

Supplementary Information
for
Highly efficient ethanol vapour detection using g-C₃N₄/ZnO
micro flower-like heterostructural composites

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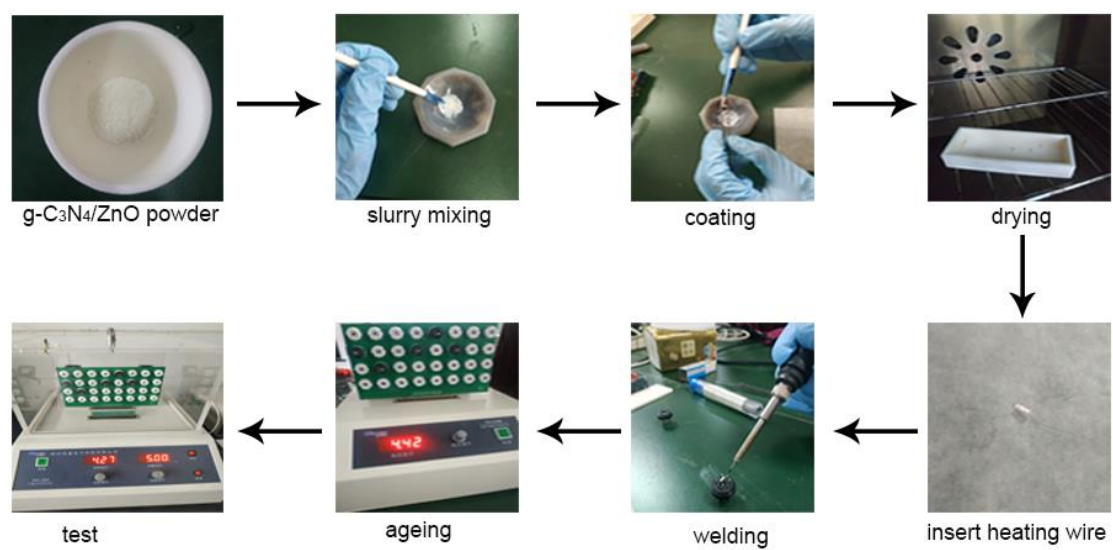


Figure S1. Schematic diagram of ethanol vapor sensor fabrication process.

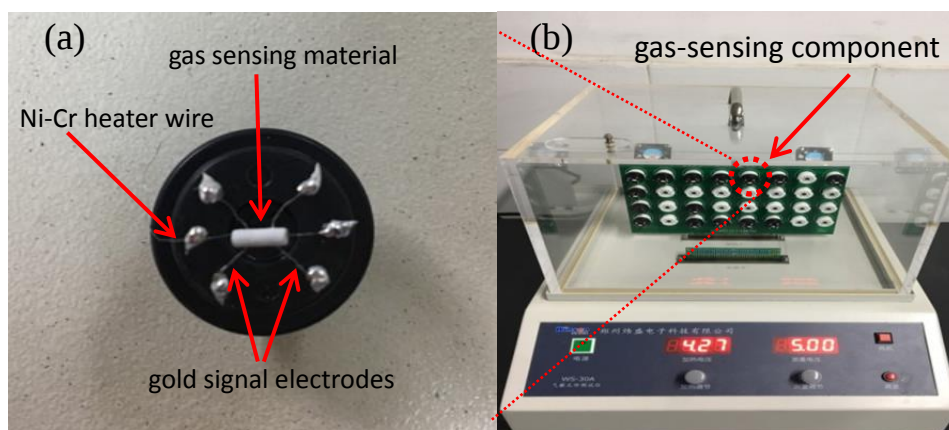


Figure S2. (a) The prepared g-C₃N₄/ZnO gas sensor component. (b) The gas-sensing component test system for examining gas-sensing properties of the synthesized materials.

