

## Electronic Supplementary Information (ESI) for

### **Water-tuned reversible spin transition with the largest hysteresis loop in 3D Hofmann frameworks pillared by flexible ligands**

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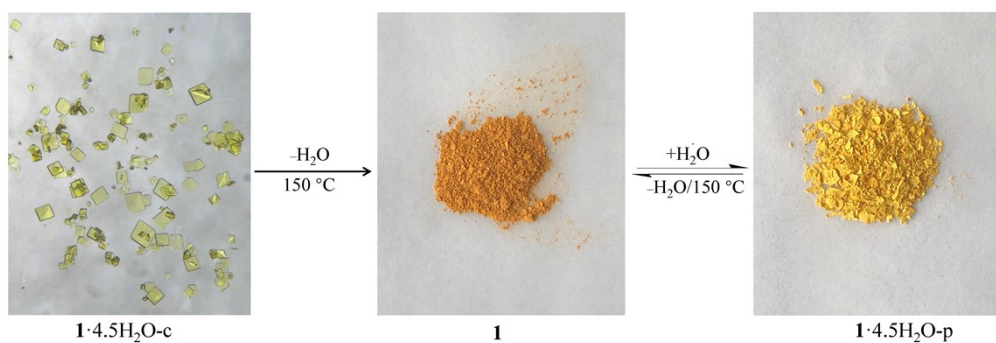
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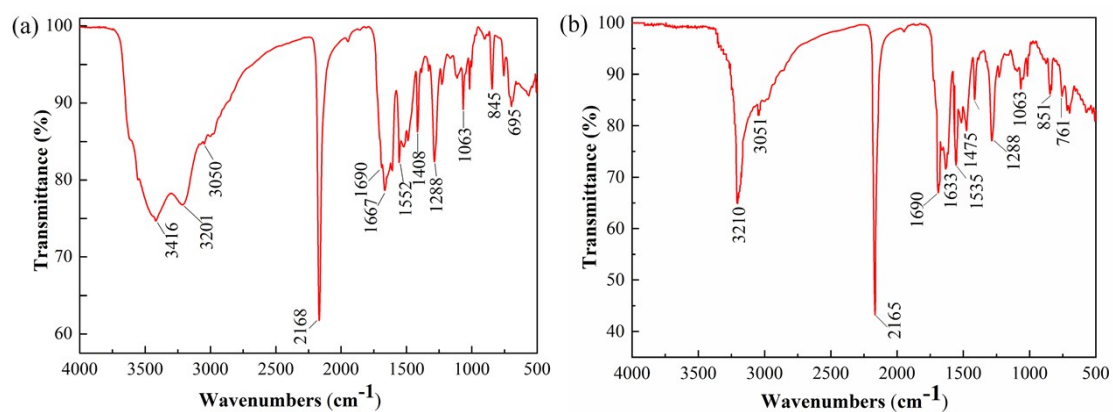
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## 1. The reversible transformation between $1 \cdot 4.5\text{H}_2\text{O-p}$ and **1**

