

Supplementary Information

Efficient hydrogen evolution by reconstruction of NiMoO₄- CoO via Mo recombination

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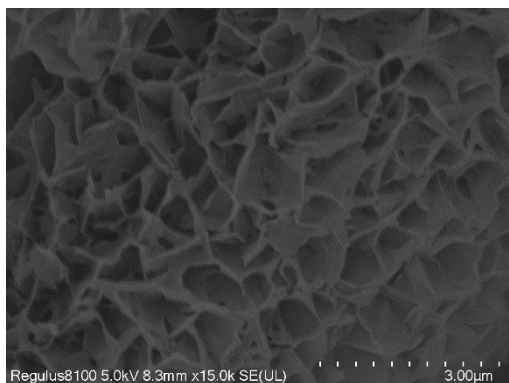


Fig. S1 SEM of NiCoMo-LDH/CC precursor.

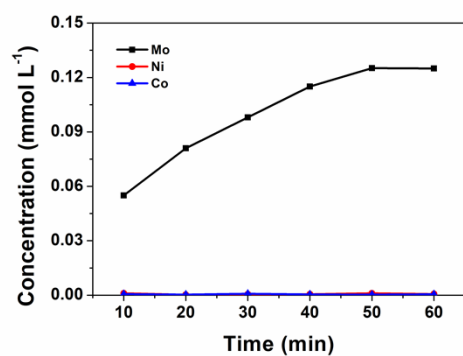


Fig. S2 Time dependent concentration of dissolved Mo, Ni, and Co in the electrolyte of NiMoO₄-CoO/CC.

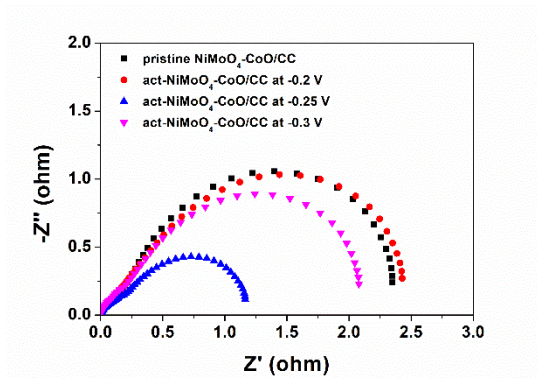


Fig. S3 Nyquist plots of pristine NiMoO₄-CoO/CC, act-NiMoO₄-CoO/CC at -0.2 V, act-NiMoO₄-CoO/CC at -0.25 V, and act-NiMoO₄-CoO/CC at -0.3 V.

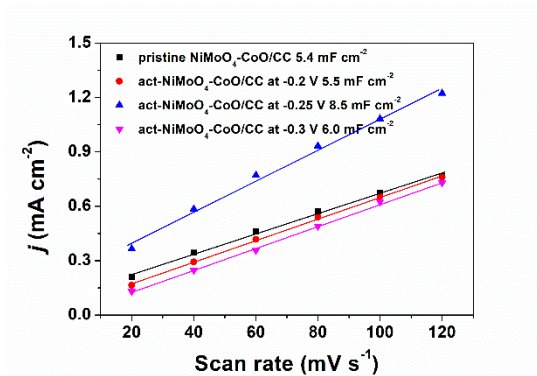


Fig. S4 The capacitive currents as a function of scan rate for pristine NiMoO₄-CoO/CC, act-NiMoO₄-CoO/CC at -0.2 V, act-NiMoO₄-CoO/CC at -0.25 V, and act-NiMoO₄-CoO/CC at -0.3 V.

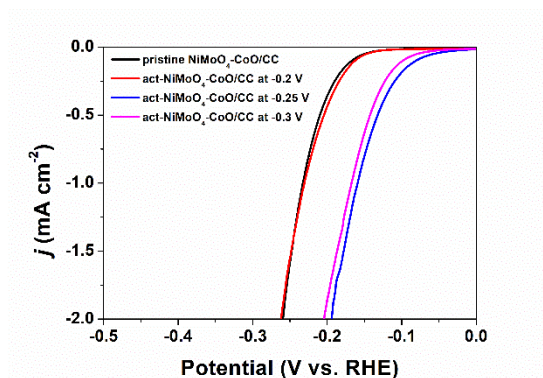


Fig. S5 ECSA normalized HER activities of pristine NiMoO₄-CoO/CC, act-NiMoO₄-CoO/CC at -0.2 V, act-NiMoO₄-CoO/CC at -0.25 V, and act-NiMoO₄-CoO/CC at -0.3 V.

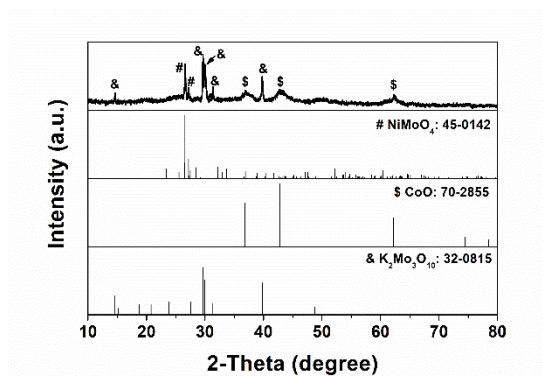


Fig. S6 XRD pattern of act-NiMoO₄-CoO/CC at -0.25 V.



Fig. S7 EDX elemental mapping of Co, K, Mo, Ni, and O elements for act-NiMoO₄-CoO/CC at -0.25 V.

