

Copper Catalyzed Cyano-Sulfonylation of Allenes via the Insertion of Sulfur Dioxide Toward the Synthesis of *(E)*- α -Cyanomethyl Vinylsulfones

Yue Li,[†] Xuemei Zhang,^{†*} and Zhong Lian^{†*}

[†]Department of Dermatology, State Key Laboratory of Biotherapy and Cancer Center, West China Hospital, Sichuan University, Chengdu 610041 (P.R. China)

Corresponding Author: Zhong Lian – Email: lianzhong@scu.edu.cn

Xuemei Zhang – Email: xuemeizhang@scu.edu.cn

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1. General information

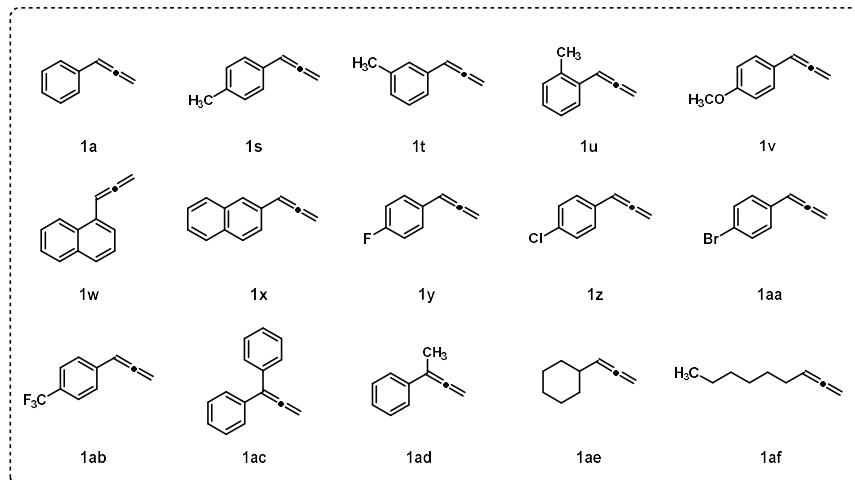
Unless otherwise noted, all reactions or reagents were obtained from commercial suppliers and used as received. Unless otherwise noted, all catalytic reactions were set up in an argon atmosphere glovebox (Vigor, SGI800-750TS-F). The substrates and reagents for catalytic reactions were degassed and stored in the glovebox, unless otherwise noted. All work-up and purification procedures were carried out with reagent-grade solvents in air.

Thin Layer Chromatography analysis was performed on silica gel coated glass plates (0.25 mm) with fluorescence indicator UV254. For detection of spots, irradiation of UV light at 254 nm or staining reagent using phosphomolybdic acid solution was used. Flash column chromatography was conducted with silica gel 60 (particle size 230–400 mesh, Huanghai) at room temperature and under elevated pressure.

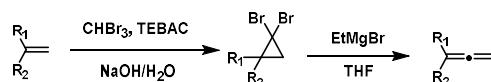
Gas chromatography (GC) analysis was conducted on a Shimadzu GC-2030 instrument equipped with a Rtx-5 column (30 m × 0.25 mm) with dodecane as an internal standard. GC-MS analysis was conducted on a Agilent 5977B GC/MSD instrument equipped with a HP-5MS UI column (30 m × 0.25 mm). ^1H NMR and ^{13}C NMR spectra were recorded at 400 MHz and 101 MHz, respectively in CDCl_3 (or $(\text{CD}_3)_2\text{SO}$) at room temperature. ^1H NMR was reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quadruplet, m = multiplet), coupling constant (J values) in Hz and integration. Chemical shifts (δ) were reported with respect to the corresponding solvent residual peak at 7.26 ppm for CDCl_3 (or 2.50 ppm for $(\text{CD}_3)_2\text{SO}$) for ^1H NMR. ^{13}C NMR spectra (^1H -broadband decoupled) were reported in ppm using the central peak of CDCl_3 (77.16 ppm). High-resolution mass spectrometric measurements were provided by the Department of The State Key Laboratory of Biotherapy, Sichuan University. The molecular ion $[\text{M}+\text{H}]^+$, $[\text{M}+\text{K}]^+$ and $[\text{M}+\text{Na}]^+$ are given in m/z units.

2. General procedure for the synthesis of Allenes

Allenes:



Method^[1]:



Step 1:

To a solution of alkene (1.0 equiv), bromoform (1.5 equiv) and BnNEt_3Cl (1 mol%) was added dropwise a solution of 50% NaOH, and the mixture was stirred at room temperature for 60 min, then heated to 60 °C and further stirred until conversion was complete as observed by TLC analysis. Water and DCM were added and the aqueous phase was extracted with DCM (3 x 50 mL). The combined organic phases were washed with saturated NaCl solution, dried over Na_2SO_4 and the solvent removed under reduced pressure. The reaction mixture was purified by column chromatography afforded dibromocyclopropane derivatives.

Step 2:

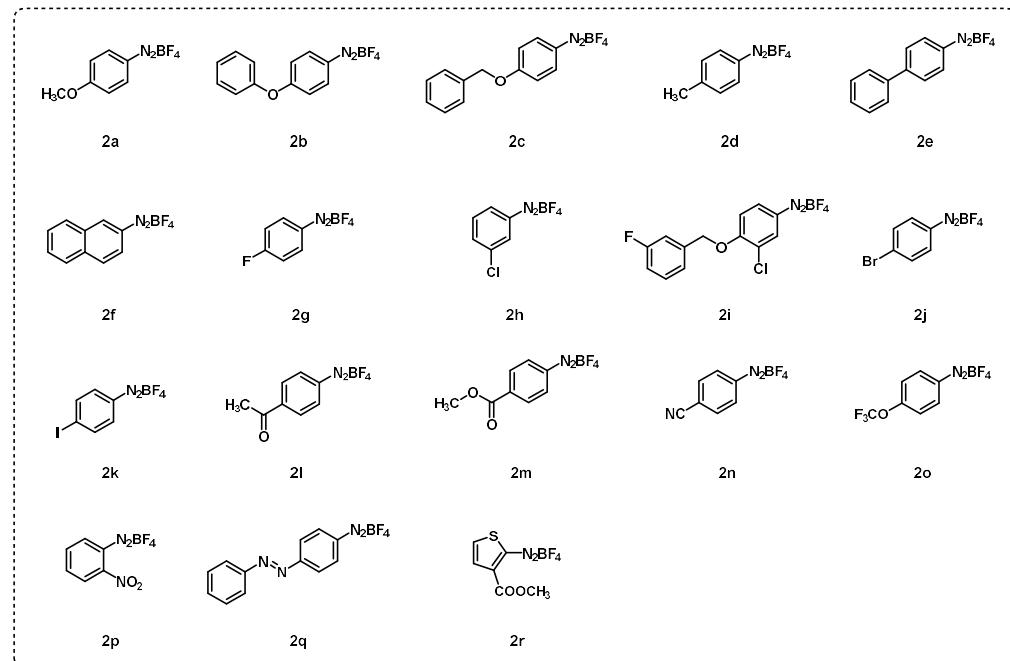
EtMgBr (3.0 M in ether, 1.5 equiv) was added dropwise to a pre-cooled (ice-bath) solution of dibromocyclopropane derivatives (1.0 equiv) in dry THF (1.0 mL/mmol) under nitrogen atmosphere. After EtMgBr was added the mixture was then slowly warmed to room temperature, and stirred at room temperature for an additional 2 hours. Then the reaction was quenched by HCl (0.5 N, 10ml) solution, water was added, and the mixture extracted with ether (3 x 50 mL). The combined organic layers were washed with brine, dried with anhydrous Na_2SO_4 and filtered.

After removing the solvent under reduced pressure, the crude product was purified by column chromatography on silica gel to afford allenes.

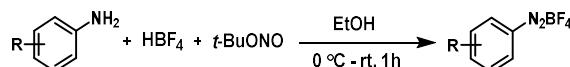
Spectral and physical data of **1a**^[1], **1s-1ad**^[1], **1ae**^[2], **1af**^[3]match literature reported values.

3. General procedure for the synthesis of aryldiazonium tetrafluoroborates

Aryldiazonium tetrafluoroborates:



Method A:

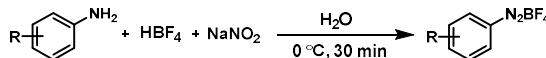


In a 50 mL round-bottom flask, the aniline (10 mmol, 1.0 equiv) was dissolved in a mixture of absolute ethanol (3.0 mL) and an aqueous solution of HBF₄ (48% aq, 2.5 mL, 13.6 mmol, 1.36 equiv), followed by dropwise addition of *t*-BuONO (2.7 mL, 23 mmol, 0.23 equiv) at 0 °C. After stirring at room temperature for 1 h, diethyl ether (20 mL) was added to precipitate the arenediazonium tetrafluoroborate. The solids were filtered off and washed with diethyl ether (3×10 mL), dried in vacuo (10⁻³ mbar) for 10 minutes, and stored in refrigerator under N₂.

atmosphere.

Arenediazonium salts **2a-2f**, **2l-2n**, **2p**, **2r** were synthesized in method A. Spectral and physical data of **2a**^[4], **2b-2f**^[8], **2m-2n**^[7], **2p**^[8], **2r**^[4] match literature reported values.

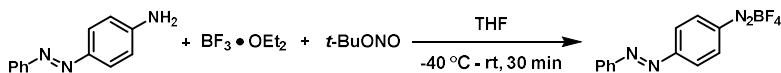
Method B:



In a 100 mL round-bottom flask, the aniline (20 mmol, 1.0 equiv) was dissolved in a mixture of H₂O (10 mL) and HBF₄ (48% aq, 6 mL, 32.7 mmol, 1.64 equiv). After stirring for 15 min, the solution of NaNO₂ (1.5 g, 22 mmol, 1.1 equiv., in 4.0 mL H₂O) was added dropwise at 0 °C. The mixture was stirred for another 30 min at 0 °C. Then, the arenediazonium tetrafluoroborate was removed by filtration and washed successively twice with Et₂O. The crude product was dried in vacuo for 20 min and was then directly used without further purification.

Arenediazonium salts **2g-2k**, **2o** were synthesized in method B. Spectral and physical data of **2g**^[5], **2h**^[5], **2i**^[5], **2j**^[5], **2k**^[5], **2o**^[6], match literature reported values.

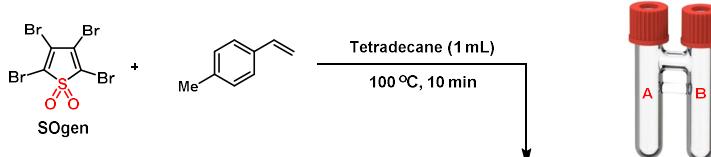
Method C:



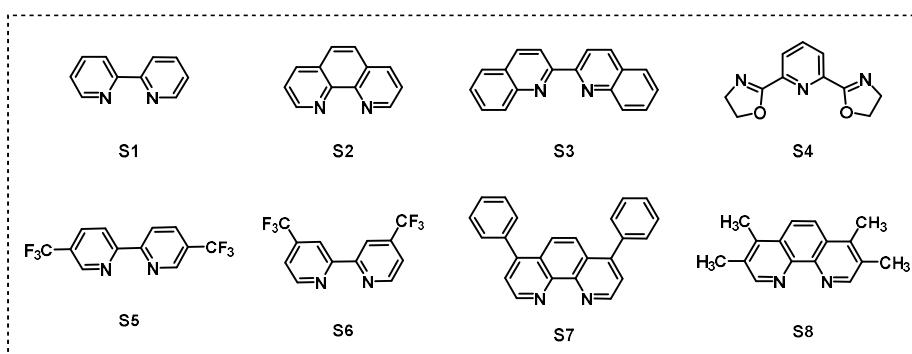
To a solution 4-aminoazobenzene (550 mg, 2.8 mmol, 1.0 equiv) in 22 mL of THF was added BF₃ · OEt₂ (1.19 g, 7.6 mmol, 2.7 equiv) at -40 °C. The mixture was stirred for 20 min and then *t*-BuONO (515 mg, 5.0 mmol, 1.8 equiv) was added dropwise. The mixture was stirred for 10 min and then gradually warmed to room temperature. Anhydrous ether was added to precipitate the diazonium salt. The mixture was filtered and the salt washed with ether copiously to give diazonium salt **2q** (385.2 mg, 1.32 mmol, 47% yield). Spectral and physical data of **2q** match literature reported values^[9].

4. Optimization of cyano-sulfonylation of allene derivatives

Chamber A: SO₂ gas generation



Chamber B: Regio- and Stereoselective Cyansulfonylation of Allene derivatives

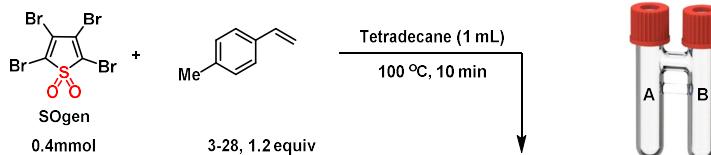


Entry	Deviation from standard conditions	Yield of 3a ^a
1	S1 instead of S2	58%
2	S2	85%
3	S3 instead of S2	26%
4	S4 instead of S2	67%
5	S5 instead of S2	57%
6	S6 instead of S2	44%
7	S7 instead of S2	36%
8	S8 instead of S2	52%

^aYields were determined by GC using dodecane as an internal standard.

5. General procedure for the synthesis of (*E*)- α - cyanomethyl vinylsulfones compounds

Chamber A: SO₂ gas generation

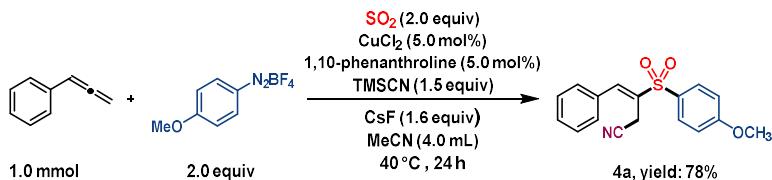


Chamber B: Regio- and Stereoselective Cyanosulfonylation of Allene derivatives



In the glovebox, Allenes (0.2 mmol, 1.0 equiv), aryldiazonium tetrafluoroborates (0.4 mmol, 2.0 equiv), CuCl₂ (1.3 mg, 5.0 mol%), 1,10-phenanthroline (1.8 mg, 5.0 mol%), TMSCN (37.5 μ L, 0.3 mmol), CsF (48.6 mg, 0.32 mmol) were added to chamber B, followed by addition of MeCN (2.5 mL). Tetrabromothiophene S,S-dioxides (0.4 mmol, 127.7 mg) in tetradecane (1.0 mL) was added to chamber A, followed by addition of 4-methylphenylene (56.6 μ L, 0.45 mmol). The two chamber system was sealed and removed out of the glovebox, then chamber A was heated to 100 °C in heat block. About ten minutes later, the sulfur dioxide was completely released since the system in chamber A became clear. Then chamber B was heated to 40 °C in heat block. After 16 hours, two chamber was cooled to room temperature. The mixture in chamber B was passed through a short silica gel pad with ethyl acetate. The filtrate was washed by ethyl acetate and H₂O (15 mL \times 3), dried by Na₂SO₄, then concentrated and the residue was purified by flash column chromatography to give the desired product.

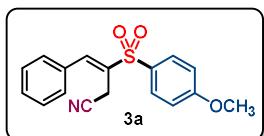
6. The procedure for the 1 mmol scale reaction



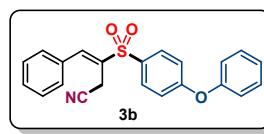
In the glovebox, Allenes (1.0 mmol, 1.0 equiv), aryldiazonium tetrafluoroborates (2.0

mmol, 2.0 equiv), CuCl₂ (6.5 mg, 5.0 mol%), 1,10-phenanthroline (9.0 mg, 5.0 mol%), TMSCN (187.5 μ L, 1.5 mmol), CsF (243 mg, 1.6 mmol) were added to chamber B, followed by addition of MeCN (4.0 mL). Tetrabromothiophene S,S-dioxides (2.0 mmol, 638.5 mg) in tetradecane (4.0 mL) was added to chamber A, followed by addition of 4-methylphenylene (283.0 μ L, 2.25 mmol). The two chamber system was sealed and removed out of the glovebox, then chamber A was heated to 100 °C in heat block. About ten minutes later, the sulfur dioxide was completely released since the system in chamber A became clear. Then chamber B was heated to 40 °C in heat block. After 16 hours, two chamber was cooled to room temperature. The mixture in chamber B was passed through a short silica gel pad with ethyl acetate. The filtrate was washed by ethyl acetate and H₂O (15 mL \times 3), dried by Na₂SO₄, then concentrated and the residue was purified by flash column chromatography to give the desired product.

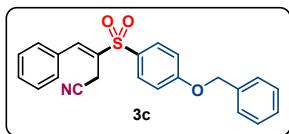
7. Characterization data of products



Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (5:1) as eluent (53.9 mg, 86%). Known compounds^[10]. ¹H NMR (400 MHz, Chloroform-d) δ 8.0 (s, 1H), 7.8 (d, J = 7.6 Hz, 2H), 7.4 – 7.3 (m, 5H), 7.0 (d, J = 7.6 Hz, 2H), 3.8 (s, 3H), 3.5 (s, 2H). ¹³C NMR (101 MHz, Chloroform-d) δ 164.3, 141.9, 132.1, 131.7, 130.7, 130.7, 129.6, 129.4, 129.3, 115.3, 115.0, 55.8, 16.3. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₇H₁₅NNaO₃S 336.0665; found: 336.0663.

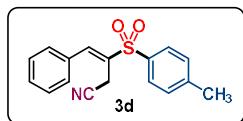


Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (8:1) as eluent (63.1mg, 84%). ¹H NMR (400 MHz, Chloroform-d) δ 8.08 (s, 1H), 7.94 (d, J = 8.9 Hz, 2H), 7.56 – 7.39 (m, 7H), 7.29 – 7.22 (m, 1H), 7.18 – 7.08 (m, 4H), 3.62 (s, 2H). ¹³C NMR (101 MHz, Chloroform-d) δ 163.1, 154.9, 142.2, 132.1, 131.50, 131.49, 130.84, 130.78, 130.3, 129.5, 129.4, 125.2, 120.4, 118.2, 115.3, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₂₂H₁₇NNaO₃S 398.0821; found: 398.0818.

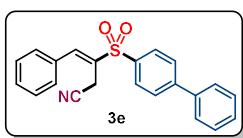


Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (8:1) as eluent (67.8 mg, 87%). ¹H NMR (400 MHz, Chloroform-d) δ 8.04 (s, 1H), 7.90 (d, J = 8.8 Hz, 2H), 7.53 – 7.32 (m, 10H), 7.14 (d, J = 8.9 Hz, 2H), 5.15 (s, 2H), 3.58 (s, 2H). ¹³C NMR (101 MHz, Chloroform-d) δ 163.5, 142.0, 135.6, 132.1,

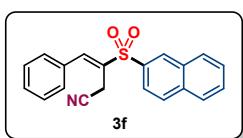
131.7, 130.7, 130.7, 129.8, 129.4, 129.3, 128.8, 128.5, 127.6, 115.8, 115.3, 70.5, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₂₃H₁₉NNaO₃S 412.0978; found: 412.0984.



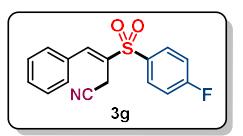
Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (7:1) as eluent (44.0 mg, 74%). ¹H NMR (400 MHz, Chloroform-d) δ 8.06 (s, 1H), 7.86 (d, *J* = 8.1 Hz, 2H), 7.51 – 7.38 (m, 7H), 3.58 (s, 2H), 2.46 (s, 3H). ¹³C NMR (101 MHz, Chloroform-d) δ 145.5, 142.4, 135.4, 132.1, 131.4, 130.8, 130.4, 129.5, 129.3, 128.5, 115.3, 21.8, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₇H₁₅NNaO₂S 320.0716; found: 320.0713.



Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (8:1) as eluent (56.1mg, 78%). ¹H NMR (400 MHz, Chloroform-d) δ 8.12 (s, 1H), 8.04 (d, *J* = 8.5 Hz, 2H), 7.81 (d, *J* = 8.5 Hz, 2H), 7.66 – 7.59 (m, 2H), 7.54 – 7.39 (m, 8H), 3.63 (s, 2H). ¹³C NMR (101 MHz, Chloroform-d) δ 147.3, 142.9, 139.0, 136.9, 132.0, 131.3, 130.9, 129.5, 129.4, 129.1, 129.0, 128.8, 128.3, 127.5, 115.3, 16.5. HRMS (ESI/Q-TOF) m/z: [M+H]⁺ calcd for C₂₂H₁₈NO₂S 360.1053; found: 360.1052.

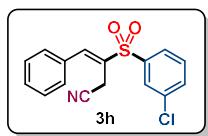


Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (8:1) as eluent (42.0mg, 63%). ¹H NMR (400 MHz, Chloroform-d) δ 8.42 (d, *J* = 2.6 Hz, 1H), 7.95 (d, *J* = 2.4 Hz, 1H), 7.84 (dd, *J* = 8.7, 2.8 Hz, 2H), 7.74 (d, *J* = 8.0 Hz, 1H), 7.69 – 7.65 (m, 1H), 7.53 – 7.42 (m, 2H), 7.30 – 7.24 (m, 4H), 7.06 (d, *J* = 2.5 Hz, 1H), 3.42 (s, 2H). ¹³C NMR (101 MHz, Chloroform-d) δ 143.1, 135.6, 135.1, 132.3, 132.0, 131.1, 130.9, 130.6, 130.2, 129.7, 129.6, 129.5, 129.4, 128.1, 128.0, 122.6, 115.2, 16.4. HRMS (ESI/Q-TOF) m/z: [M+H]⁺ calcd for C₂₀H₁₆NO₂S 334.0896; found: 334.0896.

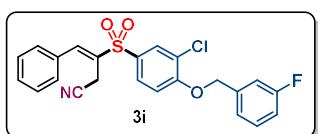


Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (5:1) as eluent (45.2 mg, 75%). ¹H NMR (400 MHz, Chloroform-d) δ 8.11 (s, 1H), 8.06 – 8.00 (m, 2H), 7.55 – 7.44 (m, 5H), 7.34 – 7.28 (m, 2H), 3.63 (s, 2H). ¹⁹F NMR (376 MHz, Chloroform-d) δ -102.00. ¹³C NMR (101 MHz, Chloroform-d) δ 166.2 (d, *J* = 257.8 Hz), 143.1, 134.5 (d, *J* = 3.1 Hz), 131.8, 131.4 (d, *J* = 9.8 Hz), 131.0, 131.0, 129.5, 129.4, 117.2 (d, *J* = 22.8

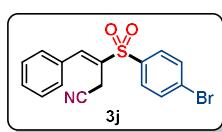
Hz), 115.1, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₆H₁₂FNNaO₂S 324.0465; found: 324.0461.



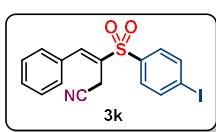
Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (8:1) as eluent (58.5mg, 92%). ¹H NMR (400 MHz, Chloroform-d) δ 8.10 (s, 1H), 7.96 (t, *J* = 1.9 Hz, 1H), 7.88 (dt, *J* = 7.8, 1.4 Hz, 1H), 7.68 – 7.64 (m, 1H), 7.59 – 7.53 (m, 1H), 7.52 – 7.44 (m, 5H), 3.61 (s, 2H). ¹³C NMR (101 MHz, Chloroform-d) δ 143.9, 140.3, 136.0, 134.5, 131.7, 131.1, 131.0, 130.5, 129.6, 129.4, 128.4, 126.6, 115.1, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₆H₁₂ClNNaO₂S 340.0169; found: 340.0168.



Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (5:1) as eluent (68.1mg, 77%). ¹H NMR (400 MHz, Chloroform-d) δ 8.06 (s, 1H), 7.99 (d, *J* = 2.3 Hz, 1H), 7.84 (dd, *J* = 8.7, 2.3 Hz, 1H), 7.53 – 7.42 (m, 5H), 7.41 – 7.34 (m, 1H), 7.24 – 7.15 (m, 2H), 7.10 (d, *J* = 8.7 Hz, 1H), 7.04 (td, *J* = 8.5, 2.5 Hz, 1H), 5.23 (s, 2H), 3.60 (s, 2H). ¹⁹F NMR (376 MHz, Chloroform-d) δ -112.08. ¹³C NMR (101 MHz, Chloroform-d) δ 163.0 (d, *J* = 246.9 Hz), 158.6, 142.9, 137.6 (d, *J* = 7.5 Hz), 131.9, 131.1 (d, *J* = 4.0 Hz), 130.9, 130.5, 130.4, 129.5, 129.4, 129.0, 124.8, 122.5 (d, *J* = 3.0 Hz), 115.5, 115.3 (d, *J* = 8.7 Hz), 114.1, 113.9, 113.5, 70.3 (d, *J* = 2.1 Hz), 16.4. HRMS (ESI/Q-TOF) m/z: [M+H]⁺ calcd for C₂₃H₁₇ClFNO₃S 442.0674; found: 442.0675.

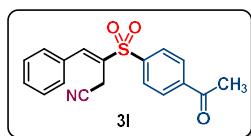


Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (8:1) as eluent (57.0 mg, 79%). ¹H NMR (400 MHz, Chloroform-d) δ 8.08 (s, 1H), 7.88 – 7.82 (m, 2H), 7.78 – 7.73 (m, 2H), 7.54 – 7.41 (m, 5H), 3.61 (s, 2H). ¹³C NMR (101 MHz, Chloroform-d) δ 143.6, 137.6, 133.1, 131.8, 131.1, 130.8, 129.87, 129.86, 129.5, 129.4, 115.1, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₆H₁₂BrNNaO₂S 383.9664; found: 383.9656.

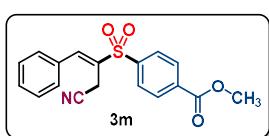


Prepared by the general procedure at the 40 °C; isolated as orange solid using petroleum/ethyl acetate (8:1) as eluent (52.4mg, 64%). ¹H NMR (400 MHz, Chloroform-d) δ 8.07 (s, 1H), 7.97 (d, *J* = 8.5 Hz, 2H), 7.68 (d, *J* = 8.4 Hz, 2H), 7.53 – 7.40 (m, 5H), 3.60 (s, 2H). ¹³C NMR (101 MHz, Chloroform-d) δ 143.6, 139.0, 138.3, 131.8, 131.1, 130.8, 129.6, 129.5, 129.4, 115.1,

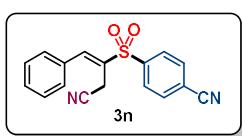
102.5, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₆H₁₂INNaO₂S 431.9526; found: 431.9525.



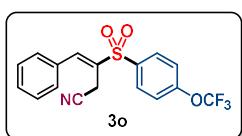
Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (5:1) as eluent (58.6 mg, 90%).
¹H NMR (400 MHz, Chloroform-d) δ 8.20 – 8.14 (m, 3H), 8.13 – 8.08 (m, 2H), 7.55 – 7.50 (m, 3H), 7.49 – 7.44 (m, 2H), 3.65 (s, 2H), 2.69 (s, 3H). ¹³C NMR (101 MHz, Chloroform-d) δ 196.5, 144.2, 142.5, 141.2, 131.7, 131.2, 130.5, 129.6, 129.5, 129.4, 128.8, 115.0, 27.0, 16.4. HRMS (ESI/Q-TOF) m/z: [M+H]⁺ calcd for C₁₈H₁₆NNaO₃S 348.0665; found: 348.0663.



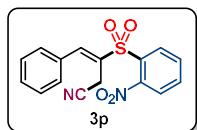
Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (3:1) as eluent (48.5 mg, 71%).
¹H NMR (400 MHz, Chloroform-d) δ 8.26 (d, *J* = 8.4 Hz, 2H), 8.12 (s, 1H), 8.06 (d, *J* = 8.4 Hz, 2H), 7.52 – 7.43 (m, 5H), 3.96 (s, 3H), 3.62 (s, 2H). ¹³C NMR (101 MHz, Chloroform-d) δ 165.3, 144.0, 142.5, 135.3, 131.8, 131.1, 130.8, 130.5, 129.6, 129.4, 128.5, 115.0, 52.8, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₈H₁₅NNaO₄S 364.0614; found: 364.0611.



