A simple supramolecular complex of boronic acid-appended  $\beta$ cyclodextrin and a fluorescent boronic acid-based probe with excellent selectivity for D-glucose in water

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Fig. S1. ICD spectra of **1/FPB-\betaCyD** in DMSO/water (2/98 in v/v) in the absence (black, free) and presence of 30 mM saccharides: D-glucose (blue, glc.), D-fructose (green, fru.), and D-galactose (red, gal.);  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \text{ mM}$ , 10 mM of carbonate buffer, pH = 10,  $T = 25^{\circ}\text{C}$  and I = 0.10 M.

### Fluorescence spectrum of 1/FPB-βCyD



Fig. S2. Fluorescence spectrum of the mixture of **1** (10  $\mu$ M) and **FPB-\betaCyD** (0.2 mM) in DMSO/water (2/98 in v/v): 10 mM of carbonate buffer, pH = 10, *T* = 25°C, *I* = 0.10 M, and  $\lambda_{ex}$  = 323 nm. \* denotes scattered light.



Fig. S3. Change in fluorescence spectra of **1/FPB-\betaCyD** in DMSO/water (2/98 in v/v) with time. Each spectrum was measured every 67 seconds:  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \ \text{mM}$ , 10 mM of carbonate buffer, pH = 10,  $T = 25^{\circ}\text{C}$ , I = 0.10 M, and  $\lambda_{\text{ex}} = 323 \text{ nm}$ .



Fig. S4. UV-vis adsorption spectra of the mixture of **1** (10  $\mu$ M) and **FPB-\betaCyD** (0.2 mM) in the absence and presence of 30 mM monosaccharides in DMSO/water (2/98 in v/v): 10 mM of carbonate buffer, pH = 10, *T* = 25°C, *I* = 0.10 M. Without monosaccharides (black), D-glucose (blue, glc.), D-fructose (green, fru.), and D-galactose (red, gal.).



Fig. S5-1. Fluorescence spectra of *1/FPB-\betaCyD* in DMSO/water (2/98 in v/v) (a) and enlarged spectra of Fig. S5-1(a) from 0 to 600 region of the fluorescence intensity (b):  $C_{\text{probe}} = 10 \ \mu\text{M}, C_{\text{FPB-}\beta\text{CyD}} = 0.2 \text{ mM}, 10 \text{ mM}$  of phosphate buffer,  $T = 25^{\circ}\text{C}, I = 0.10 \text{ M}, \text{ and } \lambda_{\text{ex}} = 323 \text{ nm}.$ 



Fig. S5-2. Fluorescence spectra of **1/FPB-\betaCyD** in the presence of D-glucose (30 mM) in DMSO/water (2/98 in v/v):  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \ \text{mM}$ , 10 mM of phosphate buffer,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \ \text{M}$ , and  $\lambda_{\text{ex}} = 323 \ \text{nm}$ .



Fig. S5-3. Fluorescence spectra of **1/FPB-\betaCyD** in the presence of D-fructose (30 mM) in DMSO/water (2/98 in v/v) (a) and enlarged spectra of Fig. S5-3(a) from 0 to 600 region of the fluorescence intensity (b):  $C_{\text{probe}} = 10 \,\mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \,\text{mM}$ , 10 mM of phosphate buffer,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \,\text{M}$ , and  $\lambda_{\text{ex}} = 323 \,\text{nm}$ .



Fig. S5-4. Fluorescence spectra of **1/FPB-** $\beta$ **CyD** in the presence of D-galactose (30 mM) in DMSO/water (2/98 in v/v) (a) and enlarged spectra of Fig. S5-4(a) from 0 to 700 region of the fluorescence intensity (b):  $C_{\text{probe}} = 10 \,\mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \,\text{mM}$ , 10 mM of phosphate buffer,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \,\text{M}$ , and  $\lambda_{\text{ex}} = 323 \,\text{nm}$ .



Fig. S5-5. Fluorescence intensities of **1/FPB-βCyD** at 413 (a) and 431 nm (b) under various pH conditions in the absence (black, free) and presence of saccharides (30 mM) in DMSO/water (2/98 in v/v) in Fig.s S5-1, S5-2, S5-3, and S5-4: D-glucose (blue, glc.), D-fructose (green, fru.), and D-galactose (red, gal.).

