

Supporting Information

Halogen Regulation Triggers Structural Transformation from Centrosymmetric to Noncentrosymmetric Switches in Tin Phosphate Halides $\text{Sn}_2\text{PO}_4\text{X}$ ($\text{X} = \text{F}, \text{Cl}$)

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Table S1. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$), and calculated Bond Valence Sum for $\text{Sn}_2\text{PO}_4\text{F}$. U_{eq} is defined as one third of the trace of the orthogonalized U_{ij} tensor.

atom	x	y	z	$U_{\text{eq}} (\text{\AA}^2)$	BVS
Sn(1)	2758.7 (19)	7281.7 (13)	8237.2 (4)	21.4 (3)	1.84
Sn(2)	7020 (2)	7392.5 (16)	5461.5 (5)	30.0 (3)	1.68
P(1)	1966 (7)	6478 (5)	6639.7 (13)	22.5 (6)	5.10
O(1)	2280 (20)	8260 (16)	7217 (4)	25.1 (17)	1.99
F(1)	6540 (20)	3662 (16)	5408 (4)	47 (2)	0.87
O(2)	3130 (20)	7576 (15)	5992 (5)	28 (2)	2.00
O(3)	3140 (20)	4046 (17)	6813 (5)	34 (2)	2.00
O(4)	-1309 (19)	6007 (16)	6492 (4)	28.0 (19)	1.97

Table S2. Atomic coordinates ($\times 10^4$) and equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$), and calculated Bond Valence Sum for $\text{Sn}_2\text{PO}_4\text{Cl}$. U_{eq} is defined as one third of the trace of the orthogonalized U_{ij} tensor $\text{Sn}_2\text{PO}_4\text{Cl}$.

atom	x	y	z	$U_{\text{eq}} (\text{\AA}^2)$	BVS
Sn(1)	1106.9 (4)	4414.0 (7)	1342.0 (8)	19.5 (3)	1.80
Sn(2)	3853.3 (4)	4448.1 (7)	1146.7 (11)	20.6 (3)	1.74
P(1)	2527.8 (13)	6110 (3)	6439 (7)	15.0 (5)	5.02
Cl(1)	4733 (2)	7389 (3)	490 (10)	36.3 (6)	0.57
O(1)	2603 (4)	7871 (7)	6410 (30)	28.2 (17)	2.01
O(2)	1576 (6)	5616 (8)	5110 (20)	27.3 (18)	1.95
O(3)	3436 (6)	5385 (9)	5150 (20)	28.0 (18)	1.93
O(4)	2506 (3)	5607 (9)	9554 (19)	20.7 (19)	2.10

Table S3. Selected Bond lengths (Å) and angles (deg) for Sn₂PO₄F.

Sn1—O1	2.122 (8)	Sn2—O2	2.109 (11)
Sn1—O3 ¹	2.162 (11)	Sn2—O4 ⁴	2.335 (8)
Sn1—O4 ²	2.241 (9)	P1—O1	1.526 (9)
Sn2—F1	2.084 (9)	P1—O2	1.528 (10)
Sn2—F1 ³	2.484 (9)	P1—O3	1.497 (10)
O1—Sn1—O3 ¹	86.5 (4)	P1—O4	1.588 (9)
O1—Sn1—O4 ²	87.9 (3)	O2—P1—O4	104.5 (5)
O3 ¹ —Sn1—O4 ²	82.1 (3)	O3—P1—O1	111.9 (5)
F1—Sn2—F1 ³	69.9 (3)	O3—P1—O2	115.1 (6)
F1—Sn2—O2	88.9 (4)	O3—P1—O4	104.5 (6)
F1—Sn2—O4 ⁴	75.8 (3)	P1—O1—Sn1	124.8 (5)
O2—Sn2—F1 ³	77.3 (4)	Sn2—F1—Sn2 ³	110.1 (3)
O2—Sn2—O4 ⁴	82.3 (3)	P1—O2—Sn2	135.3 (6)
O4 ⁴ —Sn2—F1 ³	140.1 (3)	P1—O3—Sn1 ⁵	136.4 (6)
O1—P1—O2	110.3 (5)	Sn1 ⁶ —O4—Sn2 ⁷	114.4 (4)
O1—P1—O4	110.0 (5)	P1—O4—Sn1 ⁶	113.6 (5)

Symmetry codes: (1) 1-x, 1/2+y, 3/2-z; (2) -x, 1/2+y, 3/2-z; (3) 1-x, 1-y, 1-z; (4) 1+x, +y, +z; (5) 1-x, -1/2+y, 3/2-z; (6) -x, -1/2+y, 3/2-z; (7) 1+x, +y, +z .

