

**Effect of $\text{Cu}^{2+}/\text{Zn}^{2+}$ on electrochemical performance of polyacrylamide hydrogel as
advanced flexible electrode materials**

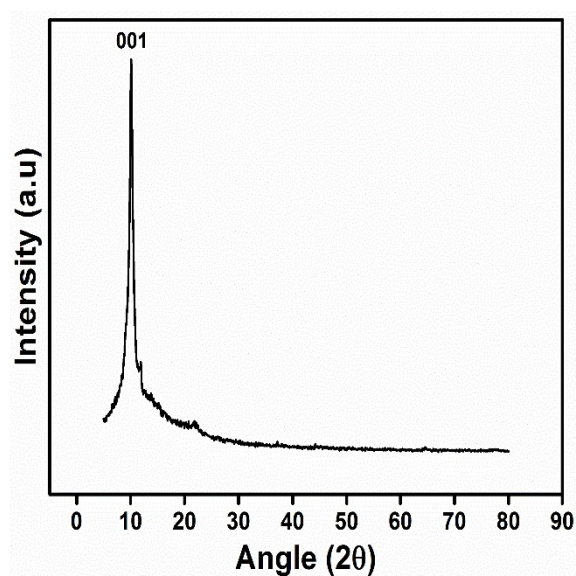


Fig. S1: XRD pattern of Graphene oxide

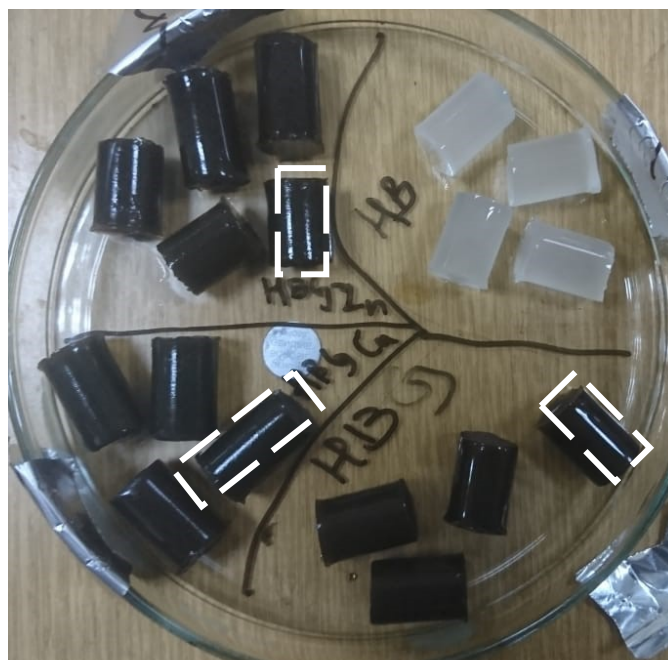


Fig. S2Light reflection from surface of Polyacrylamide based Hydrogels before reduction

