DIPEA-induced activation of OH⁻ for the synthesis of amide via photocatalysis

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1. General information

All reagents were purchased from commercial sources and used without further purification. All solvents were dried in a standard manner. The nitrile wastewater used in the study was attained from a chemical plant in China (Fujian Province). Reactions were monitored by TLC on silica gel plates. Column chromatography was performed over silica gel (200-300 mesh) and petroleum ether/ethyl acetate. All products were characterized by NMR. ¹H NMR spectra were recorded at 400 MHz and ¹³C NMR spectra were recorded at 101 MHz (Bruker DPX) with DMSO-d₆ as solvent. Chemical shifts are reported in ppm using TMS as internal standard. NMR by the services provided at the Shandong Liaocheng University. HPLC were recorded on an SHIMDZU LC-20A instrument with a HP5-MS 30 m x 0.25 mm capillary apolar columns.

2. General procedure for the catalytic reactions



In a round bottom flask equipped with a stir bar was added nitrile (0.5 mmol), DIPEA (1.0 mmol, 2.0 equiv.), and esion Y (0.05 mmol, 0.1 equiv). The mixture was stirred at room temperature for 24 h blue light LED (12W) in 3 mL H₂O solvent. The reaction mixture was extracted with ethyl acetate (3×15 mL). The combined organic layer was washed with brine (10 mL), dried over MgSO₄. The concentrated residue was purified by column chromatography on a silica gel to afford the pure product 2a.

3. General procedure for the gram scale experiment

In 50 ml round bottom flask equipped with a stir bar was added nitrile (10.0 mmol), DIPEA (15.0 mmol, 1.5 equiv), and esion Y (1.0 mmol, 0.1 equiv). The mixture was stirred at room temperature for 36 h blue light LED (12W) in 25 mL H₂O solvent. The reaction mixture was extracted with ethyl acetate (3×15 mL). After the reaction was completed the solution of the crude product was concentrated in vacuo, and the residue was purified by column chromatography on a silica gel (petroleum ether/ethyl acetate=3/1) to afford the target product as a white solid.

4. Characterization data

benzamide (2a)¹

¹H NMR (400 MHz, DMSO-*d*₆) δ 8.01 (s, 1H), 7.93–7.86 (m, 2H), 7.55–7.48 (m, 1H), 7.48–7.42 (m, 2H), 7.39 (s, 1H); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 168.42, 134.71, 131.69, 128.67, 127.93.

4-chlorobenzamide (2b)¹

¹H NMR (400 MHz, DMSO-*d*₆) δ 8.05 (s, 1H), 7.92–7.87 (m, 2H), 7.55–7.50 (m, 2H), 7.47 (s, 1H); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 167.30, 136.56, 133.49, 129.87, 128.76.

2-chlorobenzamide (2c)¹

¹H NMR (400 MHz, DMSO- d_6) δ 7.89 (s, 1H), 7.60 (s, 1H), 7.52–7.34 (m, 4H); ¹³C NMR (101 MHz, DMSO- d_6) δ 168.67, 137.60, 131.03, 130.11, 130.07, 129.13, 127.49.

2-iodobenzamide (2d)¹

¹H NMR (400 MHz, DMSO- d_6) δ 7.88 (dd, J = 7.9, 0.8 Hz, 1H), 7.83 (s, 1H), 7.52 (s, 1H), 7.43 (td, J = 7.5, 1.1 Hz, 1H), 7.35 (dd, J = 7.6, 1.7 Hz, 1H), 7.15 (td, J = 7.7, 1.8 Hz, 1H); ¹³C NMR (101 MHz, DMSO- d_6) δ 171.20, 143.57, 139.64, 131.08, 128.41, 128.26, 93.59.

2-nitrobenzamide (2e)²

¹H NMR (400 MHz, DMSO- d_6) δ 8.18 (s, 1H), 8.00 (d, J = 8.0 Hz, 1H), 7.81-7.60 (m, 4H); ¹³C NMR (101 MHz, DMSO- d_6) δ 167.74, 147.68, 133.81, 133.05, 131.08, 129.31, 124.42.

