

Supplementary information

Fabrication of Multi-responsive Photonic Crystals based on Selenium-containing Copolymers

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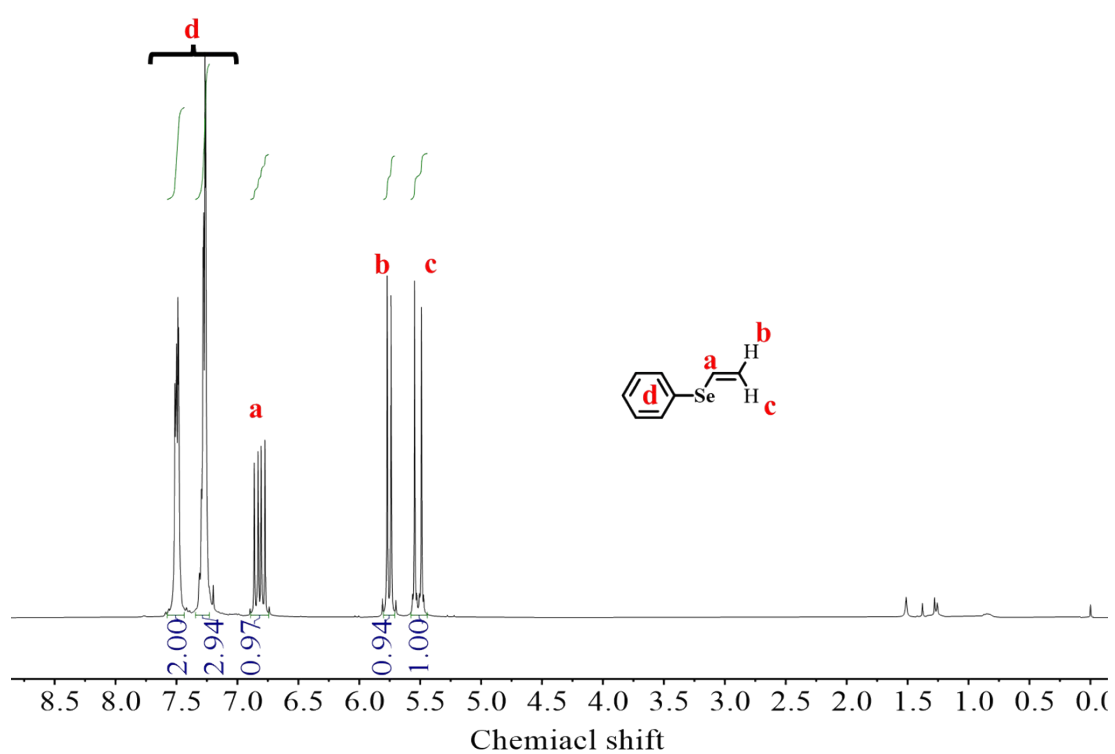


Figure S1. ^1H NMR spectrum of PVSe.

Table S1. Radical copolymerization of PVSe with 4VP in different molar ratio in bulk at 60 °C for 16 h.

Entry	[PVSe] ₀ / [4VP] ₀ / [AIBN] ₀	^{a)} Yield(%)	^{b)} $M_{n, th.}$ [g mol ⁻¹]	^{c)} $M_{n, SEC}$ [g mol ⁻¹]	^{c)} $\bar{D}$	^{d)} F_{PVSe}
1	[PVSe] ₀ / [4VP] ₀ / [AIBN] ₀ =10/90/1	98.9	27500	338000	1.70	0.042
2	[PVSe] ₀ / [4VP] ₀ / [AIBN] ₀ =30/70/1	92.4	11800	128000	1.79	0.167
3	[PVSe] ₀ / [4VP] ₀ / [AIBN] ₀ =50/50/1	50.8	7300	21700	3.19	0.285
4	[PVSe] ₀ / [4VP] ₀ / [AIBN] ₀ =70/30/1	65.0	15800	13000	3.03	0.354
5	[PVSe] ₀ / [4VP] ₀ / [AIBN] ₀ =80/20/1	49.1	8200	5500	2.30	0.498
6	[PVSe] ₀ / [4VP] ₀ / [AIBN] ₀ =90/10/1	40.3	7000	5100	1.84	0.623

^{a)} n-hexane-insoluble part.

^{b)} The theoretical molecular weight ($M_{n, th.} = [(M_W \text{ of PVSe}) \times f_{PVSe} + (M_W \text{ of 4VP}) \times f_{4VP}] \times \text{Yield.} + (M_W \text{ of CTA})$).

^{c)} Measured by size-exclusion chromatography (SEC) using polystyrene standards in DMF (10 mM LiBr).

^{d)} Determined by ¹H NMR in DMSO-*d*₆, ignoring the influence of end groups:

$$F_{PVSe} = \frac{I(6.15 - 7.73) - I(7.99 - 8.59)}{5} \div \left(\frac{I(6.15 - 7.73) - I(7.99 - 8.59)}{5} \right)$$