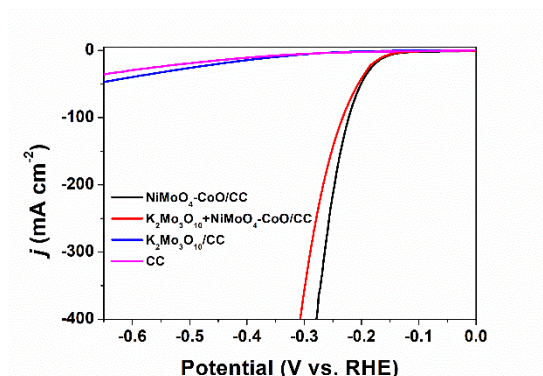


Fig. S8 LSV curves of NiMoO₄-CoO/CC, K₂Mo₃O₁₀+NiMoO₄-CoO/CC, K₂Mo₃O₁₀/CC, and CC.

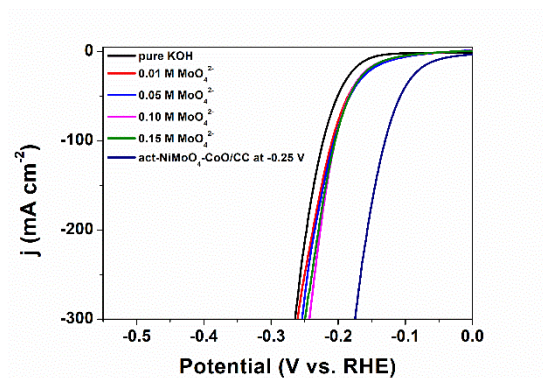


Fig. S9 LSV curves of NiMoO₄-CoO/CC in 1 M pure KOH, and 1 M KOH with the addition of 0.01, 0.05, 0.10, and 0.15 M MoO₄²⁻, as well as the LSV curve of act-NiMoO₄-CoO/CC at -0.25 V.

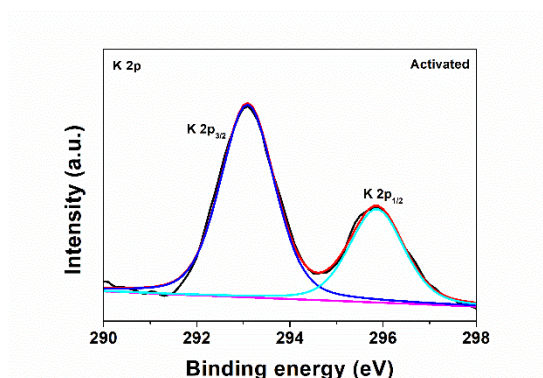


Fig. S10 High-resolution K 2p spectrum of act-NiMoO₄-CoO/CC at -0.25 V.

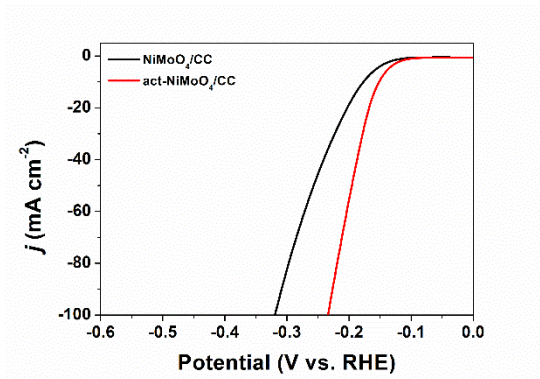


Fig. S11 LSV curves of initial NiMoO₄/CC and act-NiMoO₄/CC.

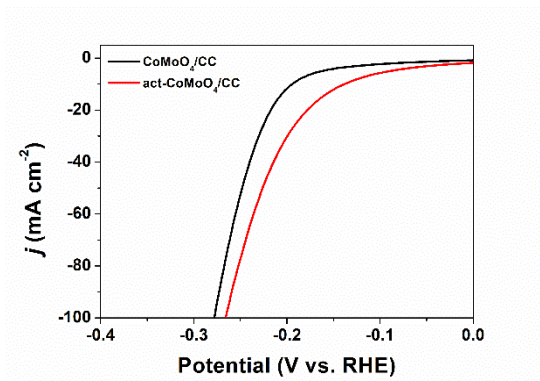


Fig. S12 LSV curves of initial CoMoO₄/CC and act-CoMoO₄/CC.

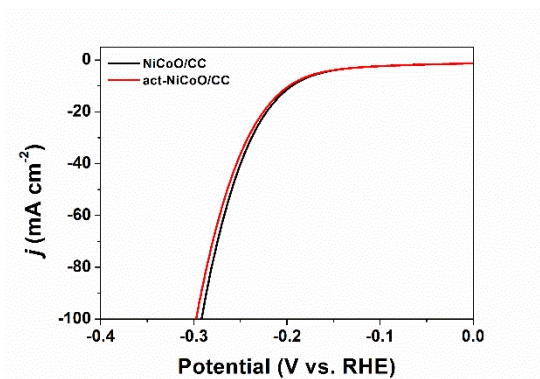


Fig. S13 LSV curves of initial NiCoO/CC and act-NiCoO/CC.

Table S1. The atomic ratios of various elements on the surface of prepared NiMoO₄-CoO/CC and act-NiMoO₄-CoO/CC at -0.25 V obtained from EDX

Element	NiMoO ₄ -CoO/CC (%)	act-NiMoO ₄ -CoO/CC at -0.25 V (%)
Ni	12.46	12.87
Co	12.72	12.17
Mo	7.83	3.31
K	N/A	3.22