3-nitrobenzamide (2f)¹

¹H NMR (400 MHz, DMSO- d_6) δ 8.72–8.66 (m, 1H), 8.40–8.28 (m, 3H), 7.76 (t, *J* = 8.0 Hz, 1H), 7.72 (s, 1H); ¹³C NMR (101 MHz, DMSO- d_6) δ 166.20, 148.24, 136.22, 134.26, 130.48, 126.31, 122.69.

4-nitrobenzamide (2g)³

¹H NMR (400 MHz, DMSO-*d*₆) δ 8.34–8.26 (m, 3H), 8.12–8.07 (m, 2H), 7.73 (s, 1H); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 166.71, 149.51, 140.43, 129.37, 123.89.

4-acetylbenzamide (2h)⁴

¹H NMR (400 MHz, DMSO-*d*₆) δ 8.15 (s, 1H), 8.05–7.96 (m, 4H), 7.57 (s, 1H), 2.62 (s, 3H); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 198.22, 167.60, 139.12, 138.56, 128.56, 128.23, 27.42.

2-bromobenzamide (2i)¹

¹H NMR (400 MHz, DMSO- d_6) δ 7.89 (s, 1H), 7.65 (d, J = 7.9 Hz, 1H), 7.59 (s, 1H), 7.42 (d, J = 4.2 Hz, 2H), 7.38–7.31 (m, 1H); ¹³C NMR (101 MHz, DMSO- d_6) δ 169.62, 139.77, 133.20, 131.15, 129.04, 127.97, 119.10.

2-methoxybenzamide (2j)¹

¹H NMR (400 MHz, DMSO- d_6) δ 7.86 (dd, J = 7.7, 1.9 Hz, 1H), 7.70 (s, 1H), 7.58 (s, 1H), 7.47 (ddd, J

= 8.4, 7.3, 1.9 Hz, 1H), 7.11 (d, *J* = 7.9 Hz, 1H), 7.03 (td, *J* = 7.6, 0.9 Hz, 1H), 3.88 (s, 1H); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 166.87, 157.74, 132.97, 131.27, 123.05, 120.87, 112.39, 56.22.

3-methoxybenzamide (2k)²

¹H NMR (400 MHz, DMSO-*d*₆) δ 8.01 (s, 1H), 7.52–7.44 (m, 2H), 7.41 (s, 1H), 7.36 (t, *J* = 7.9 Hz, 1H), 7.08 (ddd, *J* = 8.2, 2.6, 0.9 Hz, 1H), 3.79 (s, 3H); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 168.22, 159.61, 136.17, 129.80, 120.17, 117.53, 113.12, 55.65.

4-methoxybenzamide (2l)¹

¹H NMR (400 MHz, DMSO-*d*₆) δ 7.95–7.75 (m, 3H), 7.20 (s, 1H), 7.02–6.94 (m, 2H), 3.80 (s, 3H); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 167.96, 162.05, 129.82, 126.94, 113.85, 55.76.

4-methylbenzamide (2m)³

¹H NMR (400 MHz, DMSO- d_6) δ 7.92 (s, 1H), 7.79 (d, J = 8.2 Hz, 2H), 7.29 (s, 1H), 7.24 (d, J = 7.9 Hz, 2H), 2.34 (s, 3H); ¹³C NMR (101 MHz, DMSO- d_6) δ 168.35, 141.55, 131.92, 129.20, 127.98, 21.39.

3-methylbenzamide (2n)¹

¹H NMR (400 MHz, DMSO-*d*₆) δ 7.94 (s, 1H), 7.71 (s, 1H), 7.70–7.65 (m, 1H), 7.40–7.28 (m, 3H), 2.35 (s, 3H); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 168.53, 137.89, 134.72, 132.23, 128.54, 125.06, 21.41. **3-fluorobenzamide (2o)** ³

¹H NMR (400 MHz, DMSO- d_6) δ 8.08 (s, 1H), 7.77–7.71 (m, 1H), 7.67 (ddd, J = 10.1, 2.5, 1.6 Hz, 1H), 7.60–7.47 (m, 2H), 7.41–7.33 (m, 1H); ¹³C NMR (101 MHz, DMSO- d_6) δ 167.05, 167.03, 163.65, 161.22, 137.18, 137.11, 130.89, 130.81, 124.08, 124.06, 118.70, 118.49, 114.79, 114.57.

