

Supplementary Information (SI)

Diverse Continuous Photooxygenation Reactions of (+) and (-)- α - Pinenes to the Corresponding Pinocarvones or trans- Pinocarveols

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1. Complementary General Information

The CG-MS yields were determined in a SHIMADZU equipment: GCMS-QP5000 and GC17A using argon as gas (column 30 m, NA-5 MS XIL – 0.25 mm, 0.25 μm – Bonded T max: 330-350 $^{\circ}\text{C}$): samples were prepared in ethyl acetate; injection temperature 250 $^{\circ}\text{C}$, injection split ratio 19, carrier gas He, pressure 100 KPa, and column flow rate 1.3 mL/min; oven temperature setting was 100 $^{\circ}\text{C}$ for 5 min, heated at 15 $^{\circ}\text{C}\cdot\text{min}^{-1}$ to 150 $^{\circ}\text{C}$, then heated at 7.5 $^{\circ}\text{C}\cdot\text{min}^{-1}$ to 220 $^{\circ}\text{C}$, and finally heated at 10 $^{\circ}\text{C}\cdot\text{min}^{-1}$ to 280 $^{\circ}\text{C}$ and held for 7 min. Conversion percentages were analysed on the basis of chromatogram areas, with mass ion source temperature 250 $^{\circ}\text{C}$, interface temperature 280 $^{\circ}\text{C}$, and solvent cut time 5 min. The standard curve was constructed for the measured product, and quantifications were performed using dodecane as an internal standard.

Continuous-flow experiments were performed using Syrris equipment (Asia modules): two channels of a syringe pump (500 and 1000 μL), a KNAUER HPLC pump (AZURA P 4.1S), and a microchip reactor (250 μL). Two back-pressure regulators (BPR with pressures as specified). For pumping the Ph_3P solution, a Peristaltic Pump (New Era, model NE-9004C) was used.

a. Batch photoreactor construction

Details on the construction of this photoreactor can be obtained in our previous publication,^{1,2} and pictures of the steps and materials are presented in Figures S1-S2.

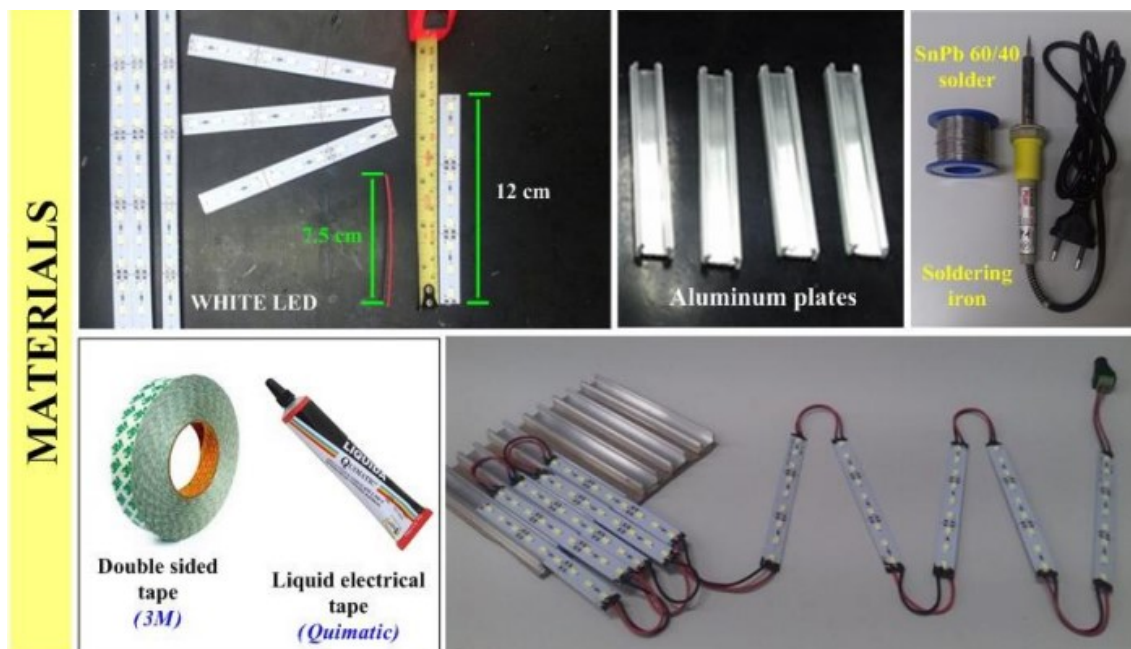


Figure S1. Materials used in the batch photoreactor.

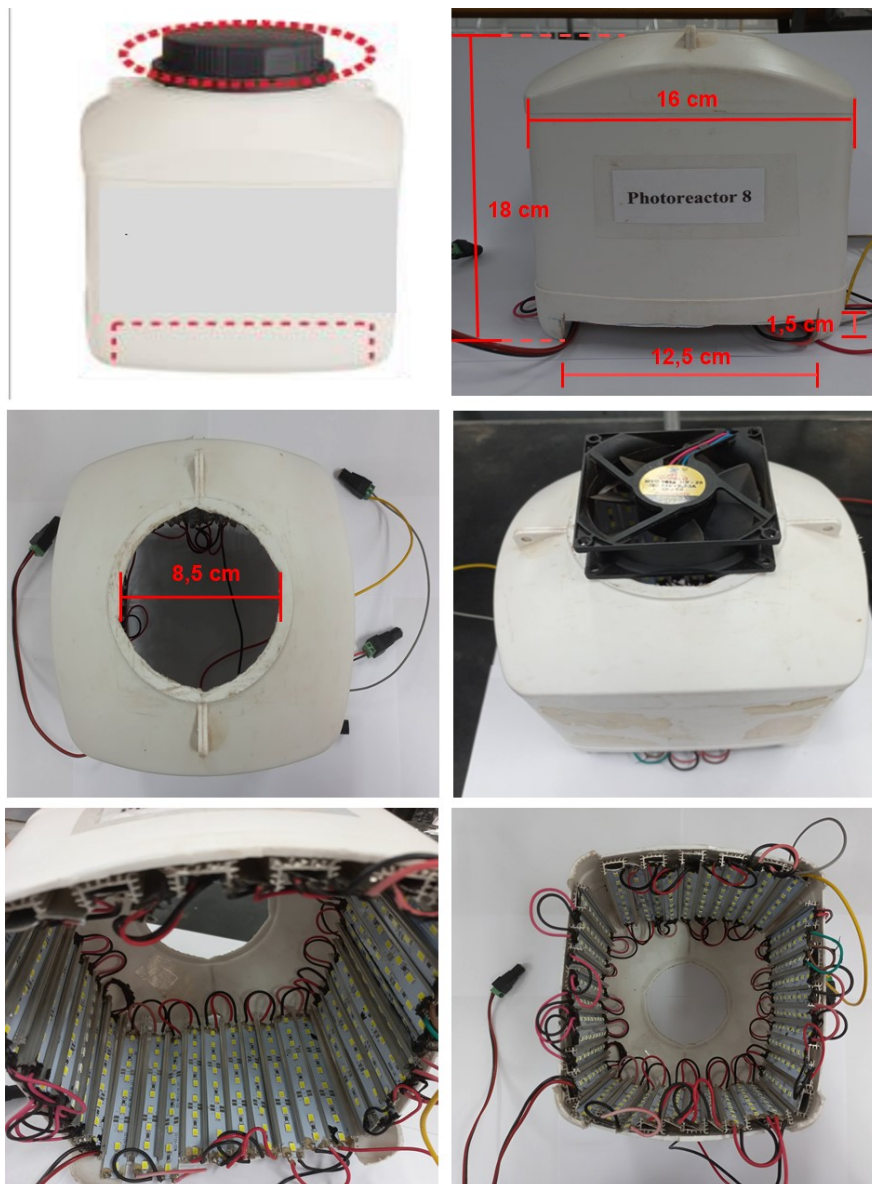


Figure S2. Step-by-Step of Batch photoreactor assembly.

b. Continuous Flow photoreactor construction

Details on the construction of this photoreactor can be obtained in our previous publication,^{1,2} and pictures of the steps and materials are presented in Figures S3.