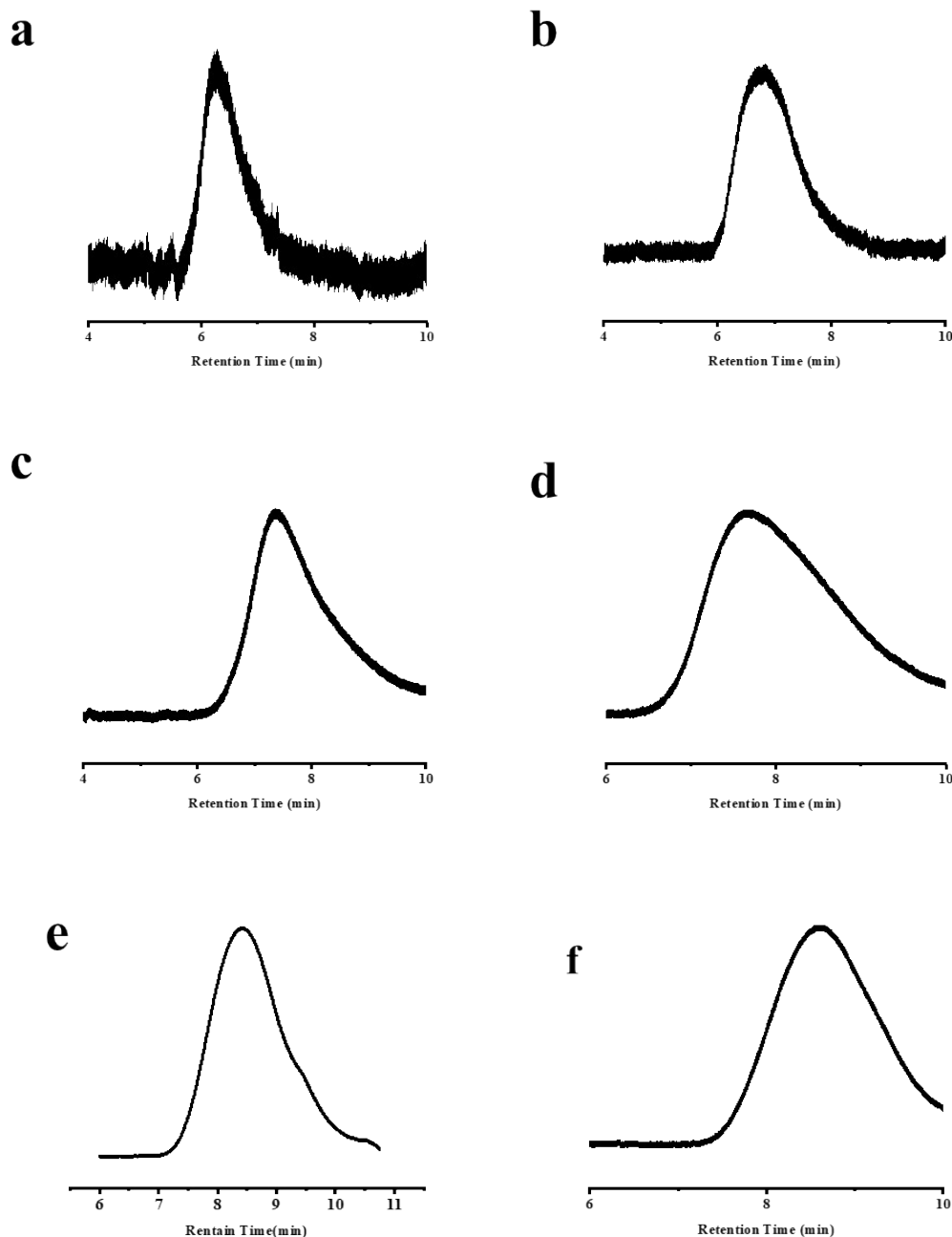


Figure S2. a) SEC traces of Entry 1 in Table S1 using polystyrene (PS) as standard in DMF (0.01M LiBr); b) SEC traces of Entry 2 in Table S1 using polystyrene (PS) as standard in DMF (0.01M LiBr); c) SEC traces of Entry 3 in Table S1 using polystyrene (PS) as standard in DMF (0.01M LiBr); d) SEC traces of Entry 4 in Table S1 using polystyrene (PS) as standard in DMF (0.01M LiBr); e) SEC traces of Entry 5 in Table S1 using polystyrene (PS) as standard in DMF (0.01M LiBr); f) SEC traces of Entry 6 in Table S1 using polystyrene (PS) as standard in DMF (0.01M LiBr).