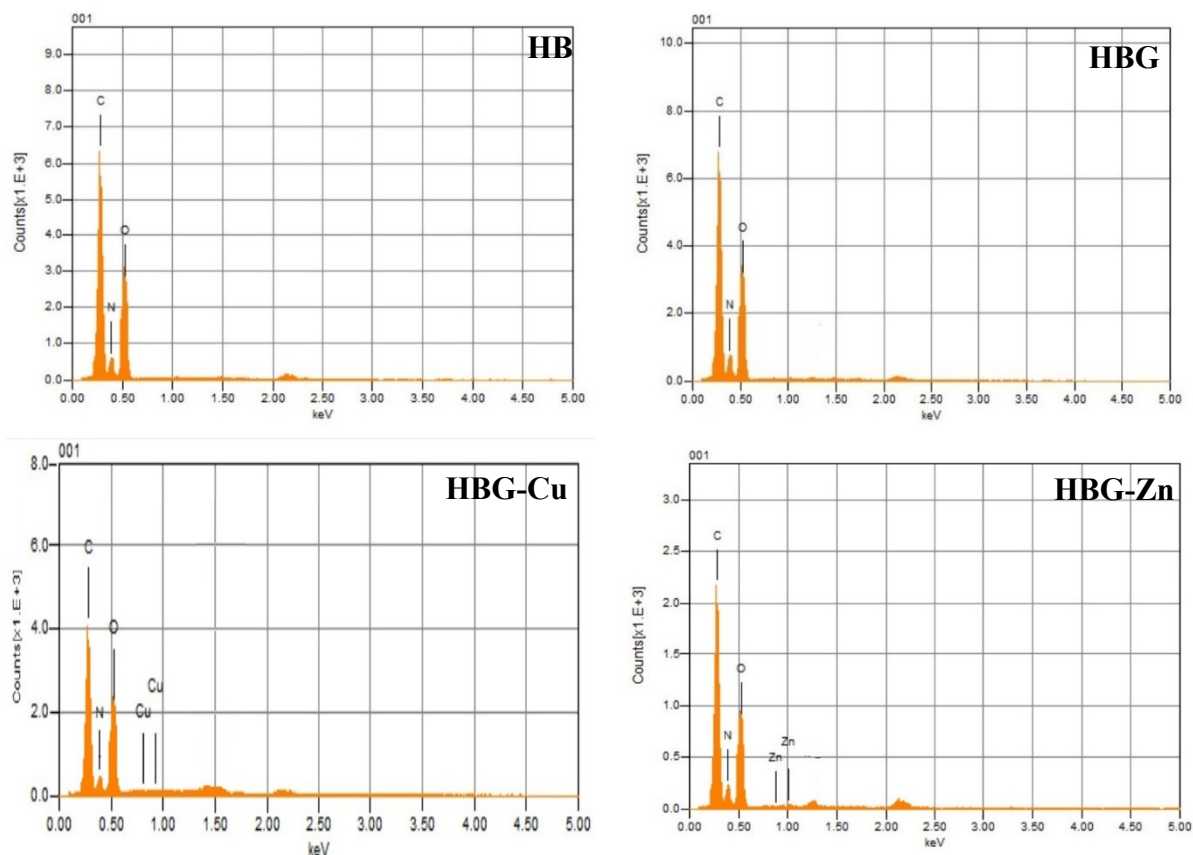


Fig. S3 EDX spectra of multi cross-linked polyacrylamide hydrogels

Table S1 Atomic percentage of each atom measured through EDX analysis for prepared multi cross-linked polyacrylamide hydrogels

| Hydrogels | Atomic % | | | | |
|---------------|----------|------|-------|------|------|
| | C | N | O | Cu | Zn |
| HB | 61.26 | 6.21 | 32.53 | - | - |
| HBG | 61.10 | 6.92 | 31.98 | - | - |
| HBG-Cu | 63.23 | 6.30 | 30.43 | 0.04 | - |
| HBG-Zn | 63.48 | 6.27 | 30.21 | - | 0.04 |