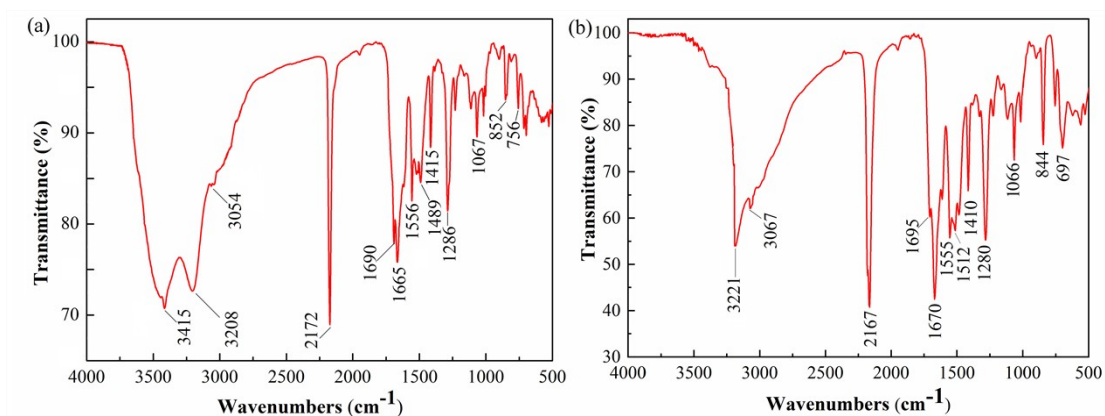


**Fig. S1** The reversible transformation between the  $1 \cdot 4.5\text{H}_2\text{O-p}$  and **1**.

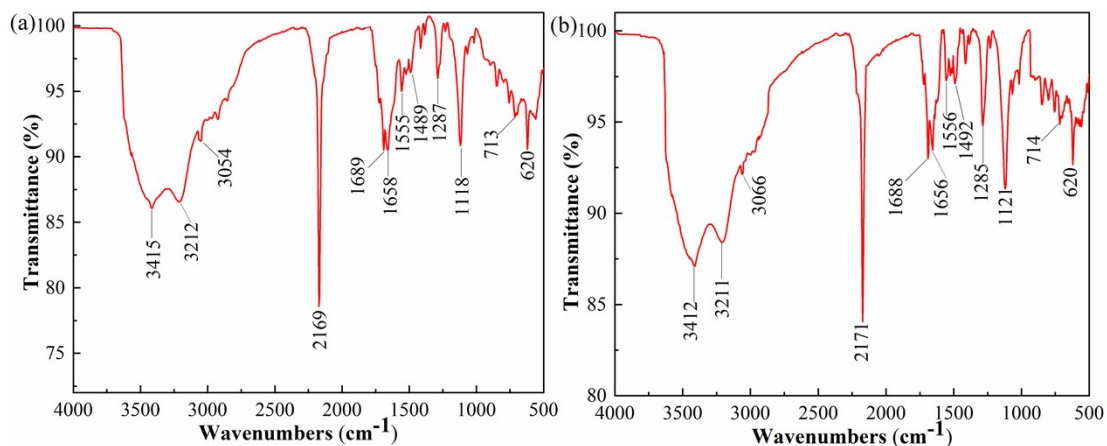
## 2. The FT-IR spectra of $1 \cdot 4.5\text{H}_2\text{O-c}$ , **1**, $2 \cdot 4.5\text{H}_2\text{O-c}$ , **2**, rehydrated $1 \cdot 4.5\text{H}_2\text{O-p}$ and $2 \cdot 4.5\text{H}_2\text{O-p}$



**Fig. S2** FT-IR spectra of (a)  $1 \cdot 4.5\text{H}_2\text{O-c}$  and (b) **1**.

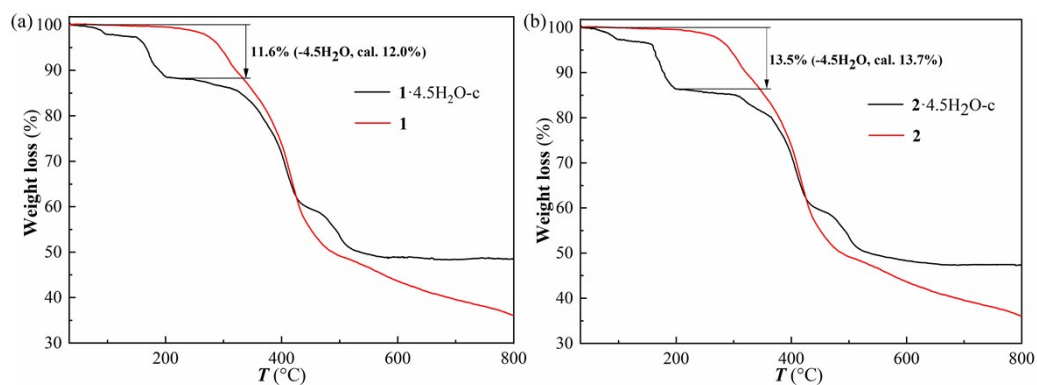


**Fig. S3** FT-IR spectra of (a)  $2 \cdot 4.5\text{H}_2\text{O-c}$  and (b) **2**.

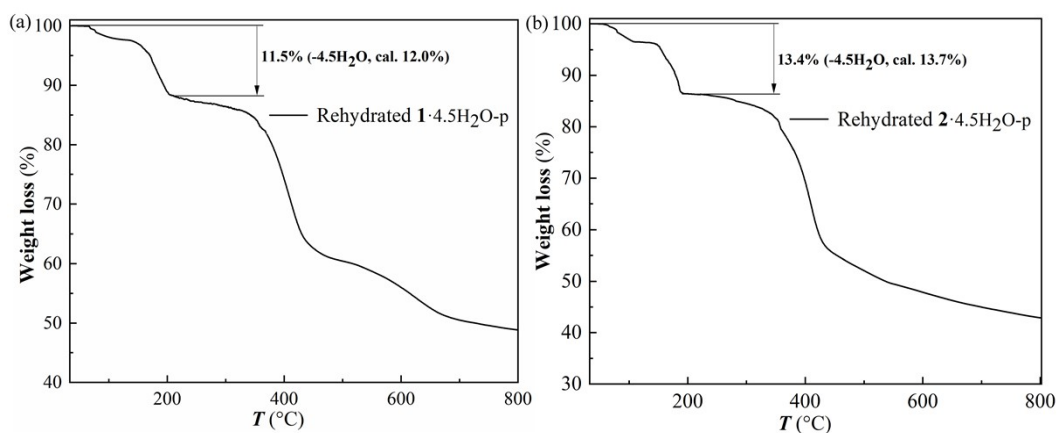


**Fig. S4** FT-IR spectra of the rehydrated  $1 \cdot 4.5\text{H}_2\text{O-p}$  (a) and  $2 \cdot 4.5\text{H}_2\text{O-p}$  (b).

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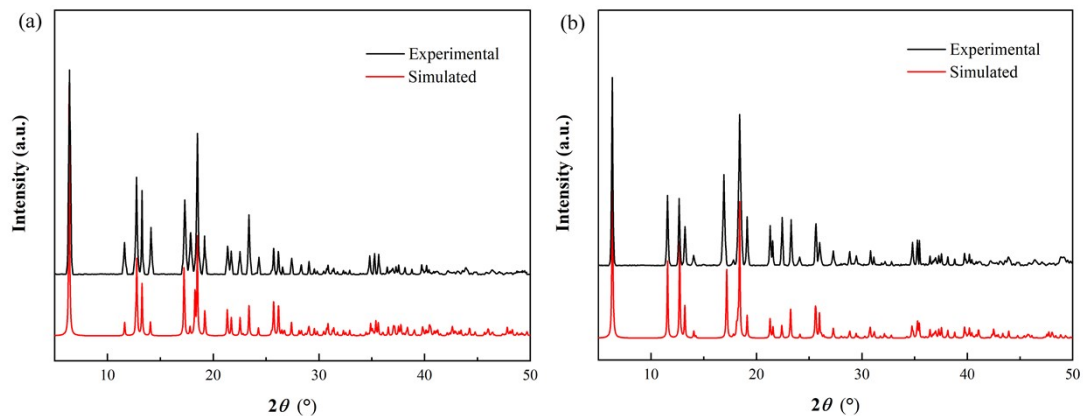


