

Electronic Supplementary Information

Modulating the electronic configuration of Co species in MOF/MXene nanosheets derived Co-based mixed spinel oxides for efficient oxygen evolution reaction

Chuming Xu,^{#a} Xifeng Yang,^{#a} Shuang Li,^{*a,b} Keke Li,^a Benjun Xi,^b Qing-Wen Han,^b
Ya-pan Wu,^{a,b} Xue-Qian Wu,^{a,b} Ru-an Chi^b and Dong-sheng Li^{*a,b}

^aCollege of Materials and Chemical Engineering, Key Laboratory of Inorganic Nonmetallic Crystalline and Energy Conversion Materials, China Three Gorges University, Yichang, 443002, P. R. China

^bHubei Three Gorges Laboratory, Yichang, Hubei 443007, P. R. China

*Corresponding Author.

E-mail address: lishmail@126.com; lidongsheng1@126.com.

[#]These authors contributed equally to this work.

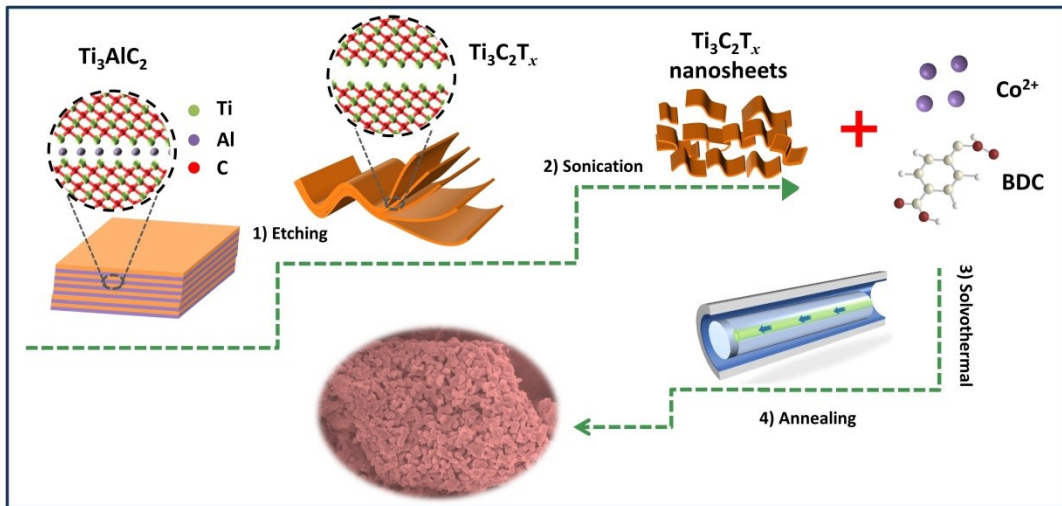


Fig. S1. The preparation process of the Co-based mixed spinel oxides.

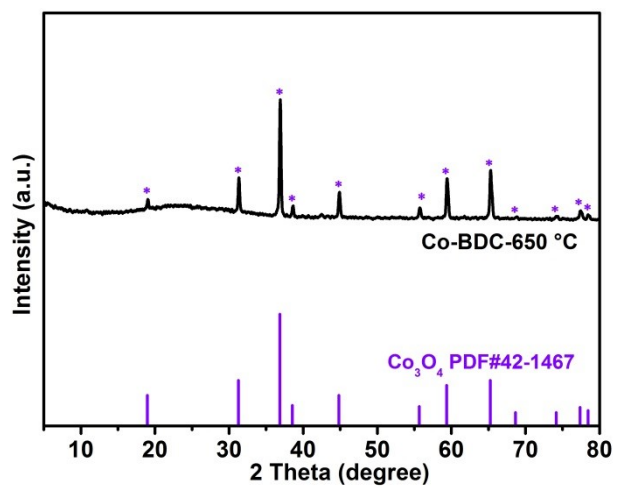


Fig. S2. XRD pattern of Co-BDC annealed at 650°C.

