

Electronic Supplementary Information

Scandium-catalyzed stereoselective block and alternating copolymerization of diphenylphosphinostyrenes and isoprene

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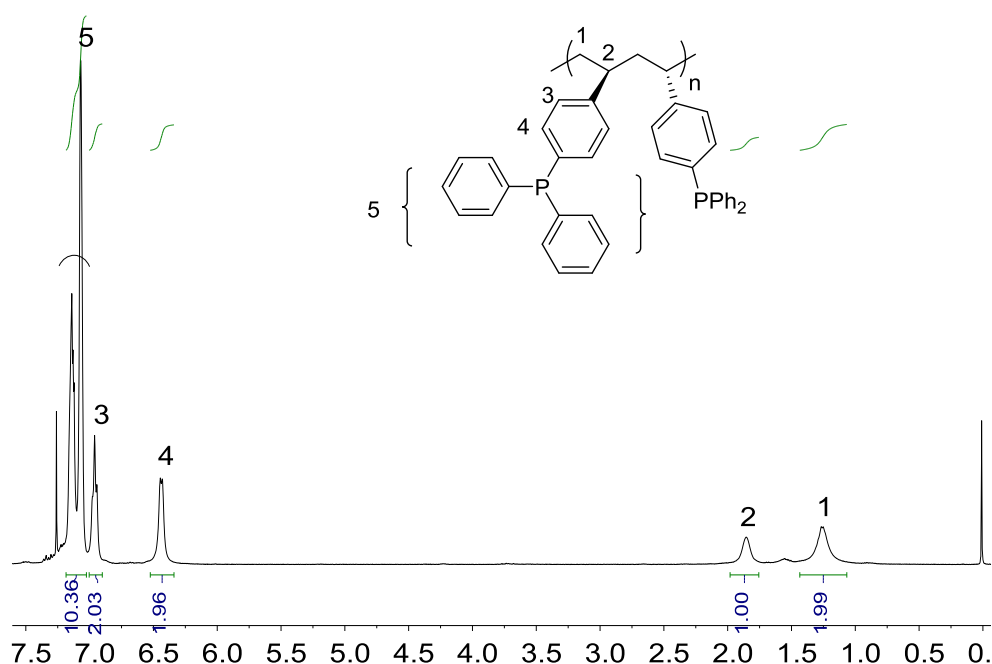


Fig. S1 ¹H-NMR spectrum of a *p*-StPPh₂ homopolymer prepared by (C₅Me₄SiMe₃)Sc(CH₂C₆H₄NMe₂-o)₂ (Table 1, Run 1).

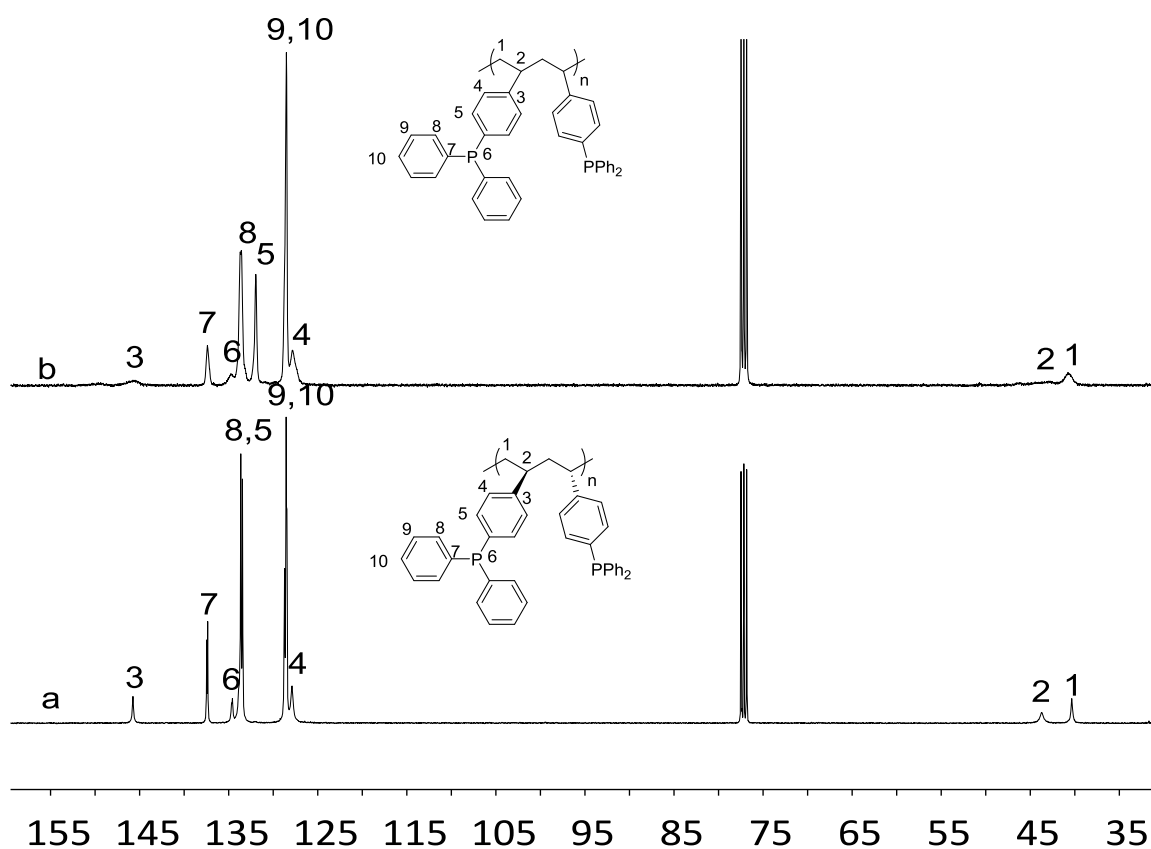
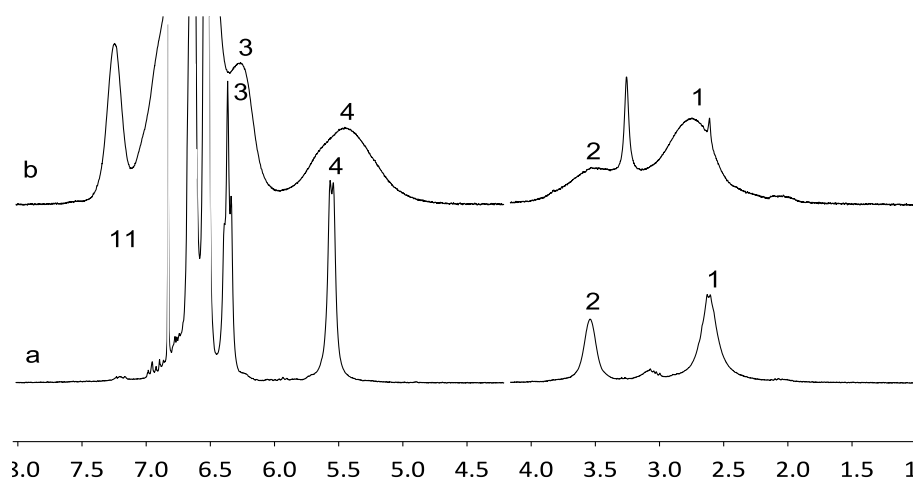


Fig. S2 ^1H , ^{13}C -NMR spectrum of a *p*-StPPh₂ homopolymer prepared by $(\text{C}_5\text{Me}_4\text{SiMe}_3)\text{Sc}(\text{CH}_2\text{C}_6\text{H}_4\text{NMe}_2\text{-}o)_2$ (a) (Table 1, Run 1) and AIBN (b) (A procedure for the polymerize of *p*-StPPh₂ by AIBN: AIBN (1.6 mg 10 μmol) in toluene (2 mL) was added under stirring to a mixture of *p*-StPPh₂ (0.58 g, 2.0 mmol) in toluene (2 mL). The polymerization was carried out at 80 $^\circ\text{C}$ for 24 h. The resulting mixture was poured into a large amount of methanol to precipitate the polymer product, which was then collected by filtration, washed with methanol, and dried under vacuum at 40 $^\circ\text{C}$ to a constant weight.)

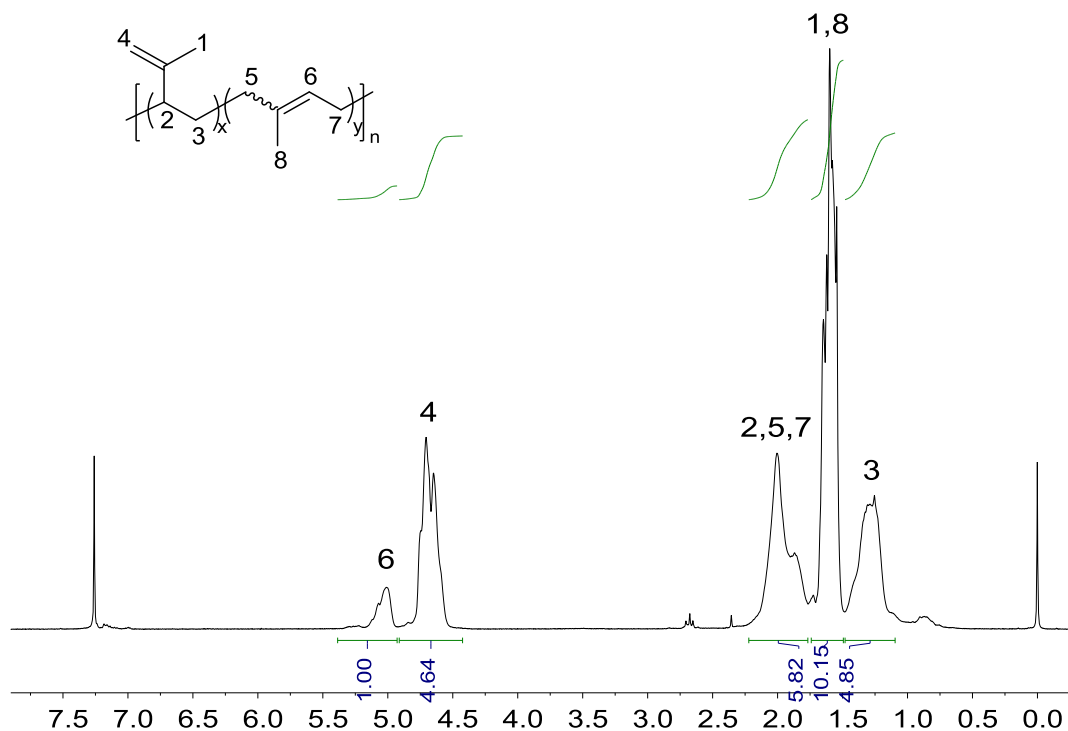


Fig. S3 ¹H-NMR spectrum of an IP homopolymer prepared by (C₅Me₄SiMe₃)Sc(CH₂C₆H₄NMe₂-o)₂ (Table 1, Run 2).