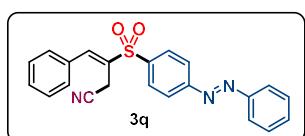
Prepared by the general procedure at the 40 °C; isolated as orange solid using petroleum/ethyl acetate (4:1) as eluent (41.9mg, 68%). ¹H NMR (400 MHz, Chloroform-d) δ 8.16 – 8.08 (m, 3H), 7.96 – 7.86 (m, 2H), 7.56 – 7.48 (m, 3H), 7.47 – 7.40 (m, 2H), 3.64 (s, 2H). ¹³C NMR (101 MHz, Chloroform-d) δ 145.0, 143.0, 133.4, 131.5, 131.4, 130.0, 129.6, 129.5, 129.1, 118.0, 117.0, 114.9, 16.5. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₇H₁₂N₂NaO₂S 331.0512; found: 331.0510.



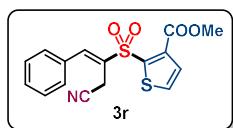
Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (8:1) as eluent (62.5mg, 85%). ¹H NMR (400 MHz, Chloroform-d) δ 8.10 (s, 1H), 8.04 (d, *J* = 8.9 Hz, 2H), 7.54 – 7.39 (m, 7H), 3.61 (s, 2H). ¹⁹F NMR (376 MHz, Chloroform-d) δ -57.64. ¹³C NMR (101 MHz, Chloroform-d) δ 153.5 (q, *J* = 2.0 Hz), 143.7, 136.8, 131.8, 131.1, 130.77, 130.75, 129.5, 129.4, 121.3, 120.2 (q, *J* = 260.1 Hz), 115.1, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₇H₁₂F₃NNaO₃S 390.0382; found: 390.0378.



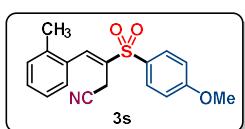
Prepared by the general procedure at the 40 °C; isolated as orange solid using petroleum/ethyl acetate (2:1) as eluent (46.6mg, 71%). ^1H NMR (400 MHz, Chloroform-*d*) δ 8.4 – 8.3 (m, 1H), 8.1 (s, 1H), 7.9 – 7.8 (m, 3H), 7.6 – 7.4 (m, 5H), 3.7 (s, 2H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 148.9, 145.3, 135.7, 132.8, 132.7, 131.8, 131.8, 131.0, 129.7, 129.5, 129.4, 125.4, 115.1, 16.5. HRMS (ESI/Q-TOF) m/z: [M+H]⁺ calcd for C₁₆H₁₃N₂O₄S 329.0591; found: 329.0596.



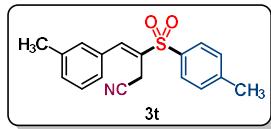
Prepared by the general procedure at the 40 °C; isolated as orange solid using petroleum/ethyl acetate (2:1) as eluent (46.5mg, 60%). ^1H NMR (400 MHz, Chloroform-*d*) δ 8.15 – 8.07 (m, 5H), 7.99 – 7.92 (m, 2H), 7.57 – 7.44 (m, 8H), 3.64 (s, 2H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 155.7, 152.4, 143.5, 139.7, 132.3, 131.9, 131.0, 130.9, 129.61, 129.56, 129.4, 129.3, 123.9, 123.4, 115.2, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₂₂H₁₇N₃NaO₂S 410.0934; found: 410.0930.



Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (3:1) as eluent (45.2mg, 65%). ^1H NMR (400 MHz, Chloroform-*d*) δ 8.22 (s, 1H), 7.76 (d, *J* = 5.3 Hz, 1H), 7.63 (d, *J* = 5.2 Hz, 1H), 7.54 – 7.42 (m, 5H), 3.87 (s, 3H), 3.67 (s, 2H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 159.3, 145.0, 141.4, 135.9, 132.2, 131.9, 130.7, 130.3, 129.5, 129.4, 129.3, 115.4, 53.1, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₆H₁₃NNaO₄S₂ 370.0178; found: 370.0174.

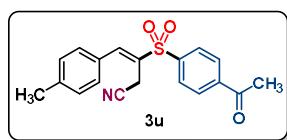


Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (10:1) as eluent (51.1 mg, 78%). ^1H NMR (400 MHz, Chloroform-*d*) δ 8.07 (s, 1H), 7.84 (d, *J* = 8.9 Hz, 2H), 7.31 – 7.11 (m, 4H), 7.00 (d, *J* = 8.9 Hz, 2H), 3.83 (s, 3H), 3.38 (s, 2H), 2.26 (s, 3H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 164.4, 141.8, 137.6, 133.0, 131.4, 130.9, 130.7, 130.3, 129.6, 127.8, 126.5, 115.3, 115.0, 55.8, 19.9, 16.1. HRMS (ESI/Q-TOF) m/z: [M+H]⁺ calcd for C₁₈H₁₈NO₃S 328.1002; found: 328.1004.

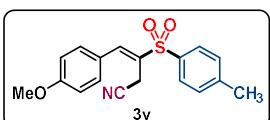


Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (8:1) as eluent (56.0 mg, 90%). ^1H NMR (400 MHz, Chloroform-*d*) δ 7.95 (s, 1H), 7.77 (d, *J* = 8.3

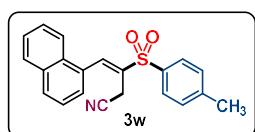
Hz, 2H), 7.32 (d, J = 8.3 Hz, 2H), 7.27 (d, J = 7.9 Hz, 1H), 7.22 – 7.12 (m, 3H), 3.50 (s, 2H), 2.38 (s, 3H), 2.31 (s, 3H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 145.5, 142.7, 139.2, 135.5, 132.0, 131.6, 131.1, 130.4, 130.2, 129.2, 128.4, 126.4, 115.4, 21.8, 21.4, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₈H₁₇NNaO₂S 334.0872; found: 334.0878.



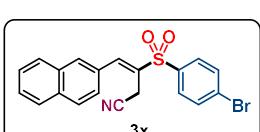
Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (5:1) as eluent (59.7mg, 88%). ^1H NMR (400 MHz, Chloroform-*d*) δ 8.14 (d, J = 8.1 Hz, 2H), 8.06 (d, J = 7.5 Hz, 3H), 7.35 (d, J = 8.0 Hz, 2H), 7.29 (d, J = 8.0 Hz, 2H), 3.63 (s, 2H), 2.65 (d, J = 1.1 Hz, 3H), 2.40 (s, 3H). ^{13}C NMR (151 MHz, Chloroform-*d*) δ 196.5, 144.1, 142.6, 142.0, 141.0, 130.1, 129.8, 129.3, 129.1, 128.9, 128.7, 115.1, 26.9, 21.5, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₉H₁₇NNaO₃S 362.0821; found: 362.0825.



Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (8:1) as eluent (39.3mg, 60%). ^1H NMR (400 MHz, Chloroform-*d*) δ 7.98 (s, 1H), 7.83 (d, J = 8.3 Hz, 2H), 7.45 – 7.35 (m, 4H), 6.99 (d, J = 8.8 Hz, 2H), 3.85 (s, 3H), 3.60 (s, 2H), 2.45 (s, 3H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 161.7, 145.3, 142.0, 135.8, 131.8, 130.3, 128.3, 128.2, 124.4, 115.4, 114.8, 55.5, 21.7, 16.4. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₈H₁₇NNaO₃S 350.0821; found: 350.0825.

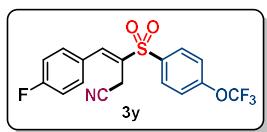


Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (10:1) as eluent (60.5 mg, 87%). ^1H NMR (400 MHz, Chloroform-*d*) δ 8.66 (s, 1H), 7.99 – 7.92 (m, 4H), 7.90 – 7.86 (m, 1H), 7.65 – 7.58 (m, 2H), 7.55 (t, J = 7.7 Hz, 1H), 7.48 (t, J = 7.2 Hz, 3H), 3.52 (s, 2H), 2.51 (s, 3H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 145.7, 141.6, 135.4, 134.1, 133.5, 131.1, 130.9, 130.5, 129.2, 128.9, 128.6, 127.5, 127.0, 126.6, 125.3, 123.9, 115.4, 21.8, 16.5. HRMS (ESI/Q-TOF) m/z: [M+H]⁺ calcd for C₂₁H₁₈NO₂S 348.1053; found: 348.1055.

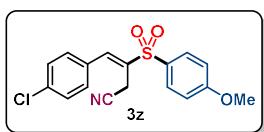


Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (8:1) as eluent (77.5 mg, 94%). ^1H NMR (400 MHz, Chloroform-*d*) δ 8.11 (s, 1H), 7.85 (d, J = 8.4 Hz, 2H), 7.78 – 7.70 (m, 3H), 7.62 (d, J = 7.3 Hz, 2H), 7.53 (d, J =

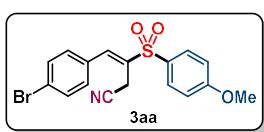
8.1 Hz, 2H), 7.51 – 7.38 (m, 2H), 3.66 (s, 2H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 143.9, 143.2, 139.5, 137.7, 133.1, 130.6, 130.28, 130.26, 129.9, 129.8, 129.1, 128.4, 128.0, 127.2, 115.2, 16.5. HRMS (ESI/Q-TOF) m/z: [M+H]⁺ calcd for C₂₀H₁₅BrNO₂S 412.0001; found: 412.0000.



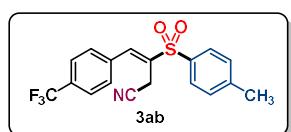
Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (5:1) as eluent (57.1mg, 74%). ^1H NMR (400 MHz, Chloroform-*d*) δ 8.05 (s, 1H), 8.04 – 8.01 (m, 2H), 7.51 – 7.39 (m, 4H), 7.24 – 7.14 (m, 2H), 3.60 (s, 2H). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -57.66, -107.08. ^{13}C NMR (101 MHz, Chloroform-*d*) δ 164.06 (d, *J* = 254.1 Hz), 153.51 (q, *J* = 1.8 Hz), 142.38, 136.63, 131.86 (d, *J* = 8.8 Hz), 130.75, 130.55 (d, *J* = 1.8 Hz), 127.90 (d, *J* = 3.5 Hz), 121.37, 120.18 (q, *J* = 260.1 Hz), 116.79 (d, *J* = 22.0 Hz), 114.93, 16.36. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₇H₁₁F₄NO₃S 408.0288; found: 408.0285.



Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (10:1) as eluent (55.0 mg, 79%). ^1H NMR (400 MHz, Chloroform-*d*) δ 7.98 (s, 1H), 7.88 (d, *J* = 8.5 Hz, 2H), 7.46 (d, *J* = 8.2 Hz, 2H), 7.37 (d, *J* = 8.2 Hz, 2H), 7.06 (d, *J* = 8.5 Hz, 2H), 3.89 (s, 3H), 3.55 (s, 2H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 164.4, 140.4, 137.0, 132.3, 130.8, 130.7, 130.5, 129.7, 129.3, 115.1, 55.8, 16.3. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₇H₁₄ClKNaO₃S 370.0275; found: 370.0271.

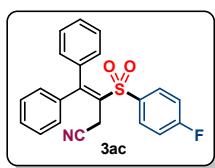


Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (10:1) as eluent (44.7 mg, 57%). ^1H NMR (400 MHz, Chloroform-*d*) δ 7.96 (s, 1H), 7.89 (d, *J* = 8.8 Hz, 2H), 7.62 (d, *J* = 8.3 Hz, 2H), 7.29 (d, *J* = 8.2 Hz, 2H), 7.06 (d, *J* = 8.8 Hz, 2H), 3.89 (s, 3H), 3.55 (s, 2H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 164.5, 140.5, 132.6, 132.5, 130.9, 130.8, 130.8, 129.2, 125.3, 115.1, 55.8, 16.3. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₇H₁₄BrNNaO₃S 413.9770; found: 413.9774.

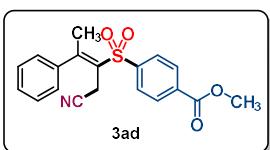


Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (8:1) as eluent (54.1 mg, 74%). ^1H NMR (400 MHz, Chloroform-*d*) δ 8.08 (s, 1H), 7.86 (d, *J* =

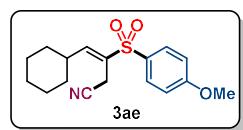
8.0 Hz, 2H), 7.75 (d, J = 8.1 Hz, 2H), 7.54 (d, J = 8.0 Hz, 2H), 7.43 (d, J = 8.0 Hz, 2H), 3.54 (s, 2H), 2.48 (s, 3H). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -63.06. ^{13}C NMR (101 MHz, Chloroform-*d*) δ 146.0, 140.6, 135.5, 134.8, 134.1, 132.3 (q, J = 33.0 Hz), 130.5, 129.5, 128.6, 126.3 (q, J = 3.7 Hz), 123.5 (q, J = 272.5 Hz), 114.8, 21.8, 16.2. HRMS (ESI/Q-TOF) m/z: [M+H]⁺ calcd for C₁₈H₁₅F₃NO₂S 366.0770; found: 366.0770.



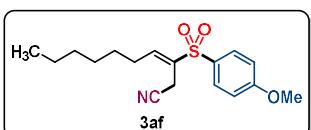
Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (8:1) as eluent (48.3 mg, 64%). ^1H NMR (400 MHz, Chloroform-*d*) δ 7.40 – 7.32 (m, 2H), 7.35 – 7.26 (m, 2H), 7.24 – 7.16 (m, 2H), 7.15 – 7.06 (m, 4H), 6.91 – 6.81 (m, 4H), 3.58 (s, 2H). ^{19}F NMR (376 MHz, Chloroform-*d*) δ -103.84. ^{13}C NMR (101 MHz, Chloroform-*d*) δ 165.3 (d, J = 256.3 Hz), 157.6, 138.2 (d, J = 219.1 Hz), 135.7 (d, J = 3.2 Hz), 133.2, 130.6 (d, J = 9.6 Hz), 129.7, 129.3, 129.2, 128.9, 128.0, 127.7, 117.6, 116.0, 115.8, 20.7. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₂₂H₁₆FNNaO₂S 400.0778; found: 400.0774.



Prepared by the general procedure at the 40 °C; isolated as white solid using petroleum/ethyl acetate (5:1) as eluent (44.8 mg, 63%). ^1H NMR (400 MHz, Chloroform-*d*) δ 8.28 (d, J = 8.3 Hz, 2H), 8.18 (d, J = 8.5 Hz, 2H), 7.48 – 7.35 (m, 3H), 7.15 (d, J = 6.6 Hz, 2H), 3.98 (s, 3H), 3.44 (s, 2H), 2.38 (s, 3H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 165.4, 158.0, 144.5, 140.6, 135.0, 130.6, 129.6, 129.4, 129.3, 127.7, 125.8, 116.9, 52.8, 23.5, 20.3. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₉H₁₇NNaO₄S 378.0770; found: 378.0777.



Prepared by the general procedure at the 40 °C; isolated as light yellow liquid using petroleum/ethyl acetate (8:1) as eluent (34.5 mg, 54%). ^1H NMR (400 MHz, Chloroform-*d*) δ 7.78 (d, J = 8.8 Hz, 2H), 7.02 (d, J = 8.8 Hz, 2H), 6.96 (d, J = 10.4 Hz, 1H), 3.87 (s, 3H), 3.36 (s, 2H), 2.39 – 2.28 (m, 1H), 1.82 – 1.68 (m, 5H), 1.36 – 1.20 (m, 5H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 164.1, 150.3, 130.4, 129.9, 129.7, 115.3, 114.9, 55.7, 38.5, 31.2, 25.4, 25.1, 14.6. HRMS (ESI/Q-TOF) m/z: [M+Na]⁺ calcd for C₁₇H₂₁NNaO₃S 342.1134; found: 342.1136.

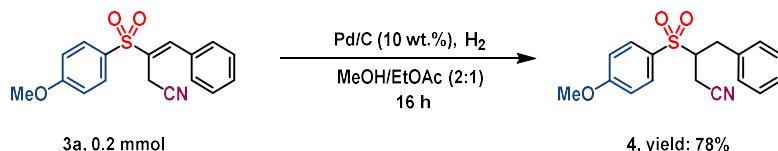


Prepared by the general procedure at the 40 °C; isolated as light yellow liquid using petroleum/ethyl acetate (8:1) as eluent (32.8 mg, 51%). ^1H NMR (400 MHz, Chloroform-*d*) δ 7.73 (d, J = 8.9 Hz, 2H), 7.06 (t, J = 7.6 Hz, 1H), 6.96 (d, J = 8.9 Hz,

2H), 3.81 (s, 3H), 3.30 (s, 2H), 2.23 (q, $J = 7.5$ Hz, 2H), 1.49 (s, 3H), 1.33 – 1.15 (m, 8H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 164.1, 146.3, 131.7, 130.4, 129.7, 114.9, 114.9, 55.7, 31.5, 29.0, 28.9, 27.9, 22.5, 14.6, 14.0. HRMS (ESI/Q-TOF) m/z: [M+H]⁺ calcd for C₁₇H₂₄NO₃S 322.1472; found: 322.1470.

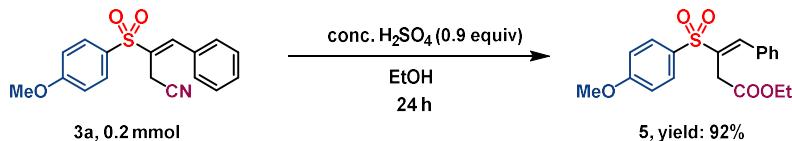
8. Transformations of (*E*)- α -cyanomethyl vinylsulfones compounds.

1) ^[11]



Compound **3a** (0.2 mmol, 62.7 mg) was dissolved in MeOH (2.5 mL) and EtOAc (1.25 mL) and Pd/C (10 wt. %, 20.9 mg, 0.02 mmol) was added. The reaction mixture was stirred under a hydrogen atmosphere for 12 h at room temperature. The reaction was filtered through celite, and the cake was washed with CH₂Cl₂ (2 × 5 mL). The organic layers were combined and concentrated in vacuo. The crude product was purified by flash column chromatography using petroleum/ethyl acetate (10:1) as eluent to afford **4** as a colorless liquid (49.2 mg, 78%). ^1H NMR (400 MHz, Chloroform-*d*) δ 7.91 (d, $J = 8.1$ Hz, 2H), 7.38 – 7.26 (m, 3H), 7.20 (d, $J = 7.3$ Hz, 2H), 7.10 (d, $J = 8.2$ Hz, 2H), 3.93 (d, $J = 1.4$ Hz, 3H), 3.54 – 3.37 (m, 2H), 2.90 (dd, $J = 13.4, 11.2$ Hz, 1H), 2.80 (dd, $J = 17.7, 4.2$ Hz, 1H), 2.61 (dd, $J = 17.7, 6.0$ Hz, 1H). ^{13}C NMR (101 MHz, Chloroform-*d*) δ 164.6, 135.1, 131.5, 129.2, 129.0, 127.7, 127.4, 115.4, 114.9, 62.3, 55.8, 33.5, 16.5. HRMS (ESI/Q-TOF) m/z: [M+H]⁺ calcd for C₁₇H₁₇NNaO₃S 338.0821; found: 338.0824.

2) ^[12]

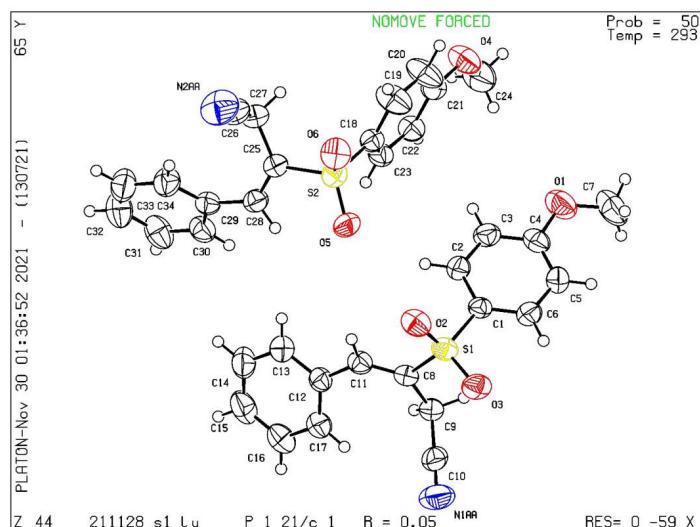


Compound **3a** (0.2 mmol, 62.7 mg) was dissolved in EtOH (2 ml) and conc. H₂SO₄ (0.18 mmol, 10 μ L) was added. The reaction mixture was stirred under a hydrogen atmosphere for 24h at 100 °C. The filtrate was washed by ethyl acetate and H₂O (15 mL×3), dried by Na₂SO₄, then concentrated and the residue was purified by flash column chromatography using petroleum/ethyl acetate (3:1) as eluent to afford **5** as a colorless liquid (66.3 mg, 92%). ^1H NMR

(400 MHz, Chloroform-*d*) δ 8.00 (s, 1H), 7.85 (d, J = 8.7 Hz, 2H), 7.49 – 7.37 (m, 5H), 7.00 (d, J = 8.8 Hz, 2H), 3.96 (q, J = 7.1 Hz, 2H), 3.87 (s, 3H), 3.52 (s, 2H), 1.15 (t, J = 7.1 Hz, 3H). ^{13}C NMR (151 MHz, Chloroform-*d*) δ 168.9, 163.7, 140.6, 135.5, 133.2, 130.7, 130.4, 129.8, 129.0, 128.8, 114.4, 61.4, 55.7, 33.0, 13.9. HRMS (ESI/Q-TOF) m/z: [M+H]⁺ calcd for C₁₉H₂₁O₅S 361.1104; found: 361.1105.

9. Crystal structure and X-ray crystallographic data of 3a.

A view of the molecular structure of compound 3a (CCDC deposition number: 2158587.)



Compound 3a was crystallized by slow evaporation of PE:EA=10:1.

Experimental

The crystal was kept at 293.15 K during data collection. Using Olex2^[13], the structure was solved with the Superflip^[14] structure solution program using Charge Flipping and refined with the ShelXL^[15] refinement package using Least Squares minimisation.

Table 1 Crystal data and structure refinement for 3a.

Identification code	211128_s1_ly
Empirical formula	C ₁₇ H ₁₅ NO ₃ S

Formula weight	313.36
Temperature/K	293.15
Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	5.1606(4)
b/Å	42.233(3)
c/Å	14.3998(11)
α/°	90
β/°	97.795(6)
γ/°	90
Volume/Å ³	3109.4(4)
Z	8
ρ _{calc} g/cm ³	1.339
μ/mm ⁻¹	0.220
F(000)	1312.0
Crystal size/mm ³	0.35 × 0.3 × 0.25
Radiation	MoKα ($\lambda = 0.71073$)
2Θ range for data collection/°	5.792 to 52.74
Index ranges	-6 ≤ h ≤ 5, -52 ≤ k ≤ 34, -13 ≤ l ≤ 17
Reflections collected	13163
Independent reflections	6234 [R _{int} = 0.0343, R _{sigma} = 0.0495]
Data/restraints/parameters	6234/0/399
Goodness-of-fit on F ²	1.041
Final R indexes [I>=2σ (I)]	R ₁ = 0.0492, wR ₂ = 0.1108
Final R indexes [all data]	R ₁ = 0.0726, wR ₂ = 0.1262
Largest diff. peak/hole / e Å ⁻³	0.21/-0.37

Table 2 Fractional Atomic Coordinates ($× 10^4$) and Equivalent Isotropic Displacement Parameters (Å $^2 × 10^3$) for 3a. Ueq is defined as 1/3 of the trace of the orthogonalised UIJ tensor.

Atom	x	y	z	U(eq)
S1	9249.1(10)	3156.5(2)	819.2(4)	45.63(17)

O1	2603(4)	3623.1(4)	-2523.7(12)	77.3(5)
O2	10249(3)	3430.6(4)	1335.3(11)	57.8(4)
O3	11064(3)	2928.7(4)	529.3(11)	59.2(4)
C1	7202(4)	3279.3(5)	-188.6(14)	40.4(5)
C2	5202(4)	3493.0(5)	-113.3(15)	48.2(6)
C3	3696(4)	3596.1(5)	-912.8(17)	53.6(6)
C4	4178(5)	3494.3(5)	-1782.1(16)	52.5(6)
C5	6133(5)	3281.7(6)	-1853.4(16)	57.7(6)
C6	7647(5)	3172.3(5)	-1050.4(16)	55.2(6)
C7	3034(6)	3531.3(8)	-3433.5(18)	95.2(11)
C8	7199(4)	2946.8(5)	1483.4(14)	40.0(5)
C9	6329(4)	2630.0(5)	1079.7(15)	46.0(5)
C10	8152(5)	2376.4(6)	1373.4(17)	57.6(6)
C11	6533(4)	3085.1(5)	2244.9(14)	45.7(5)
C12	4926(4)	2962.8(5)	2924.9(14)	45.0(5)
C13	3613(5)	3178.1(6)	3413.8(16)	62.6(7)
C14	2033(6)	3076.7(8)	4059.8(18)	77.9(8)
C15	1778(5)	2761.5(8)	4223.2(18)	73.7(8)
C16	3089(5)	2544.7(7)	3757.1(17)	66.6(7)
C17	4663(5)	2643.7(6)	3117.9(16)	56.2(6)
N1	9572(5)	2177.3(6)	1610.3(19)	96.4(9)
S2	4806.9(10)	4326.0(2)	1801.7(4)	48.22(17)
O4	-2129(4)	4555.6(4)	-1702.9(12)	71.2(5)
O5	5526(3)	3996.5(4)	1837.7(11)	58.8(4)
O6	6819(3)	4562.3(4)	1858.3(11)	61.4(5)
C18	2662(4)	4392.8(5)	770.5(15)	42.9(5)
C19	3001(5)	4650.0(7)	215.7(19)	74.4(8)
C20	1383(5)	4697.7(7)	-605(2)	78.4(9)

C21	-616(4)	4488.8(5)	-879.5(16)	51.0(6)
C22	-987(4)	4233.1(5)	-324.8(16)	50.4(6)
C23	653(4)	4186.0(5)	500.3(15)	47.8(5)
C24	-4048(6)	4330.2(7)	-2071(2)	85.4(9)
C25	2932(4)	4403.5(5)	2713.7(15)	42.6(5)
C26	2361(4)	4748.6(5)	2851.3(17)	50.8(6)
C27	4526(5)	4917.0(6)	3388.2(19)	61.6(7)
C28	2096(4)	4159.9(5)	3165.7(15)	47.4(5)
C29	585(4)	4155.9(5)	3951.7(15)	46.0(5)
C30	-1138(5)	3907.3(6)	4017.1(16)	55.5(6)
C31	-2634(5)	3894.4(7)	4725.5(18)	71.6(8)
C32	-2440(6)	4124.9(7)	5393(2)	77.4(8)
C33	-678(6)	4367.3(7)	5360.6(19)	75.7(8)
C34	824(5)	4383.5(6)	4653.4(18)	62.5(7)
N2	6207(6)	5049.7(6)	3799(2)	96.5(8)

Table 3 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 3a. The Anisotropic displacement factor exponent takes the form: - $2\pi^2[h^2a^{*2}\mathbf{U}_{11}+2hka^*\mathbf{b}^*\mathbf{U}_{12}+\dots]$.

