

# Bidentate Substrate Binding in Brønsted Acid Catalysis: Structural Space, Hydrogen Bonding and Dimerization

Johannes Gramüller, Philipp Dullinger, Dominik Horinek and Ruth M. Gschwind\*

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## 1. General information

### 1.1 Chemicals

Deuterated solvents were purchased from Deutero or Sigma Aldrich. Where dry solvents were essential, CD<sub>2</sub>Cl<sub>2</sub> was freshly distilled over CaH<sub>2</sub> and Toluene-d<sub>8</sub> was refluxed over Na/Benzophenone under argon atmosphere. Non-deuterated toluene was refluxed over Na/Benzophenone under Argon atmosphere or purchased by Sigma Aldrich (dry, stored over molecular sieve). All commercially available chemicals were purchased by Sigma Aldrich or abcr and used without further purification.

### 1.2 Sample preparation

The chiral phosphoric acid was weighed into a 5 mm NMR tube and dried at 150 °C for at least 20 min under reduced pressure. After the tube came to room temperature, the respective imine was weighed directly into the NMR tube. The tube was evacuated and flushed with Argon three times. Dry CD<sub>2</sub>Cl<sub>2</sub> (0.6 mL) was added under Argon flow and TMS atmosphere (0.5 mL) was added. The tube was closed and sealed with parafilm. The samples were stored in the fridge at -80 °C.

For the preparation of the “E-only sample”, prior to adding the solvent the tube was cooled to -80 °C in an acetone/liquid nitrogen bath. Under Argon flow, cooled CD<sub>2</sub>Cl<sub>2</sub> (0.6 mL) was added to the CPA and imine. TMS atmosphere (0.5 mL) was added and the tube was closed and sealed with parafilm. The samples were stored in the fridge at -80 °C. E-only samples were transported at -80 °C in an acetone/liquid nitrogen cooling bath and only inserted in a pre-cooled spectrometer at 180 K.

For the *in situ* UV VIS illumination experiments, a glass fiber based illumination setup developed by our group was employed.<sup>1,2</sup> The sample preparation was analogous, except only 0.35 mL of solvent were used. After solvation, the glass fiber was inserted into the ambered tube and fixated with parafilm.

### 1.3 NMR spectrometer, data procession and referencing

All NMR spectroscopic investigations on model systems were performed on a Bruker Avance DRX 600 MHz spectrometer with TBI (Triple resonance broadband inverse) 5 mm CPPBBO <sup>1</sup>H/<sup>19</sup>F-BB probe head with Z-gradient and BVT unit. Temperature was controlled in the VT-

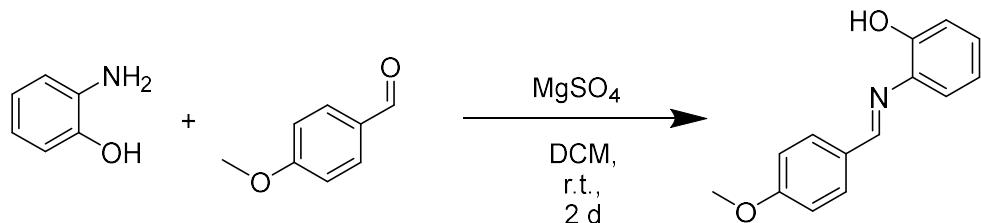
experiments by a BVT 3000 and BVT 3900 unit and liquid nitrogen. Further NMR experiments were performed on Bruker Avance III HD 400 MHz spectrometer equipped with 5 mm BBO BB-1H/D probe head with Z-Gradients. Spectrometer control and spectra processing was performed with Bruker Software TopSpin (Version 3.2 PL 1). Further plotting of the spectra was performed with Corel Draw 2020 software.  $^1\text{H}$ ,  $^{13}\text{C}$  chemical shifts were referenced to TMS or the respective solvent signals. The heteronucleus  $^{31}\text{P}$  was referenced, employing  $v(X) = v(\text{TMS}) \cdot \Xi_{\text{reference}} / 100\%$  according to Harris et al.<sup>3</sup> The following frequency ratios and reference compounds was used:  $\Xi(^{31}\text{P}) = 40.480742$  ( $\text{H}_3\text{PO}_4$ ). For all NMR measurements, 5 mm NMR tubes were used.

#### *1.4 Pulse sequences and acquisition parameters*

$^1\text{H-NMR}$ : Pulse program zg30, Relaxation delay = 2.00 s, Acquisition time = 2.54 s, SW = 22 – 24 ppm, TD = 66 K, ns = 16 – 256;  $^{13}\text{C NMR}$ : Pulse program: zgpg30, Relaxation delay = 2.00 s, Acquisition time = 0.80 s, TD = 66 K; SW = 270.0 ppm, TD = 64k, NS = 1k – 4k;  $^{31}\text{P-NMR}$ : Pulse program: zgpg30; Relaxation delay = 1.00 s, Acquisition time = 2.25 s, SW = 20 - 60.0 ppm, TD = 65k, NS = 256 - 512;  $2\text{D-}^1\text{H,}^1\text{H NOESY}$ : Pulse program: noesygpph; Relaxation delay = 5.00 s, NS = 8-16, mixing time (D8) = 300.00 ms; TD = 4096; increments = 512 - 1k;  $2\text{D-}^1\text{H,}^1\text{H ROESY}$ : Pulse program: roesyphpr.2; Relaxation delay = 5.00 s, NS = 8, mixing time (D8) = 100.00 ms; TD = 4096; increments = 1k;  $2\text{D-}^1\text{H,}^1\text{H COSY}$ : Pulse program: cosygpqf; Relaxation delay = 5.00 s, NS = 4-16, TD = 4096; increments = 512;  $2\text{D-}^1\text{H,}^{13}\text{C HSQC}$ : Pulse program: hsqce-detgpsisp2.3; Relaxation delay = 4 - 8 s, NS = 8-32,  $^1J_{\text{XH}} = 145$  Hz; TD = 4096; increments = 512 - 1k;  $2\text{D-}^1\text{H,}^{13}\text{C HMBC}$ : Pulse program: hmbcgplndqf; Relaxation delay = 4.00 s, NS = 8-16,  $^1J_{\text{XH}} = 145$  Hz,  $J_{\text{XH}}$ (long range) = 10 Hz; TD = 4096; increments = 512 - 1k;  $2\text{D-}^1\text{H,}^{31}\text{P HMBC}$ : Pulse program: inv4gplrndqf; Relaxation delay = 6.00 s, NS = 4-32, TD = 4096; increments = 256 – 512

## 2. Imine synthesis

### (E)-2-((4-methoxybenzylidene)amino)phenol (2a)

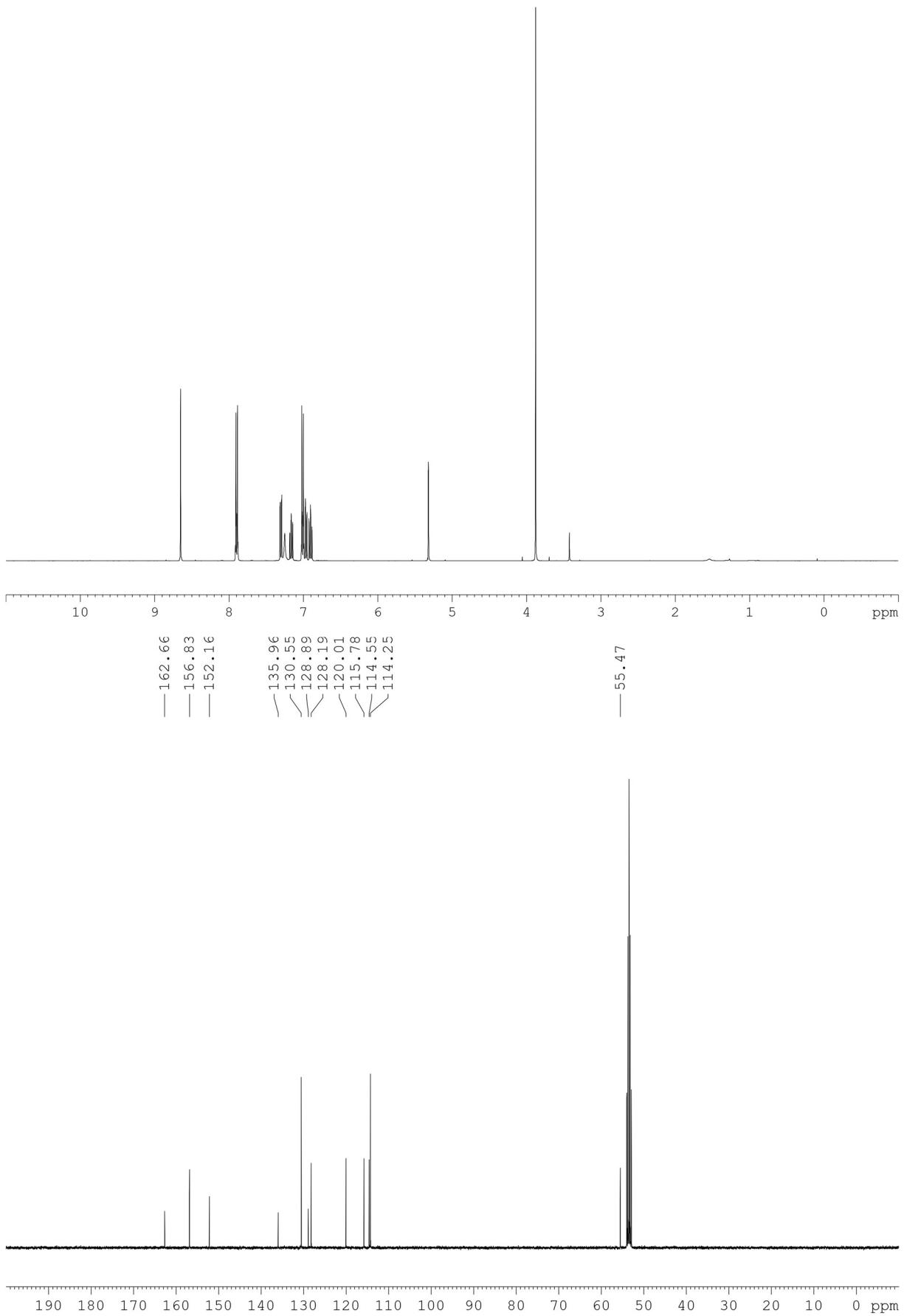


2-Aminophenol (7.00 mmol, 0.65 g, 0.55 mL, 1.7 eq), 4-Methoxybenzaldehyde (4.05 mmol, 0.55 g, 1.0 eq) and MgSO<sub>4</sub> (4.70 g) were weighed into a 50 mL Schlenk flask and dissolved in 20 mL DCM. The solution was stirred for two days at room temperature. Afterwards, the solvent was removed under reduced pressure to give a yellow solid. The crude product was recrystallized in methanol two times to give the product as yellow crystals. (0.55 g, 2.60 mmol, 37%).