4-fluorobenzamide (2p)³

¹H NMR (400 MHz, DMSO- d_6) δ 8.03 (s, 1H), 7.99–7.92 (m, 2H), 7.43 (s, 1H), 7.31–7.22 (m, 2H); ¹³C NMR (101 MHz, DMSO- d_6) δ 167.35, 165.63, 163.16, 131.20, 131.18, 130.63, 130.54, 115.65, 115.43.

2-aminobenzamide (2q)⁵

¹H NMR (400 MHz, DMSO- d_6) δ 7.73 (s, 1H), 7.54 (dd, J = 8.0, 1.4 Hz, 1H), 7.13 (ddd, J = 8.4, 7.1, 1.5 Hz, 1H), 7.07 (s, 1H), 6.69 (dd, J = 8.3, 1.1 Hz, 1H), 6.56 (s, 2H), 6.48 (ddd, J = 8.1, 7.1, 1.2 Hz, 1H); ¹³C NMR (101 MHz, DMSO- d_6) δ 171.41, 150.23, 131.98, 128.81, 116.48, 114.48, 113.75.

3-aminobenzamide (2r)³

¹H NMR (400 MHz, DMSO- d_6) δ 7.73 (s, 1H), 7.14 (s, 1H), 7.09–7.02 (m, 2H), 7.01–6.96 (m, 1H), 6.72–6.66 (m, 1H), 5.19 (s, 2H); ¹³C NMR (101 MHz, DMSO- d_6) δ 169.22, 149.04, 135.67, 129.01, 116.96, 115.15, 113.58.

4-aminobenzamide (2s)³

¹H NMR (400 MHz, DMSO-*d*₆) δ 7.62–7.56 (m, 2H), 7.52 (s, 1H), 6.85 (s, 1H), 6.57–6.49 (m, 2H), 5.60 (s, 2H); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 168.53, 152.13, 129.59, 121.41, 112.93.

thiophene-3-carboxamide (2t)⁶

¹H NMR (400 MHz, DMSO- d_{δ}) δ 8.14 (dd, J = 3.0, 1.3 Hz, 1H), 7.81 (s, 1H), 7.55 (dd, J = 5.0, 3.0 Hz, 1H), 7.50 (dd, J = 5.0, 1.3 Hz, 1H), 7.26 (s, 1H); ¹³C NMR (101 MHz, DMSO- d_{δ}) δ 164.23, 138.44, 129.52, 127.64, 127.02.

isonicotinamide (2u)¹

¹H NMR (400 MHz, DMSO- d_6) δ 8.72 (dd, J = 4.6, 1.4 Hz, 2H), 8.26 (s, 1H), 7.78 (dd, J = 4.5, 1.5 Hz, 2H), 7.74 (s, 1H); ¹³C NMR (101 MHz, DMSO- d_6) δ 166.83, 150.68, 141.74, 121.87.

2-phenylacetamide (2v)⁷

¹H NMR (400 MHz, DMSO-*d*₆) δ 7.50 (s, 1H), 7.34–7.26 (m, 4H), 7.25–7.20 (m, 1H), 6.93 (s, 1H),

3.40 (s, 2H); ¹³C NMR (101 MHz, DMSO-*d*₆) δ 172.81, 136.97, 129.54, 128.62, 126.74, 42.75.

2-(p-tolyl)acetamide (2w)¹

¹H NMR (400 MHz, DMSO- d_6) δ 7.42 (s, 1H), 7.15 (d, J = 8.1 Hz, 2H), 7.10 (d, J = 7.9 Hz, 2H), 6.86 (s, 1H), 3.32 (s, 2H), 2.27 (s, 3H); ¹³C NMR (101 MHz, DMSO- d_6) δ 172.91, 135.68, 133.90, 129.38, 129.17, 42.34, 21.11.