Atom	\mathbf{U}_{11}	\mathbf{U}_{22}	\mathbf{U}_{33}	\mathbf{U}_{23}	\mathbf{U}_{13}	\mathbf{U}_{12}
S1	39.8(3)	44.6(3)	52.6(3)	7.9(3)	6.6(2)	-2.1(3)
O1	91.0(13)	76.1(13)	59.0(11)	13.0(10)	-10.8(9)	1.2(11)
O2	57.8(9)	51.0(10)	62.1(10)	5.5(8)	-1.0(8)	-18.7(8)
O3	42.6(8)	62.0(10)	75.2(11)	10.7(9)	15.7(8)	13.3(8)
C1	40.7(11)	36.0(11)	45.5(12)	5.6(9)	9.1(9)	0.1(10)
C2	51.5(13)	46.4(13)	48.7(13)	1.1(11)	13.7(10)	3.9(11)
C3	51.1(13)	48.8(14)	60.9(15)	3.7(12)	8.2(11)	10.4(12)
C4	59.6(14)	46.0(14)	49.8(14)	9.5(11)	0.4(11)	-7.5(12)

C5	76.3(16)	54.6(15)	43.9(13)	-2.9(12)	13.9(11)	2.4(14)
C6	62.2(15)	48.3(14)	57.9(15)	2.8(12)	18.7(12)	10.1(12)
C7	127(3)	101(2)	51.5(18)	14.9(17)	-8.3(17)	-22(2)
C8	38.6(11)	35.9(11)	44.1(12)	5.1(9)	0.4(9)	-1.2(10)
C9	51.2(12)	39.6(12)	47.6(13)	2.1(10)	7.7(10)	-0.4(11)
C10	70.0(16)	43.5(14)	64.8(16)	2.9(12)	28.4(13)	3.4(13)
C11	49.8(12)	40.0(12)	45.8(12)	1.3(10)	1.1(10)	-5.6(11)
C12	46.0(12)	49.3(14)	38.3(12)	-0.6(10)	0.8(9)	0.5(11)
C13	80.6(17)	59.5(16)	48.2(14)	-1.3(12)	10.0(12)	10.4(14)
C14	89(2)	98(2)	49.4(16)	0.3(16)	19.9(14)	28.0(19)
C15	62.6(16)	109(2)	52.3(16)	18.7(17)	15.5(12)	8.7(18)
C16	69.4(16)	73.5(18)	57.9(16)	15.7(14)	12.9(13)	-7.2(15)
C17	62.6(14)	56.0(15)	52.5(14)	7.2(12)	16.6(11)	1.2(13)
N1	112(2)	67.0(17)	117(2)	22.3(15)	41.3(17)	37.6(16)
S2	39.6(3)	52.1(4)	52.9(4)	-2.2(3)	6.1(2)	1.5(3)
O4	83.2(12)	62.5(11)	61.5(11)	14.2(9)	-13.0(9)	-15.2(10)
O5	61.4(10)	53.8(10)	61.0(10)	-2.9(8)	7.9(8)	18.2(9)
O6	38.1(8)	73.7(12)	72.0(11)	-4.5(9)	5.9(7)	-13.5(8)
C18	41.6(11)	39.5(12)	48.6(12)	0.6(10)	9.6(9)	-1.8(10)
C19	71.1(18)	66.1(18)	80.3(19)	19.0(15)	-9.8(14)	-33.6(15)
C20	85.2(19)	66.0(18)	77.6(19)	30.2(15)	-12.3(15)	-34.1(16)
C21	55.2(13)	45.9(14)	51.2(14)	2.5(11)	4.9(11)	2.1(12)
C22	49.8(13)	45.7(13)	55.2(14)	0.4(11)	5.2(10)	-10.1(11)
C23	52.7(13)	41.4(12)	49.6(13)	3.8(11)	8.6(10)	-5.4(11)
C24	104(2)	76(2)	66.9(19)	-0.8(15)	-21.7(16)	-19.5(19)
C25	37.7(11)	41.1(12)	47.8(12)	-2.9(10)	1.4(9)	0.1(10)
C26	48.7(13)	41.0(13)	62.6(15)	0.7(11)	6.7(11)	1.0(11)
C27	65.3(16)	46.2(15)	72.5(18)	-6.2(13)	6.9(13)	-6.5(14)

C28	50.2(13)	40.9(13)	50.1(13)	-2.7(11)	3.5(10)	2.8(11)
C29	45.0(12)	44.8(13)	47.0(13)	3.1(10)	2.4(9)	2.5(11)
C30	65.9(15)	51.4(14)	47.0(13)	3.1(11)	-0.2(11)	-8.3(13)
C31	70.8(17)	85(2)	58.5(16)	10.2(15)	7.8(13)	-24.7(16)
C32	75.9(18)	98(2)	62.4(18)	-3.4(17)	23.0(14)	-12.5(18)
C33	95(2)	75(2)	60.8(17)	-14.3(15)	23.3(15)	-7.0(18)
C34	71.4(16)	56.2(16)	61.4(16)	-7.1(13)	14.8(13)	-10.9(14)
N2	96.5(19)	75.5(18)	112(2)	-21.9(16)	-5.9(16)	-22.0(16)

Table 4 Bond Lengths for 3a.

Atom	Atom	Length/Å	Atom	Atom	Length/Å
S1	O2	1.4334(16)	S2	O5	1.4393(16)
S1	O3	1.4432(15)	S2	O6	1.4344(16)
S1	C1	1.754(2)	S2	C18	1.751(2)
S1	C8	1.759(2)	S2	C25	1.764(2)
O1	C4	1.364(3)	O4	C21	1.358(3)
O1	C7	1.413(3)	O4	C24	1.423(3)
C1	C2	1.386(3)	C18	C19	1.374(3)
C1	C6	1.369(3)	C18	C23	1.371(3)
C2	C3	1.370(3)	C19	C20	1.366(3)
C3	C4	1.378(3)	C20	C21	1.374(3)
C4	C5	1.365(3)	C21	C22	1.372(3)
C5	C6	1.384(3)	C22	C23	1.376(3)
C8	C9	1.503(3)	C25	C26	1.505(3)
C8	C11	1.328(3)	C25	C28	1.321(3)
C9	C10	1.450(3)	C26	C27	1.455(3)
C10	N1	1.137(3)	C27	N2	1.130(3)
C11	C12	1.461(3)	C28	C29	1.459(3)

C12	C13	1.382(3)	C29	C30	1.387(3)
C12	C17	1.387(3)	C29	C34	1.388(3)
C13	C14	1.386(4)	C30	C31	1.361(3)
C14	C15	1.361(4)	C31	C32	1.362(4)
C15	C16	1.367(4)	C32	C33	1.374(4)
C16	C17	1.373(3)	C33	C34	1.363(3)

Table 5 Bond Angles for 3a.

Atom	Atom	Atom	Angle/ $^{\circ}$	Atom	Atom	Atom	Angle/ $^{\circ}$
O2	S1	O3	119.08(10)	O5	S2	C18	108.46(10)
O2	S1	C1	108.93(10)	O5	S2	C25	108.50(10)
O2	S1	C8	109.00(10)	O6	S2	O5	119.31(10)
O3	S1	C1	107.42(10)	O6	S2	C18	107.79(10)
O3	S1	C8	106.48(10)	O6	S2	C25	107.12(10)
C1	S1	C8	105.05(9)	C18	S2	C25	104.72(10)
C4	O1	C7	117.7(2)	C21	O4	C24	118.37(18)
C2	C1	S1	119.94(16)	C19	C18	S2	120.32(17)
C6	C1	S1	119.75(16)	C23	C18	S2	120.32(16)
C6	C1	C2	120.3(2)	C23	C18	C19	119.3(2)
C3	C2	C1	119.0(2)	C20	C19	C18	120.5(2)
C2	C3	C4	120.8(2)	C19	C20	C21	120.1(2)
O1	C4	C3	115.1(2)	O4	C21	C20	116.0(2)
O1	C4	C5	124.9(2)	O4	C21	C22	124.2(2)
C5	C4	C3	120.0(2)	C22	C21	C20	119.8(2)
C4	C5	C6	119.8(2)	C21	C22	C23	119.9(2)
C1	C6	C5	120.1(2)	C18	C23	C22	120.4(2)
C9	C8	S1	114.08(15)	C26	C25	S2	114.49(16)
C11	C8	S1	118.01(16)	C28	C25	S2	118.12(17)
C11	C8	C9	127.87(19)	C28	C25	C26	127.3(2)
C10	C9	C8	113.38(18)	C27	C26	C25	113.16(19)
N1	C10	C9	179.4(3)	N2	C27	C26	179.4(3)
C8	C11	C12	129.2(2)	C25	C28	C29	129.5(2)
C13	C12	C11	118.1(2)	C30	C29	C28	118.4(2)

C13	C12	C17	118.0(2)	C30	C29	C34	118.0(2)
C17	C12	C11	123.9(2)	C34	C29	C28	123.5(2)
C12	C13	C14	120.8(3)	C31	C30	C29	121.0(2)
C15	C14	C13	119.9(3)	C30	C31	C32	120.4(3)
C14	C15	C16	120.3(3)	C31	C32	C33	119.6(3)
C15	C16	C17	120.2(3)	C34	C33	C32	120.6(3)
C16	C17	C12	120.9(2)	C33	C34	C29	120.4(2)

Table 6 Torsion Angles for 3a.

A	B	C	D	Angle/ [°]	A	B	C	D	Angle/ [°]
S1	C1	C2	C3	177.27(17)	S2	C18	C19	C20	177.9(2)
S1	C1	C6	C5	-176.38(17)	S2	C18	C23	C22	-178.02(17)
S1	C8	C9	C10	86.9(2)	S2	C25	C26	C27	-80.7(2)
S1	C8	C11	C12	-178.44(17)	S2	C25	C28	C29	178.39(18)
O1	C4	C5	C6	178.5(2)	O4	C21	C22	C23	-179.9(2)
O2	S1	C1	C2	-51.63(19)	O5	S2	C18	C19	-134.4(2)
O2	S1	C1	C6	125.99(18)	O5	S2	C18	C23	44.5(2)
O2	S1	C8	C9	-171.75(14)	O5	S2	C25	C26	171.67(15)
O2	S1	C8	C11	10.45(19)	O5	S2	C25	C28	-11.2(2)
O3	S1	C1	C2	178.11(16)	O6	S2	C18	C19	-3.9(2)
O3	S1	C1	C6	-4.3(2)	O6	S2	C18	C23	174.95(17)
O3	S1	C8	C9	-42.11(17)	O6	S2	C25	C26	41.64(18)
O3	S1	C8	C11	140.09(17)	O6	S2	C25	C28	-141.28(17)
C1	S1	C8	C9	71.64(16)	C18	S2	C25	C26	-72.66(17)
C1	S1	C8	C11	-106.16(18)	C18	S2	C25	C28	104.43(19)
C1	C2	C3	C4	-1.1(3)	C18	C19	C20	C21	0.4(5)
C2	C1	C6	C5	1.2(3)	C19	C18	C23	C22	0.9(3)
C2	C3	C4	O1	-177.7(2)	C19	C20	C21	O4	179.8(3)
C2	C3	C4	C5	1.7(4)	C19	C20	C21	C22	0.3(4)

C3	C4	C5	C6	-0.8(4)	C20	C21	C22	C23	-0.4(4)
C4	C5	C6	C1	-0.6(4)	C21	C22	C23	C18	-0.2(3)
C6	C1	C2	C3	-0.3(3)	C23	C18	C19	C20	-1.0(4)
C7	O1	C4	C3	179.0(2)	C24	O4	C21	C20	173.4(3)
C7	O1	C4	C5	-0.3(4)	C24	O4	C21	C22	-7.1(4)
C8	S1	C1	C2	65.03(19)	C25	S2	C18	C19	109.9(2)
C8	S1	C1	C6	-117.35(18)	C25	S2	C18	C23	-71.22(19)
C8	C11	C12	C13	-153.5(2)	C25	C28	C29	C30	148.0(2)
C8	C11	C12	C17	27.3(4)	C25	C28	C29	C34	-33.7(4)
C9	C8	C11	C12	4.1(4)	C26	C25	C28	C29	-5.0(4)
C11	C8	C9	C10	-95.6(3)	C28	C25	C26	C27	102.6(3)
C11	C12	C13	C14	179.2(2)	C28	C29	C30	C31	-178.8(2)
C11	C12	C17	C16	-179.1(2)	C28	C29	C34	C33	179.1(2)
C12	C13	C14	C15	0.5(4)	C29	C30	C31	C32	-0.8(4)
C13	C12	C17	C16	1.7(4)	C30	C29	C34	C33	-2.6(4)
C13	C14	C15	C16	0.4(4)	C30	C31	C32	C33	-1.5(4)
C14	C15	C16	C17	-0.3(4)	C31	C32	C33	C34	1.7(4)
C15	C16	C17	C12	-0.8(4)	C32	C33	C34	C29	0.4(4)
C17	C12	C13	C14	-1.6(4)	C34	C29	C30	C31	2.9(3)

Table 7 Hydrogen Atom Coordinates ($\text{\AA} \times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 3a.

Atom	x	y	z	U(eq)
H2	4888.57	3565.34	471.16	58
H3	2330.42	3736.62	-868.18	64
H5	6446.51	3210.69	-2439.21	69
H6	8969.56	3025.8	-1096.45	66
H7A	1911.89	3651.57	-3889.75	143

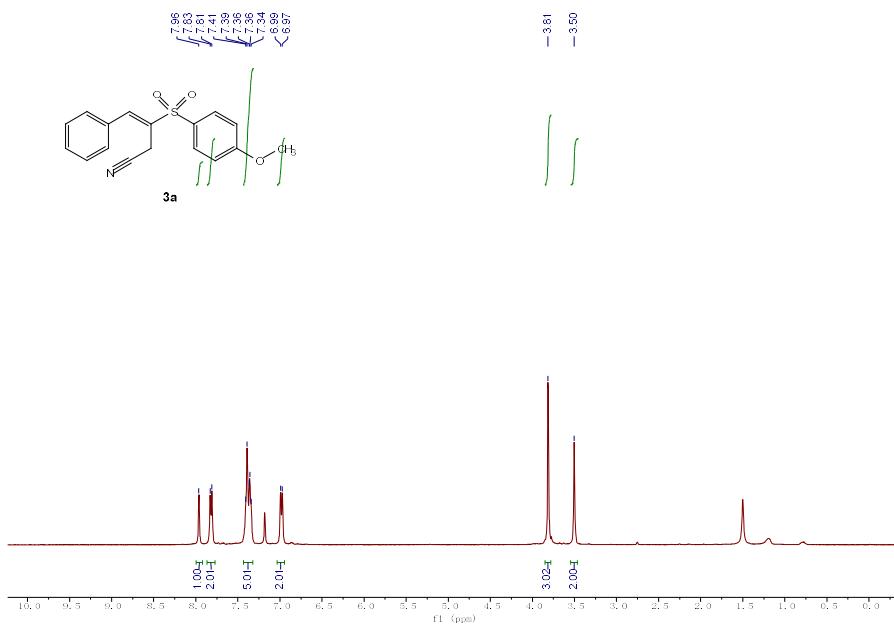
H7B	2655.05	3309.85	-3520.27	143
H7C	4826.21	3569.98	-3508.95	143
H9A	4643.05	2578.5	1267.41	55
H9B	6102.35	2644.1	401.13	55
H11	7182.83	3288.71	2361.89	55
H13	3792.54	3393.73	3307.71	75
H14	1147.89	3223.59	4381.07	93
H15	707.81	2693.36	4653.54	88
H16	2914.56	2329.73	3873.45	80
H17	5565.23	2494.67	2810.25	67
H19	4344.23	4792.9	400.2	89
H20	1634.23	4871.74	-978.09	94
H22	-2344.05	4091.85	-505.93	60
H23	396.44	4012.96	876.68	57
H24A	-4880.58	4400.45	-2672.76	128
H24B	-5334	4309.01	-1652.02	128
H24C	-3224.32	4129.22	-2137.73	128
H26A	1952	4848.51	2242.55	61
H26B	831.31	4766.48	3171.98	61
H28	2534.15	3961.87	2951.44	57
H30	-1273.25	3746.97	3570.33	67
H31	-3794.76	3727.4	4753.74	86
H32	-3493.07	4118.2	5868.54	93
H33	-509.57	4521.7	5825.32	91
H34	2017.4	4547.96	4641.01	75

10. References

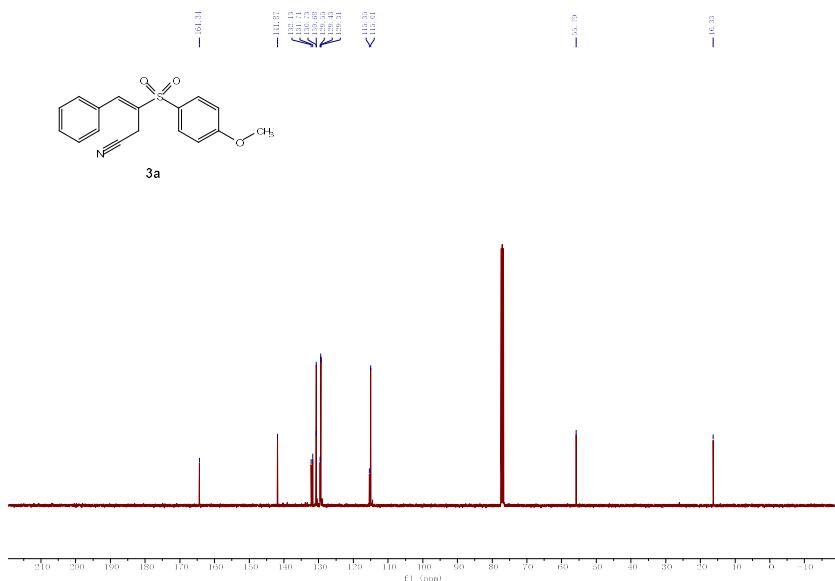
- [1] Zhao, Z.; Racicot, L.; Murphy, G. K. Fluorinative Rearrangements of Substituted Phenylallenes Mediated by (Difluoroiodo)toluene: Synthesis of α -(Difluoromethyl)styrenes. *Angew. Chem., Int. Ed.* **2017**, *56*, 11620–11623.
- [2] Deng, H.; Meng, Z.; Wang, S.; Zhang, Z.; Zhang, Y.; Shangguan, Y.; Yang, F.; Yuan, D.; Guo, H.; Zhang, C. Enantioselective Copper-Catalyzed Three-Component Carbaboronation of Allenes: Access to Functionalized Dibenzo[b,f][1,4]oxazepine Derivatives. *Adv. Synth. Catal.* **2019**, *361*, 1–7.
- [3] Hasenbeck, M.; Ahles, S.; Averdunk, A.; Becker, J.; Gellrich, U. Formation of Nucleophilic Allylboranes from Molecular Hydrogen and Allenes Catalyzed by a Pyridonate Borane that Displays Frustrated Lewis Pair Reactivity. *Angew. Chem., Int. Ed.* **2020**, *59*, 23885–23891.
- [4] Liu, Y.; Pan, Q.; Hu, X.; Guo, Y.; Chen, Q-Y; Liu, C. Rapid Access to N-Protected Sulfonimidoyl Fluorides: Divergent Synthesis of Sulfonamides and Sulfonimidamides. *Org. Lett.* **2021**, *23*, 3975–3980.
- [5] Singh, A. K.; Kandasamy, J. Palladium catalyzed stereocontrolled synthesis of C-aryl glycosides using glycals and arenediazonium salts at room temperature. *Org. Biomol. Chem.* **2018**, *16*, 5107–5112.
- [6] Liu, S.; Huang, Y.; Xu, X-H; Qing, F-L. Fluorosulfonylation of arenediazonium tetrafluoroborates with $\text{Na}_2\text{S}_2\text{O}_5$ and N-fluorobenzenesulfonimide. *J. Fluorine Chem.* **2020**, *240*, 109653.
- [7] Ramanathan, M.; Liu, S.-T. Cascade annulations of aryldiazonium salts, nitriles and halo-alkynes leading to 3-haloquinolines. *Tetrahedron* **2017**, *73*, 4317–4322.
- [8] Xing, B.; Li, L.; Ni, C.; Hu, J. Pentafluoroethylation of Arenediazonium Tetrafluoroborates Using On-Site Generated Tetrafluoroethylene. *Chin. J. Chem.* **2019**, *37*, 1131–1136.
- [9] Min, M.; Bang, G. S.; Lee, H.; Yu, B-C. A photoswitchable methylene-spaced fluorinated aryl azobenzene monolayer grafted on silicon. *Chem. Commun.*, **2010**, *46*, 5232–5234.
- [10] Cheng, C.Y.; Isobe, M. Three types of products by carbon nu-cleophiles toward methoxyphenylacetylenic sulfones. *Tetrahedron* **2011**, *67*, 9957–9965.
- [11] Yu, R. -R; Xing, Y. -D; Fang, X. -J. Regio-, Chemo-, and Enanti-oselective Ni-Catalyzed Hydrocyanation of 1,3-Dienes. *Org. Lett.* **2021**, *23*, 930–935.
- [12] Beck, H.; Jeske, M.; Thede, K.; Stoll, F.; Flamme, I.; Akbaba, M.; Ergeden, J. -K.; Karig, G.; Keldenich, J.; Oehme, F.; Militzer, H. -C.; Hartung, I. -V.; Thuss, U. Discovery of Molidustat (BAY 85-3934): A Small-Molecule Oral HIF-Prolyl Hydroxylase (HIF-PH) Inhibitor for the Treatment of Renal Anemia. *ChemMedChem* **2018**, *13*, 988–1003.

-
- [13] Dolomanov, O. V.; Bourhis, L. J.; Gildea, R. J.; Howard, J. A. K.; Puschmann, H. OLEX2: a complete structure solution, refinement and analysis program. *J. Appl. Cryst.* **2009**, *42*, 339–341.
- [14] (a) Palatinus, L.; Chapuis, G. SUPERFLIP - a computer program for the solution of crystal structures by charge flipping in arbitrary dimensions. *J. Appl. Cryst.* **2007**, *40*, 786–790; (b) Palatinus, L.; van der Lee, A. Symmetry determination following structure solution in P1. *J. Appl. Cryst.* **2008**, *41*, 975–984; (c) Palatinus, L.; Prathapa, S. J.; van Smaalen, S. EDMA: a computer program for topological analysis of discrete electron densities. *J. Appl. Cryst.* **2012**, *45*, 575–580.
- [15] Sheldrick, G. M. Crystal structure refinement with SHELXL. *Acta Cryst.* **2015**, *C71*, 3–8.

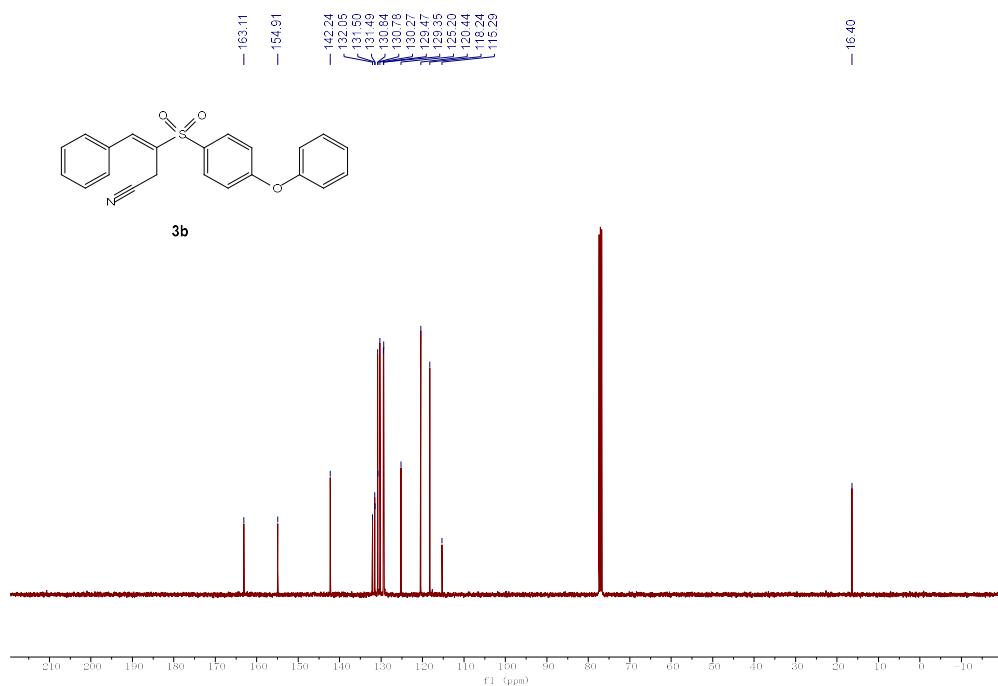
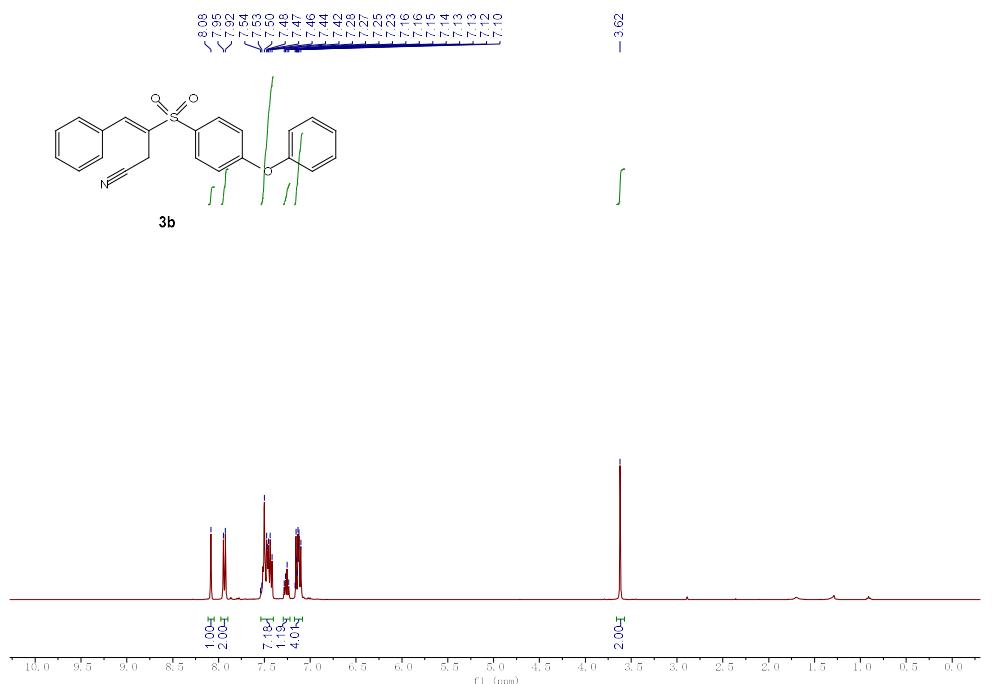
11. Copies of ^1H NMR, ^{19}F NMR and ^{13}C NMR spectra

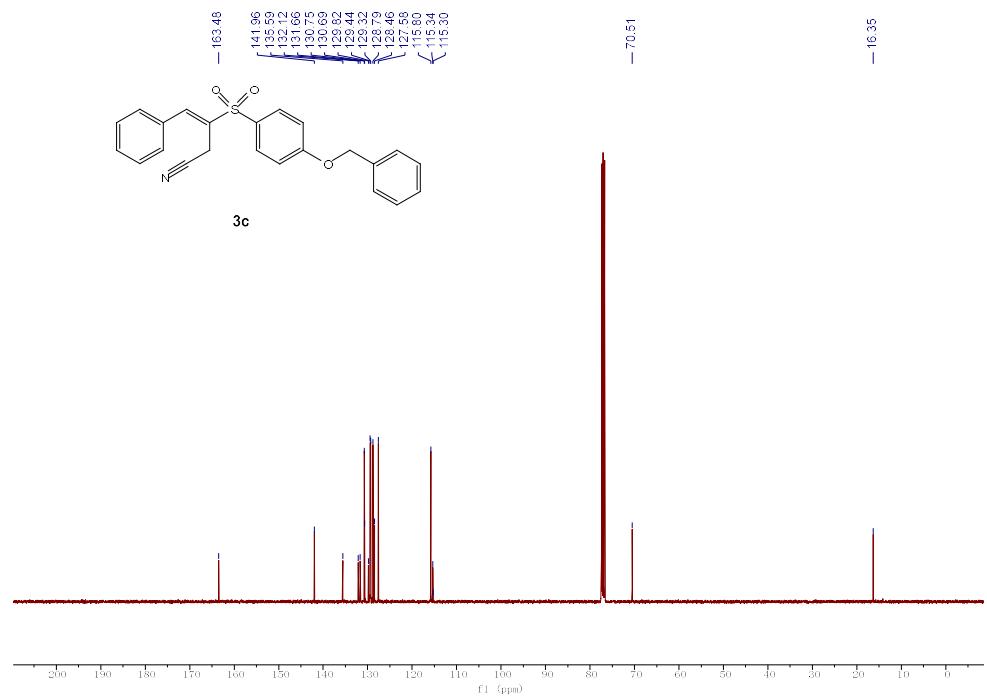
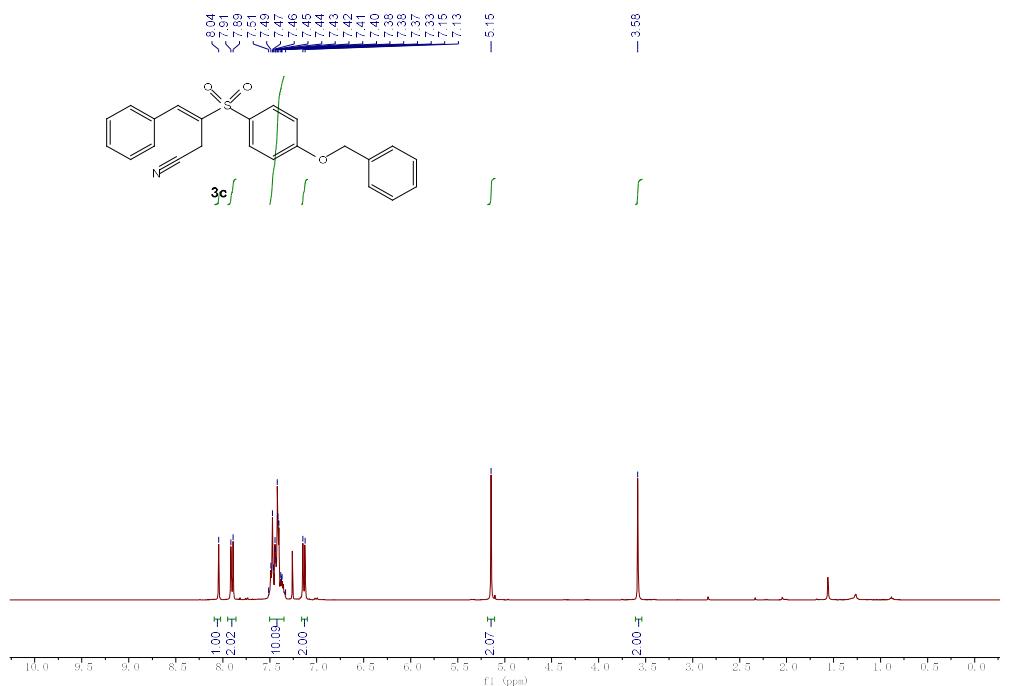