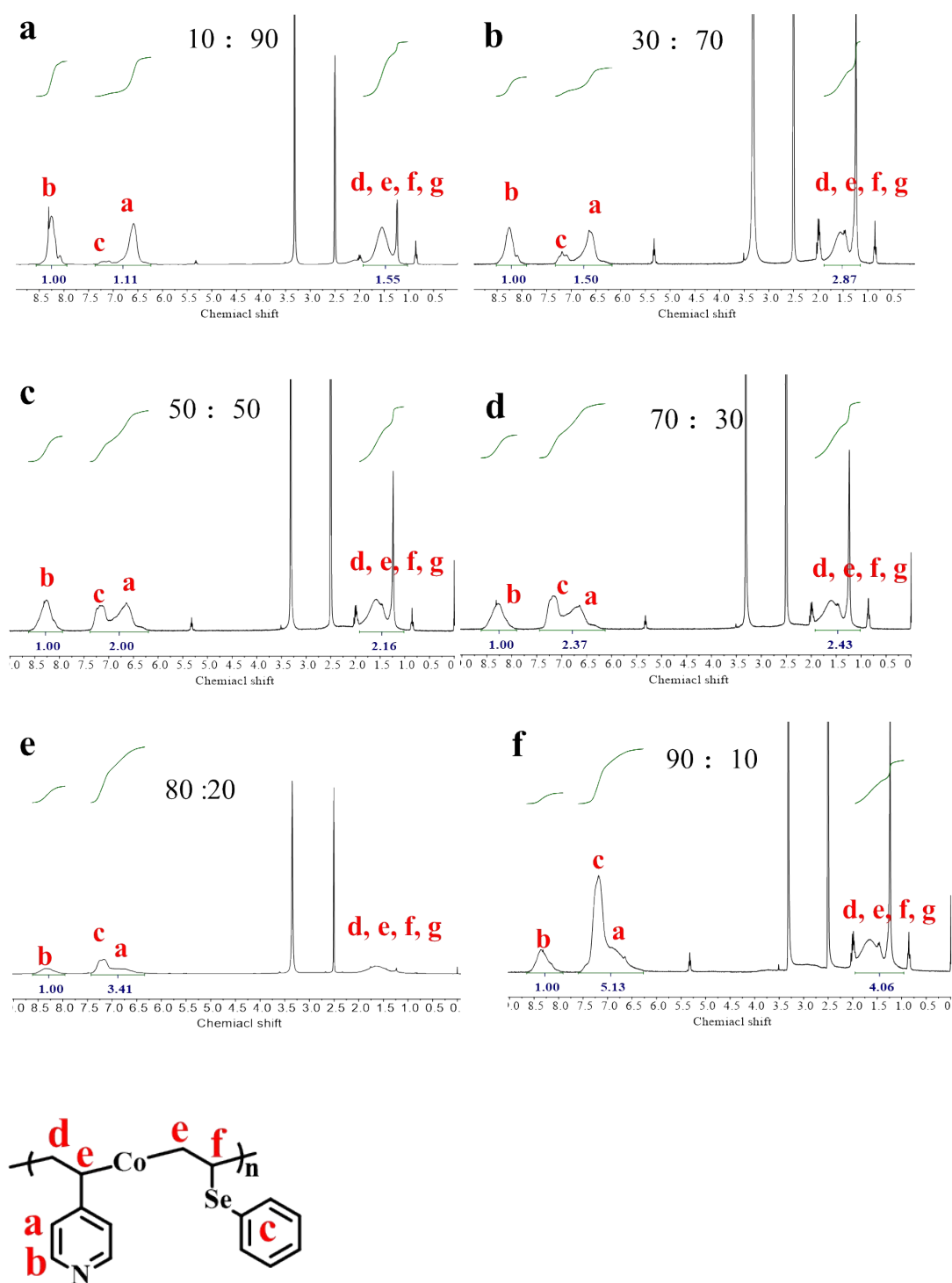


Figure S3. a) ^1H NMR spectrum of Entry 1 in Table S1; b) ^1H NMR spectrum of Entry 2 in Table S1; c) ^1H NMR spectrum of Entry 3 in Table S1; d) ^1H NMR spectrum of Entry 4 in Table S1; e) ^1H NMR spectrum of Entry 5 in Table S1. f) ^1H NMR spectrum of Entry 6 in Table S1

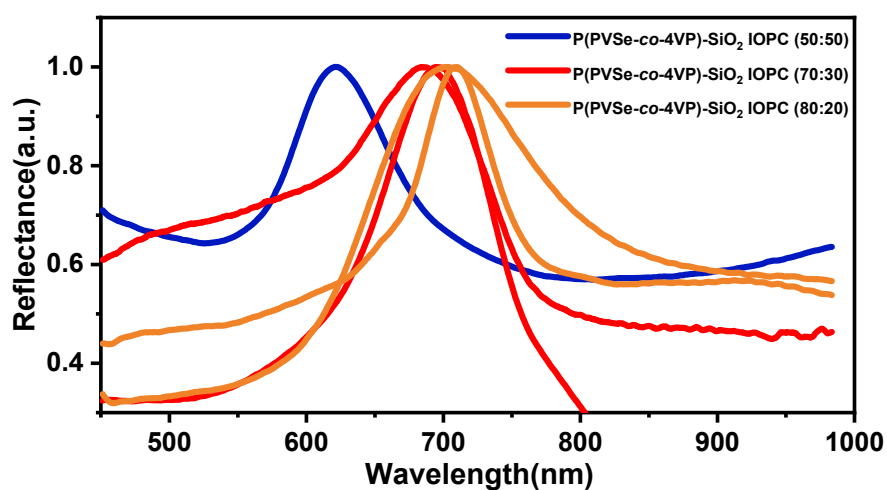


Figure S4. The normalized reflectance spectrum of the P(PVSe-co-4VP)-SiO₂ IOPC (50:50), P(PVSe-co-4VP)-SiO₂ IOPC (70: 30), P(PVSe-co-4VP)-SiO₂ IOPC (80:20) (the specific value following P(PVSe-co-4VP) SiO₂ IOPC is the molar ratio of PVSe and 4VP and P(PVSe-co-4VP)-SiO₂ IOPC (50:50) is Entry 3 in Table S1, P(PVSe-co-4VP)-SiO₂ IOPC (70:30) is Entry 4 in Table S1, P(PVSe-co-4VP)-SiO₂ IOPC (80:20) is Entry 5 in Table S1).

Table S2. RAFT copolymerization of PVSe with 4VP at 60 °C for 30 h using BPCD as CTA at different molar ratio of [PVSe]₀/[4VP]₀/[CTA]₀ in bulk ([CTA]₀/[AIBN]=1/0.5).

Entry	[PVSe] ₀ / [4VP] ₀	[M] ₀ / [CTA] ₀	^{a)} yield (%)	^{b)} <i>M</i> _{n,SEC} [g mol ⁻¹]	^{b)} <i>Đ</i>	^{c)} <i>F</i> _{PVSe}	<i>n</i> ₅₈₉
1	[PVSe] ₀ / [4VP] ₀ = 70/30	[M] ₀ / [CTA] ₀ = 100/1	36.6	2400	1.24	0.389	1.636
2	[PVSe] ₀ / [4VP] ₀ = 80/20	[M] ₀ / [CTA] ₀ = 100/1	31.6	2300	1.19	0.510	1.642
3		[M] ₀ / [CTA] ₀ = 200/1	21.5	3600	1.36	0.523	1.643
4		[M] ₀ / [CTA] ₀ = 400/1	15.4	3900	1.31	0.451	1.648
5	[PVSe] ₀ / [4VP] ₀ = 90/10	[M] ₀ / [CTA] ₀ = 100/1	11.5	1900	1.15	0.525	1.650
6		[M] ₀ / [CTA] ₀ = 200/1	6.0	1900	1.13	0.587	1.656
7		[M] ₀ / [CTA] ₀ = 400/1	3.7	2500	1.21	0.569	1.650

^{a)} n-hexane-insoluble part.

^{b)} Measured by size-exclusion chromatography (SEC) using polystyrene standards in DMF (10 mM LiBr).

^{c)} Determined by ¹H NMR in DMSO-*d*₆, ignoring the influence of end groups:

$$F_{PVSe} = \frac{I(6.15 - 7.73) - I(7.99 - 8.59)}{5} \div \left(\frac{I(6.15 - 7.73) - I(7.99 - 8.59)}{5} \right)$$

