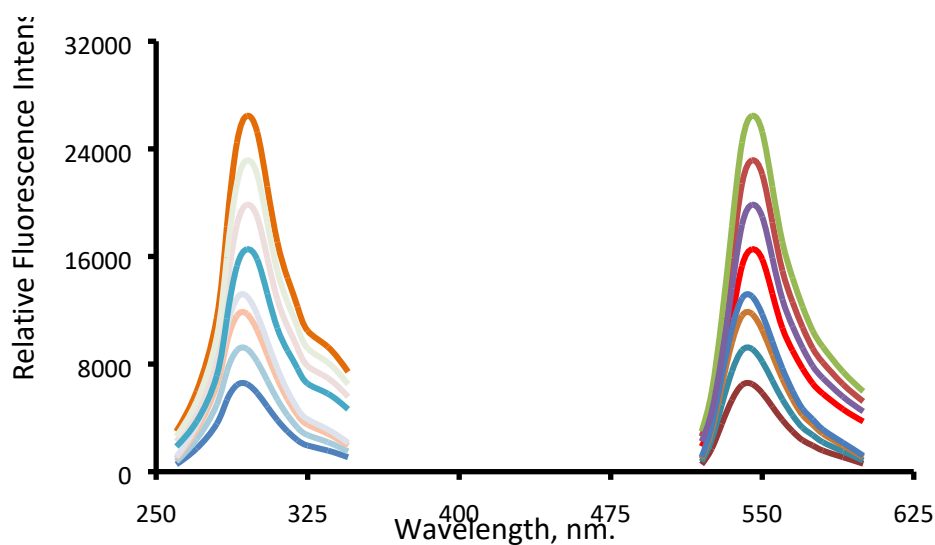
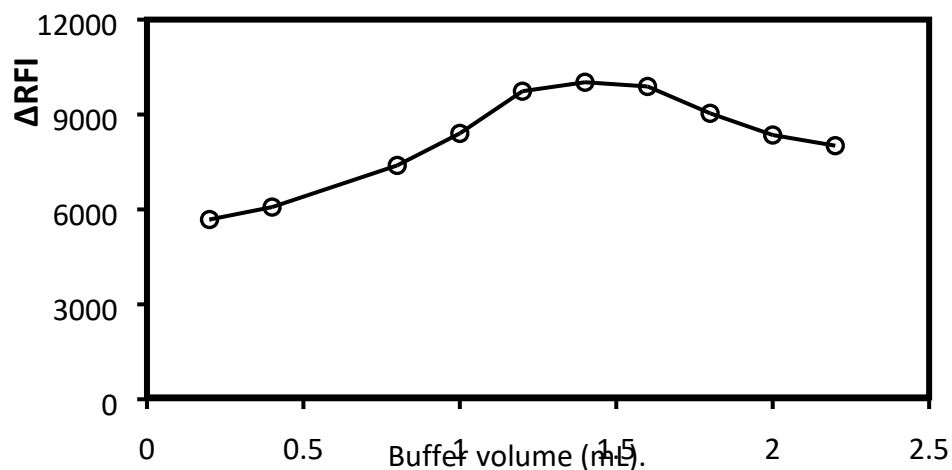


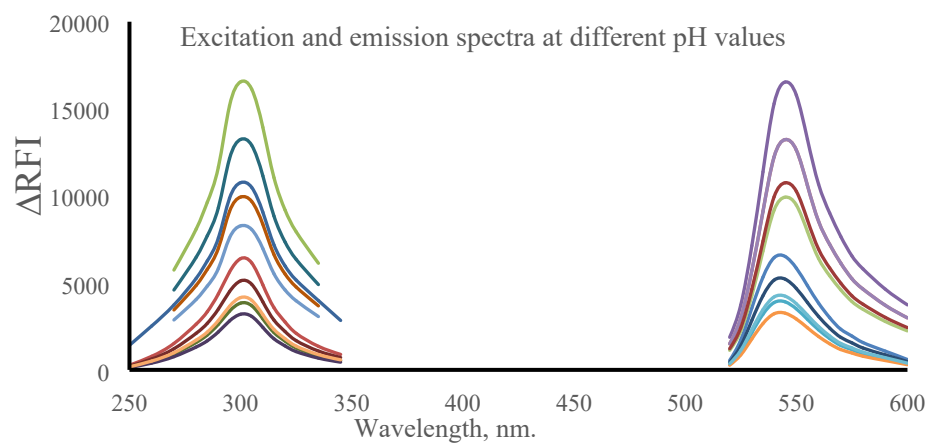
SI 1.1: Pure MTX ($1.0 \mu\text{g mL}^{-1}$) spectra (excitation and emission) in aqueous buffered solution (pH 4).



