

Supplementary Information

Anchoring polysulfides via CoS₂/NC@1T MoS₂ modified separator for high-performance lithium-sulfur batteries

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Supplementary Figures

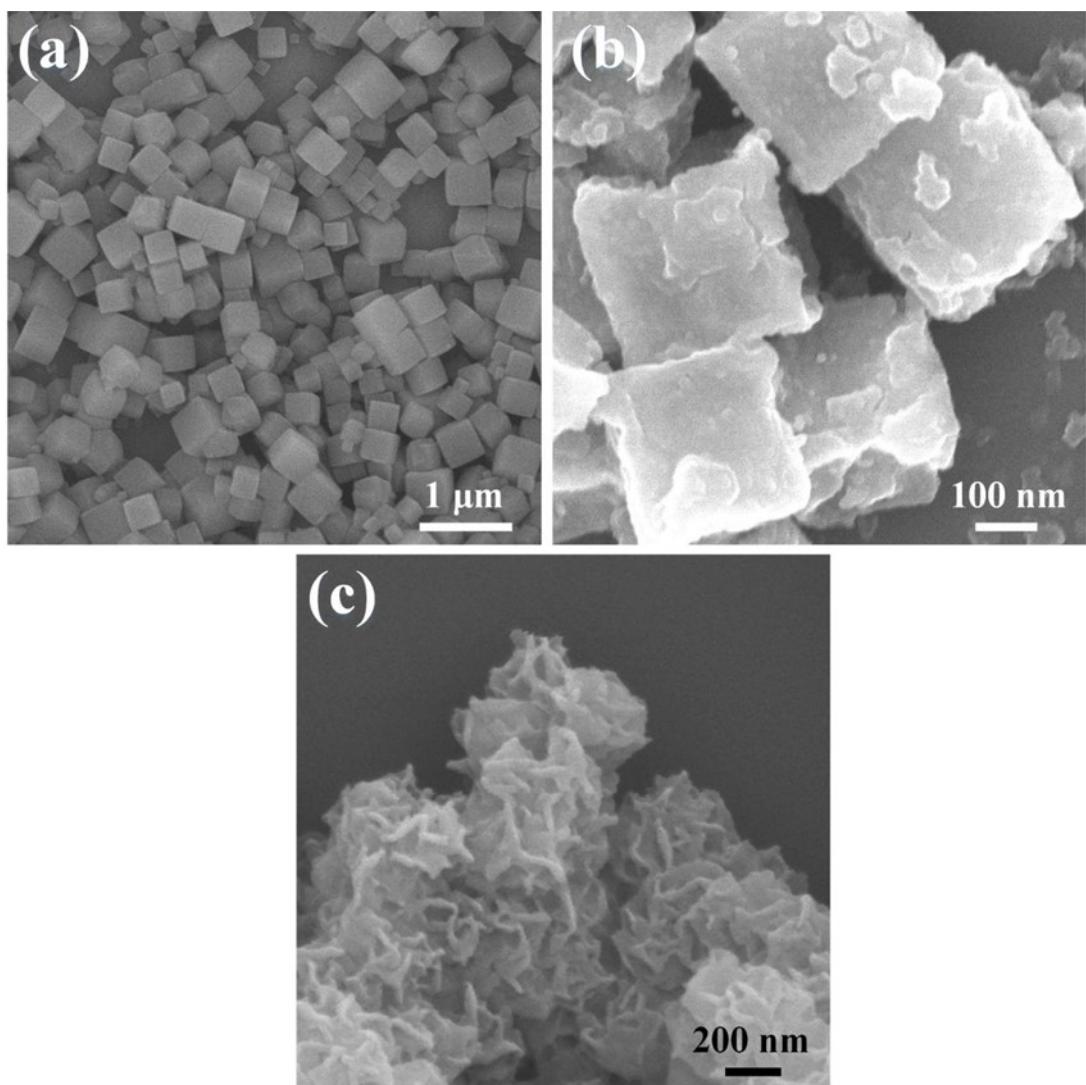


Figure S1 FESEM images of (a) ZIF-67, (b) Co/NC, and (c) CoS₂/NC@1T MoS₂.

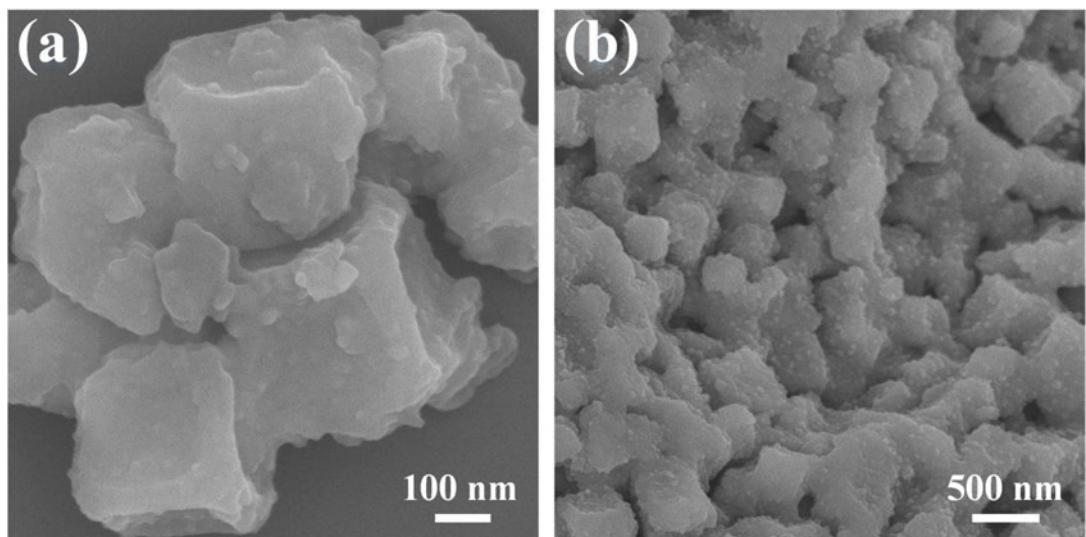


Figure S2 FESEM images of CoS₂/NC with (a) high-resolution and (b) low-resolution.

