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Supplementary Information

Synthesis and Characterization of Rhena[10]annulynes

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1. Experimental Procedures

General Information: All manipulations were carried out under an argon atmosphere using standard Schlenk techniques unless otherwise stated. Solvents were distilled under nitrogen from sodium benzophenone (tetrahydrofuran) or calcium hydride (CH₂Cl₂). Reagents were used as received from commercial sources without further purification. ReH₅(PMePh₂)₃ (1),¹ alkynol 2 (2a, R = ¹Bu; 2b, R = 1-Ad),²⁻⁴ were prepared according to the previously published procedure. ¹H, ¹³C{¹H}, and ³¹P{¹H} spectra were collected on a Bruker AVIII-400 spectrometer at room temperature. ¹H and ¹³C NMR shifts are relative to TMS, and ³¹P chemical shifts relative to 85% H₃PO₄. Two-dimensional and one-dimensional NMR spectra are abbreviated as HMBC (heteronuclear multiple bond coherence) and DEPT (distortionless enhancement by polarization transfer). The absolute values of the coupling constants are given in hertz (Hz). Multiplicities are abbreviated as s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet) and br (broad). High resolution mass spectra (HRMS) experiments were recorded on an Agilent 1290-6545XT. Elemental analyses were performed on a Flash Smart CHNS/O elemental analyzer.

Synthesis and characterization of Complex 3a: The solution of rhenium hydride ReH₅(PMePh₂)₃ (1) (1.7054 g, 2.15 mmol) and alkynol (2a) (1.0645 g, 3.45 mmol) in dichloromethane (20 mL) was added with hydrogen chloride (2.0 M in diethyl ether, 3.2 mL, 6.4 mmol), and the mixture was stirred at room temperature for 12 h to obtain a green solution. The solvent of the reaction mixture was removed under vacuum. The residue was washed with methanol (2.5 mL × 1, 2 mL × 4) to give a green solid, which was collected by filtration and dried under vacuum. Yield, 2.2075 g, 89.3 %. ³¹P{¹H} NMR (161.9 MHz, CD₂Cl₂): δ -10.73 (d, *J*_{PP} = 11.5 Hz, *P*MePh₂), -13.48 ppm (t, *J*_{PP} = 11.2 Hz, *P*MePh₂). ¹H NMR (400.0 MHz, CD₂Cl₂): 0.29 (s, 9 H, Si*M*₃), 1.08 (s, 9 H, H15), 1.90 (d, *J*_{PH} = 9.3 Hz, 3 H, *PMe*Ph₂), 2.39 (br, 6 H, *PMe*Ph₂), 4.25 (br, 1 H, H2), 6.74 (dd, *J*_{PH} = 8.6 Hz, *J*_{HH} = 7.3 Hz, 4 H, *Ph*), 6.92 (t, *J*_{HH} = 6.9 Hz, 4 H, *Ph*), 7.02-7.22 (m, 8 H, *Ph*), 7.20-7.29 (m, 2 H, H11 and H12), 7.27-7.43 (m, 6 H, *Ph*), 7.50 (d, *J*_{HH} = 7.4 Hz, 1 H, H13), 7.52-7.76 (m, 8 H, *Ph*), 8.12 ppm (d, *J*_{HH} = 7.4 Hz, 1 H, H10). ¹³C NMR (100.6 MHz, CD₂Cl₂, plus ¹³C DEPT-135, ¹H-¹³C HSQC and ¹H-¹³C HMBC): δ 262.2 (q, *J*_{PC} = 15.4 Hz, C1), 141.7 (d, *J*_{PC} = 2.2 Hz, C2), 139.6 (d, *J*_{PC} = 46.9 Hz, *Ph*), 139.6 (q, *J*_{PC} =

