Supporting information for

## **Electrochemical Deposition of Amorphous Cobalt Oxides for Oxygen Evolution Catalysis**

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Figure S1 Chronoamperogram during the electrodeposition of  $Co_3O_4$  on the CP in a 0.2-M  $Co(NO_3)_2$  aqueous solution at -0.7 V vs. Ag/AgCl (saturated KCl).



Figure S2 Equivalent circuit used in curve fittings of EIS spectra displayed in Fig. 3b and Fig. 5c.



Figure S3 XRD patterns of CoO<sub>x</sub>/VS<sub>2</sub>/CP (2 C) and CoO<sub>x</sub>/VS<sub>2</sub>/CP (8 C) acquired with a Mo target (0.7107 Å).



Figure S4 SEM images of VS<sub>2</sub>/CP and CoO<sub>x</sub>/VS<sub>2</sub>/CP with different Co deposition amounts. The scale bar corresponds to 10  $\mu$ m.



Figure S5 XRD patterns of the synthesized catalysts acquired with the Mo target (0.7107 Å).



Figure S6 Full XP spectra of VS<sub>2</sub>/C and CoO<sub>x</sub>/VS<sub>2</sub>/CP (0.5 C) before and after the cyclic voltammetry in the range of 1.23 to 1.83 V vs. RHE at a scan rate of 10 mV/s for 100 cycles.



Figure S7 Deconvoluted XP spectra in the Co  $2p_{3/2}$  and S 2p regions of Co<sub>3</sub>O<sub>4</sub>/CP (0.5 C), VS<sub>2</sub>/CP, and CoO<sub>x</sub>/VS<sub>2</sub>/CP (0.5 C) before and after the cyclic voltammetry in the range of 1.23 to 1.83 V vs. RHE at a scan rate of 10 mV/s for 100 cycles<sup>1,2</sup>.



Figure S8 (a) Cyclic voltammograms and (b) Nyquist plots of EI spectra measured for  $VS_2$  at the 1st cycle and 100th cycle.



Figure S9 Nyquist plots of EI spectra measured for  $CoO_x/VS_2/CP$  (0.5 C) at the 1st cycle and 100th cycle.

Sample	Temperature/°	Electric	Time/h	2ϑ/°	FWHM/°
	С	quantity/C			
Co <sub>3</sub> O <sub>4</sub> /CP	200	0.1	1	16.8	0.427
Co <sub>3</sub> O <sub>4</sub> /CP	200	0.1	3	16.888	0.273
Co <sub>3</sub> O <sub>4</sub> /CP	200	0.2	1	16.849	0.402
Co <sub>3</sub> O <sub>4</sub> /CP	200	0.2	3	16.919	0.225
Co <sub>3</sub> O <sub>4</sub> /CP	250	0.2	1	16.739	0.393
Co <sub>3</sub> O <sub>4</sub> /CP	300	0.2	1	16.830	0.385

Table S1 FWHMs of the 311 peaks at 16.8° for the deposited cobalt oxides  $Co_3O_4$  on the CP synthesized at different parameters.

Table S2 Curve fitting results of the Nyquist plots presented in Fig. 3b.

Sample	$R_1/\Omega$	$R_2/\Omega$	<i>C</i> <sub>2</sub> /μF	<i>R</i> <sub>3</sub> /Ω	Q <sub>3</sub> /mF	<i>a</i> <sub>3</sub>
Co <sub>3</sub> O <sub>4</sub> /CP (0.5 C)	1.582	14.62	0.3264	44.87	4.355	0.7651
Co <sub>3</sub> O <sub>4</sub> /CP (1 C)	1.633	15.97	0.3002	47.39	3.892	0.8246
Co <sub>3</sub> O <sub>4</sub> /CP (2 C)	1.333	13.33	0.3187	22	14.01	0.7018
Co <sub>3</sub> O <sub>4</sub> /CP (4 C)	1.364	14.64	0.313	34.64	9.078	0.7119
Co <sub>3</sub> O <sub>4</sub> /CP (8 C)	1.391	14.21	0.3029	17.35	43.05	0.5757

Table S3 Curve fitting results of the Nyquist plots presented in Fig. 5b.

Sample	$R_1/\Omega$	$R_2/\Omega$	<i>C</i> <sub>2</sub> /μF	$R_3/\Omega$	Q <sub>3</sub> /mF	<i>a</i> <sub>3</sub>
СР	1.059	11.93	0.4449	2114	0.1164	0.7963
VS <sub>2</sub> /CP	1.122	13.6	0.3893	135.1	0.8755	0.8864
Co <sub>3</sub> O <sub>4</sub> /CP (0.5 C)	1.582	14.62	0.3264	44.87	4.355	0.7651
CoO <sub>x</sub> /VS <sub>2</sub> /CP (0.5 C)	1.296	14.16	0.3311	9.94	197.6	1

## References

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- 2. J. Yang, H. Liu, W.N. Martens, and R.L. Frost, J. Phys. Chem. C, 2010, 114, 1, 111–119.