)-2-oxoethyl acetate (2e) ¹H NMR (500 MHz, CDCl₃)



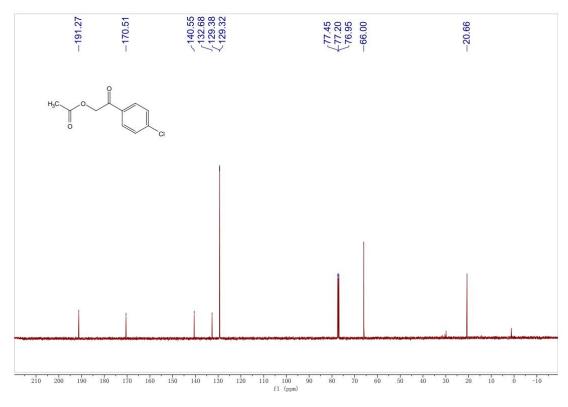
¹³C {¹H} NMR (126 MHz, CDCl₃)



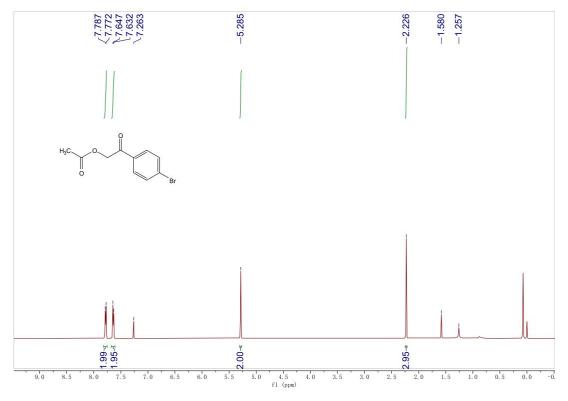
2-(4-chlorophenyl)-2-oxoethyl acetate (2f) ¹H NMR (500 MHz, CDCl₃)



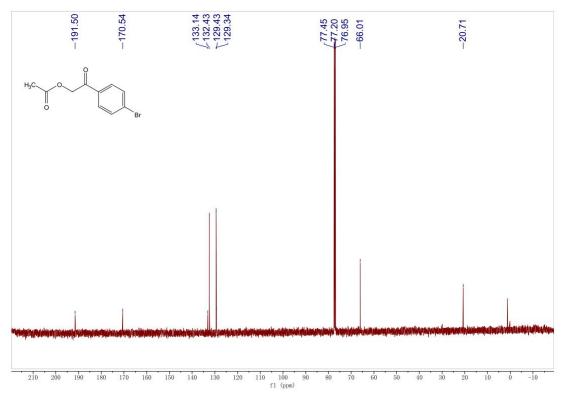
¹³C {¹H} NMR (126 MHz, CDCl₃)



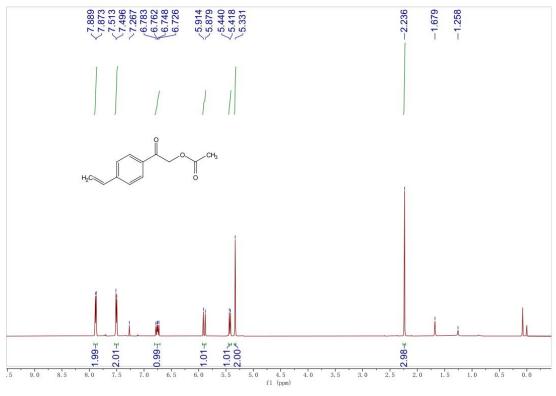
2-(4-bromophenyl)-2-oxoethyl acetate (2g) ¹H NMR (500 MHz, CDCl₃)



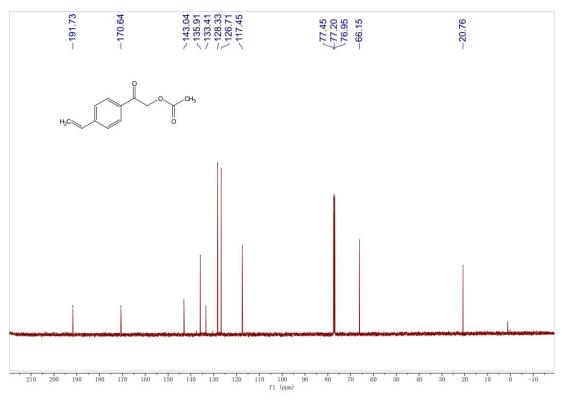
¹³C {¹H} NMR (126 MHz, CDCl₃)



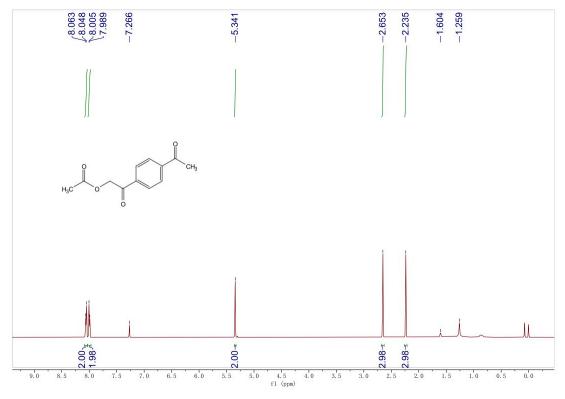
2-oxo-2-(4-vinylphenyl)ethyl acetate (2h) ¹H NMR (500 MHz, CDCl₃)



