

Supporting Information for

Ethane Oxidative Dehydrogenation with CO₂ on thiogallates

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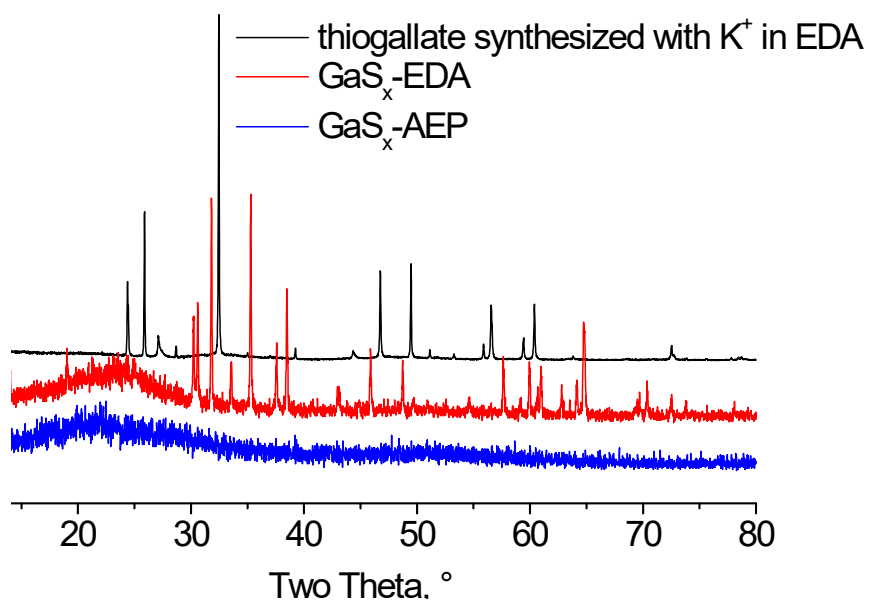


Fig. S1. PXRD patterns of the obtained phases in presence or absence of alkali cations with organic templates (EDA, AEP).

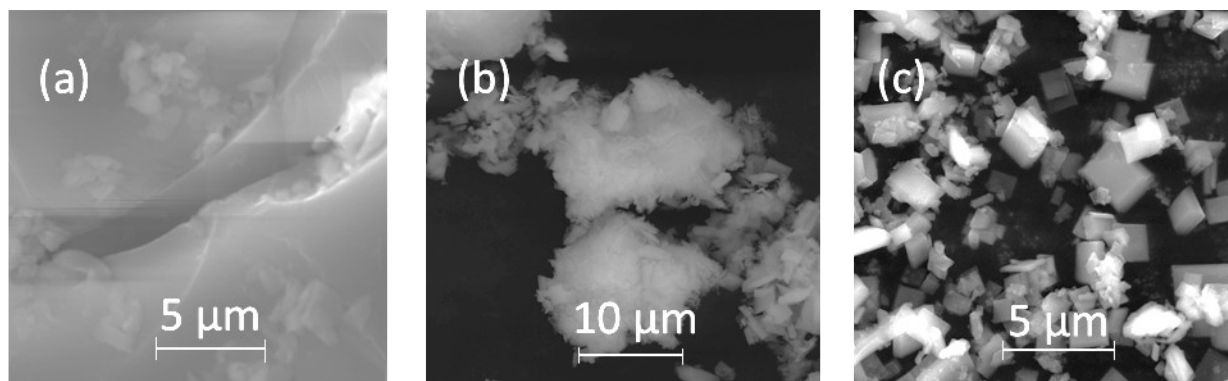


Fig. S2. SEM micrographs of the solids obtained at 190 °C for 7d from (a) Ga-S-AEP, (b) Ga-S-EDA systems, (c) Ga-S-EDA with K^+ .