1-naphthamide (2x)8

¹H NMR (400 MHz, DMSO-*d*₆) δ 8.36–8.30 (m, 1H), 8.09-7.94 (m, 3H), 7.66 (dd, *J* = 7.0, 1.2 Hz, 1H), 7.64–7.59 (m, 1H), 7.59–7.51 (m, 3H); ¹³C NMR (101 MHz,DMSO-*d*₆) δ 171.08, 135.12, 133.67, 130.25, 130.19, 128.64, 127.08, 126.60, 126.06, 125.61, 125.41.

isobutyramide (2y)³

¹H NMR (400 MHz, DMSO- d_6) δ 7.22 (s, 1H), 6.72 (s, 1H), 2.33 (hept, J = 6.9 Hz, 1H), 0.99 (d, J = 7.0 Hz, 6H); ¹³C NMR (101 MHz, DMSO- d_6) δ 179.04, 34.19, 19.88.

propionamide (2z)⁹

¹H NMR (400 MHz, DMSO- d_6) δ 7.26 (s, 1H), 6.76 (s, 1H), 2.05 (q, J = 7.6 Hz, 2H), 0.97 (t, J = 7.6 Hz, 3H); ¹³C NMR (101 MHz, DMSO) δ 175.95, 28.64, 10.11.

4-nitrophthalamide(2aa)¹¹ ¹H NMR (400 MHz, DMSO-*d*₆) δ 8.37 – 8.26 (m, 2H), 8.07 (s, 1H), 8.01 (s, 1H), 7.71 (d, *J* = 8.3 Hz, 1H), 7.63 (s, 2H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 169.21, 168.21, 147.56, 143.11, 137.68, 129.62, 124.87, 122.88.

1*H***-indole-3-carboxamide (2ab)**¹¹ ¹H NMR (400 MHz, DMSO-*d*₆) δ 11.65 – 11.52 (m, 1H), 8.21 (d, J = 7.2 Hz, 1H), 8.09 (d, J = 2.7 Hz, 1H), 7.45 (d, J = 7.8 Hz, 2H), 7.18 – 7.10 (m, 2H), 7.02 – 6.72 (m, 1H). ¹³C NMR (101 MHz, DMSO-*d*₆) δ 167.23, 136.70, 129.00, 126.72, 122.29, 121.56, 120.85, 112.27, 110.93.

pyrazinamide (2ac) ¹¹ ¹H NMR (400 MHz, DMSO- d_6) δ 9.19 (d, J = 1.5 Hz, 1H), 8.85 (d, J = 2.5 Hz, 1H), 8.71 (dd, J = 2.5, 1.5 Hz, 1H), 8.28 (s, 1H), 7.88 (s, 1H). ¹³C NMR (101 MHz, DMSO- d_6) δ 165.59, 147.85, 145.50, 144.07, 143.84.

5. References

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6. MS data of H₂¹⁸O Labeling Experiments



7. HPLC validation experiment of hydrogen peroxide

The chromatographic conditions: HPLC were recorded on an SHIMDZU LC-20A instrument with a HP5-MS 30 m x 0.25 mm Capillary Apolar Columns. Mobile phase: A phase is water, B phase is methanol; The detection wavelength was 240 nm; Injection volume: 10 μ L; The gradient elution procedure is shown in Table. HPLC results showed that H₂O₂ was detected at the retention time of 1.81

		a ze graaiene eranen preseaane	
Time/min	flow rate/(mL/min)	A phase (water) volume fraction	B phase (methanol) volume
		(%)	fraction (%)
0.0	1.0	92	8
5.0	1.0	92	8
10.0	1.0	40	60
15.0	1.0	40	60
18.0	1.0	92	8
22.0	1.0	92	8

Table 1	HPLC	gradient	elution	procedure
I abit I	III LC	Sidulette	eration	procedure



