

Supplementary Information

In situ FTIR spectroscopy study to reveal Ga modified ZSM-5 for boosting isobutene amination

Shiqing Wang, Xiaocheng Lan, Shiyong Xing, Babar Ali, and Tiefeng Wang*

Beijing Key Laboratory of Green Reaction Engineering and Technology

Department of Chemical Engineering, Tsinghua University, Beijing 100084, China

* Corresponding Author: Tel.: 86-10-62794132.

E-mail: wangtf@tsinghua.edu.cn (T. F. Wang)

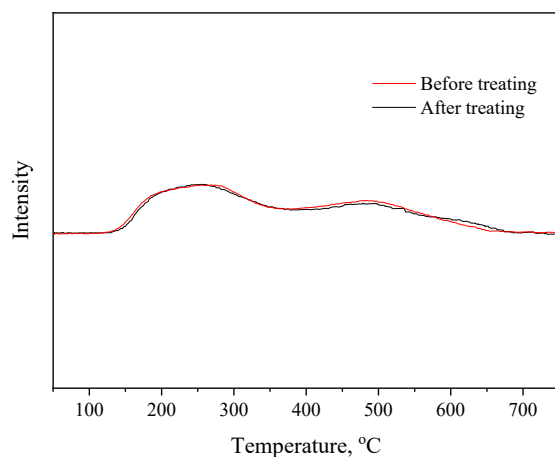
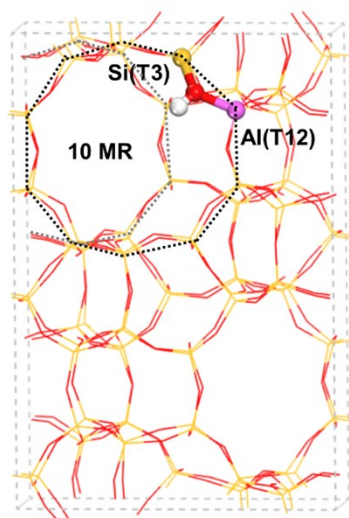


Fig. S1 NH₃-TPD



Framework	ZSM-5
Formal composition	Si ₉₆ O ₁₉₂
Acid site	Al(T12)-OH-Si(T3)
Ring sizes	10-ring
Pore size	5.3 × 5.6 Å
Si/Al ratio	95

Fig. S2 Geometry structures of H-ZSM-5 for DFT calculation. Yellow: silicon; red: oxygen; purple: aluminum; white: hydrogen; gray: carbon. 10 MR means 10-membered-ring. T represents the tetrahedral Si and Al atom.

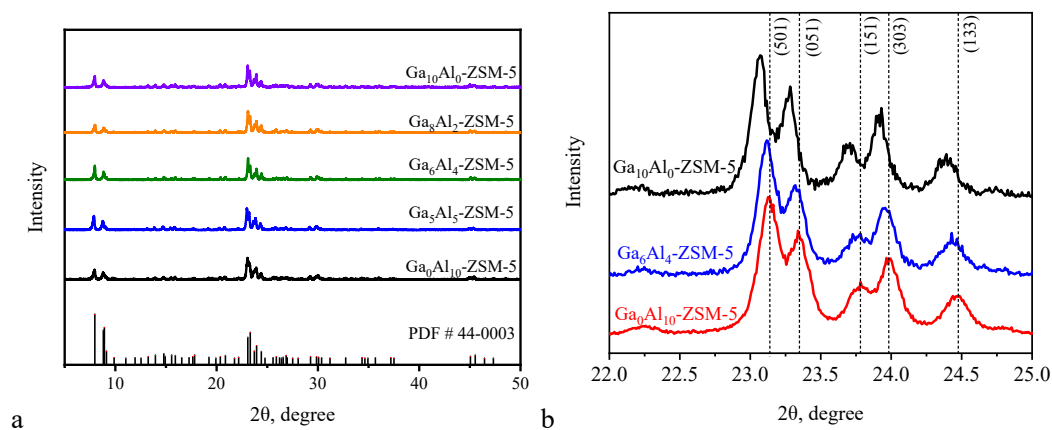


Fig. S3 (a) XRD patterns of the catalysts. (b) XRD patterns of Ga₁₀Al₀-ZSM-5, Ga₆Al₄-ZSM-5, and Ga₀Al₁₀-ZSM-5 in 2θ range of 22°-25°.

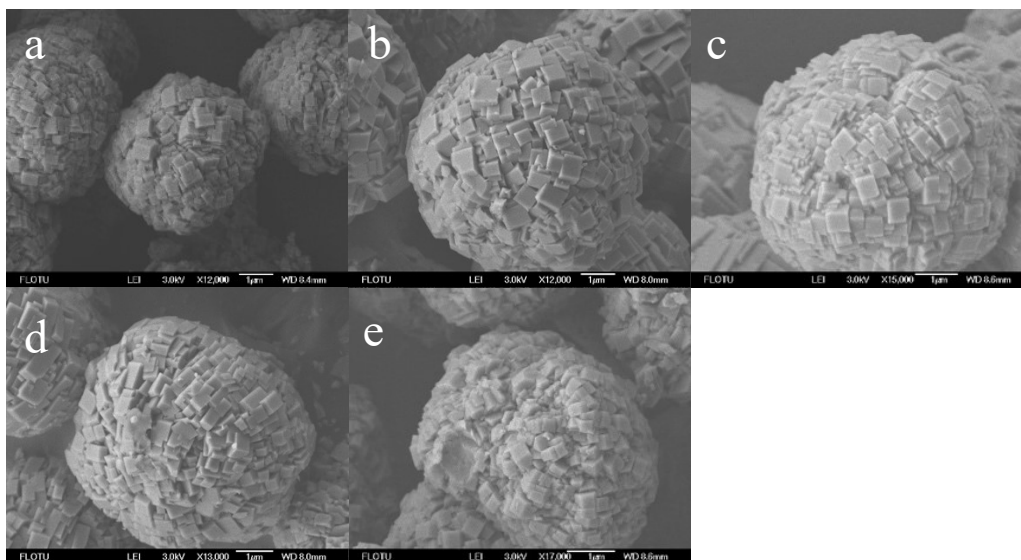


Fig. S4 SEM images of $\text{Ga}_x\text{Al}_y\text{-ZSM-5}$ catalysts. (a) $\text{Ga}_0\text{Al}_{10}\text{-ZSM-5}$. (b) $\text{Ga}_5\text{Al}_5\text{-ZSM-5}$. (c) $\text{Ga}_6\text{Al}_4\text{-ZSM-5}$. (d) $\text{Ga}_8\text{Al}_2\text{-ZSM-5}$. (e) $\text{Ga}_{10}\text{Al}_0\text{-ZSM-5}$.