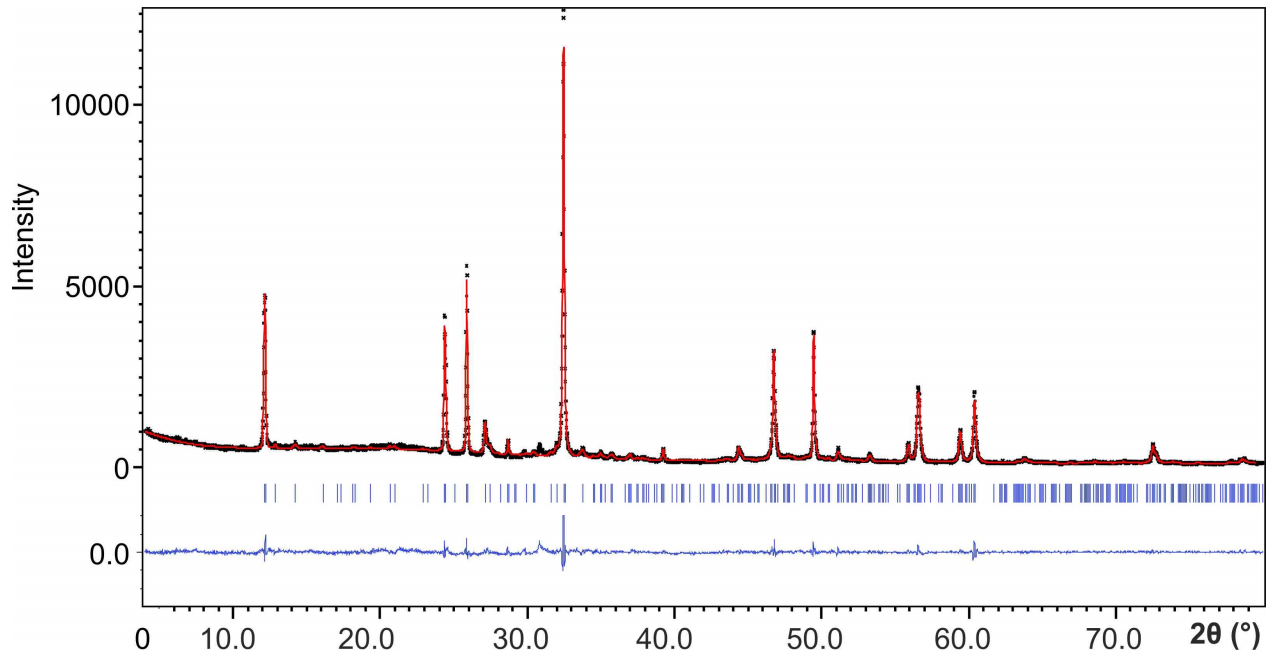


Fig. S3. PXRD patterns of as-prepared KGaS₂ with EDA attesting of the sample purity. Cross symbol is the observed signal, red line is the calculated signal, blue segments show the position of the expected reflections, blue line the differences between observed and calculated intensities.

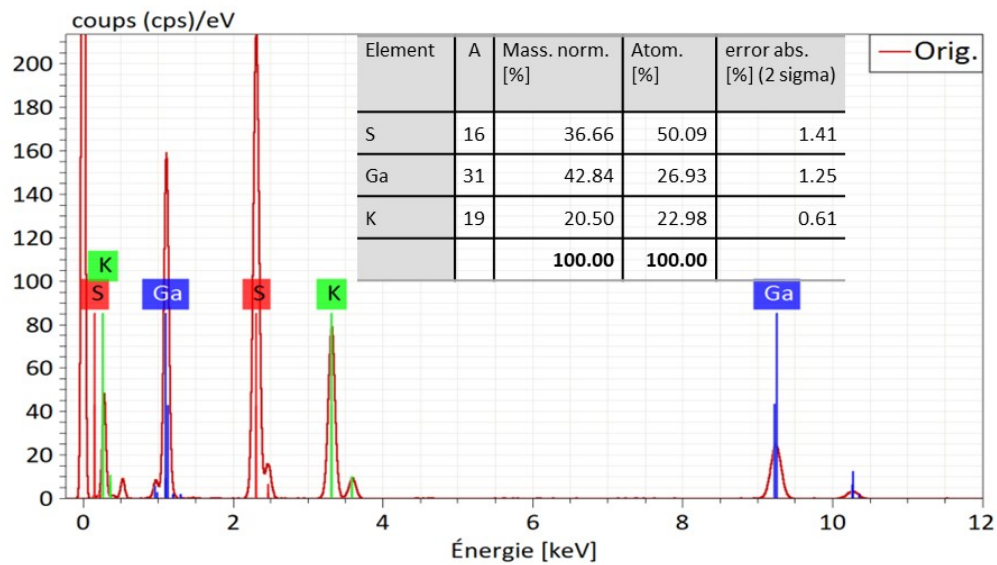


Fig. S4. EDS KGaS₂ and elemental quantitative data of the KGaS₂.