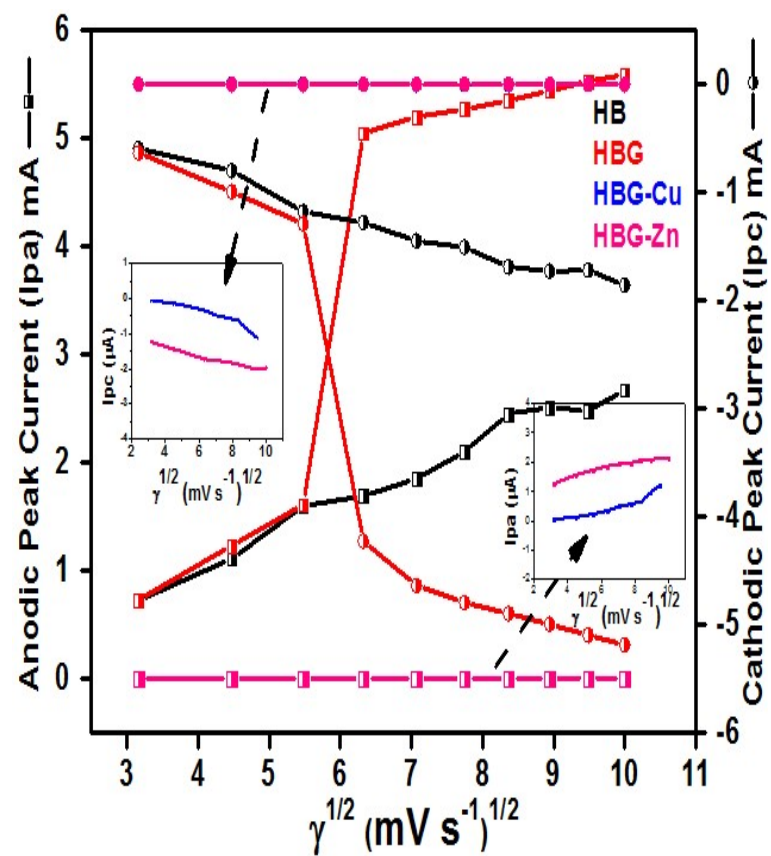


Fig. S4 Anodic and Cathodic peak current of polyacrylamide based hydrogels

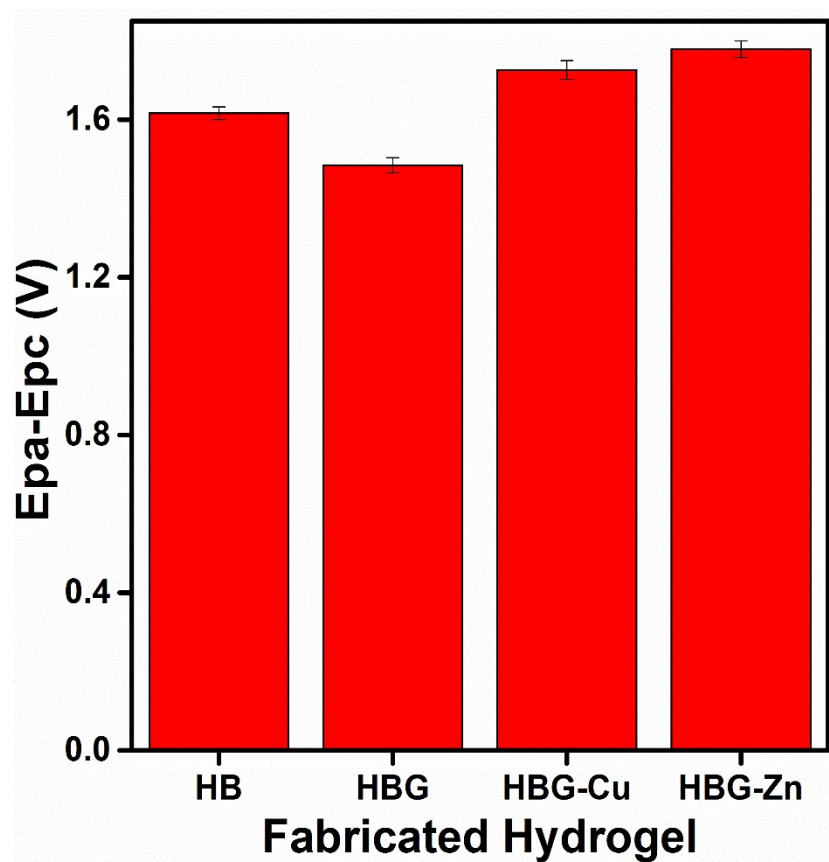


Fig. S5 Peak potential separation of polyacrylamide based hydrogels

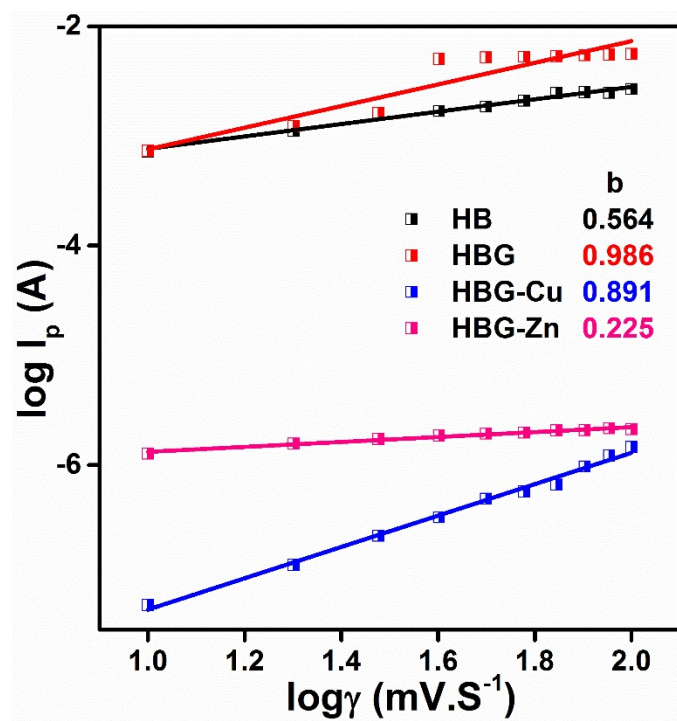


Fig. S6 Peak current at different sweep rates

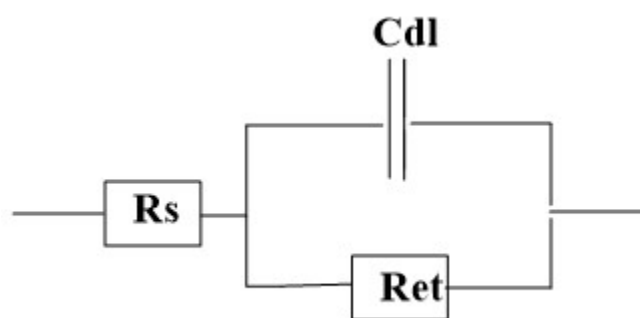