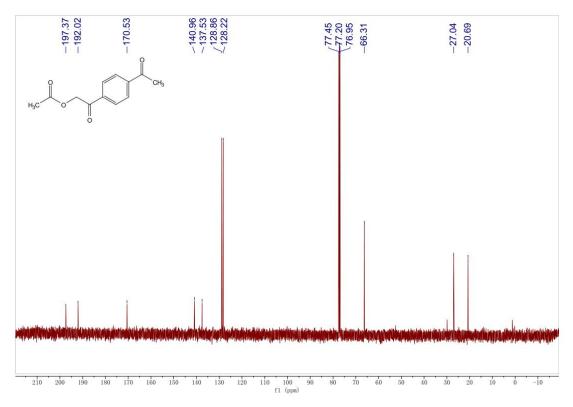
¹³C {¹H} NMR (126 MHz, CDCl₃)



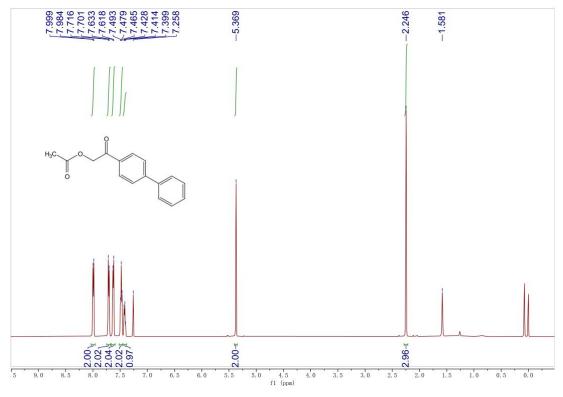
2-(4-acetylphenyl)-2-oxoethyl acetate (2i) ¹H NMR (500 MHz, CDCl₃)



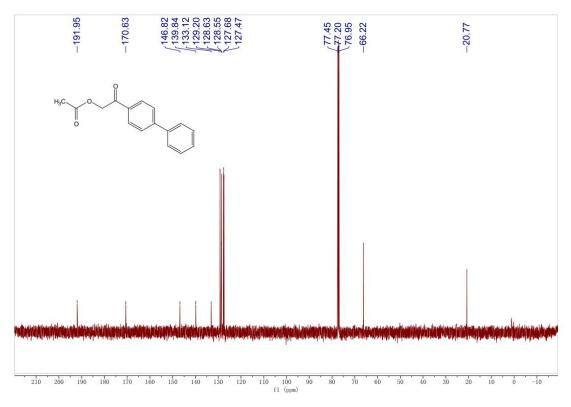
¹³C {¹H} NMR (126 MHz, CDCl₃)



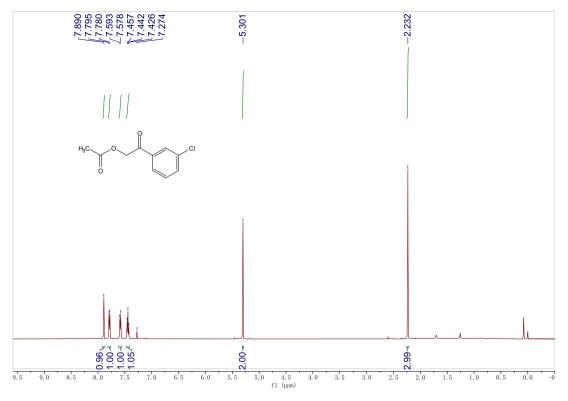
2-([1,1'-biphenyl]-4-yl)-2-oxoethyl acetate (2j) ¹H NMR (500 MHz, CDCl₃)



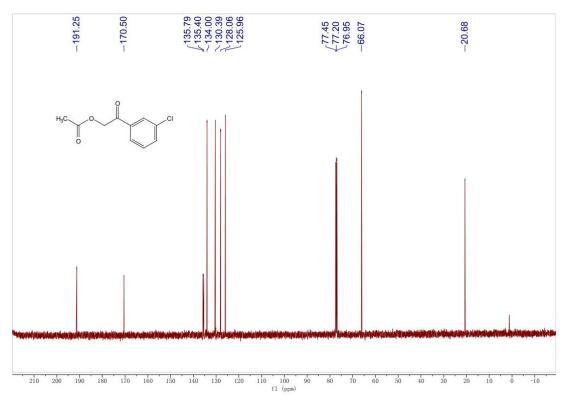
¹³C {¹H} NMR (126 MHz, CDCl₃)



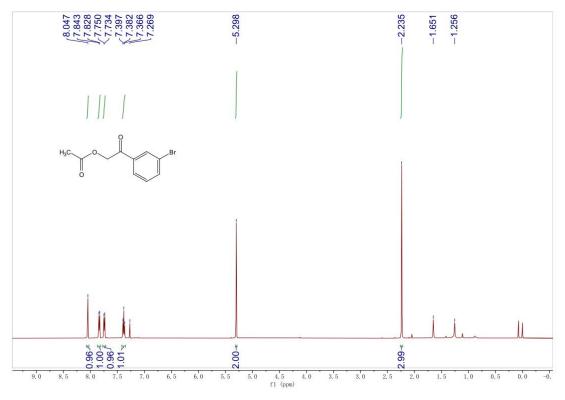
2-(3-chlorophenyl)-2-oxoethyl acetate (2k) ¹H NMR (500 MHz, CDCl₃)