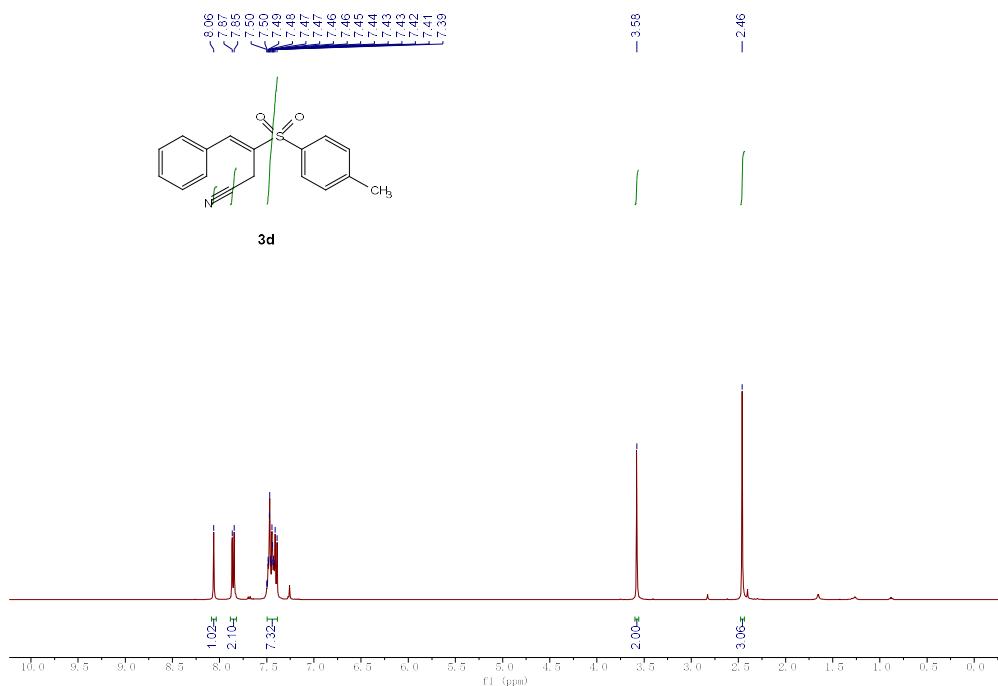
^1H NMR Spectrum of 3a (CDCl_3 , 400 MHz)



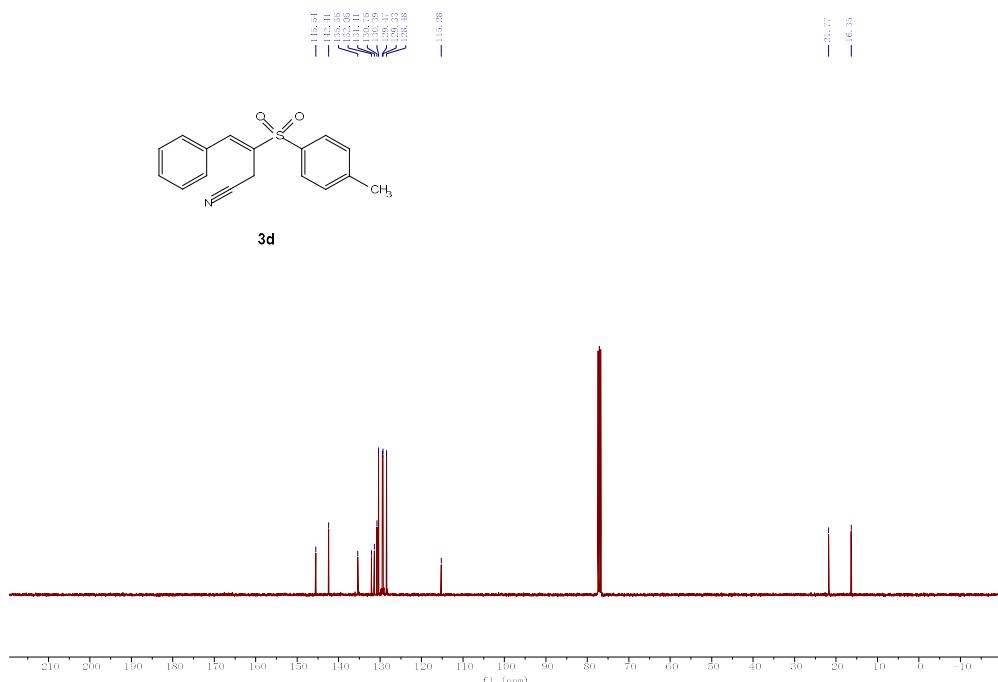
^{13}C NMR Spectrum of 3a (CDCl_3 , 101MHz)



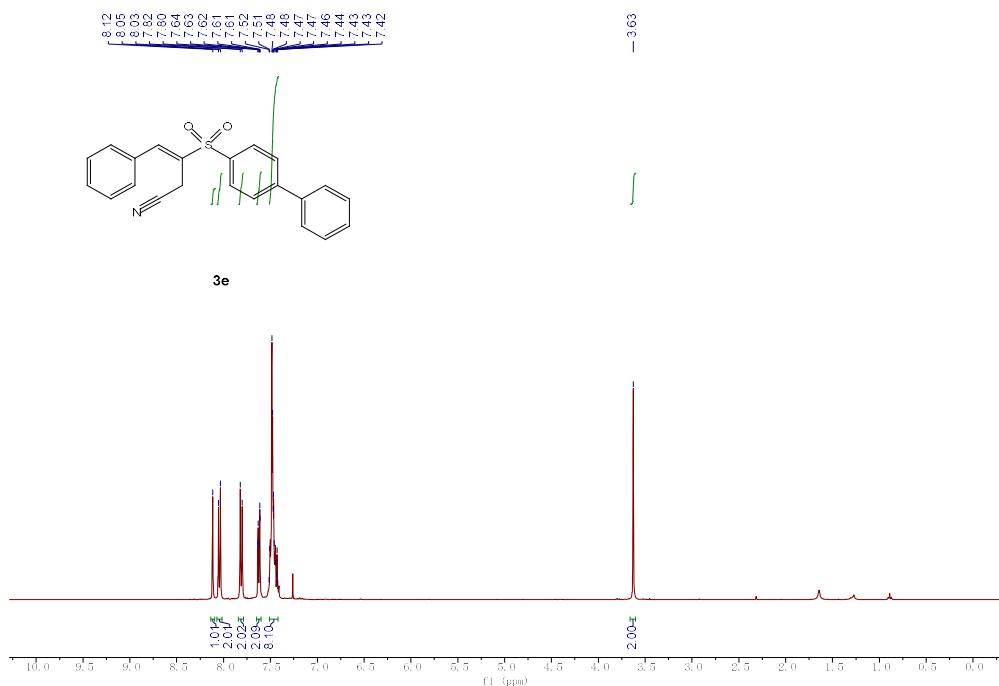




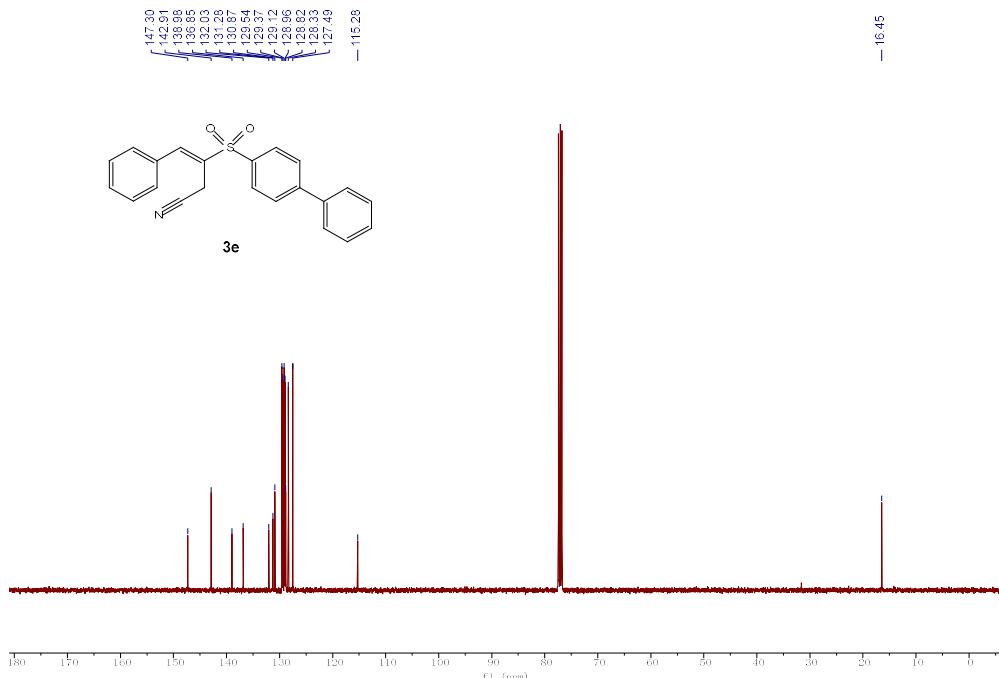
¹H NMR Spectrum of 3d (CDCl₃, 400 MHz)



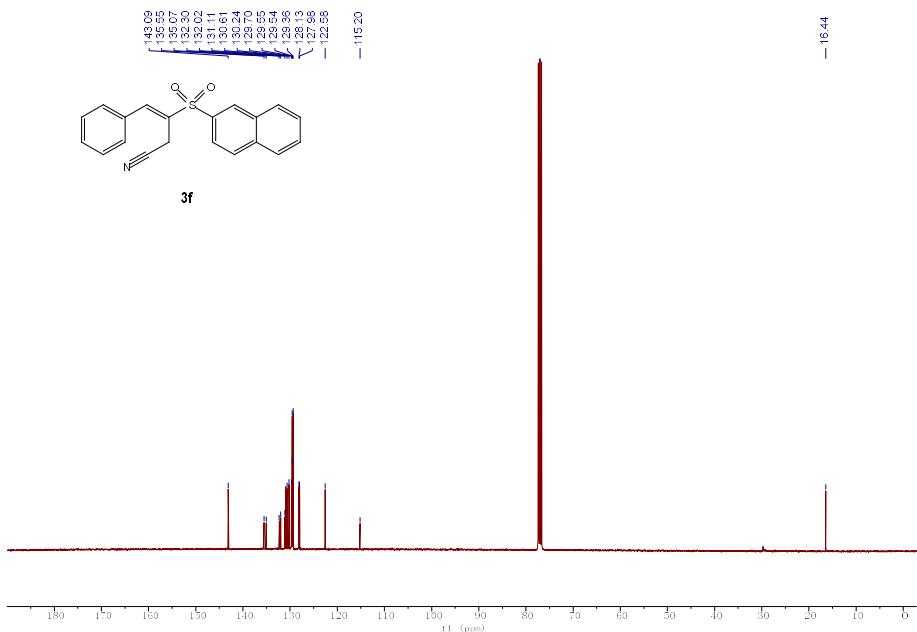
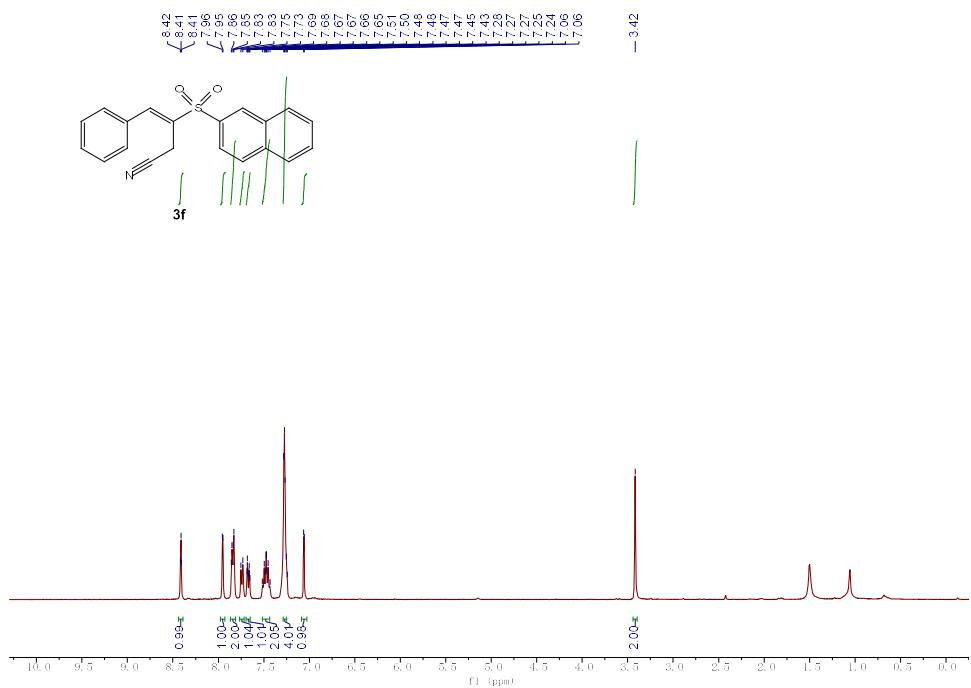
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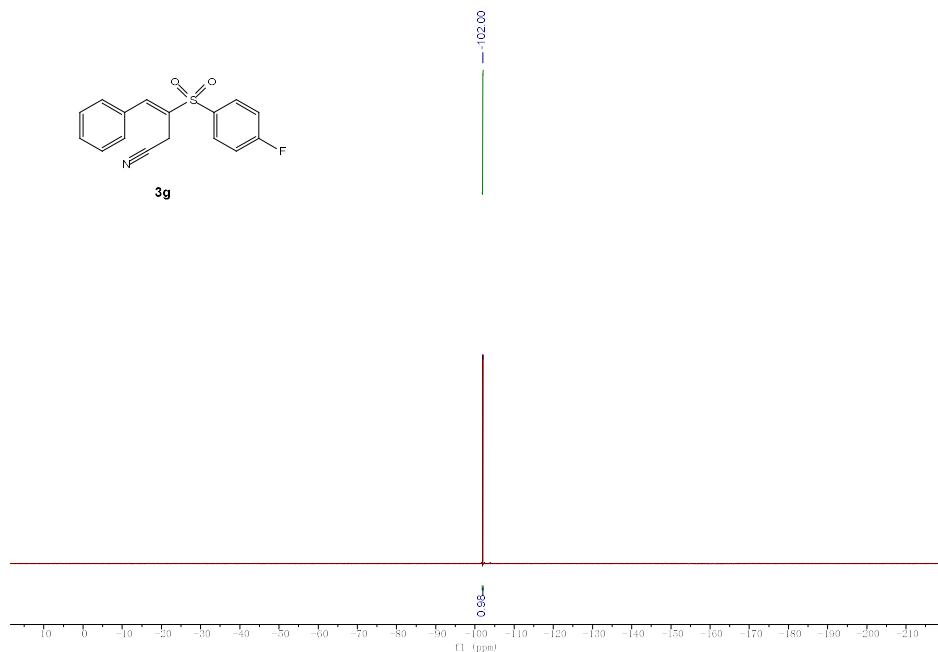
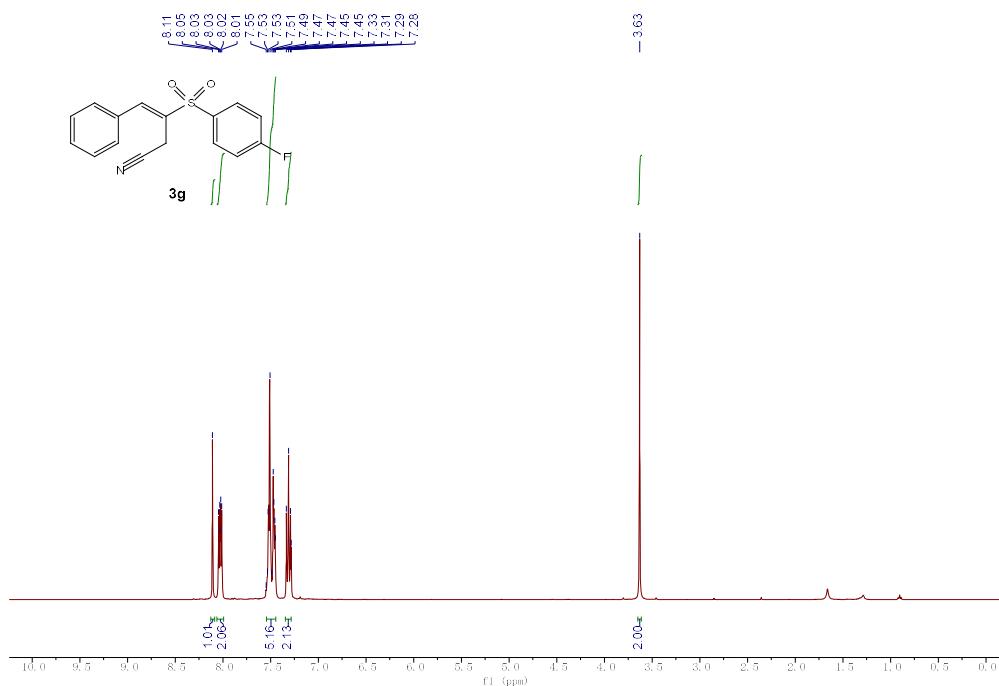


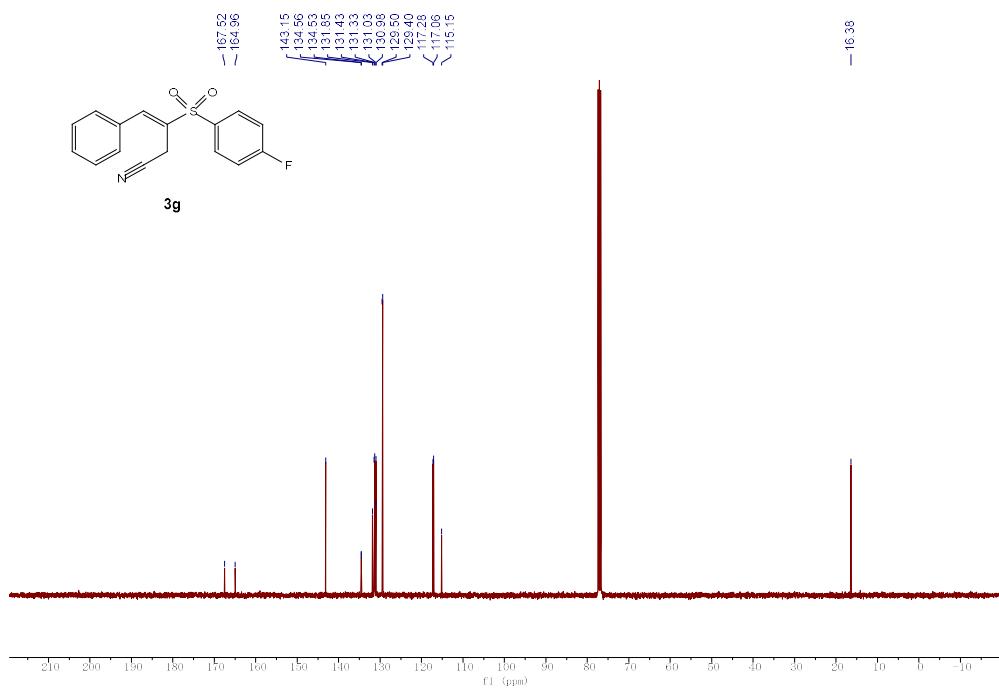
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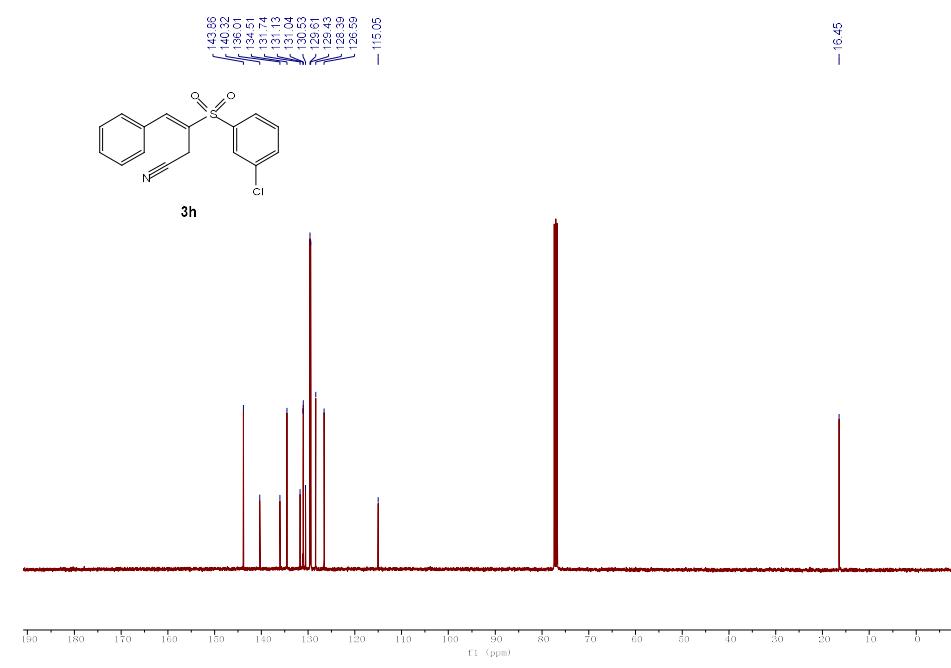
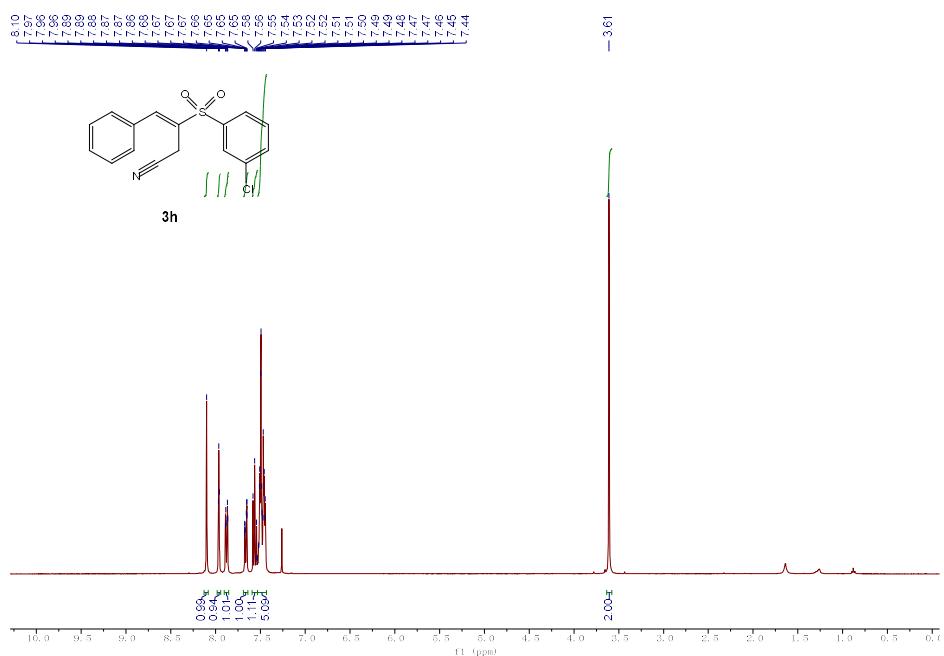


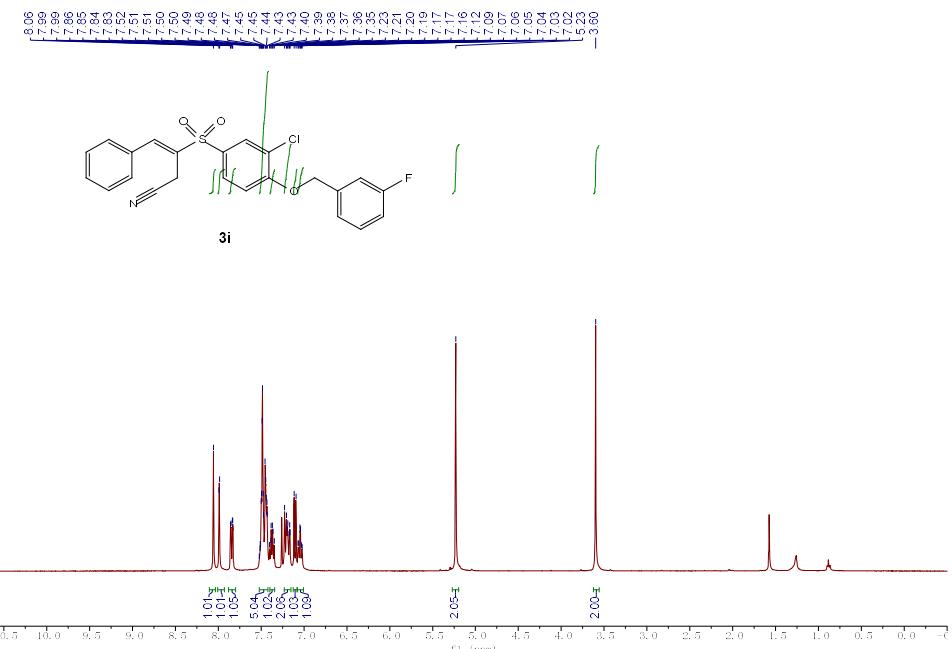
^{13}C NMR Spectrum of 3e (CDCl_3 , 101MHz)



