Supporting Information

Synthesis, characterizations and Anticancer Activities of Zn²⁺, Cu²⁺,

Co²⁺ and Ni²⁺ Complexes Involving Chiral Amino Alcohols

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Figure S3 UV spectra of complexes (I) - (VII) in the ranges of 200-800 nm, showing individual UV-

Vis absorption spectrum of each compound.

Complex (I)						
Ni1-N1	2.071(4)	Ni1-01	2.078(3)	Ni1-N2	2.085(4)	
Ni1-O3	2.076(3)	Ni1-N3	2.085(4)	Ni1-O2	2.130(3)	
N1-Ni1-O3	91.69(16)	O3-Ni1-N3	81.43(14)	O3-Ni1-N2	97.90(14)	
N1-Ni1-O1	79.95(15)	O1-Ni1-N3	97.75(15)	O1-Ni1-N2	90.69(14)	
O3-Ni1-O1	171.41(13)	N1-Ni1-N2	164.77(16)	N3-Ni1-N2	94.70(15)	
N1-Ni1-N3	98.42(16)	N1-Ni1-O2	88.32(15)	O3-Ni1-O2	88.48(14)	
01-Ni1-O2	93.19(15)	N3-Ni1-O2	167.99(14)	N2-Ni1-O2	80.17(14)	
		Complex	х (II)			
Col-N1	1.931(6)	Co1-N2	1.942(6)	Co1-N3	1.943(6)	
Co1-Co2	2.644(12)	Co2-O4	2.056(5)	Co2-O3	2.078(5)	
Co2-O5	2.088(5)	Co2-O1	2.117(5)	Co2-O6	2.127(6)	
Co2-O2	2.127(5)	Co2-Co3	2.63(13)	Co3-O4	1.897(5)	
Co3-O6	1.901(6)	Co3-O5	1.908(5)	Co3-N4	1.937(7)	
Co3-N5	1.940(7)	Co3-N6	1.948(6)			
O2-Co1-O1	85.8(2)	N3-Co1-Co2	121.0(2)	O3-Co2-Co1	46.11(14)	
O2-Co1-O3	86.3(2)	O4-Co2-O1	177.0(2)	O5-Co2-Co1	135.72(15)	
O1-Co1-O3	86.7(2)	O3-Co2-O1	77.8(2)	O1-Co2-Co1	45.88(13)	
O2-Co1-N1	87.1(2)	O5-Co2-O1	105.0(2)	O4-Co3-N4	85.5(3)	
O1-Co1-N1	93.3(2)	O4-Co2-O6	76.3(2)	O6-Co3-N4	171.0(3)	
O3-Co1-N1	173.8(2)	O3-Co2-O6	104.5(2)	O5-Co3-N4	92.0(3)	
O2-Co1-N2	86.5(2)	O5-Co2-O6	75.4(2)	O4-Co3-N5	173.7(3)	
O1-Co1-N2	172.3(3)	O1-Co2-O6	105.3(2)	O6-Co3-N5	93.5(3)	
O3-Co1-N2	92.6(2)	O4-Co2-O2	102.9(2)	O5-Co3-N5	87.3(3)	
N1-Co1-N2	93.6(3)	O3-Co2-O2	76.95(19)	N4-Co3-N5	94.9(3)	
O2-Co1-N3	172.9(3)	O5-Co2-O2	103.1(2)	O4-Co3-N6	91.6(3)	
O1-Co1-N3	92.4(3)	O1-Co2-O2	75.64(19)	O6-Co3-N6	87.0(3)	
O3-Co1-N3	86.7(2)	O6-Co2-O2	178.4(2)	O5-Co3-N6	172.0(3)	
N1-Co1-N3	93.5(3)	O4-Co2-Co3	45.62(15)	N4-Co3-N6	95.5(3)	
N2-Co1-N3	95.2(3)	O3-Co2-Co3	132.20(14)	N5-Co3-N6	94.6(3)	
O2-Co1-Co2	52.76(15)	O5-Co2-Co3	45.84(14)	O4-Co3-Co2	50.79(16)	
O1-Co1-Co2	52.39(15)	O1-Co2-Co3	137.23(14)	O6-Co3-Co2	52.92(16)	
O3-Co1-Co2	51.22(15)	O6-Co2-Co3	45.49(15)	O5-Co3-Co2	51.70(16)	
N1-Co1-Co2	124.1(18)	O2-Co2-Co3	132.99(15)	N4-Co3-Co2	119.0(2)	
N2-Co1-Co2	121.8(2)	O4-Co2-Co1	131.28(15)	N5-Co3-Co2	124.2(2)	
N6-Co3-Co2	121.6(2)	O6-Co2-Co1	136.05(16)	O2-Co2-Co1	45.51(14)	
Complex (III)						
Cu1-O5	2.000(3)	Cu1-O3	2.020(2)	Cu1-O2	2.502(3)	
Cu1-N2	2.003(3)	Cu1-O1	2.493(3)			
N1-Cu1-O5	89.41(10)	N2-Cu1-O3	88.57(11)	N1-Cu1-O2	103.47(11)	
N1-Cu1-N2	177.33(16)	N1-Cu1-O1	77.88(11)	O5-Cu1-O2	90.10(9)	
O5-Cu1-N2	92.97(14)	O5-Cu1-O1	91.94(9)	N2-Cu1-O2	77.74(11)	
N1-Cu1-O3	89.08(14)	N2-Cu1-O1	100.83(11)	O3-Cu1-O2	89.15(9)	
O5-Cu1-O3	178.11(12)	O3-Cu1-O1	88.86(9)	O1-Cu1-O2	177.57(7)	
Complex (IV)						
Cu1-N1	1.987(11)	Cu1-O1S	2.770(10)	Cu2-O2A	1.967(9)	
Cu1-N2	2.007(10)	Cu2-N1A	1.976(12)	Cu2-O3S	2.597(11)	

