## -Supplemental Information-

# Flexible organic frameworks sequester neuromuscular blocking agents in vitro and reverse neuromuscular block in vivo 

Yan Wu, ${ }^{\text {a }}$ Yue-Yang Liu, ${ }^{\text {a }}$ Hong-Kun Liu, ${ }^{\text {a }}$ Shang-Bo Yu, ${ }^{\text {b }}$ Furong Lin, ${ }^{\text {b }}$ Wei Zhou, ${ }^{* a}$ Hui Wang, ${ }^{a}$ DanWei Zhang,*a Zhan-Ting Li,*a and Da Ma,*c
${ }^{\text {a }}$ Department of Chemistry, Shanghai Key Laboratory of Molecular Catalysis and Innovative Materials, Fudan University, Shanghai 200438, China.
${ }^{\mathrm{b}}$ Key Laboratory of Synthetic and Self-Assembly Chemistry for Organic Functional Molecules, Shanghai Institute of Organic Chemistry (SIOC), Chinese Academy of Sciences, 345 Lingling Lu, Shanghai 200032, China.
${ }^{\text {c School of Pharmaceutical and Materials Engineering \& Institute for Advanced Studies, Taizhou }}$ University, 1139 Shifu Avenue, Jiaojiang, Zhejiang 318000, China.

Email: zhouw@fudan.edu.cn,zhangdw@fudan.edu.cn,ztli@fudan.edu.cn,dama@fudan.edu.cn

## Table of Contents

General experimental details ..... S3
Synthetic procedures and characterization data ..... S5
${ }^{1} \mathrm{H}$ NMR spectra of formation of FOF-SS2-3 ..... S15
DLS profile of FOF-SS2-3 ..... S17
${ }^{1}$ H NMR spectra of FOF-SS-1-3 and NMBAs ..... S18
ITC data for FOF-SS1-3 and NMBAs ..... S21
${ }^{1} \mathrm{H}$ NMR spectra of FOF-SS3 and Ach ..... S22
Concentration-absorption standard curve of Cis ..... S22
${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra of new compounds ..... S23
References ..... S30

## General experimental details

## General methods and materials

Commercial reagents were used without further purification. ${ }^{1} \mathrm{H}$ NMR and ${ }^{13} \mathrm{C}$ NMR spectra were recorded with a Bruker AVANCE III HD 400 MHz spectrometer ( 100 MHz ) in the indicated solvents at 298 K. Chemical shifts were referenced to the residual solvent peaks. Dynamic light scattering (DLS) and zeta potential experiments were conducted on a Malvern Zetasizer Nano ZS. Isothermal calorimetric experiments were conducted on a Nano ITC-TA instrument (ITC200). Quantitative analysis of NMBAs was measured by reversed-phase Agilent 1260 High Performance Liquid Chromatography. The melting point of unknown compound was determined by Mettler Toledo melting point meter.

## Isothermal titration calorimetry (ITC) experiments

The FOF-SS1-3 and NMBAs solutions were prepared in PBS (pH 7.4). The concentration of FOF-SS13 was calculated based on that of precursors T1-T3. An aqueous solution of FOF-SS1-3 ( 0.1 mM , while 0.2 mM for FOF-SS1) was placed in the sample cell ( 300 uL ). Solution of NMBAs $(0.8 \mathrm{mM}$ while 1.2 mM for Vec) was dropped in the order of the first injections ( $0.5 \mu \mathrm{~L}$ ) followed by 19 injections $(2 \mu \mathrm{~L})$, the heat evolved was recorded at $\mathrm{T}=298 \mathrm{~K}$. The biding constans (curve fitting) were fitted using the MicroCal ITC analyze software.

## Dialysis experiment

FOF-SS-1-3 and Cis were mixed in PBS ( $\mathrm{pH}=7.4$ ), while the former's concentration was $10 \mathrm{mg} / \mathrm{mL}$ and the later's concentration was $1 \mathrm{mg} / \mathrm{ml}$. The control group contained only $1 \mathrm{mg} / \mathrm{mL}$ Cis. The total volume is 1 mL , and they were put into a dialysis bag (cut-off molecular weight: 1000). The dialysis bag was placed in 20 mL PBS and incubated in a shaker at 310 K . Samples were taken every ten minutes for separation by high performance liquid chromatography. The amount of Cis remaining in the dialysis bag was quantitatively analyzed according to the standard curve of Cis ( $\mathrm{A}=30.87 \mathrm{c}-2.11, \mathrm{R}^{2}=$ 0.9996 ).

The dialysis rate was calculated following equation:
$c=\frac{A+2.11}{30.87} \times 100 \% \quad$ Equation 1
Residual Cis (\%) $=1-\frac{20 * c}{1 * 1000} \times 100 \%$

## Equation 2

in which A was the integral area in the HPLC spctra, $c$ was the concentration of Cis in external solution ( $\mu \mathrm{g} / \mathrm{mL}$ ).

## Cell cytotoxicity

Cytotoxicity was evaluated by cell counting kit-8 (CCK-8) assay. H9C2 cells and L02 cells were seeded in 96 -well plate ( $8 \times 10^{3}$ cells and $100 \mu \mathrm{~L}$ culture medium in per well) and were incubated for 24 hours at 310 K containing $5 \% \mathrm{CO}_{2}$. Then the culture medium was removed and $100 \mu \mathrm{~L}$ culture medium containing FOF-SS1-3 at different concentration was added ( $4-512 \mu \mathrm{~g} / \mathrm{mL}$ ). The culture was removed after 24 h and $100 \mu \mathrm{~L}$ fresh culture medium containing $10 \%$ CCK- 8 was added. The cells were incubated for another 1 hour and the cell viability was evaluated by absorbance at 450 nm using Allsheng AMR-100 microplate reader.