## Apparent acid dissociation constants of 1/FPB-βCyD



Weak-fluorecent

Scheme S1. Acid dissociation equilibrium of **1** (a) and the reaction of **1** with saccharides (b) in water.



Fig. S6. Fluorescence intensity of **1/FPB-\betaCyD** at 413 nm under various pH conditions in the absence (a) and presence of saccharides (30 mM) in DMSO/water (2/98 in v/v): D-glucose (b), D-fructose (c), and D-galactose (d);  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \ \text{mM}$ , 10 mM of phosphate buffer,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \ \text{M}$ , and  $\lambda_{\text{ex}} = 323 \ \text{nm}$ . Each solid curve indicates a theoretical sigmoidal curve derived from the acid dissociation model of a monobasic acid fitted by non-linear least squares analysis.



Fig. S6. Fluorescence intensity of **1/FPB-\betaCyD** at 413 nm under various pH conditions in the absence (a) and presence of saccharides (30 mM) in DMSO/water (2/98 in v/v): D-glucose (b), D-fructose (c), and D-galactose (d);  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \ \text{mM}$ , 10 mM of phosphate buffer,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \ \text{M}$ , and  $\lambda_{\text{ex}} = 323 \ \text{nm}$ . Each solid curve indicates a theoretical sigmoidal curve derived from the acid dissociation model of a monobasic acid fitted by non-linear least squares analysis. (Continued)



Fig. S7-1. Fluorescence spectra of **1/FPB-\betaCyD** at various D-glucose concentrations in DMSO/water (2/98 in v/v):  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \ \text{mM}$ , 10 mM of carbonate buffer, pH 10,  $T = 25^{\circ}\text{C}$ , I = 0.10 M, and  $\lambda_{\text{ex}} = 323 \text{ nm}$ .



Fig. S7-2. Fluorescence spectra of **1/FPB-\betaCyD** at various D-fructose concentrations in DMSO/water (2/98 in v/v) (a) and enlarged spectra of Fig. S7-2(a) from 0 to 700 region of the fluorescence intensity (b):  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \ \text{mM}$ , 10 mM of carbonate buffer, pH 10,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \ \text{M}$ , and  $\lambda_{\text{ex}} = 323 \ \text{nm}$ .



Fig. S7-3. Fluorescence spectra of **1/FPB-\betaCyD** at various D-galactose concentrations in DMSO/water (2/98 in v/v) (a) and enlarged spectra of Fig. S7-3(a) from 0 to 700 region of the fluorescence intensity (b):  $C_{\text{probe}} = 10 \,\mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \,\text{mM}$ , 10 mM of carbonate buffer, pH 10,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \,\text{M}$ , and  $\lambda_{\text{ex}} = 323 \,\text{nm}$ .



Fig. S7-4. Fluorescence spectra of **1/FPB-\betaCyD** at various D-mannose concentrations in DMSO/water (2/98 in v/v) (a) and enlarged spectra of Fig. S7-4(a) from 0 to 700 region of the fluorescence intensity (b):  $C_{\text{probe}} = 10 \,\mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \,\text{mM}$ , 10 mM of carbonate buffer, pH 10,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \,\text{M}$ , and  $\lambda_{\text{ex}} = 323 \,\text{nm}$ .



Fig. S7-5. Fluorescence spectra of **1/FPB-\betaCyD** at various D-ribose concentrations in DMSO/water (2/98 in v/v) (a) and enlarged spectra of Fig. S7-5(a) from 0 to 800 region of the fluorescence intensity (b):  $C_{\text{probe}} = 10 \,\mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \,\text{mM}$ , 10 mM of carbonate buffer, pH 10,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \,\text{M}$ , and  $\lambda_{\text{ex}} = 323 \,\text{nm}$ .