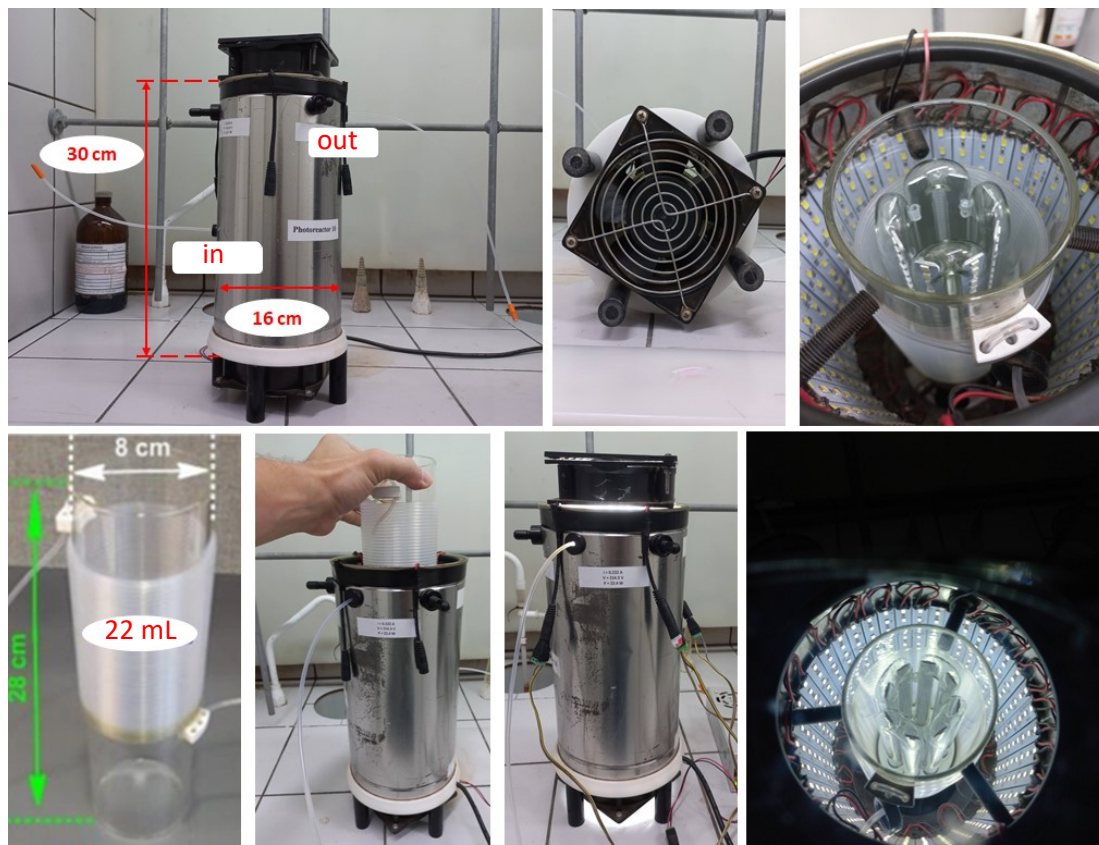


Figure S3. Final assembly of the photoreactor and PFA tubular reactor for continuous flow synthesis.

c. Tube-in-tube reactor

The tube-in-tube (TIT) reactor used to oxygenate the solutions was constructed in our lab, based on the model by the Ley group³. Details of the construction of this reactor can be obtained in our previous work,^{1,2} and pictures of the steps and materials are presented in Figures-S4.

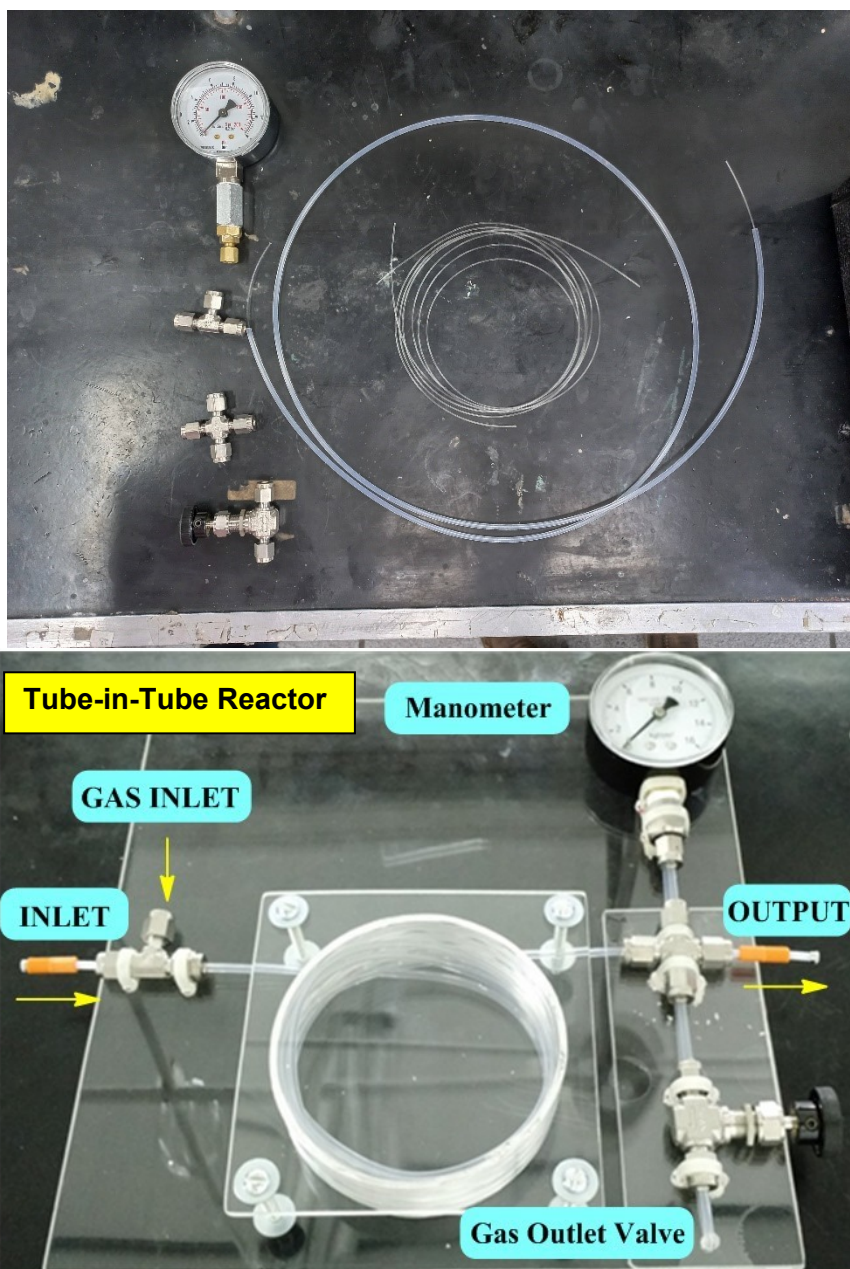


Figure S4. Our home-made tube-in-tube reactor.

