

## Supporting Information

### Alkynyl Prins Carbocyclization Cascades for the Synthesis of Linear-Fused Heterocyclic Ring Systems

Jackson J. Hernandez, Alison J. Frontier\*

Department of Chemistry, University of Rochester, 414 Hutchison Hall, 100 Trustee Road,  
Rochester, NY 14627-0216 (USA).

\*Corresponding Author: [alison.frontier@rochester.edu](mailto:alison.frontier@rochester.edu)

| Table of Contents   | Page |
|---|------|
| General Remarks   | S2   |
| Starting Material Preparation                             | S3   |
| General Conditions for Carbocyclization                   | S5   |
| Greater than 1.0 mmol Scale Experiments                   | S15  |
| Procedure for Deprotection of <i>N</i> -Tosyl Piperidines | S16  |
| Explanation of Major Byproduct Isolated                   | S17  |
| Unsuccessful Examples                                     | S19  |
| Diversification of Carbocyclization Substrates            | S19  |
| Procedure for Indole Deprotection                         | S24  |
| X-Ray Structures  | S26  |
| NMR Spectra   | S35  |
| Computational Methods                                     | S88  |
| References  | S97  |

### General Remarks:

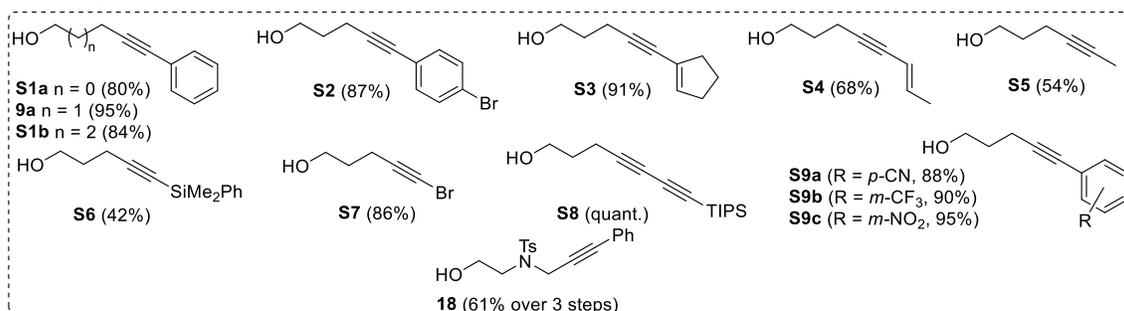
All reactions were carried out under an argon atmosphere in flame-dried glassware with magnetic stirring. Syringe needles used to dispense solvent were not flame-dried. Reagents were used as obtained from commercial suppliers without further purification. Tetrahydrofuran (THF), diethyl ether (Et<sub>2</sub>O), methylene chloride (DCM), 1,2-dichloroethane (DCE), and toluene (PhMe) were purchased from Fisher and dispensed using the Glass Contour solvent purification system. 1,1,1,3,3,3-hexafluoropropan-2-ol (HFIP) was purchased from Oakwood Chemicals and used without further drying (bottle stored in desiccator after opening). Hydrocarbon-stabilized (ChromAR®). 5 Å molecular sieves used as desiccants were purchased from Aldrich, stored in an oven, and activated under high vacuum prior to use. Celite 545 was purchased from EMD. ACS grade hexanes, toluene, ethyl acetate, and DCM were used for column chromatography. Thin-layer chromatography (TLC) was performed on pre-coated silica gel 60 F254 glass-supported plates from EMD, and visualization was performed with a UV lamp followed by staining with *p*-anisaldehyde solution followed by heating. Column chromatography was carried out on EM Science silica gel (60 Å pore size, 230-400 mesh). Preparatory thin-layer chromatography (prep-TLC) was carried out using Analtech Uniplate F254 Prep-20x20 cm TLC plates. High-performance liquid chromatography (HPLC) was performed using Prominence-*i* LC 2030 Plus with a chiral stationary phase column (Chiralpak AD-H, Daicel Corp, 0.46 cm x 0.15 cm). Deuterated chloroform was purchased from Cambridge Isotope Laboratories.

<sup>1</sup>H NMR spectra were recorded at room temperature on a 400 MHz Bruker Avance spectrometer or a 500 MHz Bruker Avance spectrometer. Chemical shifts are given in parts per million (ppm) referenced to solvent residual proton resonance ( $\delta = 7.26$  for CHCl<sub>3</sub>). NMR data are reported as: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet, dd = doublet of doublets, dq = doublet of quartets, br = broad), coupling constants (*J*) given in Hz, and integration. In cases where two stereoisomers are present in greater than 2:1 ratio, only chemical shifts from the major stereoisomer are listed. For these cases a characteristic peak from both major and minor stereoisomer is given, with proton integrations, from which the ratio of stereoisomers can be extrapolated. In cases where two stereoisomers are present in less than 2:1 ratio, all peaks are listed. <sup>13</sup>C NMR spectra were recorded at room temperature unless otherwise stated on a 125 MHz or 101 MHz Bruker Avance spectrometer with proton decoupling. Chemical shifts are given in parts per million (ppm) from referenced to solvent carbon resonance ( $\delta = 77.0$  for CHCl<sub>3</sub>). In cases where two stereoisomers are present in greater than 2:1 ratio, only chemical shifts from the major stereoisomer are listed. For these cases a characteristic peak from both major and minor stereoisomer is given, with proton integrations, from which the ratio of stereoisomers can be extrapolated. In cases where two stereoisomers are present in less than 2:1 ratio, all peaks are listed. For spectra where solvent residue is present, yields were obtained after placing sample under vacuum and bringing to a constant weight. High resolution mass spectra (HRMS) were measured at the University of Rochester Mass

Spectrometry Resource Lab. X-ray crystallography data were collected by Dr. William W. Brennessel at the X-ray Crystallographic Facility of the University of Rochester, Rochester, NY 14627 (USA).

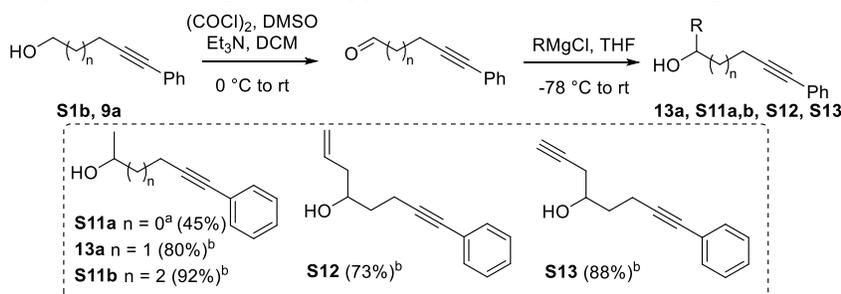
### Experimental Details:

**Primary Alcohols:** Alcohols shown in Scheme S1 were prepared according to reported experimental procedures. Sonogashira cross-couplings for **9a**, **S1a,b**, **S2**, **S3**, **S4**, and **S9a-c**.<sup>1-3</sup> Internal alkyne **S5** was made following a procedure reported by Sutherland et al.<sup>4</sup> Silyl alkyne **S6** was made following a procedure by Trost et al.<sup>5</sup> Bromo alkyne **S7** was prepared using Wang's procedure.<sup>6</sup> Dialkyne **S8** was made using Nitz's conditions.<sup>7</sup> Alcohol with a sulfonamide in the tether **S10** was made following conditions by Cardenas.<sup>8</sup> Characterization data matched the literature reports.



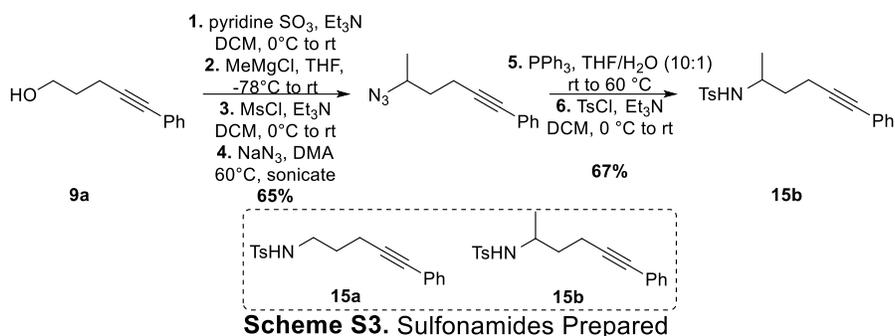
**Scheme S1:** Primary Alcohols Synthesized

**Procedures for Preparation of Secondary Alcohols:** Secondary alcohols were prepared by doing a Swern oxidation of the desired primary alcohol to afford an aldehyde to which a Grignard reagent of choice can be added to give products **13a**, **S11b**, **S12**, and **S13**.<sup>9,10</sup> Alcohol **S11a** was prepared following the Ito procedure.<sup>11</sup> Spectroscopy data matched literature reports.



**Scheme S2:** Secondary Alcohols Synthesized (a) Made using Bates's procedure<sup>11</sup>  
(b) Yield over two-steps.

**Procedure for Preparation of Sulfonamides:** Alpha-primary tosyl amine **15a** was prepared following the literature procedure. Spectra for this compound matched what has been reported in the literature.<sup>12</sup> Alpha-secondary amine **15b** was prepared following the procedure depicted in Scheme **S3**.<sup>13,14</sup>



#### *Alpha-Secondary Sulfonamide Preparation:*

**Step 1.** A solution of alcohol **9a** (3.70 g, 23.3 mmol) and Et<sub>3</sub>N (14.1 g, 19.5 mL, 140 mmol) in DCM (50 mL) was cooled to 0 °C. Pyridine SO<sub>3</sub> (10.9 g, 69.8 mmol) was dissolved in DMSO (50 mL) and added dropwise to the cooled solution over 15 minutes. After addition, the reaction was warmed to room temperature and allowed to stir until full consumption of the alcohol by TLC (using 20% ethyl acetate/hexanes as the mobile phase and *p*-anisaldehyde to stain the plates. After the reaction was complete, the reaction was diluted with Et<sub>2</sub>O, washed with 1M HCl until aqueous layer was acidic (pH < 1.0), extracted with Et<sub>2</sub>O (3 x 100 mL), washed with brine, dry over MgSO<sub>4</sub>, filter, and concentrate under vacuum. The crude aldehyde was used in Step 2.

**Step 2.** Crude aldehyde was added to a flask charged with a stir bar and purged with argon, then dissolved in dry THF and cooled to -78 °C. MeMgCl (15.5 mL, 46.6 mmol, 3 M in THF) was added dropwise, then the dry ice/isopropanol bath was removed, and the mixture was allowed to warm up to room temperature. After full consumption of the aldehyde by TLC, the mixture was cooled to 0 °C and saturated ammonium chloride solution was added dropwise to quench the reaction. When the bubbling had subsided, added 1:1 saturated ammonium chloride/water, extracted with Et<sub>2</sub>O (3 x 100 mL), washed with brine, dried over MgSO<sub>4</sub>, filtered, and concentrated under vacuum. Crude alcohol was used in the next step.

**Step 3.** The crude alcohol was dissolved in DCM (60 mL) and cooled to 0 °C. MsCl (4.5 mL, 58 mmol) was added in one portion, followed by Et<sub>3</sub>N (8.1 mL, 58 mmol), which was added dropwise. The reaction was allowed to warm to room temperature slowly overnight. It is important to note that the mesylated alcohol and the starting material are very close in R<sub>f</sub> (when 20% ethyl acetate/hexanes is used for the mobile phase). Reaction mixture was poured into a separatory funnel with ice and washed with 1 M HCl. The aqueous layer was washed with Et<sub>2</sub>O (3 x 50 mL), and the combined organic layers were washed with saturated sodium bicarbonate solution, then with brine, dried over MgSO<sub>4</sub>, filtered, and concentrated under vacuum. The crude mesylated alcohol was used in the next step.

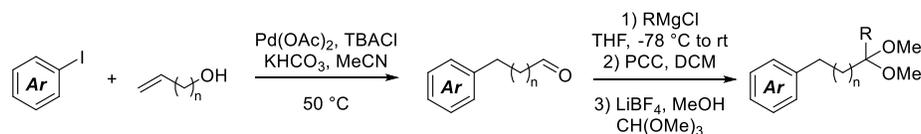
**Step 4.** The crude mesylated alcohol from Step 3 was dissolved in DMA (60 mL) and sodium azide was added (6.07 g, 93.2 mmol, 4.0 equivalents). The mixture was heated to 60 °C and sonicated for three hours, then moved to an oil bath and heated at 60 °C overnight. After consumption of the starting material by TLC, pentane was added and the reaction was washed with water, extracted with pentane, dried over MgSO<sub>4</sub>, filtered, and concentrated under vacuum. Crude was used for the next step (3.1 g, 65% crude yield over 4 steps).

**Step 5.** Crude azide from previous step (500 mg, 2.51 mmol) was dissolved in THF (16 mL). PPh<sub>3</sub> (1.35 g, 5.02 mmol) was added followed by water (2.05 mL, 120 mmol) and the reaction was allowed to stir at room temperature for an hour before heating to 60 °C overnight. Reaction mixture

was concentrated to remove the THF, diluted with Et<sub>2</sub>O, and washed with 1 M HCl (2 x 50 mL). The acidic aqueous layer was washed with EtOAc (3 x 50 mL), then basified with solid NaOH (pH > 12) and extracted with Et<sub>2</sub>O (3 x 50 mL), dried over MgSO<sub>4</sub>, filtered, and concentrated under vacuum. The crude amine was used in the next step.

**Step 6.** Crude amine from previous step was added to a vial charged with a stir bar, under argon, then dissolved in DCM (10 mL) and cooled to 0 °C. TsCl (484 mg, 2.54 mmol) was added, followed by Et<sub>3</sub>N (700 μL, 5.02 mmol). Cold bath was removed, and mixture was allowed to warm to room temperature. After the reaction was complete, water was added, extracted with Et<sub>2</sub>O, washed with brine, dried over MgSO<sub>4</sub>, filtered, and concentrated under vacuum. Crude was purified by column chromatography, using 10 to 15% EtOAc in hexanes to give sulfonamide **15b** (550 mg, 67% over 2 steps) as a yellow, viscous oil. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.77 (d, *J* = 8.0 Hz, 2H), 7.36 (d, *J* = 3.5 Hz, 2H), 7.29 (d, *J* = 3.4 Hz, 3H), 7.24 (s, 1H), 4.50 (d, *J* = 8.0 Hz, 1H), 3.55 – 3.44 (m, 1H), 2.38 (s, 3H), 2.36 (d, *J* = 7.0 Hz, 1H), 1.68 (t, *J* = 6.9 Hz, 2H), 1.13 (d, *J* = 6.5 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 143.3, 137.9, 131.5, 129.7, 128.2, 127.8, 127.1, 123.6, 88.8, 81.4, 49.5, 36.0, 21.7, 21.5, 16.0.; HRMS (ESI) *m/z*: [M+H]<sup>+</sup> Calculated for C<sub>19</sub>H<sub>21</sub>NO<sub>2</sub>S: 328.1366, found: 328.1358.

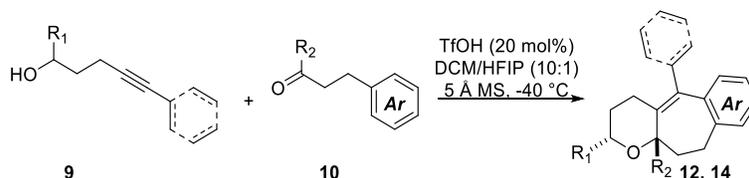
**Procedures for Preparation of Aldehydes/Ketones:** Aldehydes and ketals were prepared following previously reported procedures.<sup>15-19</sup> General reaction schemes below show the pathways used for aldehyde/ketal preparation. Spectroscopy data for aldehydes matched literature reports.



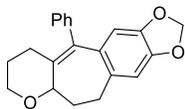
**Scheme S4:** Aldehydes/Ketals Synthetic pathways

### General Procedure for Carbocyclization Cascade:

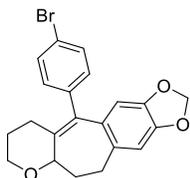
#### Scheme S5. General Procedure for Carbocyclization



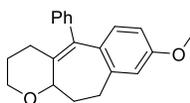
Oven dried, 5 Å molecular sieves (300 mg/mmol of limiting reagent **4**) were transferred to a flask equipped with a stir bar and dried further under vacuum while heating over a Bunsen burner for about 2 minutes. The reaction vessel was allowed to cool under a flow of argon before adding DCM/HFIP (10:1) (0.1 M with respect to the limiting reagent), followed by the alcohol **9**/**13** (1.2 equivalents) and aldehyde/ketal (1.0 equivalents for aldehydes, 1.5 equivalents for ketals). The reaction mixture was cooled to -40 °C before adding triflic acid (TfOH) (20 mol%) dropwise. Mixture was allowed to stir at -40 °C until all of the limiting reagent was consumed by TLC (using 20% ethyl acetate/hexane mixture as mobile phase and p-anisaldehyde to stain the plates). After completion, the reaction was quenched by adding solid sodium bicarbonate (300 mg/mmol of **10**) and diluting with ether (Et<sub>2</sub>O). The quenched reaction mixture was filtered through a silica plug, eluted with Et<sub>2</sub>O, and concentrated under vacuum. Pure product was isolated after preparatory TLC (using 20% ethyl acetate/hexane as mobile phase) or column chromatography using 5% ethyl acetate/hexane as mobile phase.



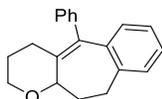
Compound **12a** was isolated as white solid (30.2 mg, 95%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.33 (t,  $J = 7.3$  Hz, 2H), 7.26 (dd,  $J = 8.7, 5.9$  Hz, 1H), 7.12 (d,  $J = 7.9$  Hz, 2H), 6.72 (s, 1H), 6.25 (s, 1H), 5.87 (s, 2H), 4.02 (dd,  $J = 11.1, 6.9$  Hz, 1H), 3.99 – 3.90 (m, 1H), 3.45 (dt,  $J = 11.0, 8.2$  Hz, 1H), 2.85 – 2.73 (m, 1H), 2.56 (ddd,  $J = 14.0, 6.7, 2.9$  Hz, 1H), 2.53 – 2.45 (m, 2H), 2.42 – 2.32 (m, 1H), 2.32 – 2.23 (m, 1H), 2.05 (dd,  $J = 11.8, 5.6$  Hz, 1H), 1.79 – 1.68 (m, 1H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  146.0, 145.7, 141.4, 138.4, 134.7, 134.3, 133.4, 129.9, 128.1, 126.7, 109.6, 108.4, 100.8, 63.3, 39.6, 30.4, 25.0, 22.8; m.p = 95-97°C; HRMS (ESI) m/z:  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{21}\text{H}_{22}\text{O}_3$ : 321.1485, found: 321.1477.



Compound **12b** was isolated as white solid (330 mg, 83%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.46 (d,  $J = 7.9$  Hz, 2H), 6.99 (d,  $J = 7.9$  Hz, 2H), 6.71 (s, 1H), 6.20 (s, 1H), 5.88 (s, 2H), 3.97 (dt,  $J = 30.7, 9.6$  Hz, 2H), 3.43 (dd,  $J = 18.3, 9.0$  Hz, 1H), 2.73 (dd,  $J = 14.9, 7.6$  Hz, 1H), 2.57 – 2.43 (m, 3H), 2.37 (dd,  $J = 20.3, 11.4$  Hz, 1H), 2.26 (dd,  $J = 19.1, 11.0$  Hz, 1H), 2.05 (s, 1H), 1.71 (s, 1H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  146.2, 145.9, 140.3, 139.2, 134.8, 133.7, 132.3, 131.6, 131.3, 120.8, 109.5, 108.5, 100.9, 77.2, 63.2, 39.6, 30.4, 24.9, 22.9; m.p = 60-63°C; HRMS (ESI) m/z:  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{21}\text{H}_{19}\text{BrO}_3$ : 399.0519, found: 399.0585.

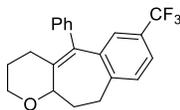


Compound **12c** was isolated as viscous oil, as a 1.7:1 of the *para-ortho* trapped products (57.8 mg, 94%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.37 – 7.30 (m, 4H), 7.30 – 7.24 (m, 5H), 7.19 (dd,  $J = 16.1, 7.9$  Hz, 2H), 7.12 (dd,  $J = 13.9, 7.6$  Hz, 6H), 6.89 (d,  $J = 7.5$  Hz, 1H), 6.79 (s, 2H), 6.74 – 6.67 (m, 3H), 6.64 (d,  $J = 8.5$  Hz, 2H), 4.11 (dt,  $J = 14.4, 7.2$  Hz, 1H), 4.05 (dt,  $J = 16.2, 8.2$  Hz, 2H), 4.00 – 3.89 (m, 3H), 3.80 (s, 6H), 3.57 – 3.43 (m, 3H), 3.32 (s, 4H), 2.93 – 2.75 (m, 4H), 2.58 (dt,  $J = 20.6, 7.0$  Hz, 5H), 2.54 – 2.20 (m, 9H), 2.12 – 1.98 (m, 3H), 1.86 – 1.69 (m, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  158.1, 157.0, 142.4, 141.8, 138.5, 137.4, 133.6, 133.3, 130.8, 130.6, 129.9, 129.4, 128.7, 128.31, 128.0, 127.4, 126.6, 125.8, 120.8, 113.6, 111.3, 110.9, 77.4, 76.5, 63.5, 62.9, 55.6, 55.1, 38.8, 38.1, 30.7, 30.5, 25.1, 24.7, 23.1, 21.6; HRMS (ESI) m/z:  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{21}\text{H}_{22}\text{O}_2$ : 307.1693, found: 307.1686.

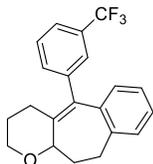


Compound **12d** was isolated as a foam (55.1 mg, 89%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.35 (t,  $J = 7.4$  Hz, 4H), 7.28 (t,  $J = 7.3$  Hz, 2H), 7.24 (d,  $J = 7.3$  Hz, 2H), 7.15 (d,  $J = 7.6$  Hz, 5H), 7.10 (dd,  $J = 14.6, 7.2$  Hz, 3H), 6.78 (d,  $J = 7.5$  Hz, 2H), 4.03 (dd,  $J = 10.8, 7.5$  Hz, 2H), 3.97 (ddd,  $J = 11.4, 8.7, 2.9$  Hz, 2H), 3.53 – 3.39 (m, 2H), 2.89 (td,  $J = 12.8, 8.1$  Hz, 2H), 2.68 – 2.56 (m, 4H), 2.53

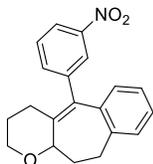
(dt,  $J = 12.8, 6.1$  Hz, 2H), 2.41 (dd,  $J = 20.0, 12.0$  Hz, 2H), 2.33 (dd,  $J = 19.7, 11.6$  Hz, 2H), 2.15 – 2.01 (m, 2H), 1.82 – 1.67 (m, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  141.6, 141.1, 140.8, 138.9, 133.5, 129.9, 129.5, 128.2, 128.1, 126.7, 126.7, 125.9, 77.2, 63.4, 39.3, 30.4, 25.0, 23.0; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{20}\text{H}_{20}\text{O}$ : 277.1587, found: 277.1583.



Reaction mixture was allowed to stir at  $-40^\circ\text{C}$  for an hour after addition of the acid, then allowed to warm up to room temperature until all of the aldehyde had been consumed. Compound **12e** was isolated as a yellow oil (21.1 mg, 31%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.40 – 7.27 (m, 5H), 7.12 (s, 1H), 7.11 (s, 1H), 7.01 (s, 1H), 3.96 (dd,  $J = 10.9, 8.2$  Hz, 2H), 3.53 – 3.34 (m, 1H), 2.88 (dd,  $J = 12.7, 8.2$  Hz, 1H), 2.68 (dd,  $J = 13.0, 6.5$  Hz, 1H), 2.64 – 2.57 (m, 1H), 2.57 – 2.47 (m, 1H), 2.38 (ddd,  $J = 28.2, 19.3, 11.4$  Hz, 2H), 2.16 – 2.02 (m, 1H), 1.82 – 1.68 (m, 1H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  144.6, 141.7, 140.6, 132.9, 129.8, 128.7, 128.4, 127.1, 126.1, 125.3, 123.4, 76.9, 63.4, 39.2, 30.4, 24.8, 23.1; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{21}\text{H}_{19}\text{F}_3\text{O}$ : 345.1461, found: 345.1457.

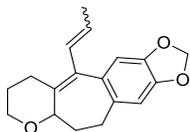


Reaction was allowed to stir at  $-40^\circ\text{C}$  for an hour after addition of TfOH, then warmed up to room temperature until full consumption of starting material was observed by TLC. Compound **12f** was isolated as a viscous, colorless oil (33.3 mg, 70%).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.56 (d,  $J = 7.7$  Hz, 1H), 7.48 (t,  $J = 7.7$  Hz, 1H), 7.41 (s, 1H), 7.34 (d,  $J = 7.6$  Hz, 1H), 7.27 (d,  $J = 6.9$  Hz, 1H), 7.15 (dt,  $J = 21.8, 7.4$  Hz, 2H), 6.73 (d,  $J = 7.6$  Hz, 1H), 4.09 – 3.93 (m, 2H), 3.47 (dd,  $J = 18.3, 9.6$  Hz, 1H), 2.89 (td,  $J = 12.8, 8.1$  Hz, 1H), 2.66 (dd,  $J = 13.0, 6.5$  Hz, 1H), 2.61 – 2.48 (m, 2H), 2.48 – 2.28 (m, 2H), 2.17 – 2.04 (m, 1H), 1.76 (ddd,  $J = 12.6, 8.3, 5.2$  Hz, 1H).  $^{13}\text{C}$  NMR (100 MHz,  $\text{CDCl}_3$ )  $\delta$  142.3, 140.9, 140.5, 140.2, 133.3, 132.3, 129.4, 128.6, 128.4, 127.1, 126.6, 126.2, 77.3, 77.2, 63.3, 39.2, 30.4, 24.9, 23.1.  $^{19}\text{F}$  NMR (250 MHz)  $\delta$  -62.7. HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{21}\text{H}_{19}\text{F}_3\text{O}$ : 345.1461, found: 345.1459.

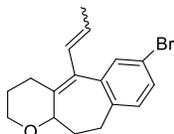


Reaction was allowed to stir at  $-40^\circ\text{C}$  for an hour after addition of TfOH, then warmed up to room temperature until full consumption of starting material was observed by TLC. Compound **12g** was isolated as a viscous oil (22.1 mg, 69%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  8.15 (d,  $J = 8.1$  Hz, 1H), 8.00 (s, 1H), 7.52 (t,  $J = 7.8$  Hz, 1H), 7.47 (d,  $J = 7.6$  Hz, 1H), 7.28 (dd,  $J = 12.0, 6.2$  Hz, 1H), 7.18 (dd,  $J = 16.7, 9.2$  Hz, 1H), 7.10 (t,  $J = 7.4$  Hz, 1H), 6.68 (d,  $J = 7.6$  Hz, 1H), 4.03 (dd,  $J = 10.9, 7.5$  Hz, 1H), 3.97 (ddd,  $J = 11.5, 8.6, 2.9$  Hz, 1H), 3.49 – 3.42 (m, 1H), 2.87 (td,  $J = 12.9, 8.0$  Hz, 1H), 2.65 (dd,  $J = 13.2, 6.4$  Hz, 1H), 2.59 – 2.48 (m, 2H), 2.48 – 2.38 (m, 1H), 2.33 (dd,  $J = 19.5, 11.6$  Hz, 1H), 2.16 – 2.05 (m, 1H), 1.78 (dtd,  $J = 10.1, 6.7, 3.0$  Hz, 1H).  $^{13}\text{C}$  NMR (126

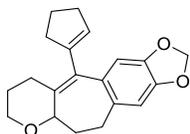
MHz, CDCl<sub>3</sub>)  $\delta$  148.3, 143.2, 141.5, 140.9, 139.7, 136.1, 131.4, 129.3, 129.1, 128.6, 127.4, 126.3, 124.8, 121.8, 77.1, 63.3, 39.2, 30.4, 24.8, 23.2. HRMS (ESI)  $m/z$ : [M+H]<sup>+</sup> Calculated for C<sub>20</sub>H<sub>19</sub>NO<sub>3</sub>: 322.1438, found: 322.1437.



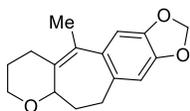
Compound **12h** was isolated as a foamy viscous oil, as a 6.3:1 mixture of *E/Z* isomers (22.4mg, 70%). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  6.72 (s, 1H), 6.67 (s, 1H), 5.93 (d, *J* = 10 Hz, 1H), 5.51 (dq, *J* = 15.2, 7.0 Hz, 1H), 3.86 – 3.76 (m, 3H), 3.36 (dt, *J* = 14.2, 7.2 Hz, 1H), 2.88 (dd, *J* = 14.7, 6.3 Hz, 1H), 2.54-2.47 (m, 1H), 2.45 – 2.10 (m, 5H), 1.99 (dt, *J* = 14.8, 7.3 Hz, 1H), 1.78 (d, *J* = 6.7 Hz, 3H), 1.67 (dt, *J* = 11.7, 6.3 Hz, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  146.0, 145.4, 135.4, 134.5, 131.5, 130.0, 128.3, 128.1, 109.7, 108.2, 100.7, 76.5, 62.8, 38.3, 29.8, 24.1, 20.8, 18.6; HRMS (ESI)  $m/z$ : [M+H]<sup>+</sup> Calculated for C<sub>18</sub>H<sub>20</sub>O<sub>3</sub>: 285.1485, found: 285.1481



Compound **12i** was isolated as a white solid, as a 7.9:1 mixture of *E/Z* isomers (22.4mg, 75%). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.35 (s, 1H), 7.27 (d, *J* = 8.4 Hz, 1H), 7.04 (d, *J* = 7.8 Hz, 1H), 6.52 (d, *J* = 15.5 Hz, 1H), 5.49 (dd, *J* = 14.9, 7.0 Hz, 1H), 3.88 – 3.76 (m, 2H), 3.38 – 3.29 (m, 1H), 2.88 (dd, *J* = 14.4, 6.0 Hz, 1H), 2.56 – 2.47 (m, 1H), 2.47 – 2.36 (m, 2H), 2.28 (dd, *J* = 12.6, 6.2 Hz, 1H), 2.21 – 2.12 (m, 1H), 2.05 – 1.94 (m, 1H), 1.80 (d, *J* = 6.1 Hz, 3H), 1.70-1.60 (m, 1H). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  141.6, 140.0, 131.8, 132.0, 129.7, 129.6, 129.0, 128.6, 127.7, 119.3, 76.2, 62.8, 37.8, 29.3, 23.9, 20.8, 18.6; m.p = 105-107°C; HRMS (ESI)  $m/z$ : [M+H]<sup>+</sup> Calculated for C<sub>17</sub>H<sub>19</sub>BrO: 319.0692, found: 319.0687.

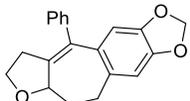


Compound **12j** was isolated as a foamy viscous oil (255 mg, 78%). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  6.67 (s, 1H), 6.60 (s, 1H), 5.99 – 5.81 (m, 2H), 5.61 (s, 1H), 3.89 (t, *J* = 9.1 Hz, 2H), 3.35 (dd, *J* = 18.4, 9.0 Hz, 1H), 2.79 – 2.68 (m, 1H), 2.59 (dd, *J* = 14.3, 7.4 Hz, 1H), 2.51 – 2.34 (m, 6H), 2.34 – 2.24 (m, 1H), 2.16 (dd, *J* = 18.2, 10.5 Hz, 2H), 2.08 – 1.97 (m, 1H), 1.94 – 1.84 (m, 2H), 1.63 (s, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  145.8 (2C), 142.6, 137.1, 134.4, 132.5, 130.1, 129.1, 108.6, 108.2, 100.7, 77.1, 63.2, 39.1, 36.0, 32.8, 30.1, 24.9, 23.8, 22.7; HRMS (ESI)  $m/z$ : [M+H]<sup>+</sup> Calculated for C<sub>20</sub>H<sub>22</sub>O<sub>3</sub>: 311.1642, found: 311.1636.

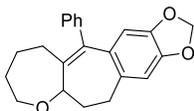


Added 40 mol% of TfOH at -40 °C, allowed to stir at this temperature for an hour and then warmed to room temperature until all of the aldehyde was consumed. Compound **12k** was isolated as a light-yellow oil, as a 3.9:1 inseparable mixture of *para/ortho*-trapped adduct (33.0 mg, 64%). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  6.74 (d, *J* = 7.9 Hz, 1H), 6.70 (s, 1H), 6.66 (d, *J* = 6.8 Hz, 1H), 5.92

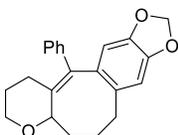
(s, 2H), 4.39 – 4.30 (m, 1H), 3.78 (td,  $J = 11.4, 3.8$  Hz, 1H), 3.67 (d,  $J = 11.4$  Hz, 1H), 2.80 (d,  $J = 14.3$  Hz, 1H), 2.70 – 2.60 (m, 1H), 2.55 (dt,  $J = 13.4, 6.9$  Hz, 1H), 2.29-2.13 (m, 3H), 1.92 (s, 3H), 1.77 – 1.67 (m, 4H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  147.7, 145.8, 139.4, 134.9, 129.9, 121.2, 108.9, 108.2, 100.8, 100.7, 72.3, 60.3, 32.7, 31.2, 25.6, 21.4, 16.2; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{16}\text{H}_{18}\text{O}_3$ : 259.1329, found: 259.1329.



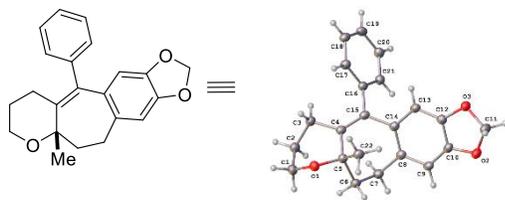
Compound **12i** was isolated as white solid (84.3 mg, 92%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.35 (t,  $J = 6.9$  Hz, 2H), 7.28 (d,  $J = 7.7$  Hz, 1H), 7.24 (t,  $J = 10.3$  Hz, 2H), 6.75 (s, 1H), 6.32 (d,  $J = 1.7$  Hz, 1H), 5.88 (d,  $J = 2.0$  Hz, 2H), 4.35 (t,  $J = 8.6$  Hz, 1H), 4.01 (dd,  $J = 10.0, 6.1$  Hz, 1H), 3.98 – 3.88 (m, 1H), 2.99 (dd,  $J = 16.2, 8.0$  Hz, 1H), 2.87 – 2.75 (m, 1H), 2.65 – 2.47 (m, 3H), 2.14 (dd,  $J = 18.9, 8.7$  Hz, 1H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  145.9, 145.8, 141.9, 141.0, 135.0, 134.6, 130.8, 129.3, 128.2, 127.9, 126.8, 109.0, 108.6, 100.8, 78.5, 67.0, 40.3, 32.2, 30.8; m.p = 39-40°C; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{20}\text{H}_{18}\text{O}_3$  : 307.1334, found: 307.1323.



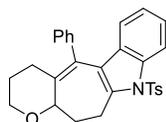
Compound **12m** was isolated as white solid (82.9 mg, 82%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.31 (t,  $J = 7.0$  Hz, 2H), 7.25 (dd,  $J = 9.1, 4.6$  Hz, 1H), 7.13 (s, 2H), 6.71 (s, 1H), 6.26 (d,  $J = 1.7$  Hz, 1H), 5.89 – 5.82 (m, 2H), 4.05 (d,  $J = 12.0$  Hz, 1H), 4.00 (dd,  $J = 10.6, 6.0$  Hz, 1H), 3.32 – 3.22 (m, 1H), 2.83 – 2.71 (m, 1H), 2.53 – 2.41 (m, 2H), 2.41 – 2.31 (m, 2H), 2.14 – 2.04 (m, 1H), 2.01 (d,  $J = 11.0$  Hz, 1H), 1.72 (s, 2H), 1.53 (d,  $J = 6.9$  Hz, 1H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  146.0, 145.7, 143.2, 141.9, 134.7, 134.7, 134.6, 134.5, 129.6, 127.9, 126.6, 109.6, 108.1, 100.7, 80.7, 71.7, 40.6, 30.7, 30.4, 29.9, 28.7; m.p = 47-49°C; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{22}\text{H}_{22}\text{O}_3$ : 335.1647, found: 335.1634.



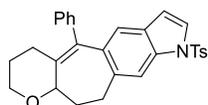
Added 40 mol% of TfOH at -40 °C, allowed to stir at this temperature for an hour and then warmed to 0 °C until all of the aldehyde was consumed. Compound **12n** was isolated as a white foam (20 mg, 34%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.31 (t,  $J = 7.4$  Hz, 2H), 7.26 – 7.19 (m, 1H), 7.09 (d,  $J = 7.1$  Hz, 2H), 6.67 (s, 1H), 6.34 (s, 1H), 5.89 – 5.82 (m, 2H), 4.02 (d,  $J = 10.9$  Hz, 1H), 3.79 (dd,  $J = 14.1, 6.8$  Hz, 1H), 3.66 (d,  $J = 11.3$  Hz, 1H), 2.77 (dd,  $J = 12.9, 7.9$  Hz, 1H), 2.64 (d,  $J = 15.0$  Hz, 1H), 2.43 (t,  $J = 12.6$  Hz, 1H), 2.35 – 2.23 (m, 1H), 2.17 – 2.06 (m, 1H), 1.99 (ddd,  $J = 16.6, 12.8, 5.4$  Hz, 1H), 1.93 – 1.72 (m, 3H), 1.53 (dt,  $J = 12.0, 7.5$  Hz, 1H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  147.1, 145.3, 141.9, 136.8, 135.5, 133.8, 132.9, 129.2, 128.1, 126.7, 108.9, 108.6, 100.8, 76.3, 61.3, 32.8, 29.7, 27.8, 27.7, 23.7.; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{22}\text{H}_{22}\text{O}_3$ : 335.1642, found: 335.1634.



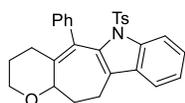
Added 1.5 equivalents of dimethyl ketal and 1.0 equivalents of alcohol **9a**. Stir at  $-40\text{ }^{\circ}\text{C}$  until all of the alcohol was consumed. Compound **12o** was isolated as white solid (38.7 mg, 48%). Recrystallized by dissolving in hot methanol and allowing to cool to room temperature in an uncapped vial.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.32 (t,  $J = 7.2$  Hz, 2H), 7.28 – 7.22 (m, 1H), 7.08 (d,  $J = 5.5$  Hz, 2H), 6.67 (s, 1H), 6.17 (s, 1H), 5.85 (s, 2H), 3.94 – 3.84 (m, 1H), 3.81 (t,  $J = 9.4$  Hz, 1H), 2.95 (dd,  $J = 13.3, 5.4$  Hz, 1H), 2.59 (d,  $J = 12.8$  Hz, 1H), 2.50 (d,  $J = 13.0$  Hz, 1H), 2.39 (ddd,  $J = 19.6, 11.4, 6.4$  Hz, 2H), 2.17 – 2.04 (m, 1H), 1.81 – 1.59 (m, 2H), 1.06 (s, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  145.9, 145.7, 143.7, 143.1, 135.0, 134.6, 133.0, 130.1, 128.1, 126.5, 110.0, 108.0, 100.7, 78.1, 58.6, 47.4, 31.5, 27.6, 25.7, 24.0; m.p =  $115\text{--}117^{\circ}\text{C}$ ; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{22}\text{H}_{22}\text{O}_3$ : 335.1642, found: 335.1634.



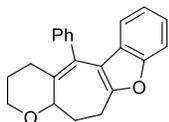
Compound **12p** was isolated as white solid (196 mg, 84%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  8.20 (d,  $J = 8.0$  Hz, 1H), 7.70 (d,  $J = 7.1$  Hz, 2H), 7.30 (d,  $J = 5.9$  Hz, 3H), 7.26 (s, 1H), 7.23 (d,  $J = 7.5$  Hz, 2H), 7.16 (t,  $J = 7.7$  Hz, 1H), 7.10 (d,  $J = 6.2$  Hz, 2H), 6.89 (t,  $J = 7.0$  Hz, 1H), 6.33 (d,  $J = 8.0$  Hz, 1H), 3.97 (t,  $J = 9.7$  Hz, 2H), 3.68 (dd,  $J = 14.7, 6.5$  Hz, 1H), 3.42 (q,  $J = 8.9$  Hz, 1H), 2.82 (dd,  $J = 21.5, 13.6$  Hz, 1H), 2.73 – 2.62 (m, 1H), 2.53 – 2.39 (m, 3H), 2.37 (s, 3H), 2.07 (d,  $J = 24.1$  Hz, 1H), 1.77 (s, 1H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  144.9, 142.1, 140.5, 139.1, 136.8, 135.8, 129.8, 129.5, 128.8, 128.5, 128.3, 127.1, 126.4, 123.7, 123.2, 121.8, 120.3, 114.9, 78.3, 63.6, 40.9, 25.3, 22.7, 21.8, 21.6; m.p =  $78\text{--}80^{\circ}\text{C}$ ; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{29}\text{H}_{27}\text{NO}_3\text{S}$ : 470.1785, found: 470.1776.



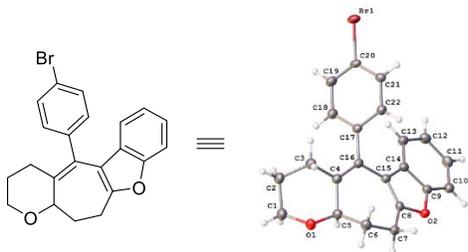
Compound **12q** was isolated as white solid (38.2 mg, 52%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.83 (s, 1H), 7.77 (d,  $J = 8.1$  Hz, 2H), 7.41 (d,  $J = 1.9$  Hz, 1H), 7.30 (t,  $J = 7.2$  Hz, 2H), 7.24 (t,  $J = 9.6$  Hz, 4H), 7.10 (d,  $J = 7.4$  Hz, 2H), 6.90 (s, 1H), 6.42 (s, 1H), 3.93 (dd,  $J = 20.6, 12.4$  Hz, 2H), 3.40 (dd,  $J = 18.4, 8.9$  Hz, 1H), 2.94 (dd,  $J = 20.7, 12.4$  Hz, 1H), 2.74 (dd,  $J = 12.9, 6.4$  Hz, 1H), 2.63 – 2.42 (m, 2H), 2.36 (d,  $J = 17.2$  Hz, 4H), 2.32 – 2.26 (m, 1H), 2.03 (s, 1H), 1.71 (s, 1H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  144.9, 141.9, 138.0, 137.1, 135.4, 133.6, 133.3, 130.0, 129.0, 128.1, 126.7, 126.0, 122.3, 112.5, 109.0, 65.8, 63.2, 38.7, 31.1, 25.0, 22.9, 21.6, 15.3; m.p =  $92\text{--}95^{\circ}\text{C}$ ; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{29}\text{H}_{27}\text{NO}_3\text{S}$ : 470.1785, found: 470.1783.



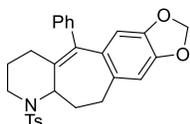
Compound **12r** was isolated as an amorphous solid, as a 1:1 mixture of conformers (96.8 mg, 41%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.68 (d,  $J = 8.1$  Hz, 20H), 7.57 (d,  $J = 8.1$  Hz, 26H), 7.40 (d,  $J = 4.3$  Hz, 44H), 7.34 (s, 34H), 7.31 (d,  $J = 13.6$  Hz, 53H), 7.26 (s, 64H), 7.21 (d,  $J = 2.3$  Hz, 15H), 7.19 – 6.96 (m, 155H), 6.80 (q,  $J = 8.1$  Hz, 20H), 6.57 (s, 7H), 6.04 (d,  $J = 7.1$  Hz, 9H), 5.94 (d,  $J = 7.5$  Hz, 7H), 5.84 (s, 9H), 5.44 (s, 10H), 4.21 – 4.10 (m, 19H), 4.01 (dd,  $J = 25.5, 9.3$  Hz, 48H), 3.90 (t,  $J = 7.1$  Hz, 8H), 3.76 (q,  $J = 7.4$  Hz, 9H), 3.67 – 3.49 (m, 17H), 2.63 – 2.50 (m, 36H), 2.46 (dd,  $J = 16.9, 10.0$  Hz, 23H), 2.32 (d,  $J = 16.8$  Hz, 107H), 2.03 (ddd,  $J = 26.3, 22.1, 9.7$  Hz, 75H), 1.94 – 1.74 (m, 92H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  143.3, 141.0, 140.3, 138.0, 137.4, 137.1, 136.4, 135.4, 135.1, 134.9, 134.9, 134.8, 131.5, 131.4, 129.8, 129.8, 129.6, 129.5, 128.9, 128.5, 128.2, 128.1, 128.0, 127.8, 127.7, 127.6, 127.5, 126.9, 126.8, 126.5, 126.4, 125.0, 124.2, 123.8, 123.7, 122.7, 122.3, 113.0, 112.4, 98.7, 89.3, 89.2, 81.2, 81.0, 75.0, 74.1, 71.4, 70.0, 68.0, 67.7, 53.4, 53.1, 29.2, 28.9, 28.4, 28.0, 27.9, 27.6, 26.6, 26.0, 25.7, 21.4, 21.4, 16.2, 16.2; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{29}\text{H}_{27}\text{NO}_3\text{S}$ : 470.1785, found: 470.1784.



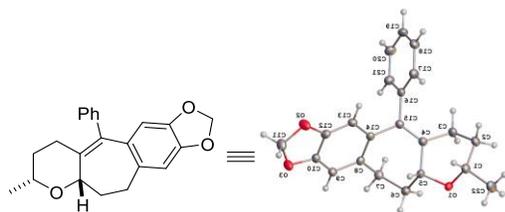
Compound **12s** was isolated as light-yellow solid (49.3 mg, 79%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.44 – 7.36 (m, 3H), 7.34 (d,  $J = 8.2$  Hz, 1H), 7.25 (s, 1H), 7.09 (t,  $J = 7.7$  Hz, 1H), 6.82 (t,  $J = 7.6$  Hz, 1H), 5.97 (d,  $J = 7.9$  Hz, 1H), 4.32 (t,  $J = 5.7$  Hz, 1H), 4.05 (dd,  $J = 14.5, 5.4$  Hz, 1H), 3.60 (td,  $J = 10.6, 6.8$  Hz, 1H), 3.27 – 3.11 (m, 2H), 2.52 – 2.42 (m, 1H), 2.42 – 2.27 (m, 3H), 2.04 (dd,  $J = 13.8, 6.9$  Hz, 1H), 1.80 – 1.67 (m, 1H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  157.3, 153.8, 140.5, 139.2, 129.8, 128.6, 128.3, 127.0, 126.0, 123.0, 121.9, 121.2, 114.8, 110.4, 79.1, 64.7, 32.7, 25.4, 25.0, 24.3; m.p = 86–88°C; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{22}\text{H}_{20}\text{O}_2$ : 317.1536, found: 317.1530.



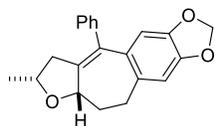
Compound **12t** was isolated as white solid (58.4 mg, 70%). Recrystallized by dissolving in hot methanol and allowing to cool to room temperature in an uncapped vial.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.52 (d,  $J = 7.7$  Hz, 2H), 7.35 (d,  $J = 8.1$  Hz, 1H), 7.11 (t,  $J = 8.4$  Hz, 3H), 6.87 (t,  $J = 7.5$  Hz, 1H), 6.06 (d,  $J = 7.8$  Hz, 1H), 4.29 (s, 1H), 4.04 (t,  $J = 9.9$  Hz, 1H), 3.56 (dd,  $J = 17.6, 10.0$  Hz, 1H), 3.25 – 3.08 (m, 2H), 2.48 – 2.39 (m, 1H), 2.33 (dd,  $J = 24.7, 6.9$  Hz, 3H), 2.04 (d,  $J = 12.0$  Hz, 1H), 1.70 (d,  $J = 3.5$  Hz, 1H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  157.5, 153.9, 140.1, 139.3, 131.5 (2C), 128.3, 124.9, 123.2, 122.1, 121.0 (2C), 114.2, 110.5, 79.0, 64.5, 32.7, 25.2, 24.9, 24.2; m.p = 126–128°C; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{22}\text{H}_{19}\text{BrO}_2$ : 395.0641, found: 395.0630.



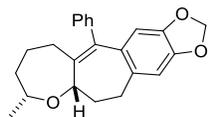
After addition of TfOH, stirred at  $-40\text{ }^{\circ}\text{C}$  for 30 minutes before warming to  $0\text{ }^{\circ}\text{C}$  until all the starting material was consumed. Compound **16a** was isolated as a white solid (24 mg, 54%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.59 (d,  $J = 8.1$  Hz, 2H), 7.30 (d,  $J = 8.0$  Hz, 2H), 7.23 (s, 2H), 6.77 (d,  $J = 6.4$  Hz, 2H), 6.75 (s, 1H), 6.23 (s, 1H), 5.93 (d,  $J = 7.0$  Hz, 2H), 4.50 (dd,  $J = 12.1, 7.3$  Hz, 1H), 3.76 – 3.62 (m, 1H), 3.40 (dd,  $J = 11.9, 6.5$  Hz, 1H), 2.64 (td,  $J = 13.1, 7.4$  Hz, 1H), 2.53 – 2.49 (m, 1H), 2.47 (s, 3H), 2.42 (dt,  $J = 19.9, 6.7$  Hz, 1H), 2.29 (td,  $J = 12.2, 7.2$  Hz, 1H), 2.17 (ddd,  $J = 15.8, 6.1, 3.4$  Hz, 1H), 1.79 – 1.70 (m, 1H), 1.59 (s, 2H), 1.48 (dd,  $J = 9.5, 4.0$  Hz, 1H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  146.6, 146.2, 142.9, 140.5, 137.5, 135.4, 134.2, 134.0, 130.1, 129.5, 127.9, 127.1, 126.9, 109.4, 108.9, 100.9, 54.5, 41.7, 41.2, 30.9, 26.0, 21.9, 21.5; m.p =  $155\text{--}157^{\circ}\text{C}$ ; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{28}\text{H}_{27}\text{NO}_4\text{S}$ : 474.1734, found: 474.1728.



Compound **14a** was isolated as white solid, 9:1 dr (61 mg, 91%). Recrystallized by dissolving in hot methanol and allowing to cool to room temperature in an uncapped vial.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.33 (t,  $J = 7.4$  Hz, 2H), 7.27 (dd,  $J = 8.8, 5.6$  Hz, 1H), 7.12 (d,  $J = 7.1$  Hz, 2H), 6.72 (d,  $J = 6.9$  Hz, 1H), 6.25 (d,  $J = 6.1$  Hz, 1H), 5.86 (s, 2H), 4.01 (dd,  $J = 10.7, 7.5$  Hz, 1H), 3.48 (dt,  $J = 6.2, 4.1$  Hz, 1H), 2.80 (td,  $J = 12.6, 7.9$  Hz, 1H), 2.56 (td,  $J = 12.3, 6.4$  Hz, 1H), 2.49 (dd,  $J = 12.9, 7.0$  Hz, 2H), 2.40 (dd,  $J = 21.4, 11.4$  Hz, 1H), 2.20 (td,  $J = 11.5, 8.2$  Hz, 1H), 1.94 – 1.82 (m, 1H), 1.73 (dd,  $J = 20.9, 9.8$  Hz, 1H), 1.24 (d,  $J = 4.8$  Hz, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  145.9, 145.7, 141.6, 139.2, 134.8, 134.3, 133.0, 129.8, 128.1, 126.7, 109.6, 108.4, 100.7, 77.9, 40.1, 33.5, 30.4, 23.0, 22.3; m.p =  $132\text{--}135^{\circ}\text{C}$ ; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{22}\text{H}_{22}\text{O}_3$ : 335.1642, found:

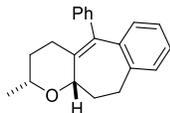


Compound **14b** isolated as a white solid, 7:1 dr (53.3 mg, 83%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.36 (t,  $J = 7.3$  Hz, 4H), 7.27 (dd,  $J = 14.5, 6.8$  Hz, 3H), 7.19 (d,  $J = 7.0$  Hz, 4H), 6.72 (s, 2H), 6.29 (s, 2H), 5.89 – 5.82 (m, 4H), 4.32 (t,  $J = 8.2$  Hz, 2H), 4.06 (dd,  $J = 10.0, 5.1$  Hz, 2H), 2.97 – 2.84 (m, 4H), 2.74 – 2.61 (m, 2H), 2.51 (dd,  $J = 12.6, 4.3$  Hz, 2H), 2.22 (dd,  $J = 14.2, 9.4$  Hz, 4H), 1.30 (d,  $J = 5.9$  Hz, 7H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  145.8, 142.5, 142.0, 134.9, 134.1, 131.1, 129.4, 128.8, 128.4, 128.0, 126.8, 108.9, 108.7, 100.8, 78.7, 75.5, 42.6, 39.6, 30.8, 20.8; m.p = sample decomposed after heating to  $100^{\circ}\text{C}$ ; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{21}\text{H}_{20}\text{O}_3$ : 321.1491, found: 321.1481.

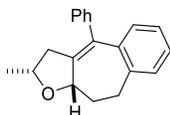


Compound **14c** isolated as a white amorphous solid, 13:1 (43.7 mg, 63%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.31 (t,  $J = 7.1$  Hz, 2H), 7.27 – 7.21 (m, 1H), 7.12 (s, 2H), 6.72 (s, 1H), 6.26 (s, 1H),

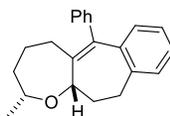
5.87 (s, 2H), 4.01 (dd,  $J = 11.3, 6.3$  Hz, 1H), 3.39 (d,  $J = 6.4$  Hz, 1H), 2.81 – 2.70 (m, 1H), 2.53 – 2.41 (m, 2H), 2.36 (s, 2H), 2.10 (t,  $J = 11.2$  Hz, 1H), 1.98 (d,  $J = 7.6$  Hz, 1H), 1.74 (d,  $J = 11.0$  Hz, 1H), 1.51 (dd,  $J = 22.8, 14.1$  Hz, 2H), 1.15 (d,  $J = 6.2$  Hz, 3H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  145.9, 145.7, 143.9, 141.9, 134.9, 134.6, 134.1, 129.7, 127.9, 126.6, 109.6, 108.1, 100.7, 79.6, 77.5, 40.7, 37.2, 30.4, 29.6, 28.3, 23.2; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{23}\text{H}_{24}\text{O}_3$ : 349.1804, found: 349.1792.