Fig. S7-6. Fluorescence spectra of **1/FPB-\betaCyD** at various D-xylose concentrations in DMSO/water (2/98 in v/v) (a) and enlarged spectra of Fig. S7-6(a) from 0 to 700 region of the fluorescence intensity (b):  $C_{\text{probe}} = 10 \,\mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \,\text{mM}$ , 10 mM of carbonate buffer, pH 10,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \,\text{M}$ , and  $\lambda_{\text{ex}} = 323 \,\text{nm}$ .

# <u>Change in fluorescence spectra of 1/FPB-βCyD with time in the presence of</u> <u>saccharides</u>

![](_page_20_Figure_1.jpeg)

Fig. S8-1. Change in fluorescence spectra of **1/FPB-\betaCyD** with time after the addition of D-glucose in DMSO/water (2/98 in v/v):  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{Dglu}} = 30 \ \text{mM}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \ \text{mM}$ , 10 mM of carbonate buffer, pH 10,  $T = 25^{\circ}\text{C}$ , I = 0.10 M, and  $\lambda_{\text{ex}} = 323 \ \text{nm}$ . Each spectrum was measured every 67 seconds.

![](_page_20_Figure_3.jpeg)

Fig. S8-2. Change in fluorescence spectra of **1/FPB-\betaCyD** with time after the addition of D-fructose in DMSO/water (2/98 in v/v):  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{Dfru}} = 30 \ \text{mM}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \ \text{mM}$ , 10 mM of carbonate buffer, pH 10,  $T = 25^{\circ}\text{C}$ , I = 0.10 M, and  $\lambda_{\text{ex}} = 323 \ \text{nm}$ . Each spectrum was measured every 67 seconds.

![](_page_21_Figure_0.jpeg)

Fig. S8-3. Change in fluorescence spectra of **1/FPB-\betaCyD** with time after the addition of D-galactose in DMSO/water (2/98 in v/v):  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{Dgal}} = 30 \ \text{mM}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \ \text{mM}$ , 10 mM of carbonate buffer, pH 10,  $T = 25^{\circ}\text{C}$ , I = 0.10 M, and  $\lambda_{\text{ex}} = 323 \ \text{nm}$ . Each data was measured every 67 seconds.

![](_page_22_Figure_0.jpeg)

Fig. S8-4. Changes in fluorescence intensities of **1/FPB-\betaCyD** at 413 and 431 nm with time after the addition of saccharides in DMSO/water (2/98 in v/v):  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \ \text{mM}$ , 10 mM of carbonate buffer, pH 10,  $T = 25^{\circ}\text{C}$ , I = 0.10 M, and  $\lambda_{\text{ex}} = 323 \text{ nm}$ . Each data was measured every 67 seconds. The data were taken from Fig.s S3, S8-1, S8-2, and S8-3.

#### Determination of the conditional equilibrium constant

The fluorescence intensity at 431 nm was measured at various D-glucose concentrations. The data were analyzed by using KaleidaGraph programme according to a theoretical equation derived from the 1:1 binding model (Eq. S1).

$$I - I_0 = \frac{I_{\rm lim} - I_0}{2C_{\rm probe}} \left\{ C_{\rm probe} + C_{\rm glucose} + \frac{1}{K'} - \left[ \left( C_{\rm probe} + C_{\rm sugar} + \frac{1}{K'} \right)^2 - 4C_{\rm probe} \cdot C_{\rm glucose} \right]^{\frac{1}{2}} \right\}$$
(S1)

where  $C_{\text{glucose}}$  is the total concentration of D-glucose; *I* and *I*<sub>0</sub> represent the fluorescence intensity at 431 nm in the presence and absence of D-glucose, respectively; *I*<sub>lim</sub> is the fluorescence intensity at 431 nm when the change of  $I (= \Delta F.I. = I - I_0)$  reaches saturation; *K*' is the conditional equilibrium constant for the reaction of  $I/\text{FPB-}\beta\text{CyD}$  with D-glucose.

![](_page_23_Figure_4.jpeg)

Fig. S9. Difference in the fluorescence intensity of **1/FPB-\betaCyD** at 431 nm ( $\Delta$ F.I. =  $I - I_0$ ) at various D-glucose concentrations in DMSO/water (2/98 in v/v):  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \text{ mM}$ , 10 mM of carbonate buffer, pH 10,  $T = 25^{\circ}\text{C}$ , I = 0.10 M, and  $\lambda_{\text{ex}} = 323 \text{ nm}$ . I and  $I_0$  denote the fluorescence intensities at 431 nm in the presence and absence of saccharides, respectively.

