

## Supporting Information

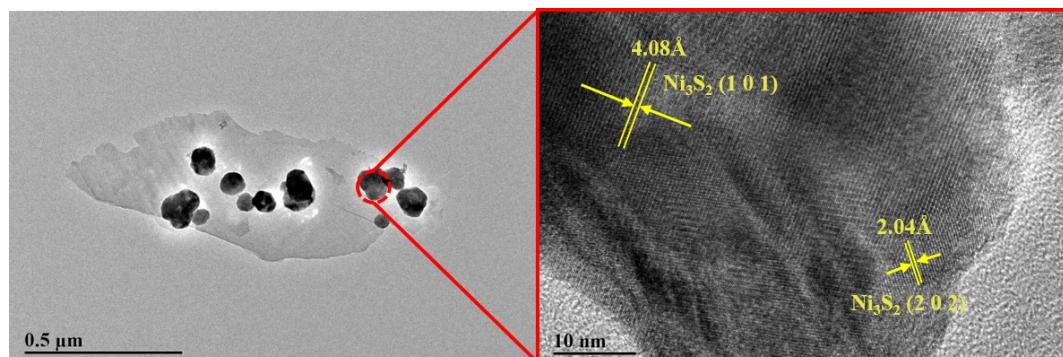
### In-situ formation of Nickel sulfide quantum dots embedded into two-dimensional metal-organic framework for water splitting

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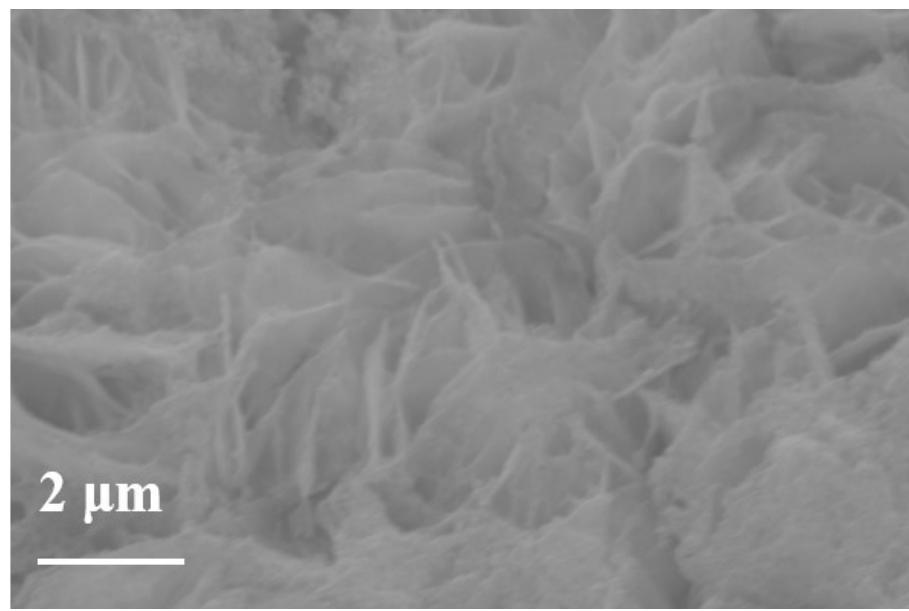
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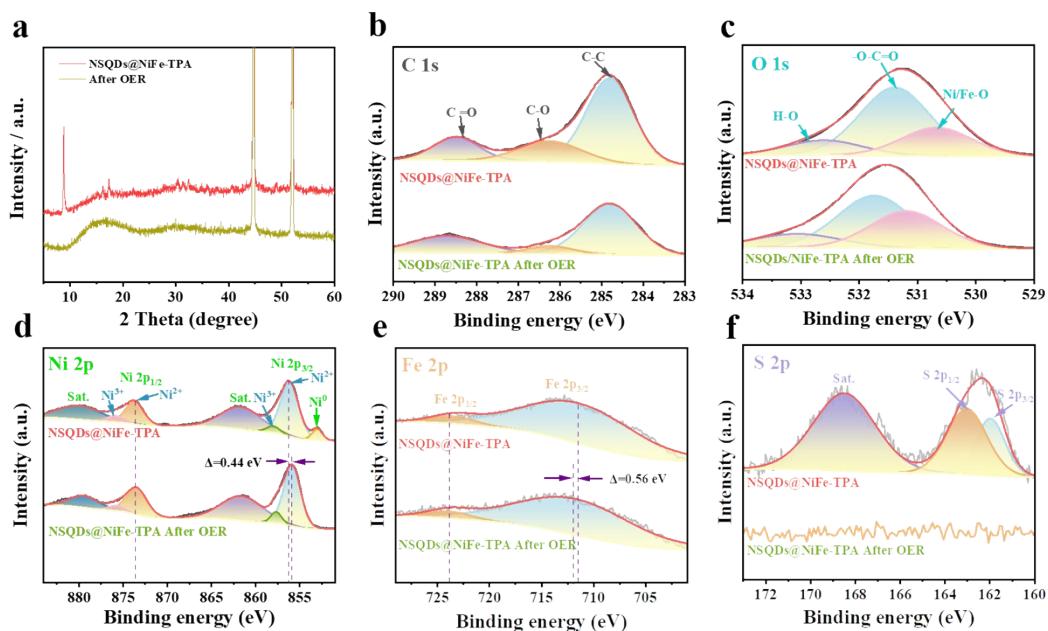