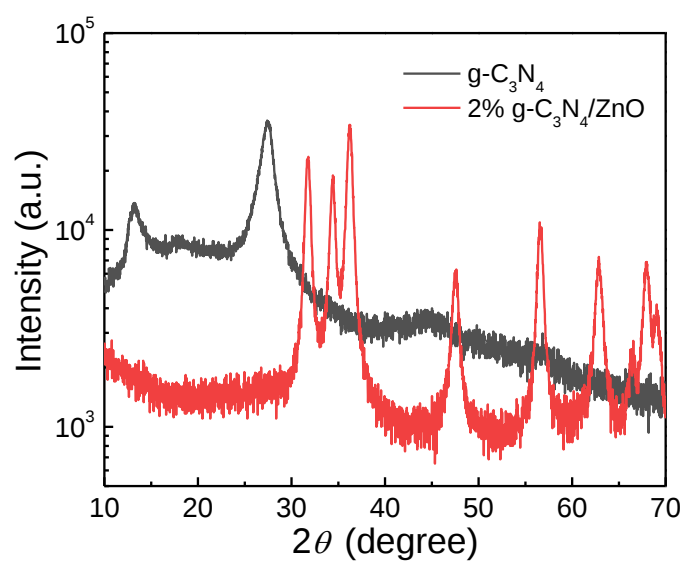


Figure S3. XRD patterns of the $\text{g-C}_3\text{N}_4$ and 2% $\text{g-C}_3\text{N}_4/\text{ZnO}$ composite in logarithmic scale.

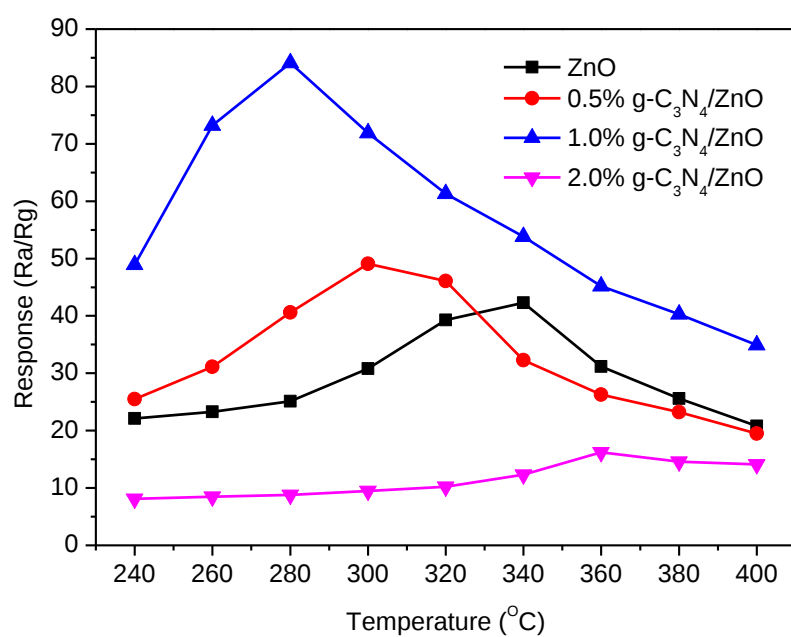


Figure S4. Response of ZnO and C₃N₄/ZnO composite gas sensors to 100 ppm ethanol vapour at different working temperatures.

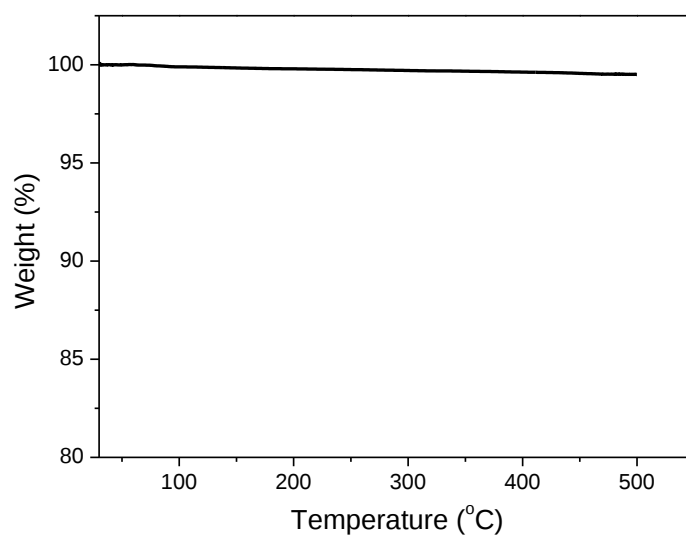


Figure S5. TGA curves of 1%g-C₃N₄/ZnO composite.

