

Electronic Supplementary Information

Synthesis and characterization of polysulfides formed by the inverse vulcanisation of cyclosiloxanes with sulfur

Kun Woo Park,^{ab} Elizabeth A Tafili,^{ab} Flora Fan,^{ac} Zoran Zujovic^a and Erin M Leita^{*ab}

^a School of Chemical Sciences, University of Auckland, Private Bag, 92019, Auckland, 1142, New Zealand

^b The MacDiarmid Institute for Advanced Materials and Nanotechnology, Victoria University of Wellington, Wellington 6140, New Zealand

^c Department of Chemistry, University of California, Berkeley, Berkeley, California 94720, United States



Figure S1: Photo of X-poly(S-*r*-TVTSi) series ($X = \text{wt\% S}$). From left to right: $X = 20, 30, 40, 50, 60, 70$ and 80 .

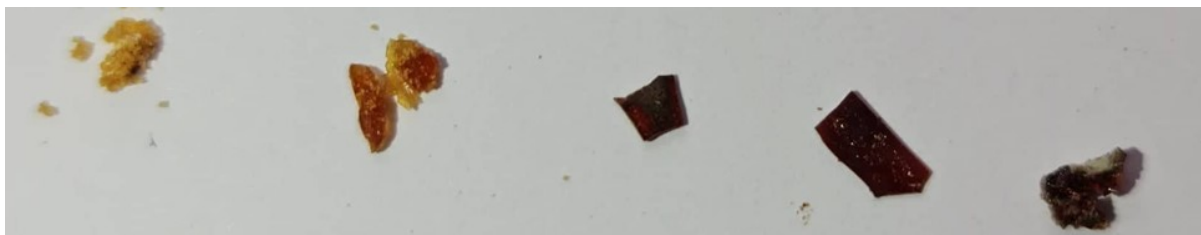


Figure S2: Photo of X-poly(S-*r*-PVPSi) series ($X = \text{wt\% S}$). From left to right: $X = 10, 20, 30, 40, 50$.

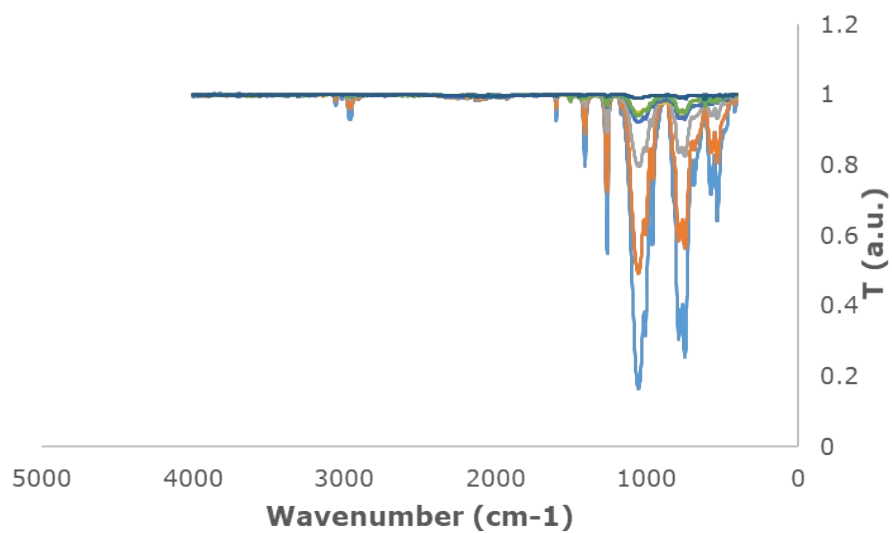


Figure S3: FTIR spectra of X-poly(S-*r*-TVTSi) (X = wt % S), X = 20 (light blue), X = 30 (orange), X = 40 (grey), X = 50 (yellow), X = 60 (blue), X = 70 (green), X = 80 (navy)

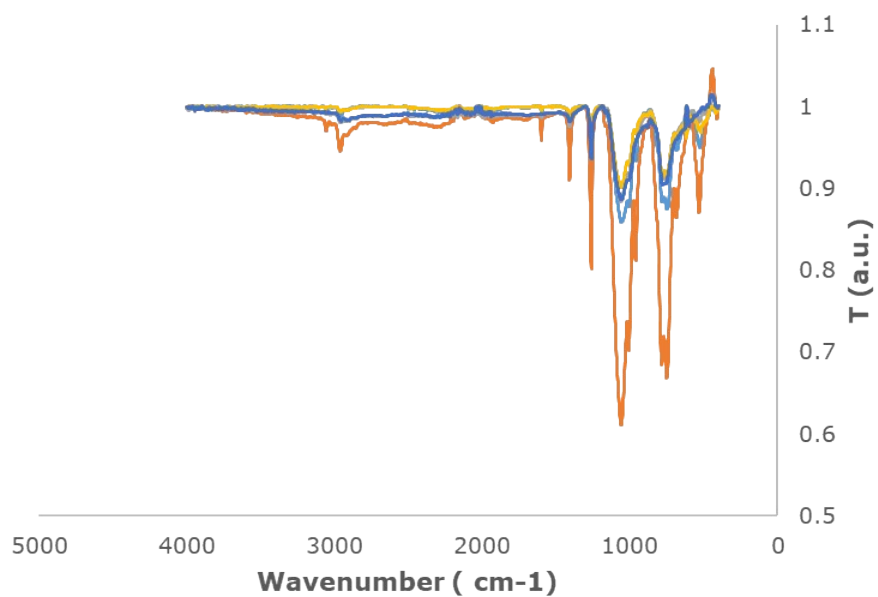


Figure S4: FTIR spectra of X-poly(S-*r*-PVPSi) (X = wt % S), X = 10 (light blue), X = 20 (orange), X = 30 (grey), X = 40 (yellow), X = 50 (dark blue).

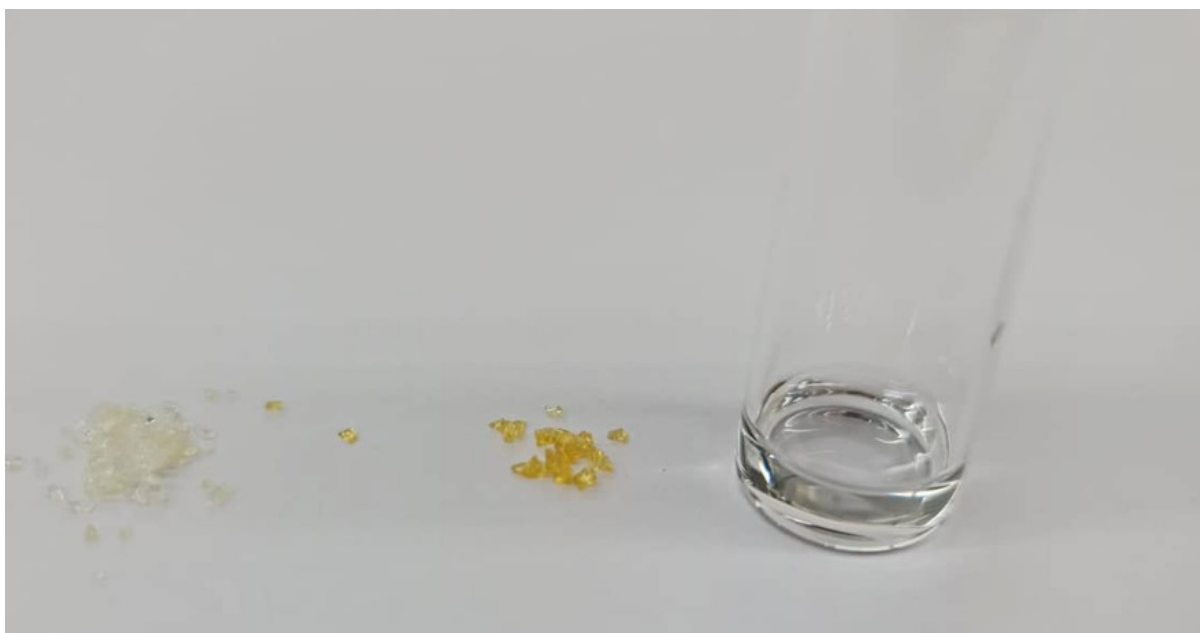


Figure S5: Tetravinyl-tetramethyl-cyclotetrasiloxane (TVTSi) after heating at 230°C for 3h (left), pentavinyl-pentamethyl-cyclopentasiloxane (PVPSi) after heating at 230°C for 3 h (middle) and unheated TVTSi (right).

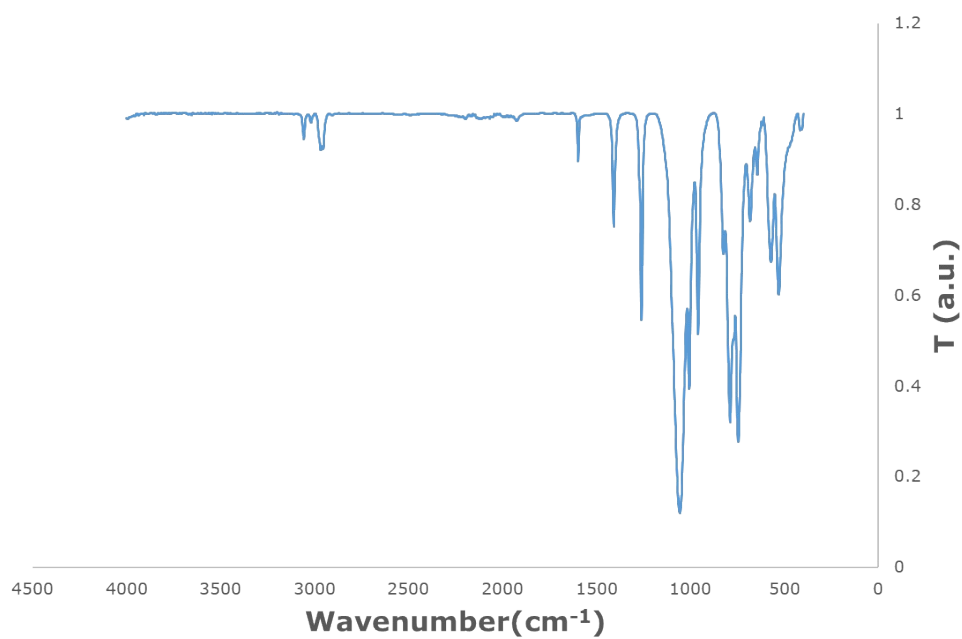


Figure S6: FTIR spectrum of TVTSi.

