Supporting Information

Chemical and physical properties of low-molecularweight poly(2,6-dimethyl-1,4-phenylene oxide) (LMW-PPO) synthesized by peroxydisulfate and metal/non-metal catalysts

Yi-Fang Lu^a, Song-Hai Wu^a, Cong Wang^b, Yong Liu^c, Fu-Gen Huang^d, Bao-Dong

Song^{a,*} and Xu Han^{a,*}

^a Tianjin Key Laboratory of Chemical Process Safety and Equipment Technology, School of Chemical Engineering and Technology, Tianjin University, Tianjin 300350, P.R. China

^b Heibei Key Laboratory of Hazardous Chemicals Safety and Control Technology,

School of Chemical and Environmental Engineering, North China Institute of Science and Technology, Langfang 065201, Hebei, P.R. China

^c School of Chemistry and Chemical Engineering, Tianjin University of Technology, Tianjin 300384, P.R. China

^d Industrial Resources BU, Sinochem International Corporation, Beijing 100045, P.R. China

*Corresponding Author Telephone: <u>+86-13920225599</u> E-mail address: <u>bdsong@tju.edu.cn</u> <u>*Corresponding Author</u> <u>Telephone: +86-15222072695</u> <u>E-mail address: xuhan@tju.edu.cn</u>

Entry	Solvent	PPO yield/%	DPQ yield/%	
1	H ₂ O	42.3	15.2	
2	H ₂ O/CH ₃ CN (v/v=1/1)	61.6	5.1	

Table S1. The influence of solvents on the yields of PPO and DPQ.^a

^a Polymerization was carried out by PDS and CuCl₂-MSPDC catalyst in the air.

Table S2. The dielectric properties of PPO.

	Commercial PPO	Cu(I)-O ₂ ^a	Cu(II)-PDS ^b	MSPDC-PDS °	MA-PDS ^d
D_{κ} (at 1 GHz)	2.45	2.74	2.61	2.42	2.32
$D_{\rm f}$ (at 1 GHz)	0.00174	0.00528	0.00307	0.00171	0.00132

^a PPO was synthesized in the O₂ atmosphere in toluene by CuCl-amine catalyst;

 $^{\rm b}$ PPO was synthesized in the mixture of CH_3CN/H_2O (v/v=9/1) by PDS and CuCl_2-MSPDC catalyst;

^c PPO was synthesized in the mixture of CH₃CN/H₂O (v/v=9/1) by PDS and MSPDC catalyst;

 d PPO was synthesized in the mixture of CH_3CN/H_2O (v/v=9/1) by PDS and MA catalyst.



Figure S1. GPC traces of the obtained PPO. I and II were synthesized in the O₂ atmosphere in toluene by CuCl-amine catalyst with the respective molar ratios of PDS:DMP of 0 and 1:1; III was synthesized in the mixture of CH₃CN/H₂O (v/v=9/1) by PDS and CuCl₂-MSPDC; and IV was synthesized in the mixture of CH₃CN/H₂O (v/v=9/1) by PDS and MSPDC catalyst.



Figure S2. FTIR spectra of the obtained PPO synthesized in the mixture of CH_3CN/H_2O (v/v=9/1) by PDS and CuCl₂-MSPDC catalyst with different molar ratios of PDS:DMP (2:1 and 1:1).



Figure S3. TGA curves of the obtained PPO. I-V: PPO synthesized in the O₂ atmosphere in toluene by CuCl-amine catalyst with the respective molar ratio of PDS:DMP (0, 1:8, 1:4, 1:2 and 1:1), [DMP]=0.2 M, [CuCl]=50 mM, [amine]=0.2 M, temperature: 25°C, time: 2 h; VI: PPO synthesized in the mixture of CH₃CN/H₂O (v/v=9/1) with PDS and CuCl₂-MSPDC catalyst, [DMP]=50 mM, [PDS]=50 mM, [CuCl₂]=5 mM, [MSPDC]=0.05 M, temperature: 50°C, time: 2 h; VII and VIII: PPO synthesized in the mixture of CH₃CN/H₂O (v/v=9/1) by PDS and metal-free catalyst, [DMP]=50 mM, [PDS]=50 mM, temperature: 25°C, time: 2 h, VII: [MSPDC]=5 mM, VIII: MA/CH₃CN/H₂O (v/v=2/7/1).