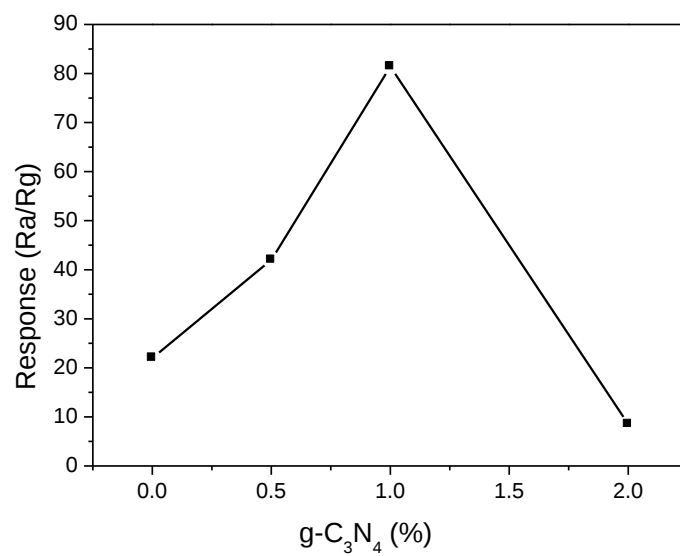


Figure S6. Response value relationship of g-C₃N₄/ZnO gas sensor with different g-C₃N₄ doping content to 100 ppm ethanol.

Table S1. The Ra values of ZnO and g-C₃N₄/ZnO composites at different operating temperatures.

Working temperatures (°C)	Ra values of ZnO (KΩ)	Ra values of 1%g-C ₃ N ₄ /ZnO composites (KΩ)
240	41332.09	572256.71
260	32718.03	363443.70
280	28211.37	131746.58
300	24329.13	58058.68
340	21242.80	29589.12
380	13925.73	10776.73
400	11847.18	8552.31