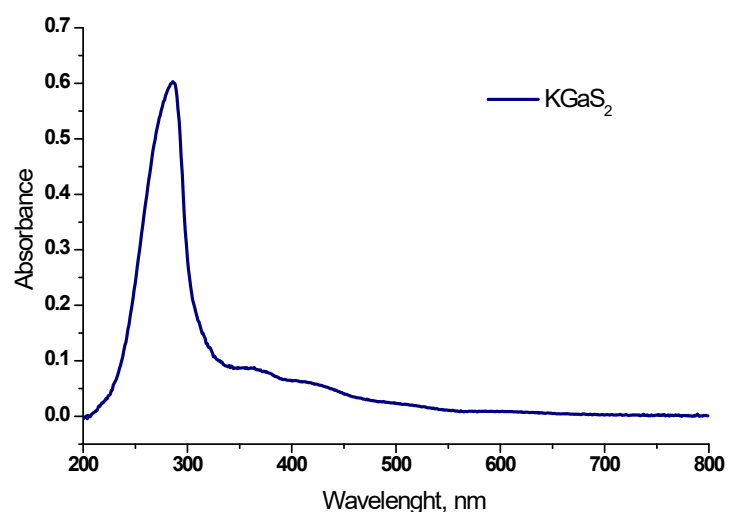


Fig. S5. UV-Vis absorption spectra of the EDA synthesized.

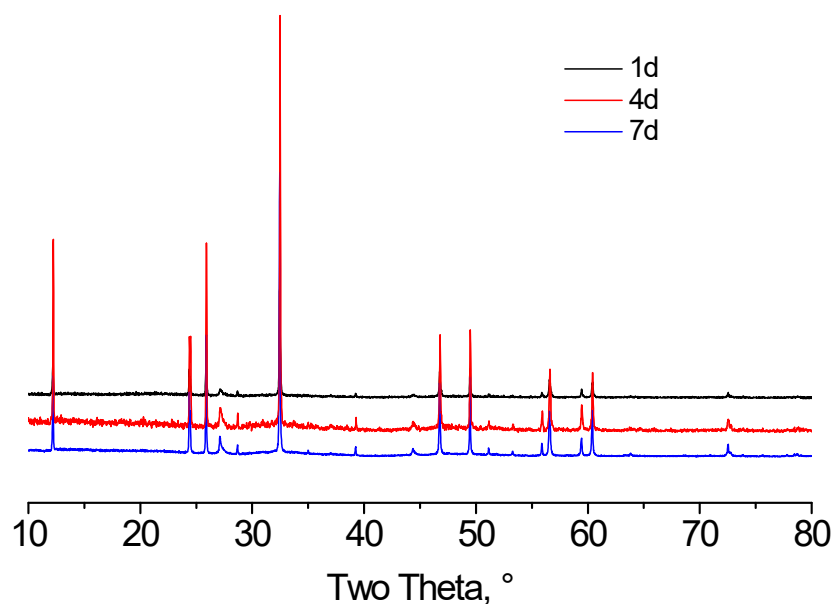


Fig. S6. PXRD patterns of as-prepared KGaS₂ with various crystallization time (1-7 days).

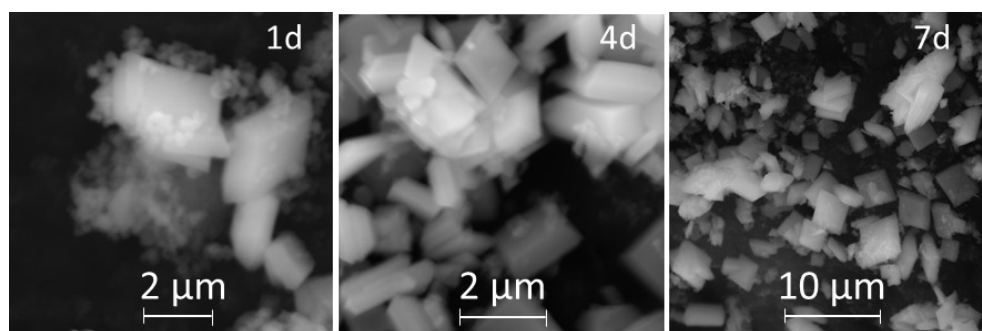


Fig. S7. SEM micrographs of KGaS₂ with ethylenediamine at 190 °C were prepared in the period of 1-7d.