Table S4. Selected Bond lengths (Å) and angles (deg) for Sn₂PO₄Cl.

Sn1—O1 ²	2.218 (6)	Sn2—Cl1	2.871 (3)
Sn1—O2	2.182 (9)	Sn2—Cl1 ⁴	3.256 (4)
Sn1—O4 ³	2.332 (6)	P1—O1	1.556 (7)
Sn2—O1 ²	2.419 (6)	P1—O2	1.504 (8)
Sn2—O3	2.157 (10)	P1—O3	1.518 (8)
Sn2—O4 ³	2.228 (6)	P1—O4	1.551 (10)
Sn1—Cl1 ¹	2.899 (4)	O2—P1—O3	113.8 (6)
O1 ¹ —Sn1—O4 ³	68.8 (2)	O2—P1—O4	107.8 (4)
O2—Sn1—Cl1 ²	81.4 (2)	O3—P1—O1	111.3 (5)
O2—Sn1—O1 ¹	93.2 (4)	O3—P1—O4	106.4 (4)
O2—Sn1—O4 ³	81.1 (3)	O4—P1—O1	107.2 (6)
O4 ³ —Sn1—Cl1 ²	147.9 (2)	Sn1 ⁵ —O1—Sn2 ⁵	107.0 (3)
Cl1—Sn2—Cl1 ⁴	105.88 (8)	P1—O1—Sn1 ⁵	131.6 (3)
O1 ¹ —Sn2—Cl1 ⁴	99.3 (2)	P1—O1—Sn2 ⁵	121.4 (3)
O4 ³ —Sn2—Cl1 ⁴	158.3 (2)	P1—O3—Sn2	137.2 (5)
O4 ³ —Sn2—O1 ¹	67.1 (2)	Sn2 ⁶ —O4—Sn1 ⁶	109.7 (3)
O2—P1—O1	109.9 (4)	P1—O4—Sn1 ⁶	119.7 (3)

Symmetry codes (bond lengths): (1) 1/2-x, -1/2+y, 1/2+z; (2) 1/2-x, -1/2+y, -1/2+z; (3) +x, +y, -1+z; (4) 1-x, 1-y, 1/2+z

Symmetry codes (angles): (1) 1/2-x, -1/2+y, -1/2+z; (2) 1/2-x, -1/2+y, 1/2+z; (3) +x, +y, -1+z; (4) 1-x, 1-y, 1/2+z; (5) 1/2-x, 1/2+y, 1/2+z; (6) +x, +y, 1+z.

Table S5. Crystal data and structure refinement for Sn₂PO₄F and Sn₂PO₄Cl.

Formula	Sn ₂ PO ₄ F	Sn ₂ PO ₄ Cl
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Formula weight	351.35	367.80
Crystal system	Monoclinic	Orthorhombic
Space group	$P2_1/c$	$Pna2_1$
a (Å)	4.6907 (2)	13.5701 (8)
b (Å)	5.5469 (2)	8.8164 (6)
c (Å)	19.9752 (9)	4.7717 (3)
V (Å ³)	519.73 (4)	570.88 (6)
α (°)	90	90
β (°)	89.744 (4)	90
γ (°)	90	90
Z	4	4
$D_{\text{calcd}}(\text{g}\cdot\text{cm}^{-3})$	4.490	4.279
Temperature (K)	296.69 (10)	297.50 (10)
λ (Å)	0.71073	0.71073
$F(000)$	624.0	656.0
μ (mm ⁻¹)	9.860	9.416
GOF on F^2	1.097	1.130
Flack para	/	0.07(4)
R_1, wR_2 ($I > 2\sigma(I)$) ^a	0.0637/0.1760	0.0403/0.1017
R_1, wR_2 (all data)	0.0685/0.1808	0.0460/0.1052

$$^a R_1(F) = \sum ||F_o| - |F_c|| / \sum |F_o|, wR_2(F_o^2) = [\sum w(F_o^2 - F_c^2)^2 / \sum w(F_o^2)^2]^{1/2}$$

(a)



(b)



Fig. S1. Photographs of (a) $\text{Sn}_2\text{PO}_4\text{F}$ and (b) $\text{Sn}_2\text{PO}_4\text{Cl}$.