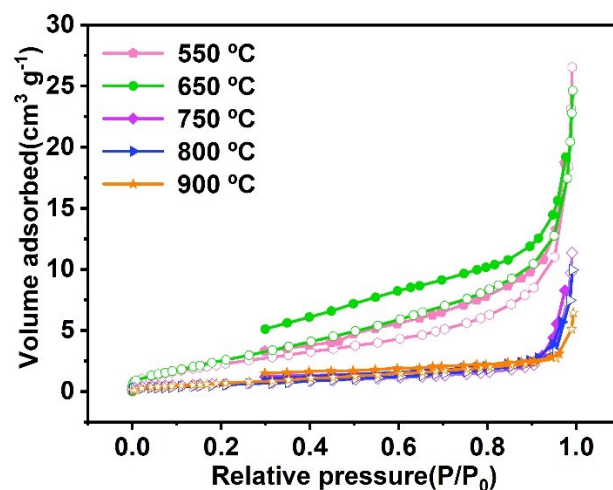


Fig. S3. Nitrogen adsorption and desorption curves of the $\text{CoTiO}_x\text{-T}$. All samples show type-IV isotherms with a very distinct hysteresis loop of typical H3. The BET surface area of the $\text{CoTiO}_x\text{-650}$ is $12.3 \text{ m}^2 \text{ g}^{-1}$, which is larger than that of $\text{CoTiO}_x\text{-550}$ ($9.5 \text{ m}^2 \text{ g}^{-1}$), $\text{CoTiO}_x\text{-750}$ ($2.3 \text{ m}^2 \text{ g}^{-1}$), $\text{CoTiO}_x\text{-800}$ ($2.5 \text{ m}^2 \text{ g}^{-1}$), $\text{CoTiO}_x\text{-900}$ ($3.1 \text{ m}^2 \text{ g}^{-1}$).

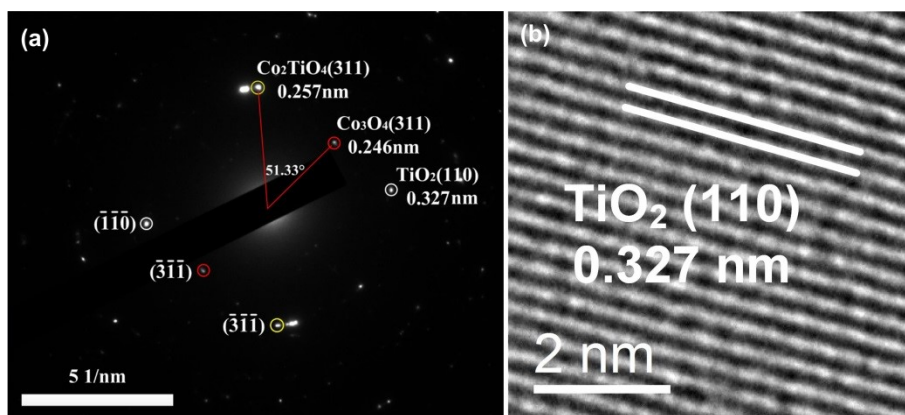


Fig. S4. (a) The selected area electron diffraction (SAED) image and (b) HRTEM image of CoTiO_x -650.

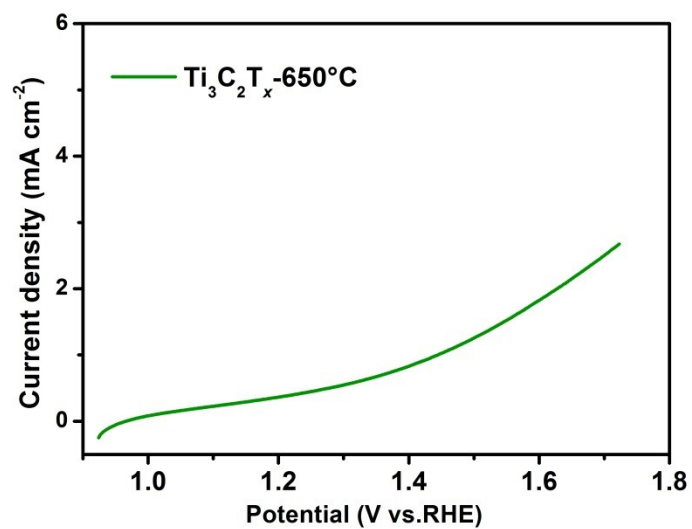


Fig. S5. The polarization curve of OER on $\text{Ti}_3\text{C}_2\text{T}_x$ annealed at 650°C . It can be seen that the annealed product TiO_2 has almost no OER activity.

