

Supplementary Data

Immobilization of Ni (II) complex on the surface of mesoporous modified-KIT-6 as a new, reusable and highly efficient nanocatalyst for the synthesis of tetrazole and pyranopyrazole derivatives

Mitra Darabi, Mohsen Nikoorazm, Bahman Tahmasbi, Arash Ghorbani-Choghamarani

^aDepartment of Chemistry, Faculty of Science, Ilam University, P. O. Box 69315516, Ilam, Iran.

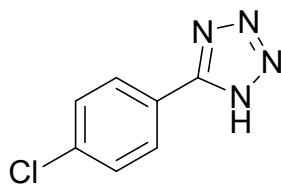
**E-mail: e_nikoorazm@yahoo.com*

^bDepartment of Organic Chemistry, Faculty of Chemistry, Bu-Ali Sina University, Hamedan, 6517838683, Iran.

ABSTRACT: In this paper, KIT-6@SMTU@Ni was successfully synthesized *via* a new Ni (II) complex stabilization on the modified mesoporous KIT-6, as a novel and green heterogeneous catalyst. The obtained catalyst (KIT-6@SMTU@Ni) was characterized using Fourier transform infrared spectroscopy (FT-IR), Brunauer- Emmett- Teller (BET), X-Ray diffraction (XRD), atomic absorption spectroscopy (AAS), energy-dispersive X-ray spectroscopy (EDS), X-ray mapping, thermogravimetric analysis (TGA) techniques and scanning electron microscopy (SEM). After complete characterization of the catalyst, it was successfully used for the synthesis of 5-substituted 1*H*-tetrazoles and pyranopyrazoles. Moreover, tetrazoles were synthesized from benzonitrile derivatives and sodium azide (NaN₃). All tetrazole products were synthesized with high TON, TOF and excellent yields (88-98 %) and reasonable time (0.13-8.08 h), which shows the efficiency and practicality of KIT-6@SMTU@Ni catalyst. Furthermore, pyranopyrazoles were prepared through the condensation reaction of benzaldehyde derivatives with malononitrile, hydrazine hydrate and ethyl acetoacetate with high TON, TOF and excellent yields (87-98 %) and appropriate time (2-10.5 h). KIT-6@SMTU@Ni catalyst can be reused again for five runs without any re-activation. Significantly, this protocol has prominent benefits such as applying green solvents, use of commercially available and low-cost materials, excellent separation and reusability of the catalyst, short reaction time, high yield of products and facile work-up.

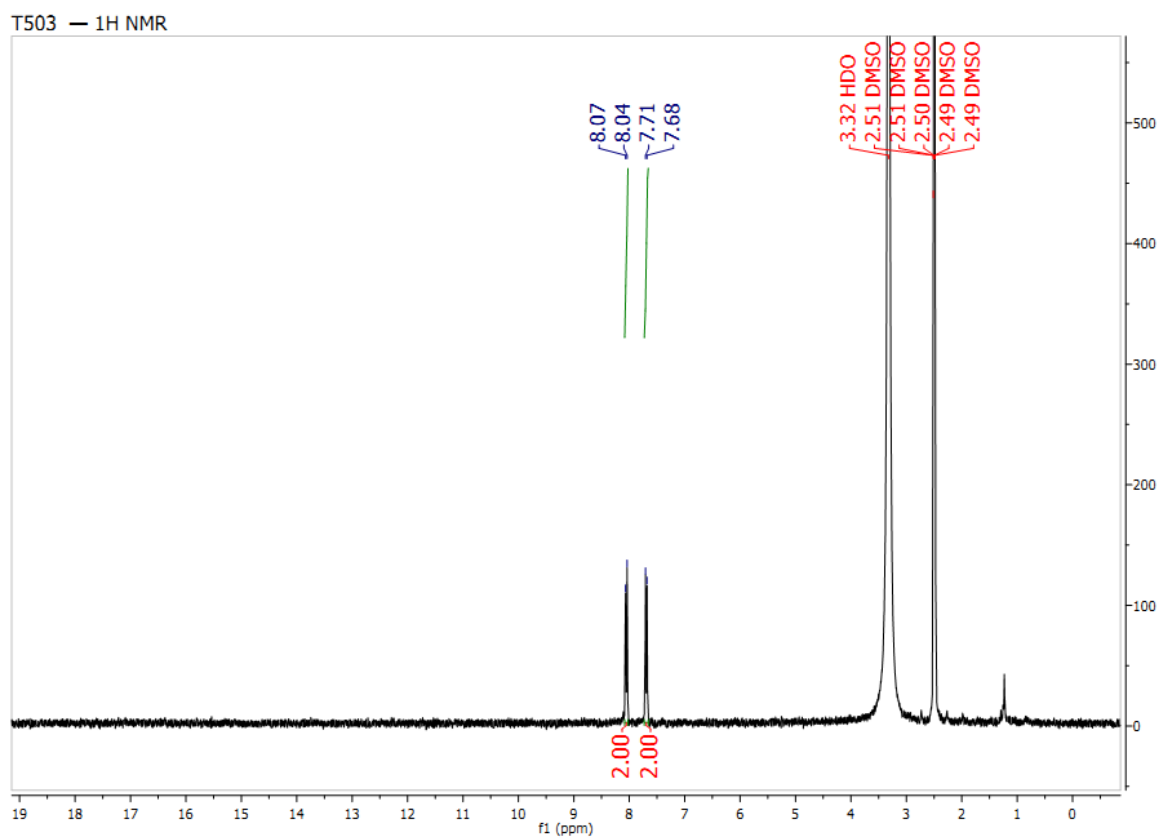
Keywords: mesoporous silica, KIT-6, nanocatalyst, tetrazoles, pyranopyrazoles.