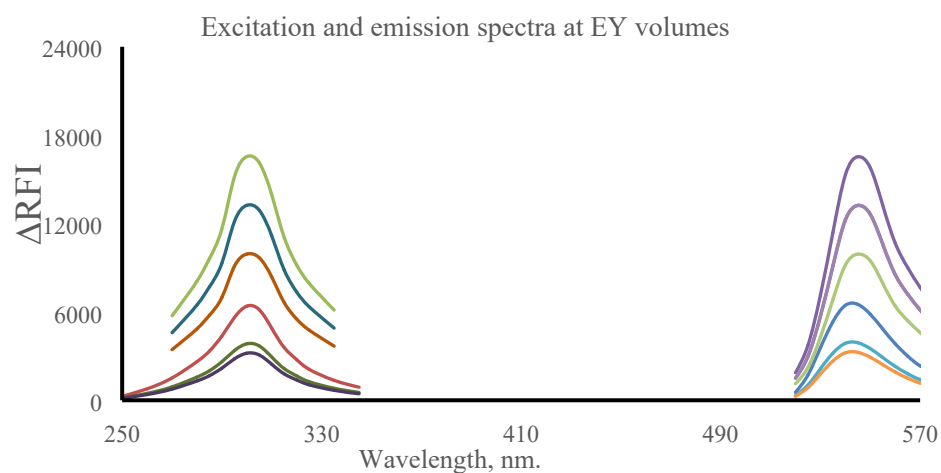
SI 1.2: Full spectra (excitation and emission) of the fully optimized MTX-EosinY system conditions at different MTX concentrations within the operating range.



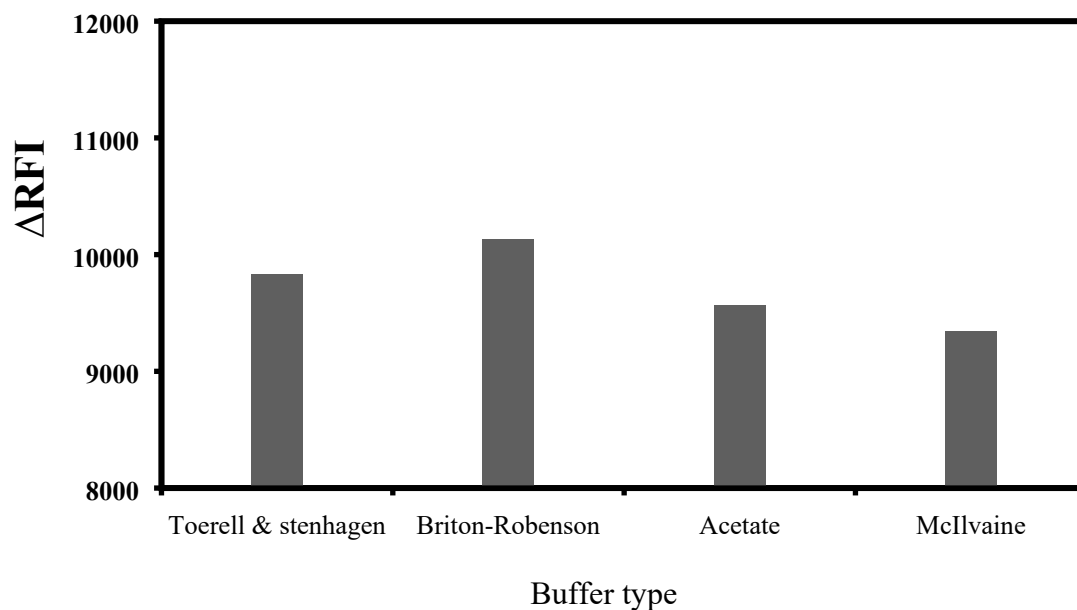
SI 2.1: Effect of buffer volume on the fluorescence quenching of the formed association complex between MTX ($1.0 \mu\text{g mL}^{-1}$) and Eosin Y.