**Fig. S5** TGA curves of (a)  $1 \cdot 4.5\text{H}_2\text{O-c}$  and **1** and (b)  $2 \cdot 4.5\text{H}_2\text{O-c}$  and **2**.

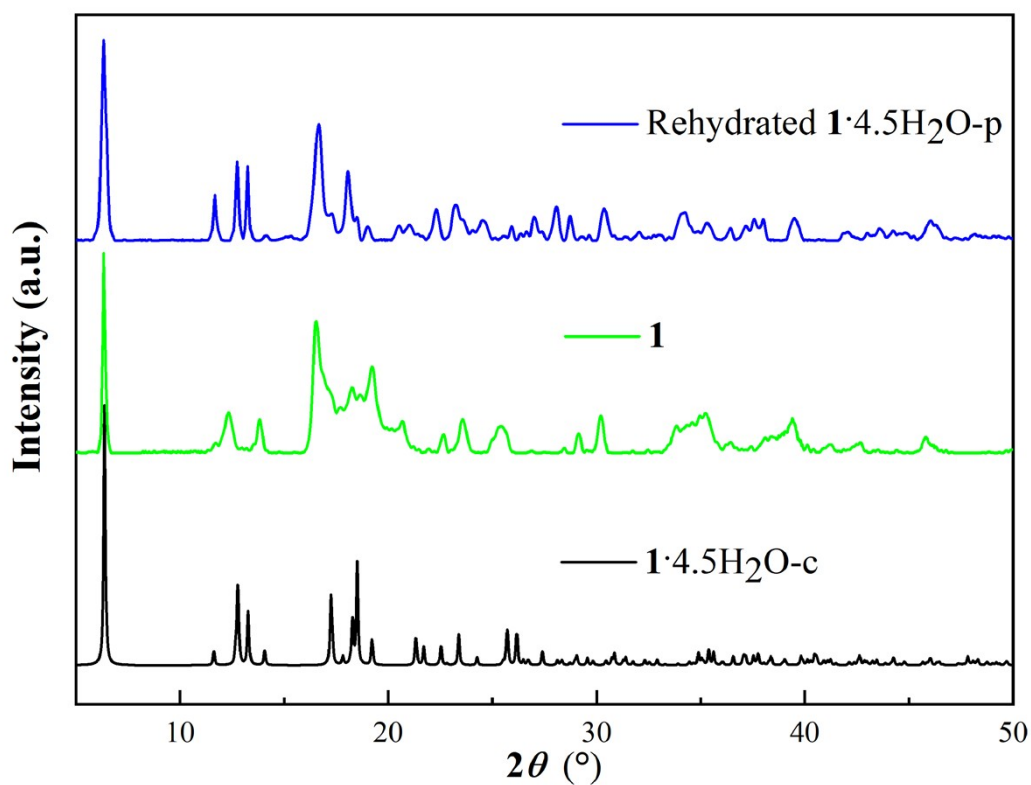


**Fig. S6** TGA curves of rehydrated  $1 \cdot 4.5\text{H}_2\text{O-p}$  (a) and  $2 \cdot 4.5\text{H}_2\text{O-p}$  (b).

**4. The PXRD patterns of  $1 \cdot 4.5\text{H}_2\text{O-c}$ ,  $2 \cdot 4.5\text{H}_2\text{O-c}$ , **1**, **2**, rehydrated  $1 \cdot 4.5\text{H}_2\text{O-p}$  and  $2 \cdot 4.5\text{H}_2\text{O-p}$**



**Fig. S7** PXRD patterns of (a)  $1 \cdot 4.5\text{H}_2\text{O-c}$  and (b)  $2 \cdot 4.5\text{H}_2\text{O-c}$ .



**Fig. S8** PXRD patterns of  $1 \cdot 4.5\text{H}_2\text{O-c}$ , **1** and rehydrated  $1 \cdot 4.5\text{H}_2\text{O-p}$ .

