

Electronic Supplementary Information

2,7-Diazapyrenes: a brief review on synthetic strategies and application opportunities

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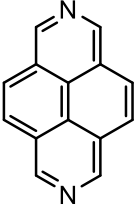
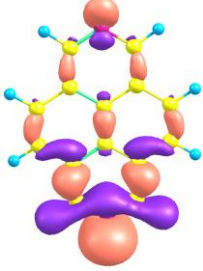
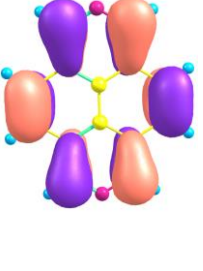
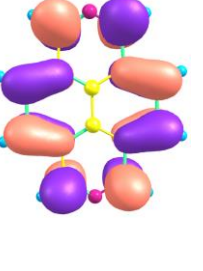
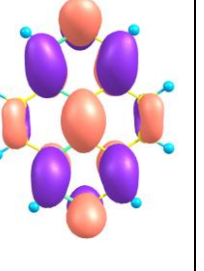
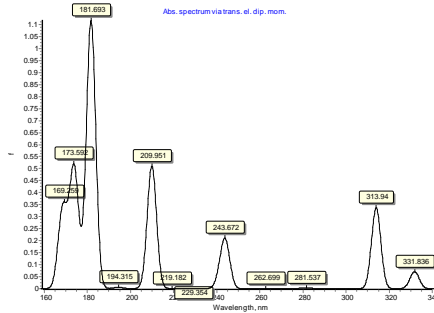
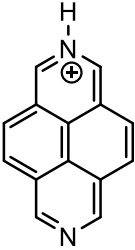
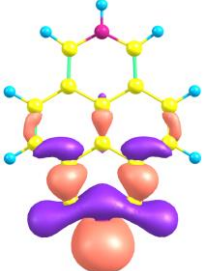
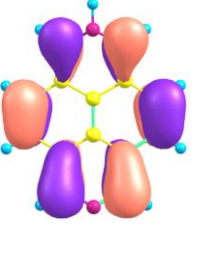
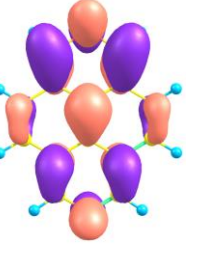
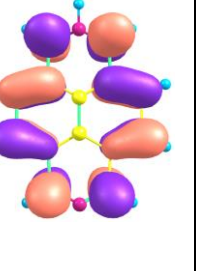
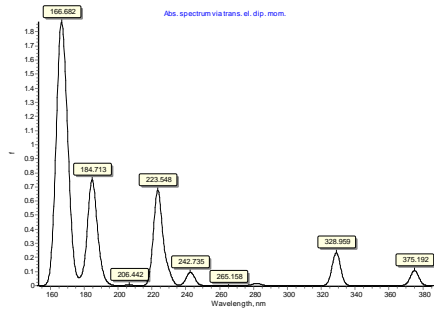
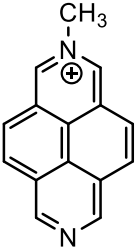
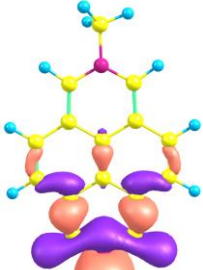
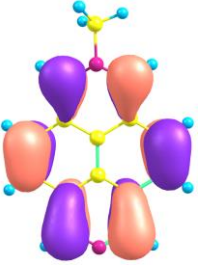
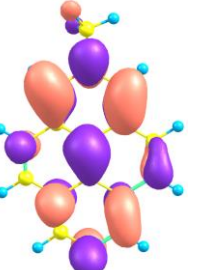
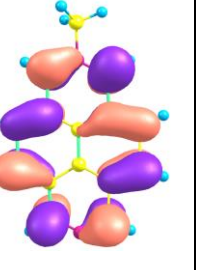
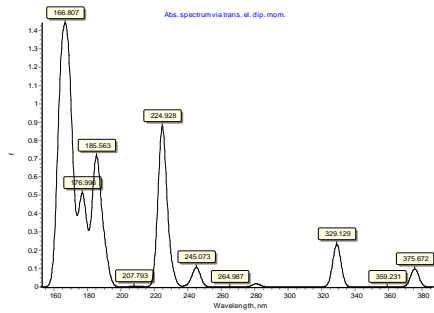