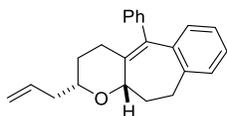
Compound **14d** isolated as an amorphous white solid, >20:1 dr (28 mg, 96%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.34 (t,  $J = 7.2$  Hz, 5H), 7.31 – 7.25 (m, 3H), 7.23 (d,  $J = 7.1$  Hz, 2H), 7.14 (d,  $J = 6.7$  Hz, 6H), 7.09 (dd,  $J = 14.8, 7.2$  Hz, 3H), 6.77 (d,  $J = 7.4$  Hz, 2H), 4.01 (dd,  $J = 9.8, 7.4$  Hz, 2H), 3.55 – 3.43 (m, 2H), 2.88 (t,  $J = 11.4$  Hz, 2H), 2.60 (dt,  $J = 12.0, 6.5$  Hz, 4H), 2.55 – 2.48 (m, 2H), 2.43 (dd,  $J = 21.5, 11.1$  Hz, 2H), 2.31 – 2.20 (m, 2H), 1.94 – 1.83 (m, 2H), 1.74 (dd,  $J = 20.5, 10.0$  Hz, 2H), 1.24 (d,  $J = 5.0$  Hz, 3H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  141.8, 141.1, 140.9, 139.7, 133.2, 129.9, 129.5, 128.2, 128.1, 126.6, 125.9, 77.9, 70.6, 39.7, 33.5, 30.4, 23.3, 22.3; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{21}\text{H}_{22}\text{O}$ : 291.1749, found: 291.1742.



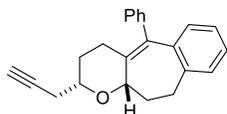
Compound **14e** isolated as an amorphous solid, >20:1dr (55.0 mg, 99%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.35 (t,  $J = 7.2$  Hz, 2H), 7.27 (t,  $J = 6.8$  Hz, 1H), 7.13 – 7.03 (m, 2H), 6.80 (d,  $J = 7.2$  Hz, 1H), 4.32 (t,  $J = 8.2$  Hz, 1H), 4.09 – 3.99 (m, 1H), 3.04 – 2.88 (m, 2H), 2.74 – 2.65 (m, 1H), 2.62 (d,  $J = 12.6$  Hz, 1H), 2.24 (dd,  $J = 22.5, 12.7$  Hz, 2H), 1.30 (d,  $J = 4.9$  Hz, 3H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  142.6, 140.8, 140.7, 131.3, 129.5, 128.7, 128.6, 128.4, 126.7, 126.5, 125.9, 78.7, 75.4, 42.2, 39.7, 30.8, 20.8; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{20}\text{H}_{20}\text{O}$ : 277.1592, found: 277.1584.



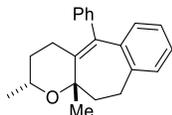
Compound **14f** isolated as an amorphous solid, 8:1dr (50.4 mg, 99%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.31 (t,  $J = 7.0$  Hz, 108H), 7.24 (dd,  $J = 14.6, 6.8$  Hz, 125H), 7.16 – 7.07 (m, 203H), 6.77 (d,  $J = 7.3$  Hz, 46H), 4.00 (dd,  $J = 11.2, 7.0$  Hz, 46H), 3.38 (d,  $J = 6.4$  Hz, 43H), 2.90 – 2.78 (m, 48H), 2.60 – 2.53 (m, 55H), 2.53 – 2.44 (m, 44H), 2.38 (d,  $J = 11.5$  Hz, 101H), 2.20 – 2.08 (m, 48H), 2.00 (d,  $J = 11.0$  Hz, 48H), 1.74 (d,  $J = 12.4$  Hz, 52H), 1.63 – 1.45 (m, 101H), 1.14 (d,  $J = 6.0$  Hz, 150H), 1.06 (t,  $J = 10.1$  Hz, 8H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  144.2, 142.1, 141.8, 140.8, 134.2, 129.8, 129.4, 127.9, 127.8, 126.7, 126.5, 126.0, 79.4, 77.5, 40.4, 37.2, 30.5, 29.5, 28.3, 23.2; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{22}\text{H}_{24}\text{O}$ : 305.1905, found: 305.1897.



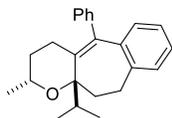
Compound **14g** isolated as a viscous oil, 16:1dr (155 mg, 98%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.37 (t,  $J = 7.3$  Hz, 2H), 7.31 (t,  $J = 7.0$  Hz, 1H), 7.26 (s, 1H), 7.21 – 7.09 (m, 4H), 6.81 (d,  $J = 7.4$  Hz, 1H), 5.86 (ddt,  $J = 17.0, 10.3, 6.9$  Hz, 1H), 5.13 (d,  $J = 17.2$  Hz, 1H), 5.07 (d,  $J = 10.3$  Hz, 1H), 4.04 (dd,  $J = 10.9, 6.8$  Hz, 1H), 3.43 (dt,  $J = 9.1, 6.5$  Hz, 1H), 2.93 (dt,  $J = 14.6, 7.4$  Hz, 1H), 2.70 – 2.62 (m, 2H), 2.62 – 2.54 (m, 1H), 2.51 – 2.38 (m, 2H), 2.35 – 2.23 (m, 2H), 1.95 – 1.80 (m, 2H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  141.7, 141.1, 140.8, 139.7, 134.8, 133.2, 129.9, 129.4, 128.2, 128.1, 126.6, 125.9, 116.6, 78.0, 74.0, 40.9, 39.6, 31.5, 30.4, 23.0; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{23}\text{H}_{24}\text{O}$ : 317.1900, found: 317.1898.



Compound **14h** isolated as a viscous oil, 3.4:1dr (56.6 mg, 90%) or 2.8:1 dr (282mg, 90%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.35 (t,  $J = 7.1$  Hz, 1H), 7.29 (dd,  $J = 13.4, 6.5$  Hz, 1H), 7.23 (t,  $J = 6.8$  Hz, 1H), 7.18 – 7.07 (m, 2H), 6.78 (d,  $J = 7.4$  Hz, 1H), 4.05 (dt,  $J = 18.3, 9.2$  Hz, 1H), 3.61 – 3.51 (m, 1H), 2.96 – 2.85 (m, 1H), 2.70 – 2.34 (m, 3H), 2.28 (dt,  $J = 19.6, 9.8$  Hz, 1H), 2.05 – 1.85 (m, 1H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  141.6, 140.9, 140.8, 138.9, 133.5, 130.0, 129.9, 129.5, 128.2, 128.1, 126.7, 125.9, 80.9, 78.1, 72.6, 69.5, 39.6, 31.2, 30.3, 26.2, 23.0. HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{23}\text{H}_{24}\text{O}$ : 315.1744, found: 315.1741.

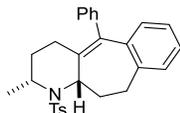


Added 1.5 equivalents of dimethyl ketal and 1.0 equivalent of alcohol **13a**. Stir at  $-40^\circ\text{C}$  until all of the alcohol was consumed. Compound **14i** was isolated as an amorphous solid, 1.2:1 dr (70.6 mg, 51%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.43 – 7.38 (m, 0.8H, *minor diastereomer*), 7.36 – 7.29 (m, 4H), 7.29 – 7.21 (m, 2H), 7.20 – 7.14 (m, 2H), 7.13 – 7.04 (m, 4H), 7.01 (dd,  $J = 13.8, 6.1$  Hz, 1H), 6.90 (d,  $J = 7.2$  Hz, 0.8H, *minor diastereomer*), 6.68 (t,  $J = 7.6$  Hz, 1H), 4.11 – 4.02 (m, 1H), 3.99 – 3.92 (m, 0.8H, *minor diastereomer*), 3.29 (d,  $J = 7.3$  Hz, 0.8H, *minor diastereomer*), 3.13 – 2.97 (m, 1H), 2.82 (d,  $J = 13.9$  Hz, 0.8H, *minor diastereomer*), 2.69 – 2.56 (m, 1H), 2.49 – 2.28 (m, 4H), 2.16 – 2.05 (m, 1.8H), 1.78 – 1.64 (m, 3H), 1.56 – 1.41 (m, 2H), 1.22 (dd,  $J = 9.4, 6.3$  Hz, 4H), 1.15 (s, 3H), 1.00 (s, 2.5H, *minor diastereomer*).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  144.5, 144.4, 143.3, 143.2, 142.1, 141.8, 141.0, 140.1, 133.4, 133.0, 130.5, 130.4, 129.9, 129.6, 128.6, 128.2, 128.2, 128.2, 127.9, 127.7, 127.5, 127.3, 126.6, 126.5, 126.4, 126.3, 126.0, 125.5, 78.5, 78.4, 71.7, 64.3, 63.3, 49.3, 46.4, 34.4, 33.2, 32.3, 31.1, 30.4, 30.3, 29.9, 26.8, 26.2, 25.8, 24.8, 23.1, 22.8, 20.9, 17.0; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{22}\text{H}_{24}\text{O}$ : 305.1900, found: 305.1897.

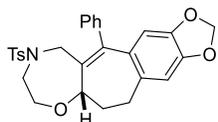


Added 1.5 equivalents of dimethyl ketal and 1.0 equivalents of alcohol **13a**. Stir at  $-40^\circ\text{C}$  for an hour, then warmed to  $0^\circ\text{C}$  until all of the alcohol was consumed. Compound **14j** was obtained as a crystalline solid, 1.6:1 dr (21.6mg, 28%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.36 (d,  $J = 6.8$  Hz, 2H), 7.31-7.27 (m, 1H), 7.18-7.11 (m, 4H), 7.01-6.96 (m, 1H), 6.82 (d,  $J = 7.8$  Hz, 1H), 5.73 (t,  $J = 7.0$  Hz, 1H), 3.67 – 3.58 (m, 1H), 3.15 (dd,  $J = 12.4, 7.9$  Hz, 1H), 2.93 – 2.84 (m, 1H), 2.67 – 2.57 (m,

1H), 2.54 – 2.38 (m, 2H), 1.52 – 1.40 (m, 3H), 1.32 (d,  $J = 8.8$  Hz, 2H), 1.08 (dd,  $J = 6.3, 3.7$  Hz, 3H), 1.04 (d,  $J = 6.1$  Hz, 3H), 0.98 (d,  $J = 6.1$  Hz, 3H, corresponds to the alpha methyl of the minor diastereomer), 0.81 (d,  $J = 6.8$  Hz, 3H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  143.8, 143.8, 143.4, 143.2, 141.8, 141.1, 140.9, 140.1, 140.0, 138.9, 130.6, 130.5, 130.2, 130.0, 128.3, 128.1, 126.9, 126.7, 126.6, 125.5, 124.8, 122.4, 122.4, 68.0, 67.1, 39.4, 39.1, 33.7, 29.9, 29.8, 29.7, 28.2, 27.9, 23.3, 23.1, 23.0, 22.8, 20.5; m.p = 108-110°C; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{24}\text{H}_{28}\text{O}$ : 333.2213, found: 333.2210.



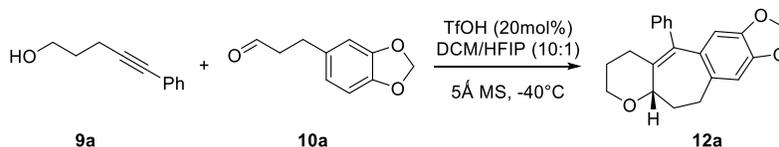
After addition of TfOH, stirred at  $-40$  °C for 30 minutes before warming to  $0$  °C until all the starting material was consumed. Compound **16b** was isolated as an amorphous white solid, >20:1 dr (50 mg, 56%).  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  8.04 (d,  $J = 7.7$  Hz, 1H), 7.64 (d,  $J = 7.9$  Hz, 2H), 7.43 (d,  $J = 7.3$  Hz, 1H), 7.37 (t,  $J = 7.6$  Hz, 1H), 7.33 – 7.24 (m, 4H), 7.24 – 7.15 (m, 4H), 7.08 (d,  $J = 7.5$  Hz, 2H), 4.43 (s, 1H), 3.32 (t,  $J = 4.5$  Hz, 1H), 2.40 (dd,  $J = 14.8, 6.3$  Hz, 1H), 2.31 (d,  $J = 6.5$  Hz, 3H), 2.15 (dddd,  $J = 28.8, 16.9, 14.6, 4.9$  Hz, 6H), 1.38 (dd,  $J = 12.1, 5.8$  Hz, 1H), 1.30 – 1.19 (m, 2H), 1.12 (d,  $J = 6.8$  Hz, 3H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  145.0, 143.7, 142.2, 136.1, 135.2, 133.7, 129.5, 128.4, 128.2, 128.0, 126.6, 125.8, 124.8, 122.5, 122.2, 51.5, 48.4, 32.4, 30.6, 23.5, 21.4, 18.7, 17.8. HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{28}\text{H}_{29}\text{NO}_2\text{S}$ : 444.1992, found: 444.1984.

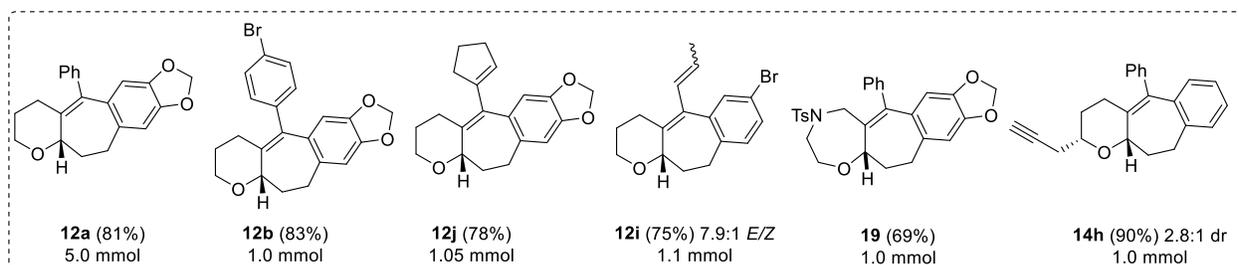


Compound **19** was isolated as a white solid (394 mg, 89%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.62 (d,  $J = 8.1$  Hz, 2H), 7.34 (d,  $J = 6.8$  Hz, 3H), 7.30 (d,  $J = 8.0$  Hz, 3H), 7.22 (d,  $J = 8.5$  Hz, 1H), 7.18 (t,  $J = 7.3$  Hz, 1H), 7.13 (t,  $J = 7.4$  Hz, 1H), 6.80 (d,  $J = 7.7$  Hz, 1H), 4.37 (d,  $J = 11.5$  Hz, 1H), 4.01 (dd,  $J = 11.7, 6.9$  Hz, 1H), 3.94 – 3.86 (m, 1H), 3.69 (d,  $J = 13.4$  Hz, 1H), 3.54 (d,  $J = 11.3$  Hz, 2H), 3.00 – 2.92 (m, 1H), 2.83 – 2.73 (m, 1H), 2.56 (dd,  $J = 13.1, 7.1$  Hz, 1H), 2.48 (dd,  $J = 12.9, 6.8$  Hz, 1H), 2.43 (s, 3H), 2.03 (dd,  $J = 20.4, 11.8$  Hz, 1H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  143.5, 141.8, 140.7, 140.4, 139.6, 134.7, 134.0, 130.5, 129.8, 129.6, 128.0, 127.8, 127.7, 127.7, 126.3, 79.3, 68.8, 49.4, 47.9, 40.1, 30.0, 21.5; m.p = compound turns brown after  $95$  °C; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{28}\text{H}_{27}\text{NO}_5\text{S}$ : 490.1683, found: 490.1676.

### Greater than 1.0 mmol Scale Experiments:

#### Scheme S6. General Procedure for Carbocyclizations (>1.0 mmol scale)



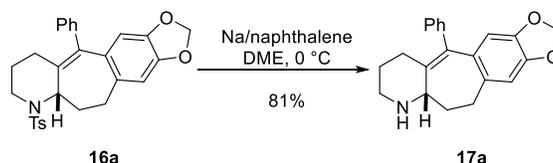


**Scheme S6.** Greater than 1.0 mmol Scale Substrates

Oven dried, 5 Å molecular sieves (1.5 g) were transferred to a 100 mL flask equipped with a stir bar and dried further under vacuum while heating over a Bunsen burner for about 3 minutes. The reaction vessel was allowed to cool under a flow of argon before adding DCM/HFIP (10:1) (50 mL), followed by the alcohol **9a** (961 mg, 6.0 mmol, 1.2 equivalents) and aldehyde **10a** (891 mg, 5.0 mmol, 1.0 equivalents). The reaction mixture was cooled to -40 °C before adding triflic acid (TfOH) (88 µL, 20 mol%) dropwise. Mixture was allowed to stir at -40 °C until all of the limiting reagent was consumed by TLC (using 20% ethyl acetate/hexanes mixture as mobile phase and *p*-anisaldehyde to stain the plates). After completion, the reaction was quenched by adding solid sodium bicarbonate (1.5 g) and diluting with ether (Et<sub>2</sub>O). The quenched reaction mixture was filtered through a silica plug, eluted with Et<sub>2</sub>O, and concentrated under vacuum. Crude was purified by column chromatography, using 5% ethyl acetate/hexane as mobile phase. Pure product **12a** was obtained as a pale yellow amorphous solid (1.3 g, 81%).

#### Procedure for Deprotection of *N*-Tosyl Piperidines:

#### Scheme S7. General Procedure for Piperidine Detosylation



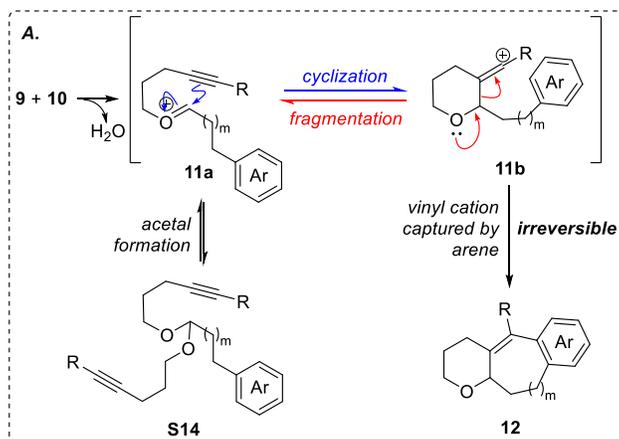
In a flame-dried 25 mL round bottom flask under argon, prepared a 1.0M solution of naphthalene (640 mg, 5.0 mmol) in dry dimethoxyethane (DME, 5.0 mL) and cooled to 0 °C. Na (140 mg, 6.0 mmol) that had been cut into small cubes was added and the mixture was stirred vigorously at 0 °C for about 15 minutes, then warmed up to room temperature and allowed to stir for 1 hour. The deep green solution (162 µL, 0.162 mmol, 4.0 equivalents) was added dropwise to a 0.2 M solution of sulfonamide **16a** (19.2 mg, 0.0405 mmol) in DME (0.4 mL). The drops were added slowly, allowing time for the coloration to disappear before the next addition. After 5 minutes, 1M HCl was added to quench the reaction. Acidic mixture was washed with Et<sub>2</sub>O (2 x 15 mL) and the ethereal layer was discarded. The aqueous layer was then basified by adding solid NaOH and extracted with Et<sub>2</sub>O (3 x 15 mL), dried over MgSO<sub>4</sub>, filtered, and concentrated under vacuum to afford amine **17a** as a yellow oil, no further purification needed. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.31 (t, *J* = 7.4 Hz, 2H), 7.24 (s, 1H), 7.10 (d, *J* = 7.5 Hz, 2H), 6.71 (s, 1H), 6.25 (s, 1H), 5.86 (s, 2H), 3.42 – 3.36 (m, 1H), 3.00 (dt, *J* = 12.6, 6.2 Hz, 1H), 2.75 (ddd, *J* = 25.4, 13.2, 7.4 Hz, 2H), 2.60 (dt, *J* = 14.9, 4.8 Hz, 1H), 2.49 (dd, *J* = 13.2, 6.1 Hz, 1H), 2.37 – 2.19 (m, 3H), 2.04 (s, 1H), 1.86 – 1.76 (m, 1H), 1.65 (td, *J* = 12.0, 5.9 Hz, 1H). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 145.9, 145.8, 141.7, 138.6, 135.0, 134.8, 134.1, 129.9, 128.1, 126.6, 109.7, 108.5, 100.7, 55.8, 41.2, 40.4, 31.3, 25.4, 24.8. HRMS (ESI) *m/z*: [M+H]<sup>+</sup> Calculated for C<sub>21</sub>H<sub>21</sub>NO<sub>2</sub>: 320.1645, found: 320.1638.

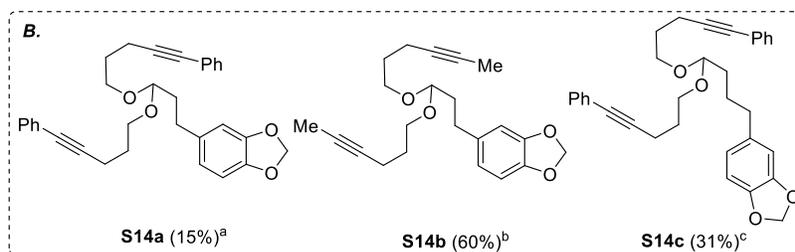
## Explanation of Acetal Byproduct (Observed in the absence of HFIP, and for cases 12k and 12n):

In less efficient alkynyl Prins reactions, it is notable that acetal **S14** and tricycle **12** are the only products observed in significant quantities. During optimization, the acetal **S14a** was isolated in 15% yield when HFIP was left out of the reaction mixture. Using the optimized conditions (HFIP present), the acetal was also observed in two other cases, namely **12k** and **12n**. For alkyne **9k** (R=Me; n=1), the reaction produces the corresponding acetal **S14b** exclusively when left at -40°C. Higher temperatures are required for the alkynyl Prins/carbocyclization to occur. For the extended chain arenyne **9n** (R=Ph; n=2) acetal **S14c** is isolated as the major product when the reaction is left to stir at -40 °C. Once again, warming is required to push the acetal to undergo cyclization (Scheme S8B). The combination of HFIP<sup>20</sup> and the adjacent  $\pi$ -system in R (arene/alkene) helps stabilize **11b-2** (-1.2 kcal/mol relative to the oxocarbenium, Scheme S8C), improving the efficiency of arene capture and disfavoring the reversibility of the Prins, back to **11a-2**. The vinyl cation produced from **9k** (R = Me), however, has no adjacent  $\pi$ -system, which could account for the reduced efficiency (+6.4 kcal/mol relative to the oxocarbenium, Scheme S8C).<sup>21</sup> For the **12n** case (m = 2), it seems logical that the longer tether makes capture of the vinyl cation less efficient. This, when **11b** is either unstable, or has a poorly aligned arene, oxocarbenium **11a** is trapped by a second molecule of alcohol **9**. We also note that in these inefficient cases, numerous other undesired byproducts form over the course of the experiment, ultimately compromising the yield of the reaction.

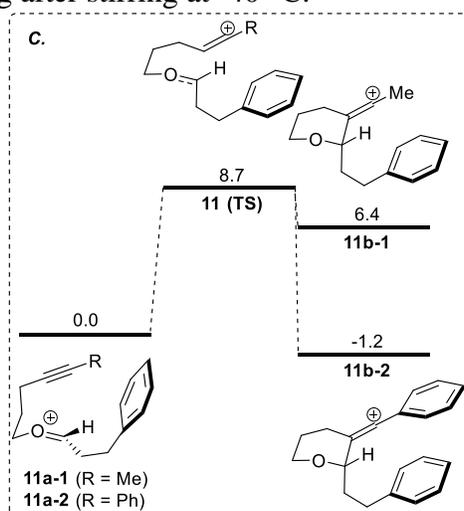
A possible explanation for these observations is shown in Scheme S8. The reversal of the alkyne/oxocarbenium ring closure is similar to the Eshenmoser class of fragmentations,<sup>22</sup> which could establish an equilibrium between oxocarbenium **11a** and vinyl cation **11b** (Scheme S8A). This is supported by DFT calculations, which show that generation of the vinyl cation **11b-1**, is endergonic for **9k**, and slightly exergonic in the case of **11b-2** for arenyne **9a** (Scheme S8C). Either way, the energetic barrier is readily available in both directions. At the vinyl cation stage **11b**, it is plausible for the oxacycle to open and regenerate the oxocarbenium **11a**, or to undergo carbocyclization after aryl trapping of the vinyl cation, terminating the process.

**Scheme S8. A.** Explanation of Acetal Byproducts Observed. **B.** Acetals Isolated **C.** DFT Backing for Plausible Reversibility of Alkynyl Prins Step (Theory level: M062X/Def2TZVP, Solvation: CPCM: Dichloromethane, Energy in kcal/mol).





**a)** Isolated from the reaction of **9a** with aldehyde **10a**, in the absence of HFIP (See Table 1, Entry 1) **b)** Isolated from the reaction of **9k** with aldehyde **10a**, using 40 mol% TfOH and quenching after stirring at -40 °C **c)** Isolated from the reaction of **9n** with the longer-tether aldehyde, using 40 mol% TfOH and quenching after stirring at -40 °C.



**Compound S14a.** Acetal isolated as a yellow solid after preparatory TLC with 20% ethyl acetate in hexanes as the mobile phase (14 mg, 15%). **S14a** was observed as the major byproduct of subjecting **9a** and **10a** to 20 mol% TfOH in dichloromethane (no HFIP), in the presence of molecular sieves. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.37 (s, 3H), 7.26 (s, 9H), 6.72 – 6.58 (m, 3H), 5.90 (d, *J* = 4.3 Hz, 2H), 4.51 (dd, *J* = 12.8, 5.0 Hz, 1H), 3.76 (d, *J* = 5.5 Hz, 1H), 3.64 – 3.55 (m, 1H), 2.62 (s, 2H), 2.58 – 2.50 (m, 3H), 1.98 – 1.81 (m, 5H), 1.58 (s, 1H), 1.22 (t, *J* = 5.6 Hz, 2H). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 147.6, 145.6, 135.5, 135.5, 131.5, 128.2, 127.6, 123.9, 121.1, 108.9, 108.1, 102.4, 102.2, 100.7, 89.4, 81.0, 64.0, 63.8, 61.2, 35.3, 35.2, 30.7, 29.0, 16.3, 15.4. HRMS (ESI) *m/z*: [M+H]<sup>+</sup> Calculated for C<sub>32</sub>H<sub>32</sub>O<sub>4</sub>: 481.2374, found: 481.2366.

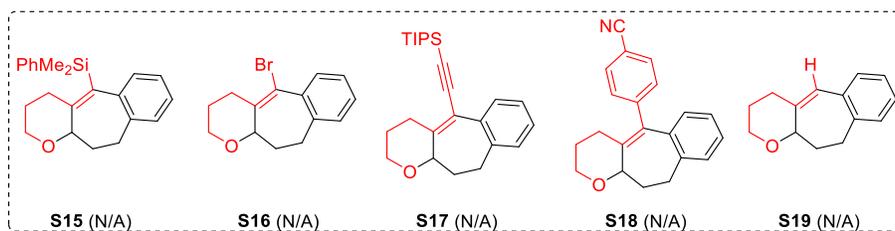
**Compound S14b.** Acetal isolated as a yellow oil after preparatory TLC with 20% ethyl acetate in hexanes as the mobile phase (43 mg, 60%). **S14b** was observed as the major product when the Prins/carbocyclization with alcohol **9n** was quenched after stirring at -40 °C and full consumption of the aldehyde was observed by TLC. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 6.72 (d, *J* = 7.9 Hz, 1H), 6.69 (s, 1H), 6.64 (d, *J* = 7.8 Hz, 1H), 5.91 (s, 2H), 4.46 (t, *J* = 5.7 Hz, 1H), 3.66 (dt, *J* = 9.2, 6.2 Hz, 2H), 3.50 (dt, *J* = 9.2, 6.2 Hz, 2H), 2.66 – 2.56 (m, 2H), 2.24 (dd, *J* = 9.1, 6.7 Hz, 4H), 1.88 (dd, *J* = 15.1, 6.2 Hz, 2H), 1.77 (d, *J* = 1.9 Hz, 6H), 1.73 (dd, *J* = 13.2, 6.6 Hz, 3H). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 147.5, 145.6, 135.6, 121.1, 108.9, 108.1, 102.2, 100.7, 78.4, 75.8, 64.0, 35.2, 30.7, 29.2, 15.6, 3.4. HRMS (ESI) *m/z*: [M+H]<sup>+</sup> Calculated for C<sub>22</sub>H<sub>28</sub>O<sub>4</sub>: 357.2061, found: 357.2059.

**Compound S14c.** Acetal isolated as a yellow oil after preparatory TLC with 20% ethyl acetate in hexanes as the mobile phase (19 mg, 27%). **S14c** was observed as the major product when the

reaction was quenched after stirring at  $-40\text{ }^{\circ}\text{C}$  and full consumption of the aldehyde was observed by TLC.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.44 – 7.35 (m, 2H), 7.29 – 7.26 (m, 2H), 6.84 (s, 1H), 6.54 (s, 1H), 5.88 (d,  $J = 6.5$  Hz, 2H), 4.31 (t,  $J = 4.4$  Hz, 1H), 3.68 (dt,  $J = 9.1, 6.2$  Hz, 1H), 3.58 – 3.47 (m, 1H), 2.72 (dt,  $J = 16.3, 5.4$  Hz, 1H), 2.61 (dd,  $J = 14.8, 7.0$  Hz, 1H), 2.45 (t,  $J = 6.8$  Hz, 2H), 2.00 – 1.63 (m, 9H).  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  146.9, 145.7, 131.5, 131.0, 130.2, 128.1, 127.9, 127.5, 124.0, 119.9, 108.9, 108.4, 100.6, 90.0, 80.8, 75.5, 68.0, 29.4, 29.4, 27.9, 25.6, 19.2, 19.0. HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{33}\text{H}_{34}\text{O}_4$ : 495.2530, found: 495.2522.

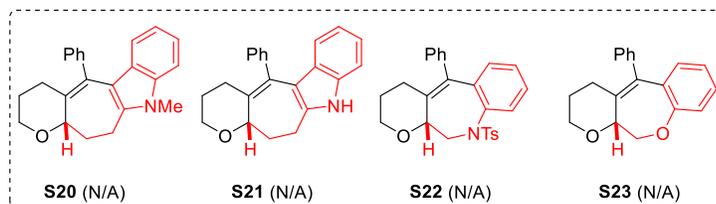
### Unsuccessful Carbocyclization Substrates:

Alkyne alcohols (highlighted in red) **S6-S9a** failed to give carbocyclization adducts **S15-S18**. Complex product mixtures were observed after subjecting these alcohols to the reaction conditions. Terminal alkynes also did not react cleanly and no **S18** was isolated.



**Scheme S9.** Products not accessible through this approach

Carbocyclization adducts **S20-S23** did not form under the reaction conditions (aldehyde portion highlighted in red). Aldehyde decomposition was observed in the cases of **S20** and **S21**. In the cases **S22** and **S23**, where the aldehyde had a heteroatom in the tether, recovery of unreacted starting material was observed and decomposition to complex mixtures.



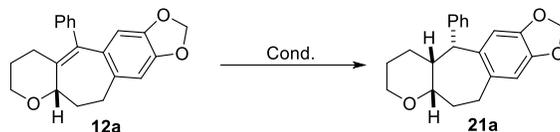
**Scheme S10.** Products not accessible through this approach

### Diversification of Carbocyclization Products:

#### Hydrogenations:

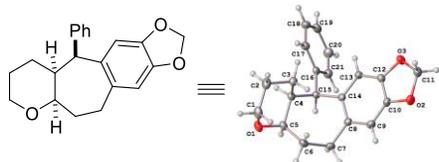
A few different hydrogenation catalysts were explored. Since the olefin at hand was tetrasubstituted, we started with Crabtree catalyst  $[\text{Ir}(\text{cod})(\text{PCy}_3)(\text{pyr})]\text{PF}_6$ , which is known to reduce highly hindered alkenes.<sup>23, 24</sup> These conditions led to recovery of starting material **12a** (Entry 1, Table S1). More recent conditions by Chirk et al. are successful at reducing difficult olefins using a Ni catalyst that is generated *in situ* after mixing  $\text{Ni}(\text{octanoate})_2$  with HBPIn and an  $\alpha$ -diimine ligand ( $^i\text{PrDI}$ ) in benzene.<sup>25</sup> These conditions, however, failed to afford hydrogenated product **21a**. To our delight, using 20 mol% of 5% palladium on carbon (Pd/C) with 55 atmospheres of hydrogen. See below.

**Table S1.** Screen of Catalysts for Hydrogenation of **12a**

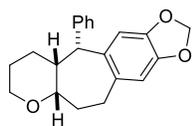


| Entry           | Catalyst   | Solvent                       | H <sub>2</sub> (atm) | Result            |
|-----------------|--|-------------------------------|----------------------|-------------------|
| 1 <sup>23</sup> | [Ir(cod)(PCy <sub>3</sub> )(pyr)]PF <sub>6</sub>     | DCM                           | 1-35                 | 100% <b>12a</b>   |
| 2 <sup>25</sup> | Ni(octanoate) <sub>2</sub> , HBpin,<br><i>i</i> PrDI | C <sub>6</sub> H <sub>6</sub> | 1-4                  | 100% <b>12a</b>   |
| 3 <sup>a</sup>  | Pd/C (20 mol %)                                      | MeOH/DCM                      | 55                   | Quant. <b>21a</b> |

**General Procedure for hydrogenation:** To a scintillation vial charged with stir bar and compound **12a** (32 mg, 0.1 mmol), added DCM (0.3 mL) followed by Pd/C (42 mg, 0.02 mmol Pd, 5% Pd on C). Some additional DCM was added to wash the Pd/C off the sides of the vial. Added MeOH (1.7 mL) and the vial was placed in a bomb reactor, on a stir plate. Pressure was brought up to 55 atm, then purged, refilled back up to 55 atm, and allowed to stir at room temperature for 48 hours. After this time, the hydrogen gas was released and the mixture was carefully filtered through celite, washed with DCM, concentrated, and purified by preparatory TLC using 30% EtOAc in hexanes as the mobile phase. Product was obtained as a 1:1 mixture of separable diastereomers (32 mg, 100% combined yield).

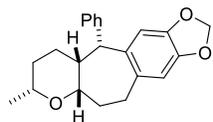


Compound **21a-1** was obtained as a white, crystalline solid (15.4 mg, 48%). Recrystallized after dissolving in hot methanol and allowing to cool to room temperature in an uncapped vial. <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.24 (d, *J* = 7.2 Hz, 2H), 7.19 (s, 3H), 6.62 (s, 1H), 6.49 (s, 1H), 5.88 (d, *J* = 5.9 Hz, 2H), 4.02 (s, 1H), 3.90 (d, *J* = 10.2 Hz, 1H), 3.28 (t, *J* = 10.7 Hz, 2H), 3.00 (d, *J* = 11.7 Hz, 1H), 2.75 (dd, *J* = 15.7, 8.8 Hz, 1H), 2.34 – 2.25 (m, 1H), 2.22 (d, *J* = 10.6 Hz, 1H), 1.92 (d, *J* = 12.2 Hz, 1H), 1.75 (tt, *J* = 24.6, 12.3 Hz, 4H), 1.66 – 1.54 (m, 2H). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 145.9, 145.7, 142.0, 134.5, 129.7, 127.8, 126.1, 111.5, 110.7, 100.8, 78.7, 67.9, 56.1, 43.7, 34.3, 31.1, 30.7, 30.3, 27.2, 26.6; m.p = 127-130°C; HRMS (ESI) *m/z*: [M+H]<sup>+</sup> Calculated for C<sub>21</sub>H<sub>22</sub>O<sub>3</sub>: 323.1642, found: 323.1639.



Compound **21a-2** was obtained as an amorphous solid (16.6 mg, 52% yield). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.35 (t, *J* = 7.3 Hz, 2H), 7.28 (d, *J* = 7.1 Hz, 2H), 6.67 (s, 1H), 5.98 (s, 1H), 5.84 (d, *J* = 5.0 Hz, 2H), 4.24 (s, 1H), 4.09 – 4.01 (m, 1H), 3.62 (d, *J* = 11.2 Hz, 2H), 2.92 (t, *J* = 13.5 Hz, 1H), 2.81 (d, *J* = 12.5 Hz, 1H), 2.69 (dd, *J* = 14.3, 6.5 Hz, 1H), 2.15 – 2.02 (m, 1H), 1.94 – 1.87 (m, 1H), 1.85 – 1.75 (m, 1H), 1.70 (d, *J* = 13.8 Hz, 1H), 1.51 (d, *J* = 12.9 Hz, 1H). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 145.6, 145.3, 143.7, 135.6, 134.4, 128.6, 128.5, 126.3, 109.4, 108.9, 100.6, 78.3,

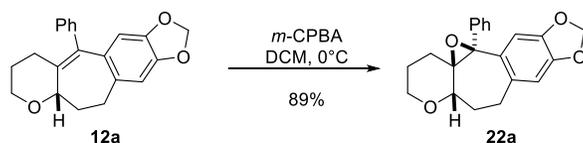
60.8, 46.7, 40.9, 30.8, 26.7, 26.1, 22.4; HRMS (ESI)  $m/z$ :  $[M+H]^+$  Calculated for  $C_{21}H_{22}O_3$ : 323.1642, found: 323.1639.



Compound **S23** was isolated as an amorphous solid (as a 5:1 mixture of diastereomers) after subjecting **14a** to the hydrogenation conditions (13.7 mg, 78%).  $^1H$  NMR (500 MHz,  $CDCl_3$ )  $\delta$  7.34 (t,  $J = 7.5$  Hz, 2H), 7.28 (s, 1H), 7.24 (s, 1H), 6.66 (s, 1H), 5.98 (s, 1H), 5.84 (d,  $J = 6.3$  Hz, 2H), 4.27 (s, 1H), 4.08 (dd,  $J = 11.2, 5.4$  Hz, 1H), 3.69 (dd,  $J = 9.4, 6.0$  Hz, 1H), 2.91 (t,  $J = 13.5$  Hz, 1H), 2.75 (d,  $J = 13.3$  Hz, 1H), 2.68 (dd,  $J = 14.4, 6.5$  Hz, 1H), 2.07 (dd,  $J = 25.0, 12.3$  Hz, 1H), 1.95 (dd,  $J = 12.5, 6.3$  Hz, 1H), 1.69 (d,  $J = 13.6$  Hz, 1H), 1.58 (d,  $J = 10.5$  Hz, 1H), 1.41 (dd,  $J = 17.0, 5.0$  Hz, 1H), 1.10 (d,  $J = 5.8$  Hz, 3H), 0.92 (ddd,  $J = 26.9, 13.4, 3.5$  Hz, 1H);  $^{13}C$  NMR (126 MHz,  $CDCl_3$ )  $\delta$  145.6, 145.3, 143.7, 135.7, 134.5, 129.5, 128.7, 128.5, 127.9, 126.3, 109.4, 108.8, 100.6, 79.0, 65.5, 46.4, 40.5, 34.2, 30.9, 30.8, 26.9, 22.6, 22.3; HRMS (ESI)  $m/z$ :  $[M+H]^+$  Calculated for  $C_{22}H_{24}O_3$ : 337.1798, found: 337.1794.

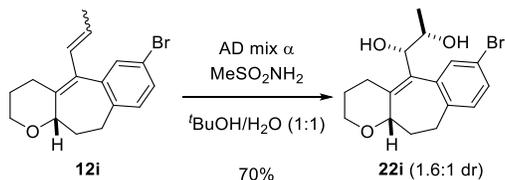
#### Oxidations:

#### **Scheme S11.** General Procedure for *m*-CPBA Oxidation of **12a**<sup>26</sup>



To a solution of **12a** (32 mg, 0.1 mmol) in DCM (0.4 mL) at 0°C, added *m*-CPBA (26 mg, 0.15 mmol) in one portion. The reaction was allowed to stir at 0°C until all the starting material had been consumed by TLC (using 20% EtOAc in hexanes, and *p*-anisaldehyde to stain the plates). Reaction was diluted with more DCM (1.0 mL), washed with 10% NaOH (2 x 15 mL), then washed with Brine (20 mL), dried over  $MgSO_4$ , filtered, and concentrated to afford **22a** as a white solid (29.9 mg, 89%). No purification needed. Recrystallized after dissolving in hot methanol and slowly cooling to room temperature in an uncapped vial.  $^1H$  NMR (500 MHz,  $CDCl_3$ )  $\delta$  7.28 (dd,  $J = 13.0, 6.2$  Hz, 5H), 6.70 (s, 1H), 6.65 (s, 1H), 5.89 (d,  $J = 9.9$  Hz, 2H), 3.79 (d,  $J = 11.8$  Hz, 1H), 3.70 (t,  $J = 11.4$  Hz, 1H), 3.50 – 3.41 (m, 1H), 3.04 (td,  $J = 13.3, 7.3$  Hz, 1H), 2.70 (dd,  $J = 13.8, 6.5$  Hz, 1H), 2.28 (td,  $J = 12.2, 7.5$  Hz, 1H), 1.98 (dd,  $J = 13.2, 9.3$  Hz, 1H), 1.89 – 1.77 (m, 1H), 1.66 (dt,  $J = 19.9, 7.2$  Hz, 2H), 1.41 (d,  $J = 14.6$  Hz, 1H).  $^{13}C$  NMR (126 MHz,  $CDCl_3$ )  $\delta$  147.4, 146.9, 137.9, 131.8, 130.4, 129.8, 128.0, 127.4, 127.0, 109.1, 108.5, 101.0, 77.5, 66.0, 65.4, 60.9, 29.4, 24.7, 23.8, 22.8; m.p = 143-146°C; HRMS (ESI)  $m/z$ :  $[M+H]^+$  Calculated for  $C_{21}H_{20}O_4$ : 337.1435, found: 337.1431.

#### **Scheme S12.** AD-mix $\alpha$ Dihydroxylation of **12i**

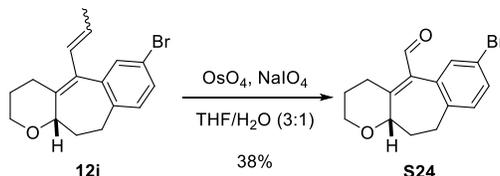


A mixture of **12i** (32 mg, 0.1 mmol), methanesulfonamide (MeSO<sub>2</sub>NH<sub>2</sub>) (9.5 mg, 0.1 mmol), and AD-mix  $\alpha$  (140 mg, 1.4g/mmol) was placed in a scintillation vial charged with a stir bar. Added <sup>t</sup>BuOH (0.5 mL) followed by water (0.5 mL), and the reaction mixture was allowed to stir vigorously under air for 4 days (until the starting material had been fully consumed by TLC, using 20% EtOAc in hexanes, and *p*-anisaldehyde to stain the plates). After this time, added solid sodium sulfite (Na<sub>2</sub>SO<sub>3</sub>) (140 mg) and vigorously stirred for an hour. Added more water (2 mL) extracted with EtOAc (4 x 15 mL), dry over MgSO<sub>4</sub>, filter, and concentrate. The crude was purified by preparatory TLC, using 50% EtOAc in hexanes as the mobile phase to afford **22i** as a 1.6:1 mixture of separable diastereomers (70% combined yield).

Compound **22i-1** was isolated as an amorphous solid (9.6 mg, 27%). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.13 (d, *J* = 8.0 Hz, 1H), 7.10 (s, 1H), 6.90 (d, *J* = 8.1 Hz, 1H), 4.21 (d, *J* = 8.4 Hz, 1H), 3.95 – 3.86 (m, 1H), 3.75 – 3.67 (m, 1H), 3.53 (dd, *J* = 11.1, 7.5 Hz, 1H), 3.10 (dd, *J* = 19.3, 8.4 Hz, 1H), 2.99 (dd, *J* = 12.7, 5.9 Hz, 1H), 2.42 – 2.28 (m, 4H), 2.28 – 2.17 (m, 2H), 2.03 (dd, *J* = 19.5, 11.6 Hz, 1H), 1.92 (dt, *J* = 17.1, 8.7 Hz, 1H), 1.50 (dd, *J* = 12.0, 4.1 Hz, 1H), 0.88 (t, *J* = 9.5 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  143.3, 141.4, 139.4, 130.3, 130.2, 130.1, 129.3, 120.2, 79.6, 76.8, 71.0, 63.0, 38.7, 30.3, 29.7, 24.0, 21.5, 19.2; HRMS (ESI) *m/z*: [M+H]<sup>+</sup> Calculated for C<sub>17</sub>H<sub>21</sub>BrO<sub>3</sub>: 353.0747, found: 353.0783.

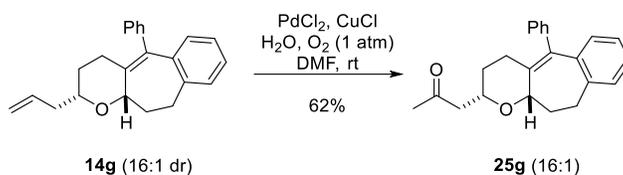
Compound **22i-2** was isolated as an amorphous solid (15.2 mg, 43%). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  7.95 (d, *J* = 1.4 Hz, 1H), 7.32 – 7.27 (m, 1H), 7.03 (d, *J* = 8.0 Hz, 1H), 4.57 (d, *J* = 8.8 Hz, 1H), 3.89 – 3.77 (m, 3H), 3.69 – 3.59 (m, 1H), 3.34 (dt, *J* = 11.6, 7.8 Hz, 1H), 2.91 (dd, *J* = 13.8, 6.1 Hz, 1H), 2.84 (s, 1H), 2.44 (dd, *J* = 16.6, 10.1 Hz, 4H), 2.40 – 2.24 (m, 3H), 2.20 – 2.12 (m, 1H), 2.08 (dd, *J* = 18.7, 10.5 Hz, 1H), 1.67 (ddd, *J* = 24.3, 12.2, 5.4 Hz, 2H), 0.92 (d, *J* = 6.2 Hz, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  144.3, 139.4, 138.9, 131.5, 130.1, 129.9, 127.9, 119.8, 76.3, 75.2, 69.6, 62.8, 38.1, 30.3, 29.7, 29.5, 24.2, 20.9, 18.8; HRMS (ESI) *m/z*: [M+H]<sup>+</sup> Calculated for C<sub>17</sub>H<sub>21</sub>BrO<sub>3</sub>: 353.0747, found: 353.0783.

### Scheme S13. Oxidative Cleavage of **12i**



To a solution of **12i** (40 mg, 0.13 mmol) in THF/H<sub>2</sub>O (3:1, 0.65 mL), added OsO<sub>4</sub> (16  $\mu$ L, 4% H<sub>2</sub>O solution, 2.0 mol%) followed by NaIO<sub>4</sub> (13 mg/0.5 hr for 2 hrs; 88 mg, 0.26 mmol total). The reaction was allowed to stir for 48 hrs, then added saturated Na<sub>2</sub>S<sub>2</sub>O<sub>3</sub> and stirred for 30 minutes. After this time, added more water (15 mL), extracted with EtOAc (3 x 15 mL), dried over MgSO<sub>4</sub>, filtered, and concentrated. The crude material was purified by preparatory TLC using 30% EtOAc in hexanes to afford aldehyde **S24** as a foamy solid (14 mg, 38%). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>)  $\delta$  10.24 (s, 1H), 7.39 (s, 1H), 7.36 (d, *J* = 8.1 Hz, 1H), 7.08 (d, *J* = 8.0 Hz, 1H), 4.00 – 3.90 (m, 2H), 3.54 (dd, *J* = 14.4, 6.5 Hz, 1H), 3.35 (dd, *J* = 19.7, 8.1 Hz, 1H), 2.70 (dd, *J* = 23.6, 10.3 Hz, 1H), 2.52 (d, *J* = 10.6 Hz, 3H), 2.24 (td, *J* = 15.2, 7.3 Hz, 2H), 1.82 – 1.70 (m, 1H), 0.87 (d, *J* = 6.4 Hz, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>)  $\delta$  187.7, 164.4, 139.0, 135.1, 132.5, 131.6, 131.0, 129.8, 119.9, 77.1, 62.9, 38.0, 28.9, 23.7, 19.9. HRMS (ESI) *m/z*: [M+H]<sup>+</sup> Calculated for C<sub>15</sub>H<sub>15</sub>BrO<sub>2</sub>: 307.0328, found: 307.0327.

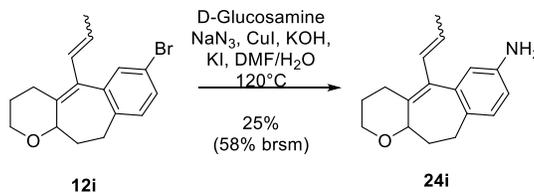
### Scheme S14. Wacker Oxidation of **14g**<sup>27</sup>



A solution of **14g** (50 mg, 0.158 mmol) in DMF (3.0 mL) was sparged with oxygen for 5 minutes. After this time, added PdCl<sub>2</sub> (6 mg, 32 μmol, 20 mol%), CuCl (16 mg, 0.158 mmol, 1.0 equiv.), and H<sub>2</sub>O (128 μL, 7.11 mmol, 45 equiv.). The reaction was allowed to stir under an oxygen atmosphere, at room temperature until all of the starting material had been consumed by TLC (using 20% EtOAc in hexanes, and *p*-anisaldehyde to stain the plates). The reaction mixture was diluted with EtOAc (20 mL), filtered through celite, washed with water (3 x 20 mL), then Brine, dried over MgSO<sub>4</sub>, filtered, and concentrated under vacuum to give **25g** as a viscous yellow oil (32 mg, 62% yield). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.33 (t, *J* = 7.4 Hz, 2H), 7.28 (d, *J* = 7.3 Hz, 1H), 7.21 (d, *J* = 7.3 Hz, 1H), 6.76 (d, *J* = 7.5 Hz, 1H), 4.02 (dd, *J* = 10.6, 7.2 Hz, 1H), 3.80 (dd, *J* = 14.5, 6.7 Hz, 1H), 2.93 – 2.82 (m, 1H), 2.77 (dd, *J* = 16.0, 7.9 Hz, 1H), 2.64 – 2.48 (m, 4H), 2.41 (dd, *J* = 22.7, 10.0 Hz, 1H), 2.25 – 2.19 (m, 1H), 2.17 (d, *J* = 6.5 Hz, 3H), 1.95 (dq, *J* = 10.9, 7.4 Hz, 1H), 1.74 (dd, *J* = 21.1, 10.0 Hz, 1H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 207.0, 141.6, 140.9, 140.7, 138.7, 133.7, 130.0, 129.9, 129.6, 128.2, 128.1, 126.7, 126.7, 126.0, 78.2, 70.7, 50.4, 39.5, 31.6, 30.8, 30.3, 23.0; HRMS (ESI) *m/z*: [M+H]<sup>+</sup> Calculated for C<sub>23</sub>H<sub>24</sub>O<sub>2</sub> : 333.1849, found: 333.1847.

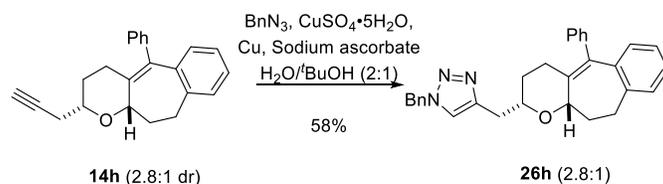
#### Cross Coupling and Click Reactions:

#### **Scheme S15.** Ammonia Cross-Coupling with **12i**



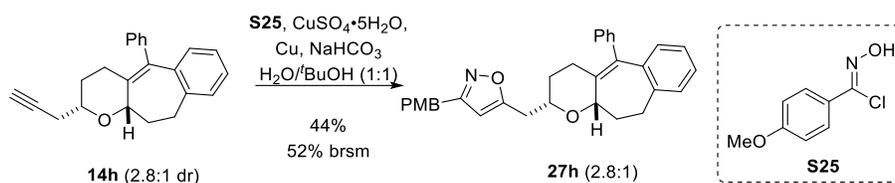
A mixture of D-glucosamine (4.4 mg, 0.02 mmol, 0.1 equiv.), NaN<sub>3</sub> (39 mg, 0.6 mmol, 3.0 equiv.), CuI (3.8 mg, 0.02 mmol, 0.1 equiv.), KOH (11.2 mg, 0.2 mmol, 1 equiv.), and KI (33.2 mg, 0.2 mmol, 1 equiv.) in DMF/H<sub>2</sub>O (v/v 1:1, 0.6 mL) was added to a sealable tube. The mixture was sparged with argon for about 10 minutes then added **12i** (63.8 mg, 0.2 mmol, 1.0 equiv.) and heated to 120°C for 18 hours. After this time, the reaction was cooled to room temperature, diluted with water, extracted with DCM (3 x 15 mL), washed with brine, dried over MgSO<sub>4</sub>, filtered, and concentrated under vacuum. The crude material was purified by preparatory TLC using 30% EtOAc in hexanes with 1% Et<sub>3</sub>N to give 13.0 mg of **24i** as an orange oil, and 35.8 mg of recovered **12i** (25% yield, 58% yield based on recovered starting material). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 6.96 (t, *J* = 7.2 Hz, 1H), 6.58 (s, 1H), 6.52 (dd, *J* = 8.7, 6.9 Hz, 2H), 5.56 (dq, *J* = 13.5, 6.6 Hz, 1H), 3.94 – 3.86 (m, 1H), 3.86 – 3.78 (m, 1H), 3.53 (bs, 2H), 3.35 (dt, *J* = 11.4, 7.6 Hz, 1H), 2.82–2.84 (m, 2H), 2.54 – 2.32 (m, 4H), 2.24 (tt, *J* = 12.9, 6.5 Hz, 1H), 2.10 (dd, *J* = 19.8, 11.7 Hz, 1H), 2.00 (dd, *J* = 13.7, 6.7 Hz, 1H), 1.78 (d, *J* = 6.5 Hz, 3H), 1.72 – 1.58 (m, 2H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 143.9, 139.3, 135.6, 131.2, 130.2, 128.7, 128.2, 128.0, 116.4, 114.0, 76.7, 62.8, 37.8, 28.9, 24.2, 20.8, 18.6; HRMS (ESI) *m/z*: [M+H]<sup>+</sup> Calculated for C<sub>15</sub>H<sub>15</sub>BrO<sub>2</sub> : 307.0328, found: 307.1804.