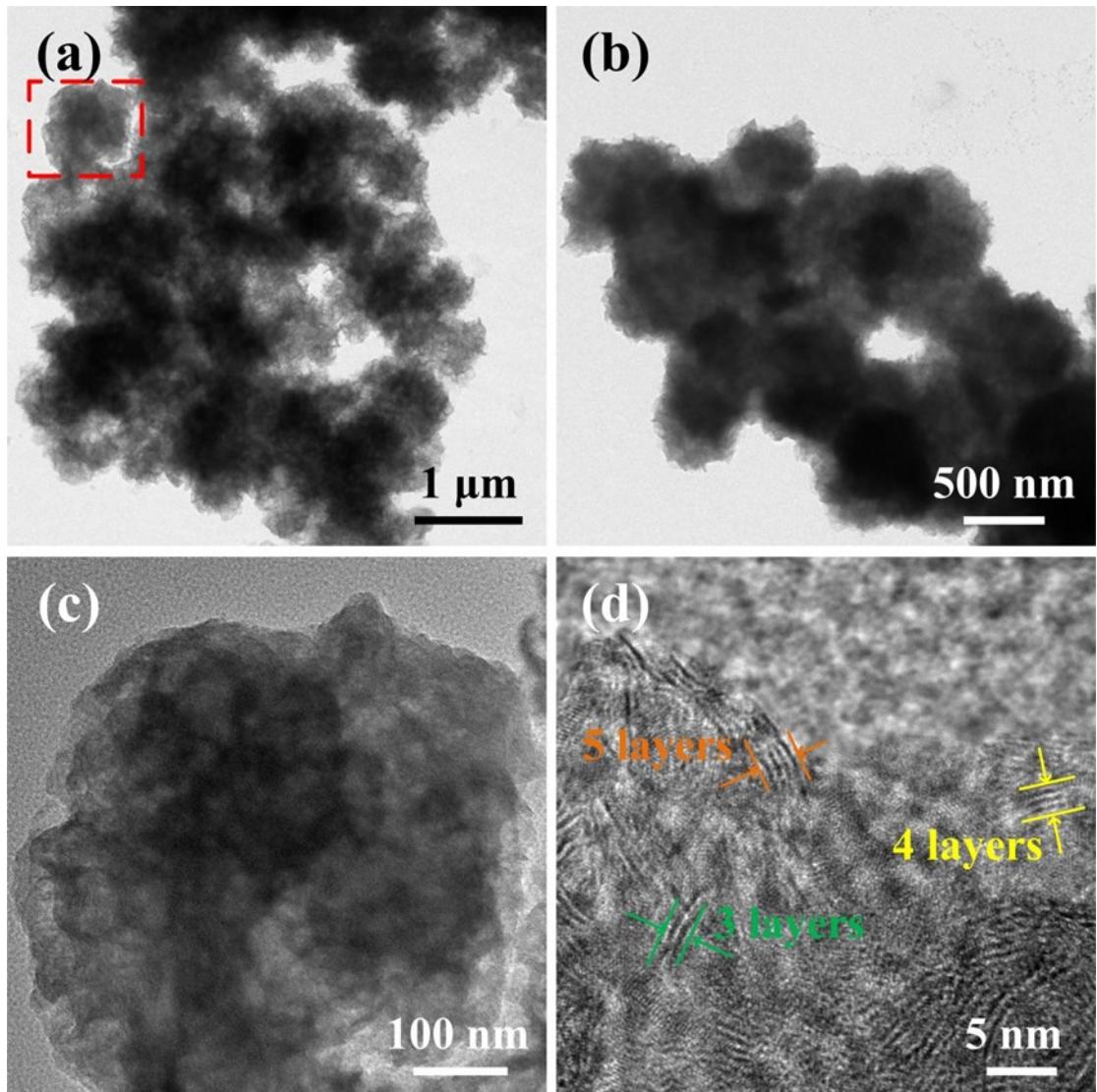


Figure S3 (a,b) TEM images of CoS₂/NC@1T MoS₂ with different magnifications. (c) Enlarged TEM image of the red frame in (a). (d) HRTEM image of CoS₂/NC@1T MoS₂.

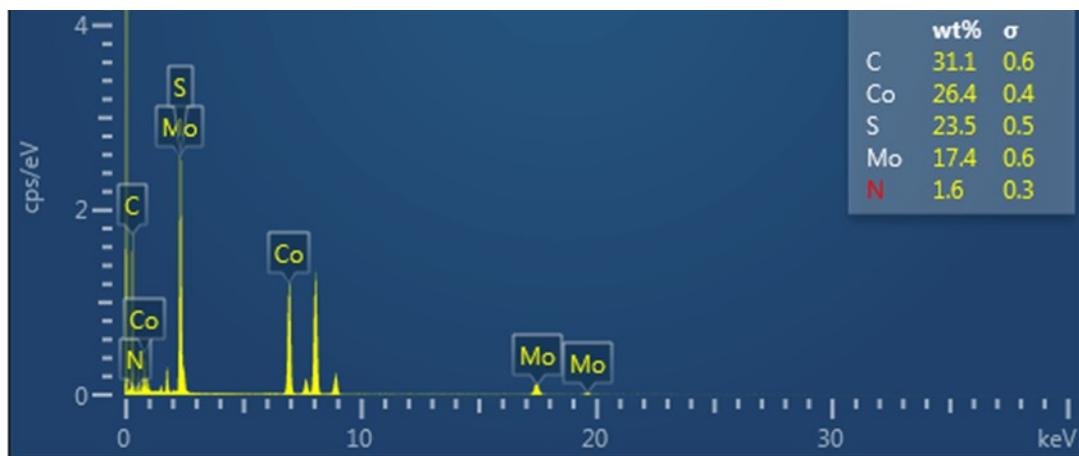


Figure S4 EDS elemental composition diagram of $\text{CoS}_2/\text{NC}@\text{1T MoS}_2$.

According to the EDS elemental composition diagram of TEM (Figure S4), the total amount of N and C elements in $\text{CoS}_2/\text{NC}@\text{1T MoS}_2$ is 32.7 wt%. Since Co, Mo, and S elements exist in the form of chemical compounds, we use more accurate inductively coupled plasma emission spectrometry to determine their elemental composition.

To prepare the ICP sample, a certain amount of $\text{CoS}_2/\text{NC}@\text{1T MoS}_2$ was dissolved in nitric acid, and treated overnight at 60 °C. The as-obtained solution was diluted with deionized water into a concentration of 0.5 mg/L solution. The ICP results show that the concentrations of Mo and Co in $\text{CoS}_2/\text{NC}@\text{1T MoS}_2$ are 0.0751 and 0.0154 mg/L, respectively. The contents of 1T MoS₂ and CoS₂ are calculated by formulas as follows:

$$\text{1T MoS}_2 \text{ wt\%} = \frac{0.0751 \text{ mg/L} \times 160/96}{0.0751 \text{ mg/L} \times 160/96 + 0.0154 \text{ mg/L} \times 123/59} \times 67.3\% = 53.6\%$$

$$\text{CoS}_2 \text{ wt\%} = \frac{0.0154 \text{ mg/L} \times 123/59}{0.0751 \text{ mg/L} \times 160/96 + 0.0154 \text{ mg/L} \times 123/59} \times 67.3\% = 13.7\%$$

According to Figure S4 and the above calculations, the contents of each component in $\text{CoS}_2/\text{NC}@\text{1T MoS}_2$ are 53.6 wt% for 1T MoS₂, 13.7 wt% for CoS₂, and 32.7 wt% for NC, respectively.

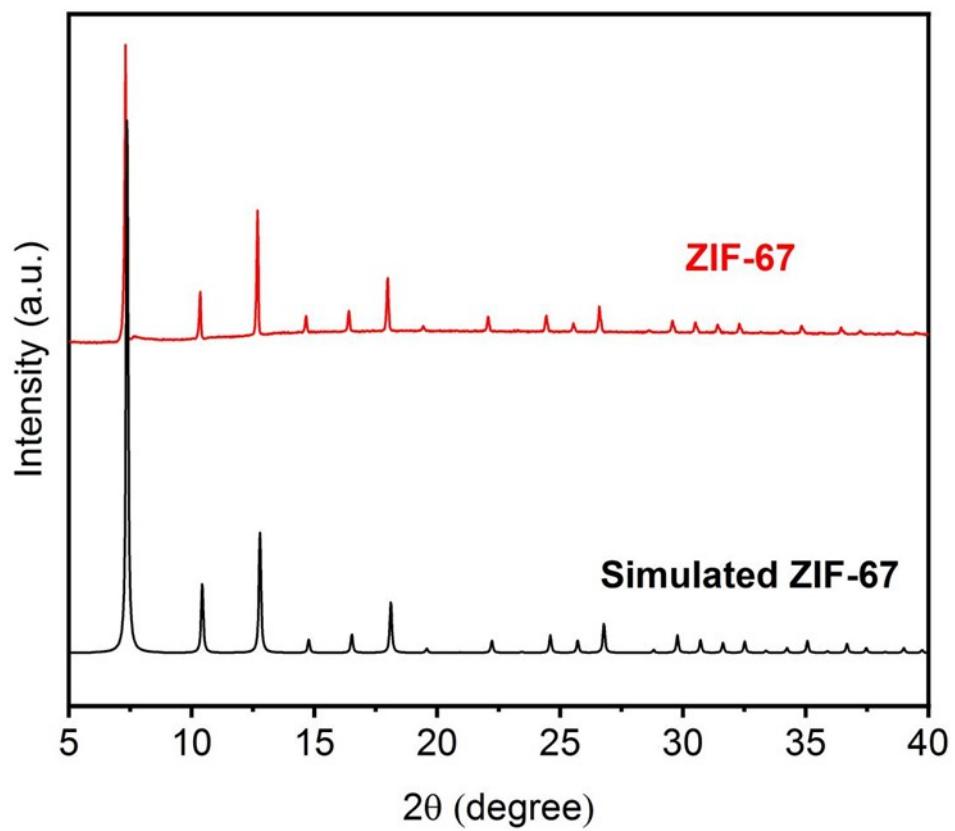


Figure S5 XRD pattern of ZIF-67.