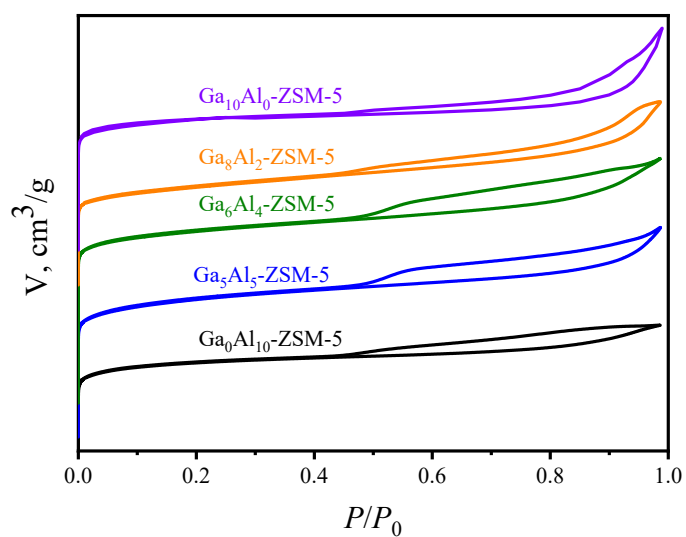


Fig. S5 N_2 adsorption-desorption isotherms of the catalysts.

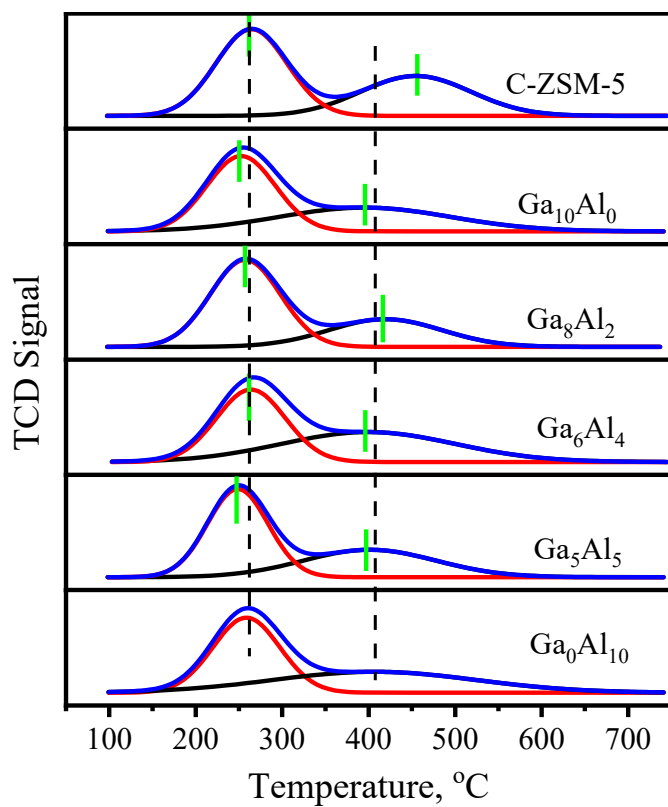


Fig. S6 NH₃-TPD profiles of the catalysts.

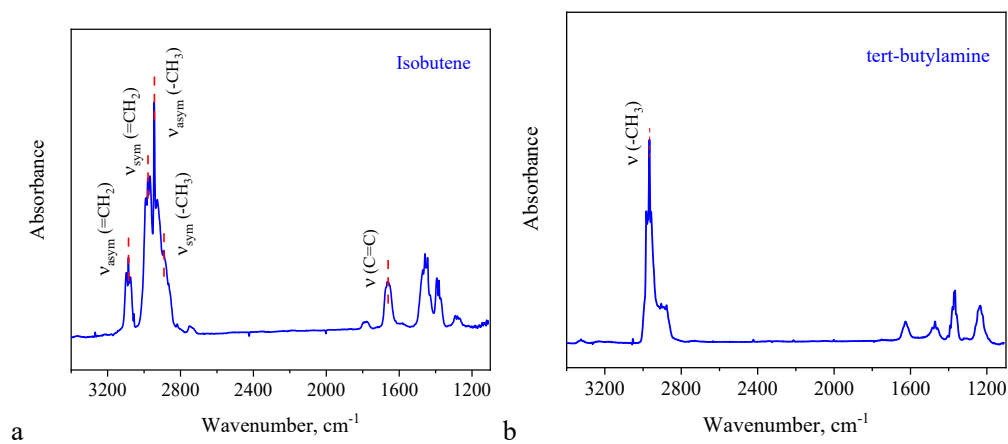


Fig. S7 IR spectra of the gas phase chemicals and corresponding vibration.

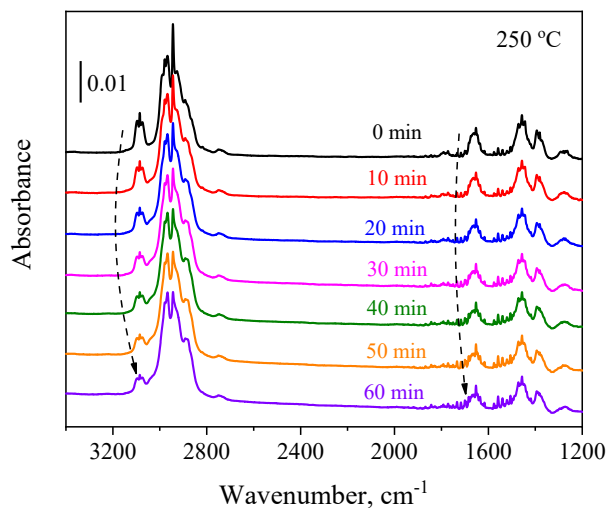


Fig. S8 Evolution of in situ FTIR spectra for isobutene protonation over $\text{Ga}_6\text{Al}_4\text{-ZSM-5}$.