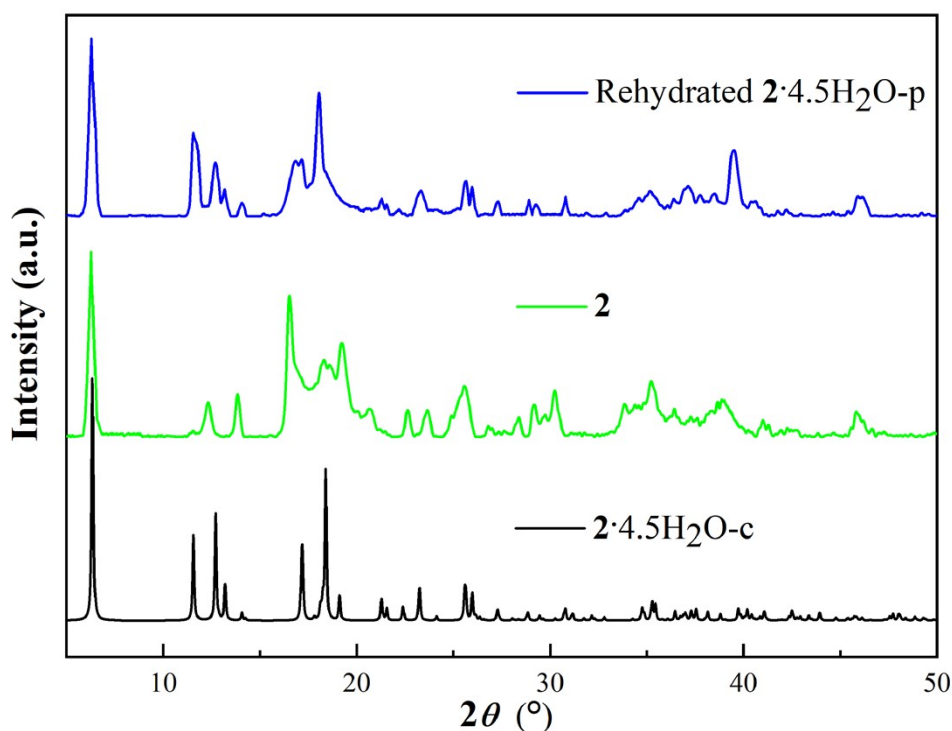


Fig. S9 PXRD patterns of  $2 \cdot 4.5\text{H}_2\text{O-c}$ , **2** and rehydrated  $2 \cdot 4.5\text{H}_2\text{O-p}$ .

### 5. Molecular structures of $1 \cdot 4.5\text{H}_2\text{O-c}$ and $2 \cdot 4.5\text{H}_2\text{O-c}$ at 296 and 100 K

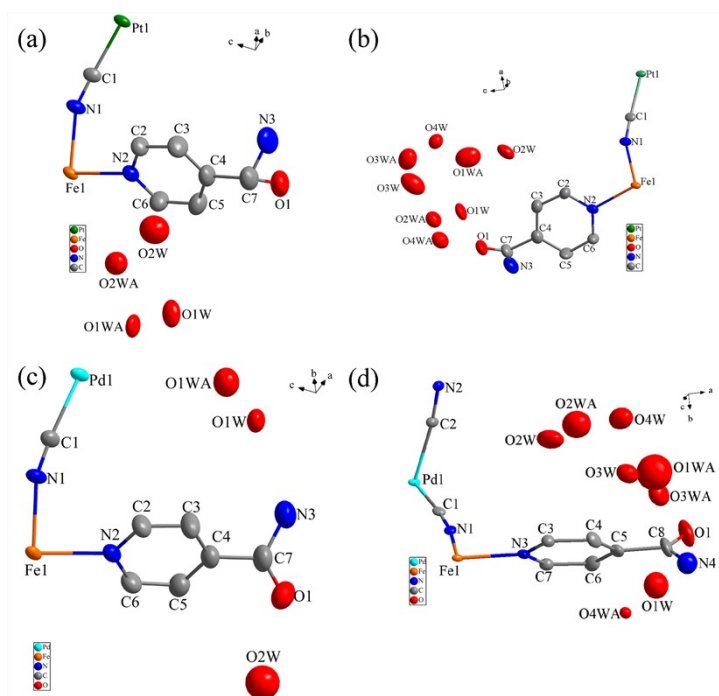
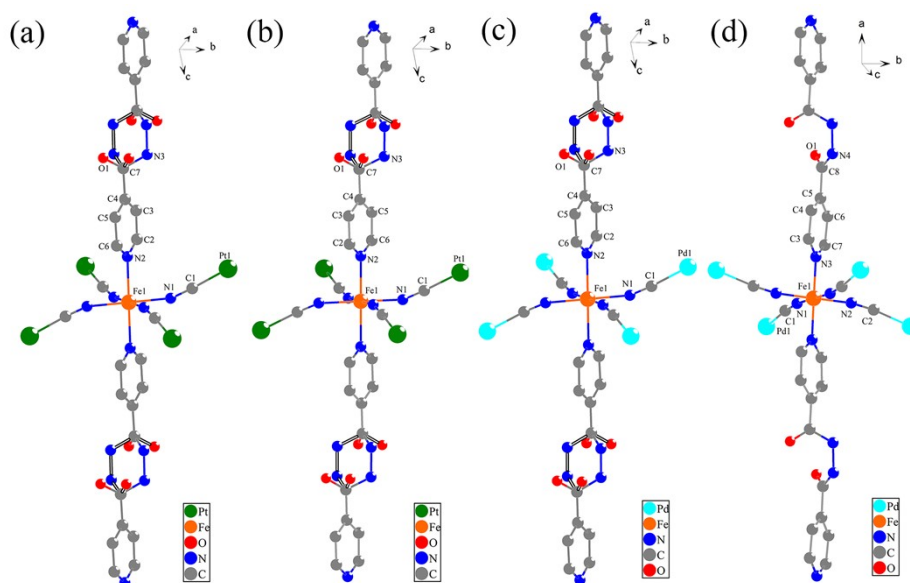
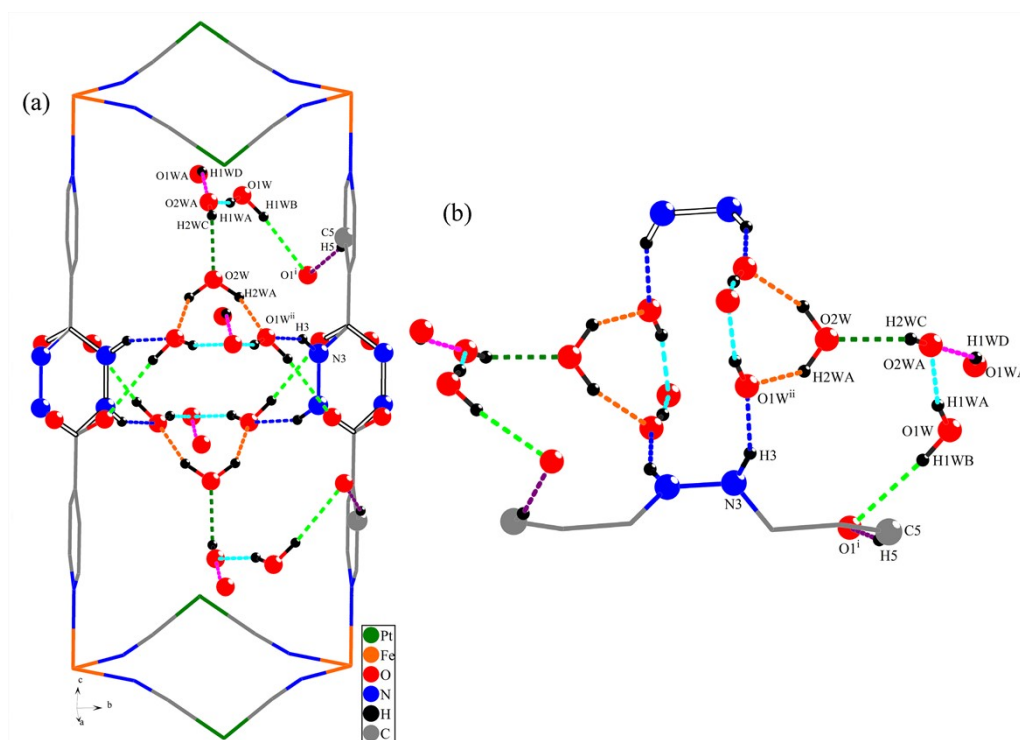