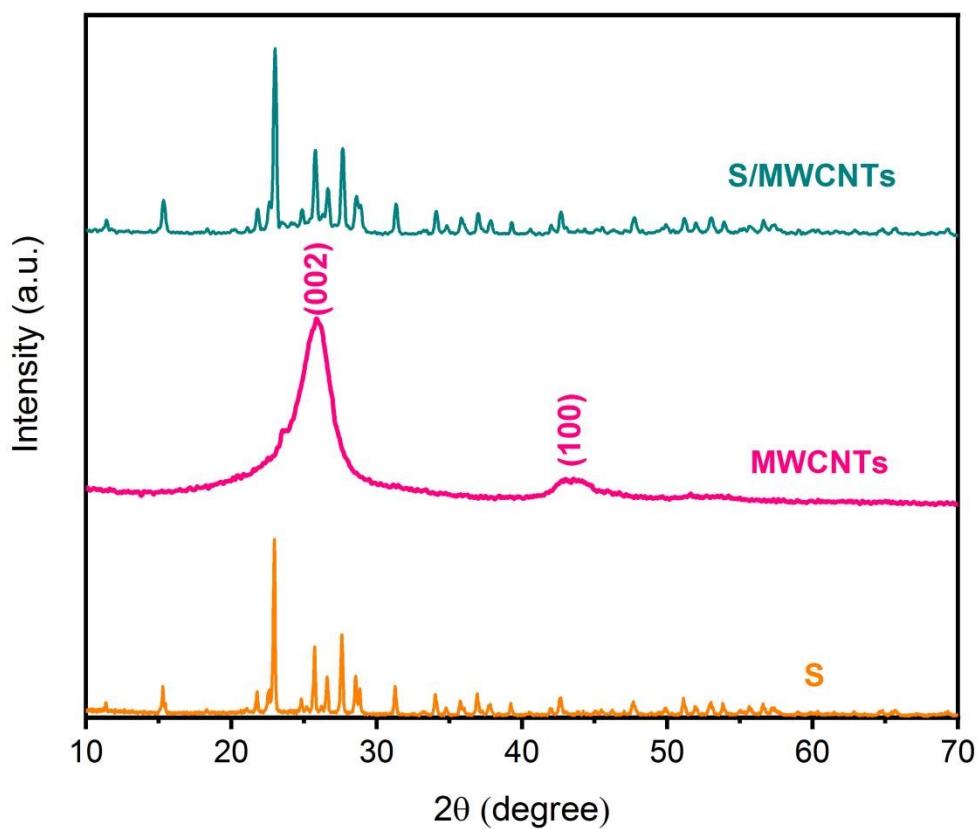


Figure S6 XRD patterns of S, MWCNTs, and S/MWCNTs.

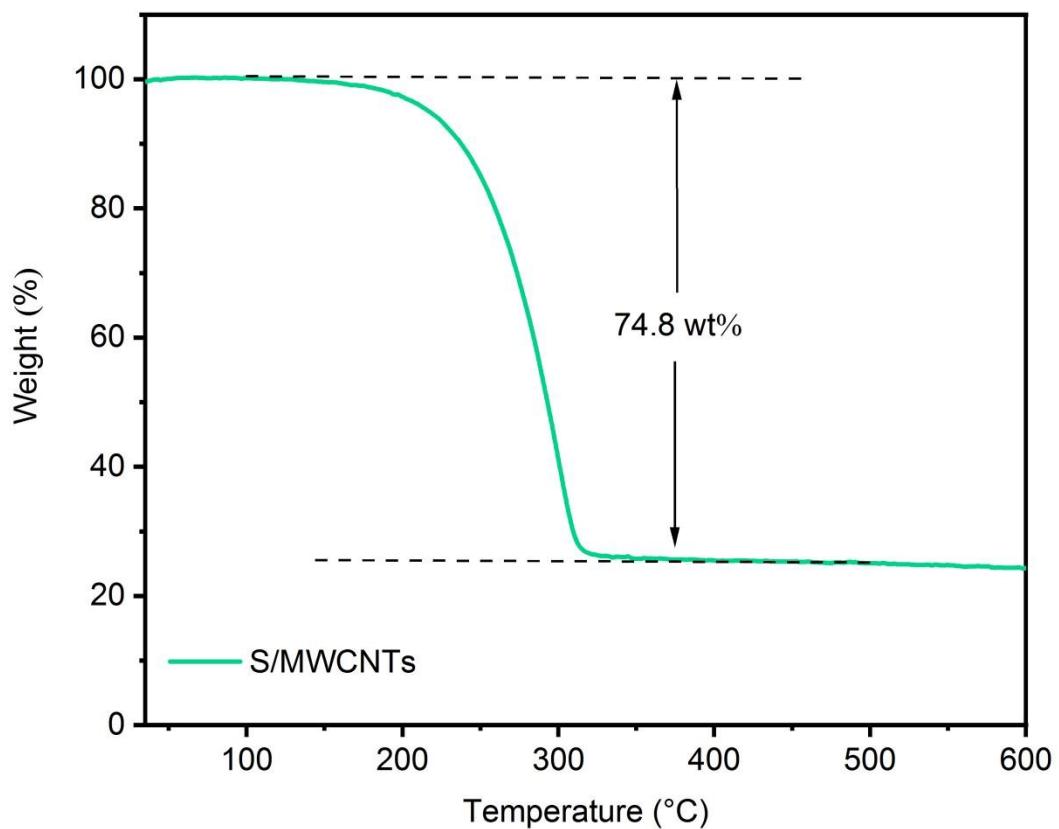


Figure S7 TGA curve of S/MWCNTs.

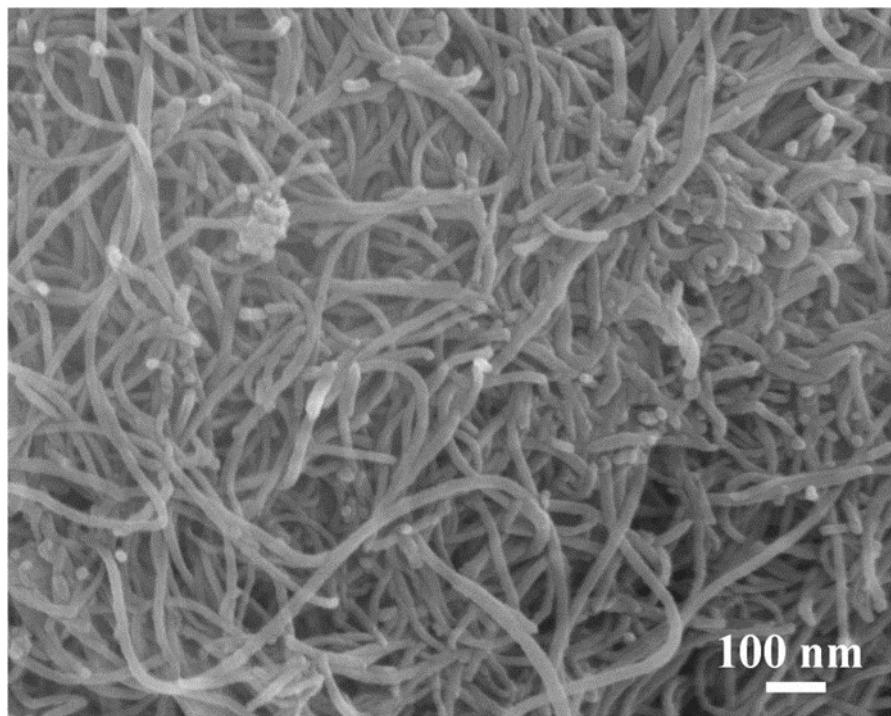


Figure S8 FESEM image of S/MWCNTs.

