Supporting Information

Chromium Complexes Supported by NNO-Tridentate Ligands: An Unprecedent Activity with the Requirement of a small amount of MAO

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Figure S1. ¹H NMR spectrum of L1 in CDCl₃.



Figure S2. ¹³C NMR spectrum of L1 in CDCl₃.







Figure S4. ¹³C NMR spectrum of L2 in CDCl₃.



Figure S5. ¹H NMR spectrum of L3 in CDCl₃.



Figure S6. ¹³C NMR spectrum of L3 in CDCl₃.



Figure S7. ¹H NMR spectrum of L4 in CDCl₃.



Figure S8. ¹³C NMR spectrum of L4 in CDCl₃.



Figure S9. ¹H NMR spectrum of L5 in CDCl₃.



Figure S10. ¹³C NMR spectrum of L5 in CDCl₃.



Figure S12. ¹³C NMR spectrum of L6 in CDCl₃.



Figure S13. HRMS spectrum of Cr1(THF) (m/z 575.27, [M-2C1-THF+OH+Na]⁺).



Figure S14. HRMS spectrum of Cr2(THF) (m/z 519.21, [M-2Cl-THF+OH+Na]⁺).



Figure S15. HRMS spectrum of Cr3(THF) (m/z 491.18, [M-2C1-THF+OH+Na]⁺).



Figure S16. HRMS spectrum of Cr4(THF) (m/z 527.16, [M-2C1-THF+OH+Na]⁺).



Figure S17. HRMS spectrum of Cr5(THF) (m/z 407.08, [M-2C1-THF+OH+Na]⁺).



Figure S18. HRMS spectrum of Cr6(THF) (m/z 475.01, [M-2C1-THF+OH+Na]⁺).



Figure S19. GC analysis of the liquid fraction obtained from polymerization in Table 1 run 1.



Figure S20. ¹H NMR spectrum of polyethylene sample obtained in Table 2 run 10 (120

 $^{\circ}$ C in C₂D₂Cl₄).



Figure S21. ¹³C NMR spectrum of polyethylene sample obtained in Table 2 run 10 (120 °C in $C_2D_2Cl_4$).



Figure S22. GPC curve of polyethylene obtained in Table 3 run 1 ($M_w = 14.4 \times 10^4$

 $g \cdot mol^{-1}, D = 2.3$).



Figure S23. GPC curve of polyethylene obtained in Table 3 run 2 ($M_w = 38.9 \times 10^4$ g·mol⁻¹, D = 1.5).



Figure S24. GPC curve of polyethylene obtained in Table 3 run 3 ($M_w = 22.3 \times 10^4$

 $g \cdot mol^{-1}, D = 4.2$).



Figure S25. GPC curve of polyethylene obtained in Table 3 run 4 ($M_w = 18.0 \times 10^4$ g·mol⁻¹, D = 10).



Figure S26. GPC curve of polyethylene obtained in Table 3 run 5 ($M_{\rm w} = 45.6 \times 10^4$

 $g \cdot mol^{-1}, D = 19$).



Figure S27. GPC curve of polyethylene obtained in Table 3 run 6 ($M_w = 53.4 \times 10^4$ g·mol⁻¹, D = 19).

Complex	Cr2(THF)	Cr4(THF)
Empirical formula	$C_{33}H_{43}Cl_2CrN_2O_2$	C ₃₁ H ₃₇ Cl ₂ CrF ₂ N ₂ O ₂
Formula weight	622.59	630.52
Temperature / K	193	213
Crystal system	triclinic	monoclinic
Space group	P-1	$P2_1/c$
	16.326(3),	14.4764(4),
a / Å, b / Å, c / Å	16.389(3),	14.5398(4),
	16.464(3)	15.5208(4)
	70.953(7),	90,
$lpha/^{\circ},eta/^{\circ},\gamma/^{\circ}$	89.935(10),	109.5390(10),
	71.859(7)	90
Volume / Å ³	3932.1(13)	3078.75(15)
Z	4	4
$ ho_{calc}$ / mg mm ⁻³	1.052	1.360
μ / mm^{-1}	2.591	3.395
F(000)	1316.0	1316.0
Crystal size / mm ³	$0.05 \times 0.03 \times 0.03$	$0.07 \times 0.07 \times 0.05$
2Θ range for data collection	5.866 to 110.332	8.222 to 110.554
	$-19 \le h \le 17$,	$-17 \le h \le 17$,
Index ranges	$-20 \le k \le 19$,	$-17 \le k \le 11$,
	$-20 \le 1 \le 20$	$-18 \le 1 \le 18$
Reflections collected	56466	26800
Independent reflections	14951 [$R_{int} = 0.0787$]	5858 [$R_{int} = 0.0458$]
Data/restraints/parameters	14951/0/737	5858/18/367
Goodness-of-fit on F ²	1.054	1.072
Final R indexes [I>2 σ (I)]	$R_1 = 0.0736,$	$R_1 = 0.0518,$
	$wR_2 = 0.2105$	$wR_2 = 0.1382$
Final R indexes [all data]	$R_1 = 0.1177,$	$R_1 = 0.0664,$
	$wR_2 = 0.2400$	$wR_2 = 0.1507$
Largest diff. peak/hole / e Å ⁻³	0.35/-0.67	0.78/-0.56

Table S1. Crystal data and structure refinement for Cr2(THF) and Cr4(THF).