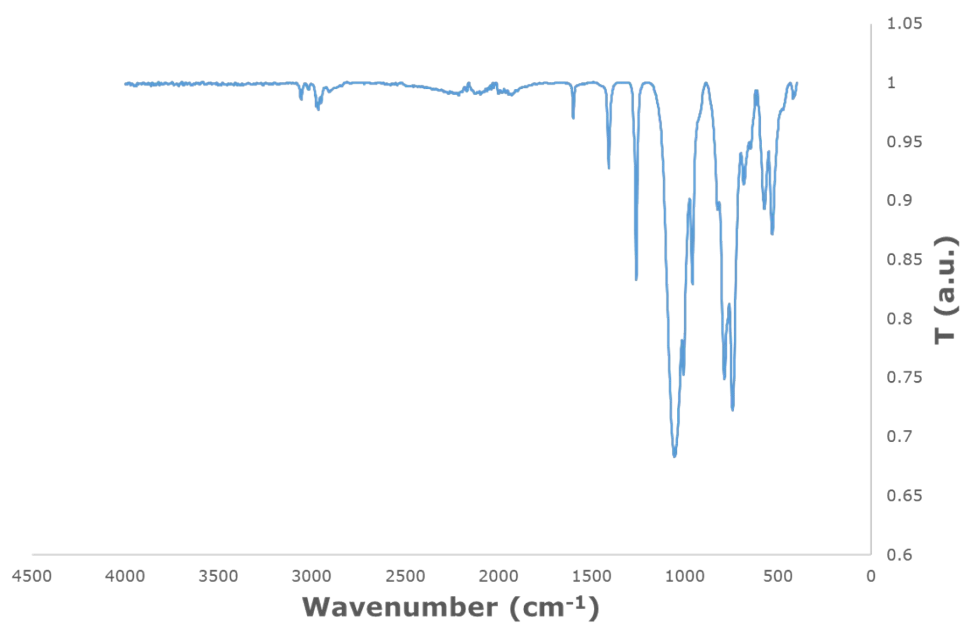


Figure S7: FTIR spectrum of TVTSi heated.

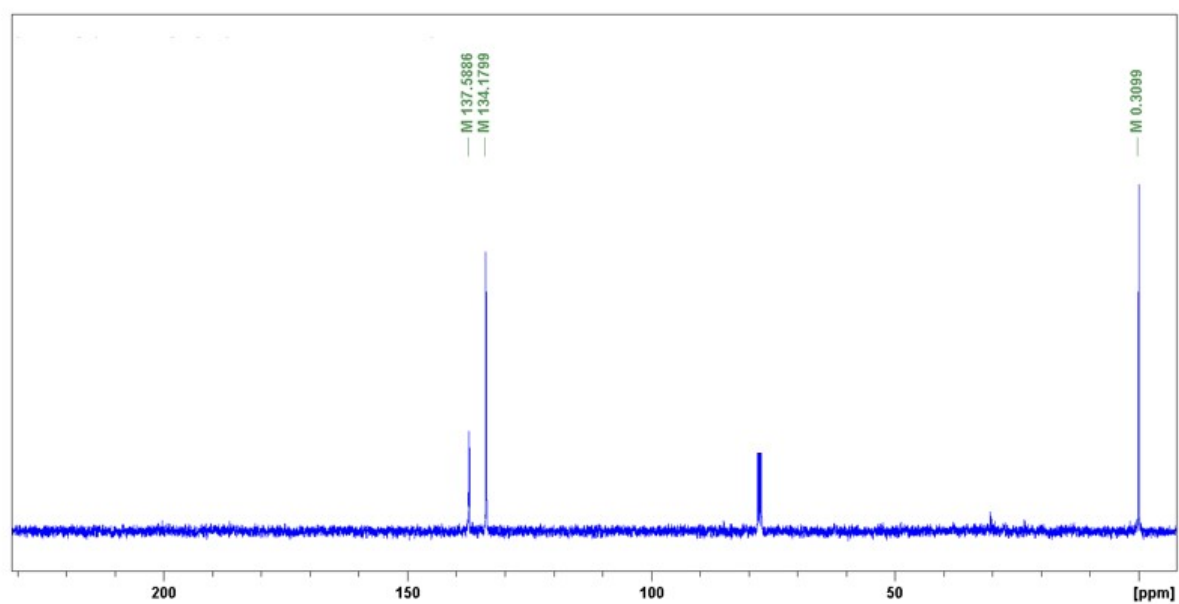


Figure S8: ¹³C{¹H} NMR spectrum of tetravinyl-tetramethyl-cyclotetrasiloxane in CDCl₃.

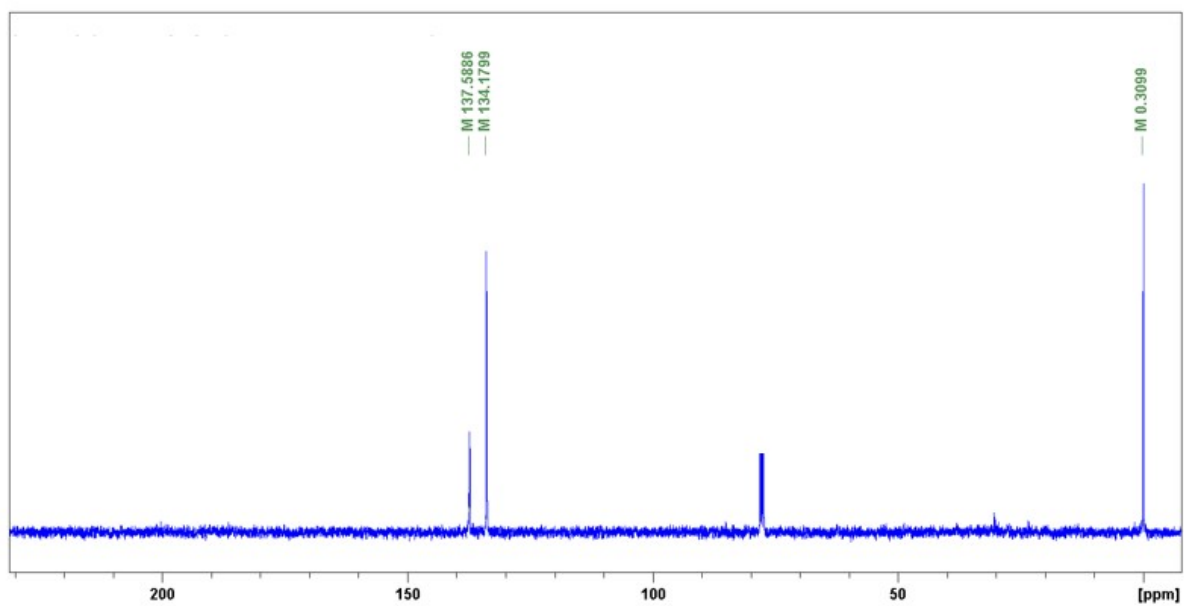


Figure S9: $^{13}\text{C}\{^1\text{H}\}$ NMR spectrum of pentavinyl-pentamethyl-cyclopentasiloxane in CDCl_3 .