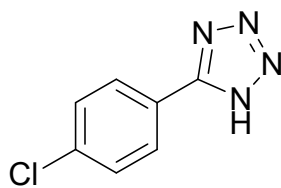
¹H NMR spectral data



5-(4-chlorophenyl)-1H-tetrazole

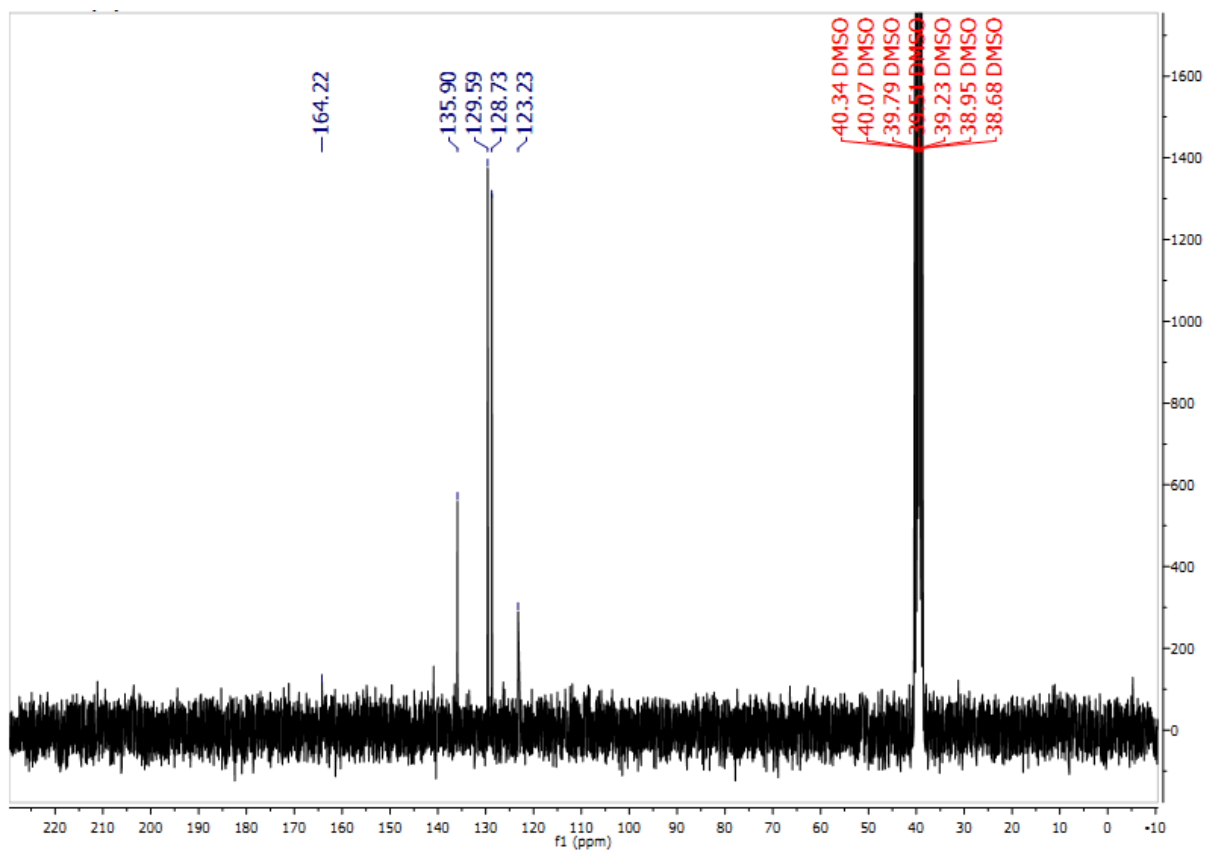
¹H NMR (400 MHz, DMSO_{d6}): δ_H = 8.07-8.04 (d, *J* = 12 Hz, 2H), 7.71-7.68 (d, *J* = 12 Hz, 2H) ppm.

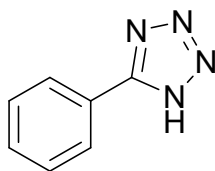




5-(4-chlorophenyl)-1H-tetrazole

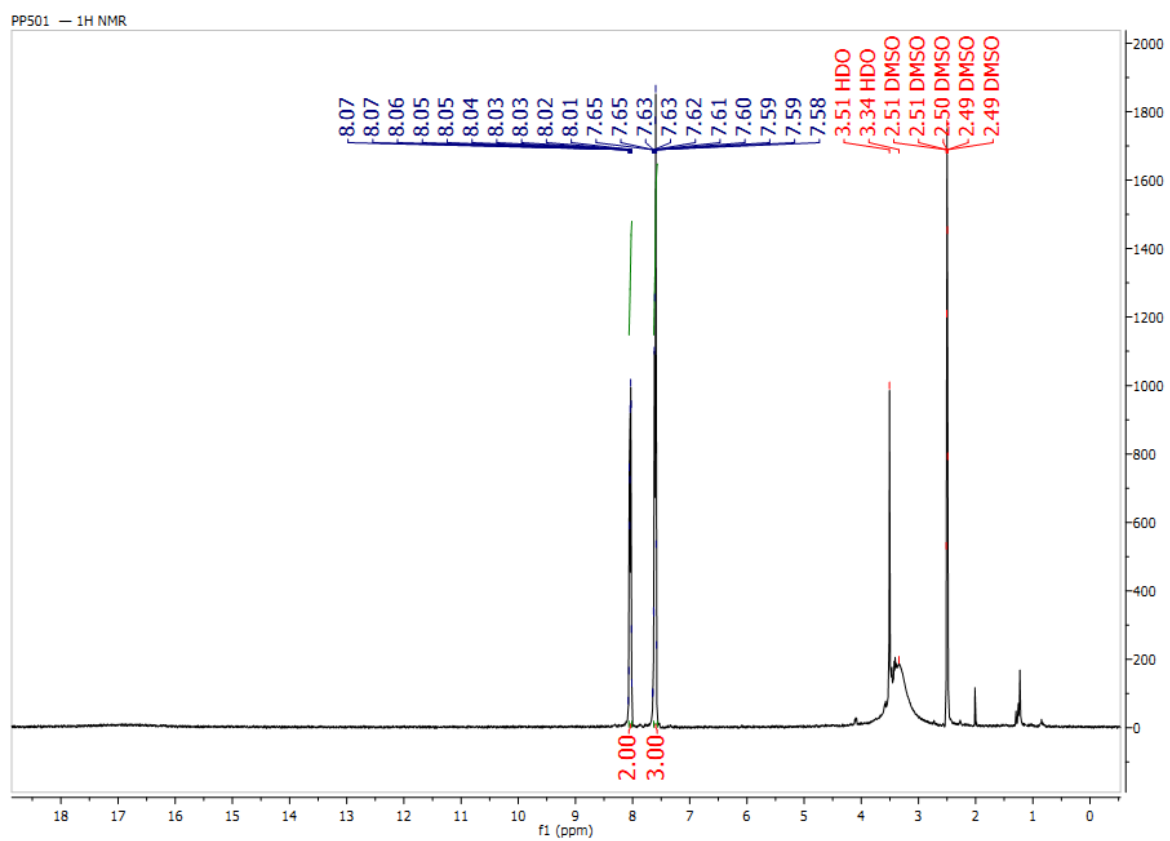
^{13}C NMR (100 MHz, DMSO-d_6): $\delta_{\text{H}} = 164.2, 135.9, 129.6, 128.7, 123.2$ ppm.

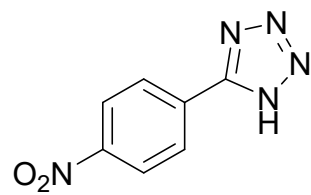




5-phenyl-1H-tetrazole

$^1\text{H NMR}$ (400 MHz, DMSO-d_6): $\delta_{\text{H}} = 8.07\text{-}8.01$ (m, 2H), $7.65\text{-}7.58$ (m, 3H) ppm.

