

Supporting information

Precursor-converted Formation of Bimetal-Organic Framework

Nanosheets for Efficient Oxygen Evolution Reaction

*Lei Xia^{a,b}, Crystal Bowers^c, Pei Dong^c, Mingxin Ye^{*a}, Jianfeng Shen^{*a}*

^a Institute of special materials and technology, Fudan University, Shanghai 200433, China

^b Department of Chemistry, Fudan University, Shanghai 200433, China.

^c Department of Mechanical Engineering, George Mason University, VA 22030, USA

Corresponding author: mxye@fudan.edu.cn; jfshen@fudan.edu.cn

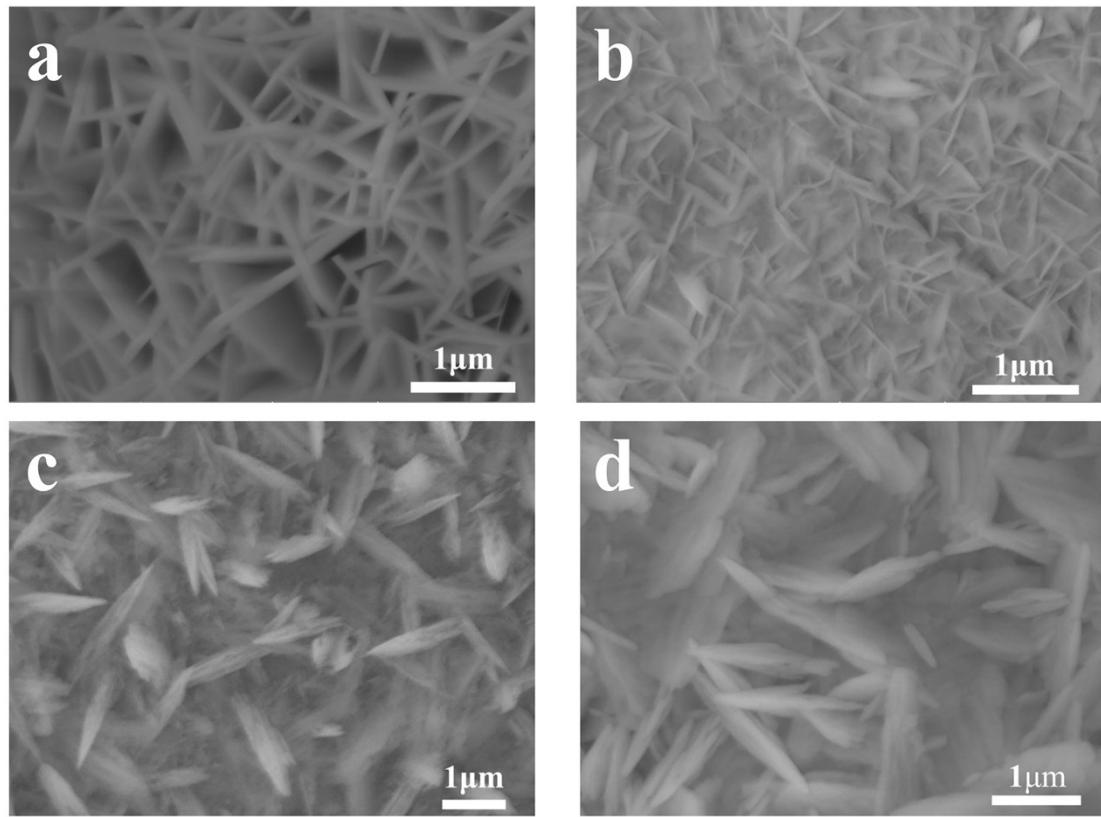


Figure S1. SEM images of NiFe-MOF@NF under different reaction times. a) 0 h, b) 1 h, c) 3 h and d) 6 h.