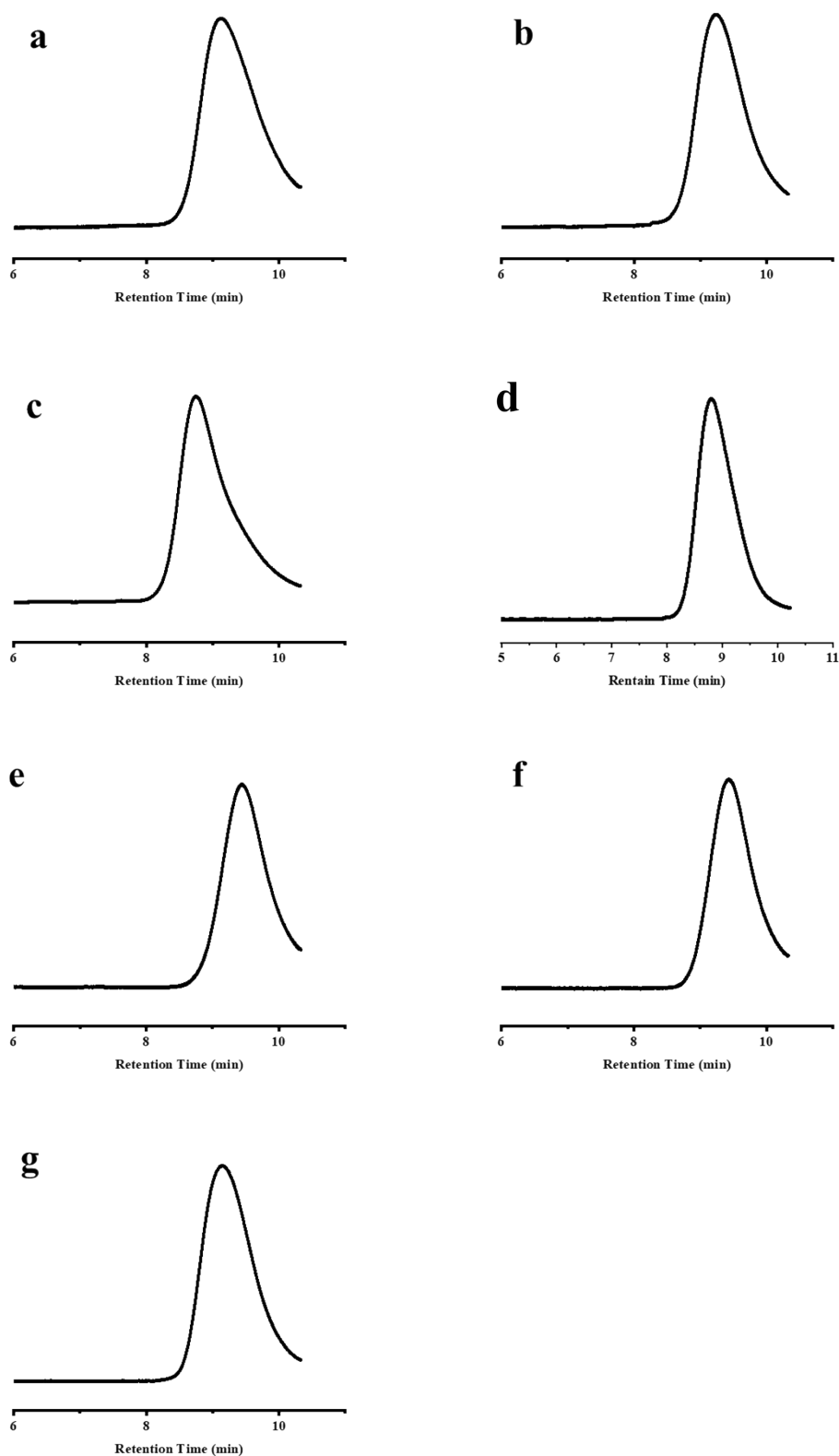


Figure S5. a) SEC trace of P(PVSe-co-4VP) with the molar ratio of $[PVSe]_0/[4VP]_0/[CTA]_0/[AIBN]_0=70/30/1/0.5$; b) SEC trace of P(PVSe-co-4VP) with the molar ratio of $[PVSe]_0/[4VP]_0/[CTA]_0/[AIBN]_0=80/20/1/0.5$; c) SEC trace of P(PVSe-co-4VP) with the molar ratio of $[PVSe]_0/[4VP]_0/[CTA]_0/[AIBN]_0=160/400/1/0.5$; d) SEC trace of

P(PVSe-co-4VP) with the molar ratio of $[PVSe]_0/[4VP]_0/[CTA]_0/[AIBN]_0=320/80/1/0.5$;
e) SEC trace of P(PVSe-co-4VP) with the molar ratio of $[PVSe]_0/[4VP]_0/[CTA]_0/[AIBN]_0=90/10/1/0.5$; f) SEC trace of P(PVSe-co-4VP) with the molar ratio of $[PVSe]_0/[4VP]_0/[CTA]_0/[AIBN]_0=180/20/1/0.5$; g) SEC trace of P(PVSe-co-4VP) with the molar ratio of $[PVSe]_0/[4VP]_0/[CTA]_0/[AIBN]_0=360/40/1/0.5$.