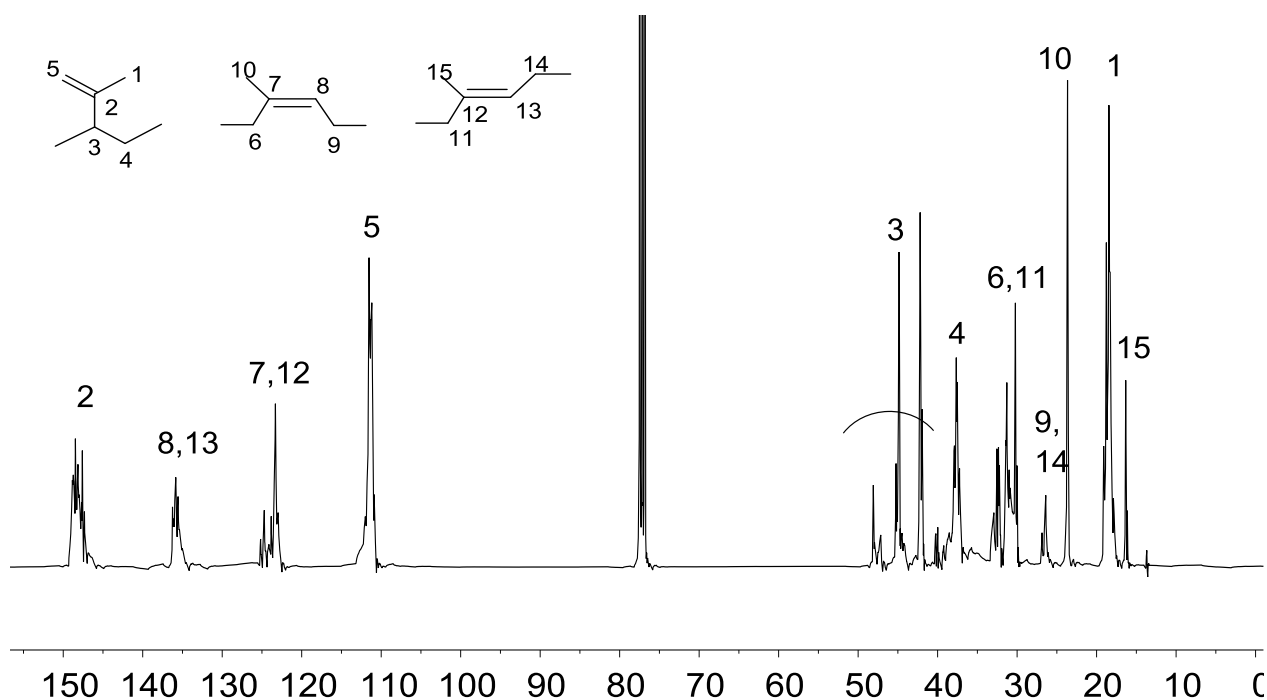


Fig. S4 ¹³C-NMR spectrum of an IP homopolymer prepared by (C₅Me₄SiMe₃)Sc(CH₂C₆H₄NMe₂-o)₂ (Table 1, Run 2).

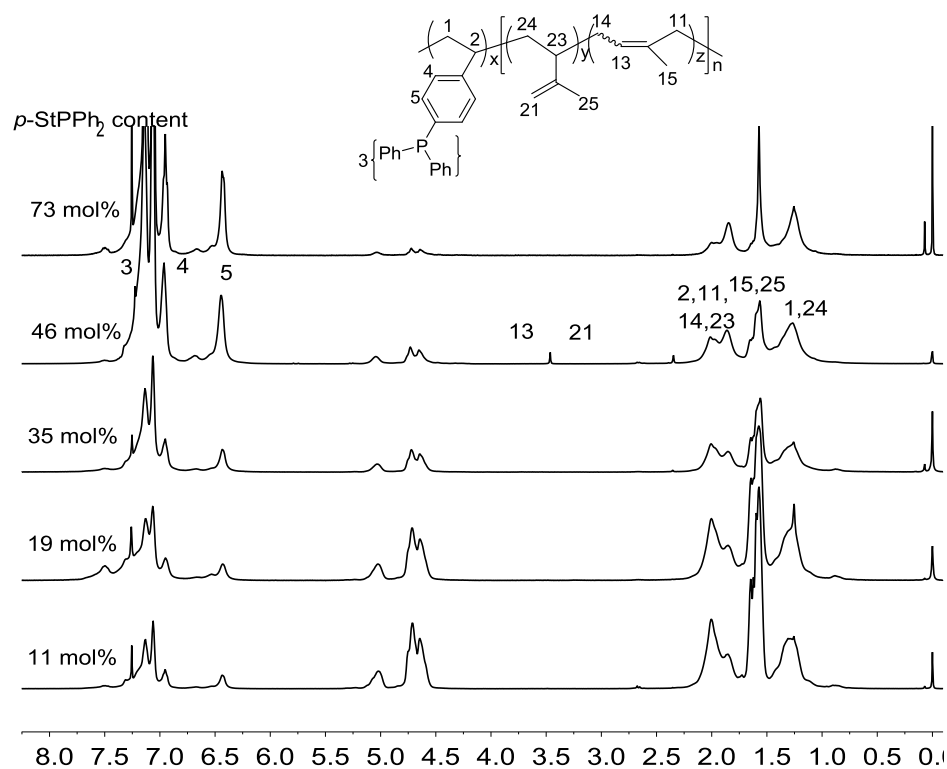


Fig. S5 ^1H -NMR spectra of $p\text{-StPPh}_2$ -IP copolymers with different $p\text{-StPPh}_2$ content prepared by $(\text{C}_5\text{Me}_4\text{SiMe}_3)\text{Sc}(\text{CH}_2\text{C}_6\text{H}_4\text{NMe}_2\text{-}o)_2$.

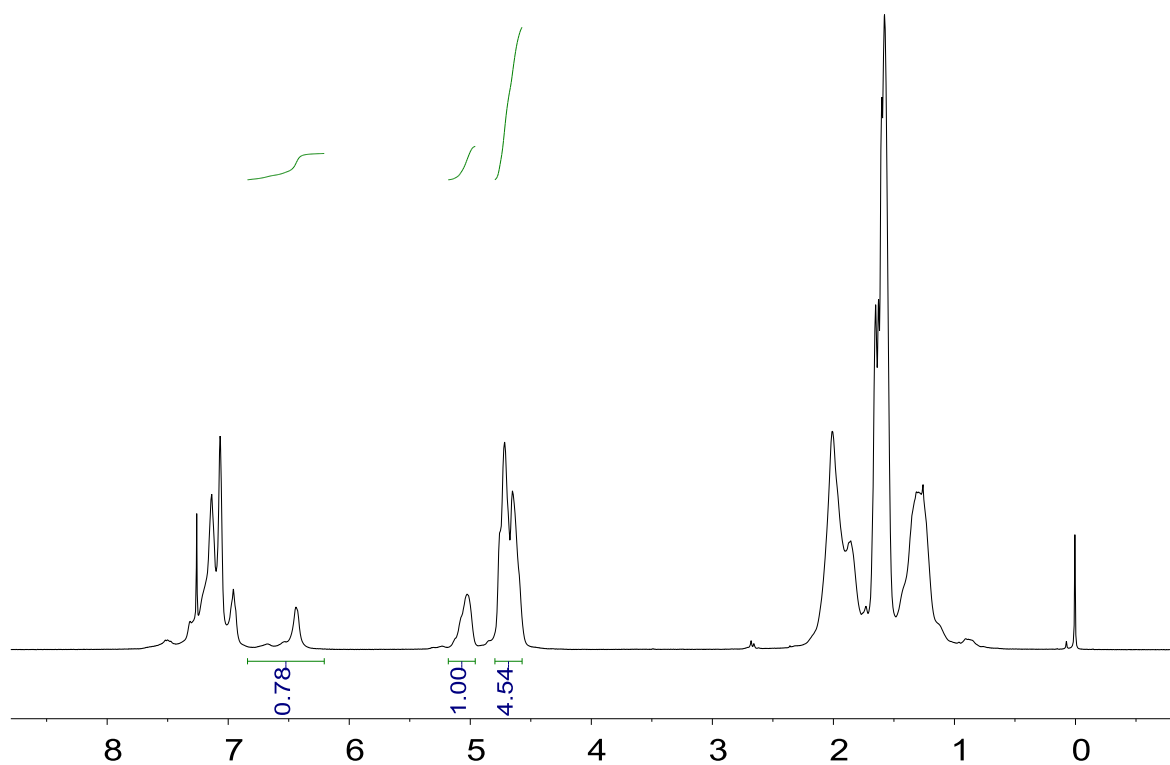


Fig. S6 ^1H -NMR spectrum of a $p\text{-StPPh}_2$ -IP copolymer (Table 1, Run 3).

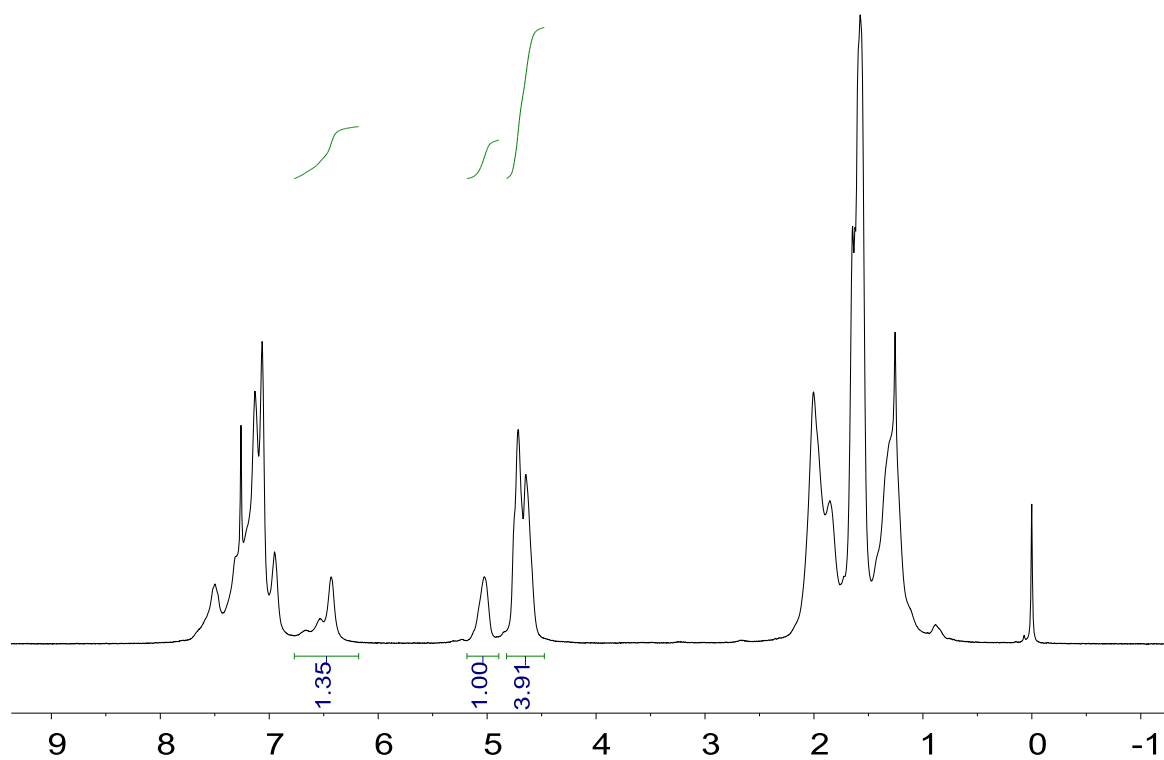


Fig. S7 ¹H-NMR spectrum of a *p*-StPPh₂-IP copolymer (Table 1, Run 4).