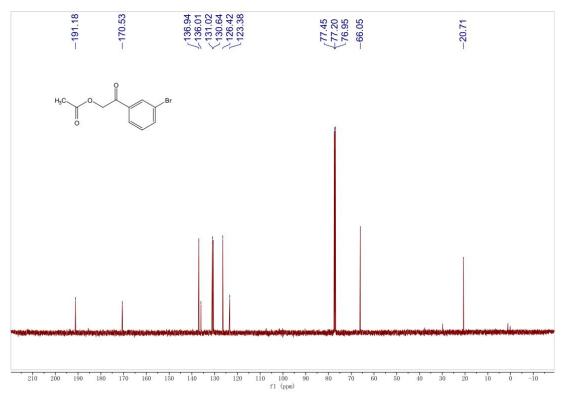
¹³C {¹H} NMR (126 MHz, CDCl₃)



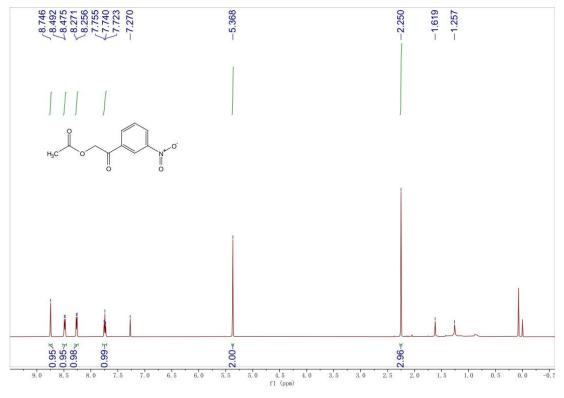
2-(3-bromophenyl)-2-oxoethyl acetate (2l) ¹H NMR (500 MHz, CDCl₃)



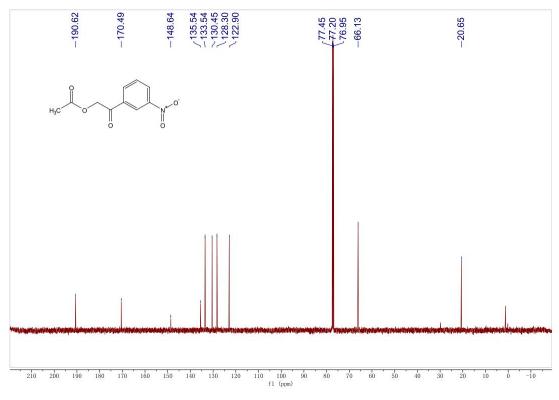
¹³C {¹H} NMR (126 MHz, CDCl₃)



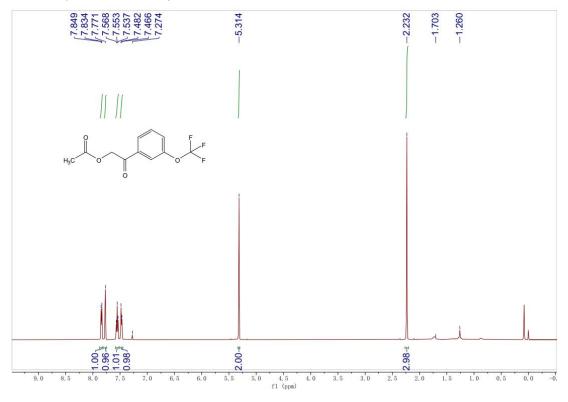
2-(3-nitrophenyl)-2-oxoethyl acetate (2m) ¹H NMR (500 MHz, CDCl₃)



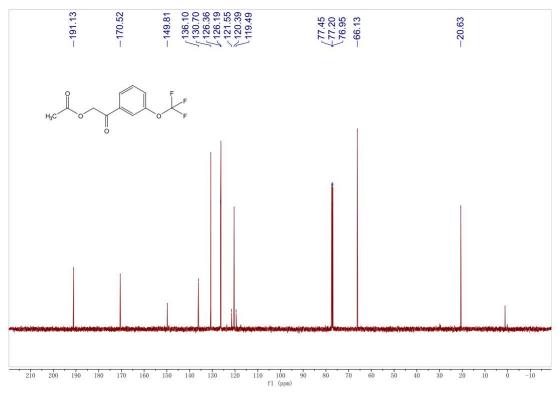
¹³C {¹H} NMR (126 MHz, CDCl₃)



