

Support Information

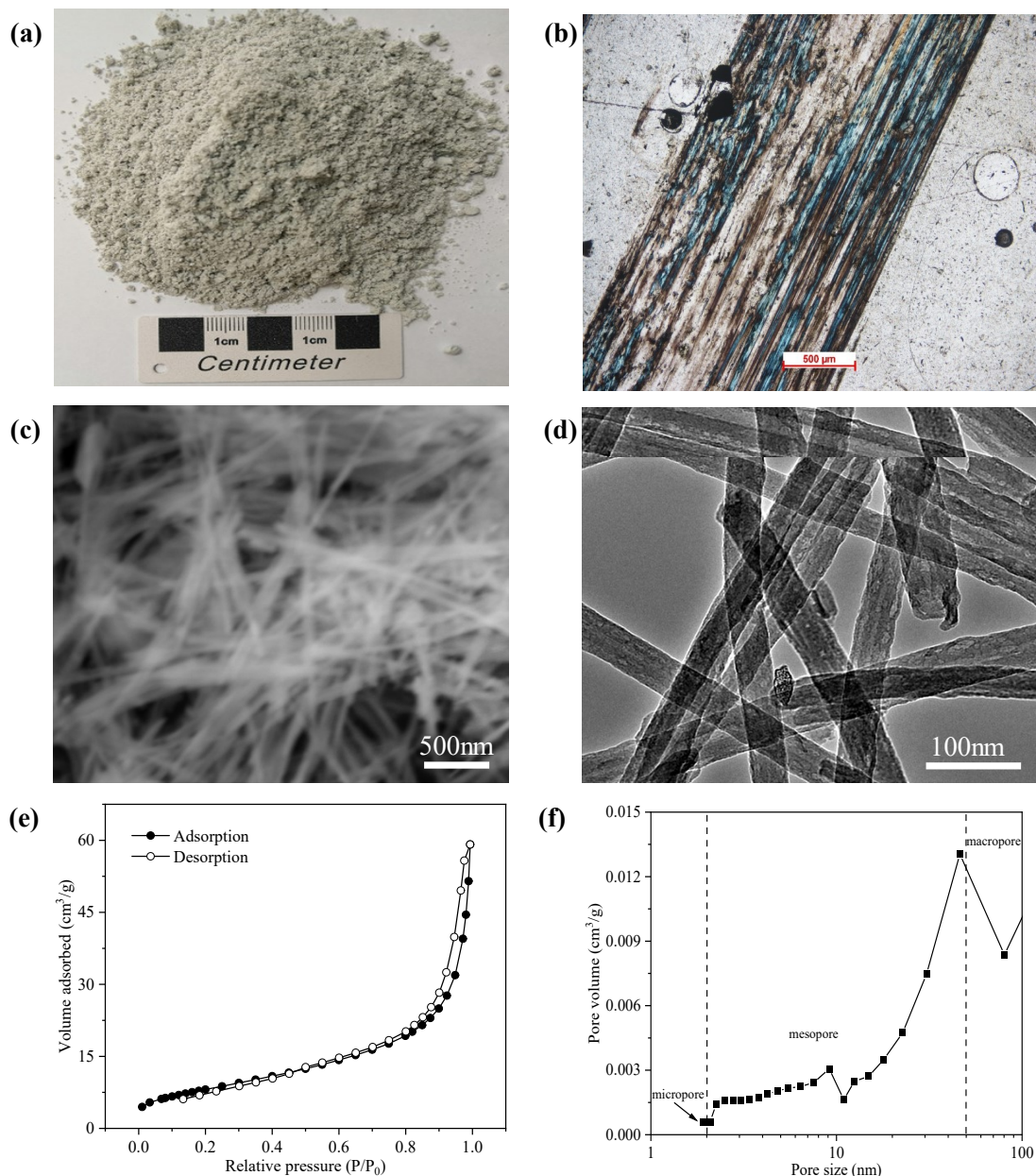
**Hierarchical Hollow Tubular Fibrous Brucite-Templated Carbons Obtained by KOH Activation for Supercapacitor**

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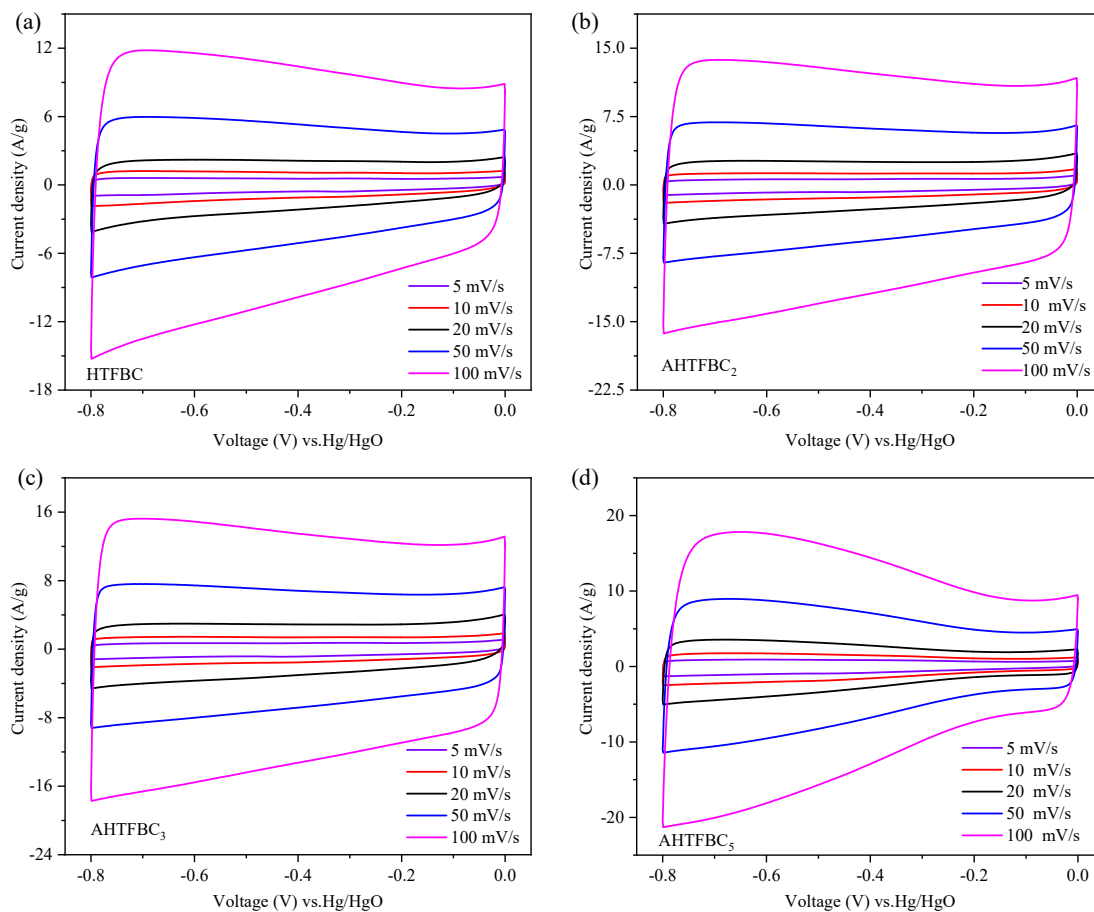
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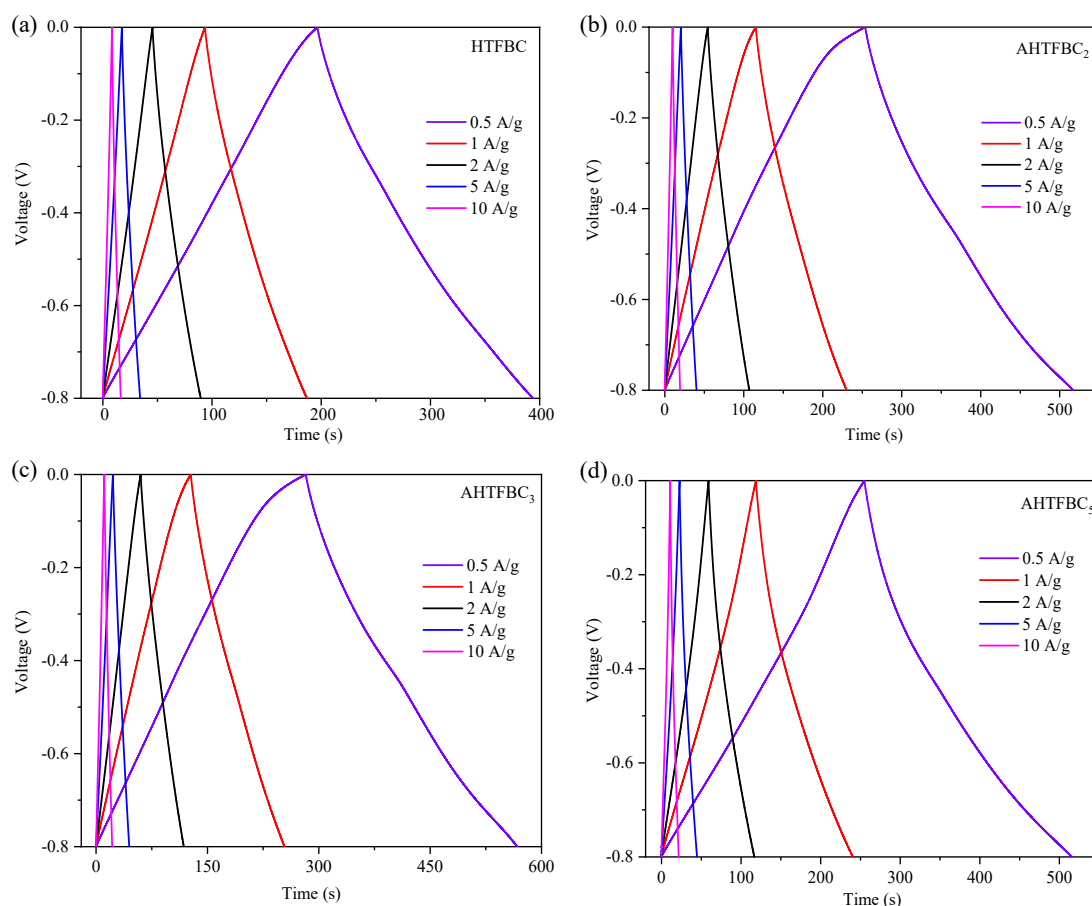