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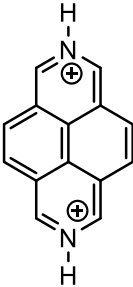
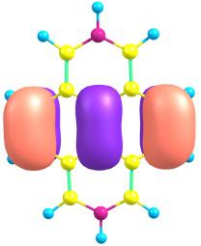
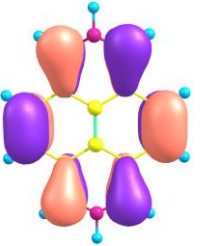
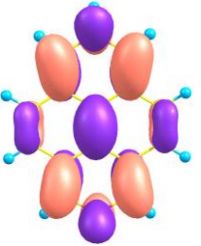
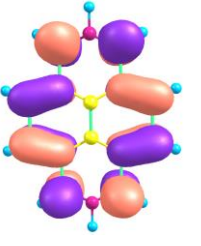
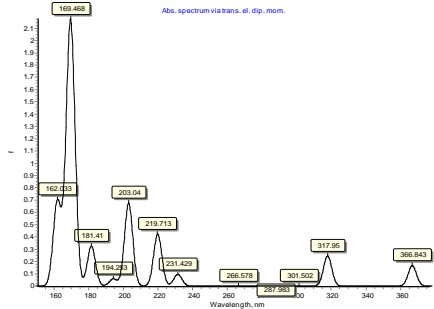
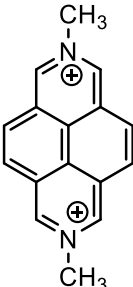
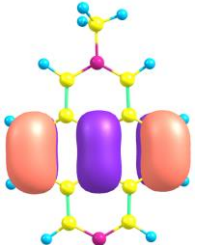
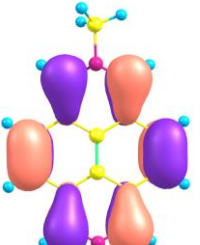
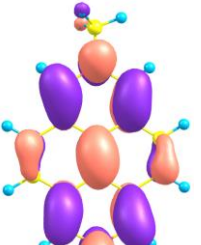
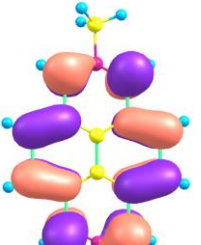
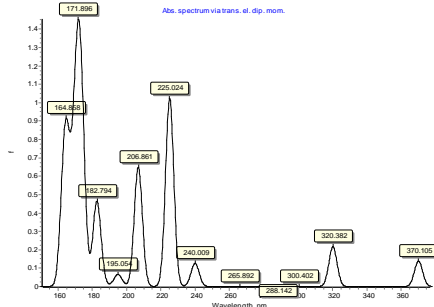
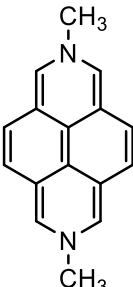
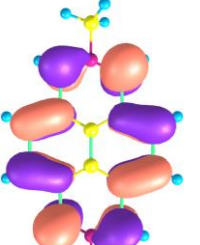
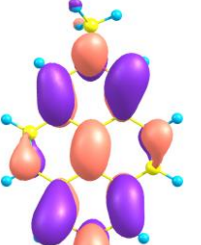
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The calculations were carried out as reported earlier [1-4] in the Orca 4.0.1 software package using the DFT B3LYP, 6-311G * method. The calculations in a solvent were carried out by the CPCM solvation method [5].