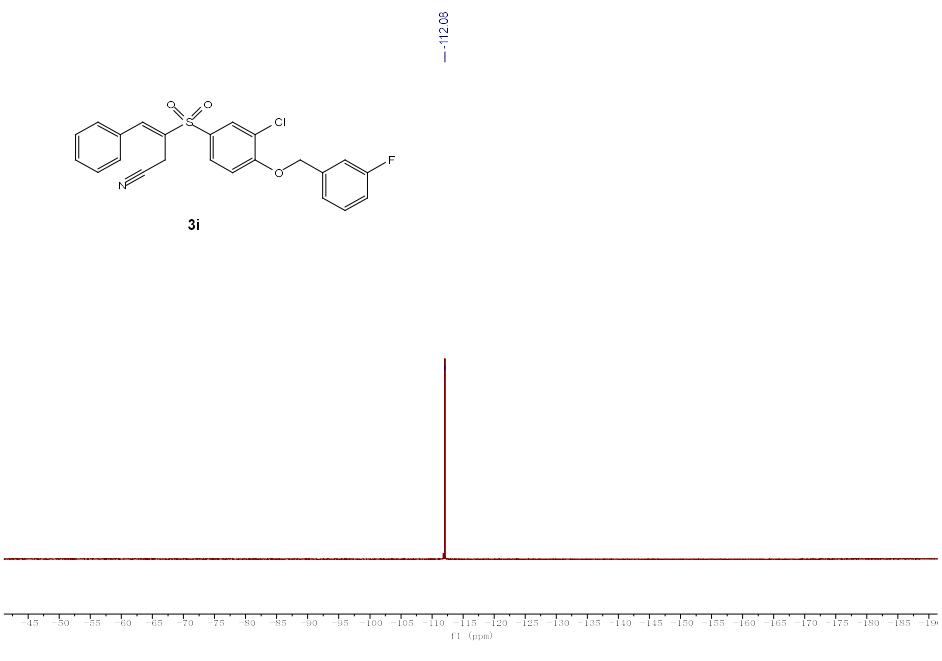




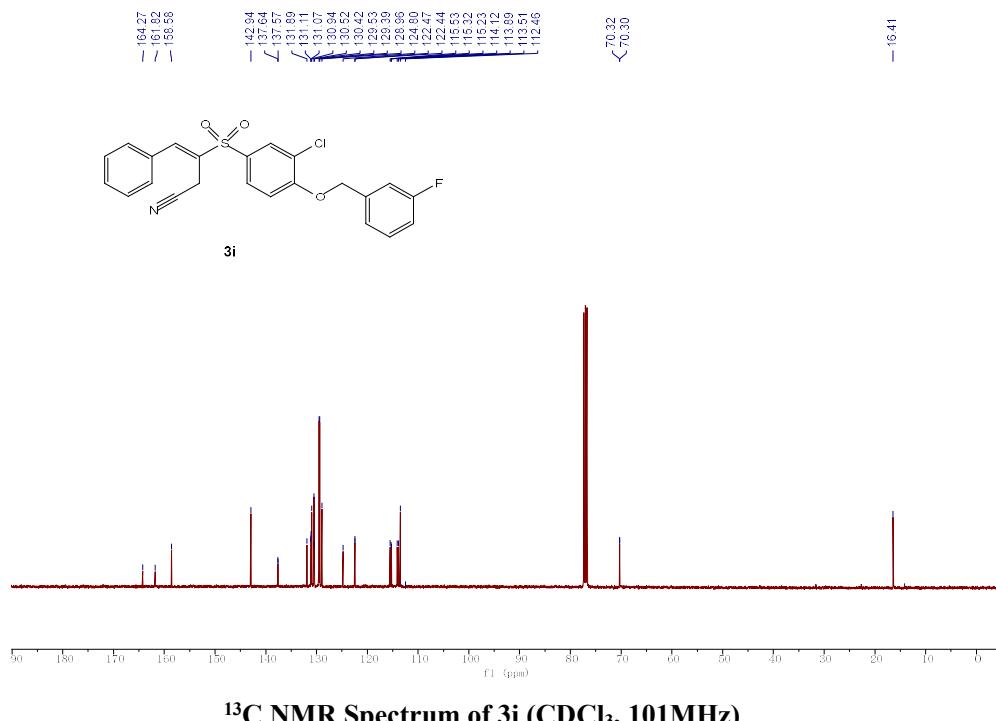




^1H NMR Spectrum of 3i (CDCl_3 , 400 MHz)



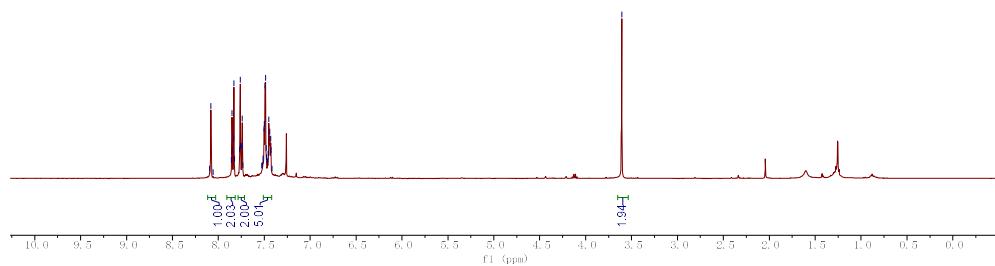
^{19}F NMR Spectrum of 3i (CDCl_3 , 376 MHz)





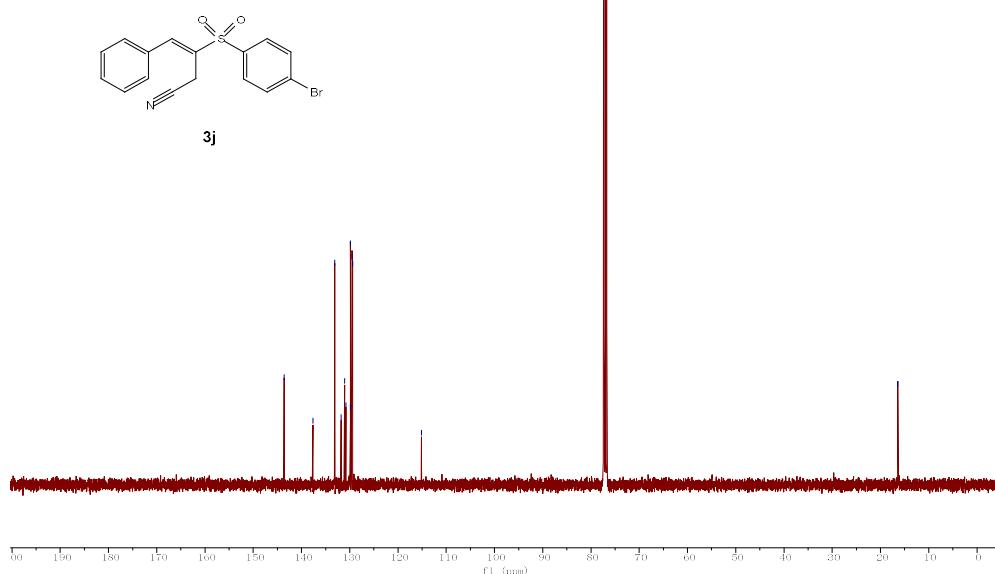
3j

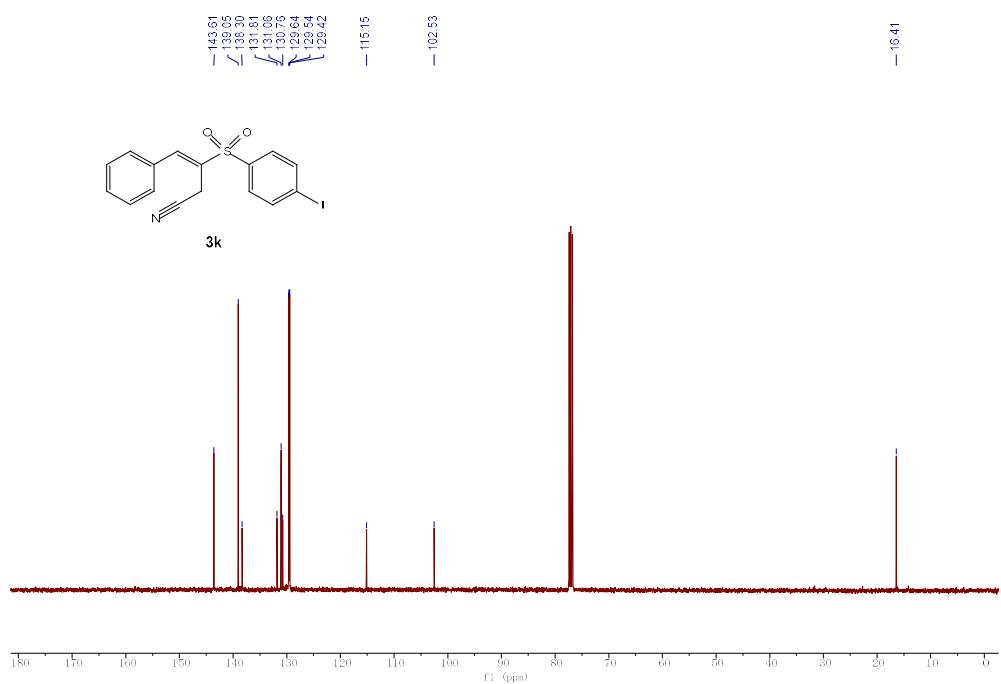
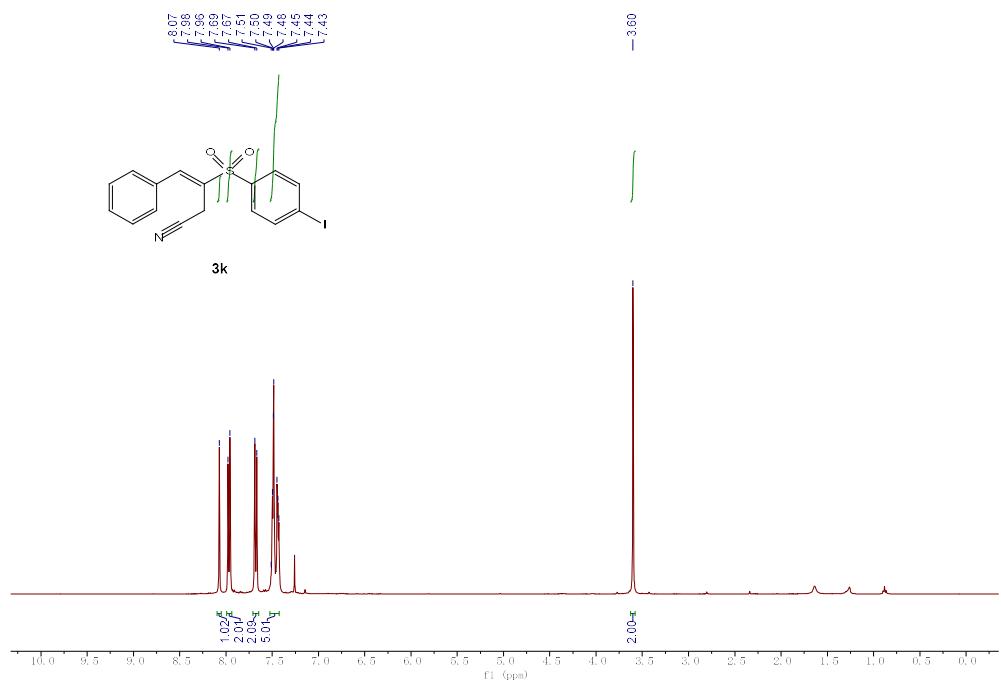
— 3.61

**1^H NMR Spectrum of 3j (CDCl₃, 400 MHz)**

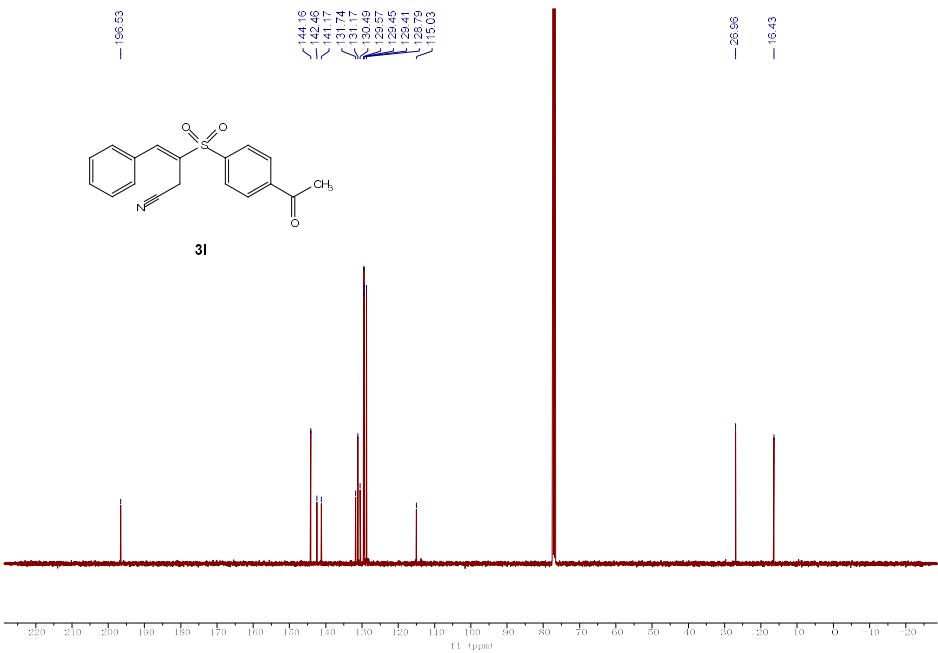
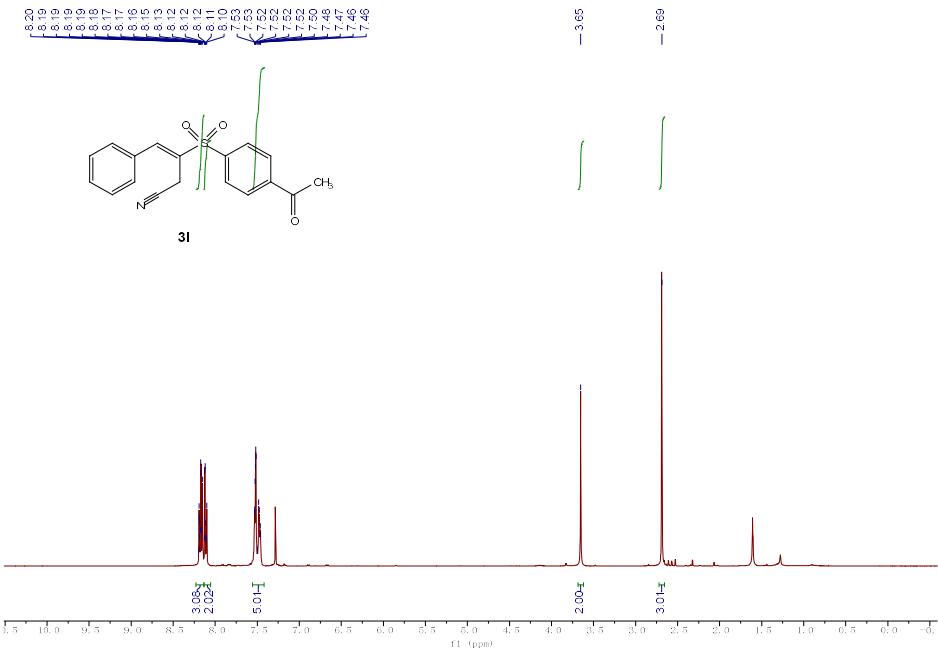
3j

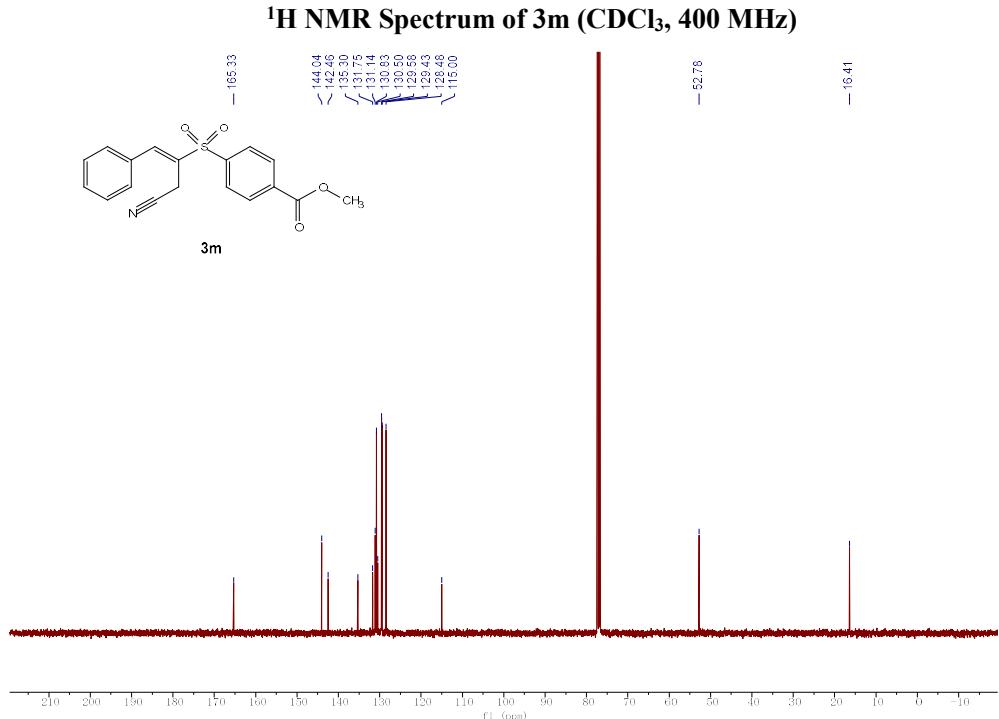
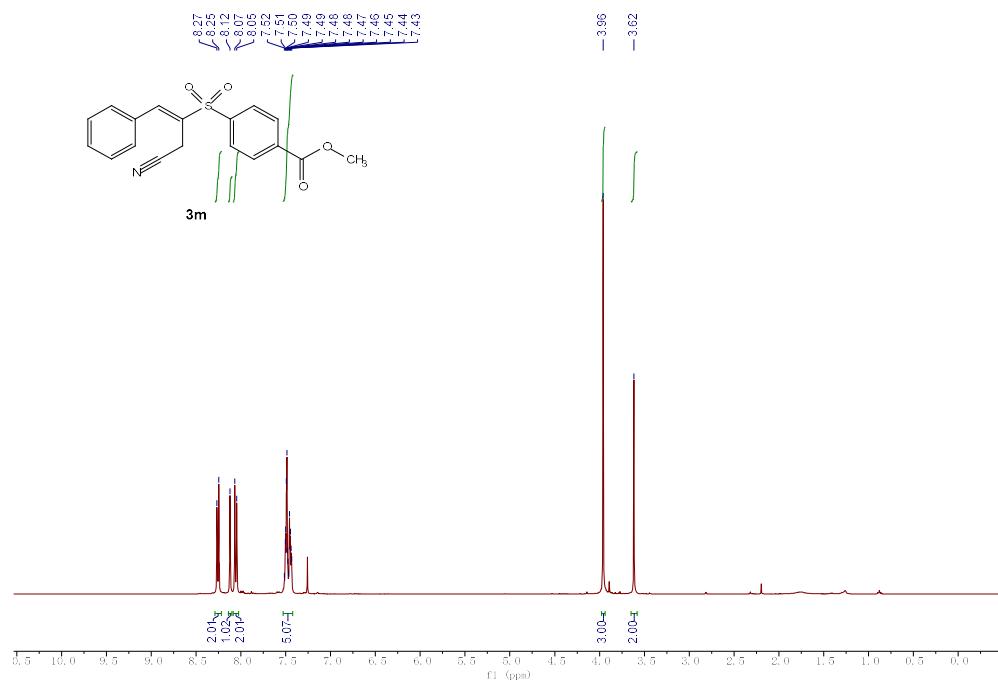
— 16.40

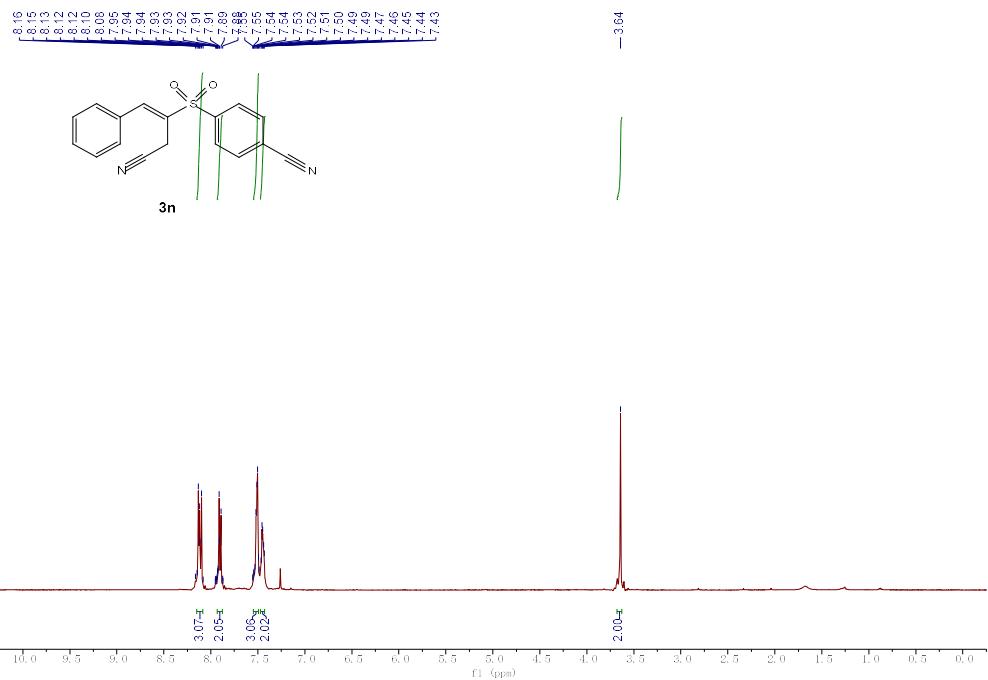
**1³C NMR Spectrum of 3j (CDCl₃, 101MHz)**



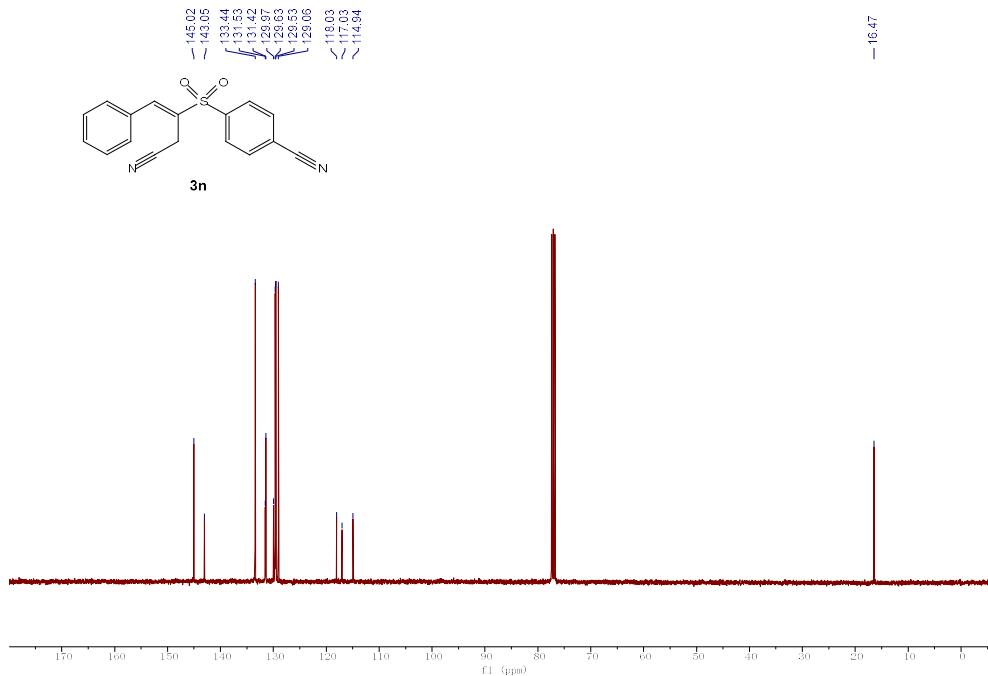
¹³C NMR Spectrum of 3k (CDCl₃, 101MHz)



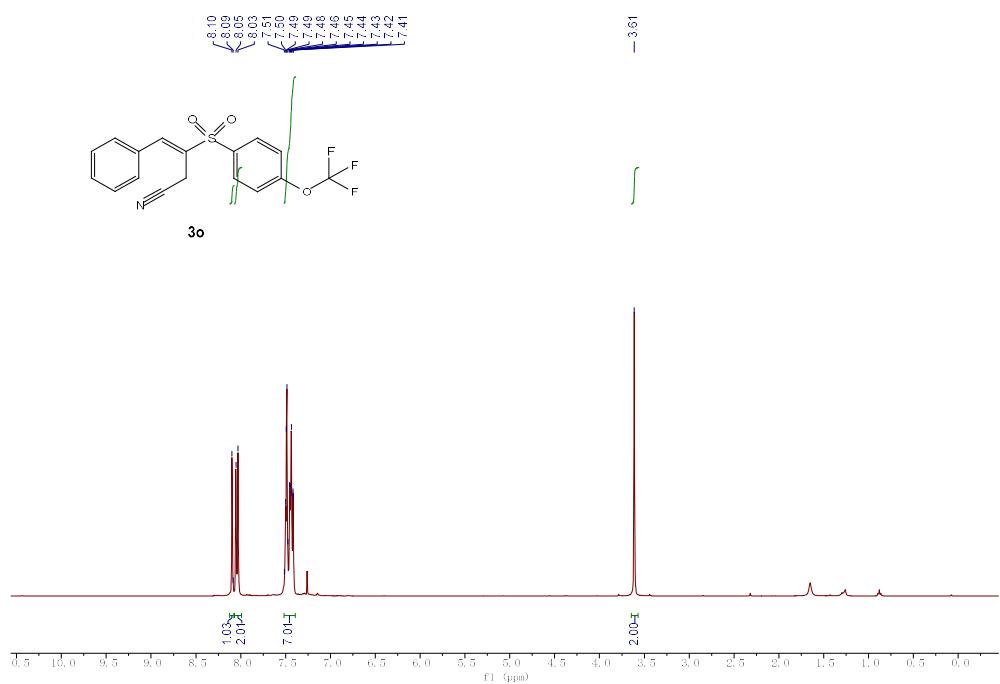




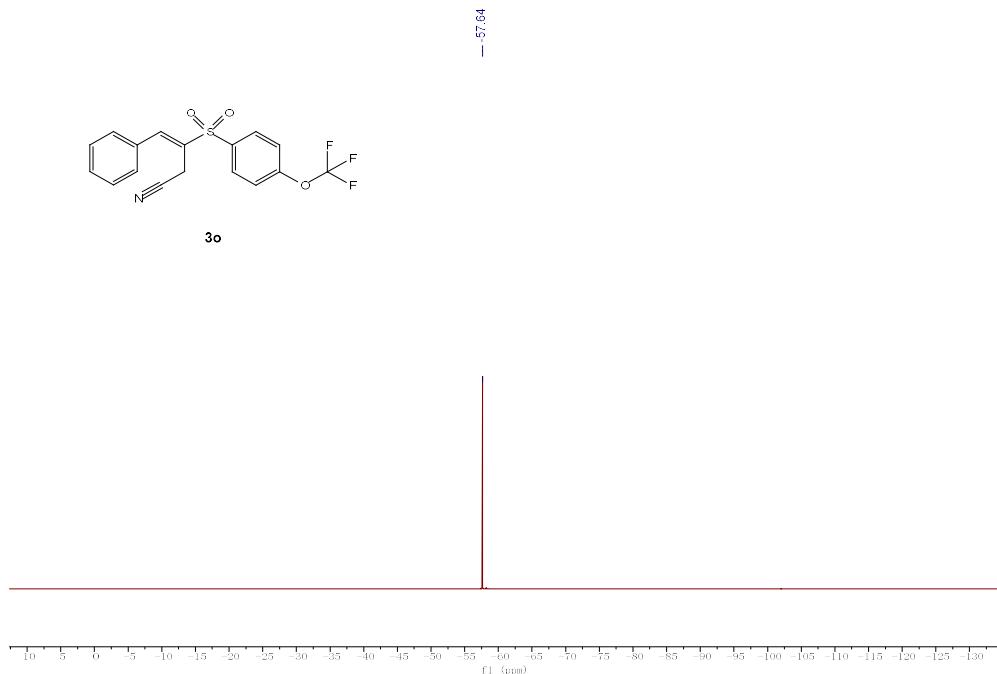
¹H NMR Spectrum of 3n (CDCl₃, 400 MHz)



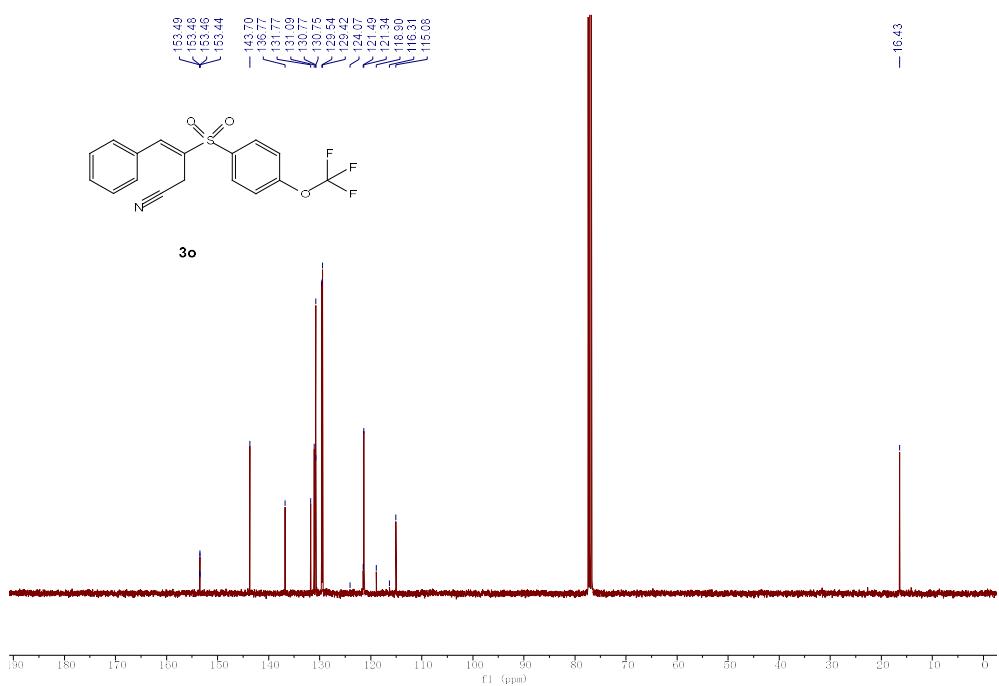
¹³C NMR Spectrum of 3n (CDCl₃, 101MHz)