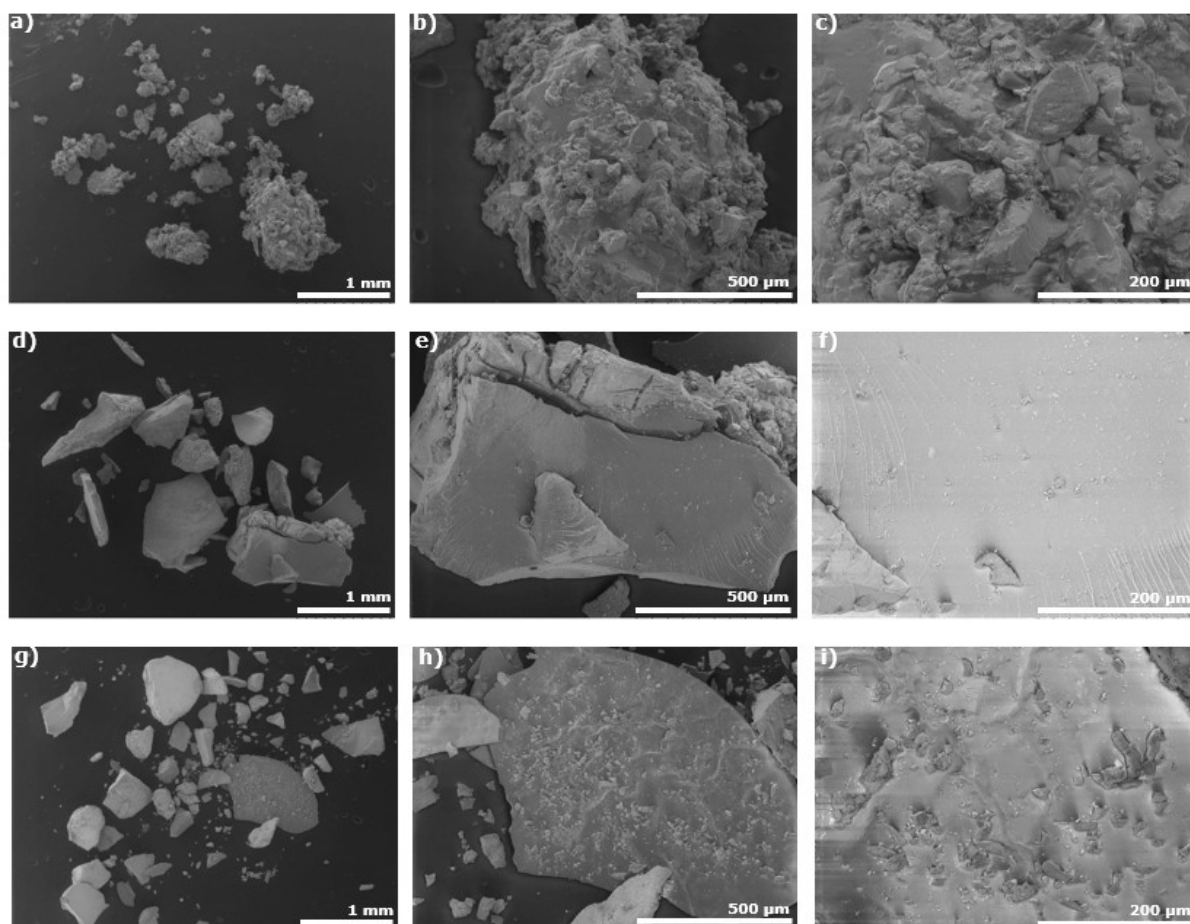


Figure S10: Representative SEM images for different wt% S polysulfides; top row: 20-poly(S-*r*-TVTSi), middle row: 40-poly(S-*r*-TVTSi), bottom row: 80-poly(S-*r*-TVTSi).

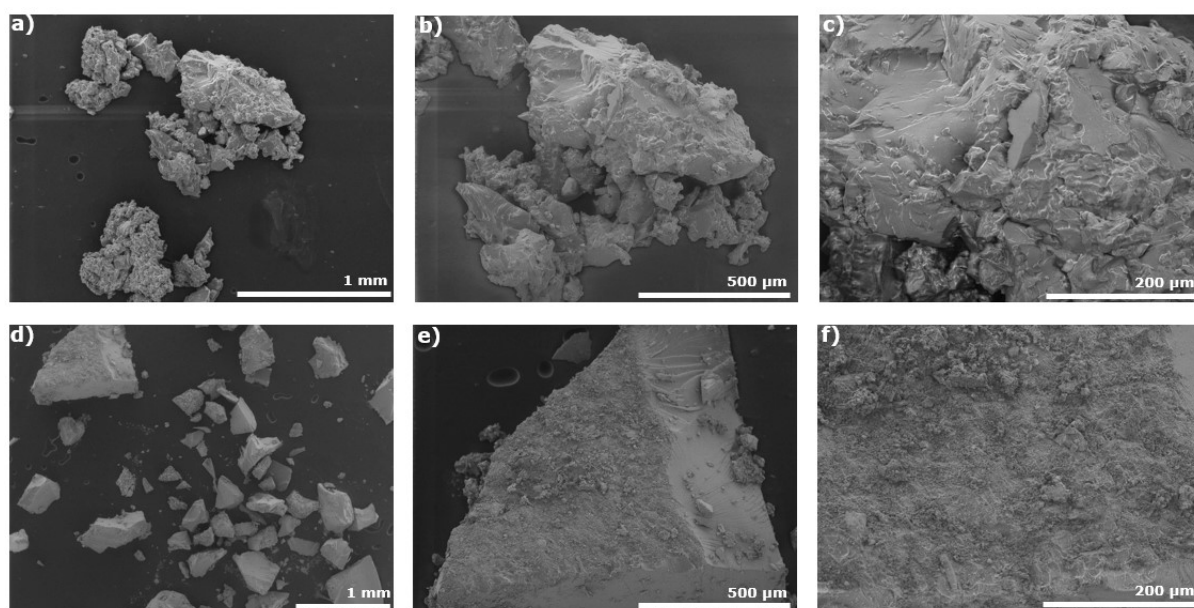


Figure S11: Representative SEM images for different wt % S polysulfides; top row: 20-poly(S-*r*-PVPSi), bottom row: 40-poly(S-*r*-PVPSi)

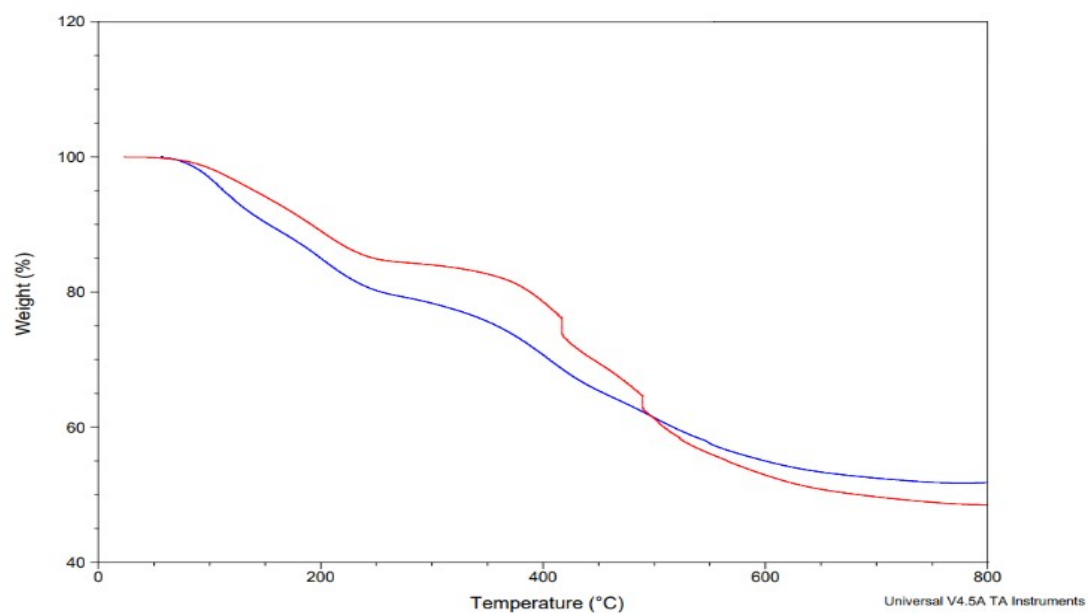


Figure S12: Thermogram of 20-poly(S-*r*-TVTSi) repeated twice.

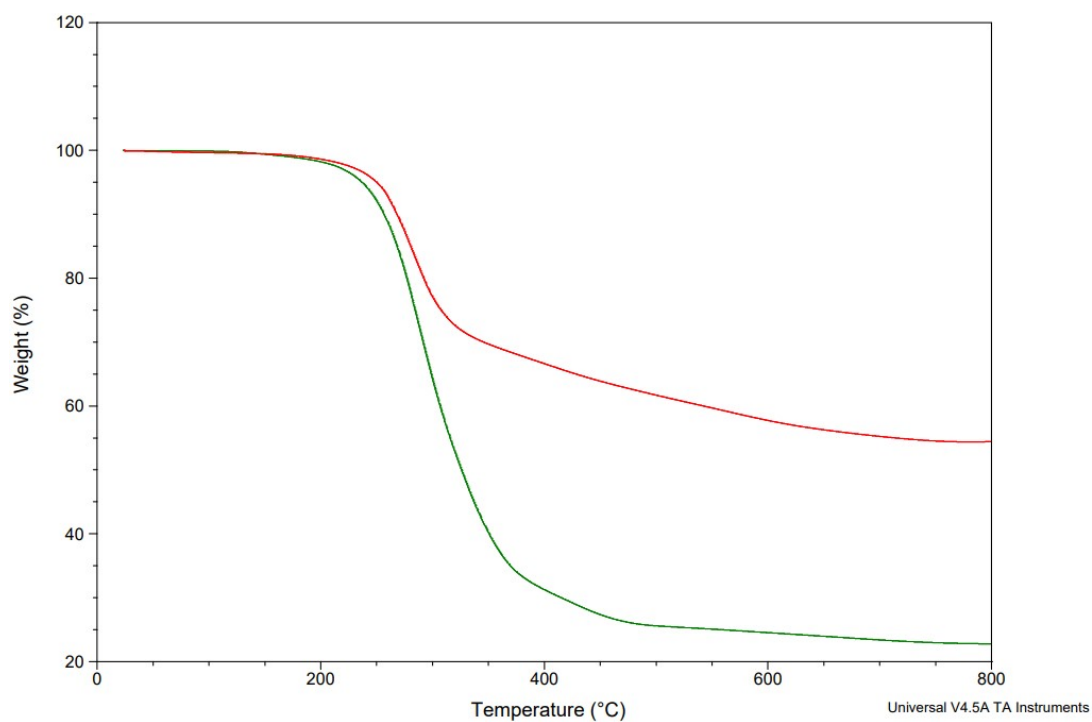


Figure S13: Thermogram comparison of 80-poly(S-*r*-TVTSi) before (green) and after (red) washing with CS₂.