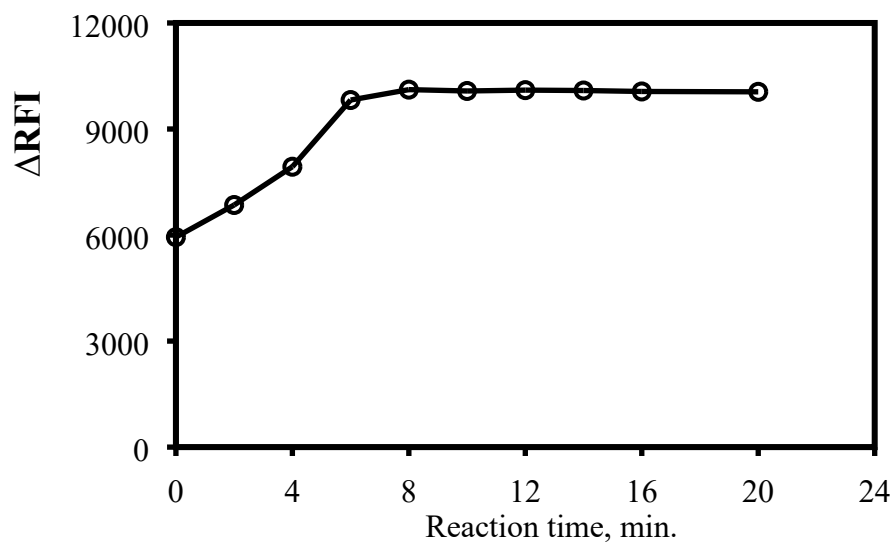
SI 2.2: Excitation and emission spectra at different pH values and buffer volumes for the formed association complex between MTX ($1.0 \mu\text{g mL}^{-1}$) and Eosin Y.



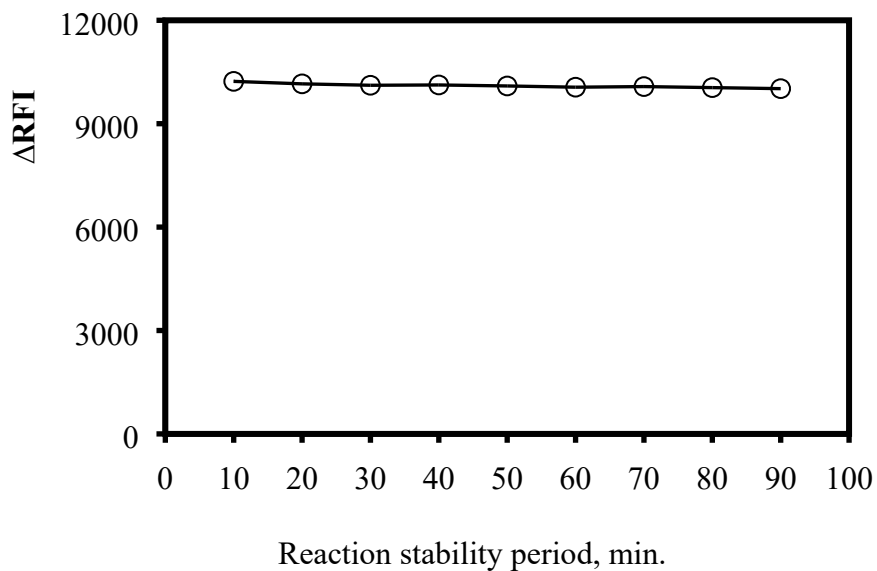
SI 2.3: Excitation and emission spectra at different volumes of EY (0.02% w/v) for the formed association complex between MTX ($1.0 \mu\text{g mL}^{-1}$) and Eosin Y.



