

Supporting Information

g-C₃N₅-dots as fluorescence probes prepared by alkali-assisted
hydrothermal method for cell imaging

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Table S1. Comparison of different methods for the preparation of CN quantum dots.

Precursors	methods	synthesized time	references
bulk g-C ₃ N ₄	acid treatment and hydrothermal method	26 h	1
g-C ₃ N ₄ sheets	ultrasonication treatment	48 h	2
bulk g-C ₃ N ₄	electrochemical “tailoring” method	3 h	3
bulk g-C ₃ N ₄	acid treatment and ultrasonication treatment	14 h	4
GH and EDAT	hydrothermal method	24 h	5
bulk g-C ₃ N ₄	acid treatment and sonication treatment	22 h	6
bulk g-C ₃ N ₄	ultrasonic treatment and hydrothermal method	6 h	7
bulk g-C ₃ N ₄	NH ₃ treatment and liquid exfoliation	20 h	8
DAMN and 4-amino-TEMPO	hydrothermal method	6 h	9
bulk g-C ₃ N ₅	alkali assisted-hydrothermal method	60 min	present work

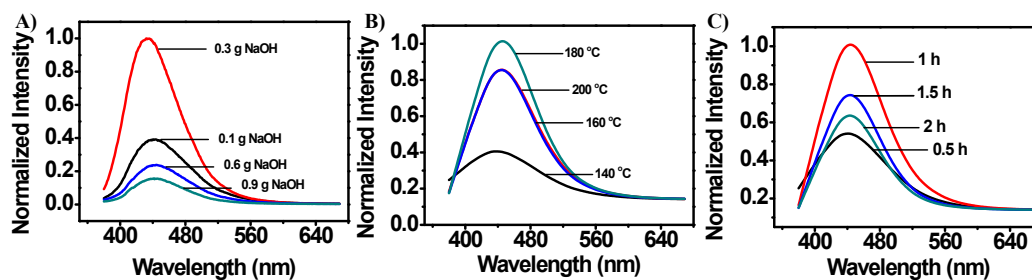


Figure S1. (A) The FL spectra of g-C₃N₅-dots synthesized by hydrothermal method using the different mass ratio of g-C₃N₅ powder and NaOH (0.03:0.1, 0.03:0.3, 0.03:0.6 and 0.03:0.9). (B) The FL spectra of g-C₃N₅-dots synthesized by hydrothermal method under different hydrothermal temperature (140 °C, 160 °C, 180 °C and 200 °C). (C) The FL spectra of g-C₃N₅-dots synthesized by hydrothermal method under different hydrothermal time (0.5 h, 1 h, 1.5 h and 2 h). The concentration of the prepared product is 0.05 mg mL⁻¹. The excitation wavelength is 360 nm. The slit widths of emission and excitation are 5 nm.

Table S2. Comparison of g-C₃N₅-dots prepared at different reaction conditions.

Factor	<i>T</i> (°C)	<i>t</i> (h)	r _{g-C₃N₅/NaOH}	FLQY (%)
1	180	1	0.03:0.1	5.1
2	180	1	0.03:0.3	12.0
3	180	1	0.03:0.6	3.6
4	180	1	0.03:0.9	2.4
5	140	1	0.03:0.3	4.7
6	160	1	0.03:0.3	10.2
7	200	1	0.03:0.3	10.9
8	180	0.5	0.03:0.3	6.8
9	180	1.5	0.03:0.3	9.4
10	180	2	0.03:0.3	7.7

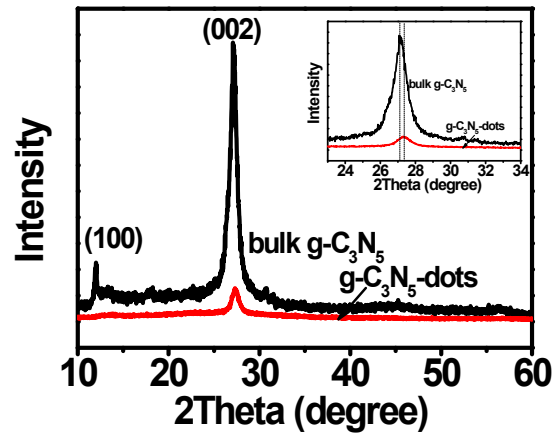


Figure S2. XRD patterns of bulk $g\text{-C}_3\text{N}_5$ and $g\text{-C}_3\text{N}_5\text{-dots}$.