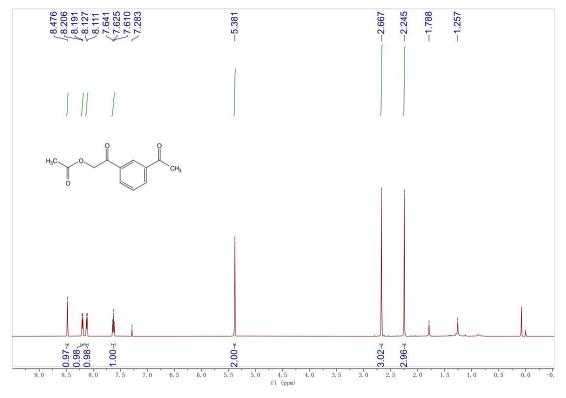
2-oxo-2-(3-(trifluoromethoxy)phenyl)ethyl acetate (2n) ¹H NMR (500 MHz, CDCl₃)



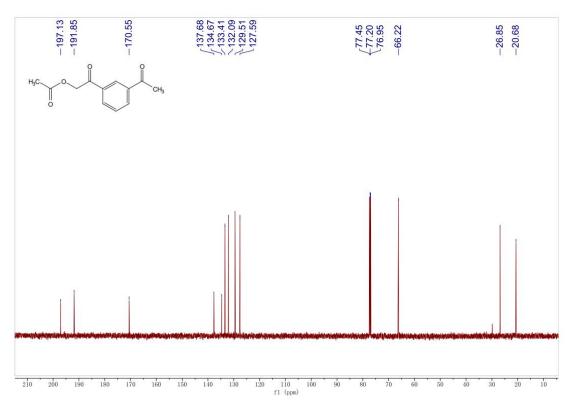
¹³C {¹H} NMR (126 MHz, CDCl₃)



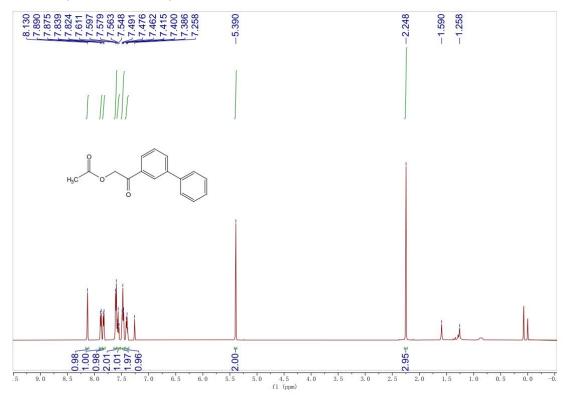
2-(3-acetylphenyl)-2-oxoethyl acetate (20) ¹H NMR (500 MHz, CDCl₃)



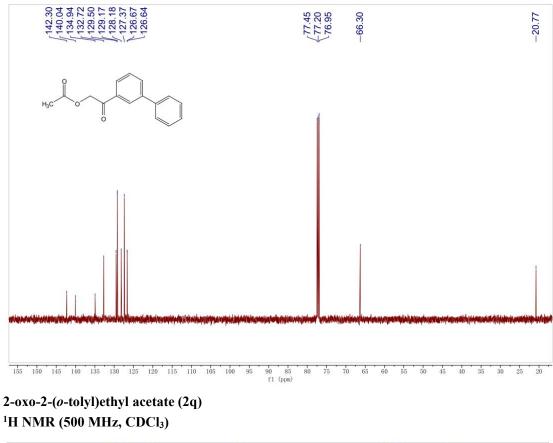
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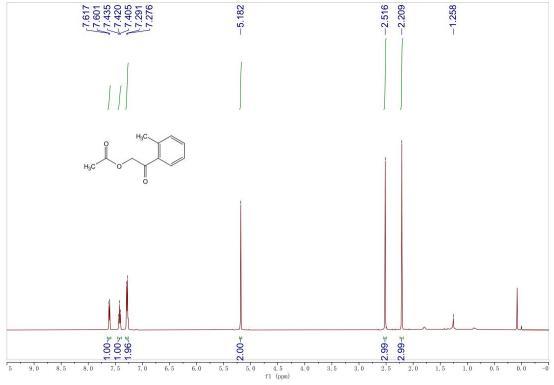


2-([1,1'-biphenyl]-3-yl)-2-oxoethyl acetate (2p) ¹H NMR (500 MHz, CDCl₃)

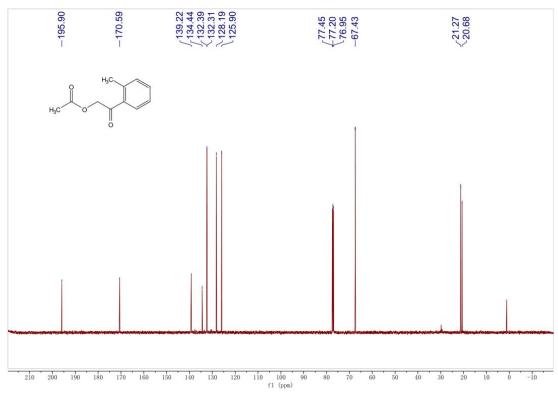


¹³C {¹H} NMR (126 MHz, CDCl₃)