¹H NMR Spectrum of 3o (CDCl₃, 400 MHz)

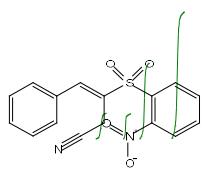


¹⁹F NMR Spectrum of 3o (CDCl₃, 376 MHz)

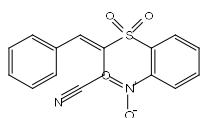
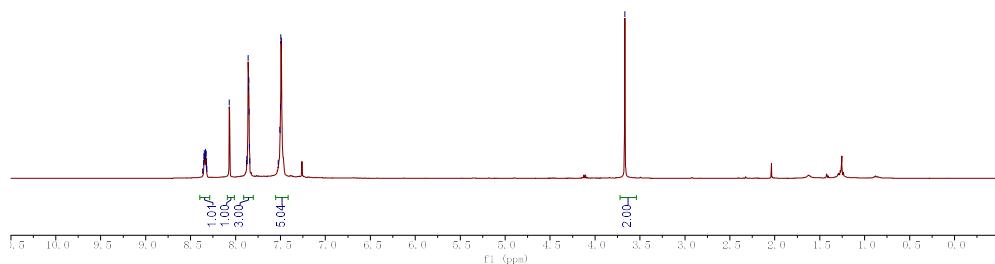


^{13}C NMR Spectrum of **3o (CDCl_3 , 101MHz)**

8.35
8.35
8.34
8.34
8.34
8.33
8.33
8.32
8.07
7.88
7.87
7.86
7.86
7.85
7.84
7.84
7.83
7.82
7.51
7.50
7.50
7.49

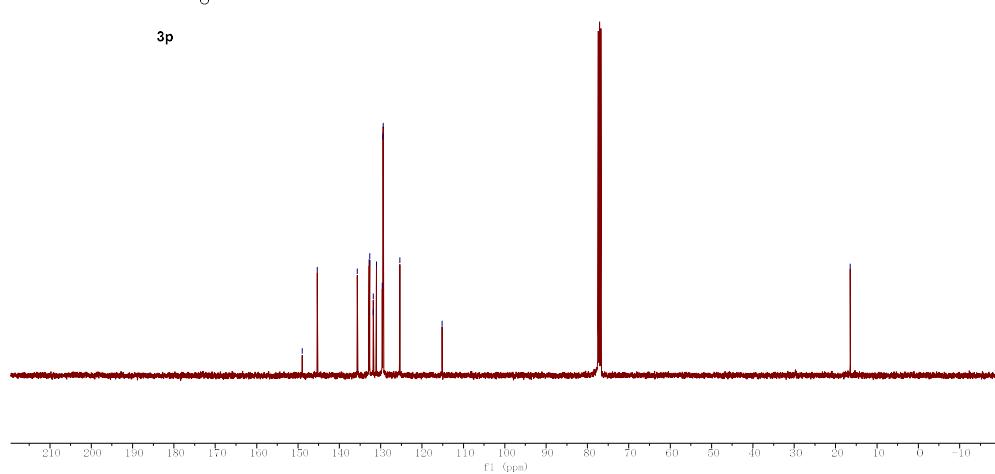


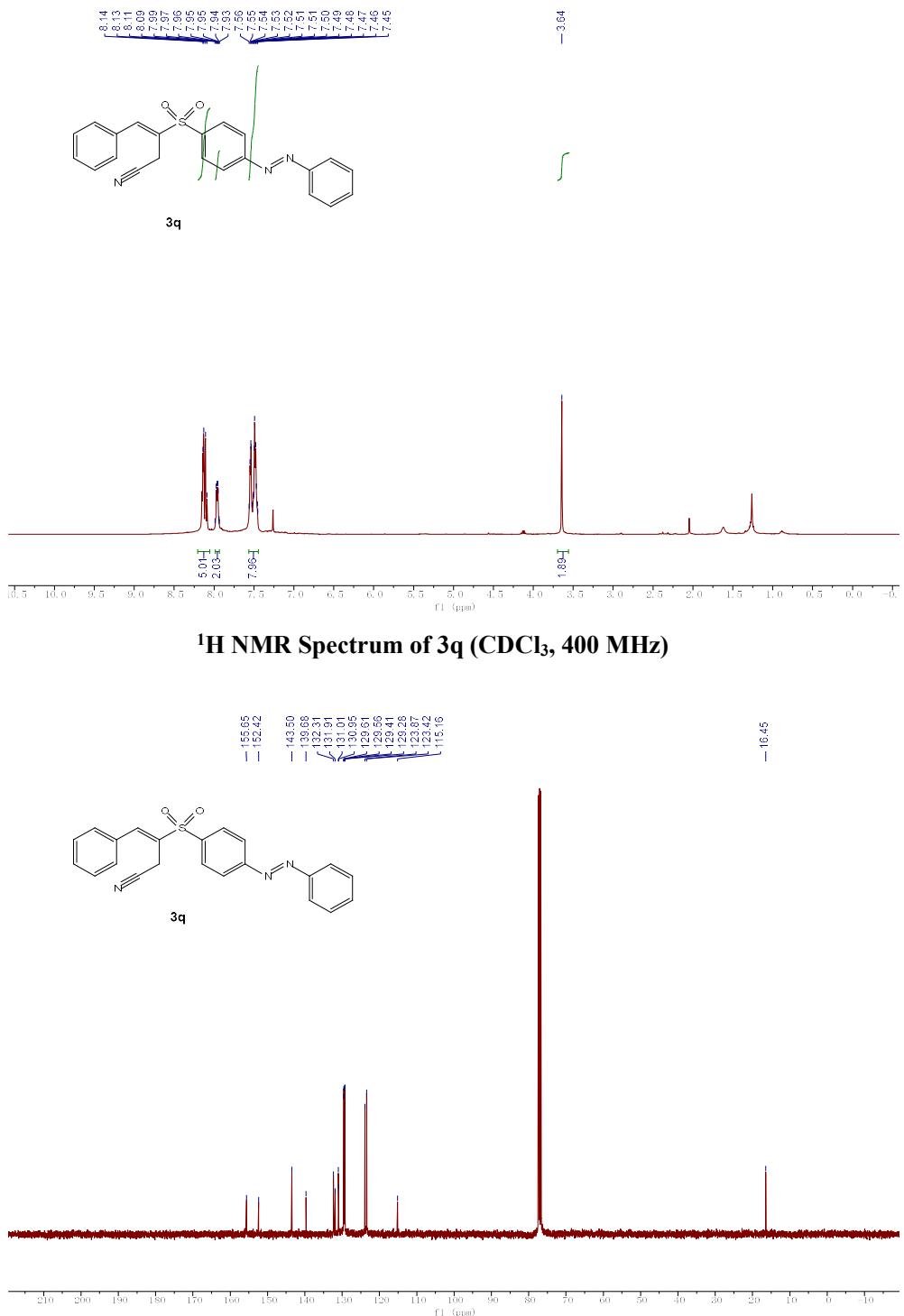
-3.67

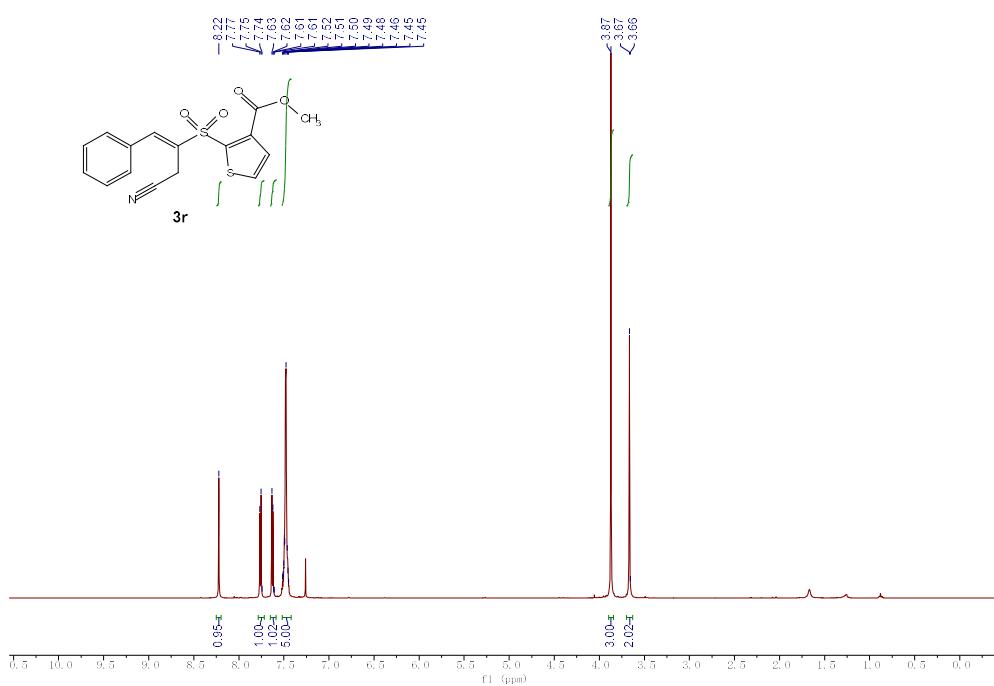


3p

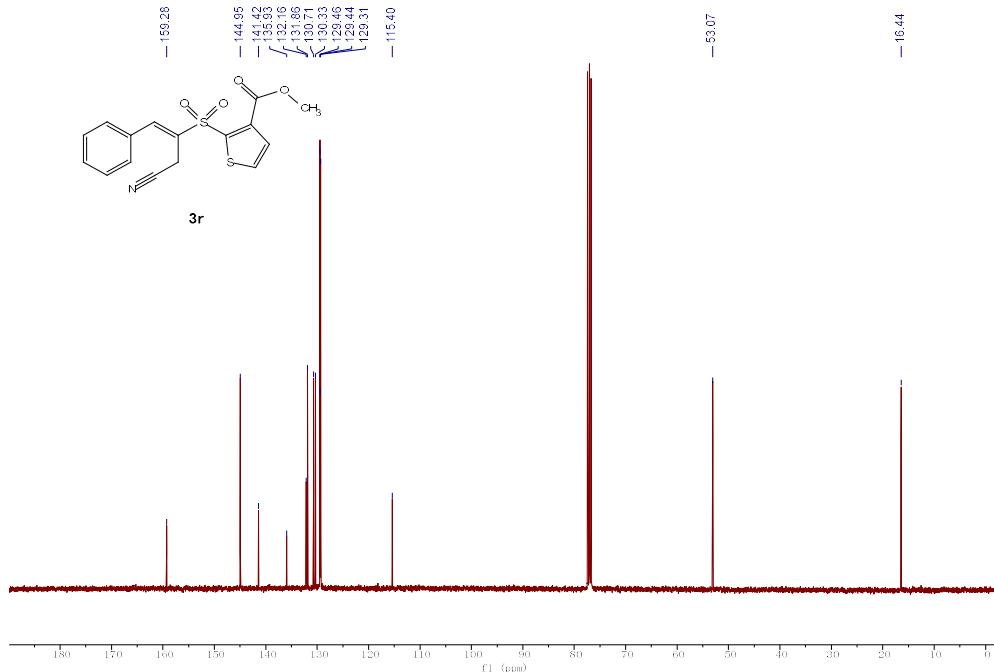
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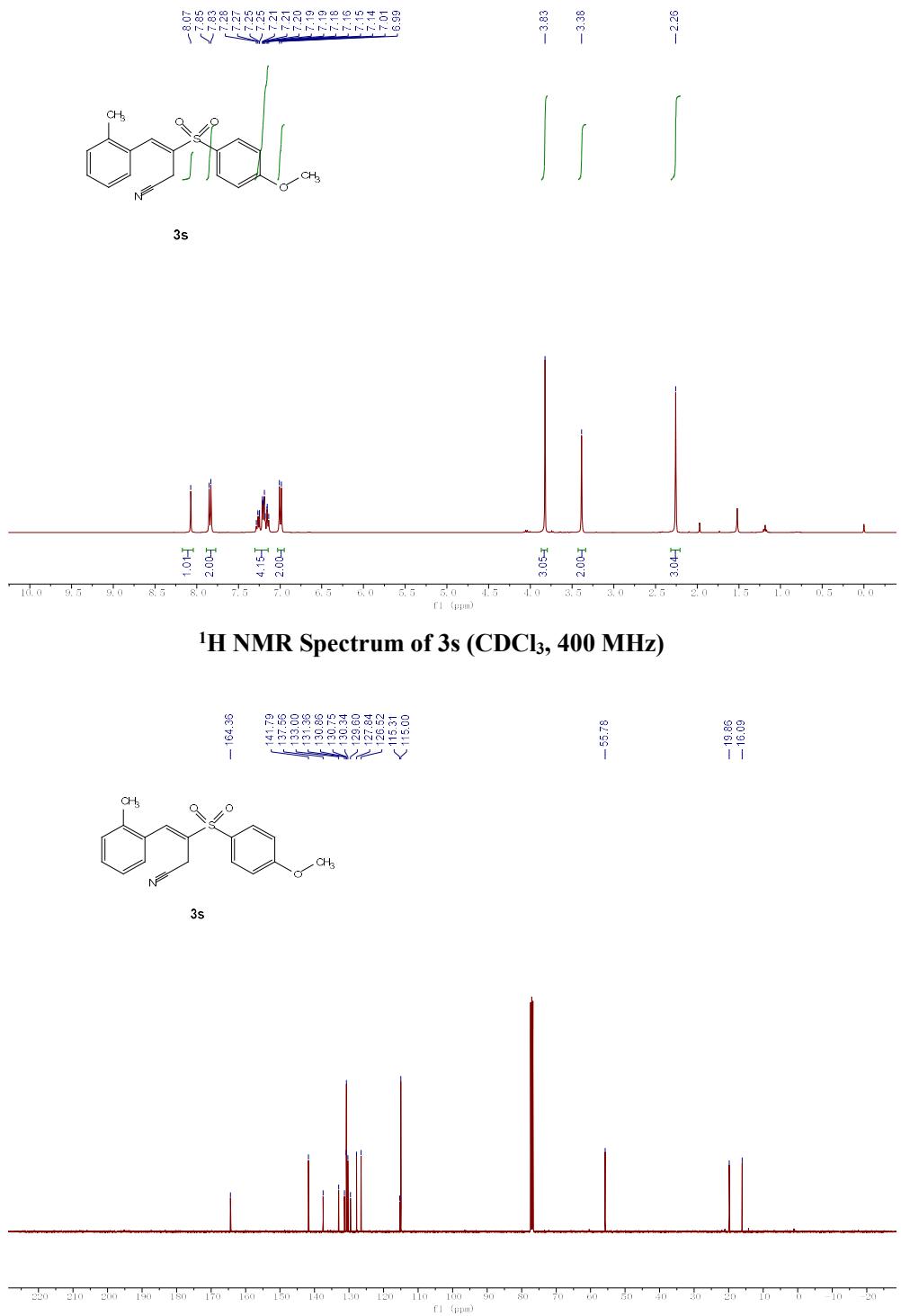


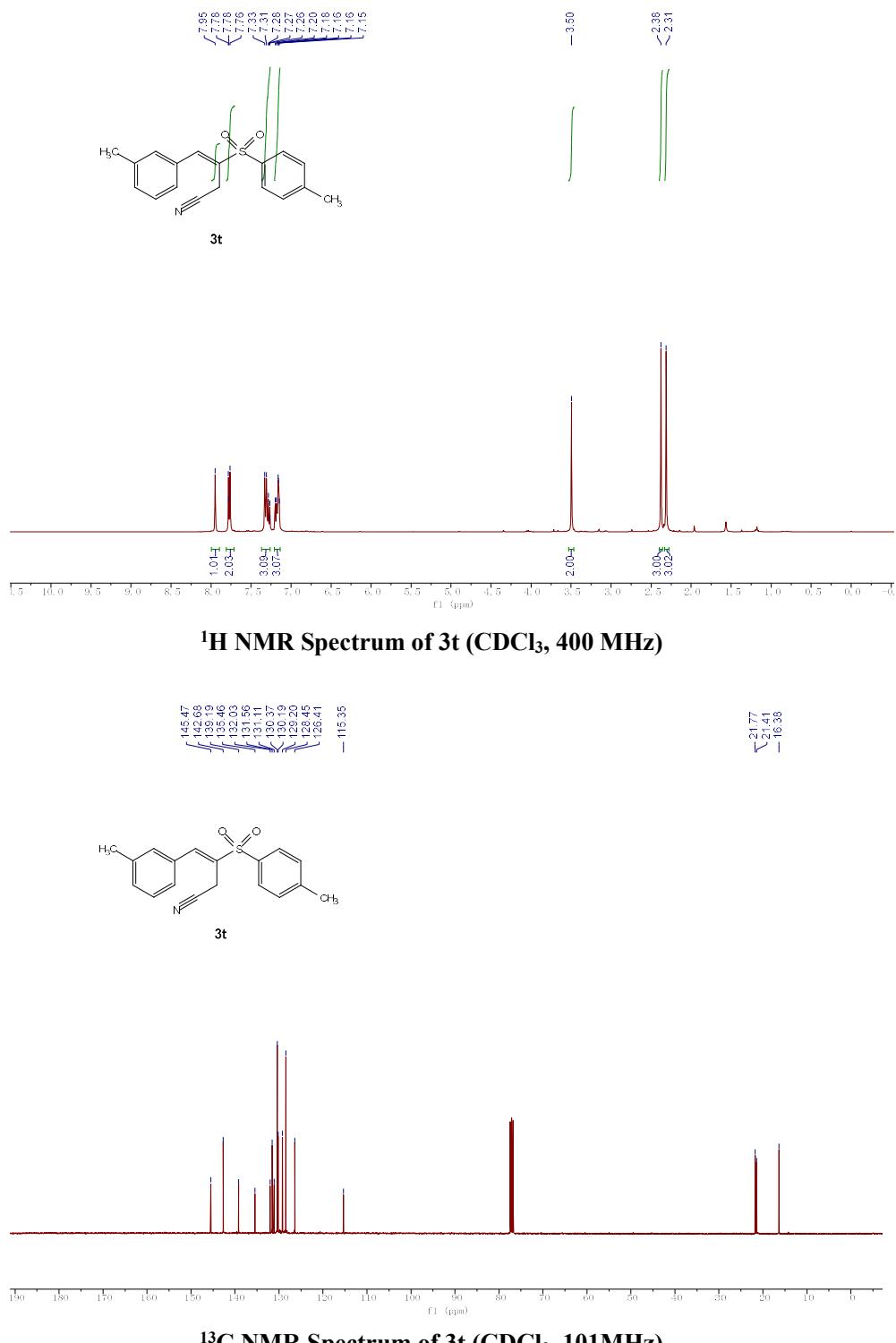


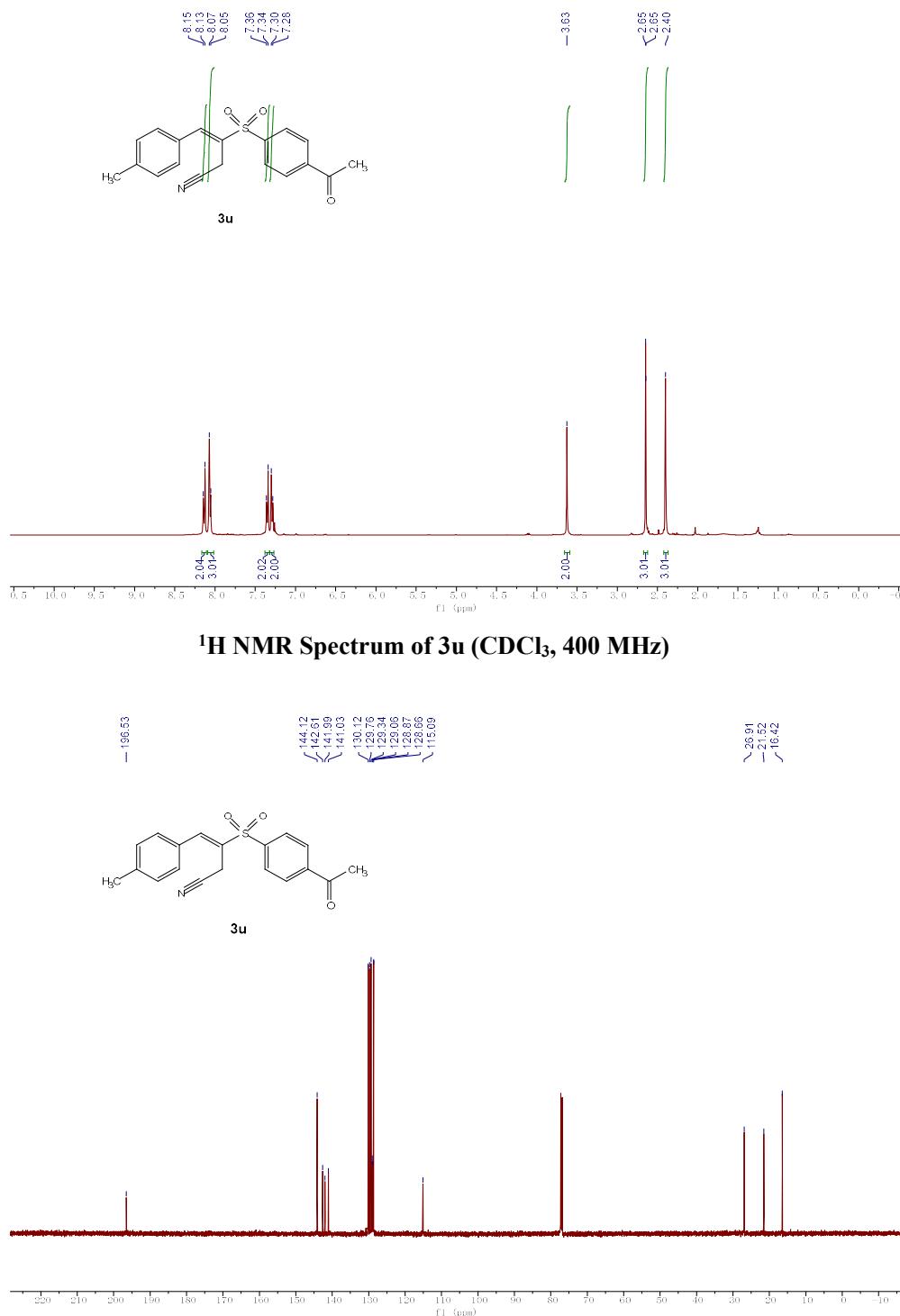
¹H NMR Spectrum of 3r (CDCl₃, 400 MHz)

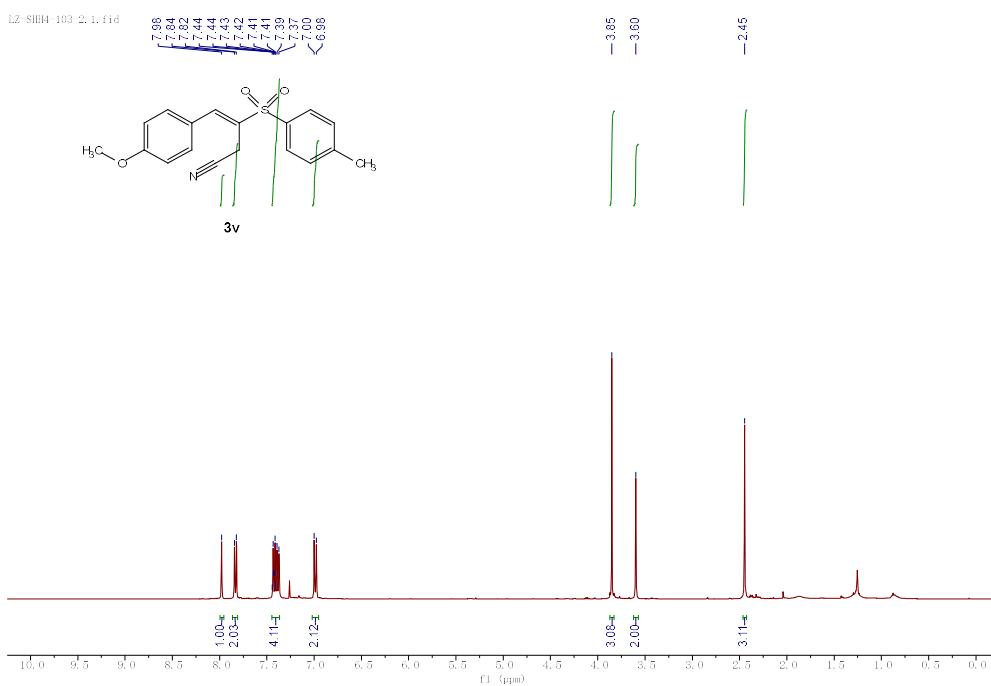


¹³C NMR Spectrum of 3r (CDCl₃, 101MHz)