## Hemolysis experiment

$5 \%$ human and mouse red blood cells preserved in Alsever's solution were centrifuged in a $1000 \mathrm{r} / \mathrm{min}$ centrifuge for ten minutes to abtain red blood cells. Cells were diluted with equal volume of isotonic saline. The mixed solution ( $140 \mu \mathrm{~L}$ ) were mixed with saline ( $560 \mu \mathrm{~L}$, negative control), deionized water ( $560 \mu \mathrm{~L}$, positive control), or FOF-SS1-3 solution at different concentrations ( $4-512 \mu \mathrm{~g} / \mathrm{mL}$ ). After incubation for 1 h at 310 K , the samples were centrifugated at $3000 \mathrm{r} / \mathrm{min}$ for 10 minutes at 277 K and
the supernatants were obtained, and their hemolysis ratio was evaluated by absorbance at 545 nm using Biotek Synergy H1 microplate reader. The hemolysis rate was calculated following equation:
Hemolysis rate (\%) $=\frac{A_{s}-A_{n}}{A_{p}-A_{n}} \times 100 \% \quad$ Equation 3
in which $A_{s}, A_{n}$ and $A_{p}$ are the absorbances at 545 nm of the samples, negative control and positive control.

## In vivo experiment

Ethical statement: the animal experiments were carried out based on protocols approved by ethical board of Fudan University.
General surgical procedures: A total of 144 Sprague-Dawley rats whose weights ranged from 180 to 280 g were used ( $n=6$, contained 3 male and 3 female rats). Surgical rats were anesthetized with 5\% isoflurane and then maintained anesthesia with $1 \%$ isoflurane while the flow rate of isoflurane was controlled at $0.6 \mathrm{~L} / \mathrm{min}$. Endotracheal intubation was conducted for mechanical ventilation before injection of NMBAs. Constant airflow containing $1 \%$ isoflurane was pumped to maintain anesthesia to eliminate stress response. Rats were positioned supine on a thermostatic heating plate to maintain the rectal temperature at $37.0 \pm 1{ }^{\circ} \mathrm{C} . .^{17,18}$
Drug administration: Cis ( $0.6 \mathrm{mg} / \mathrm{kg}$, two times ED90), Roc ( $3.5 \mathrm{mg} / \mathrm{kg}$, two times ED90), Vec ( 0.7 $\mathrm{mg} / \mathrm{kg}$, two times ED90), neostigmine/glycopyrrolate ( $0.06 / 0.012,0.12 / 0.024$ and $0.24 / 0.048 \mathrm{mg} / \mathrm{kg}$ ) were diluted in 0.5 ml water. FOF-SS1-3 $(40 \mathrm{mg} / \mathrm{kg}, 80 \mathrm{mg} / \mathrm{kg}$ and $120 \mathrm{mg} / \mathrm{kg})$ were diluted in 0.5 ml $5 \%$ glucose.
Experimental Protocol: The right leg was shaved and pasted on electrode patches to stimulate the rat quadriceps femories muscle. The twitch response of the quadriceps muscle was measured by acceleration sensor of the Slgo TOF-Watch Monitor. The transducer was fixed on the tibia of the right leg where the swing amplitude is the largest, which is convenient for accurate measurement of acceleration. The TOF Watch was switched to cal mode, we stimulated the femoral nerve continuously at 1 Hz with 10 mA current until twitch height reached a stable state. Then the TOF Watch was switched to the train-of-four (TOF) mode, and continued to stimulate the femoral nerve at 2 Hz with a 15 s-internal until the T1 stabilized while TOF was $>90 \%$. Rats were intubated and mechanical ventilation with the tidal volume of 8.0 mL at 80 Hz . Corresponding doses of NMBAs was injected into rats through the tail vein. After complete NMB was induced, either placebo ( 0.5 mL ), mentioned dose of neostigmine/glycopyrrolate and FOF-SS1-3 were administered at maximum twitch depression (T1 = 0 ). The time of TOF-recovery ( $>0.9$ ) was measureded to evaluate the effect of reversing NMB in vivo.

Synthetic Procedures




FOF-SS1

Scheme S1. Synthesis of T1 and FOF-SS1.