Fig. S7 Electrochemical circuit fit model for Polyacrylamide based hydrogels

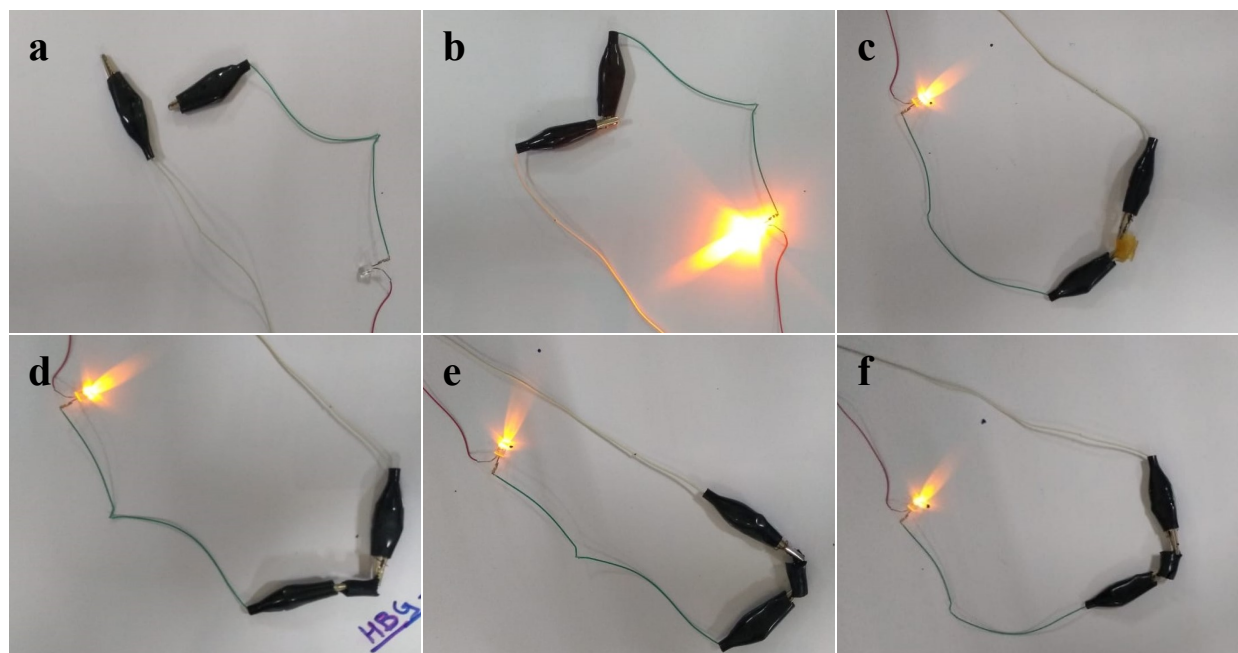


Fig. S8 Light response towards applied external DC current i.e. not connected (a), terminals connected in the absence of Hydrogel (b), and in the presence of Hydrogels i.e. HB (c), HBG (d), HBG-Cu (e) and HBG-Zn (f)