#### **Scheme S16.** Click Reaction with **14h** and Benzyl Azide for Triazole Synthesis<sup>28</sup>



A mixture of **14h** (63 mg, 0.2 mmol),  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  (1 mg, 4  $\mu\text{mol}$ ), Cu powder (1 mg, 2  $\mu\text{mol}$ ), sodium ascorbate (4 mg, 20  $\mu\text{mol}$ ), and benzyl azide ( $\text{BnN}_3$ ) (32  $\mu\text{L}$ , 40 mg, 0.3 mmol), in a 2:1 mixture of  $\text{H}_2\text{O}/\text{BuOH}$ , was stirred at room temperature until all of the starting material **14h** was consumed by TLC (using 20% EtOAc in hexanes as the mobile phase, and *p*-anisaldehyde to stain the plates). Product **26h** was obtained as a white solid after purification by preparatory TLC using 45% EtOAc in hexanes (52.3 mg, 58%).  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.36 – 7.32 (m, 4H), 7.31 (d,  $J = 4.5$  Hz, 2H), 7.28 (d,  $J = 6.6$  Hz, 1H), 7.24–7.20 (m, 2H), 7.11 (d,  $J = 7.5$  Hz, 3H), 6.74 (d,  $J = 7.6$  Hz, 1H), 5.49 (s, 2H), 3.95 (dd,  $J = 10.6, 7.5$  Hz, 1H), 3.71 – 3.59 (m, 1H), 2.96–2.83 (m, 3H), 2.62–2.56 (m, 1H), 2.55 – 2.45 (m, 2H), 2.39 (dd,  $J = 21.7, 10.7$  Hz, 1H), 2.22 (dd,  $J = 19.5, 11.9$  Hz, 1H), 1.92 – 1.81 (m, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  145.3, 141.6, 141.0, 140.7, 139.1, 134.9, 133.5, 129.8, 128.9, 128.1, 128.1, 127.8, 126.7, 126.6, 125.9, 122.0, 78.0, 73.6, 53.9, 39.5, 32.9, 31.4, 30.3, 23.1; m.p = 145–147°C; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{30}\text{H}_{29}\text{N}_3\text{O}$  : 448.2384, found: 448.2386.

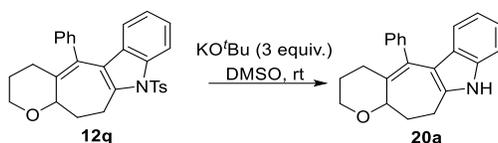
#### Scheme S17. Click Reaction with Imidoyl Chloride for Isoxazole Synthesis<sup>28</sup>



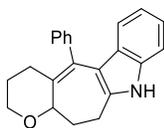
A mixture of **14h** (63 mg, 0.2 mmol),  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  (1 mg, 4  $\mu\text{mol}$ ), Cu powder (1 mg, 2  $\mu\text{mol}$ ), sodium bicarbonate (72 mg, 0.86 mmol), and **S25** (45 mg, 0.24 mmol), in a 1:1 mixture of  $\text{H}_2\text{O}/\text{BuOH}$ , was stirred at room temperature for 24 hours. Product **27h** was obtained as a white solid after purification by preparatory TLC using 45% EtOAc in hexanes (52.2 mg, 44%, 52% brsm). Recrystallization was achieved by dissolving in hot hexanes with toluene and slowly allowing to cool to room temperature in an uncapped vial.  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.72 (d,  $J = 7.2$  Hz, 2H), 7.34 (t,  $J = 7.3$  Hz, 2H), 7.29 (d,  $J = 6.5$  Hz, 1H), 7.22 (t,  $J = 8.1$  Hz, 1H), 7.13 (d,  $J = 7.0$  Hz, 3H), 7.09 (t,  $J = 7.4$  Hz, 1H), 6.96 (d,  $J = 8.4$  Hz, 2H), 6.77 (d,  $J = 7.8$  Hz, 1H), 6.32 (s, 1H), 4.05 (dd,  $J = 10.5, 6.9$  Hz, 1H), 3.85 (s, 3H), 3.07 (dd,  $J = 15.5, 7.6$  Hz, 1H), 2.99 – 2.88 (m, 2H), 2.65 – 2.53 (m, 3H), 2.46 (dd,  $J = 22.9, 10.2$  Hz, 1H), 2.28 (dd,  $J = 20.6, 9.6$  Hz, 1H), 1.92 (dt,  $J = 20.7, 9.0$  Hz, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  170.5, 162.0, 160.8, 141.5, 140.9, 140.7, 138.6, 133.8, 130.0, 129.8, 129.6, 128.3, 128.1, 126.8, 126.8, 126.0, 114.2, 100.0, 78.3, 72.2, 55.3, 39.5, 34.0, 31.5, 30.3, 23.0; m.p = 112–115°C; HRMS (ESI)  $m/z$ :  $[\text{M}+\text{H}]^+$  Calculated for  $\text{C}_{32}\text{H}_{31}\text{NO}_3$  : 464.2220, found: 464.2220.

#### Procedure for Indole Substrate Deprotection:

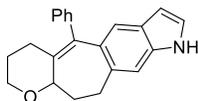
#### Scheme S18: General Conditions for Indole Deprotections



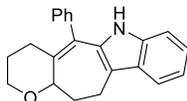
Indole substrates were deprotected following the procedure from Buden et al.<sup>29</sup> To a flame dried vial under argon and charged with a stir bar, added indole adduct **12p** (21 mg, 0.045 mmol) followed by DMSO (0.5mL). To this mixture, added KO<sup>t</sup>Bu (15 mg, 0.13 mmol) that was kept in the glovebox. After addition of the base, the black reaction mixture was covered with aluminum foil and allowed to stir at room temperature for 3 hours. After this time, saturated ammonium chloride was added to quench the reaction, extracted with EtOAc (3 x 15mL), dried over MgSO<sub>4</sub>, filtered, and concentrated. Product was purified by preparatory TLC, using 30% EtOAc in hexane with 2% Et<sub>3</sub>N to afford deprotected indole **20a** as a pale grey amorphous solid (10.5 mg, 74% yield).



Compound **20a** was obtained as a grey amorphous solid (10.5 mg, 74%). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 8.04 (s, 1H), 7.39 – 7.29 (m, 4H), 7.23 (d, *J* = 7.2 Hz, 3H), 7.00 (t, *J* = 7.5 Hz, 1H), 6.72 (t, *J* = 7.6 Hz, 1H), 6.16 (d, *J* = 8.1 Hz, 1H), 4.30 (t, *J* = 6.9 Hz, 1H), 4.06 – 3.97 (m, 1H), 3.57 (td, *J* = 10.8, 6.1 Hz, 1H), 3.11 (dt, *J* = 15.6, 7.8 Hz, 1H), 2.87 (dt, *J* = 15.5, 6.3 Hz, 1H), 2.54 – 2.34 (m, 5H), 2.02 (d, *J* = 4.9 Hz, 1H), 1.73 (dd, *J* = 12.6, 6.2 Hz, 1H). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 141.6, 139.8, 135.4, 135.0, 129.9, 128.2, 127.9, 126.7, 120.9, 120.4, 119.4, 112.4, 110.2, 79.1, 65.1, 36.2, 30.3, 26.5, 24.8, 23.5; HRMS (ESI) *m/z*: [M+H]<sup>+</sup> Calculated for C<sub>22</sub>H<sub>21</sub>NO : 316.1696, found: 316.1687.



Compound **20b** was obtained as a grey amorphous solid (6.9 mg, 51%). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 8.06 (s, 1H), 7.34 (t, *J* = 7.3 Hz, 2H), 7.31 – 7.25 (m, 2H), 7.18 (d, *J* = 7.2 Hz, 2H), 7.10 (s, 1H), 7.05 (s, 1H), 6.36 (s, 1H), 4.08 (dd, *J* = 10.6, 7.8 Hz, 1H), 3.94 (t, *J* = 8.4 Hz, 1H), 3.48 – 3.40 (m, 1H), 3.01 (td, *J* = 12.8, 8.0 Hz, 1H), 2.69 (dd, *J* = 13.1, 6.5 Hz, 1H), 2.60 (dd, *J* = 13.9, 4.1 Hz, 1H), 2.55 – 2.44 (m, 1H), 2.40 (dd, *J* = 20.3, 12.0 Hz, 1H), 2.30 (dd, *J* = 19.5, 11.5 Hz, 1H), 2.13 – 2.00 (m, 1H), 1.76 (s, 1H). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 142.6, 136.4, 135.3, 134.9, 134.3, 133.9, 130.1, 128.0, 126.4, 126.3, 123.9, 122.1, 109.8, 102.7, 77.3, 63.3, 38.5, 31.0, 25.2, 22.8; HRMS (ESI) *m/z*: [M+H]<sup>+</sup> Calculated for C<sub>22</sub>H<sub>21</sub>NO : 316.1696, found: 316.1695.

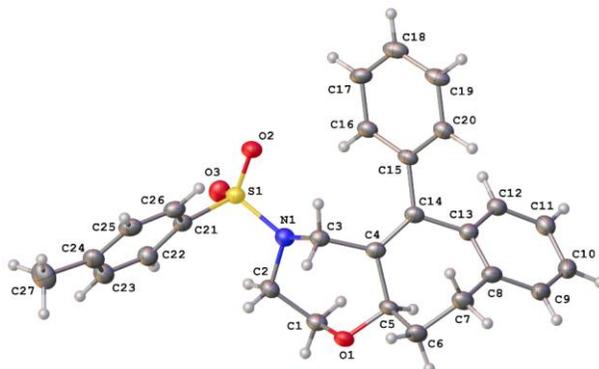


Compound **20c** was obtained as a white solid (20.4 mg, 55%). <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.56 (d, *J* = 6.3 Hz, 1H), 7.45 (dq, *J* = 14.5, 7.1 Hz, 3H), 7.29 – 7.26 (m, 2H), 7.16 (s, 1H), 7.13-7.05 (m, 3H), 4.31 (dd, *J* = 7.3, 2.9 Hz, 1H), 4.14 – 4.04 (m, 1H), 3.70 (td, *J* = 10.8, 5.1 Hz, 1H), 3.20 – 2.99 (m, 2H), 2.44 – 2.14 (m, 4H), 1.97 – 1.83 (m, 1H), 1.66 (dd, *J* = 11.5, 5.2 Hz, 1H). <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 139.9, 134.8, 132.7, 129.7, 128.9, 128.3, 127.5, 126.5, 122.0, 119.3,

118.3, 115.7, 110.5, 80.3, 66.5, 33.4, 28.6, 26.9, 20.5; m.p = 140-141°C; HRMS (ESI) m/z: [M+H]<sup>+</sup> Calculated for C<sub>22</sub>H<sub>21</sub>NO : 316.1696, found: 316.1693.

### X-Ray Crystal Structures for Compounds 19, 12o, 12t, 14a, 21a-1, 22a, and 27h:

#### Compound 19:

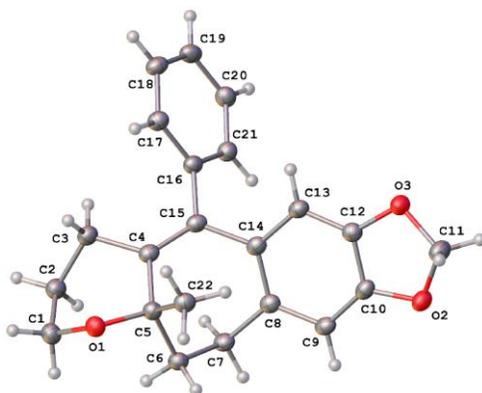


**Table S2.** Crystal data and structure refinement for **19**

|                                 |   |                             |
|---------------------------------|---|-----------------------------|
| Identification code             | <b>19</b>   |                             |
| Empirical formula               | C <sub>27</sub> H <sub>27</sub> N O <sub>3</sub> S          |                             |
| Formula weight                  | 445.55  |                             |
| Temperature                     | 100.01(10) K  |                             |
| Wavelength                      | 1.54184 Å   |                             |
| Crystal system                  | monoclinic  |                             |
| Space group                     | <i>Pc</i>   |                             |
| Unit cell dimensions            | <i>a</i> = 12.17620(10) Å                                   | $\alpha = 90^\circ$         |
|                                 | <i>b</i> = 11.57040(10) Å                                   | $\beta = 94.5220(10)^\circ$ |
|                                 | <i>c</i> = 7.98290(10) Å                                    | $\gamma = 90^\circ$         |
| Volume                          | 1121.158(19) Å <sup>3</sup>                                 |                             |
| <i>Z</i>                        | 2   |                             |
| Density (calculated)            | 1.320 Mg/m <sup>3</sup>                                     |                             |
| Absorption coefficient          | 1.516 mm <sup>-1</sup>                                      |                             |
| <i>F</i> (000)                  | 472   |                             |
| Crystal color, morphology       | colourless, needle  |                             |
| Crystal size                    | 0.489 x 0.09 x 0.018 mm <sup>3</sup>                        |                             |
| Theta range for data collection | 3.641 to 80.188°  |                             |
| Index ranges                    | -15 ≤ <i>h</i> ≤ 15, -14 ≤ <i>k</i> ≤ 14, -9 ≤ <i>l</i> ≤ 8 |                             |
| Reflections collected           | 34948   |                             |
| Independent reflections         | 4549 [ <i>R</i> (int) = 0.0717]                             |                             |

|  |                                    |
|--|------------------------------------|
| Observed reflections                   | 4400                               |
| Completeness to theta = 74.504°        | 100.0%                             |
| Absorption correction                  | Multi-scan                         |
| Max. and min. transmission             | 1.00000 and 0.62538                |
| Refinement method                      | Full-matrix least-squares on $F^2$ |
| Data / restraints / parameters         | 4549 / 2 / 290                     |
| Goodness-of-fit on $F^2$               | 1.097                              |
| Final $R$ indices [ $I > 2\sigma(I)$ ] | $R1 = 0.0438$ , $wR2 = 0.1208$     |
| $R$ indices (all data)                 | $R1 = 0.0450$ , $wR2 = 0.1219$     |
| Absolute structure parameter           | -0.002(16)                         |
| Largest diff. peak and hole            | 0.383 and -0.464 e.Å <sup>-3</sup> |

### Compound 12o:



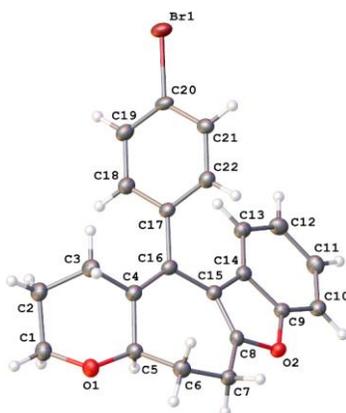
**Table S3.** Crystal data and structure refinement for **12o**

---

|                      |                             |                          |
|----------------------|-----------------------------|--------------------------|
| Identification code  | <b>12o</b>                  |                          |
| Empirical formula    | C22 H22 O3                  |                          |
| Formula weight       | 334.39                      |                          |
| Temperature          | 100.00(10) K                |                          |
| Wavelength           | 1.54184 Å                   |                          |
| Crystal system       | monoclinic                  |                          |
| Space group          | $P2_1/n$                    |                          |
| Unit cell dimensions | $a = 17.2159(2)$ Å          | $a = 90^\circ$           |
|                      | $b = 10.13720(10)$ Å        | $b = 110.5460(10)^\circ$ |
|                      | $c = 20.2267(2)$ Å          | $g = 90^\circ$           |
| Volume               | $3305.44(6)$ Å <sup>3</sup> |                          |

|   |   |
|---|---|
| Z   | 8   |
| Density (calculated)                                | 1.344 Mg/m <sup>3</sup>                                       |
| Absorption coefficient                              | 0.703 mm <sup>-1</sup>  |
| <i>F</i> (000)                                      | 1424  |
| Crystal color, morphology                           | colourless, block   |
| Crystal size  | 0.168 x 0.13 x 0.082 mm <sup>3</sup>                          |
| Theta range for data collection                     | 4.179 to 80.392°  |
| Index ranges  | -22 ≤ <i>h</i> ≤ 16, -12 ≤ <i>k</i> ≤ 12, -25 ≤ <i>l</i> ≤ 25 |
| Reflections collected                               | 35734   |
| Independent reflections                             | 7100 [ <i>R</i> (int) = 0.0387]                               |
| Observed reflections                                | 6280  |
| Completeness to theta = 74.504°                     | 100.0%  |
| Absorption correction                               | Multi-scan  |
| Max. and min. transmission                          | 1.00000 and 0.87854   |
| Refinement method                                   | Full-matrix least-squares on <i>F</i> <sup>2</sup>            |
| Data / restraints / parameters                      | 7100 / 0 / 453  |
| Goodness-of-fit on <i>F</i> <sup>2</sup>            | 1.059   |
| Final <i>R</i> indices [ <i>I</i> > 2σ( <i>I</i> )] | <i>R</i> 1 = 0.0442, <i>wR</i> 2 = 0.1209                     |
| <i>R</i> indices (all data)                         | <i>R</i> 1 = 0.0487, <i>wR</i> 2 = 0.1248                     |
| Largest diff. peak and hole                         | 0.377 and -0.225 e.Å <sup>-3</sup>                            |

### Compound 12t:



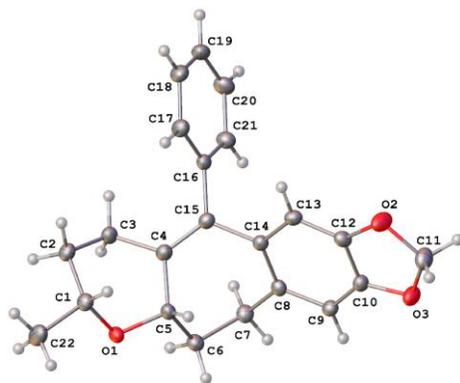
**Table S4.** Crystal data and structure refinement for **12t**

Identification code

**12t**

|   |  |
|---|--|
| Empirical formula                                   | C <sub>22</sub> H <sub>19</sub> Br O <sub>2</sub>  |
| Formula weight                                      | 395.28   |
| Temperature   | 100.00(10) K   |
| Wavelength  | 1.54184 Å  |
| Crystal system                                      | triclinic  |
| Space group   | <i>P</i> -1  |
| Unit cell dimensions                                | <i>a</i> = 7.14240(10) Å <i>a</i> = 93.7290(10)°<br><i>b</i> = 9.29770(10) Å <i>b</i> = 99.1970(10)°<br><i>c</i> = 13.9152(2) Å <i>g</i> = 107.703(2)° |
| Volume  | 862.63(2) Å <sup>3</sup>   |
| <i>Z</i>  | 2  |
| Density (calculated)                                | 1.522 Mg/m <sup>3</sup>  |
| Absorption coefficient                              | 3.338 mm <sup>-1</sup>   |
| <i>F</i> (000)                                      | 404  |
| Crystal color, morphology                           | colourless, block  |
| Crystal size  | 0.155 x 0.113 x 0.078 mm <sup>3</sup>  |
| Theta range for data collection                     | 3.241 to 80.002°   |
| Index ranges  | -9 ≤ <i>h</i> ≤ 9, -11 ≤ <i>k</i> ≤ 10, -17 ≤ <i>l</i> ≤ 17  |
| Reflections collected                               | 28144  |
| Independent reflections                             | 3680 [ <i>R</i> (int) = 0.0421]  |
| Observed reflections                                | 3503   |
| Completeness to theta = 74.504°                     | 99.7%  |
| Absorption correction                               | Multi-scan   |
| Max. and min. transmission                          | 1.00000 and 0.88565  |
| Refinement method                                   | Full-matrix least-squares on <i>F</i> <sup>2</sup>   |
| Data / restraints / parameters                      | 3680 / 0 / 226   |
| Goodness-of-fit on <i>F</i> <sup>2</sup>            | 1.081  |
| Final <i>R</i> indices [ <i>I</i> > 2σ( <i>I</i> )] | <i>R</i> 1 = 0.0325, <i>wR</i> 2 = 0.0764  |
| <i>R</i> indices (all data)                         | <i>R</i> 1 = 0.0340, <i>wR</i> 2 = 0.0772  |
| Largest diff. peak and hole                         | 0.371 and -0.694 e.Å <sup>-3</sup>   |

**Compound 14a:**

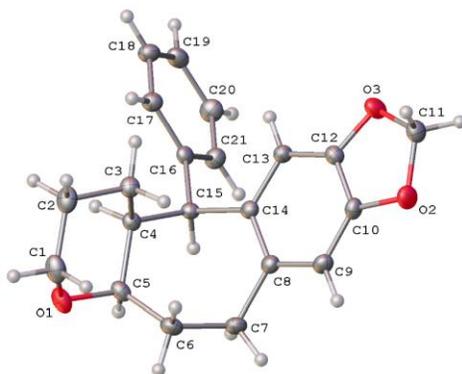


**Table S5.** Crystal data and structure refinement for **14a**

|                                 |   |                          |
|---------------------------------|---|--------------------------|
| Identification code             | <b>14a</b>  |                          |
| Empirical formula               | C <sub>22</sub> H <sub>22</sub> O <sub>3</sub>                |                          |
| Formula weight                  | 334.39  |                          |
| Temperature                     | 100.00(10) K  |                          |
| Wavelength                      | 1.54184 Å   |                          |
| Crystal system                  | monoclinic  |                          |
| Space group                     | <i>P</i> 2 <sub>1</sub> / <i>c</i>                            |                          |
| Unit cell dimensions            | <i>a</i> = 11.11210(10) Å                                     | <i>a</i> = 90°           |
|                                 | <i>b</i> = 14.44820(10) Å                                     | <i>b</i> = 100.5810(10)° |
|                                 | <i>c</i> = 11.06860(10) Å                                     | <i>c</i> = 90°           |
| Volume                          | 1746.85(3) Å <sup>3</sup>                                     |                          |
| <i>Z</i>                        | 4   |                          |
| Density (calculated)            | 1.271 Mg/m <sup>3</sup>                                       |                          |
| Absorption coefficient          | 0.665 mm <sup>-1</sup>  |                          |
| <i>F</i> (000)                  | 712   |                          |
| Crystal color, morphology       | colourless, block   |                          |
| Crystal size                    | 0.234 x 0.139 x 0.108 mm <sup>3</sup>                         |                          |
| Theta range for data collection | 4.047 to 80.156°  |                          |
| Index ranges                    | -14 ≤ <i>h</i> ≤ 14, -18 ≤ <i>k</i> ≤ 18, -12 ≤ <i>l</i> ≤ 13 |                          |
| Reflections collected           | 20536   |                          |
| Independent reflections         | 3744 [ <i>R</i> (int) = 0.0366]                               |                          |
| Observed reflections            | 3460  |                          |
| Completeness to theta = 74.504° | 99.9%   |                          |
| Absorption correction           | Multi-scan  |                          |

|  |                                    |
|--|------------------------------------|
| Max. and min. transmission             | 1.00000 and 0.89800                |
| Refinement method                      | Full-matrix least-squares on $F^2$ |
| Data / restraints / parameters         | 3744 / 106 / 256                   |
| Goodness-of-fit on $F^2$               | 1.052                              |
| Final $R$ indices [ $I > 2\sigma(I)$ ] | $R1 = 0.0417$ , $wR2 = 0.1039$     |
| $R$ indices (all data)                 | $R1 = 0.0442$ , $wR2 = 0.1057$     |
| Largest diff. peak and hole            | 0.228 and -0.249 e.Å <sup>-3</sup> |

### Compound 21a-1:



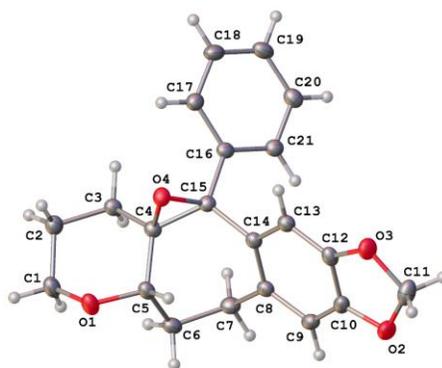
**Table S6.** Crystal data and structure refinement for **21a-1**

---

|                        |  |                |
|------------------------|--|----------------|
| Identification code    | <b>21a -1</b>                                  |                |
| Empirical formula      | C <sub>21</sub> H <sub>22</sub> O <sub>3</sub> |                |
| Formula weight         | 322.38   |                |
| Temperature            | 99.97(10) K                                    |                |
| Wavelength             | 1.54184 Å                                      |                |
| Crystal system         | orthorhombic                                   |                |
| Space group            | $Pna2_1$                                       |                |
| Unit cell dimensions   | $a = 14.11320(10)$ Å                           | $a = 90^\circ$ |
|                        | $b = 19.7985(2)$ Å                             | $b = 90^\circ$ |
|                        | $c = 5.92430(10)$ Å                            | $c = 90^\circ$ |
| Volume                 | $1655.37(3)$ Å <sup>3</sup>                    |                |
| $Z$                    | 4  |                |
| Density (calculated)   | 1.294 Mg/m <sup>3</sup>                        |                |
| Absorption coefficient | 0.680 mm <sup>-1</sup>                         |                |
| $F(000)$               | 688  |                |

|   |   |
|---|---|
| Crystal color, morphology                           | colourless, block   |
| Crystal size  | 0.284 x 0.227 x 0.1 mm <sup>3</sup>                         |
| Theta range for data collection                     | 3.846 to 80.166°  |
| Index ranges  | -17 ≤ <i>h</i> ≤ 17, -24 ≤ <i>k</i> ≤ 25, -6 ≤ <i>l</i> ≤ 7 |
| Reflections collected                               | 19326   |
| Independent reflections                             | 3305 [ <i>R</i> (int) = 0.0399]                             |
| Observed reflections                                | 3221  |
| Completeness to theta = 67.684°                     | 100.0%  |
| Absorption correction                               | Multi-scan  |
| Max. and min. transmission                          | 1.00000 and 0.87289   |
| Refinement method                                   | Full-matrix least-squares on <i>F</i> <sup>2</sup>          |
| Data / restraints / parameters                      | 3305 / 1 / 217  |
| Goodness-of-fit on <i>F</i> <sup>2</sup>            | 1.048   |
| Final <i>R</i> indices [ <i>I</i> > 2σ( <i>I</i> )] | <i>R</i> 1 = 0.0312, <i>wR</i> 2 = 0.0777                   |
| <i>R</i> indices (all data)                         | <i>R</i> 1 = 0.0319, <i>wR</i> 2 = 0.0783                   |
| Absolute structure parameter                        | -0.06(10)   |
| Largest diff. peak and hole                         | 0.138 and -0.196 e.Å <sup>-3</sup>                          |

**Compound 22a:**



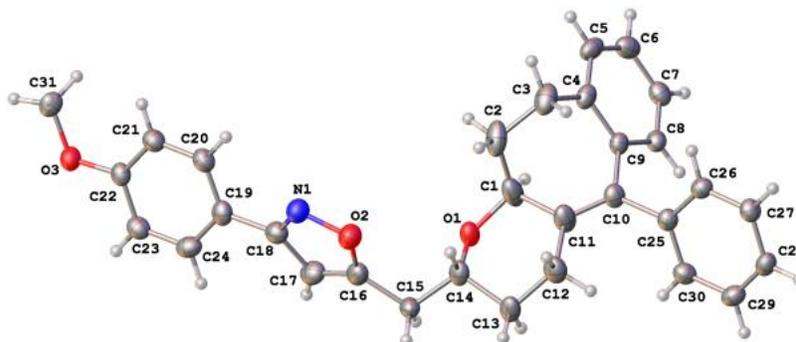
**Table S7.** Crystal data and structure refinement for **22a**

---

|                     |  |
|---------------------|--|
| Identification code | <b>22a</b>                                     |
| Empirical formula   | C <sub>21</sub> H <sub>20</sub> O <sub>4</sub> |
| Formula weight      | 336.37   |
| Temperature         | 100.00(10) K                                   |
| Wavelength          | 1.54184 Å                                      |

|  |   |                         |
|--|---|-------------------------|
| Crystal system                         | monoclinic  |                         |
| Space group                            | $P2_1/n$  |                         |
| Unit cell dimensions                   | $a = 11.33020(10) \text{ \AA}$                                    | $a = 90^\circ$          |
|  | $b = 9.30400(10) \text{ \AA}$                                     | $b = 92.0390(10)^\circ$ |
|  | $c = 15.25110(10) \text{ \AA}$                                    | $g = 90^\circ$          |
| Volume                                 | $1606.69(2) \text{ \AA}^3$  |                         |
| $Z$                                    | 4   |                         |
| Density (calculated)                   | $1.391 \text{ Mg/m}^3$  |                         |
| Absorption coefficient                 | $0.776 \text{ mm}^{-1}$   |                         |
| $F(000)$                               | 712   |                         |
| Crystal color, morphology              | colourless, block   |                         |
| Crystal size                           | $0.126 \times 0.118 \times 0.057 \text{ mm}^3$                    |                         |
| Theta range for data collection        | $4.782$ to $80.320^\circ$   |                         |
| Index ranges                           | $-14 \leq h \leq 14$ , $-9 \leq k \leq 11$ , $-19 \leq l \leq 19$ |                         |
| Reflections collected                  | 28132   |                         |
| Independent reflections                | 3482 [ $R(\text{int}) = 0.0354$ ]                                 |                         |
| Observed reflections                   | 3218  |                         |
| Completeness to theta = $74.504^\circ$ | 100.0%  |                         |
| Absorption correction                  | Multi-scan  |                         |
| Max. and min. transmission             | 1.00000 and 0.94365   |                         |
| Refinement method                      | Full-matrix least-squares on $F^2$                                |                         |
| Data / restraints / parameters         | 3482 / 0 / 226  |                         |
| Goodness-of-fit on $F^2$               | 1.061   |                         |
| Final $R$ indices [ $I > 2\sigma(I)$ ] | $R1 = 0.0358$ , $wR2 = 0.0917$                                    |                         |
| $R$ indices (all data)                 | $R1 = 0.0384$ , $wR2 = 0.0935$                                    |                         |
| Largest diff. peak and hole            | $0.234$ and $-0.208 \text{ e.\AA}^{-3}$                           |                         |

### Compound 27h:



**Table S8.** Crystal data and structure refinement for **27h**.

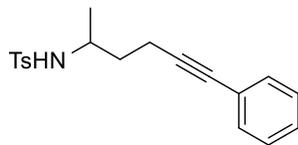
---

|   |   |                         |
|---|---|-------------------------|
| —   |   |                         |
| Identification code                                 | <b>27h</b>  |                         |
| Empirical formula                                   | C <sub>31</sub> H <sub>29</sub> N O <sub>3</sub>            |                         |
| Formula weight                                      | 463.55  |                         |
| Temperature   | 99.99(10) K   |                         |
| Wavelength  | 1.54184 Å   |                         |
| Crystal system                                      | monoclinic  |                         |
| Space group   | <i>P</i> 2 <sub>1</sub> / <i>n</i>                          |                         |
| Unit cell dimensions                                | <i>a</i> = 15.9276(3) Å                                     | <i>a</i> = 90°          |
|   | <i>b</i> = 5.55609(9) Å                                     | <i>b</i> = 98.7546(18)° |
|   | <i>c</i> = 27.3254(5) Å                                     | <i>g</i> = 90°          |
| Volume  | 2389.99(7) Å <sup>3</sup>                                   |                         |
| <i>Z</i>  | 4   |                         |
| Density (calculated)                                | 1.288 Mg/m <sup>3</sup>                                     |                         |
| Absorption coefficient                              | 0.651 mm <sup>-1</sup>                                      |                         |
| <i>F</i> (000)                                      | 984   |                         |
| Crystal color, morphology                           | colourless, needle  |                         |
| Crystal size  | 0.2 x 0.104 x 0.077 mm <sup>3</sup>                         |                         |
| Theta range for data collection                     | 3.273 to 80.341°  |                         |
| Index ranges  | -20 ≤ <i>h</i> ≤ 20, -7 ≤ <i>k</i> ≤ 7, -34 ≤ <i>l</i> ≤ 34 |                         |
| Reflections collected                               | 47366   |                         |
| Independent reflections                             | 9452 [ <i>R</i> (int) = 0.044]                              |                         |
| Observed reflections                                | 8533  |                         |
| Completeness to theta = 74.504°                     | 99.5%   |                         |
| Absorption correction                               | Multi-scan  |                         |
| Max. and min. transmission                          | 1.00000 and 0.63440   |                         |
| Refinement method                                   | Full-matrix least-squares on <i>F</i> <sup>2</sup>          |                         |
| Data / restraints / parameters                      | 9452 / 36 / 355   |                         |
| Goodness-of-fit on <i>F</i> <sup>2</sup>            | 1.099   |                         |
| Final <i>R</i> indices [ <i>I</i> > 2σ( <i>I</i> )] | <i>R</i> 1 = 0.0706, <i>wR</i> 2 = 0.1831                   |                         |
| <i>R</i> indices (all data)                         | <i>R</i> 1 = 0.0762, <i>wR</i> 2 = 0.1865                   |                         |
| Largest diff. peak and hole                         | 0.617 and -0.408 e.Å <sup>-3</sup>                          |                         |

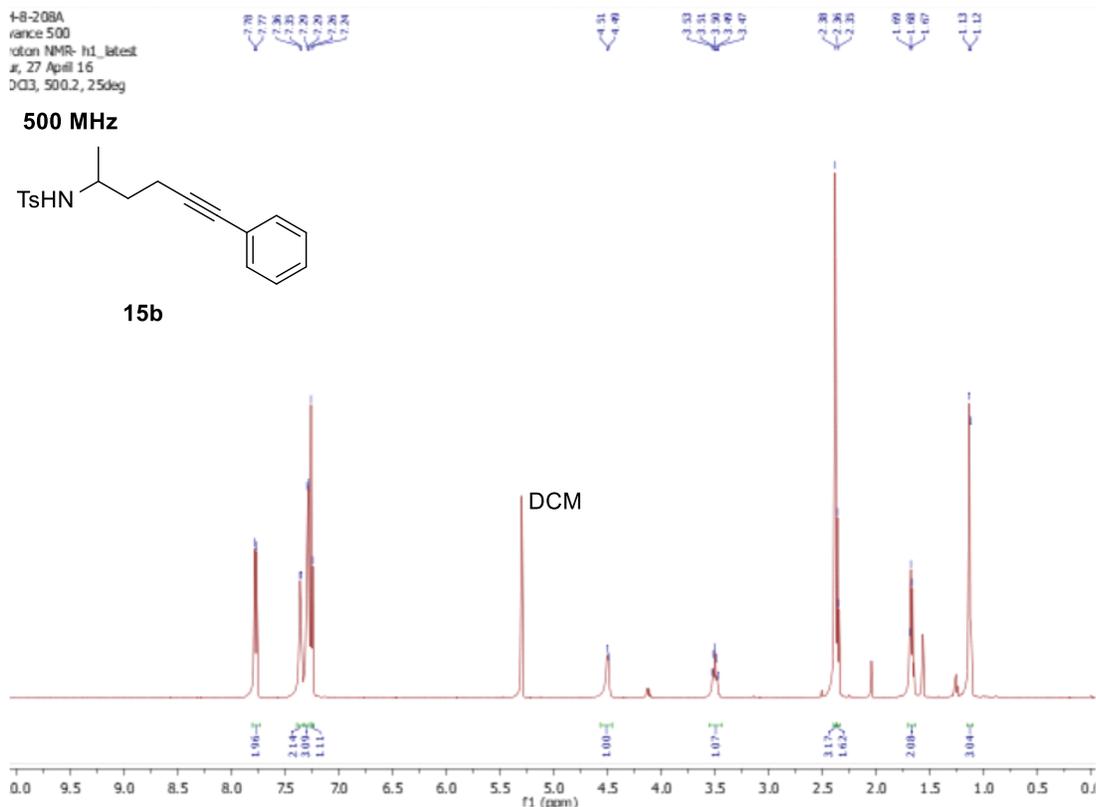
## NMR Spectra:

4-8-20BA  
 rance 500  
 otion NMR- h1\_latest  
 ar, 27 April 16  
 DCl3, 500.2, 25deg

500 MHz

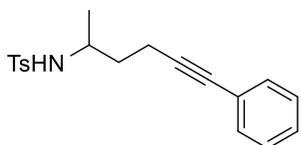


15b

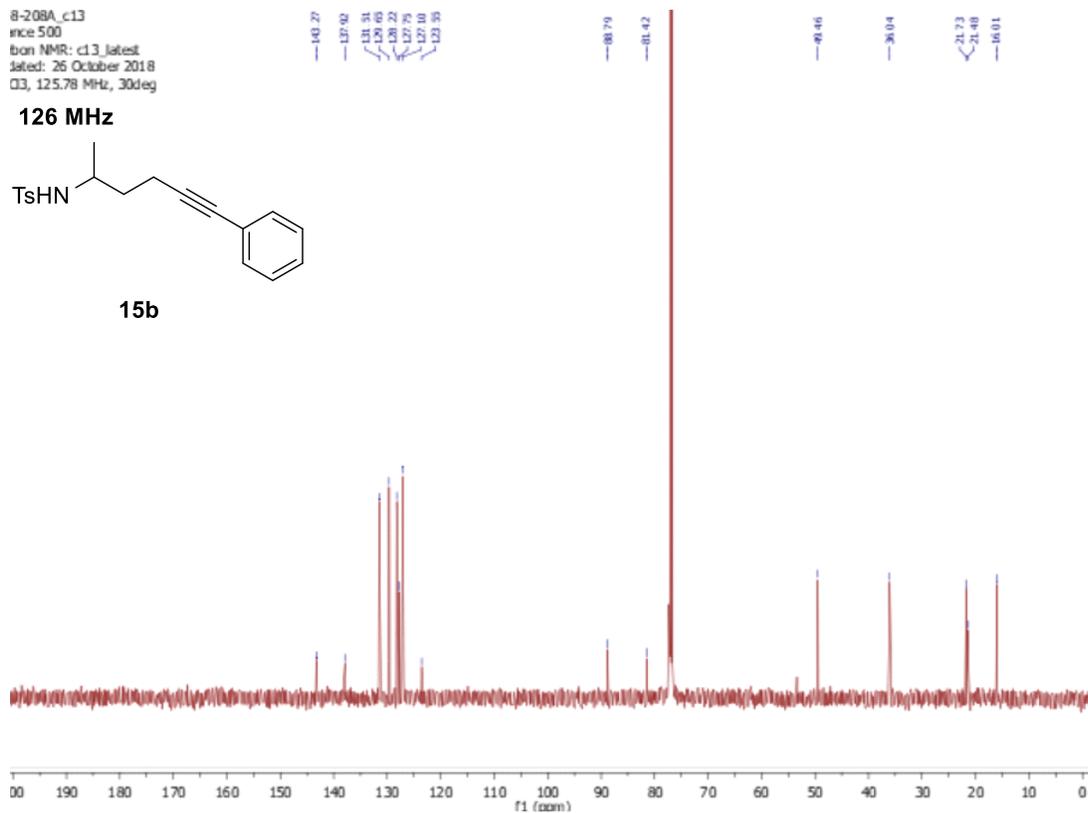


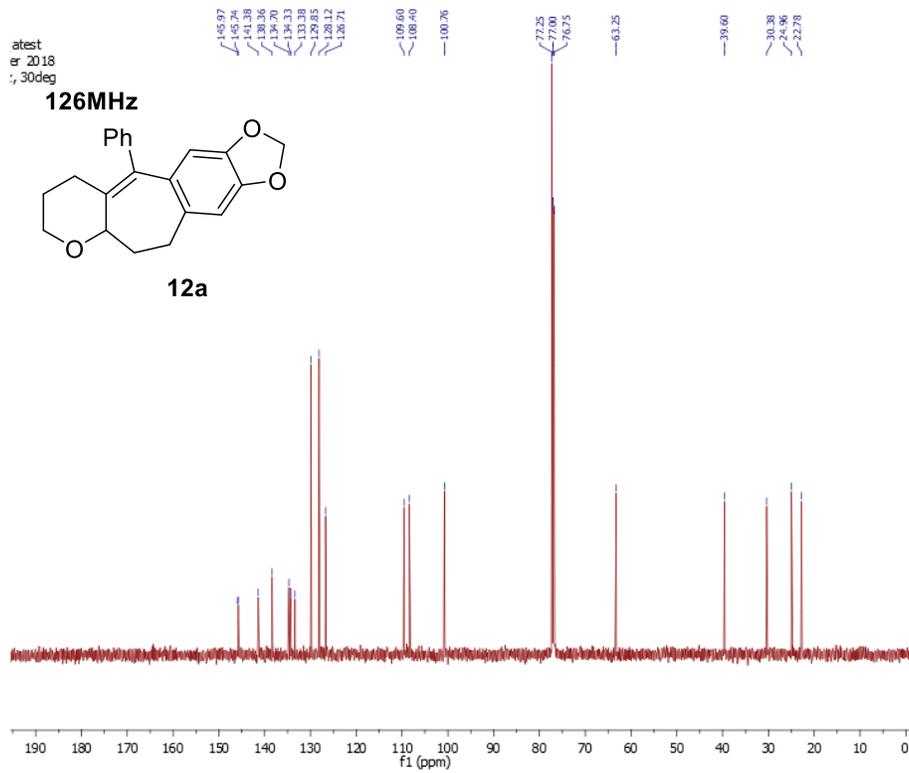
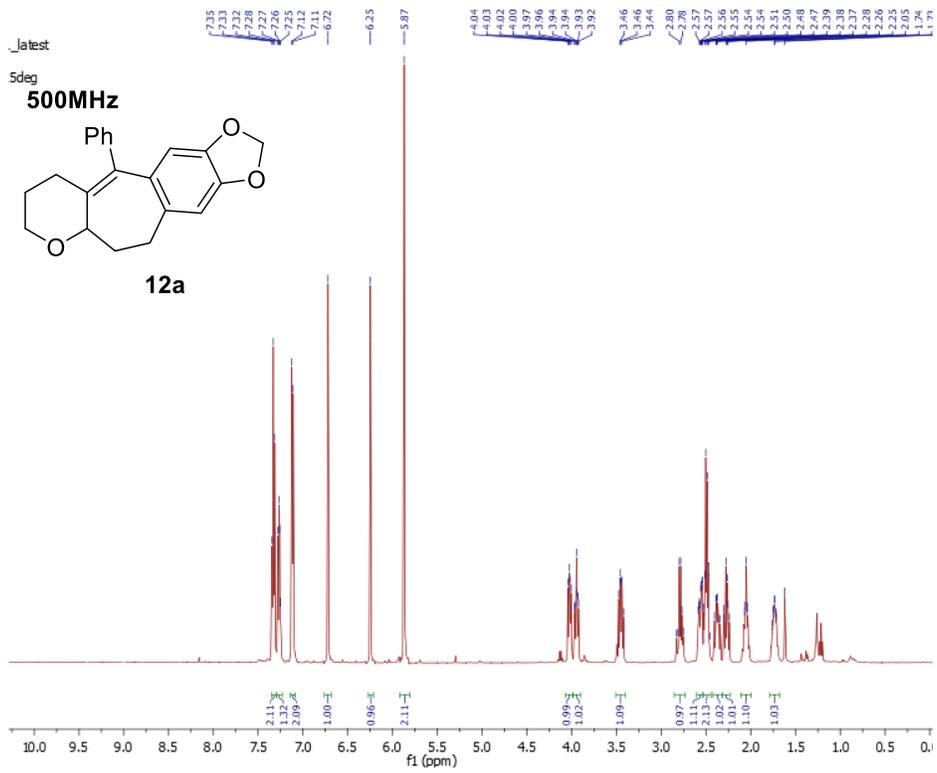
8-20BA\_c13  
 rance 500  
 ion NMR: c13\_latest  
 dated: 26 October 2018  
 Cl3, 125.78 MHz, 30deg

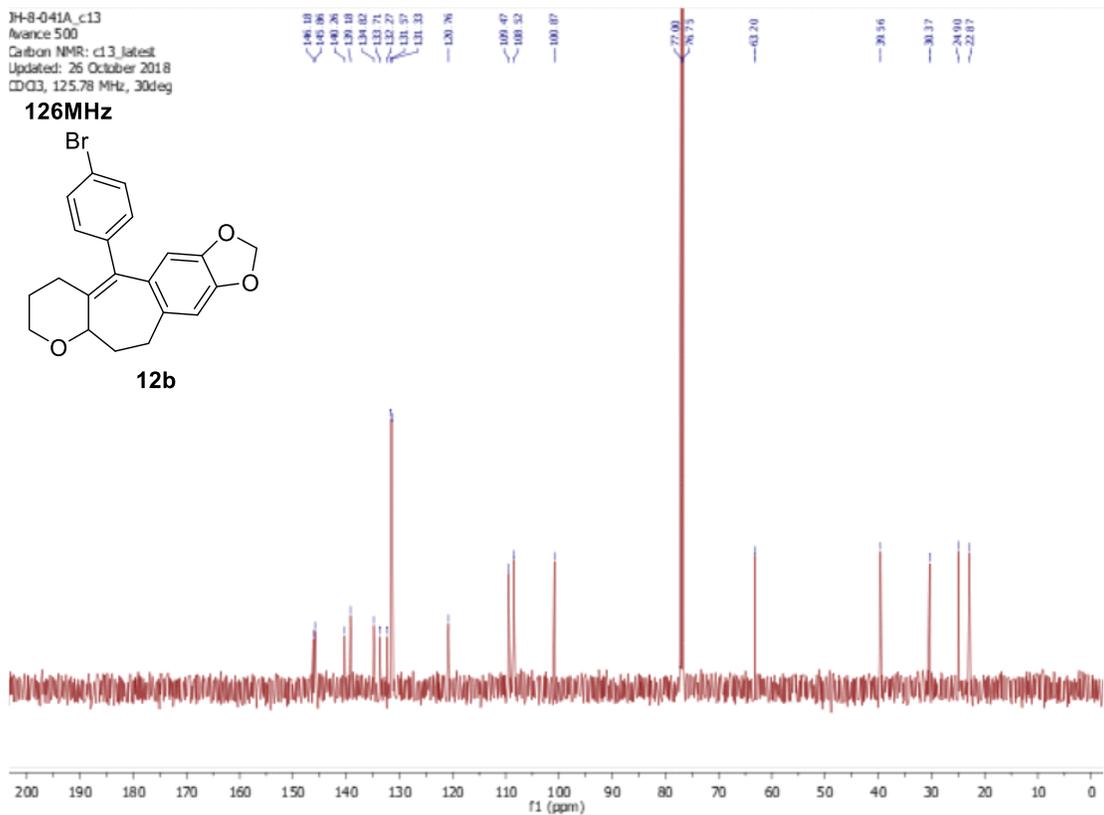
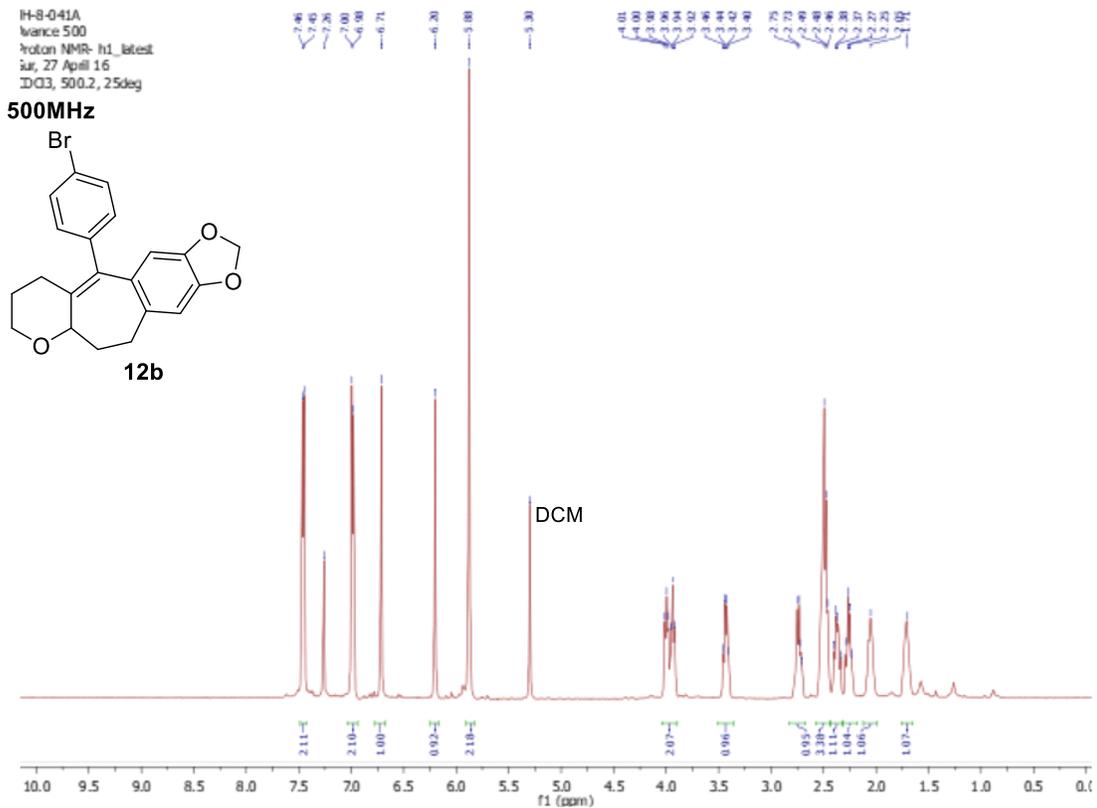
126 MHz



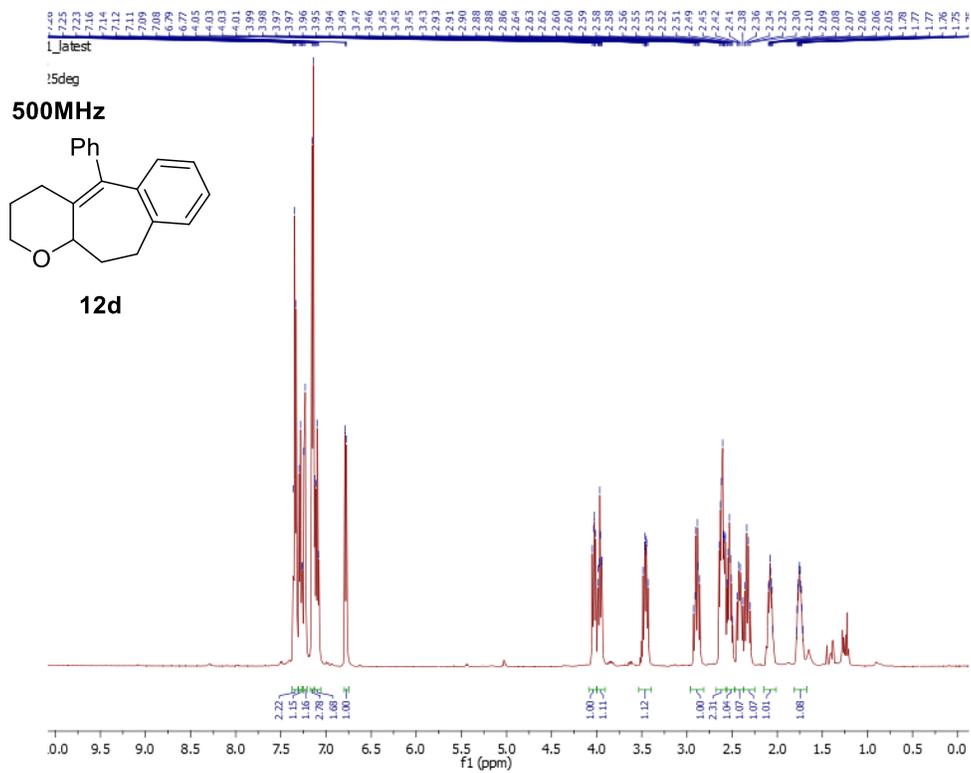
15b





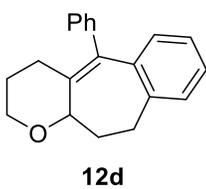


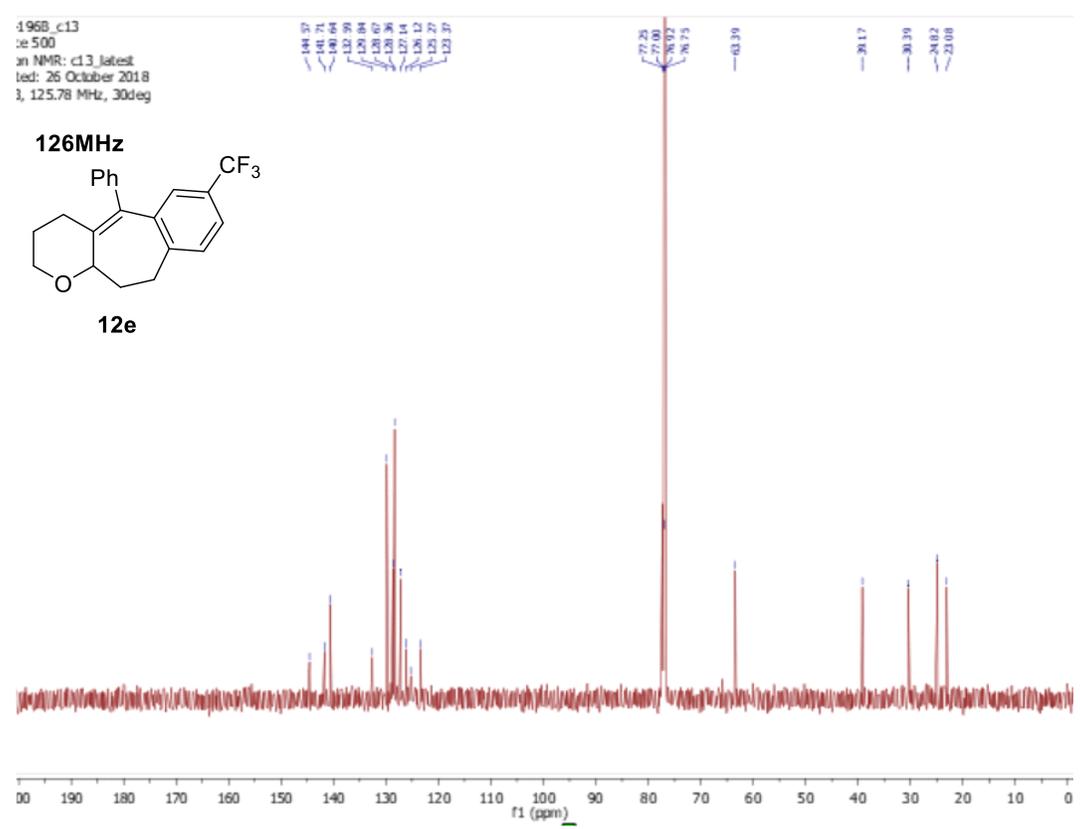
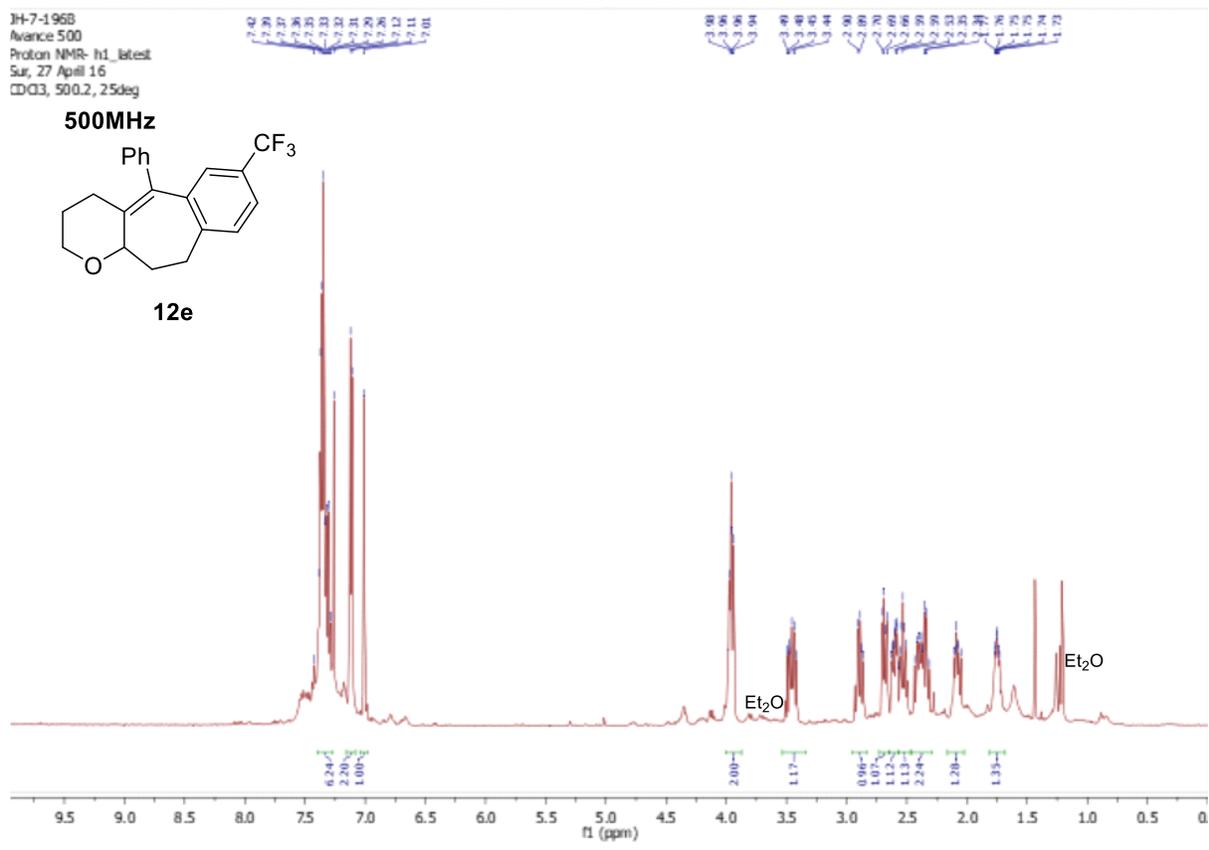