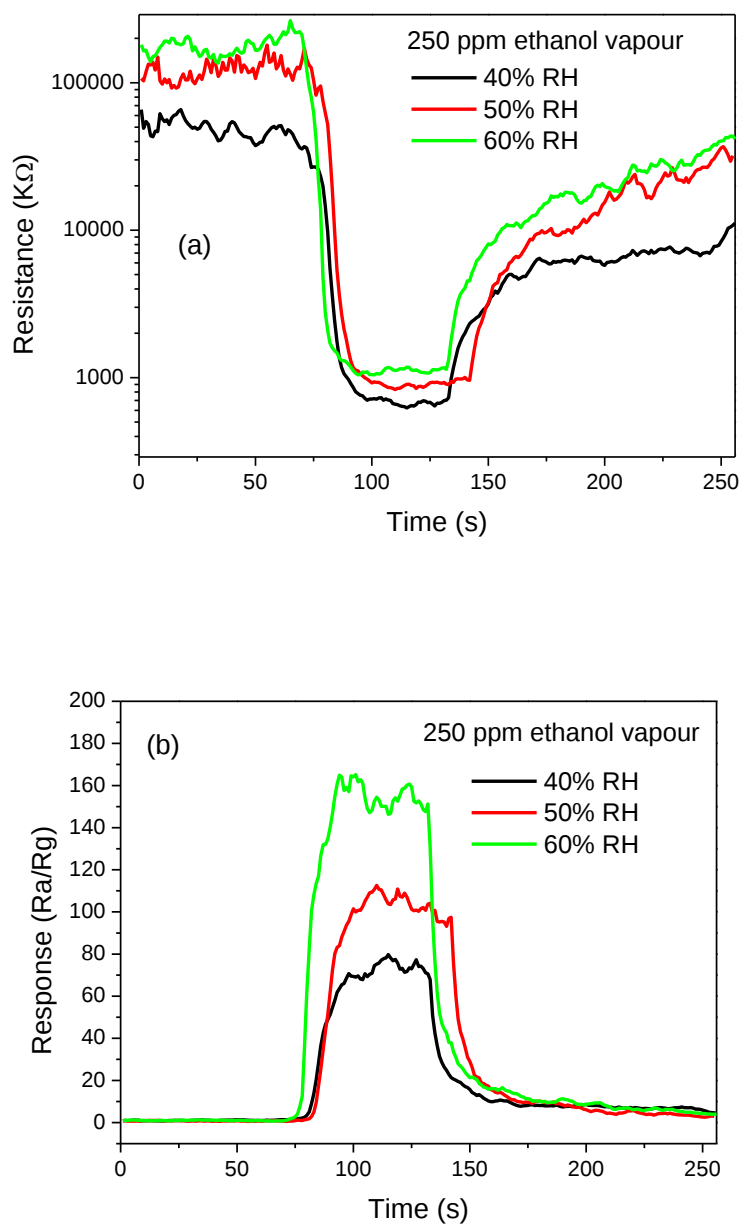


Figure S7. (a) The dynamic resistance curve of the 1% g-C₃N₄/ZnO composite to 250 ppm ethanol vapour at optimal working temperature at various relative humidities. (b) Dynamic response at various humidity levels.

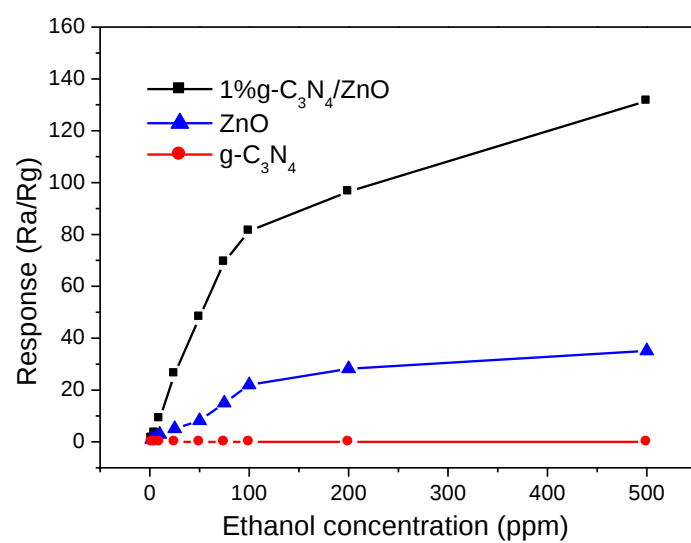


Figure S8. Response of the 1% g-C₃N₄/ZnO, ZnO, and g-C₃N₄ gas sensors to different concentrations of ethanol vapour.

Table S2. The Ra (or Rg) values of 1%g-C₃N₄/ZnO composites and corresponding response value to different ethanol vapour concentrations.

Concentrations of ethanol vapour (ppm)	Ra (or Rg) values of g-C ₃ N ₄ /ZnO composites (KΩ)	Response (Ra/Rg)
0	133786.96	1
2	88571.26	1.51
5	38639.36	3.46
10	14854.55	9.01
25	5076.20	26.36
50	2777.45	48.17
75	1926.81	69.43
100	1643.57	81.40
200	1385.60	96.56
500	1017.54	131.48

