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Supporting Information

Low-temperature Activable, Carbon Dioxide based, Highly Adhesive and Degradable Oligo-urethane and its Potential Application as Drug Release and Auto-detachable Dressing

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Table S1. Compositions and characterization of the synthesized OUs

Batch No.	PPCG (g)	Chain extende r (g)	IPDI (g)	Catalyst (g)	M _n /10 ³ (g /mol)	M _w /10 ⁴ (g /mol)	PDI
OU-1	16.00	0.50	2.80	0.25	6.75	1.28	1.89
OU-2	16.00	1.00	3.73	0.25	5.70	1.01	1.77
OU-3	16.00	1.50	4.67	0.25	8.04	1.23	1.23
OU-4	16.00	2.00	5.60	0.25	5.76	0.91	1.60

Note: 1) Mn=2000 g/mol (PPCG). 2) Catalyst: Dibutyltin dilaurate.

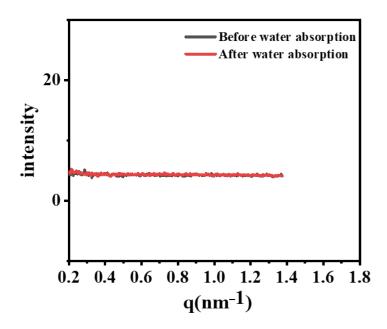


Figure S1. 1D-SAXS curves of the OU-3 films before and after water absorption.

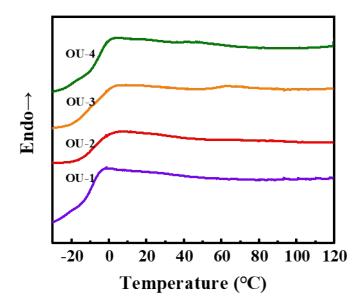


Figure S2. DSC curves of OUs $(1^{\sim}4)$ with various chain extender contents.

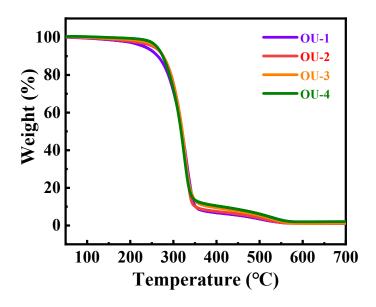


Figure S3. TGA curves of OUs (1^4) with various chain extender contents.

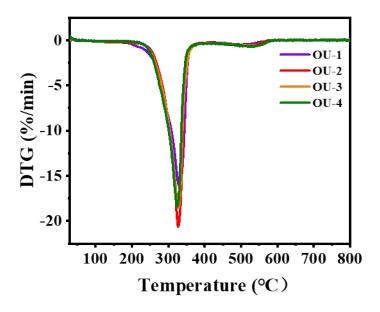


Figure S4. DTG curves of OUs (1 $^{\sim}$ 4) with various chain extender contents.

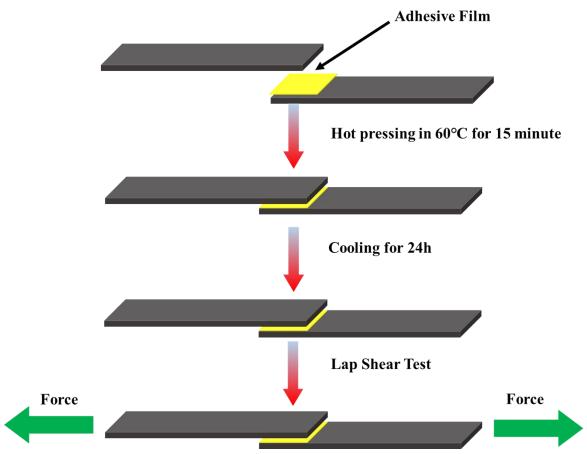
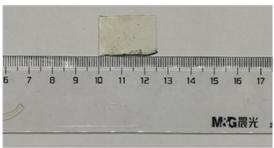


Figure S5. Schematic diagram of the adhesion and lap shear experiment procedure. The hot pressing is carried out at 60 $^{\circ}$ C.







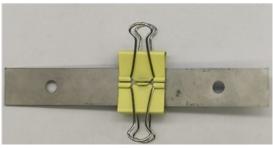


Figure S6. (a) A film of OU-3 for adhesion experiment. (b) Two pieces of stainless steel and OU-3 held together by two paper clips.

Table S2. Mechanical properties of OU samples

Name	Tensile strength	Elongation	Young's Modulus (MPa)	Fracture Toughness	
	(MPa)	(%)		(MJ/m³)	
OU-1	0.27±0.06	1079.3±28.2	0.65±0.05	2.05±0.53	
OU-2	0.36±0.10	980.8±19.2	1.40±0.03	2.41±0.12	
OU-3	2.45±0.06	317.4±19.7	9.70±1.52	6.43±0.68	
OU-4	2.23±0.22	270.7±18.7	6.71±1.17	5.43±0.09	

Table S3. Summary of the assignment of the deconvoluted subpeaks in the FTIR C=O absorption bands for the OU-1, OU-2, OU-3 and OU-4.