**Fig. S1** Mineralogical characteristics analysis of fibrous brucite used in experiments. (a) Fibrous brucite powder. (b) Photomineralogical features under cross-polarized. (c) SEM. (d) TEM micrograph. (e) N<sub>2</sub> adsorption-desorption isotherms and (f) DFT pore size distribution.

Table S1 Specific surface areas and pore texture of fibrous brucite

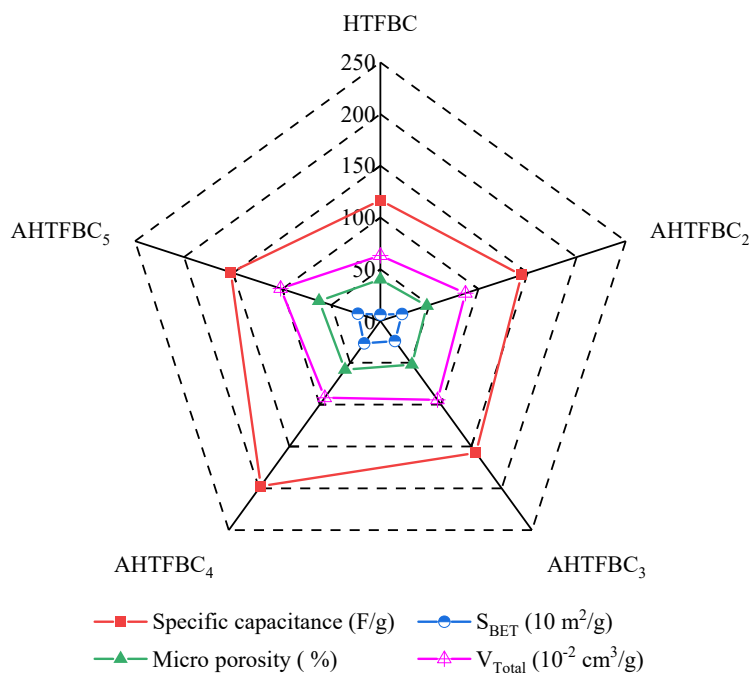
Sample	$D_{ap}$ (nm)	$S_{Total}$ ( $m^2/g$ )	$V_{Total}$ ( $cm^3/g$ )	$V_{micro}$ ( $cm^3/g$ )	$V_{meso}$ ( $cm^3/g$ )
FB	8.2	33	0.09	0.001	0.06



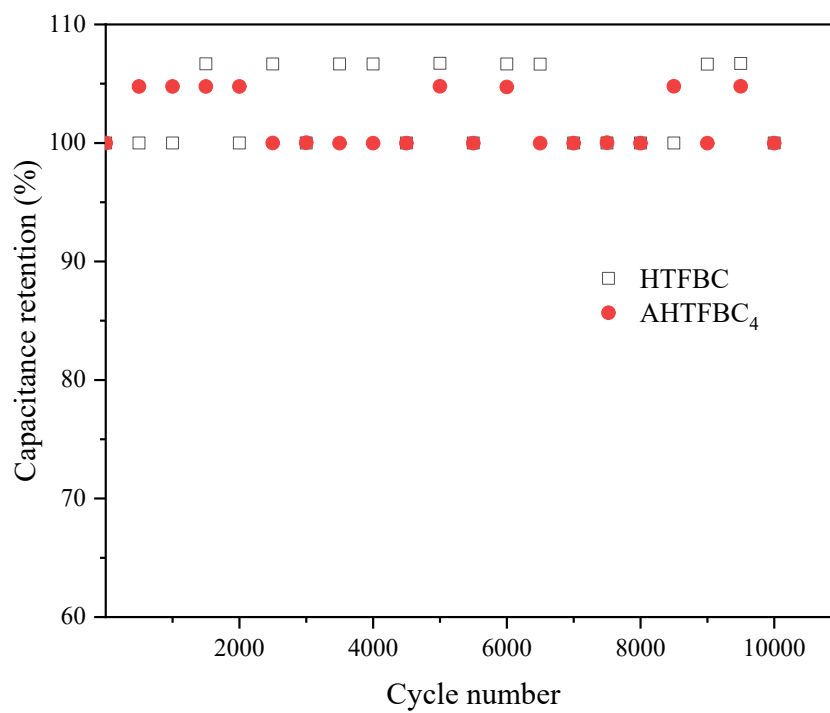
**Fig. S2** CV curves of the HTFBC and AHTFBCs measured in a three-electrode