1H NMR and 13C NMR spectra for the products

2a





2b

8.053 7.912 77.906 77.901 77.889 77.589 77.589 77.513 77.513 77.513 77.513 77.513 77.513 77.513 77.513





2c



-168.67 137.60 137.60 137.60 137.60 137.60 137.60 127.49



2d





2e





-167.74 -147.68 -147.68 -133.81 -133.05 133.05 133.05 132.42



2f



-166.20 -148.24 -148.24 -136.22 -130.48 -130.48 -126.31



2g













2k

8.008 7.495 7.495 7.476 7.472 7.477 7.472 7.472 7.472 7.472 7.472 7.473 7.478 7.473 7.452 7.745 7.7458 7.7458 7.7458 7.7458 7.7359 7.05











2n

7.341 7.714 7.689 7.689 7.689 7.689 7.689 7.680 7.589 7.567 7.352 7.352 7.352 7.352 7.332 7.332 7.332 7.332 7.332 7.332 7.332 7.332 7.332 -2.346







2p







2q

7.731 7.550 7.550 7.551 7.550 7.550 7.150 7.1150 7.1150 7.1150 7.1150 7.1150 7.1129 7.1129 7.1129 7.1128 7.1128 7.1128 7.1128 7.1128 7.1128 7.1128 7.1128 7.1150 7.





2r



-169.22 -149.04 -135.67 -129.01 -129.01 -115.15 7113.58



2s

7.608 7.501 7.501 7.536 7.538 7.573 7.573 7.573 7.573 7.573 7.573 6.533 6.533 6.533 6.533 6.533 6.533 6.533 6.533 6.553 6.553 6.553





2t







2u







2v

7.503 7.325 7.325 7.325 7.325 7.327 7.327 7.307 7.247 7.247 7.242 7.233 7.723 7.7242 7.7242 7.7242 7.7242 7.7242 7.7242 7.7242 7.7242 7.7242 7.7242 7.7242 7.7233 7.7242 7.7233 7.7242 7.7233 7.7242 7.7233 7.7233 7.7242 7.7233 7.7242 7.7233 7.7242 7.7233 7.7242 7.7234 7.7242 7.7234 7.7247 7.7247 7.7275 7.7233









-171.08 135.12 135.12 133.67 133.67 130.25 130.19 125.64 125.61 125.61 NH₂ 210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 f1 (ppm) 50 40 30 20 10 0 -10 2у -7.222 -6.722 2.382 2.365 2.348 2.331 2.331 2.331 2.331 2.236 2.236 2.2396 2.2396 2.2296 0.994 NH₂ Ĭ 1.00-1 100.9 1.014 7 6 f1 (ppm) 2 1 4 8 3 -3 16 15 13 9 -2

5

4

Ó

-1

14

12 11











210 200 190 180 170 160 150 140 130 120 110 100 90 80 70 60 50 40 30 20 10 0 -10 f1 (ppm)

7 Optimized Structures and Cartesian Coordinates



A

Energies= -371.141599 a.u

NCHCHCHHHCHHHCHHHCHHCHHCHHCHH

-0.08551700	-0.10559000	-0.22262200
-1.44164100	0.46145100	-0.35971700
-1.34408100	1.25603800	-1.11031100
1.00824600	0.89217900	-0.30155600
0.53271100	1.83365800	-0.60428900
2.00855700	0.53675200	-1.41742800
1.47708000	0.36054400	-2.35799700
2.72650600	1.35282200	-1.57133200
2.57766100	-0.36809800	-1.18041400
1.74123900	1.18981500	1.02062000
2.44556000	2.01630900	0.87190800
1.04533900	1.48179100	1.81454500
2.32029500	0.33058400	1.37728600
-2.42695900	-0.55914900	-0.95060200
-2.00223100	-1.01962700	-1.84814100
-2.67931800	-1.35785700	-0.24351100
-3.36566500	-0.06217500	-1.22218600
-2.00934500	1.10584200	0.92210700
-1.34408300	1.88999200	1.30088800
-2.98820700	1.56205300	0.72995000
-2.14164900	0.36211400	1.71699100
0.04841200	-1.17687200	0.76517200
-0.92657600	-1.66516100	0.85127300
0.28222000	-0.80160600	1.77650800
1.07125200	-2.25475700	0.38511900
0.86217600	-2.64265600	-0.61740700
2.09788400	-1.87507300	0.39281700
1.02444100	-3.08609200	1.09947400



