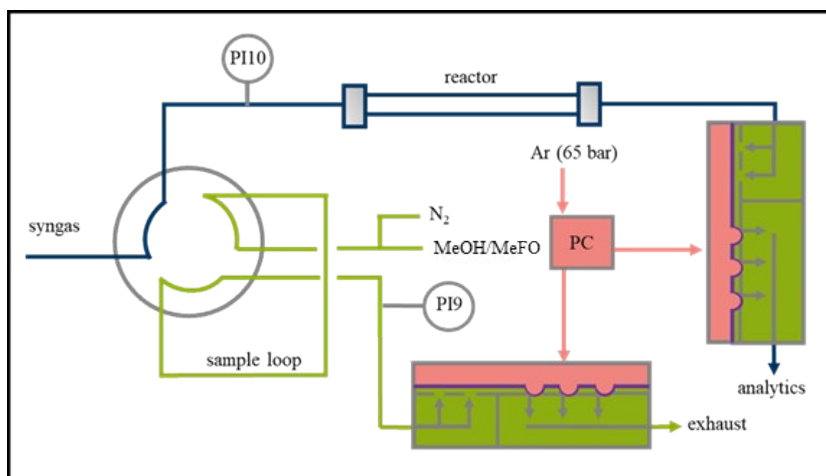
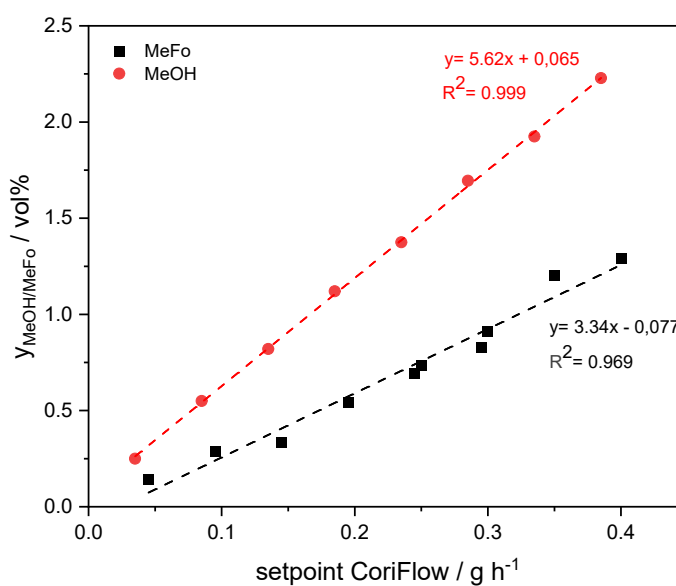


