## **Supplemental Material**

## Simultaneous detection of acetaminophen, catechol and hydroquinone using

### graphene-assisted electrochemical sensor

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## Table S1 Analytical results for Phenols of ITO/APTES /r-GO@Au electrode

Phenols	Fitting equation	Dynamic range (µM)	LOD (µM)	LOQ (µM)	R <sup>2</sup>
Acetaminophen	<i>Y</i> =713.59+49.57 <i>X</i>	1-500	0.82	2.73	0.9844
Catechol	<i>Y</i> =-1163.62+59.21 <i>X</i>	5-500	1.41	4.71	0.996
Hydroquinone	<i>Y</i> =2023.84+67.05 <i>X</i>	8-700	1.95	6.51	0.9912

Table S1 Analytical results for separate determination of phenols

## Table S2. Analytical results for simultaneous determination of phenols

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Phenols	Fitting equation	Dynamic range (µM)	$LOD  (\mu M)$	LOQ (µM)	<b>R</b> <sup>2</sup>
acetaminophen	<i>Y</i> =68.22 <i>X</i> +2325.89	1-180	0.12	0.43	0.997
catechol	<i>Y</i> =75.50 <i>X</i> +2535.50	5-140	0.13	0.42	0.9949
hydroquinone	<i>Y</i> =59.04 <i>X</i> +5721.67	8-200	0.11	0.36	0.9948

Table S2 Analytical results for simultaneous determination of phenols

# Table S3 Recovery results for phenolic compounds at ITO/APTES /r-GO@Au electrode

electrode						
Phenols	Added (µM)	Found (µM)	Recovery (%)	RSD (%)		
Catechol	60	63.86	106.43	0.63		
Acetaminophen	65	68.86	105.94	0.10		
Hydroquinone	88	95.54	108.57	0.31		

Table S3 Recovery results for phenolic compounds at ITO/APTES /r-GO@Au

# Table S4 Drug content determined by HPLC and EC

	HPLC				EC		
Samples	Real concentration (µM)	Detected concentrati on (µM)	Content	Real concentrat ion (µM)	Detected concentration (µM)	Content	
Acetaminoph en	231.82	200.33	86.42%	107.58	85.62	79.59%	

Table S4 Drug content determined by HPLC and EC

# Table S5 Comparison of analytical performance of phenolic compounds

Mathada	Dynamic range	LOD	Dof
Methods	(µM)	(µM)	Kel
This work	1-500	0.82	
A flow injection chemiluminescence	5 50	1.0	
method	5 - 50	1.8	1
A MIP electrochemical sensor	10 - 8000	1	2
This work	5-500	1.41	
An expanded graphite electrode			
modified with intercalated	10-1000	1.13	3
montmorillonite			
Electrodeposited molecularly			
imprinted chitosan film on BDD	0-80	0.69	4
electrodes		<ol> <li>1.41</li> <li>1.13</li> <li>0.69</li> <li>1.95</li> <li>0.66</li> </ol>	
This work	8-700	1.95	
MOF-rGO modified carbon paste	4 1000	0.66	~
electrode	4-1000	0.66	5
A nanometer cobalt/l-glutamate-	2.05.1200	0.407	<i>r</i>
modified electrode	3.85-1300	0.497	6
	MethodsThis workA flow injection chemiluminescencemethodA MIP electrochemical sensorThis workAn expanded graphite electrodemodified with intercalatedmontmorilloniteElectrodeposited molecularlyimprinted chitosan film on BDDelectrodesThis workMOF-rGO modified carbon pasteelectrodeananometer cobalt/l-glutamate-modified electrode	Methods         Dynamic range (μM)           This work         1-500           A flow injection chemiluminescence method         3           A flow injection chemiluminescence         5-50           method         10 - 8000           A MIP electrochemical sensor         10 - 8000           This work         5-500           An expanded graphite electrode         4000           modified with intercalated         10-1000           montmorillonite         4000           Electrodeposited molecularly         4000           imprinted chitosan film on BDD         0-80           electrodes         4-1000           MOF-rGO modified carbon paste         4-1000           A nanometer cobalt/1-glutamate-         3.85-1300           modified electrode         3.85-1300	MethodsDynamic range $(\mu M)$ LOD $(\mu M)$ This work1-5000.82A flow injection chemiluminescence method $5 - 50$ 1.8Methods $5 - 50$ 1.8A MIP electrochemical sensor10 - 80001This work5-5001.41An expanded graphite electrode modified with intercalated10-10001.13Mornmorillonite10-10001.13Electrodeposited molecularly imprinted chitosan film on BDD0-800.69electrodes1.95MOF-rGO modified carbon paste electrode4-10000.66A nanometer cobalt/1-glutamate- 

Table S5 Comparison of analytical performance of phenolic compounds





**Fig. S1. Comparison of HPLC and EC method.** A. Plot of various concentrations of AP *vs* peak area (HPLC). B. Plot of various concentrations of AP *vs* square of current (EC).

## Figure S2 Raman spectra of different electrodes



Fig. S2. Raman spectra of electrodes. Raman spectra of ITO glass, ITO/r-GO glass, ITO/r-GO@Au glass, ITO/APTES /r-GO glass and ITO/APTES / r-GO@Au glass electrodes.

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