system in the 6 M KOH electrolyte. (a) HTFBC; (b) AHTFBC<sub>2</sub>; (c) AHTFBC<sub>3</sub>; (d) AHTFBC<sub>5</sub>



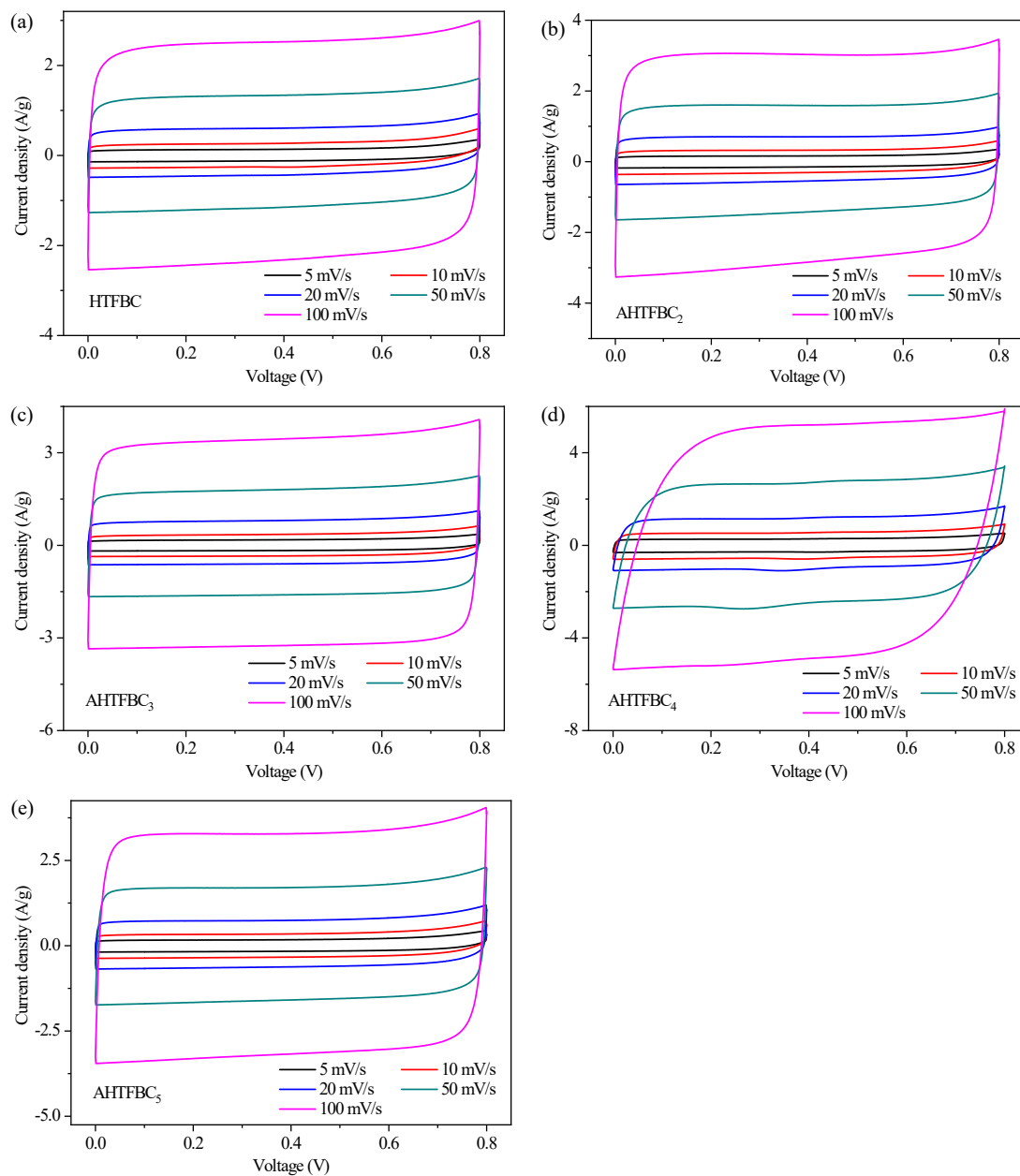
**Fig. S3** GCD curves of the HTFBC and AHTFBCs measured in a three-electrode system in the 6 M KOH electrolyte. (a) HTFBC; (b) AHTFBC<sub>2</sub>; (c) AHTFBC<sub>3</sub>; (d) AHTFBC<sub>5</sub>