Eosin Y^{•+}

Energies=-11439.750934 a.u

С	3.62873800	-0.98598600	-0.09124300
С	2.40903800	-1.67598900	-0.03367800
С	1.22157000	-0.94894900	-0.09104800
С	1.20031400	0.47395600	-0.21039100
С	2.44936300	1.15079000	-0.26515300
С	3.61964800	0.44341000	-0.20828500
С	-0.04546500	1.14223900	-0.24148600
С	-1.23771600	0.37198800	-0.21524900
С	-1.13271100	-1.04084600	-0.09588500
С	-2.25610200	-1.85819800	-0.04385200
Č	-3.61160800	-1.27249100	-0.11538700
Č	-3.66860300	0.20603500	-0.24661500
Č	-2.54165600	0.96549000	-0.29155600
Ĥ	2.46769800	2.22968500	-0.35121500
H	-2.61486200	2.04138500	-0.38798800
Br	2.38277900	-3.54929000	0.11786800
Br	5.29652100	1.32733900	-0.27945500
Br	-5.37092600	0.97779700	-0.34823000
Br	-2.10347600	-3.69862300	0.11442900
Õ	-4.61689300	-1.97343400	-0.07107800
Č	-0.10471500	2.62314700	-0.43502400
Č	-0.23963200	3.52068700	0.64720400
Č	-0.03726100	3.12673800	-1.73909900
Č	-0.30844100	4.89763600	0.40312900
č	-0.10941500	4.50309600	-1.96960700
H	0.06348000	2.44290400	-2.57699000
Ĉ	-0 24591200	5 38808600	-0.90028800
H	-0 40946000	5 57649100	1 24231700
Ĥ	-0.05942700	4 87762000	-2.98746800
Ĥ	-0 30204700	6 45715300	-1 07887000
Ċ	-0.29461800	2 96831600	2 02791500
ŏ	-0.22153400	1 77649700	2.27775900
ŏ	-0.22135400 -0.43070400	3 90895900	2 97425300
й	-0 45891600	3 45527800	3 83846200
ö	0 07546400	-1 64693000	-0 03020600
ŏ	4 74957000	-1 68825000	-0.03598000
й	5 52596400	-1.00023700	-0.03370000
11	3.34370400	-1.0/20/000	-0.00-09300



	Fosin V		
Energies= -11439.968326 a.u			
С	-0.07007500	1.13160800	-0.25205800
С	-1.21905700	-1.01964300	-0.08169200
С	-1.25266100	0.42429000	-0.20409300
С	-2.55307800	1.04578100	-0.25220100
Н	-2.60842800	2.12413800	-0.33652200
С	-3.68467500	0.30813900	-0.19395400
С	-3.69275900	-1.17517400	-0.07513900
Č	-2.35309400	-1.77412200	-0.02341000
Č	-0.06463200	2.61581000	-0.44619800
Č	0.05212500	3.54568000	0.61225700
Č	-0.18238600	3.09860100	-1.75606300
Č	0.04470600	4.92056800	0.33242100
Č	-0.18433400	4.46805300	-2.02283600
Й	-0.27693700	2.38760200	-2.57154300
Ĉ	-0.06997000	5 38221100	-0 97590000
й	0 12757200	5 62217700	1 15409000
Ĥ	-0 27871300	4 81584800	-3 04754800
Ĥ	-0.07360700	6 44960500	-1 17531700
n C	0 17507400	3 06876700	2 01958900
ŏ	0.17507400	1 90682500	2.01750700
ŏ	0.18813500	1.90002500	2.30171000
Ч	0.10015500	3 66251700	3 7935/000
II Br	5 38325800	1 1/513600	0.26306100
		1 92670000	-0.20500100
0	-4./2031900	-1.620/9000	-0.02095500
U Dr	-0.01554200	-1.00291000	-0.02390000
	-2.23300400	-3.03301000	
\mathbf{C}	3.5/199100		-0.11540900
	3.39833000		-0.23//0400
C C	2.44231/00	1.04358400	-0.28013900
C C	1.18489100	0.41/2/100	-0.21290000
C C	1.155/2400	-0.98840300	-0.09360000
<u> </u>	2.3248/000	-1./4886900	-0.04555000
H	2.50328400	2.12057200	-0.37960500
Br	5.30/15300	1.1491/200	-0.33858600
Br	2.22981200	-3.62842500	0.11045800
Q	4.68268600	-1.86806800	-0.06750900
H	5.46078300	-1.28128200	-0.12573800