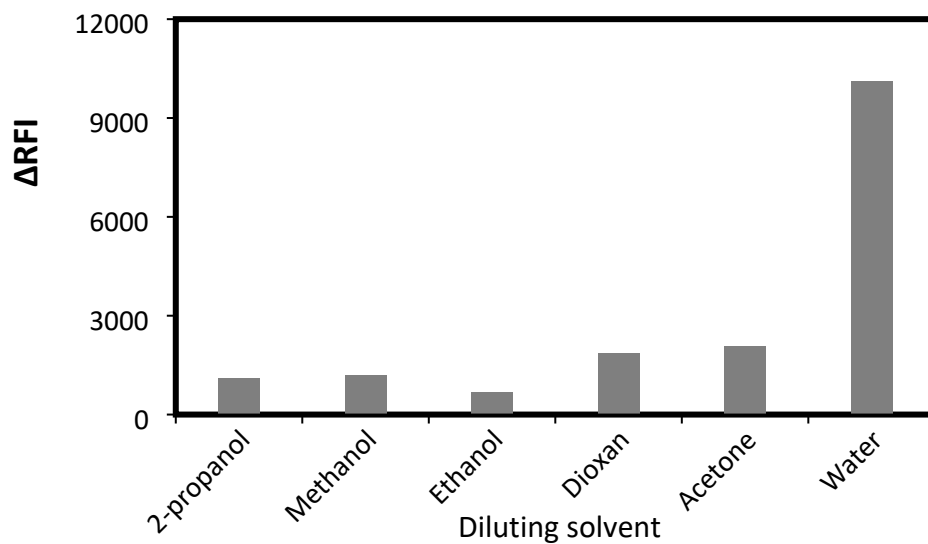
SI 3: Effect of buffer type on the fluorescence quenching of the formed association complex between MTX ($1.0 \mu\text{g mL}^{-1}$) and Eosin Y.



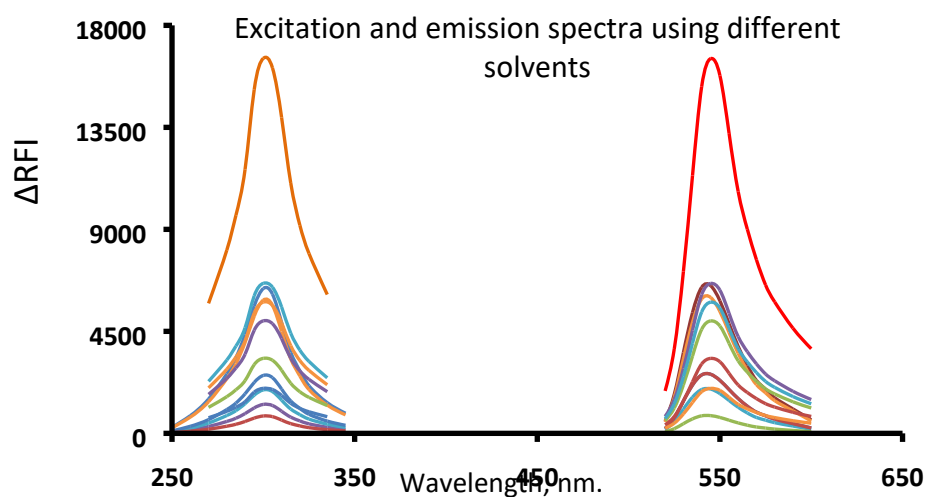
SI 4: Effect of reaction time on the fluorescence quenching of the formed association complex between MTX ($1.0 \mu\text{g mL}^{-1}$) and Eosin Y.