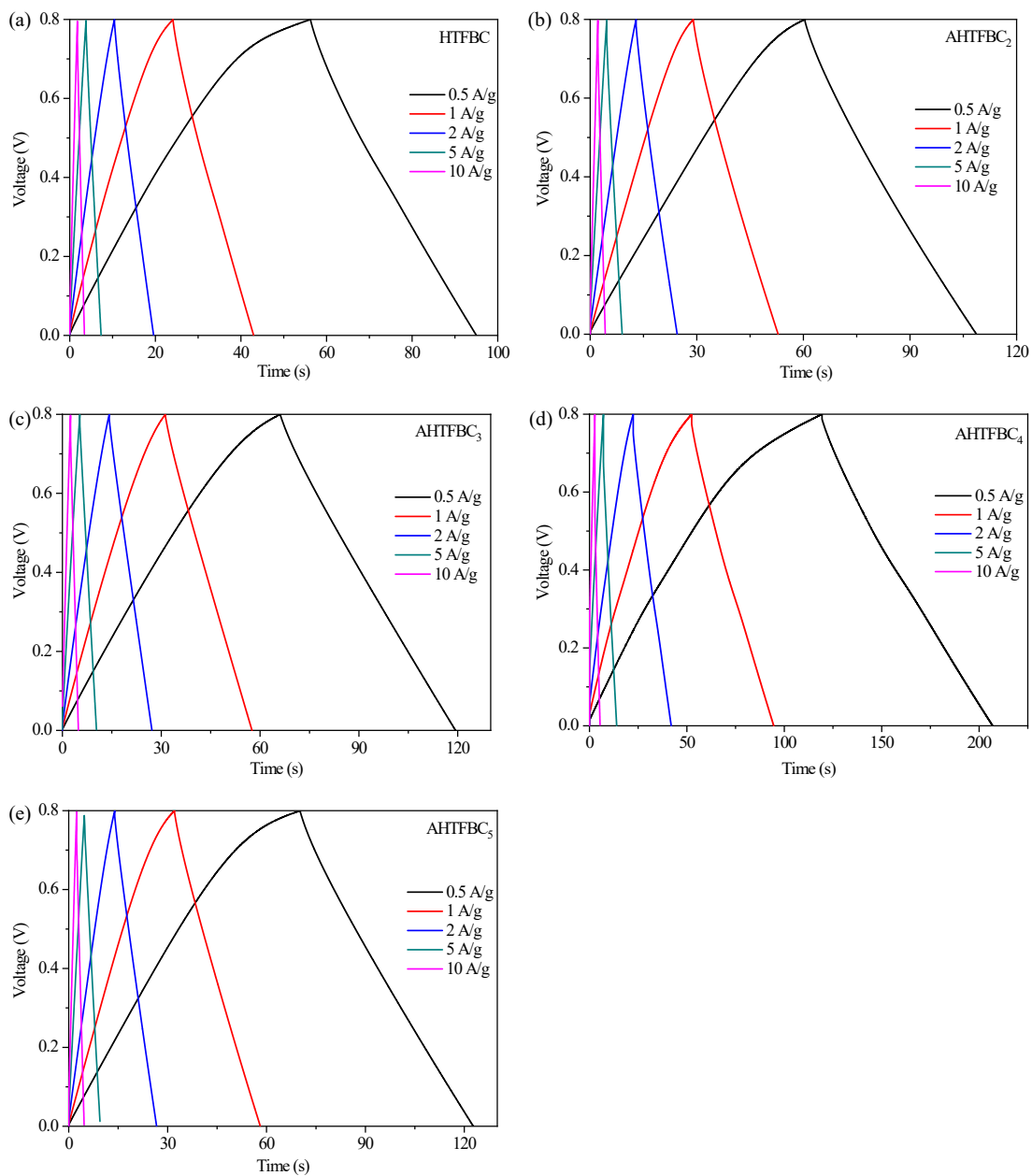
**Fig. S4** The cobweb map of specific capacitance, specific surface area, pore volume and micro porosity of HTFBC and AHTFBCs



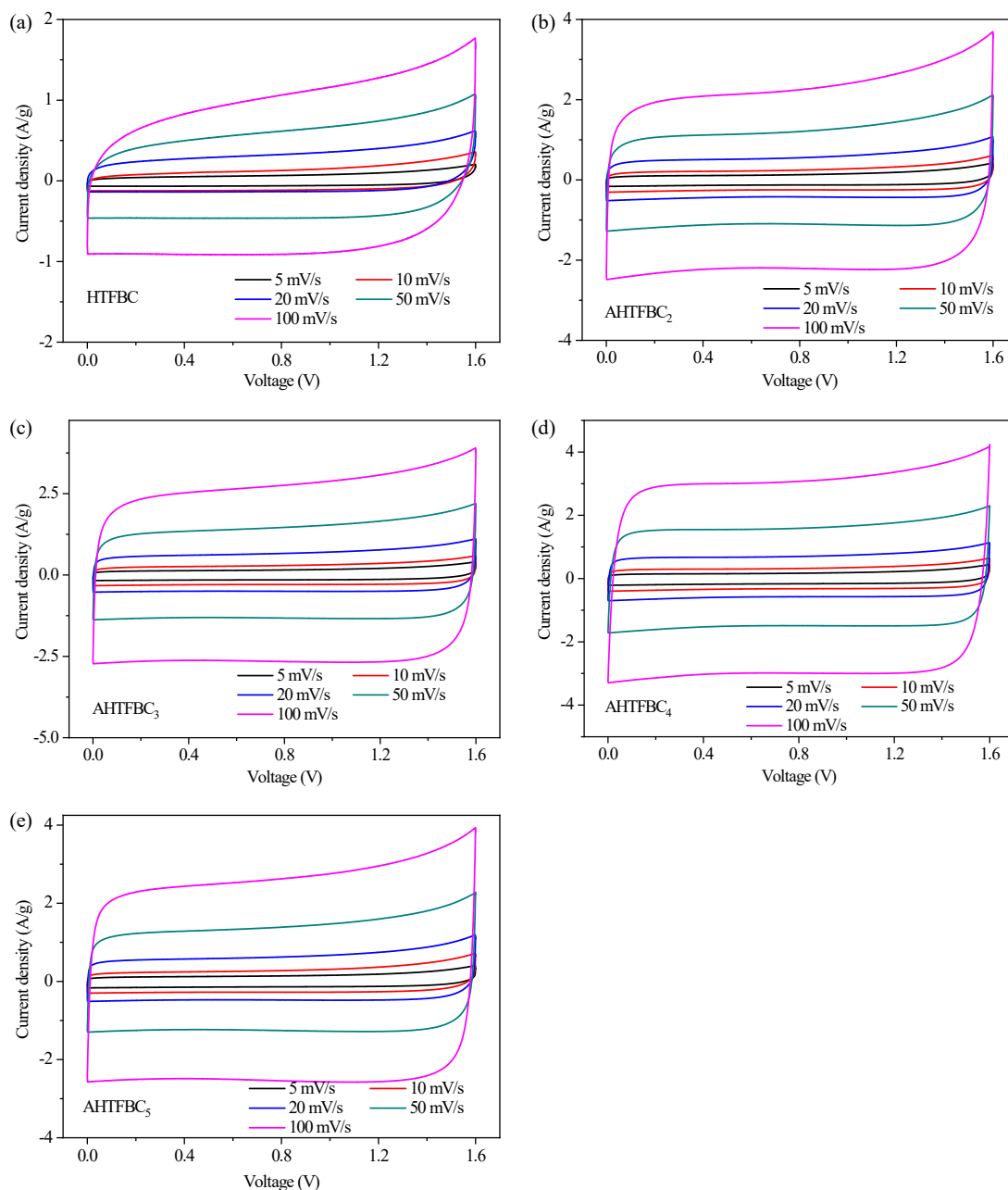
**Fig. S5** Cycling performance at 5 A/g of HTFBC and AHTFBC<sub>4</sub> measured in a three-electrode system in the 