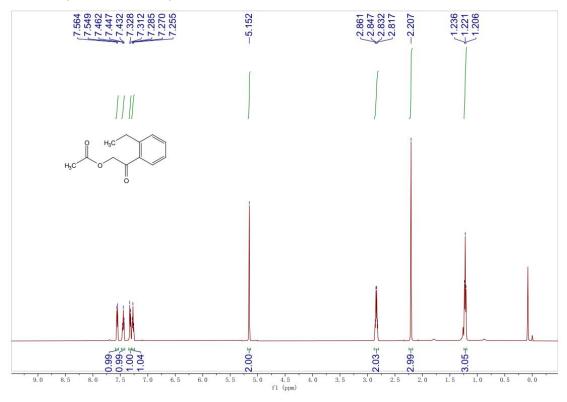




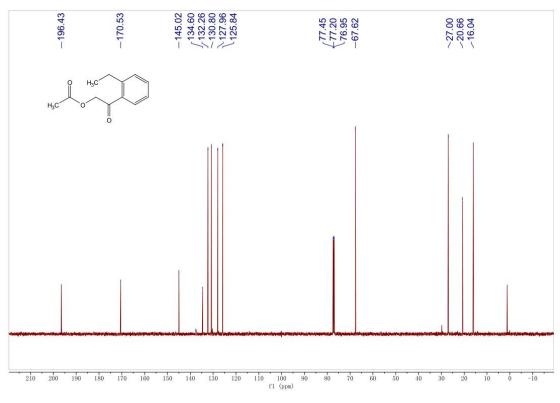
¹³C {¹H} NMR (126 MHz, CDCl₃)



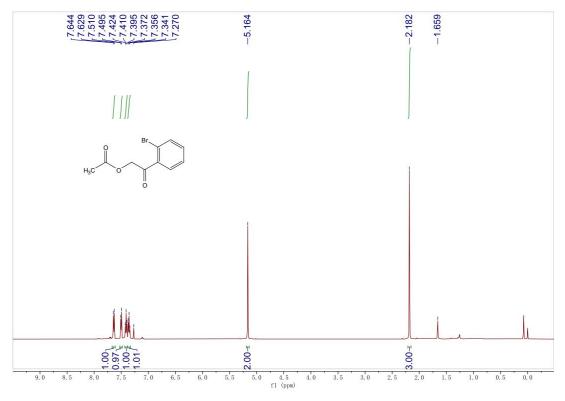
2-oxo-2-(*o*-tolyl)ethyl acetate (2r) ¹H NMR (500 MHz, CDCl₃)



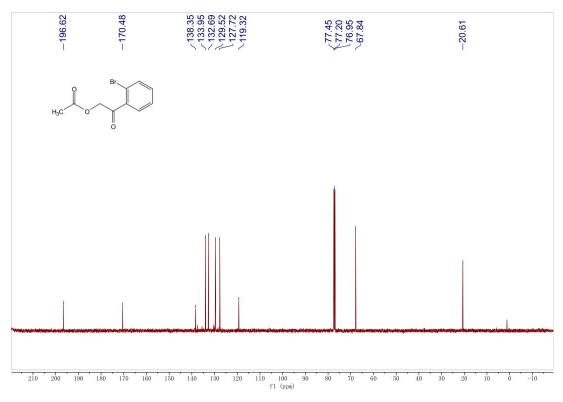
¹³C {¹H} NMR (126 MHz, CDCl₃)



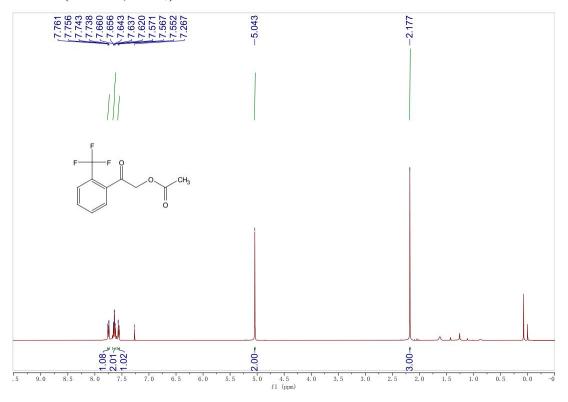
2-(2-bromophenyl)-2-oxoethyl acetate (2s) ¹H NMR (500 MHz, CDCl₃)



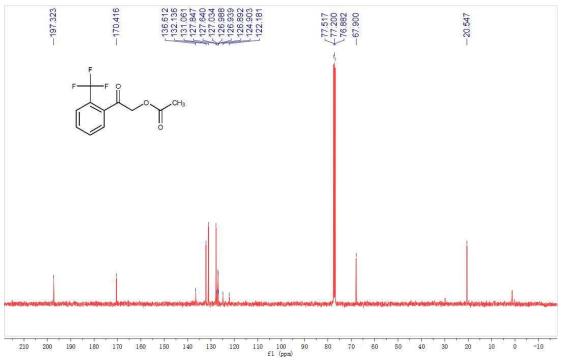
¹³C {¹H} NMR (126 MHz, CDCl₃)



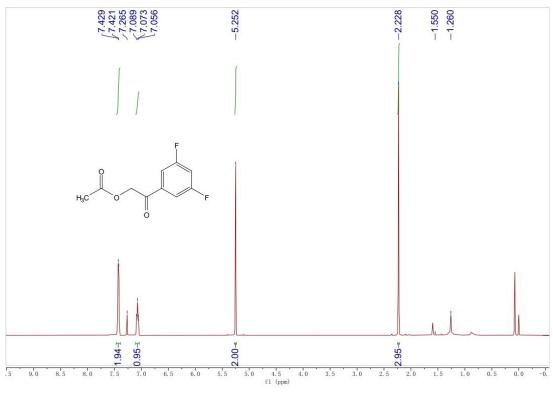
2-oxo-2-(2-(trifluoromethyl)phenyl)ethyl acetate (2t) ¹H NMR (400 MHz, CDCl₃)