2. Batch results - setup S1

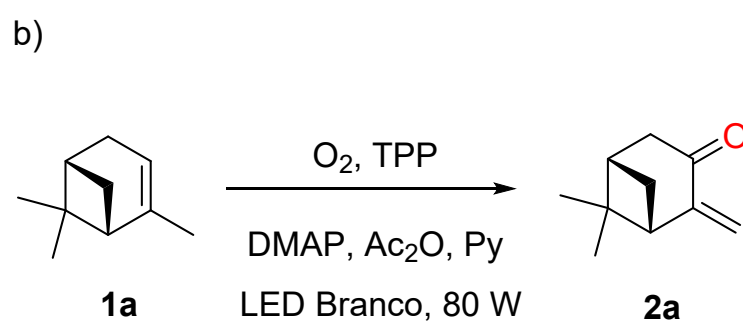


Figure S5 – a) Batch setup reactor for the synthesis of (-)-pinocarvone. b) Conversion of (+)- α -pinene into (-)-pinocarvone.

3. Flow setup S1

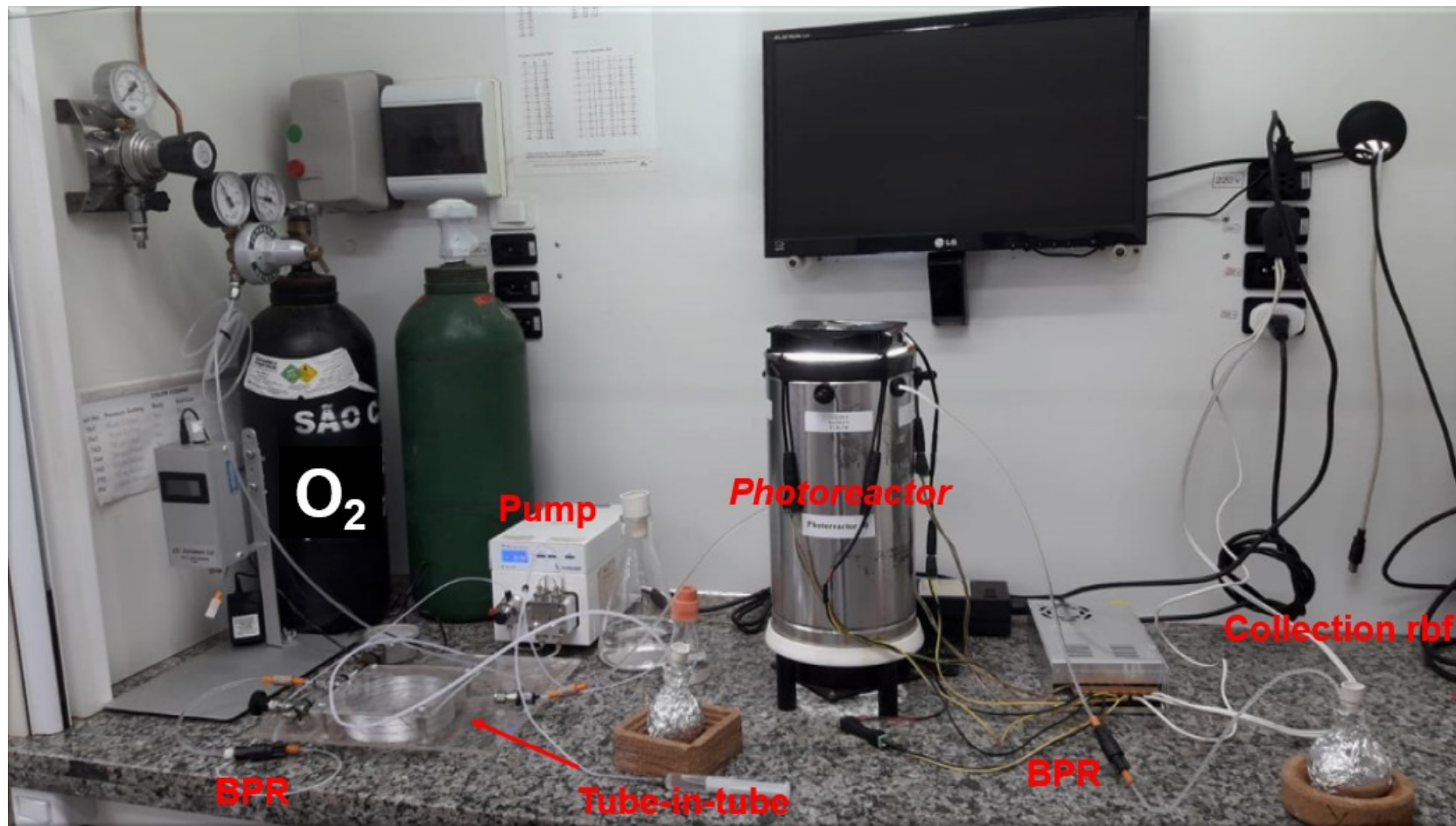


Figure S6 - Flow setup 1 for the synthesis of (-)-pinocarvone (2a).

4. Flow Setup S2



Figure S7 – Flow setup 2 for the synthesis of pinocarvones (2a-b)