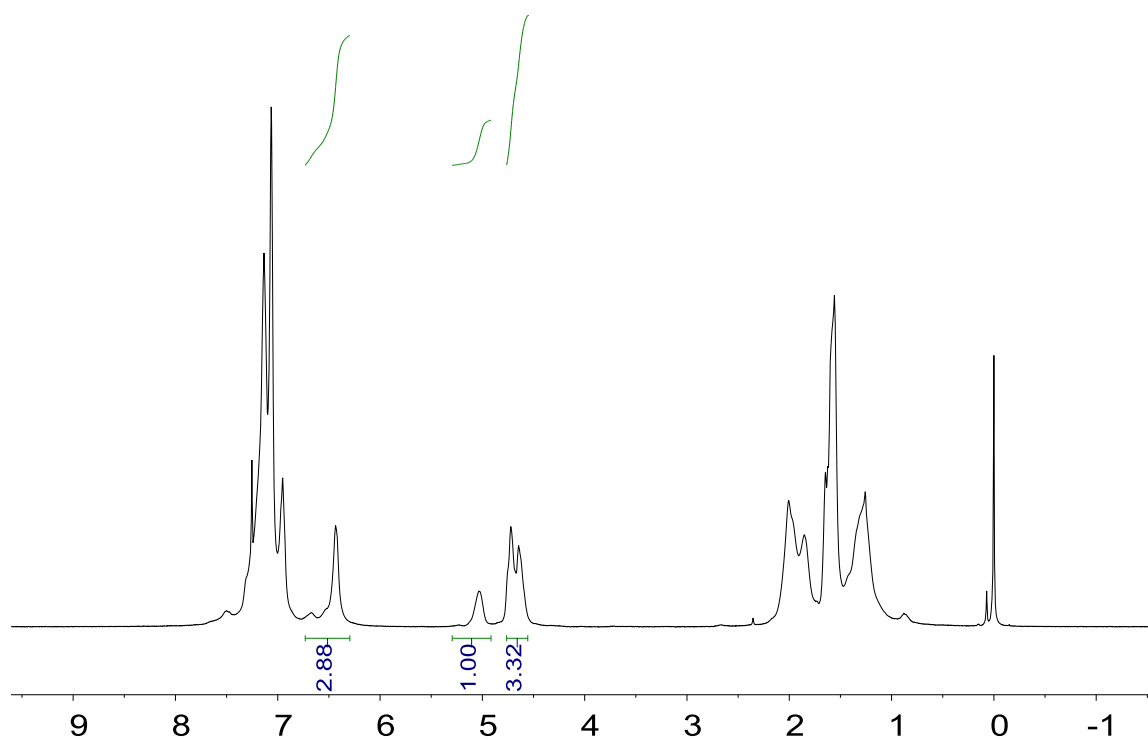


Fig. S8 ¹H-NMR spectrum of a *p*-StPPh₂-IP copolymer (Table 1, Run 5).

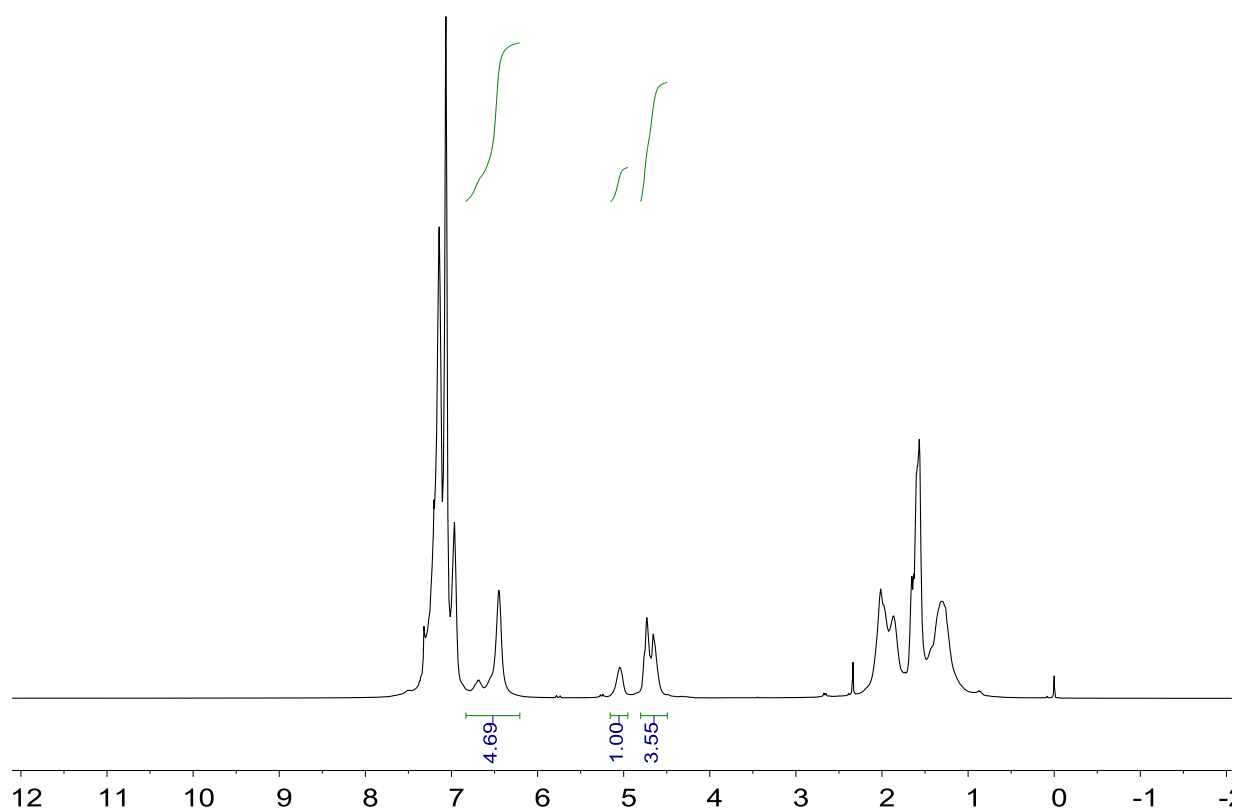


Fig. S9 ¹H-NMR spectrum of a *p*-StPPh₂-IP copolymer (Table 1, Run 6).

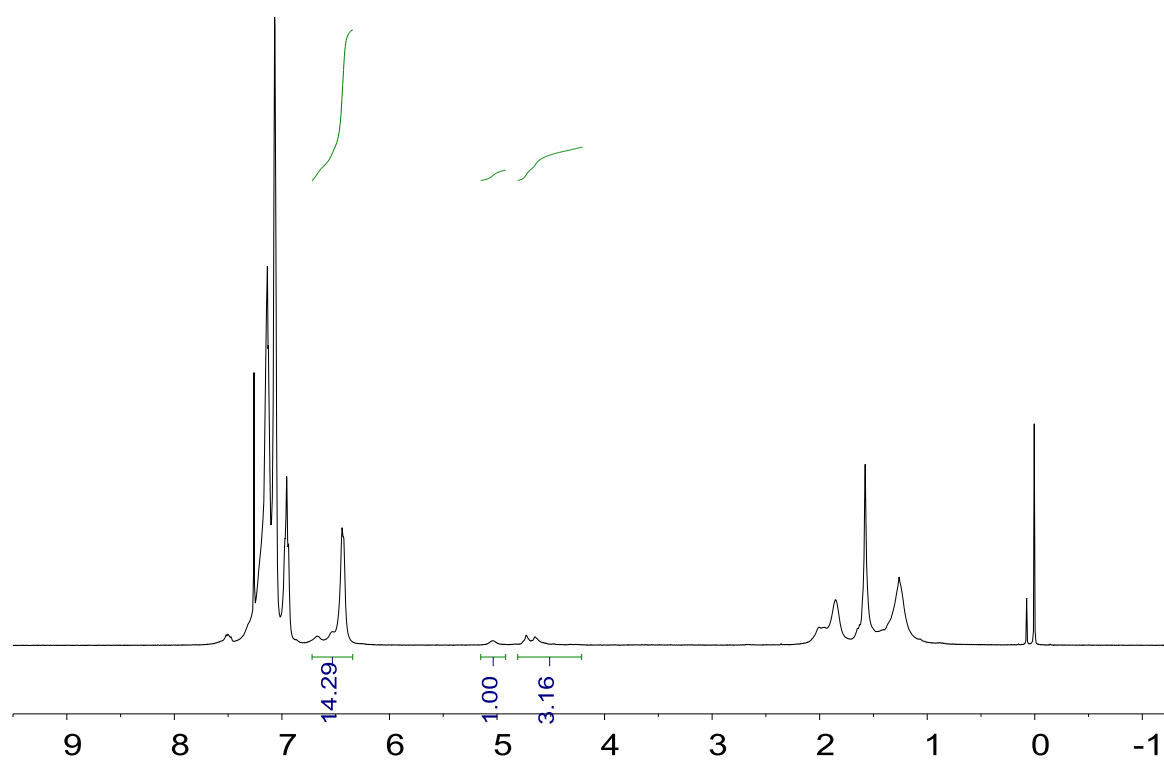
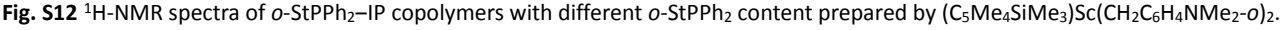
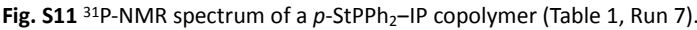


Fig. S10 ¹H-NMR spectrum of a *p*-StPPh₂-IP copolymer (Table 1, Run 7).



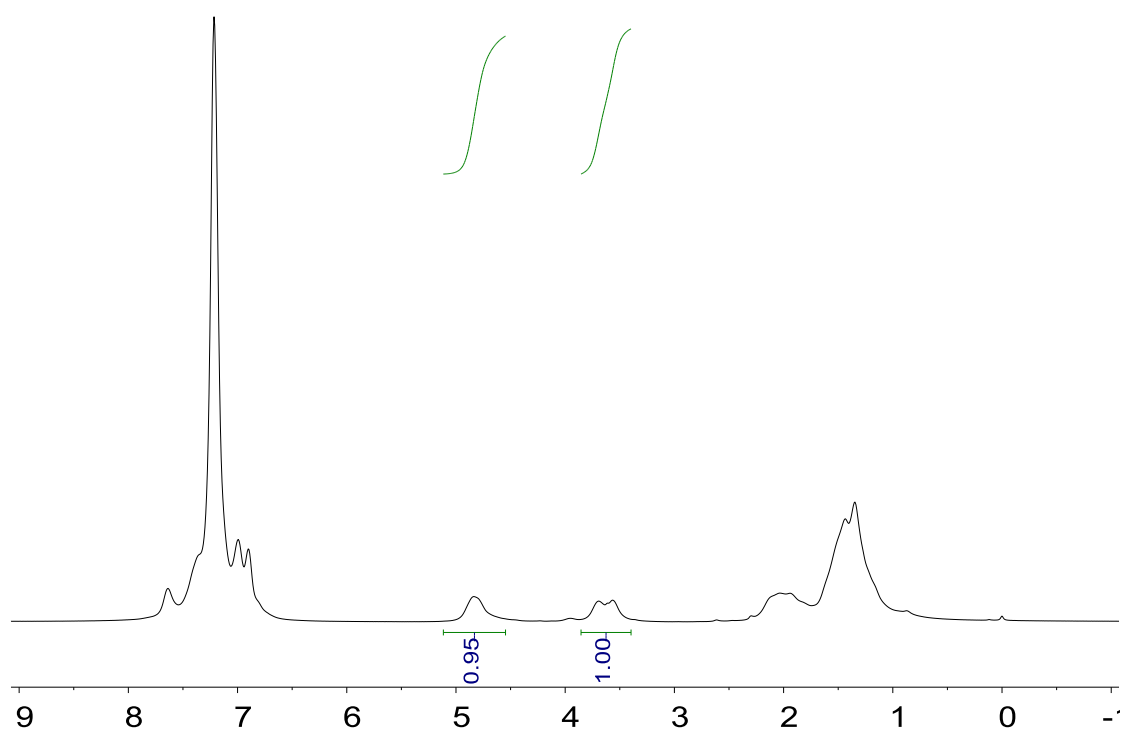


Fig. S13 ^1H -NMR spectrum of an *o*-StPPh₂-IP copolymer (Table 2, Run 2).