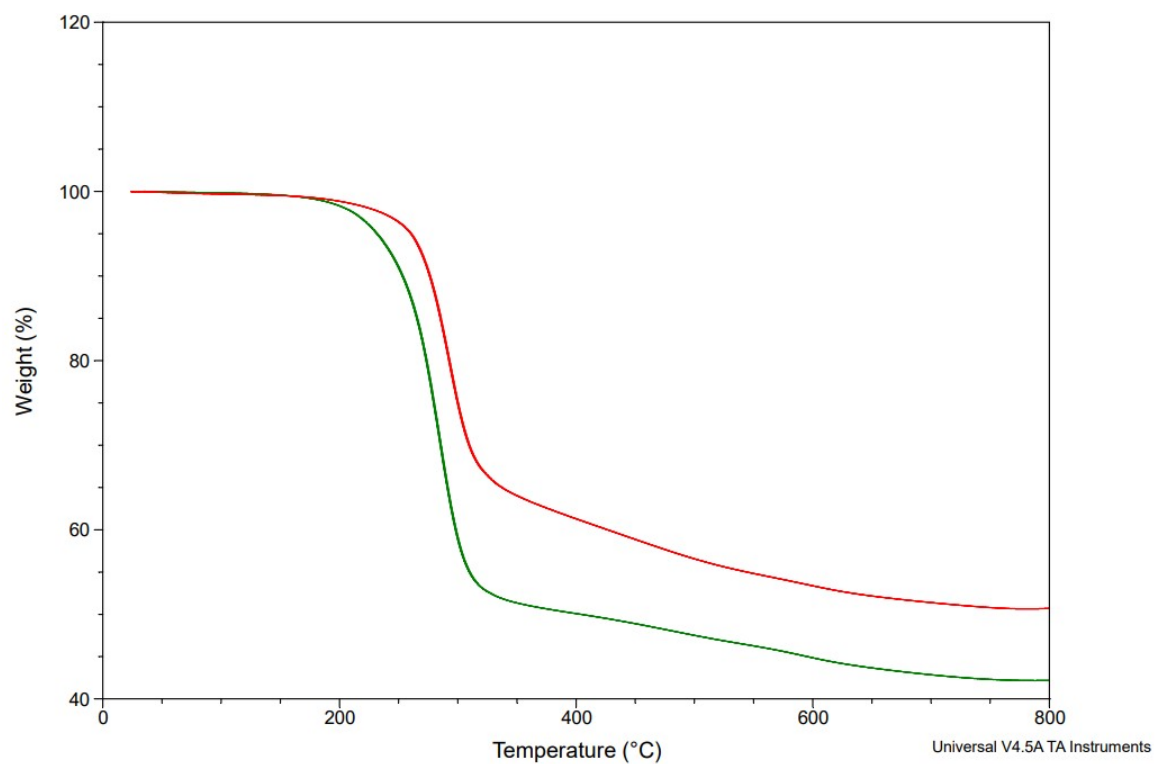


Figure S14: Thermogram comparison of 50-poly(S-*r*-PVPSi) before (green) and after (red) washing with CS₂.

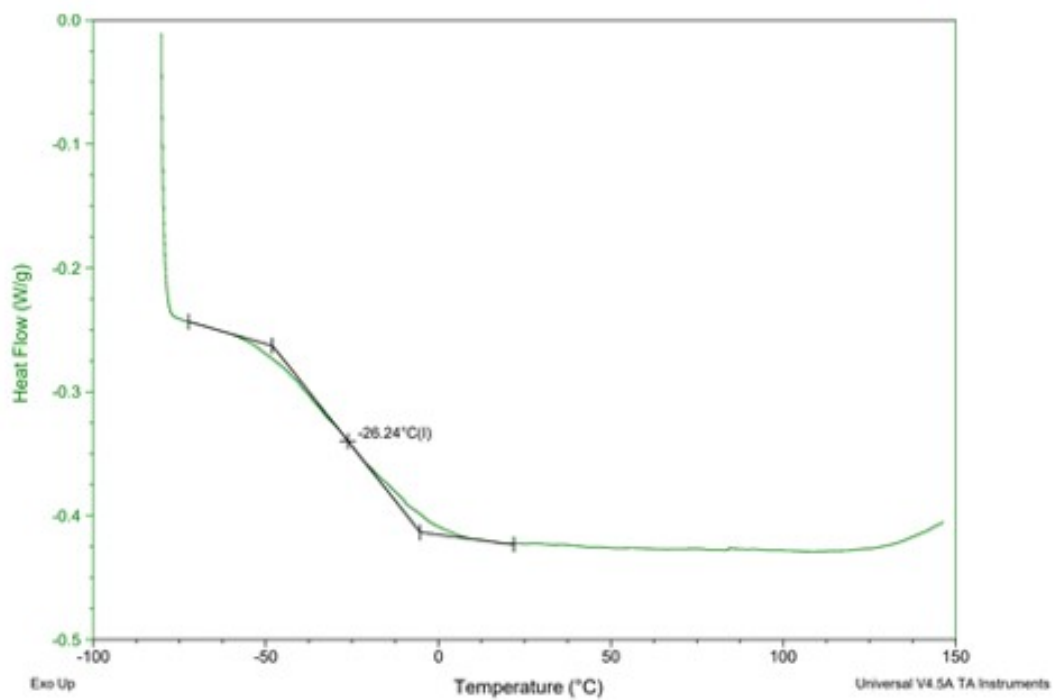


Figure S15: DSC scan of 20-poly(S-*r*-TVTSi) with T_g measured at -26.24°C

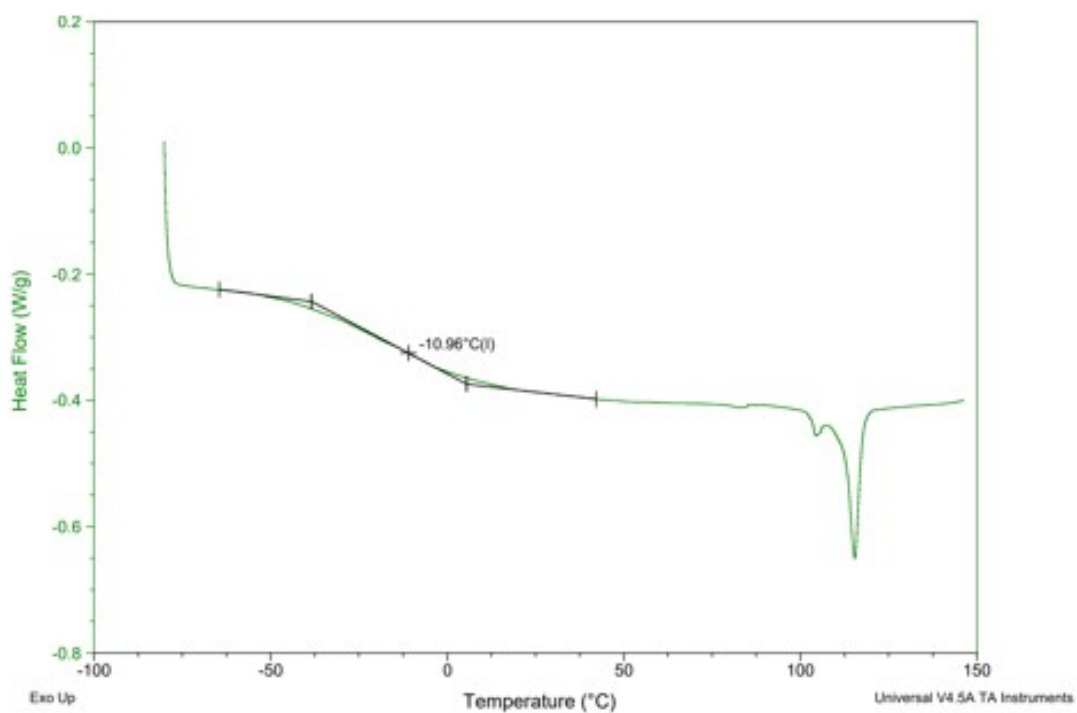


Figure S16: DSC scan of 30-poly(S-*r*-TVTSi) with T_g measured at -10.96°C and elemental sulfur shown to melt at 120°C .

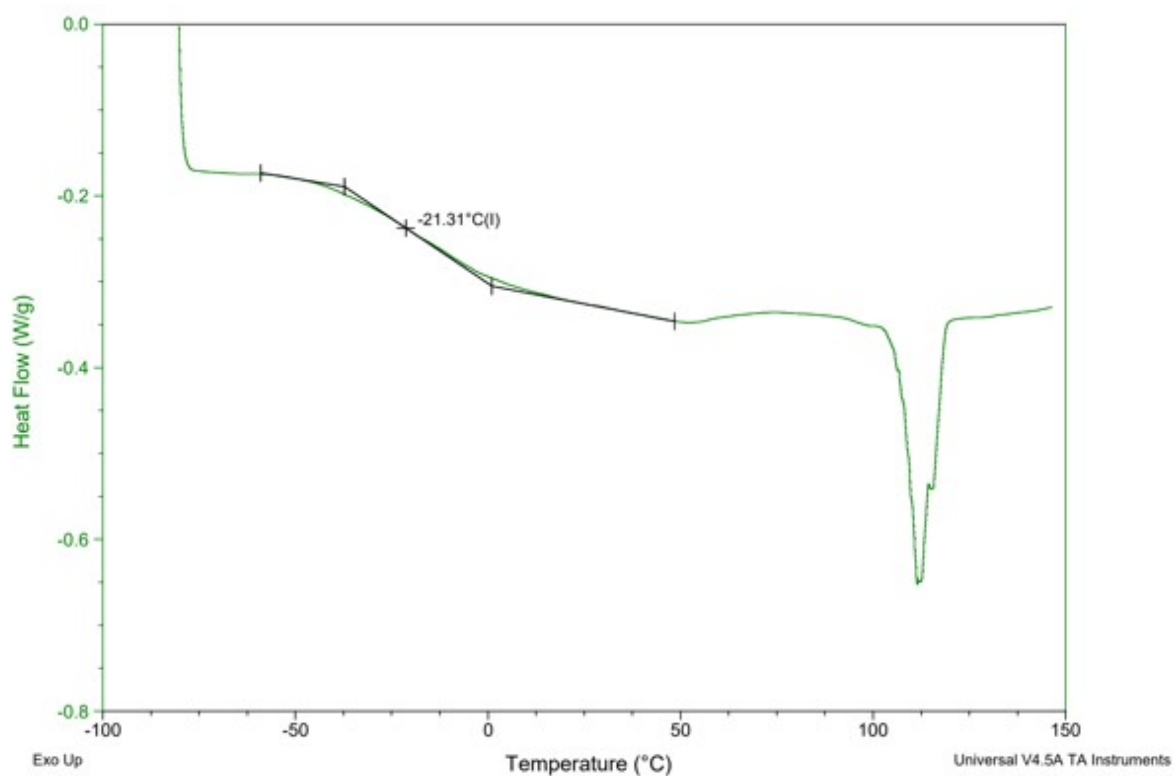


Figure S17: DSC scan of 40-poly(S-*r*-TVTSi) with T_g measured at -21.31°C and elemental sulfur shown to melt at 120°C .