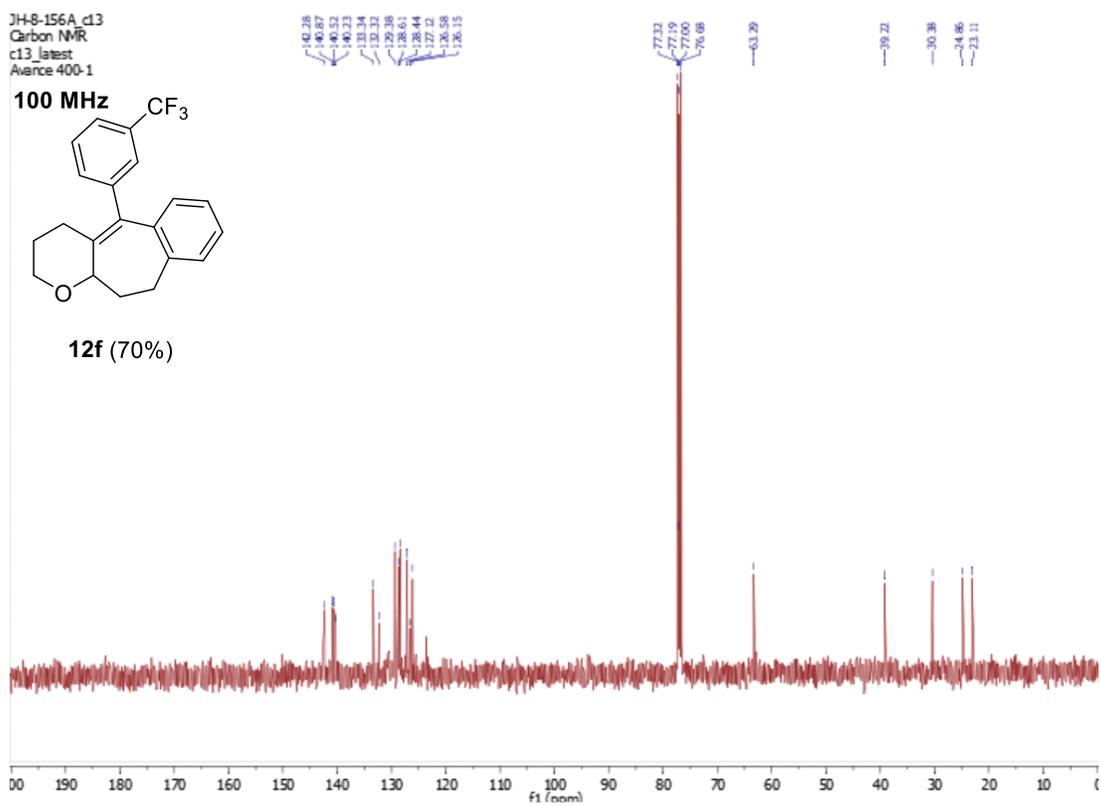
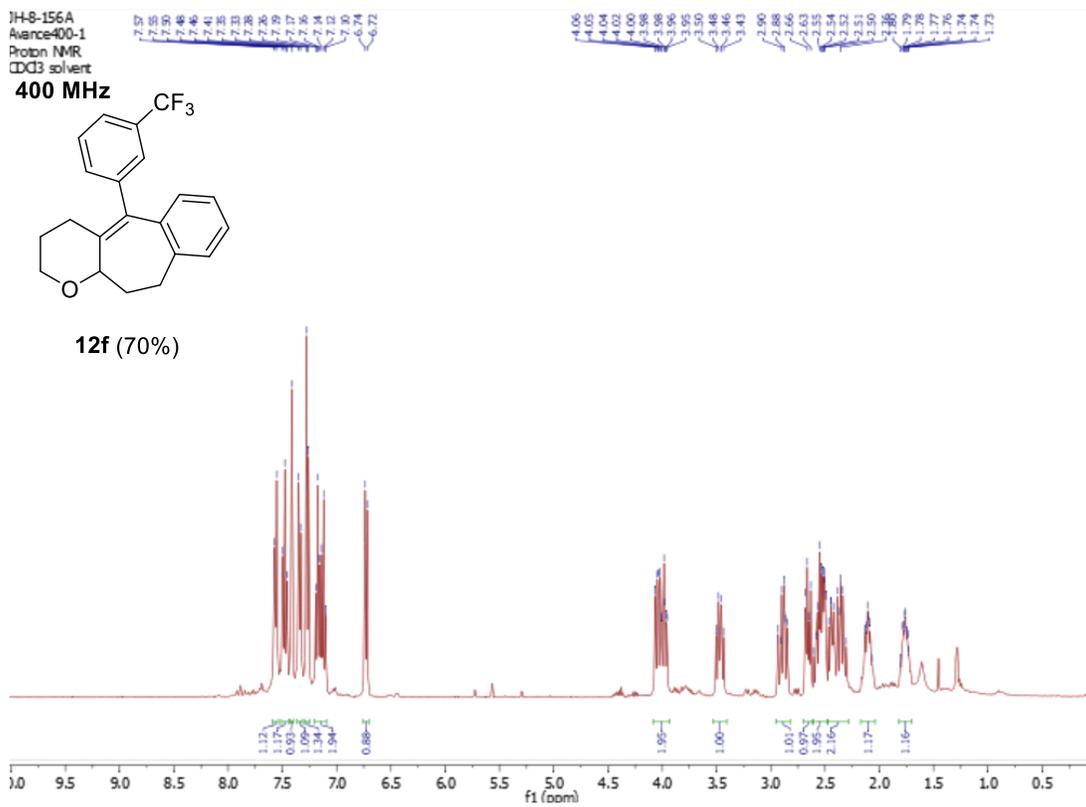


8A\_c13  
500  
NMR: c13\_latest  
f: 26 October 2018  
125.78 MHz, 30deg

**126MHz**

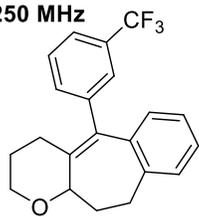




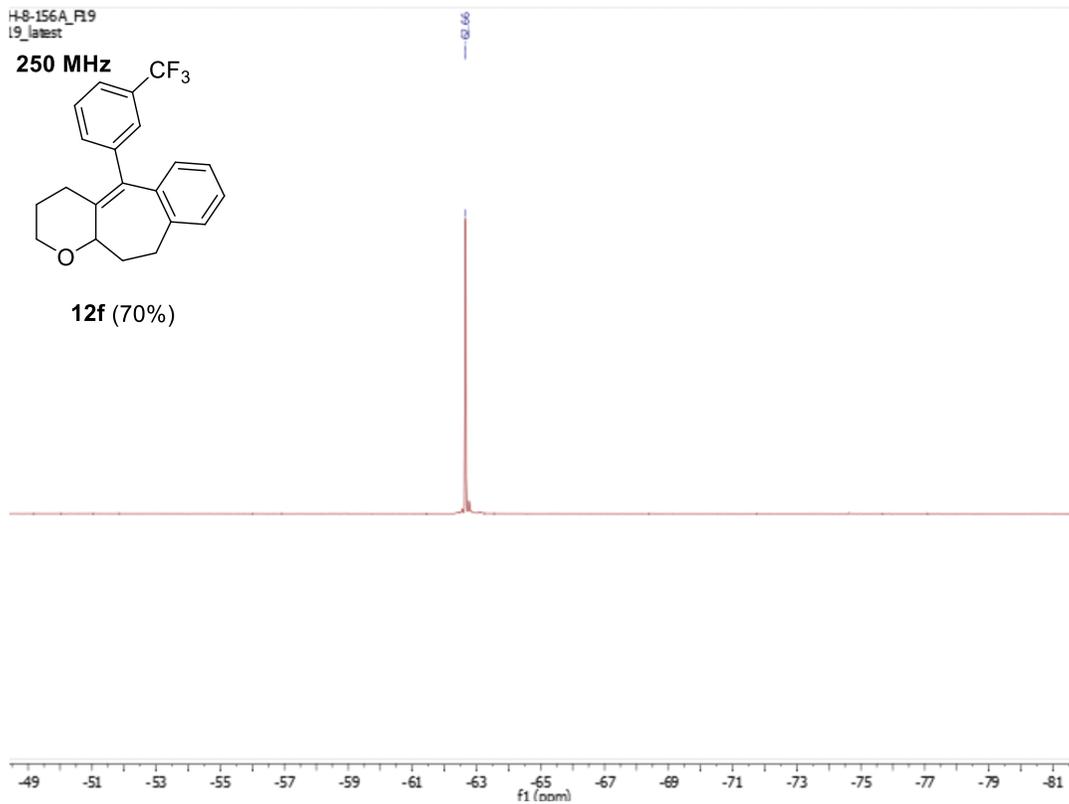


H-8-156A\_F19  
19\_latest

250 MHz

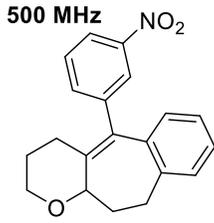


12f (70%)

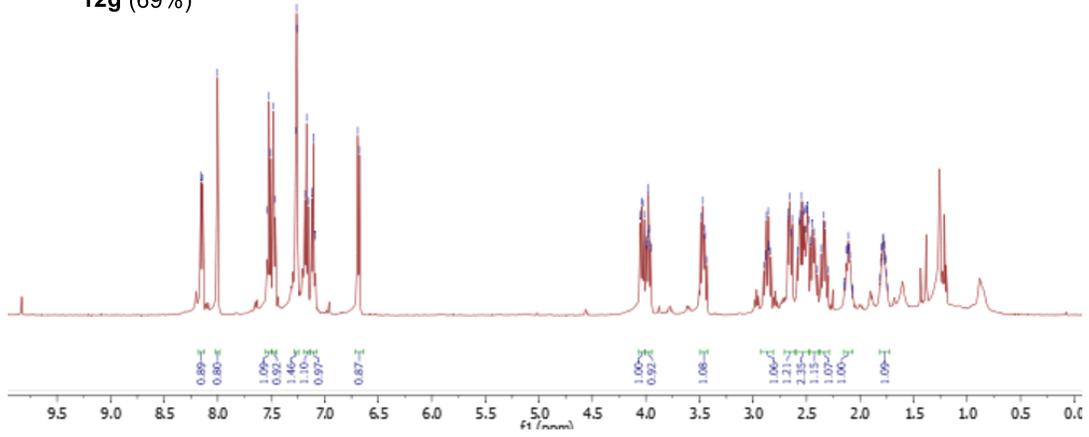


*This space was left blank intentionally*

1H-160B\_c13  
 Varianc 500  
 Proton NMR- h1\_latest  
 Sat 27 April 16  
 CDCl3, 500.2, 25deg



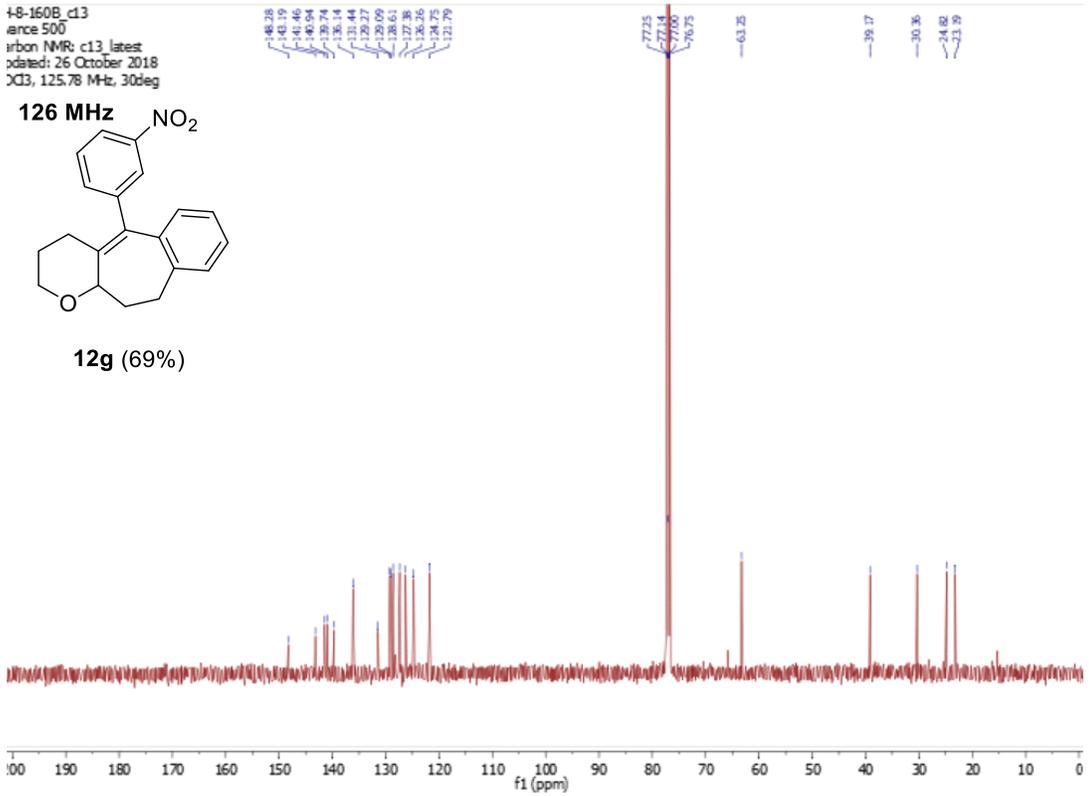
12g (69%)



13-160B\_c13  
 Varianc 500  
 Carbon NMR- c13\_latest  
 dated: 26 October 2018  
 CDCl3, 125.76 MHz, 30deg

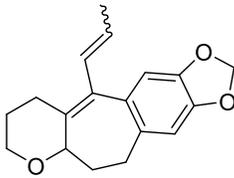


12g (69%)

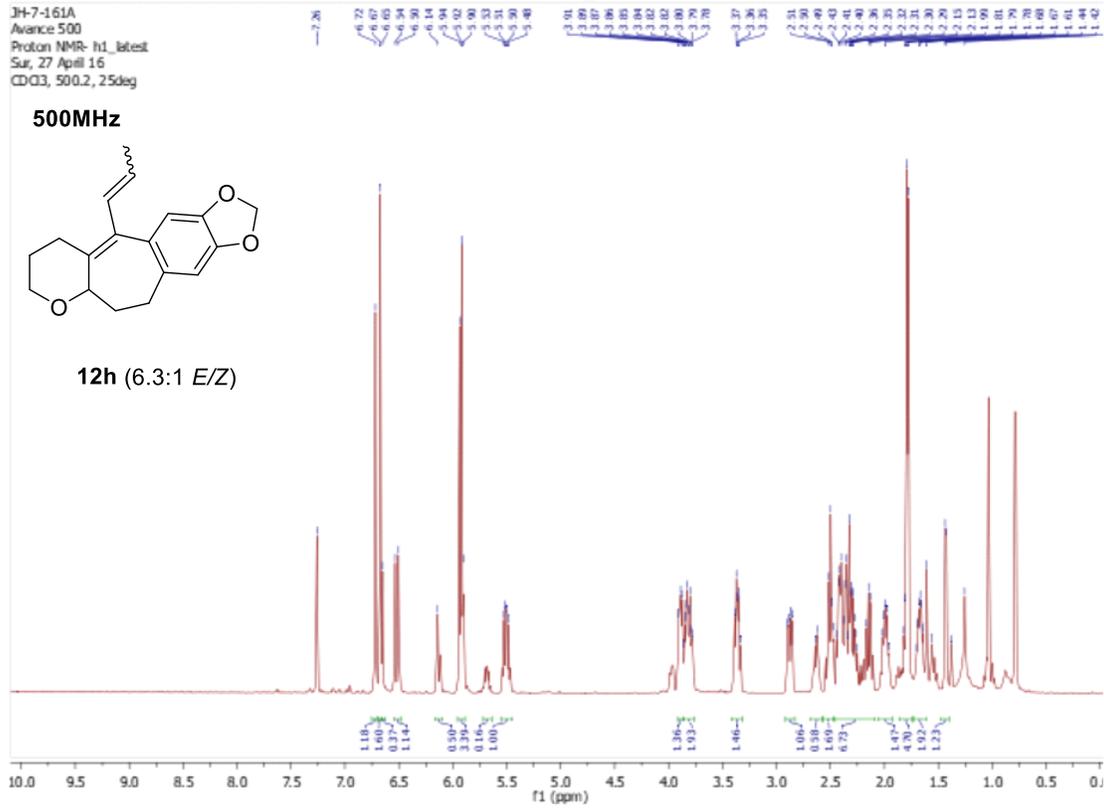


JH-7-161A  
Avenice 500  
Proton NMR- h1\_test1  
Sun, 27 April 16  
CDCl3, 500.2, 25deg

500MHz

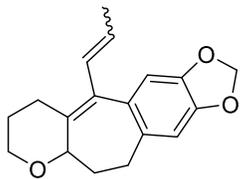


12h (6.3:1 E/Z)

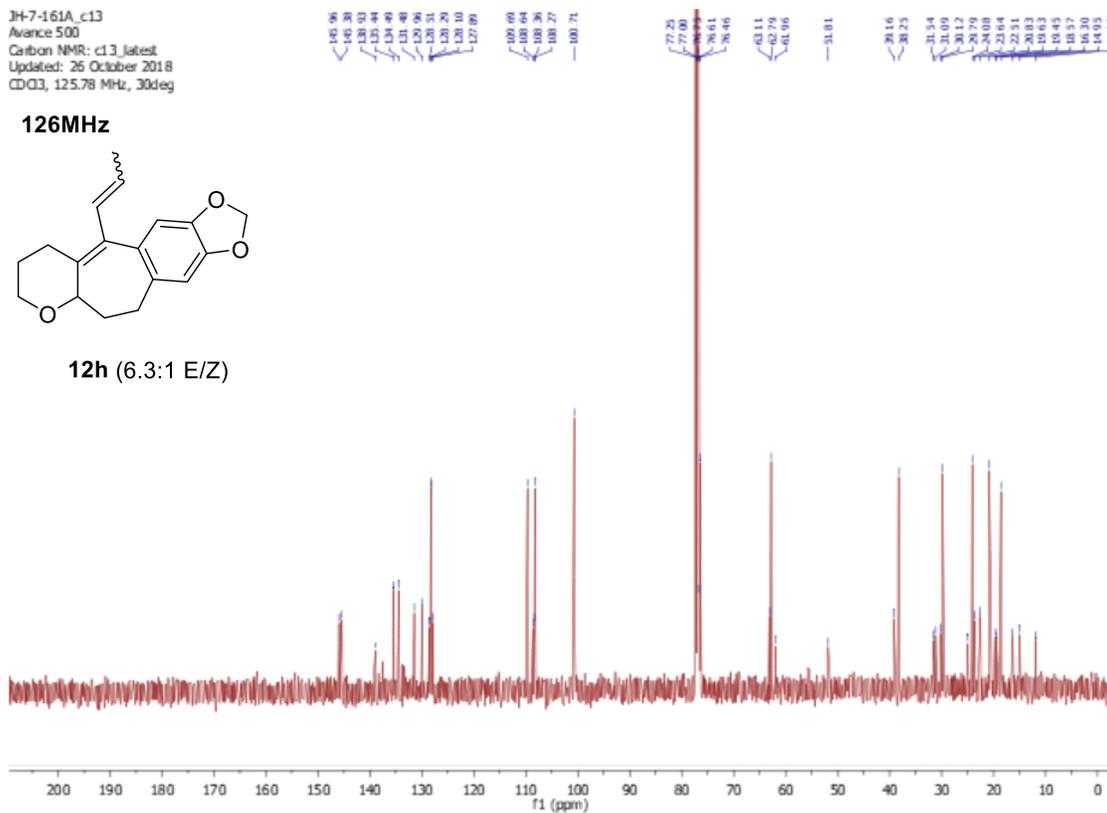


JH-7-161A\_c13  
Avenice 500  
Carbon NMR: c13\_test1  
Updated: 26 October 2018  
CDCl3, 125.78 MHz, 30deg

126MHz

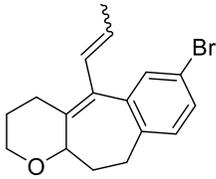


12h (6.3:1 E/Z)

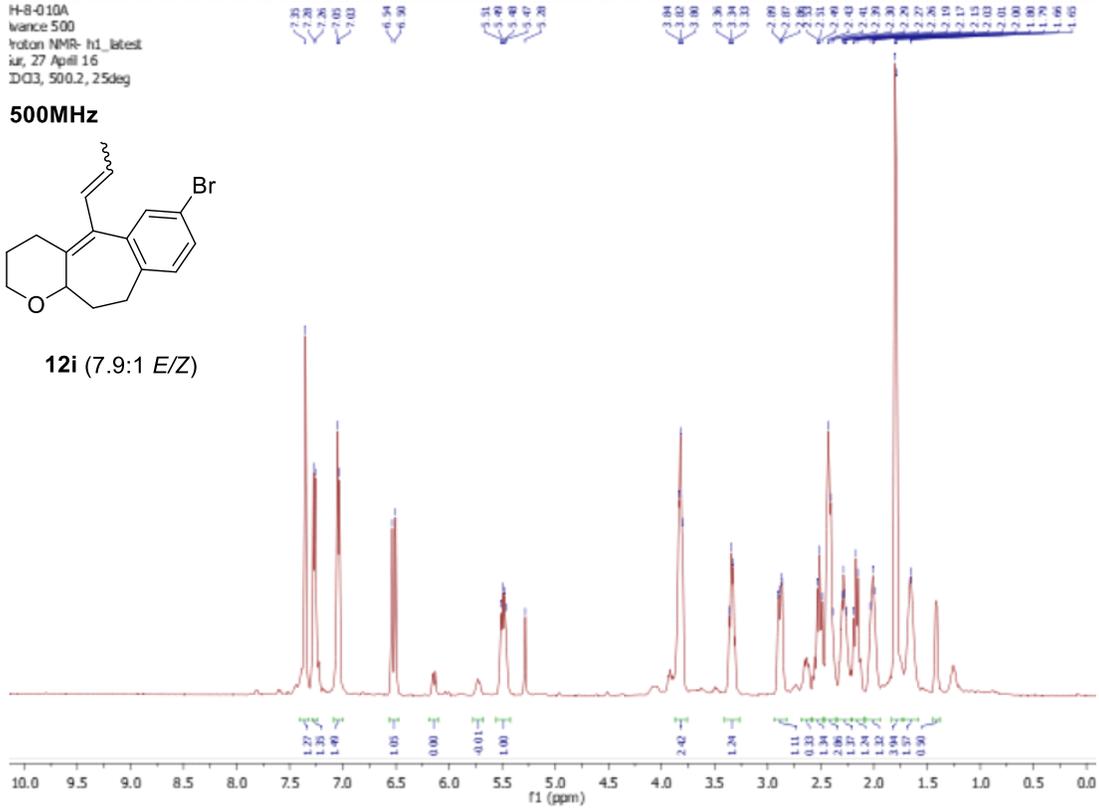


H-8-010A  
vance 500  
1 proton NMR- h1\_latest  
sur, 27 April 16  
DQ3, 500.2, 25deg

500MHz

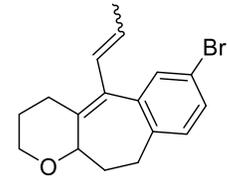


12i (7.9:1 E/Z)

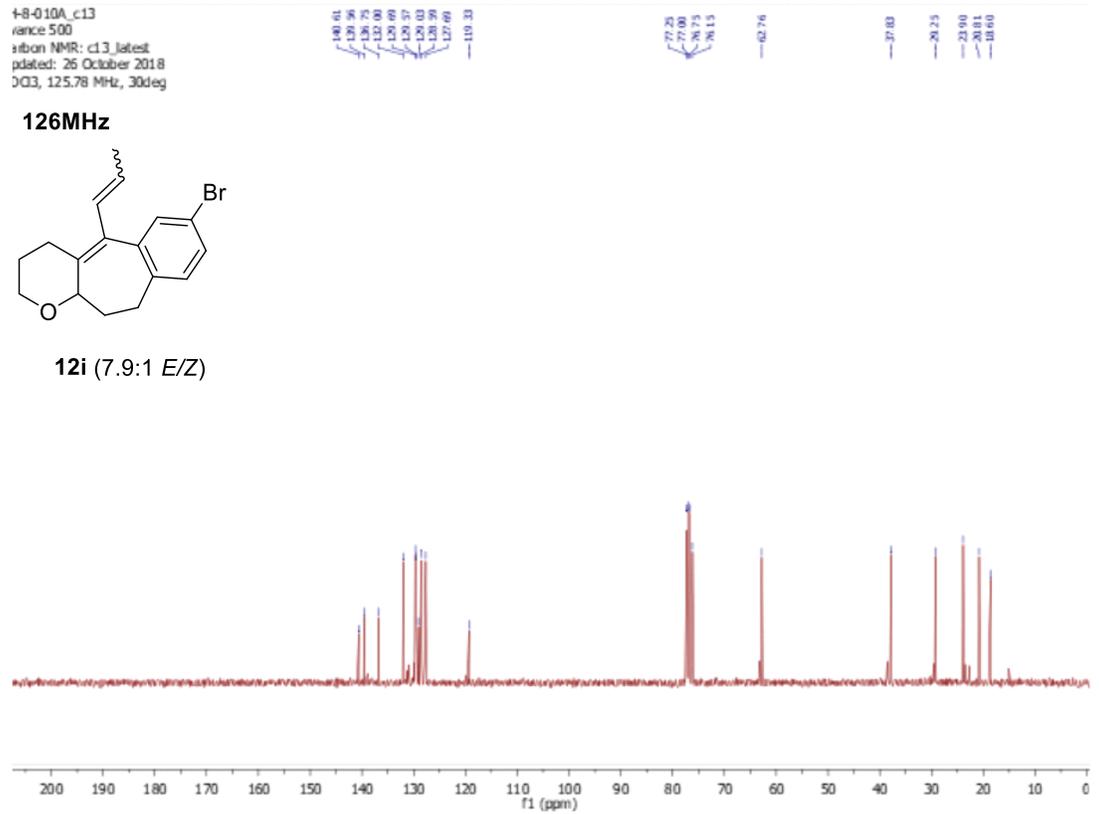


H-8-010A\_c13  
vance 500  
13C NMR- c13\_latest  
dated: 26 October 2018  
DQ3, 125.78 MHz, 30deg

126MHz



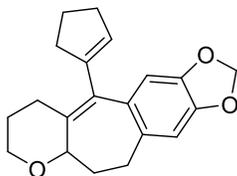
12i (7.9:1 E/Z)



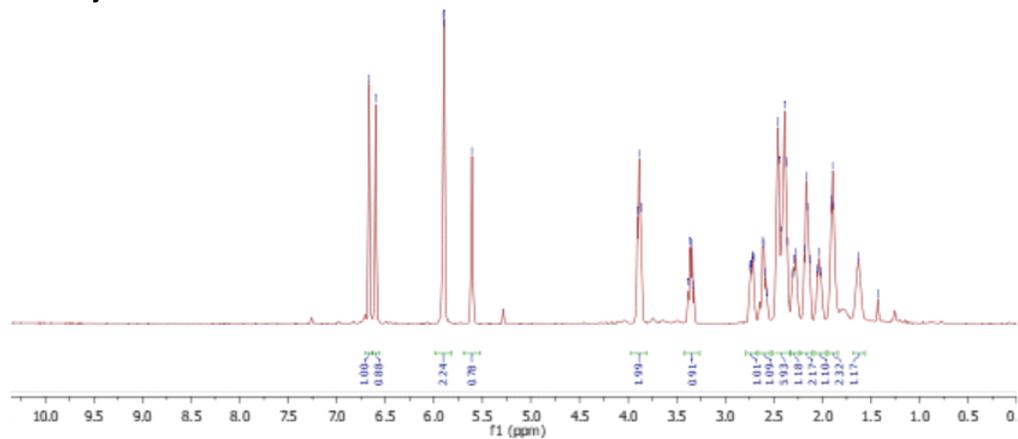
17A  
500  
NMR- h1\_latest  
April 16  
500.2, 25deg

6.67  
6.68  
5.90  
5.90  
5.90  
5.91  
3.90  
3.89  
3.87  
3.88  
3.88  
3.89  
2.75  
2.74  
2.74  
2.72  
2.71  
2.71  
2.62  
2.60  
2.46  
2.45  
2.45  
2.38  
2.36  
2.20  
2.20  
2.19  
2.17  
2.13  
2.03

500MHz



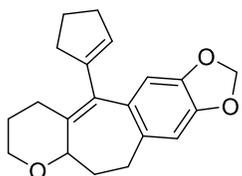
12j



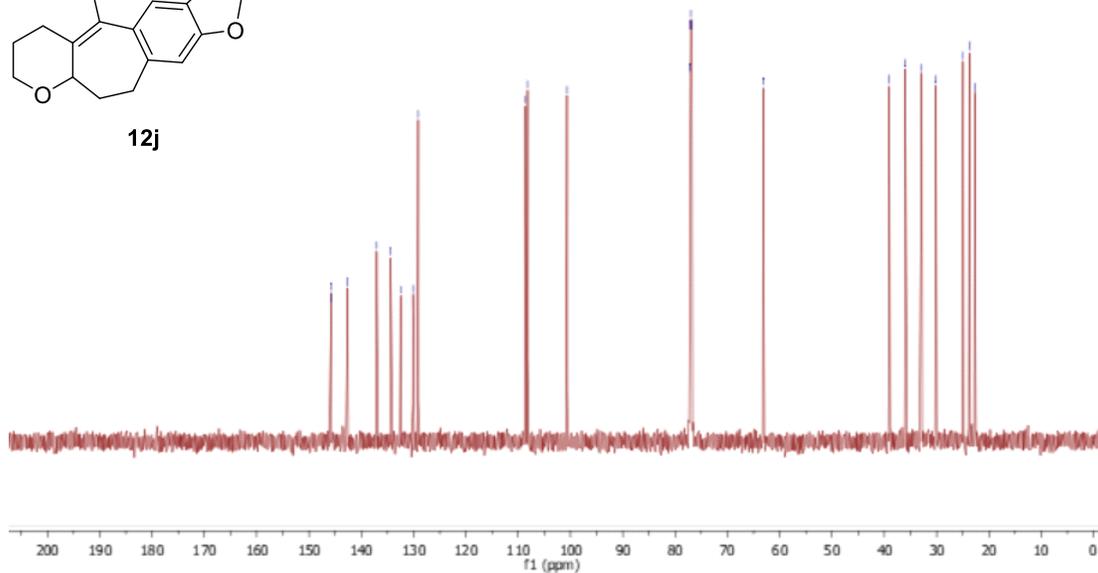
H-8-007A\_c13  
vance 500  
arbon NMR: c13\_latest  
dated: 26 October 2018  
DCI3, 125.78 MHz, 30deg

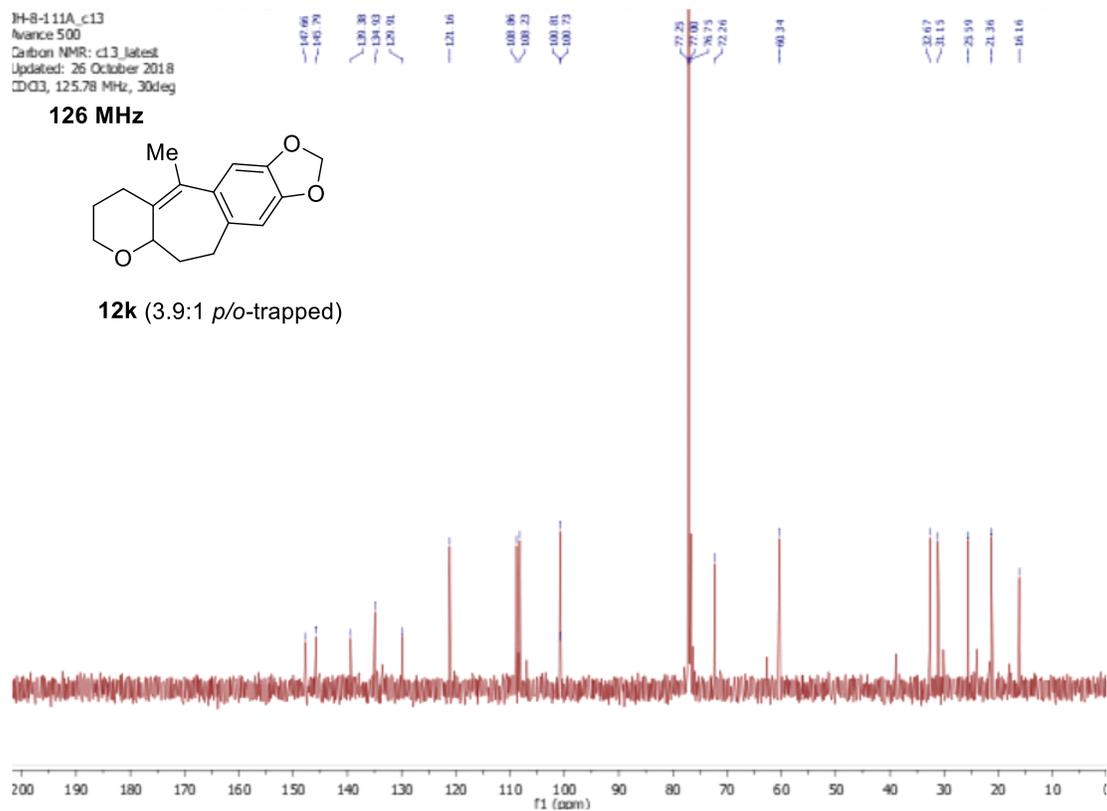
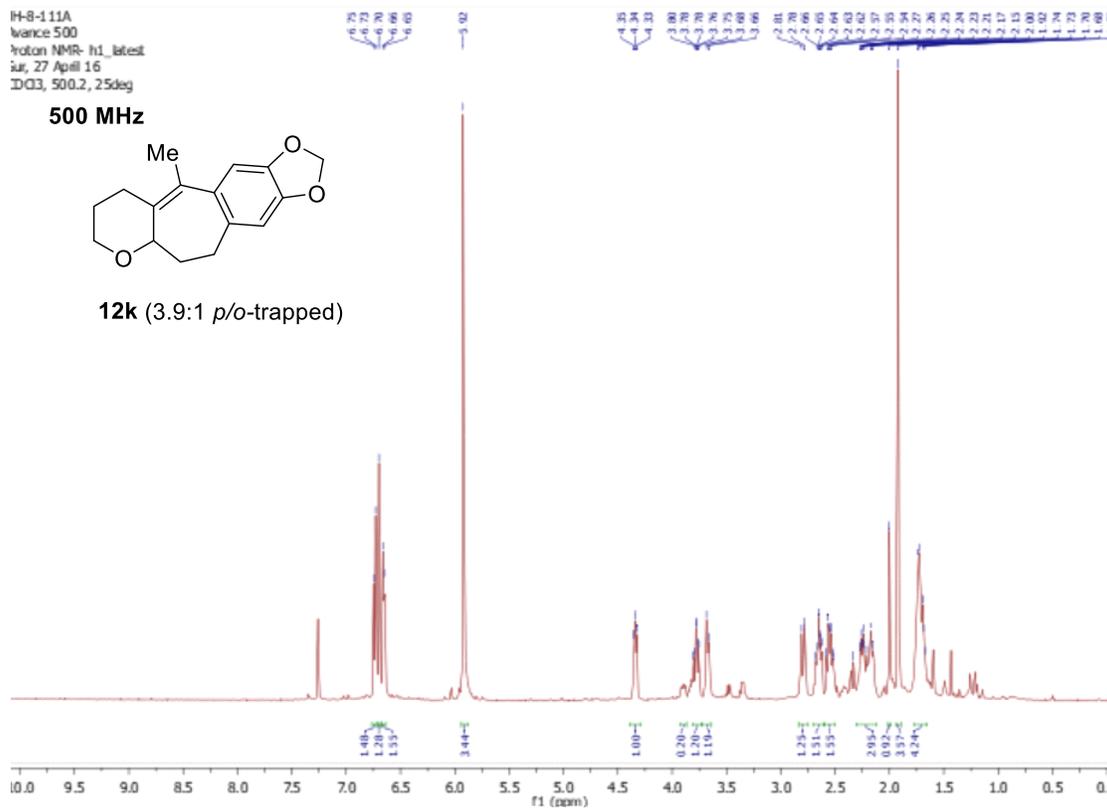
145.81  
145.75  
142.60  
137.05  
134.37  
133.49  
132.13  
108.99  
108.21  
100.67  
77.26  
77.00  
76.84  
76.73  
-0.17  
38.04  
35.90  
32.84  
30.13  
28.94  
23.76  
22.71

126MHz



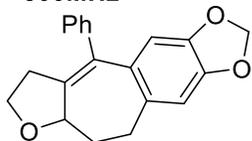
12j



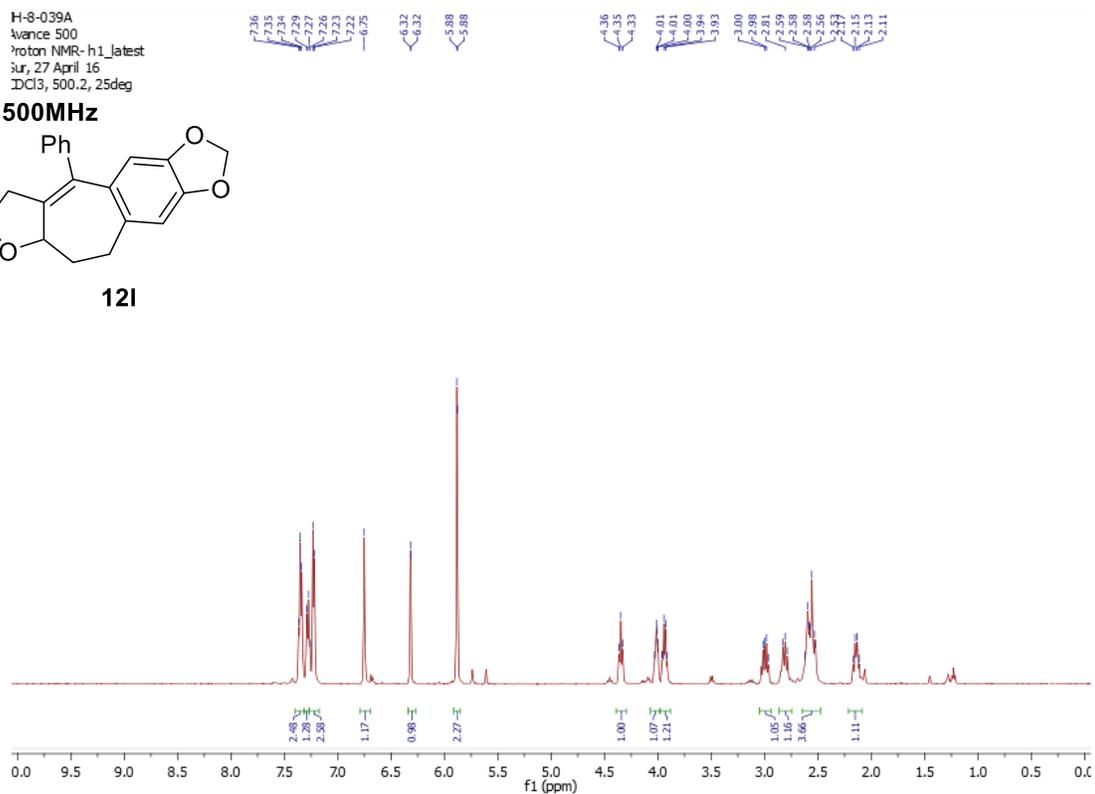


H-8-039A  
 vance 500  
 1H NMR- h1\_latest  
 sur, 27 April 16  
 CDCl3, 500.2, 25deg

500MHz

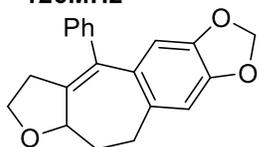


12I

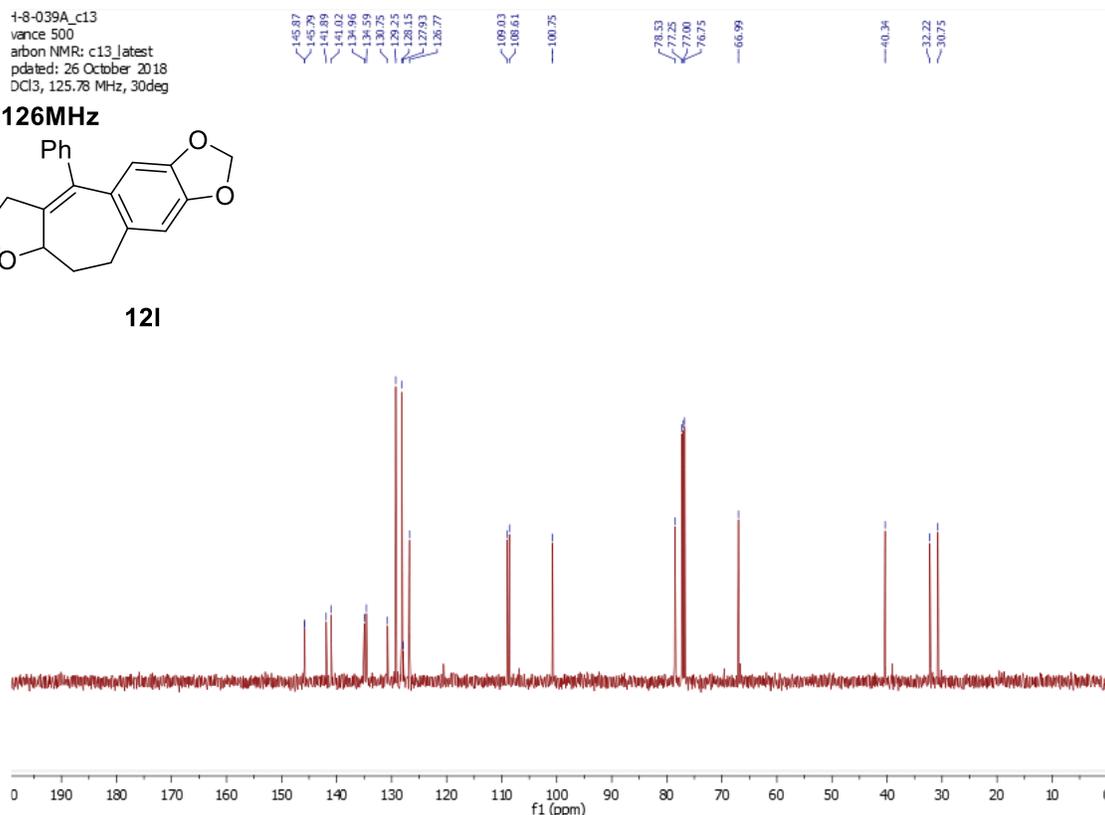


1-8-039A\_c13  
 vance 500  
 carbon NMR: c13\_latest  
 pdated: 26 October 2018  
 CDCl3, 125.78 MHz, 30deg

126MHz

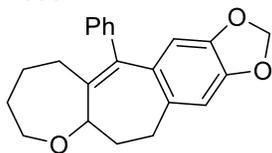


12I

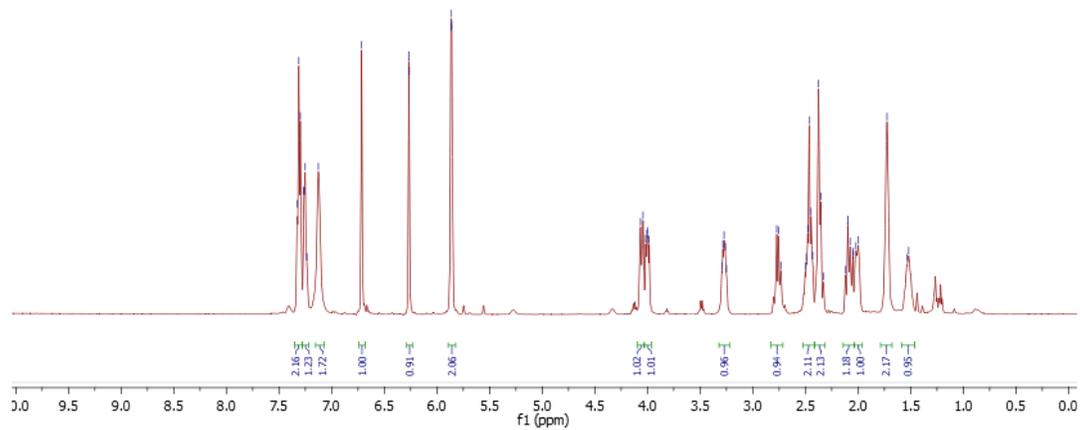


H-8-040A  
vance 500  
Proton NMR- h1\_latest  
Acq, 27 April 16  
DCl3, 500.2, 25deg

500MHz

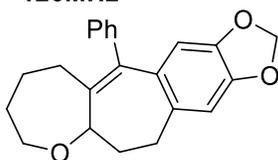


12m

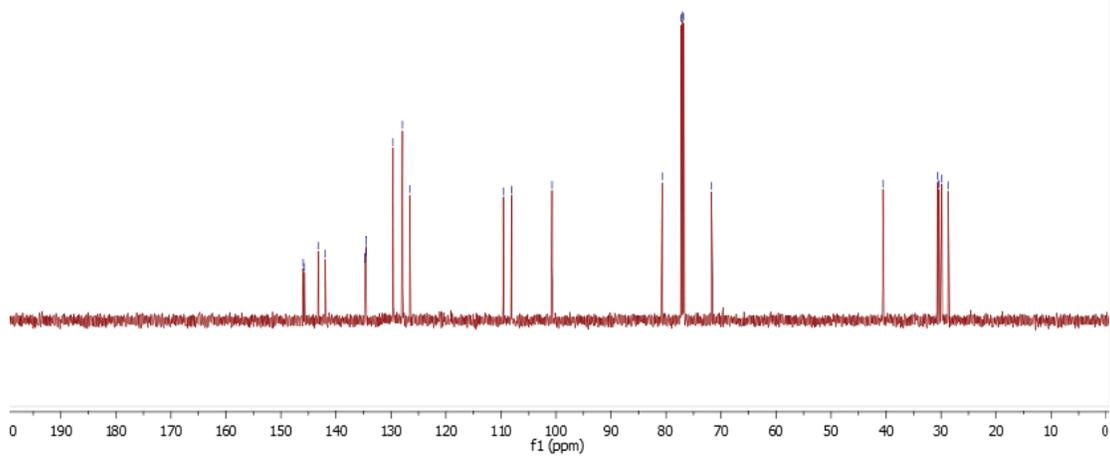


H-8-040A\_c13  
vance 500  
Carbon NMR: c13\_latest  
Updated: 26 October 2018  
DCl3, 125.78 MHz, 30deg

126MHz

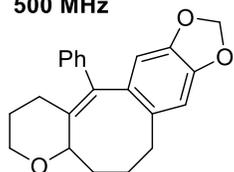


12m

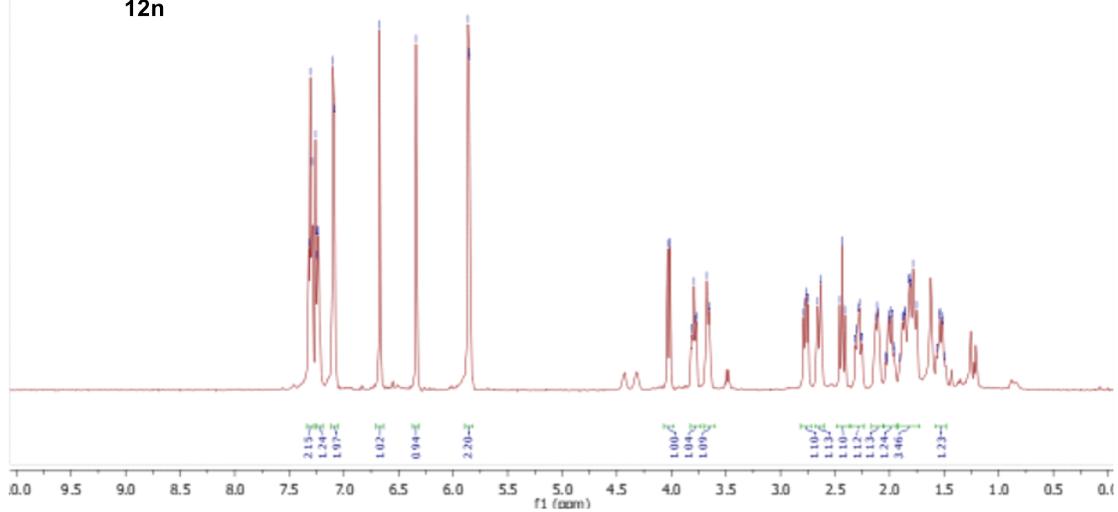


JH-9-0508  
Avance 500  
Proton NMR- h1\_latest  
Sun, 27 April 16  
CDCl3, 500.2, 25deg

500 MHz

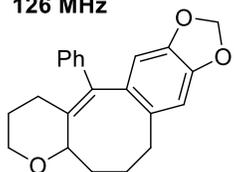


12n

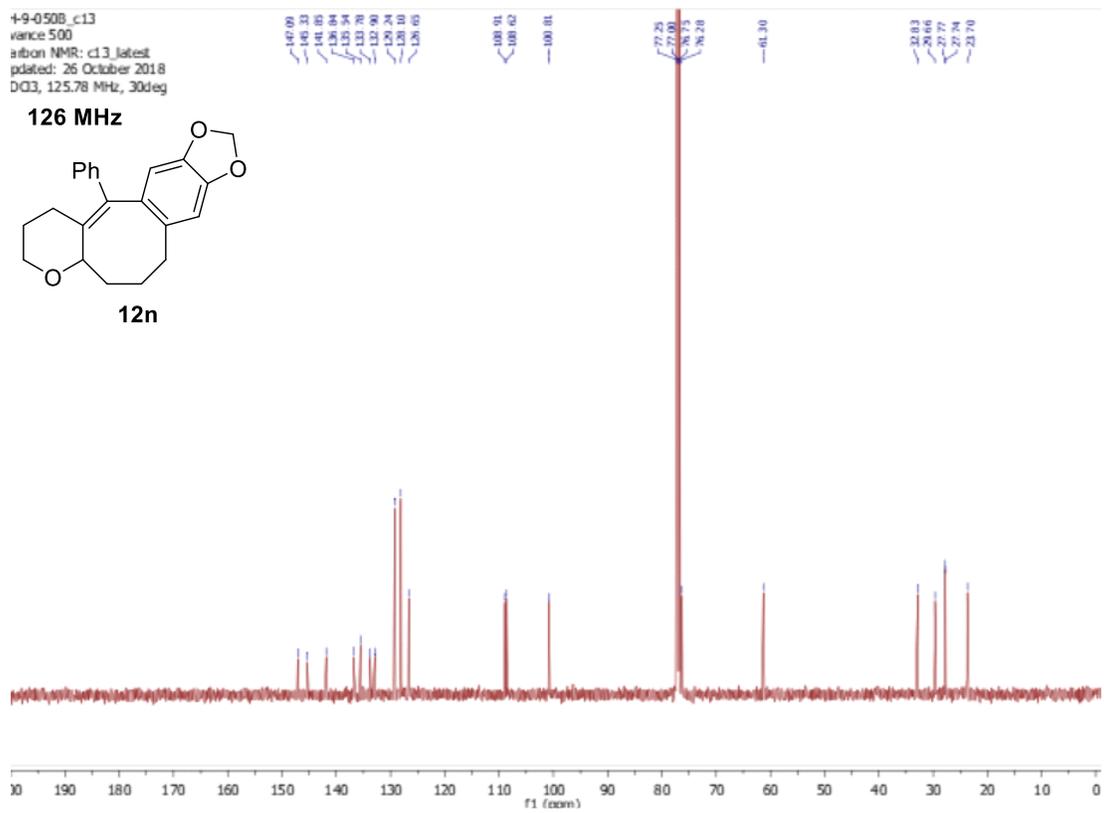


+9-0508\_c13  
Avance 500  
Carbon NMR- c13\_latest  
Updated: 26 October 2018  
CDCl3, 125.78 MHz, 30deg