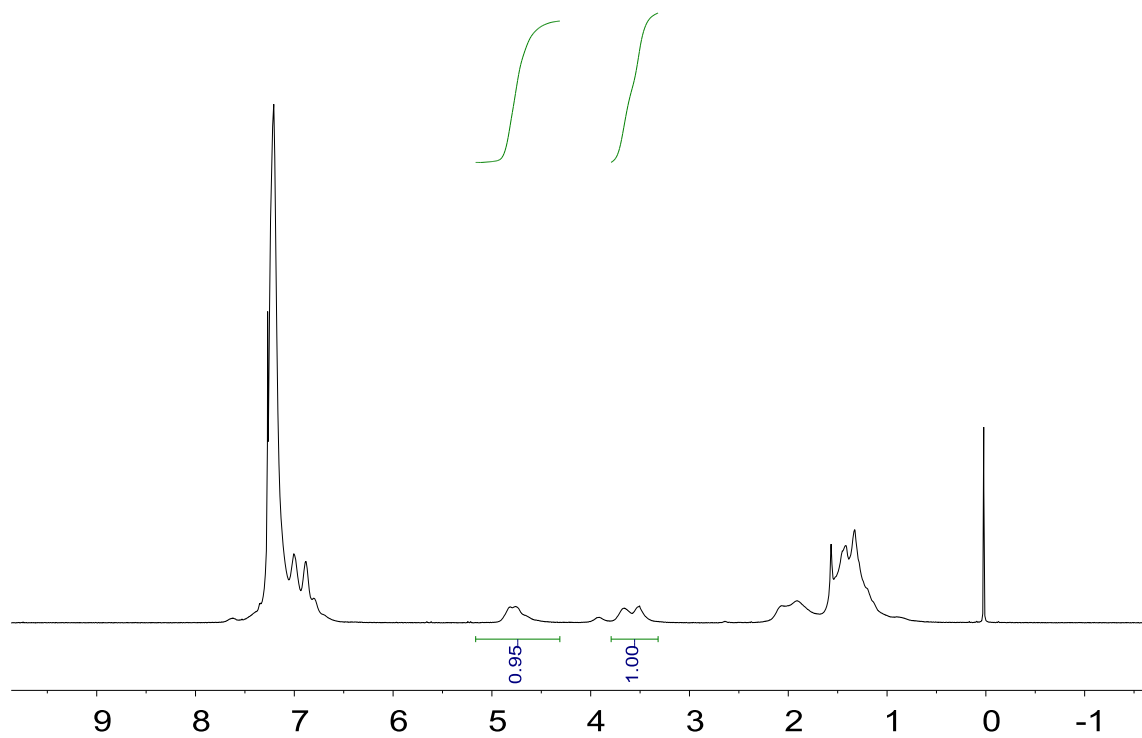


Fig. S14 ^1H -NMR spectrum of an *o*-StPPh₂-IP copolymer (Table 2, Run 3).

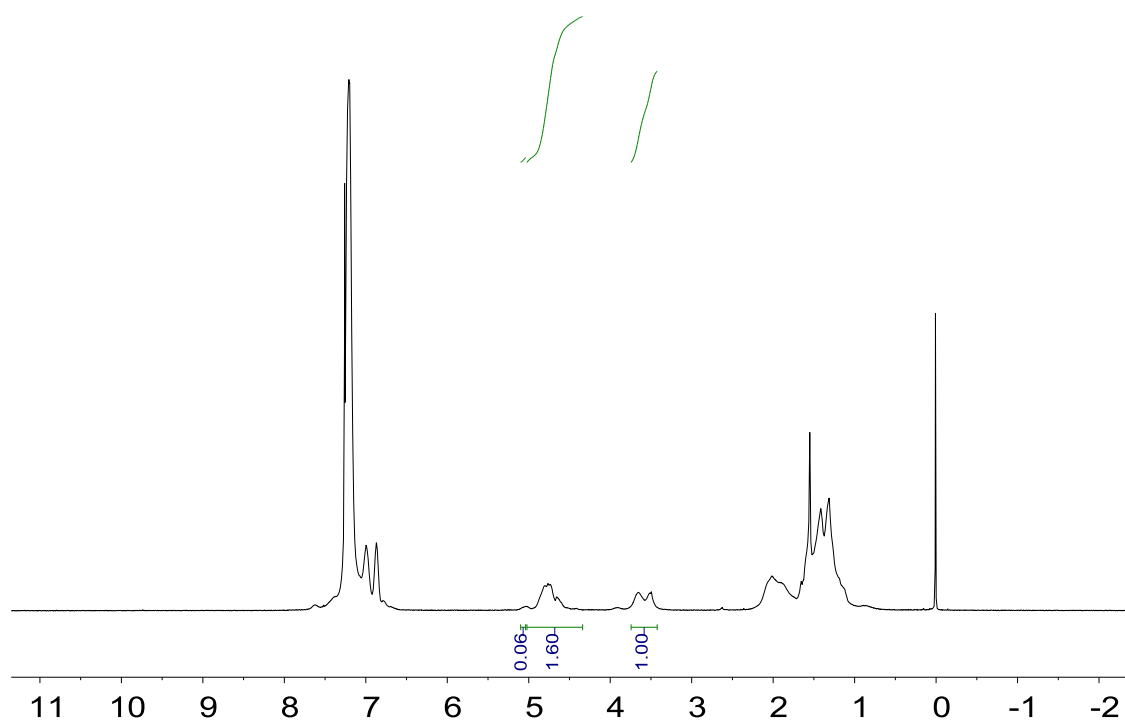


Fig. S15 ^1H -NMR spectrum of an *o*-StPPh₂-IP copolymer (Table 2, Run 4).

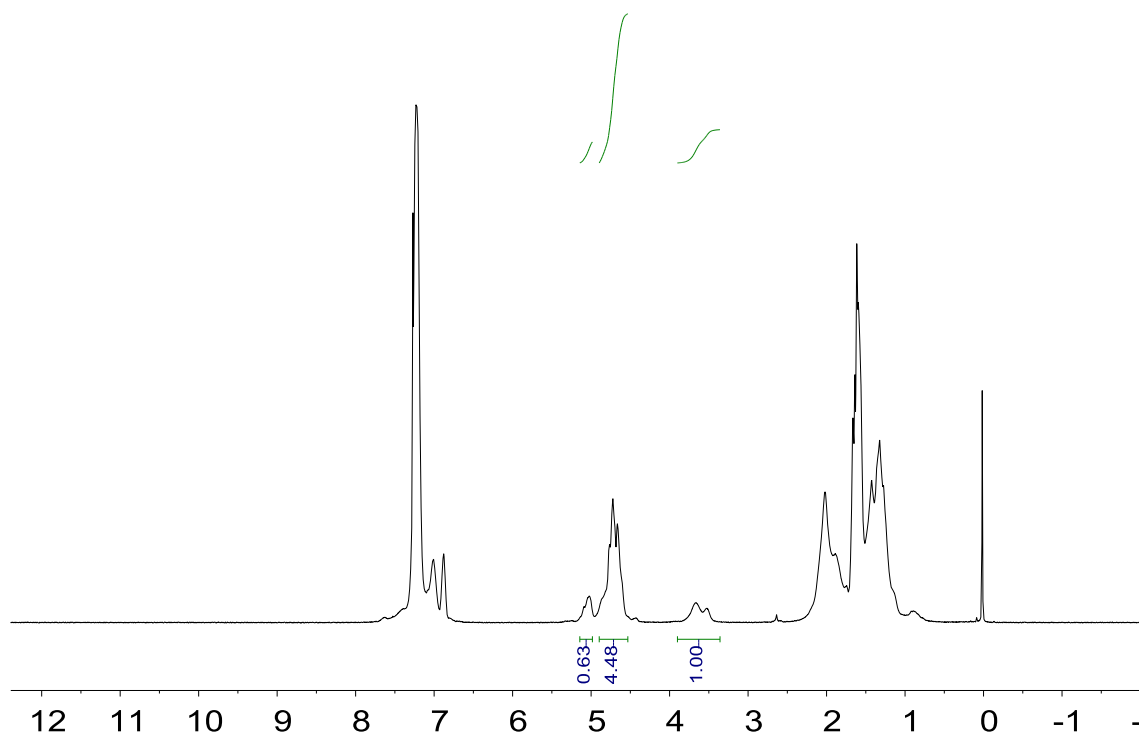


Fig. S16 ^1H -NMR spectrum of an *o*-StPPh₂-IP copolymer (Table 2, Run 5).

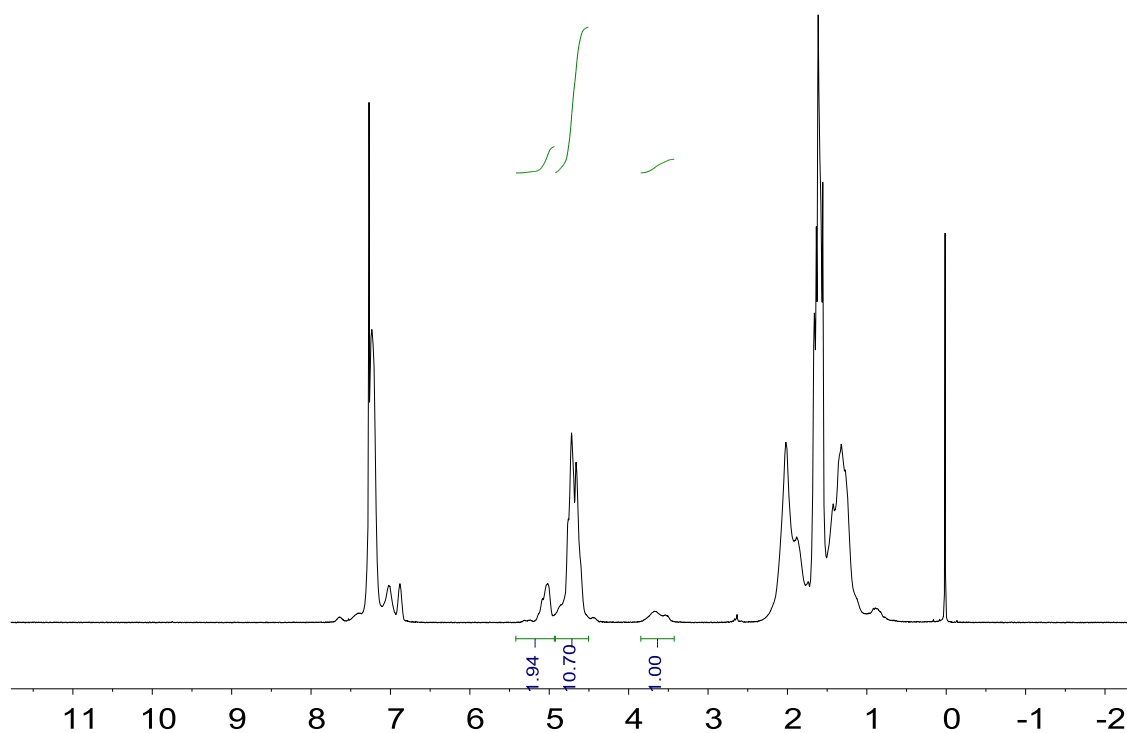


Fig. S17 ^1H -NMR spectrum of an *o*-StPPh₂-IP copolymer (Table 2, Run 6).

