

## Supporting Information

### **Synthesis of novel hybrid mesoporous gold iron oxide nanoconstructs for enhanced catalytic reduction and remediation of toxic organic pollutants**

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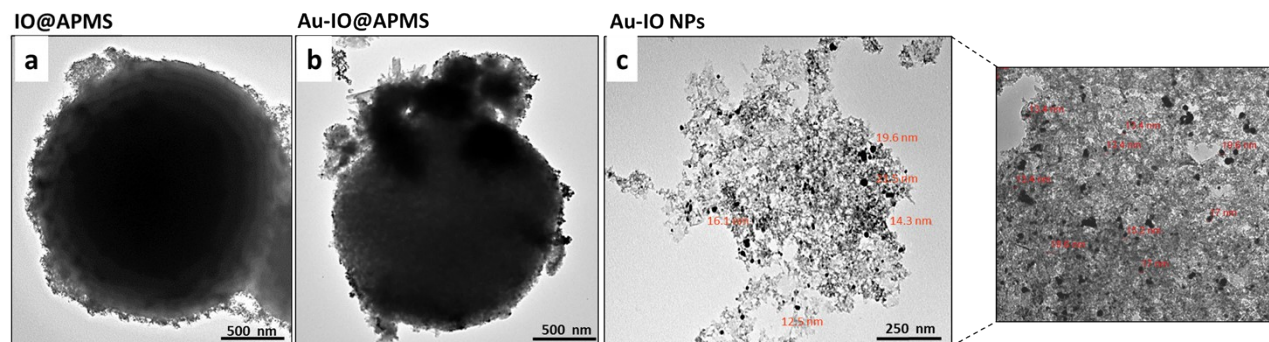
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## TEM



**Figure S1.** TEM images of the mesoporous Au-IO samples: a) IO@APMS-Amine, b) Au-grafted IO@APMS, and c) Au-IO NPs. The images clearly show the grafting of AuNPs (15-20 nm) onto IO@APMS-Amine and the successful etching to afford Au-IO mesostructures.

## EDX Elemental Analysis

### Au-IO@APMS

Element	Weight %	Atomic %
O	43.7	72.0
Si	11.7	11.1
Fe	32.1	15.2
Au	12.5	1.7

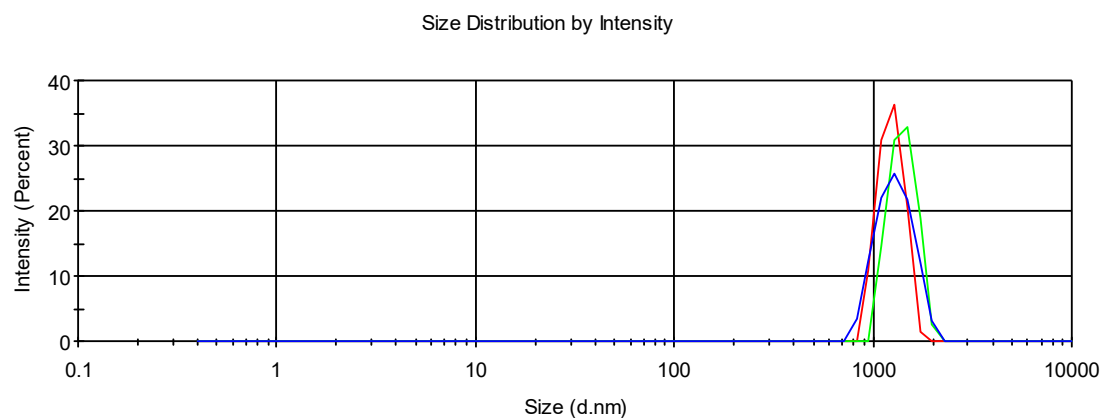
### Au-IO NPs

Element	Weight %	Atomic %
O	36.6	70.6
Si	0.5	0.5
Fe	48.3	26.6
Au	14.6	2.3

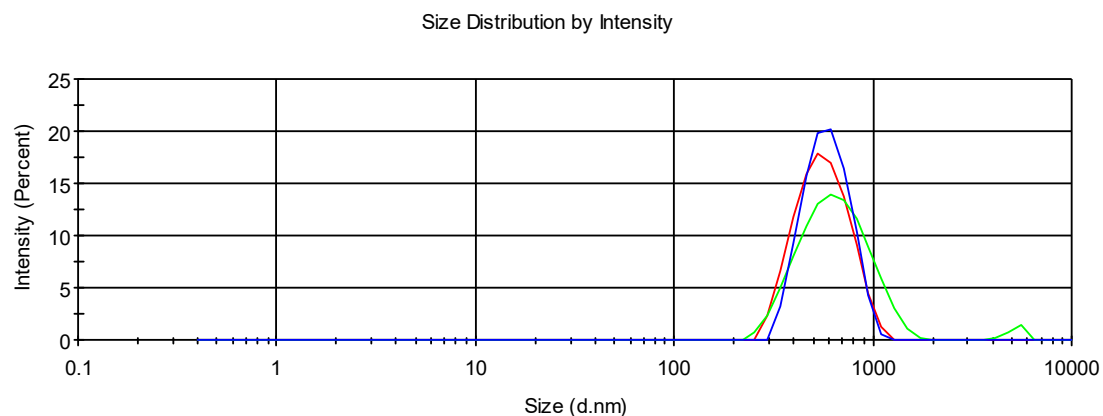
**Table S1.** EDX elemental analysis of the Au-incorporated IO samples selected from their respective SEM images.