Table S1 Selected bond lengths [Å] and angles [°] for complexes (I) – (VII).

Cu1-O1	1.954(9)	Cu2-N2A	1.986(14)	Cu2-O9S	2.71(3)
Cu1-O2	1.935(8)	Cu2-O1A	1.911(11)		
N1-Cu1-N2	96.3(4)	O2-Cu1-N1	173.6(4)	N1A-Cu2-O9S	96.9(8)
N1-Cu1-O1S	84.4(4)	O2-Cu1-N2	86.0(4)	N2A-Cu2-O3S	85.6(6)
N2-Cu1-O1S	81.1(4)	O2-Cu1-O1	93.3(4)	N2A-Cu2-O9S	82.9(11)
O1-Cu1-N1	85.4(4)	O2-Cu1-O1S	90.1(3)	O1A-Cu2-N1A	84.9(5)
O1-Cu1-N2	170.6(4)	N1A-Cu2-N2A	98.3(6)	O1A-Cu2-N2A	175.1(7)
O1-Cu1-O1S	108.3(3)	N1A-Cu2-O3S	88.4(5)	O1A-Cu2-O2A	91.6(5)
O1A-Cu2-O3S	98.3(4)	O1A-Cu2-O9S	93.0(10)	O2A-Cu2-N1A	173.5(5)
O2A-Cu2-N2A	85.6(5)	O2A-Cu2-O3S	86.8(4)	O2A-Cu2-O9S	88.7(8)
O3S-Cu2-O9S	167.9(8)				
		Complex	K (V)		
Zn1-O2	2.028(11)	Zn1-N2	2.085(12)	Zn2-Cl3	2.227(5)
Zn1-N1	2.047(13)	Zn1-O3	2.310(10)	Zn2-Cl1	2.283(4)
Zn1-N3	2.068(13)	Zn2-O2	1.918(11)	Zn2-Cl2	2.345(4)
O2-Zn1-N1	109.5(5)	N1-Zn1-N2	122.0(5)	N3-Zn1-O3	73.9(5)
O2-Zn1-N3	97.2(5)	N3-Zn1-N2	127.3(5)	N2-Zn1-O3	93.1(4)
N1-Zn1-N3	107.6(5)	O2-Zn1-O3	165.5(5)	O2-Zn2-Cl3	114.5(3)
O2-Zn1-N2	83.1(4)	N1-Zn1-O3	84.5(5)	O2-Zn2-Cl1	102.0(3)
Cl3-Zn2-Cl1	119.37(19)	O2-Zn2-Cl2	109.3(4)	Cl3-Zn2-Cl2	110.68(18)
Cl1-Zn2-Cl2	99.61(16)				
		Complex	(VI)		
Col-O3	1.883(9)	Co1-N1	1.946(12)	Co2-O3	2.095(10)
Col-Ol	1.900(10)	Co1-N3	1.958(12)	Co2-O5	2.111(9)
Co1-O2	1.920(10)	Co1-Co2	2.650(2)	Co2-O2	2.112(9)