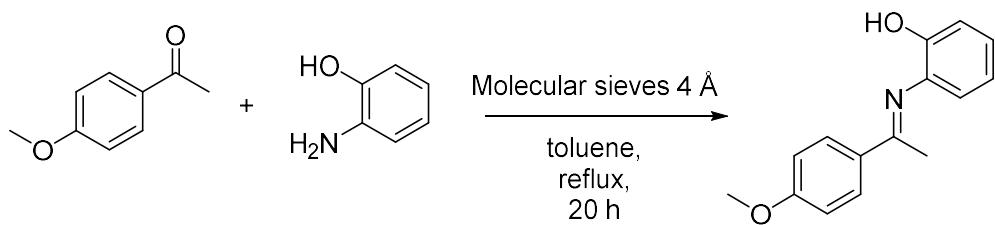
<sup>1</sup>H-NMR: (400.1 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ<sub>H</sub> = 8.65 (s, 1H), 7.90 (m, 2H), 7.30 (dd, 1H, <sup>3</sup>J = 8.0 Hz, <sup>4</sup>J = 1.5 Hz), 7.25 (s, 1H), 7.16 (m, 1H), 7.01 (m, 2H), 6.97 (dd, 1H, <sup>3</sup>J = 8.1 Hz, <sup>4</sup>J = 1.4 Hz), 6.90 (m, 1H), 3.88 (s, 1H).

<sup>13</sup>C-NMR: (100.6 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ<sub>C</sub> = 162.7, 156.8, 152.2, 135.9, 130.5, 128.9, 128.2, 120.0, 115.8, 114.6, 114.3, 55.5.

HR-MS (EI, *m/z*): found 226.08607 (M-H)<sup>+</sup> (calculated 226.08626 for C<sub>14</sub>H<sub>12</sub>NO<sub>2</sub> ); Diff(ppm) = -0.81.



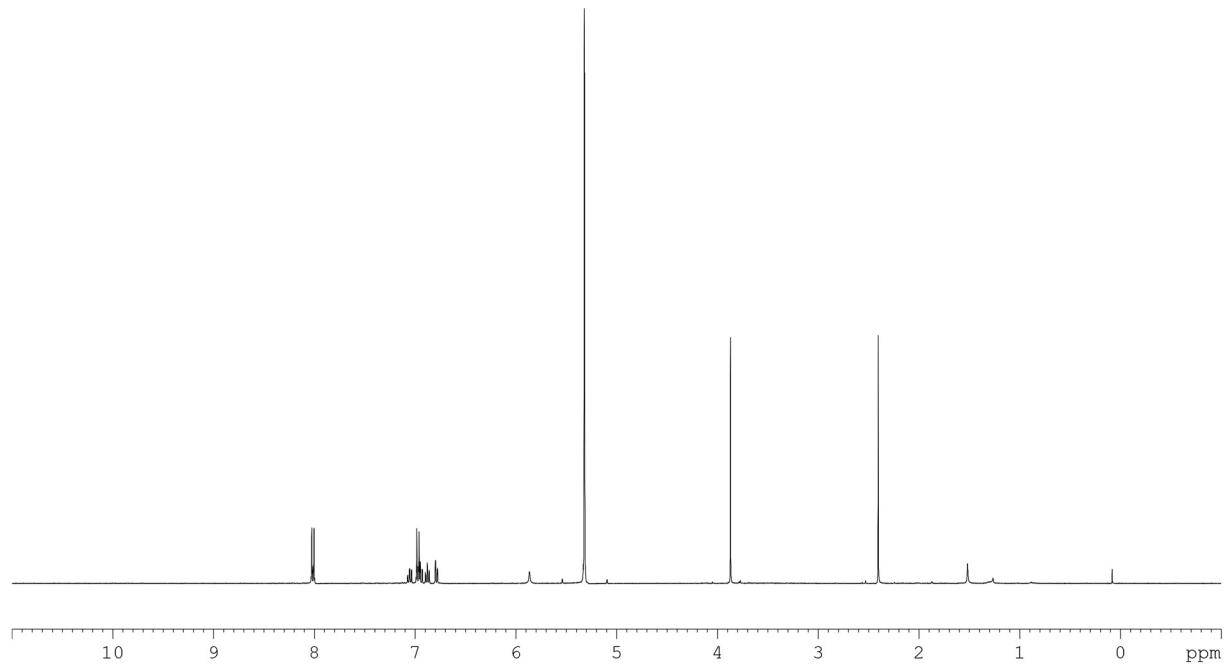
**(E)-2-((1-(4-methoxyphenyl)ethylidene)amino)phenol (2b)**



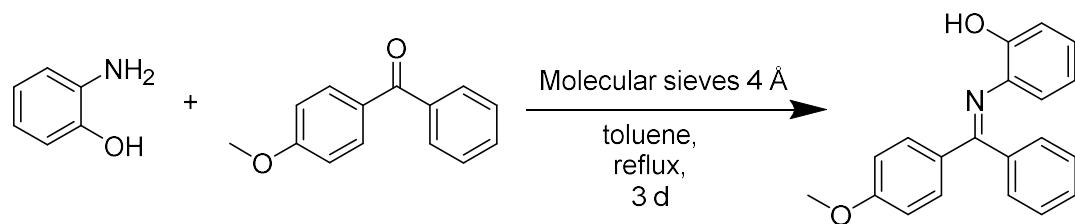
The molecular sieves 4 Å (approx. 4 g) were weighed into a 50 mL Schlenk flask and dried with a heat gun at 350 °C for 30 min under reduced pressure. 2-Aminophenol (30.0 mmol, 3.27 g, 3.0 eq) and 1-(4-methoxyphenyl)ethan-1-one (10 mmol, 1.50 g, 1.0 eq.) were added under argon flow and dissolved in 20 mL anhydrous toluene. Under argon flow, a reflux condenser was added to the setup. After flushing it for 3 min with argon, a drying tube containing CaCl<sub>2</sub> was added to the setup and the solution was refluxed for 20 h. The reaction mixture was allowed to cool down and the solvent was removed under reduced pressure. The crude product was obtained as an orange solid and was purified two times using a bulb-to-bulb distillation (0.01 mbar, 170 °C) to give the imine **2b** as an orange solid (0.313 g, 1.30 mmol, 13 %).

<sup>1</sup>H-NMR: (400.1 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ<sub>H</sub> = 8.02 (m, 2H), 7.06 (m, 1H), 6.97 (m, 2H), 6.94 (m, 1H, <sup>3</sup>J = 8.1 Hz, <sup>4</sup>J = 1.3 Hz), 6.88 (td, 1H, <sup>3</sup>J = 7.6 Hz, <sup>4</sup>J = 1.5 Hz), 6.79 (dd, 1H, <sup>3</sup>J = 7.8 Hz, <sup>4</sup>J = 1.5 Hz), 5.86 (s, 1H), 3.87 (s, 3H), 2.40 (s, 3H).

**HR-MS** (EI, *m/z*): found 240.10166 (M-H)<sup>+</sup> (calculated 240.10191 for C<sub>15</sub>H<sub>14</sub>NO<sub>2</sub> ); Diff(ppm) = -1.04.



**(E)-2-(((4-methoxyphenyl)(phenyl)methylene)amino)phenol (2c)**

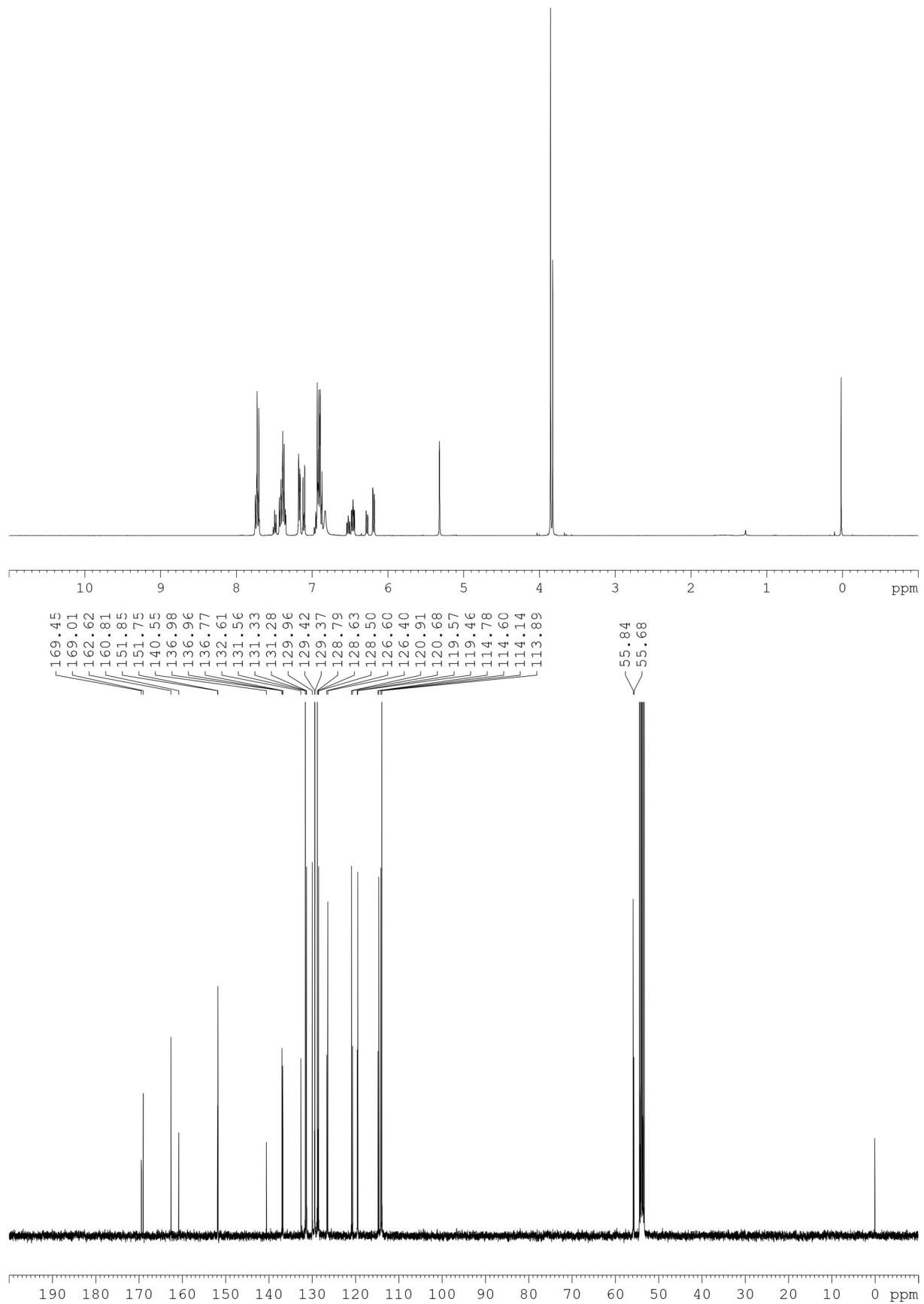


The molecular sieves 4 Å (approx. 4 g) were weighed into a 50 mL Schlenk flask and dried at 350 °C under reduced pressure for 30 min. 2-Aminophenol (30.0 mmol, 3.27 g, 3.0 eq) and (4-methoxy-phenyl)(phenyl)methanone (10 mmol, 2.12 g, 1.0 eq.) were added under argon flow and dissolved in 20 mL anhydrous toluene. Under argon flow, a reflux condenser was added to the setup and flushed with argon for 3 min. A drying tube filled with CaCl<sub>2</sub> was added to the setup and the solution was refluxed for 3 d. After cooling down, the mixture was filtrated, and the solvent was removed under reduced pressure to give an orange solid. The crude product was purified two times using a bulb-to-bulb distillation (0.01 mbar, 190 °C) to give the imine **2c** as an orange solid (0.516 g, 1.7 mmol, 17 %) as a mixture of *E* and *Z*-isomer (ratio major/minor = 1.9 : 1).