Boc-Cys(Trt)-Asp-(OMe)2: To a solution of Boc-Cys-Trt (9.28g, 20 mmol ) in $100 \mathrm{~mL} \mathrm{CH}_{2} \mathrm{Cl}_{2}, \mathrm{HOBt}$ $(2.97 \mathrm{~g}, 22 \mathrm{mmol})$, EDCI $(4.22 \mathrm{~g}, 22 \mathrm{mmol})$ and Asp-(OMe) $2(3.95 \mathrm{~g}, 20 \mathrm{mmol})$ were dissolved. The pH was adjusted to 9 with N -methylmorpholine. The mixture was stirred at room temperature for 12 hours. Then the mixture was washed with saturated NaCl solution ( $50 \mathrm{~mL} \times 2$ ) and $5 \% \mathrm{KHSO}_{4}(50 \mathrm{~mL} \times 2)$. The organic phase was separated, and anhydrous sodium sulfate was added to remove water. The crude product was separated by column chromatography (EA: Hexane $=1: 2$ ) and the Boc-Cys(Trt)-Asp(OMe) 2 was obtained as white solid ( $10.70 \mathrm{~g}, 88 \%$ ). M.P. $95^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO- $d_{6}$ ): $\delta$ $8.17(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 1 \mathrm{H}), 7.40-7.20(\mathrm{~m}, 15 \mathrm{H}), 7.00(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 1 \mathrm{H}), 4.6(\mathrm{q}, J=6.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.92(\mathrm{q}, J$ $=7.5 \mathrm{~Hz}, 1 \mathrm{H}), 3.56(\mathrm{~s}, 3 \mathrm{H}), 3.54(\mathrm{~s}, 3 \mathrm{H}), 2.72(\mathrm{qd}, J=16.6,6.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.37-2.28(\mathrm{~m}, 2 \mathrm{H}), 1.38(\mathrm{~s}$, $9 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 100 MHz, DMSO- $d_{6}$ ): $\delta 171.18,170.85,170.55,155.36,144.77,129.55,128.51$, $127.23,78.96,66.38,53.76,52.63,52.08,48.87,35.87,34.17,28.58$. HR-MS (ESI): calcd for $\mathrm{C}_{33} \mathrm{H}_{38} \mathrm{~N}_{2} \mathrm{O}_{7} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}: 607.2472$,found 607.2462 .
Cys(Trt)-Asp-(OMe)2: Boc-Cys(Trt)-Asp-(OMe)2 $(2.428 \mathrm{~g}, 4 \mathrm{mmol})$ was dissolved in 20 mL formic acid, and stirred at room temperature for 4 h . Then the solution was concentrated by rotary evaporation. The residual oil mixture was dissolved in $20 \mathrm{mLCH}_{2} \mathrm{Cl}_{2}$, and washed with saturated $\mathrm{NaHCO}_{3}$ solution to neutralize the residual formic acid until the pH approached to 7 . The organic phase was separated, and anhydrous sodium sulfate was added to remove water. $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ was removed by rotary evaporation to obtain the Cys(Trt)-Asp-(OMe) $)_{2}$ as an oil. $(1.93 \mathrm{~g}, 96 \%) .{ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right): 87.87(\mathrm{~d}, J=$ $8.5 \mathrm{~Hz}, 1 \mathrm{H}), 7.47-7.41(\mathrm{~m}, 6 \mathrm{H}), 7.32-7.26(\mathrm{~m}, 6 \mathrm{H}), 7.21(\mathrm{t}, J=7.3 \mathrm{~Hz}, 3 \mathrm{H}), 4.76(\mathrm{dt}, J=8.9,4.7 \mathrm{~Hz}$, $1 \mathrm{H}), 3.03-2.93(\mathrm{~m}, 2 \mathrm{H}), 2.78(\mathrm{dd}, J=17.0,4.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.69-2.57(\mathrm{~m}, 2 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , $\left.\mathrm{CDCl}_{3}\right): \delta 173.04,171.32,171.14,144.67,129.71,128.11,126.94,67.08,53.96,52.84,52.10,48.36$, 37.45, 36.24. HR-MS (ESI): calcd for $\mathrm{C}_{28} \mathrm{H}_{30} \mathrm{~N}_{2} \mathrm{O}_{5} \mathrm{~S}[\mathrm{M}+\mathrm{H}]^{+}: 507.1948$, found 507.1943.