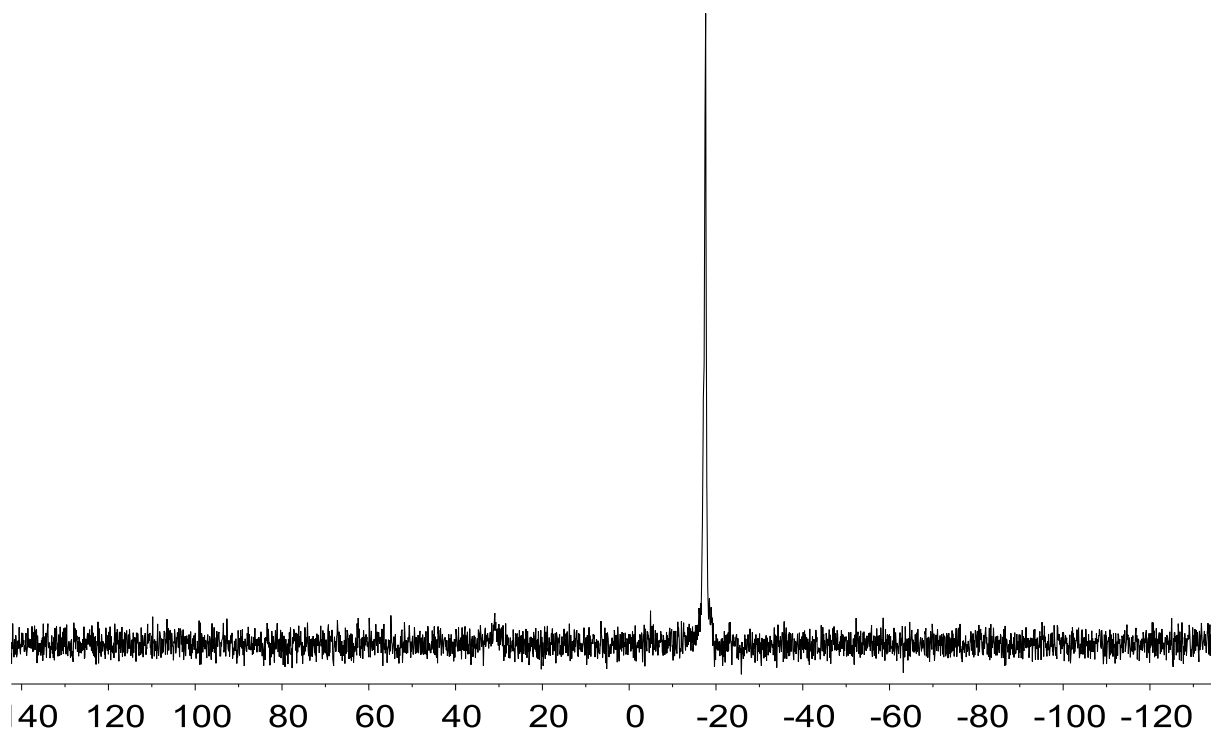


Fig. S18 ^{13}P -NMR spectrum of an *o*-StPPh₂-IP copolymer (Table 2, Run 2).

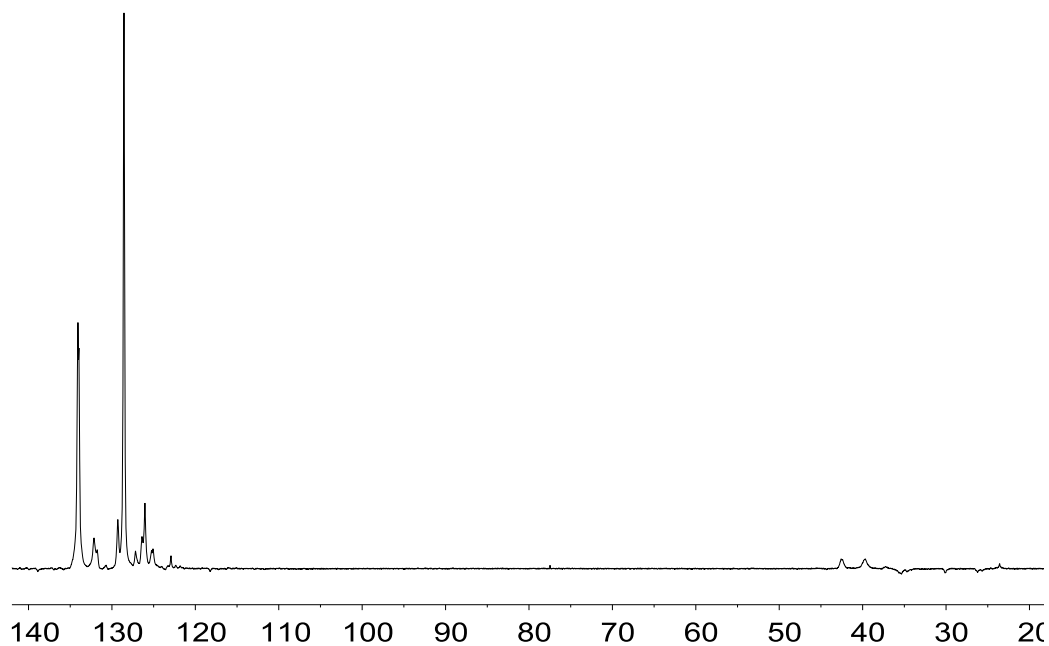


Fig. S19 DEPT135-¹³C NMR spectrum of a poly(*o*-StPPh₂-*alt*-IP) prepared by (C₅Me₄SiMe₃)Sc(CH₂C₆H₄NMe₂-*o*)₂.

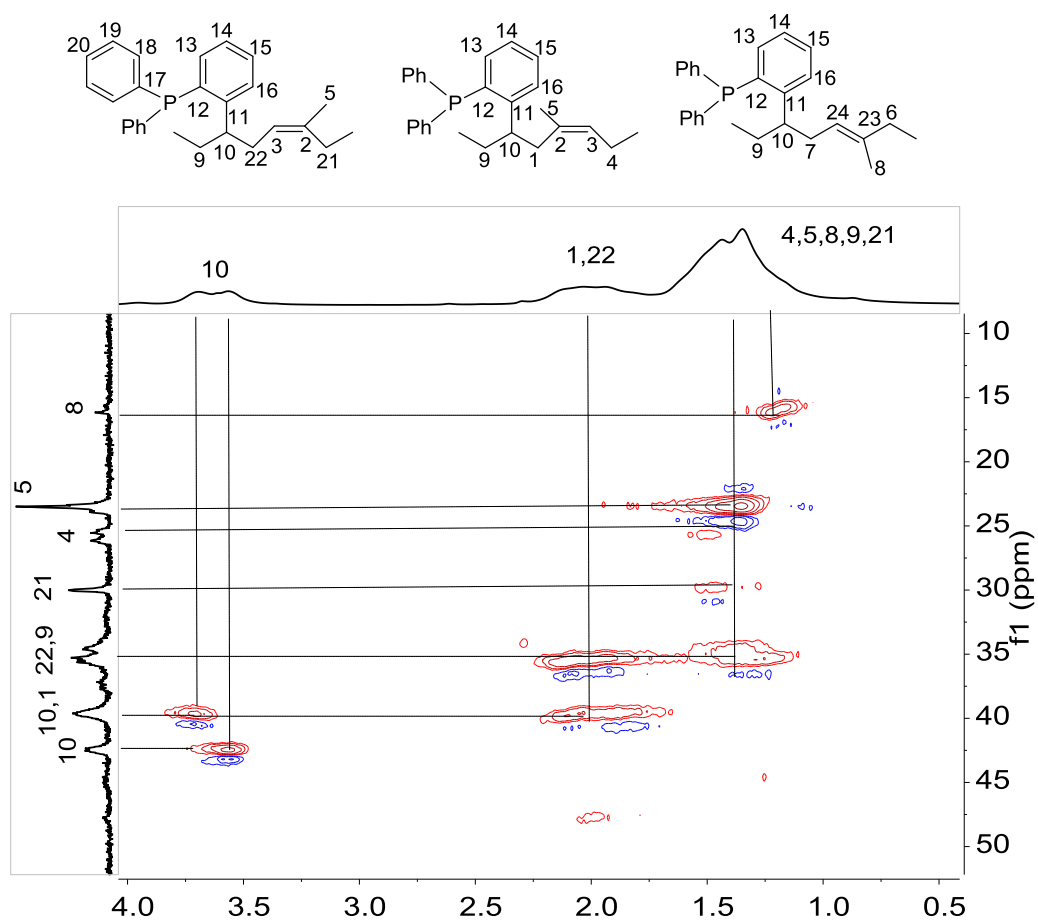


Fig. S20 HSQC NMR spectrum of a poly(*o*-StPPh₂-*alt*-IP) prepared by (C₅Me₄SiMe₃)Sc(CH₂C₆H₄NMe₂-*o*)₂.

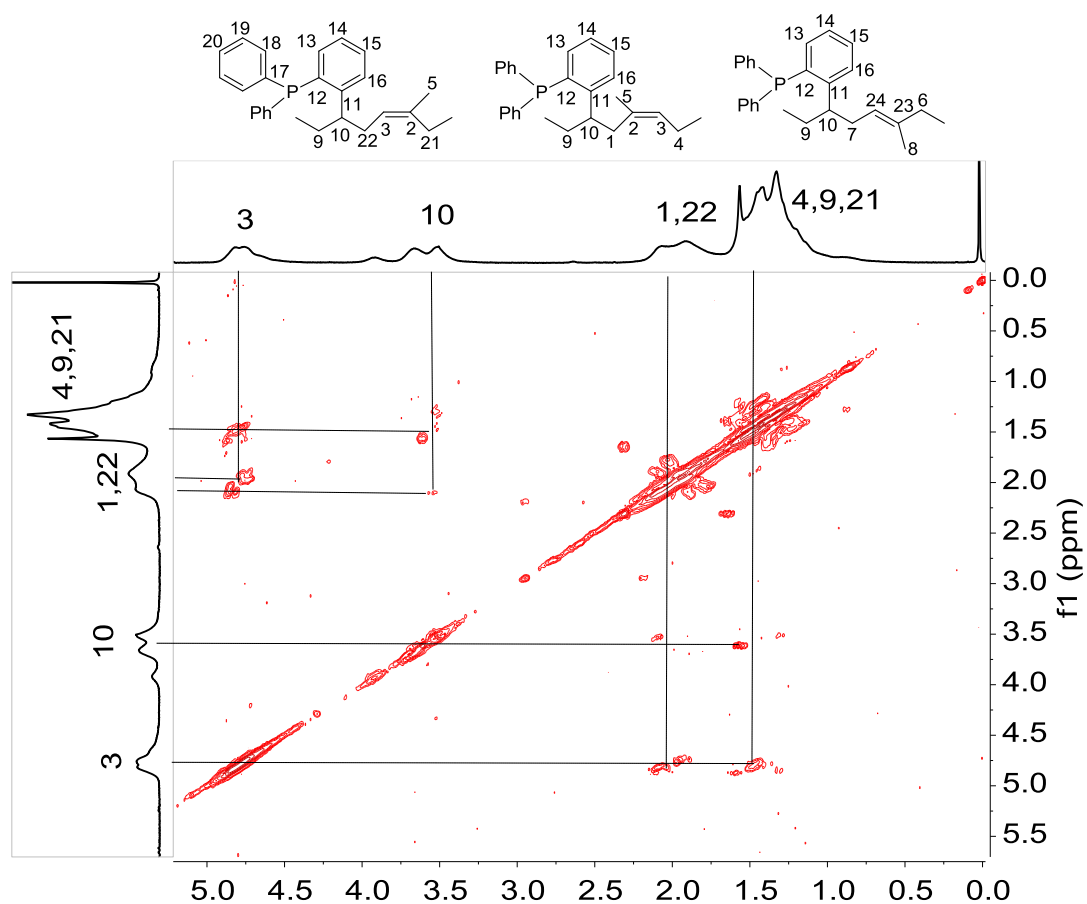


Fig. S21 gCOSY ^1H - ^1H NMR spectrum of a poly(*o*-StPPh₂-*alt*-IP) prepared by $(\text{C}_5\text{Me}_4\text{SiMe}_3)\text{Sc}(\text{CH}_2\text{C}_6\text{H}_4\text{NMe}_2\text{-}o)_2$.