В

Energies= -370.968345a.u

Ν	0.04586900	0.07594500	-0.18355600
C	-1.14338000	0.91231200	0.07664500
Ĥ	-0.77713900	1.94229100	-0.02295400
Ē	1.35697700	0.60410200	0.24344600
Ĥ	1.12951100	1.44825000	0.90063500
$\overline{\mathbf{C}}$	-1.58698600	0.72237700	1.55230800
Ĥ	-0.76827200	0.90310900	2.25430100
Ĥ	-2.37771300	1.44929600	1.75899900
Ĥ	-1.98541100	-0.28107900	1.72018500
Ĉ	-2.28965800	0.69366200	-0.91097100
Ĥ	-3.06589800	1.43293500	-0.69365200
Ĥ	-1.97247300	0.84635800	-1.94725900
Ĥ	-2.74644800	-0.29561100	-0.81571500
Ē	2.18859600	-0.42406300	1.02327000
Ă	1.65271200	-0.79274000	1.90242900
Ĥ	2.49519900	-1.27399700	0.40693200
Ĥ	3.09872600	0.07497600	1.36904200
$\overline{\mathbf{C}}$	2.09961900	1.14837400	-1.00121300
Ĥ	1.50150300	1.89177800	-1.53616700
Н	3.01924000	1.63177900	-0.65815300
Н	2.37323800	0.34555000	-1.69220700
С	-0.03868900	-1.20649900	-0.87684100
Н	-0.68789000	-1.07166700	-1.74633800
Н	0.96167200	-1.45951300	-1.23051900
С	-0.58864100	-2.34868300	0.00852000
Н	0.02838300	-2.49874400	0.89719200
Η	-1.62135100	-2.16836100	0.31467900
Н	-0.56569600	-3.26571500	-0.58752400



Energies= -76.471995 a.u			
0	0.00000000	0.00000000	0.11972000
Н	0.00000000	0.76156000	-0.47887900
Н	0.00000000	-0.76156000	-0.47887900



02'-

0.00000000	0.00000000	0.67654300
0.00000000	0.00000000	-0.67654300

Energies= -150.519076 a.u O O



 H_2O_2

Energies= -151.617242 a.u			
0	-0.71823600	-0.11814100	-0.05445400
Н	-1.01376800	0.66962200	0.43555200
0	0.71823800	0.11813000	-0.05445400
Н	1.01375400	-0.66953900	0.43571200



С

Energies= -370.395517 a.u N C H C H C H

-0.24763000	-0.42606000	-0.05107100
-0.64528000	0.98790000	0.33851100
-1.60808400	0.86019900	0.83515400
1.19309000	-0.84352200	-0.33746000
1.07320000	-1.80789300	-0.83902000

