

## Supporting Information

### Preparation and Lithium Storage of Core-Shell Honeycomb-Like $\text{Co}_3\text{O}_4$ @C Microspheres

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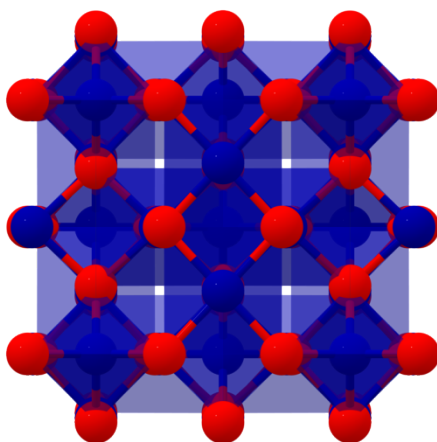


Figure S1. crystal structure of  $\text{Co}_3\text{O}_4$

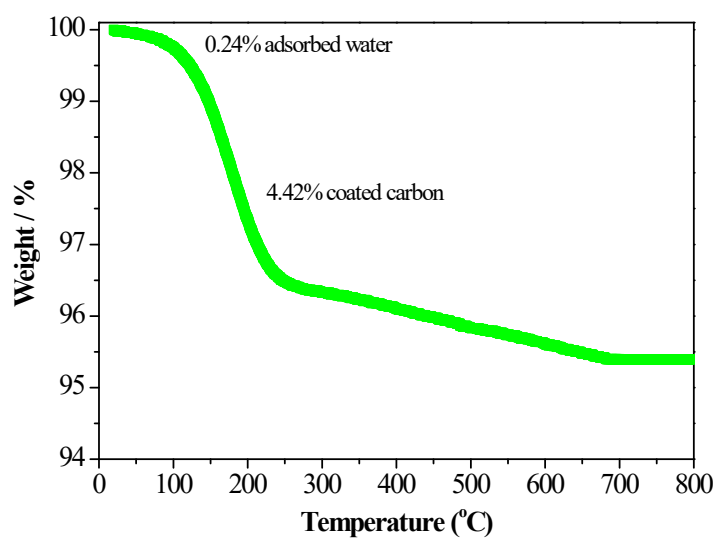


Figure S2. Thermogravimetric curve of CSHCo<sub>3</sub>O<sub>4</sub>@C microspheres

Figure S3. (a-c) SEM images, (d) HRTEM image of CSHCo<sub>3</sub>O<sub>4</sub>@C microspheres

Figure S4. (a) SEM image of CSHCo<sub>3</sub>O<sub>4</sub>@C microsphere, EDX mapping of (b) Co, (c) O, (d) C

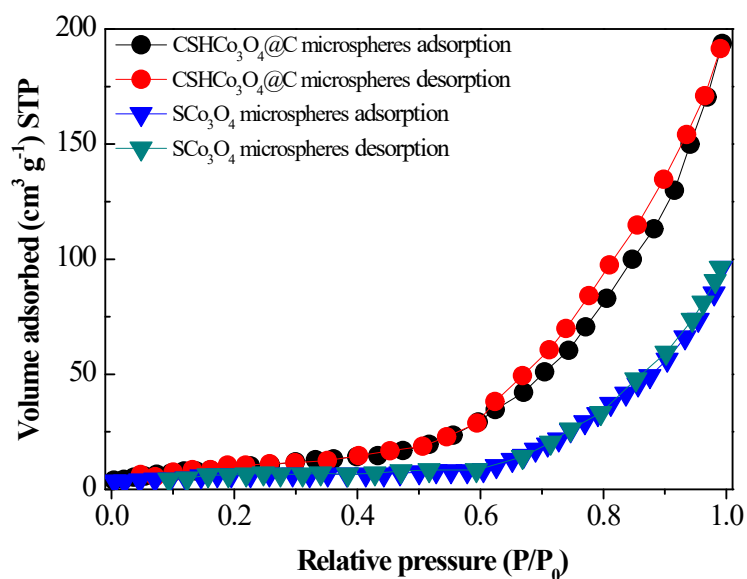


Figure S5. N<sub>2</sub> adsorption/desorption curves of CSHCo<sub>3</sub>O<sub>4</sub>@C and