126 MHz

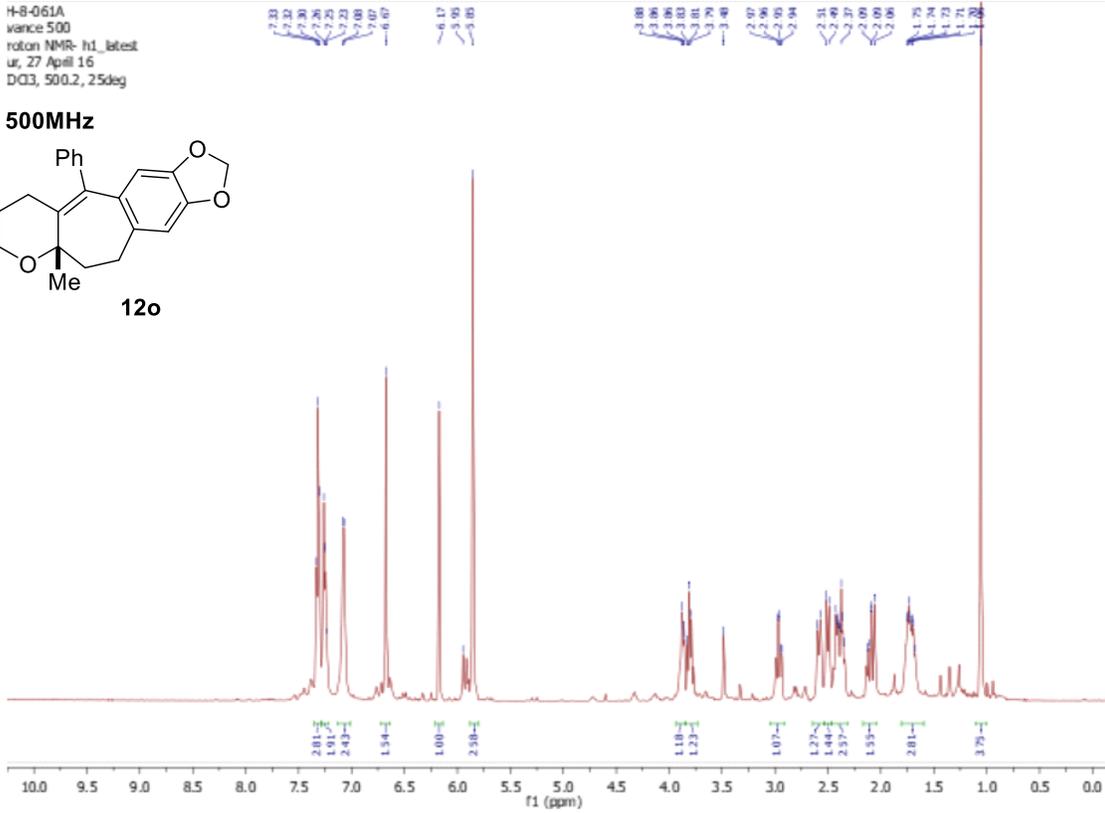
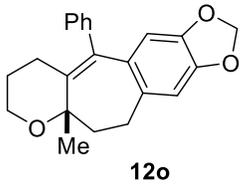


12n



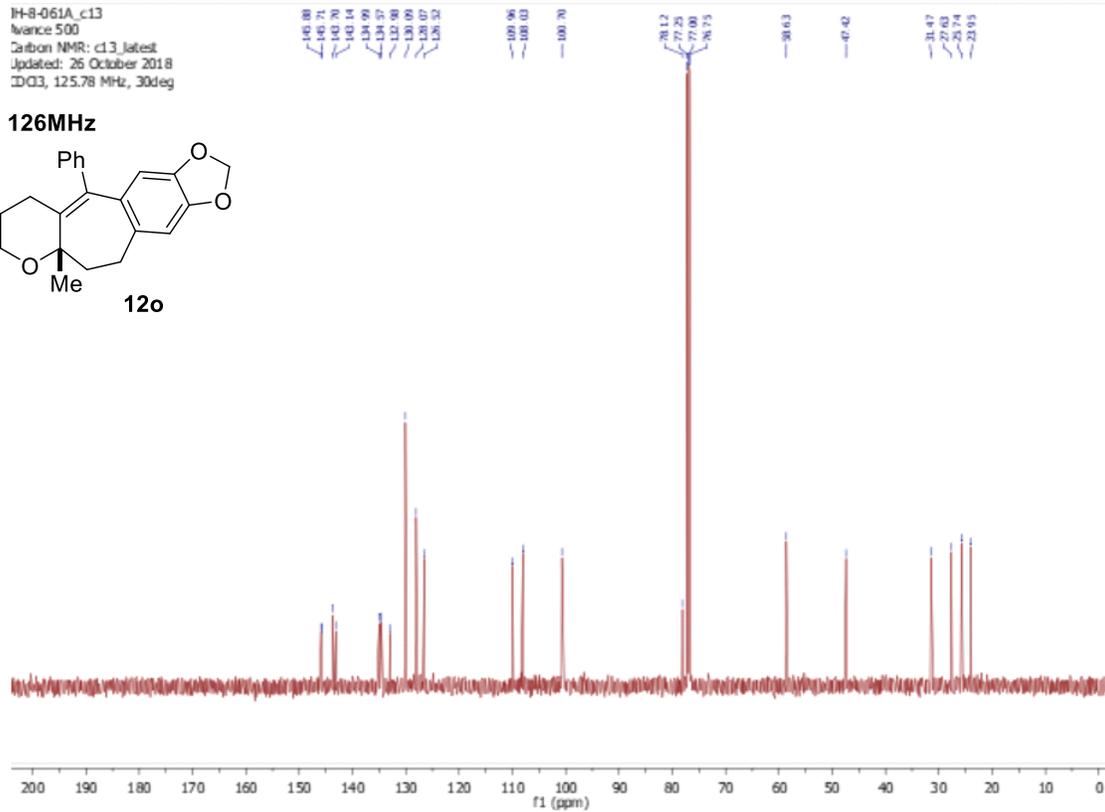
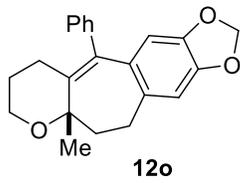
H-8-061A  
avance 500  
rotor NMR- h1\_test  
ur, 27 April 16  
DCl3, 500.2, 2.5deg

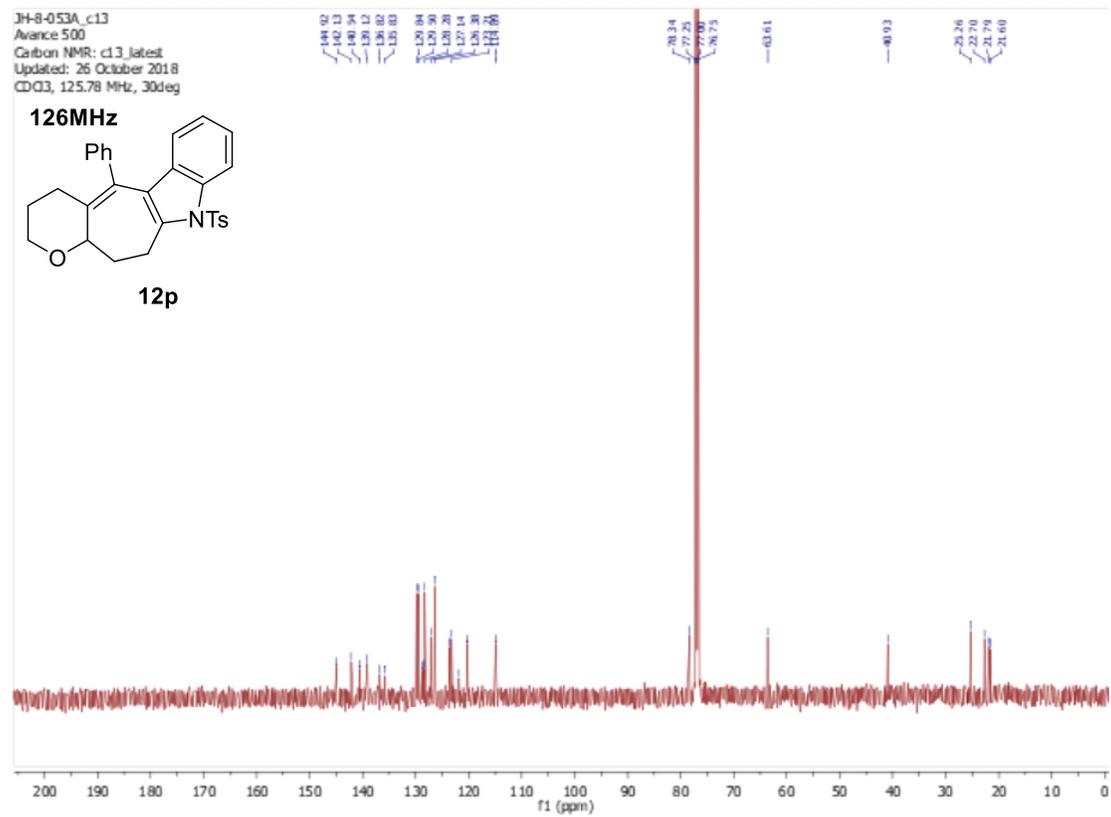
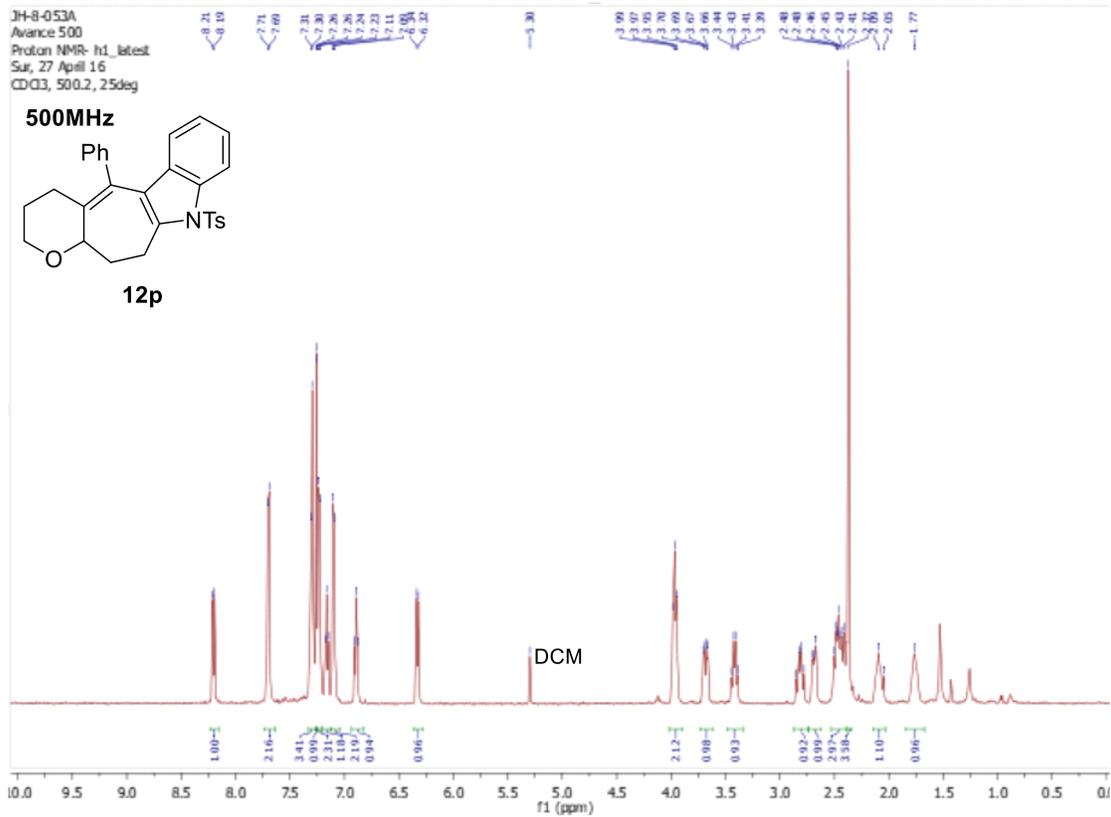
500MHz



H-8-061A\_c13  
avance 500  
Carbon NMR: c13\_test  
Updated: 26 October 2018  
DCl3, 125.78 MHz, 30deg

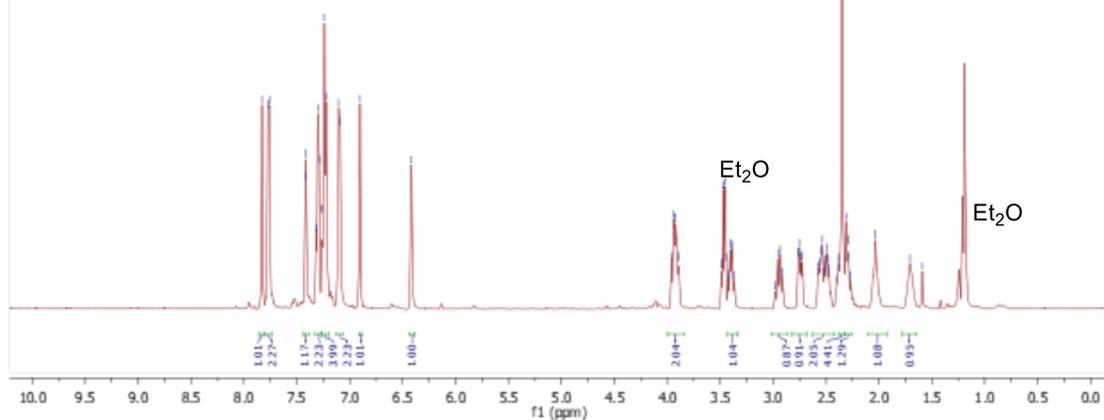
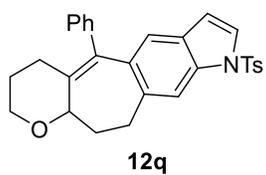
126MHz





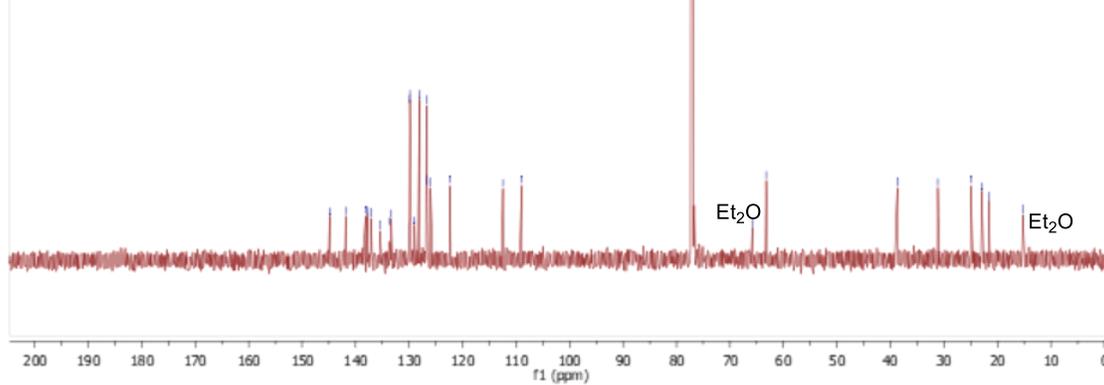
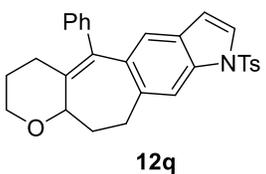
JH-8-073A  
Avance 500  
Proton NMR- h1\_test  
Sun, 27 April 16  
CDCl3, 500.2, 2.5deg

500MHz



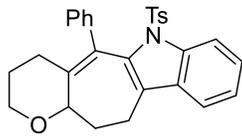
JH-8-073A\_c13  
Avance 500  
Carbon NMR: c13\_test  
Updated: 26 October 2018  
CDCl3, 125.78 MHz, 30deg

126MHz

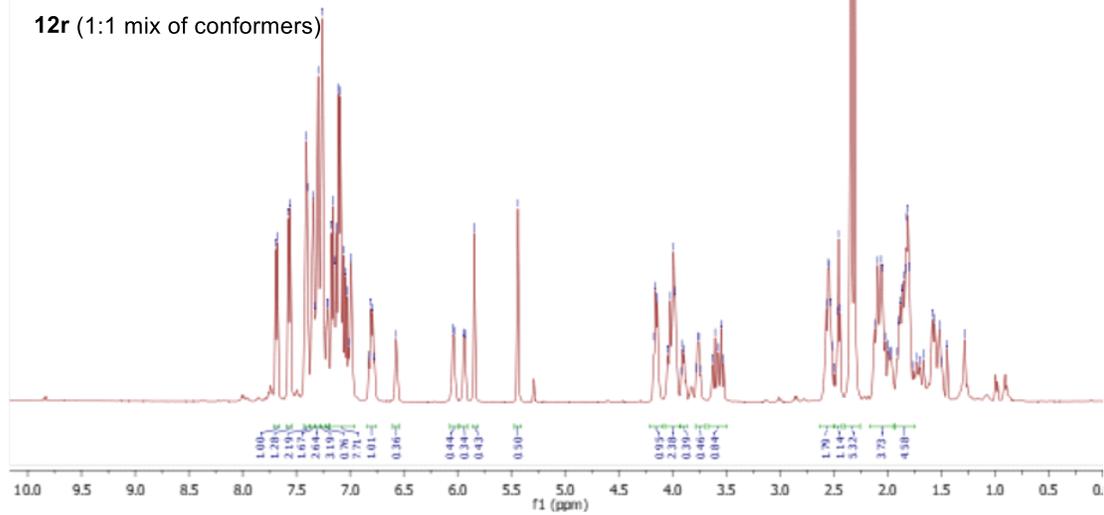


JH-8-974a  
 Name: 500  
 Proton NMR- h1\_test  
 Sur, 27 April 16  
 CDCl3, 500.2, 25deg

500MHz

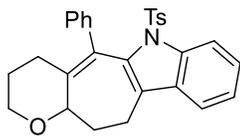


12r (1:1 mix of conformers)

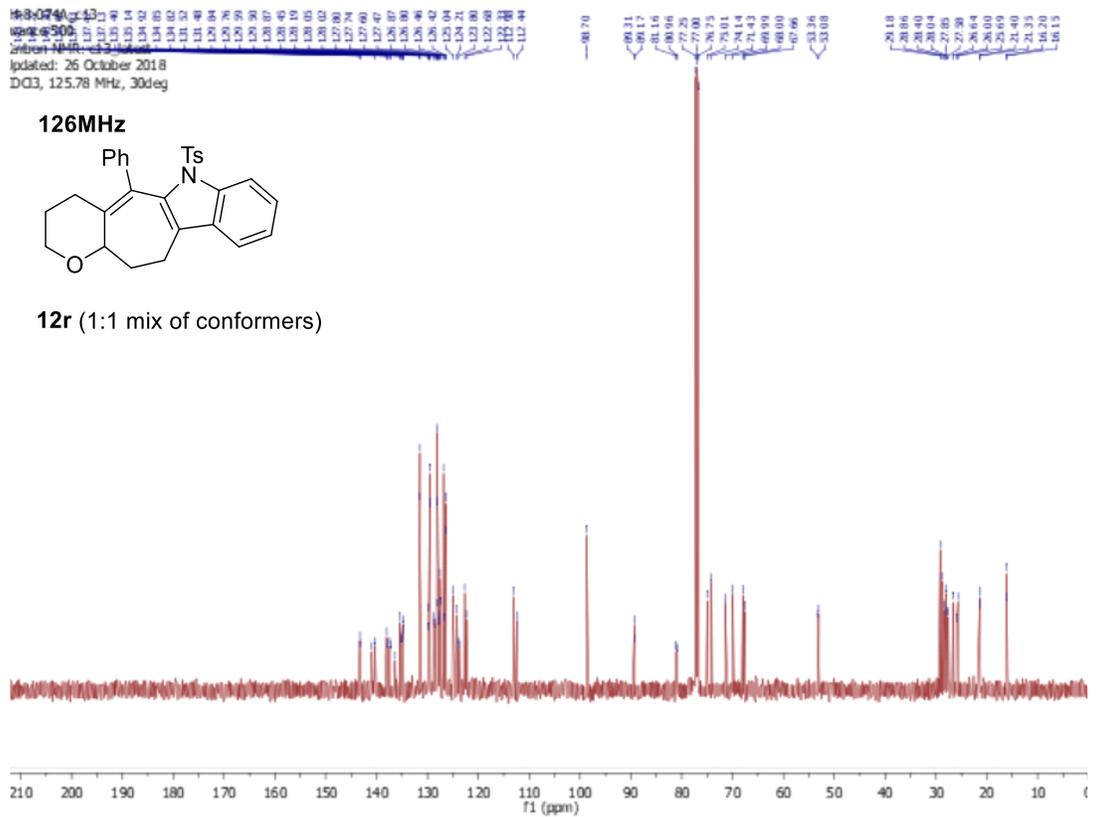


JH-8-974a-43  
 Name: 500  
 Carbon NMR- h1  
 Updated: 26 October 2018  
 CDCl3, 125.78 MHz, 30deg

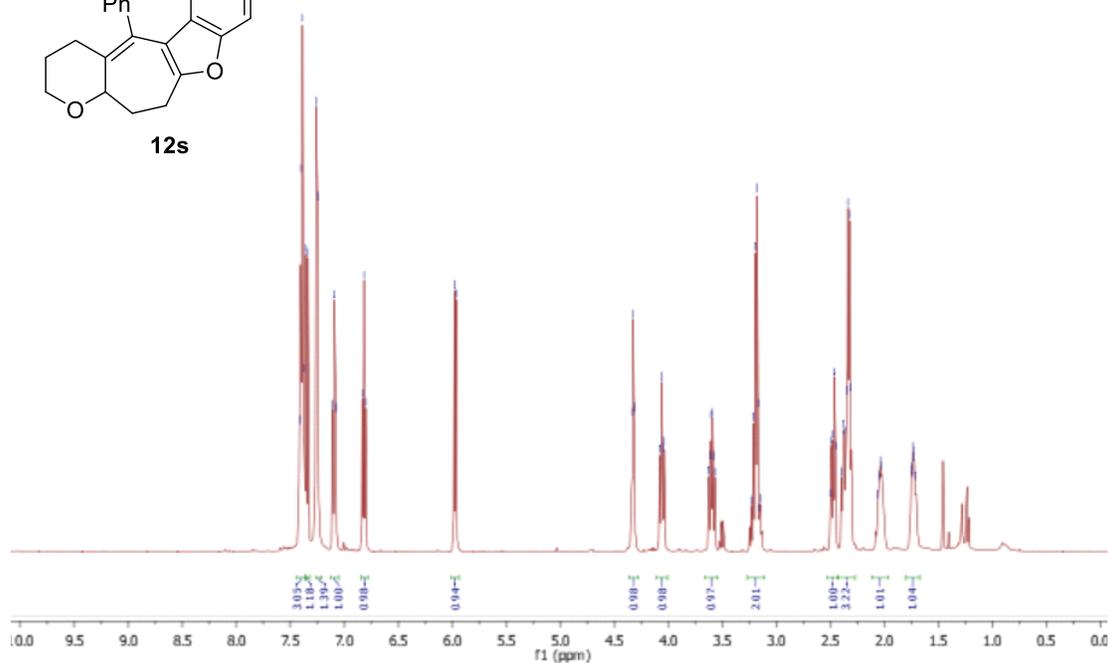
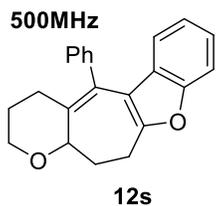
126MHz



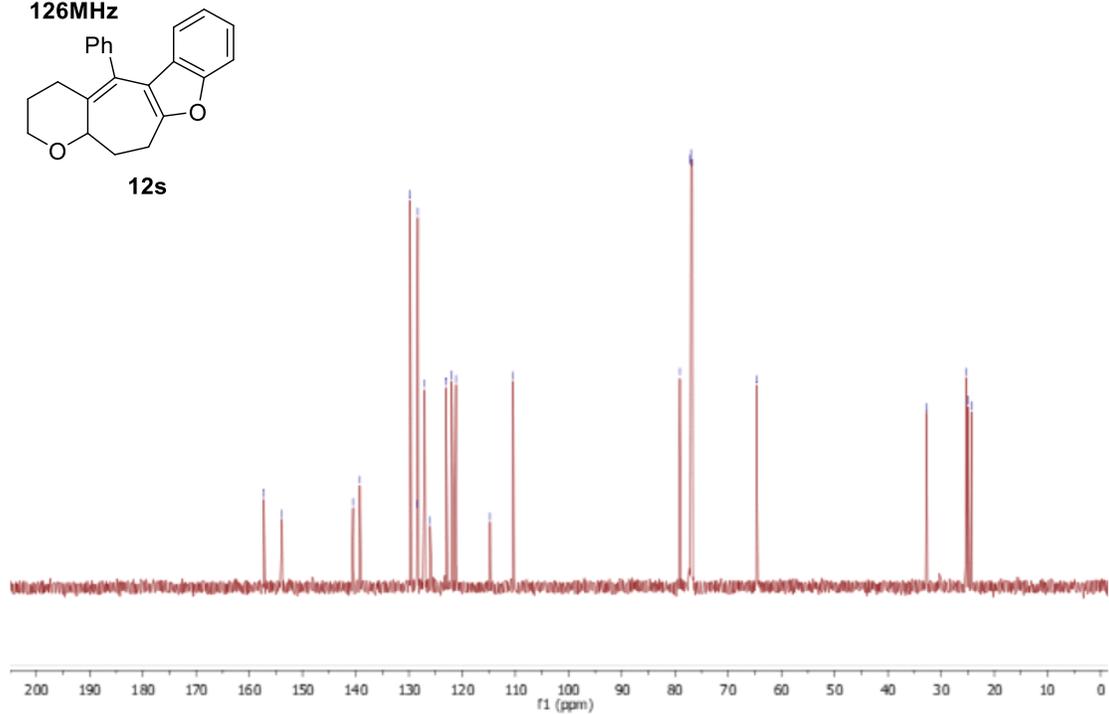
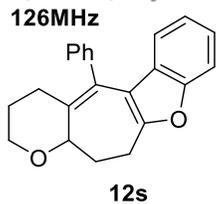
12r (1:1 mix of conformers)



JH-7-201A  
Avance 500  
Proton NMR- h1\_latest  
Sun, 27 April 16  
CDCl3, 500.2, 25deg



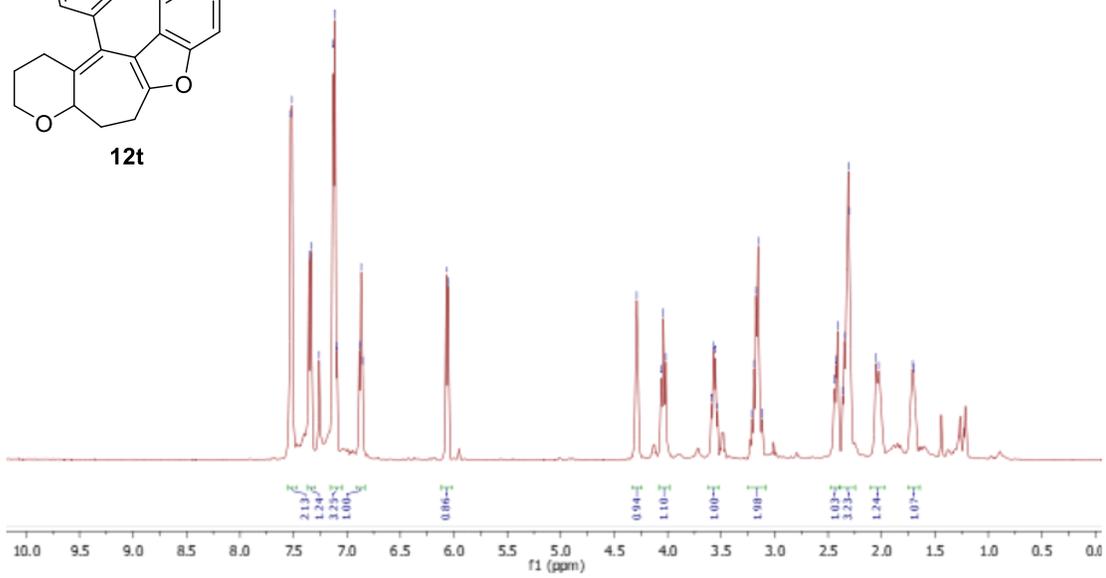
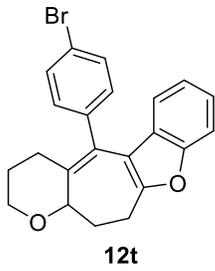
JH-7-201A\_c13  
Avance 500  
Carbon NMR: c13\_latest  
Updated: 26 October 2018  
CDCl3, 125.78 MHz, 30deg



H-7-2078\_1  
vance 500  
Proton NMR- h1\_test  
ur, 27 April 16  
DCl3, 500.2, 25deg



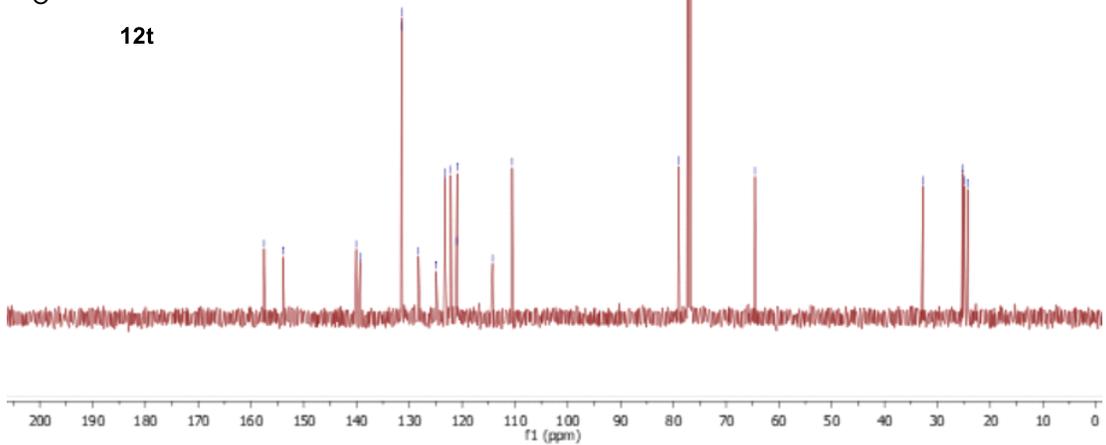
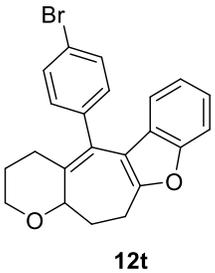
500MHz



H-7-2078\_1\_c13  
vance 500  
Carbon NMR- c13\_test  
Updated: 26 October 2018  
DCl3, 125.78 MHz, 30deg



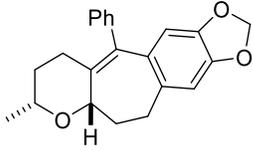
126MHz



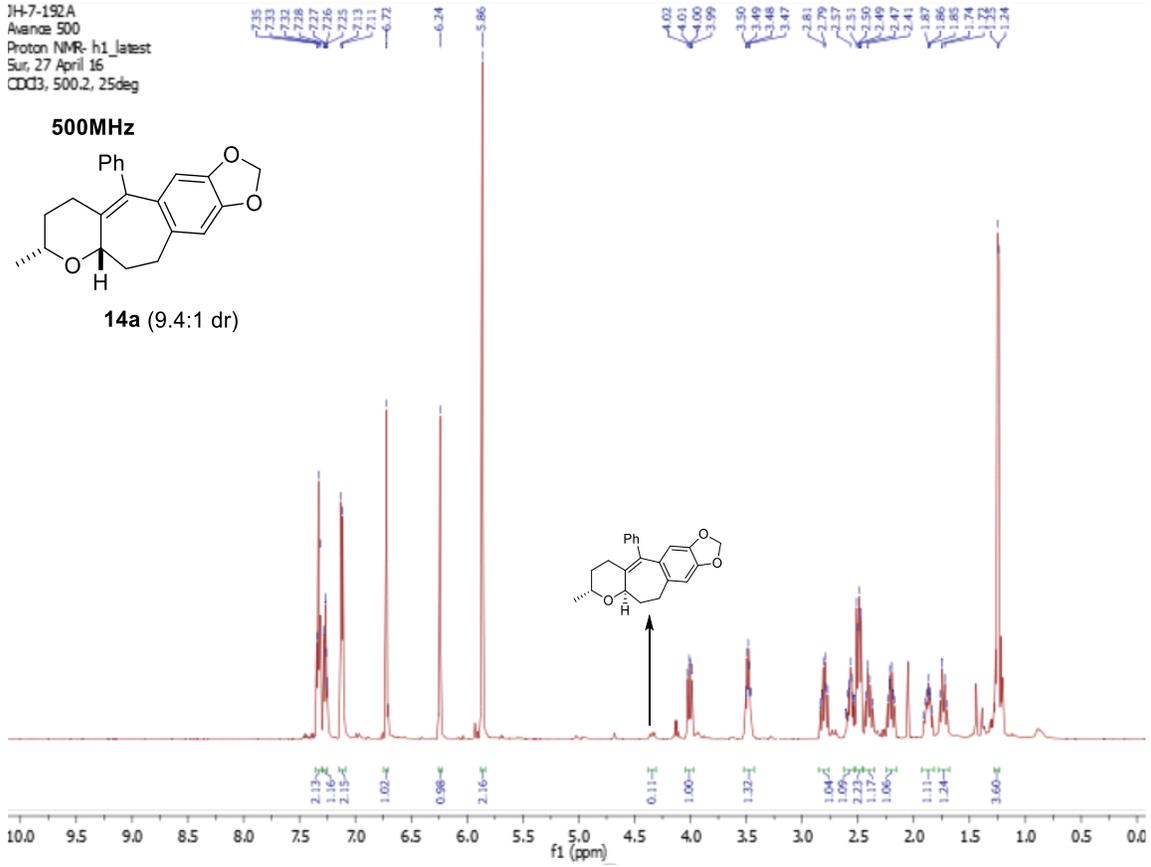


JH-7-192A  
Avance 500  
Proton NMR- h1\_latest  
Sun, 27 April 16  
CDCl3, 500.2, 25deg

500MHz

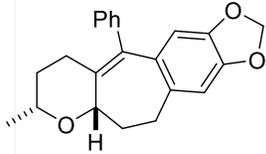


14a (9.4:1 dr)

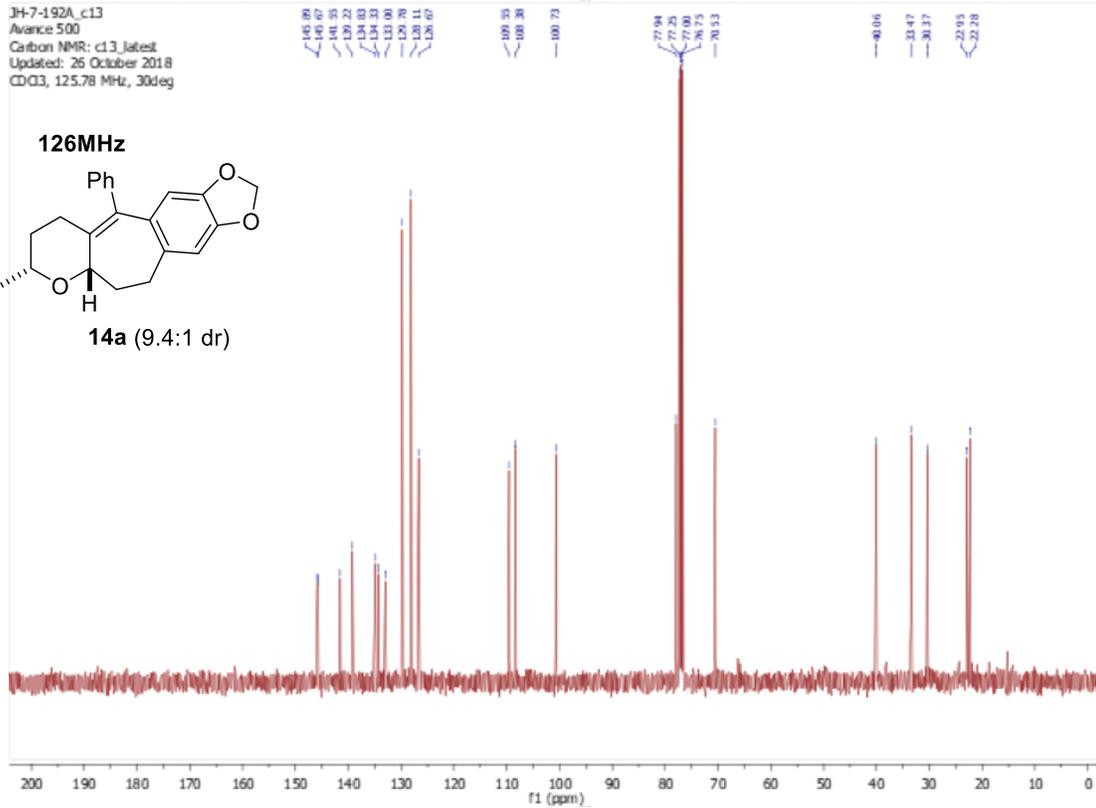


JH-7-192A\_c13  
Avance 500  
Carbon NMR: c13\_latest  
Updated: 26 October 2018  
CDCl3, 125.78 MHz, 30deg

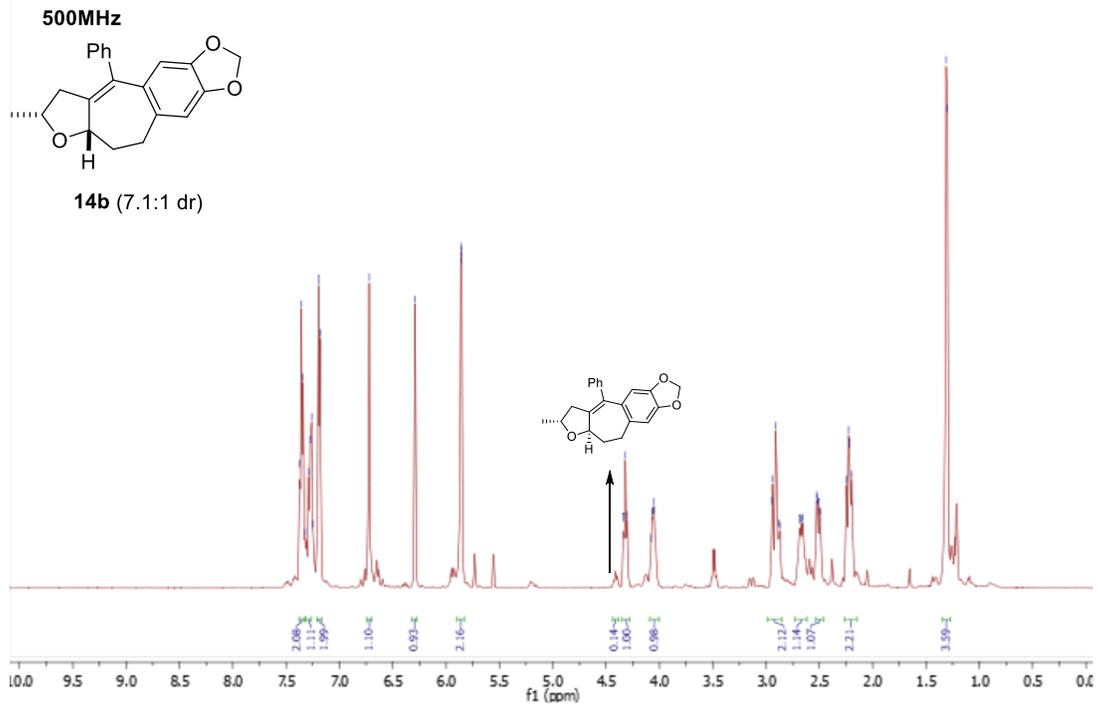
126MHz



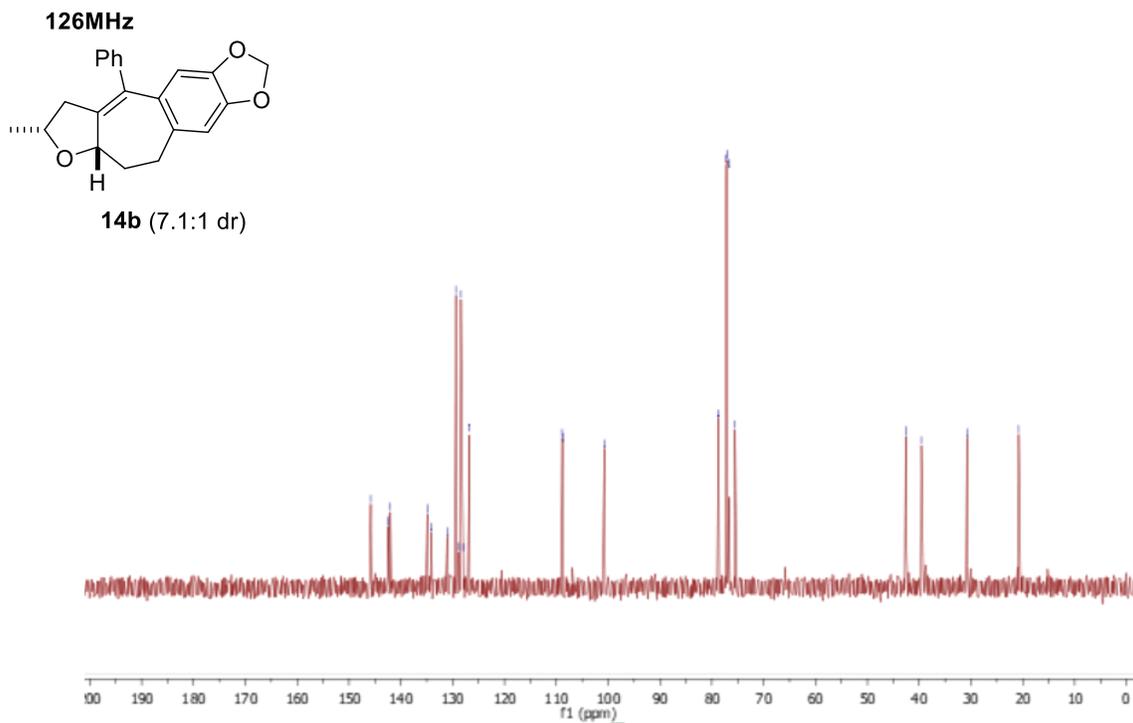
14a (9.4:1 dr)



JH-8-066A  
 Avance 500  
 Proton NMR: h1\_latest  
 Sat, 27 April 16  
 CDCl3, 500.2, 25deg



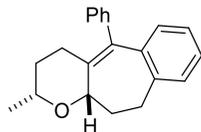
5A\_c13  
 100  
 NMR: c13\_latest  
 : 26 October 2018  
 25.78 MHz, 30deg



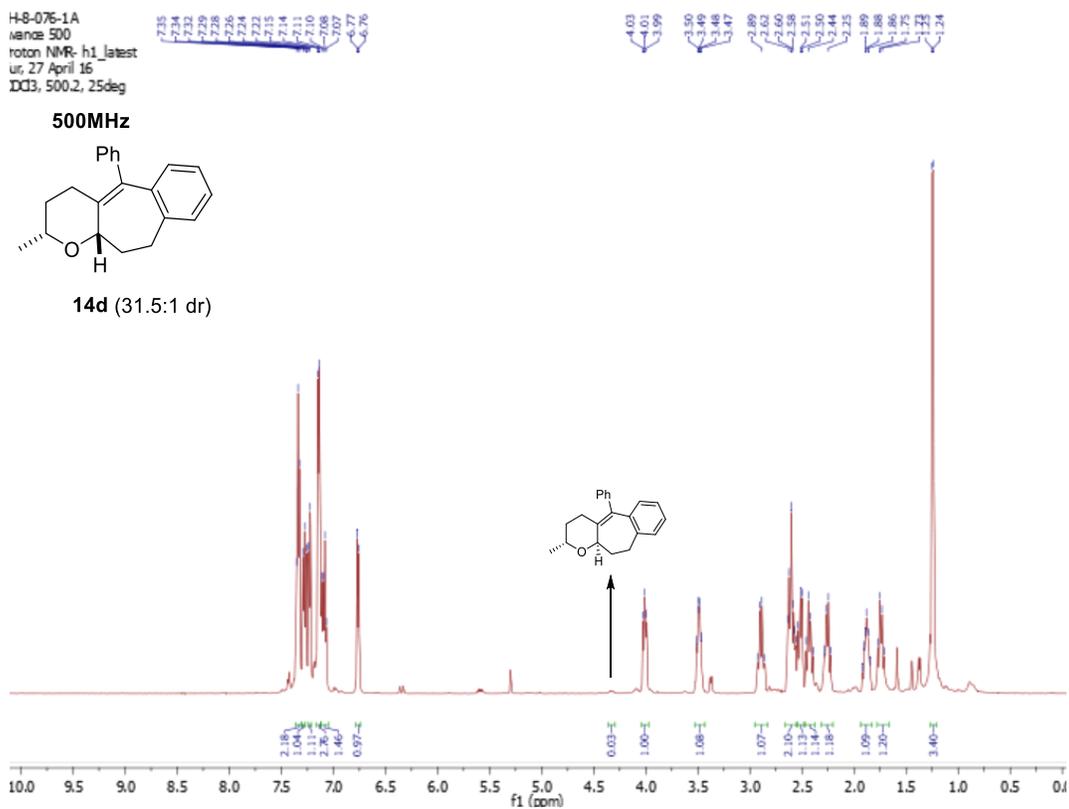


H-8-076-1A  
Avance 500  
1H NMR- h1\_latest  
ur, 27 April 16  
CDCl3, 500.2, 25deg

500MHz

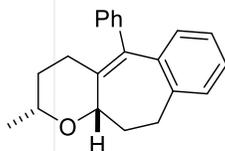


14d (31.5:1 dr)

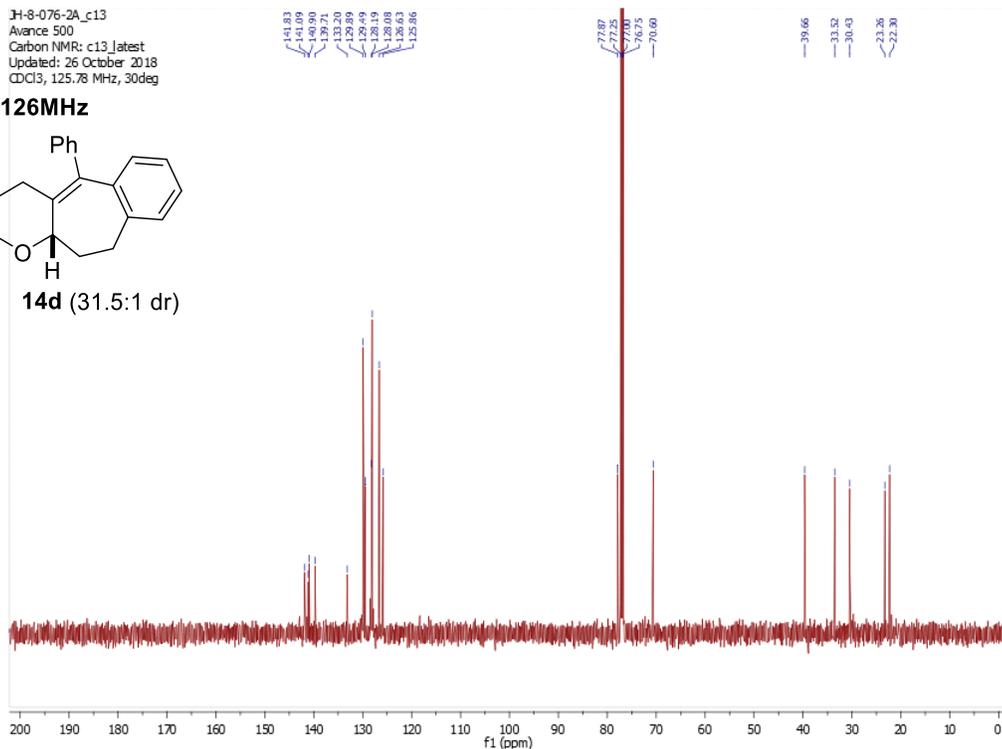


H-8-076-2A\_c13  
Avance 500  
Carbon NMR: c13\_latest  
Updated: 26 October 2018  
CDCl3, 125.78 MHz, 30deg

126MHz

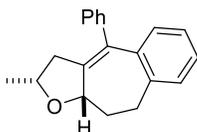


14d (31.5:1 dr)

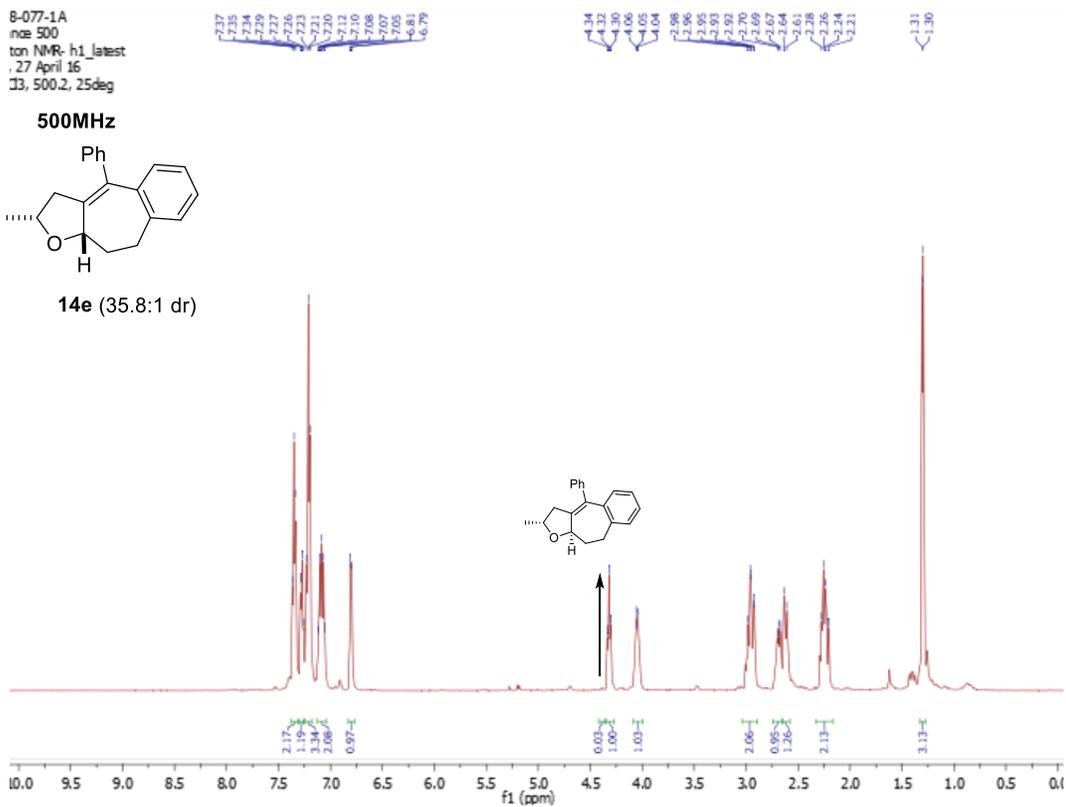


8-077-1A  
nve 500  
ton NMR- h1\_latest  
. 27 April 16  
13, 500.2, 25deg

500MHz

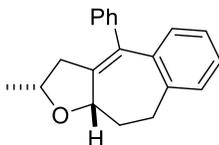


14e (35.8:1 dr)

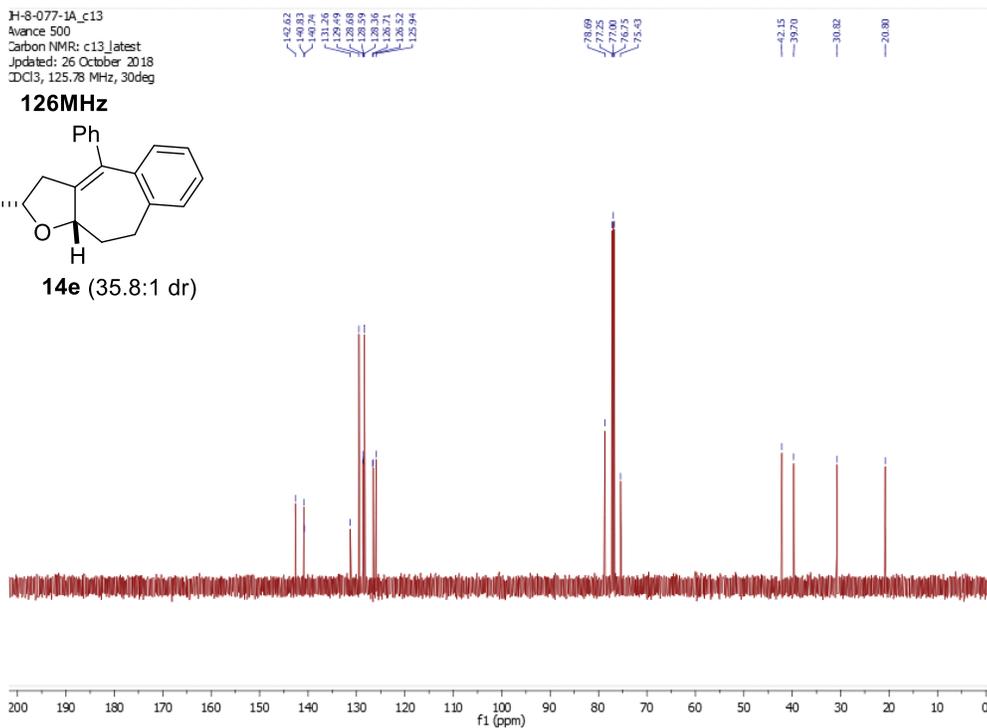


H-8-077-1A\_c13  
Nve 500  
Carbon NMR: c13\_latest  
Updated: 26 October 2018  
CDCl3, 125.78 MHz, 30deg

126MHz



14e (35.8:1 dr)

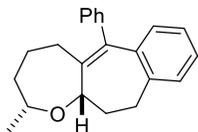


JH-8-077-2A  
Avance 500  
Proton NMR- h1\_latest  
Sur. 27 April 16  
CDCl3, 500.2, 25deg