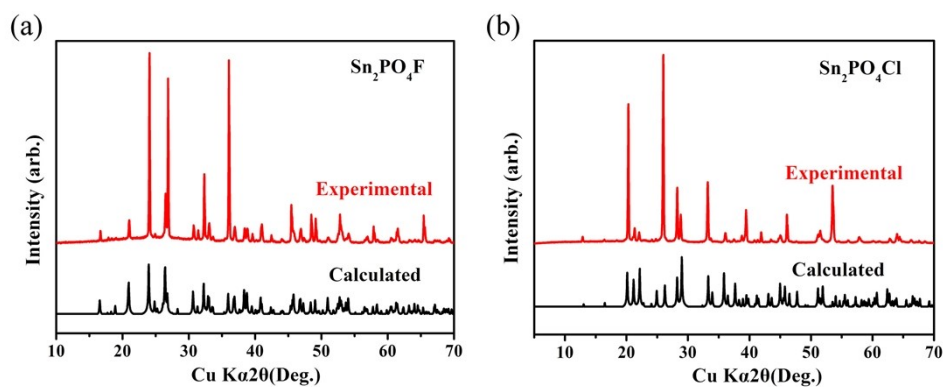


Fig. S2. Experimental and calculated XRD patterns for (a) $\text{Sn}_2\text{PO}_4\text{F}$ and (b) $\text{Sn}_2\text{PO}_4\text{Cl}$.

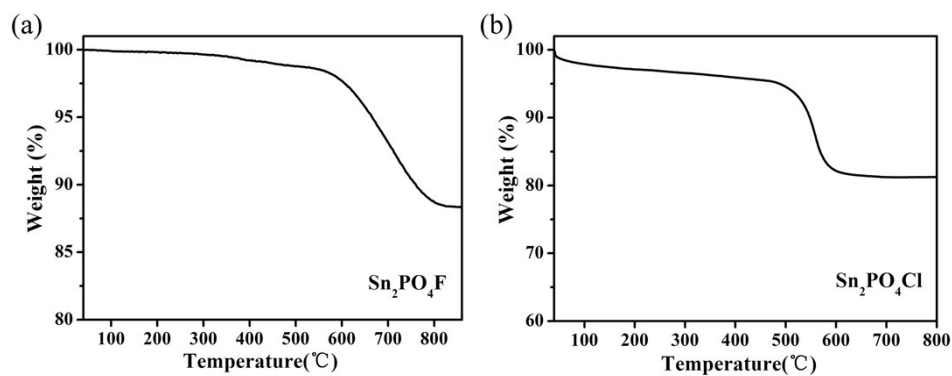


Fig. S3. TGA curves of compounds (a) $\text{Sn}_2\text{PO}_4\text{F}$ and (b) $\text{Sn}_2\text{PO}_4\text{Cl}$ under N_2 air atmosphere.

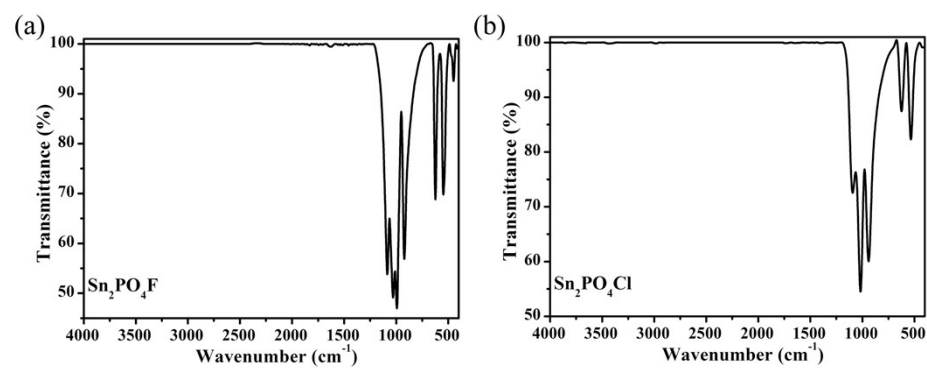


Fig. S4. The IR spectra of compounds (a) $\text{Sn}_2\text{PO}_4\text{F}$ and (b) $\text{Sn}_2\text{PO}_4\text{Cl}$.