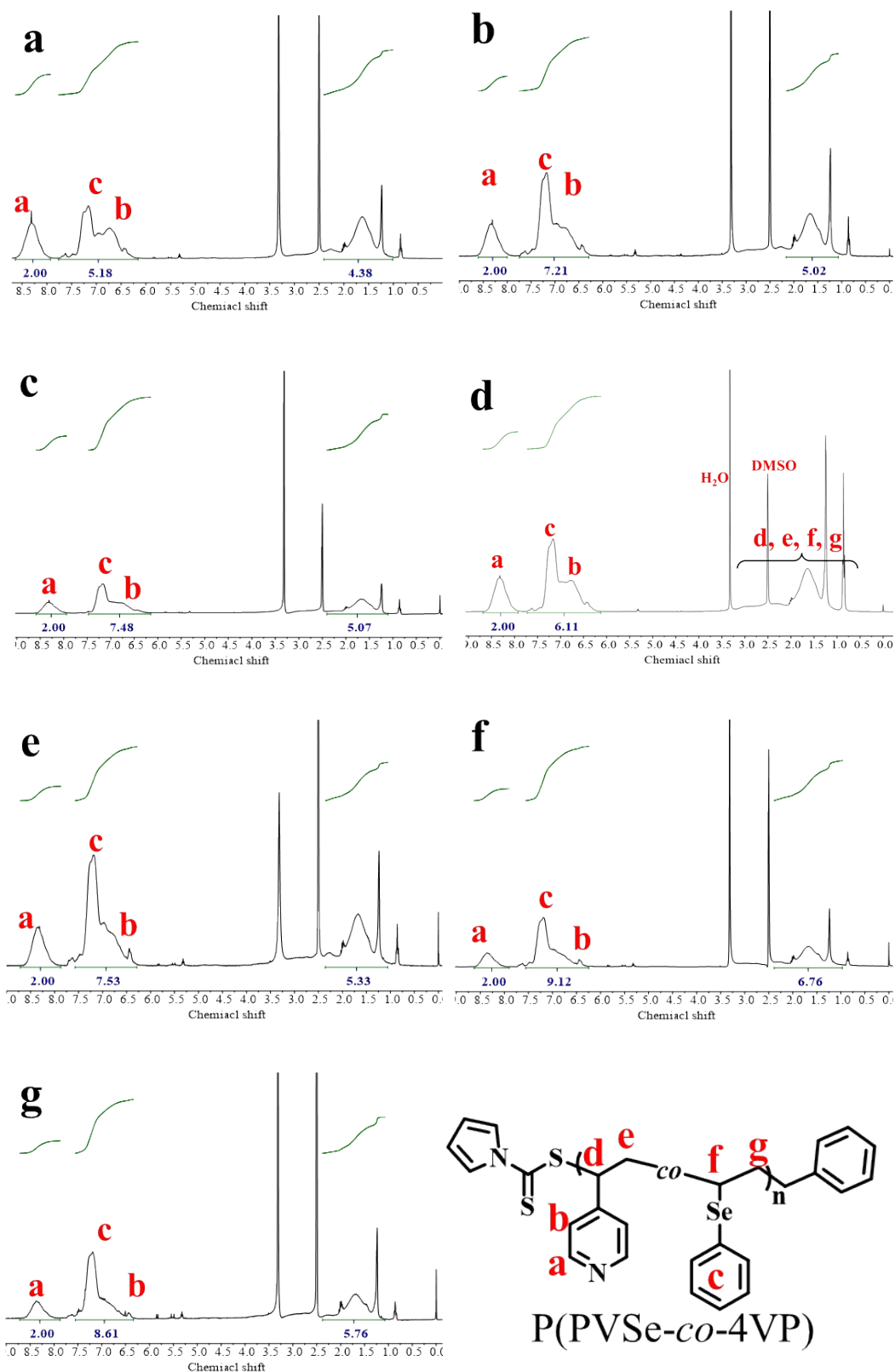


Figure S6. a) ^1H NMR spectrum of P(PVSe-co-4VP) with the molar ratio of $[\text{PVSe}]_0/[\text{4VP}]_0/[\text{CTA}]_0/[\text{AIBN}]_0=70/30/1/0.5$; b) ^1H NMR spectrum of P(PVSe-co-4VP) with the molar ratio of $[\text{PVSe}]_0/[\text{4VP}]_0/[\text{CTA}]_0/[\text{AIBN}]_0=80/20/1/0.5$; c) ^1H NMR spectrum of P(PVSe-co-4VP) with the molar ratio of $[\text{PVSe}]_0/[\text{4VP}]_0/[\text{CTA}]_0/[\text{AIBN}]_0=90/10/1/0.5$; d) ^1H NMR spectrum of P(PVSe-co-4VP) with the molar ratio of $[\text{PVSe}]_0/[\text{4VP}]_0/[\text{CTA}]_0/[\text{AIBN}]_0=95/5/1/0.5$; e) ^1H NMR spectrum of P(PVSe-co-4VP) with the molar ratio of $[\text{PVSe}]_0/[\text{4VP}]_0/[\text{CTA}]_0/[\text{AIBN}]_0=99/1/1/0.5$; f) ^1H NMR spectrum of P(PVSe-co-4VP) with the molar ratio of $[\text{PVSe}]_0/[\text{4VP}]_0/[\text{CTA}]_0/[\text{AIBN}]_0=99.5/0.5/1/0.5$; g) ^1H NMR spectrum of P(PVSe-co-4VP) with the molar ratio of $[\text{PVSe}]_0/[\text{4VP}]_0/[\text{CTA}]_0/[\text{AIBN}]_0=100/0/1/0.5$.

[AIBN]₀=160/400/1/0.5; d) ¹H NMR spectrum of P(PVSe-co-4VP) with the molar ratio of [PVSe]₀/ [4VP]₀/ [CTA]₀/ [AIBN]₀=320/80/1/0.5; e) SEC trace of P(PVSe-co-4VP) with the molar ratio of [PVSe]₀/ [4VP]₀/ [CTA]₀/ [AIBN]₀=90/10/1/0.5; f) ¹H NMR spectrum of P(PVSe-co-4VP) with the molar ratio of [PVSe]₀/ [4VP]₀/ [CTA]₀/ [AIBN]₀=180/20/1/0.5; g) ¹H NMR spectrum of P(PVSe-co-4VP) with the molar ratio of [PVSe]₀/ [4VP]₀/ [CTA]₀/ [AIBN]₀=360/40/1/0.5.

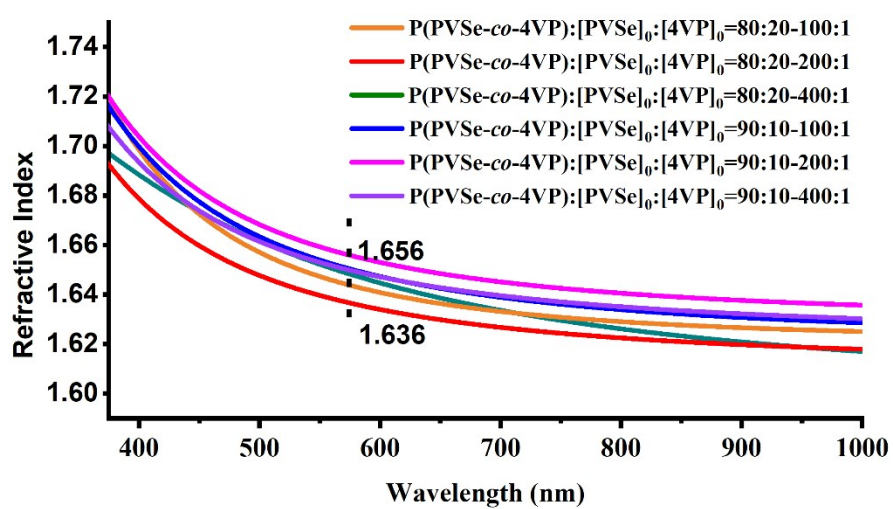


Figure S7. Refractive index curves of P(PVSe-co-4VP) using different molar ratio.