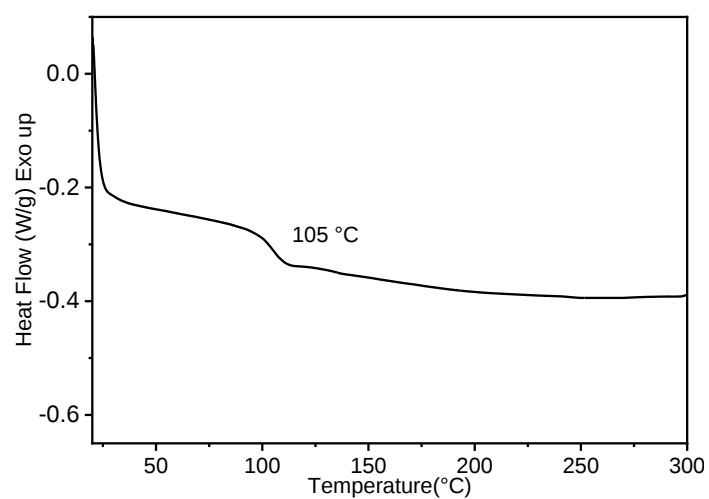


Fig. S22 DSC curve of a *p*-StPPh₂ homopolymer prepared by $(\text{C}_5\text{Me}_4\text{SiMe}_3)\text{Sc}(\text{CH}_2\text{C}_6\text{H}_4\text{NMe}_2\text{-}o)_2$.

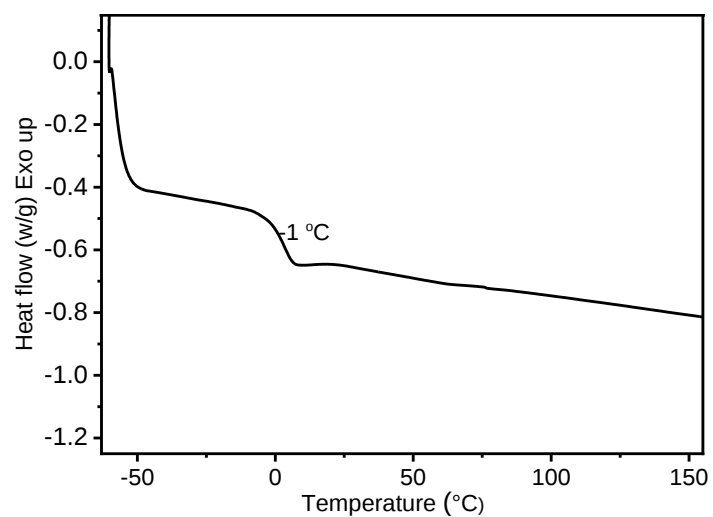


Fig. S23 DSC curve of an IP homopolymer prepared by $(C_5Me_4SiMe_3)Sc(CH_2C_6H_4NMe_2-o)_2$.

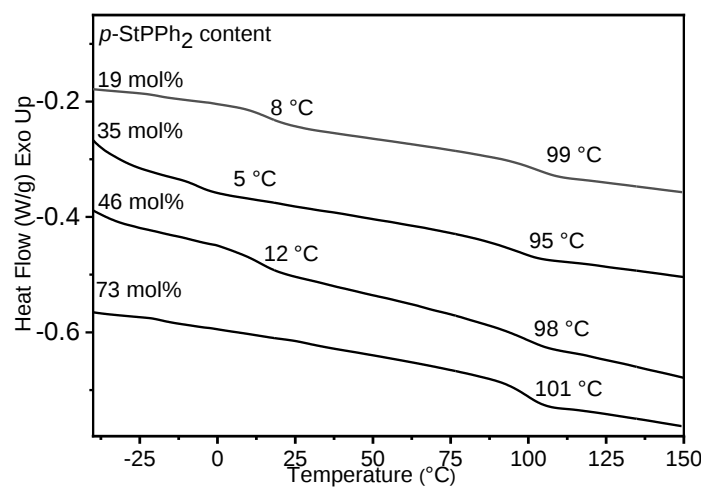


Fig. S24 DSC curves of *p*-StPPh₂-IP block copolymers with different composition.

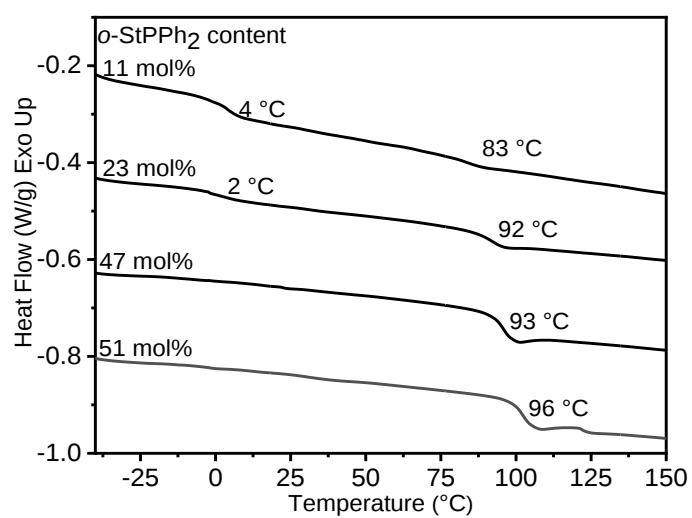


Fig. S25 DSC curves of *o*-StPPh₂-IP copolymers with different composition.

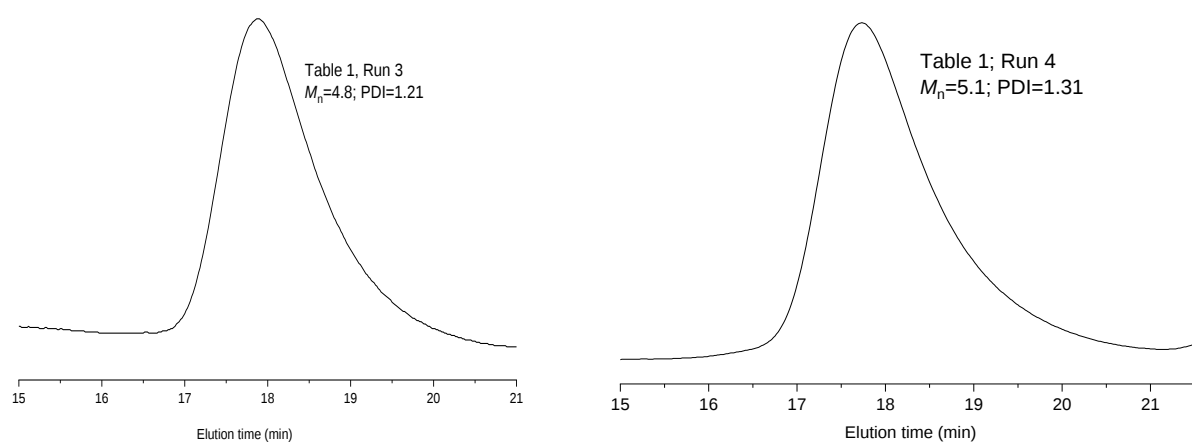


Fig. S26 GPC curves of *p*-StPPh₂-IP copolymers with RI detection mode using THF as eluent..

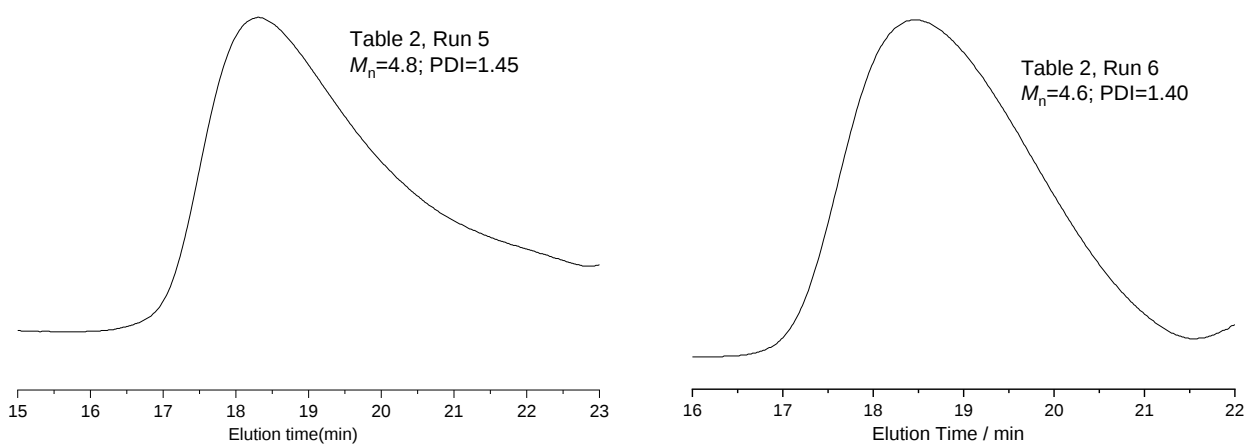


Fig. S27 GPC curves of *o*-StPPh₂-IP copolymers with RI detection mode using THF as eluent.

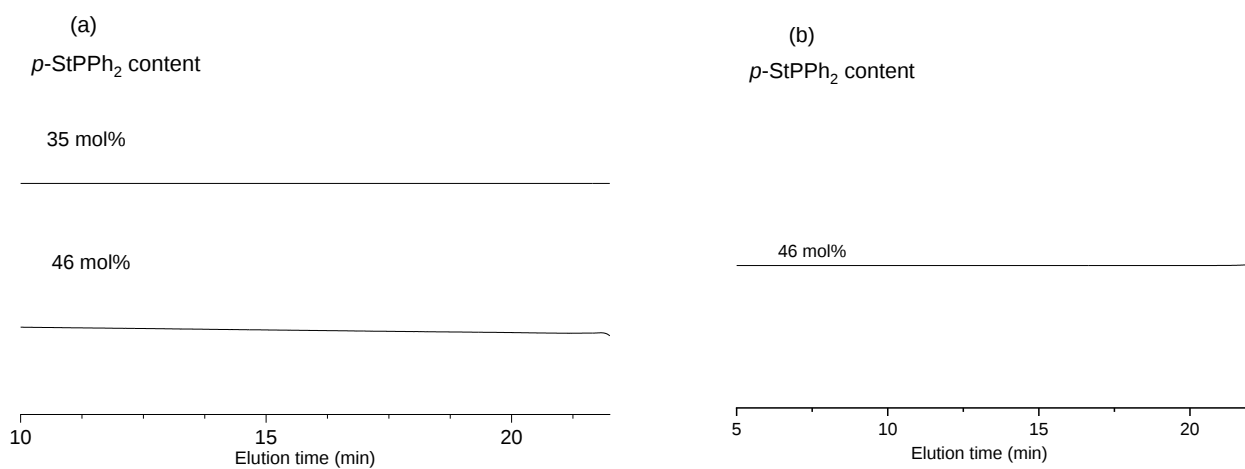


Fig. S28 GPC curves of *p*-StPPh₂-IP copolymers with RI detection mode using THF (a) and chloroform (b) as eluent.