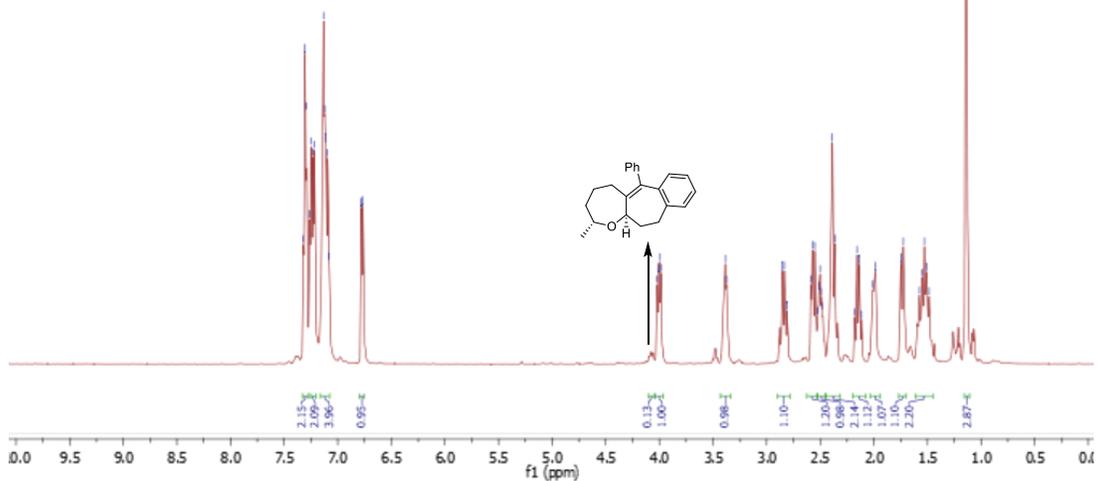
7.32  
7.31  
7.30  
7.26  
7.25  
7.23  
7.22  
7.13  
7.12  
7.10  
7.09  
7.08  
6.77

4.02  
4.01  
4.00  
3.99  
3.39  
3.37  
2.85  
2.83  
2.57  
2.55  
2.50  
2.39  
2.36  
2.16  
2.14  
1.99  
1.75  
1.72  
1.55  
1.53  
1.51  
1.13

500MHz



14f (7.9:1 dr)



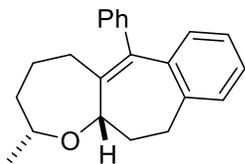
H-8-077-2A\_c13  
Avance 500  
Carbon NMR- c13\_latest  
Updated: 26 October 2018  
CDCl3, 125.78 MHz, 30deg

144.21  
152.07  
140.75  
140.75  
138.19  
129.77  
129.38  
127.85  
127.83  
126.67  
126.50  
125.96

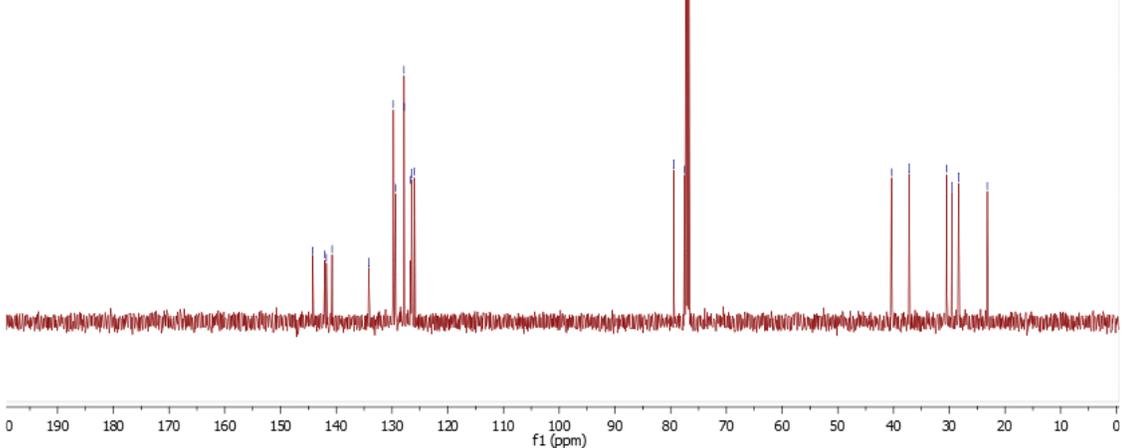
79.43  
77.32  
77.25  
77.00  
76.75

40.39  
37.22  
30.47  
28.51  
28.34  
23.17

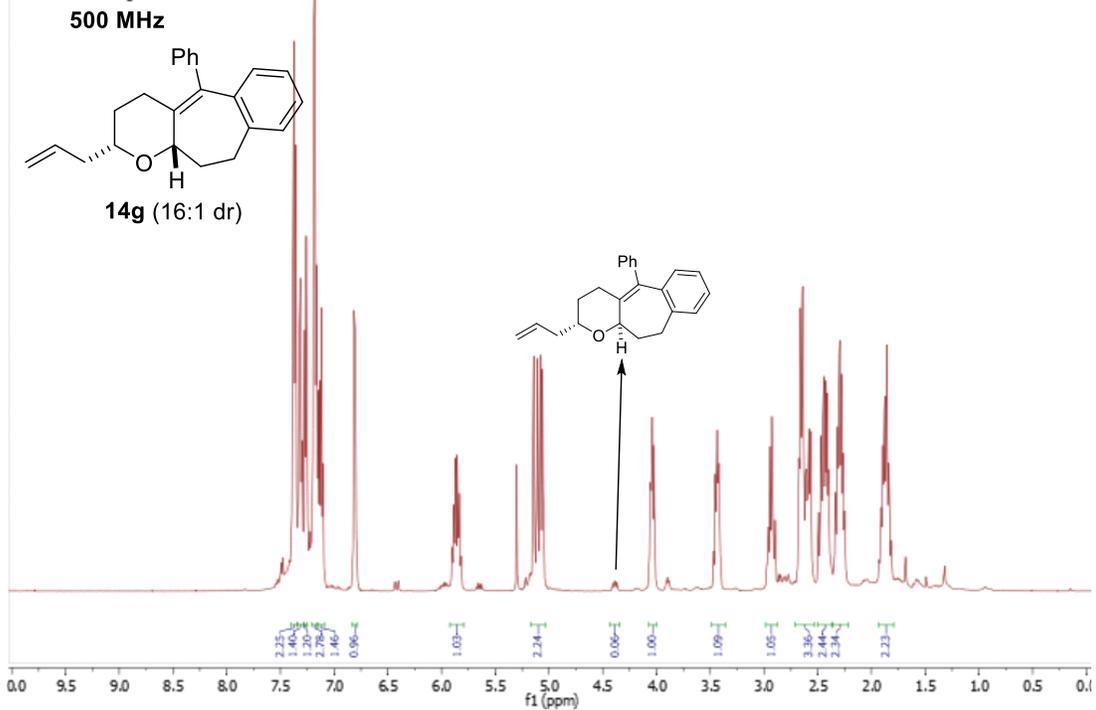
126MHz



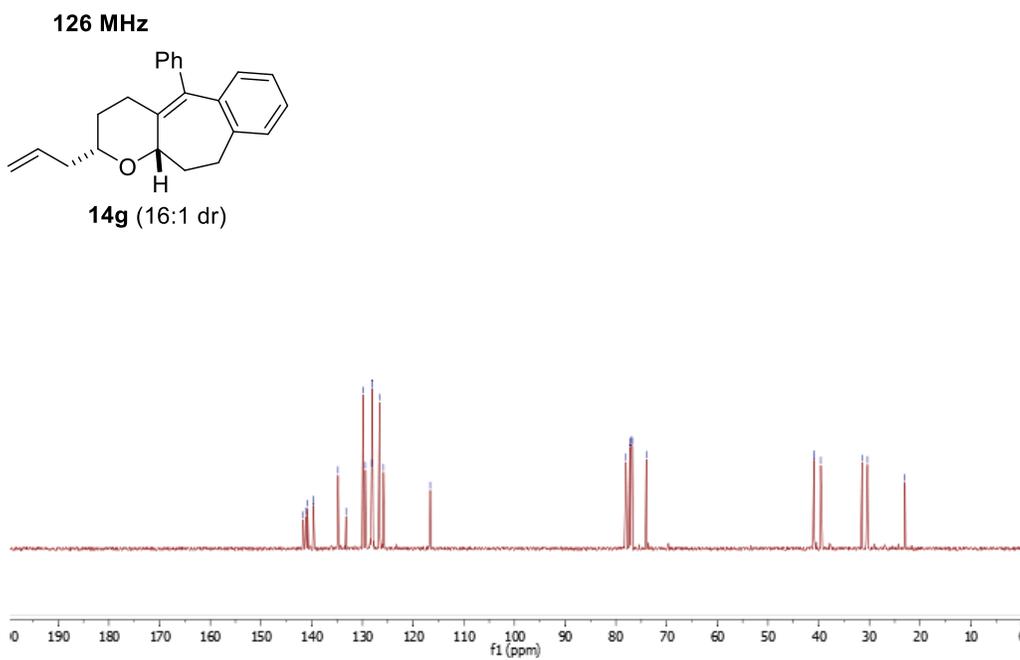
14f (7.9:1 dr)

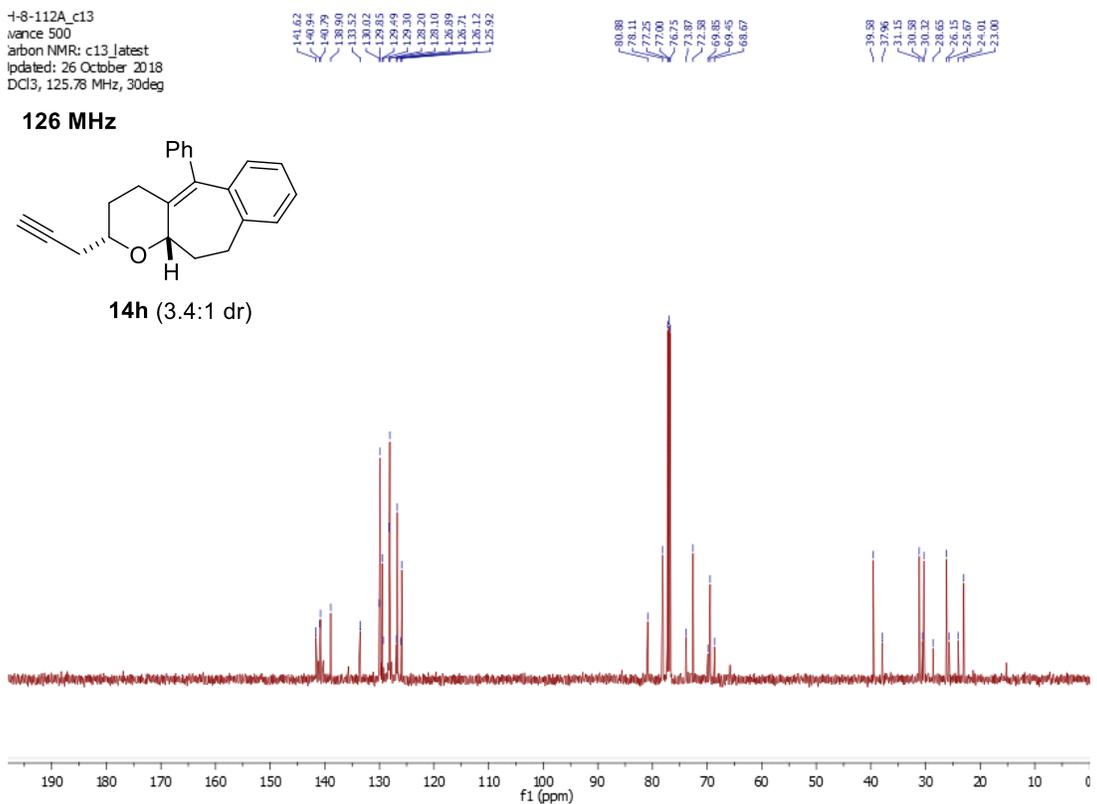
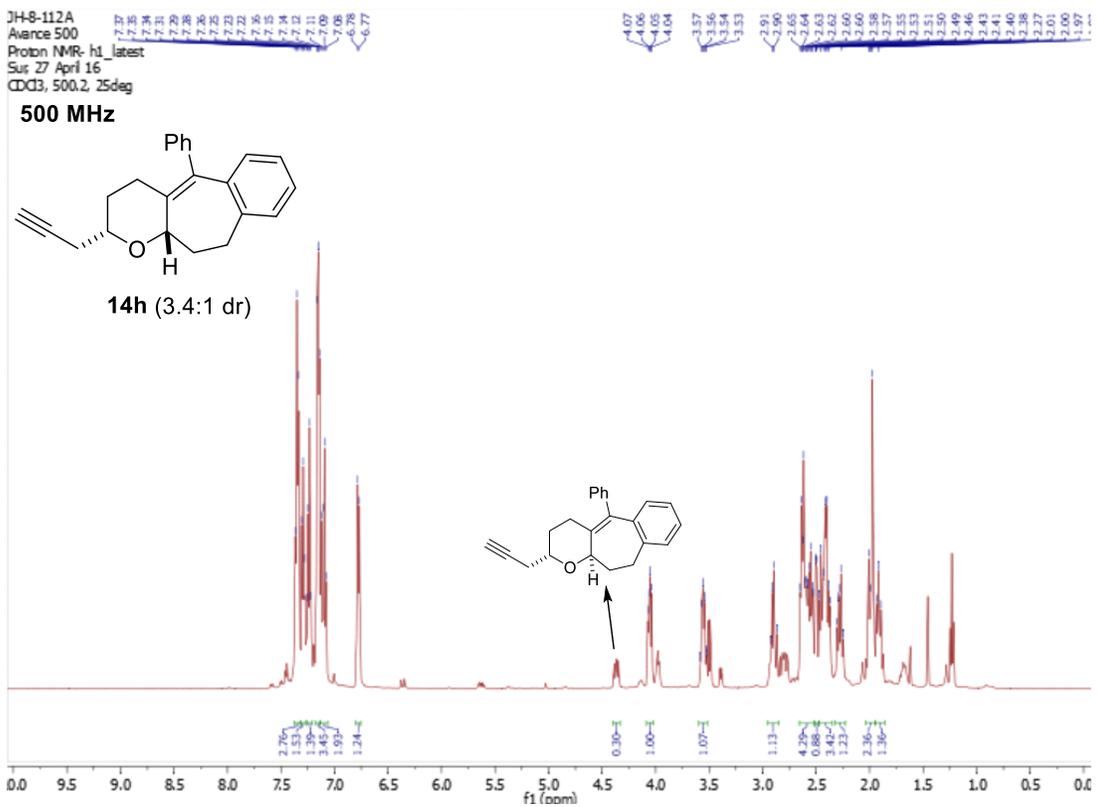


JH-8-084A  
Avance 500  
Proton NMR- f1\_latest  
Sun 27 April 16  
CDCl3, 500.2, 25deg



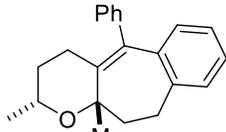
JH-8-084A\_c13  
Avance 500  
Carbon NMR- c13\_latest  
Updated: 26 October 2018  
CDCl3, 125.78 MHz, 30deg



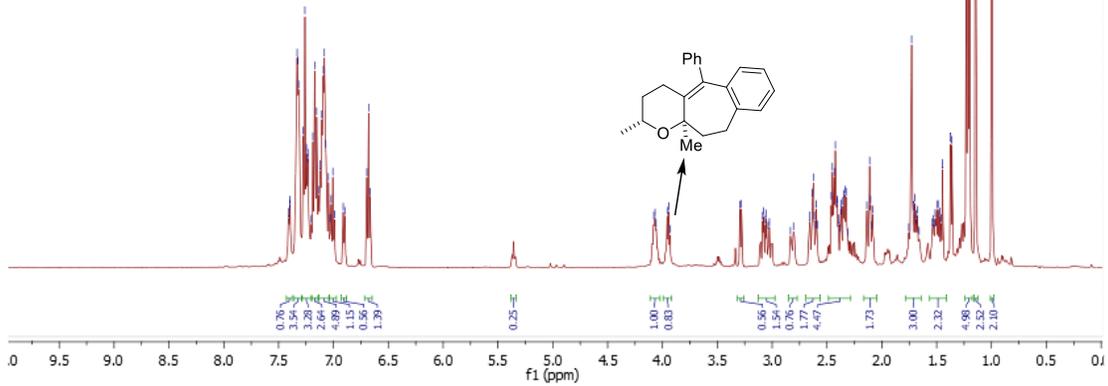


H-8-998A\_2  
 vance 500  
 1H NMR- h1\_latest  
 ur, 27 April 16  
 DCl3, 500.2, 25deg

500 MHz

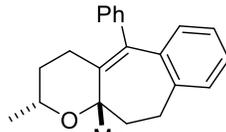


14i (1.2:1 dr)

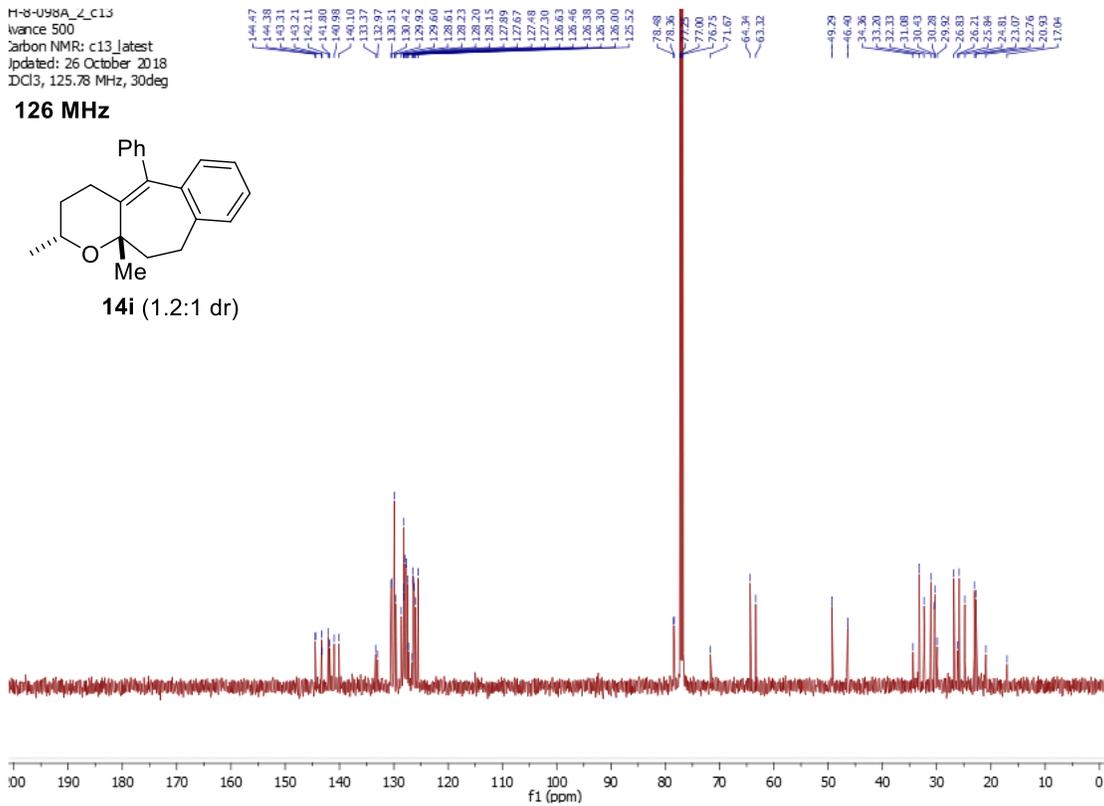


H-8-U98A\_2\_C13  
 vance 500  
 Carbon NMR: c13\_latest  
 Updated: 26 October 2018  
 DCl3, 125.78 MHz, 30deg

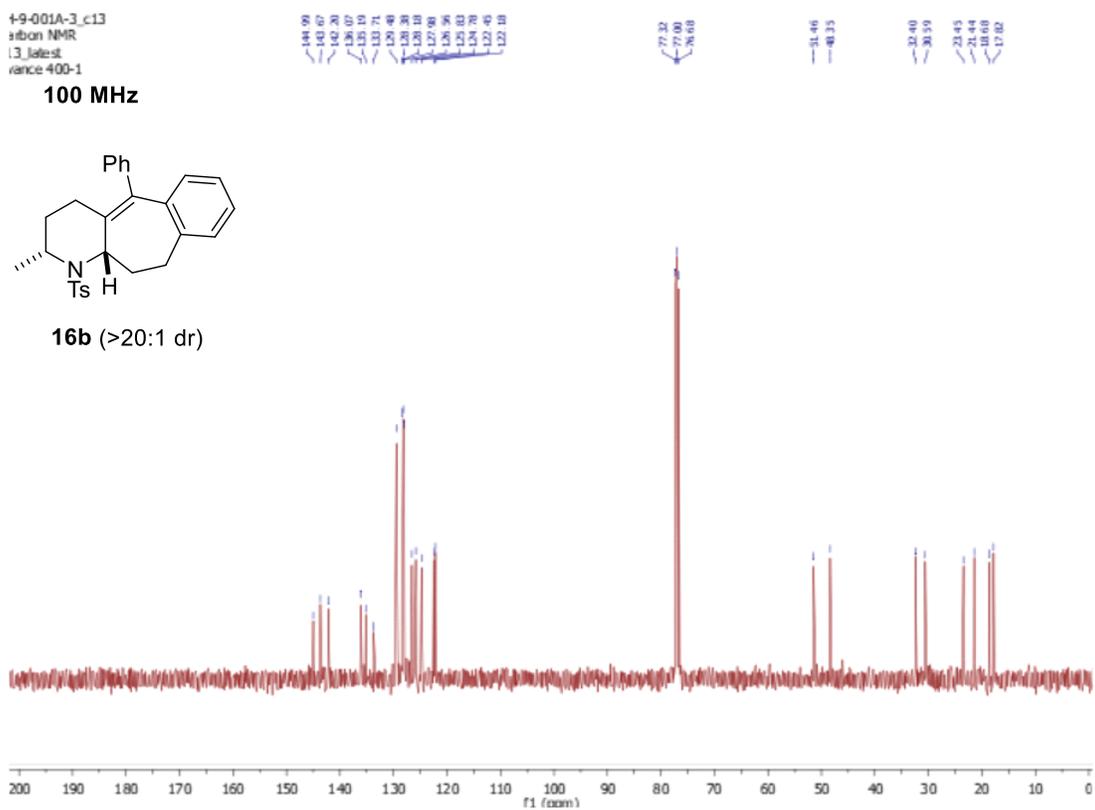
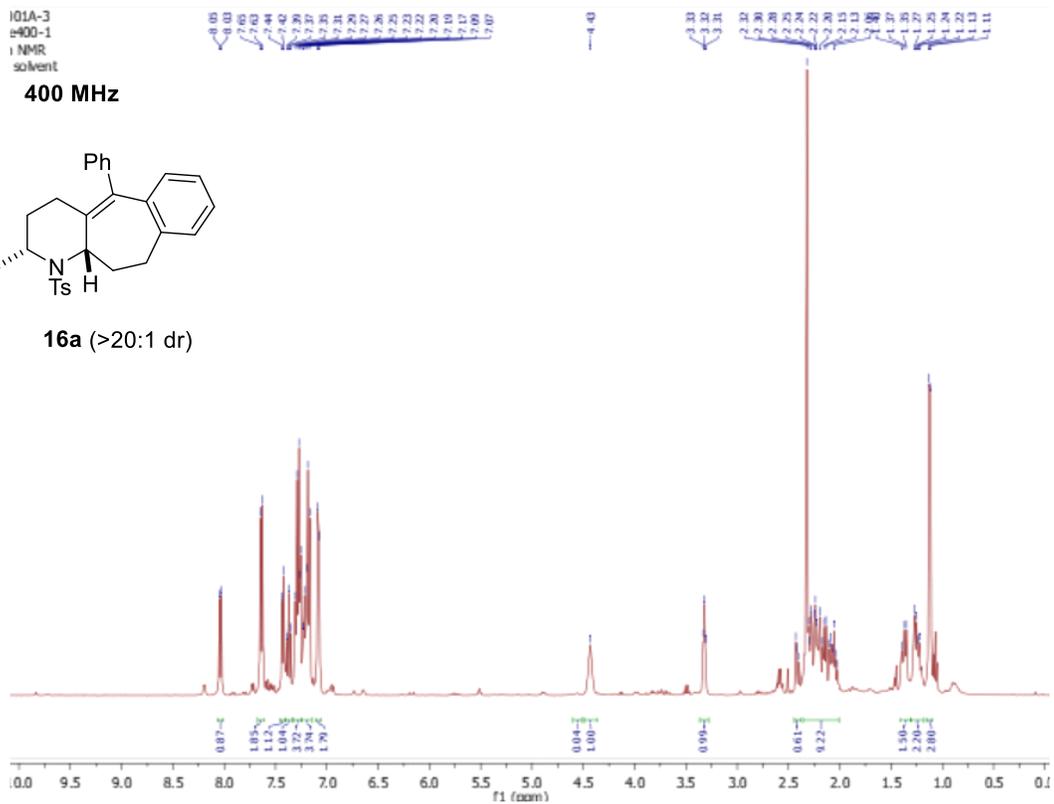
126 MHz



14i (1.2:1 dr)

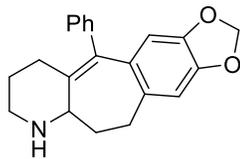




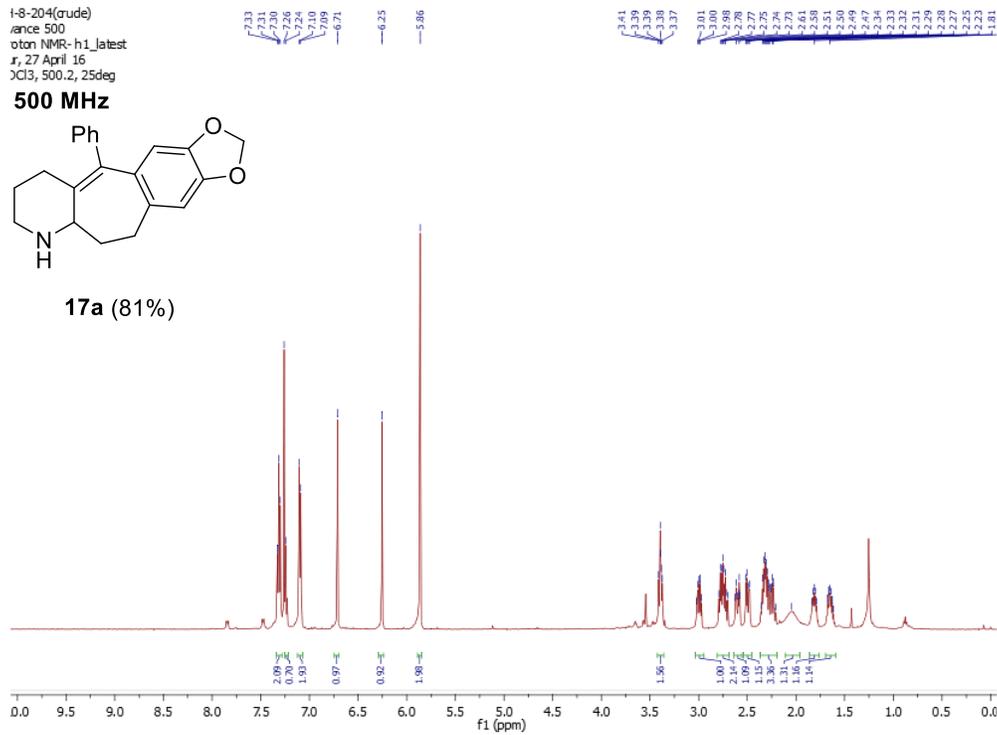


1-8-204(crude)  
 rance 500  
 on NMR- h1\_latest  
 r, 27 April 16  
 Cl3, 500.2, 25deg

**500 MHz**

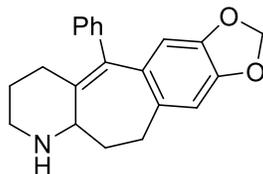


**17a (81%)**

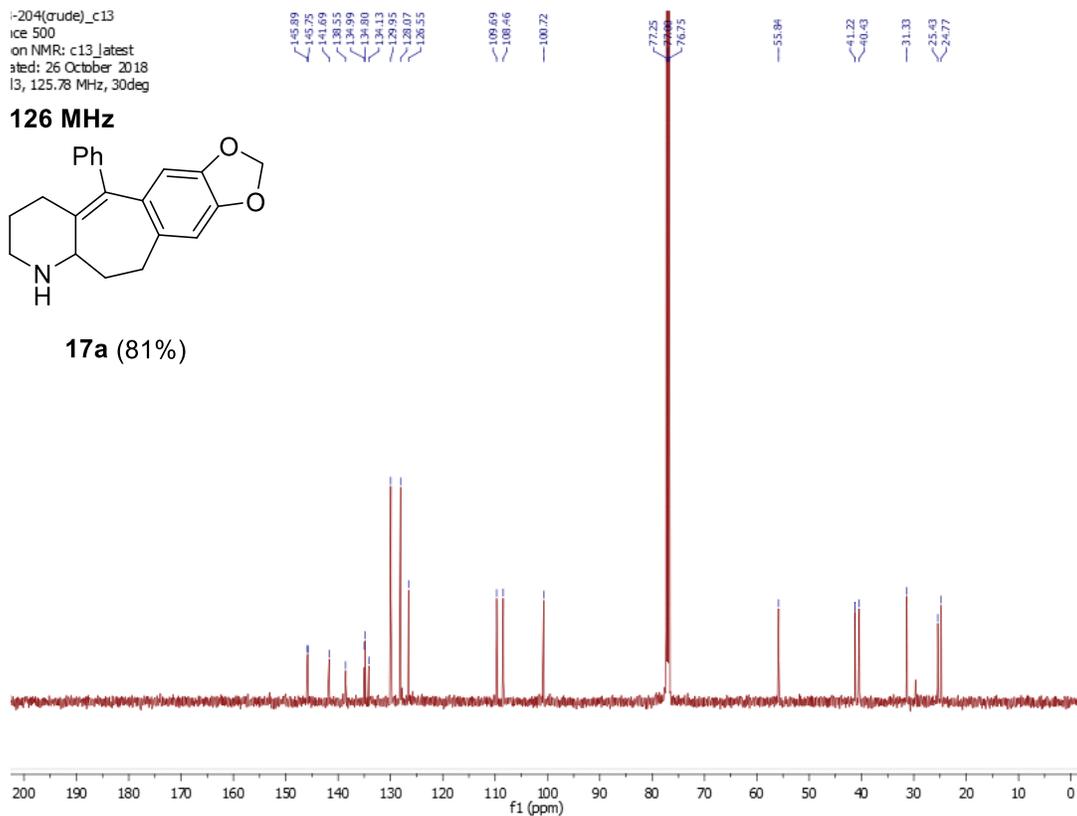


1-204(crude)\_c13  
 ice 500  
 on NMR: c13\_latest  
 sted: 26 October 2018  
 l3, 125.78 MHz, 30deg

**126 MHz**

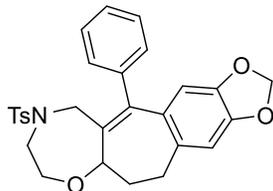


**17a (81%)**

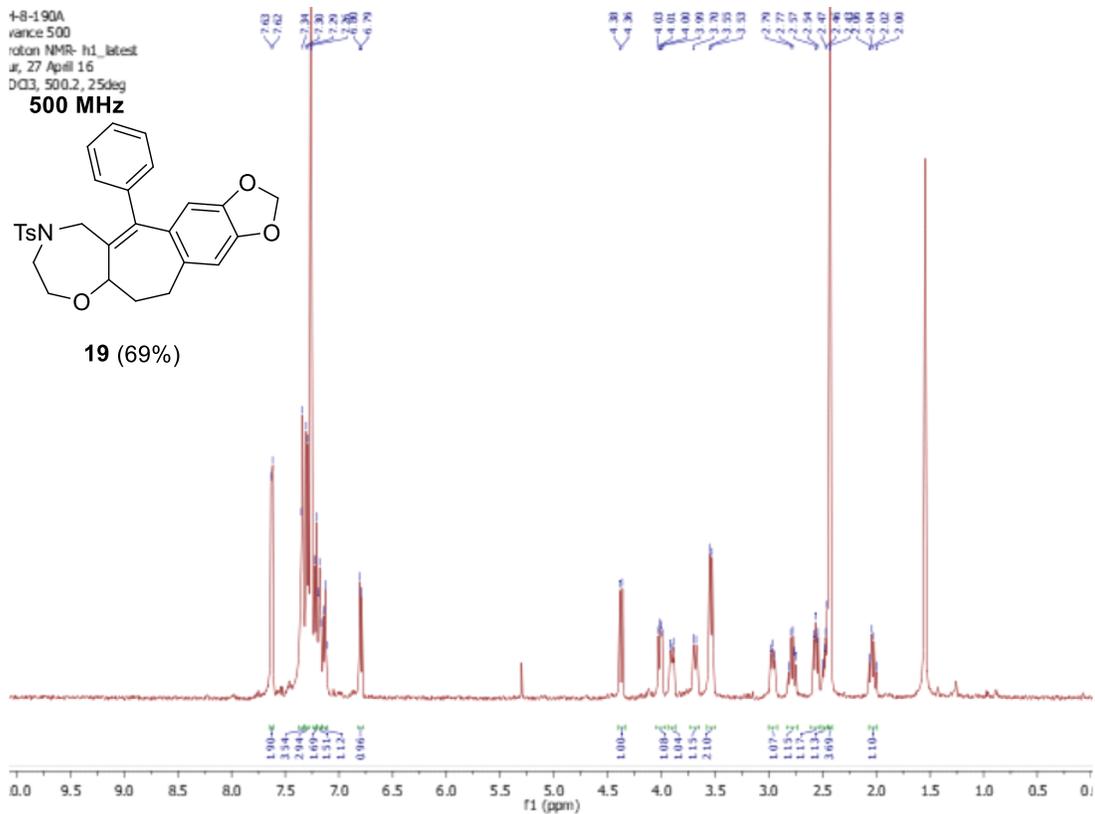


1-8-190A  
vance 500  
rotor NMR- h1\_latest  
at, 27 April 16  
DCl3, 500.2, 25deg

**500 MHz**

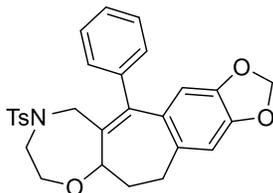


**19 (69%)**

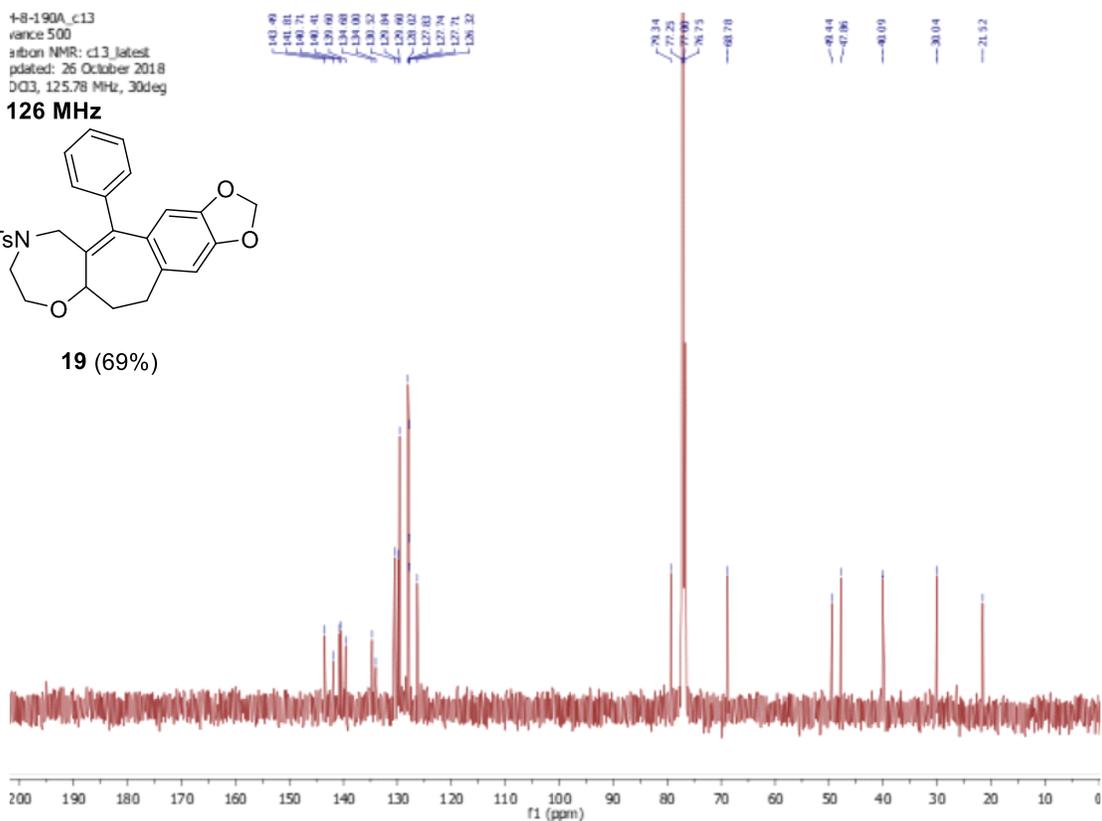


1-8-190A\_c13  
vance 500  
carbon NMR: c13\_latest  
dated: 26 October 2018  
DCl3, 125.78 MHz, 30deg

**126 MHz**

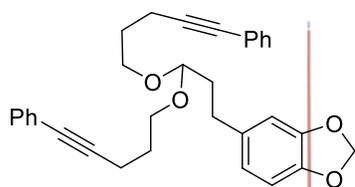


**19 (69%)**

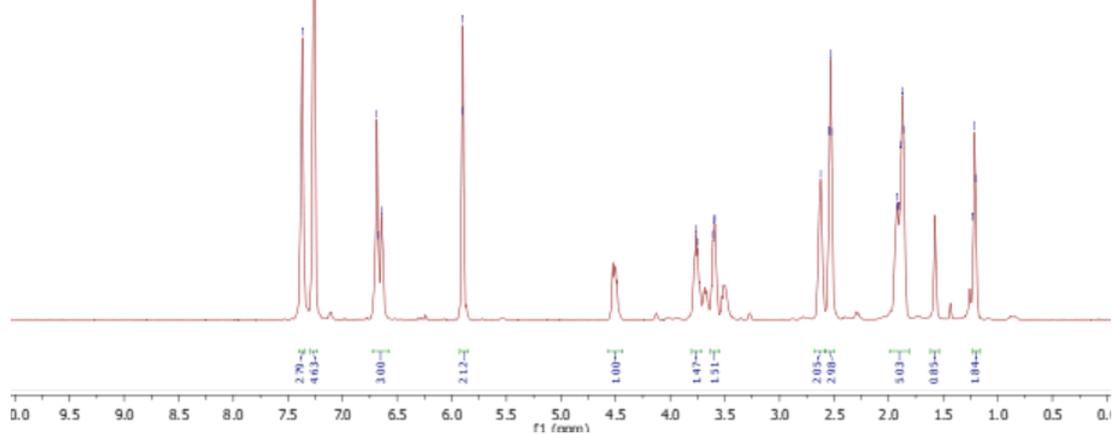


1H-9-044B  
Avance 500  
Proton NMR- h1\_test  
Sun, 27 April 16  
CDCl3, 500.2, 25deg

500 MHz

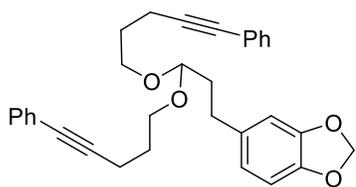


S14a

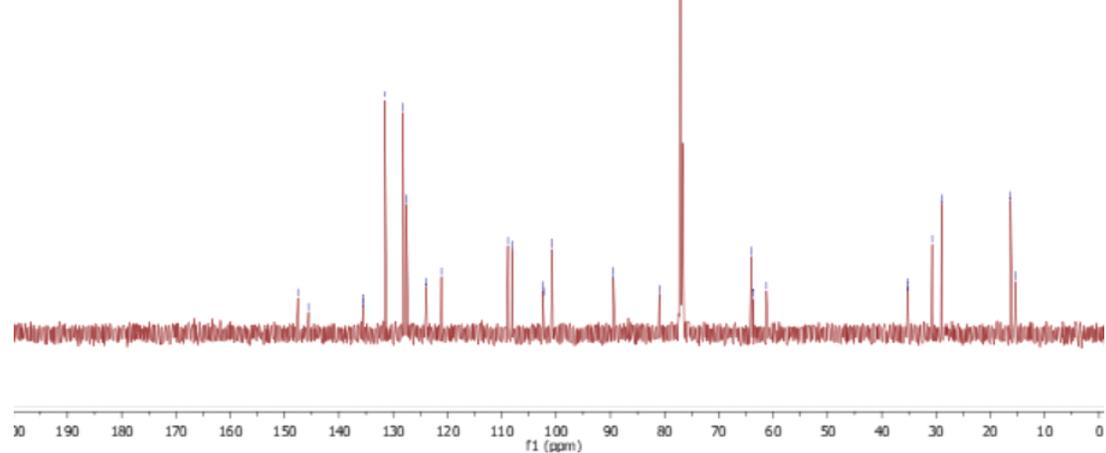


1H-9-044B\_c13  
Avance 500  
Carbon NMR- c13\_test  
dated: 26 October 2018  
CDCl3, 125.78 MHz, 30deg

126 MHz

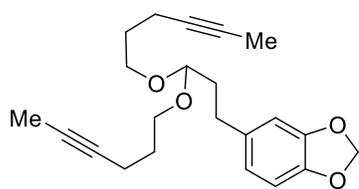


S14a

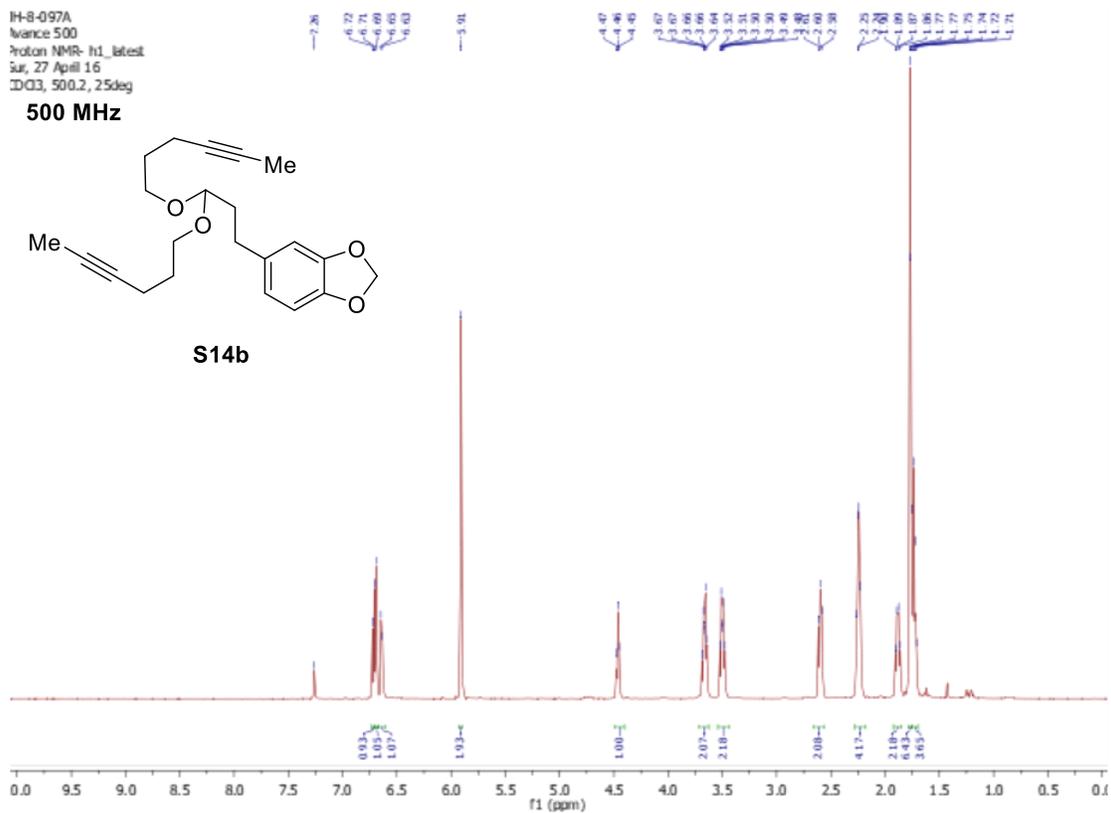


JH-8-097A  
 Avance 500  
 Proton NMR- h1\_latest  
 sur, 27 April 16  
 CDCl3, 500.2, 25deg

500 MHz

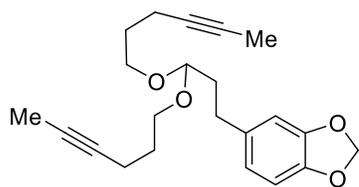


S14b

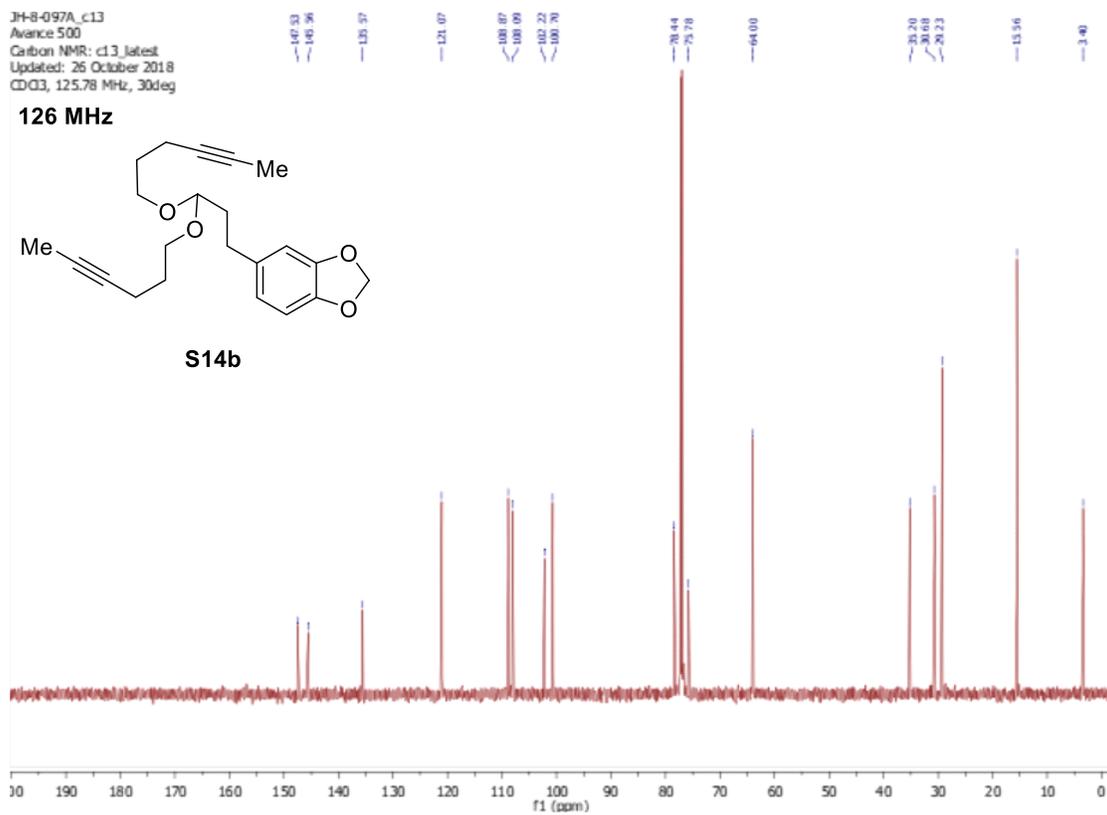


JH-8-097A\_c13  
 Avance 500  
 Carbon NMR: c13\_latest  
 Updated: 26 October 2018  
 CDCl3, 125.78 MHz, 30deg

126 MHz



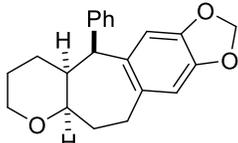
S14b



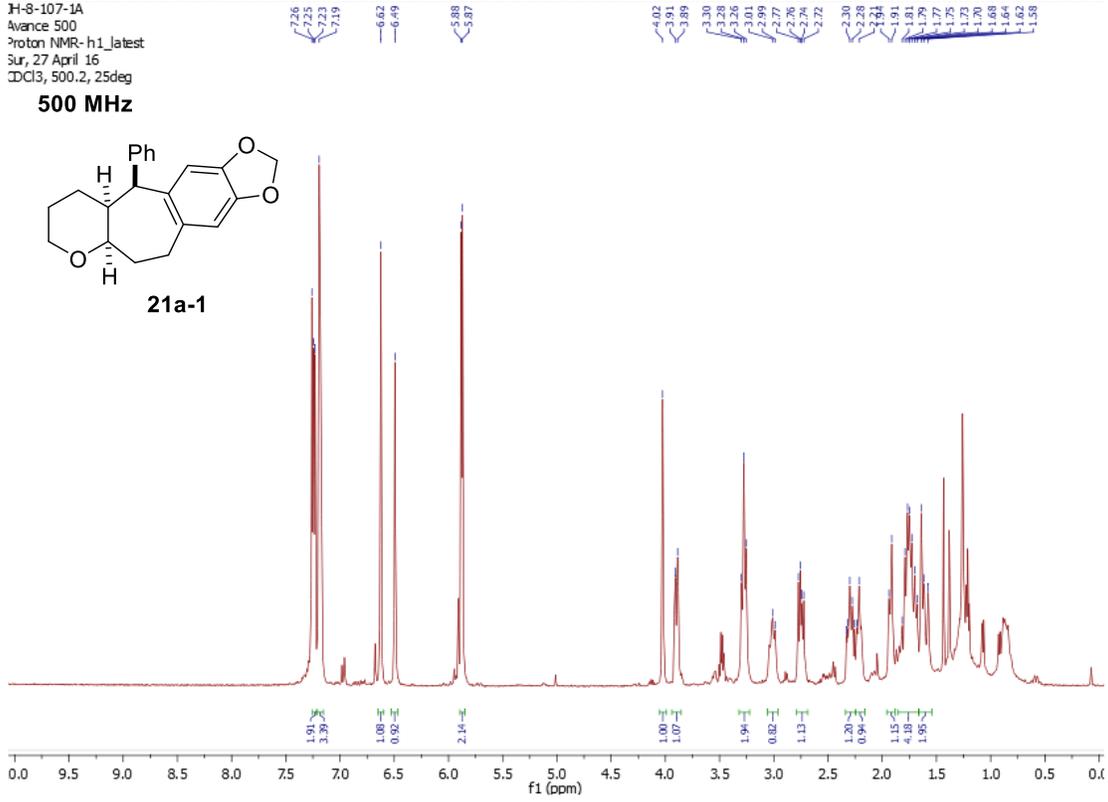


H-8-107-1A  
 Avance 500  
 Proton NMR- h1\_latest  
 Sat, 27 April 16  
 CDCl<sub>3</sub>, 500.2, 25deg

500 MHz

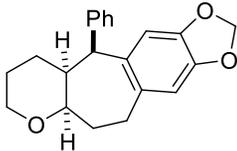


21a-1

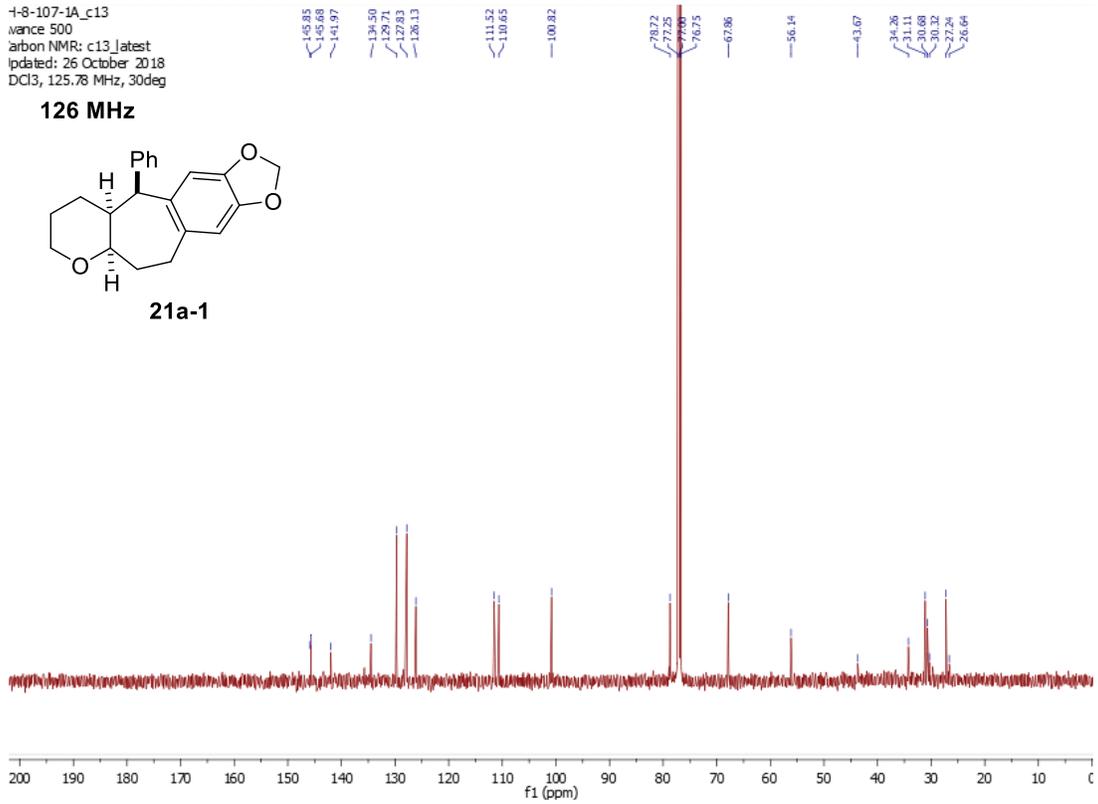


H-8-107-1A\_c13  
 Avance 500  
 Carbon NMR: c13\_latest  
 Updated: 26 October 2018  
 CDCl<sub>3</sub>, 125.78 MHz, 30deg