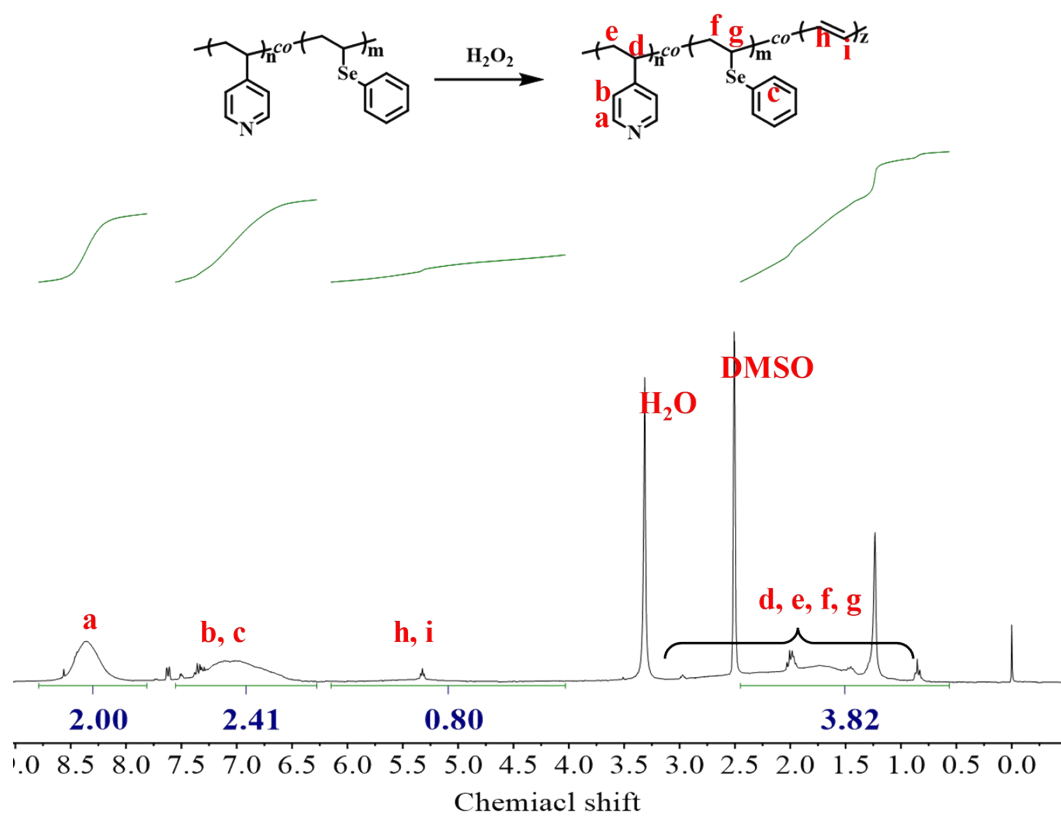


Figure S8. ^1H NMR spectrum of P(PVSe-co-4VP) after oxidated by H_2O_2 .

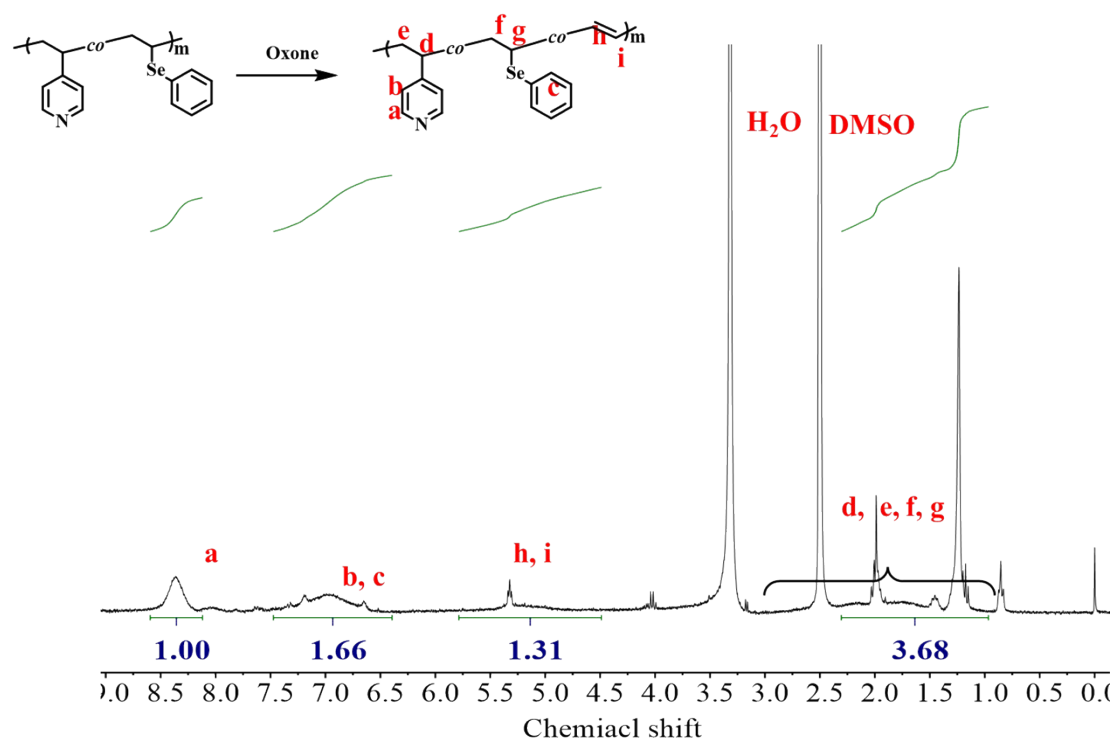


Figure S9. ^1H NMR spectrum of P(PVSe-co-4VP) after oxidated by OXONE.