3.0 Hz, C3), 138.4 (t, $J_{PC} = 20.3$ Hz, *Ph*), 138.1 (t, $J_{PC} = 22.8$ Hz, *Ph*), 134.8 (s, C10), 133.9 (t, $J_{PC} = 4.2$ Hz, *Ph*), 133.5 (s, C13), 133.2 (t, $J_{PC} = 4.6$ Hz, *Ph*), 133.0 (d, $J_{PC} = 9.1$ Hz, *Ph*), 129.6 (s *Ph*), 129.4 (s *Ph*), 129.2 (s, C11), 129.1 (s, *Ph*), 128.7 (s, C12), 128.3 (t, $J_{PC} = 3.9$ Hz, *Ph*), 127.6-127.9 (m, *Ph*), 125.4 (s, C6), 125.3 (s, C7), 104.7 (s, C8), 99.0 (s, C5), 98.6 (s, C9), 94.9 (s, C4), 37.4 (s, C14), 29.0 (s, C15), 20.8 (d, $J_{PC} = 35.8$ Hz, *PMe*Ph₂), 17.5 (t, $J_{PC} = 16.6$ Hz, *PMe*Ph₂), 0.1 ppm (s, Si*Me*₃). Elemental analysis calcd (%) for C₅₉H₆₂Cl₂P₃ReSi: C, 61.66; H, 5.44. Found: C, 61.79; H, 5.49.

Synthesis and characterization of Complex 3b. The solution of rhenium hydride ReH₅(PMePh₂)₃ (1) (1.7785 g, 2.25 mmol) and alkynol (2b) (1.2998 g, 3.37 mmol) in dichloromethane (20 mL) was added with hydrogen chloride (2.0 M in diethyl ether, 3.2 mL, 6.4 mmol), and the mixture was stirred at room temperature for 12 h to obtain a green solution. The solvent of the reaction mixture was removed under vacuum. The residue was washed with methanol (4 mL \times 5) to give a green solid, which was collected by filtration and dried under vacuum. Yield, 1.9261 g, 69.9 %.³¹P{¹H} NMR (161.9 MHz, CD_2Cl_2): δ -10.38 (d, J_{PP} = 11.4 Hz, *P*MePh₂), -13.29 ppm (t, J_{PP} = 11.4 Hz, *P*MePh₂). ¹H NMR (400.0 MHz, CD₂Cl₂): 0.31 (s, 9 H, SiMe₃), 1.57-1.96 (m, 15 H, Ad), 2.05 (br, 3 H, $PMePh_2$), 2.36 (br, 6 H, $PMePh_2$), 4.28 (br, 1 H, H2), 6.78 (dd, $J_{PH} = 8.4$ Hz, $J_{HH} = 7.1$ Hz, 4 H, Ph), 6.95 (t, J_{HH} = 7.1 Hz, 4 H, Ph), 7.10-7.45 (m, 16 H, Ph), 7.51 (d, J_{HH} = 7.2 Hz, 1 H, H13), 7.55-7.77 (m, 8 H, *Ph*), 7.93 ppm (d, $J_{\rm HH}$ = 7.4 Hz, 1 H, H10). ¹³C NMR (100.6 MHz, CD₂Cl₂, plus ¹³C DEPT-135, ¹H-¹³C HSQC and ¹H-¹³C HMBC): δ 262.2 (q, $J_{PC} = 15.5$ Hz, C1), 142.0 (d, $J_{PC} = 3.7$ Hz, C2), 139.5 (d, $J_{PC} = 46.5$ Hz, Ph), 139.5 (q, $J_{PC} = 3.8 \text{ Hz}, \text{ C3}$, 138.3 (t, $J_{PC} = 20.5 \text{ Hz}, Ph$), 137.9 (t, $J_{PC} = 22.8 \text{ Hz}, Ph$), 134.5 (s, C10), 133.8 (t, $J_{PC} = 4.6$ Hz, Ph), 133.6 (s, C13), 133.1 (t, $J_{PC} = 4.8$ Hz, Ph), 132.9 (d, J_{PC} = 8.7 Hz, Ph), 129.4 (s Ph), 129.2 (s Ph), 129.0 (s, Ph), 128.8 (s, C11), 128.5 (s, C12), 128.1 (t, $J_{PC} = 4.2$ Hz, Ph), 127.5-127.8 (m, Ph), 125.1 (s, C6), 124.8 (s, C7), 104.7 (s, C8), 98.7 (s, C5), 98.7 (s, C9), 94.2 (s, C4), 40.7 (s, Ad), 38.7 (s, Ad), 36.7 (s, Ad), 28.7 (s, Ad), 20.5 (d, $J_{PC} = 35.6$ Hz, $PMePh_2$), 17.3 (t, $J_{PC} = 16.6$ Hz, $PMePh_2$), 0.0 ppm (s, SiMe₃). Elemental analysis calcd (%) for $C_{65}H_{68}Cl_2P_3ReSi$: C, 63.61; H, 5.58. Found: C, 63.76; H, 5.59.