## DLS and Zeta Potential

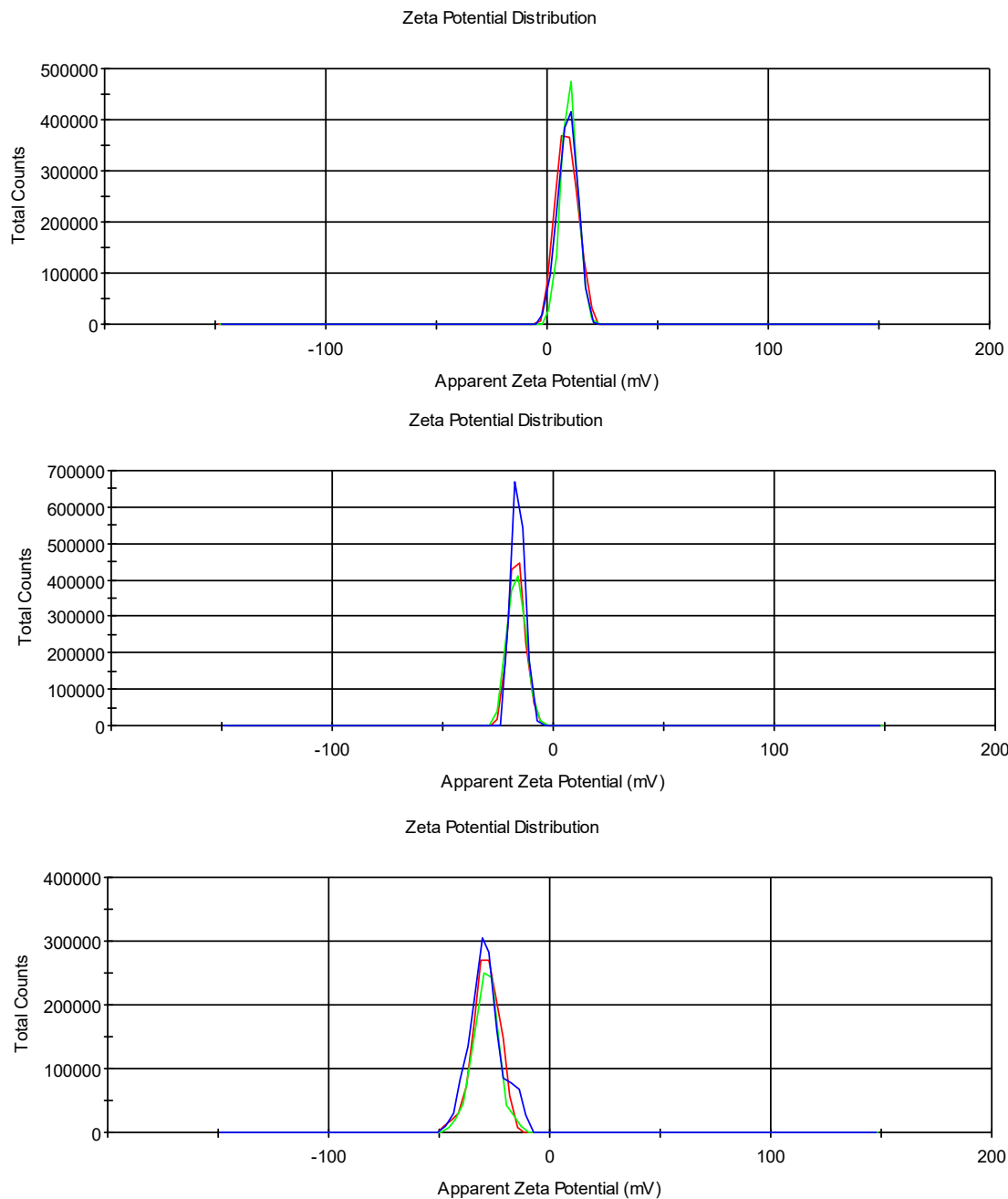
### IO@APMS-Amine & Au-IO@APMS



### Mesoporous Au-IO NPs

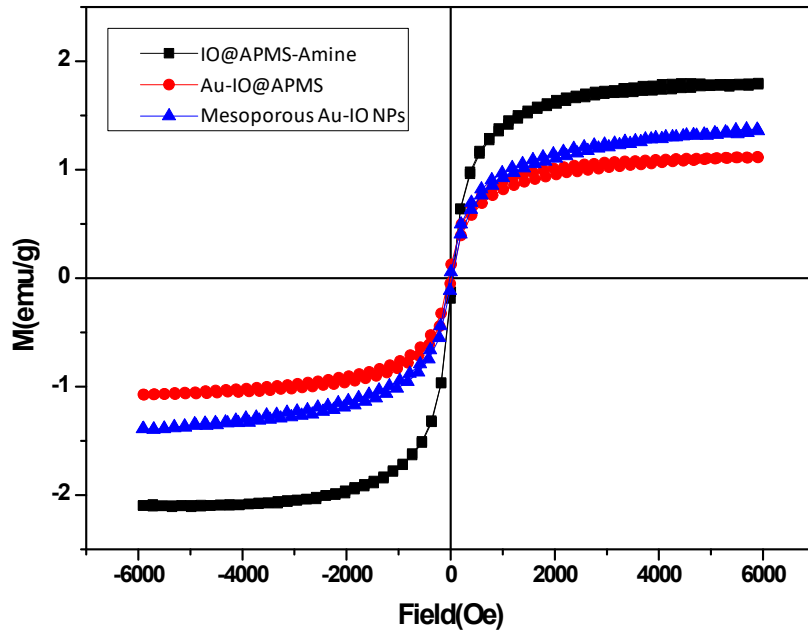


**Figure S2.** A) DLS measurements of the three different mesoporous IO@APMS-Amine, Au-IO@APMS, and Au-IO samples dispersed in aqueous media. IO@APMS-Amine ( $D_H = 1475 \pm 117$  nm), Au-IO@APMS ( $D_H = 1565 \pm 182$  nm), and Au-IO mesostructures ( $D_H = 710 \pm 55$  nm). Three independent