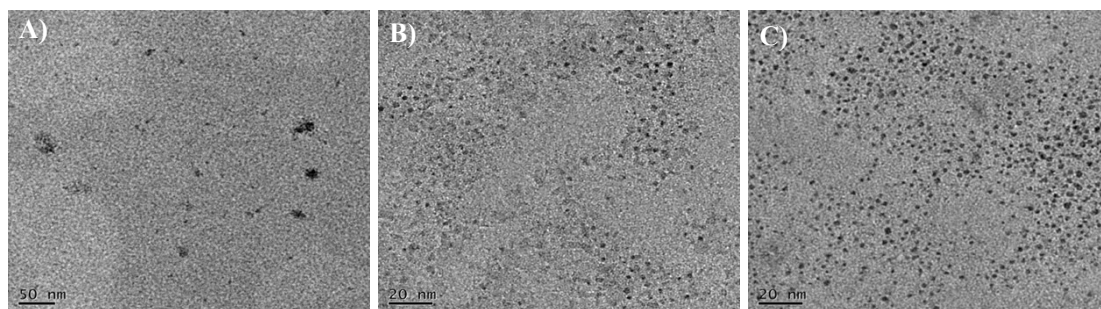


Figure S3. The TEM of as-synthesized g-C₃N₅-dots in the (A) EA, (B) CHCl₃ and (C) DMSO.

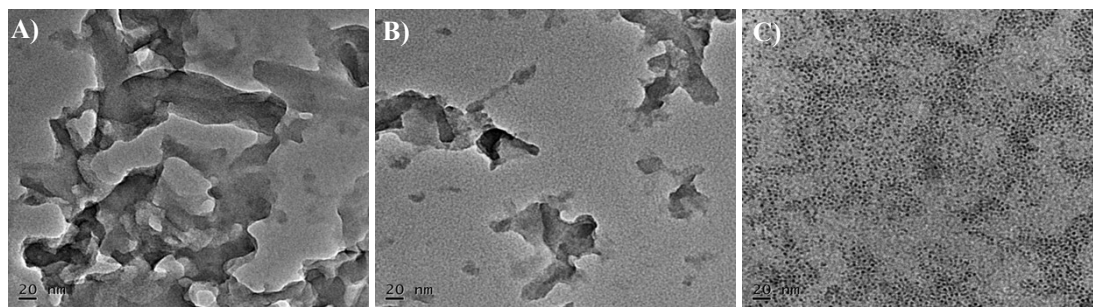


Figure S4. (A) TEM images of synthesized products prepared by hydrothermal method under (A) H_2O , (B) 1.5 M HCl and (C) 1.5 M NaOH conditions, respectively.

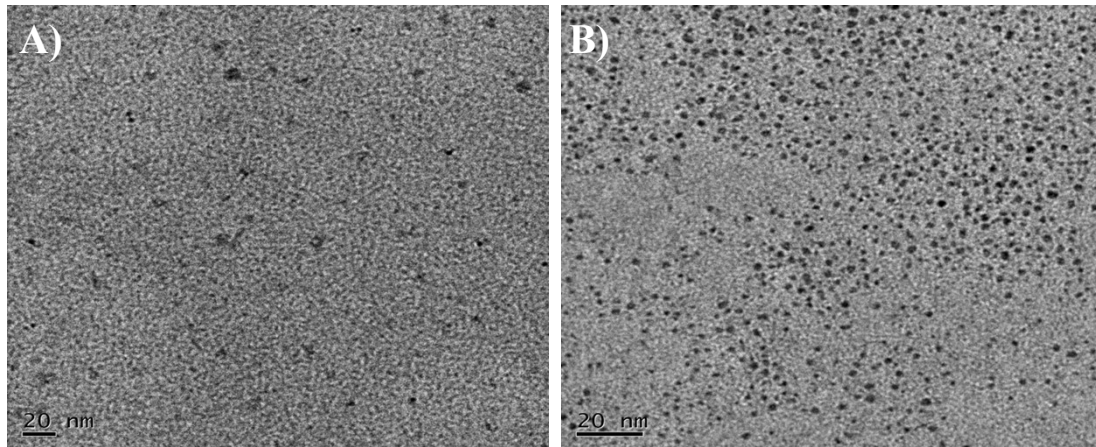


Figure S5 TEM images of the synthesized g-C₃N₅-dots products at reaction time (A) 30 min, (B) 60 min.