Synthesis and characterization of Complex 4a. To a solution of (3a) (0.3003 g, 0.26 mmol) in THF (6 mL) was added tetrabutylammonium flouride (TBAF) (1.0 M in THF, 0.32 mL, 0.32 mmol). The mixture was stirred at room temperature for 3 h to give a yellow green solution. The solvent of reaction mixture was dried under vacuum to give a solid. The resulting solid was dissolved in dichloromethane (1.5 mL) and then was loaded on a silica gel column. The column was flashed with dichloromethane and eluted with dichloromethane/tetrahydrofuran (20:1) to give an orange solution. The solvent of the orange solution was dried under vacuum to give an orange solid, which was washed with tetrahydrofuran (0.1 mL \times 2) and hexane (0.6 mL \times 4) and dried under vacuum. Yield, 139.9 mg, 61.0 %. ³¹P{¹H} NMR (161.9 MHz, CD₂Cl₂): δ 1.12, -0.25 ppm (ABq, J_{AB} = 224.3 Hz, RePMePh₂). ¹H NMR plus ¹H $\{^{31}P\}$ (400.0 MHz, CD₂Cl₂): δ 1.08 (s, 9 H, H15), 2.10 (d, $J_{PH} = 8.4$ Hz, 3 H, $PMePh_2$), 2.41 (d, $J_{PH} = 8.7$ Hz, 3 H, $PMePh_2$), 5.10 (t, $J_{PH} =$ 5.5 Hz, 1 H, H8), 5.86 (br, 1 H, H2), 6.83-7.89 (m, 24 H, *Ph*), 7.02 (t, *J*_{HH} = 7.0 Hz, 1 H, H12 mentioned above), 7.16 (t, $J_{\rm HH}$ = 7.5 Hz, 1 H, H11 mentioned above), 7.25 (d, $J_{\rm HH}$ = 9.0 Hz, 1 H, H13 mentioned above), 7.34 ppm (d, $J_{\rm HH}$ = 7.8 Hz, 1 H, H10 mentioned above). ¹³C NMR (100.6 MHz, CD₂Cl₂, plus ¹³C DEPT-135, ¹H-¹³C HSQC and ¹H-¹³C HMBC): δ 332.6 (t, J_{PC} = 11.7 Hz, C1), 276.5 (t, J_{PC} = 13.7 Hz, C9), 156.3 (t, J_{PC} = 3.5 Hz, C3), 137.1 (s, C2), 135.5 (dt, *J*₁ = 43.2 Hz, *J*₂ = 9.2 Hz, *Ph*), 134.2 (dd, *J*₁ = 41.1 Hz, $J_2 = 8.0$ Hz, Ph), 133.6 (s, C7), 133.5 (d, $J_{PC} = 8.8$ Hz, Ph), 133.0 (d, $J_{PC} = 7.8$ Hz, Ph), 132.6 (d, J_{PC} = 16.5 Hz), 132.6 (s, *Ph*), 131.8 (s, C11), 130.5 (d, J_{PC} = 23.0 Hz, *Ph*), 130.3 (s, C12), 130.0 (s, *Ph*), 128.3 (dd, $J_1 = 13.4$ Hz, $J_2 = 8.6$ Hz *Ph*), 127.6 (dd, $J_1 = 21.6$ Hz, $J_2 = 8.3$ Hz, Ph), 127.2 (s, C13), 125.7 (s, C10), 121.6 (s, C6), 113.5 (s, C8), 104.3 (s, C5), 93.3 (s, C4), 36.4 (s, C14), 29.1 (s, C15), 15.0 (dd, $J_1 = 30.4$ Hz, $J_2 = 5.5$ Hz, $PMePh_2$), 12.8 ppm (dd, $J_1 = 29.2$ Hz, $J_2 = 5.9$ Hz, $PMePh_2$). Elemental analysis (%) calcd for C₄₃H₄₁Cl₂P₂Re: C 58.90, H 4.71; found: C 58.86, H 4.65.