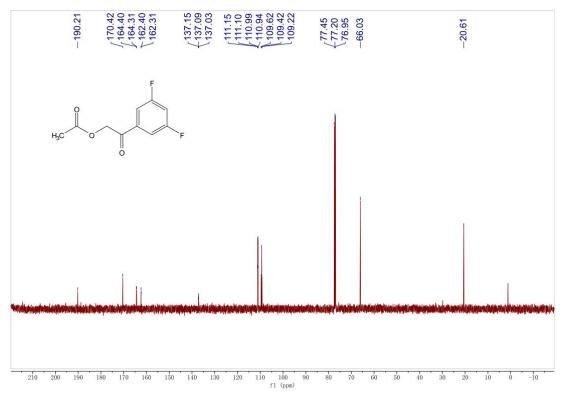
¹³C {¹H} NMR (101 MHz, CDCl₃)



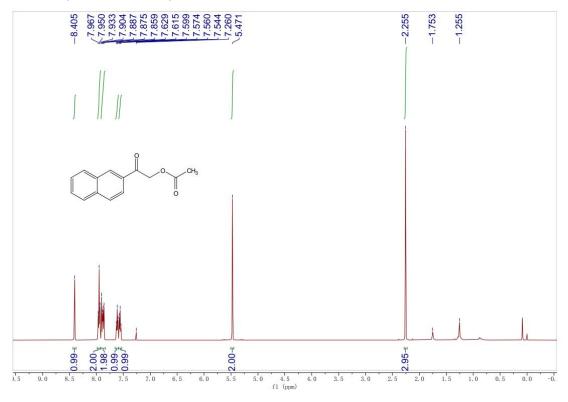
2-(3,5-difluorophenyl)-2-oxoethyl acetate (2u) ¹H NMR (500 MHz, CDCl₃)



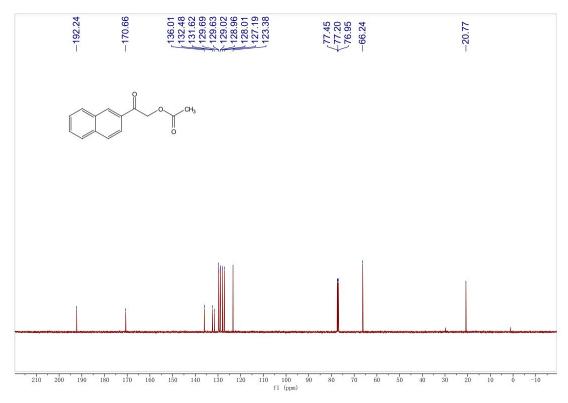
¹³C {¹H} NMR (126 MHz, CDCl₃)



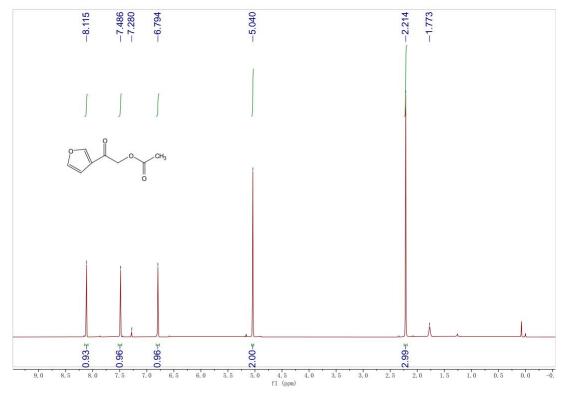
2-(naphthalen-2-yl)-2-oxoethyl acetate (2v) ¹H NMR (500 MHz, CDCl₃)



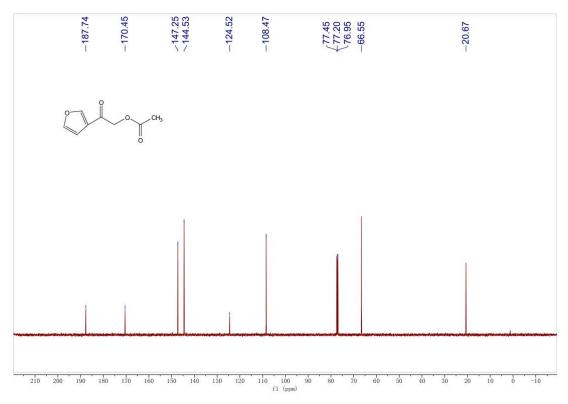
¹³C {¹H} NMR (126 MHz, CDCl₃)