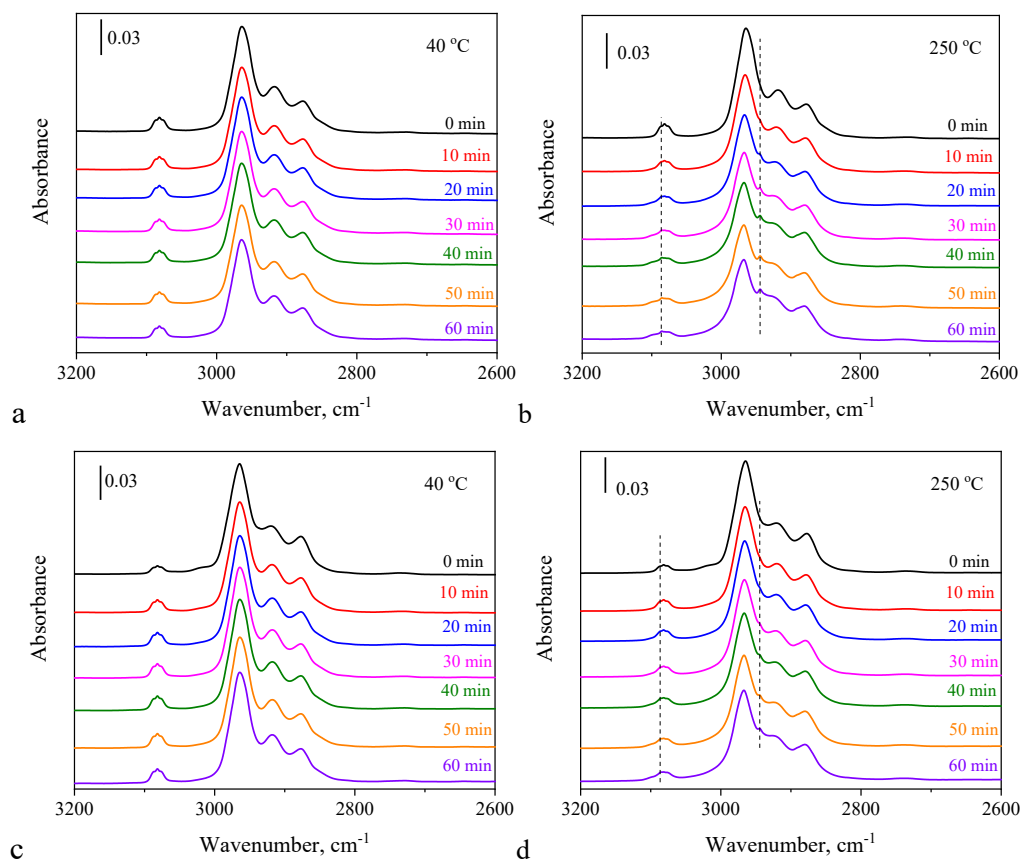


Fig. S9 Dynamic changes of in situ FTIR spectra for the chemicals over $\text{Ga}_6\text{Al}_4\text{-ZSM-5}$. (a) 2,4,4-trimethyl-1-pentene at 40 °C. (b) 2,4,4-trimethyl-1-pentene at 250 °C. (c) 2,4,4-trimethyl-2-pentene at 40 °C. (d) 2,4,4-trimethyl-2-pentene at 250 °C.

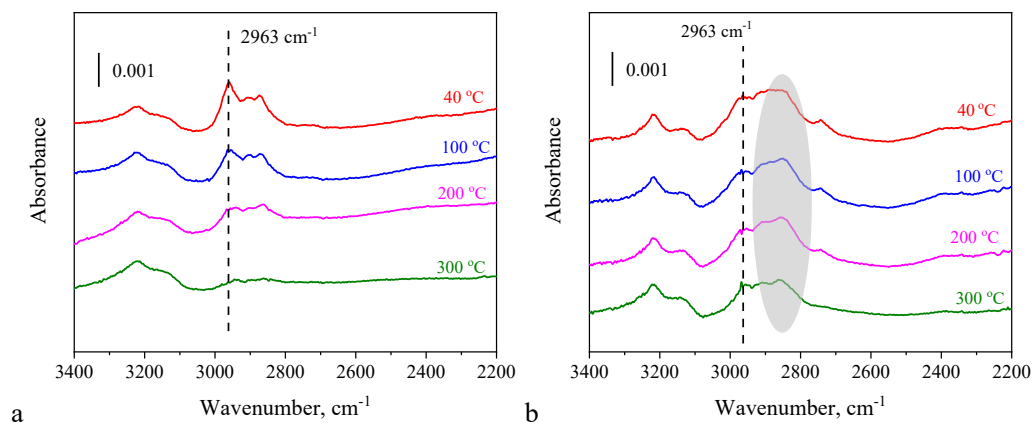


Fig. S10 Desorption of surface species over Ga₆Al₄-ZSM-5. (a) after in situ FTIR experiment of C8-1 at 40 °C. **(b)** after in situ FTIR experiment of C8-1 at 250 °C.

After in situ FTIR experiment of C8-1 for 60 min (as shown in **Fig. S9**), the temperature of the catalyst was first set to 40 °C and then the IR cell was evacuated by a turbo pump to remove the gas phase. With the pressure decreasing to 5×10^{-3} Pa, IR was collected as the spectrum of 40 °C. After that, the temperature of the catalyst was set to desired value (T) and kept for 1 min and then was set back to 40 °C to collect the IR spectra of T .

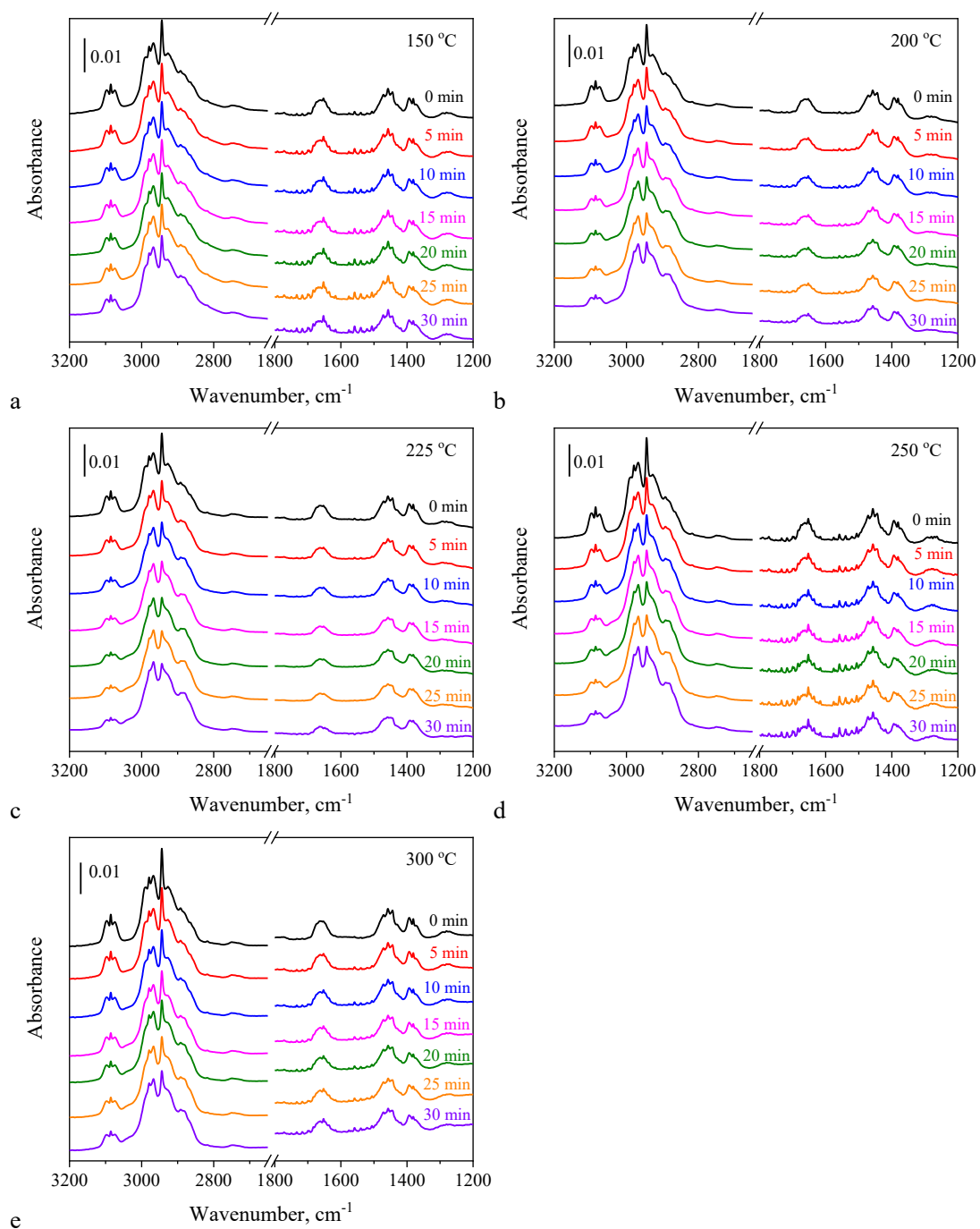


Fig. S11 In situ FTIR spectra of isobutene protonation over $\text{Ga}_6\text{Al}_4\text{-ZSM-5}$ at different temperature. (a) 150 °C. (b) 200 °C. (c) 225 °C. (d) 250 °C. (e) 300 °C.