Co1-N2	1.937(11)	Co2-O6	2.058(8)	Co2-O1	2.113(9)
Co2-O4	2.125(9)	Co2-Co3	2.649(2)	Co3-O6	1.882(9)
Co3-N6	1.902(11)	Co3-O4	1.905(9)	Co3-O5	1.945(8)
Co3-N5	1.958(10)	Co3-N4	1.988(11)		
O3-Co1-O1	86.0(4)	O3-Co1-Co2	51.7(3)	O2-Co2-O1	75.8(3)
O3-Co1-O2	86.7(4)	O1-Co1-Co2	52.2(3)	O6-Co2-O4	75.9(3)
O1-Co1-O2	85.6(4)	O2-Co1-Co2	52.1(3)	O3-Co2-O4	100.5(3)
O3-Co1-N2	173.8(5)	N2-Co1-Co2	122.6(4)	O5-Co2-O4	77.4(3)
O1-Co1-N2	91.9(5)	N1-Co1-Co2	121.5(3)	O2-Co2-O4	175.6(4)
O2-Co1-N2	87.3(4)	N3-Co1-Co2	121.1(4)	O1-Co2-O4	106.9(3)
O3-Co1-N1	91.5(4)	O6-Co2-O3	108.0(4)	O6-Co2-Co3	45.0(2)
01-Co1-N1	86.9(4)	O6-Co2-O5	76.3(3)	O3-Co2-Co3	134.6(2)
O2-Co1-N1	172.4(5)	O3-Co2-O5	174.7(3)	O5-Co2-Co3	46.5(2)
N2-Co1-N1	94.3(5)	O6-Co2-O2	101.7(3)	O2-Co2-Co3	134.8(3)
O3-Co1-N3	88.0(4)	O3-Co2-O2	76.7(4)	O1-Co2-Co3	134.5(3)
O1-Co1-N3	173.2(4)	O5-Co2-O2	105.7(4)	O4-Co2-Co3	45.4(2)
O2-Co1-N3	90.8(5)	O6-Co2-O1	175.2(4)	O6-Co2-Co1	135.3(3)
N2-Co1-N3	93.7(5)	O3-Co2-O1	75.6(4)	O3-Co2-Co1	44.9(2)
N1-Co1-N3	96.6(5)	O5-Co2-O1	100.3(4)	O5-Co2-Co1	133.9(2)
O2-Co2-Co1	45.9(3)	O6-Co3-N6	88.6(4)	N6-Co3-O5	173.2(4)
O1-Co2-Co1	45.3(3)	O6-Co3-O4	85.6(3)	O4-Co3-O5	86.9(4)
O4-Co2-Co1	133.9(2)	N6-Co3-O4	92.5(4)	O6-Co3-N5	91.6(5)
Co3-Co2-Co1	179.29(9)	O6-Co3-O5	84.6(4)	N6-Co3-N5	92.9(5)
O4-Co3-N5	173.8(4)	O5-Co3-N5	87.3(4)	O6-Co3-N4	171.6(4)