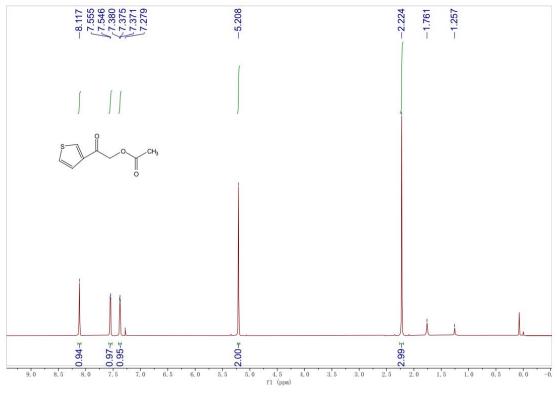
2-(furan-3-yl)-2-oxoethyl acetate (2w) ¹H NMR (500 MHz, CDCl₃)



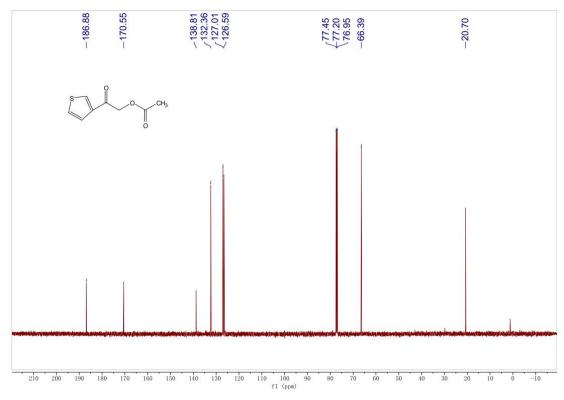
¹³C {¹H} NMR (126 MHz, CDCl₃)



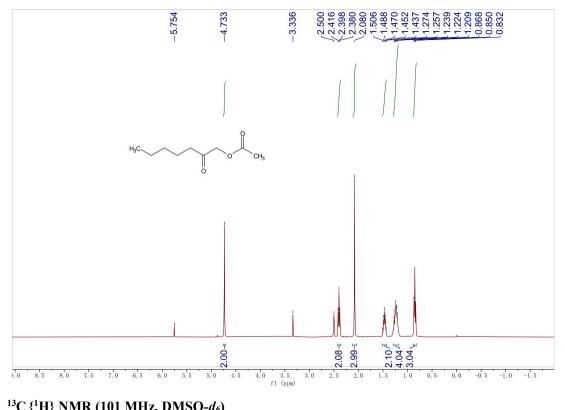
2-oxo-2-(thiophen-3-yl)ethyl acetate (2x) ¹H NMR (500 MHz, CDCl₃)



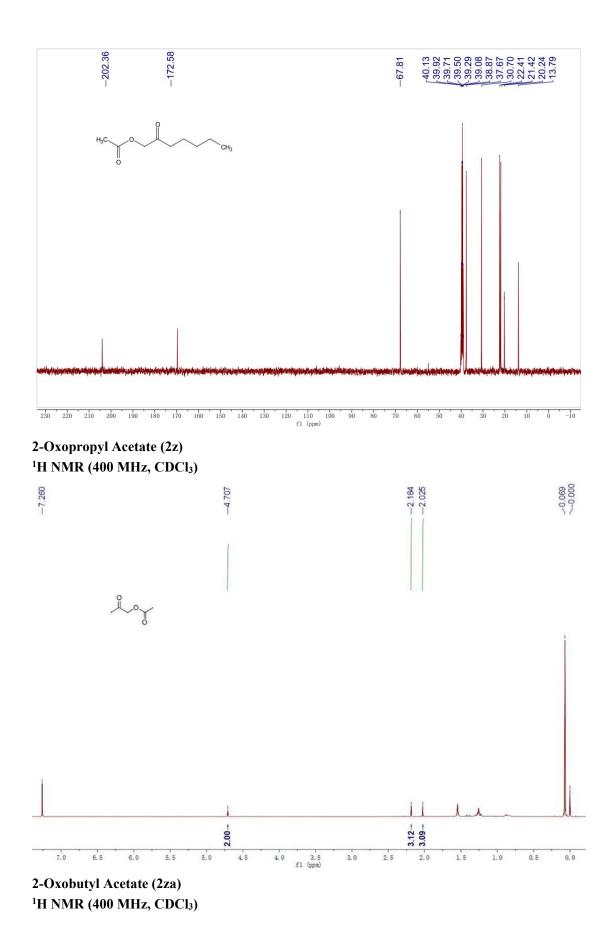
¹³C {¹H} NMR (126 MHz, CDCl₃)



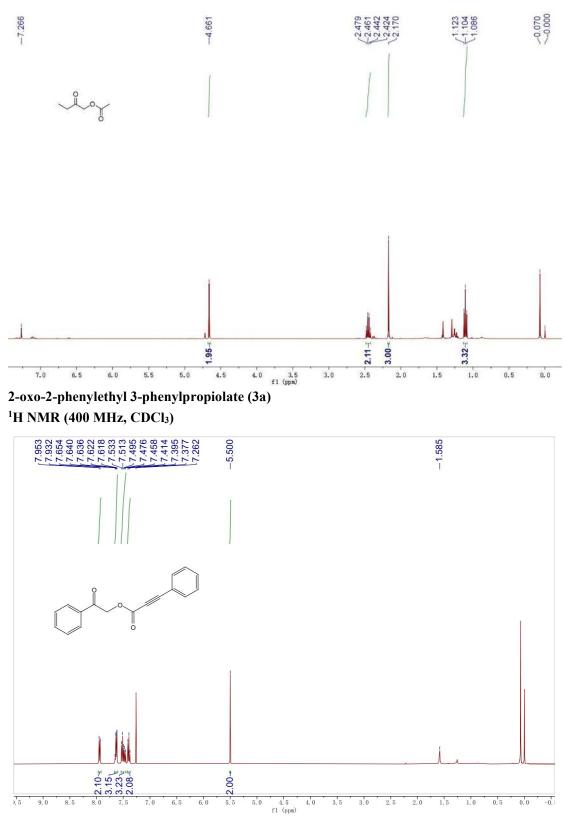
2-oxoheptyl acetate (2y) ¹H NMR (400 MHz, DMSO-*d*₆)