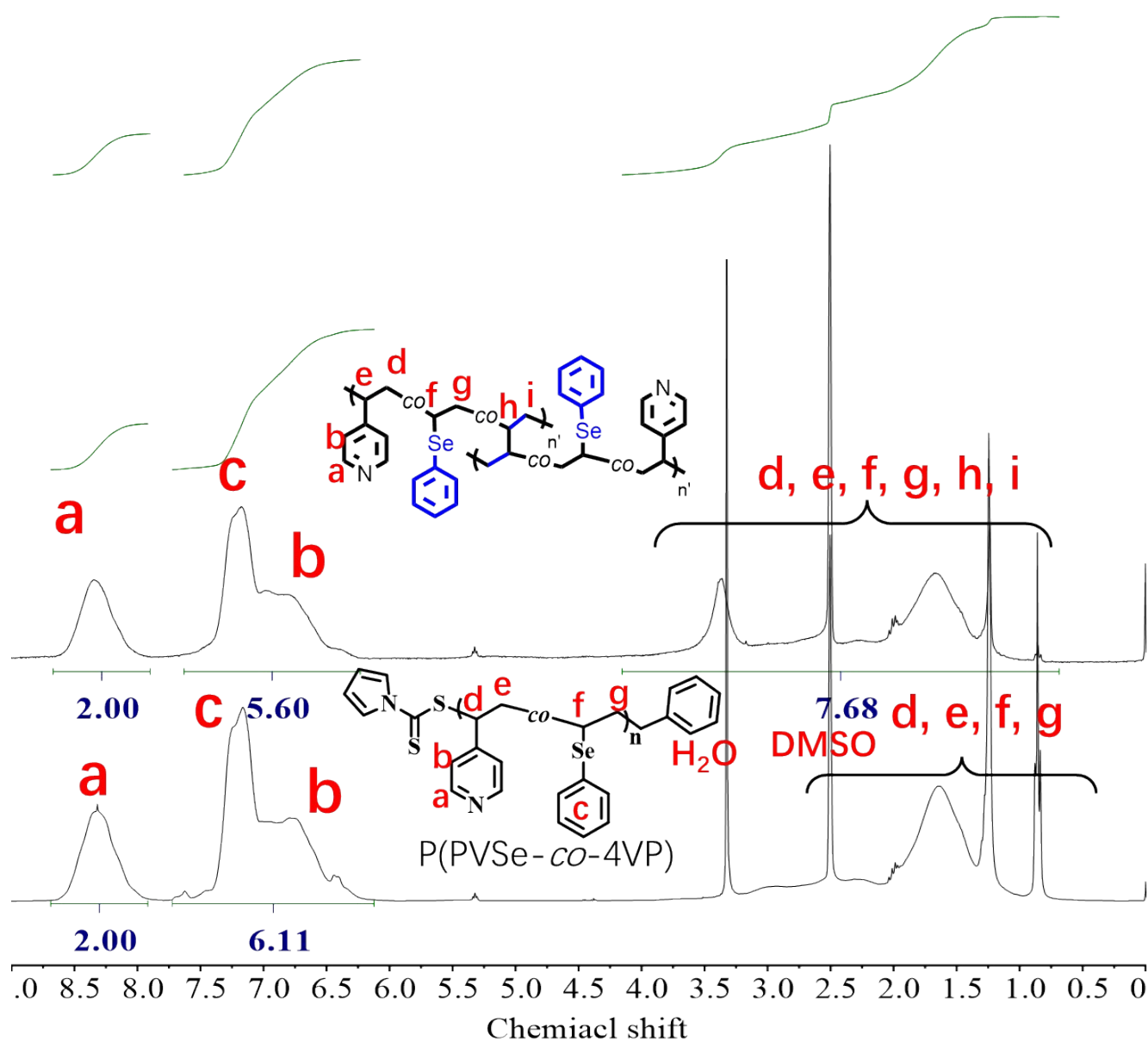


Figure S10. ^1H NMR spectrum of P(PVSe-co-4VP) before and after irradiated by ultraviolet (365 nm) for one night.