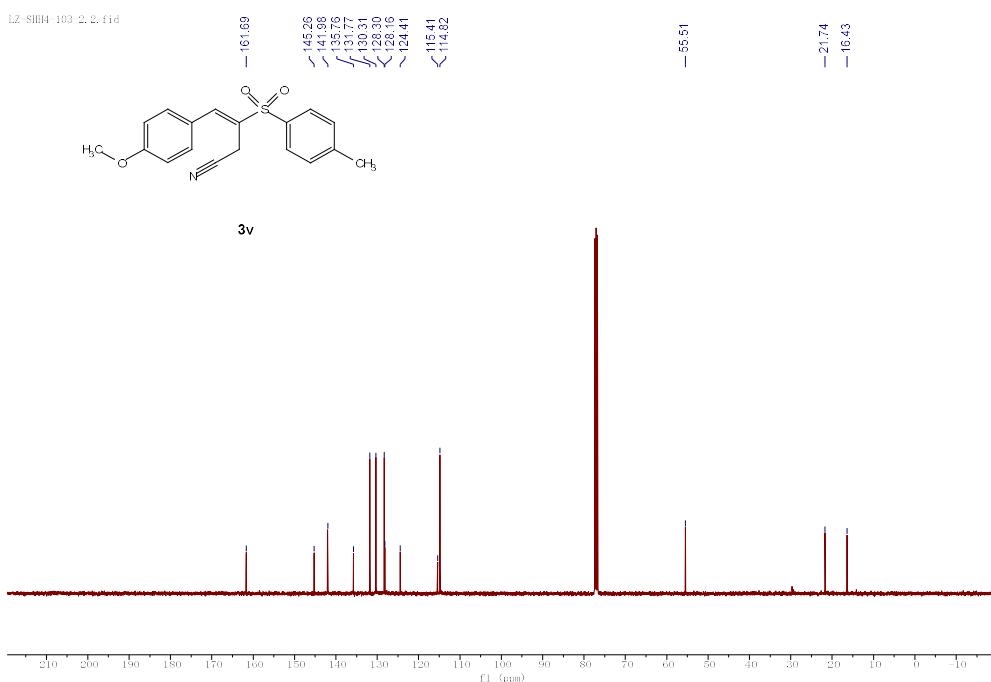




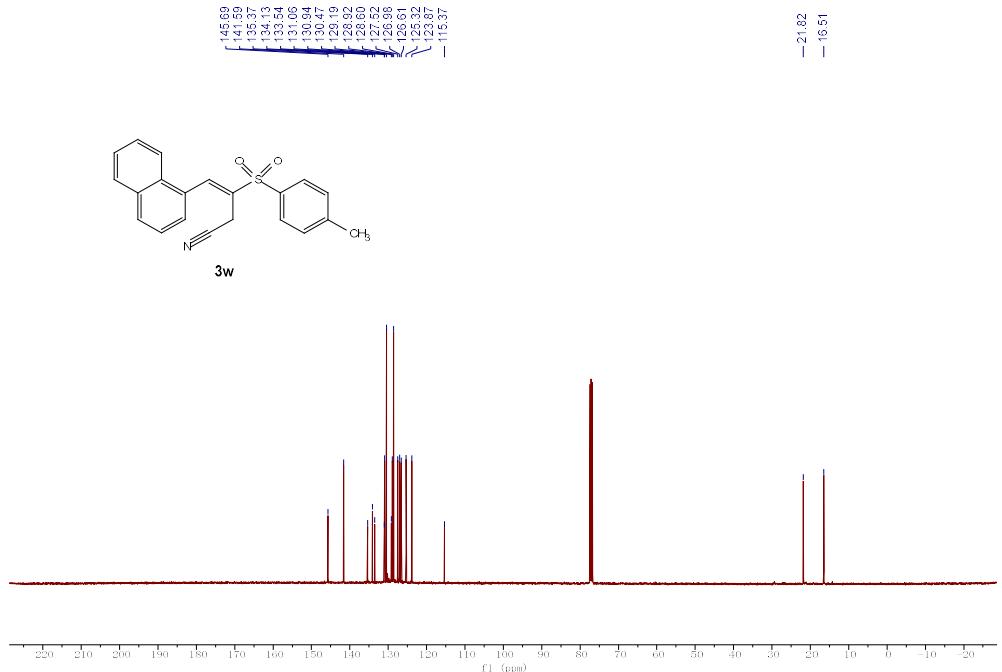
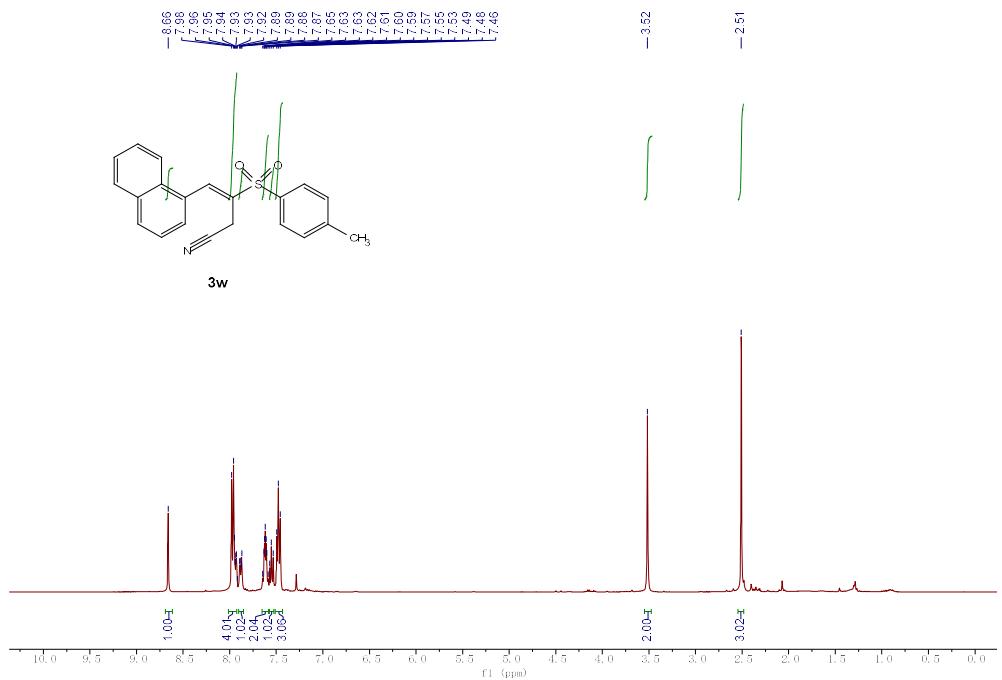


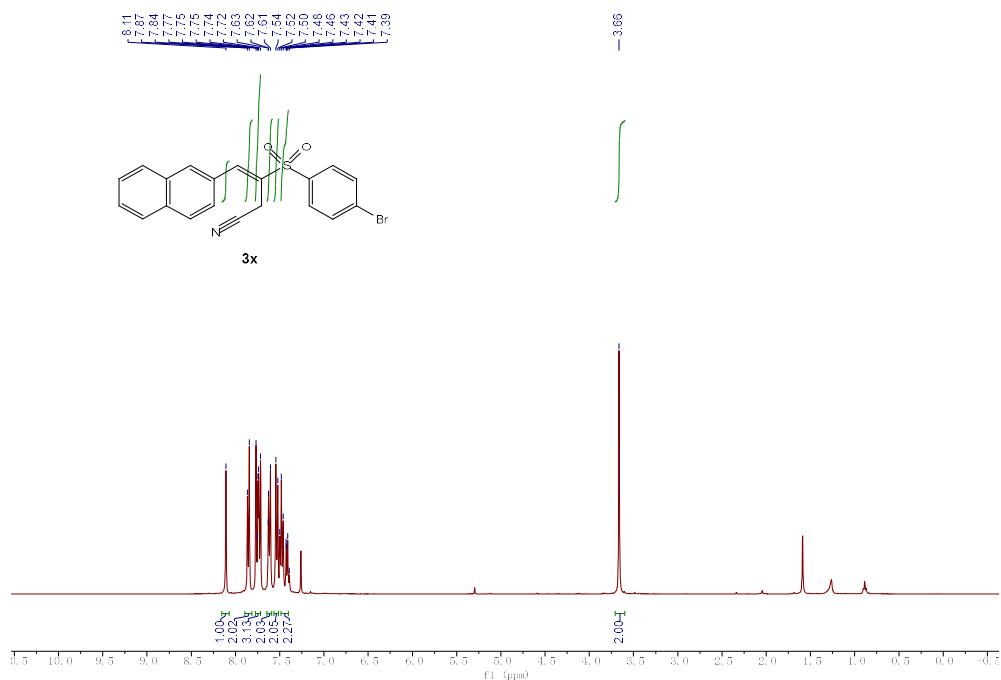


¹H NMR Spectrum of 3v (CDCl₃, 400 MHz)

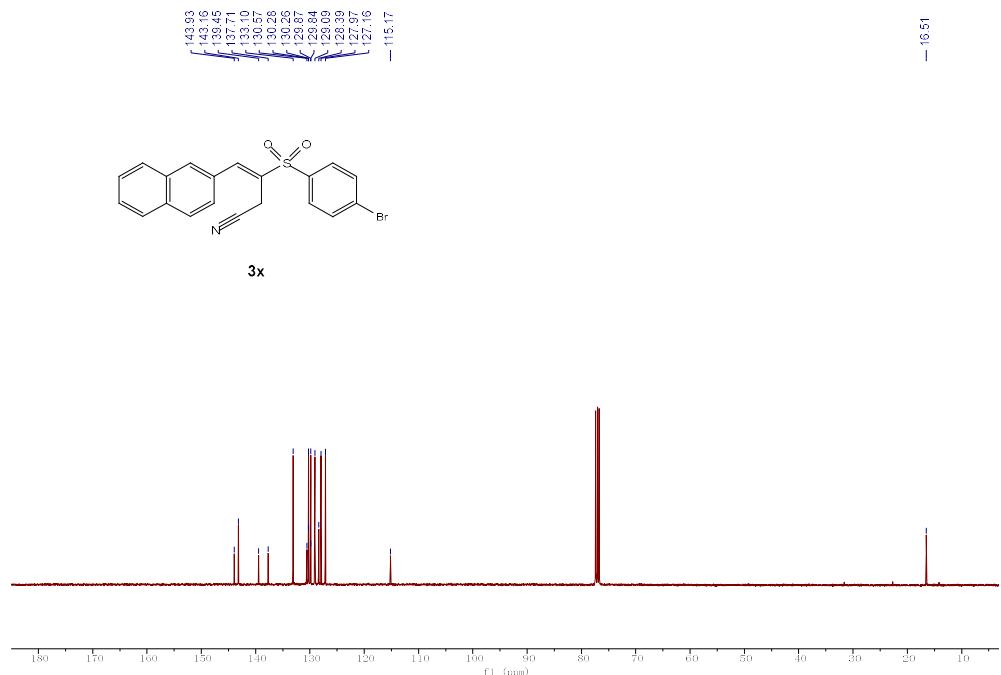


¹³C NMR Spectrum of 3v (CDCl₃, 101MHz)

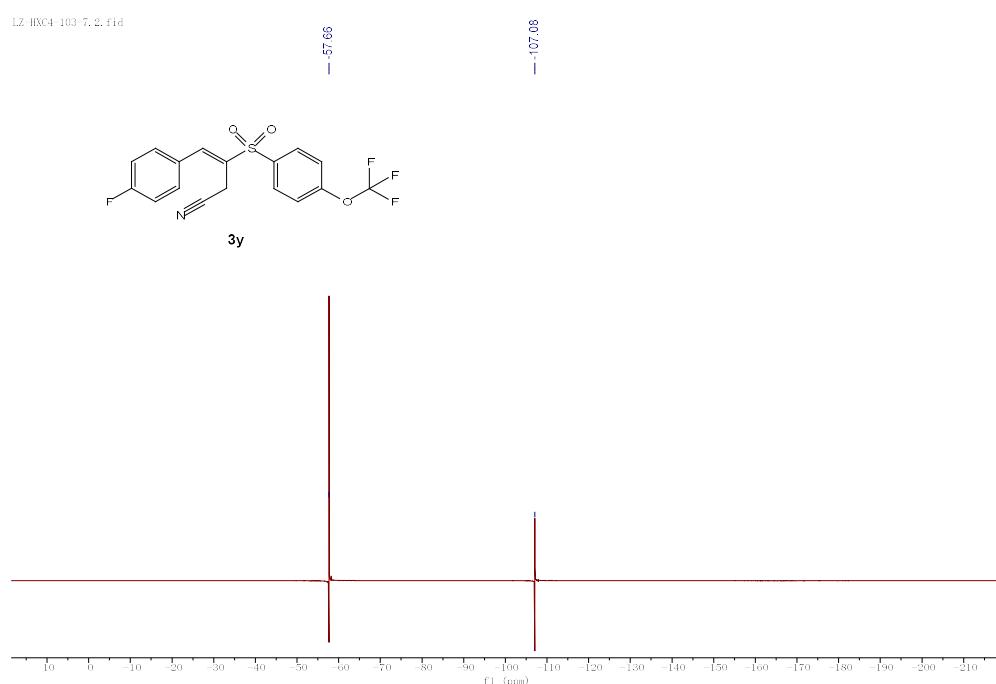
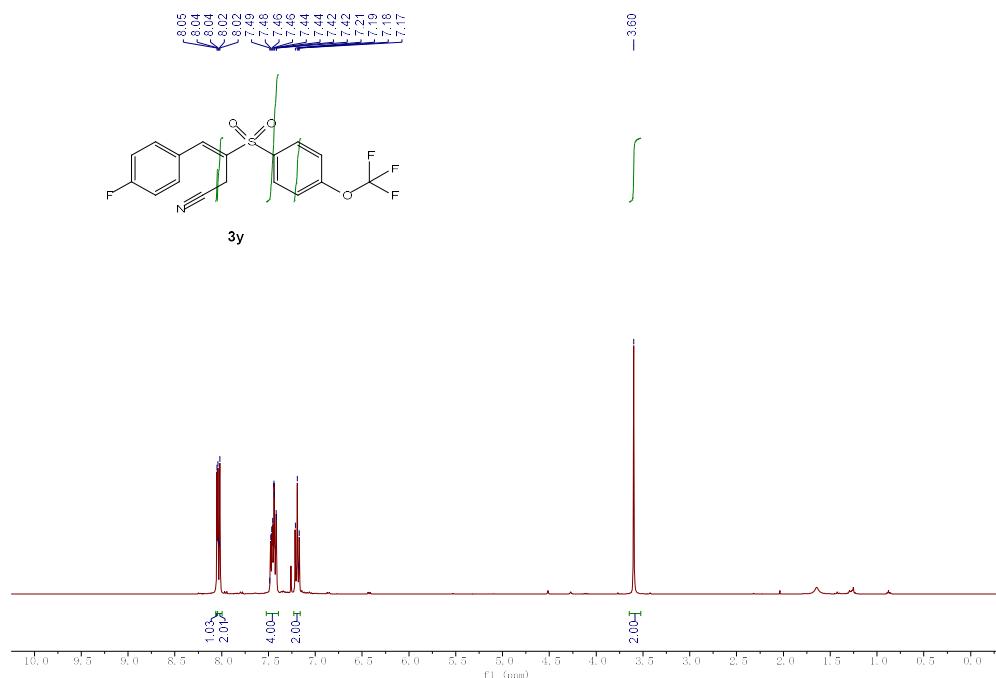


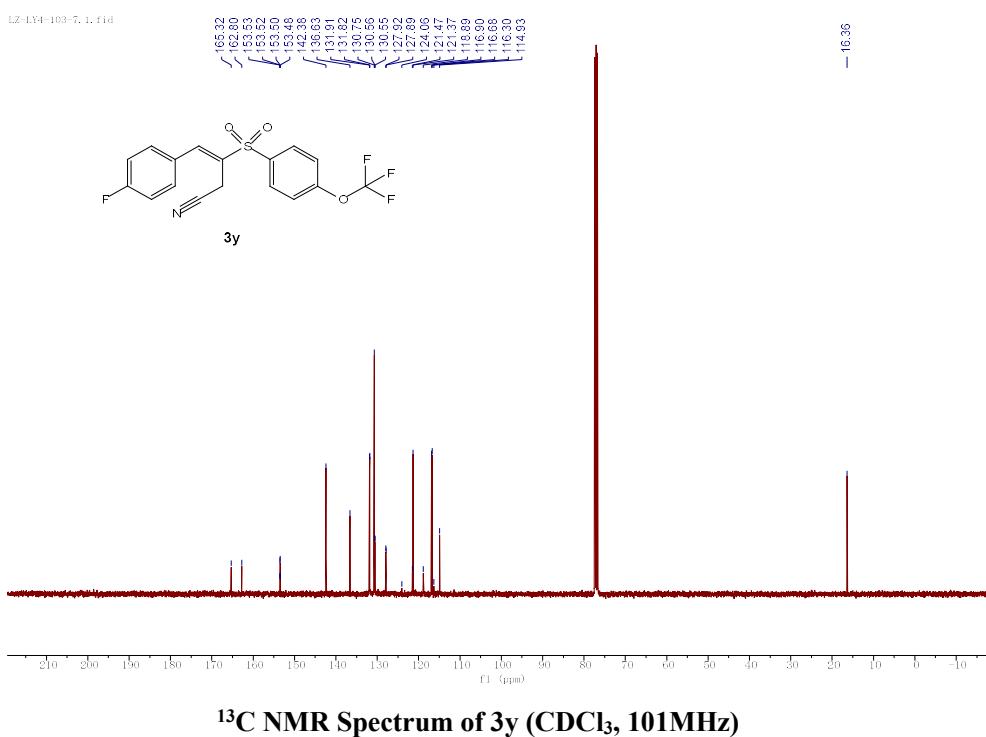


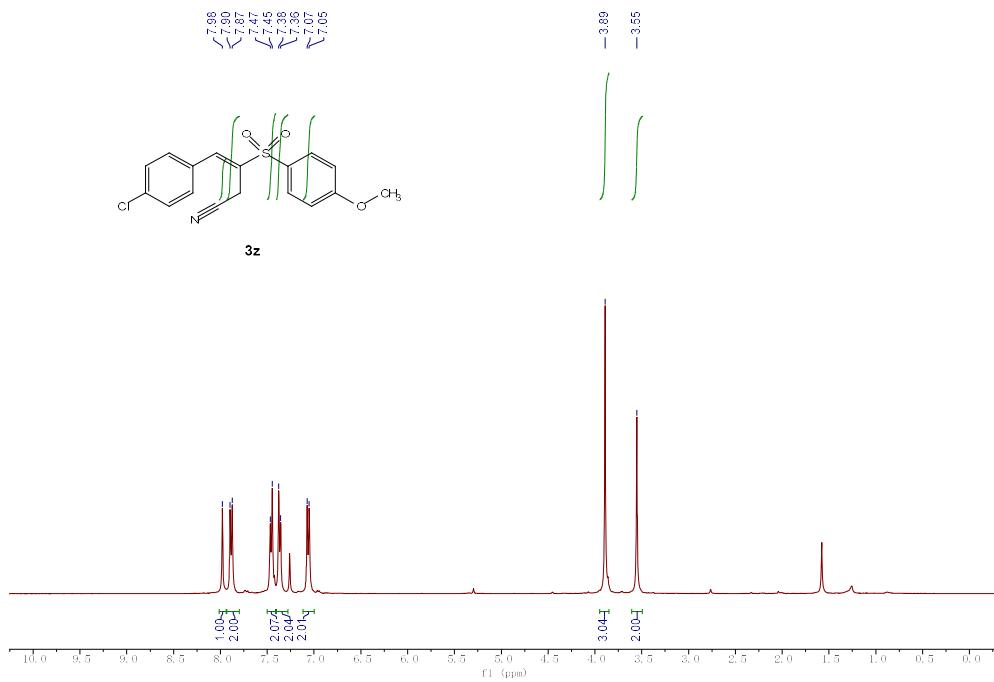
^1H NMR Spectrum of 3x (CDCl_3 , 400 MHz)



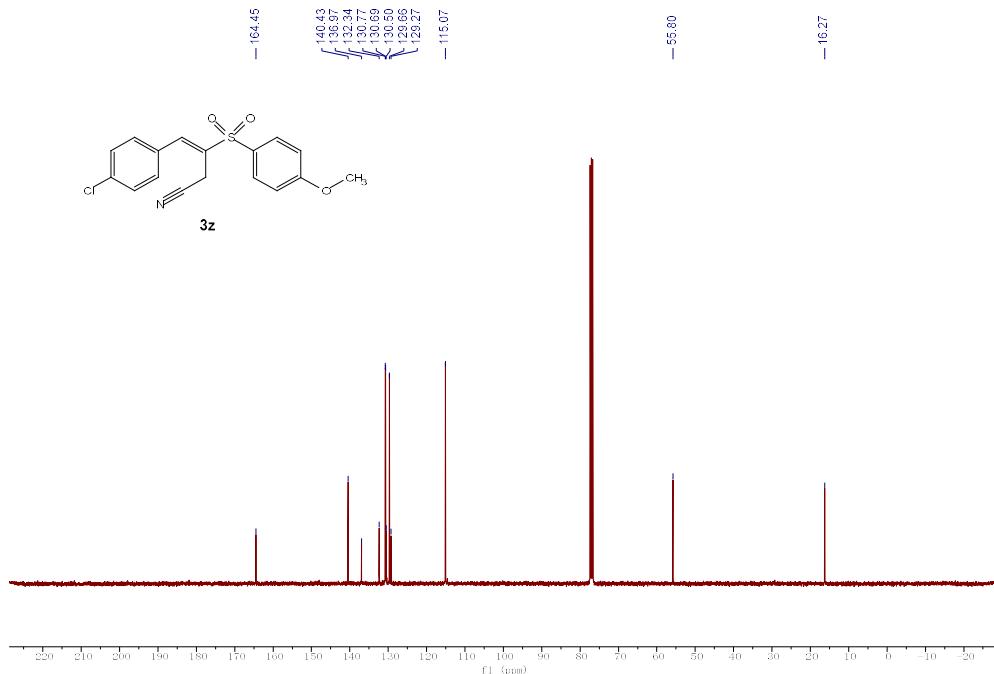
^{13}C NMR Spectrum of 3x (CDCl_3 , 101MHz)



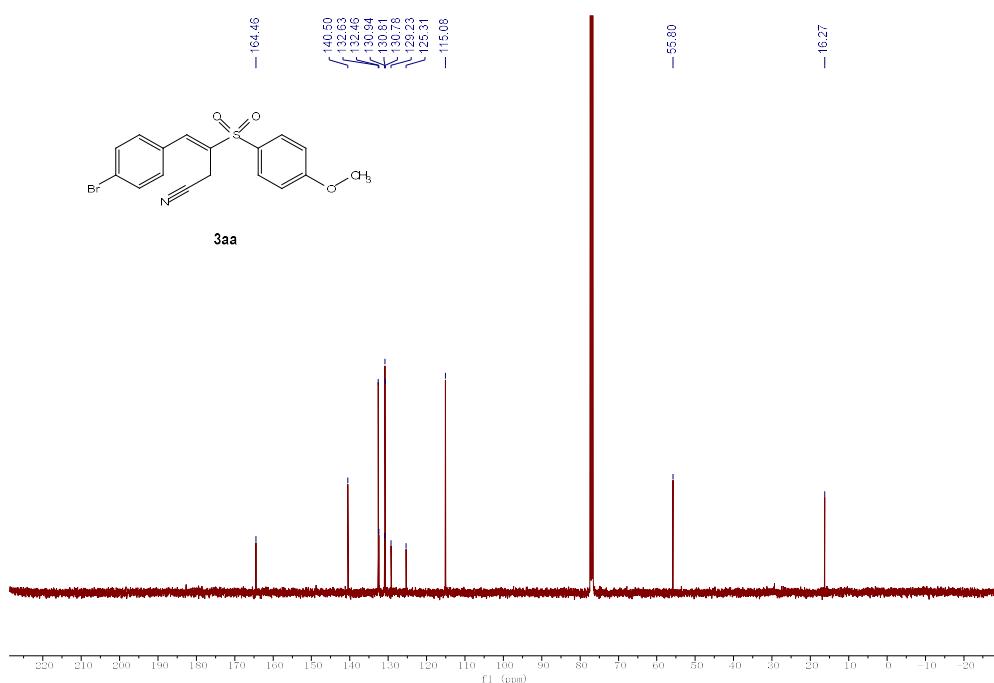
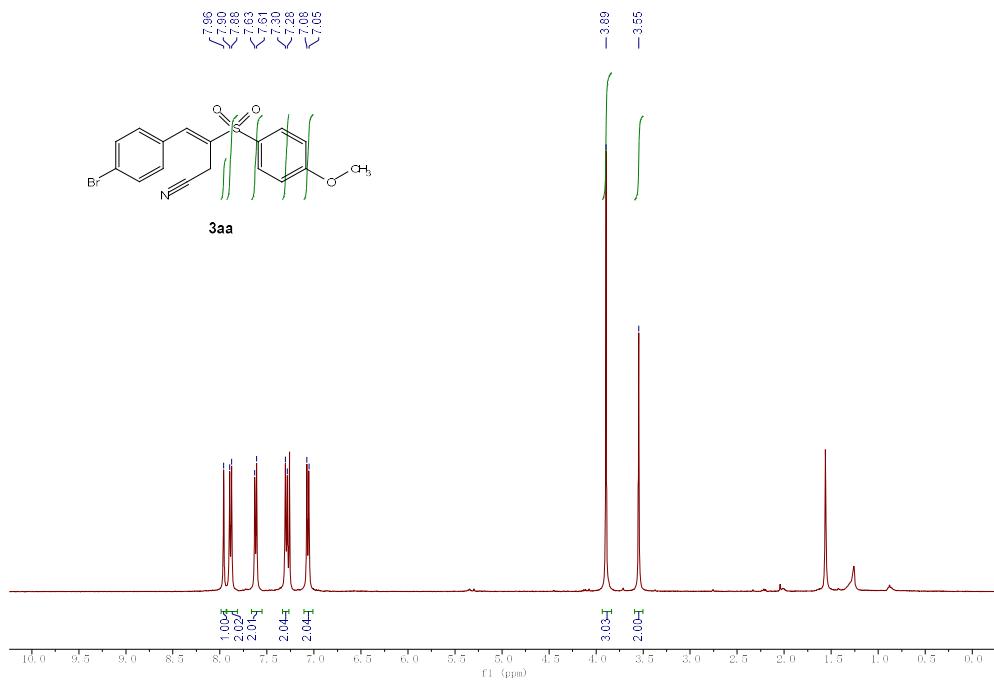


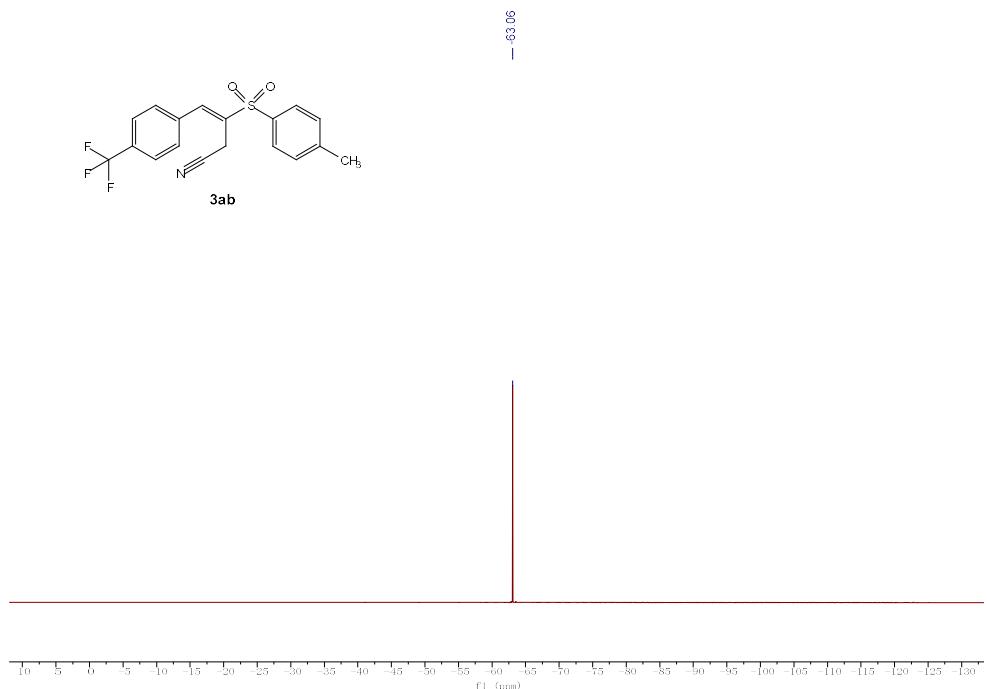
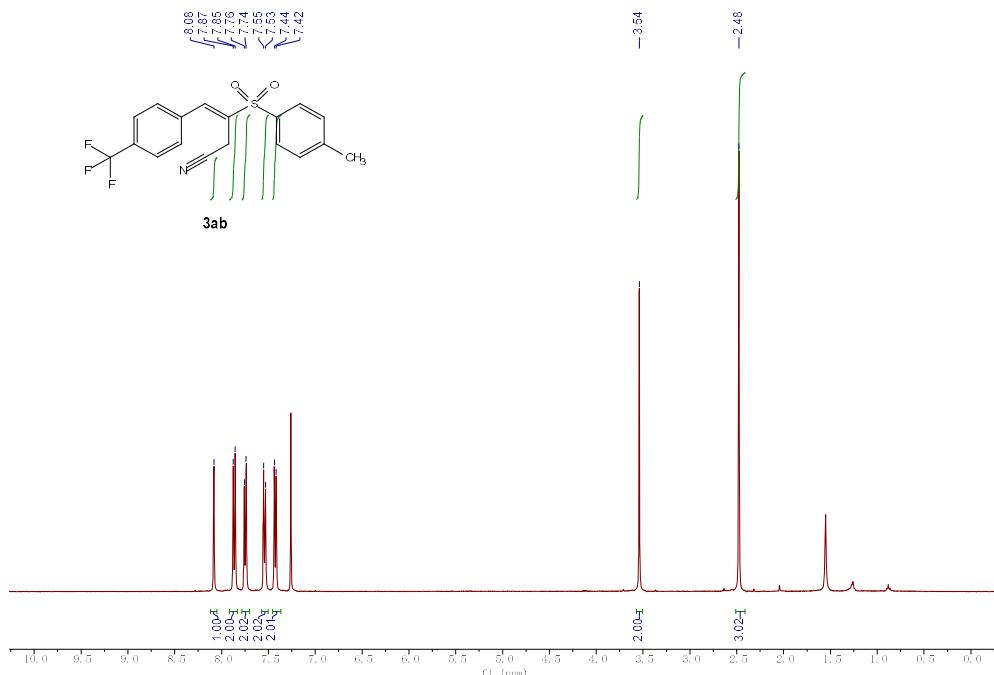