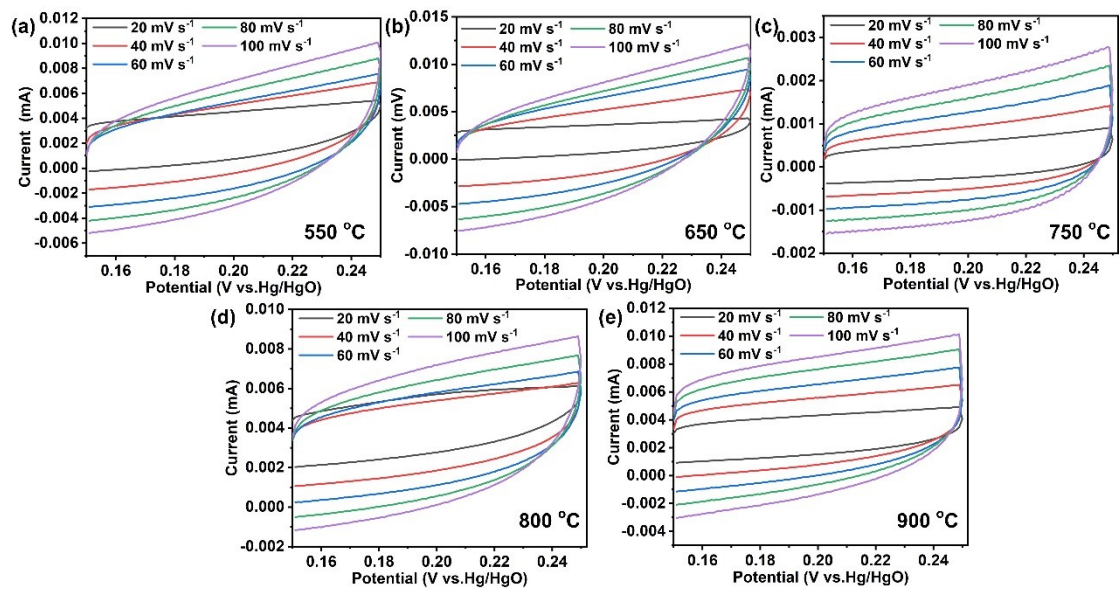


Fig. S6. Cyclic voltammetric curves of different samples.

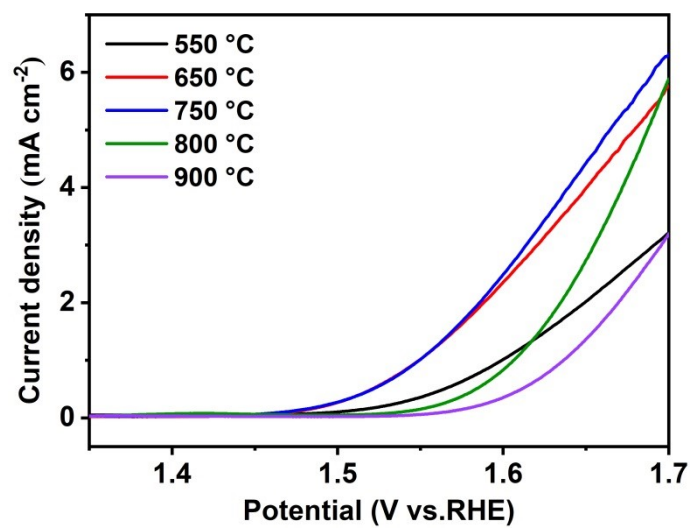


Fig. S7. C_{dl} -normalized polarization curves for CoTiO_x-T catalysts.