## SCo<sub>3</sub>O<sub>4</sub> microspheres

Figure S6. Charge and discharge curves of (a) CSHCo<sub>3</sub>O<sub>4</sub>@C, (b) FCo<sub>3</sub>O<sub>4</sub>, and (c) SCo<sub>3</sub>O<sub>4</sub> microspheres

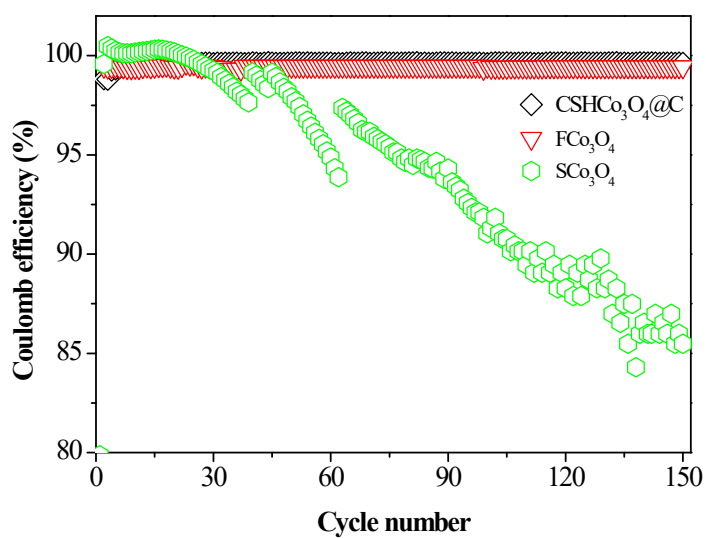


Figure S7. The coulombic efficiency of CSHCo<sub>3</sub>O<sub>4</sub>@C, SCo<sub>3</sub>O<sub>4</sub>, and FCo<sub>3</sub>O<sub>4</sub> microspheres at 0.2 C for 150 cycles

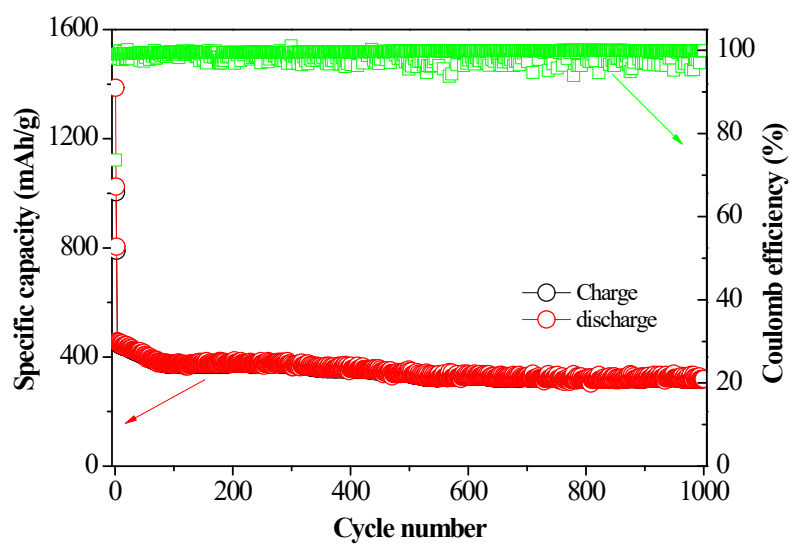


Figure S8. Cycling performance of CSHCo<sub>3</sub>O<sub>4</sub>@C microspheres at 5 C after 1000 cycles

Figure S9.  $\sigma$  values of CSHCo<sub>3</sub>O<sub>4</sub>@C, SCo<sub>3</sub>O<sub>4</sub>, and FCo<sub>3</sub>O<sub>4</sub> microspheres