measurements for three different concentrations were conducted reported as average means  $\pm$  Std. Dev.



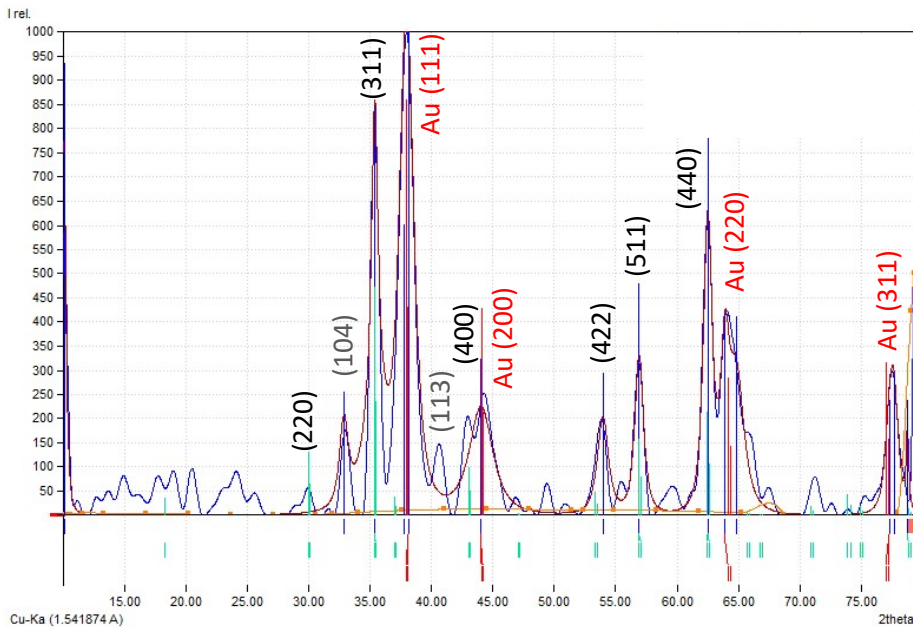
**Figure S2. B)** Zeta potential ( $\xi$ ) measurements of IO@APMS-Amine ( $\xi = +9.40 \pm 0.56$  mV), Au-IO@APMS ( $\xi = -16.8 \pm 0.50$  mV), and Au-IO mesostructures ( $-26.4 \pm 0.98$  mV). Three independent measurements for three different concentrations were conducted reported as average means  $\pm$  Std. Dev.