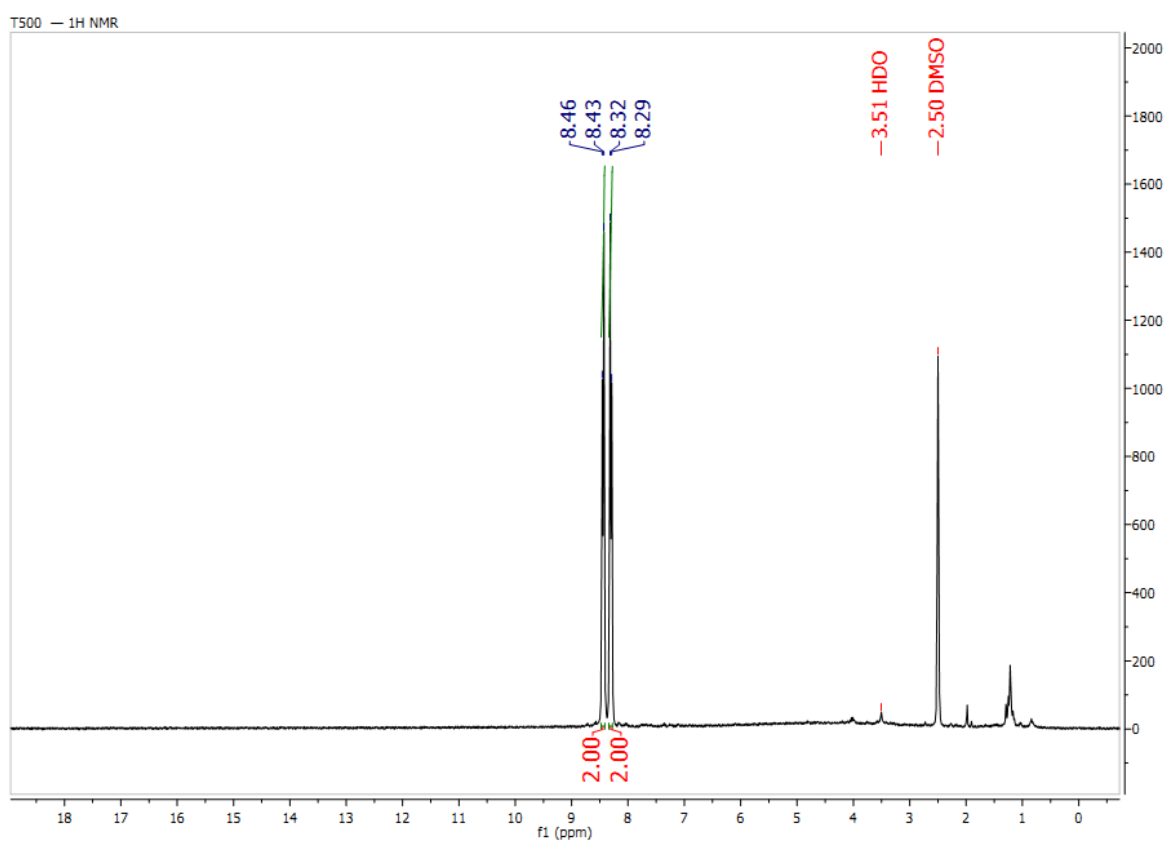


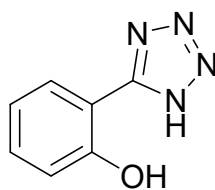


5-(4-nitrophenyl)-1H-tetrazole

$^1\text{H NMR}$ (400 MHz, DMSO-d_6): $\delta_{\text{H}} = 8.46-8.43$ (d, $J = 12$ Hz, 2H), $8.32-8.29$ (d, $J = 12$ Hz, 2H)

ppm.

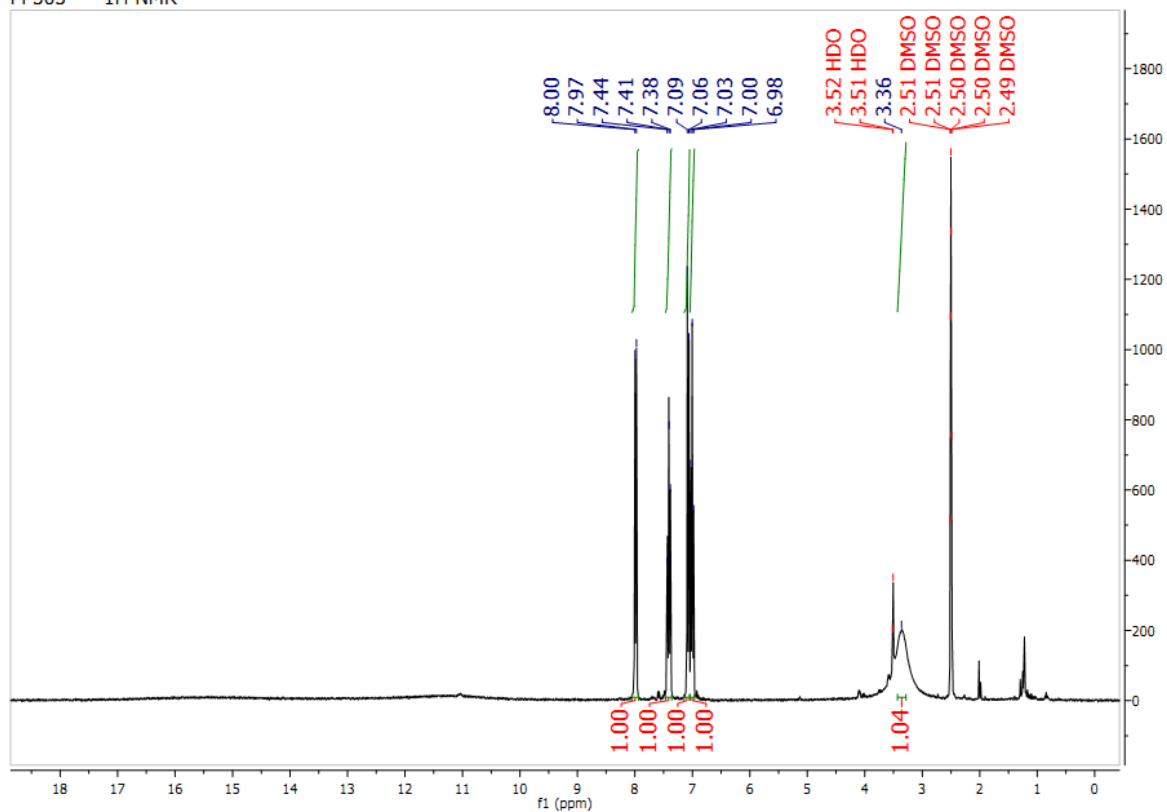


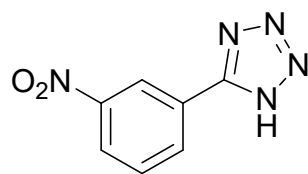


2-(1H-tetrazol-5-yl)phenol

^1H NMR (400 MHz, DMSO-d_6): $\delta_{\text{H}} = 8.00-7.97$ (d, $J = 12$ Hz, 1H), $7.44-7.38$ (t, $J = 12$ Hz, 1H), $7.09-7.06$ (d, $J = 12$ Hz, 1H), $7.03-6.98$ (t, $J = 12$ Hz, 1H), 3.36 (br, 1H) ppm.