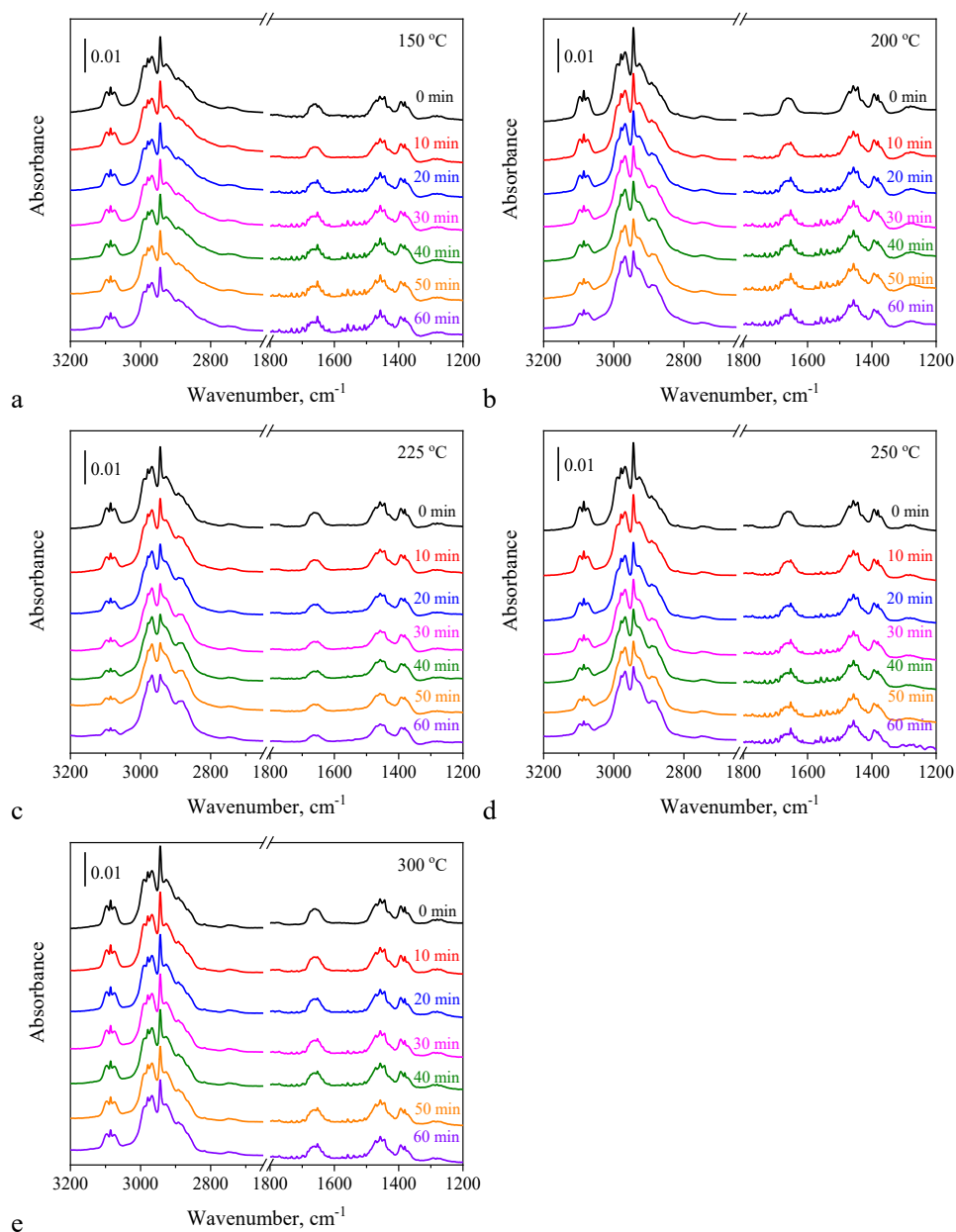


Fig. S12 In situ FTIR spectra of isobutene protonation over $\text{Ga}_{10}\text{Al}_0\text{-ZSM-5}$ at different temperature. (a) 150 °C. (b) 200 °C. (c) 225 °C. (d) 250 °C. (e) 300 °C.