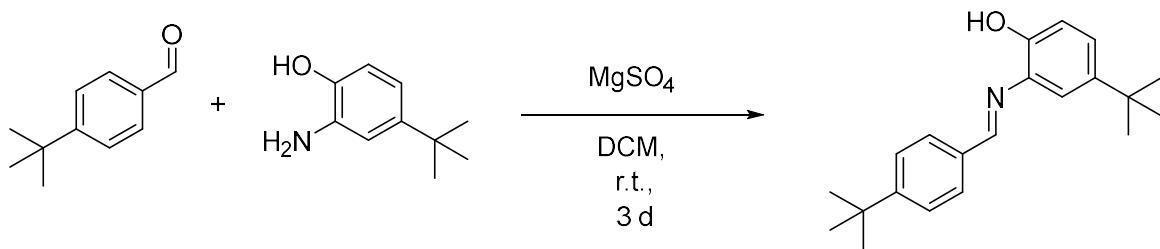
<sup>1</sup>H-NMR: (400.1 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ<sub>H</sub> = 7.74 (m, 2H, minor), 7.72 (m, 2H, major), 7.49 (m, 1H, minor), 7.44 – 7.33 (m, both), 7.17 (m, 2H, major), 7.11 (m, 2H, minor), 6.99 - 6.86 (m, both), 6.83 (s, 1H, broad, both, OH group), 6.52 (m, 1H, minor), 6.45 (m, 1H, major), 6.26 (dd, 1H, <sup>3</sup>J = 8.0 Hz, <sup>4</sup>J = 1.4 Hz minor), 6.17 (dd, 1H, <sup>3</sup>J = 8.0 Hz, <sup>4</sup>J = 1.4 Hz, major), 3.84 (s, 3H, major), 3.81 (s, 3H, minor).

<sup>13</sup>C-NMR: (100.6 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ<sub>C</sub> = 169.4 (minor), 169.0 (major), 162.6 (major), 160.8 (minor), 151.9 (minor), 151.7 (major), 140.6, 140.0, 136.9, 136.8, 132.6, 131.6, 131.3, 131.3, 130.0, 129.4 (minor), 129.4 (major), 128.8 (major), 128.6, 128.5 (minor), 126.6 (minor), 126.4 (major), 120.9 (major), 120.7 (minor), 119.6 (minor), 119.5 (major), 114.8 (minor), 114.6 (major), 114.1 (minor), 113.9 (major), 55.8 (major), 55.7 (minor).

**HR-MS (EI, *m/z*):** found 302.11714 (M-H)<sup>+</sup> (calculated 302.11756 for C<sub>20</sub>H<sub>16</sub>NO<sub>2</sub> ); Diff(ppm) = -1.36.



**(E)-4-(tert-butyl)-2-((4-(tert-butyl)benzylidene)amino)phenol (3a)**

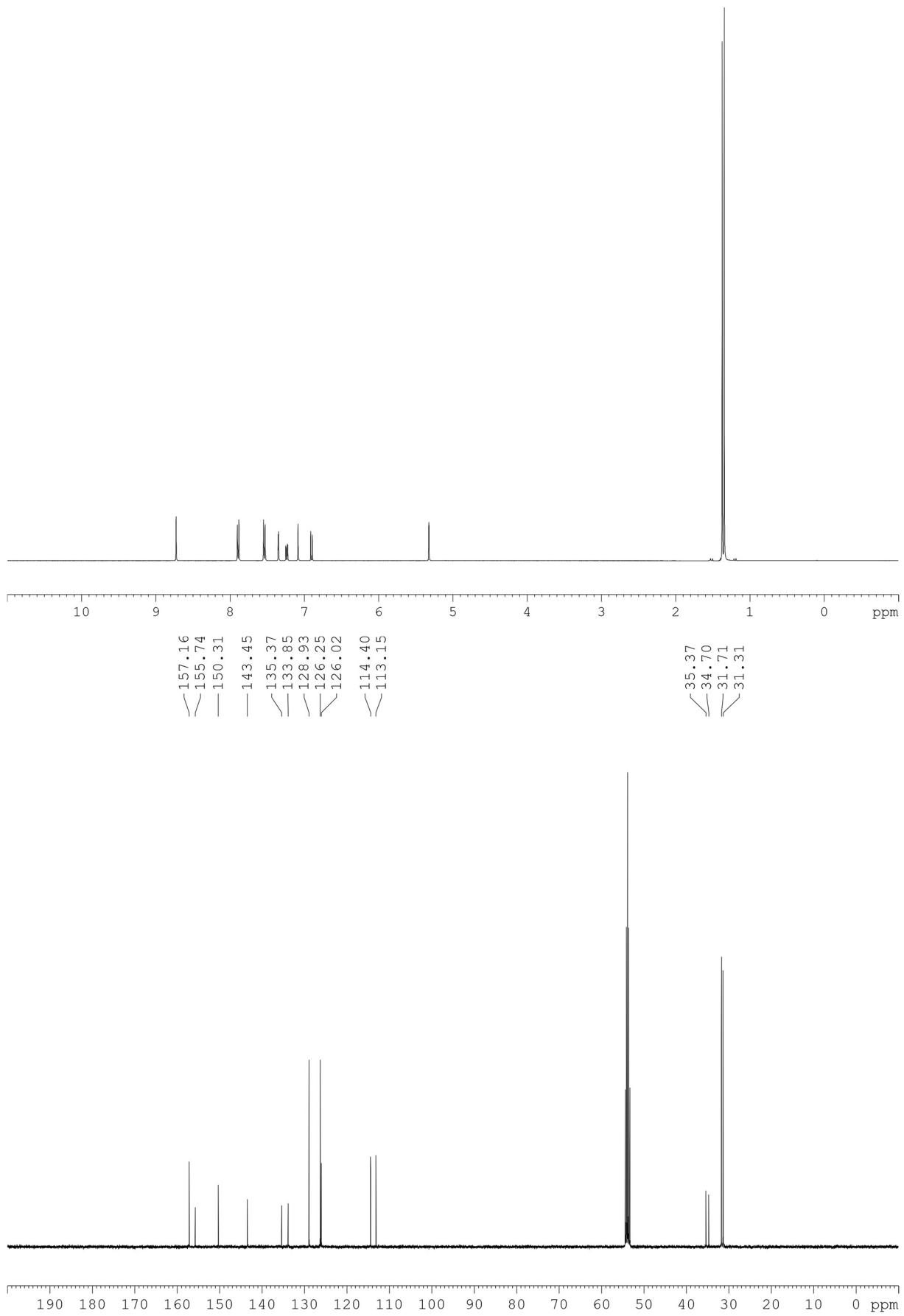


The aldehyde (0.811 g, 0.836 mL, 5.0 mmol, 1.0 eq.) and the hydroxy aniline (0.826 g, 5.0 mmol, 1.0 eq.) and MgSO<sub>4</sub> (5 g) were dissolved in 20 mL DCM and stirred for 3 days at room temperature. Afterwards, MgSO<sub>4</sub> was filtrated off and the solvent was removed under reduced pressure. The crude product was purified by bulb-to-bulb distillation (160 °C, 0.1 mbar) to give imine **3a** (0.883 mg, 2.85 mmol, 57%) as yellow oil which solidified as yellow to white solid after several days/weeks.

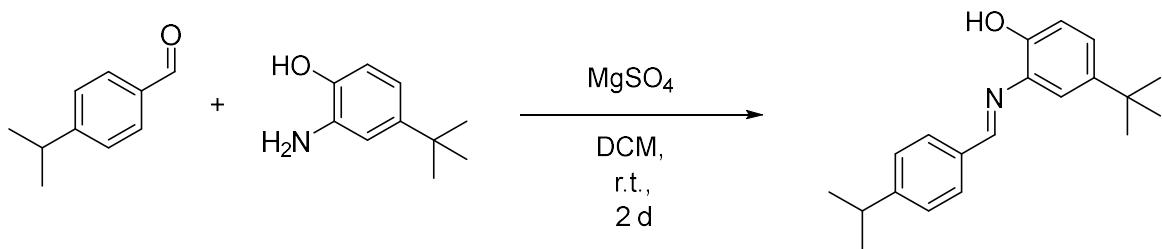
<sup>1</sup>H-NMR: (400.1 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ<sub>H</sub> = 8.72 (s, 1H), 7.89 (m, 2H), 7.54 (m, 2H), 7.35 (d, 1H, <sup>4</sup>J = 2.3 Hz), 7.23 (dd, 1H, <sup>3</sup>J = 8.5 Hz, <sup>4</sup>J = 2.3 Hz), 7.08 (s, 1H), 6.9 (d, 1H, <sup>3</sup>J = 8.5 Hz), 1.37 (s, 9H), 1.34 (s, 9H).

<sup>13</sup>C-NMR: (100.6 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ<sub>C</sub> = 157.2, 155.7, 150.3, 143.5, 135.4, 133.8, 128.9, 126.3, 126.0, 114.4, 113.1, 35.4, 34.7, 31.7, 31.3.

**HR-MS** (EI, *m/z*): found 308.20058 (M-H)<sup>+</sup> (calculated 308.20089 for C<sub>21</sub>H<sub>16</sub>NO); Diff(ppm) = - 1.02.