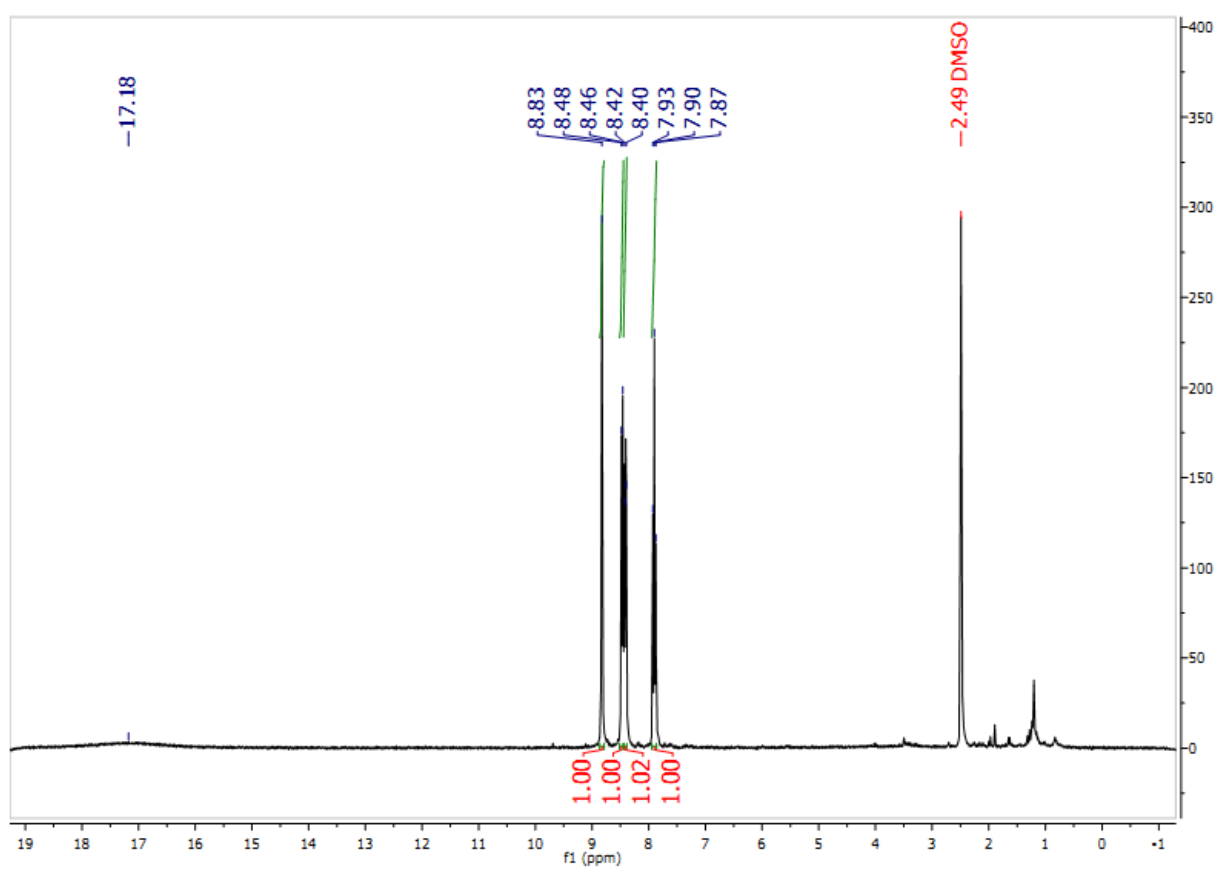
PP505 — ^1H NMR

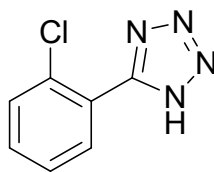




5-(3-nitrophenyl)-1H-tetrazole

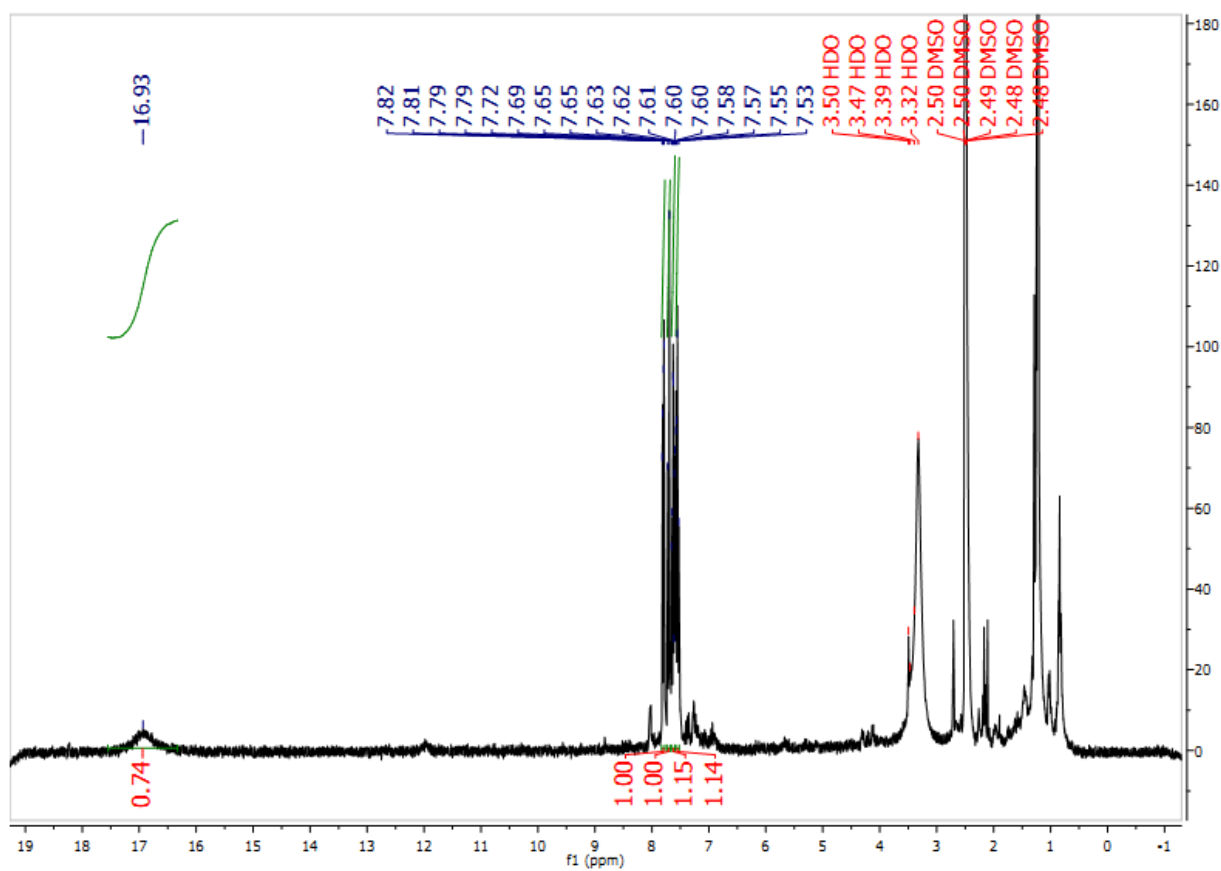
^1H NMR (400 MHz, DMSO-d_6): δ_{H} = 8.83 (s, 1H), 8.48-8.46 (d, J = 8 Hz, 1H), 8.42-8.40 (d, J = 8 Hz, 1H), 7.93-6.87 (t, J = 12 Hz, 1H) ppm.