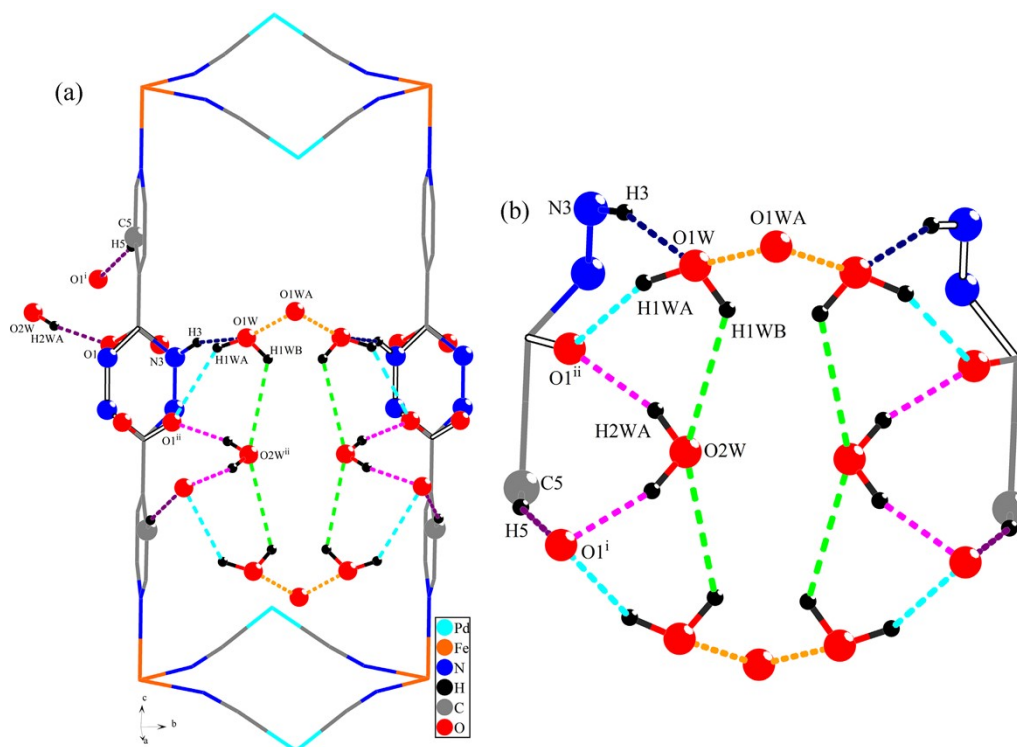
Fig. S10 The asymmetric unit of  $1 \cdot 4.5\text{H}_2\text{O-c}$  at 296 K (a) and 100 K (b), and  $2 \cdot 4.5\text{H}_2\text{O-c}$  at 296 K (c) and 100 K (d) with 50% thermal ellipsoids probability, all hydrogen atoms are omitted for clarity.