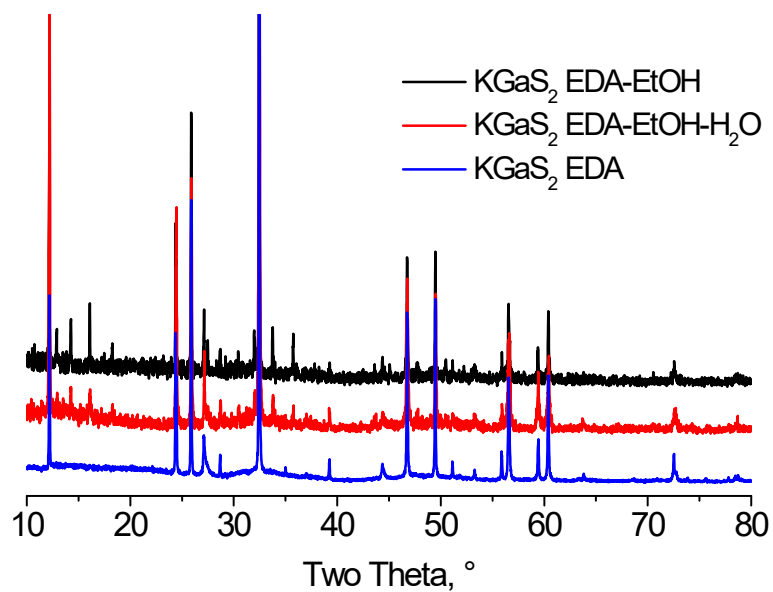


Fig. S8. PXRD patterns of KGaS_2 prepared with different solvents at 190°C for 7 days.

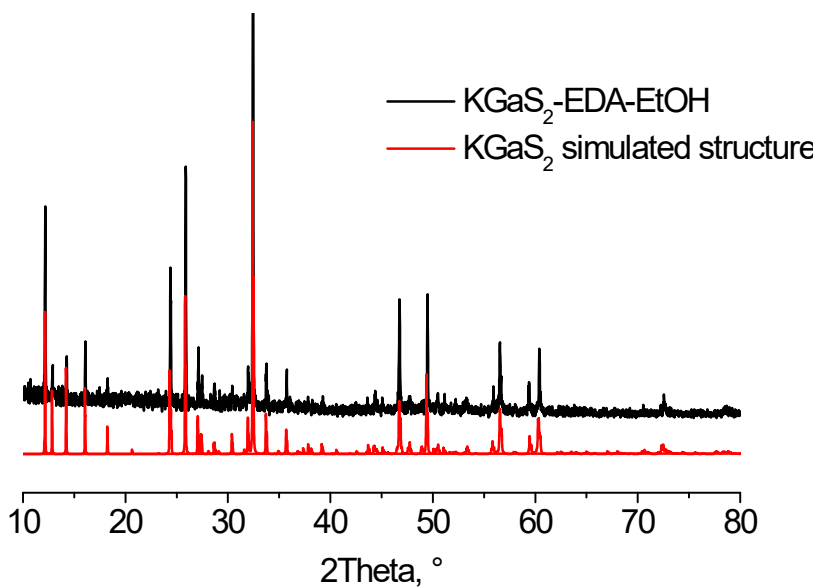


Fig. S9. PXRD patterns of as-prepared KGaS_2 with EDA-EtOH and simulated structure.

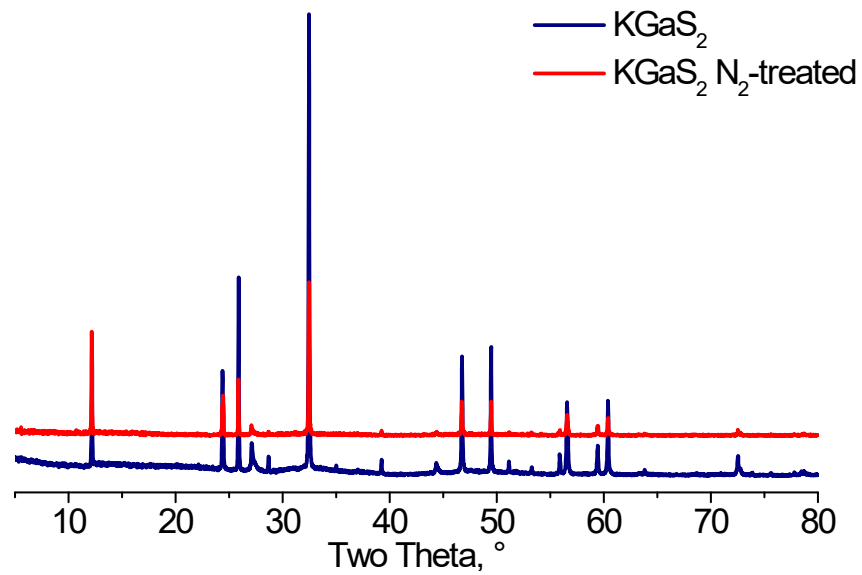


Fig. S10. PXRD patterns of KGaS₂ as-prepared and N₂-treated sample at 800 °C.