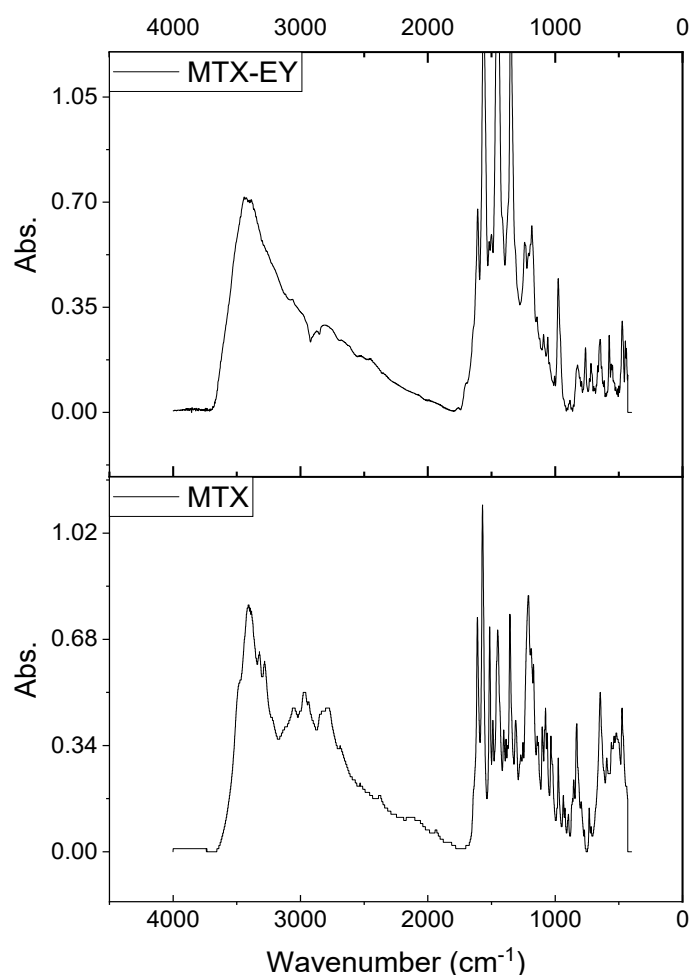
SI 5: Effect of stability time on the fluorescence quenching of the formed association complex between MTX ($1.0 \mu\text{g mL}^{-1}$) and Eosin Y.



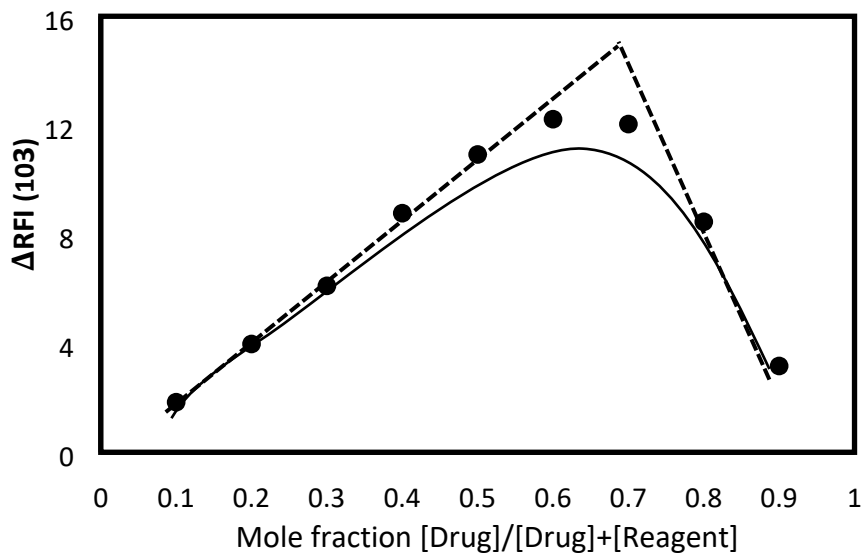
SI 6.1: Effect of diluting solvent on the fluorescence quenching of the formed association complex between MTX ($1.0 \mu\text{g mL}^{-1}$) and Eosin Y.