### **Competitive experiments**

![](_page_24_Figure_1.jpeg)

Fig. S10-1. Fluorescence spectra of **1/FPB-\betaCyD** with 5 mM D-glucose in the presence of various 0.1 mM saccharides in DMSO/water (2/98 in v/v). The spectra of **1/FPB-\betaCyD** without saccharides was also shown:  $C_{\text{probe}} = 10 \,\mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \,\text{mM}$ , 10 mM of carbonate buffer, pH 10,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \,\text{M}$ , and  $\lambda_{\text{ex}} = 323 \,\text{nm}$ .  $I \,\text{and} \, I_0$  denote the fluorescence intensities in the presence and absence of saccharides, respectively. The abbreviations, glc., fru., gal., man., rib., and xyl., denote D-glucose, D-fructose, D-galactose, D-mannose, D-ribose, and D-xylose, respectively.

![](_page_25_Figure_0.jpeg)

Fig. S10-2. The ratio of fluorescence intensities of **1**/**FPB**- $\beta$ **CyD** (*I*/*I*<sub>0</sub>) with 5 mM D-glucose in the presence of various 0.1 mM saccharides in Fig. S10-1. *I* and *I*<sub>0</sub> denote the fluorescence intensities at 431 nm in the presence and absence of saccharides, respectively.

![](_page_26_Figure_1.jpeg)

Fig. S11-1. Fluorescence spectra of **1/PB-\betaCyD** under various pH conditions in the absence (a) and presence of saccharides (30 mM) in DMSO/water (2/98 in v/v): D-glucose (b), D-fructose (c), and D-galactose (d);  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \ \text{mM}$ , 10 mM of phosphate buffer,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \ \text{M}$ , and  $\lambda_{\text{ex}} = 323 \ \text{nm}$ .

![](_page_27_Figure_0.jpeg)

Fig. S11-2. Fluorescence intensities of **1/PB-\betaCyD** at 413, 431 and 447 nm under various pH conditions in the absence (a) and presence of saccharides (30 mM) in DMSO/water (2/98 in v/v): D-glucose (b), D-fructose (c), and D-galactose (d):  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \text{ mM}$ , 10 mM of phosphate buffer,  $T = 25^{\circ}\text{C}$ , I = 0.10 M, and  $\lambda_{\text{ex}} = 323 \text{ nm}$ .

### Apparent acid dissociation constants of 1/PB-βCyD

![](_page_28_Figure_1.jpeg)

Fig. S12. Fluorescence intensity of **1/PB-\betaCyD** at 413 nm under various pH conditions in the absence (a) and presence of saccharides (30 mM) in DMSO/water (2/98 in v/v): D-glucose (b), D-fructose (c), and D-galactose (d);  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \ \text{mM}$ , 10 mM of phosphate buffer,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \ \text{M}$ , and  $\lambda_{\text{ex}} = 323 \ \text{nm}$ . Each solid curve indicates a theoretical sigmoidal curve derived from the acid dissociation model of a monobasic acid fitted by non-linear least squares analysis.

![](_page_29_Figure_0.jpeg)

Fig. S12. Fluorescence intensity of **1/PB-\betaCyD** at 413 nm under various pH conditions in the absence (a) and presence of saccharides (30 mM) in DMSO/water (2/98 in v/v): D-glucose (b), D-fructose (c), and D-galactose (d);  $C_{\text{probe}} = 10 \ \mu\text{M}$ ,  $C_{\text{FPB-}\beta\text{CyD}} = 0.2 \ \text{mM}$ , 10 mM of phosphate buffer,  $T = 25^{\circ}\text{C}$ ,  $I = 0.10 \ \text{M}$ , and  $\lambda_{\text{ex}} = 323 \ \text{nm}$ . Each solid curve indicates a theoretical sigmoidal curve derived from the acid dissociation model of a monobasic acid fitted by non-linear least squares analysis. (Continued)