## Magnetic Properties

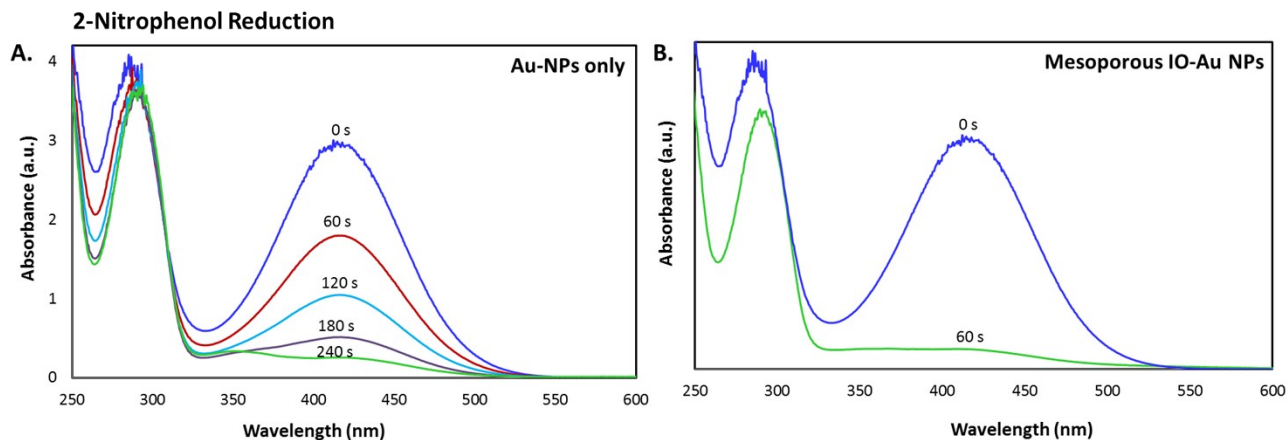


**Figure S3.** Field-dependent magnetization ( $M$ - $H$ ) hysteresis loops for the three different mesoporous samples clearly showing their superparamagnetic nature ( $M_s$  of IO@APMS-Amine = 1.85 emu/g, Au-IO@APMS = 1.09 emu/g, and mesoporous Au-IO = 1.36 emu/g).

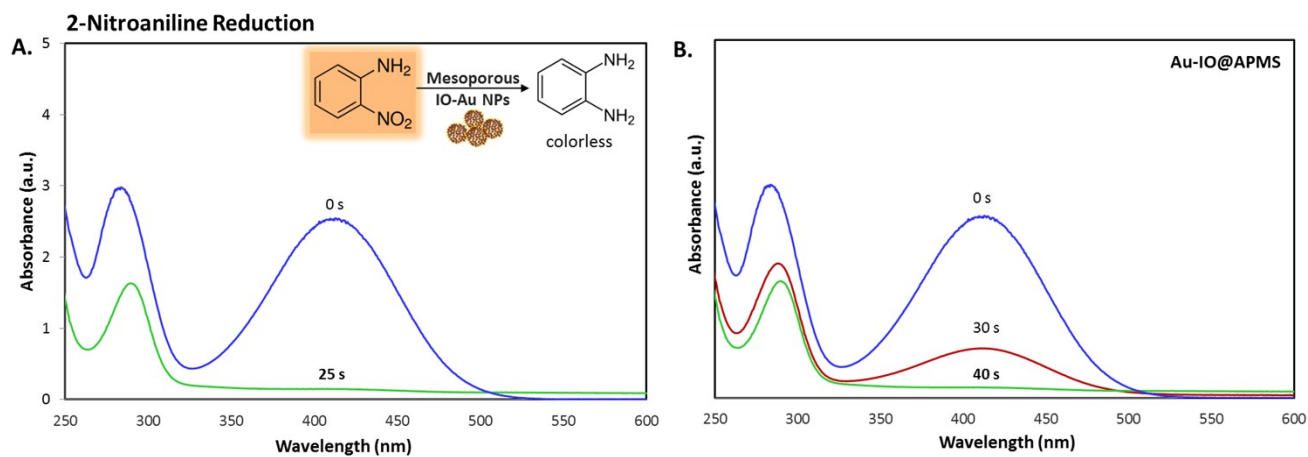
## XRD



**Figure S4.** X-ray diffraction patterns showing  $\gamma$ -Fe<sub>2</sub>O<sub>3</sub> maghemite (black) and Au phases (red).



**Figure S5.** Time progressive UV-vis absorption spectra of 2-nitrophenol reduction using same concentrations of Au-NPs only and mesoporous IO-Au NPs. Reaction conditions: 2.5 mM of nitrophenol solution, 25 mM of freshly prepared  $\text{NaBH}_4$  (excess), 1.5 mg of catalyst, total volume = 1 mL.



**Figure S6.** Time progressive UV-vis absorption spectra of 2-nitroaniline reduction using mesoporous nanocatalysts a) IO-Au NP and b) Au-IO@APMS. Reaction conditions: 2.5 mM of nitroaniline solution, 25 mM of freshly prepared  $\text{NaBH}_4$  (excess), 1.5 mg of catalyst, total volume = 1 mL.