**Figure S1** (a) TEM and (b) HRTEM images of  $\text{Ni}_3\text{S}_2@\text{NiFe-TPA}$

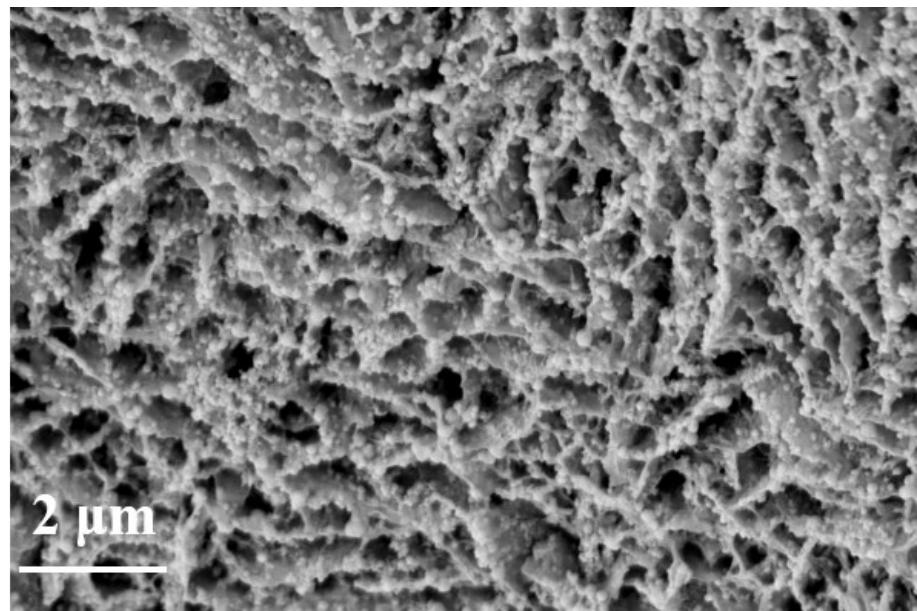
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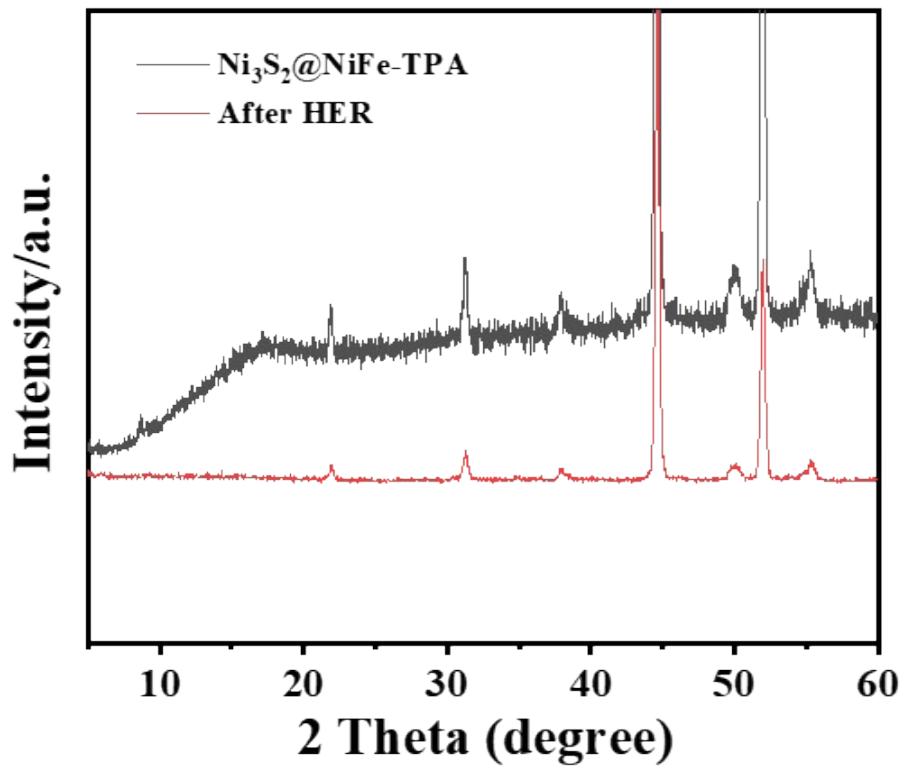
**Figure S2** the SEM curve of NSQDs@NiFe-TPA after the stability test of OER