N6-Co3-N4	93.1(5)	O4-Co3-N4	86.1(4)	O5-Co3-N4	93.6(4)
N5-Co3-N4	96.6(5)	O6-Co3-Co2	50.6(2)	N6-Co3-Co2	122.8(3)
O4-Co3-Co2	52.6(3)	O5-Co3-Co2	52.0(3)	N5-Co3-Co2	121.6(4)
N4-Co3-Co2	122.3(3)				
		Complex	(VII)		
Zn1-N1	2.094(6)	Zn1-O3	2.191(6)	Zn1-O2	2.194(5)
Zn1-N3	2.101(6)	Zn1-O1	2.203(5)	Zn1-N2	2.105(6)
N1-Zn1-N3	98.6(3)	N2-Zn1-O3	94.8(2)	N1-Zn1-O1	78.7(2)
N1-Zn1-N2	100.1(2)	N1-Zn1-O2	94.5(2)	N3-Zn1-O1	94.2(2)
N3-Zn1-N2	99.3(3)	N3-Zn1-O2	166.9(3)	N2-Zn1-O1	166.5(2)
N1-Zn1-O3	165.1(2)	N2-Zn1-O2	79.4(2)	O3-Zn1-O1	86.8(2)
N3-Zn1-O3	78.8(3)	O3-Zn1-O2	88.2(2)	O2-Zn1-O1	87.2(2)

Table S2 Hydrogen bond lengths (Å) and bond angles (°) for complexes (I) - (VII).

D –H···A	d(D-H)	d(H···A)	d(D…A)	ZDHA		
Complex (I)						
O10-H10DO4	0.85(3)	2.19(9)	2.752(7)	123(9)		
O11-H11DO4	0.85	2.08	2.816(6)	144.7		
O10-H10EO8	0.85	2.31	3.119(6)	159.1		
C12-H12O5#1	0.98	2.57	3.517(7)	163.8		
N3-H3BO6#2	0.97	2.41	3.273(6)	147.6		
N3-H3AO7#3	0.97	2.32	3.258(6)	162.6		
N2-H2BO6#2	0.97	2.18	3.065(6)	151.8		
N2-H2AO5#1	0.97	2.22	3.128(6)	155.0		
N1-H1BO7	0.97	2.11	3.065(6)	168.9		
N1-H1AN5#3	0.97	2.70	3.657(7)	171.0		
N1-H1AO9 ^{#3}	0.97	2.17	3.104(6)	162.1		
O3-H3O11	0.856(12)	1.80(2)	2.610(5)	158(5)		
O2-H2O8	0.870(13)	1.92(2)	2.767(5)	165(4)		
O1-H1O10 ^{#2}	0.848(12)	1.766(17)	2.608(5)	172(6)		
O1-H1O10 ^{#2}	0.848(12)	1.766(17)	2.608(5)	172(6)		
O2-H2O8	0.870(13)	1.92(2)	2.767(5)	165(4)		
O3-H3O11	0.856(12)	1.80(2)	2.610(5)	158(5)		
N1-H1AO9#3	0.97	2.17	3.104(6)	162.1		
N1-H1AN5 ^{#3}	0.97	2.70	3.657(7)	171.0		
N1-H1BO7	0.97	2.11	3.065(6)	168.9		
N2-H2AO5#1	0.97	2.22	3.128(6)	155.0		
N2-H2BO6#2	0.97	2.18	3.065(6)	151.8		
N3-H3AO7#3	0.97	2.32	3.258(6)	162.6		
N3-H3BO6#2	0.97	2.41	3.273(6)	147.6		
C12-H12O5#1	0.98	2.57	3.517(7)	163.8		
O10-H10EO8	0.85	2.31	3.119(6)	159.1		
O11-H11DO4	0.85	2.08	2.816(6)	144.7		
O10-H10DO4	0.85(3)	2.19(9)	2.752(7)	123(9)		
O10-H10DO4	0.85(3)	2.19(9)	2.752(7)	123(9)		