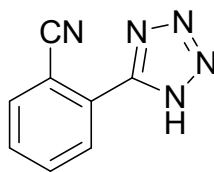




5-(2-chlorophenyl)-1H-tetrazole

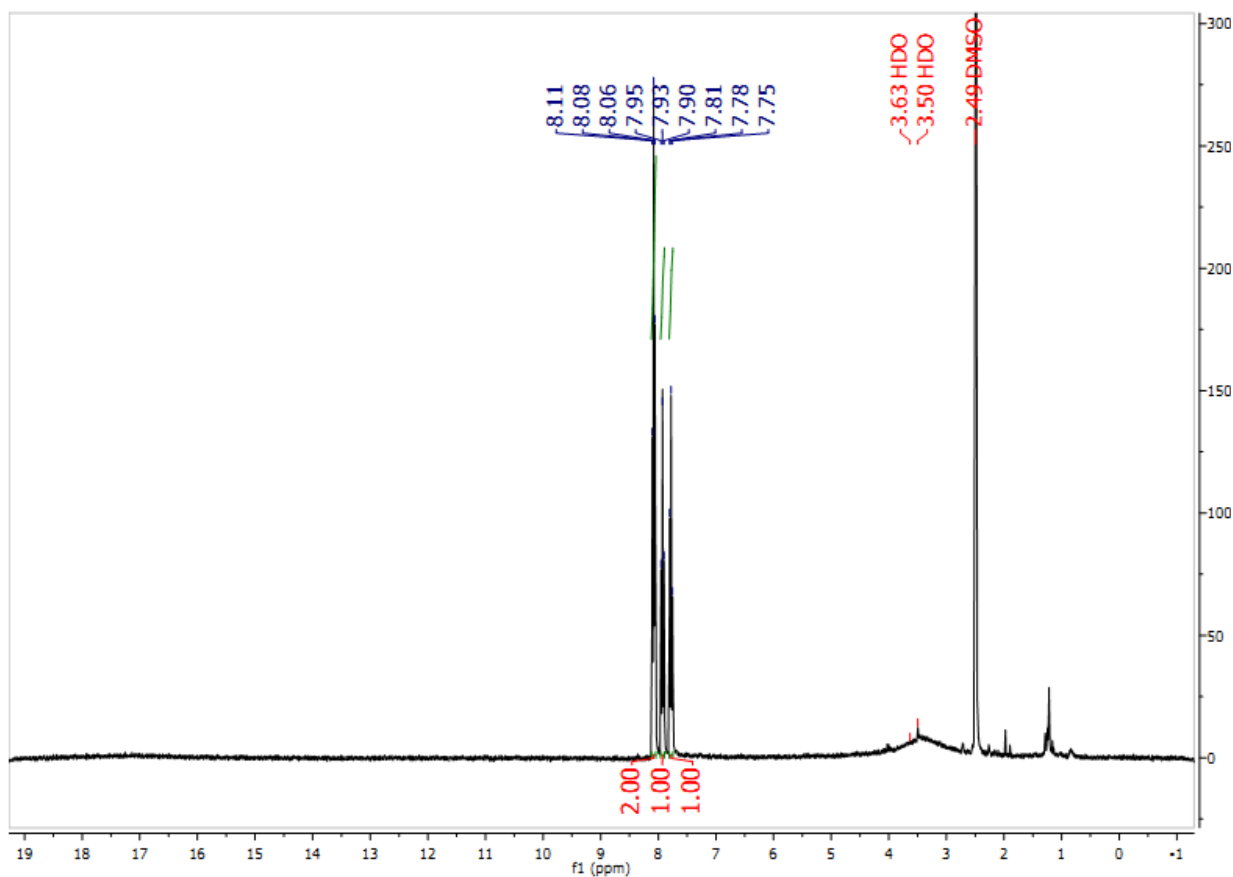
^1H NMR (400 MHz, DMSO-d_6): $\delta_{\text{H}} = 16.93$ (br, 1H), 7.82-7.79 (d of d, $J = 8$ Hz, 1H), 7.72-7.69 (d, $J = 8$ Hz, 1H), 7.65-7.60 (t of d, $J = 8$ Hz, 1H), 7.58-7.53 (t of d, $J = 8$ Hz, 1H) ppm.