T1-(Trt)4-(OMe)8: To a solution of TCPM $(0.248 \mathrm{~g}, 0.5 \mathrm{mmol})$ in $12 \mathrm{~mL} \mathrm{SOCl}{ }_{2}$ was added two drops of DMF. The mixture was heated to reflux for 4 hours and cooled to room temperature, $\mathrm{SOCl}_{2}$ was removed by rotary evaporation. Cys(Trt)-Asp-(OMe) $2(1.115 \mathrm{~g}, 2.2 \mathrm{mmol})$ dissolved in 40 mL dry $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and dry triethylamine ( $0.606 \mathrm{~g}, 6 \mathrm{mmol}$ ) was added and the resulting mixture was stirred at room temperature under $\mathrm{N}_{2}$ atmosphere for 2 hours. Then the mixture was washed with saturated NaCl solution ( $30 \mathrm{~mL} \times 3$ ). The organic phase was separated, and anhydrous sodium sulfate was added to remove water. The crude product was separated by column chromatography $\left(\mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ : methanol $\left.=50: 1\right)$ and recrystallized in acetonitrile to obtain $\mathrm{T} 1-(\mathrm{Trt})_{4}-(\mathrm{OMe})_{8}$ as light yellow solid $(0.686 \mathrm{~g}, 56 \%)$. M.P. $205{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ): $\delta 8.65$ (d, $J=8.2 \mathrm{~Hz}, 4 \mathrm{H}$ ), 8.29 (d, $J=8.0 \mathrm{~Hz}, 4 \mathrm{H}$ ), 7.83 (d, $J$ $=8.5 \mathrm{~Hz}, 8 \mathrm{H}), 7.41(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 8 \mathrm{H}), 7.35-7.17(\mathrm{~m}, 60 \mathrm{H}), 4.59(\mathrm{q}, J=6.6 \mathrm{~Hz}, 4 \mathrm{H}), 4.51-4.43(\mathrm{~m}$, $4 \mathrm{H}), 3.52(\mathrm{~s}, 12 \mathrm{H}), 3.48(\mathrm{~s}, 12 \mathrm{H}), 2.69(\mathrm{ddd}, J=53.1,16.5,6.6 \mathrm{~Hz}, 8 \mathrm{H}), 2.56-2.46(\mathrm{~m}, 8 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO- $d_{6}$ ): $\delta 169.56,169.22,168.58,164.53,147.86,143.14,130.44,128.86,127.94$, $126.89,126.41,125.60,64.81,63.86,51.14,51.01,50.42,47.32,34.18,32.30$. HR-MS (ESI): calcd for $\mathrm{C}_{141} \mathrm{H}_{132} \mathrm{~N}_{8} \mathrm{O}_{42} \mathrm{~S}_{4}[\mathrm{M}+\mathrm{H}]^{+}: 2449.8310$, found 2449.8388 .
T1-(OMe)s: To a solution of T1-(Trt) $)_{4}(\mathrm{OMe})_{8}(0.32 \mathrm{~g}, 0.13 \mathrm{mmol})$ in 8 mL trifluoroacetic acid was added triethylsilane ( $166 \mu \mathrm{~L}, 1.044 \mathrm{mmol})$. The mixture was stirred at room temperature for 4 hours. Then the insoluble by-product triphenylmethane was filtered out. Hexane ( $5 \mathrm{~mL} \times 2$ ) was added to the filtrate for removing the slightly dissolved triphenylmethane in trifluoroacetic acid. The trifluoroacetic acid was removed by rotary evaporation to obtain $\mathrm{T} 1-(\mathrm{OMe})_{8}(0.149 \mathrm{~g}$, yield $79 \%)$. M.P. $137{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ): $\delta 8.51(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 8 \mathrm{H}), 7.85(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 8 \mathrm{H}), 7.41(\mathrm{~d}, J=8.5 \mathrm{~Hz}$, $8 \mathrm{H}), 4.66(\mathrm{q}, J=6.9 \mathrm{~Hz}, 4 \mathrm{H}), 4.55(\mathrm{td}, J=8.7,4.6 \mathrm{~Hz}, 4 \mathrm{H}), 3.61$ (s, 12H), 3.57 (s, 12H), 2.97-2.69 (m, 16 H ), 2.37 ( $\mathrm{t}, J=8.5 \mathrm{~Hz}, 4 \mathrm{H}$ ). ${ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO- $d_{6}$ ): $\delta$ 170.91, 170.38, 170.04, 166.13, $148.91,131.85,130.00,127.51,64.98,55.86,52.24,51.67,48.56,35.43,25.85$. HR-MS (ESI): calcd for $\mathrm{C}_{65} \mathrm{H}_{76} \mathrm{~N}_{8} \mathrm{O}_{24} \mathrm{~S}_{4},[\mathrm{M}-\mathrm{H}]:$ : 1479.3772 ,found 1479.3776 .
T1: To a solution of T1-(OMe) $8(0.149 \mathrm{~g}, 0.1 \mathrm{mmol})$ in 3 mL methanol and 3 mL deionized water was added sodium hydroxide $(0.083 \mathrm{~g}, 2.08 \mathrm{mmol})$. The resulting mixture was stirred at room temperature for 12 hours under $\mathrm{N}_{2}$. Then methanol was removed by rotary evaporation, and 5 mL deionized water was added. The mixed solution was filtered to obtain light yellow solution. 2 M HCl solution was gradually added to the resulting filtrate until the pH approached to $3 \sim 4$. Precipitate was filtered and dried, and the target molecule $\mathbf{T 1}$ was obtained as light yellow solid ( $0.123 \mathrm{~g}, 91 \%$ ). M.P. $>300^{\circ} \mathrm{C}$
(decomposed). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz , DMSO- $\mathrm{d}_{6}$ ): $\delta 12.72$ ( $\mathrm{s}, 8 \mathrm{H}$ ), 8.50 (d, $J=8.2 \mathrm{~Hz}, 4 \mathrm{H}$ ), 8.31 (d, $J=$ $7.9 \mathrm{~Hz}, 4 \mathrm{H}), 7.85(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 8 \mathrm{H}), 7.40(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 8 \mathrm{H}), 4.59-4.50(\mathrm{~m}, 8 \mathrm{H}), 2.97-2.75$ (m, 8H), $2.64(\mathrm{qd}, J=16.8,6.4 \mathrm{~Hz}, 8 \mathrm{H}), 2.37(\mathrm{t}, J=8.5 \mathrm{~Hz}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO- $d_{6}$ ): $\delta 172.65$, 172.13, 170.32, 166.68, 149.33, 132.48, 130.54, 127.95, 65.40, 56.37, 49.14, 36.44, 26.54. HR-MS (ESI): calcd for $\mathrm{C}_{57} \mathrm{H}_{60} \mathrm{~N}_{8} \mathrm{O}_{24} \mathrm{~S}_{4},[\mathrm{M}-\mathrm{H}]:$ : 1367.2520 , found 1367.2512.
FOF-SS1: To a solution of compound $\mathbf{T 1}(0.137 \mathrm{~g}, 0.1 \mathrm{mmol})$ in 8.6 mL deionized water was added NaOH solution $(0.8 \mathrm{~mL}, 1 \mathrm{M})$, NaI solution $(0.2 \mathrm{~mL}, 0.1 \mathrm{M}), \mathrm{H}_{2} \mathrm{O}_{2}$ solution ( $0.4 \mathrm{~mL}, 1 \mathrm{M}$ ). The mixture was stirred at room temperature for 1 hour. The mixture was transferred to dialysis bags to remove excess $\mathrm{H}_{2} \mathrm{O}_{2}$ and inorganic salt. FOF-SS1 was obtained by rotary evaporation under low temperature as light yellow solid ( $0.141 \mathrm{~g}, 91 \%$ ).