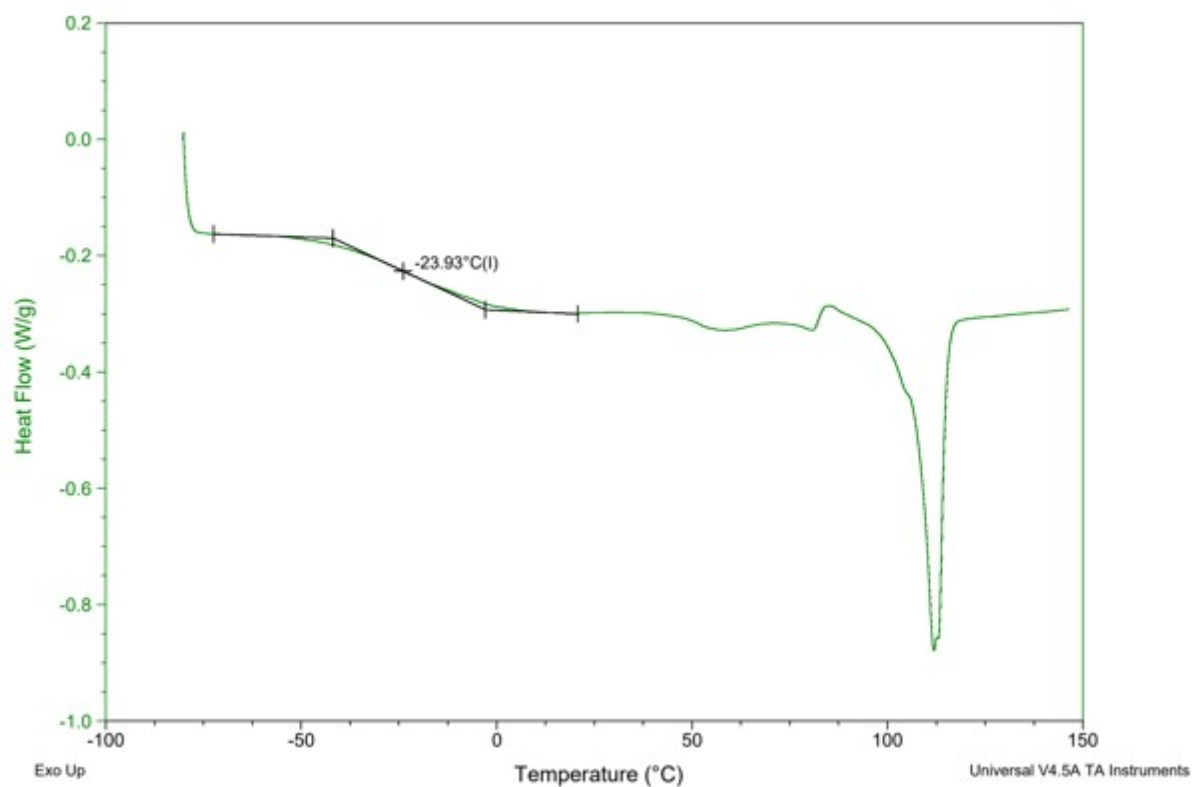


Figure S18: DSC scan of 50-poly(S-*r*-TVTSi) with T_g measured at -23.93°C and elemental sulfur shown to melt at 120°C.

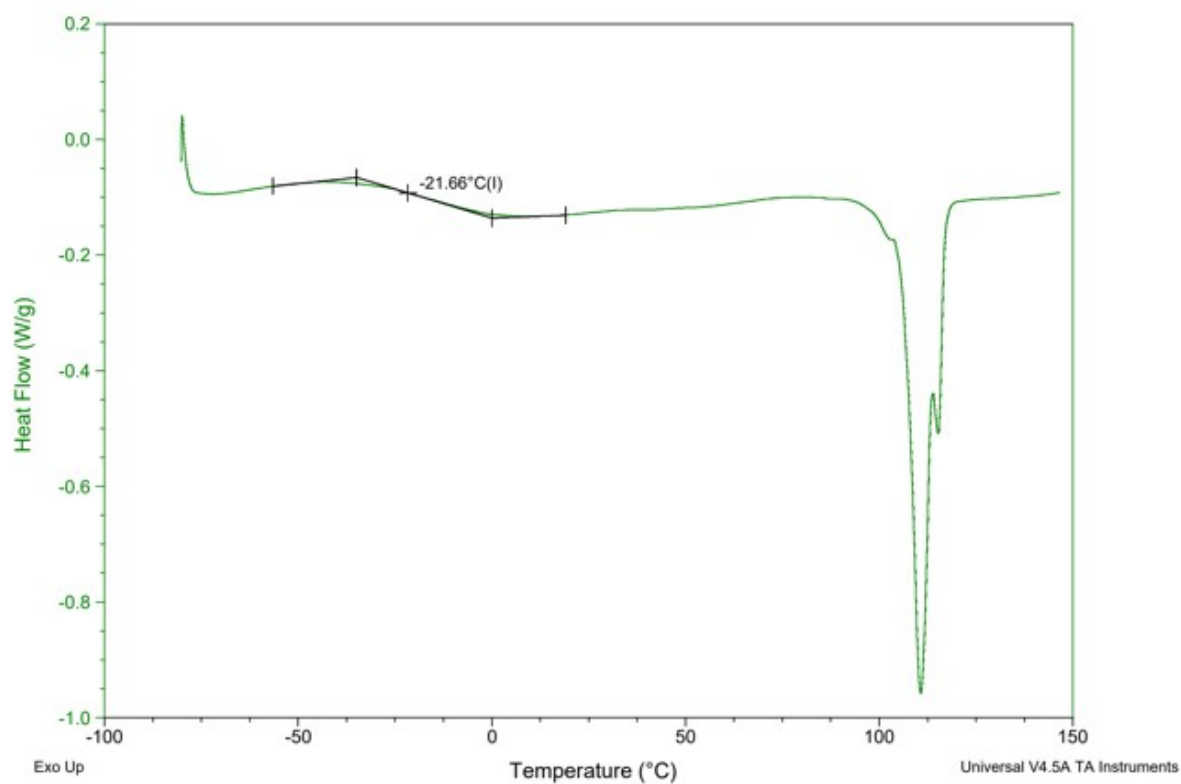


Figure S19: DSC scan of 60-poly(S-*r*-TVTSi) with T_g measured at -21.66°C and elemental sulfur shown to melt at 120°C.

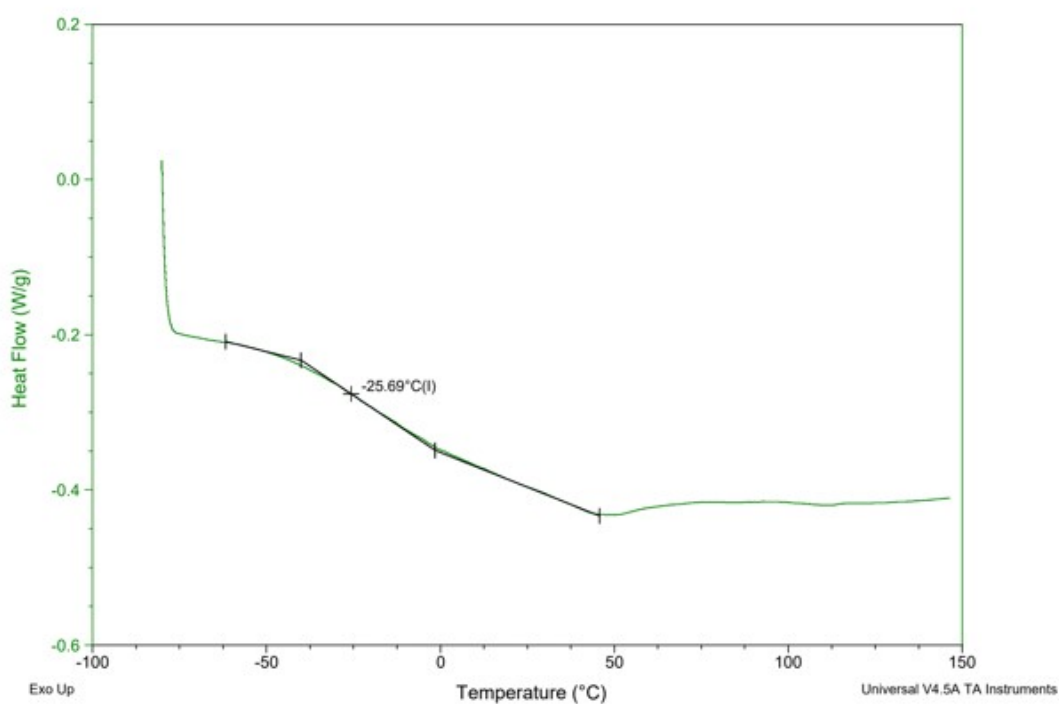


Figure S20: DSC scan of 70-poly(S-*r*-TVTSi) with T_g measured at -25.69°C.