	Comp	lex (III)		
N1-H1BO1#1	0.85(3)	2.21(4)	3.039(4)	166(3)
N2-H2BO2#2	0.83(3)	2.30(3)	3.066(4)	154(3)
N2-H2AO5#2	0.87(3)	2.64(3)	3.088(5)	113(2)
C14-H14BO6#1	0.98	2.43	3.374(5)	160.8
C12-H12AO4#2	0.98	2.39	3.350(4)	165.8
С8-Н8О5	1.00	2.48	3.251(4)	133.5
C6-H6AO6#3	0.99	2.52	3.502(4)	171.7
C2-H2CO3	1.00	2.59	3.151(4)	115.6
O2-H2O4	0.84	1.83	2.634(3)	159.0
O1-H1O6	0.84	1.78	2.618(3)	174.0
N1-H1BO1#1	0.85(3)	2.21(4)	3.039(4)	166(3)
N2-H2BO2#2	0.83(3)	2.30(3)	3.066(4)	154(3)
N2-H2AO5#2	0.87(3)	2.64(3)	3.088(5)	113(2)
C14-H14BO6#1	0.98	2.43	3.374(5)	160.8
C12-H12AO4#2	0.98	2.39	3.350(4)	165.8
С8-Н8О5	1.00	2.48	3.251(4)	133.5
C6-H6AO6#3	0.99	2.52	3.502(4)	171.7
C2-H2CO3	1.00	2.59	3.151(4)	115.6
O2-H2O4	0.84	1.83	2.634(3)	159.0
O1-H1O6	0.84	1.78	2.618(3)	174.0
O1-H1O6	0.84	1.78	2.618(3)	174.0
O2-H2O4	0.84	1.83	2.634(3)	159.0
C2-H2CO3	1.00	2.59	3.151(4)	115.6
C6-H6AO6#3	0.99	2.52	3.502(4)	171.7
С8-Н8О5	1.00	2.48	3.251(4)	133.5
C12-H12AO4#2	0.98	2.39	3.350(4)	165.8
C14-H14BO6#1	0.98	2.43	3.374(5)	160.8
N2-H2AO5#2	0.87(3)	2.64(3)	3.088(5)	113(2)
N2-H2BO2#2	0.83(3)	2.30(3)	3.066(4)	154(3)
N1-H1BO1#1	0.85(3)	2.21(4)	3.039(4)	166(3)
O1-H1O6	0.84	1.78	2.618(3)	174.0
O2-H2O4	0.84	1.83	2.634(3)	159.0
C2-H2CO3	1.00	2.59	3.151(4)	115.6
C6-H6AO6#3	0.99	2.52	3.502(4)	171.7
С8-Н8О5	1.00	2.48	3.251(4)	133.5
C12-H12AO4#2	0.98	2.39	3.350(4)	165.8
C14-H14BO6#1	0.98	2.43	3.374(5)	160.8
N2-H2AO5#2	0.87(3)	2.64(3)	3.088(5)	113(2)
N2-H2BO2#2	0.83(3)	2.30(3)	3.066(4)	154(3)
N1-H1B01 ^{#1}	0.85(3)	2.21(4)	3.039(4)	166(3)
	Comp	lex (IV)		
N1-H1AO4S#1	0.91	2.19	3.015(15)	150.8
N1-H1BO2#2	0.91	1.98	2.880(14)	169.6

Complex (III)

