

Supporting information for

**The synergetic effect of Pd, In and Zr on the mechanism of
Pd/In₂O₃-ZrO₂ for CO₂ hydrogenation to methanol**

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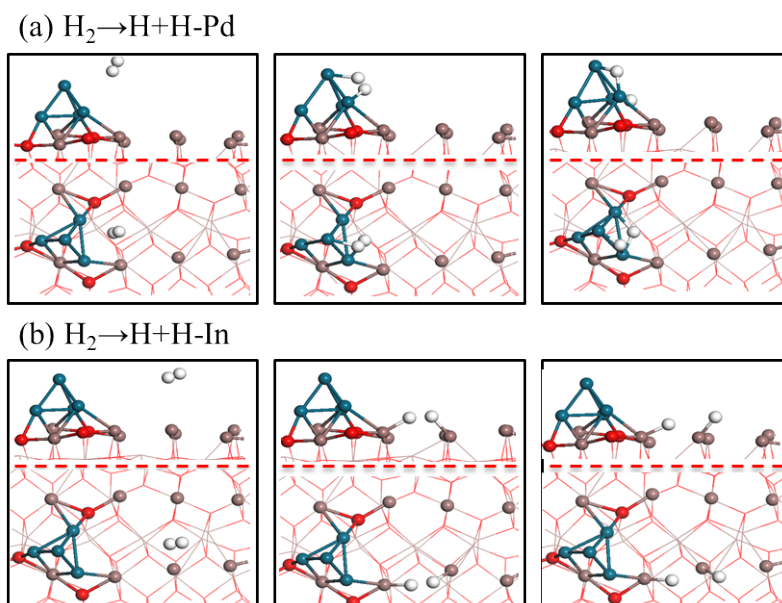


Fig. S1. The initial, transition and final states of H_2 dissociation on (a) Pd_4 cluster and (b) In site of $\text{Pd}_4/\text{In}_2\text{O}_3(110)\text{-3O}$ model

Table S1 Calculated reaction energies, E_r (eV), and activation barriers, E_a (eV), of H_2 dissociations

Reaction site	E_a	E_r
(a) Pd site on $\text{Pd}_4/\text{In}_2\text{O}_3(110)\text{-3O}$	0.16	-0.75
(b) In site on $\text{Pd}_4/\text{In}_2\text{O}_3(110)\text{-3O}$	0.69	0.46
(c) In site on $\text{In}_2\text{O}_3(110)\text{-In}$	0.96	0.71

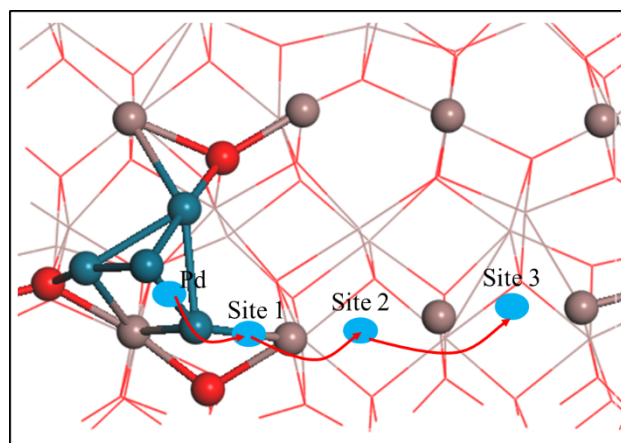


Fig. S2. The transfer pathway of H on $\text{Pd}_4/\text{In}_2\text{O}_3(110)\text{-3O}$ surface

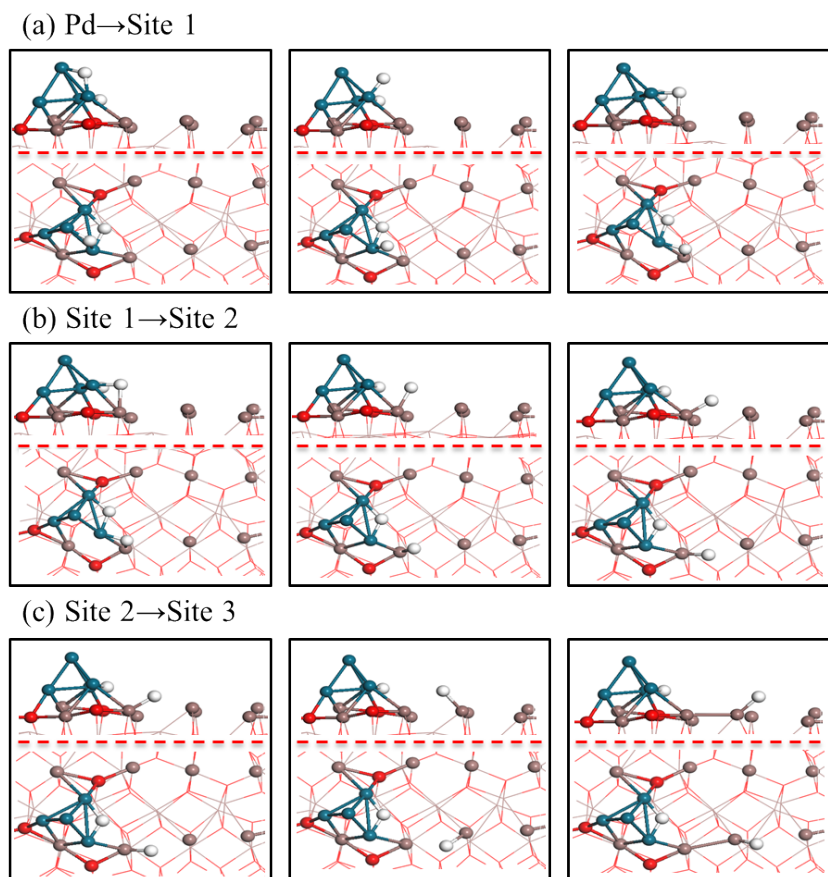


Fig. S3. The initial, transitional and final states of the H transfer on Pd₄/In₂O₃(110)-3O surface

Table S2 Calculated reaction energies, E_r (eV), and activation barriers, E_a (eV), of H transfers

Diffusion site	E_a	E_r
(a) Pd site → Site 1	0.63	0.34
(b) Site 1 → Site 2	0.55	0.24
(c) Site 2 → Site 3	0.66	0.03

Table S3 Calculated reaction energies, E_r (eV), and activation barriers, E_a (eV), of elementary reaction steps in CO₂ conversion on In₂O₃(110)-In surface

Elementary reaction step	E_a	E_r
(a) CO ₂ + H → HCOO	0.76	-0.87
(b) CO ₂ + H → COOH	0.99	0.01

Table S4 The adsorption energies of intermediates, E_{ab} (eV), on Pd₄/In₂O₃(110)-4O/Zr model

Species	E_{ab}
b-CO ₂	-0.78
p-CO ₂	-0.62
CO	-1.26
HCOO	-2.99
H ₂ COO	-5.42
H ₂ COOH	-3.04
H ₂ CO	-1.21
H ₃ CO	-2.93
H ₃ COH	-0.51
COOH	-2.06
HCOOH	-0.71
HCO	-2.13