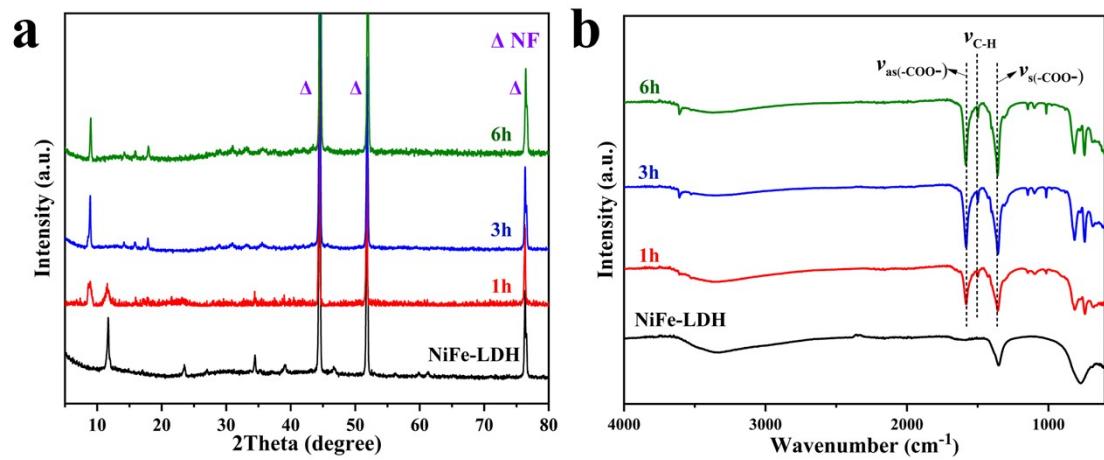


Figure S2. a) XRD patterns and b) FT-IR spectra of NiFe-MOF@NF under different reaction times of 0 h, 1 h, 3 h, and 6 h.

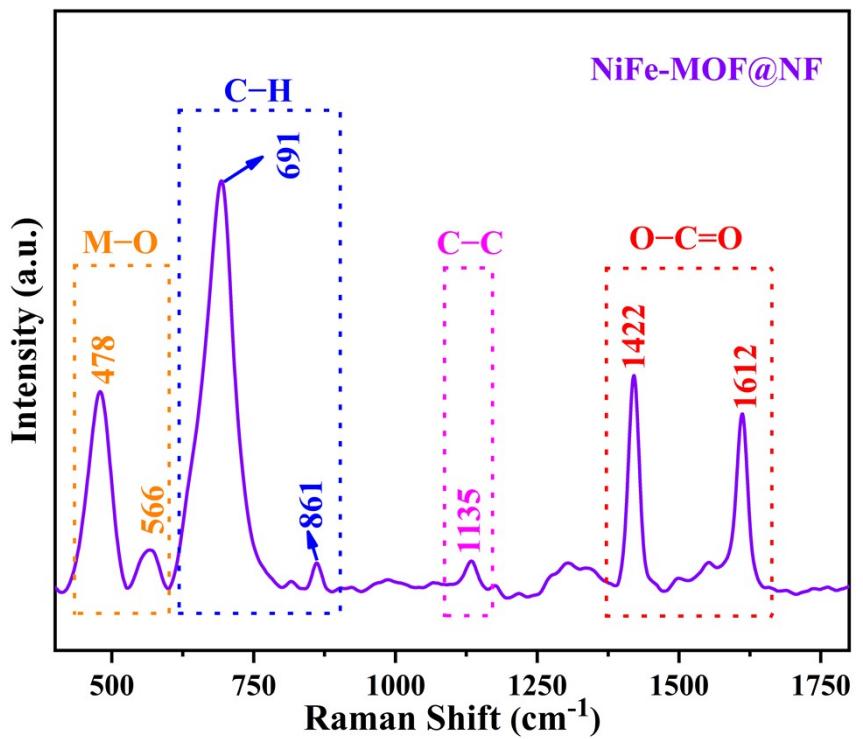


Figure S3. Raman spectrum of NiFe-MOF@NF.