**(E)-4-(tert-butyl)-2-((4-isopropylbenzylidene)amino)phenol (3b)**

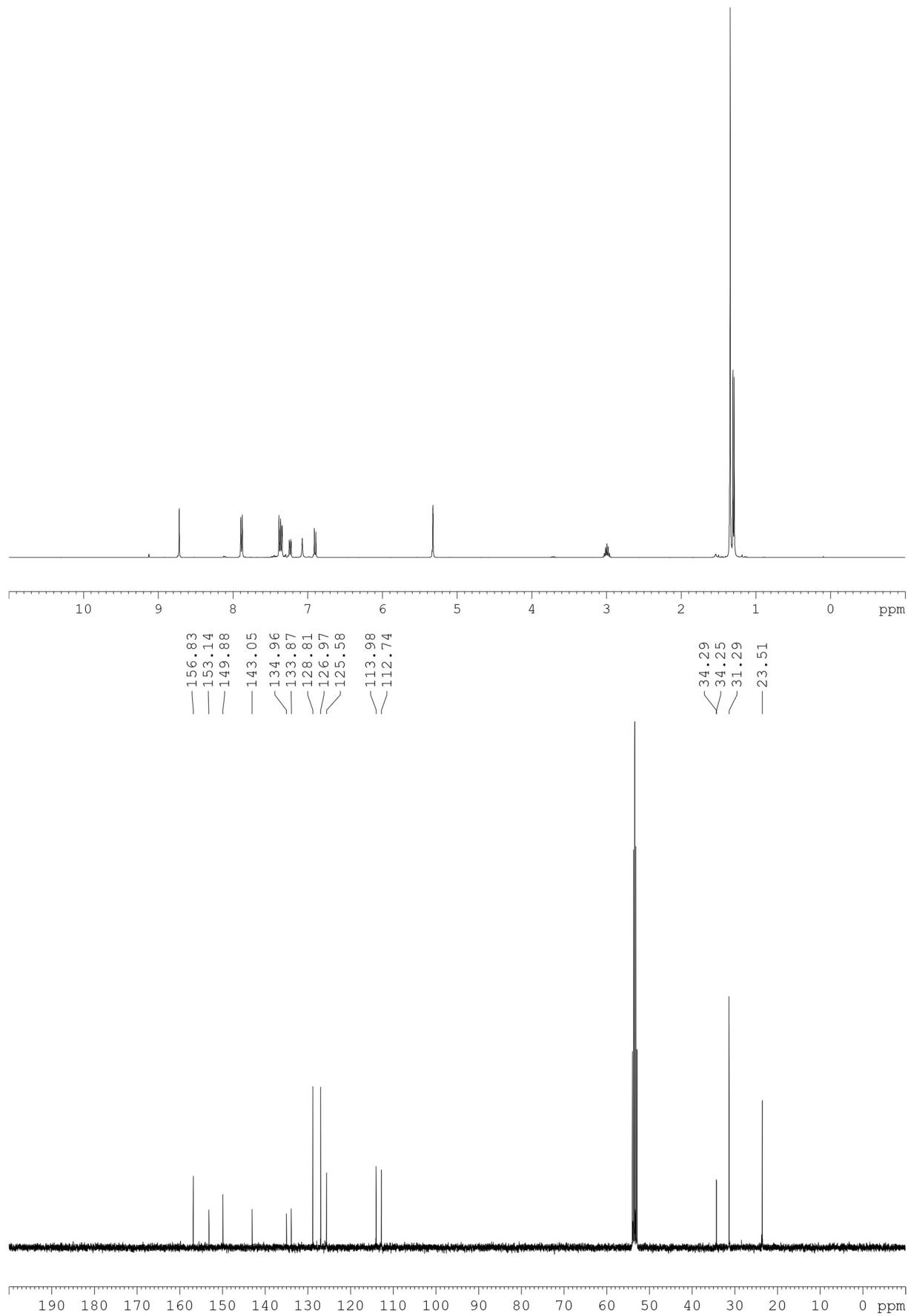


The aldehyde (1.48 g, 1.52 mL, 10.0 mmol, 1.0 eq.) and the hydroxy aniline (1.65 g, 10.0 mmol, 1.0 eq.) and MgSO<sub>4</sub> (6 g) were dissolved in 20 mL dry DCM and stirred for 2 days at room temperature. Afterwards, MgSO<sub>4</sub> was filtrated off and the solvent was removed under reduced pressure. The crude product was purified by bulb-to-bulb distillation (180 °C, 0.1 mbar) to give imine **3b** (0.93 g, 3.2 mmol, 32 %) as a yellow oil.

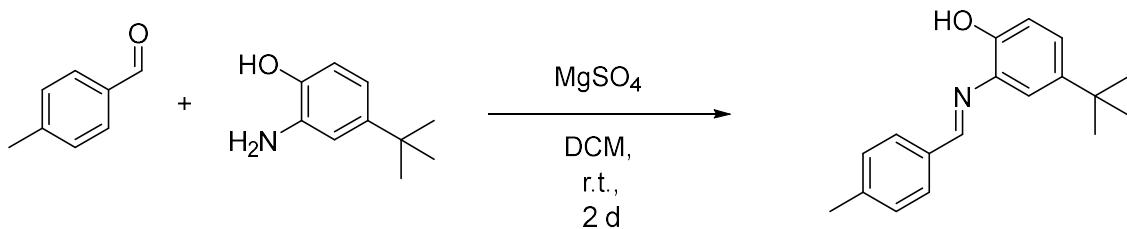
<sup>1</sup>H-NMR: (400.1 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ<sub>H</sub> = 8.72 (s, 1H), 7.89 (m, 2H), 7.37 (m, 2H), 7.35 (d, <sup>4</sup>J = 2.3 Hz), 7.23 (dd, 1H, <sup>3</sup>J = 8.4 Hz, <sup>4</sup>J = 2.3 Hz), 7.07 (s, 1H), 6.90 (d, 1H, <sup>3</sup>J = 8.4 Hz), 3.00 (sept, 1H, <sup>3</sup>J = 6.9 Hz), 1.34 (s, 9H), 1.30 (d, 6H, <sup>3</sup>J = 6.9 Hz).

<sup>13</sup>C-NMR: (100.6 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ<sub>C</sub> = 156.8, 153.1, 149.9, 143.0, 135.0, 133.9, 128.8, 127.0, 125.6, 114.0, 112.7, 34.3, 34.2, 31.3, 23.5.

**HR-MS (EI, *m/z*):** found 295.19322 (M)<sup>+</sup> (calculated 295.19307 for C<sub>20</sub>H<sub>25</sub>NO); Diff(ppm) = + 0.53.



**(E)-4-(tert-butyl)-2-((4-methylbenzylidene)amino)phenol (3c)**

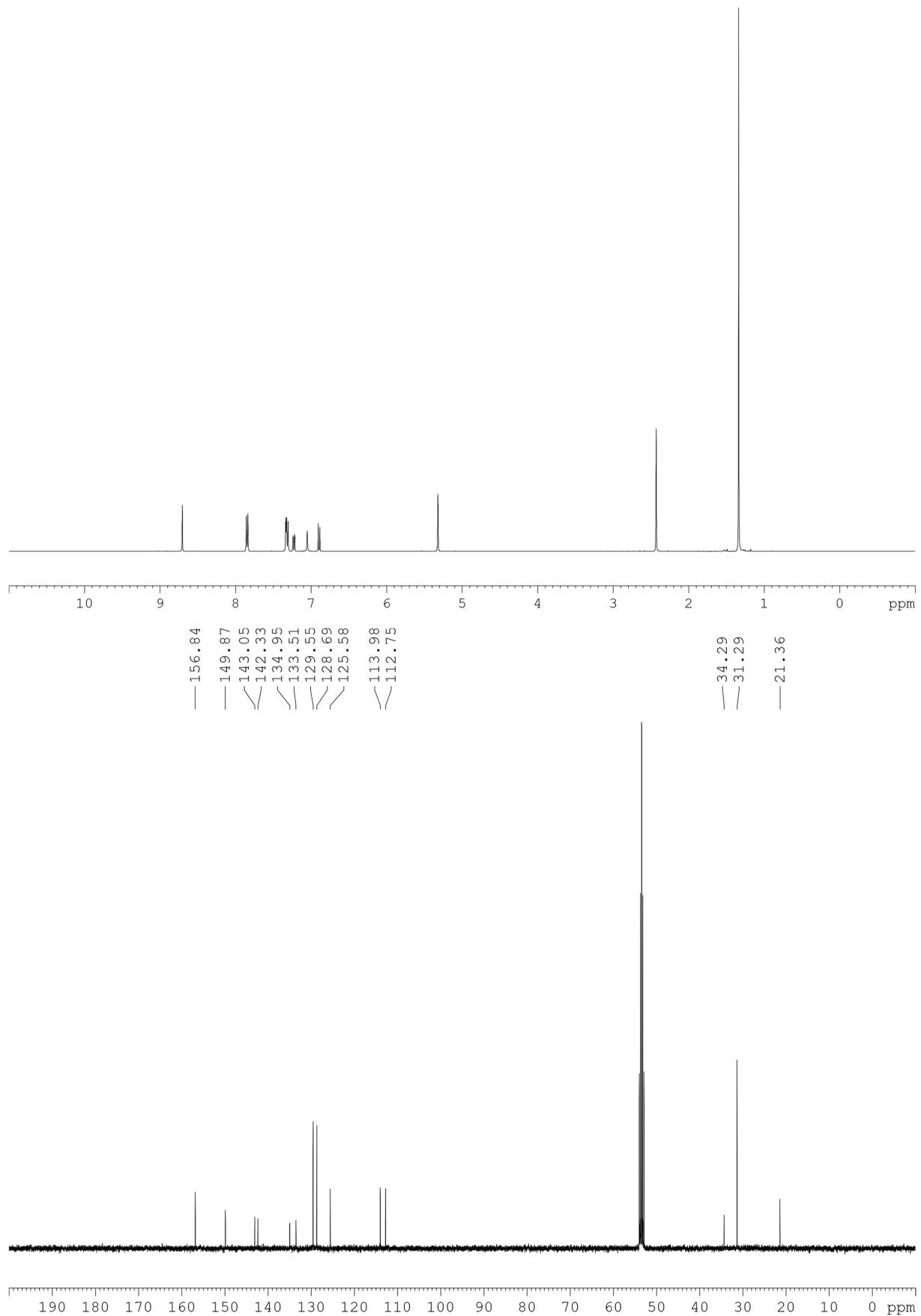


The aldehyde (1.20 g, 1.02 mL, 10.0 mmol, 1.0 eq.) and the hydroxy aniline (1.65 g, 10.0 mmol, 1.0 eq.) and  $\text{MgSO}_4$  (7 g) were dissolved in 20 mL dry  $\text{DCM}$  and stirred for 2 days at room temperature. Afterwards,  $\text{MgSO}_4$  was filtrated off and the solvent was removed under reduced pressure. The crude product was purified by bulb-to-bulb distillation ( $175\text{ }^\circ\text{C}$ , 0.1 mbar) to give imine **3c** (1.76 g, 6.0 mmol, 60 %) as a yellow oil.