Table S1. HOMO-LUMO distributions of 2,7-diazapyrenes in a gas phase

#	Structure	HOMO-1	HOMO	LUMO	LUMO+1	E ₀₋₀	Absorption bands
1.		-7.1346 	-5.9814 	-2.1156 	-1.6774 	3.8658	
2.		-10.5139 	-9.9152 	-6.3504 	-6.1311 	3.5648	
3.		-10.3920 	-9.7740 	-6.1724 	-5.9879 	3.6016	

4.		-15.0642 	-13.9682 	-10.4358 	-10.0694 	3.5324	
5.		-14.6932 	-13.5723 	-9.9681 	-9.6769 	3.6042	
6.			-2.5108 	-1.4594 		1.0514	

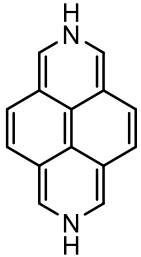
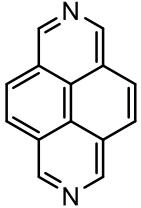
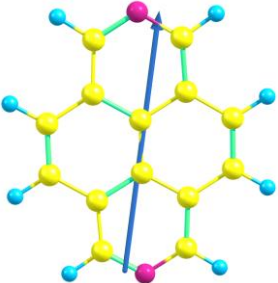
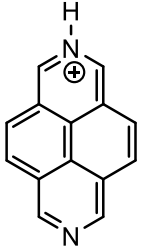
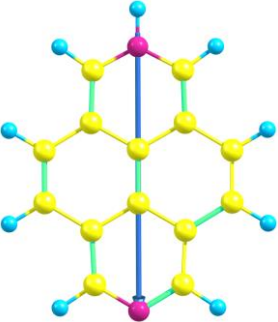
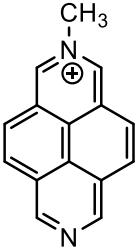
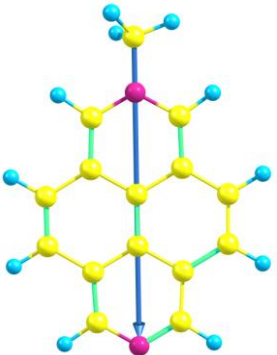
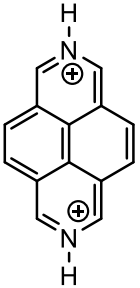
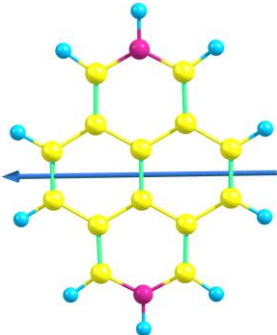
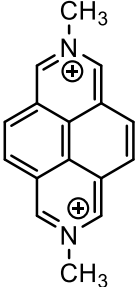
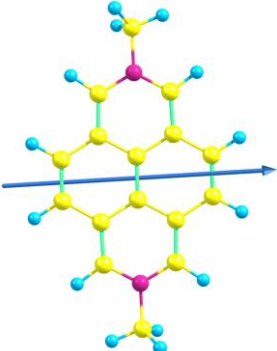
7.			-2.6134	-1.4975		1.1159	
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Table S2. Dipole moments of 2,7-diazapyrenes in a gas phase

#	Structure	Dipole moment, Debye
1		0.00000 
2		12.21599 

3	 <p>CH₃</p>	<p>11.76324</p> 
4	 <p>H</p> <p>H</p>	<p>0.00000</p> 
5	 <p>CH₃</p> <p>CH₃</p>	<p>0.00000</p> 