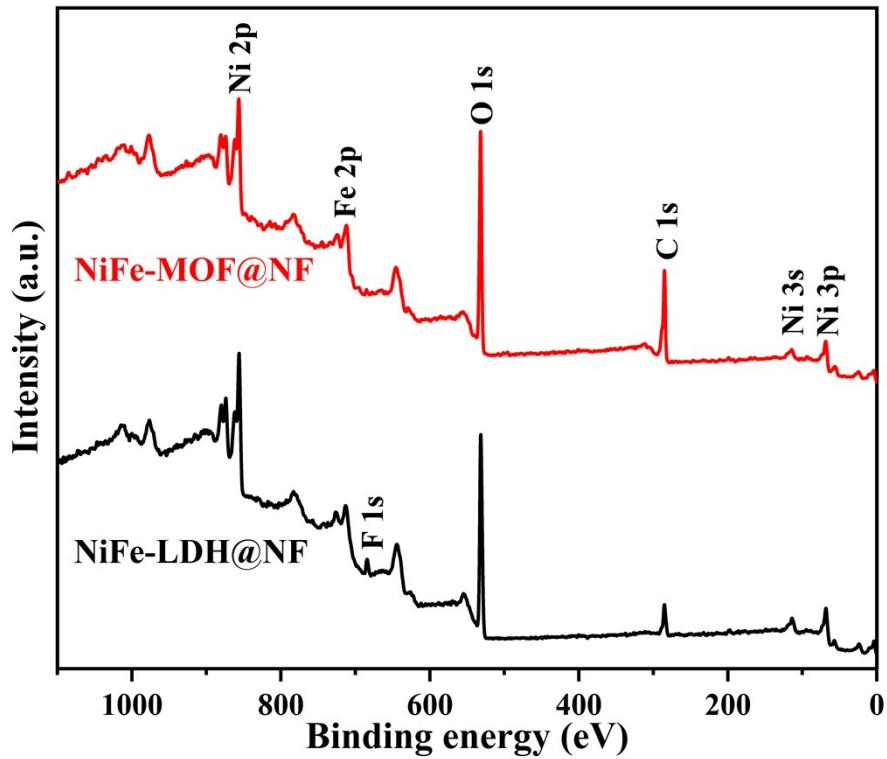


Figure S4. The survey XPS spectra of NiFe-MOF@NF and pristine NiFe-LDH@NF.

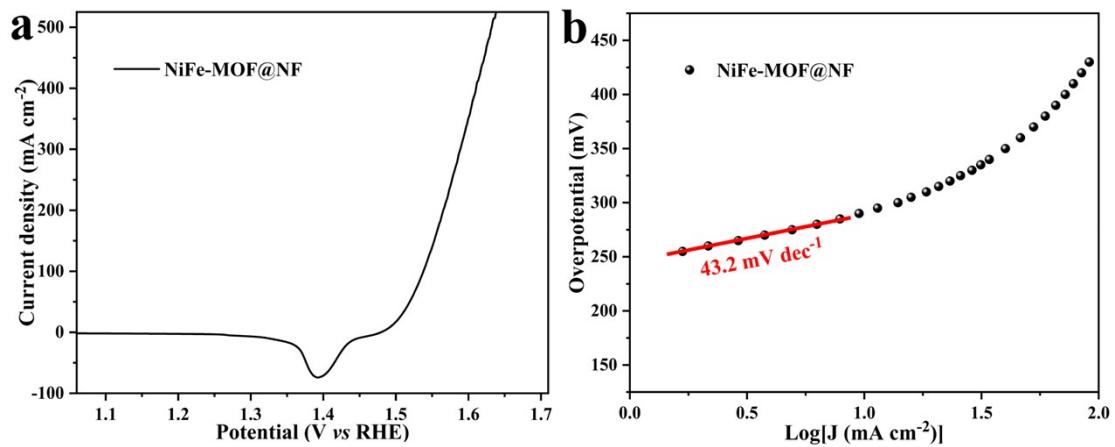


Figure S5. a) The reverse scan from 1.66 to 1.06 V (vs. RHE) for NiFe MOF@NF in 1 M KOH electrolyte; b) Tafel plot of NiFe-MOF@NF recorded using steady state test.