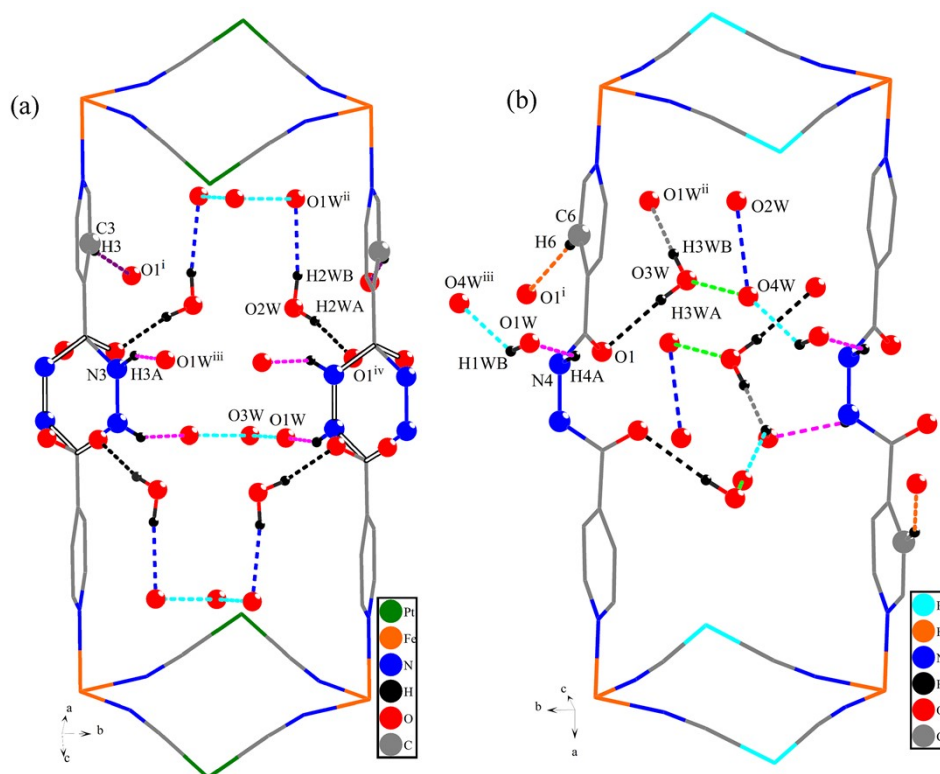
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**Fig. S12** (a) The hydrogen bond interactions in  $1 \cdot 4.5\text{H}_2\text{O-c}$  at 296 K and (b) the hydrogen-bonding networks formed in the 1D channels of  $1 \cdot 4.5\text{H}_2\text{O-c}$  (the disorder atoms are linked by the hollow lines).



**Fig. S13** (a) The hydrogen bond interactions in  $2 \cdot 4.5\text{H}_2\text{O-c}$  at 296 K and (b) the hydrogen-bonding networks formed in the 1D channels of  $2 \cdot 4.5\text{H}_2\text{O-c}$  (the disorder atoms are linked by the hollow lines).



**Fig. S14** The hydrogen bond interactions in (a)  $1 \cdot 4.5\text{H}_2\text{O-c}$  and (b)  $2 \cdot 4.5\text{H}_2\text{O-c}$  at 100 K (the disorder atoms are linked by the hollow lines).

## 6. Selected bond distances and angles and hydrogen bond interactions for 1·4.5H<sub>2</sub>O-c and 2·4.5H<sub>2</sub>O-c at 296 and 100 K

**Table S1** Selected bond distances (Å) and angles (°) for 1·4.5H<sub>2</sub>O-c and 2·4.5H<sub>2</sub>O-c at 296 and 100 K