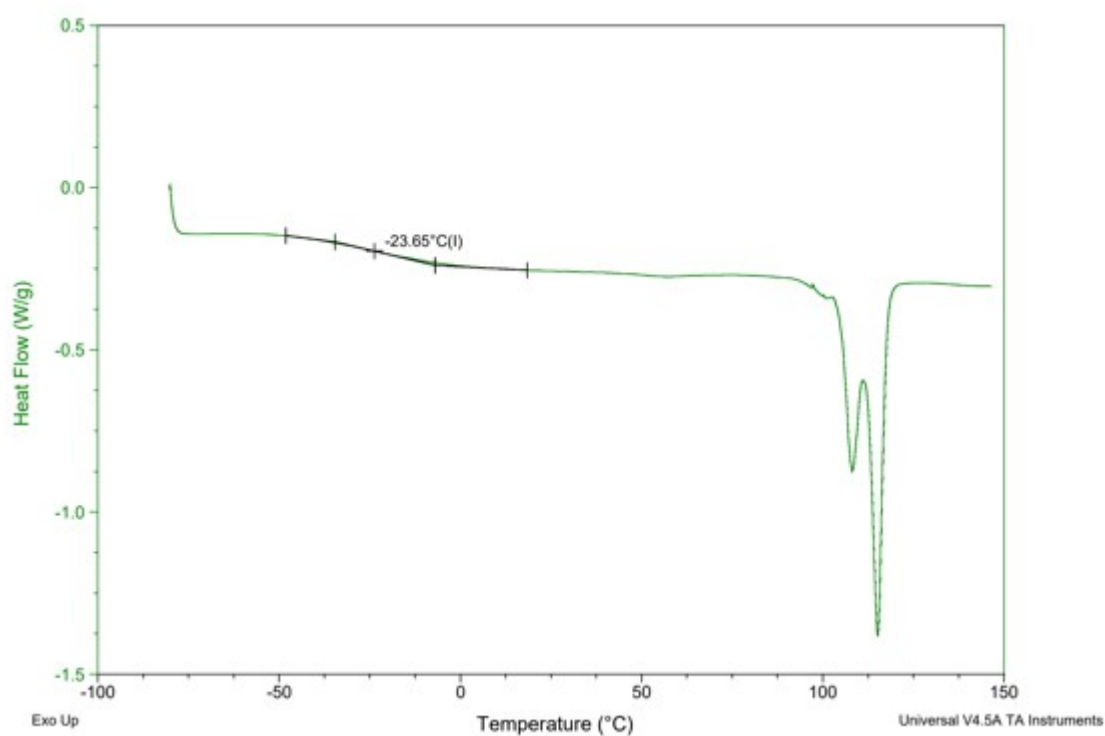


Figure S21: DSC scan of 80-poly(S-*r*-TVTSi) with T_g measured at -23.65°C and elemental sulfur shown to melt at 120°C.

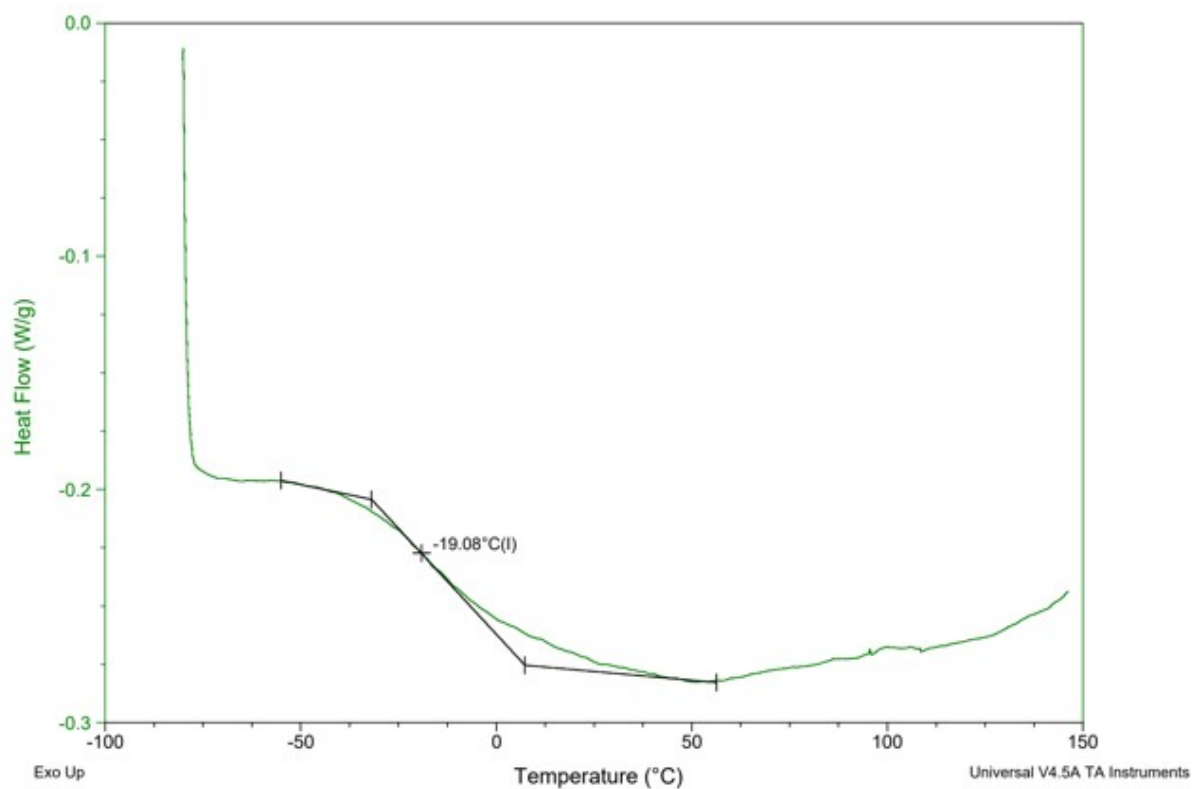


Figure S22: DSC scan of 10-poly(S-r-PVPSi) with T_g measured at -19.08°C.

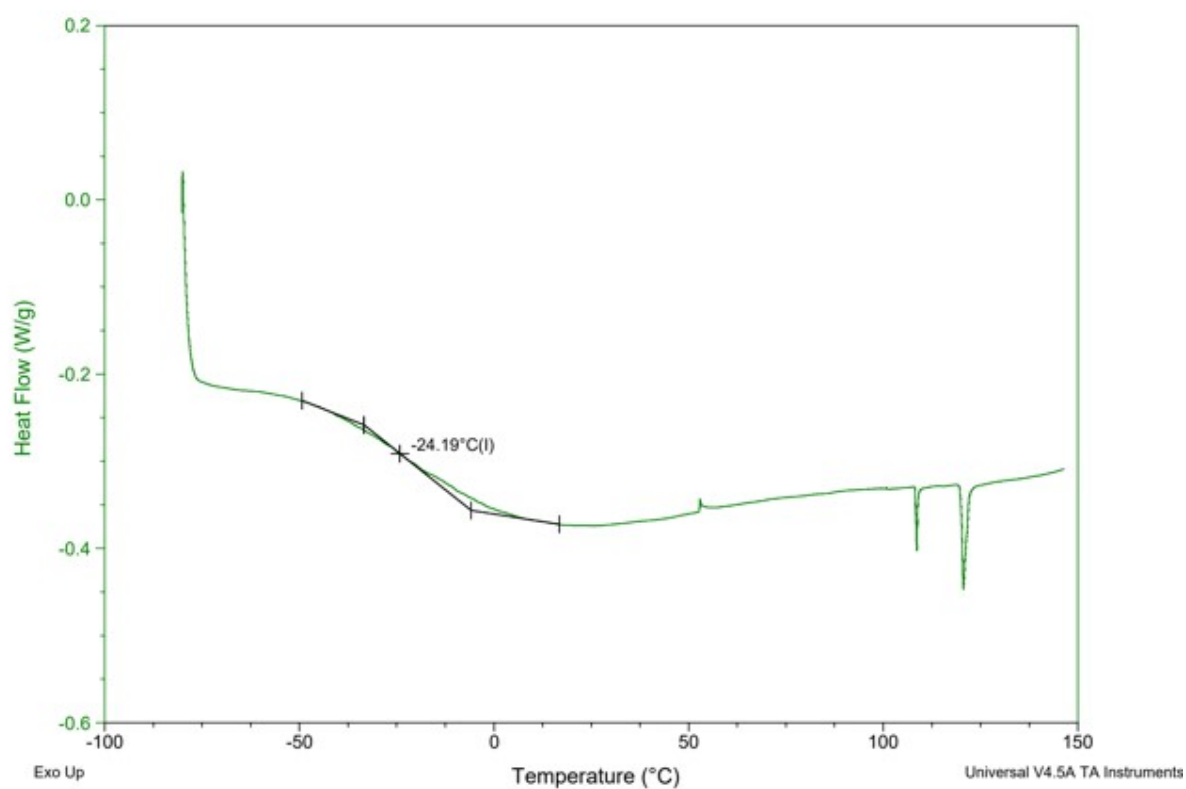


Figure S23: DSC scan of 20-poly(S-r-PVPSi) with T_g measured at -24.19°C and elemental sulfur shown to melt at 120°C.

