

Supporting Information

KLi₂CO₃F: a Beryllium-free KBBF-type Deep-UV Carbonate with Enhanced Interlayer Interaction and Large Birefringence

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Table of contents

Sections	Titles	Pages
Table S1	Atomic coordinates and equivalent isotropic displacement parameters for KLi ₂ CO ₃ F.	S2
Table S2	Selected Bond lengths (Å) and angles (deg) for KLi ₂ CO ₃ F.	S3
Fig. S1	The coordination mode of K ⁺ cations in KBBF.	S3
Fig. S2	The XRD analysis of KLi ₂ CO ₃ F before and after melting.	S4
Fig. S3	The birefringence measurements for KLi ₂ CO ₃ F.	S4

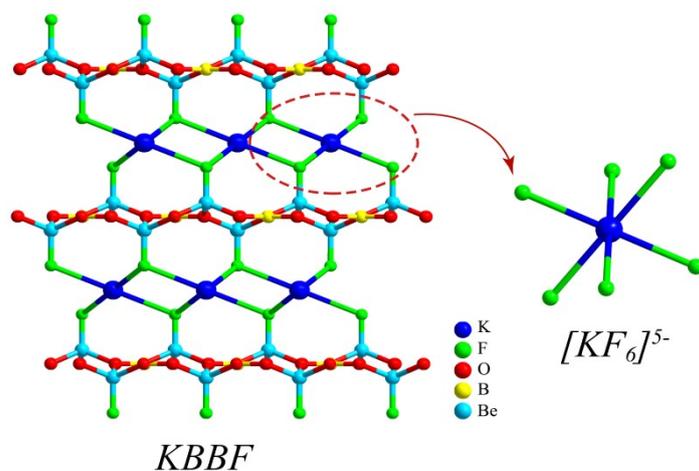
Table S1. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for $\text{KLi}_2\text{CO}_3\text{F}$. $U_{(\text{eq})}$ is defined as one third of the trace of the orthogonalized U_{ij} tensor.

atom	x	y	z	$U_{\text{eq}} (\text{\AA}^2)$	BVS
K1	3333	6667	2500	27.4(3)	0.733
F2	6667	3333	2500	34.1(7)	0.950
O3	2653(3)	2653(3)	5000	24.1(5)	2.071
C4	0	0	5000	13.3(7)	4.027
Li5	6667	3333	4297(4)	16.5(9)	1.163

Table S2. Selected Bond lengths [Å] and angles [°] for $\text{KLi}_2\text{CO}_3\text{F}$.

K1—F2^1	2.7841 (5)	$\text{K1}^{11}\text{—F2—K1}$	120.0
K1—O3	3.0841 (7)	$\text{O3}^7\text{—K1—O3}$	126.337 (14)
F2—K1^{10}	2.7841 (5)	$\text{O3}^5\text{—K1—O3}$	60.51 (3)
F2—Li5	1.803 (4)	Li5—F2—K1^{10}	90.0
O3—K1^{13}	3.0841 (7)	$\text{Li5}^{12}\text{—F2—Li5}$	180.0
O3—C4	1.2791 (16)	C4—O3—K1	111.53 (3)
O3—Li5^8	1.9281 (18)	C4—O3—Li5	125.95 (5)
C4—O3^{14}	1.2791 (16)	Li5—O3—K1	79.20 (10)
Li5—O3^{16}	1.9280 (18)	Li5—O3—Li5^8	108.09 (10)
$\text{F2}^1\text{—K1—F2}^2$	120.0	O3—C4—O3^{14}	120.0
$\text{F2}^1\text{—K1—O3}^3$	63.168 (7)	F2—Li5—O3^{16}	111.47 (11)
F2—K1—O3^4	84.71 (3)	$\text{O3}^6\text{—K1—O3}^5$	114.15 (5)
$\text{K1}^{10}\text{—F2—K1}$	120.0	$\text{Li5}^{12}\text{—F2—K1}^{11}$	90.0
$\text{O3}^{16}\text{—Li5—O3}$	107.40 (12)		

Symmetry codes: (1)+X,1+Y,+Z; (2)-1+X,+Y,+Z; (3)1-Y,1+X-Y,+Z; (4)-Y+X,+X,-1/2+Z; (5)+Y-X,1-X,+Z; (6)+Y,1-X+Y,-1/2+Z; (7)1-X,1-Y,-1/2+Z; (8)+Y,+X,1-Z; (9)-Y,1-X,1/2-Z; (10)1+X,+Y,+Z; (11)+X,-1+Y,+Z; (12)1-Y,1-X,1/2-Z; (13)1-X,1-Y,1/2+Z; (14)+Y-X,-X,+Z; (15)-Y,+X-Y,+Z; (16)1+Y-X,1-X,+Z; (17)1-Y,+X-Y,+Z; (18)1+Y,+X,1-Z; (19)+Y,-1+X,1

**Fig. S1** The coordination mode of K^+ cations in KBBF .

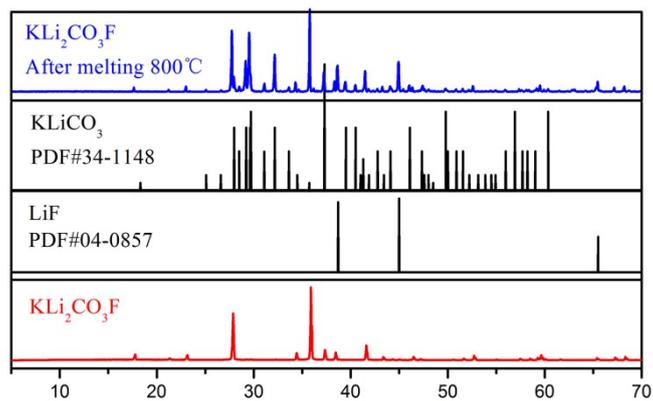


Fig. S2 The XRD analysis of $\text{KLi}_2\text{CO}_3\text{F}$ before and after melting.



Fig. S3 The birefringence measurements for $\text{KLi}_2\text{CO}_3\text{F}$.