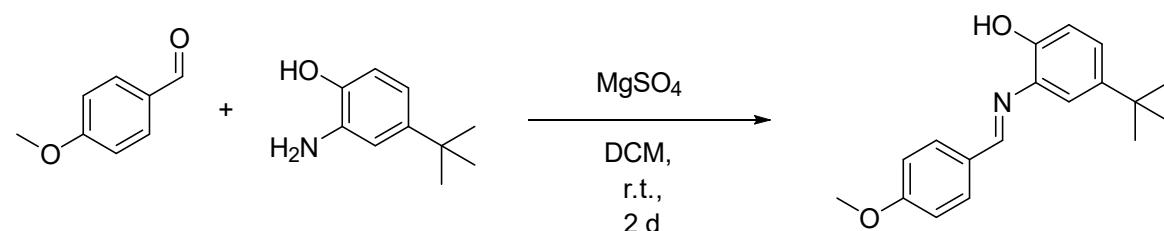
$^1\text{H-NMR}$ : (400.1 MHz,  $\text{CD}_2\text{Cl}_2$ ):  $\delta_{\text{H}} = 8.71$  (s, 1H), 7.85 (m, 2H), 7.34 (d, 1H,  $^4\text{J} = 2.3$  Hz), 7.32 (d, 2H,  $^3\text{J} = 7.9$  Hz), 7.23 (dd,  $^3\text{J} = 8.4$  Hz,  $^4\text{J} = 2.3$  Hz), 7.05 (s, 1H), 6.90 (d, 1H,  $^3\text{J} = 8.4$  Hz), 2.43 (s, 3H), 1.34 (s, 9H).

$^{13}\text{C-NMR}$ : (100.6 MHz,  $\text{CD}_2\text{Cl}_2$ ):  $\delta_{\text{C}} = 156.8, 149.9, 143.1, 142.3, 134.9, 133.5, 129.5, 128.7, 125.6, 114.0, 112.8, 34.3, 31.3, 21.4$ .

**HR-MS** (EI,  $m/z$ ): found 266.15369 ( $\text{M-H}^+$ ) (calculated 266.15394 for  $\text{C}_{18}\text{H}_{20}\text{NO}$ ); Diff(ppm) = -0.95.

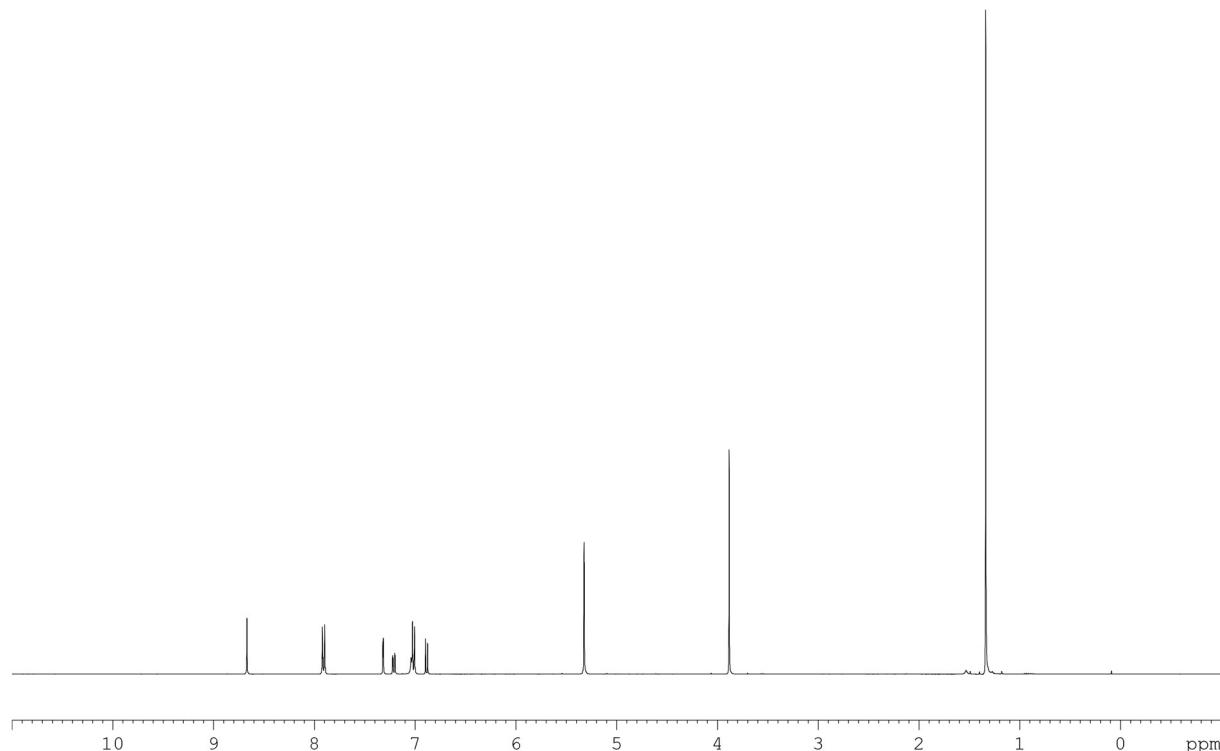


**(E)-4-(tert-butyl)-2-((4-methoxybenzylidene)amino)phenol (3d)**

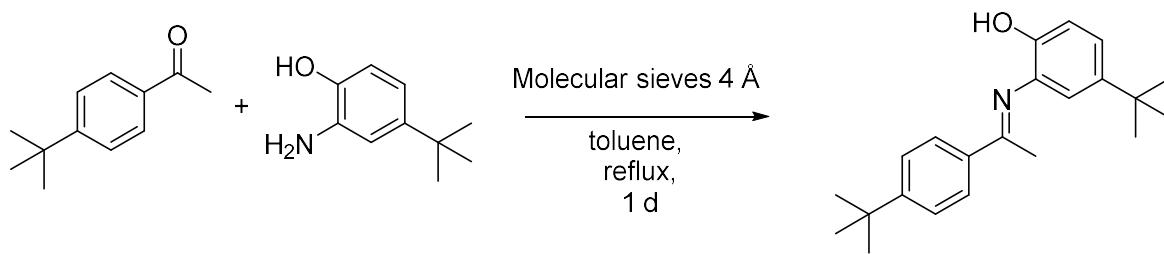


The aldehyde (0.68 g, 0.55 mL, 5.0 mmol, 1.0 eq.) and the hydroxy aniline (0.83 g, 5.0 mmol, 1.0 eq.) and MgSO<sub>4</sub> (7 g) were dissolved in 20 mL dry DCM and stirred for 2 days at room temperature. Afterwards, MgSO<sub>4</sub> was filtrated off and the solvent was removed under reduced pressure. The crude product was recrystallized in methanol to give imine **3d** (0.55 g, 2.4 mmol, 48 %) as a pale orange solid.

<sup>1</sup>H-NMR: (400.1 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ<sub>H</sub> = 8.71 (s, 1H), 7.85 (m, 2H), 7.34 (d, 1H, <sup>4</sup>J = 2.3 Hz), 7.32 (d, 2H, <sup>4</sup>J = 7.9 Hz), 7.23 (dd, 1H, <sup>3</sup>J = 8.6 Hz, <sup>4</sup>J = 2.3 Hz), 7.05 (s, 1H), 6.90 (d, 1H, <sup>3</sup>J = 8.6 Hz), 2.43 (s, 3H), 1.34 (s, 9H).



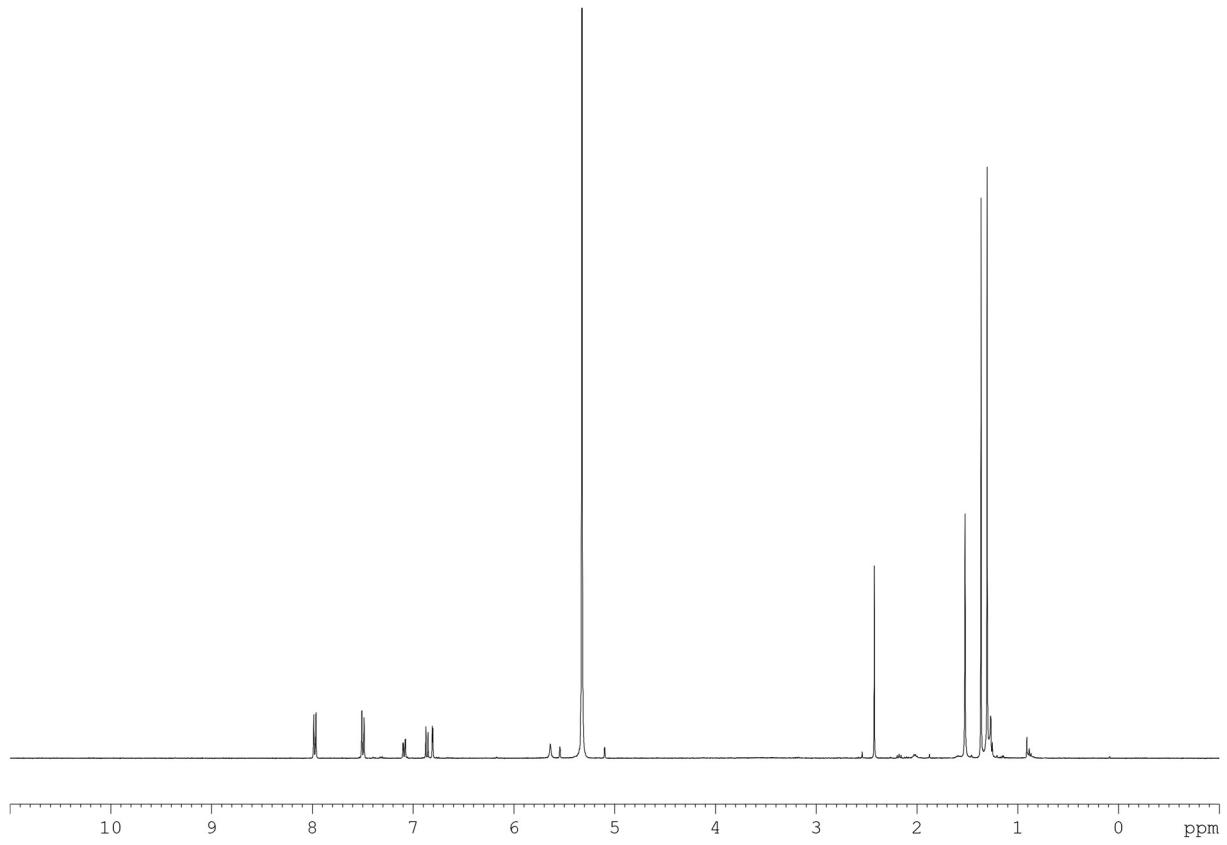
**(E)-4-(tert-butyl)-2-((1-(4-(tert-butyl)phenyl)ethylidene)amino)phenol (3e)**



The molecular sieves 4 Å (approx. 4 g) were weighed into a 50 mL Schlenk flask and dried with a heat gun at 350 °C for 30 min under reduced pressure. The aminophenol (1.65 g, 10.0 mmol, 3.0 eq.) and the ketone (0.881 g, 5 mmol, 1.0 eq.) were added under argon flow and dissolved in 20 mL anhydrous toluene. Under argon flow, a reflux condenser was added to the setup. After flushing it for 3 min with argon, a drying tube containing CaCl<sub>2</sub> was added to the setup and the solution was refluxed for 1 d. The reaction mixture was allowed to cool down and the solvent was removed under reduced pressure. The crude product was obtained as an dark brown solution and purified by recrystallization in Et<sub>2</sub>O/MeOH (4:1) to give the imine **3e** (0.108 g, 0.33 mmol, 6.5%) as a colorless powder.