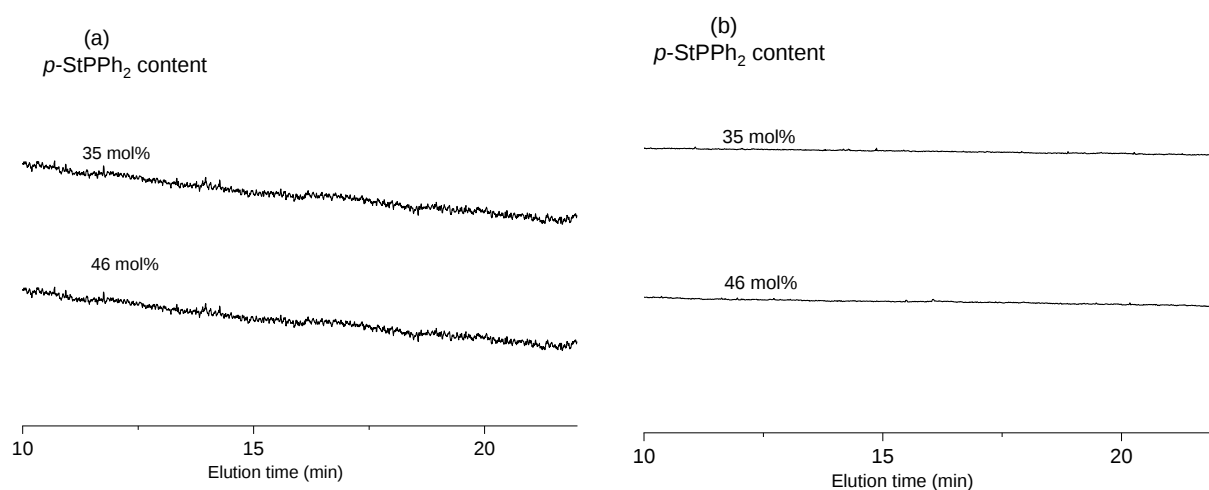


Fig. S29 GPC curves of p -StPPh₂-IP copolymers with RALS (a) and LALS (b) detection mode using THF as eluent.

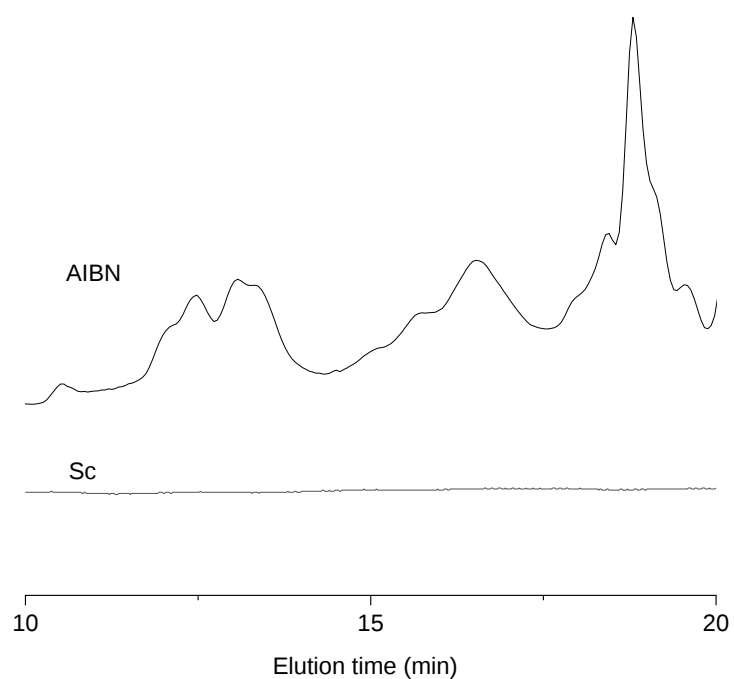


Fig. S30 GPC curves of p -StPPh₂ homopolymers prepared by $(C_5Me_4SiMe_3)Sc(CH_2C_6H_4NMe_2-o)_2$ and AIBN with RI detection mode using THF as eluent.

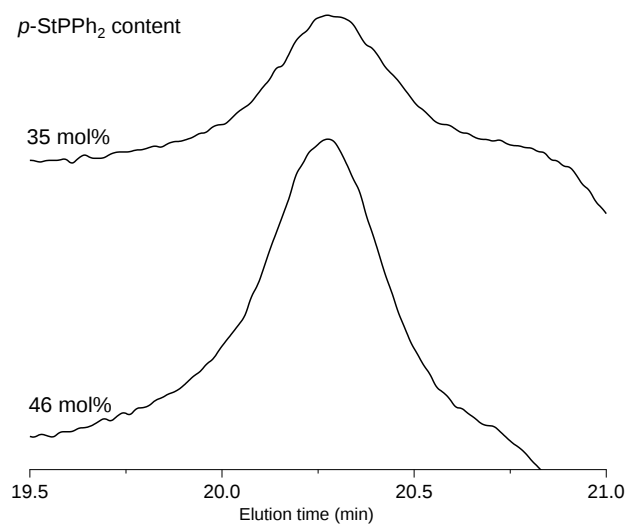


Fig. S31 GPC curves of *p*-StPPh₂-IP copolymers with UV detection mode using THF as eluent.

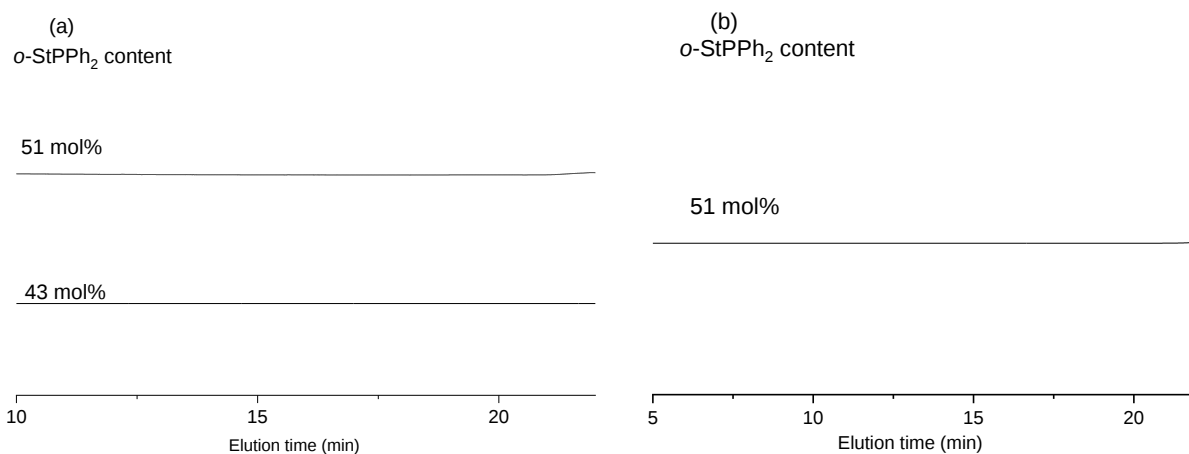


Fig. S32 GPC curves of *o*-StPPh₂-IP copolymers with RI detection mode THF (a) and chloroform (b) as eluent.

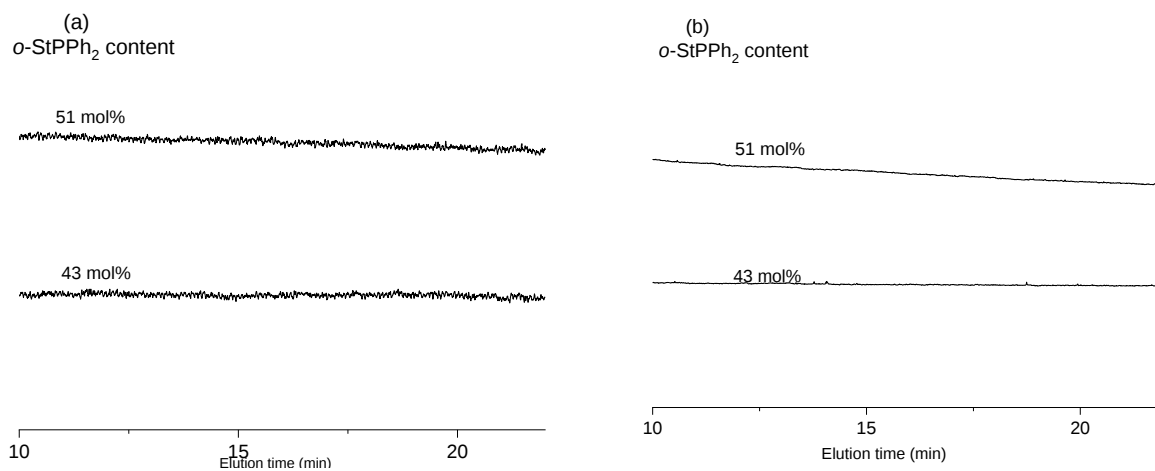


Fig. S33 GPC curves of *o*-StPPh₂-IP copolymers with RALS (a) and LALS (b) detection mode using THF as eluent.

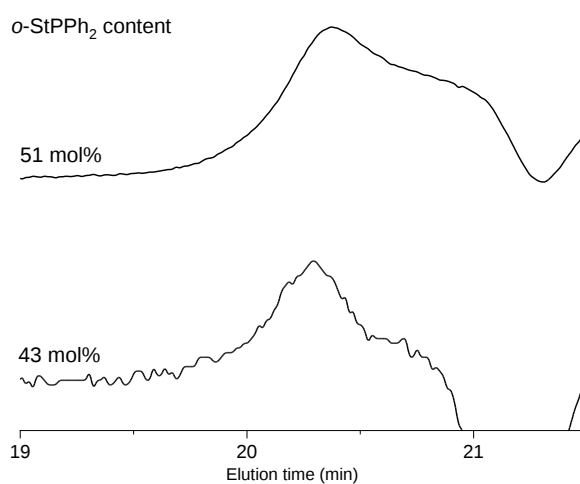


Fig. S34 GPC curves of *o*-StPPh₂-IP copolymers with UV detection mode using THF as eluent.

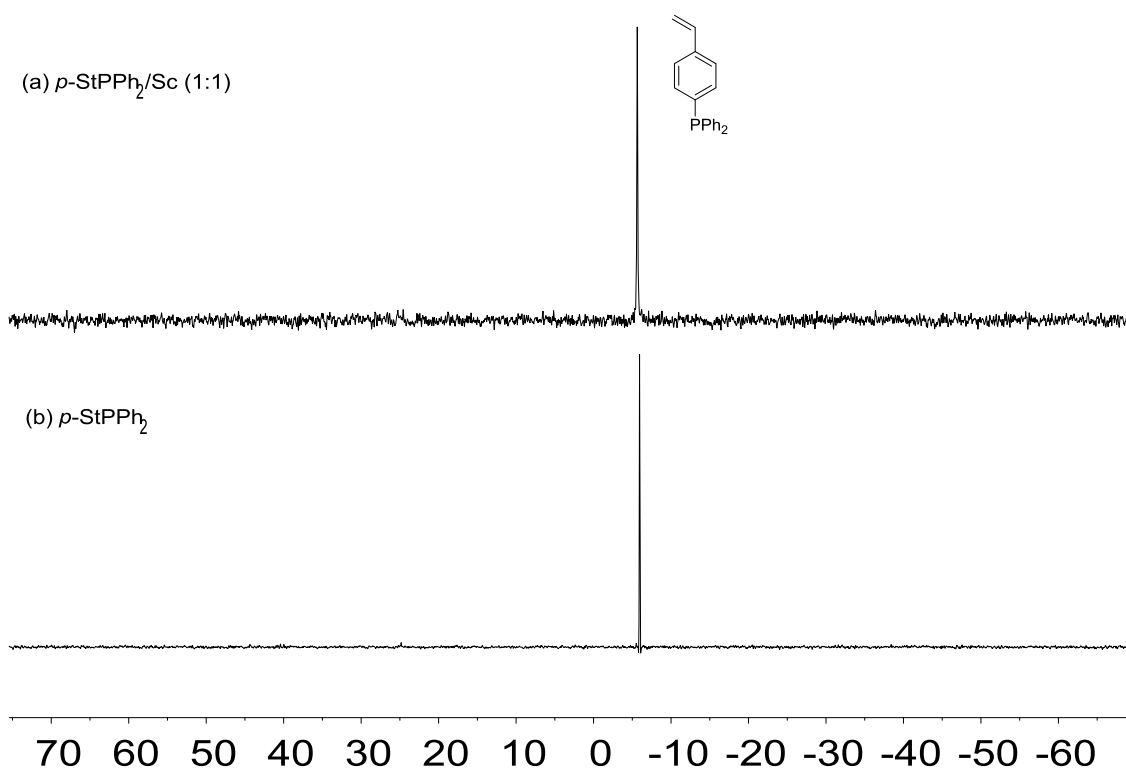


Fig. S35 ³¹P-NMR spectra of (a) *p*-StPPh₂ monomer with (C₅Me₄SiMe₃)Sc(CH₂C₆H₄NMe₂-*o*)₂/[Ph₃C][B(C₆F₅)₄] (molar ratio 1/1) and (b) *p*-StPPh₂ monomer only.