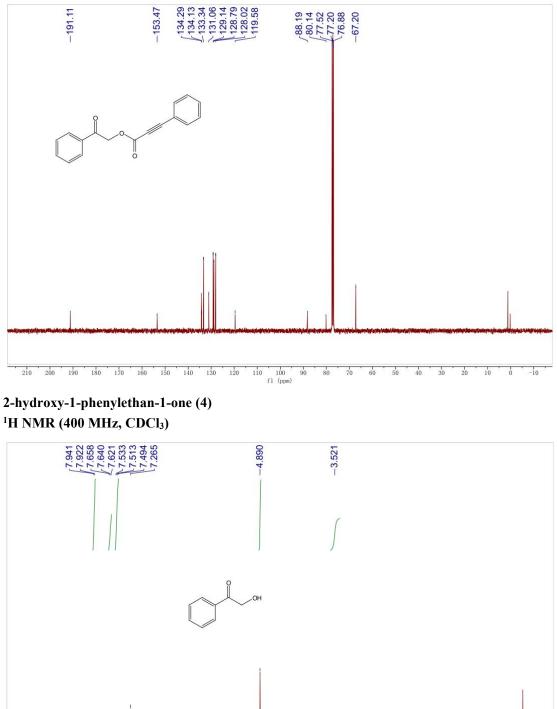
¹³C {¹H} NMR (101 MHz, DMSO-*d*₆)

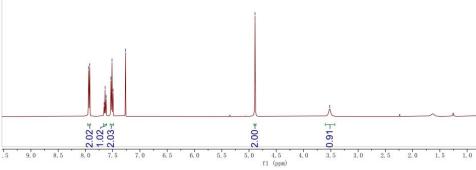


S42



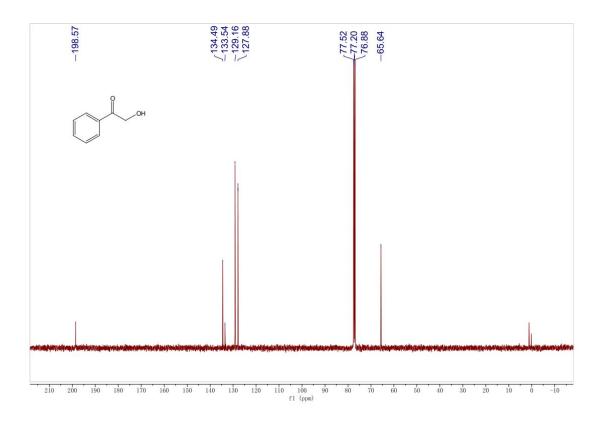
¹³C {¹H} NMR (101 MHz, CDCl₃)





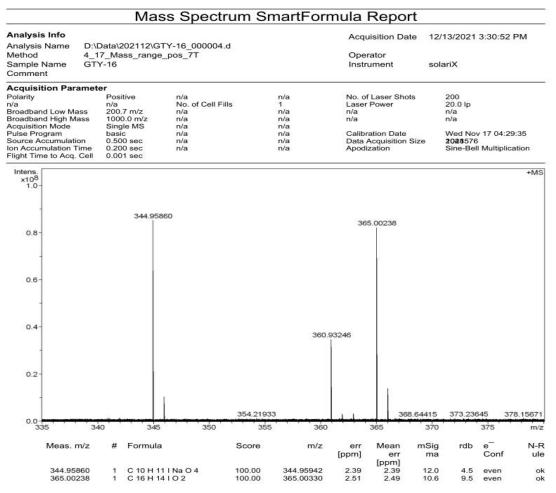
¹³C {¹H} NMR (101 MHz, CDCl₃)

0.5 0.0



5. Mass spectrometry for mechanism study

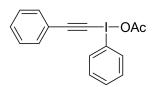
Propiolic acid **1a** (0.50 mmol, 1.0 equivalent), H₂O (22 μ L, 1.25 mmol, 2.5 equivalent), PhI(OAc)₂ (483 mg, 1.5 mmol, 3.0 equivalent) and 8 mL AcOH were added in a round-bottom flask. The mixture was allowed to stir at 80 °C (oil bath temperature) under air for 0.5 h. After cooling to room temperature, the mixture was characterized by high resolution mass spectroscopy and intermediate **C** was detected.



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Page 1 of 1

Intermediate C



HRMS (ESI-TOF) m/z: calcd for C₁₆H₁₃IO₂ [M+H] ⁺ 365.0033, found 365.0024.

6. References

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