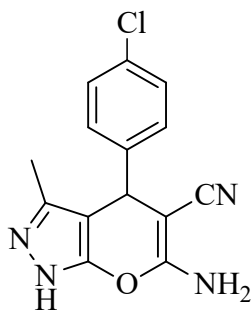




2-(1H-tetrazol-5-yl)benzonitrile

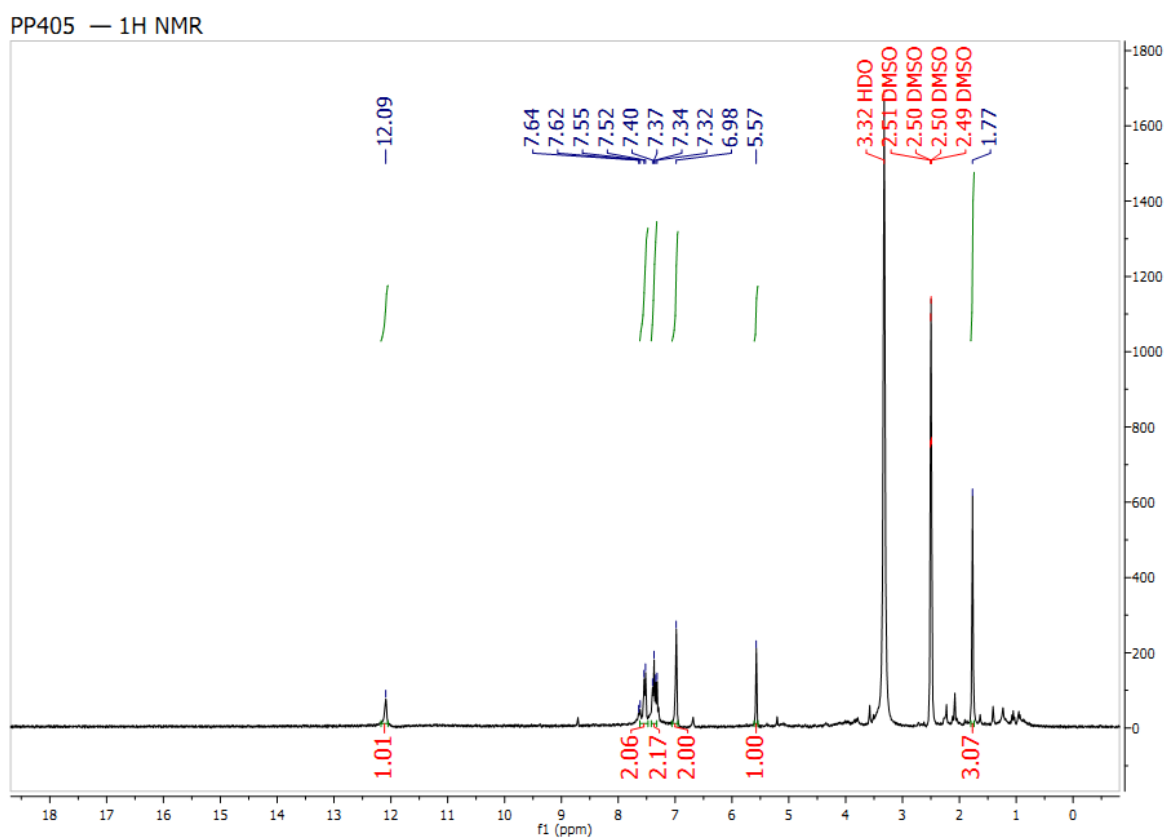
$^1\text{H NMR}$ (400 MHz, DMSO-d_6): $\delta_{\text{H}} = 8.11\text{-}8.06$ (t, $J = 8$ Hz, 2H), $7.95\text{-}7.90$ (t, $J = 8$ Hz, 1H), $7.81\text{-}7.75$ (t, $J = 8$ Hz, 1H) ppm.

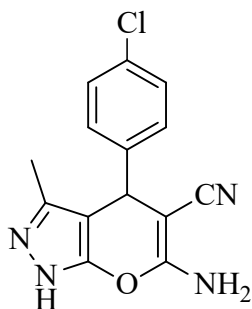




6-amino-4-(4-chlorophenyl)-3-methyl-1,4-dihydropyranopyrazole-5-carbonitrile

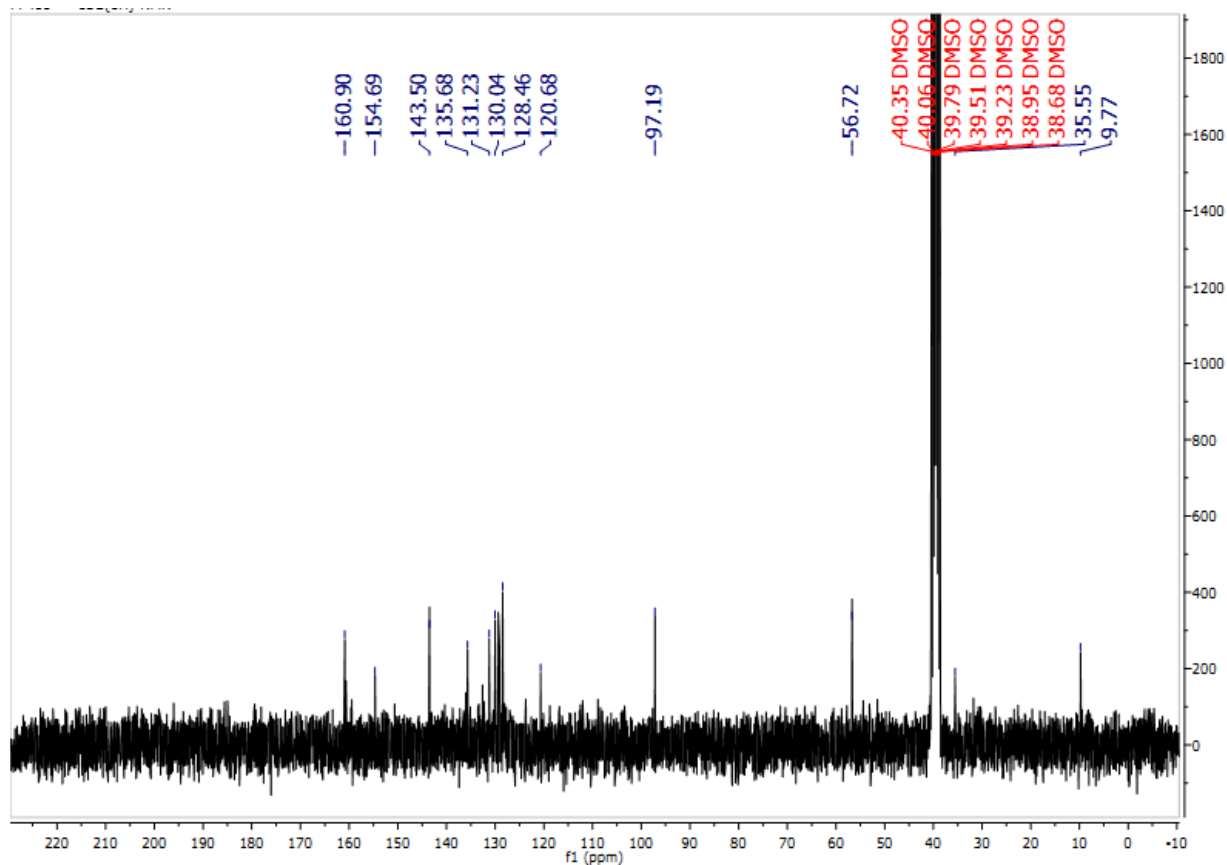
^1H NMR (400 MHz, DMSO-d_6): $\delta_{\text{H}} = 12.09$ (s, 1H), 7.64-7.52 (m, 2H), 7.40-7.32 (m, 2H), 6.98 (br, 2H), 5.57 (s, 1H), 1.77 (s, 3H) ppm.

