

Supporting Information for

Engineering thiospinel-based hollow heterostructured nanoarrays for boosting electrocatalytic oxygen evolution reaction

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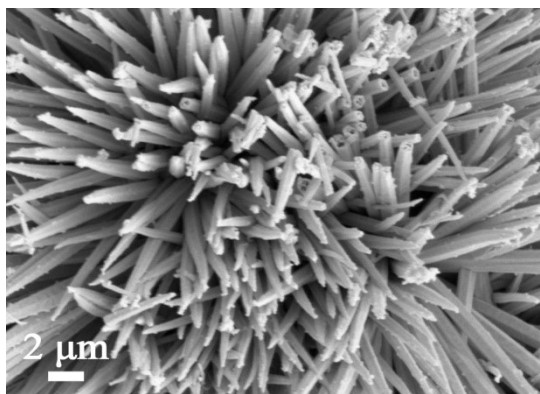


Fig.S1 SEM image of Ni₃S₂/NiCo₂S₄@NF.

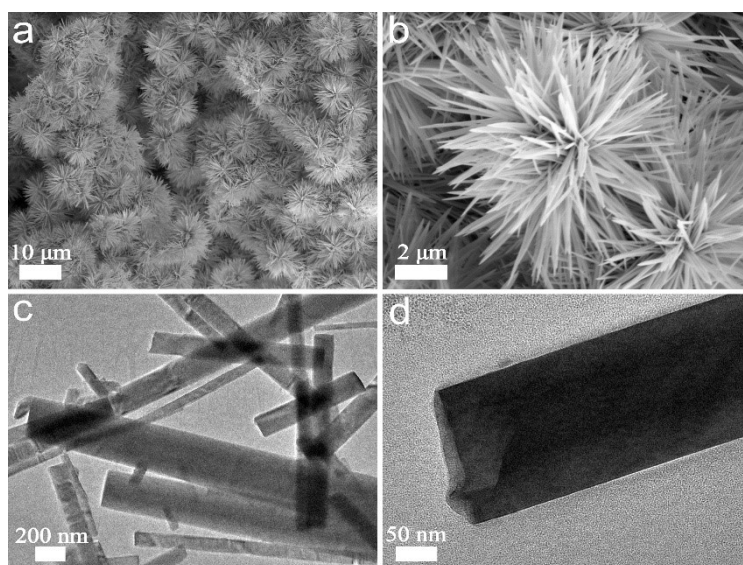


Fig.S2 (a, b) SEM images and (c, d) TEM images of the NiCo-LDH@NF.

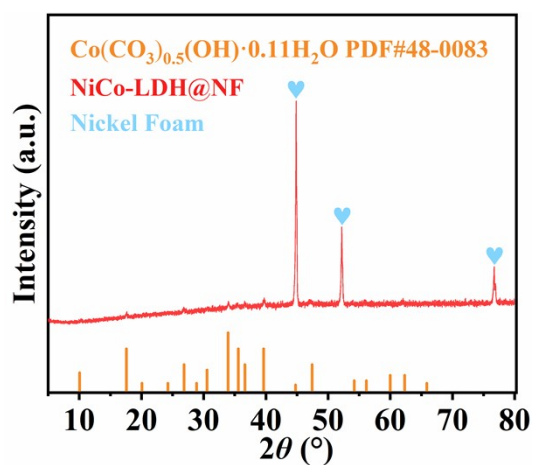


Fig.S3 XRD pattern of NiCo-LDH@NF.

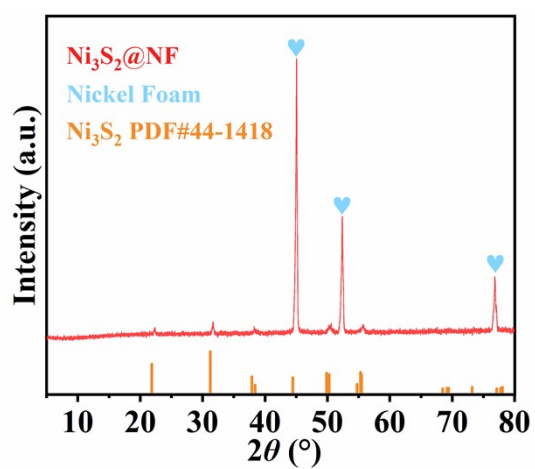


Fig.S4 XRD pattern of Ni₃S₂@NF.