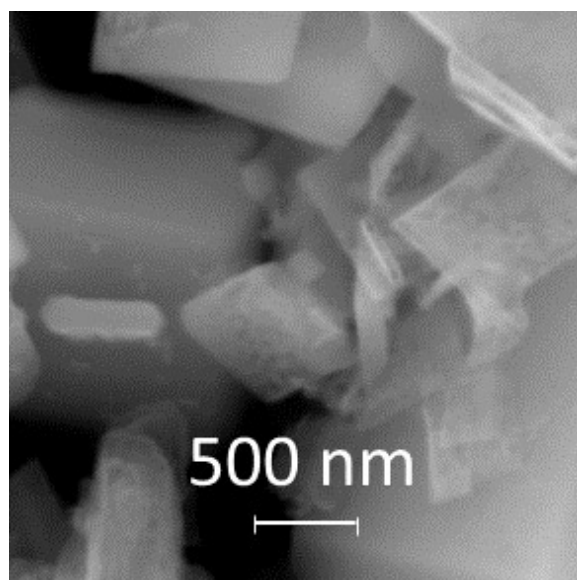


Fig. S11. SEM images of KGaS₂ N₂-treated at 800 °C.

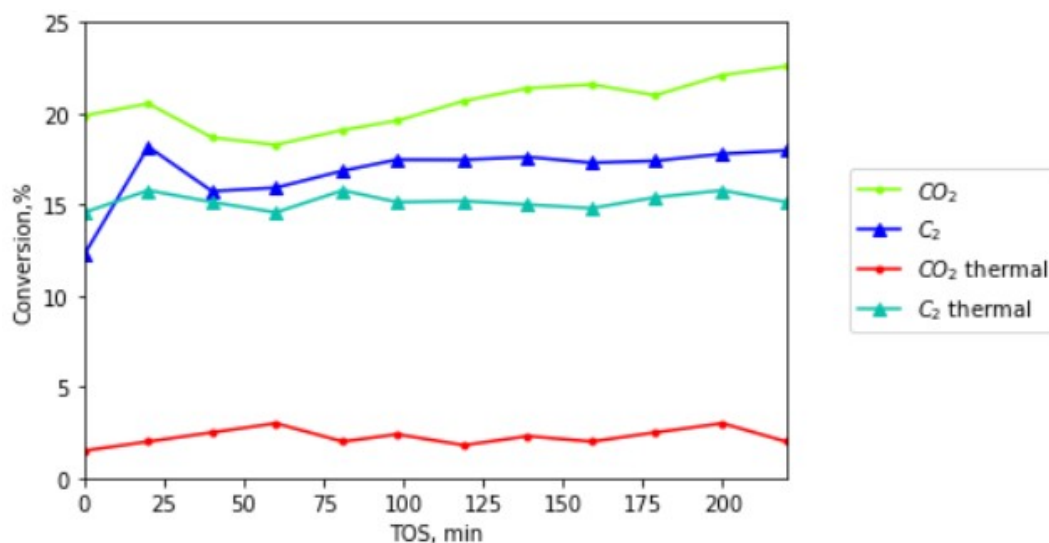


Fig. S12. C₂H₆ and CO₂ conversion on KGaS₂ based catalysts (P = 1 atm, C₂H₆/CO₂ = 1/0.74, WHSV = 5.9 h⁻¹, 700 °C).