	1·4.5H <sub>2</sub> O-c				2·4.5H <sub>2</sub> O-c				
	296 K		100 K		296 K		100 K		
Fe1-N1	2.139(4)	Fe1-N1	2.140(4)	Fe1-N1	2.148(3)	Fe1-N1	1.941(4)	Fe1-N2 <sup>ii</sup>	1.938(4)
Fe1-N2	2.210(6)	Fe1-N2	2.215(5)	Fe1-N2	2.210(4)	Fe1-N3	1.989(4)	Pd1-C1/C2	1.990(5)
Pt1-C1	1.981(5)	Pt1-C1	1.989(5)	Pd1-C1	2.001(4)	C1-N1	1.158(7)	C1-N1	1.158(7)
C1-N1	1.150(7)	C1-N1	1.144(6)	C1-N1	1.136(5)	C2-N2	1.160(7)	N3-C7	1.350(7)
N2-C2	1.330(10)	N2-C2	1.324(9)	N2-C2	1.311(7)	C8-N4	1.355(8)	C8-O1	1.206(8)
N2-C6	1.317(11)	N2-C6	1.333(10)	N2-C6	1.347(7)	Fe···Fe <sup>iii</sup>	10.082(2)	Fe···Fe <sup>iv</sup>	15.154(2)
C7-N3	1.362(15)	C7-N3	1.403(14)	C7-N3	1.384(11)				
C7-O1	1.177(13)	C7-O1	1.190(12)	C7-O1	1.196(9)				
Fe···Fe <sup>iii</sup>	10.318(2)	Fe···Fe <sup>iii</sup>	10.315(2)	Fe···Fe <sup>iii</sup>	10.361(2)				
Fe···Fe <sup>iv</sup>	15.657(2)	Fe···Fe <sup>iv</sup>	15.679(2)	Fe···Fe <sup>iv</sup>	15.682(2)				
N1-Fe1-N2	88.27(16)	N1-Fe1-N2	91.58(15)	N1-Fe1-N2	87.92(12)	N1-Fe1-N3	92.54(16)		
N1-Fe1-N2 <sup>i</sup>	91.74(16)	N1-Fe1-N2 <sup>i</sup>	88.42(15)	N1-Fe1-N2 <sup>i</sup>	92.08(12)	N1-Fe1-N3 <sup>i</sup>	88.29(16)		
N1-Fe1-N1 <sup>ii</sup>	89.65(2)	N1-Fe1-N1 <sup>ii</sup>	90.85(4)	N1-Fe1-N1 <sup>ii</sup>	88.98(17)	N1-Fe1-N1 <sup>i</sup>	88.3(2)		
N1-Fe1-N1 <sup>i</sup>	90.35(2)	N1-Fe1-N1 <sup>i</sup>	89.1(2)	N1-Fe1-N1 <sup>i</sup>	91.02(17)	N1-Fe1-N2 <sup>ii</sup>	91.93(18)		
C1-N1-Fe1	157.78(4)	C1-N1-Fe1	157.5(4)	C1-N1-Fe1	159.42(3)	C1-N1-Fe1	168.1(4)		
C2-N2-Fe1	122.89(5)	C2-N2-Fe1	119.0(5)	C2-N2-Fe1	123.96(4)	C3-N3-Fe1	121.3(3)		
C6-N2-Fe1	119.37(5)	C6-N2-Fe1	122.5(5)	C6-N2-Fe1	119.46(4)	C7-N3-Fe1	122.1(3)		
C2-N2-C6	117.74(6)	C2-N2-C6	118.5(6)	C2-N2-C6	116.58(5)	N1-C1-Pd1	174.8(4)		
N1-C1-Pt1	176.07(4)	N1-C1-Pt1	176.6(4)	N1-C1-Pd1	175.32(3)	N2-C2-Pd1	173.4(4)		
C4-C7-N3	112.97(8)	C4-C7-N3	113.4(7)	C4-C7-N3	116.02(6)	C5-C8-N4	112.6(6)		
C4-C7-O1	121.61(8)	C4-C7-O1	121.6(7)	C4-C7-O1	121.41(7)	C5-C8-O1	124.2(6)		

Symmetry codes: 1·4.5H<sub>2</sub>O-c at 296 K: i) 1-x, y, 1-z; ii) x, 1-y, z; iii) 1+x, 1+y, z; iv) 1+x, y, z-1; 100 K: i) 1-x, y, -z; ii) x, -y, z; iii) 1+x, 1+y, z; iv) x-1, y 1+z; 2·4.5H<sub>2</sub>O-c at 296 K: i) -x, y, 1-z; ii) x, 1-y, z; iii) 1+x, 1+y, z; iv) 1+x, y, z-1; 100 K: i) 1-x, y, 1/2-z; ii) x, 1+y, z; iii) 1+x, y, z; iv) 1-x, 2-y, -z.

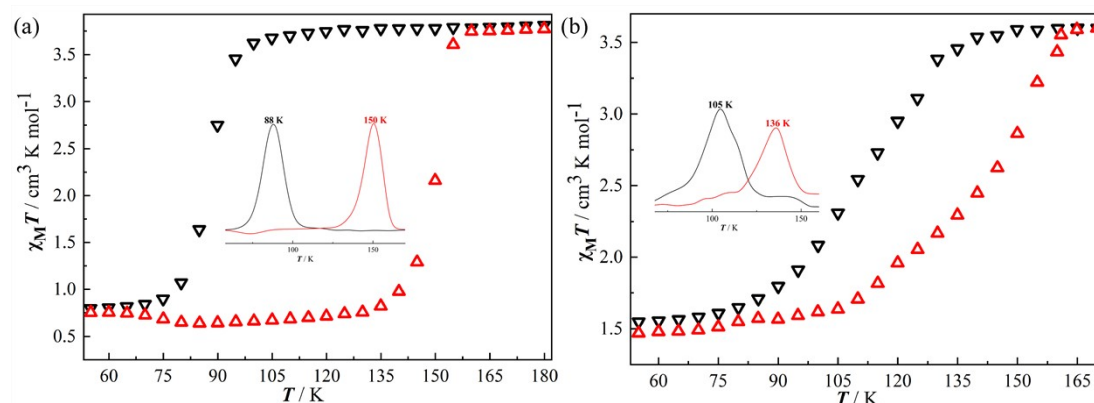
**Table S2** Hydrogen-bond geometry (Å, °) for 1·4.5H<sub>2</sub>O-c and 2·4.5H<sub>2</sub>O-c at 296 and 100 K