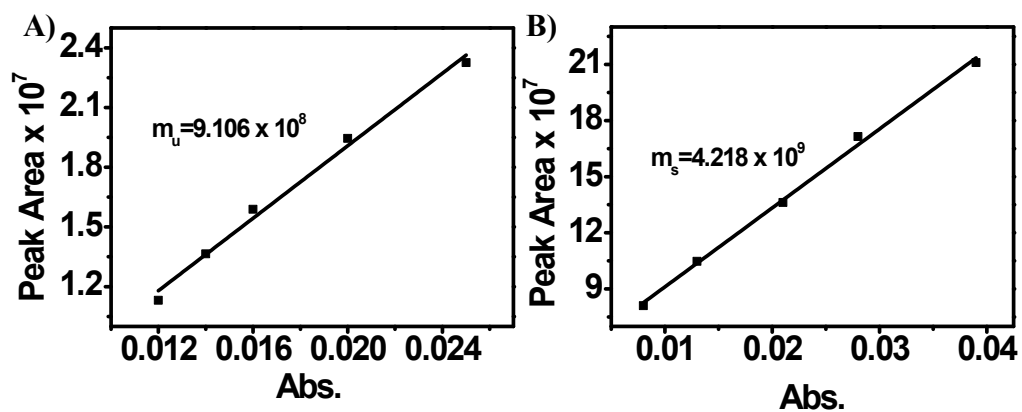


Figure S6. (A) The area of FL spectra of g-C₃N₅-dots under different absorbance. The excitation is 360 nm. Distilled water is served as solvent. (B) The area of the FL spectra of quinine sulfate under different absorbance. The excitation is 350 nm. 0.1 M H₂SO₄ is served as solvent.