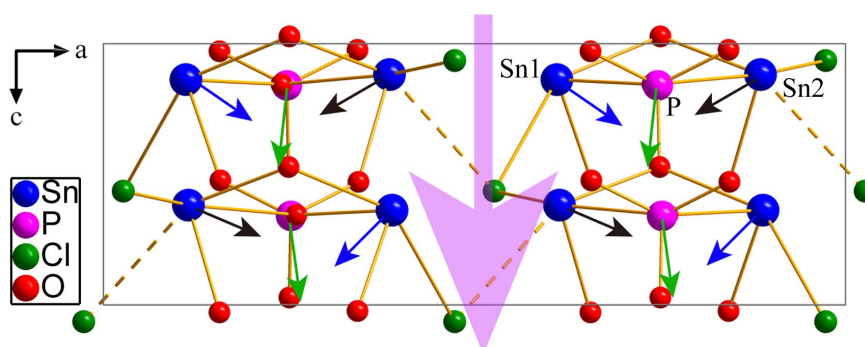


Fig. S5. Projection of the arrangement and dipole orientation of $[\text{SnOCl}]$ polyhedra and PO_4 units in the ac -plane for $\text{Sn}_2\text{PO}_4\text{Cl}$. The arrows in black, blue and green represent the local moments and the large bright purple arrow represents the net macroscopic polarization direction.

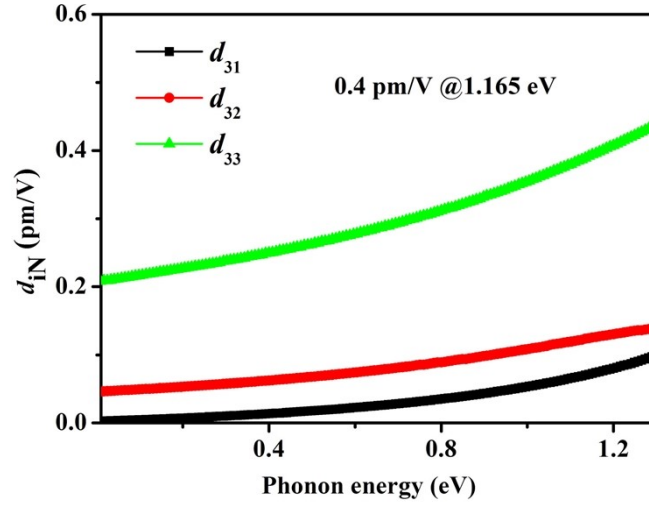


Fig. S6. The calculated frequency dependent second harmonic generation coefficients for $\text{Sn}_2\text{PO}_4\text{Cl}$.

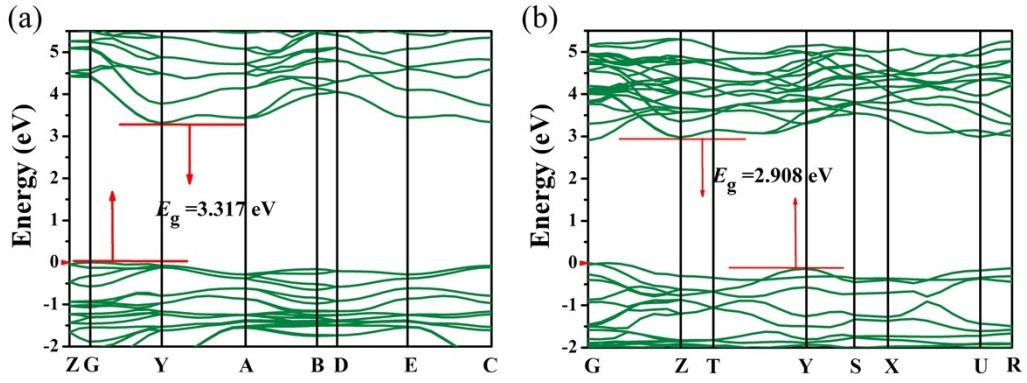


Fig. S7. Calculated band structure for (a) $\text{Sn}_2\text{PO}_4\text{F}$ and (b) $\text{Sn}_2\text{PO}_4\text{Cl}$ (the Fermi level is set at 0 eV).