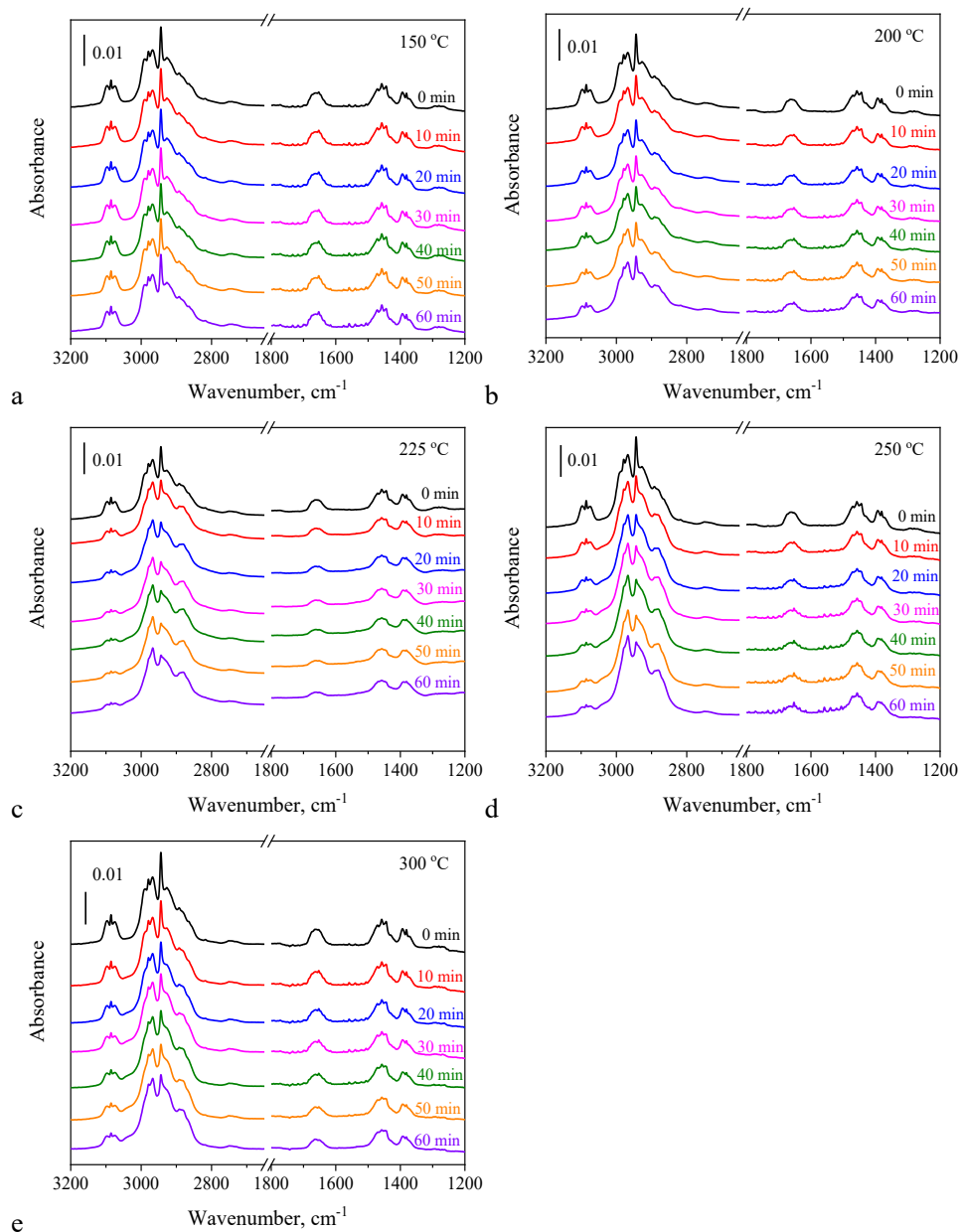


Fig. S13 In situ FTIR spectra of isobutene protonation over $\text{Ga}_0\text{Al}_{10}\text{-ZSM-5}$ at different temperature. (a) 150 °C. (b) 200 °C. (c) 225 °C. (d) 250 °C. (e) 300 °C.

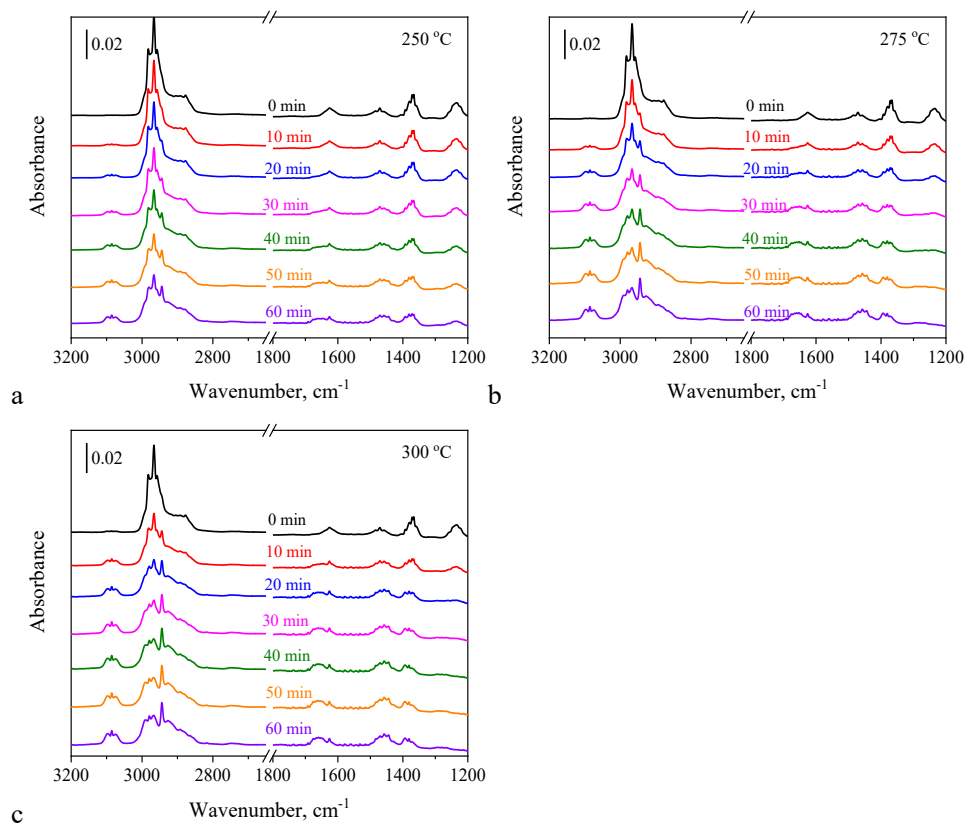


Fig. S14 In situ FTIR spectra of Hoffman elimination over Ga₆Al₄-ZSM-5 at different temperature. (a) 250 °C. (b) 275 °C. (c) 300 °C.

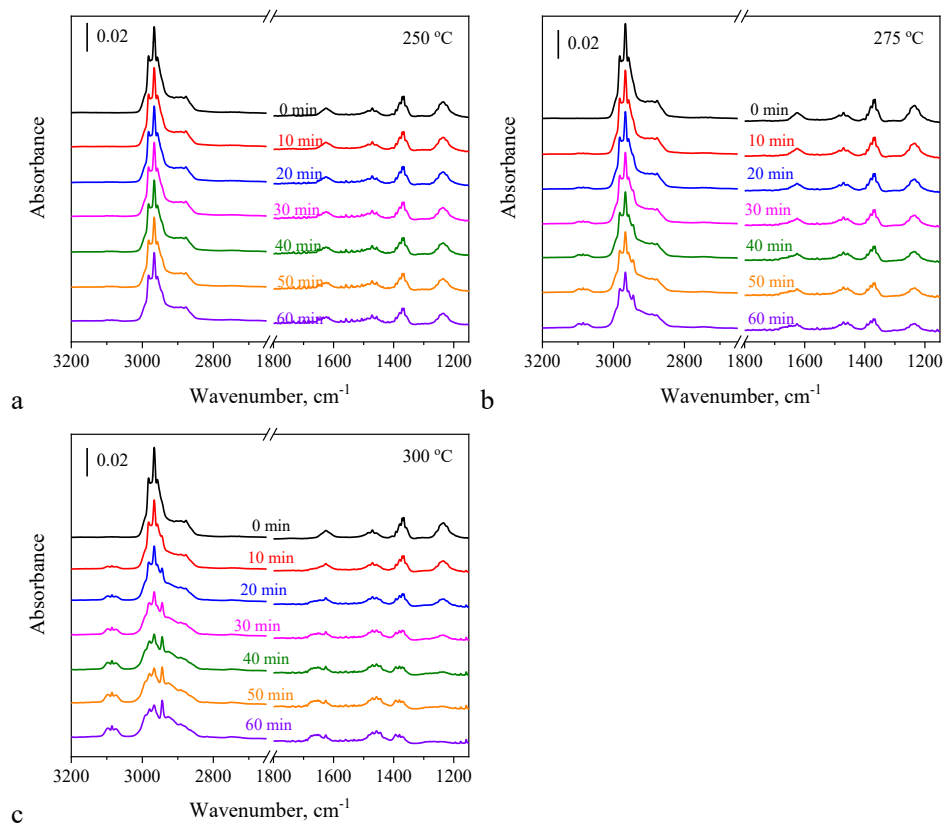


Fig. S15 In situ FTIR spectra of Hoffman elimination over Ga₁₀Al₀-ZSM-5 at different temperature. (a) 250 °C. (b) 275 °C. (c) 300 °C.