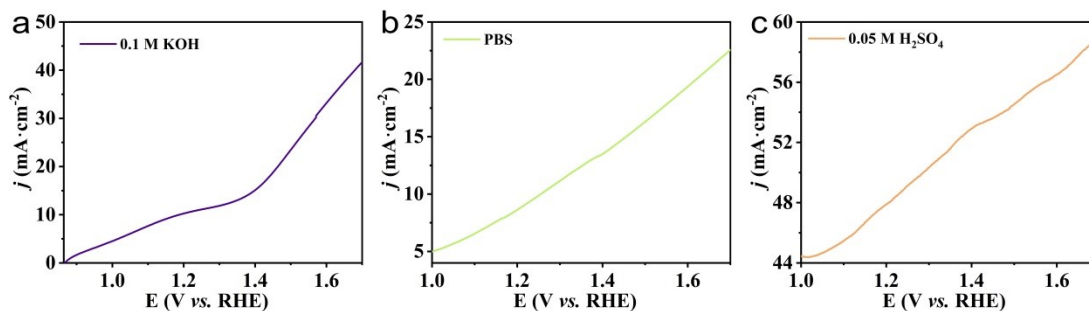


Fig.S5 OER polarization curves of $\text{Ni}_3\text{S}_2/\text{NiCo}_2\text{S}_4@\text{NF}$ in (a) 0.1 M KOH, (b) PBS and (c) 0.05 M H_2SO_4 electrolyte.

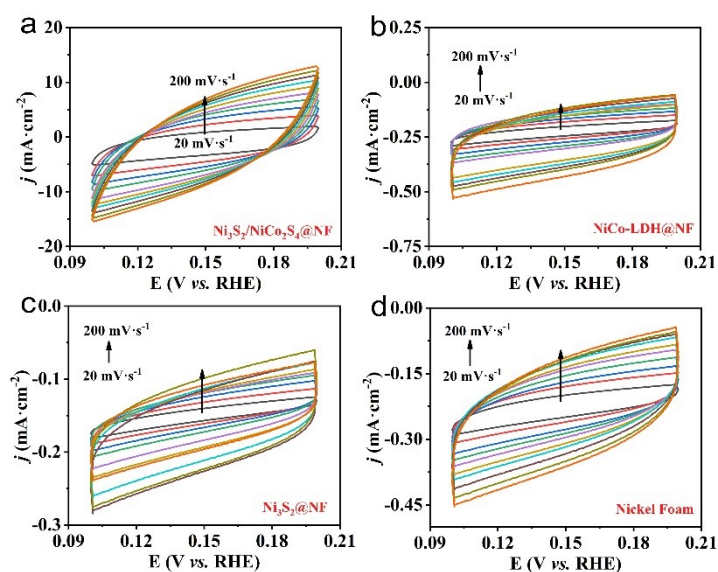


Fig.S6 Cyclic voltammograms in a capacitive current region at a scan rate ranging from 20 to 200 $\text{mV}\cdot\text{s}^{-1}$. (a) $\text{Ni}_3\text{S}_2/\text{NiCo}_2\text{S}_4@\text{NF}$, (b) $\text{NiCo-LDH}@\text{NF}$, (c) $\text{Ni}_3\text{S}_2@\text{NF}$ and (d) NF.