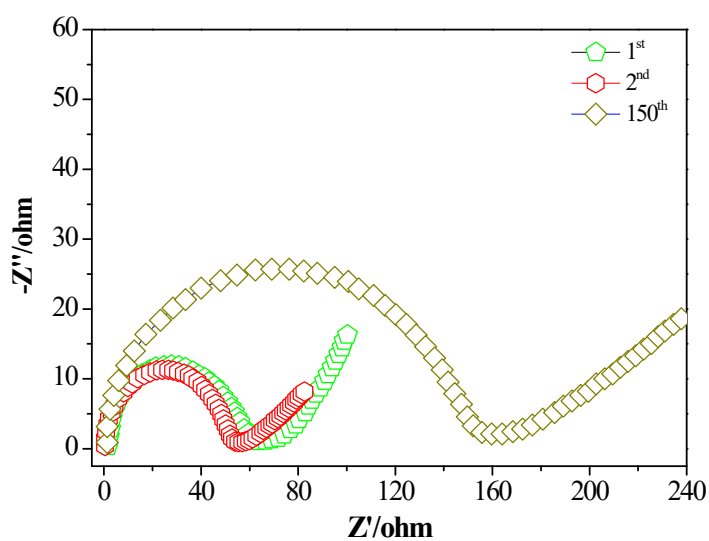


Figure S10. The 1<sup>st</sup>, 2<sup>nd</sup>, and 150<sup>th</sup> electrochemical impedance spectroscopy of CSHCo<sub>3</sub>O<sub>4</sub>@C microspheres

Table S1. The rate capability comparison of the reported cobalt oxide materials and CSHCo<sub>3</sub>O<sub>4</sub>@C microspheres

Materials	Specific capacity (mA h g <sup>-1</sup> )	Current density (mA g <sup>-1</sup> )	Reference
CoO@N-C nanocubes	309	1000	[1]
G-Co <sub>3</sub> O <sub>4</sub> rose-spheres	462.3	4450	[2]
CNFs/Co <sub>3</sub> O <sub>4</sub>	867	2000	[3]
3D hierarchical porous Co <sub>3</sub> O <sub>4</sub>	987	1200	[4]
<b>CSHCo<sub>3</sub>O<sub>4</sub>@C</b>	<b>318.9</b>	<b>8900</b>	<b>This work</b>

Table S2. Electrochemical performance comparison of the reported cobalt oxide materials with different structure and CSHCo<sub>3</sub>O<sub>4</sub>@C microspheres

Materials	Specific capacity (mA h g <sup>-1</sup> )	Current density (mA g <sup>-1</sup> )	Cycles	Reference
Co <sub>3</sub> O <sub>4</sub> hexapods	166	90	100	[5]
CoO nanoparticles	458	200	80	[6]
Pristine CoO nanorods	259	71.6	50	[7]
Co <sub>3</sub> O <sub>4</sub> /carbon nanowires	534	100	20	[8]
<b>CSHCo<sub>3</sub>O<sub>4</sub>@C</b>	<b>1091.2</b>	<b>178</b>	<b>150</b>	<b>This work</b>

Table S3. The resistance values of CSHCo<sub>3</sub>O<sub>4</sub>@C, FCo<sub>3</sub>O<sub>4</sub>, SCo<sub>3</sub>O<sub>4</sub> microspheres after fitting of EIS data

Materials	R <sub>s</sub> (ohm/cm <sup>2</sup> )	Q <sub>1</sub> (μF/cm <sup>2</sup> )	R <sub>ct</sub> (ohm/cm <sup>2</sup> )	Q <sub>2</sub> (μF/cm <sup>2</sup> )
CSHCo <sub>3</sub> O <sub>4</sub> @C	1.38	1.86	82.15	2534.18
FCo <sub>3</sub> O <sub>4</sub>	1.09	1.57	112.36	5867.39
SCo <sub>3</sub> O <sub>4</sub>	1.21	1.73	135.78	1037.47

Table S4. The resistance values of CSHCo<sub>3</sub>O<sub>4</sub>@C microspheres during cycling

CSHCo <sub>3</sub> O <sub>4</sub> @C	R <sub>s</sub> (ohm/cm <sup>2</sup> )	Q <sub>1</sub> (μF/cm <sup>2</sup> )	R <sub>ct</sub> (ohm/cm <sup>2</sup> )	Q <sub>2</sub> (μF/cm <sup>2</sup> )
1 <sup>st</sup>	3.57	0.76	63.37	3123.56
2 <sup>nd</sup>	2.83	0.37	53.85	3908.51
150 <sup>th</sup>	1.69	1.35	157.63	7325.56

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