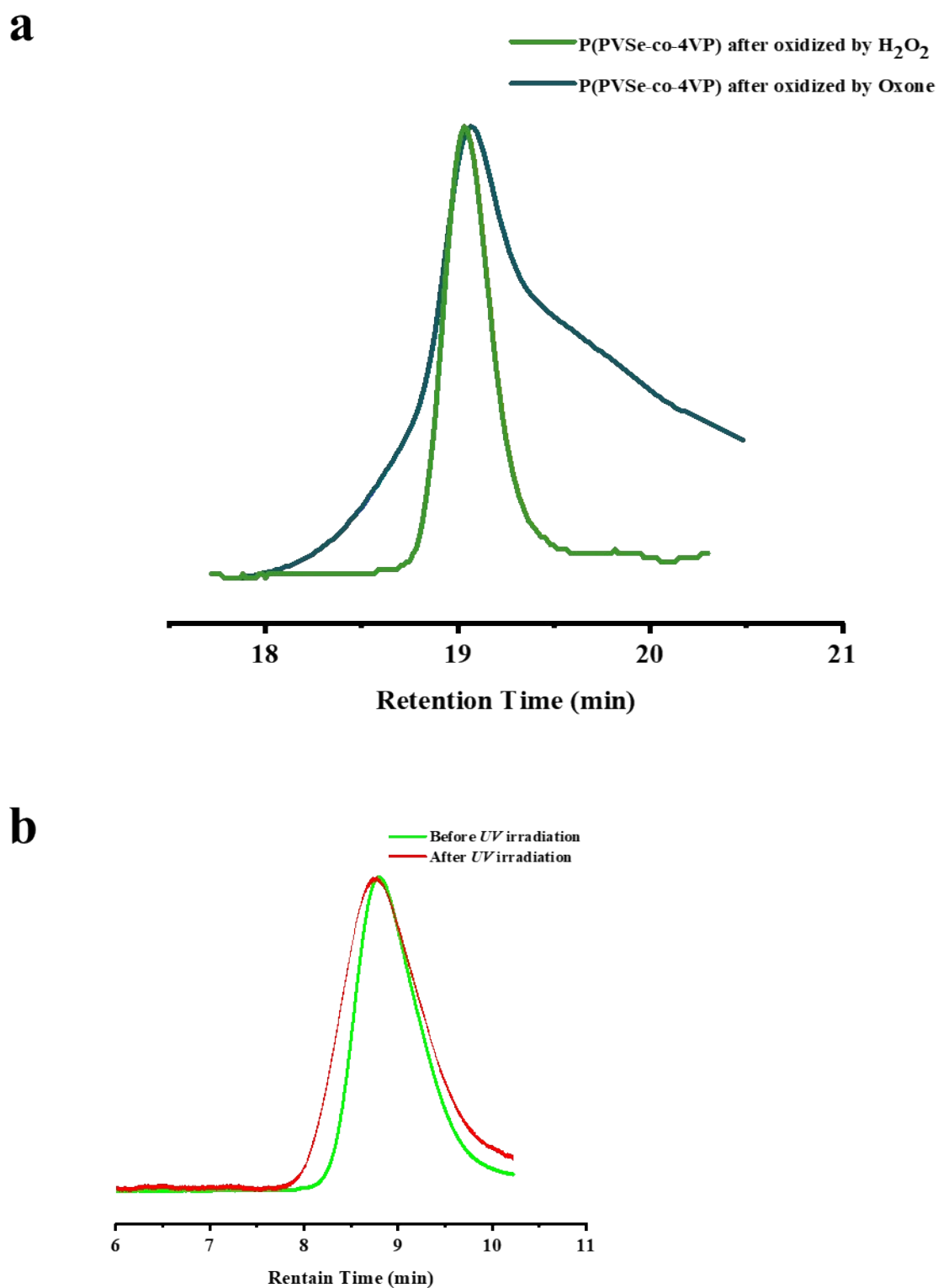


Figure S11. a) SEC traces (water with 0.2 M NaCl+0.03 M NaN₃) of P(PVSe-co-4VP) after oxidated by H₂O₂.and Oxone; b) SEC traces of P(PVSe-co-4VP) before and after irradiated by ultraviolet (365 nm) for one night.

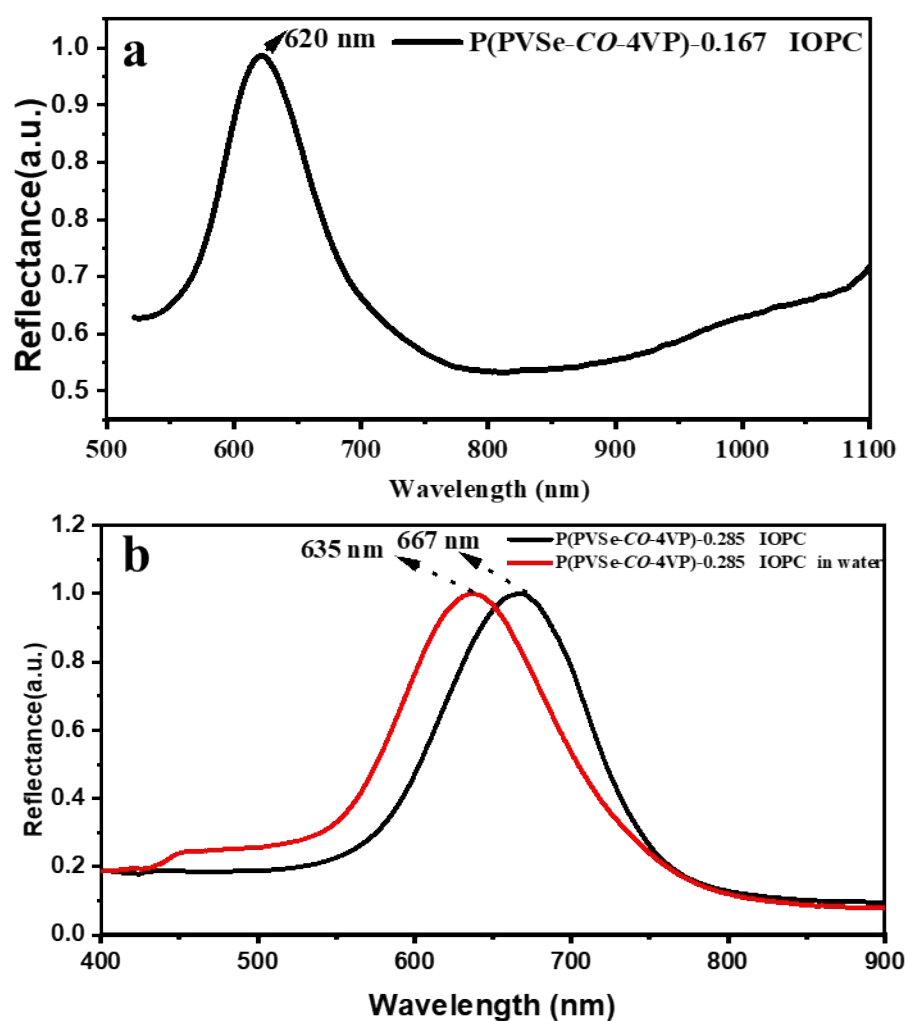


Figure S12. a). The normalized reflectance spectrum of the P(PVSe-co-4VP)-0.167 IOPC (the number following P(PVSe-co-4VP) is F_{PVSe}); b). The normalized reflectance spectrum of the P(PVSe-co-4VP)-0.285 IOPC and when it is in water.

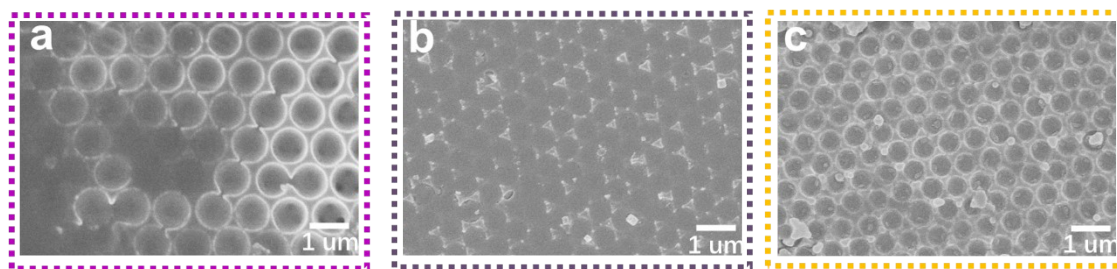


Figure S13. a) SEM image of P(PVSe-co-4VP)-SiO₂ PC after oxidated by 1 wt% Oxone; b) SEM image of P(PVSe-co-4VP)-SiO₂ PC after oxidated by 1 wt% H₂O₂; c) SEM image of P(PVSe-co-4VP)-SiO₂ PC after irradiated by Uv.