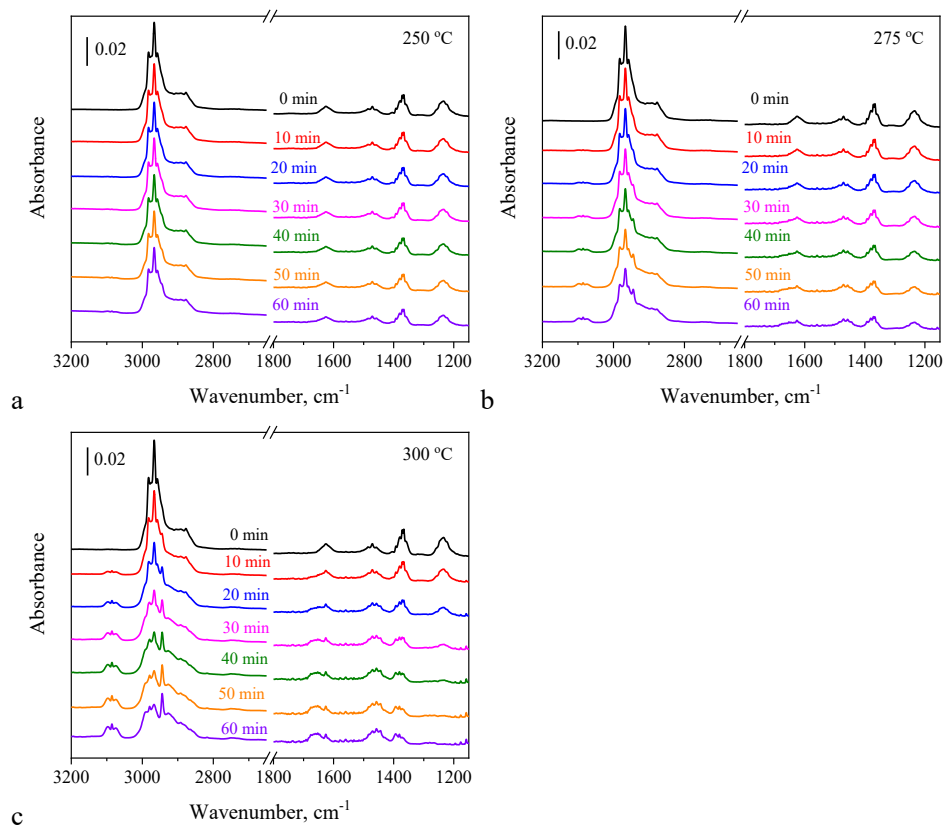
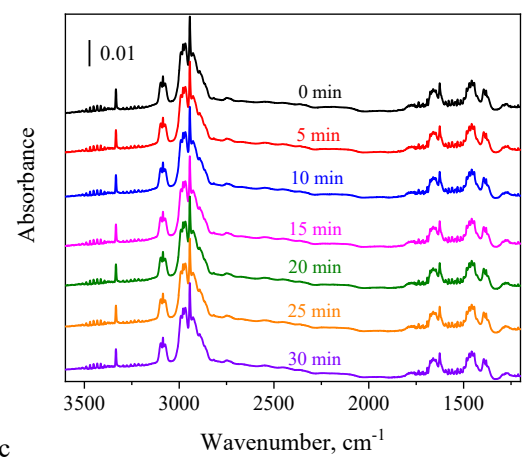
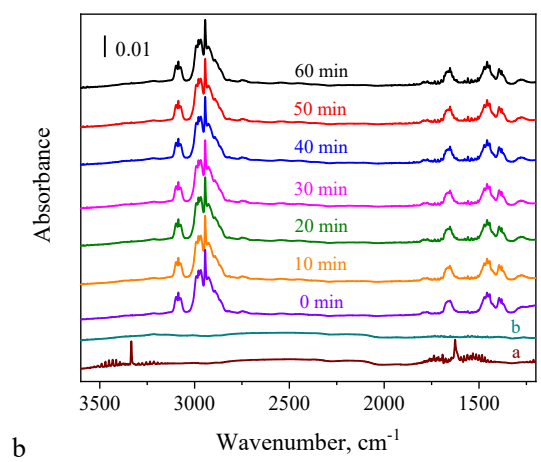
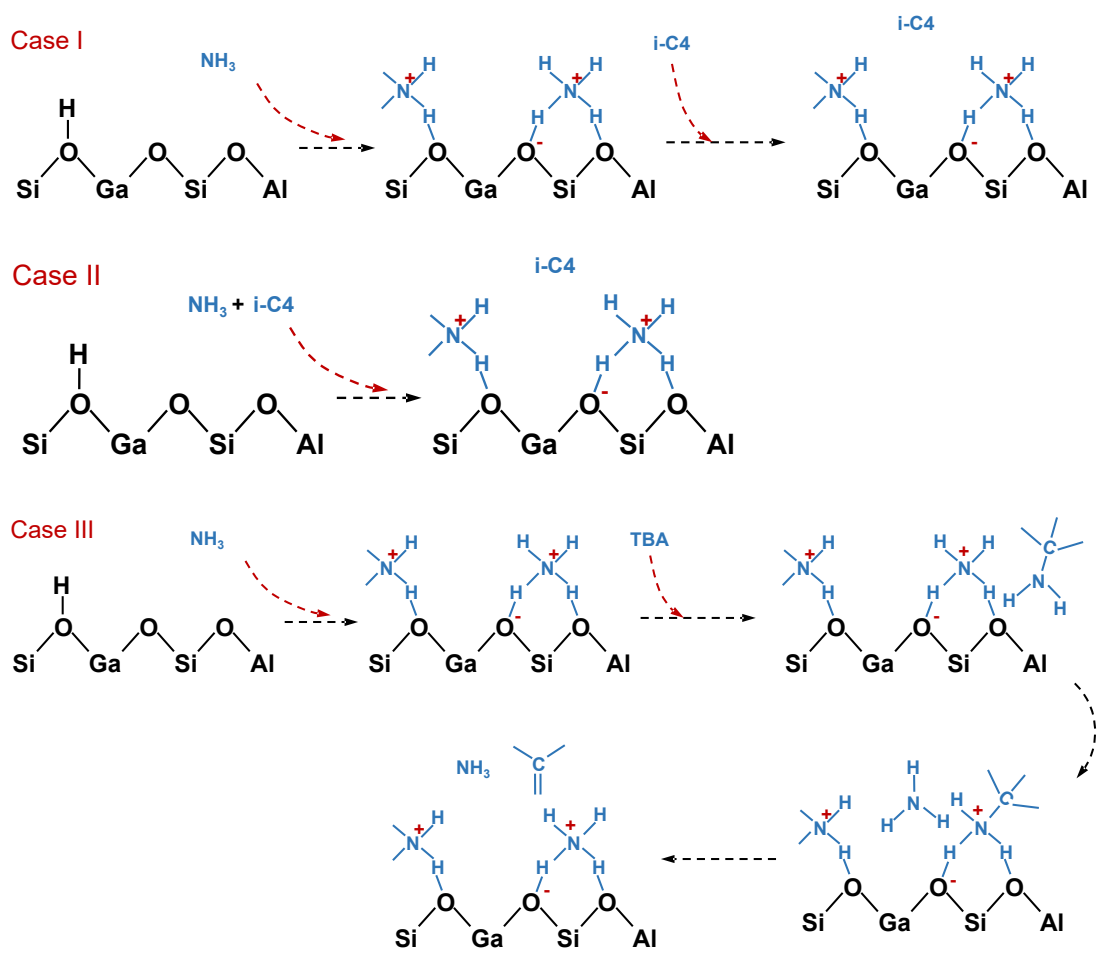