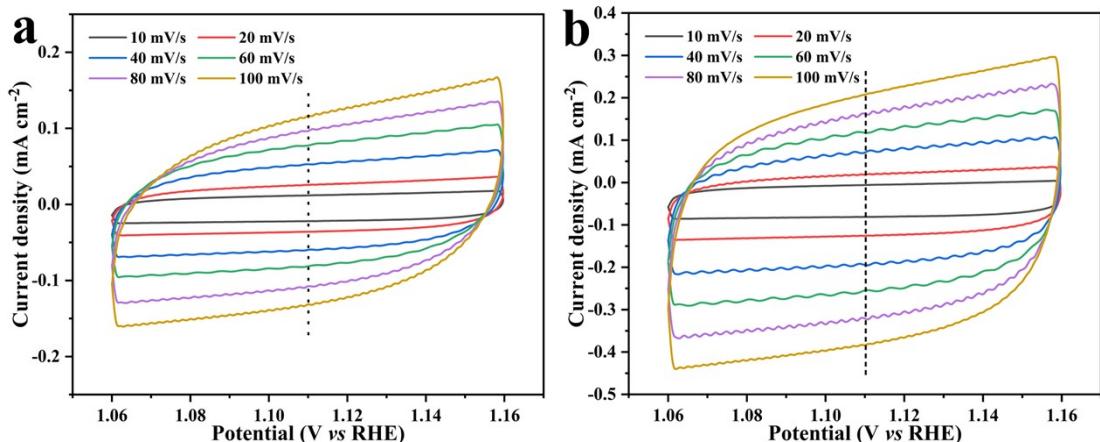


Figure S6. CV curves of (a) NiFe-LDH@NF, (b) NiFe-MOF@NF at different scan rates from 10 to 100 mV s^{-1} .

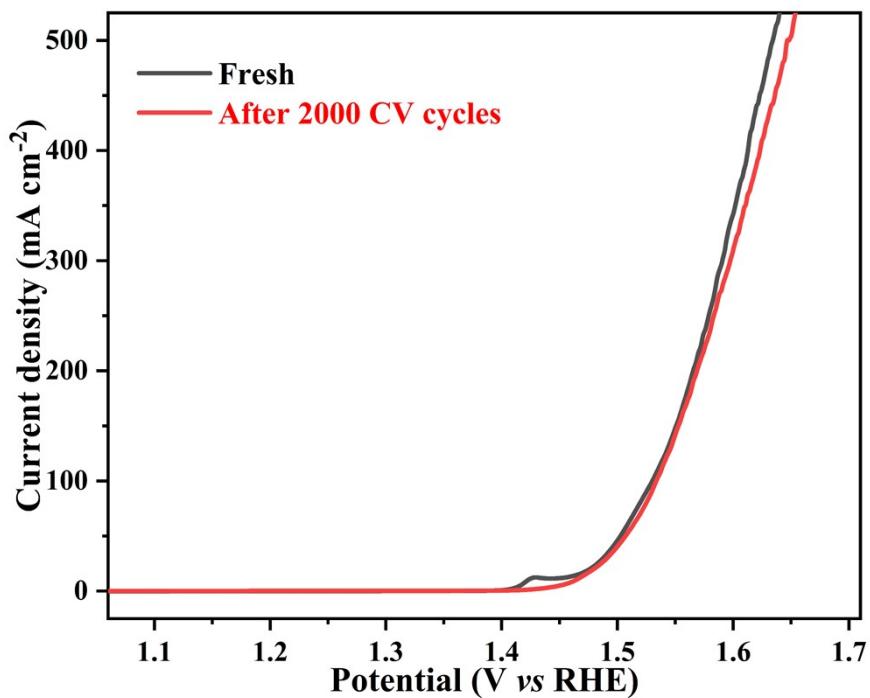


Figure S7. LSV curves before and after 2000 CV of NiFe-MOF@NF.