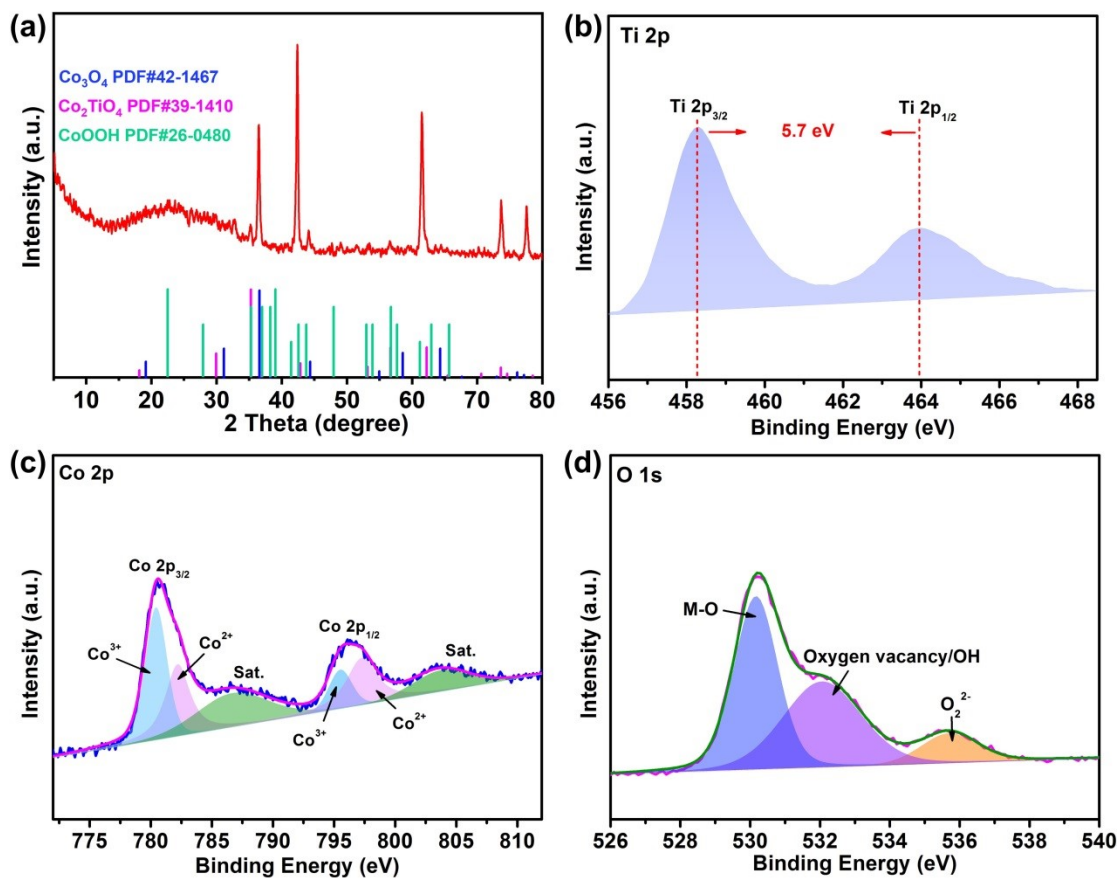


Fig. S8. The XRD and XPS of CoTiO_x-650 after stability test.

Table S1. Comparison of the overpotential at 10 mA cm⁻² of present work and the other OER catalysts in a three-electrode system

Catalyst	Overpotential 10 mA cm ⁻² (mV)	Electrolyte	Support	Reference
CoTiO_x-650	260	1M KOH	NF	This work
	280	1M KOH	GC	
NNU-23 (Fe ₂ Ni-MOF)	365	0.1m KOH	CC	Angew. Chem. Int. Ed., 2018, 57, 9660
Co-BDC nanosheets	371	1M KOH	GC	Nat. Energy, 2016, 1, 16184
NiO/CoN PINWs	300	0.1M KOH	CC	ACS Nano, 2017, 11, 2275
Ni-MOF@Fe-MOF powder	265	1M KOH	GC	Adv. Funct. Mater., 2018, 28, 1801554
FeNi-BTC	270	0.1 M KOH	NF	ACS Appl. Mater. Interfaces, 2016, 8 , 16736
Co-PB/Pt	300	1M KOH	GC	ACS Sustainable Chem. Eng., 2017, 5, 11577
Co(TCNQ) ₂	310	1M KOH	Co foil	Chem. Eur. J., 2018, 24, 2075
TiC ₂ TX-CoBDC	410	0.1M KOH	GC	ACS Nano, 2017, 11, 5800
Fe:2D-Co-NS	282	0.1M KOH	GC	Angew. Chem. Int. Ed., 2018, 57, 4632
Co ₃ O ₄ -based catalysts	260	1M KOH	CC	Angew. Chem. Int. Ed., 2020, 59, 6929-6935
Co ₃ O ₄ -Ag@B	270	1M KOH	GC	Applied Catalysis B: Environmental, 2021, 298, 120529
Co ₃ O ₄ /Ti ₃ C ₂	300	1M KOH	GC	Sci. Bull., 2020, 65, 460
Ni-MOF@Fe-MOF powder	265	1M KOH	GC	Adv. Funct. Mater., 2018, 28, 1801554
n-Co ₃ O ₄	380	1M KOH	GC	ACS Appl. Energy Mater., 2020, 3, 5439
Fe-Co-O nanosheets	260	1M KOH	GC	Small, 2020, 16, 2001571
Co ₃ O ₄ /rGO	290	1M KOH	CF	Chem. Eur. J., 2017, 23, 4010
Co@Co ₃ O ₄	333	1M KOH	GC	ACS Catal., 2018, 8, 7879
Co ₃ O ₄ @CMC	290	1M KOH	GC	Electrochim. Acta, 2021, 398, 139338