**Figure S3** (a) the XRD curve and the high resolution XPS of (b) C 1s, (c) O 1s, (d) Ni 2p (e) Fe 2p and (f) S 2p of NSQDs@NiFe-TPA and NSQDs@NiFe-TPA after OER.



**Figure S4** the SEM curve of  $\text{Ni}_3\text{S}_2@\text{NiFe-TPA}$  after HER.



**Figure S5** XRD patterns of  $\text{Ni}_3\text{S}_2@\text{NiFe-TPA}$  and  $\text{Ni}_3\text{S}_2@\text{NiFe-TPA}$  after HER.

**Table S1** A performance comparison with recently published non-noble metal-based OER catalysts

Electrocatalyst	Electrolyt e	Overpotential (mV vs. RHE)	Ref.
		10 mA cm <sup>-2</sup>	
NSQDs@NiFe-TPA	1 M KOH	219	This work
Co(Zn)S <sub>2</sub> /CC	1 M KOH	248	[1]
Fe-NiCo-S	1 M KOH	247	[2]
CoFe-MS/MOF	1 M KOH	264	[3]
N, P-Co <sub>9</sub> S <sub>8</sub> /CoS <sub>2</sub> /Co <sub>1-x</sub> S	1 M KOH	285	[4]
Ti-CoS <sub>x</sub> HSS	1 M KOH	249	[5]
Ni–Co–S/NSC	1 M KOH	309	[6]

**Table S1** A performance comparison with recently published non-noble metal-based HER catalysts

Electrocatalyst	Electrolyt e	Overpotential (mV vs. RHE)	Ref.
		10 mA cm <sup>-2</sup>	
Ni <sub>3</sub> S <sub>2</sub> @NiFe-TPA	1 M KOH	109	This work
Ni <sub>3</sub> S <sub>2</sub> @2D Co-MOF	1 M KOH	140	[7]
40% MoS <sub>x</sub> /Co-MOF-74	1 M KOH	147	[8]
Ni-M@C-130	1 M KOH	123	[9]
CoS <sub>1.097</sub> -160	1 M KOH	163	[10]
MoS <sub>x</sub> /Ni-MOF-74	1 M KOH	114	[11]
Ni–Co–S/NSC	1 M KOH	177	[12]

## References

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