Fig. S16 In situ FTIR spectra of Hoffman elimination over Ga₀Al₁₀-ZSM-5 at different temperature. (a) 250 °C. (b) 275 °C. (c) 300 °C.

a



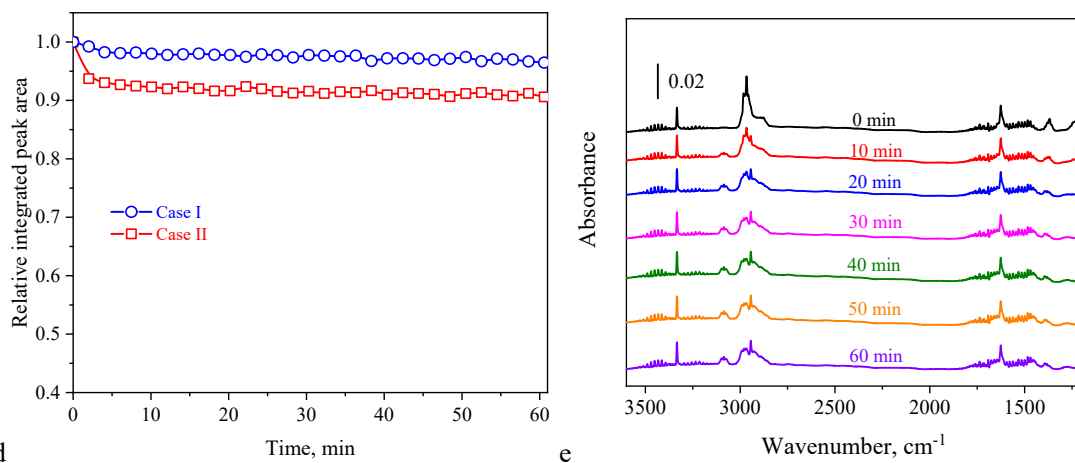


Fig. S17 Investigation of the relative adsorption stability of the chemicals. (a) schemes of the cases. **(b)** the in situ FTIR spectra of case I ($T = 250\text{ }^{\circ}\text{C}$). **(c)** the in situ FTIR spectra of case II ($T = 250\text{ }^{\circ}\text{C}$). **(d)** the integrated peak area of band at 3050-3125 cm^{-1} of case I and II. **(e)** the in situ FTIR spectra of case III ($T = 300\text{ }^{\circ}\text{C}$).

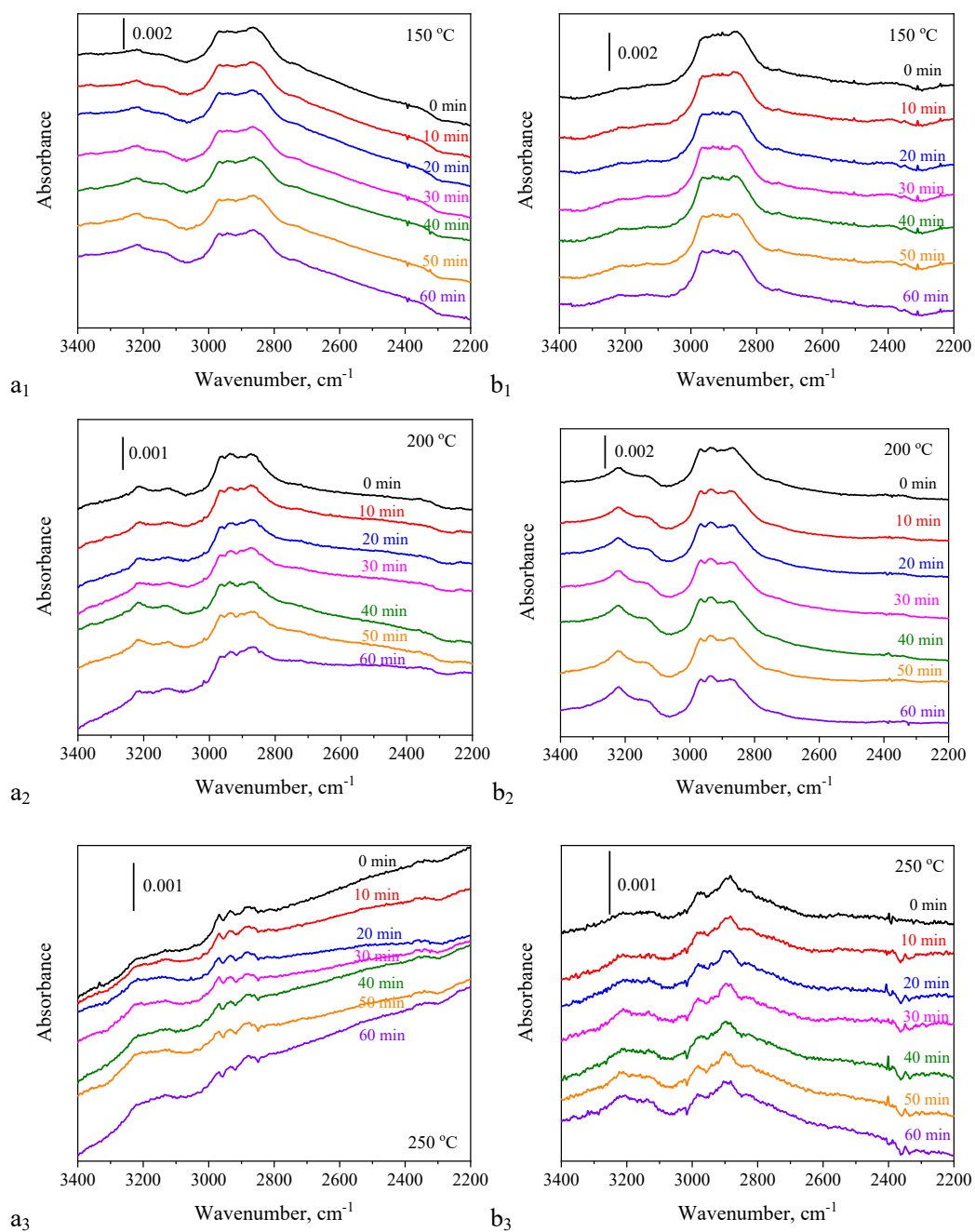
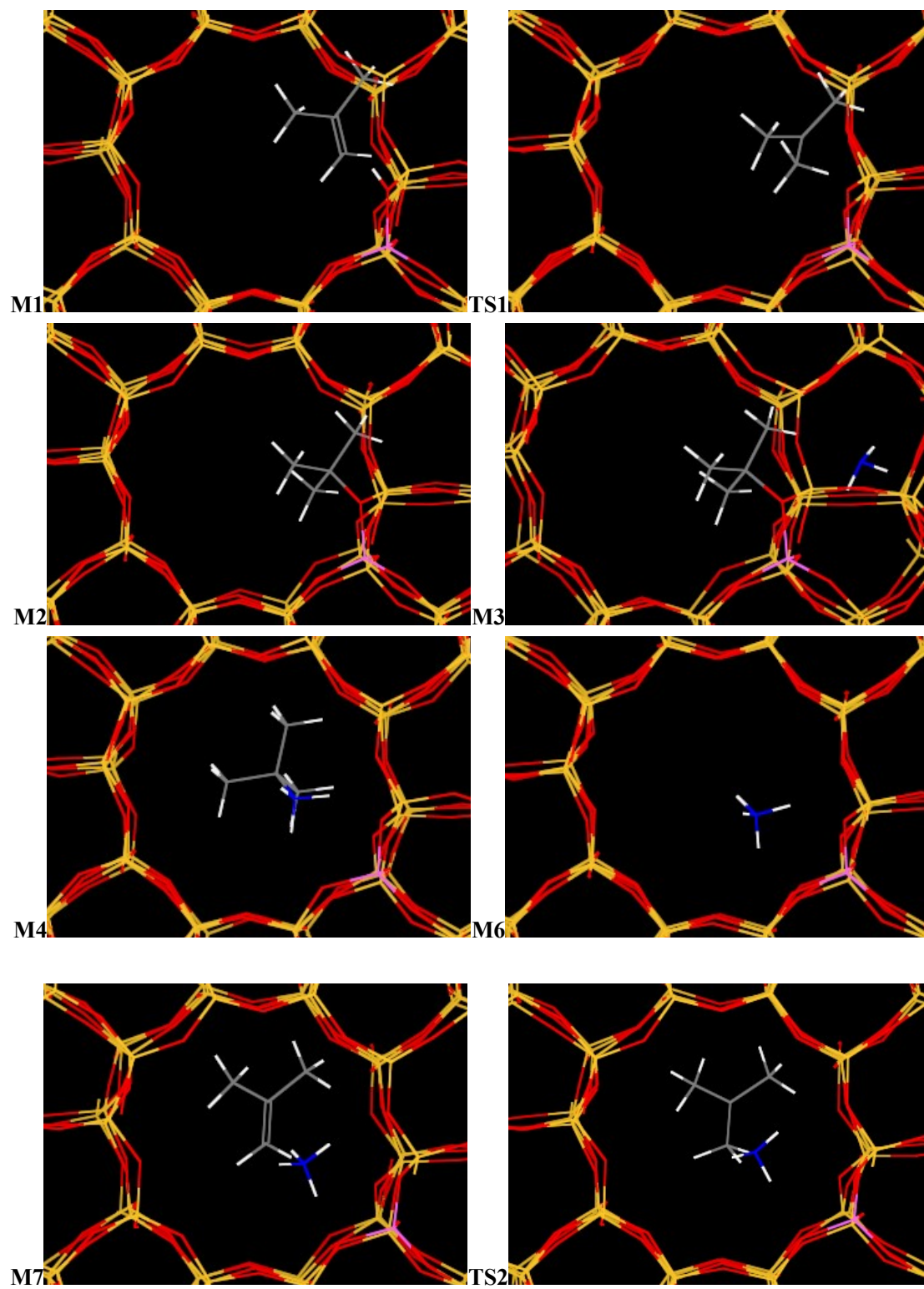