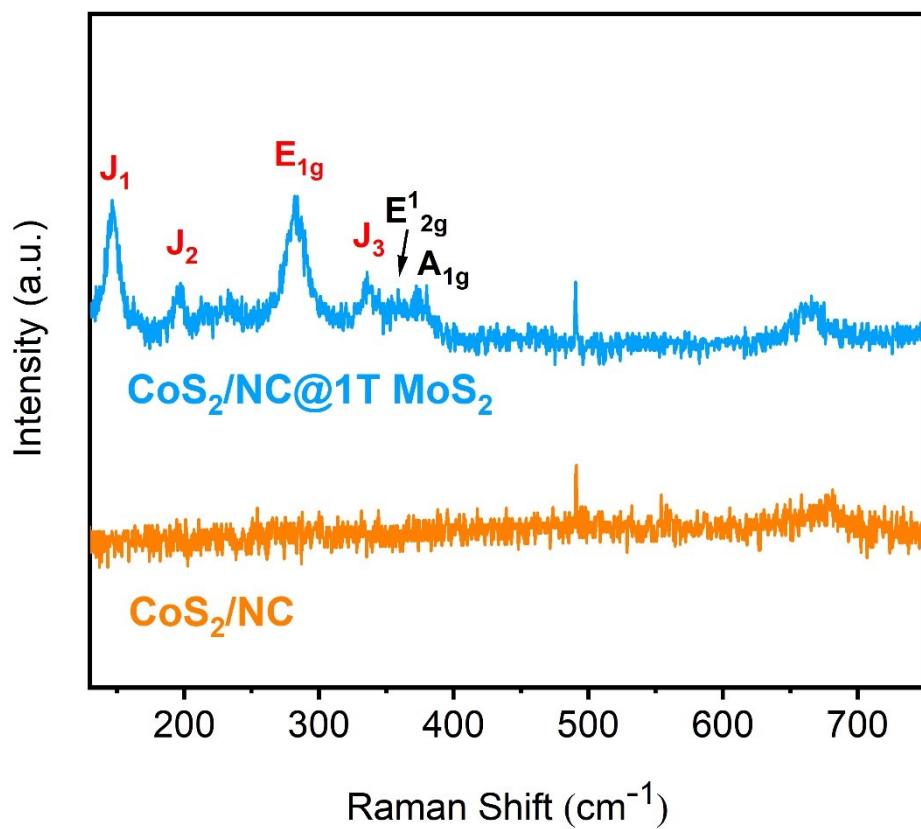


Figure S9 Raman spectra of CoS₂/NC and CoS₂/NC@1T MoS₂ in Raman shift range of 130-750 cm⁻¹.

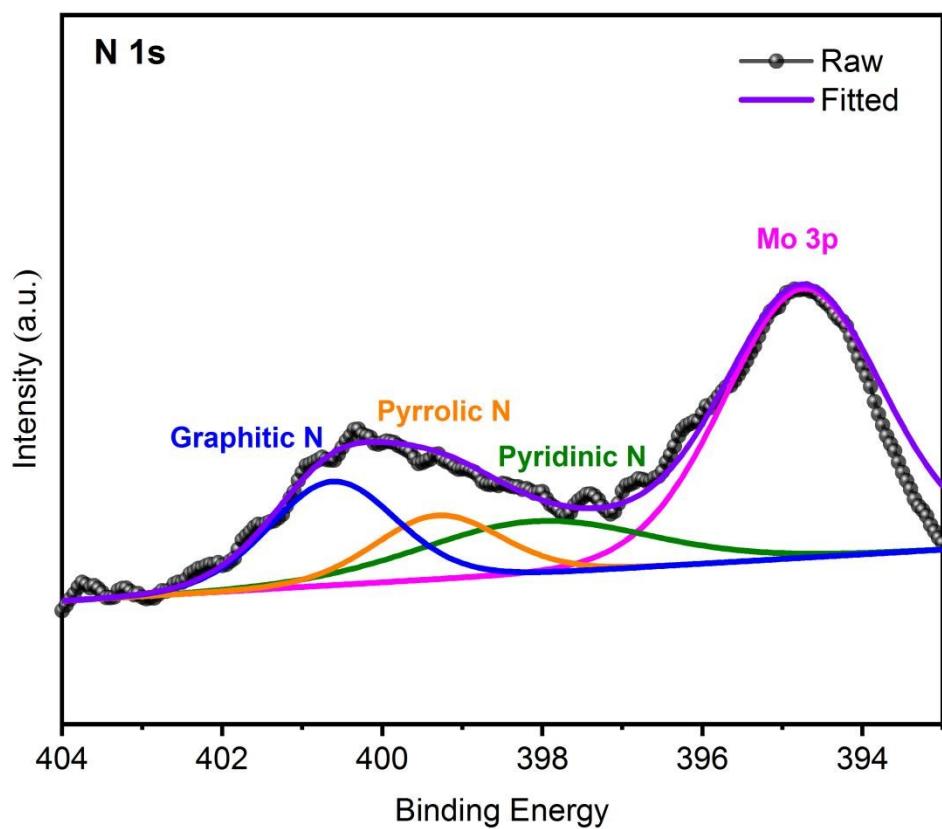


Figure S10 High-resolution XPS spectrum of N 1s of CoS₂/NC@1T MoS₂.

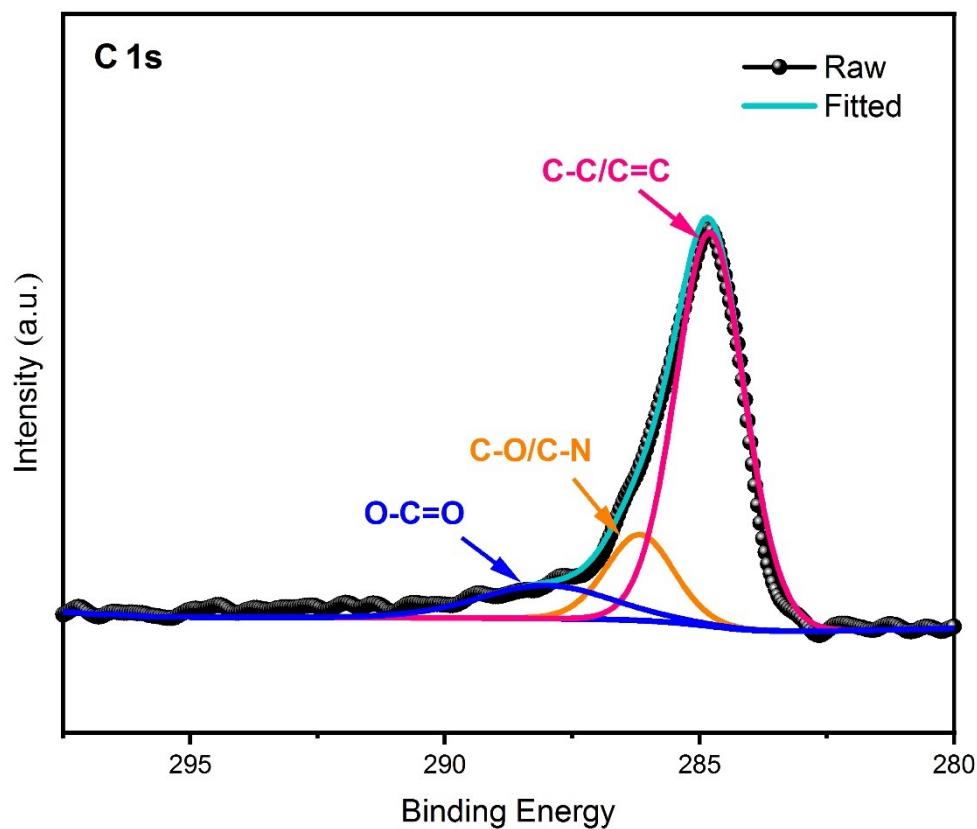


Figure S11 High-resolution XPS spectrum of C 1s of CoS₂/NC@1T MoS₂.