<sup>1</sup>H-NMR: (400.1 MHz, CD<sub>2</sub>Cl<sub>2</sub>): δ<sub>H</sub> = 7.98 (m, 2H), 7.50 (m, 2H), 7.09 (dd, 1H, <sup>3</sup>J = 8.4 Hz, <sup>4</sup>J = 2.3 Hz), 6.86 (d, 1H, <sup>3</sup>J = 8.4 Hz), 6.87 (d, 1H, <sup>4</sup>J = 2.3 Hz), 5.64 (s, 1H), 2.43 (s, 3H), 1.36 (s, 9H), 1.30 (s, 9H).

**HR-MS** (EI, *m/z*): found 322.21654 (M-H)+ (calculated 322.21654 for C<sub>18</sub>H<sub>20</sub>NO); Diff(ppm) = -0.01.



### 3. Diffusion ordered spectroscopy (DOSY)

DOSY measurements were performed with the convection suppressing DSTE (double stimulated echo) pulse sequence developed by Jerschow and Müller in a pseudo 2D mode.<sup>4</sup> Smoothed square (SMSQ10.100) gradient shapes and a linear gradient ramp with 20 increments between 5% and 95% of the maximum gradient strength (5.35 G/mm) were used. The diffusion time delay was set to 45 ms. For the homospoil gradient strengths, values of 100, -13.17, 20 and -17.13 % were used. Gradient pulse lengths (p16) were first optimized to obtain a sigmoidal signal decay for increasing gradient strength (3.0 ms for TMS, 6.0 – 6.5 ms for CPA/imine complexes). NMR spectra were processed with Bruker TopSpin 3.2 (T1/T2 relaxation package) and diffusion coefficients were derived according to Jerschow and Müller<sup>4</sup>. Tetramethylsilan was added to the samples to reference chemical shifts and the viscosity of each sample.

The molecular radii were derived by the Stokes-Einstein equation<sup>5</sup> using Chens correction.<sup>6</sup>

$$D_i = \frac{k_B T}{6\pi\eta r_H} * (1 + 0.695 * \left(\frac{r_{solv}}{r_H}\right)^{2.234})$$

$D_i$  is the self-diffusion coefficient derived by the measurement,  $\eta$  is the viscosity of the solvent,  $r_H$  is the hydrodynamic radius of the observed molecule and  $r_{solv}$  the radius of the solvent. No form factor correction was applied. The viscosity was determined by measuring the diffusion coefficient of the reference tetramethylsilane (TMS) and solving the equation for  $\eta$  with the literature value<sup>7</sup> of the radius of 2.96 Å.

To estimate the error, diffusion coefficients for different signals of the same complex were averaged and the standard deviation was determined. The molecular radii were calculated based on the averaged diffusion coefficient, the averaged diffusion coefficient plus the standard deviation and the averaged diffusion coefficient minus the standard deviation. The resulting radii are given as the radius derived from the averaged diffusion coefficient and the error range is given by:

$$\text{Error range} = \frac{1}{2} * [(r_i^{aver} - r_i^{aver+StDev}) + (r_i^{aver-StDev} - r_i^{aver})]$$

Where  $r_i^{aver}$  is the radius derived from the averaged diffusion coefficient,  $r_i^{aver+StDev}$  is the minimum radius derived from the averaged diffusion coefficient plus the standard deviation

and  $r_i^{aver-StDe}$  is the maximum radius derived from the averaged diffusion coefficient minus the standard deviation.

**Table S1:** Measured self diffusion coefficients of **1a/2a** at a concentration of 50 mM at 180 K in CD<sub>2</sub>Cl<sub>2</sub> for signals 1-5 which were unambiguously assigned to one of the three different species A, B or C (see chemical shift assignment Figure S1). The determined coefficients are similar for all three species within the precision of the measurement (signal intensities were low).

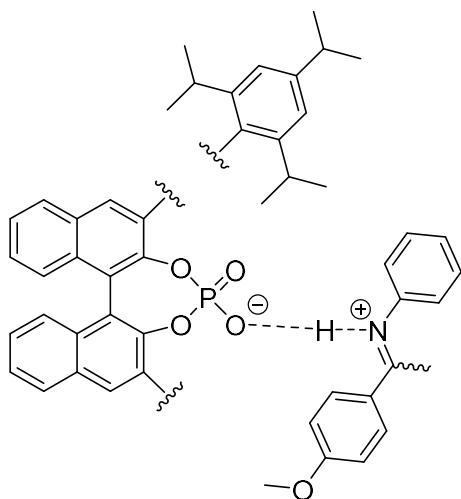
Entry	Chem. Shift [ppm]	Assignment	Self diffusion coefficient [m <sup>2</sup> /s]
1	9.58	Species C	3.21E-11
2	8.63	Species A	3.72E-11
3	8.57	Species B	3.55E-11
4	6.69	Species B	3.10E-11
5	3.55	Species C	3.09E-11
6	0.00	TMS	1.90E-10

The self diffusion coefficients determined for signals 1-5 are similar within the error of the experiment, considering the low signal intensities. This clearly shows, that all three species are similar in size, i.e. [CPA/imine]<sub>2</sub> dimers. An average hydrodynamic radius of 11.8 ± 0.87 Å was derived based on the averaged self diffusion coefficient of entries 1-5.

As a reference for monomeric CPA/imine systems, the system depicted below was used. The chemical shift assignment and structural investigations on this system were done in our previous research.<sup>8</sup> As probe signals, one signal of the CPA for the CPA/E and CPA/Z complex, as well as one signal of the imine for the CPA/E and CPA/Z complex was selected. All signals were similar and based on the averaged self diffusion coefficient, the average hydrodynamic radius of 8.8 ± 0.15 Å was determined.

**Table S2:** Measured self diffusion coefficients of the monomeric reference system (see below) at a concentration of 50 mM at 180 K in CD<sub>2</sub>Cl<sub>2</sub>.

Entry	Chem. Shift [ppm]	Assignment	Self diffusion coefficient [m <sup>2</sup> /s]
1	7.88	CPA/E (CPA)	4.76E-11
2	7.75	CPA/Z (CPA)	4.93E-11
3	3.80	CPA/Z (Imin)	4.98E-11
4	2.54	CPA/E (Imin)	4.81E-11
5	0.00	TMS	2.02E-10



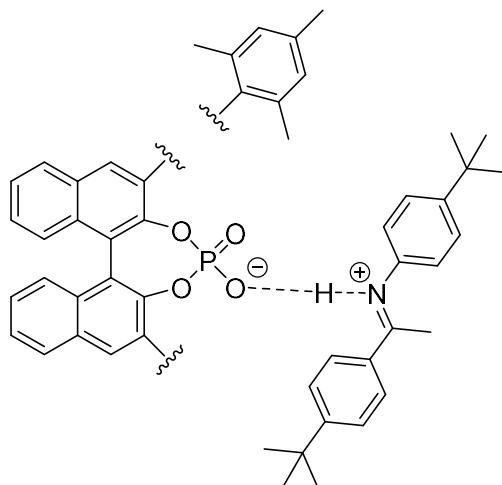
**Table S3:** Measured self diffusion coefficients of **1b/3a** at a concentration of 25 mM at 180 K in CD<sub>2</sub>Cl<sub>2</sub>.

Entry	Chem. Shift [ppm]	Assignment	Self diffusion coefficient [m <sup>2</sup> /s]
1	8.81	Imine	4.08E-11
2	8.03	CPA	3.90E-11
3	7.92	CPA	3.96E-11
4	6.29	CPA	3.86E-11
5	5.90	CPA	3.87E-11
6	1.44	Imine	3.88E-11
7	0.98	Imine	3.90E-11
8	0.00	TMS	2.06E-10

The self diffusion coefficients of entries 1, 6 and 7 (imine) are identical to the ones of the CPA (2-5), showing that both molecules are part of the same complex. Hence, the self diffusion coefficients were averaged and a hydrodynamic radius of 10.93 ± 0.15 Å was determined.

**Table S4:** Measured self diffusion coefficients of the monomeric reference system depicted below at a concentration of 100 mM and a 1:1 ratio at 180 K in CD<sub>2</sub>Cl<sub>2</sub>.

Entry	Chem. Shift [ppm]	Assignment	Self diffusion coefficient [m <sup>2</sup> /s]
1	7.77	CPA/E (CPA)	4.00E-11
2	2.64	CPA/E (Imine)	4.16E-11
3	1.16	CPA/E (Imine)	4.12E-11
4	0.00	TMS	1.99E-10



To put the found hydrodynamic radius of **1b/3a** (see Table S3) into relation, the reference system depicted above was selected to mimic the steric properties and hydrodynamic radius of a monomeric CPA/imine complex. The chemical shift analysis for this system was done previously<sup>9</sup> and it was also clarified that the CPA/imine complex is monomeric. Based on the averaged self-diffusion coefficients, a hydrodynamic radius of 10.2 ± 0.16 Å was derived. The obtained radius of the monomeric reference system is close to the one of **1b/3a**, validating that **1b/3a** is indeed a monomeric CPA/imine complex.