Fig. S1. ${ }^{1} \mathrm{H}$ NMR spectrum ( 400 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of compound T1.


Fig. S2. ${ }^{13} \mathrm{C}$ NMR spectrum ( 100 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of compound T1.


Fig. S3. HR-MS (ESI) spectra of compound T1.



Scheme S2. Synthesis of T2 and FOF-SS2.
TIPM, TMTBA-OMe and TMTBA were prepared according to the literature procedures. ${ }^{[1]}$
T2-(Trt)4-(OMe)8 was prepared as a light yellow solid in $35 \%$ yield using the procedures described for T1-(Trt)4-(OMe) 8. M.P. $233 ~^{\circ}$ C. ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 8.55$ (d, $\left.J=7.9 \mathrm{~Hz}, 4 \mathrm{H}\right), 8.44(\mathrm{~d}, J=$ $8.1 \mathrm{~Hz}, 4 \mathrm{H}), 7.54(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 8 \mathrm{H}), 7.41(\mathrm{~d}, J=15.6 \mathrm{~Hz}, 4 \mathrm{H}), 7.38-7.01(\mathrm{~m}, 76 \mathrm{H}), 6.76(\mathrm{~d}, J=15.8 \mathrm{~Hz}$, $4 \mathrm{H}), 4.64-4.52(\mathrm{~m}, 8 \mathrm{H}), 3.55(\mathrm{~s}, 12 \mathrm{H}), 3.54(\mathrm{~s}, 12 \mathrm{H}), 2.85-2.64(\mathrm{~m}, 8 \mathrm{H}), 2.37(\mathrm{~m}, 8 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (100 $\left.\mathrm{MHz}, \mathrm{DMSO}-d_{6}\right): \delta 171.14,170.80,170.15,165.10,147.55,144.72,139.17,133.30,131.33,129.55$, $128.53,127.81,127.24,122.43,66.31,64.91,52.63,52.12,51.75,49.00,35.79,34.63$. HR-MS (ESI): calcd for $\mathrm{C}_{149} \mathrm{H}_{140} \mathrm{~N}_{8} \mathrm{O}_{24} \mathrm{~S}_{4},[\mathrm{M}+2 \mathrm{Na}]^{2+}$ 1299.4324, found 1299.4348 .

T2-(OMe)8 was prepared as a yellow solid yield using the procedures described for T1-(OMe)8. M.P. $144{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR (400 MHz, DMSO- $d_{6}$ ): $\delta 8.63(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 4 \mathrm{H}), 8.38(\mathrm{~d}, J=8.0 \mathrm{~Hz}, 4 \mathrm{H}), 7.55(\mathrm{~d}, J$ $=8.3 \mathrm{~Hz}, 8 \mathrm{H}), 7.43(\mathrm{~d}, J=15.7 \mathrm{~Hz}, 4 \mathrm{H}), 7.26(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 8 \mathrm{H}), 6.78(\mathrm{~d}, J=15.8 \mathrm{~Hz}, 4 \mathrm{H}), 4.66(\mathrm{q}, J=$ $7.0 \mathrm{~Hz}, 4 \mathrm{H}), 4.57(\mathrm{q}, J=7.4 \mathrm{~Hz}, 4 \mathrm{H}), 3.63(\mathrm{~s}, 12 \mathrm{H}), 3.61(\mathrm{~s}, 12 \mathrm{H}), 2.90-2.67(\mathrm{~m}, 16 \mathrm{H}), 2.26(\mathrm{t}, J=8.4$ $\mathrm{Hz}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (100 MHz, DMSO-d ${ }_{6}$ ): $\delta 171.34,170.79,170.25,165.31,147.50,139.03,133.22$, $131.26,127.77,122.42,64.87,55.15,52.70,52.16,49.03,35.79,26.84$. HR-MS (ESI): calcd for $\mathrm{C}_{73} \mathrm{H}_{84} \mathrm{~N}_{8} \mathrm{O}_{24} \mathrm{~S}_{4},[\mathrm{M}+\mathrm{Na}]^{+}$1607.4373, found 1607.4373.

Compound T2 was prepared as a yellow solid yield using the procedures described for T1 (yield 74\% for 2 steps). M.P. $>300^{\circ} \mathrm{C}\left(\right.$ decomposed). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ) $\delta 12.68(\mathrm{~s}, 8 \mathrm{H}), 8.46(\mathrm{~d}, J=7.8$ $\mathrm{Hz}, 4 \mathrm{H}), 8.37(\mathrm{~d}, J=7.9 \mathrm{~Hz}, 4 \mathrm{H}), 7.54(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 8 \mathrm{H}), 7.42(\mathrm{~d}, J=15.6 \mathrm{~Hz}, 4 \mathrm{H}), 7.25(\mathrm{~d}, J=8.2 \mathrm{~Hz}$, $8 \mathrm{H}), 6.78(\mathrm{~d}, J=15.8 \mathrm{~Hz}, 4 \mathrm{H}), 4.61-4.52(\mathrm{~m}, 8 \mathrm{H}), 2.88-2.57(\mathrm{~m}, 16 \mathrm{H}), 2.27(\mathrm{t}, J=8.4 \mathrm{~Hz}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR (100 MHz, DMSO- $d_{6}$ ) $\delta 172.64,172.04,170.15,165.39,147.53,139.01,133.32,131.31,127.81,122.60$, 64.91, 55.30, 49.19, 36.28, 27.07. HR-MS (ESI):calcd for $\mathrm{C}_{65} \mathrm{H}_{68} \mathrm{~N}_{8} \mathrm{O}_{24} \mathrm{~S}_{4}$, $[\mathrm{M}+2 \mathrm{Na}]^{2+} 737.1687$,found 737.1680.