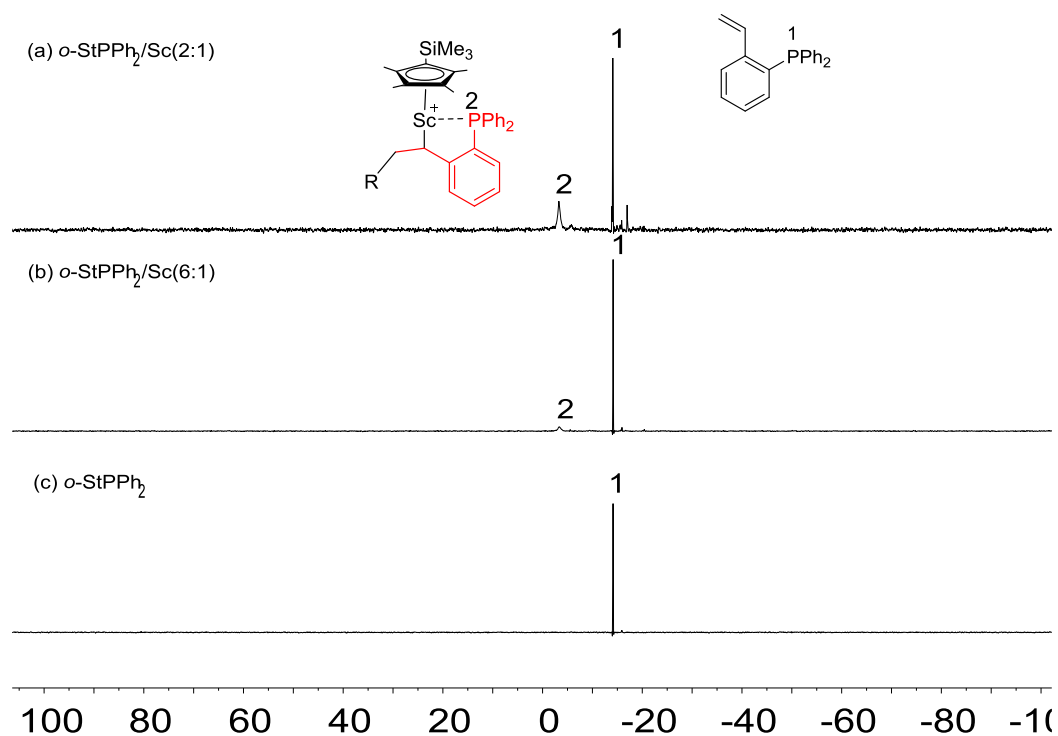
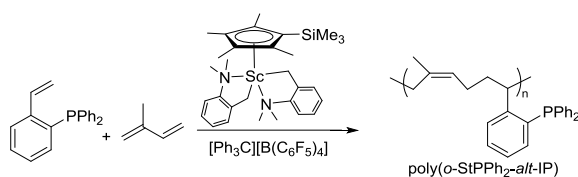


Fig. S36 ³¹P-NMR spectra of (a) *o*-StPPh₂ monomer with (C₅Me₄SiMe₃)Sc(CH₂C₆H₄NMe₂-*o*)₂/[Ph₃C][B(C₆F₅)₄] (molar ratio 2/1), (b) *o*-StPPh₂ monomer with (C₅Me₄SiMe₃)Sc(CH₂C₆H₄NMe₂-*o*)₂/[Ph₃C][B(C₆F₅)₄] (molar ratio 6/1) and (c) *o*-StPPh₂ monomer only.

Table S1 A kinetics investigation of *p*-StPPh₂/IP (feed ratio 250/250) copolymerization^a

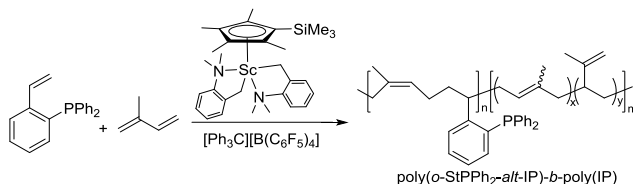
Run	Time (min)	Yield (%)	Composition (mol%) ^b		Conversions (mol%)	
			<i>p</i> -StPPh ₂	IP (1,4/3,4)	<i>p</i> -StPPh ₂	IP
1	5	15	0	100(30/70)	0	82
2	10	17	0	100(32/68)	0	92
3	15	19	0	100(34/66)	0	99
4	20	21	3	97(34/66)	3	99
5	25	28	10	90(34/66)	11	99
6	35	48	27	73(34/66)	36	99
7	50	78	42	58(34/66)	74	99
8	60	91	47	53(34/66)	91	99

^a Conditions: [Sc] (5 μmol), [Ph₃C][B(C₆F₅)₄] (5 μmol), toluene (3 mL), 25 °C. ^b Determined by ¹H-NMR.

Table S2 A kinetics investigation of *o*-StPPh₂/IP (feed ratio 250/250) copolymerization^a

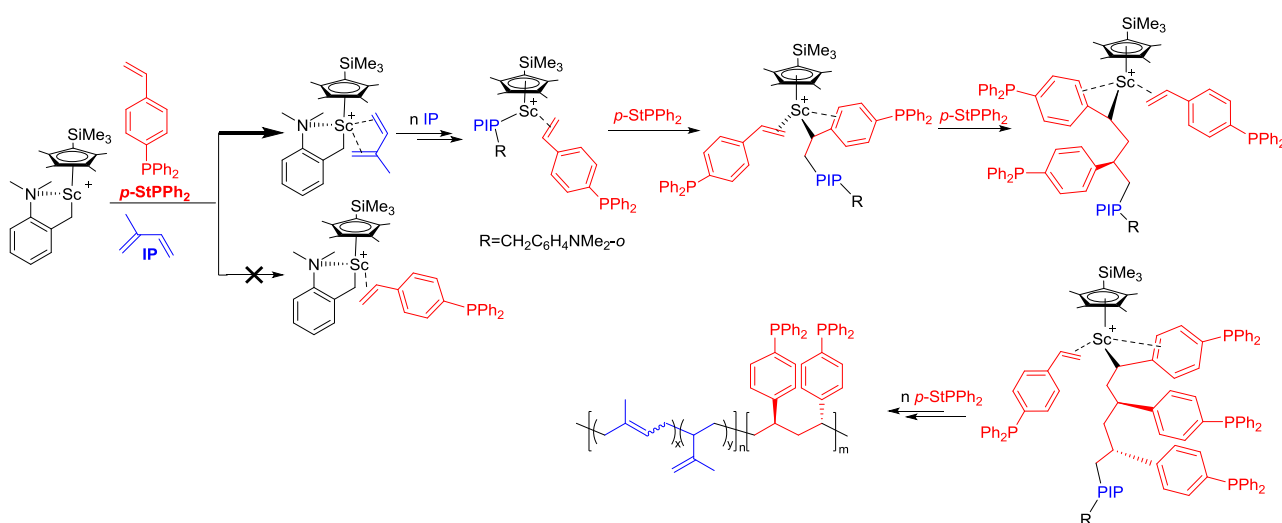
Run	Time (min)	Yield (%)	Composition (mol%) ^b		Conversions (mol%)	
			<i>o</i> -StPPh ₂	IP(1,4/3,4)	<i>o</i> -StPPh ₂	IP
1	60	42	51	49(100/0)	42	40
2	120	66	51	49(100/0)	66	64
3	240	97	51	49(100/0)	97	94

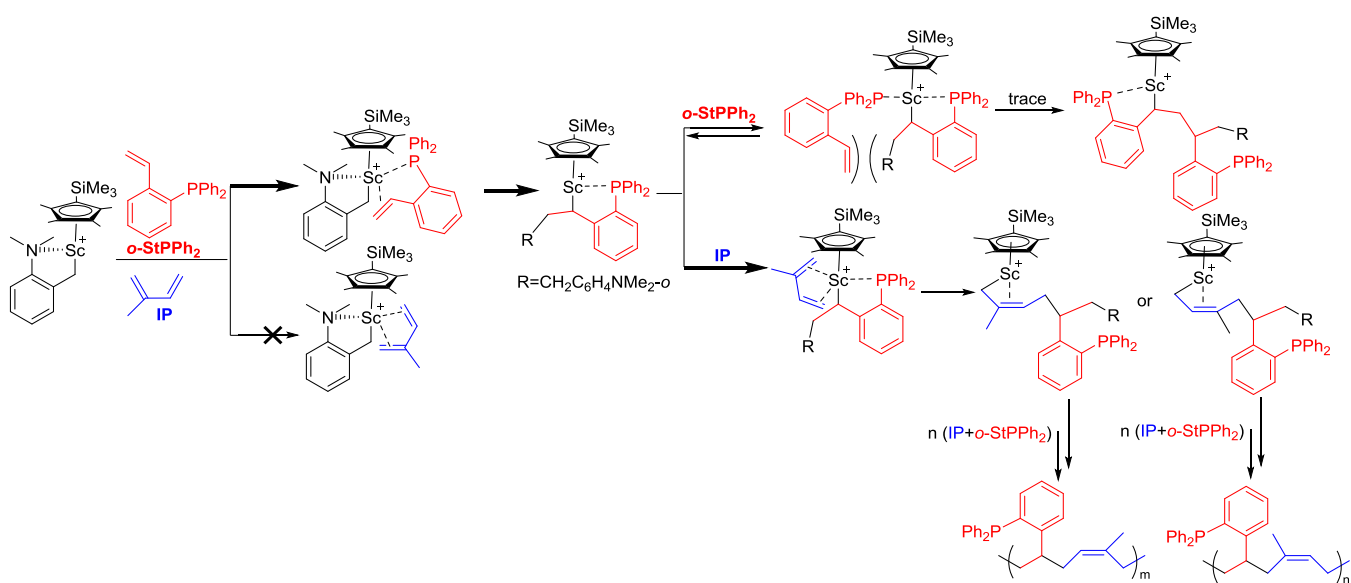
^a Conditions: [Sc] (5 μmol), [Ph₃C][B(C₆F₅)₄] (5 μmol), toluene (3 mL), 25 °C. ^b Determined by ¹H-NMR.

Table S3 A kinetics investigation of *o*-StPPh₂/IP (feed ratio 250/375) copolymerization^a

Run	Time (min)	Yield (%)	Composition (mol%) ^b		Conversions (mol%)	
			<i>o</i> -StPPh ₂	IP(1,4/3,4)	<i>o</i> -StPPh ₂	IP
1	0.2	90	51	49(100/0)	99	65
2	0.5	94	44	56(81/19)	99	83
3	1	96	42	58(77/23)	99	93
4	8	98	38	62(71/29)	99	96
5	10	98	38	62(71/29)	99	96

^a Conditions: [Sc] (5 μmol), [Ph₃C][B(C₆F₅)₄] (5 μmol), toluene (3 mL), 25 °C. ^b Determined by ¹H-NMR.

**Scheme S1** Possible mechanism of the copolymerization of *p*-StPPh₂ and IP by (C₅Me₄SiMe₃)Sc(CH₂C₆H₄NMe₂-*o*)₂.



Scheme S2 Possible mechanism of the alternating copolymerization of *o*-StPPh₂ and IP by $(C_5Me_4SiMe_3)Sc(CH_2C_6H_4NMe_2-o)_2$.