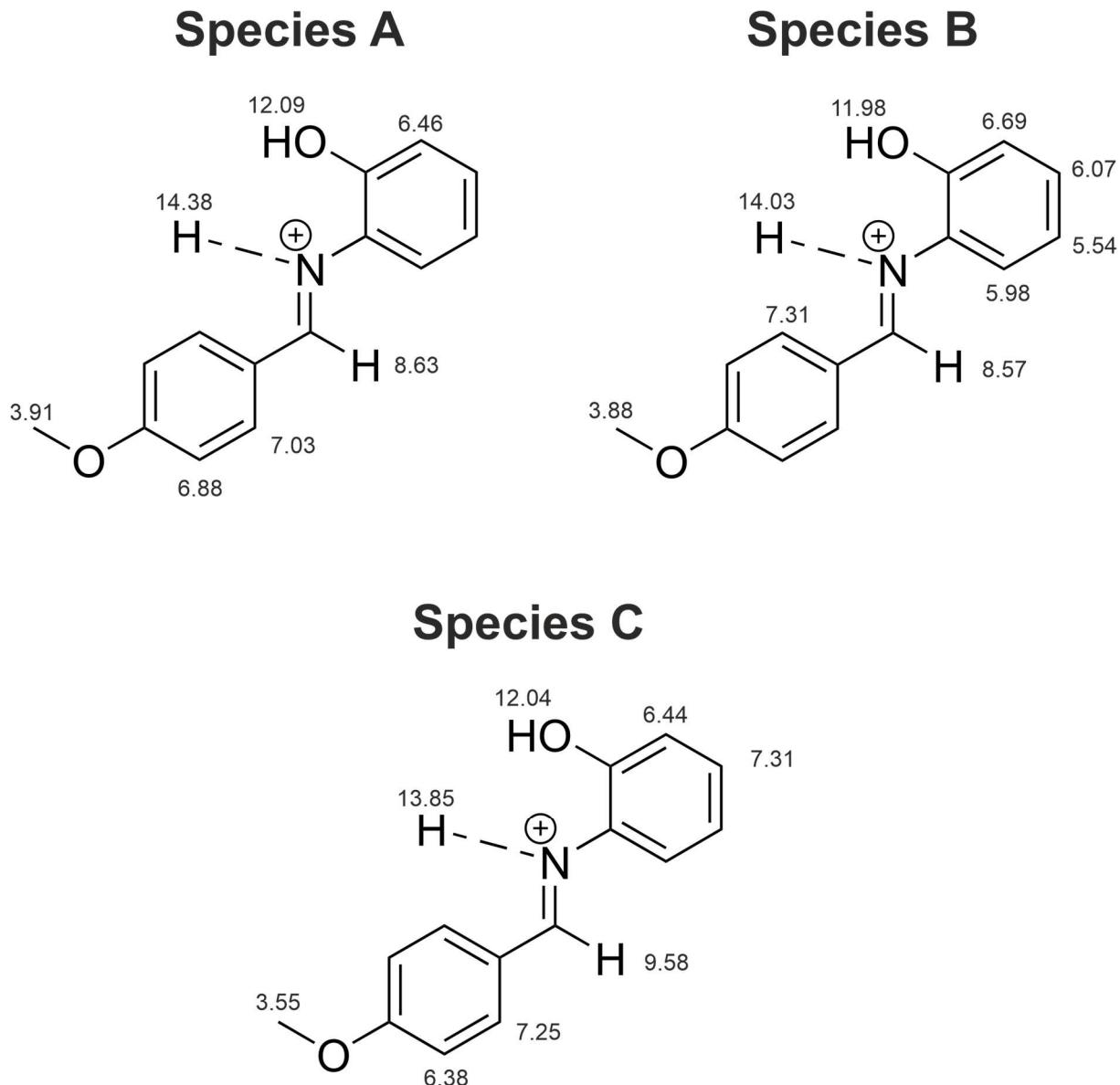
**Table S5:** Measured self diffusion coefficients of **1c/2a** at a 1:1 ratio and a concentration of 10 mM at 180 K in CD<sub>2</sub>Cl<sub>2</sub>.

Entry	Chem. Shift [ppm]	Assignment	Self diffusion coefficient [m <sup>2</sup> /s]
1	4.02	Imine	3.56E-11
2	0.00	TMS	2.13E-10

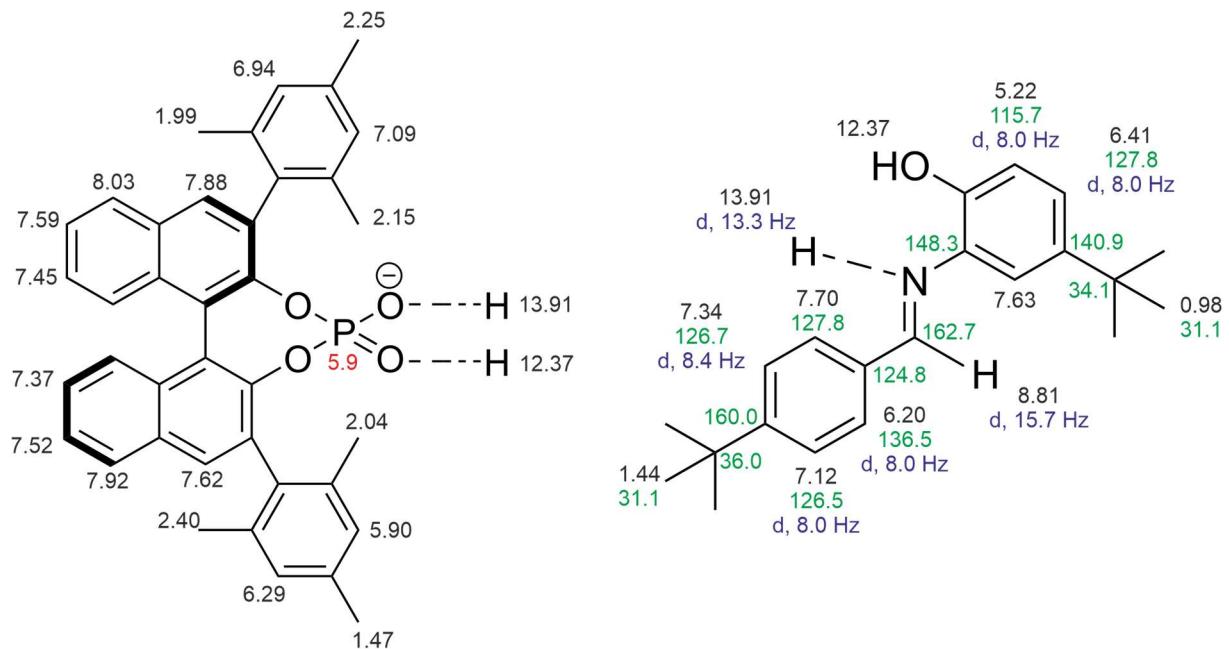
As a probe signal for the **1c/2a** complex, the methoxy group of **2a** was selected. Based on the self diffusion coefficient, a hydrodynamic radius of ~ 12.36 Å was determined.

#### 4. Chemical Shift assignments

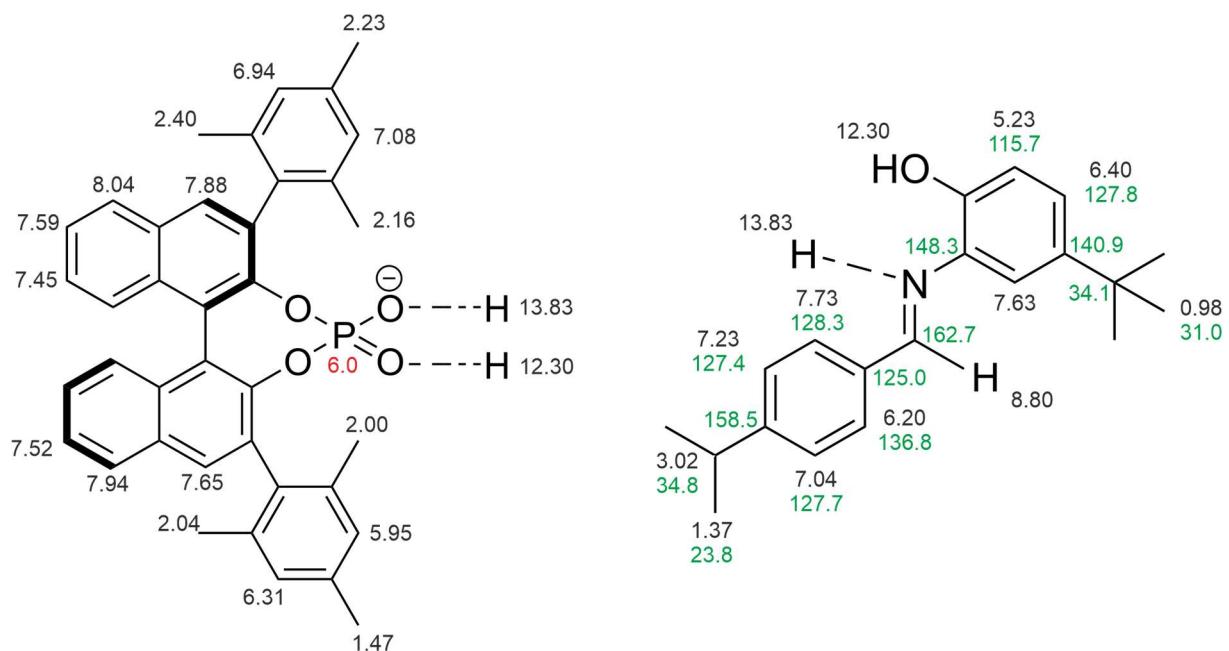
Chemical shifts of CPA/imine complexes were assigned using  $^1\text{H}$ - $^{13}\text{C}$ - $^{31}\text{P}$ -NMR,  $^1\text{H}$   $^{13}\text{C}$  HSQC,  $^1\text{H}$   $^{13}\text{C}$  HMBC,  $^1\text{H}$   $^{31}\text{P}$  HMBC,  $^1\text{H}$  COSY,  $^1\text{H}$  TOCSY and  $^1\text{H}$  NOESY spectra.  $^1\text{H}$  chemical shifts are given in black,  $^{13}\text{C}$  chemical shifts are given in green,  $^{31}\text{P}$  chemical shifts are given in red, multiplicities and coupling constants are given in blue.



**Figure S1:** Proton chemical shift assignment of **1a/2a** at a concentration of 50 mM in  $\text{CD}_2\text{Cl}_2$  at 180 K and 600 MHz.



**Figure S2:** Proton chemical shift assignment of **1b/3a** at a concentration of 25 mM in  $\text{CD}_2\text{Cl}_2$  at 180 K and 600 MHz.