FOF-SS2 was prepared as a yellow solid yield using the procedures described for FOF-SS1 (yield 89\%).


Fig. S4. ${ }^{1} \mathrm{H}$ NMR spectrum ( 400 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of compound $\mathbf{T 2}$.


Fig. S5. ${ }^{13} \mathrm{C}$ NMR spectrum ( 100 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of compound $\mathbf{T} 2$.


Fig. S6. HR-MS (ESI) spectra of compound T2.




Scheme S3. Synthesis of T3 and FOF-SS3.
TMTBC-OMe and TMTBC were prepared according to the literature procedures. ${ }^{[2]}$
T3-(Trt)4-(OMe) $\mathbf{8}_{\mathbf{8}}$ was prepared as a light yellow solid in $52 \%$ yield using the procedures described for T1-(Trt) $)_{4}(\mathrm{OMe})_{8} . \mathrm{M} . \mathrm{P} .198{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{DMSO}_{6}$ ): $\delta 8.74(\mathrm{~d}, J=8.1 \mathrm{~Hz}, 4 \mathrm{H}), 8.35(\mathrm{~d}, J=$ $8.0 \mathrm{~Hz}, 4 \mathrm{H}$ ), 7.99 (d, $J=8.3 \mathrm{~Hz}, 8 \mathrm{H}$ ), $7.84(\mathrm{~d}, J=8.3 \mathrm{~Hz}, 8 \mathrm{H}), 7.79(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 8 \mathrm{H}), 7.47(\mathrm{~d}, J=8.2$ $\mathrm{Hz}, 8 \mathrm{H}), 7.38-7.20(\mathrm{~m}, 60 \mathrm{H}), 4.64(\mathrm{q}, ~ J=6.6 \mathrm{~Hz}, 4 \mathrm{H}), 4.53(\mathrm{q}, J=8.5 \mathrm{~Hz}, 4 \mathrm{H}), 3.55(\mathrm{~s}, 12 \mathrm{H}), 3.54(\mathrm{~s}$, 12 H ), 2.85-2.65 (m, 8H), 2.64-2.53 (m, 8H). ${ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO- $d_{6}$ ): $\delta$ 171.18, 170.83, $170.25,166.26,146.51,144.75,142.73,137.20,132.99,131.48,129.53,128.74,128.49,127.20,126.97$, 126.80, 66.41, 64.37, 52.80, 52.61, 52.05, 48.93, 35.80, 33.91. HR-MS (ESI): calcd for $\mathrm{C}_{165} \mathrm{H}_{148} \mathrm{~N}_{8} \mathrm{O}_{24} \mathrm{~S}_{4}$,, $[\mathrm{M}+2 \mathrm{Na}]^{2+}$ 1399.4637, found 1399.4681.

T3-(OMe)8 was prepared as a yellow solid yield using the procedures described for T1-(OMe)8. M.P. $159{ }^{\circ} \mathrm{C} .{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}$ ): $\delta 8.62-8.52(\mathrm{~m}, 8 \mathrm{H}), 8.01(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 8 \mathrm{H}), 7.84(\mathrm{~d}, J=8.4$ $\mathrm{Hz}, 8 \mathrm{H}), 7.80(\mathrm{~d}, J=8.5 \mathrm{~Hz}, 8 \mathrm{H}), 7.46(\mathrm{~d}, J=8.4 \mathrm{~Hz}, 8 \mathrm{H}), 4.69(\mathrm{q}, J=6.7 \mathrm{~Hz}, 4 \mathrm{H}), 4.59(\mathrm{td}, J=8.6,4.9$ $\mathrm{Hz}, 4 \mathrm{H}), 3.63(\mathrm{~s}, 12 \mathrm{H}), 3.60(\mathrm{~s}, 12 \mathrm{H}), 3.02-2.70(\mathrm{~m}, 16 \mathrm{H}), 2.41(\mathrm{t}, J=8.5 \mathrm{~Hz}, 4 \mathrm{H}) .{ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO- $d_{6}$ ): $\delta 171.43,170.89,170.63,166.70,146.54,142.73,137.27,133.23,131.53,128.75,127.00$,
126.83, 64.40, 56.42, 52.74, 52.19, 49.08, 35.95, 26.36. HR-MS (ESI): calcd for $\mathrm{C}_{89} \mathrm{H}_{92} \mathrm{~N}_{8} \mathrm{O}_{24} \mathrm{~S}_{4},[\mathrm{M}+\mathrm{H}]^{+}$ 1785.5180, found 1785.5178 .