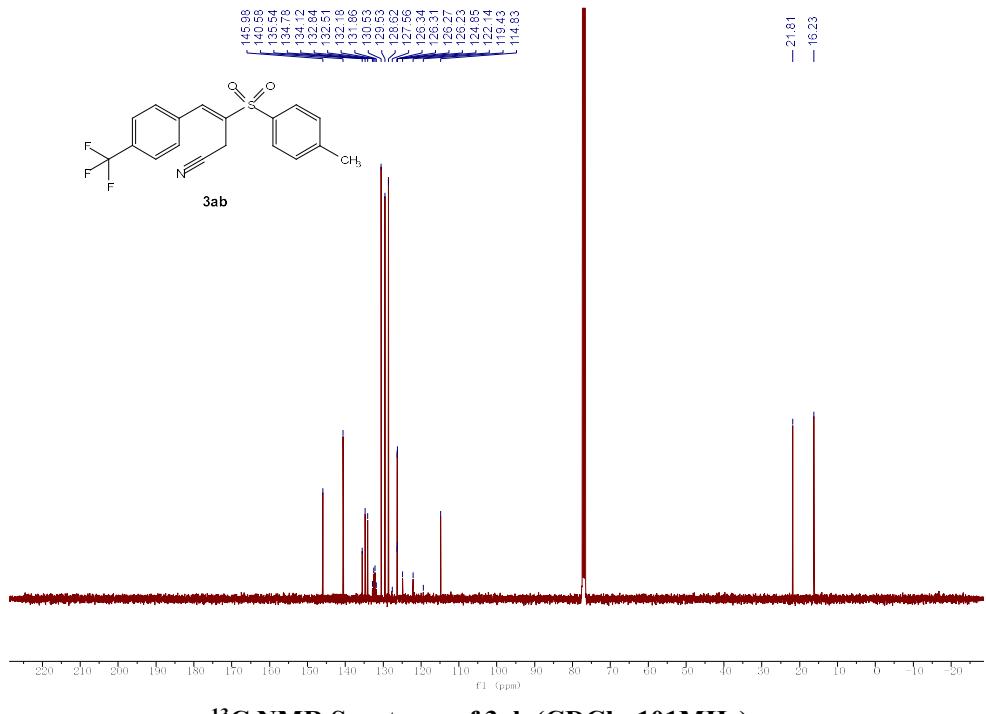
¹H NMR Spectrum of 3z (CDCl₃, 400 MHz)

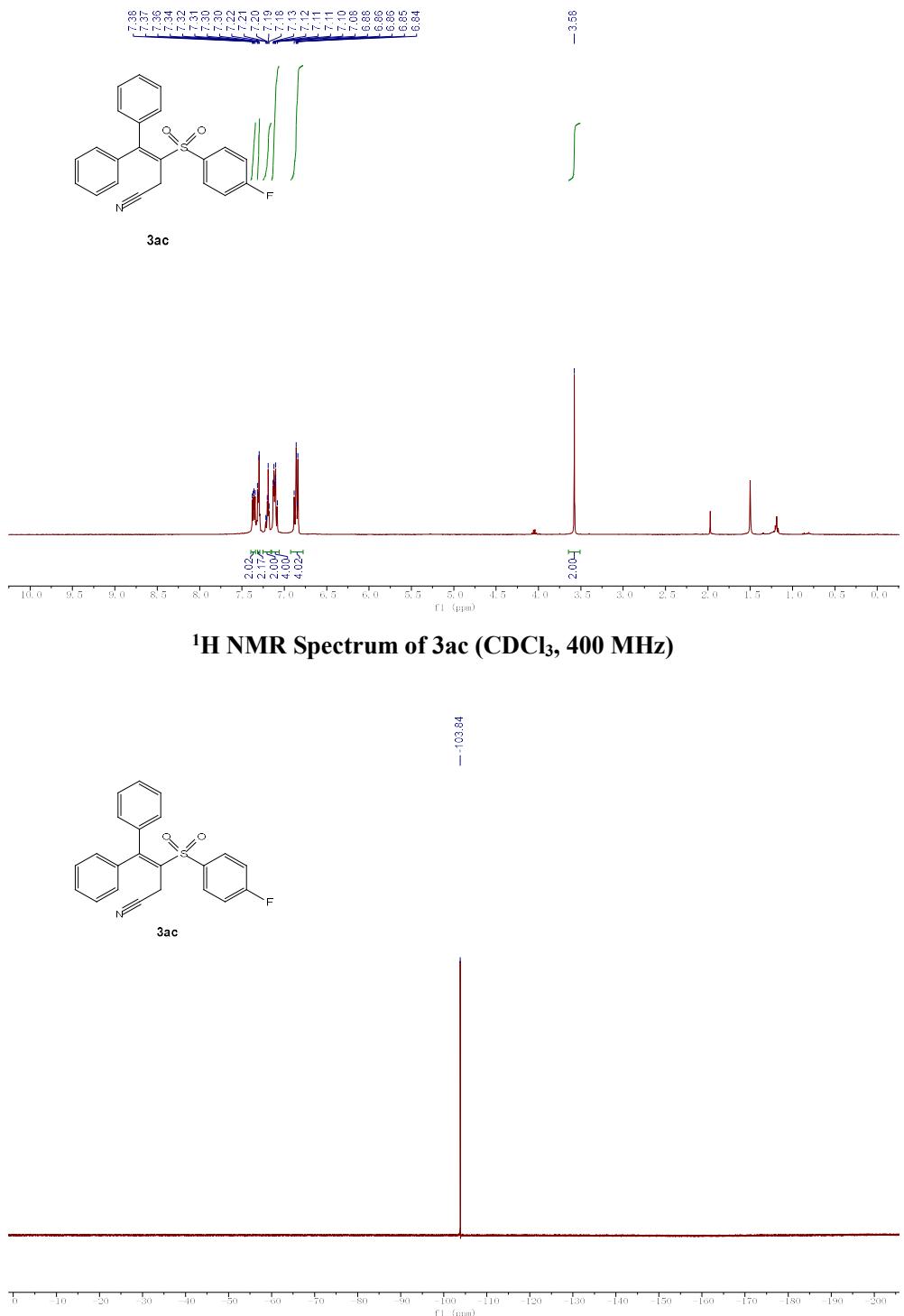


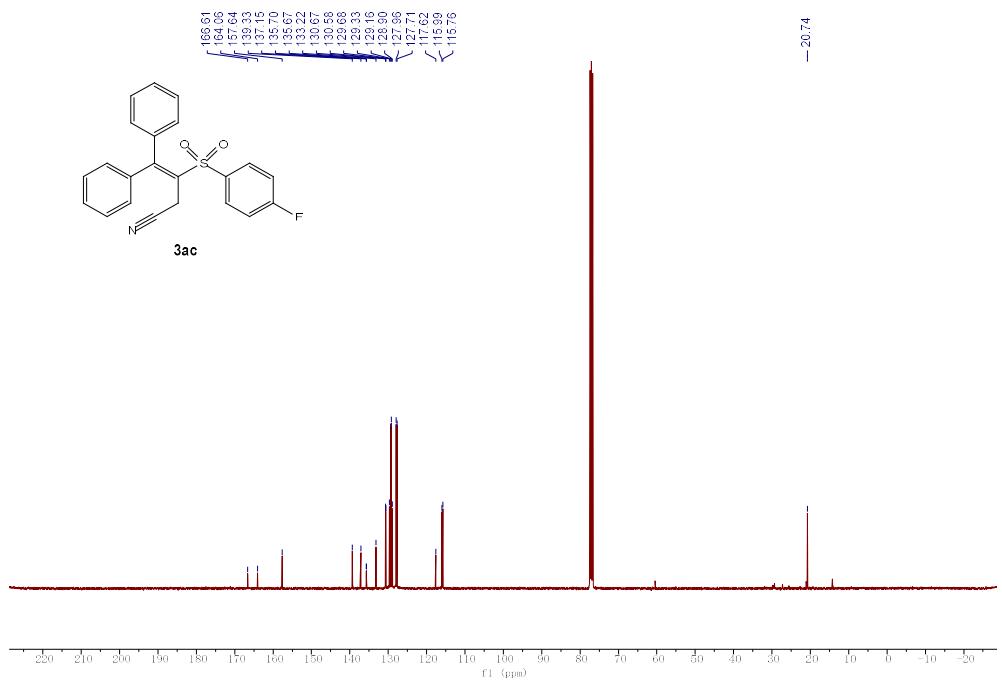
¹³C NMR Spectrum of 3z (CDCl₃, 101MHz)



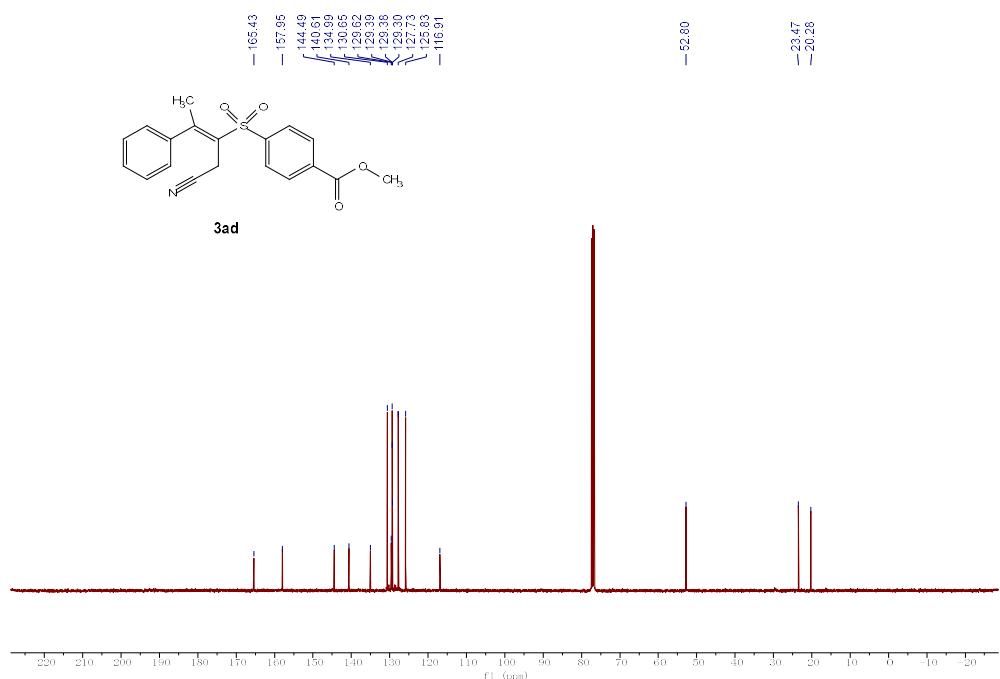
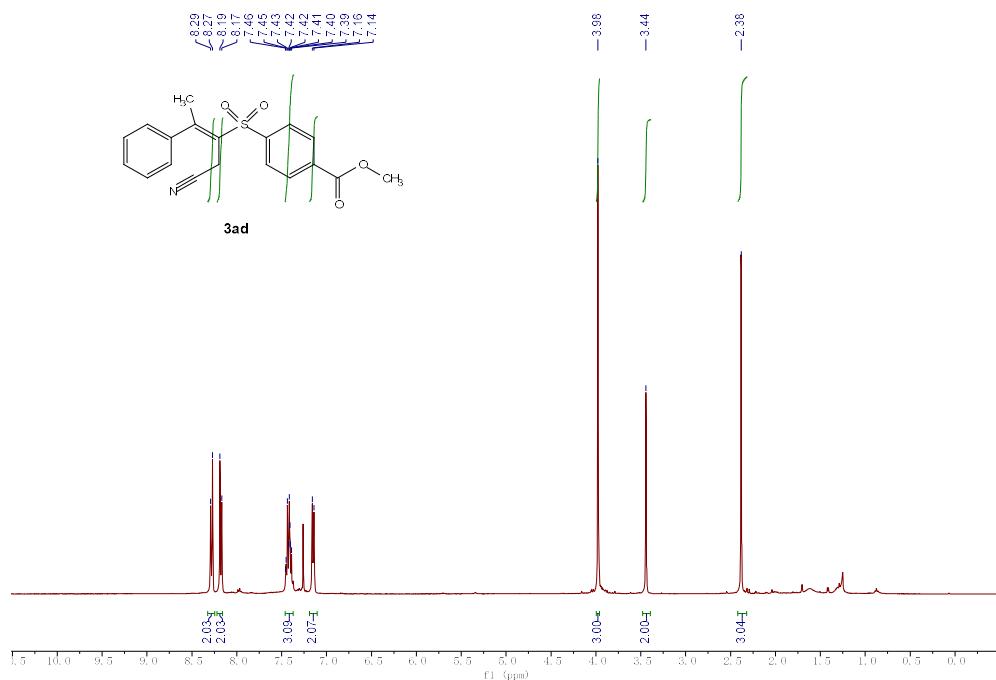




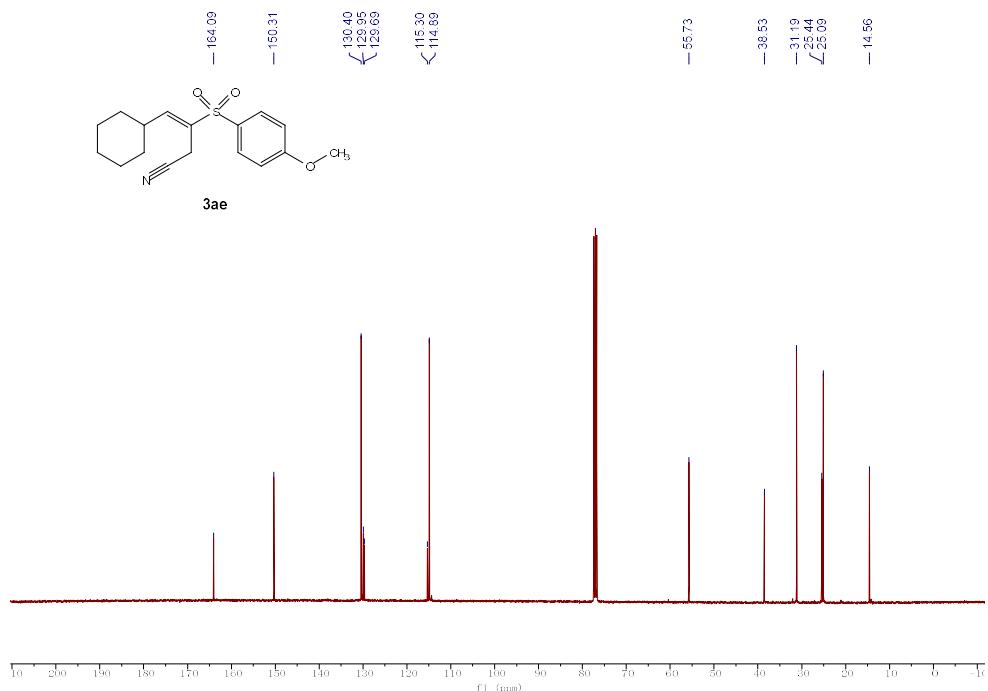
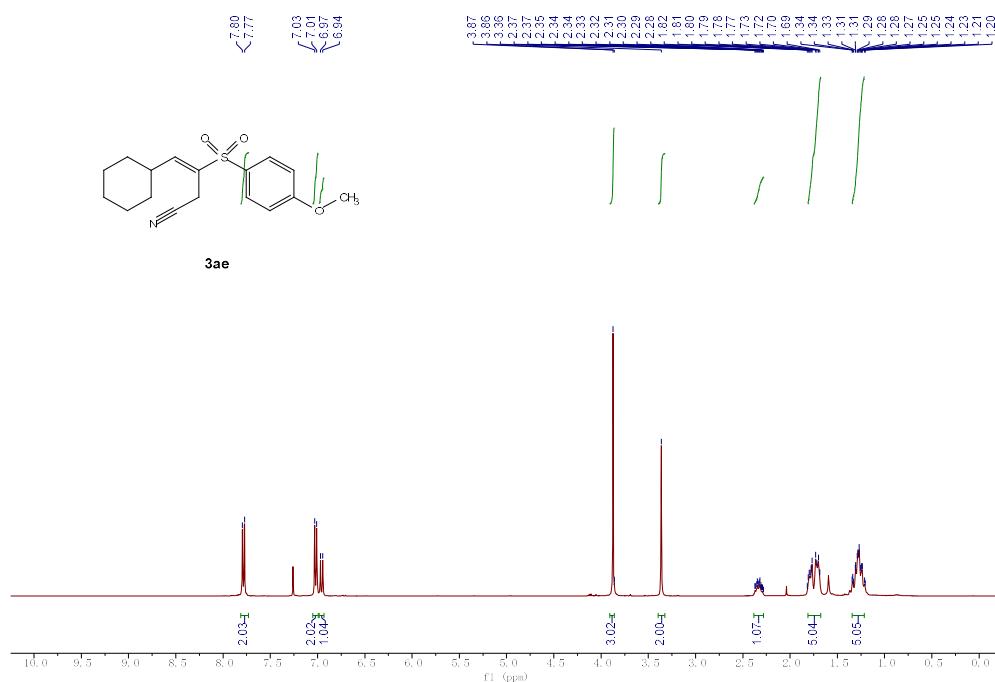


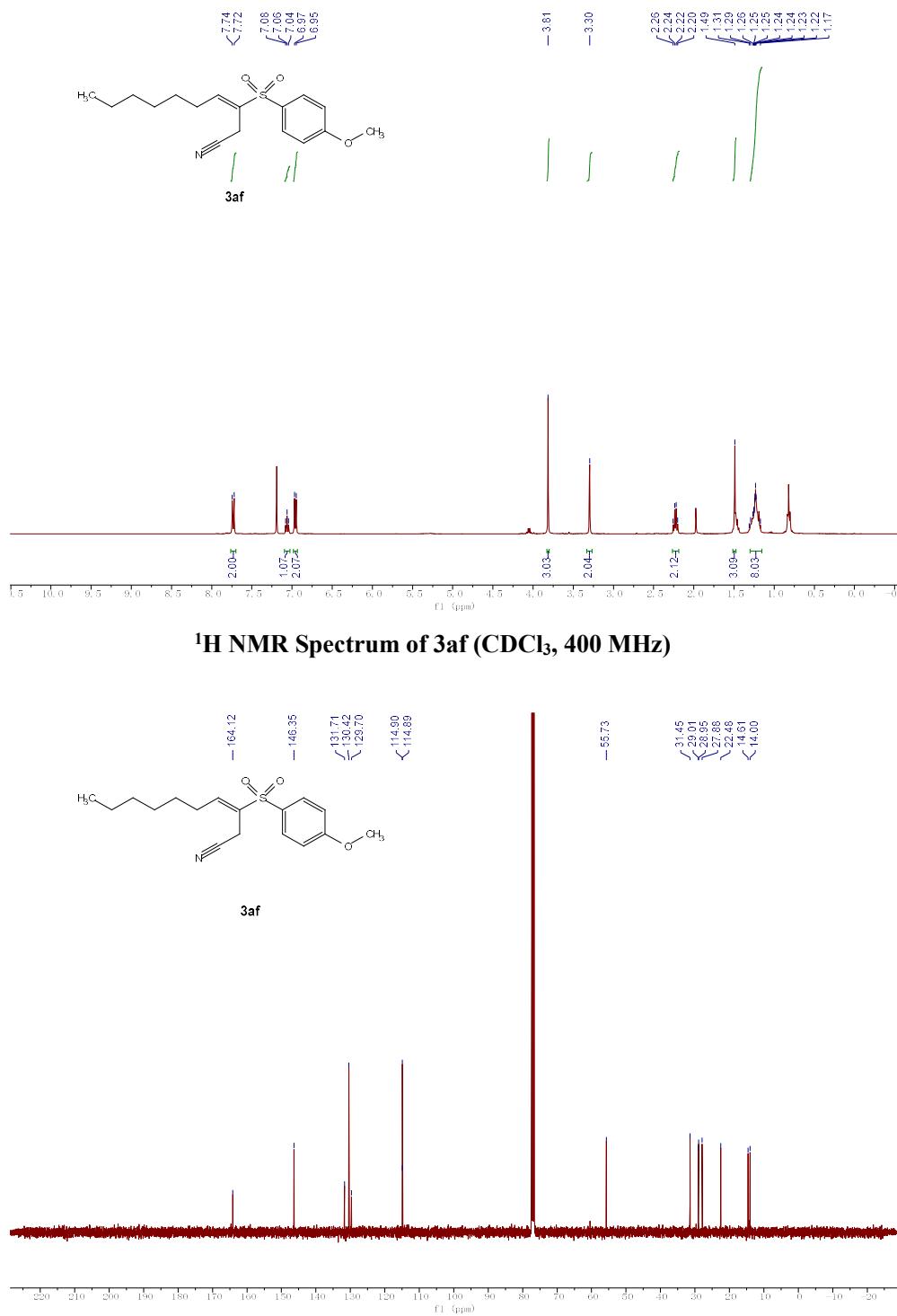


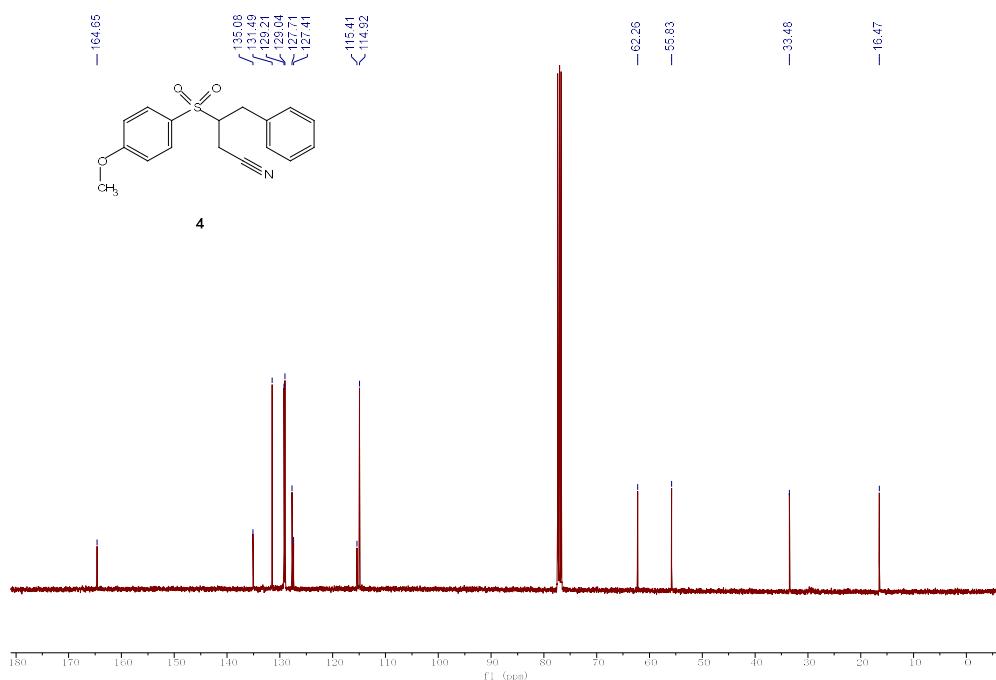
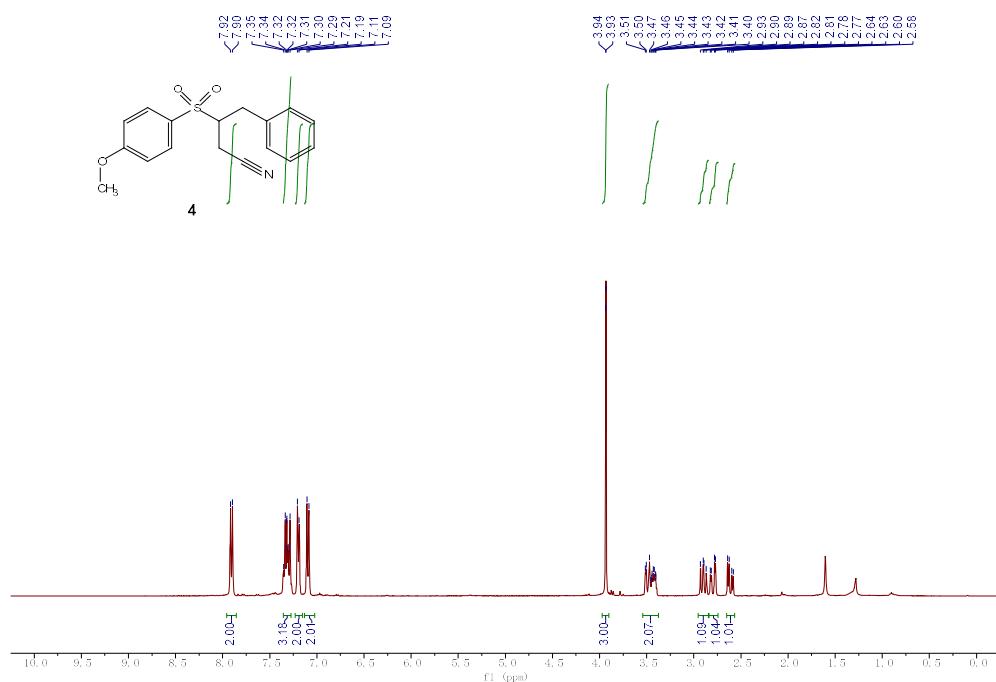
¹³C NMR Spectrum of 3ac (CDCl₃, 101MHz)

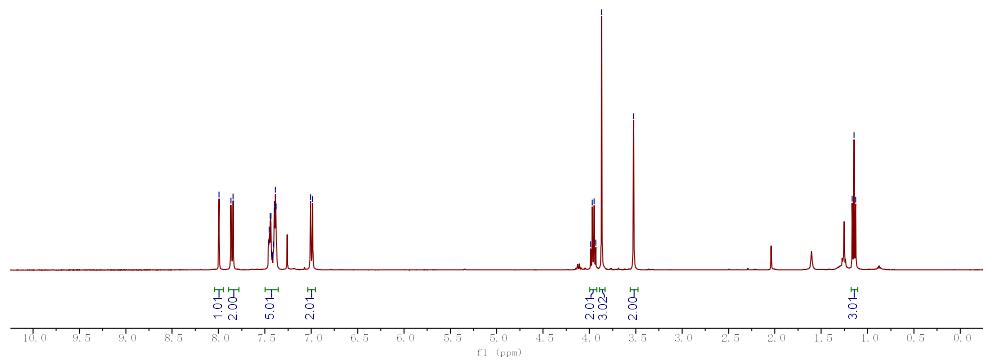
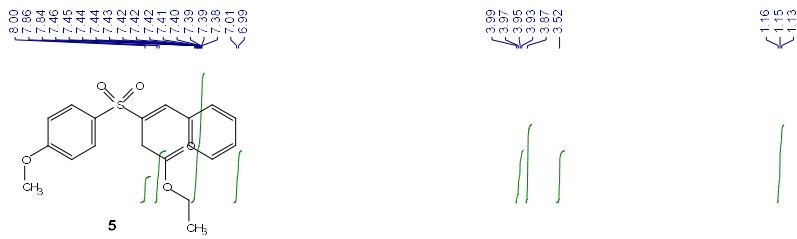


¹³C NMR Spectrum of 3ad (CDCl_3 , 101MHz)

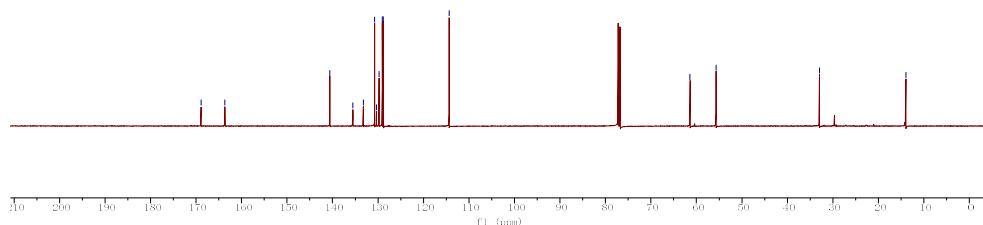
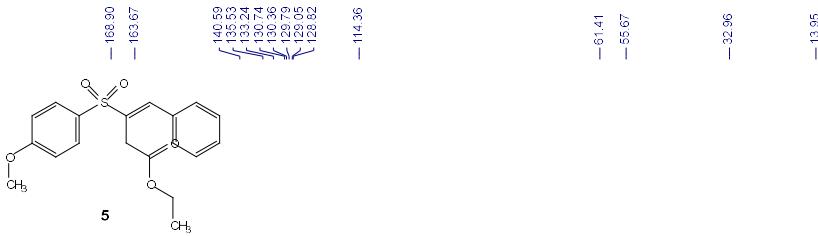








^1H NMR Spectrum of 5 (CDCl_3 , 400 MHz)



^{13}C NMR Spectrum of 5 (CDCl_3 , 101MHz)