Flow setup S3

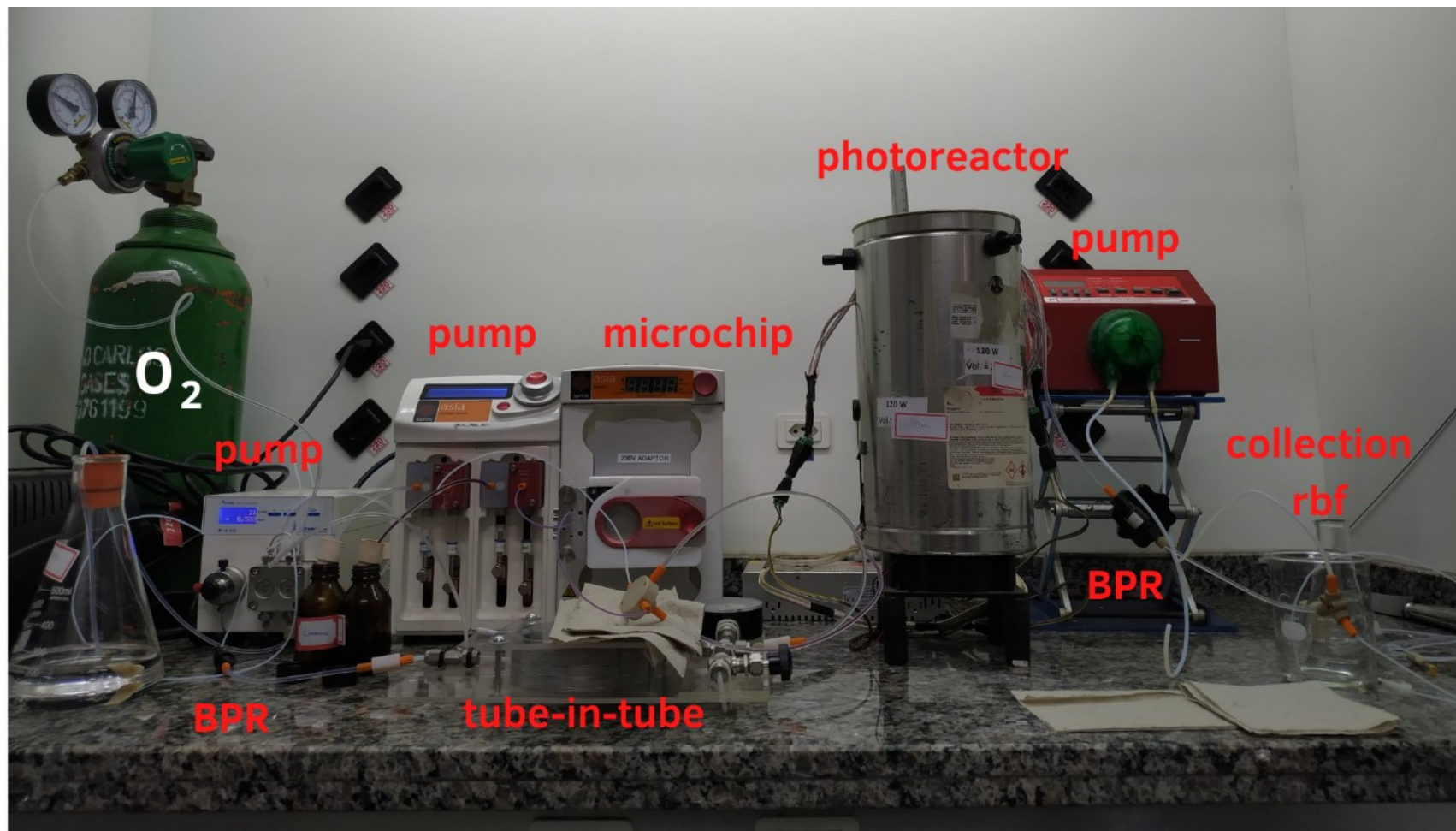
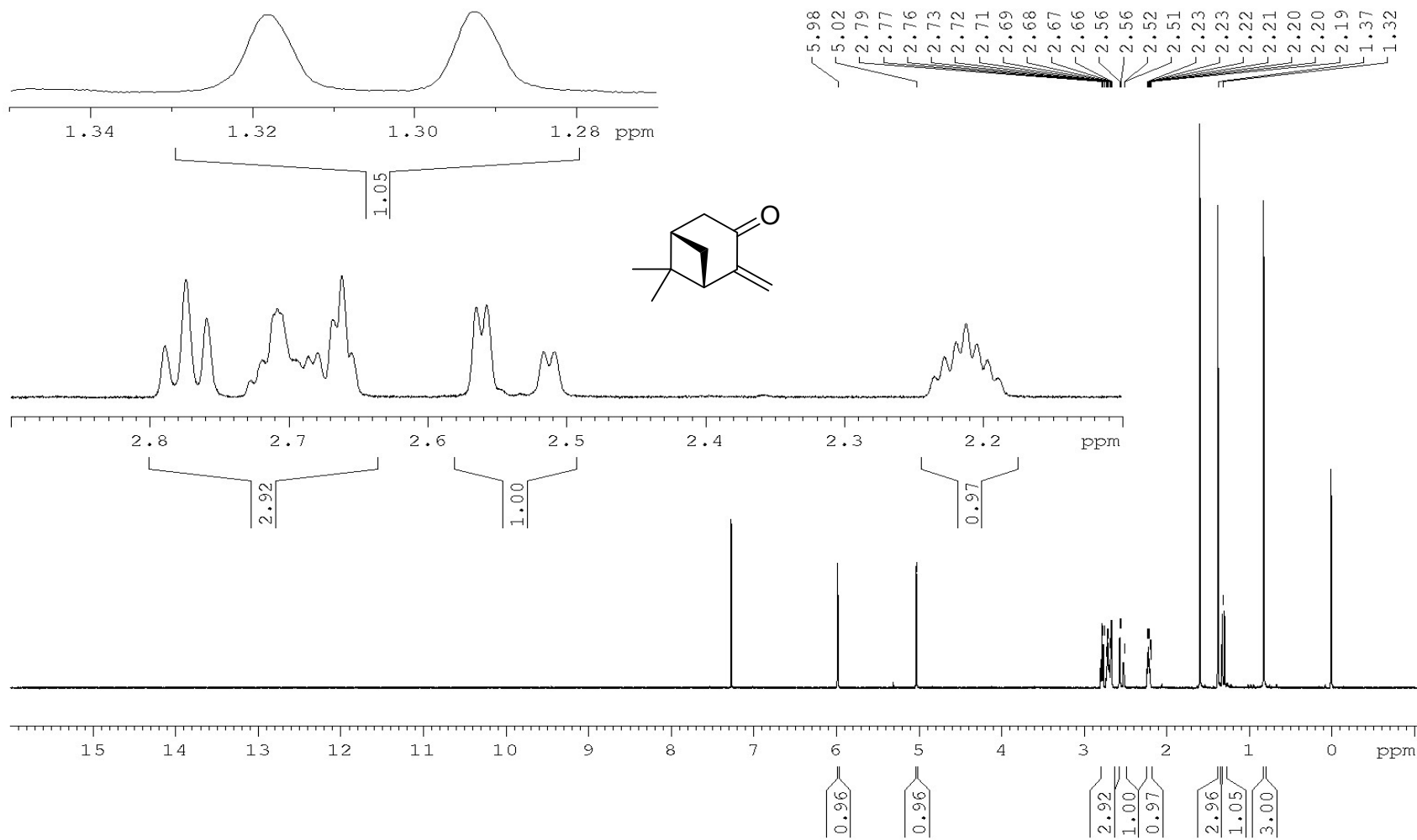