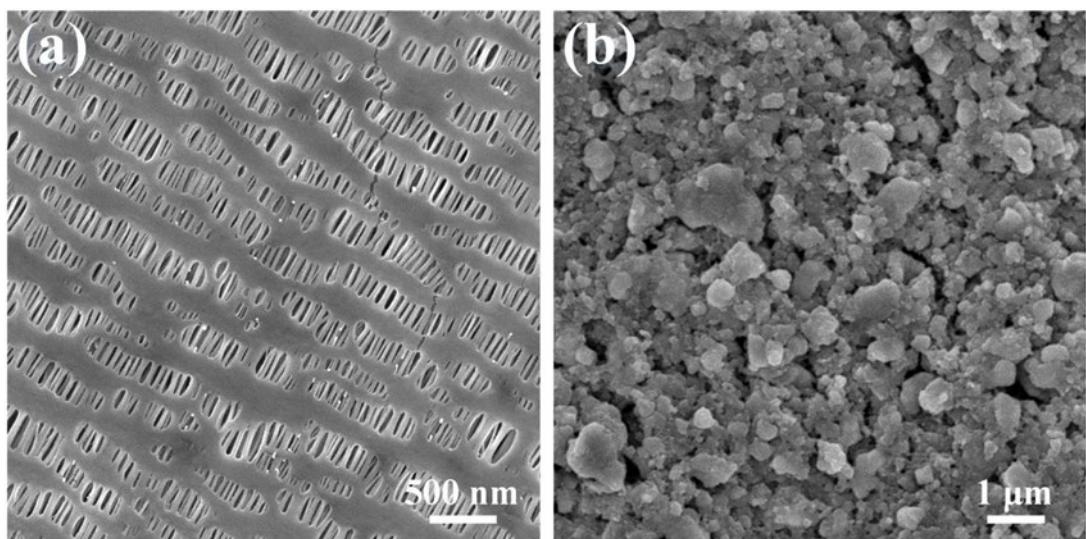


Figure S12 FESEM images of surfaces of (a) pure PP and (b) CoS₂/NC@1T MoS₂-PP separators before cycle.

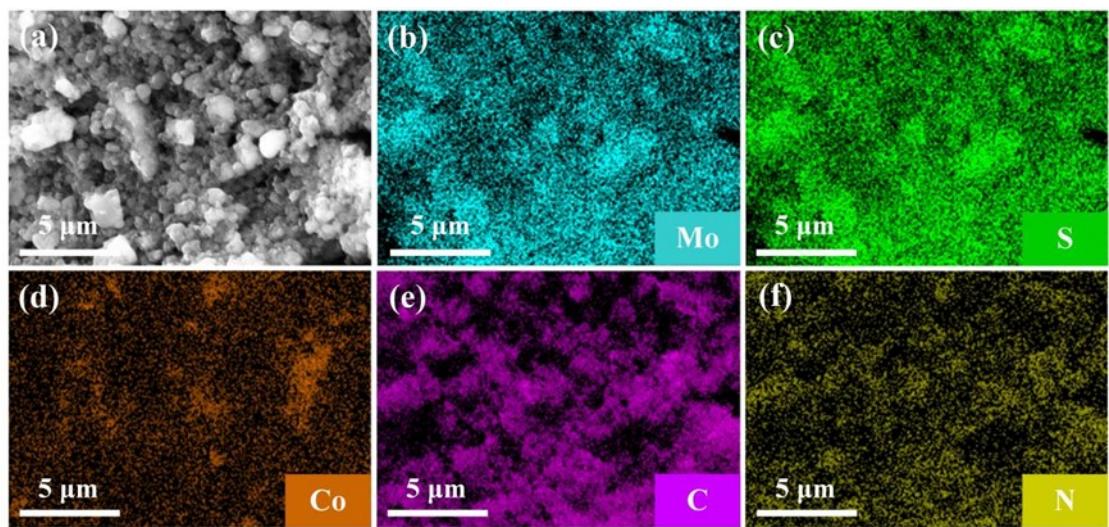


Figure S13 (a) Top STEM image of $\text{CoS}_2/\text{NC}@\text{1T MoS}_2\text{-PP}$ separator, and (b-f) corresponding elemental mappings of Mo, S, Co, C, and N.



Figure S14 Photographs of PP and CoS₂/NC@1T MoS₂-PP separators after heat treatment.