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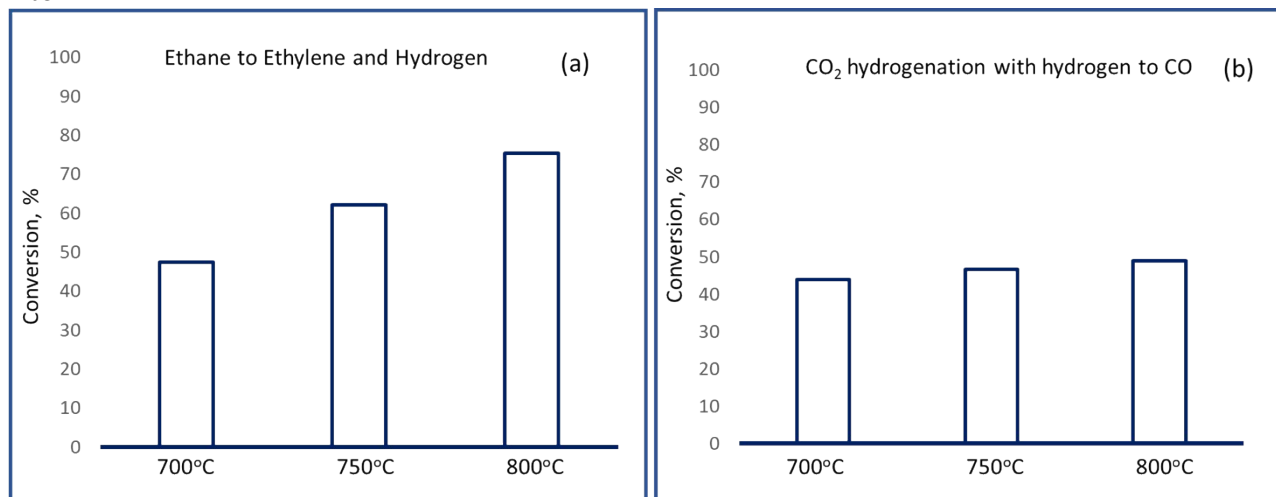


Fig. S13. Thermodynamic equilibrium conversions at different temperature, assuming isothermal temperature profile for a) ethane conversion to ethylene and hydrogen at atmospheric pressure b) CO₂ hydrogenation with hydrogen (CO₂/H₂ -1) to CO at atmospheric pressure (calculated with AspenHysys v12)

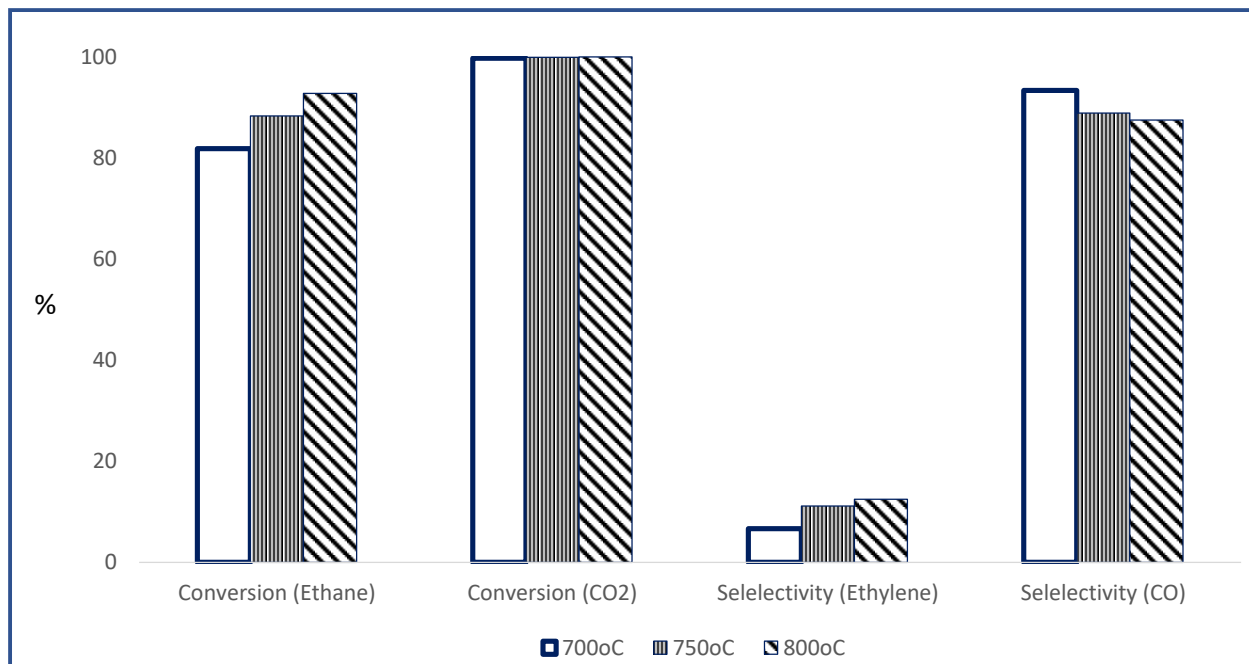


Fig. S14. Equilibrium product distribution at different temperature, assuming isothermal temperature profile in the system (CO, CO₂, ethane, ethylene, hydrogen) containing initially the equal molar composition of CO₂ and ethane at atmospheric pressure (calculated with AspenHysys v12)