

Supporting information

Highly effective removal of Hg(II) solution using corn bract@MoS₂ as a new biomass adsorbent

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Materials and instruments

CB come from farmers' markets; Ammonium molybdate tetrahydrate solution (AR 20%) and thiourea were purchased from Tianjin Daomao Chemical Reagent Company (Tianjin, China); Mercurium nitrate was supplied by Guizhou Tongren Tailuier Chemical Plant (Guizhou, China). FT-IR spectrum was recorded on the PerkinElmer spectrum One (B) spectrometer using KBr particles in the range of 4000-500 cm⁻¹. Using the Bruker D8 device, X-ray powder diffraction (XRD) spectrum of the Cu-K α radiation ($\lambda=1.54\text{\AA}$) sample under 40kv and 40ma was obtained in the range of 1~10 $^{\circ}$ (2 θ). The morphology of the sample was observed with a scanning

electron microscope (SEM, JEOL-6500F). Using Netzsch 209C under N₂ flow conditions, the heating rate is 20 °C min⁻¹ to perform thermogravimetric analysis (TGA) on the stability of the sample. X-ray photoelectron spectroscopy (XPS) used ESCALAB250 to detect the surface composition of the sample.

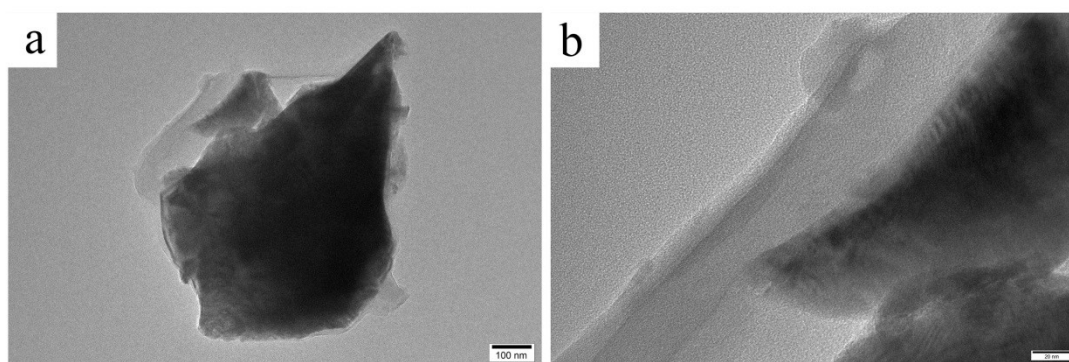


Figure S1 TEM images of CB@MoS₂ (a,b)

Table S1 Langmuir and Freundlich isotherm parameters

Models	Parameters	value
Langmuir	$Q_{m,cal}$ (mg/g)	990.10
	K_L (min ⁻¹)	0.2583
	R^2	0.9992
Freundlich	$\ln K_F$	6.0181
	n	5.4177
	R^2	0.9088

Table S2 Pseudo-first-order, pseudo-second-order and intraparticle diffusion model parameters

Models	Parameters	value
Pseudo-first-order equation	$Q_{e, \text{exp}} (mg/g)$	332.50
	$Q_{e, \text{cal}} (mg/g)$	295.75
	$k_1 (\text{min}^{-1})$	0.0659
	R^2	0.8281
Pseudo-second-order equation	$Q_{e, \text{cal}} (mg/g)$	341.30
	$k_2 (\text{g} \cdot \text{mg}^{-1} \cdot \text{min}^{-1})$	0.0006
	R^2	0.9991
Intraparticle diffusion	$k_{p1} (\text{mg g}^{-1} \text{min}^{-0.5})$	45.104
	C_1	80.156
	R^2_1	0.9762
	$K_{p2} (\text{mg g}^{-1} \text{min}^{-0.5})$	13.627
	C_2	200.34
	R^2_2	0.9630
	$K_{p3} (\text{mg g}^{-1} \text{min}^{-0.5})$	2.5264
C_3	302.03	
	R^2_3	0.5639

Table S3 Thermodynamic parameters at different temperatures

T(K)	298	308	318	328
$\Delta G(\text{kJ/mol})$	-6.69	-10.00	-13.32	-16.64

notes: $\Delta H^0 = 92.108 \text{ kJ/mol}$, $\Delta S^0 = 331.534 \text{ J/mol}$

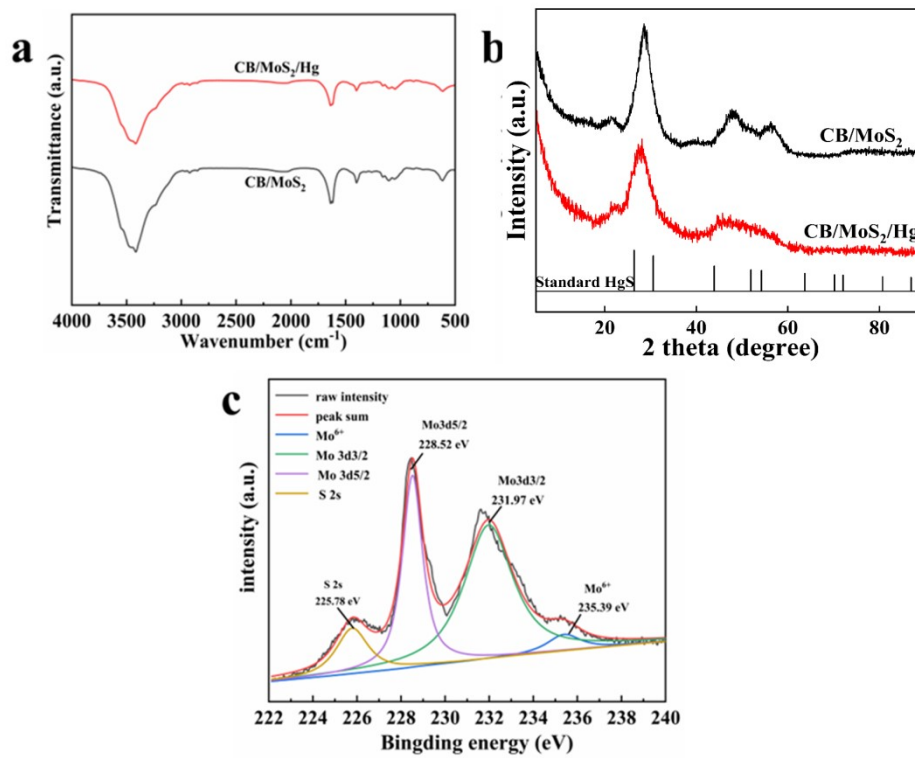


Figure S2 (a) FT-IR spectra of CB@MoS₂ and CB@MoS₂@HgS; (b) XRD patterns of CB@MoS₂ and CB@MoS₂@HgS; (c) High-resolution XPS spectrum of Mo3d