С	-0.86353200	1.85194100	-0.91256400
Η	-1.52662400	1.36565400	-1.63509100
Η	-1.33573700	2.78897700	-0.60170500
Н	0.07342900	2.10124800	-1.41351600
С	0.29148800	1.63405600	1.36160000
Ĥ	-0.16657400	2.58096500	1.66393600
Н	0.40472100	1.02229500	2.26019900
Н	1.27791200	1.86670600	0.95559200
C	1.90417800	0.08908800	-1.31910100
Ĥ	1.33773300	0.21682400	-2.24555400
Н	2.12601900	1.07013000	-0.89351600
Η	2.86053300	-0.37604900	-1.57800300
С	1.97668300	-1.09095500	0.95735600
Н	1.41785900	-1.72592700	1.65204800
Н	2.89930400	-1.61795100	0.69418300
Н	2.25741000	-0.16896600	1.46785800
С	-1.15768200	-1.33760900	-0.15867800
Н	-0.80087900	-2.32565500	-0.44396400
С	-2.62286200	-1.18769600	0.07558600
Н	-3.14460400	-2.06134400	-0.31976600
Н	-2.83075200	-1.13798500	1.15404200
Н	-3.03795700	-0.28802500	-0.38688500



OH-

Energies= -75.978159 a.u O H

0.00000000 0.0000000 0.10970700 0.00000000 0.00000000 -0.87765300



1a

Energies= -324.583672 a.u

С	-1.48403400	-1.21086500	-0.00000100
С	-0.09146600	-1.21746700	0.00001600
С	0.61027900	-0.00004600	-0.00000200
С	-0.09142400	1.21744800	0.00000700
С	-1.48395900	1.21091100	0.00000700
С	-2.18066100	0.00002400	-0.00001300
H	-2.02616600	-2.15194500	-0.00001200
Η	0.45872800	-2.15284100	0.00001200
H	0.45886700	2.15276500	0.00000300
Η	-2.02609200	2.15199000	0.00000200
Н	-3.26703600	0.00007600	-0.00001500
С	2.04504100	-0.00002600	-0.00000800
Ν	3.20843600	0.00001200	-0.00000400



D

-1.90954000	1.16605900	0.00000600
-2.56010800	-0.07539500	-0.00013600
-1.79227000	-1.24159600	-0.00010000
-0.39524500	-1.17338300	0.00013800
0.26951400	0.06617800	0.00028800
-0.51961700	1.23037400	0.00016300
-2.49635000	2.08494600	0.00009400
-3.64813000	-0.12989000	-0.00039700
-2.28530500	-2.21432000	-0.00035700
0.20197500	-2.07859900	0.00028900
0.02658500	2.17133100	0.00011900
1.76357500	0.23499500	0.00004400
2.42676100	1.27562900	-0.00026100
2.54869500	-1.140/4500	-0.00003800
3.28647900	-0.88030200	-0.00004000



2a

	2a		
Energies= -401.085404 a.u			
C	-0.22024800	-0.02240200	-0.01853400
Č	0.51521500	-1.20800400	0.10919500
Ĉ	1.90718200	-1.17382800	0.13726800
Č	2.57989700	0.04573800	0.02399100
Č	1.85483900	1.22978100	-0.12273500
Ĉ	0.46044700	1.19702900	-0.14331100
Ĥ	-0.02826700	-2.14414100	0.18391700
Н	2.46926200	-2.09760500	0.24437800
Н	3.66619500	0.07251000	0.04218400
Ĥ	2.37443900	2.17813000	-0.22971300
Н	-0.09205500	2.12060800	-0.29338200
С	-1.71841200	-0.13669100	-0.03399900
Õ	-2.28789100	-1.19439000	-0.27170800
Ň	-2.42077600	1.02379800	0.20674200
H	-3.41776600	0.90053300	0.32715100
Ĥ	-1.99676300	1.76876800	0.74069300

8 Computational methods

Density functional theory (DFT) calculations were carried out by using Gaussian 16 program^[1]. All the structures were optimized by using the combination of Becke's hybrid 3-parameter exchange functional^[2] and Lee-Yang-Parr's correlation functional^[3] known as B3LYP method in conjunction with the 6-31G(d) basis set to ensure these structure without imaginary frequencies. Then, the energy of the reaction mechanism is calculated at the B3LYP/6-311++G** level based on the optimized structures, and the solvation model based on electron density (SMD)^[4] with H₂O solvent attached was used throughout. Dimensional plots of molecular configurations were generated with the GaussView program^[5].

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