N2-H2A01#2	0.91	2.03	2.925(13)	166.5			
N2-H2AO1S	0.91	2.72	3.158(14)	110.6			
01-H101A	0.84	1.73	2.429(14)	139.0			
N1A-H1AAO6S	0.91	2.31	3.02(3)	135.1			
N2A-H2ABO9S	0.91	2.72	3.16(4)	110.6			
N2A-H2ACO5S	0.91	2.26	3.16(3)	176.0			
O2A-H2ADO2	0.877(14)	2.05(15)	2.445(13)	106(12)			
O9S-H9SBO6S#3	0.87	2.38	3.15(4)	148.7			
	Comp	lex (V)					
C24-H24BO1#1	0.98	2.43	3.338(18)	153.4			
C23-H23Cl1 ^{#1}	0.99	2.75	3.576(17)	141.9			
C7-H7Cl3#2	0.99	2.83	3.782(16)	161.8			
N3-H3BCl2	0.98	2.77	3.567(13)	138.8			
N3-H3BCl1	0.98	2.83	3.512(13)	127.4			
N3-H3ACl2#3	0.98	2.43	3.338(12)	153.1			
N2-H2BCl3#2	0.98	2.69	3.613(13)	156.2			
N1-H1BCl1	0.98	2.46	3.295(13)	142.8			
N1-H1ACl1 ^{#1}	0.98	2.48	3.418(12)	160.4			
O3-H3Cl2#2	0.866(13)	2.49(9)	3.221(11)	142(12)			
O1-H1Cl2 ^{#1}	0.83	2.53	3.281(12)	150.5			
O1-H1Cl2#1	0.83	2.53	3.281(12)	150.5			
O3-H3Cl2#2	0.866(13)	2.49(9)	3.221(11)	142(12)			
N1-H1ACl1 ^{#1}	0.98	2.48	3.418(12)	160.4			
N1-H1BCl1	0.98	2.46	3.295(13)	142.8			
N2-H2BCl3#2	0.98	2.69	3.613(13)	156.2			
N3-H3ACl2#3	0.98	2.43	3.338(12)	153.1			
N3-H3BCl1	0.98	2.83	3.512(13)	127.4			
N3-H3BCl2	0.98	2.77	3.567(13)	138.8			
C7-H7Cl3#2	0.99	2.83	3.782(16)	161.8			
C23-H23Cl1#1	0.99	2.75	3.576(17)	141.9			
C24-H24BO1#1	0.98	2.43	3.338(18)	153.4			
Complex (VII)							
N3-H3BCl2 ^{#1}	0.97	2.63	3.456(8)	143.4			
N3-H3ACl1#2	0.97	2.46	3.378(7)	157.3			
N2-H2BCl2#1	0.97	2.51	3.361(7)	145.9			
N2-H2ACl1	0.97	2.42	3.334(7)	157.2			
N1-H1BCl2 ^{#1}	0.97	2.40	3.268(7)	148.7			
N1-H1ACl2	0.97	2.37	3.296(7)	159.9			
O3-H3Cl1	0.863(13)	2.56(8)	3.057(6)	118(7)			
O2-H2Cl2	0.867(13)	2.43(6)	3.058(6)	130(6)			
O1-H1Cl1#2	0.860(13)	2.53(7)	3.071(7)	122(7)			
O1-H1Cl1 ^{#2}	0.860(13)	2.53(7)	3.071(7)	122(7)			
O2-H2Cl2	0.867(13)	2.43(6)	3.058(6)	130(6)			
O3-H3Cl1	0.863(13)	2.56(8)	3.057(6)	118(7)			

N1-H1ACl2	0.97	2.37	3.296(7)	159.9
N1-H1BCl2#1	0.97	2.40	3.268(7)	148.7
N2-H2ACl1	0.97	2.42	3.334(7)	157.2
N2-H2BCl2#1	0.97	2.51	3.361(7)	145.9
N3-H3ACl1#2	0.97	2.46	3.378(7)	157.3
N3-H3BCl2#1	0.97	2.63	3.456(8)	143.4

Symmetry transformations used to generate equivalent atoms: For (I), #1 = x + 1/2, -y + 3/2, -z + 1; #2 = x + 1, y, z; #3 = x + 1/2, -y + 1/2, -z + 1. For (III), #1 = x + 1, y, z; #2 = x - 1, y, z; #3 = x, y, z - 1. For (IV), #1 = x + 1/2, -y + 3/2, -z + 1; #2 = x - 1/2, -y + 3/2, -z + 1; #2 = x - 1/2, -y + 3/2, -z + 1; #2 = x - 1/2, -z + 1; #3 = x + 1, y, z. For (V), #1 = -x + 2, y - 1/2, -z + 1; #2 = x, y - 1, z; #3 = -x + 1, y - 1/2, -z + 1. For (VII), #1 = x + 1/2, -y + 3/2, -z + 1; #2 = -x + 1, y + 1/2, -z + 1/2.





(c)



Figure S1 The crystal packing of complexes (I) - (VII), showing the formation of the threedimensional network by hydrogen bonding interactions, (a) for (I), (b) for (II), (c) for (III), (d) for (IV), (e) for (V), (f) for (VI) and (g) for (VII), respectively.



Figure S2 IR spectrums of complexes (I) - (VII).



Figure S3 UV spectrum of complexes (I) - (VII).