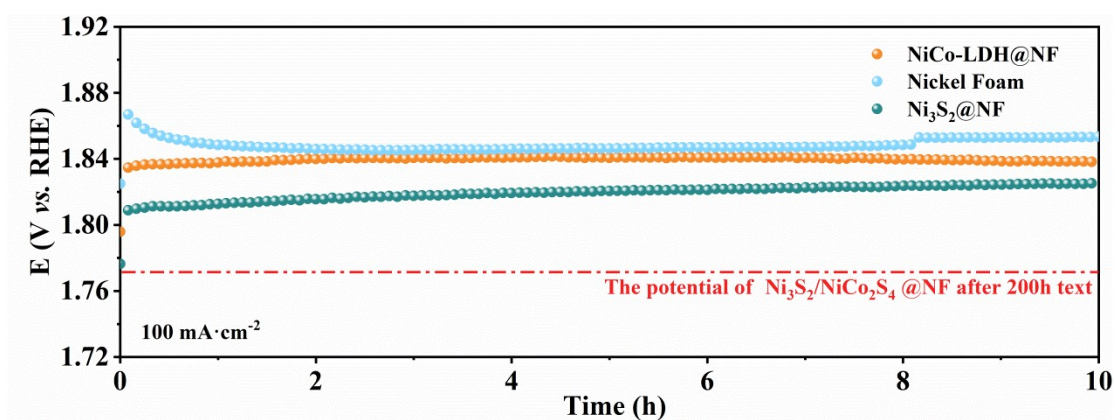


Fig.S7 Chronopotentiometry test of $\text{Ni}_3\text{S}_2@\text{NF}$, $\text{NiCo-LDH}@\text{NF}$, and NF at a constant current density of $100 \text{ mA}\cdot\text{cm}^{-2}$.

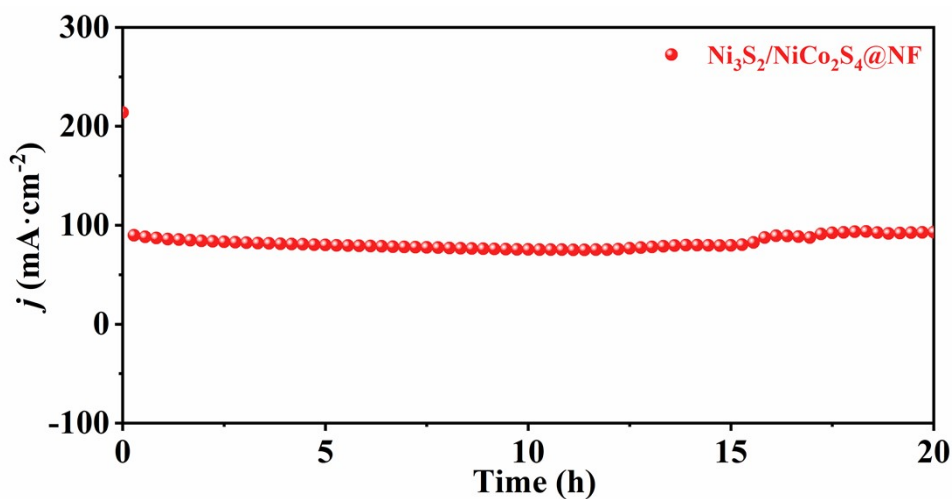


Fig.S8 Chronoamperometry test of $\text{Ni}_3\text{S}_2/\text{NiCo}_2\text{S}_4@\text{NF}$.

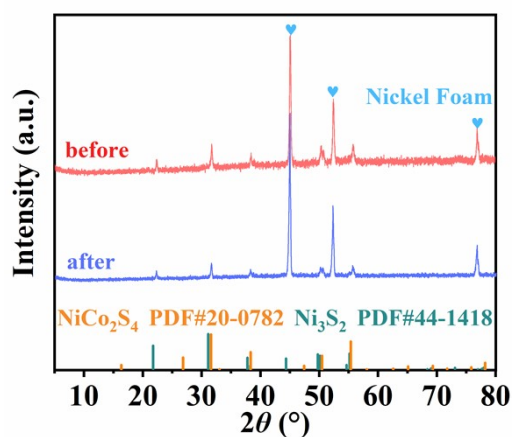


Fig.S9 XRD pattern of $\text{Ni}_3\text{S}_2/\text{NiCo}_2\text{S}_4@\text{NF}$ before and after a continuous 200 h stability test.

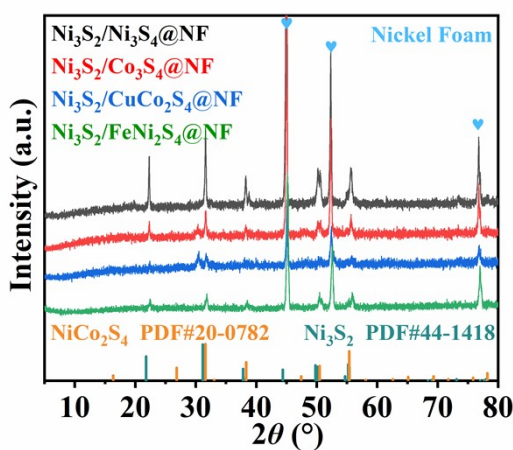


Fig.S10 XRD pattern of $\text{Ni}_3\text{S}_2/\text{Ni}_3\text{S}_4@\text{NF}$, $\text{Ni}_3\text{S}_2/\text{Co}_3\text{S}_4@\text{NF}$, $\text{Ni}_3\text{S}_2/\text{CuCo}_2\text{S}_4@\text{NF}$ and $\text{Ni}_3\text{S}_2/\text{FeNi}_2\text{S}_4@\text{NF}$.