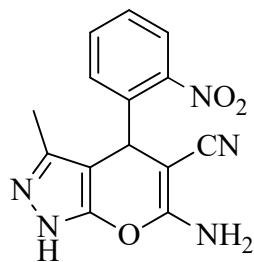




6-amino-4-(4-chlorophenyl)-3-methyl-1,4-dihydropyrano[2,3-c]pyrazole-5-carbonitrile

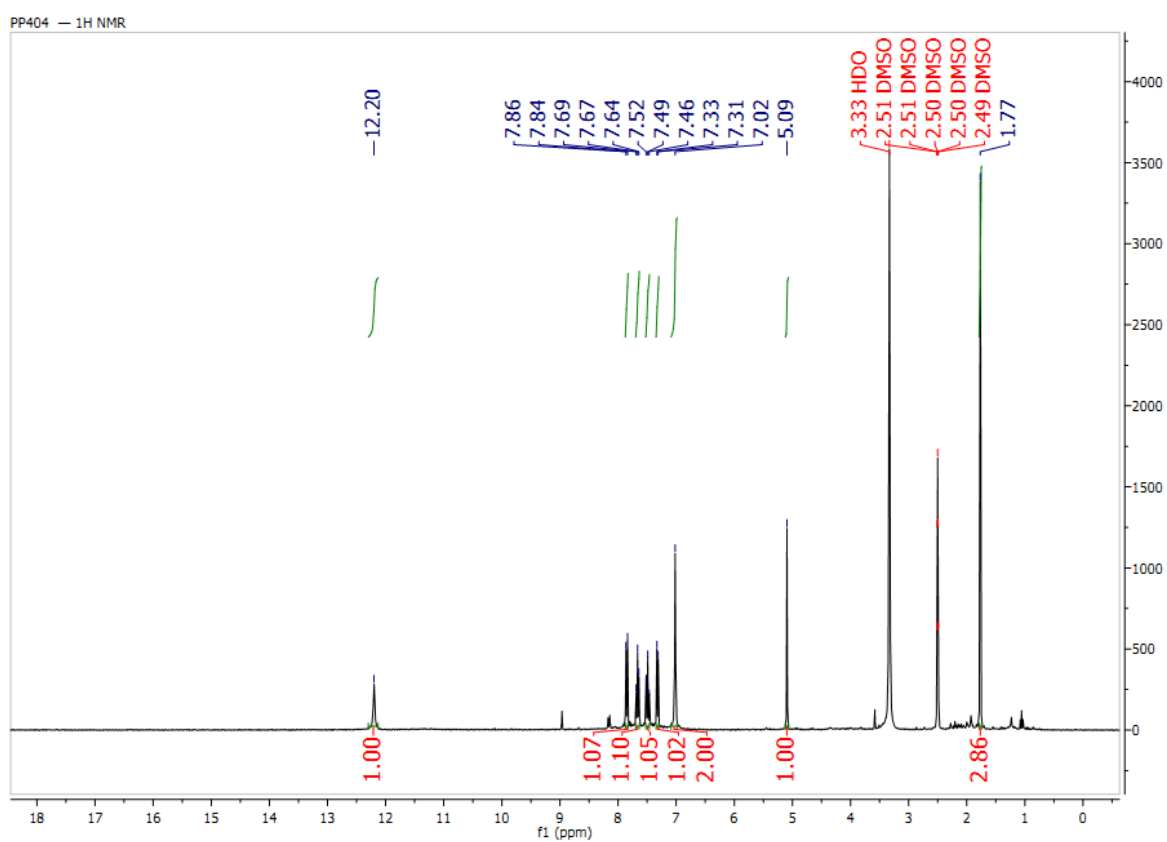
^{13}C NMR (100 MHz, DMSO-d_6): δ_{H} = 160.9, 154.7, 143.5, 135.7, 131.2, 130.0, 128.5, 120.7, 97.2, 56.7, 35.5, 9.7 ppm.

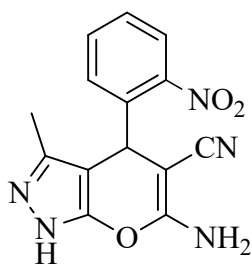




6-amino-3-methyl-4-(2-nitrophenyl)-1,4-dihydropyranopyrazole-5-carbonitrile

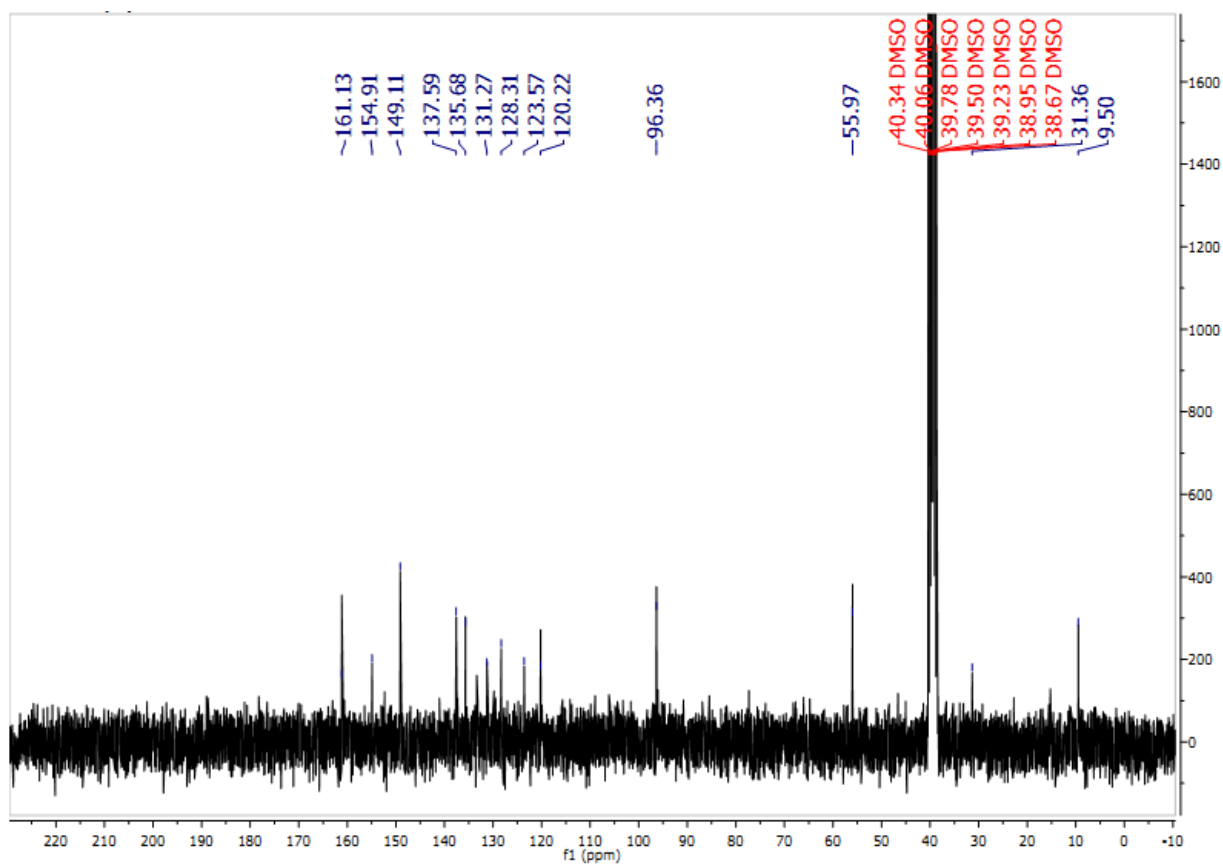
^1H NMR (400 MHz, DMSO-d_6): δ_{H} = 12.20 (s, 1H), 7.86-7.84 (d, J = 8 Hz, 1H), 7.69-7.64 (t, J = 12 Hz, 1H), 7.52-7.46 (t, J = 12 Hz, 1H), 7.33-7.31 (d, J = 8 Hz, 1H), 7.02 (s, 2H), 5.09 (s, 1H), 1.77 (s, 3H) ppm.