6 M KOH electrolyte



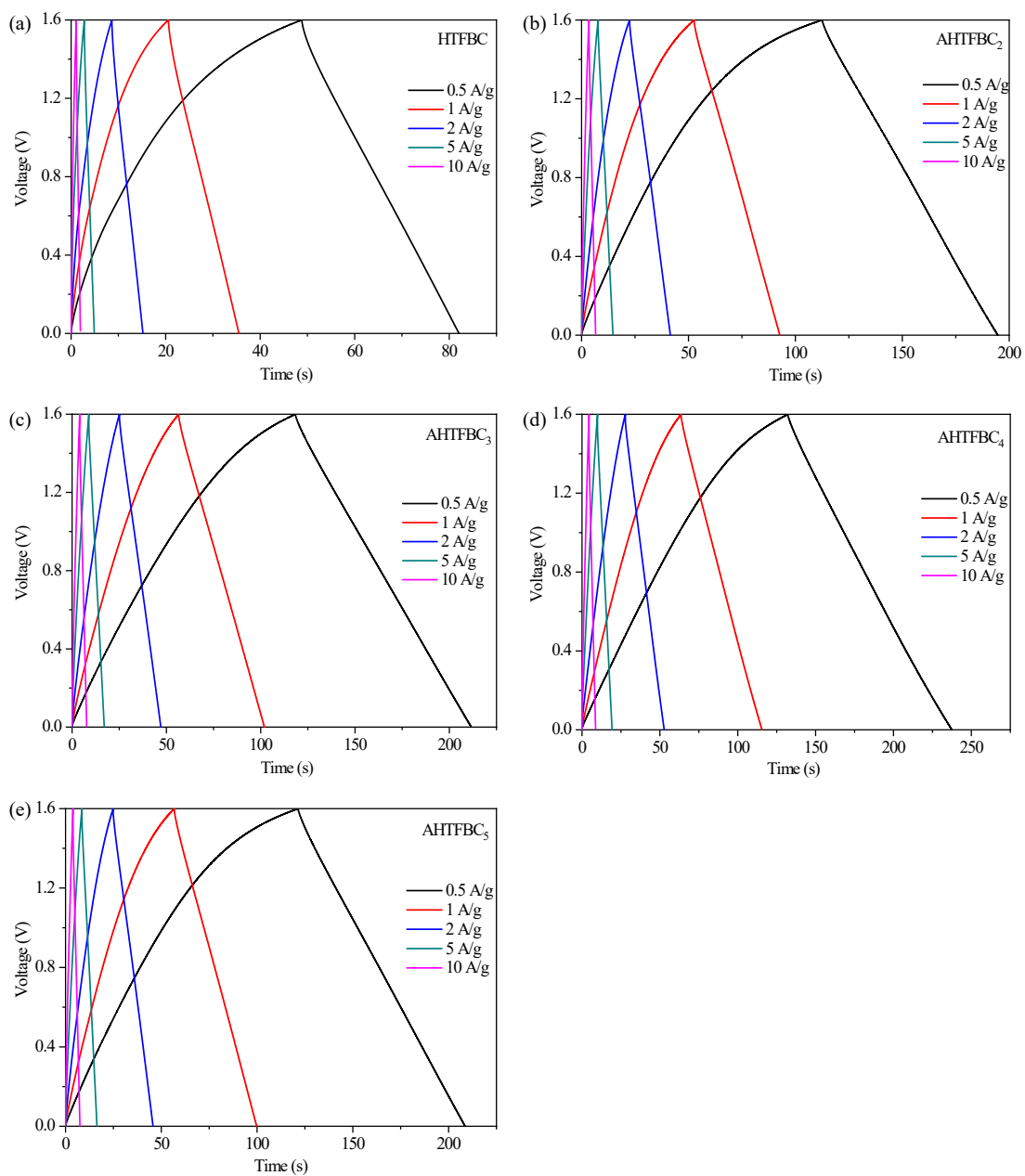
**Fig. S6** CV curves of the HTFBC and AHTFBCs measured in a two-electrode symmetrical system in the 6 M KOH electrolyte. (a) HTFBC; (b) AHTFBC<sub>2</sub>; (c) AHTFBC<sub>3</sub>; (d) AHTFBC<sub>4</sub>; (e) AHTFBC<sub>5</sub>



**Fig. S7** GCD curves of the HTFBC and AHTFBCs measured in a two-electrode symmetrical system in the 6 M KOH electrolyte. (a) HTFBC; (b) AHTFBC<sub>2</sub>; (c) AHTFBC<sub>3</sub>; (d) AHTFBC<sub>4</sub>; (e) AHTFBC<sub>5</sub>



**Fig. S8** CV curves of the HTFBC and AHTFBCs measured in a two-electrode symmetrical system in the 1 M  $\text{Na}_2\text{SO}_4$  electrolyte. (a) HTFBC; (b) AHTFBC<sub>2</sub>; (c) AHTFBC<sub>3</sub>; (d) AHTFBC<sub>4</sub>; (e) AHTFBC<sub>5</sub>



**Fig. S9** CV curves of the HTFBC and AHTFBCs measured in a two-electrode symmetrical system in the 1 M  $\text{Na}_2\text{SO}_4$  electrolyte. (a) HTFBC; (b) AHTFBC<sub>2</sub>; (c) AHTFBC<sub>3</sub>; (d) AHTFBC<sub>4</sub>; (e) AHTFBC<sub>5</sub>