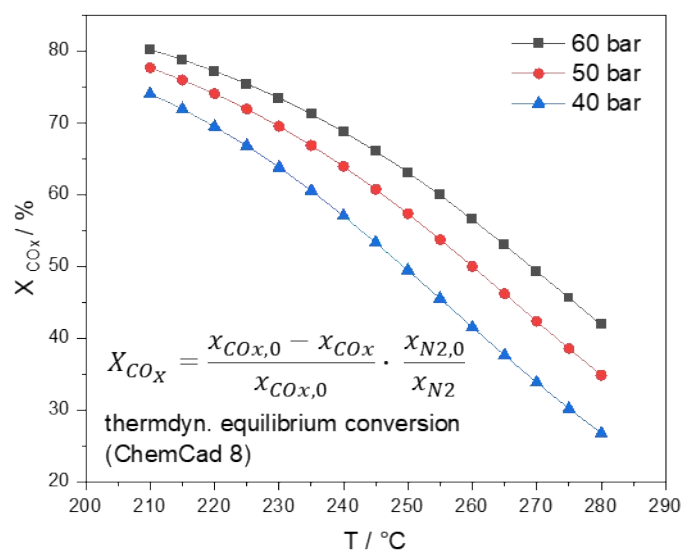
## Supporting information



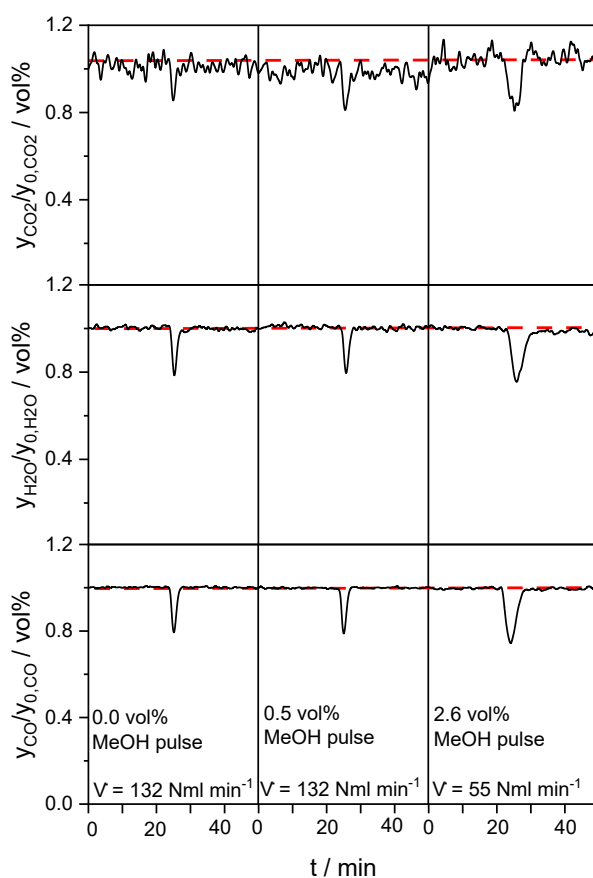
**Fig. S1.** Schematic drawing of the high-pressure pulse unit (HPPU) containing a 6 port, a 2 position dosing valve equipped with a 1 mL sample loop and two identical back pressure regulators (BPR) connected to a process pressure controller.



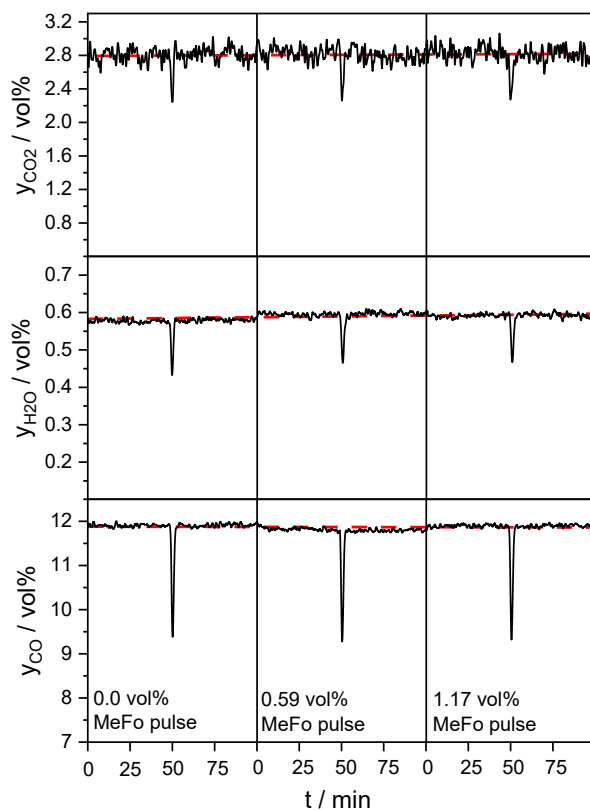
**Fig. S2.** Calibration of methyl formate and methanol for the high-pressure evaporator system. The evaporator temperature was set to 165°C and the diluting N<sub>2</sub> gas flow was 200 Nml min<sup>-1</sup>.