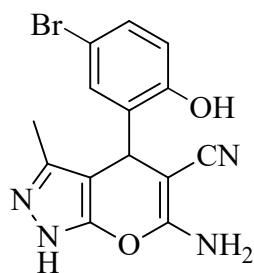




6-amino-3-methyl-4-(2-nitrophenyl)-1,4-dihydropyranopyrazole-5-carbonitrile

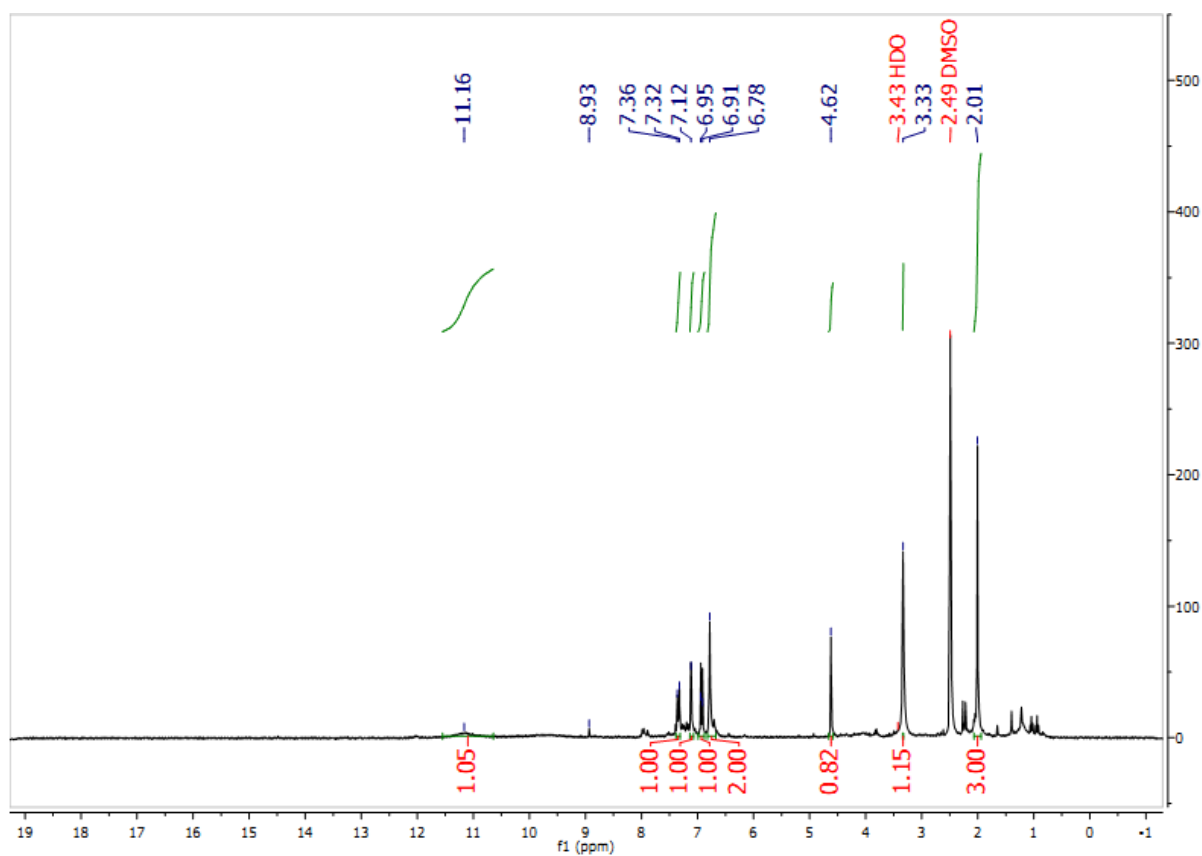
^{13}C NMR (100 MHz, DMSO-d_6): $\delta_{\text{H}} = 161.1, 154.9, 149.1, 137.6, 135.7, 131.3, 128.3, 123.6, 120.2, 96.4, 55.9, 31.4, 90.5$ ppm.

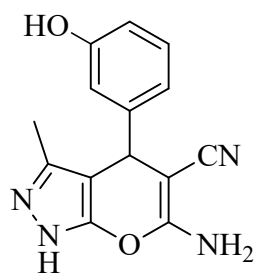




6-amino-4-(5-bromo-2-hydroxyphenyl)-3-methyl-1,4-dihydropyrano[2,3-c]pyrazole-5-carbonitrile

^1H NMR (400 MHz, DMSO-d_6): $\delta_{\text{H}} = 11.16$ (br, 1H), 8.93 (s, 1H), 7.36-7.32 (d, $J = 16$ Hz, 1H), 7.12 (s, 1H), 6.95-6.91 (d, $J = 16$ Hz, 1H), 6.78 (br, 2H), 4.62 (s, 1H), 3.33 (s, 1H), 2.01 (s, 3H) ppm.





6-amino-4-(3-hydroxyphenyl)-3-methyl-1,4-dihydropyranopyrazole-5-carbonitrile

$^1\text{H NMR}$ (400 MHz, DMSO-d_6): $\delta_{\text{H}} = 12.09$ (br, 1H), 9.32 (s, 1H), 7.11-7.06 (d, $J = 8$ Hz, 1H), 6.86 (br, 2H), 6.62-6.58 (m, 2H), 6.52 (s, 1H), 4.47 (s, 1H), 3.36 (br, 1H), 1.81 (s, 3H) ppm.