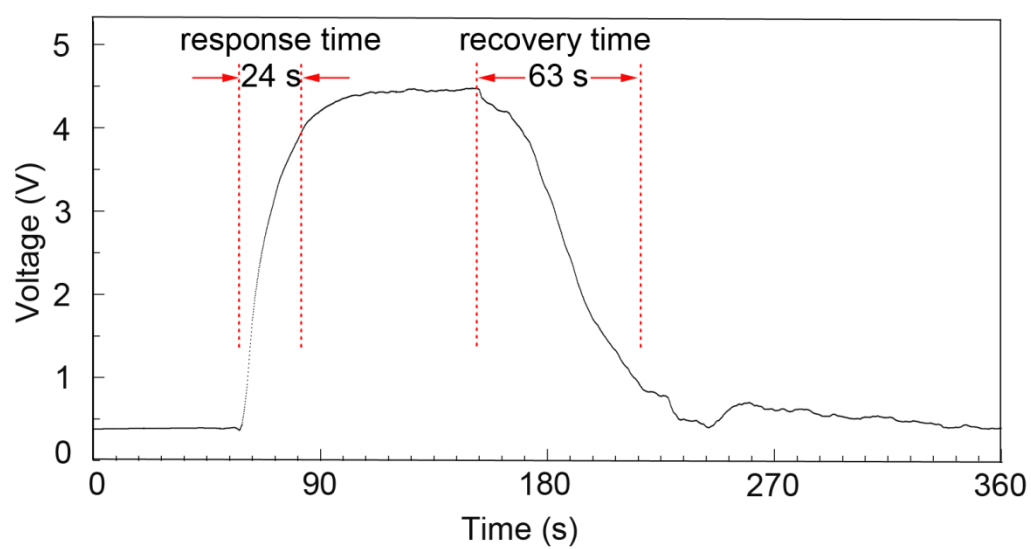


Figure S9. Response recovery curve of 1% g-C₃N₄/ZnO gas sensor to 100 ppm ethanol vapour.

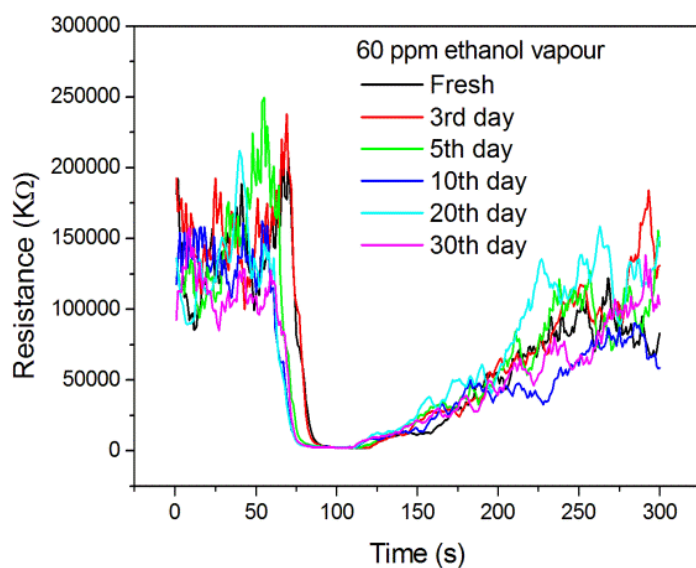


Figure S10. The dynamic resistance curves of the fresh sensor and the sensor stored for different time.

Table S3. The R_a values, R_g values and Response value of the fresh sensor and the sensor stored for different time.

Storage time (day)	R_a values of $g-C_3N_4/ZnO$ composites ($K\Omega$)	R_g values of $g-C_3N_4/ZnO$ composites ($K\Omega$)	Response value ($S=R_a/R_g$)
fresh	140525.70	2658.95	52.85
3rd	138528.53	2655.32	52.17
5th	136853.08	2655.23	51.54
10th	135601.89	2733.58	49.60
20th	133662.10	2769.60	48.26
30th	114604.20	2504.10	45.76

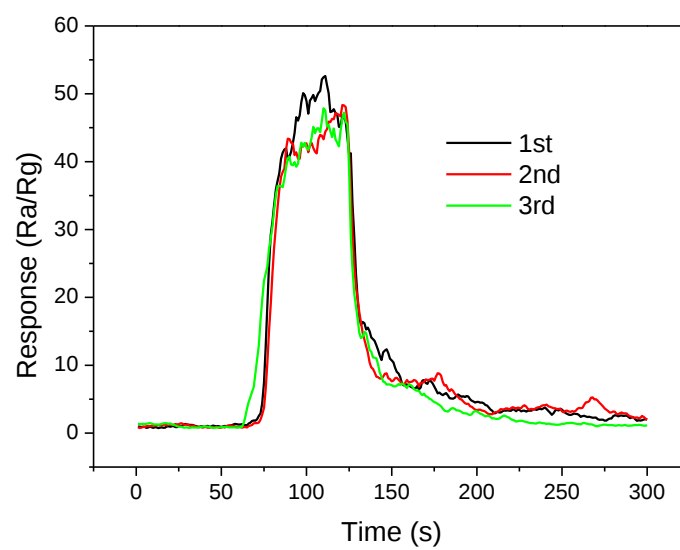


Figure S11. The repeatability of the 1% g-C₃N₄/ZnO composite based sensor to 50 ppm ethanol vapour.