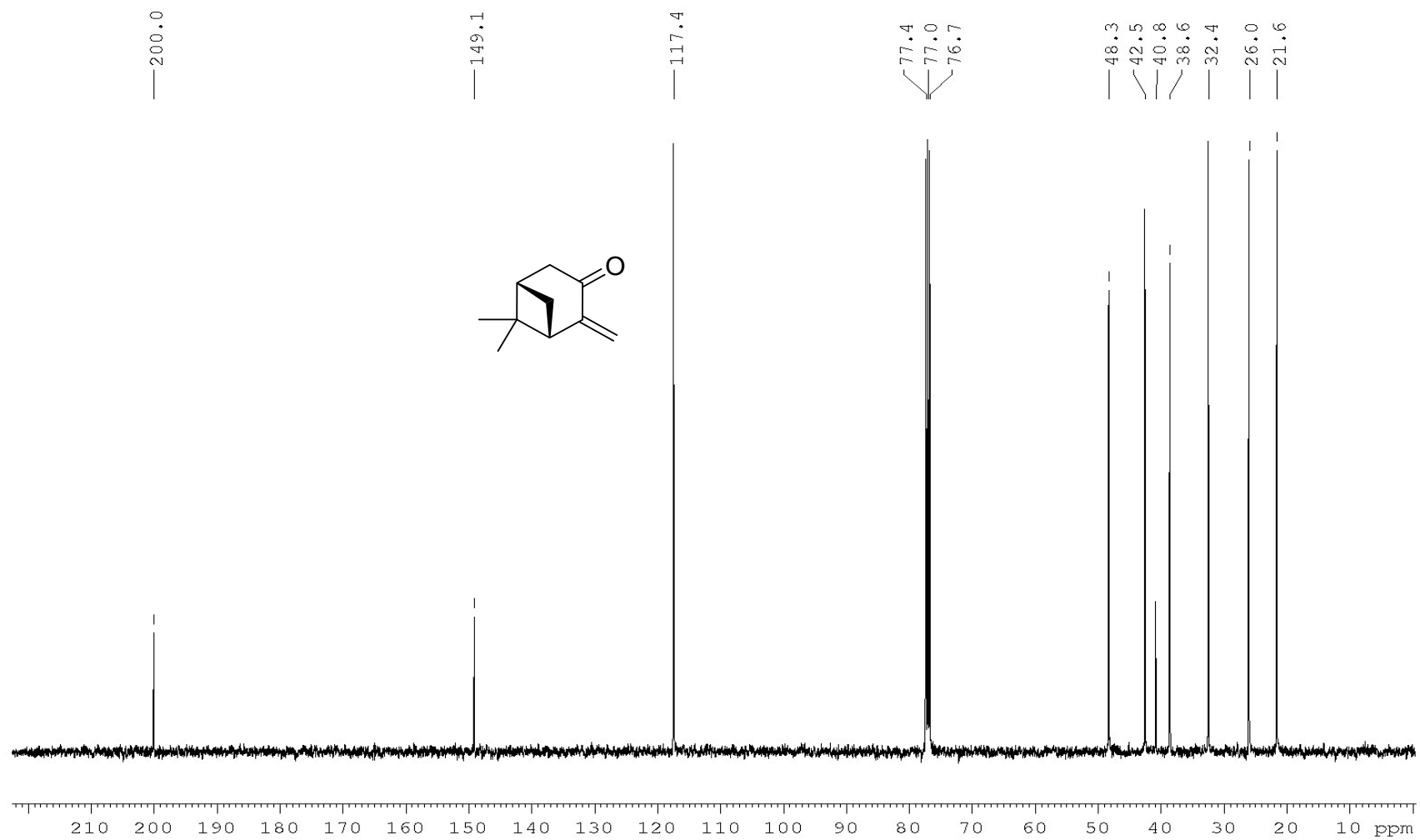
Figure S8 - Flow setup 3 for the synthesis of the trans-pinocarveols (3a-b)..