Synthesis and characterization of Complex 4b. To a solution of (**3b**) (0.5000 g, 0.41 mmol) in THF (10 mL) was added tetrabutylammonium flouride (TBAF) (1.0 M in THF, 0.61 mL, 0.61 mmol). The mixture was stirred at room temperature for 3 h to give a yellow green solution. The solvent of reaction mixture was dried under vacuum to give a solid. The resulting solid was dissolved in dichloromethane (2 mL) and then was loaded on a

silica gel column. The column was flashed with dichloromethane and eluted with dichloromethane/tetrahydrofuran (20:1) to give an orange solution. The solvent of the orange solution was dried under vacuum to give an orange solid. The resulted solid was redissolved with tetrahydrofuran (1.5 mL) and then hexane (20 mL) was added to give an orange precipitate, which was collected by filtration, washed with hexane $(3 \text{ mL} \times 2)$ and dried under vacuum. Yield, 175.4 mg, 45.0 %. ³¹P{¹H} NMR (161.9 MHz, CD₂Cl₂): δ 1.29, -0.60 ppm (ABq, $J_{AB} = 224.6$ Hz, RePMePh₂). ¹H NMR plus ¹H{³¹P} (400.0 MHz, CD₂Cl₂,): 1.56-1.80 (m, 12 H, Ad), 2.01 (br, 3 H, Ad), 2.07 (d, *J*_{PH} = 9.0 Hz, 3 H, *PMePh*₂), 2.38 (d, *J*_{PH} = 9.4 Hz, 3 H, P*Me*Ph₂), 5.04 (t, *J*_{PH} = 5.5 Hz, 1 H, H8), 5.82 (br, 1 H, H2), 6.83-7.89 (m, 24 H, Ph), 7.00 (t, $J_{\rm HH}$ = 7.0 Hz, 1 H, H12 mentioned above), 7.13 (t, $J_{\rm HH}$ = 7.4 Hz, 1 H, H11 mentioned above), 7.22 (d, $J_{\rm HH}$ = 8.0 Hz, 1 H, H13 mentioned above), 7.32 ppm (d, $J_{\rm HH}$ = 7.7 Hz, 1 H, H10 mentioned above). ¹³C NMR (100.6 MHz, CD₂Cl₂, plus ¹³C DEPT-135, ¹H-¹³C HSQC and ¹H-¹³C HMBC): δ 332.6 (t, J_{PC} = 11.8 Hz, C1), 276.8 (t, *J*_{PC} = 13.1 Hz, C9), 156.2 (t, *J*_{PC} = 3.5 Hz, C3), 136.8 (s, C2), 135.5 (dt, *J*₁ = 45.9 Hz, *J*₂ = 4.9 Hz, *Ph*), 134.4 (dd, *J*₁ = 42.3 Hz, *J*₂ = 5.1 Hz, *Ph*), 134.2 (dd, *J*₁ = 43.1 Hz, J₂ = 5.4 Hz, Ph), 133.5 (s, C7), 133.3 (dd, J₁ = 53.0 Hz, J₂ = 8.3 Hz, Ph), 132.6 (t, J_{PC} = 8.0 Hz, Ph), 131.8 (s, C10), 130.5 (d, $J_{PC} = 27.7$ Hz, Ph), 130.2 (s, C12), 130.0 (d, J_{PC} =7.7 Hz, *Ph*), 128.3 (dd, *J*₁ = 15.9 Hz, *J*₂ = 9.0 Hz, *Ph*), 127.6 (dd, *J*₁ = 26.2 Hz, *J*₂ = 8.7 Hz, Ph), 127.1 (s, C13), 125.6 (s, C11), 121.5 (s, C6), 113.4 (s, C8), 104.6 (s, C5), 93.0 (s, C4), 41.3 (s, Ad), 38.3 (s, Ad), 36.6 (s, Ad), 28.8 (s, Ad), 14.9 (dd, *J*₁ = 32.2 Hz, *J*₂ = 3.4 Hz, PMePh₂), 12.8 ppm (dd, $J_1 = 31.3$ Hz, $J_2 = 3.5$ Hz, PMePh₂). Elemental analysis (%) calcd for C₄₉H₄₇Cl₂P₂Re: C 61.63, H 4.96; found: C 61.90, H 5.27.