Compound T3 was prepared as a light yellow solid yield using the procedures described for $\mathbf{T 1}$ (yield $84 \%$ for 2 steps). M.P. $>300{ }^{\circ} \mathrm{C}$ (decomposed). ${ }^{1} \mathrm{H}$ NMR ( 400 MHz, DMSO- $d_{6}$ ): $\delta 12.67$ ( $\mathrm{s}, 8 \mathrm{H}$ ), $8.58(\mathrm{~d}$, $J=7.4 \mathrm{~Hz}, 4 \mathrm{H}), 8.38(\mathrm{~d}, J=7.8 \mathrm{~Hz}, 4 \mathrm{H}), 8.00(\mathrm{~d}, J=7.5 \mathrm{~Hz}, 8 \mathrm{H}), 7.84-7.78(\mathrm{~m}, 16 \mathrm{H}), 7.45(\mathrm{~d}, J=7.2$ $\mathrm{Hz}, 8 \mathrm{H}), 4.63-4.52(\mathrm{~m}, 8 \mathrm{H}), 3.03-2.78(\mathrm{~m}, 8 \mathrm{H}), 2.66(\mathrm{qd}, J=16.6,6.1 \mathrm{~Hz}, 8 \mathrm{H}), 2.40(\mathrm{t}, J=8.3 \mathrm{~Hz}, 4 \mathrm{H})$. ${ }^{13} \mathrm{C}$ NMR ( 100 MHz , DMSO- $d_{6}$ ): $\delta 172.74,172.17,170.31,166.65,146.48,142.65,137.24,133.24$, 131.49, 128.71, 126.95, 126.78, 64.35, 56.51, 49.11, 36.88, 26.52. HR-MS (ESI): calcd for $\mathrm{C}_{81} \mathrm{H}_{76} \mathrm{~N}_{8} \mathrm{O}_{24} \mathrm{~S}_{4},[\mathrm{M}-2 \mathrm{H}]^{2-}$ 835.1844, found 835.1836.

FOF-SS3 was prepared as a yellow solid yield using the procedures described for FOF-SS1 (yield 92\%).


Fig. S7. ${ }^{1} \mathrm{H}$ NMR spectrum ( 400 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of compound T3.


Fig. S8. ${ }^{13} \mathrm{C}$ NMR spectrum ( 100 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of compound T3.


Fig. S9. HR-MS (ESI) spectra of compound T3.


Fig. S10. ${ }^{1} \mathrm{H}$ NMR spectra ( $400 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}, 298 \mathrm{~K}$ ) of the formation of $\mathbf{F O F - S S 2}$ as oxidant was added.


Fig. S11. ${ }^{1} \mathrm{H}$ NMR spectra ( $400 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}, 298 \mathrm{~K}$ ) of the formation of $\mathbf{F O F - S S 3}$ as oxidant was added.


Fig. S12. ${ }^{1} \mathrm{H}$ NMR spectra ( 400 MHz , DMSO- $d_{6}$, 298 K ) of a) T-1, b) acidulated FOF-SS1.


Fig. S13. ${ }^{1} \mathrm{H}$ NMR spectra ( 400 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of a) T-2, b) acidulated FOF-SS2.


Fig. S14. ${ }^{1} \mathrm{H}$ NMR spectra ( 400 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of a) T-3, b) acidulated FOF-SS3.


Fig. S15. DLS profile of a) FOF-SS2 and b) FOF-SS3 at different concentrations in water (calculated based on [T2]-[T3]).


Fig. S16. ${ }^{1} \mathrm{H}$ NMR spectra ( $400 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}, 298 \mathrm{~K}$ ) of a) FOF-SS1 ( 2 mM ), b) Cis ( 2 mM ) and FOFSS1 (2 mM), and c) Cis ( 2 mM ).


Fig. S17. ${ }^{1} \mathrm{H}$ NMR spectra ( $400 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}, 298 \mathrm{~K}$ ) of a) FOF-SS2 ( 2 mM ), b) Cis ( 2 mM ) and FOFSS2 ( 2 mM ) , and c) Cis ( 2 mM ).


Fig. S18. ${ }^{1} \mathrm{H}$ NMR spectra ( $400 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}, 298 \mathrm{~K}$ ) of a) FOF-SS3 ( 2 mM ), b) Roc ( 2 mM ) and FOFSS3 ( 2 mM ), and c) Roc ( 2 mM ).


Fig. S19. ${ }^{1} \mathrm{H}$ NMR spectra ( $400 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}, 298 \mathrm{~K}$ ) of a) FOF-SS3 ( 2 mM ), b) Vec ( 2 mM ) and FOFSS3 ( 2 mM ), and c) Vec ( 2 mM ).


Fig. S20. ${ }^{1} \mathrm{H}$ NMR spectra ( $400 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}, 298 \mathrm{~K}$ ) of a) FOF-SS3 ( 2 mM ), b) Panc ( 2 mM ) and FOFSS3 ( 2 mM ), and c) Panc ( 2 mM ).


Fig. S21. Isothermal titration thermograms of FOF-SS1 $(300 \mu \mathrm{~L}, 0.2 \mathrm{mM})$ titrated with the aqueous solution of a) Cis ( $40 \mu \mathrm{~L}, 0.8 \mathrm{mM}$ ), b) Roc ( $40 \mu \mathrm{~L}, 0.8 \mathrm{mM}$ ), c) Vec ( $40 \mu \mathrm{~L}, 1.2 \mathrm{mM}$ ), d) Panc ( $40 \mu \mathrm{~L}$, 0.8 mM ) at 298 K .