D-H···A	d(D-H)	d(H···A)	d(D···A)	∠D-H···A
1·4.5H <sub>2</sub> O-c (296 K)				
C5-H5···O1 <sup>i</sup>	0.93	2.45	3.351(7)	164
N3-H3···O1W <sup>ii</sup>	0.90	1.91	2.757(7)	156
O1W-H1WA···O2WA	0.85	2.54	3.357(4)	163
O1W-H1WB···O1 <sup>i</sup>	0.85	2.15	2.987(5)	169
O1WA-H1WD···O2WA	0.85	1.84	2.603(9)	148
O2W-H2WA···O1W <sup>ii</sup>	0.85	1.99	2.629(6)	131
1·4.5H <sub>2</sub> O-c (100 K)				
C3-H3···O1 <sup>i</sup>	0.93	2.46	3.358(6)	164
N3-H3A···O1W <sup>iii</sup>	0.90	1.94	2.772(2)	153
O2W-H2WA···O1 <sup>iv</sup>	0.85	1.83	2.669(2)	171
O2W-H2WB···O1W <sup>ii</sup>	0.85	2.01	2.847(2)	171
O1W···O3W			2.695(3)	
2·4.5H <sub>2</sub> O-c (296 K)				
C5-H5···O1 <sup>i</sup>	0.93	2.45	3.357(13)	164
N3-H3···O1W	0.90	1.92	2.777(6)	158
O1W-H1WA···O1 <sup>ii</sup>	0.85	2.43	3.017(4)	127
O1W-H1WB···O2W <sup>ii</sup>	0.85	2.59	3.151(5)	124
O2W-H2WA···O1	0.85	1.70	2.52(5)	162
2·4.5H <sub>2</sub> O-c (100 K)				
N4-H4···O1W	0.89	2.18	2.784(7)	124
O3W-H3WB···O1W <sup>ii</sup>	0.85	1.71	2.562(6)	177
O3W-H3WA···O1	0.85	2.08	2.926(6)	176
C6-H6···O1 <sup>i</sup>	0.95	2.54	3.447(3)	161
O1W-H1WB···O4W <sup>iii</sup>	0.85	1.95	2.529(6)	125



Symmetry codes: **1**·4.5H<sub>2</sub>O-c at 296 K: i) 1-x, 1-y, -z; ii) 1+x, y, z; at 100 K: i) 1-x, -y, 1-z; ii) 1-x, y, 1-z; iii) -x, 1-y, 1-z; iv) x, 1+y, z. **2**·4.5H<sub>2</sub>O-c at 296 K: i) -x, y, -z; ii) 1-x, 1-y, -z; at 100 K: i) 2-x, 1-y, 1-z; ii) x, 1-y, 0.5+z; iii) x, 1+y, z.

## 7. Second magnetic measurements for the hysteretic SCO behaviors of 1·4.5H<sub>2</sub>O-c and 2·4.5H<sub>2</sub>O-c



**Fig. S15** Variable temperature magnetic susceptibility ( $\chi_M T$ ) for (a) 1·4.5H<sub>2</sub>O-c and (b) 2·4.5H<sub>2</sub>O-c measured again in the range of hysteresis loop (black: cooling; red: heating; inset:  $\partial(\chi_M T)/\partial T$  showing the  $T_{1/2}$  values).

## 8. Comparison of the spin transition parameters of some representative 3D Hofmann SCO frameworks

**Table S3** Critical temperature ( $T_{1/2}$ ) and hysteresis width ( $\Delta T$ ) of some representative 3D Hofmann SCO frameworks [FeLM(CN)<sub>4</sub>]·G and [FeL{M'(CN)<sub>2</sub>}<sub>2</sub>]·G with the rigid pillars except bph (M = Ni, Pt, Pd; M' = Ag, Au; L = pillar ligand; G = guest molecule)

L	M	G	$T_{1/2}^{\downarrow}$ (K)	$T_{1/2}^{\uparrow}$ (K)	$\Delta T$ (K)	Ref.
bpe	Ag		120	215	95	1
dpb	Au	0.7naphthalene	141	214	73	2
pz	Pt	0.5thiourea	213	277	64	3
bph	Pt	4.5H <sub>2</sub> O	90(88)	150	60(62)*	this work
bpac	Pt	0.5bpac	251	300	49	4
bpy	Ni	<i>x</i> (CD <sub>3</sub> ) <sub>2</sub> CO	103	148	45	5
pz	Pd	2.5H <sub>2</sub> O	233	266	33	6
bpac	Pd	0.5bpac	283	315	32	4
bph	Pd	4.5H <sub>2</sub> O	105	135(136)	30(31)*	this work
bpb	Pt	nitrobenzene	210	237	27	7
pz	Ni	2H <sub>2</sub> O	280	305	25	6
2,5-bpp	Au	<i>s</i> BuOH	186/171	209/189	23/18	8
azpy	Pd		181	202	21	9
bpac	Pt	H <sub>2</sub> O·0.5bpac	301	322	21	4
bpd	Au		158/128	179/149	21/21	10
pz	Pt	2H <sub>2</sub> O	220	240	20	6
bpan	Au		242/143	252/163	10/20	11
4-abpt	Ag	<i>x</i> EtOH	264	281	17	12

azpy	Pt		175	190	15	9
dpe	Pt	0.5dpe	135	150	15	13
bpn	Ag	azobenzene	182/171/132/ 118	184/177/147/ 128	2/6/15/ 10	14
azpy	Pt	H <sub>2</sub> O	275	285	10	9
azpy	Pd	H <sub>2</sub> O	287	296	9	9
dpoda	Ag	1.5naphthalene	250/228/190/ 181	252/232/194/ 187	2/4/4/6	15
dpt	Pt	1.5H <sub>2</sub> O·dpt	210/127	212/132	2/5	16
Hbpt	Pt	0.5Hbpt·0.5MeOH·2.5H <sub>2</sub> O	244/158/124	249/158/124	5/0/0	17
bipytz	Au		273	277	4	18
bpac	Pd	H <sub>2</sub> O·0.5bpac	307	310	3	4
dpe	Pt	H <sub>2</sub> O·0.5dpe	275/243	275/243	0/0	13
dpni	Ag	4CH <sub>3</sub> CN	196/160	196/160	0/0	19

\* The data were obtained from the second magnetic measurements.

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