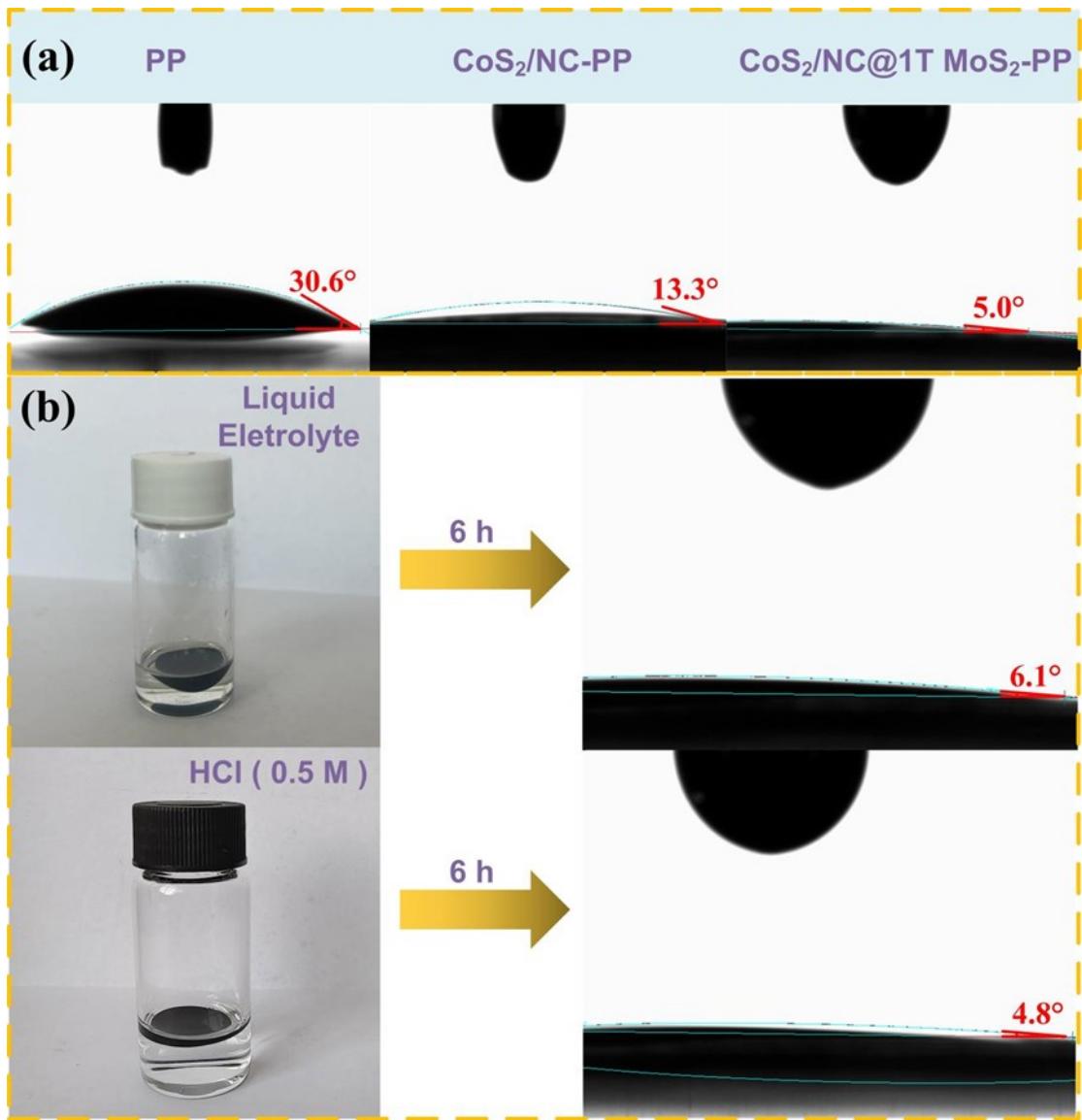


Figure S15 (a) Wettability of pure PP, $\text{CoS}_2/\text{NC-PP}$, and $\text{CoS}_2/\text{NC@1T MoS}_2\text{-PP}$ separators. (b) Chemical stability of $\text{CoS}_2/\text{NC@1T MoS}_2\text{-PP}$ separator.

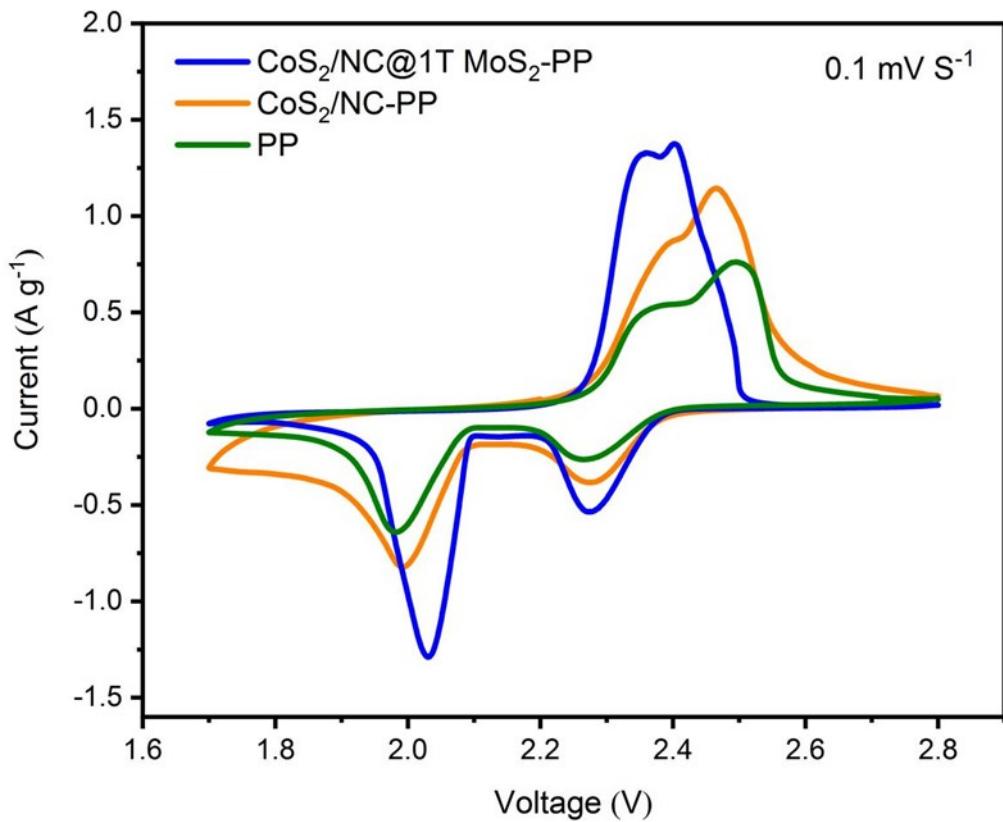


Figure S16 CV curves of lithium-sulfur batteries with $\text{CoS}_2/\text{NC}@1\text{T MoS}_2$ -PP, CoS_2/NC -PP, and pure PP separators at a scan rate of 0.1 mV S^{-1} .

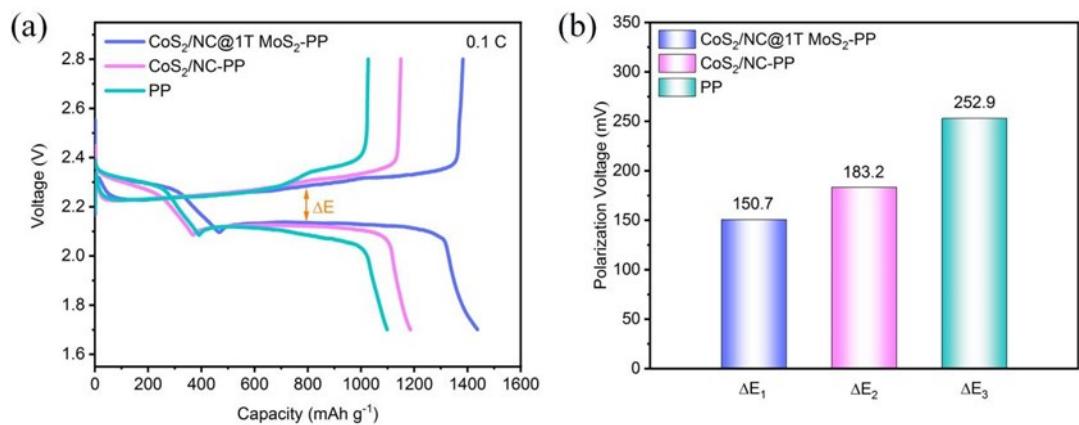


Figure S17 (a) Initial charge/discharge curves and (b) corresponding polarization voltage bar graphs of lithium-sulfur batteries with $\text{CoS}_2/\text{NC@1T MoS}_2\text{-PP}$, $\text{CoS}_2/\text{NC-PP}$, and pure PP separators at 0.1 C.

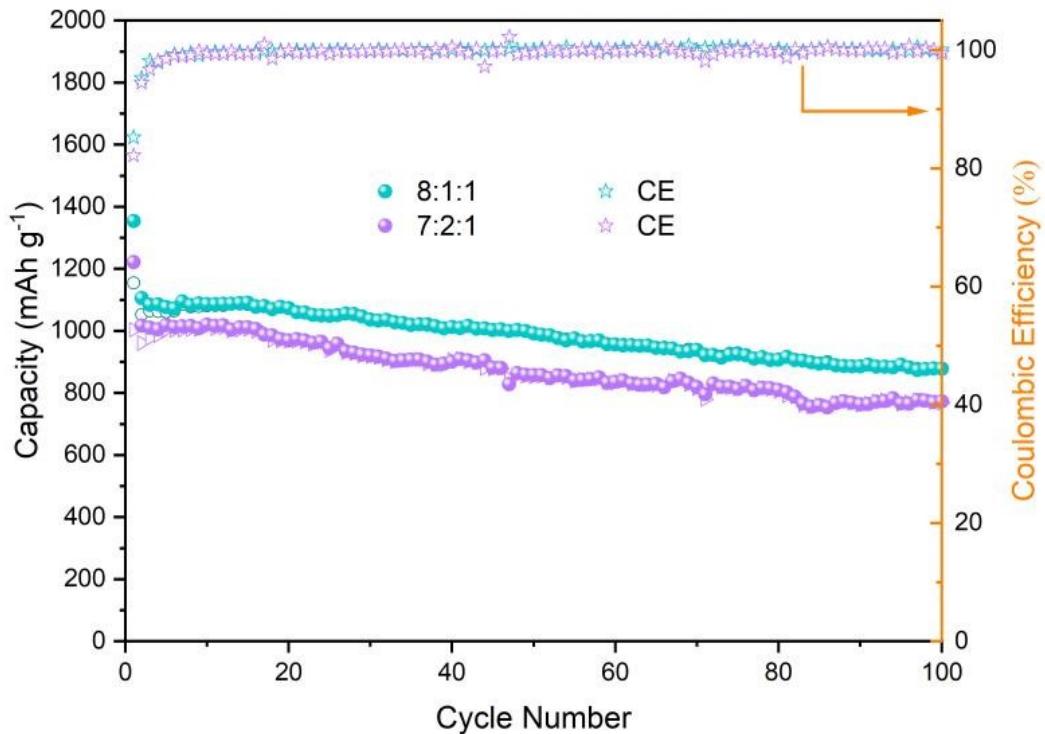


Figure S18 Cycle performance and coulombic efficiency of lithium-sulfur batteries with CoS₂/NC@1T MoS₂-PP separator for 100 cycles at a current density of 0.2 C at different positive slurry ratios.

