Supporting information

Dehydrogenation of the liquid organic hydrogen carrier perhydrodibenzyltoluene – reaction pathway over Pt/Al₂O₃

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Fig. S1 Dehydrogenation of H18-DBT at different temperature



Fig. S2 GC-MS results of dehydrogenation products at 270 °C and 300 °C



Fig. S3 Hx-DBT species from GC-MS results: composition change during dehydrogenation reaction for H18-DBT species to H12-, H6-, and H0-DBT species versus time.



Fig. S4. The experimental ¹H NMR results of dehydrogenation products of Pt/Mo-Al₂O₃ catalysts (fivepointed stars) and the different dehydrogenation pathways (solid lines)

Table S1	The average	residual	between e	each exper	imental po	oint and	each	dehyd	rogenation	1 path	way	over
Pt/Mo-Al ₂ O ₃												

	Average residual							
Plots	SSM	MSS	SMS	Statistical				
$f_{CS[0.8-1.8]} \sim f_{CS[2-2.5]}$	0.016	0.089	0.021	0.043				
$f_{CS[0.8-1.8]} \sim f_{CS[3.8-4.2]}$	0.074	0.010	0.010	0.031				
$f_{CS[2-2.5]} \sim f_{CS[3.8-4.2]}$	0.020	0.043	0.014	0.029				
$f_{CS[6.8-7.4]} \sim f_{CS[0.8-1.8]}$	0.079	0.083	0.012	0.035				
$f_{CS[6.8-7.4]} \sim f_{CS[2-2.5]}$	0.033	0.122	0.014	0.052				
$f_{CS[6.8-7.4]} \sim f_{CS[3.8-4.2]}$	0.122	0.012	0.009	0.032				