5. NMR Spectra



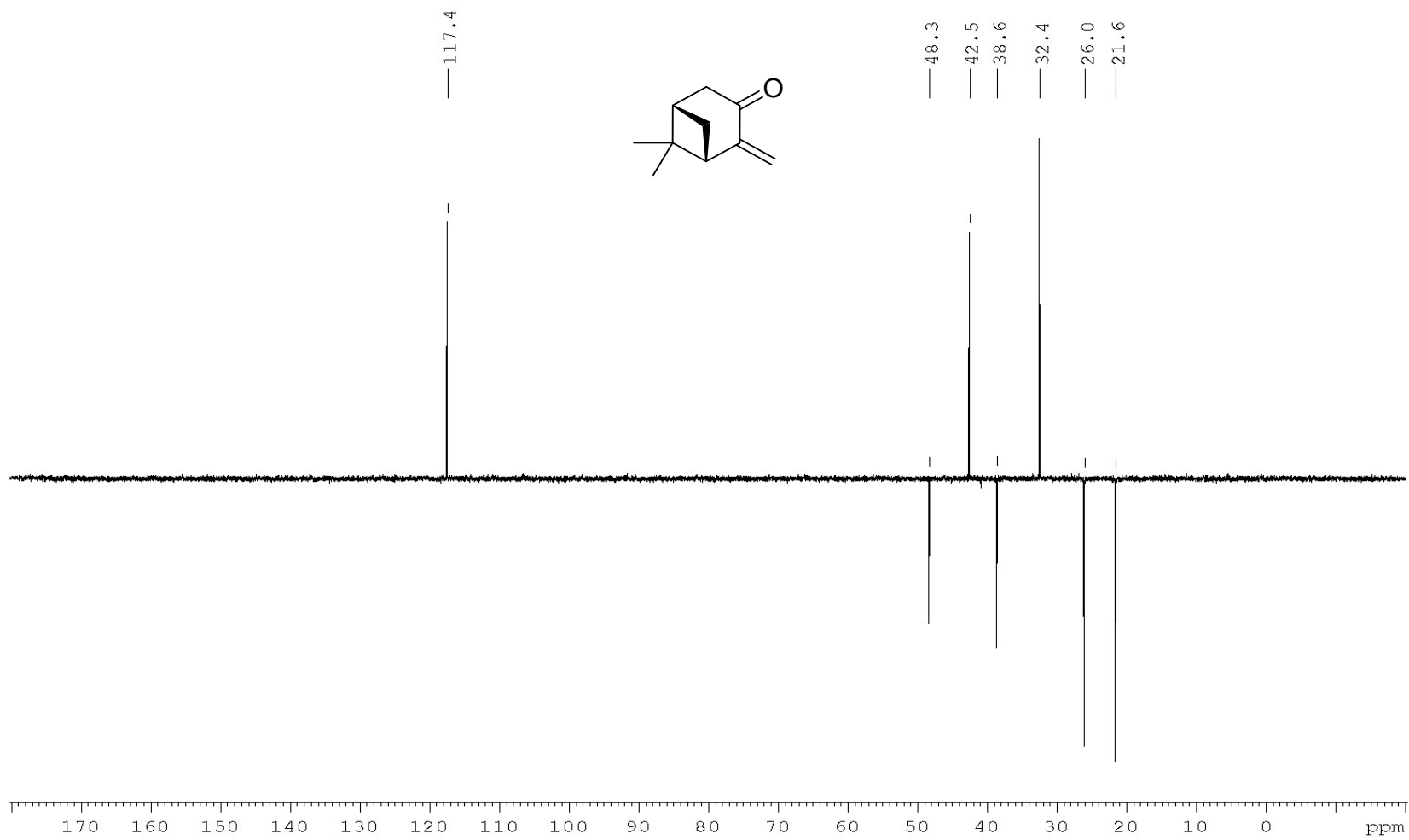
^1H NMR (CDCl_3) – 400 MHz

S11

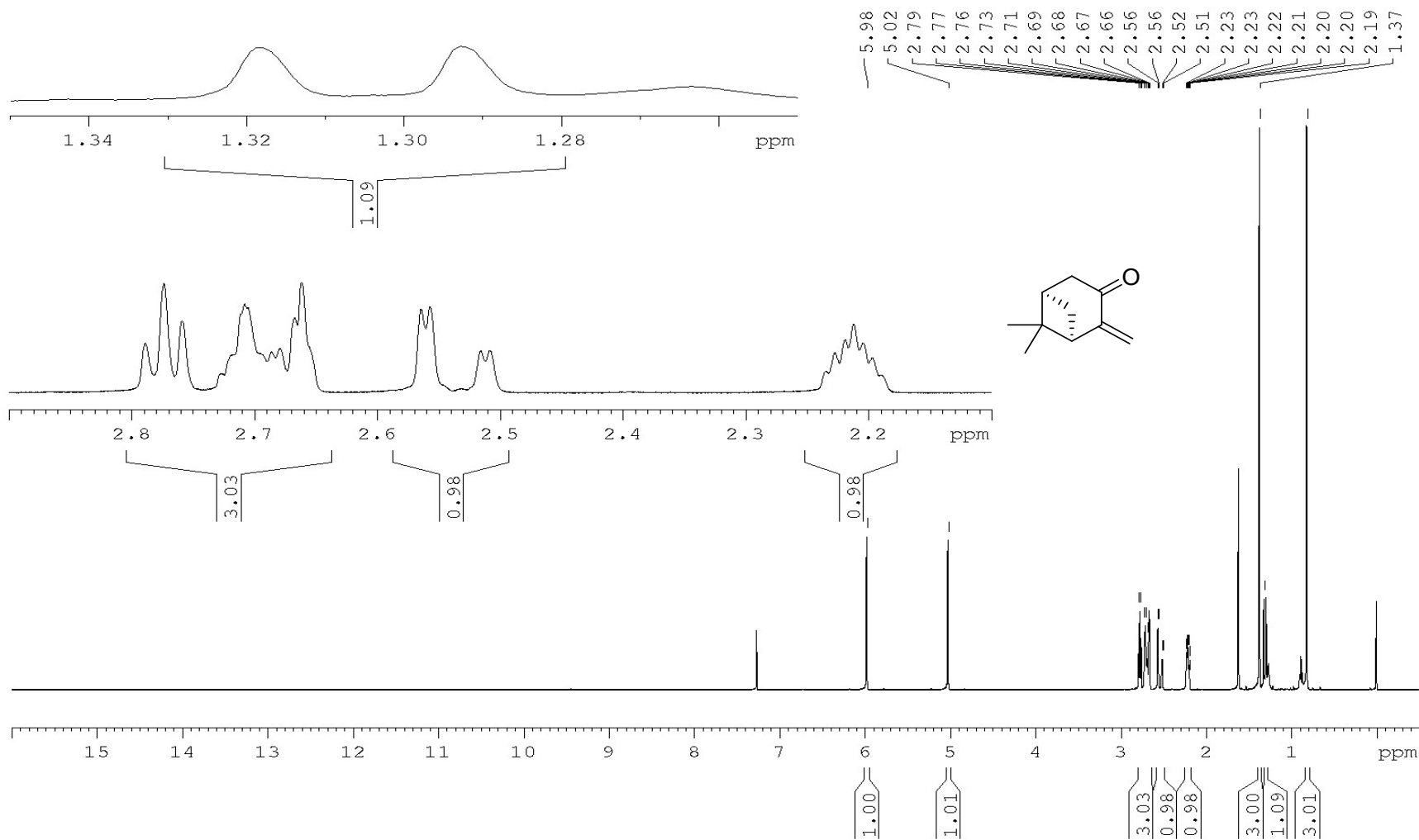


$^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl₃) – 100 MHz

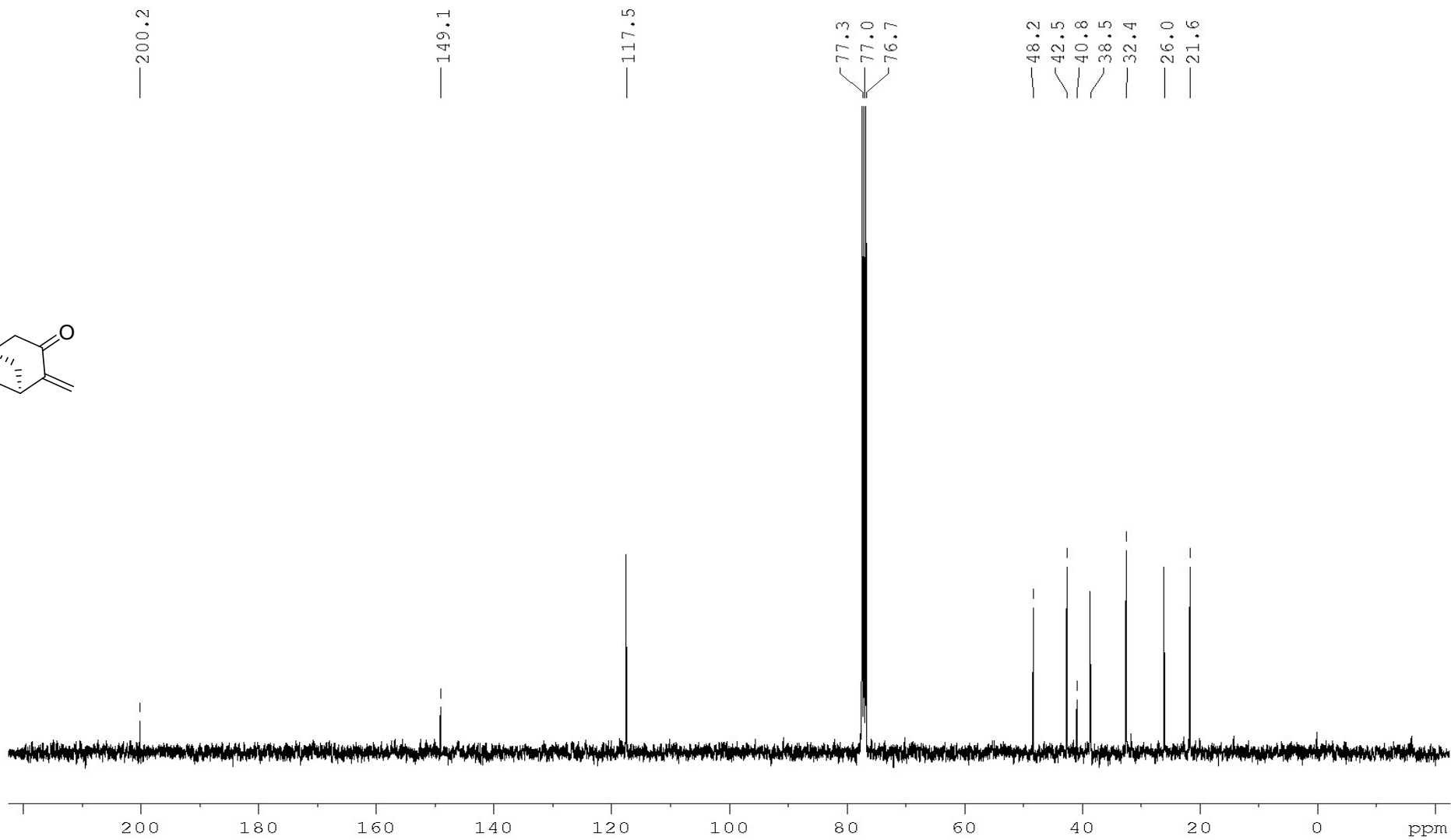
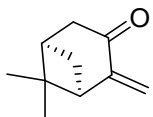
S12