	Wavenumber (cm ⁻¹)				Area (%)				
Assignment		OU-1	OU-2	OU-3	OU-4	OU-1	OU-2	OU-3	OU-4
ν(C=O) urethane amide	Free	I (1747)	I (1742)	I (1743)	I (1742)	53.5%	55.4%	52.0%	49.26%
	H-bonded (Ordered)	II (1719)	II (1712)	II (1713)	II (1715)	12.1%	18.9%	12.5%	14.81%
v(C=O) urea amide	Free	III(1696)	III(1692)	III(1697)	III(1692)	12.0%	7.1%	2.9%	11.85%
	H-bonded (Disordered)	IV(1661)	IV(1672)	IV(1677)	IV (1661)	14.5%	8.6%	11.25%	16.30%
	H-bonded (Ordered)	V (1642)	V (1654)	V (1658)	V (1642)	4%	7.2%	5.83%	6.10%
v(C=O) amide	H-bonded (Ordered)	VI(1611)	VI(1642)	VI(1646)	VII(1608)	3.1%	2.6%	15.41%	1.50%
Total degree of H-bonded						34.4%	37.4%	45.0%	38.8%

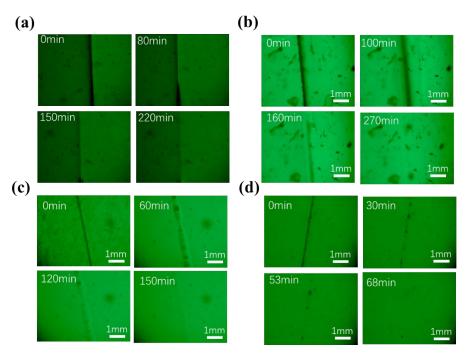


Figure S7. The self-healing processes of OU-3 at different temperatures. (a) 40 $^{\circ}$ C (b) 50 $^{\circ}$ C (c) 60 $^{\circ}$ C (d) 70 $^{\circ}$ C.

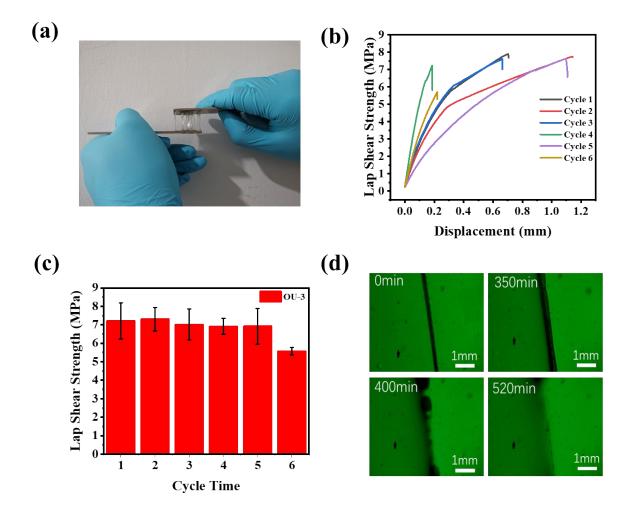


Figure S8. (a) Debonding of the two stainless steel plates at an elevated temperature. (b) Lap shear strength-displacement curves. (c) Lap shear strength of OU-3 in recycling test with stainless steel as matrix. (d) Optical microscopic images of a self-healing process of a cut-off OU-3 film at room temperature.

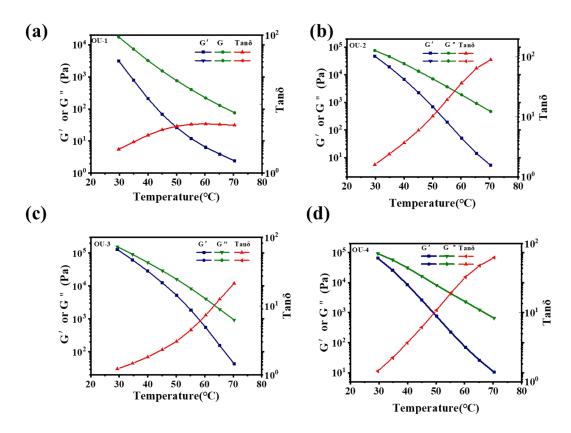


Figure S9. The rheological temperature sweep experiment from 70 $^{\circ}\mathrm{C}$ to 30 $^{\circ}\mathrm{C}$ of OUs 1-4.

 Table S4. Absorbance of aquatic gentamicin solutions with different concentrations

Sample	Concentration (mg/ml)	Absorbance (a.u)		
1	0.005	0.03157		
2	0.010	0.06236		
3	0.015	0.09722		
4	0.020	0.14012		

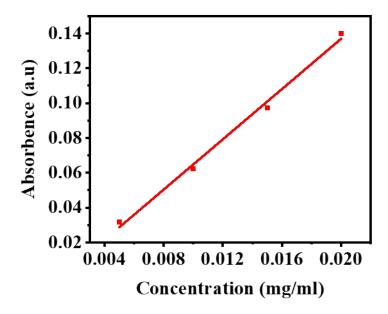


Figure \$10. The linear absorbance line of gentamicin solutions.

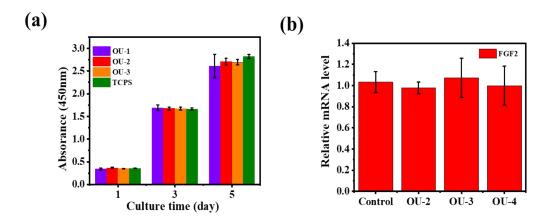


Figure S11. (a) The CCK-8 test of the OU films; (b) The FGF2 level of the OU films.