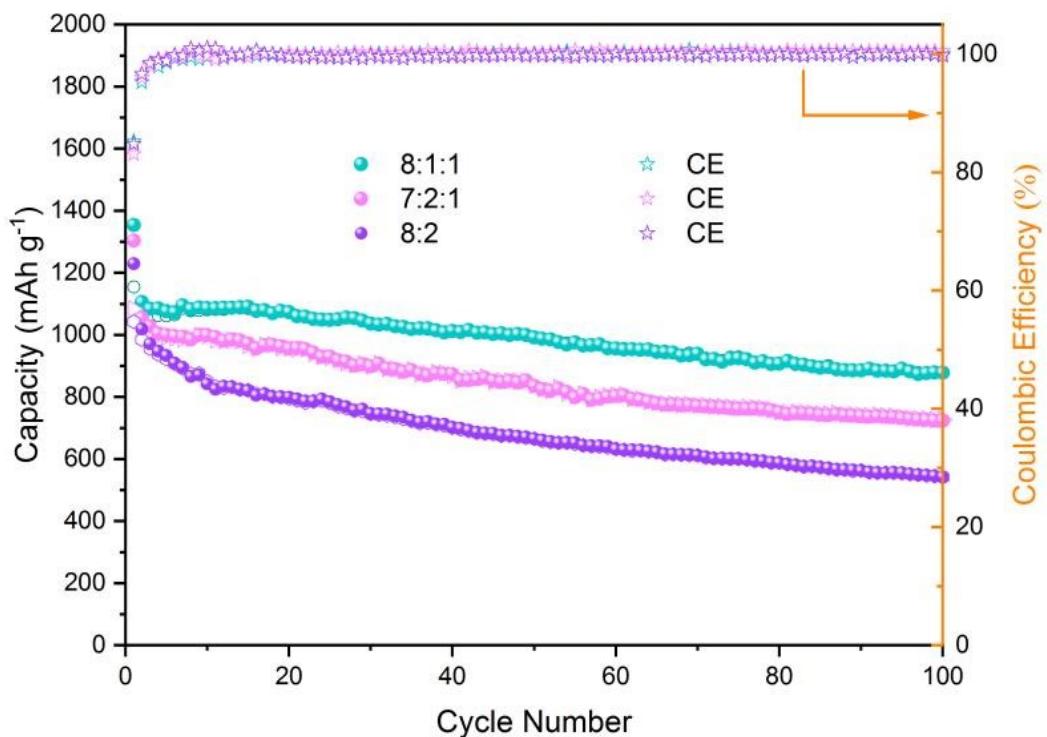


Figure S19 Cycle performance and coulombic efficiency of lithium-sulfur batteries with $\text{CoS}_2/\text{NC}@\text{1T MoS}_2\text{-PP}$ separator for 100 cycles at a current density of 0.2 C at different separator slurry ratios (8:2, no acetylene black).

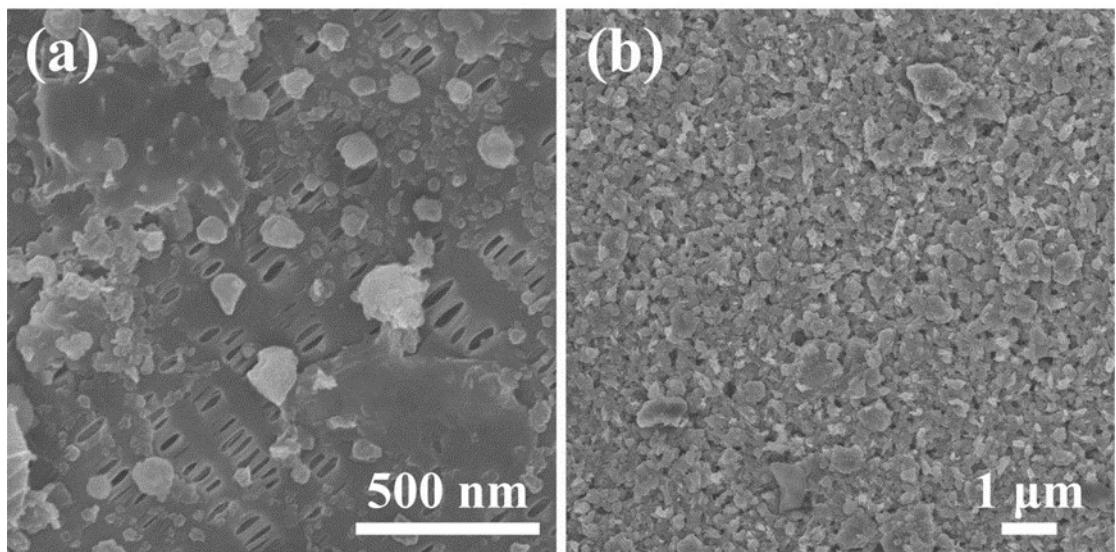


Figure S20 SEM images of surfaces of (a) pure PP and (b) CoS₂/NC@1T MoS₂-PP separators after cycles.

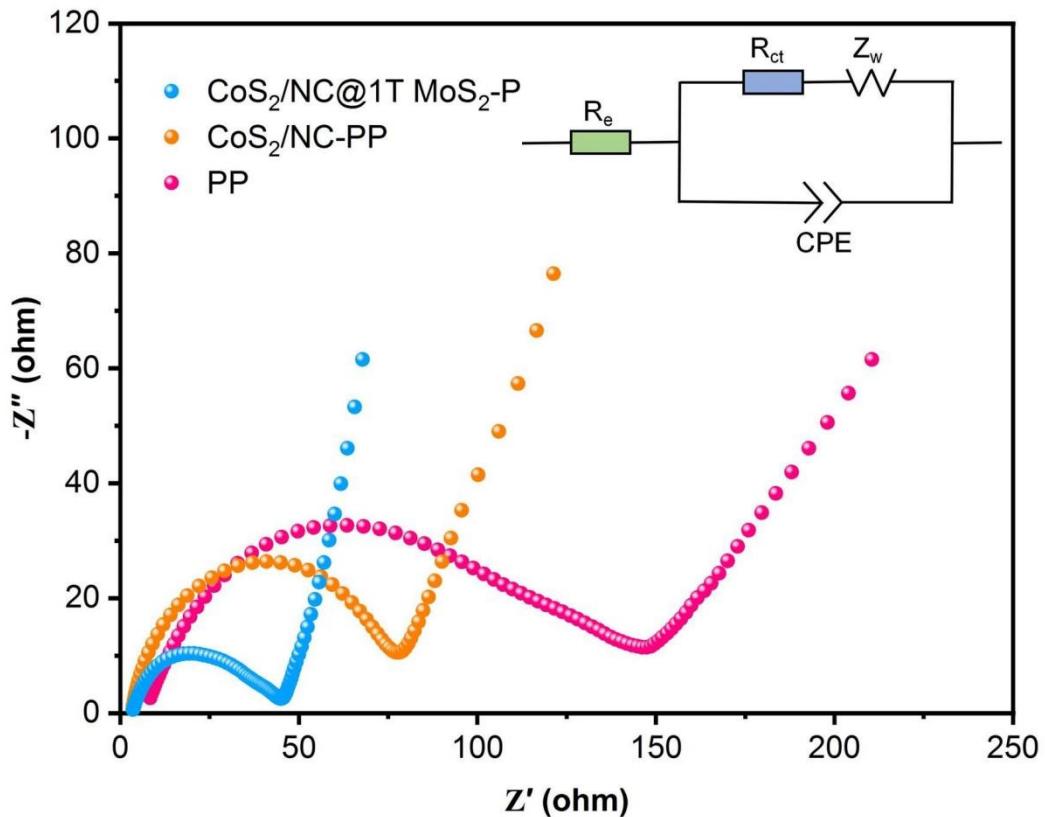


Figure S21 AC impedance spectra of lithium-sulfur batteries with $\text{CoS}_2/\text{NC}@\text{1T MoS}_2\text{-PP}$, $\text{CoS}_2/\text{NC-PP}$, and pure PP separators before cycle (inset: equivalent circuit).