Fig. S18 In situ FTIR dynamic changes of alkoxide. (a) under 4 kPa NH₃ atmosphere at 150 °C (a₁), 200 °C (a₂), and 250 °C (a₃). **(b)** under He atmosphere at 150 °C (b₁), 200 °C (b₂), and 250 °C (b₃).



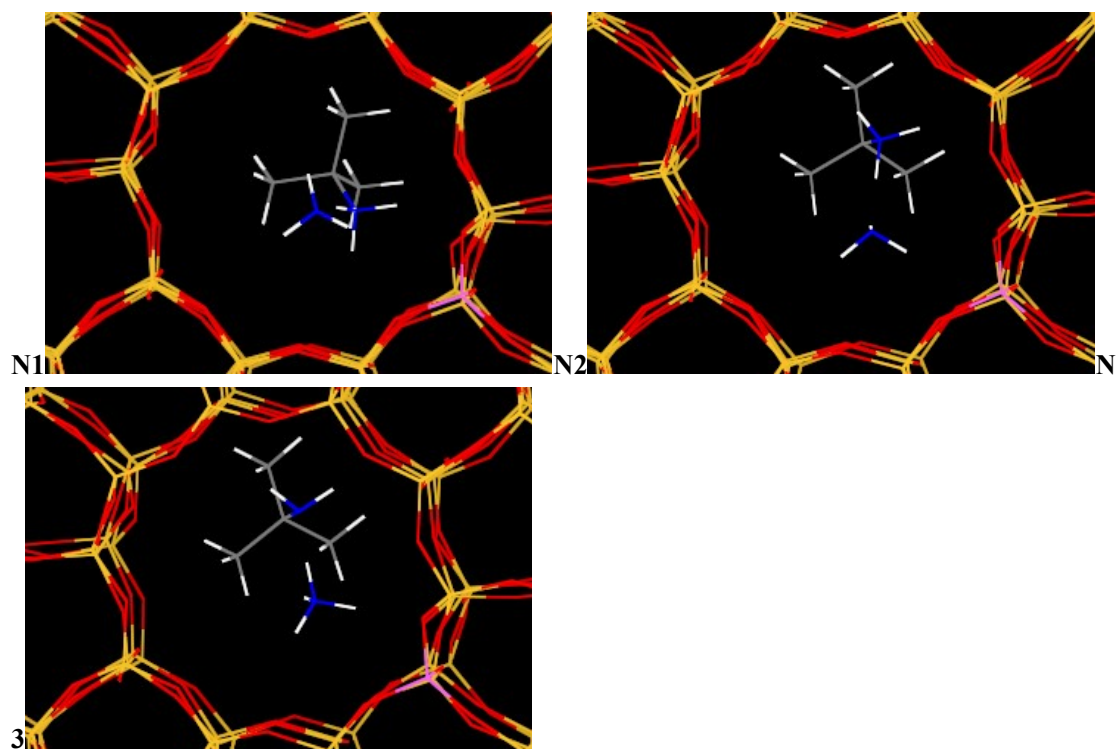


Fig. S19 Geometry structures of the intermediates along reaction coordinate over ZSM-5.

Table S1. Characterization of the catalysts.

Samples	^a Si/(Al+Ga)	^b Surface area, m ² /g	^c Acid amount, mmol/g
Ga ₀ Al ₁₀ -ZSM-5	18.1	367	0.285
Ga ₅ Al ₅ -ZSM-5	20.2	326	0.285
Ga ₆ Al ₄ -ZSM-5	17.5	306	0.256
Ga ₈ Al ₂ -ZSM-5	21.0	302	0.214
Ga ₁₀ Al ₀ -ZSM-5	21.3	241	0.380

^a. Measured by XRF; ^b. Calculated by BET and t-plot methods; ^c. Mid-strong acid amount measured by NH₃-TPD between 350 and 650 °C.

Table S2. Adsorption energy of the chemicals over Al-MFI and Ga-MFI.

Model	NH ₃ /eV	IBE/eV	TBA/eV
Ga-12T	-1.43	-1.08	-2.48
Al-12T	-1.47	-1.09	-2.52