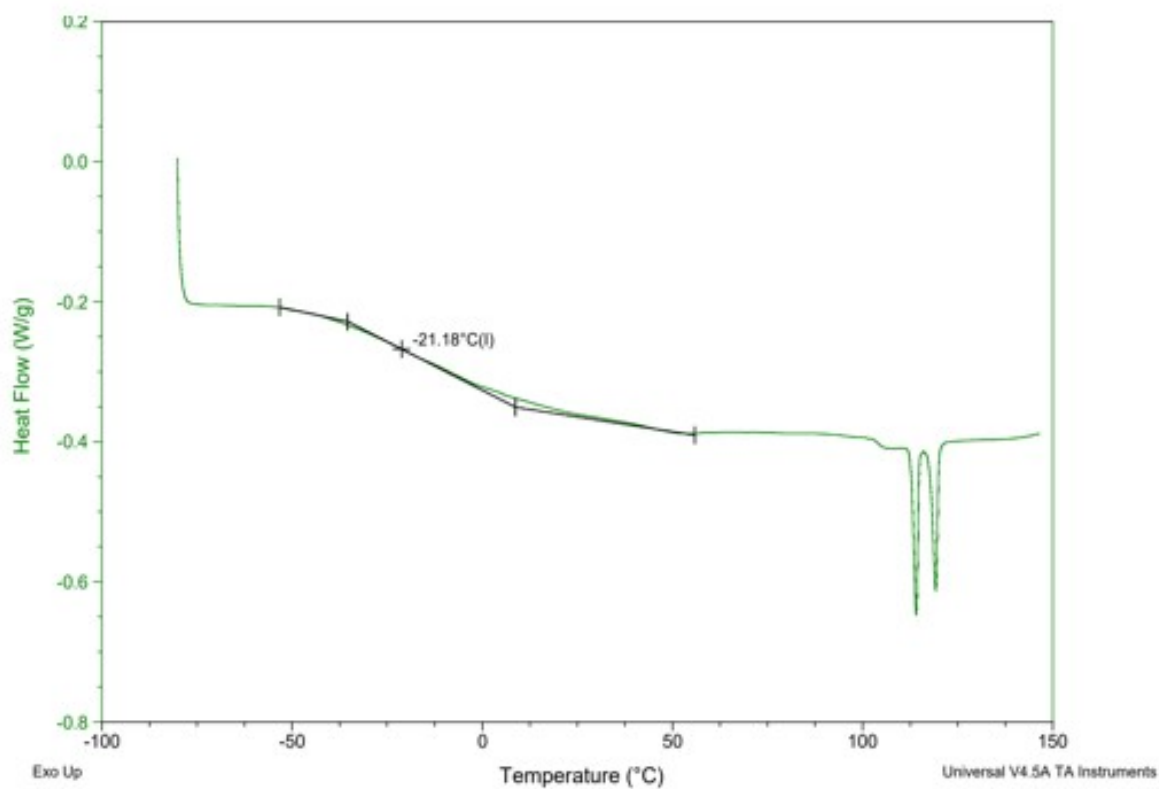


Figure S24: DSC scan of 30-poly(S-*r*-PVPSi) with T_g measured at -21.18°C and elemental sulfur shown to melt at 120°C .

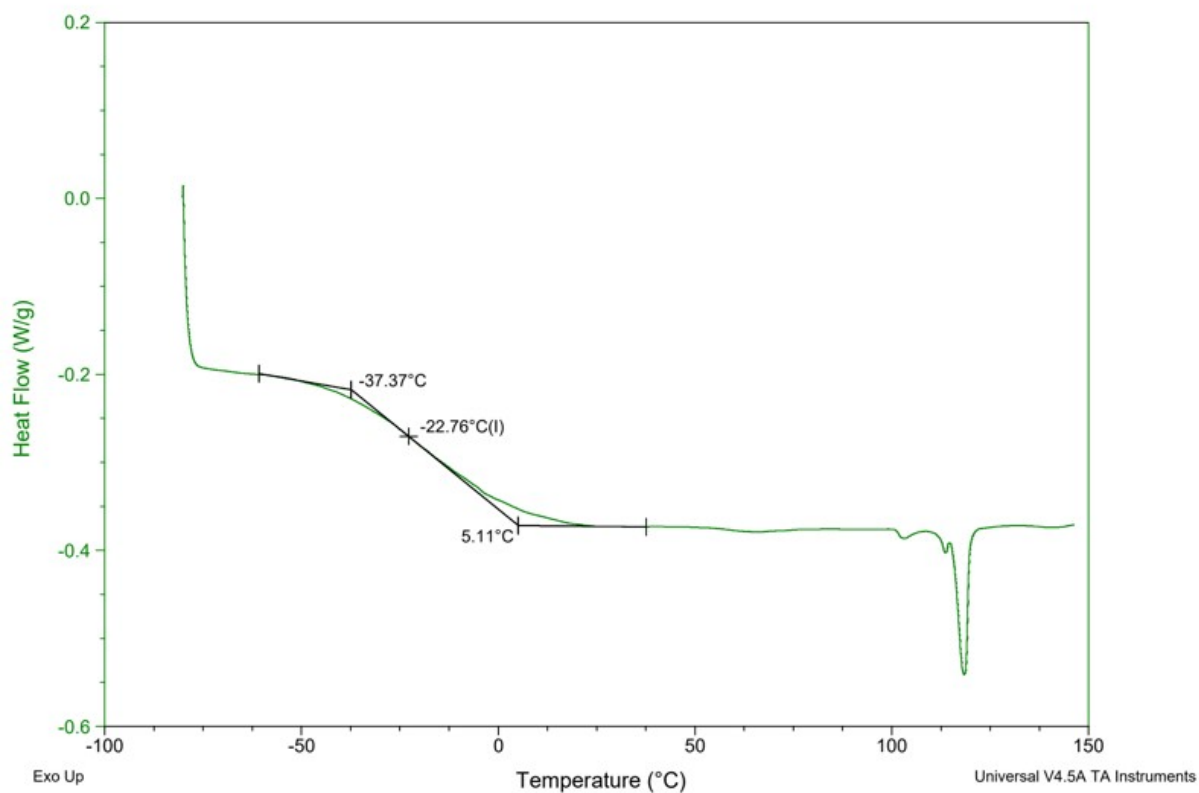


Figure S25: DSC scan of 40-poly(S-*r*-PVPSi) with T_g measured at -22.76°C and elemental sulfur shown to melt at 120°C .

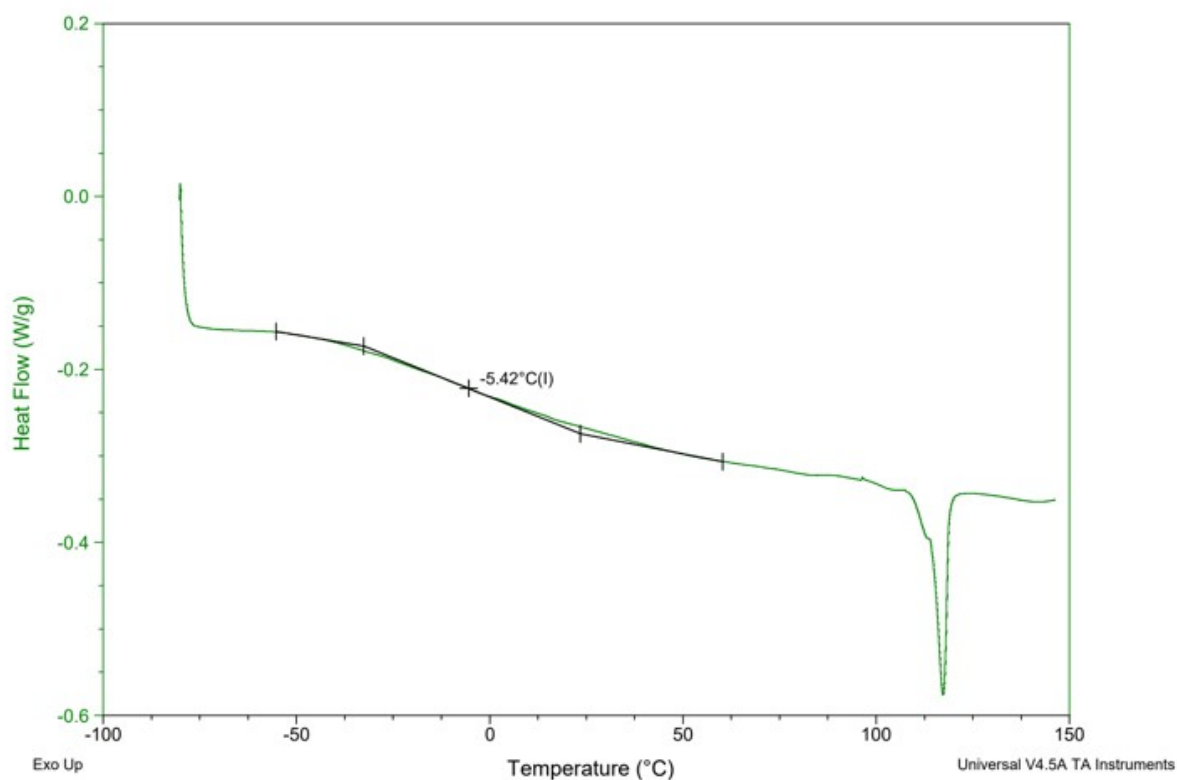


Figure S26: DSC scan of 50-poly(S-*r*-PVPSi) with T_g measured at -5.42°C and elemental sulfur shown to melt at 120°C.