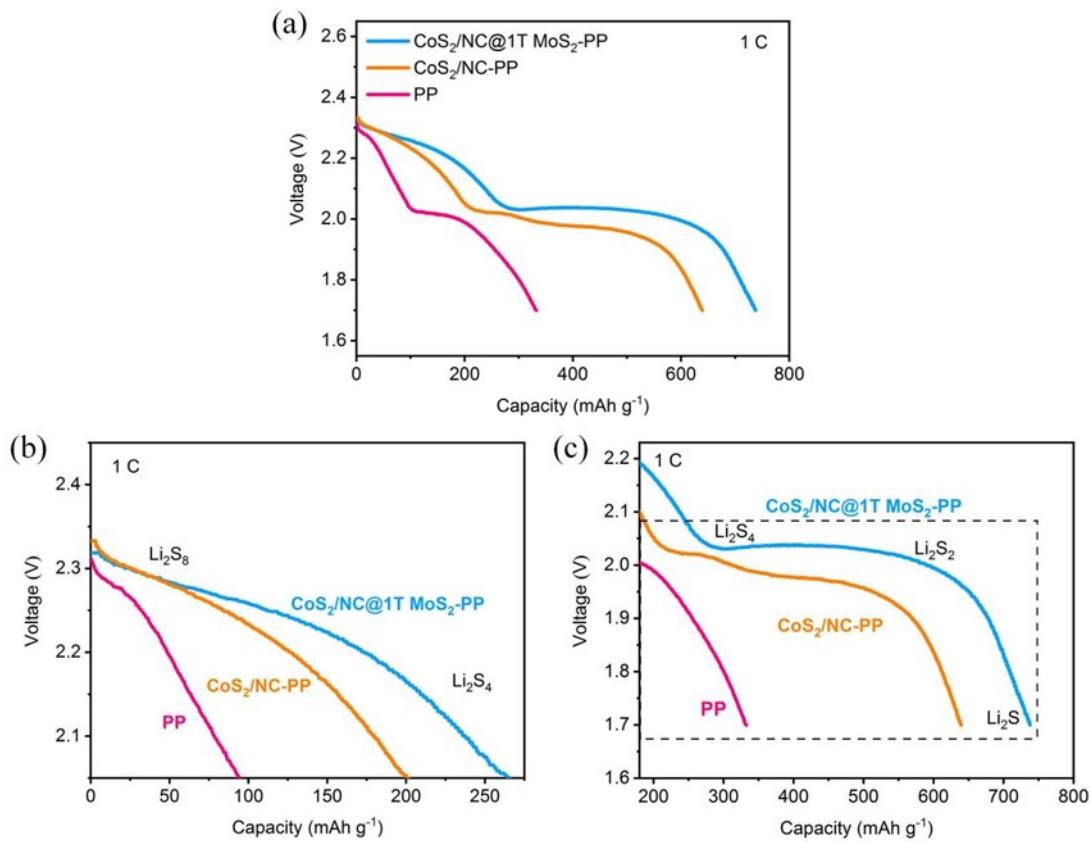


Figure S22 (a) Discharge curves of lithium-sulfur batteries with CoS₂/NC@1T MoS₂-PP, CoS₂/NC-PP, and pure PP separators at a current density of 1 C, and comparison of discharge curves at (b) high voltage and (c) low voltage at a current density of 1 C.

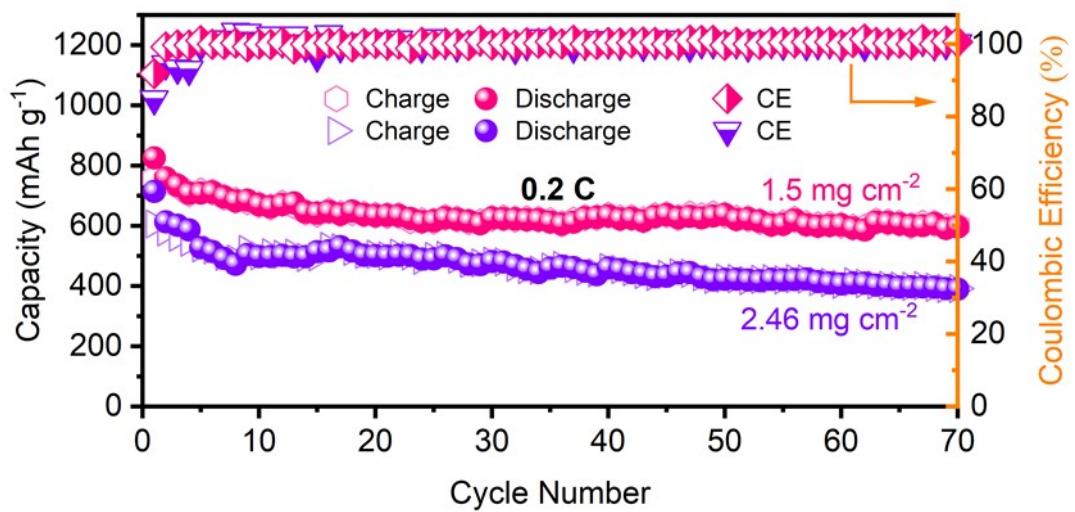


Figure S23 Cycle performance of LSBs with different high sulfur loadings at 0.2 C.



Figure S24 Digital photos of a light-emitting diode (LED) four-color lamp string lit up by CR2025-type lithium-sulfur batteries with $\text{CoS}_2/\text{NC}@\text{1T MoS}_2\text{-PP}$ separator.

Table S1 Comparison of electrochemical properties of LSBs with CoS₂/NC@1T MoS₂-PP separator with previously reported LSBs.

Sample	Current density	Cycle number	Discharge capacity (mAh g ⁻¹)	Capacity decay rate per cycle (%)	Ref.
CC/VN/Co@NCNTs	0.1 C	100	864.1	0.35	S1
N-MIMEC	0.1 C	100	971.3	0.253	S2
CoSe@NC	0.1 C	100	804.7	0.39	S3
MnO ₂ @SnO ₂	0.1 C	100	806	0.39	S4
3D P-MoS ₂	0.5 C	300	629.4	0.144	S5
TiO ₂	0.5 C	400	612	0.104	S6
CP@NCNT@CoS ₃	0.83 C	400	680	0.119	S7
CoS ₂ @NGCNs	1 C	300	519.4	0.075	S8
CoS ₂ /HPGC	1 C	500	519	0.07	S9
Co-NC@N-HCS	1 C	450	399.7	0.13	S10
CoS ₂ /ACCF	1 C	500	580	0.082	S11
FSC/MoS ₂ /CNTs	1 C	500	615.3	0.059	S12
CoS₂/NC@1T MoS₂	0.1 C	100	972	0.316	This
	1 C	500	616	0.07	work

Notes and references

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