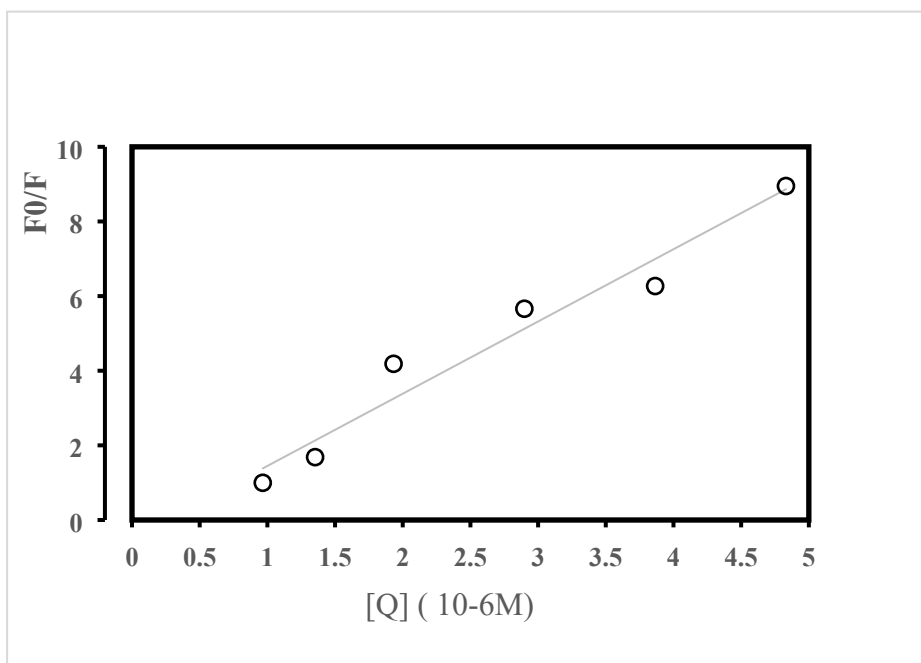
SI 6.2: Full excitation and emission spectra showing the effect of different diluting solvents on the formed association complex between MTX ($1.0 \mu\text{g mL}^{-1}$) and Eosin Y.



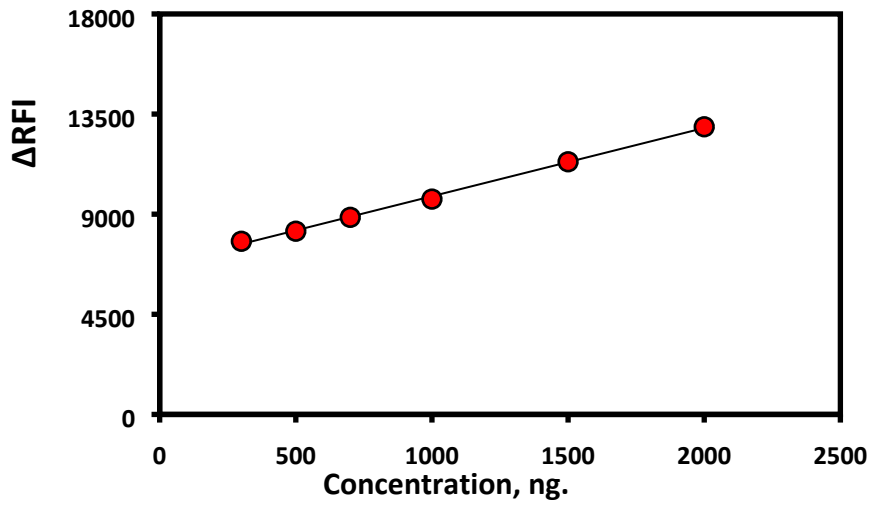
SI 7: FT-IR spectra for MTX and formed MTX-EY complex