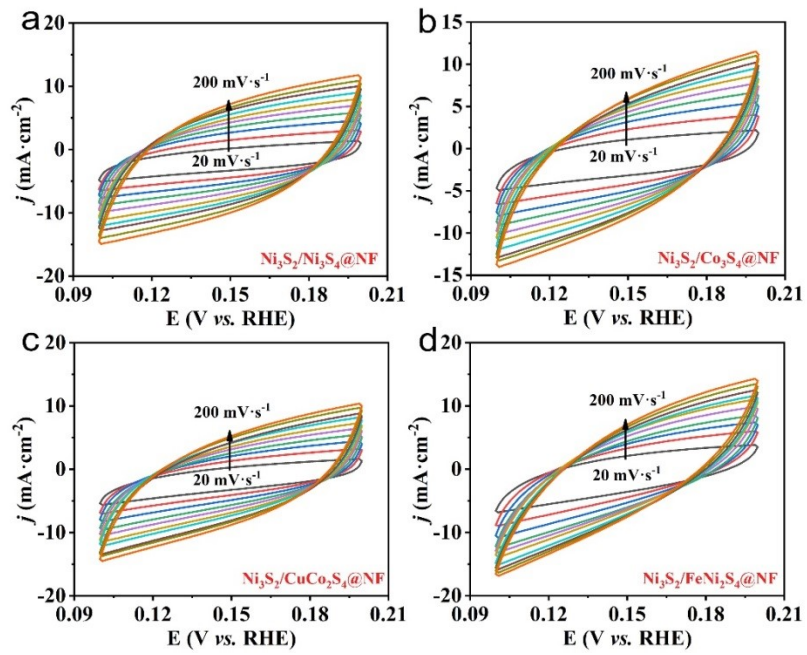


Fig.S11 Cyclic voltammograms in a capacitive current region at a scan rate from 20 to 200 mV·s⁻¹. (a) Ni₃S₂/Ni₃S₄@NF, (b) Ni₃S₂/Co₃S₄@NF, (c) Ni₃S₂/CuCo₂S₄@NF and (d) Ni₃S₂/FeNi₂S₄@NF.

Table S1 OER activity comparison of different catalysts in alkaline condition.

Catalysts	Electrolyte	Overpotential	Reference
Ni₃S₂/NiCo₂S₄@NF	1.0 M KOH	177 mV at 100 mA·cm⁻²	This Work
NiCo ₂ S ₄ @NiFe-LDH@NF	1.0 M KOH	201 mV at 60 mA·cm ⁻²	1
P-NiCo ₂ S ₄ @NF	1.0 M KOH	300 mV at 50 mA·cm ⁻²	2
Mn-NiCo ₂ S ₄ @NF	1.0 M KOH	220 mV at 10 mA·cm ⁻²	3
NiCo ₂ S ₄ /NiFeP@NF	1.0 M KOH	293 mV at 100 mA·cm ⁻²	4
NiCo ₂ S ₄ @NF	1.0 M KOH	279 mV at 50 mA·cm ⁻²	5
NiCo ₂ S ₄ @N-rGO@NF	1.0 M KOH	230 mV at 10 mA·cm ⁻²	6
MoS ₂ /NiCo ₂ S ₄ @NF	1.0 M KOH	220 mV at 10 mA·cm ⁻²	7
Ru-NiCo ₂ S _{4-x} @NF	1.0 M KOH	330 mV at 100 mA·cm ⁻²	8
CuCo ₂ S ₄ /NiCo ₂ S ₄ @NF	1.0 M KOH	271 mV at 10mA·cm ⁻²	9
NiCo ₂ S ₄ @NF	1.0 M KOH	319 mV at 100 mA·cm ⁻²	10
NiCo ₂ S ₄ @NF	1.0 M KOH	260 mV at 10 mA·cm ⁻²	11

References

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