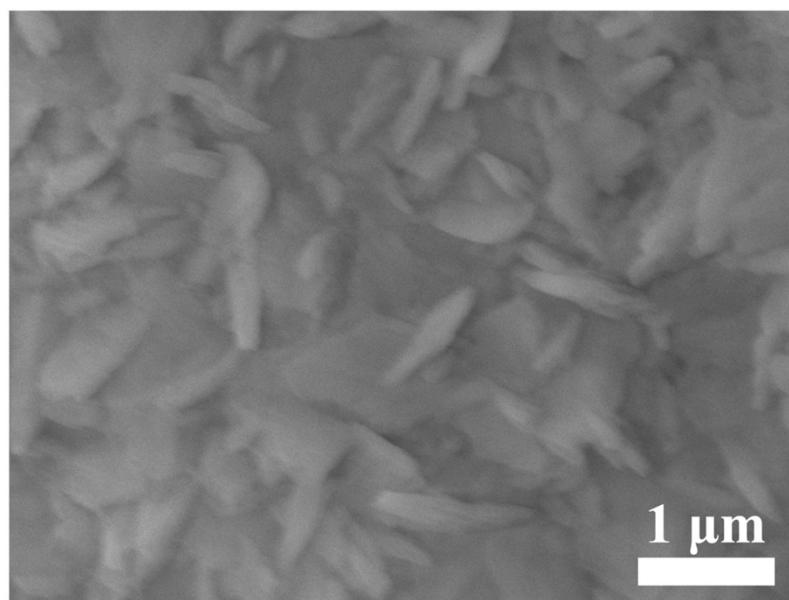


Figure S8. The SEM image of NiFe-MOF@NF electrode after the stability test.

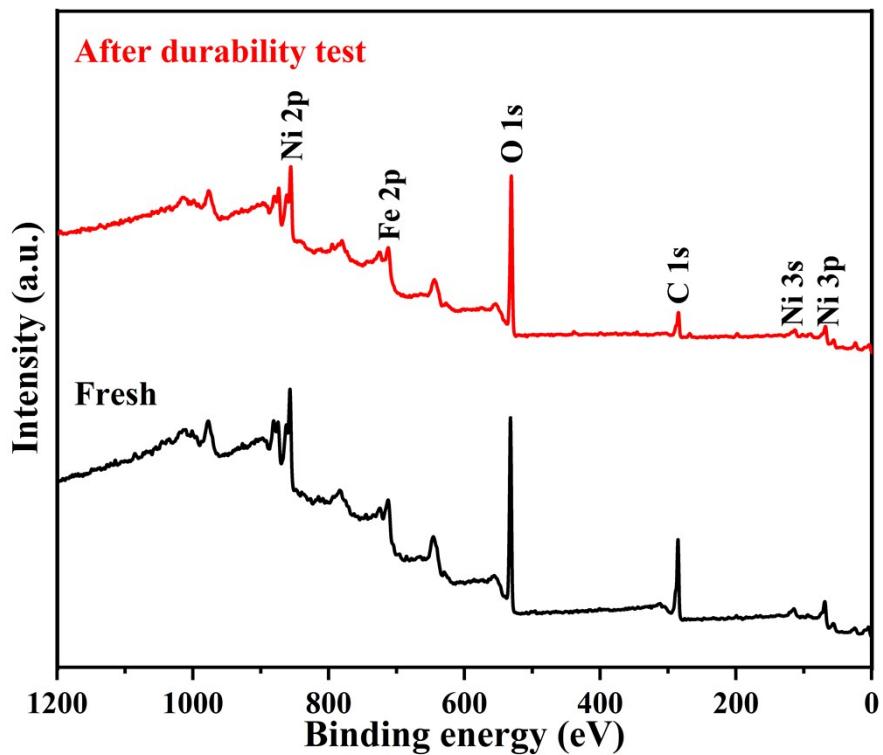


Figure S9. The survey XPS spectra of NiFe-MOF@NF before and after the durability test.