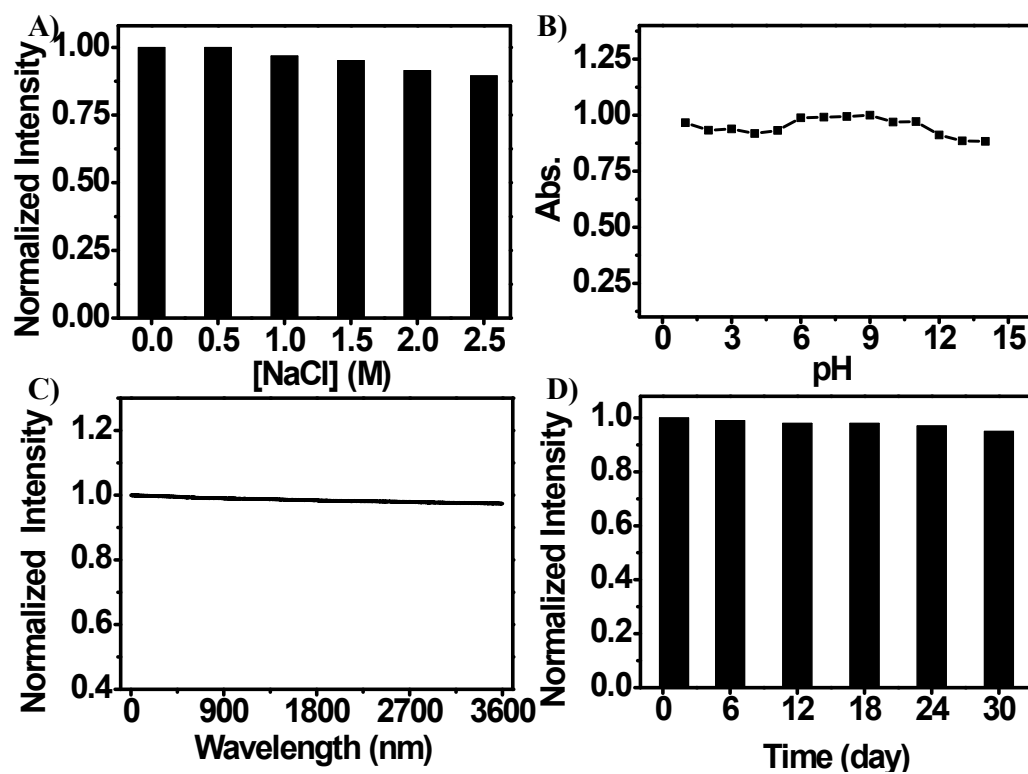


Figure S7. (A) The effect of NaCl and (B) the effect of pH value for the normalized FL intensity of the as-prepared g-C₃N₅-dots. The pH value was adjusted by 0.1 M HCl and 0.1 M NaOH. The g-C₃N₅-dots solution was kept at room temperature. (C) Time-course plot of FL intensity from g-C₃N₅-dots during continuous irradiation by a Xe lamp at 360 nm. (D) The changes of normalized FL intensity of g-C₃N₅-dots solution within 30 days.

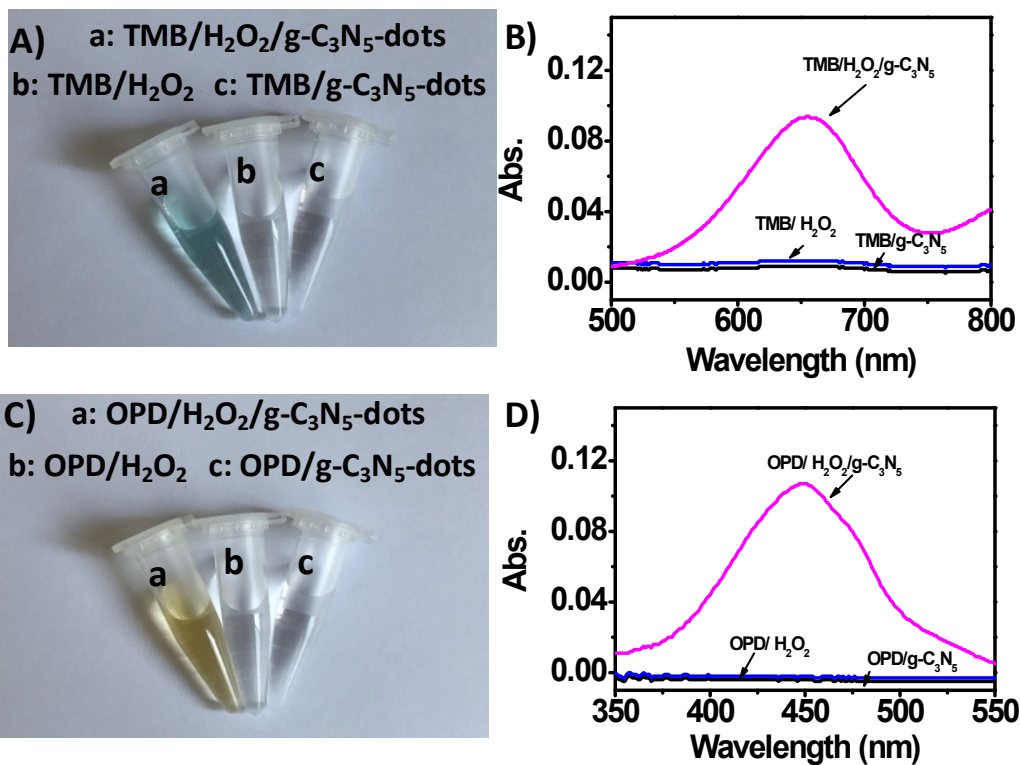


Figure S8. (A, C) the optical photographs and (B, D) the UV-vis spectra of (a) TMB (OPD)/H₂O₂/g-C₃N₅-dots, aqueous (b) OPD/H₂O₂ aqueous and (c) TMB (OPD)/g-C₃N₅-dots aqueous, respectively. (B) Size distribution histograms of g-C₃N₅-dots. [g-C₃N₅-dots]=0.05 mg mL⁻¹. [H₂O₂]=0.01 mM. [TMB]=0.5 mM and [OPD]=0.5 mM. The incubation temperature is 37 °C. The incubation time is 30 min.

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