polysulfide mass before (g)	solvent	solvent volume (mL)	time/h	polysulfide mass after(g)	mass difference (g)	swelling (%)	volume absorbed(mL)
0.02	Et2O	1	24	0.0268	0.0068	34	0.009537167
0.0198	Et2O	1	24	0.0369	0.0171	86.36363636	0.02398317
0.0199	Et2O	1	24	0.0362	0.0163	81.90954774	0.02286115
0.0205	DCM	1	24	0.065	0.0445	217.0731707	0.033458647
0.0203	DCM	1	24	0.0753	0.055	270.9359606	0.041353383
0.0203	DCM	1	24	0.0731	0.0528	260.0985222	0.039699248
0.0191	hexane	1	24	0.0248	0.0057	29.84293194	0.00870229
0.0205	hexane	1	24	0.0272	0.0067	32.68292683	0.010229008
0.0206	hexane	1	24	0.0273	0.0067	32.52427184	0.010229008
0.0198	toluene	1	24	0.0258	0.006	30.3030303	0.006920415
0.0209	toluene	1	24	0.0265	0.0056	26.79425837	0.006459054
0.0202	toluene	1	24	0.0269	0.0067	33.16831683	0.007727797
0.0201	benzene	1	24	0.0386	0.0185	92.039801	0.021118721
0.0204	benzene	1	24	0.0381	0.0177	86.76470588	0.020205479
0.0203	benzene	1	24	0.044	0.0237	116.7487685	0.027054795
0.0206	octane	1	24	0.0247	0.0041	19.90291262	0.005832148
0.0206	octane	1	24	0.0242	0.0036	17.47572816	0.00512091
0.0197	octane	1	24	0.0225	0.0028	14.21319797	0.00398293
0.02	water	1	24	0.02	0	0	0
0.0208	water	1	24	0.0208	0	0	0
0.0207	water	1	24	0.0208	1E-04	0.483091787	1E-04
0.0202	ethanol	1	24	0.0202	0	0	0
0.0203	ethanol	1	24	0.0203	0	0	0
0.02	ethanol	1	24	0.02	0	0	0
0.02	methanol	1	24	0.02	0	0	0
0.0204	methanol	1	24	0.0204	0	0	0
0.0197	methanol	1	24	0.0197	0	0	0

Table S1: Data for the initial absorbance testing of 20-poly-(S-*r*-TVTSi)

0.02	Et2O	1	24	0.0298	0.0098	49	0.013744741
0.0203	Et2O	1	24	0.0289	0.0086	42.36453202	0.012061711
0.0201	Et2O	1	24	0.0297	0.0096	47.76119403	0.013464236
0.0199	DCM	1	24	0.038	0.0181	90.95477387	0.013609023
0.0202	DCM	1	24	0.0382	0.018	89.10891089	0.013533835
0.0203	DCM	1	24	0.0376	0.0173	85.22167488	0.013007519
0.02	hexane	1	24	0.0266	0.0066	33	0.010076336
0.02	hexane	1	24	0.0262	0.0062	31	0.009465649
0.0201	hexane	1	24	0.0284	0.0083	41.29353234	0.012671756
0.0202	toluene	1	24	0.0386	0.0184	91.08910891	0.021222607
0.0201	toluene	1	24	0.0415	0.0214	106.4676617	0.024682814
0.0202	toluene	1	24	0.0419	0.0217	107.4257426	0.025028835
0.02	benzene	1	24	0.0292	0.0092	46	0.010502283
0.02	benzene	1	24	0.0329	0.0129	64.5	0.014726027
0.0203	benzene	1	24	0.0315	0.0112	55.17241379	0.012785388
0.02	octane	1	24	0.0316	0.0116	58	0.016500711
0.0199	octane	1	24	0.0315	0.0116	58.29145729	0.016500711
0.02	octane	1	24	0.0305	0.0105	52.5	0.014935989
0.02	water	1	24	0.02	0	0	0
0.0206	water	1	24	0.0206	0	0	0
0.0207	water	1	24	0.0207	0	0	0
0.02	ethanol	1	24	0.02	0	0	0
0.0196	ethanol	1	24	0.0196	0	0	0
0.0201	ethanol	1	24	0.0201	0	0	0
0.02	methanol	1	24	0.02	0	0	0
0.0203	methanol	1	24	0.0203	0	0	0
0.0209	methanol	1	24	0.0211	0.0002	0.956937799	0.000252525

Table S2: Data for the initial absorbance testing of 20-poly-(S-*r*-PVPSi)

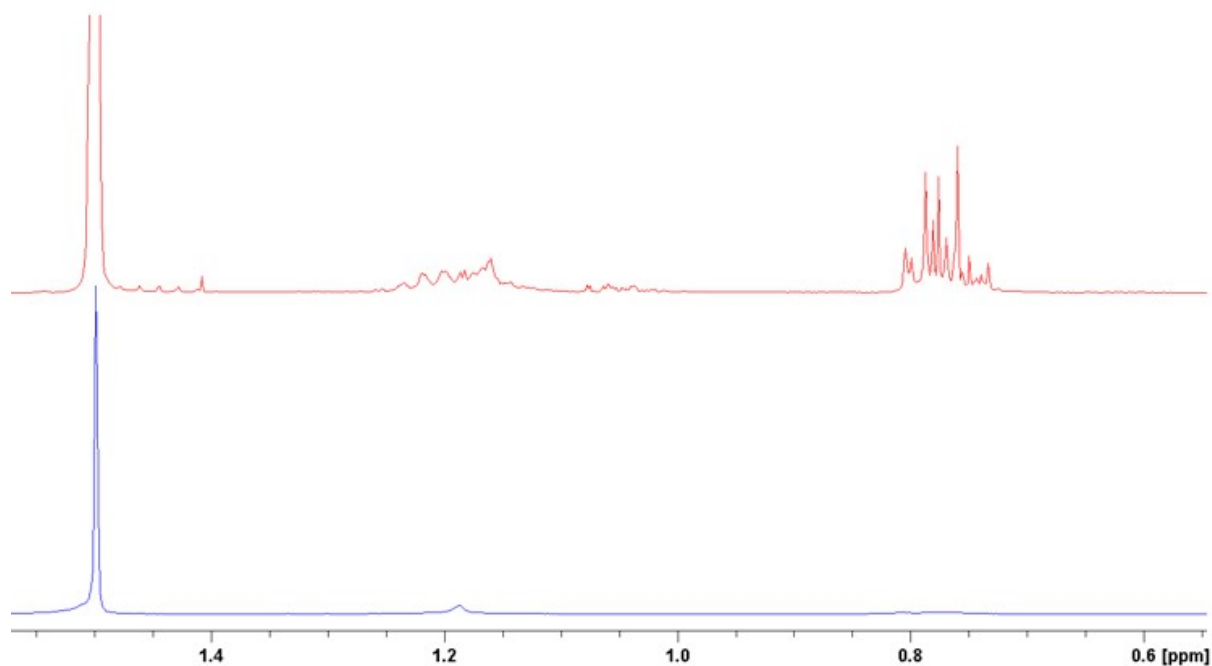


Figure S27: Comparison of ^1H NMR spectra (CDCl_3) of a 5000 ppm hexane solution in water before (top, red) and after (bottom, blue) the addition of 20-poly-(S-*r*-TVTSi) for 24 hours.

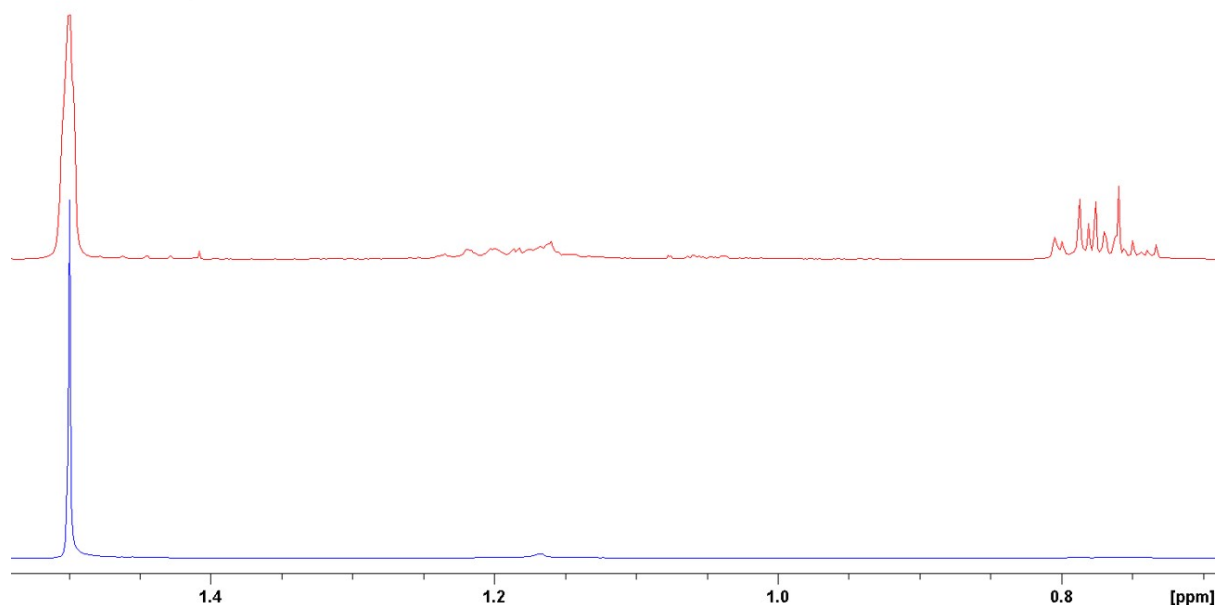


Figure S28: Comparison of ^1H NMR spectra (CDCl_3) of a 5000 ppm hexane solution in water before (top, red) and after (bottom, blue) the addition of 20-poly-(*S-r*-PVPSi) for 24 hours.

Sample	Integration at 1.28ppm	Difference of hexane integration from stock (%)
5000 ppm Stock solution	1.627	0
1st	0.4883	69.98770744
2nd	0.4487	72.42163491
3rd	0.4718	71.00184388
4th	0.4604	71.70251998

Table S3: Analysis and calculation % hexanes removed for recycling testing of 20-poly(*S-r*-PVPSi) in different batches of contaminated water.

Sample	Integration at 1.28ppm	Difference of hexane integration from previous addition (%)	Difference of hexane integration from stock (%)
5000 ppm Stock solution	1.627	0	0
1st	0.4928	69.71112477	69.71112477
2nd	0.4829	2.008928571	70.31960664
3rd	0.4819	0.207082212	70.38106945
4th	0.4779	0.830047728	70.62692071

Table S4: Analysis and calculation of total % hexanes removed for multiple cycles of adding 20-poly(*S-r*-PVPSi) to the same 5000 ppm hexanes in water solution.