2. X-ray Crystallographic Analysis

Single crystals of **3a** and **4a** suitable for X-ray diffraction were grown from dichloromethane (for **3a**) or toluene (for **4a**) solution layered with hexane. The single crystal X-ray diffraction data of **3a** was collected on a SuperNova Dual Cu at home/near Atlas diffractometer using Mo K α radiation ($\lambda = 0.71073$ Å). The single crystal X-ray diffraction data of **4a** was collected on a XtaLAB Synergy Dualflex HyPix diffractometer using Cu K α radiation ($\lambda = 1.54184$ Å). Using Olex2,⁵ the structures of **3a** and **4a** were solved with the Shelxt⁶ structure solution program using Intrinsic Phasing and refined

with the Shelxt⁶ refinement package using least-squares minimization. All non-hydrogen atoms were refined anisotropically unless otherwise stated. Hydrogen atoms were placed at idealized positions and assumed the riding model. Some of the solvent molecules and phenyl groups were disordered and refined with suitable restrains. The X-ray crystal structures have been deposited in the Cambridge Crystallographic Data Centre under the deposition numbers CCDC-2152928 (complex 3a) and CCDC-2152927 (complex 4a). The data can be obtained free of charge from the CCDC (www.ccdc.cam.ac.uk/data request/cif).

	3a	4a
Empirical formula	$C_{60}H_{64}Cl_4P_3ReSi$	$C_{93.58}H_{90.67}Cl_4P_4Re_2$
Color & habit	pink, plate	yellow, block
Crystal size (mm ³)	0.12 x 0.11 x 0.05	0.1 x 0.08 x 0.05
Temperature (K)	100.00(10)	100.00(19)
Crystal system	monoclinic	orthorhombic
Space group	P2 ₁ /n	Pna2 ₁
a (Å)	16.2507(3)	24.4910(4)
b (Å)	20.9637(2)	12.5355(2)
c (Å)	18.1840(3)	28.4851(5)
α (°)	90	90
β (°)	115.618(2)	90
γ (°)	90	90
V (Å ³), Z	5585.86(17), 4	8745.1(3), 4
D _{cal} (g/cm ³)	1.467	1.408
μ (mm ⁻¹)	2.512	7.472
2θ range for data collection (°)	6.786 to 61.03	6.206 to 130
Reflections collected	30022	40523
Indep. Reflection, R_{int} , R_{sigma}	14449, 0.0219, 0.0331	12591, 0.0468, 0.0477
Data/ restraints/ parameters	14449/0/631	12583/385/980
Goodness-of-fit on F ²	1.042	1.034
$R_1 [I > 2sigma(I)], wR_2$	0.0237, 0.0518	0.0414, 0.1105
R1 (all data), wR2	0.0275, 0.0537	0.0454, 0.1144
Largest diff. peak and hole ($e{\cdot} {\rm \AA}^{-3})$	1.49/-1.22	1.45/-0.59

 Table S1. Crystallographic data and refinement details for 3a and 4a.

3. NMR and HRMS Spectra













Figure S8. The HRMS spectrum of [3a-Cl]⁺ measured in dichloromethane.







Figure S12. The ${}^{13}C{}^{1}H$ NMR spectrum (100.6 MHz) of **3b** in CD₂Cl₂.













Figure S18. The ¹H NMR spectrum (400.0 MHz) of 4a in CD₂Cl₂.







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Figure S28. The ${}^{1}H{}^{31}P{}$ NMR spectrum (400.0 MHz) of 4b in CD₂Cl₂.





Figure S32. The $^{1}H^{-13}C$ HMBC spectrum of 4b in $CD_{2}Cl_{2}$.



Figure S34. The HRMS spectrum of [4b-Cl]⁺ measured in dichloromethane.

4. Computational Details

All the optimizations were performed with the Gaussian 16 Ver. C01 software package.⁷ All of the structures evaluated were optimized at the B3LYP/def2-TZVP level of density functional theory (DFT).⁸⁻¹¹ The anisotropy of the induced current density (AICD) calculations were carried out with the AICD program.^{12,13} The analysis of the iso-chemical shielding surfaces (ICSS)^{14,15} and the π -type localized orbital locator (LOL- π)¹⁶ were performed using the wavefunction analysis code Multiwfn 3.8¹⁷ and the GaussView 6.1.1 software.¹⁸

5. The Calculated Cartesian Coordinates with Electronic Energies

Н

С

Η



E = -2070.56120876 a.u

C -4.22199100 -2.54441700 -0.08055700 H -4.49864100 -2.67270300 0.96912900 H -5.09948500 -2.15296500 -0.59834700 H -3.97591300 -3.52400300 -0.48859200 Ph.MeP H