Table S4. The comparative analysis of the sensing performances between this work and the previously reported results.

Sensing materials	Ethanol concentration (ppm)	Temperature (°C)	Response (Ra/Rg)	Ref.
3D SnO ₂ nanoflower	100	240	7	[S1]
Au/3D SnO ₂ microstructure	150	340	30	[S2]
RGO/hollow SnO ₂	100	300	70.4	[S3]
Pt/3D SnO ₂ nanoflower	100	320	8	[S4]
SnO ₂ /g-C ₃ N ₄	100	300	105	[S5]
Cocoon-like ZnO/ g-C ₃ N ₄	100	350	15.8	[S6]
g-C ₃ N ₄ /ZnO micro flower	100	280	81.4	This work

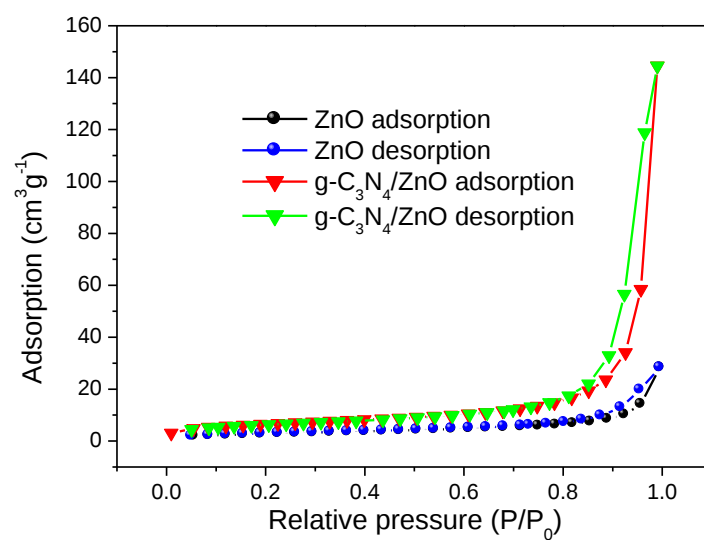


Figure S12. N₂ adsorption-desorption isotherms for the ZnO and 1%g-C₃N₄/ZnO composite.

References:

- S1. L. Wang, S. Wang, Y. Wang, H. Zhang, Y. Kang, W. Huang, Synthesis of hierarchical SnO₂ nanostructures assembled with nanosheets and their improved gas sensing properties, *Sens. Actuators B*, 2013, **188**, 85–93.
- S2. J. Guo, J. Zhang, H. B. Gong, D. X. Ju, B. Q. Cao, Au nanoparticle-functionalized 3D SnO₂ microstructures for high performance gas sensor, *Sens. Actuators B*, 2016, **226**, 266–272.
- S3. C. A. Zito, T. M. Perfecto, D. P. Volanti, Impact of reduced graphene oxide on the ethanol sensing performance of hollow SnO₂ nanoparticles under humid atmosphere, *Sens. Actuators B*, 2017, **244**, 466–474.
- S4. Y. Liu, J. Huang, J. Yang, S. Wang, Pt nanoparticles functionalized 3D SnO₂ nanoflowers for gas sensor application, *Solid State Electron.*, 2017, **130**, 20–27.
- S5. J. L. Cao, C. Qin, Y. Wang. Synthesis of g-C₃N₄ nanosheets decorated flower-like tin oxide composites and their improved ethanol gas sensing properties. *J. Alloy. Compd.*, 2017, **728**, 1101–1109.
- S6. J. Cao, Y. Gong, Y. Wang, B. Zhang, H. Zhang, G. Sun, H. Bala, Z. Zhang, Cocoon-like ZnO decorated graphitic carbon nitride nanocomposite: Hydrothermal synthesis and ethanol gas sensing application, *Mater. Lett.*, 2017, **198**, 76–80.