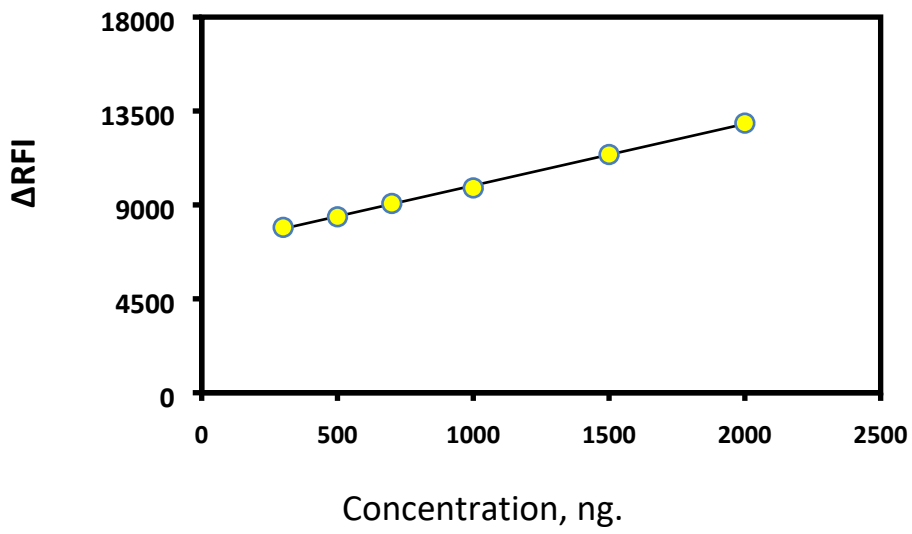
SI 8: Job's plot for association complex formation between MTX and eosin using an equimolar concentration (2.8×10^{-4} M) of master solutions.



SI 9: Stern–Volmer plot describing Eosin Y quenching caused by MTX.



SI 10: Calibration curves for MTX in plasma.



SI 11: Calibration curves for MTX in urine

SI Table 1: Comparable cases showing quenching of fluorescence upon association complex formation with MTX and other compounds with Eosin Y and other fluorescence active dye.

drug	Dye or reagent	method	mechanism	application	$\lambda_{ex}/\lambda_{em}$ (nm)	Ref.
MTX	MSA-CdT-QDs as fluorescence probe	Fluorimetry	Fluorescence quenching	Pure form and human serum sample	365/500-700	¹
MTX	AuNPs probe by nanometal surface energy transfer.	Fluorimetry	quenching nature of the MTX-	Live cells biological samples	633	²
MTX	molybdate	RRS	RRS based on an ion-association complex between MTX and molybdate	serum and urine	365/365	³
ABZ, FBZ & MBZ	Erythrosine B	Fluorimetry	Fluorescence quenching	Bulk powder, tablets, suspension and human plasma	517/544	⁴⁻⁶
Mebeverine	Eosin y	Fluorimetry	Fluorescence quenching	commercial tablets	390/540	⁷
Fluoxetine and paroxetine	Eosin Y	Fluorimetry	Fluorescence quenching	Bulk powder and pharmaceutical formulations	301/545	⁸
Amlodipine and nicardipine	Eosin Y	fluorimetry	Fluorescence quenching	Powder and tablets	549	⁹

SI Table 2: polarity index and dielectric constant for solvents used.

Solvent	Polarity index	Dielectric constant
Water	9.0	80.2
Methanol	6.6	32
Acetone	5.4	20.7
Ethanol	5.2	24.8
Dioxane	4.8	2.2
2-propanol	4.2	19.9

SI Table 3: Comparison between the sensitivity of the current fluorescence method and the other techniques in utilized in the determination of MTX.

Technique	Range	Ref
HPLC	1-2000 ng mL ⁻¹ (serum)	10
HPLC	5-1000 ng ml ⁻¹ (plasma) 25-2000 ng ml ⁻¹ (mouse tissue)	11
HPLC	5–1000 ngml ⁻¹ (plasm) 25–2500 ngml ⁻¹ (liver) 12.5–2500 ng ml ⁻¹ (other tissue)	12
CE	$7.0 \times 10^{-8} - 1.0 \times 10^{-6}$ M	13
CL	$5.0 \times 10^{-9} - 1.0 \times 10^{-7}$ g/ml	14
Polarography	$5 \times 10^{-7} - 2.5 \times 10^{-5}$ M	15
Voltammetry	$2.0 \times 10^{-7} - 6.0 \times 10^{-6}$ mol dm ⁻³	16
ELISA	0.25-50 ng ml ⁻¹	17
Spectrophotometry	4-10 mg L ⁻¹	18
Spectrofluorimetry	0.0675–0.337 μM	1
SERRS	$2.5 \times 10^{-9} - 1 \times 10^{-6}$ M.	19
Spectrofluorimetry	70–2500 ng ml ⁻¹ 0.3–2 μg ml ⁻¹ (biofluids)	Current method

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