-4.00111700 2.92075700 0.41460200

-2.14762300 3.02387500 -0.59606500

-2.31988900 4.08129300 -0.76700500

Re	0.62345000	-0.07738800	-0.10364300
Cl	2.12434200	-1.81773200	0.92046000
Cl	2.28717800	1.61287300	0.79510900
Р	2.13922500	-0.47414000	-1.99602200
Р	-0.12106000	0.22294700	2.21493700
С	-0.95444800	2.49541400	-1.15932100
Н	-0.47550700	3.07880800	-1.94391500
С	-0.32647200	1.37086700	-0.78310700
С	-3.05782600	-1.59858200	-0.20300200
С	-3.21485900	-0.25400300	0.09883800
С	-1.81666800	-2.03051200	-0.66258900
Н	-1.72175900	-3.02465000	-1.09099400
С	-3.21347800	0.96806500	0.16387200
С	-0.69234700	-1.24215800	-0.54513100
Н	2.36757200	-1.81165200	-2.35446200
Н	3.45114300	-0.02685300	-1.77854900
Н	1.88814700	0.08311200	-3.26536700
Н	-1.14524000	-0.59352500	2.73217500
Н	0.89431900	0.00546000	3.15350300
Н	-0.59426100	1.48879500	2.59358100
С	-3.15369500	2.34524200	0.05857500



4a

E = -334	5.469958 a.u.		
Re	-0.41666600	-0.14924600	-0.95970200
Cl	-1.03812700	-2.07727000	-2.48914500
Cl	-0.34936500	1.34452000	-3.05194400
Р	-2.89085200	0.29628300	-0.96182000
Р	1.92217000	-0.82946100	-1.71489600
С	-5.21950500	-0.54884500	0.40026500
Н	-5.54709400	0.48011500	0.33539900
С	-3.96579900	-0.92126600	-0.09501100
С	-3.57458800	-2.25924100	-0.00228900
Н	-2.62383500	-2.56939600	-0.41414000
С	-3.35671300	1.90374700	-0.21810300
С	0.17727400	2.62762500	0.52911100
Н	-0.65541600	3.32616600	0.44851500

С	-0.01240400	1.45593800	-0.04933200
С	-6.05541900	-1.48938400	0.98964400
Н	-7.02050000	-1.18240400	1.37309400
С	-5.65503500	-2.81722800	1.08438800
Н	-6.30721700	-3.55047600	1.54232100
С	0.41232400	-1.24894700	2.86514900
С	3.09810000	1.70058400	-1.58216800
Н	2.09410600	2.06594800	-1.75047300
С	-4.41660500	-3.19874300	0.58260000
Н	-4.10212400	-4.23343000	0.63967800
С	1.17716000	-0.08111800	2.66600800
С	5.45380500	2.14074500	-1.31527100
Н	6.27613000	2.84289400	-1.25647200
С	0.00255900	-0.98195200	5.31254900
Н	-1.04245400	-0.71307400	5.14688300
Н	0.59102600	-0.06412600	5.32331200
Н	0.08184000	-1.45096700	6.29582600
С	1.72777500	4.46220200	1.10936900
Н	1.13686800	5.09925300	0.46277600
С	1.36198900	3.11399600	1.23459600
С	4.15707700	2.59816800	-1.51404800
Н	3.96301300	3.65851100	-1.61203100
С	-0.42028000	-1.64579000	1.85241500
Н	-1.09398300	-2.48236500	2.00104300
С	0.50240500	-1.95461000	4.22320600
С	-0.34028300	-3.23448000	4.27125000
Н	-0.02927600	-3.95234500	3.51007400
Н	-1.40365500	-3.02836900	4.13643600
Н	-0.22130400	-3.71100100	5.24602900
С	3.60405500	4.16312800	2.59007400
Н	4.47038600	4.56414900	3.09998700
С	1.69843100	0.98429200	2.41221100
С	2.12599400	2.29688700	2.11445100
С	3.32668200	0.32844100	-1.45545000
С	3.24667400	2.84178500	2.76846200
Н	3.82011500	2.20444500	3.42856800
С	-0.44598600	-1.03523000	0.58875000
С	-3.61014100	0.32102800	-2.64102600
Н	-3.47056400	-0.67285200	-3.06259700
Н	-3.06439700	1.02825700	-3.26224400
Н	-4.67154100	0.56525900	-2.59778300
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