Fig. S22. Isothermal titration thermograms of FOF-SS2 $(300 \mu \mathrm{~L}, 0.1 \mathrm{mM})$ titrated with the aqueous solution of a) Cis ( $40 \mu \mathrm{~L}, 0.8 \mathrm{mM}$ ), (b) Roc ( $40 \mu \mathrm{~L}, 0.8 \mathrm{mM}$ ), (c) Vec ( $40 \mu \mathrm{~L}, 1.2 \mathrm{mM}$ ), (d) Panc ( 40 $\mu \mathrm{L}, 0.8 \mathrm{mM})$ at 298 K .


Fig. S23. Isothermal titration thermograms of FOF-SS3 ( $300 \mu \mathrm{~L}, 0.1 \mathrm{mM}$ ) titrated with the aqueous solution of a) Roc ( $40 \mu \mathrm{~L}, 0.8 \mathrm{mM})$, b) Vec $(40 \mu \mathrm{~L}, 1.2 \mathrm{mM})$, c) Panc $(40 \mu \mathrm{~L}, 0.8 \mathrm{mM})$ at 298 K .


Fig. S24. ${ }^{1} \mathrm{H}$ NMR spectra ( $400 \mathrm{MHz}, \mathrm{D}_{2} \mathrm{O}, 298 \mathrm{~K}$ ) of a) Cis (2 mM), b) FOF-SS3 (2 mM), Cis (2 $\mathrm{mM})$, Ach $(6 \mathrm{mM})$, c) Ach ( 2 mM ), FOF-SS3 $(2 \mathrm{~m} \mathrm{M})$ and d) Ach ( 2 mM ).
a)


Fig. S25. Concentration-absorption standard curve of Cis, the red line represents the linear fit of the data $\left(\mathrm{R}^{2}=0.9996, \mathrm{~A}=30.867 \mathrm{c}-2.113\right)$.


Fig. S26. ${ }^{1} \mathrm{H}$ NMR spectrum ( 400 MHz , DMSO- $d_{6}$, 298 K ) of Boc-Cys(Trt)-Asp-(OMe) $\mathbf{2}^{2}$


Fig. S27. ${ }^{13} \mathrm{C}$ NMR spectrum ( 100 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of Boc-Cys(Trt)-Asp-(OMe)2.


Fig. S28. ${ }^{1} \mathrm{H}$ NMR spectrum ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, 298 \mathrm{~K}$ ) of Cys(Trt)-Asp-(OMe)2.


Fig. S29. ${ }^{13} \mathrm{C}$ NMR spectrum ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}, 298 \mathrm{~K}$ ) of $\mathrm{Cys}(\mathrm{Trt})$-Asp-(OMe) $)_{2}$.


Fig. S30. ${ }^{1} \mathrm{H}$ NMR spectrum ( 400 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of T1-(Trt) $4-(\mathrm{OMe})_{8}$.


Fig. S31. ${ }^{13} \mathrm{C}$ NMR spectrum ( 100 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of T1-(Trt) $)_{4}$-(OMe) 8 .


Fig. S32. ${ }^{1} \mathrm{H}$ NMR spectrum ( 400 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of T1-(OMe) 8 .

$\stackrel{\bar{\sigma}}{\stackrel{\infty}{\dot{\infty}}}$

$\mathrm{CO}_{2} \mathrm{Me}$





Fig. S33. ${ }^{13} \mathrm{C}$ NMR spectrum ( 100 MHz , DMSO- $d_{6}$, 298 K ) of T1-(OMe) 8 .


Fig. S34. ${ }^{1} \mathrm{H}$ NMR spectrum ( 400 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of T2-(Trt) $)_{4}(\mathrm{OMe})_{8}$.


Fig. S35. ${ }^{13} \mathrm{C}$ NMR spectrum ( 100 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of T2-(Trt) $)_{4}-(\mathrm{OMe})_{8}$.


Fig. S36. ${ }^{1} \mathrm{H}$ NMR spectrum ( 400 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of T2-(OMe) $)_{8}$.


Fig. S37. ${ }^{13} \mathrm{C}$ NMR spectrum ( 100 MHz , DMSO- $d_{6}$, 298 K ) of T2-(OMe) 8 .


Fig. S38. ${ }^{1}$ H NMR spectrum ( 400 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of T3-(Trt) $)_{4}-(\mathrm{OMe})_{8}$.


Fig. S39. ${ }^{13} \mathrm{C}$ NMR spectrum ( 100 MHz , DMSO- $d_{6}, 298 \mathrm{~K}$ ) of T3-(Trt) $)_{4}$-(OMe) 8 .


Fig. S40. ${ }^{1} \mathrm{H}$ NMR spectrum ( $400 \mathrm{MHz}, \mathrm{DMSO}-d_{6}, 298 \mathrm{~K}$ ) of T3-(OMe) $8_{8}$.


Fig. S41. ${ }^{13} \mathrm{C}$ NMR spectrum ( 100 MHz , DMSO- $d_{6}$, 298 K ) of T3-(OMe) 8 .

## References

[1] a) L. Felix, U. Sezer, M. Arndt, M. Mayor, Eur. J. Org. Chem. 2014, 31, 6884-6895; b) D. Liu, Z. Xie, L. Ma, W. Lin, Inorg. Chem. 2010, 49, 9107-9109.
[2] a) L. Ma, A, Jin, Z. Xie, W. Lin, Angew. Chem. Int. Ed. 2009, 48, 9905-9908.