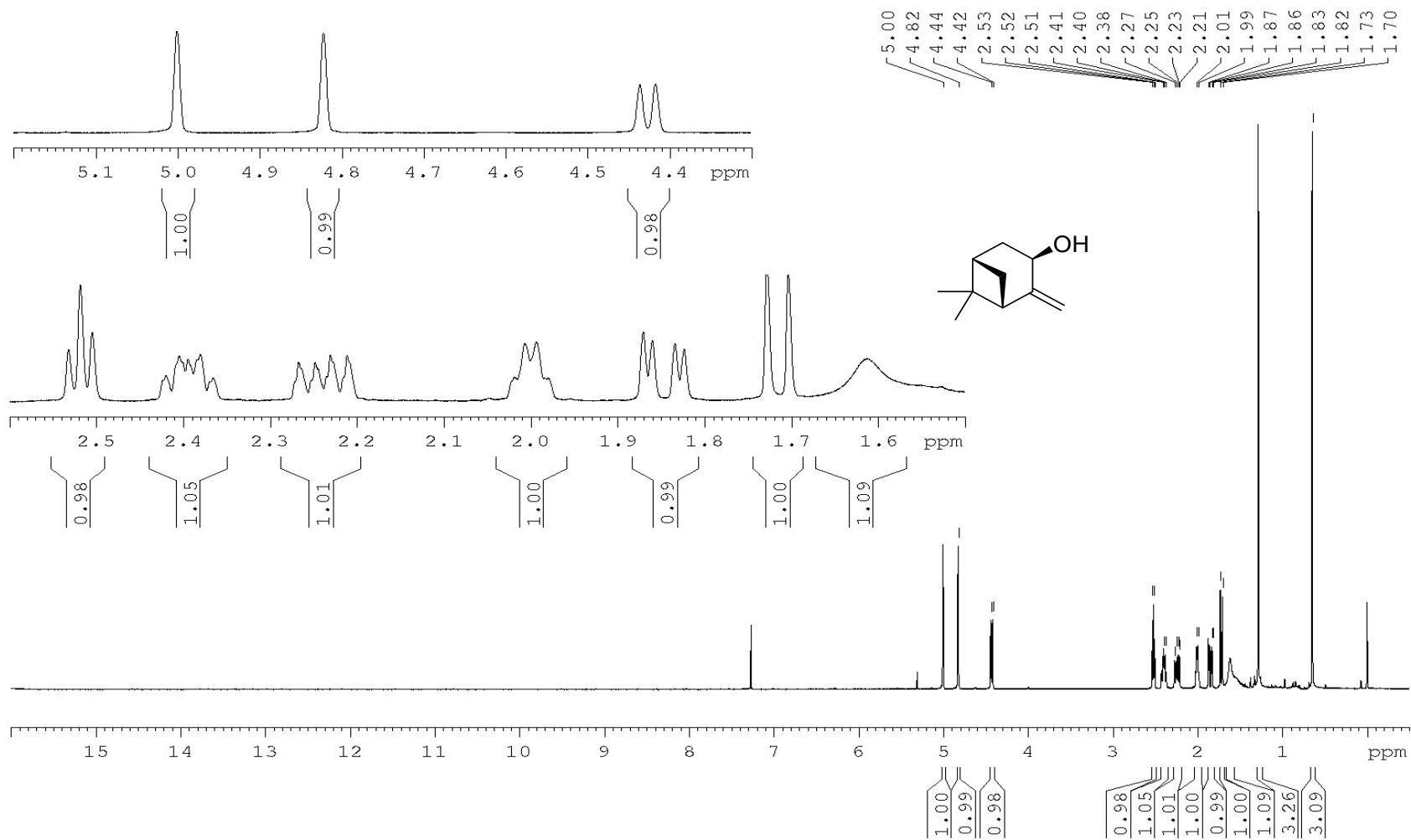
DEPT-135 ¹³C NMR (CDCl₃) – 100 MHz



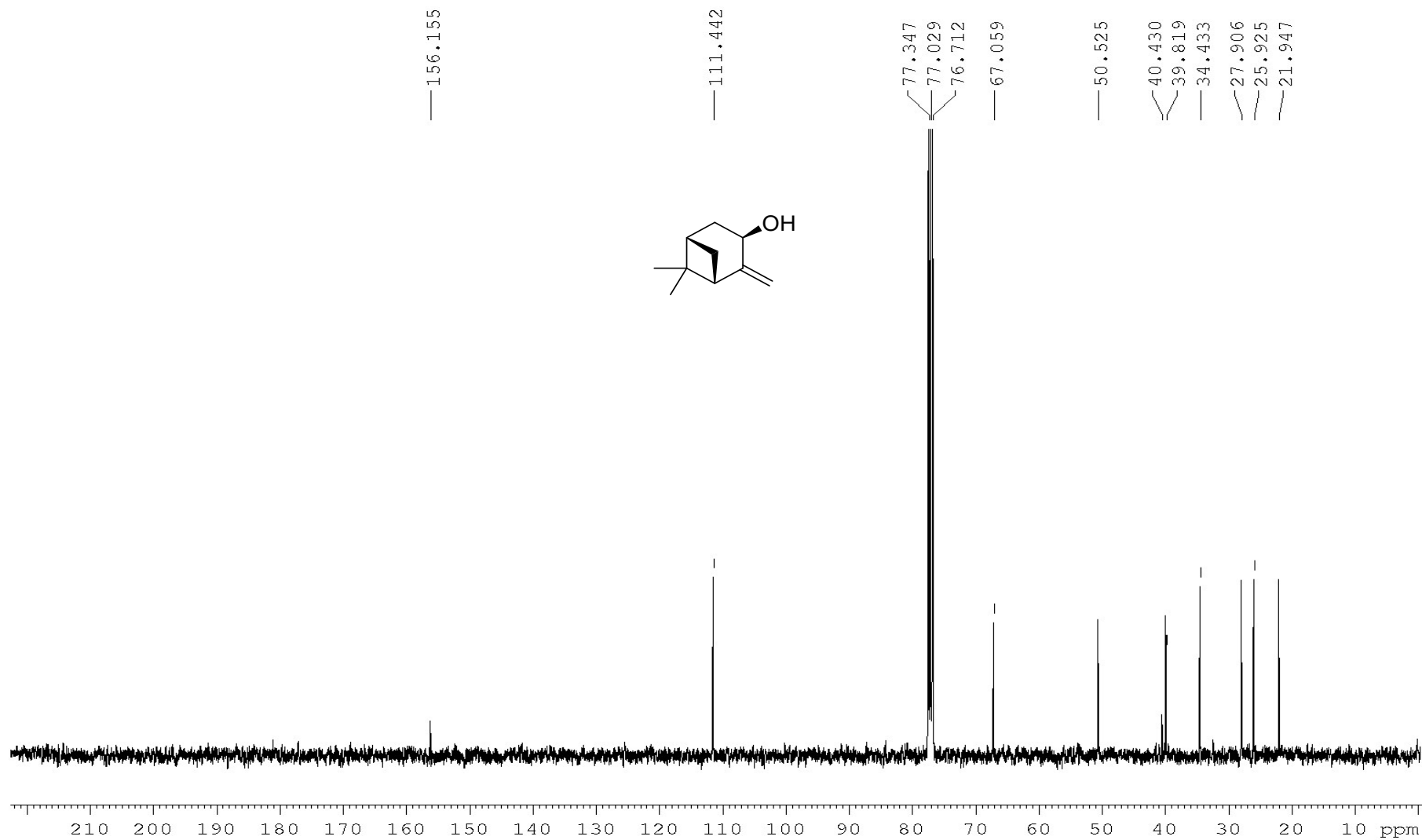
^1H NMR (CDCl_3) – 400 MHz



$^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3) – 100 MHz

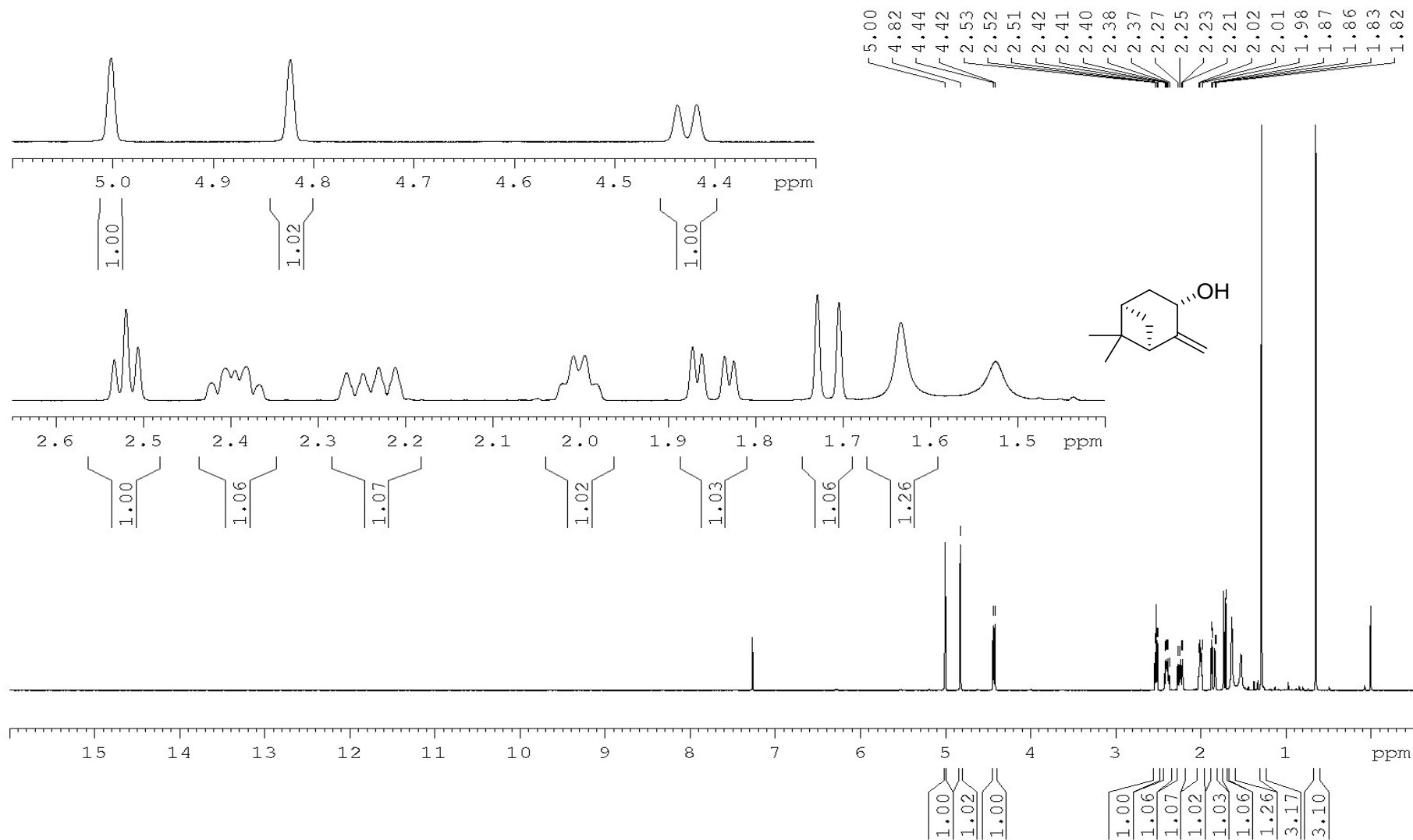


¹H NMR (CDCl₃) – 400 MHz
S17

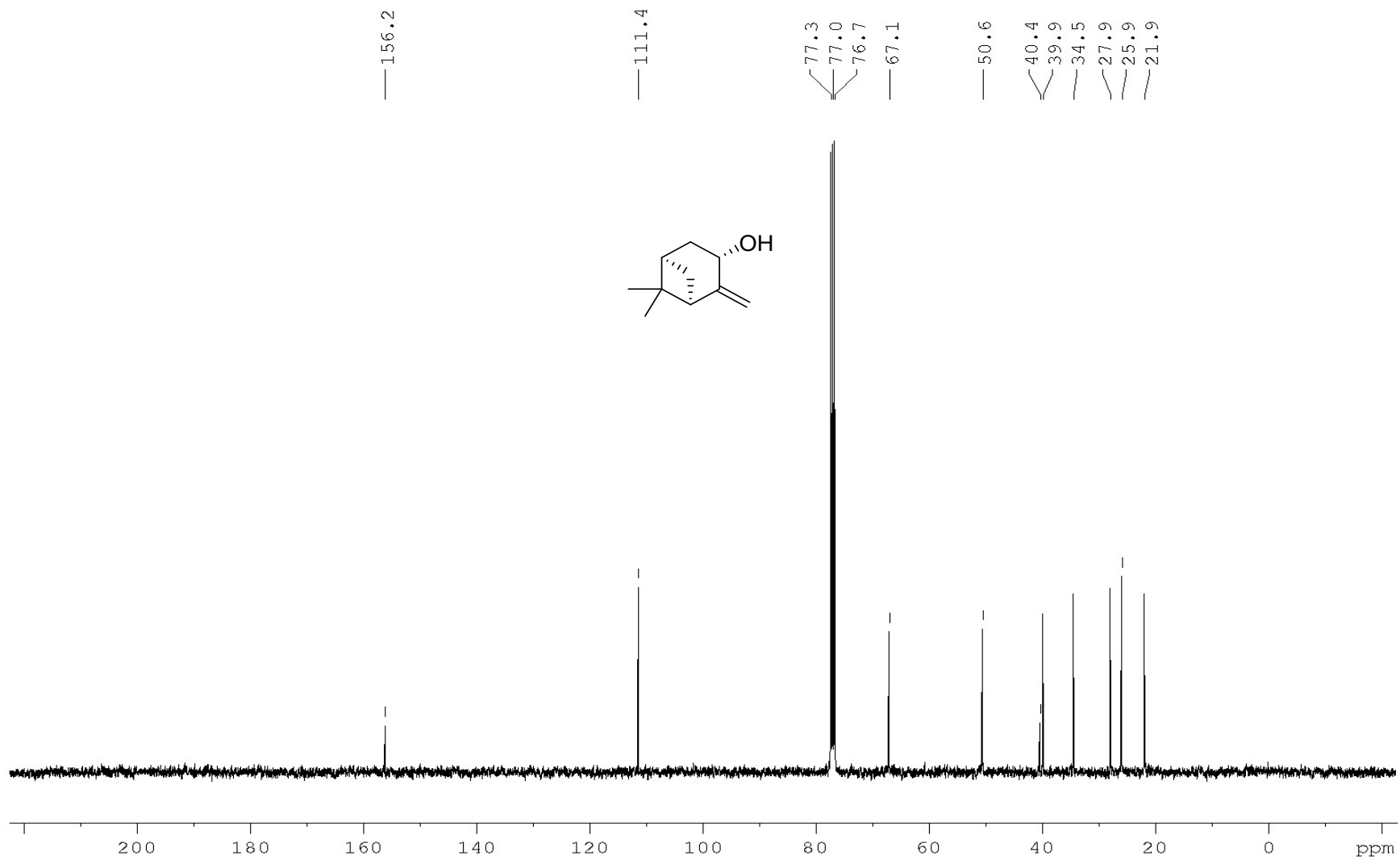


$^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl₃) – 100 MHz

S18



¹H NMR (CDCl₃) – 400 MHz



$^{13}\text{C}\{^1\text{H}\}$ NMR (CDCl_3) – 100 MHz

S20

6. References

- 1 J. M. De Souza, T. J. Brocksom, D. T. McQuade and K. T. De Oliveira, *J. Org. Chem.*, 2018, **83**, 7574–7585.
- 2 A. A. N. De Souza, N. S. Silva, A. V Müller, A. S. Polo, T. J. Brocksom and K. T. De Oliveira, 2018, **83**, 15077–15086.
- 3 P. B. Cranwell, M. O’Brien, D. L. Browne, P. Koos, A. Polyzos, M. Peña-López and S. V. Ley, *Org. Biomol. Chem.*, 2012, **10**, 5774–5779.