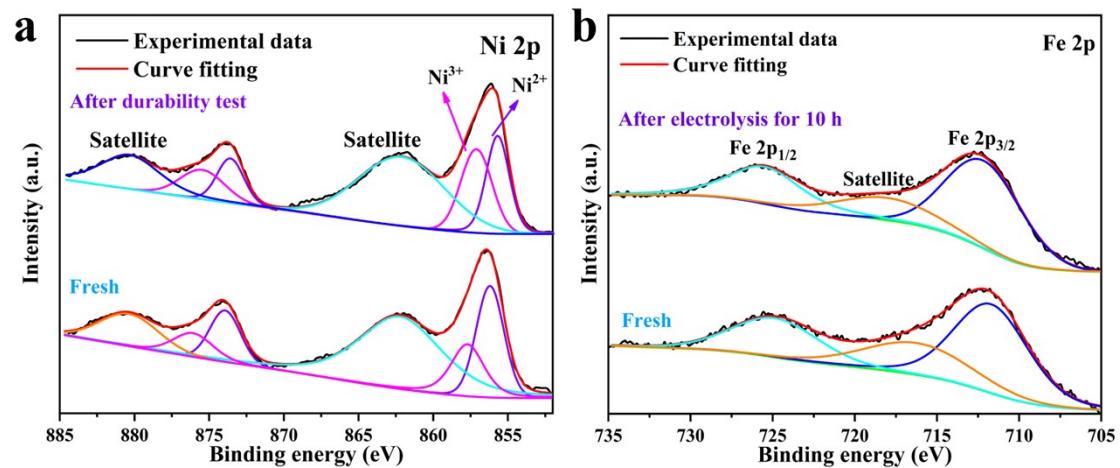


Figure S10. a) Ni 2p and b) Fe 2p spectra of fresh NiFe-MOF@NF and corresponding catalyst after the durability test.

Table S1. OER performances comparison of the electrocatalysts and NiFe-MOF@NF.

Electrocatalyst	Overpotential (mV)	Tafel slope (dec ⁻¹)	Electrolyte	Reference
NiFe-MOF@NF	265@10mA cm ⁻²	38.1	1M KOH	This work
NiCo-BDC BMNSs	230@10mA cm ⁻²	61	1M KOH	1
MCCF/NiMn-MOFs	280@10mA cm ⁻²	86	1M KOH	2
NiFe-BTC-GNPs MOF	220@10mA cm ⁻²	51	1M KOH	3
Fe/Ni _{2.4} /Co _{0.4} -MIL-53	236@20mA cm ⁻²	52.2	1M KOH	4
CoNi-MOFNA	215@10mA cm ⁻²	51.6	1M KOH	5
MIL-53(FeNi)/NF	233@50mA cm ⁻²	31.3	1M KOH	6
Co ₃ Fe-MOF	280@10mA cm ⁻²	38	1M KOH	7
Ni–Fe–MOF NSs	221@10mA cm ⁻²	56	1M KOH	8
MOF-Fe/Co(1:2)	238@10mA cm ⁻²	52	1M KOH	9
D-Ni-MOF NSA	219@10mA cm ⁻²	48.2	1M KOH	10
CoZn MOF/CC	287@10mA cm ⁻²	76.3	1M KOH	11
CoNiBDC/CA-350	192@10mA cm ⁻²	39	1M KOH	12

Reference

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