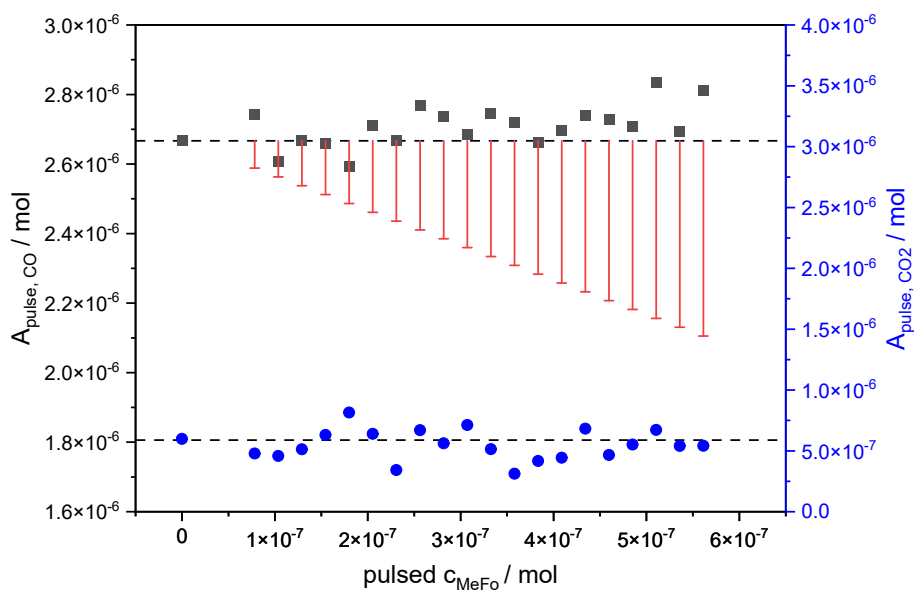
**Fig. S3.** Thermodynamic equilibrium conversion for methanol synthesis from CO/CO<sub>2</sub>/H<sub>2</sub>/N<sub>2</sub> (13.5 vol%/ 3.5 vol%/ 73.5 vol%/ 9.5 vol%) synthesis gas at 210°C and 60 bar 40, 50 and 60 bar depending on temperature. All calculations were performed using ChemCad 8.



**Fig. S4.** Recorded mole fractions of CO, CO<sub>2</sub> and H<sub>2</sub>O normalized to the initial mole fraction before pulses during the methanol pulse series. Methanol synthesis was performed from CO/CO<sub>2</sub>/H<sub>2</sub>/N<sub>2</sub> (13.5 vol%/ 3.5 vol%/ 73.5 vol%/ 9.5 vol%) synthesis gas at 210°C and 60 bar. The mole fractions of the dosed methanol pulses and the respective gas flow rates are shown below the pulses.



**Fig. S5.** Recorded mole fractions of CO, CO<sub>2</sub> and H<sub>2</sub>O during the methyl formate pulse series. Methanol synthesis was performed from CO/CO<sub>2</sub>/H<sub>2</sub>/N<sub>2</sub> (13.5 vol%/ 3.5 vol%/ 73.5 vol%/ 9.5 vol%) synthesis gas at 210°C and 60 bar. The mole fractions of the dosed methyl formate pulses are shown below the pulses.



**Fig. S6.** Molar amounts of CO (black squares) and CO<sub>2</sub> (blue dots) detected as responses to methyl formate pulses and the missing molar amount of CO per pulse in red.

**Table S1.** Partial pressures of methanol in the 1 ml sample loop at 60 bar.

pulsed $n_{\text{MeOH}}$ / mol	pulsed $n_{\text{N}_2}$ / mol	pulsed $n_{\text{MeOH}+\text{N}_2}$ / mol	$p_{\text{MeOH}}$ / bar
3.66E-07	3.01E-05	5.92E-03	0.72
4.09E-07	3.01E-05	5.92E-03	0.80
4.61E-07	3.00E-05	5.92E-03	0.91
5.14E-07	3.00E-05	5.92E-03	1.01
5.92E-07	3.00E-05	5.92E-03	1.16
5.77E-07	2.99E-05	5.92E-03	1.13
6.19E-07	2.99E-05	5.92E-03	1.22
6.63E-07	2.98E-05	5.92E-03	1.30
7.05E-07	2.98E-05	5.92E-03	1.39
7.48E-07	2.97E-05	5.92E-03	1.47
7.91E-07	2.97E-05	5.92E-03	1.56