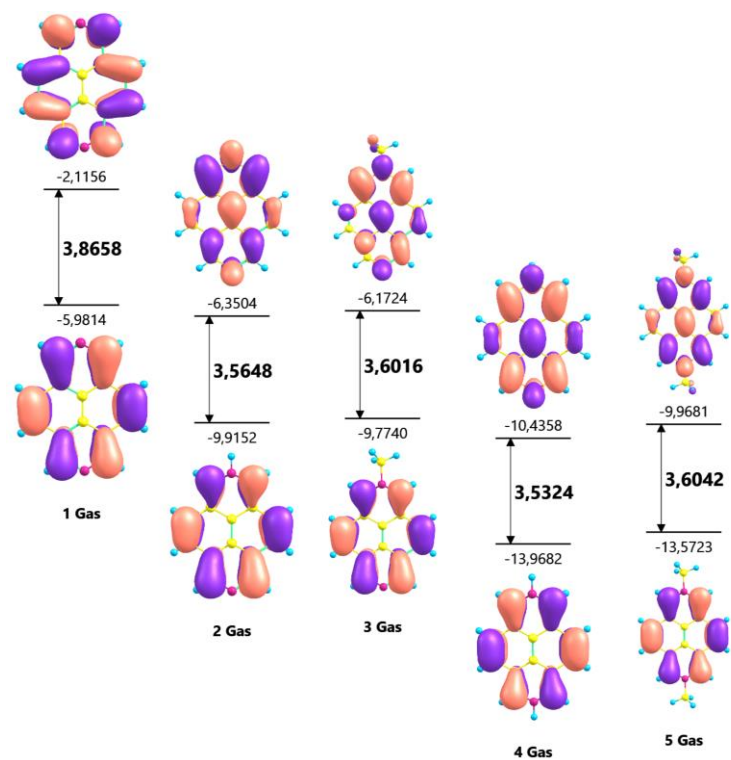
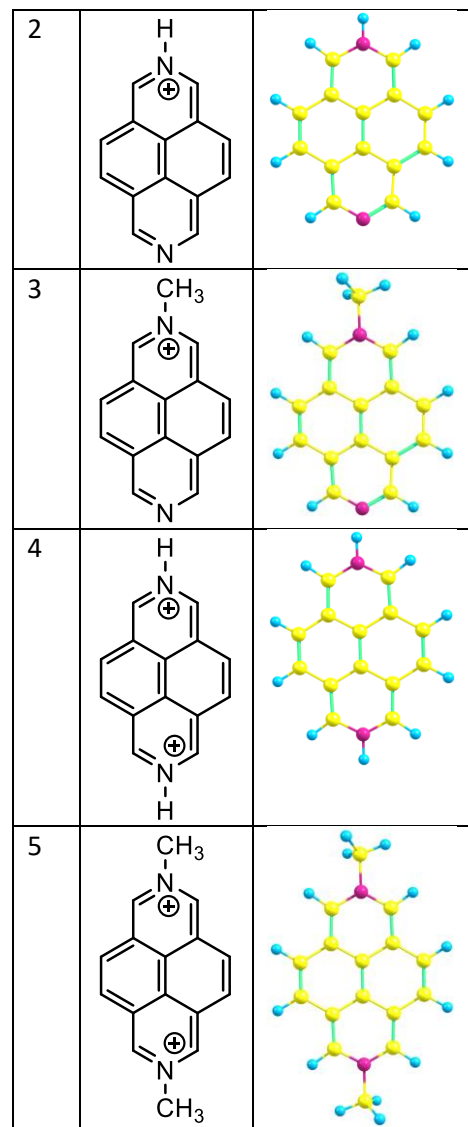
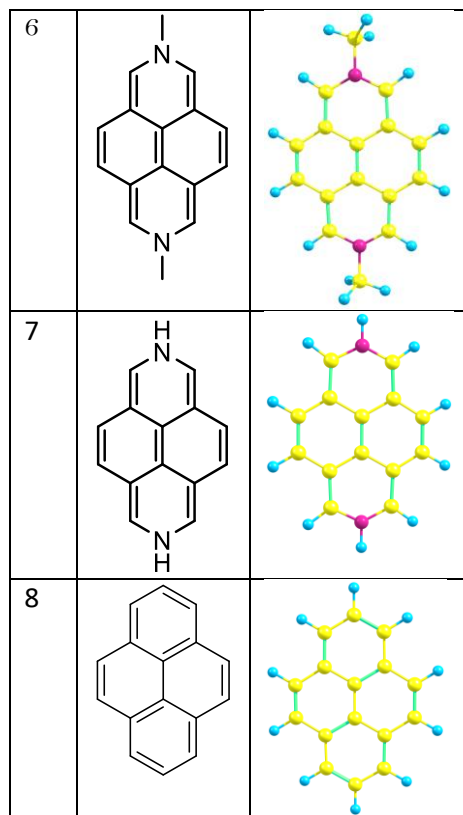


Figure S1. HOMO-LUMO distributions of 2,7-diazapyrenes **1-5**

Table S3. Structure optimization of 2,7-diazapyrenes in a gas phase

#	Structure	Optimization
1		





References:

1. R. Krishnan, J. S. Binkley, R. Seeger and J. A. Pople, *J. Chem. Phys.* 72, 650 (1980)
2. A. D. McLean and G. S. Chandler *J. Chem. Phys.* 72, 5639, (1980)
3. T. Clark, J. Chandrasekhar and P. v. R. Schleyer, *J. Comp. Chem.* 4, 294 (1983)
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6. Neese, F. "The ORCA program system" *Wiley interdisciplinary Reviews - Computational Molecular Science*, 2012, Vol 2., Issue 1, Pages 73-78