126 MHz

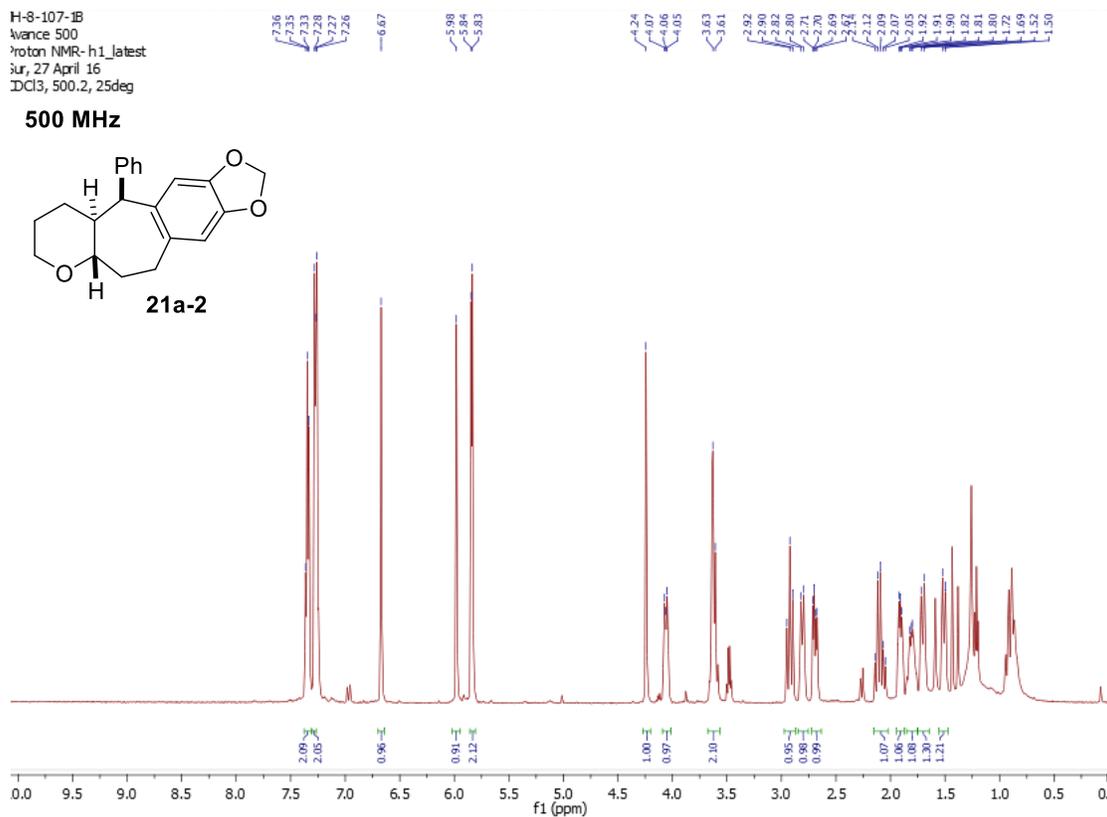
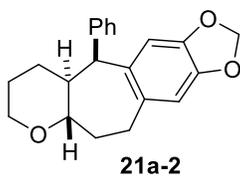


21a-1



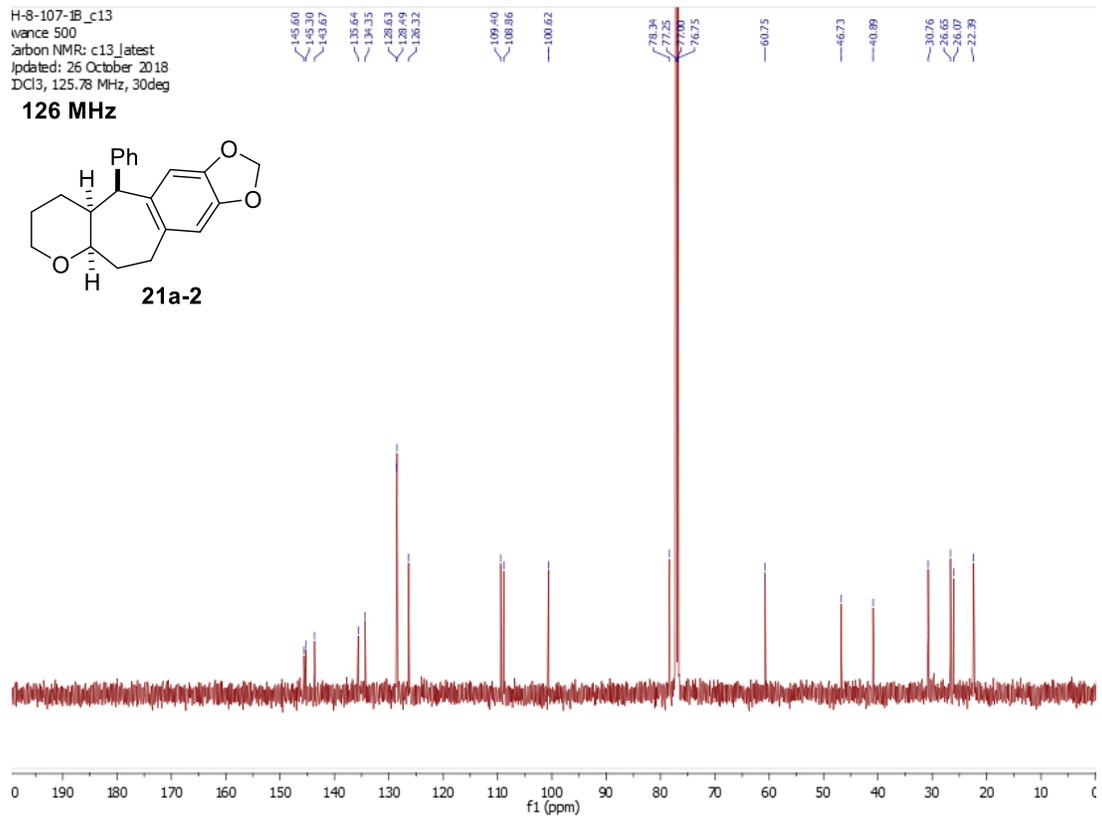
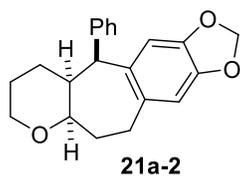
H-8-107-1B  
 Avance 500  
 Proton NMR- h1\_latest  
 sur, 27 April 16  
 DCl3, 500.2, 25deg

500 MHz



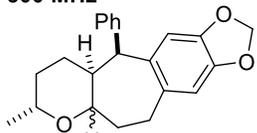
H-8-107-1B\_c13  
 Avance 500  
 Carbon NMR: c13\_latest  
 Updated: 26 October 2018  
 DCl3, 125.78 MHz, 30deg

126 MHz

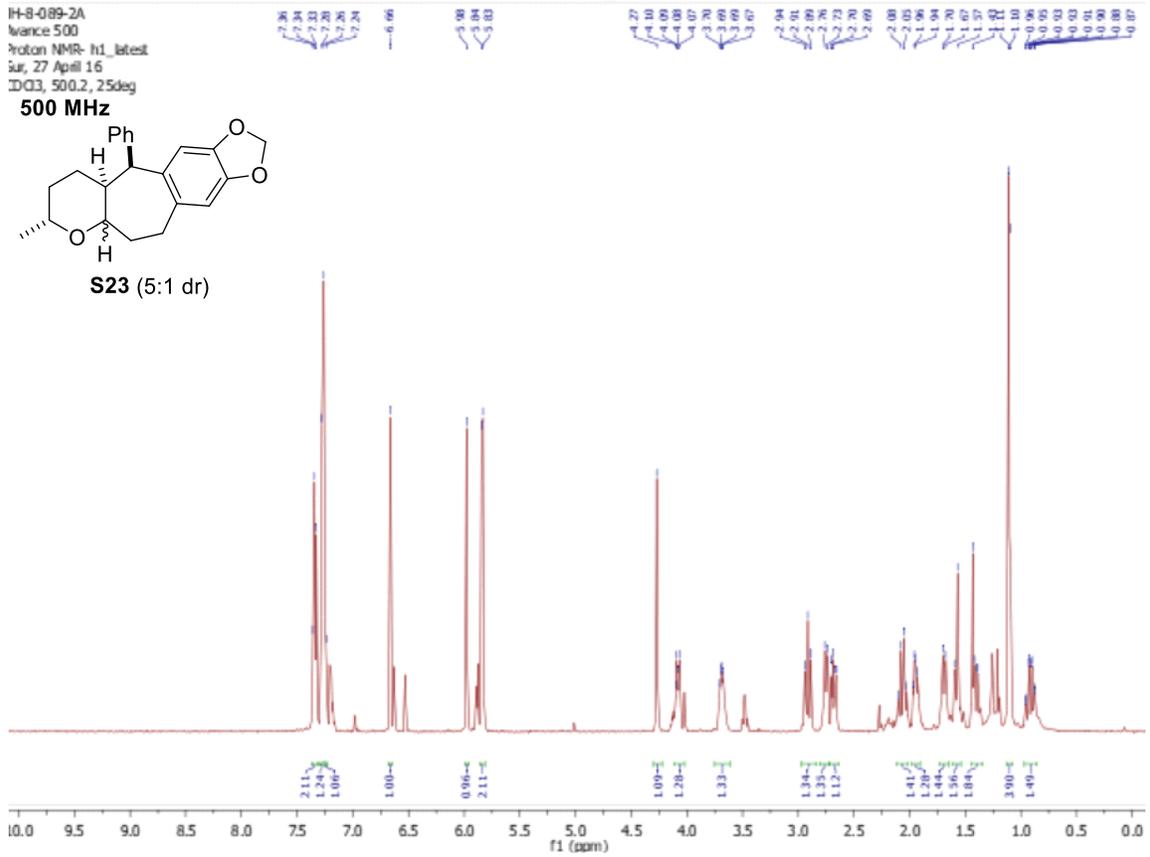


JH-8-089-2A  
Avance 500  
Proton NMR- f1\_1atest  
Sur, 27 April 16  
CDCl<sub>3</sub>, 500.2, 2.5deg

500 MHz

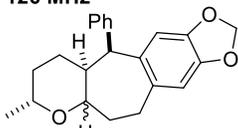


S23 (5:1 dr)

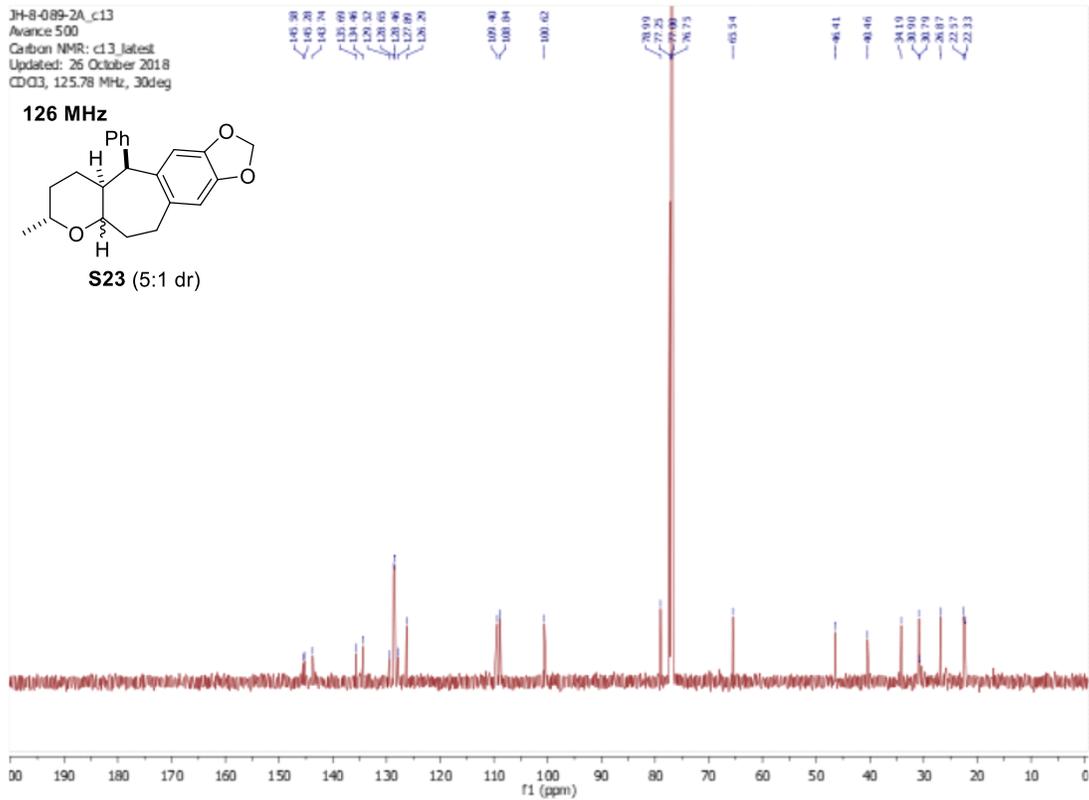


JH-8-089-2A\_c13  
Avance 500  
Carbon NMR- c13\_1atest  
Updated: 26 October 2018  
CDCl<sub>3</sub>, 125.78 MHz, 30deg

126 MHz

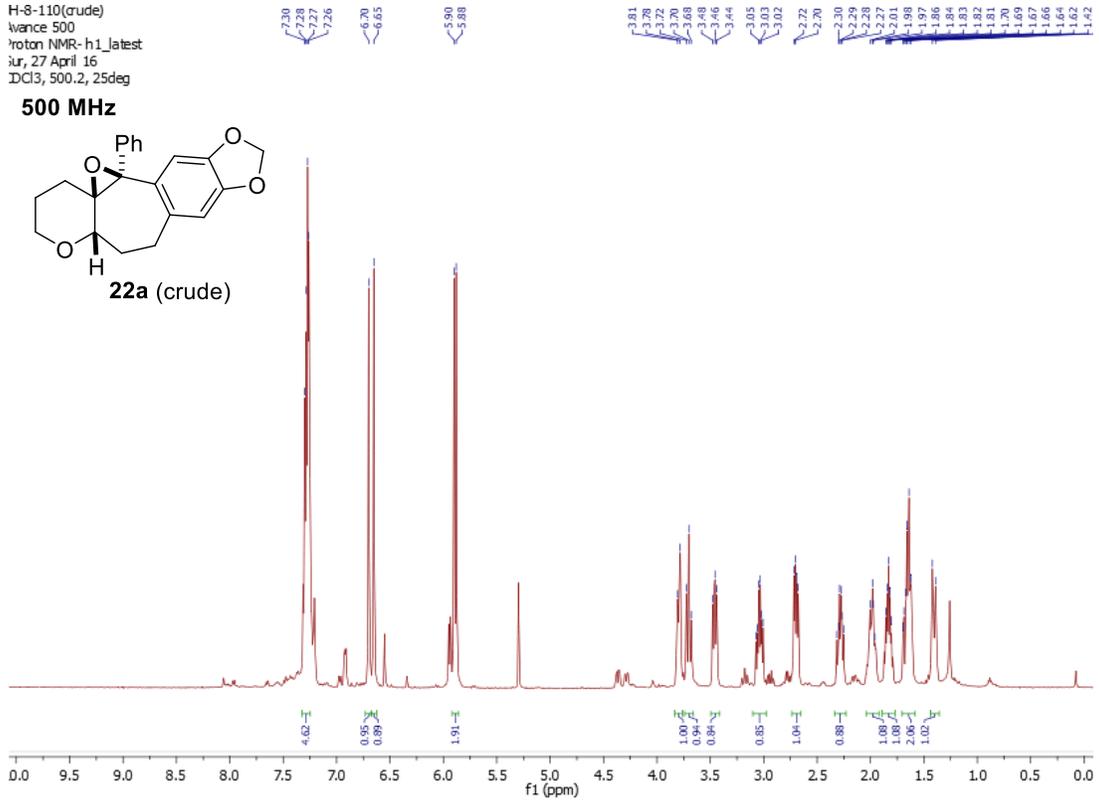
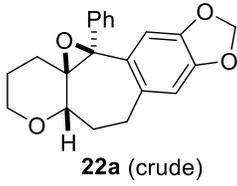


S23 (5:1 dr)



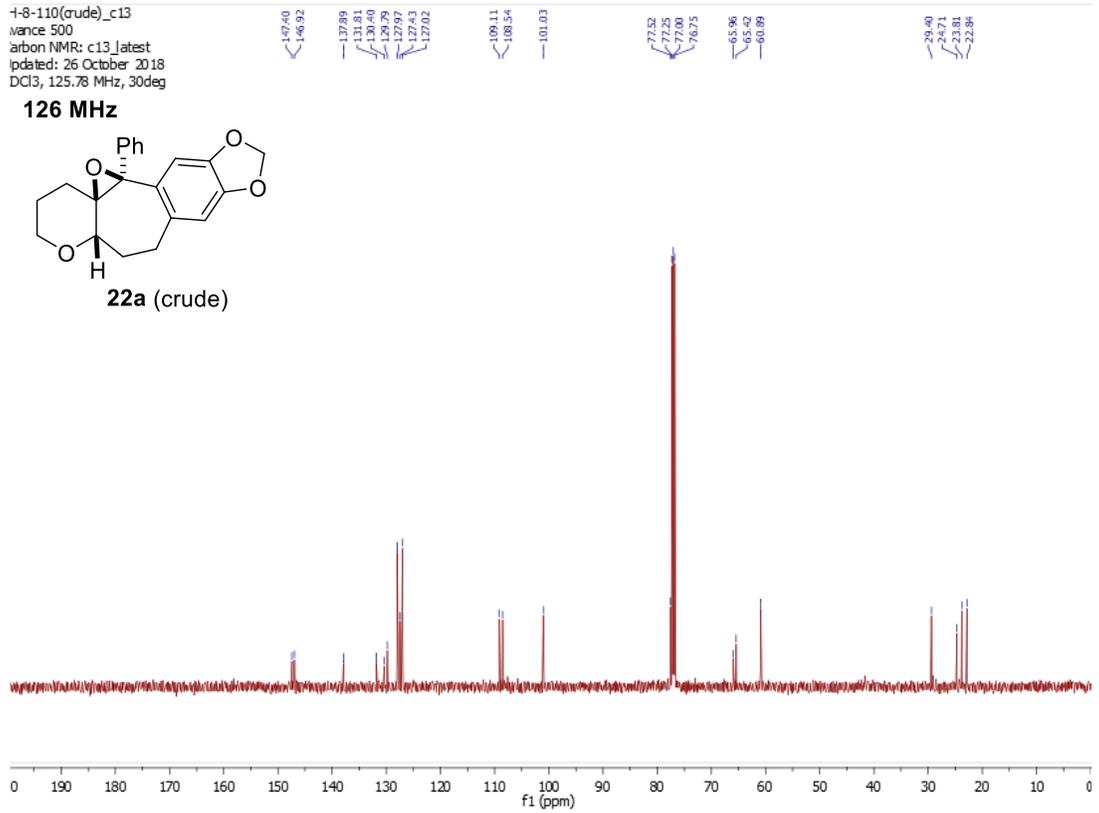
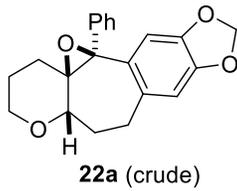
t-8-110(crude)  
 vance 500  
 proton NMR- h1\_latest  
 ur, 27 April 16  
 DCl3, 500.2, 25deg

500 MHz



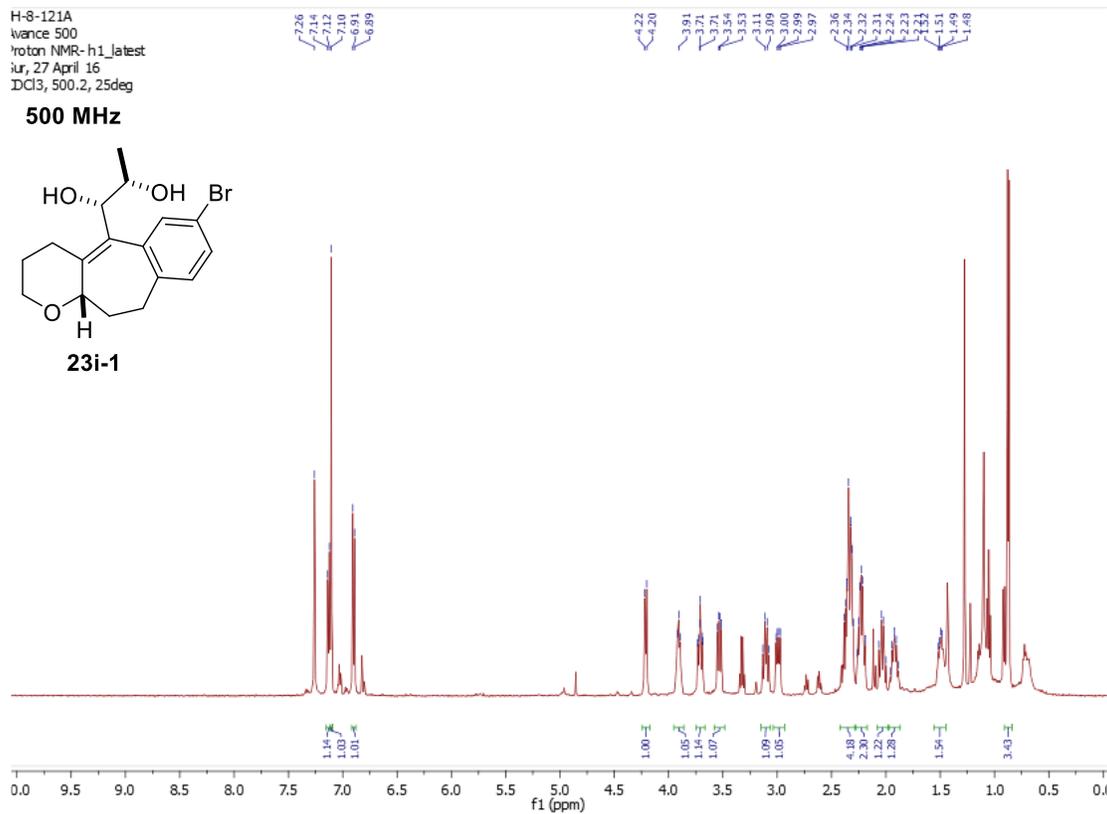
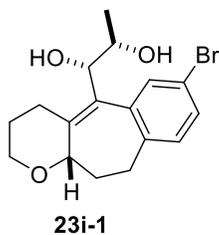
t-8-110(crude)\_c13  
 vance 500  
 carbon NMR: c13\_latest  
 pdated: 26 October 2018  
 DCl3, 125.78 MHz, 30deg

126 MHz



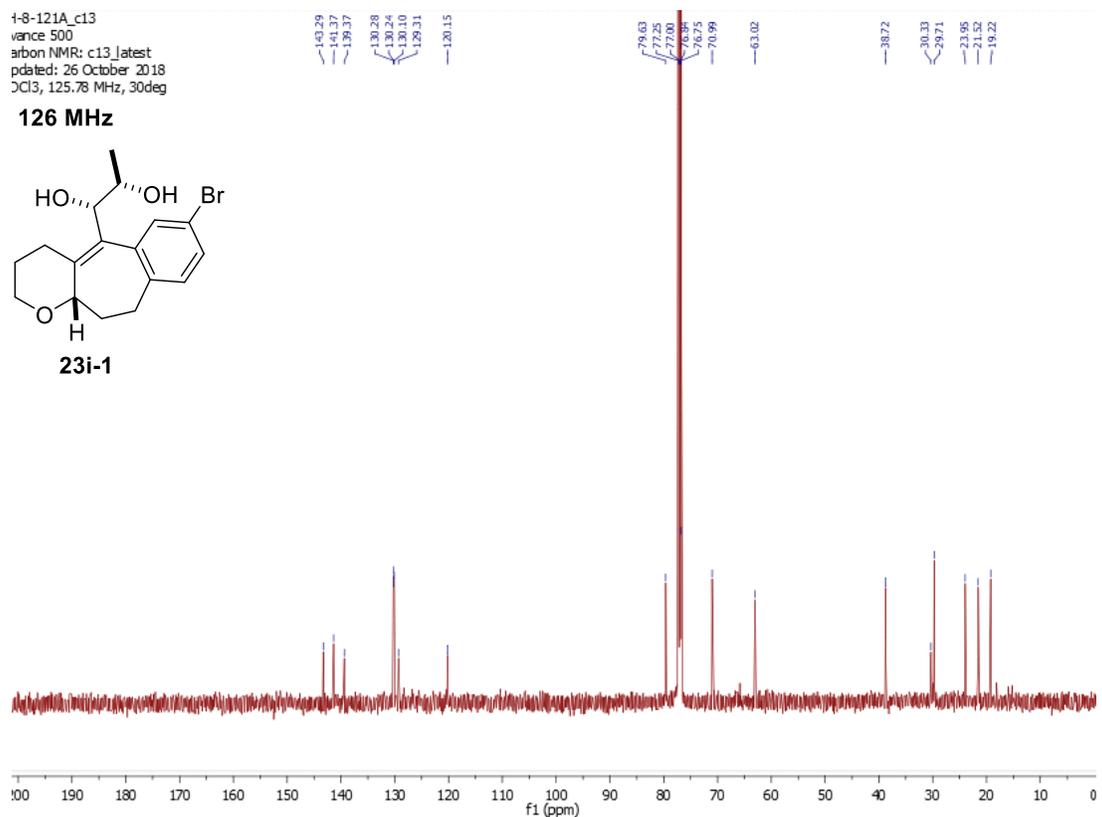
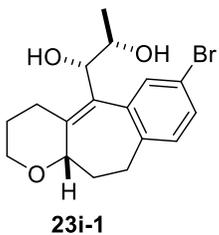
H-8-121A  
 vance 500  
 1H NMR- h1\_latest  
 sur, 27 April 16  
 DCl3, 500.2, 25deg

500 MHz



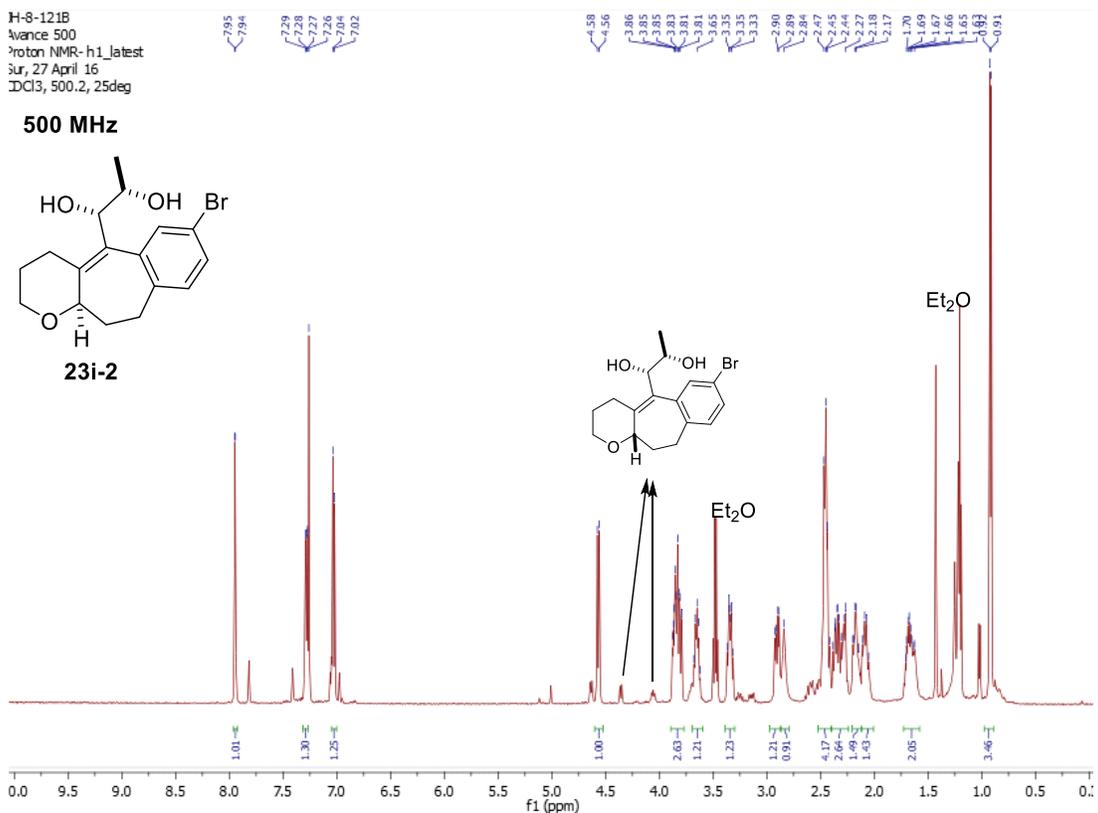
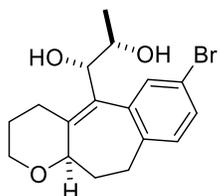
t-8-121A\_c13  
 vance 500  
 13C NMR: c13\_latest  
 dated: 26 October 2018  
 DCl3, 125.78 MHz, 30deg

126 MHz



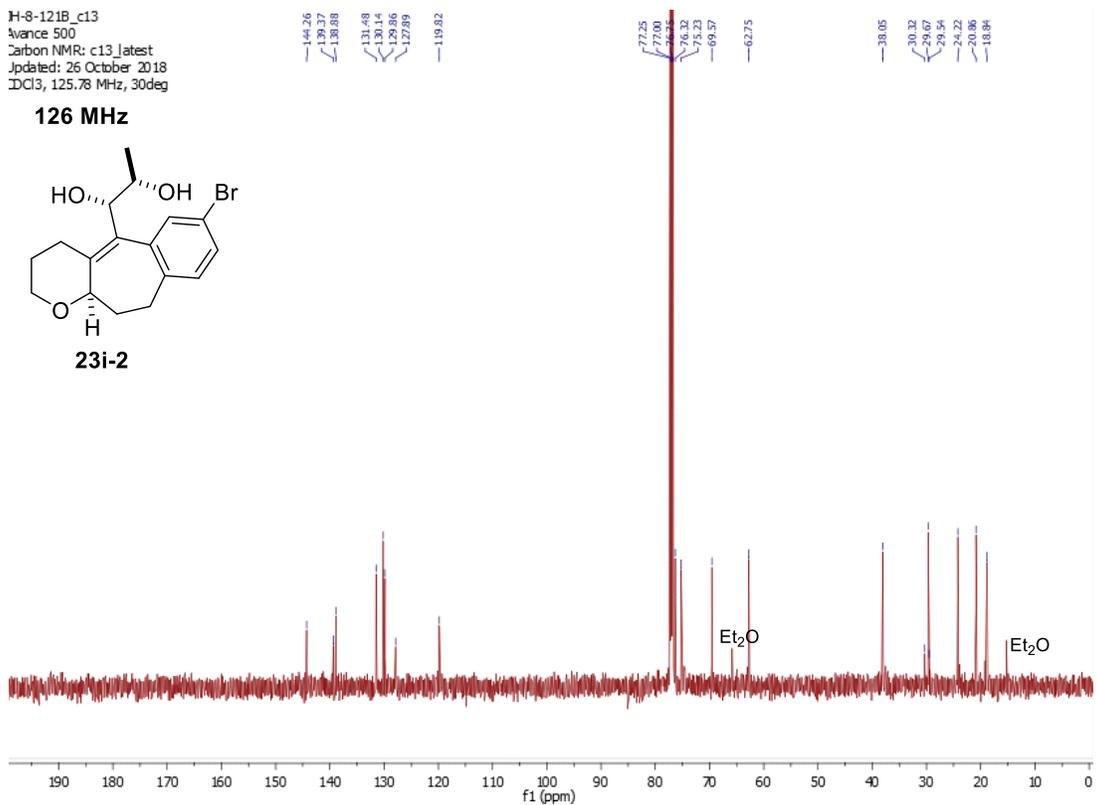
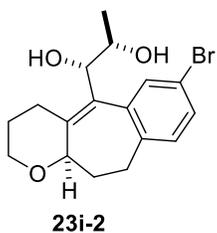
H-8-121B  
 Avance 500  
 Proton NMR: h1\_latest  
 Sur, 27 April 16  
 CDCl<sub>3</sub>, 500.2, 25deg

500 MHz



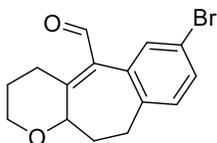
H-8-121B\_c13  
 Avance 500  
 Carbon NMR: c13\_latest  
 Jdated: 26 October 2018  
 CDCl<sub>3</sub>, 125.78 MHz, 30deg

126 MHz

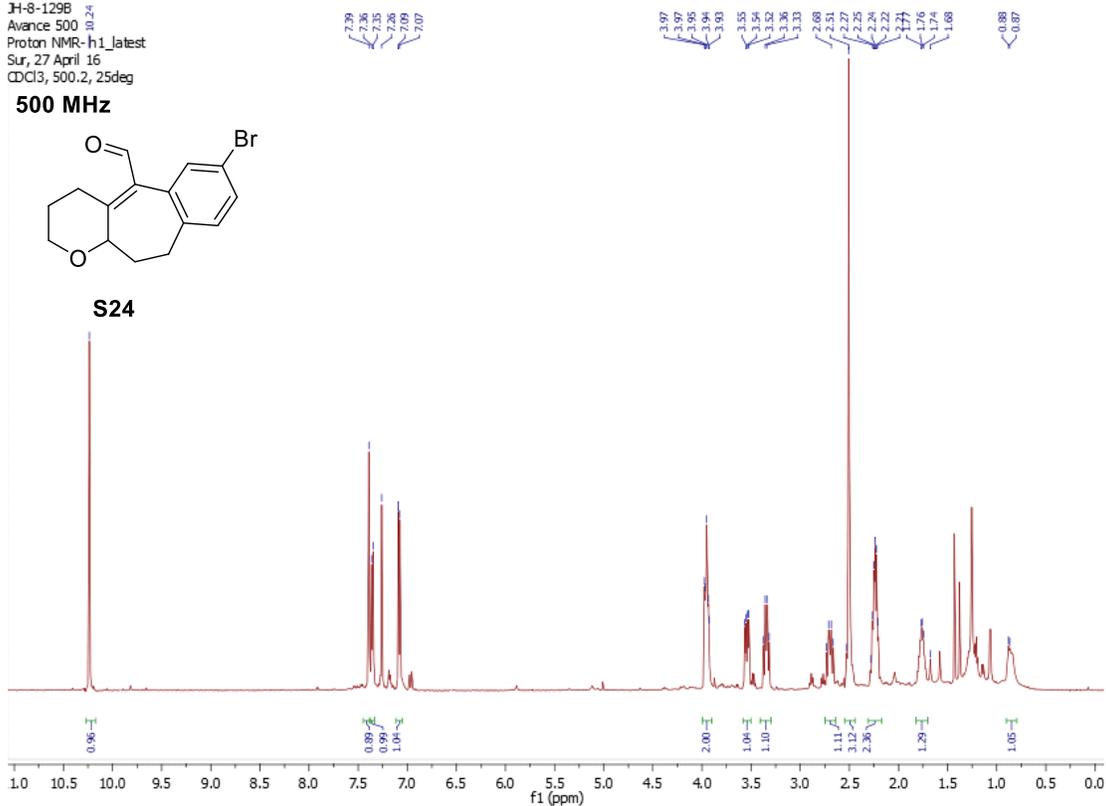


H-8-129B  
Avance 500  
Proton NMR- h1\_latest  
Sur, 27 April 16  
CDCl3, 500.2, 25deg

500 MHz

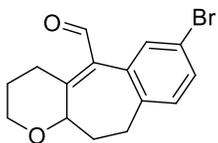


S24

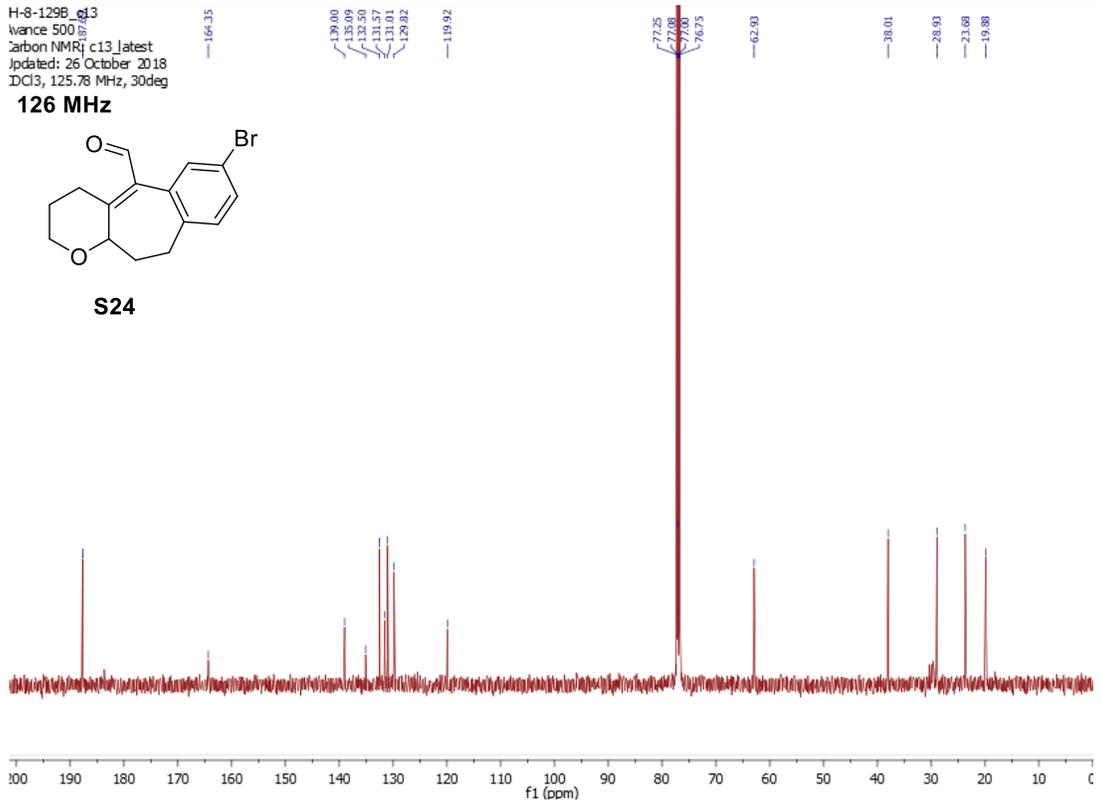


H-8-129B\_c13  
Avance 500  
Carbon NMR c13\_latest  
Jdated: 26 October 2018  
CDCl3, 125.76 MHz, 30deg

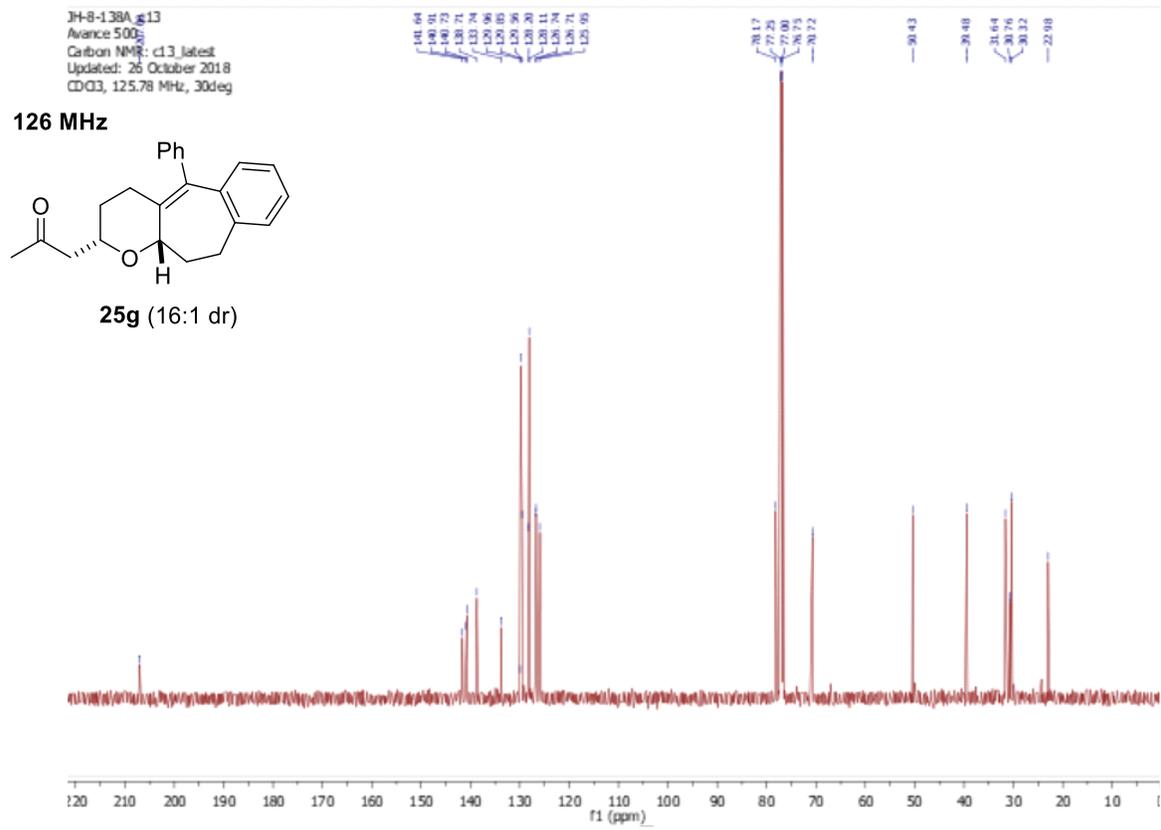
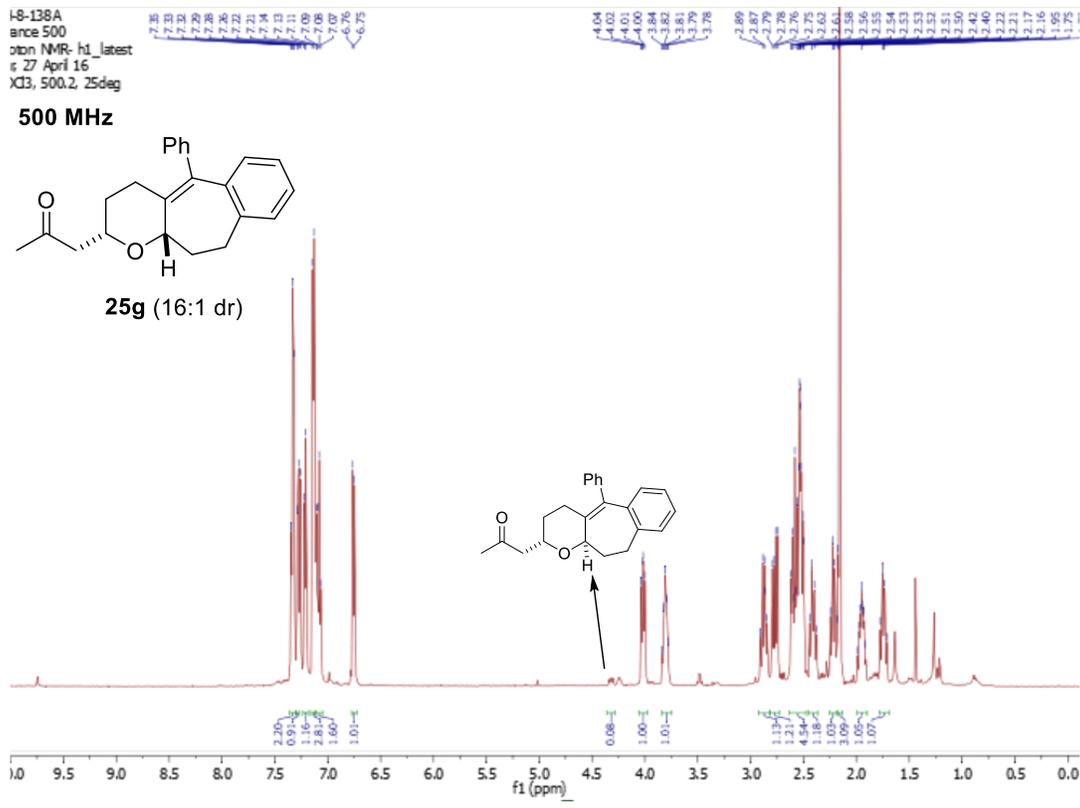
126 MHz

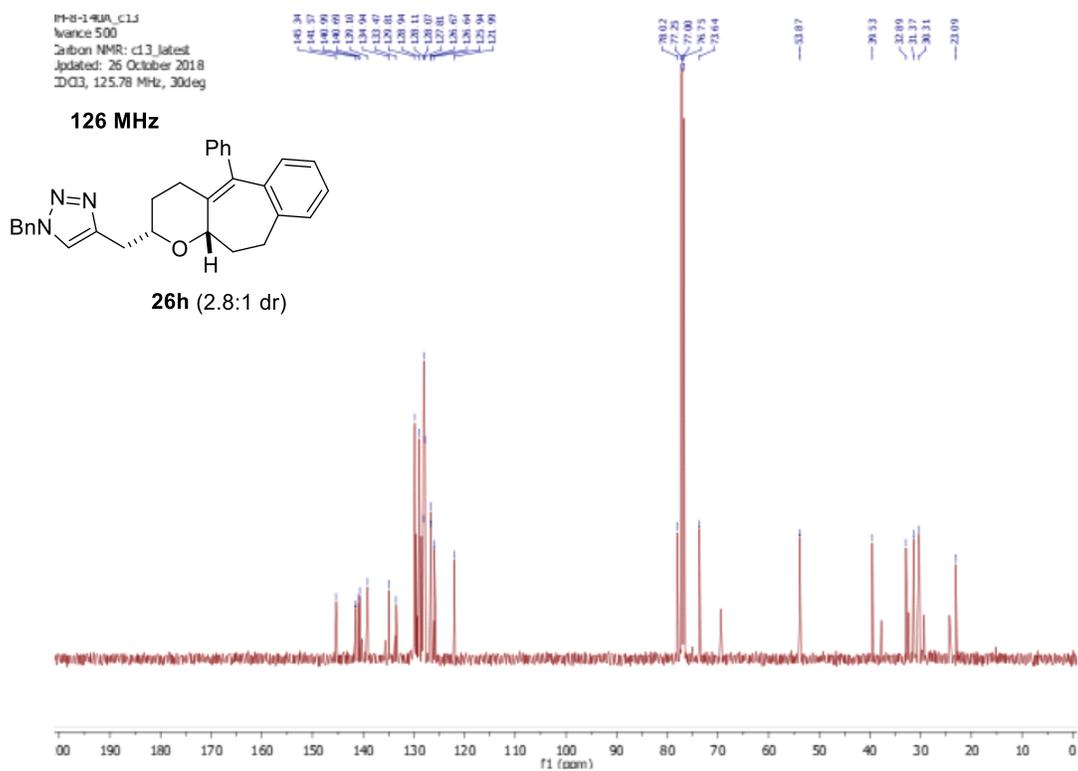
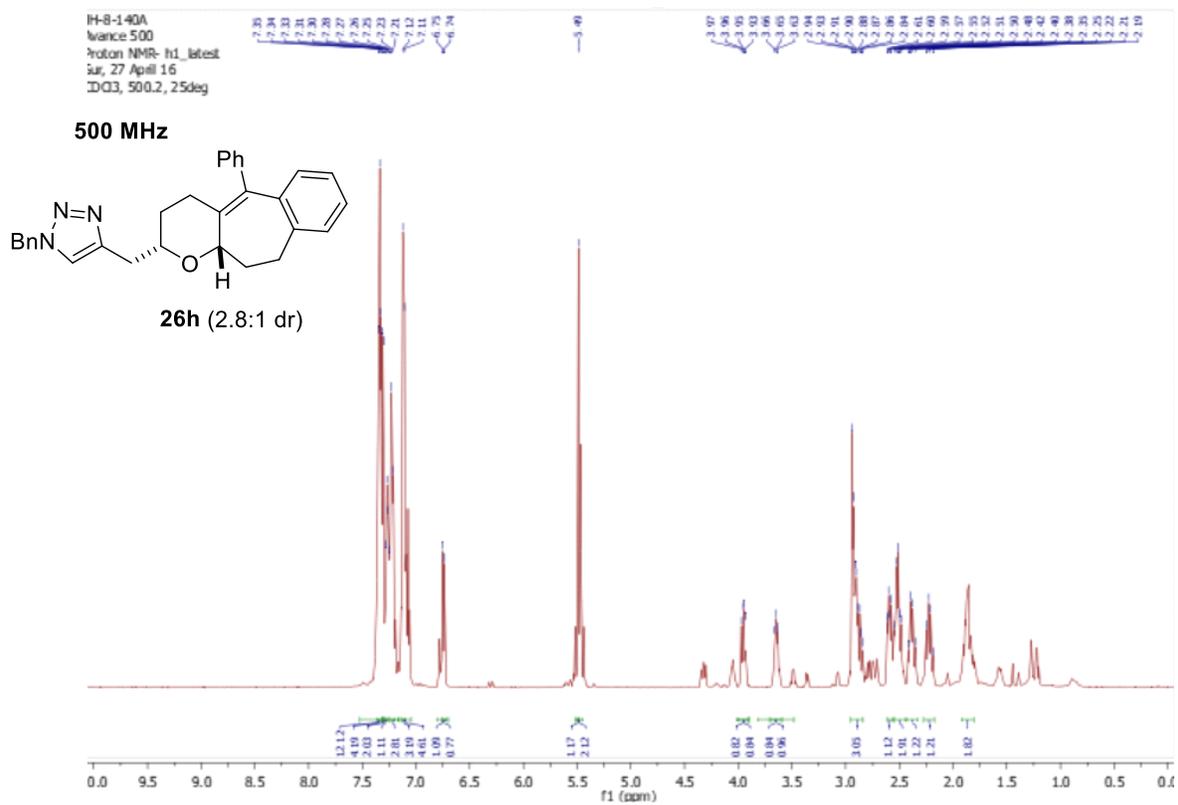


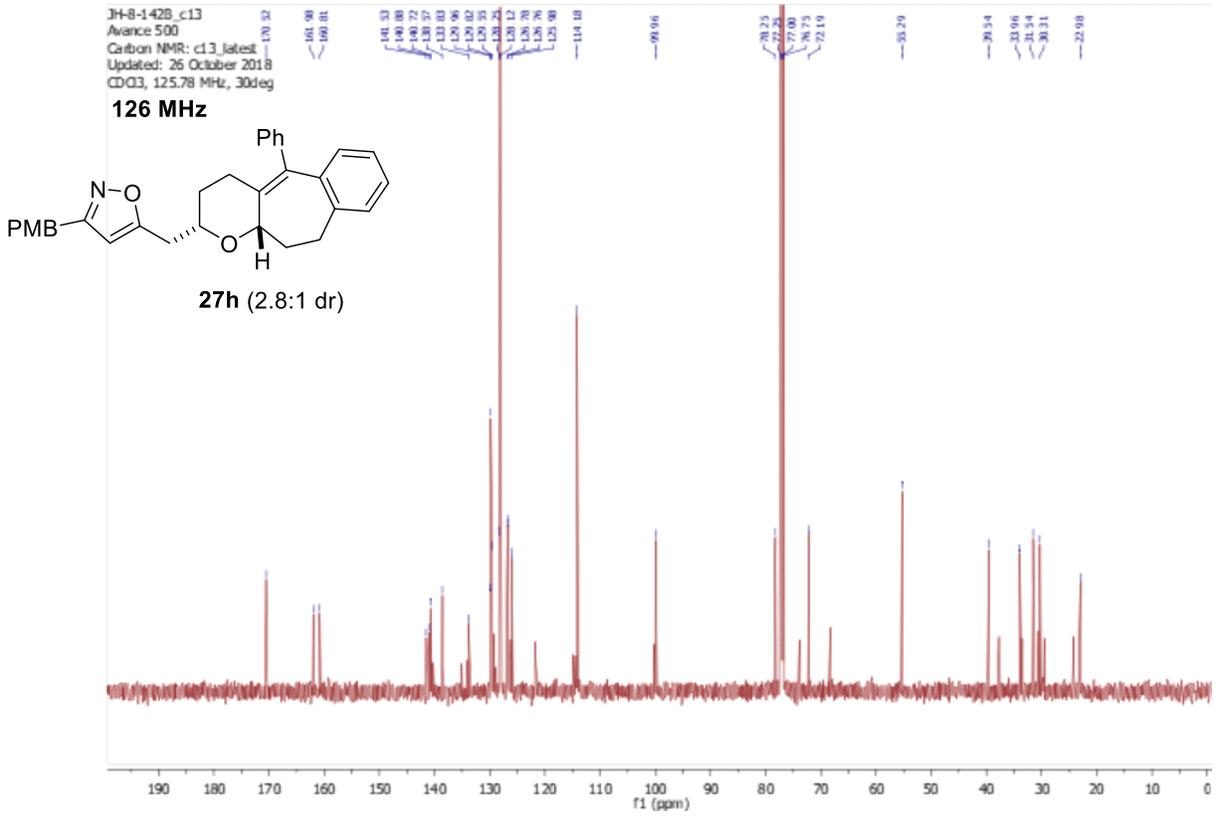
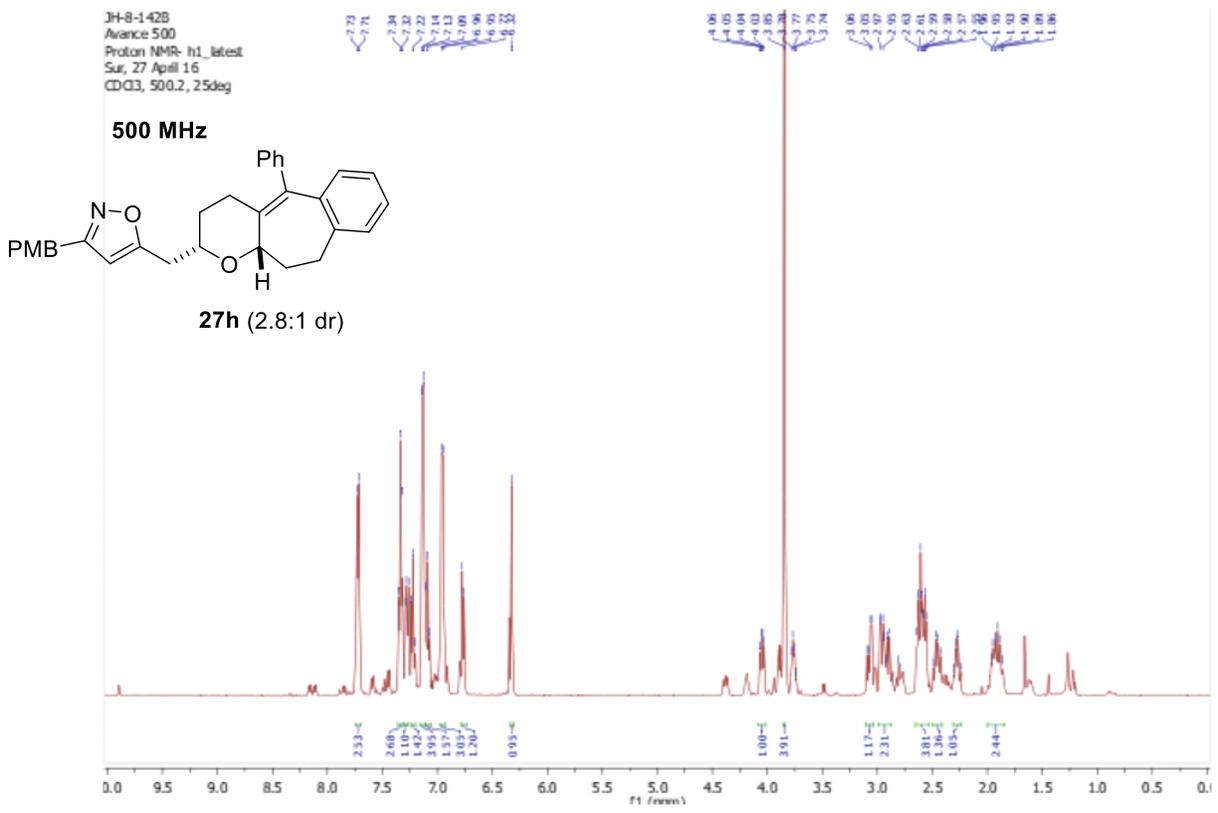
S24





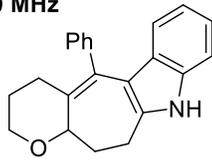




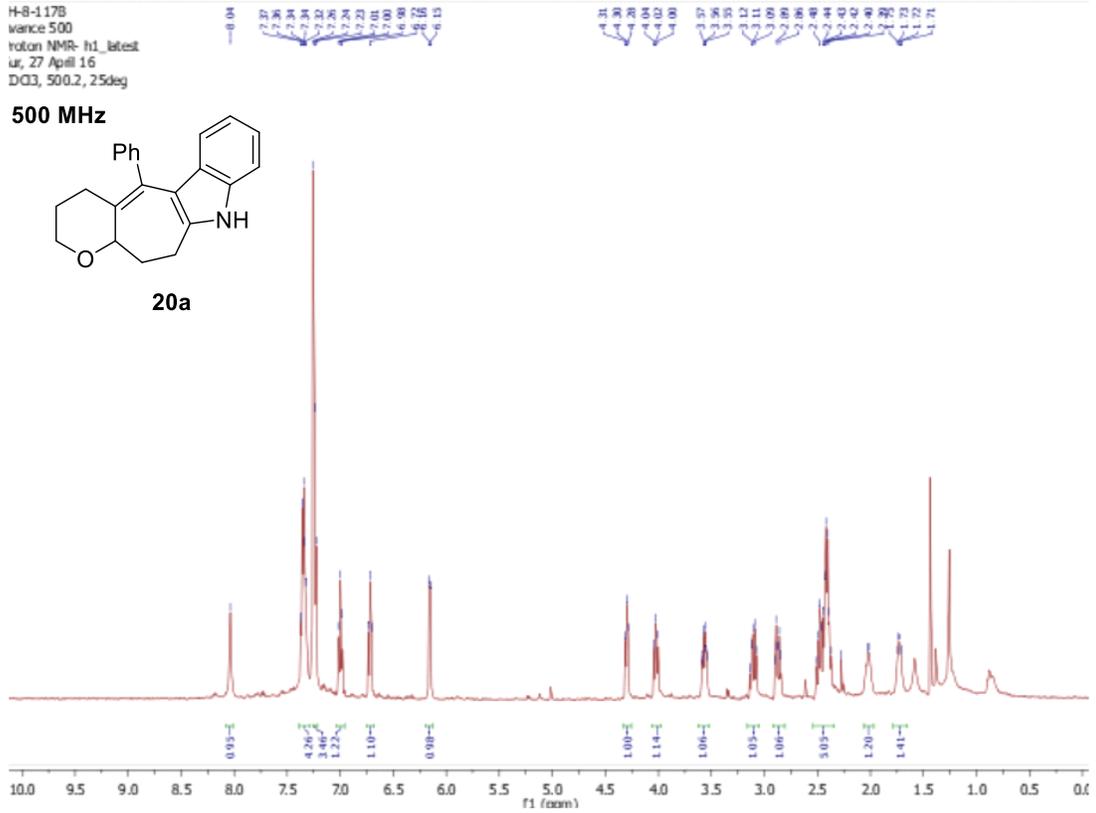


H-8-117B  
vance 500  
rotan NMR- h1\_latest  
ur, 27 April 16  
DQ3, 500.2, 2.5deg

500 MHz

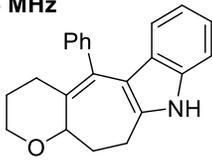


20a

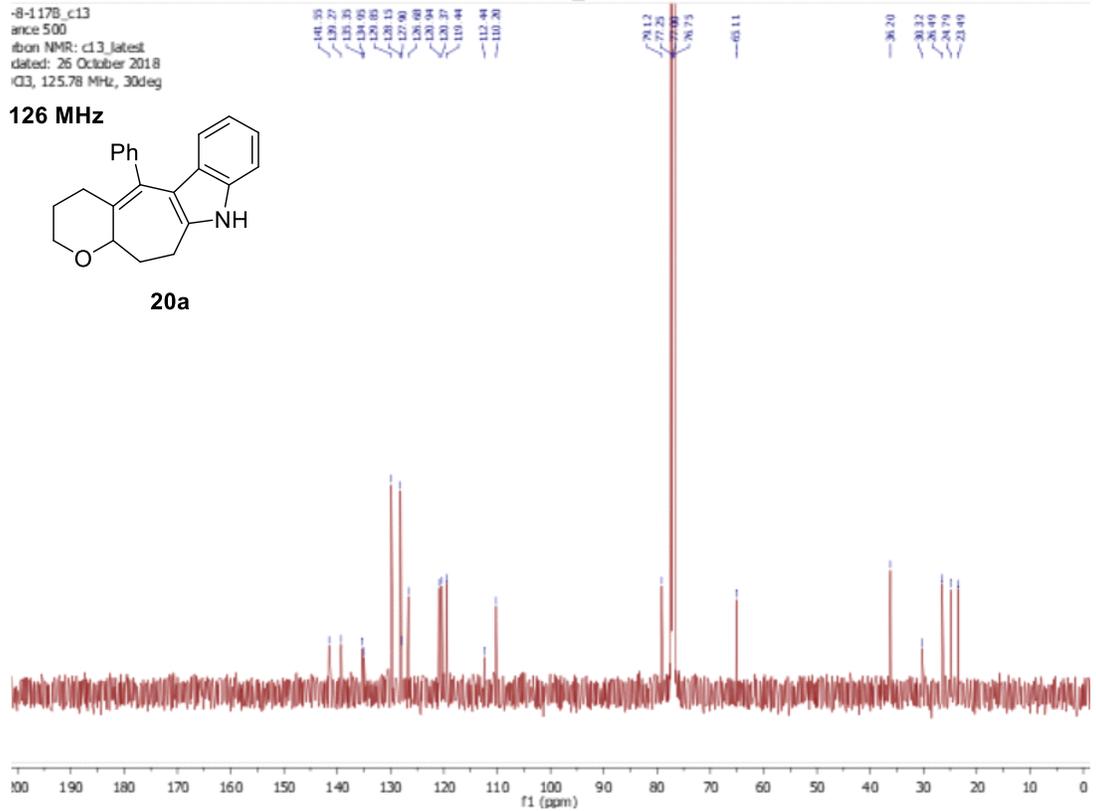


-8-117B\_c13  
ance 500  
rbon NMR: c13\_latest  
dated: 26 October 2018  
Q3, 125.78 MHz, 30deg

126 MHz

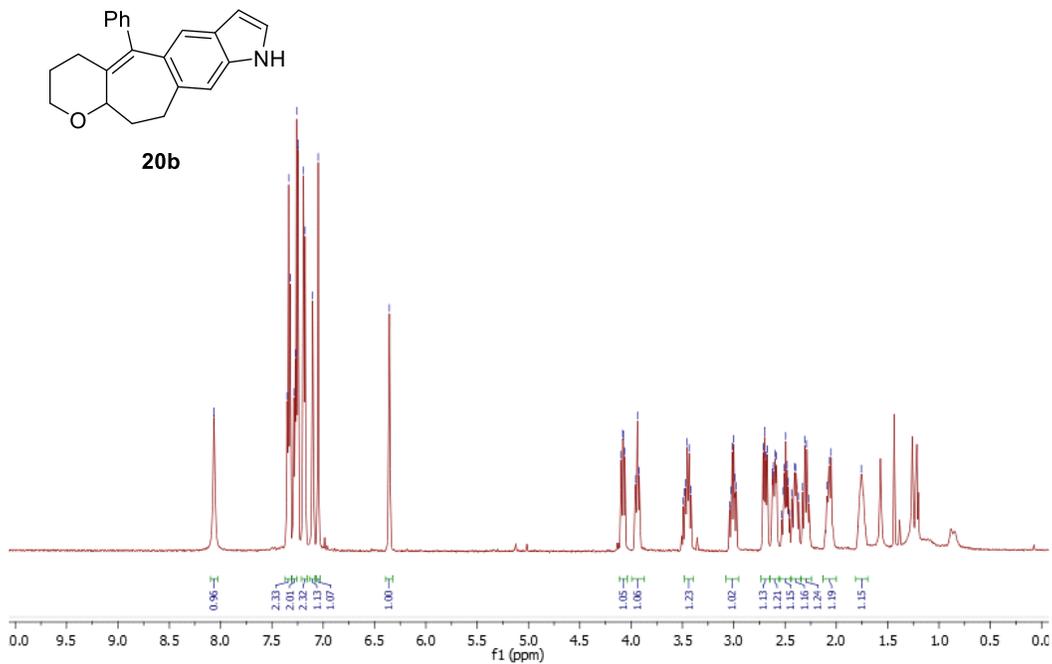


20a



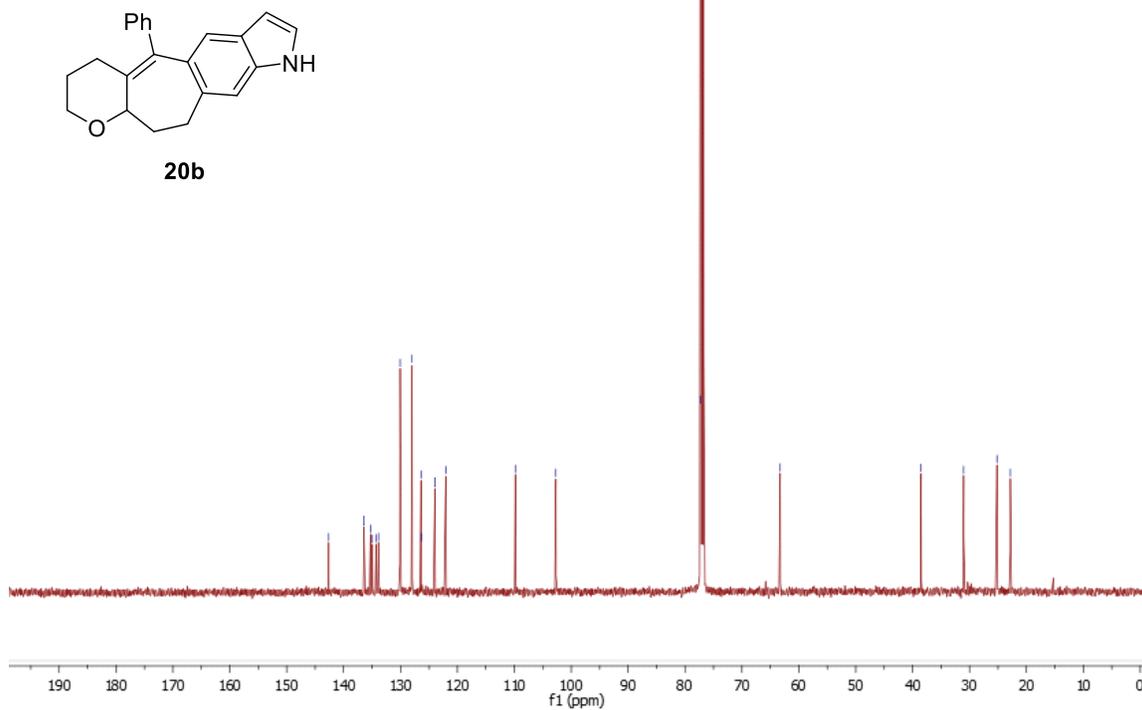
116A  
e 500  
n NMR- h1\_latest  
7 April 16  
3, 500.2, 25deg

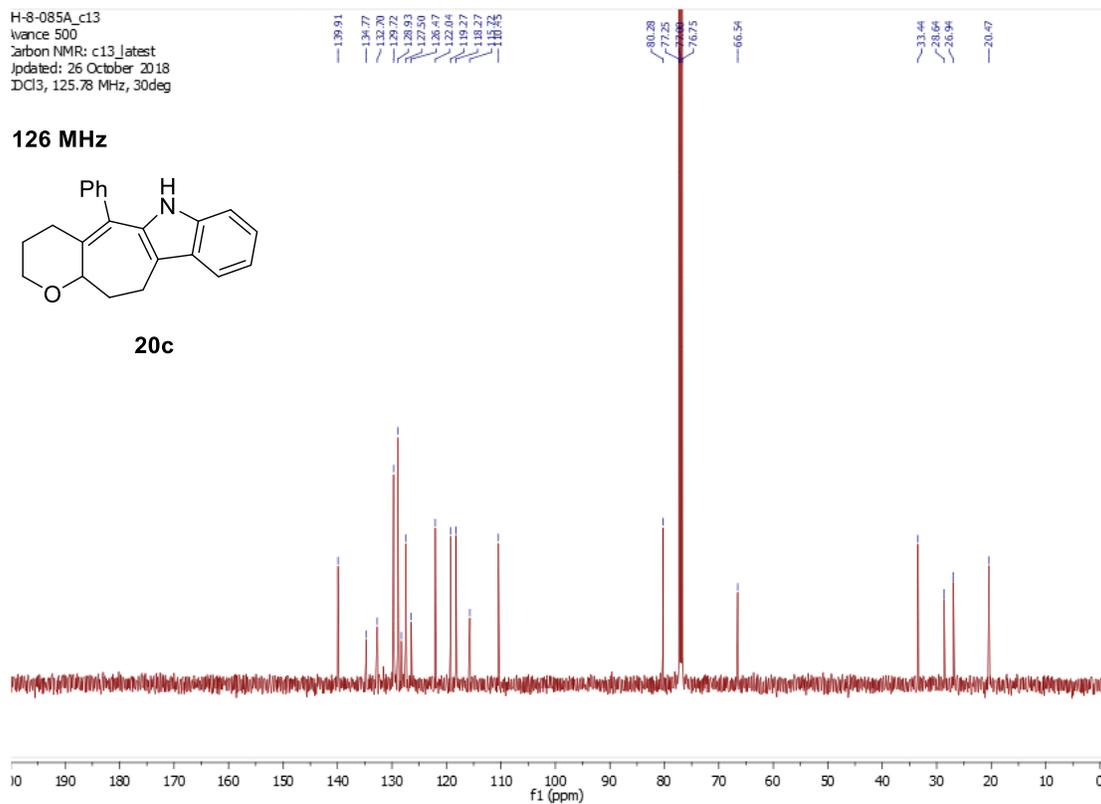
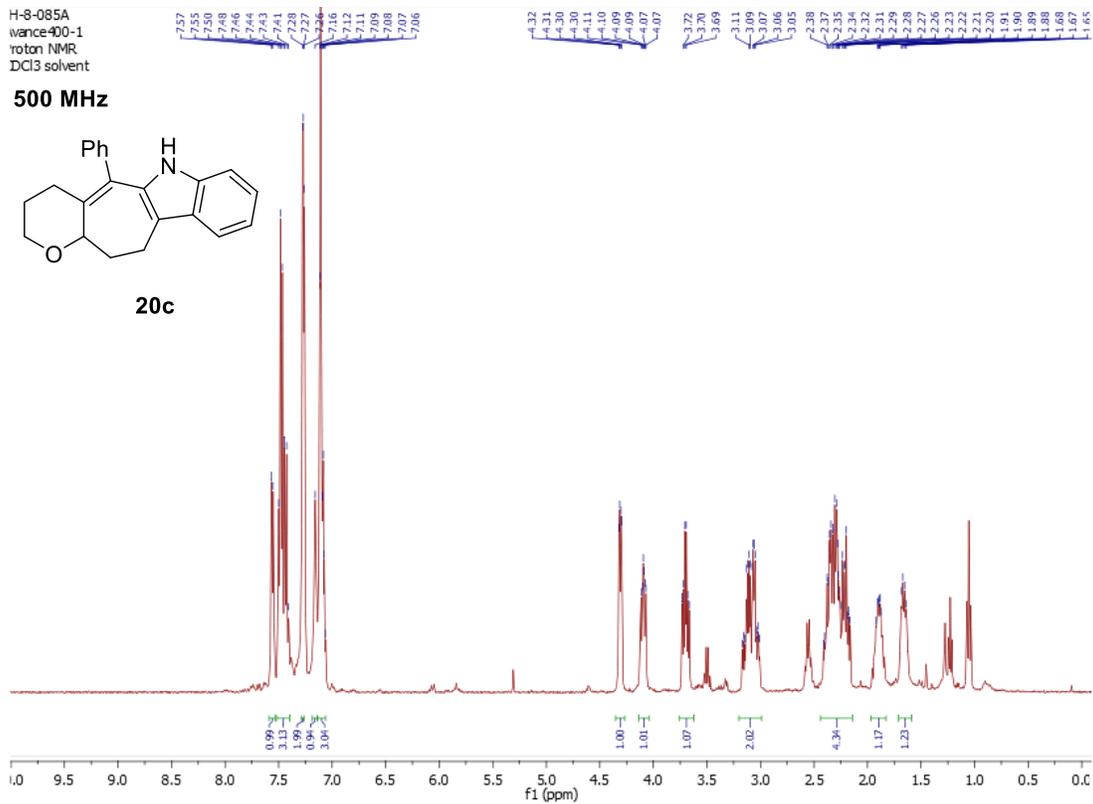
500 MHz



H-8-116A\_c13  
Avance 500  
Carbon NMR: c13\_latest  
Updated: 26 October 2018  
CDCl3, 125.78 MHz, 30deg

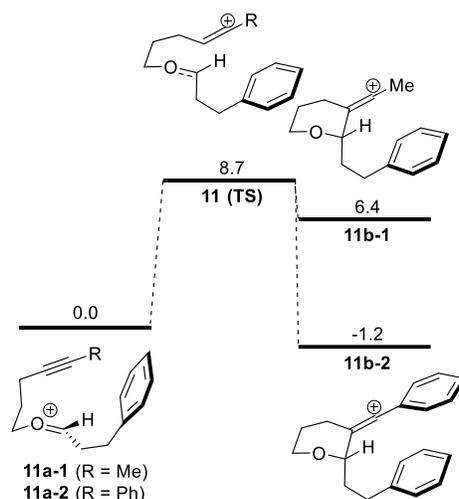
126 MHz





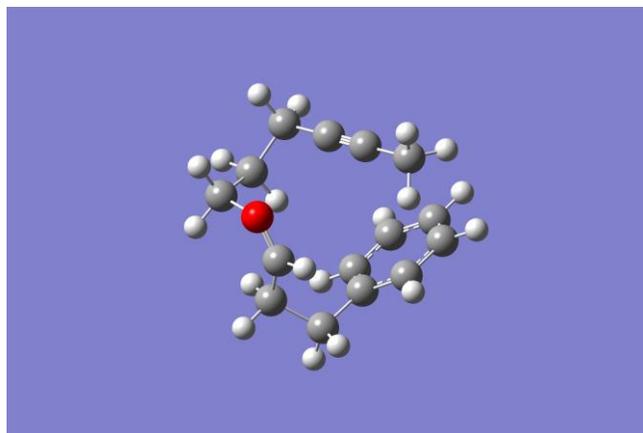
## Computational Methods:

Input files for ground state or transition state intermediates of a given reaction were prepared locally using GaussView 6.0 and transferred to the University of Rochester Blue Hive Linux cluster where DFT calculations were carried out in the Gaussian 16 suite of programs. Optimization calculations (either to a ground state or a transition state) were first carried out at the M06-2x functional and Def2TZVP basis set level of theory,<sup>30, 31</sup> with implicit solvation using the Polarizable Continuum Model (PCM = dichloromethane).<sup>32</sup> Intrinsic reaction coordinate (IRC)iv calculations were carried out on the transition state structures to verify that they connected to the associated reactant and product structures. Thereafter, frequency DFT calculations at the M06-2x/Def2TZVP level (PCM = dichloromethane), of the obtained optimized structures were carried out to confirm that ground state structures had zero imaginary frequencies and that transition states had a single imaginary frequency. All shown free energies are ZPE and thermally corrected (reported in Hartree units, at 298.15 K and 1 atm) and were obtained from the frequency calculations.



## DFT Optimized Structures

### 11a-1

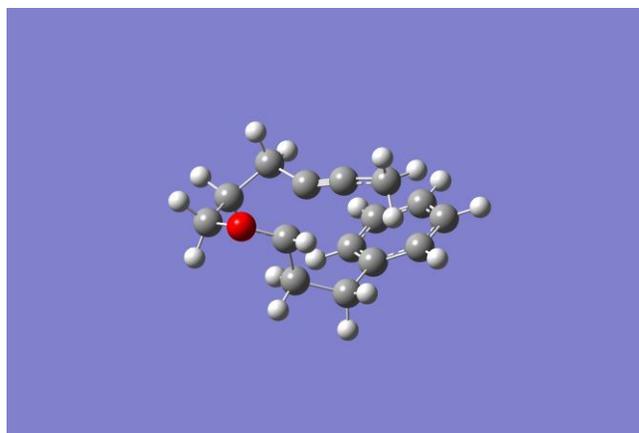


Imaginary Frequencies: 0

ZPE and thermally corrected free energy: -657.679999 Ha

| Center<br>Number | Atomic<br>Number | Forces (Hartrees/Bohr) |              |              |
|------------------|------------------|------------------------|--------------|--------------|
|                  |                  | X                      | Y            | Z            |
| 1                | 6                | -0.033778506           | -0.023072397 | -0.026009855 |
| 2                | 1                | 0.001406698            | -0.001659511 | 0.013578900  |
| 3                | 1                | 0.012515378            | -0.001945708 | -0.003722202 |
| 4                | 6                | -0.012274634           | -0.027203809 | 0.015798334  |
| 5                | 1                | 0.002616691            | 0.012184343  | 0.004740189  |
| 6                | 1                | 0.001268686            | 0.001443613  | -0.015793244 |
| 7                | 6                | 0.036677908            | 0.032561971  | 0.019031334  |
| 8                | 6                | -0.015930393           | -0.051451601 | -0.031740989 |
| 9                | 6                | -0.011130619           | 0.059638130  | 0.080600156  |
| 10               | 1                | -0.010469710           | 0.005858830  | -0.025360129 |
| 11               | 1                | 0.017108042            | -0.009269989 | -0.030087377 |
| 12               | 1                | 0.020445405            | -0.020550836 | -0.001647008 |
| 13               | 6                | 0.029734389            | 0.023631618  | -0.006333732 |
| 14               | 1                | -0.006209053           | -0.009285478 | 0.011874592  |
| 15               | 1                | -0.007949863           | 0.006398891  | 0.004085740  |
| 16               | 8                | -0.033402757           | -0.013127737 | -0.018190281 |
| 17               | 6                | 0.015900952            | 0.062734729  | 0.006326405  |
| 18               | 1                | -0.004300629           | -0.013896594 | 0.006236565  |
| 19               | 6                | 0.004431825            | -0.041608670 | -0.014194430 |
| 20               | 1                | -0.013815710           | 0.005904664  | 0.010695588  |
| 21               | 1                | 0.006897498            | 0.007219079  | -0.000424150 |
| 22               | 6                | 0.028199153            | -0.014913520 | 0.008593356  |
| 23               | 1                | -0.011941710           | -0.006198985 | -0.004177735 |
| 24               | 1                | -0.003474219           | 0.011105178  | -0.004671257 |
| 25               | 6                | -0.015449116           | -0.010871535 | 0.008631716  |
| 26               | 6                | -0.016667191           | 0.007928171  | 0.005722941  |
| 27               | 6                | 0.002118518            | 0.018605551  | -0.000127838 |
| 28               | 6                | 0.020096129            | 0.011220562  | -0.005655882 |
| 29               | 6                | -0.009242774           | 0.005551181  | -0.000865474 |
| 30               | 6                | 0.002861905            | -0.021715584 | 0.000884717  |
| 31               | 1                | 0.007156067            | 0.004846961  | -0.003101196 |
| 32               | 1                | 0.007071283            | -0.004495433 | -0.003003921 |
| 33               | 1                | 0.000505680            | -0.009173036 | 0.000070205  |
| 34               | 1                | -0.009552347           | -0.006286357 | -0.000797693 |
| 35               | 1                | -0.001422973           | 0.009893309  | -0.000966346 |

11-1 (TS)



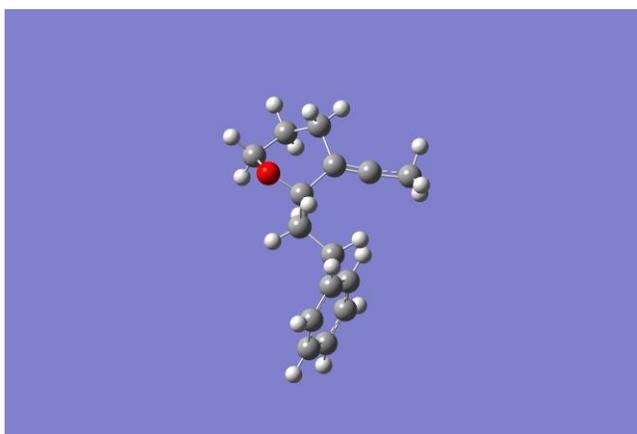
Imaginary Frequencies: 1

ZPE and thermally corrected free energy: -657.666073 Ha ( $\Delta G^\ddagger = 8.7$  kcal/mol for this transition state)

| Center Number | Atomic Number | Forces (Hartrees/Bohr) |              |              |
|---------------|---------------|------------------------|--------------|--------------|
|               |               | X                      | Y            | Z            |
| 1             | 6             | 0.000000120            | -0.000005403 | -0.000001616 |
| 2             | 1             | 0.000001965            | 0.000001738  | -0.000001930 |
| 3             | 1             | 0.000001040            | 0.000000564  | -0.000001605 |
| 4             | 6             | -0.000002986           | 0.000001650  | -0.000002029 |
| 5             | 1             | 0.000001644            | 0.000001914  | -0.000001638 |
| 6             | 1             | 0.000000335            | -0.000000424 | -0.000002412 |
| 7             | 6             | -0.000217714           | 0.000028284  | -0.000082695 |
| 8             | 6             | -0.000001496           | 0.000000517  | 0.000002922  |
| 9             | 6             | -0.000006039           | 0.000002592  | -0.000000500 |
| 10            | 1             | 0.000001274            | 0.000000432  | 0.000003852  |
| 11            | 1             | 0.000001453            | -0.000000013 | 0.000000894  |
| 12            | 1             | 0.000000527            | 0.000002831  | 0.000002291  |
| 13            | 6             | 0.000002582            | 0.000004411  | -0.000003697 |
| 14            | 1             | 0.000000494            | 0.000000476  | -0.000001156 |
| 15            | 1             | 0.000000896            | -0.000000238 | -0.000001661 |
| 16            | 8             | 0.000000807            | 0.000001841  | -0.000000960 |
| 17            | 6             | 0.000221624            | -0.000017745 | 0.000084756  |
| 18            | 1             | 0.000000524            | -0.000001707 | 0.000001750  |
| 19            | 6             | -0.000000126           | 0.000001733  | 0.000002200  |
| 20            | 1             | -0.000000127           | -0.000000494 | 0.000000028  |
| 21            | 1             | 0.000001104            | -0.000000815 | 0.000000314  |
| 22            | 6             | -0.000005456           | -0.000002181 | -0.000002330 |
| 23            | 1             | 0.000000433            | -0.000001566 | 0.000002196  |
| 24            | 1             | 0.000001181            | -0.000001410 | 0.000001185  |
| 25            | 6             | 0.000004357            | -0.000005913 | -0.000000103 |

|    |   |              |              |              |
|----|---|--------------|--------------|--------------|
| 26 | 6 | 0.000000163  | 0.000004516  | 0.000000374  |
| 27 | 6 | -0.000007261 | -0.000004719 | 0.000000798  |
| 28 | 6 | 0.000007272  | -0.000006457 | 0.000001989  |
| 29 | 6 | -0.000000142 | 0.000006040  | -0.000000300 |
| 30 | 6 | -0.000005157 | -0.000003407 | -0.000001035 |
| 31 | 1 | -0.000000591 | -0.000001286 | -0.000001651 |
| 32 | 1 | -0.000000512 | -0.000002283 | -0.000000155 |
| 33 | 1 | -0.000001009 | -0.000001768 | 0.000002148  |
| 34 | 1 | -0.000001969 | -0.000000801 | 0.000000059  |
| 35 | 1 | 0.000000788  | -0.000000908 | -0.000000284 |

### 11b-1



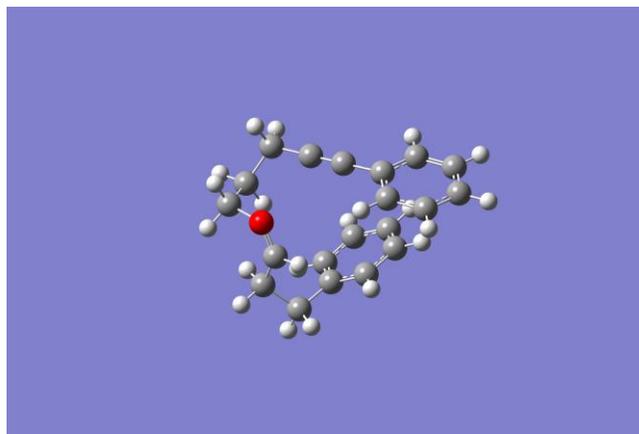
Imaginary Frequencies: 0

ZPE and thermally corrected free energy: -657.669757 Ha (6.4 kcal/mol vs **11a-1**)

| Center Number | Atomic Number | Forces (Hartrees/Bohr) |              |              |
|---------------|---------------|------------------------|--------------|--------------|
|               |               | X                      | Y            | Z            |
| 1             | 6             | -0.018960152           | -0.028604950 | 0.007079143  |
| 2             | 1             | 0.008898174            | 0.009116311  | -0.012071224 |
| 3             | 1             | -0.008978384           | 0.008195809  | 0.007648726  |
| 4             | 6             | 0.020074834            | 0.008022633  | -0.012770079 |
| 5             | 1             | 0.005018561            | 0.003143474  | 0.011767332  |
| 6             | 1             | -0.001715886           | 0.002121048  | 0.017425819  |
| 7             | 6             | -0.038696617           | 0.066918221  | -0.055769239 |
| 8             | 6             | 0.103743812            | 0.000519527  | -0.017217817 |
| 9             | 6             | 0.048734220            | -0.021325780 | -0.027954420 |
| 10            | 1             | -0.011003612           | -0.002767951 | -0.002527881 |
| 11            | 1             | 0.003108052            | 0.012095353  | 0.014718633  |
| 12            | 1             | 0.008617017            | 0.009581789  | -0.011215492 |
| 13            | 6             | -0.030460541           | 0.012006541  | 0.000680910  |
| 14            | 1             | 0.011041942            | 0.005516965  | -0.000114454 |

|    |   |              |              |              |
|----|---|--------------|--------------|--------------|
| 15 | 1 | 0.003438413  | -0.016269657 | -0.001886373 |
| 16 | 8 | -0.010154658 | 0.001051233  | -0.002989670 |
| 17 | 6 | 0.033747146  | -0.005475536 | 0.018348257  |
| 18 | 1 | 0.004352294  | -0.000211356 | -0.012745627 |
| 19 | 6 | -0.022078742 | 0.014480031  | -0.013346422 |
| 20 | 1 | 0.010364800  | -0.008317500 | -0.008503908 |
| 21 | 1 | -0.003280429 | -0.014930565 | -0.002295862 |
| 22 | 6 | 0.007104047  | 0.013605311  | 0.032753271  |
| 23 | 1 | -0.004089309 | 0.003407264  | -0.014924465 |
| 24 | 1 | -0.007065911 | -0.010178585 | -0.002536586 |
| 25 | 6 | 0.008089404  | 0.011500432  | -0.014909333 |
| 26 | 6 | 0.008365222  | -0.010476281 | -0.012100029 |
| 27 | 6 | 0.000186660  | -0.019915839 | 0.001172420  |
| 28 | 6 | -0.054787753 | -0.020915834 | 0.064251151  |
| 29 | 6 | 0.001582783  | -0.007891993 | -0.011235608 |
| 30 | 6 | 0.000361728  | 0.024866322  | -0.000276187 |
| 31 | 1 | -0.003435745 | -0.003999709 | 0.006660603  |
| 32 | 1 | -0.003301252 | 0.004581443  | 0.006321227  |
| 33 | 1 | -0.001356714 | 0.009941461  | -0.000017051 |
| 34 | 1 | -0.067143924 | -0.039855427 | 0.049641435  |
| 35 | 1 | -0.000319481 | -0.009534206 | -0.001061200 |

### 11a-2



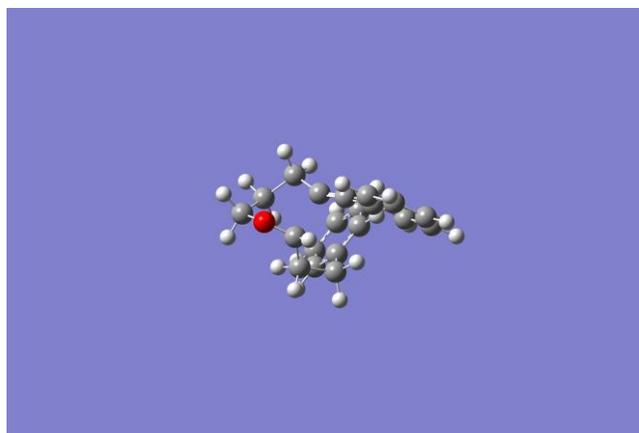
Imaginary Frequencies: 0

ZPE and thermally corrected free energy: -849.364613 Ha

| Center<br>Number | Atomic<br>Number | Forces (Hartrees/Bohr) |             |              |
|------------------|------------------|------------------------|-------------|--------------|
|                  |                  | X                      | Y           | Z            |
| 1                | 6                | 0.000074051            | 0.000034380 | 0.000019754  |
| 2                | 1                | -0.000187887           | 0.000195494 | -0.000334373 |
| 3                | 1                | -0.000347788           | 0.000250583 | -0.000479096 |

|    |   |              |              |              |
|----|---|--------------|--------------|--------------|
| 4  | 6 | -0.000221724 | -0.000664105 | -0.000440037 |
| 5  | 1 | -0.000112575 | 0.000297270  | -0.000065871 |
| 6  | 1 | -0.000081377 | -0.000105463 | 0.000053352  |
| 7  | 6 | 0.001571325  | -0.000411093 | 0.002328520  |
| 8  | 6 | -0.033455342 | 0.015246665  | -0.045982752 |
| 9  | 6 | 0.000291942  | -0.000118509 | -0.000136902 |
| 10 | 1 | -0.000001088 | -0.000054578 | -0.000030575 |
| 11 | 1 | -0.000160165 | -0.000000862 | 0.000009635  |
| 12 | 8 | 0.000193782  | 0.000419536  | 0.000153693  |
| 13 | 6 | -0.000567651 | -0.000452193 | 0.000277300  |
| 14 | 1 | -0.000000788 | -0.000595965 | 0.000288994  |
| 15 | 6 | 0.000176016  | 0.000056817  | -0.000124051 |
| 16 | 1 | 0.000112238  | -0.000040806 | -0.000050367 |
| 17 | 1 | 0.000114055  | -0.000043123 | 0.000154074  |
| 18 | 6 | -0.000109904 | -0.000055719 | 0.000039285  |
| 19 | 1 | 0.000029663  | 0.000072675  | -0.000062552 |
| 20 | 1 | 0.000068824  | -0.000048652 | -0.000024215 |
| 21 | 6 | 0.000231722  | 0.000063279  | 0.000075525  |
| 22 | 6 | 0.000005215  | -0.000122775 | -0.000185442 |
| 23 | 6 | -0.000764818 | -0.000273210 | -0.000018911 |
| 24 | 6 | -0.000016235 | -0.000106553 | -0.000174439 |
| 25 | 6 | 0.000019469  | 0.000245840  | -0.000115315 |
| 26 | 6 | 0.000169232  | -0.000067207 | 0.000235245  |
| 27 | 1 | 0.000000882  | -0.000031660 | 0.000047285  |
| 28 | 1 | -0.000072911 | 0.000083774  | 0.000030416  |
| 29 | 1 | 0.000369076  | -0.000102983 | -0.000051158 |
| 30 | 1 | 0.000129715  | 0.000340598  | 0.000009364  |
| 31 | 1 | -0.000027241 | -0.000002162 | -0.000033812 |
| 32 | 6 | 0.028564481  | -0.012748768 | 0.039440740  |
| 33 | 6 | -0.004782598 | 0.005779384  | 0.006461551  |
| 34 | 6 | 0.005671760  | -0.005681632 | -0.004594782 |
| 35 | 6 | -0.002675974 | 0.002239062  | 0.001041491  |
| 36 | 1 | 0.003065017  | -0.004634352 | -0.010249645 |
| 37 | 6 | 0.001114326  | -0.001991327 | -0.003613265 |
| 38 | 1 | -0.008674721 | 0.007771932  | 0.001767314  |
| 39 | 6 | -0.003726890 | 0.001462372  | -0.004782786 |
| 40 | 1 | 0.009447128  | -0.007073723 | -0.000976208 |
| 41 | 1 | -0.002143644 | 0.003876310  | 0.011027552  |
| 42 | 1 | 0.006711404  | -0.003008550 | 0.009065465  |

**11-2 (TS)**



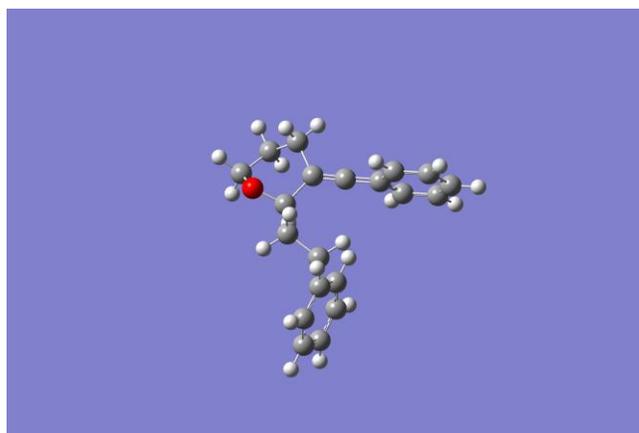
Imaginary Frequencies: 1

ZPE and thermally corrected free energy: -849.350748 Ha ( $\Delta G^\ddagger = 8.7$  kcal/mol for this transition state)

| Center Number | Atomic Number | Forces (Hartrees/Bohr) |              |              |
|---------------|---------------|------------------------|--------------|--------------|
|               |               | X                      | Y            | Z            |
| 1             | 6             | -0.001356407           | 0.000710171  | 0.000735663  |
| 2             | 1             | -0.000166144           | 0.000026885  | -0.000309697 |
| 3             | 1             | 0.000370374            | 0.000291529  | 0.000183803  |
| 4             | 6             | 0.000149202            | -0.000126488 | -0.000048005 |
| 5             | 1             | -0.000033509           | 0.000135617  | 0.000007677  |
| 6             | 1             | -0.000022705           | 0.000040227  | -0.000053537 |
| 7             | 6             | 0.000202679            | -0.006783279 | -0.004213658 |
| 8             | 6             | -0.005529553           | -0.008157009 | 0.000140949  |
| 9             | 6             | -0.000424824           | -0.000642014 | -0.000313562 |
| 10            | 1             | 0.000063254            | 0.000105991  | 0.000023948  |
| 11            | 1             | 0.000153478            | 0.000227393  | 0.000027891  |
| 12            | 8             | 0.003563091            | 0.000905545  | -0.000791349 |
| 13            | 6             | -0.003927382           | 0.000540465  | 0.004233962  |
| 14            | 1             | -0.002934389           | -0.003912506 | -0.000308122 |
| 15            | 6             | 0.001193147            | -0.000770402 | -0.000015499 |
| 16            | 1             | -0.000268191           | 0.000236311  | -0.000235996 |
| 17            | 1             | -0.000016354           | 0.000190415  | -0.000134410 |
| 18            | 6             | 0.000096621            | 0.000219870  | -0.000115960 |
| 19            | 1             | 0.000519338            | -0.000220970 | -0.000182232 |
| 20            | 1             | -0.000197968           | 0.000029635  | 0.000043225  |
| 21            | 6             | 0.000007347            | -0.000207695 | -0.000096855 |
| 22            | 6             | -0.000072415           | 0.000024240  | 0.000034036  |
| 23            | 6             | -0.000146671           | -0.000150202 | 0.000268818  |
| 24            | 6             | 0.000364252            | 0.000237778  | 0.000206996  |
| 25            | 6             | 0.000226518            | 0.000136457  | -0.000315768 |

|    |   |              |              |              |
|----|---|--------------|--------------|--------------|
| 26 | 6 | -0.000153423 | -0.000024371 | -0.000097922 |
| 27 | 1 | 0.000002238  | -0.000028525 | 0.000002878  |
| 28 | 1 | 0.000017096  | -0.000038323 | -0.000044726 |
| 29 | 1 | -0.000075806 | 0.000100476  | -0.000019340 |
| 30 | 1 | -0.000538757 | 0.000399380  | -0.000523255 |
| 31 | 1 | 0.000007750  | -0.000102036 | -0.000000713 |
| 32 | 6 | 0.005748093  | 0.013272561  | 0.001868112  |
| 33 | 6 | 0.010898742  | -0.002820310 | -0.014018863 |
| 34 | 6 | -0.004950443 | 0.010324857  | 0.012344838  |
| 35 | 6 | 0.000582755  | -0.001269999 | -0.002510665 |
| 36 | 1 | -0.001898224 | 0.007820327  | 0.008866218  |
| 37 | 6 | -0.001796900 | 0.000704262  | 0.003757778  |
| 38 | 1 | 0.010730498  | 0.007075708  | -0.008953293 |
| 39 | 6 | 0.002785327  | 0.004455758  | 0.000001163  |
| 40 | 1 | -0.008381915 | -0.003410056 | 0.008567424  |
| 41 | 1 | 0.001203885  | -0.009143054 | -0.008368086 |
| 42 | 1 | -0.005993706 | -0.010404622 | 0.000356135  |

### 11b-2



Imaginary Frequencies: 0

ZPE and thermally corrected free energy: -849.366505 Ha (-1.2 kcal/mol vs **11a-2**)

| Center<br>Number | Atomic<br>Number | Forces (Hartrees/Bohr) |              |              |
|------------------|------------------|------------------------|--------------|--------------|
|                  |                  | X                      | Y            | Z            |
| 1                | 6                | -0.003178310           | 0.001313664  | 0.006652844  |
| 2                | 1                | -0.001277528           | -0.000881186 | -0.001410797 |
| 3                | 1                | 0.000210115            | 0.000205687  | 0.000062185  |
| 4                | 6                | 0.000505260            | 0.000072365  | 0.000495412  |
| 5                | 1                | 0.000021331            | 0.000112663  | -0.000043206 |
| 6                | 1                | 0.000200993            | -0.000007869 | 0.000000502  |
| 7                | 6                | -0.001065710           | -0.012859666 | 0.000127445  |

|    |   |              |              |              |
|----|---|--------------|--------------|--------------|
| 8  | 6 | 0.002047379  | 0.024734608  | -0.001012505 |
| 9  | 6 | 0.000085739  | 0.000397657  | 0.000534073  |
| 10 | 1 | -0.000056544 | -0.000066560 | -0.000066775 |
| 11 | 1 | 0.000110375  | -0.000037802 | -0.000023727 |
| 12 | 8 | -0.000757328 | -0.001918582 | -0.000279913 |
| 13 | 6 | 0.002370859  | 0.003607000  | -0.006376094 |
| 14 | 1 | -0.000280643 | 0.000481614  | -0.000110518 |
| 15 | 6 | 0.000265143  | -0.001831999 | 0.001249925  |
| 16 | 1 | 0.000589902  | -0.000201870 | -0.000193722 |
| 17 | 1 | -0.000181325 | -0.000002282 | 0.000046148  |
| 18 | 6 | 0.000361861  | 0.001239164  | -0.000514229 |
| 19 | 1 | 0.000307709  | -0.000549649 | 0.000038020  |
| 20 | 1 | 0.000084326  | -0.000143839 | 0.000151439  |
| 21 | 6 | 0.000029316  | -0.000017878 | 0.000042884  |
| 22 | 6 | 0.000061064  | 0.000040343  | -0.000053428 |
| 23 | 6 | -0.000000743 | 0.000001198  | 0.000023113  |
| 24 | 6 | 0.000101240  | 0.000001279  | -0.000064180 |
| 25 | 6 | -0.000300988 | -0.000325253 | 0.000482698  |
| 26 | 6 | 0.000205037  | 0.000009587  | -0.000217062 |
| 27 | 1 | 0.000030398  | -0.000016711 | 0.000001165  |
| 28 | 1 | 0.000016479  | -0.000003924 | -0.000028183 |
| 29 | 1 | 0.000025633  | -0.000005862 | -0.000030966 |
| 30 | 1 | -0.000055630 | 0.000004225  | 0.000044267  |
| 31 | 1 | -0.000031026 | 0.000025049  | 0.000009429  |
| 32 | 6 | -0.001868082 | -0.018294377 | 0.000590152  |
| 33 | 6 | -0.005214486 | 0.014392338  | 0.026350973  |
| 34 | 6 | 0.010998822  | 0.012362299  | -0.026133627 |
| 35 | 6 | 0.000878542  | -0.005721700 | 0.008964540  |
| 36 | 1 | 0.001211847  | 0.005580706  | -0.010537532 |
| 37 | 6 | -0.003373706 | -0.004808465 | -0.009449733 |
| 38 | 1 | 0.001157871  | 0.005045454  | 0.010756113  |
| 39 | 6 | 0.000757405  | 0.003903013  | 0.000448723  |
| 40 | 1 | -0.000644016 | -0.007009675 | -0.010555118 |
| 41 | 1 | -0.002186409 | -0.007155809 | 0.010302010  |
| 42 | 1 | -0.002162171 | -0.011668957 | -0.000272747 |

## References:

1. Alachouzos, G.; Frontier, A. J., Diastereoselective Construction of Densely Functionalized 1-Halocyclopentenes Using an Alkynyl Halo-Prins/Halo-Nazarov Cyclization Strategy. *Angew. Chem., Int. Ed.* **2017**, *56*, 15030-15034.
2. Holt, C.; Alachouzos, G.; Frontier, A. J., Leveraging the Halo-Nazarov Cyclization for the Chemodivergent Assembly of Functionalized Haloindenes and Indanones. *J. Am. Chem. Soc.* **2019**, *141*, 5461-5469.
3. Abdul-Rashed, S.; Alachouzos, G.; Brennessel, W. W.; Frontier, A. J., One-Pot Double-Annulation Strategy for the Synthesis of Unusual Fused Bis-Heterocycles. *Org. Lett.* **2020**, *22*, 4350-4354.
4. Mostafa, M. A. B.; Grafton, M. W.; Wilson, C.; Sutherland, A., A one-pot, three-step process for the diastereoselective synthesis of aminobicyclo[4.3.0]nonanes using consecutive palladium(ii)- and ruthenium(ii)-catalysis. *Org. Biomol. Chem.* **2016**, *14*, 3284-3297.
5. Trost, B. M.; Machacek, M. R.; Faulk, B. D., Sequential Ru–Pd Catalysis: A Two-Catalyst One-Pot Protocol for the Synthesis of N- and O-Heterocycles. *J. Am. Chem. Soc.* **2006**, *128*, 6745-6754.
6. Xie, H.; Zhang, S.; Li, H.; Zhang, X.; Zhao, S.; Xu, Z.; Song, X.; Yu, X.; Wang, W., Total Synthesis of Polyene Natural Product Dihydroxerulin by Mild Organocatalyzed Dehydrogenation of Alcohols. *Chemistry – A European Journal* **2012**, *18*, 2230-2234.
7. Park, H.; Edgar, L. J.; Lumba, M. A.; Willis, L. M.; Nitz, M., Organotellurium scaffolds for mass cytometry reagent development. *Org. Biomol. Chem.* **2015**, *13*, 7027-7033.
8. Cabrera-Lobera, N.; Quirós, M. T.; Buñuel, E.; Cárdenas, D. J., Atom-economical regioselective Ni-catalyzed hydroborylative cyclization of enynes: development and mechanism. *Catalysis Science & Technology* **2019**, *9*, 1021-1029.
9. Bates, R. W.; Maiti, T. B., Propylene Glycol Cyclic Sulfate as a Substitute for Propylene Oxide in Reactions with Acetylides. *Synth. Commun.* **2003**, *33*, 633-640.
10. Alachouzos, G.; Holt, C.; Frontier, A. J., Stereochemical Relay through a Cationic Intermediate: Helical Preorganization Dictates Direction of Conrotation in the halo-Nazarov Cyclization. *Org. Lett.* **2020**, *22*, 4010-4015.
11. Ito, H.; Kubota, K., Copper(I)-Catalyzed Boryl Substitution of Unactivated Alkyl Halides. *Org. Lett.* **2012**, *14*, 890-893.
12. Yin, Y.; Ma, W.; Chai, Z.; Zhao, G., Et<sub>2</sub>Zn-Catalyzed Intramolecular Hydroamination of Alkynyl Sulfonamides and the Related Tandem Cyclization/Addition Reaction. *J. Org. Chem.* **2007**, *72*, 5731-5736.
13. Kavala, M.; Mathia, F.; Kožíšek, J.; Szolcsányi, P., Efficient Total Synthesis of (+)-Dihydropinidine, (–)-Epidihydropinidine, and (–)-Pinidinone. *J. Nat. Prod.* **2011**, *74*, 803-808.
14. Wallace, K. J.; Hanes, R.; Anslyn, E.; Morey, J.; Kilway, K. V.; Siegel, J., Preparation of 1,3,5-Tris(aminomethyl)-2,4,6-triethylbenzene from Two Versatile 1,3,5-Tri(halosubstituted) 2,4,6-Triethylbenzene Derivatives. *Synthesis* **2005**, *2005*, 2080-2083.
15. Simmons, E. M.; Hartwig, J. F., Catalytic functionalization of unactivated primary C–H bonds directed by an alcohol. *Nature* **2012**, *483*, 70-73.
16. Williams, C. W.; Shenje, R.; France, S., Catalytic, Interrupted Formal Homo-Nazarov Cyclization with (Hetero)arenes: Access to  $\alpha$ -(Hetero)aryl Cyclohexanones. *J. Org. Chem.* **2016**, *81*, 8253-8267.
17. Auzzas, L.; Larsson, A.; Matera, R.; Baraldi, A.; Deschênes-Simard, B.; Giannini, G.; Cabri, W.; Battistuzzi, G.; Gallo, G.; Ciacci, A.; Vesci, L.; Pisano, C.; Hanessian, S., Non-

Natural Macrocyclic Inhibitors of Histone Deacetylases: Design, Synthesis, and Activity. *J. Med. Chem.* **2010**, *53*, 8387-8399.

18. Häfner, M.; Sokolenko, Y. M.; Gamedinger, P.; Stempel, E.; Gaich, T., Enantioselective Synthesis of Cyclohepta[b]indoles via Pd-Catalyzed Cyclopropane C(sp<sup>3</sup>)-H Activation as a Key Step. *Org. Lett.* **2019**, *21*, 7370-7374.
19. Yuan, K.; Soulé, J.-F.; Dorcet, V.; Doucet, H., Palladium-Catalyzed Cascade sp<sup>2</sup> C-H Bond Functionalizations Allowing One-Pot Access to 4-Aryl-1,2,3,4-tetrahydroquinolines from N-Allyl-N-arylsulfonamides. *ACS Catal.* **2016**, *6*, 8121-8126.
20. HFIP, which is known to stabilize carbocationic species (Chem. Rev. 2022, 122, 12544-12747.), improves reaction outcomes (see Table 1).
21. Corcoran, J. C.; Guo, R.; Xia, Y.; Wang, Y.-M., Vinyl cation-mediated intramolecular hydroarylation of alkynes using pyridinium reagents. *Chem. Commun.* **2022**, *58*, 11523-11526.
22. Drahl, M. A.; Manpadi, M.; Williams, L. J., C-C Fragmentation: Origins and Recent Applications. *Angew. Chem. Int. Ed.* **2013**, *52*, 11222-11251.
23. Crabtree, R. H.; Felkin, H.; Morris, G. E., Cationic iridium diolefin complexes as alkene hydrogenation catalysts and the isolation of some related hydrido complexes. *J. Organomet. Chem.* **1977**, *141*, 205-215.
24. Crabtree, R., Iridium compounds in catalysis. *Acc. Chem. Res.* **1979**, *12*, 331-337.
25. Léonard, N. G.; Chirik, P. J., Air-Stable  $\alpha$ -Diimine Nickel Precatalysts for the Hydrogenation of Hindered, Unactivated Alkenes. *ACS Catal.* **2018**, *8*, 342-348.
26. Okamoto, A.; Snow, M. S.; Arnold, D. R., Photosensitized (electron transfer) carbon-carbon bond cleavage of radical cations: The diphenylmethyl system. *Tetrahedron* **1986**, *42*, 6175-6187.
27. Zúñiga, A.; Pérez, M.; Pazos, G.; Gómez, G.; Fall, Y., A Concise, Enantioselective Synthesis of (+)-Decarestrictine L from Tri-O-acetyl-d-glucal. *Synthesis* **2010**, *2010*, 2446-2450.
28. Himo, F.; Lovell, T.; Hilgraf, R.; Rostovtsev, V. V.; Noodleman, L.; Sharpless, K. B.; Fokin, V. V., Copper(I)-Catalyzed Synthesis of Azoles. DFT Study Predicts Unprecedented Reactivity and Intermediates. *J. Am. Chem. Soc.* **2005**, *127*, 210-216.
29. Heredia, M. D.; Guerra, W. D.; Barolo, S. M.; Fornasier, S. J.; Rossi, R. A.; Budén, M. E., Transition-Metal-Free and Visible-Light-Mediated Desulfonylation and Dehalogenation Reactions: Hantzsch Ester Anion as Electron and Hydrogen Atom Donor. *J. Org. Chem.* **2020**, *85*, 13481-13494.
30. Zhao, Y.; Truhlar, D. G., The M06 suite of density functionals for main group thermochemistry, thermochemical kinetics, noncovalent interactions, excited states, and transition elements: two new functionals and systematic testing of four M06-class functionals and 12 other functionals. *Theor. Chem. Acc.* **2008**, *120*, 215-241.
31. Zheng, J.; Xu, X.; Truhlar, D. G., Minimally augmented Karlsruhe basis sets. *Theor. Chem. Acc.* **2011**, *128*, 295-305.
32. Tomasi, J.; Mennucci, B.; Cammi, R., Quantum Mechanical Continuum Solvation Models. *Chem. Rev.* **2005**, *105*, 2999-3094.