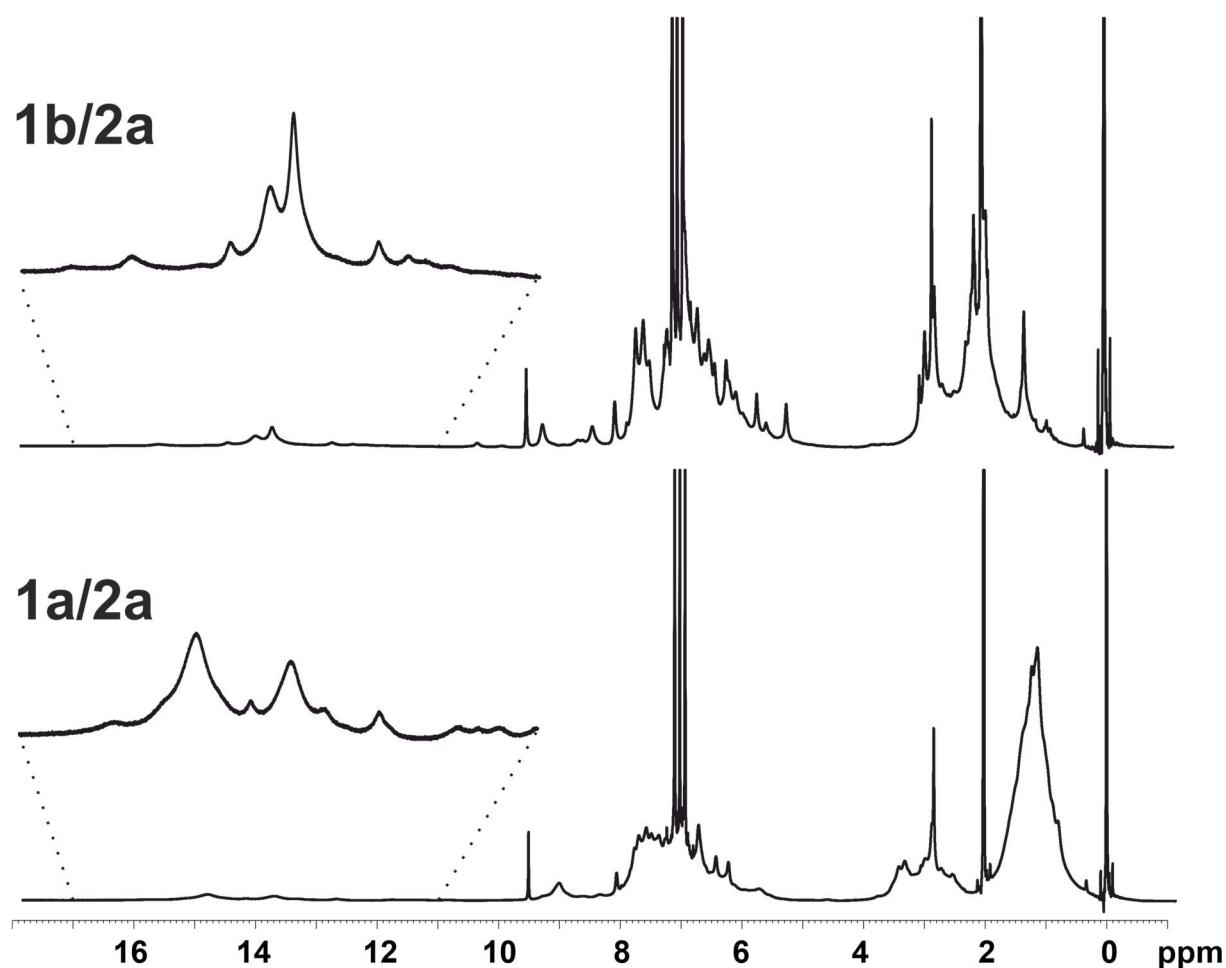


**Figure S3:** Proton chemical shift assignment of **1b/3b** at a concentration of 25 mM in  $\text{CD}_2\text{Cl}_2$  at 180 K and 600 MHz.

## 5. Initial system screening

### 5.1 Solvent screening

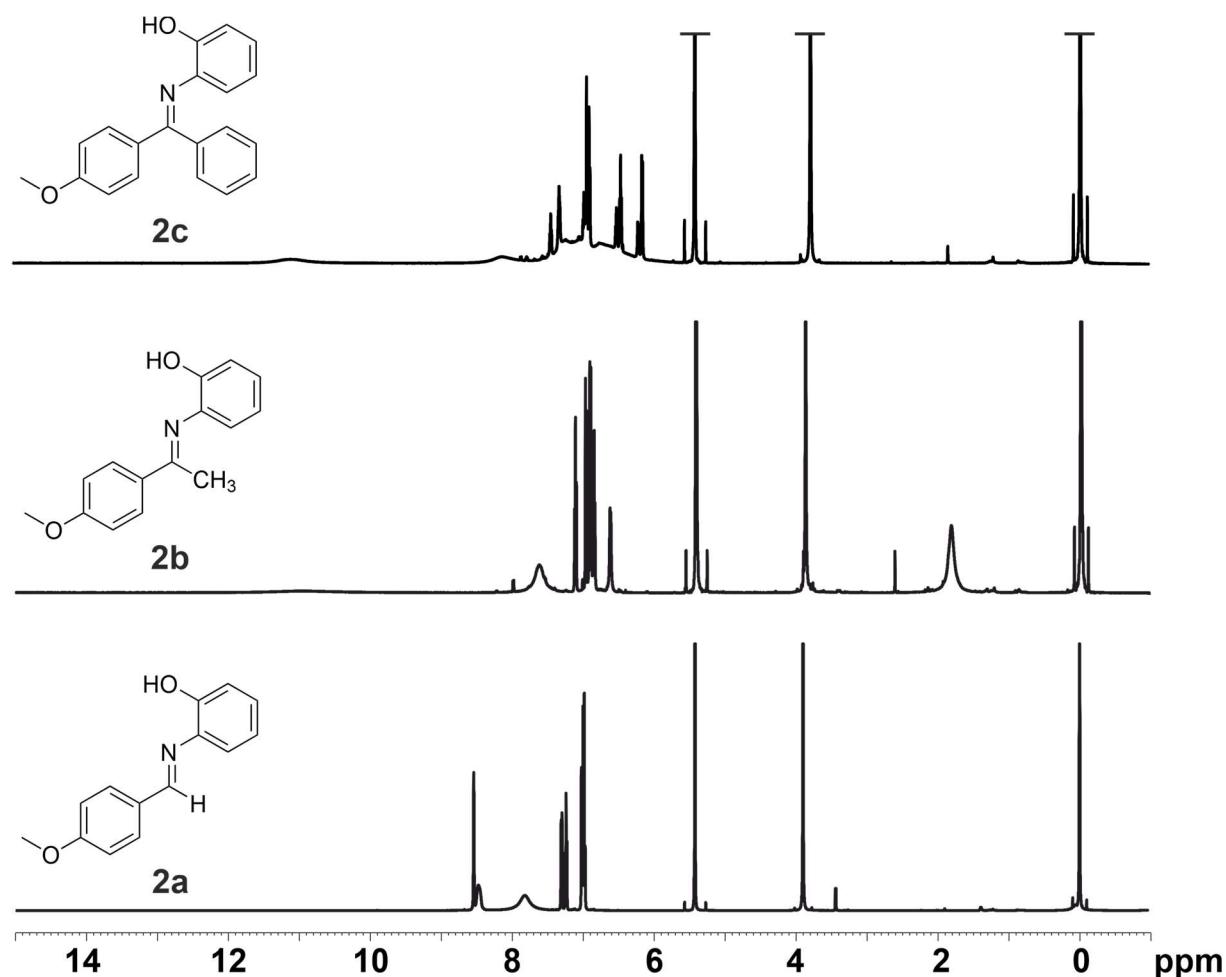
Initially, toluene-d<sub>8</sub> was tried as solvent to investigate the CPA/imine complexes, because in this solvent the best stereoselectivities were obtained in the optimization screening by Akiyama.<sup>10</sup> However, the measured spectra showed severe line broadening and signal overlap. This is similar to our previous NMR investigations, where toluene was not suitable as a solvent.<sup>11</sup> We assume, that changing the solvent from toluene to dichloromethane does not drastically change the structural space of the investigated CPA/imine systems. However, it should be noted that during the reaction optimization by Akiyama on the example of one N-(*ortho*-hydroxyphenyl) imine and one CPA, the enantioselectivity in toluene (87% ee) was significantly higher than in dichloromethane (13% ee).



**Figure S4:** <sup>1</sup>H NMR spectra of **1a/2a** and **1b/2a** in toluene-d<sub>8</sub> at a temperature of 180 K and 600 MHz.

## 5.2 Intramolecular hydrogen bonding

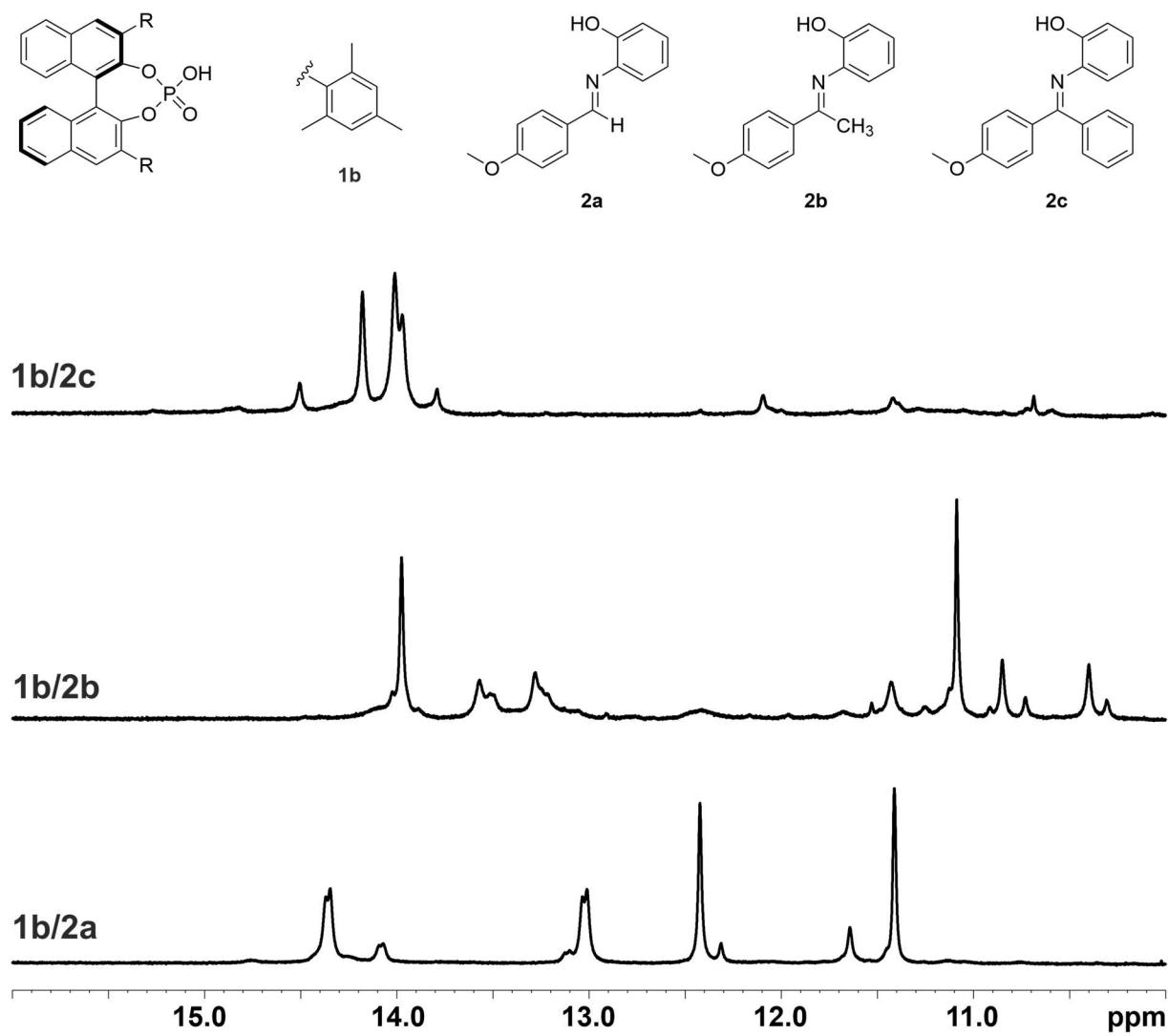
As imines **2a-c** have both a hydrogen bond donor (-OH) and acceptor (C=N), the formation of intra- and intermolecular hydrogen bonds is possible. To confirm, that all observed hydrogen bonded proton signals (see Figure 2 and Figure S6) stem from CPA/imine hydrogen bonds, the proton spectra of imines **2a-c** (see Figure S5) were recorded under analogous conditions (temperature, concentration, sample, sample preparation) as the CPA/imine samples. The resulting spectra clearly show, that none of the monitored hydrogen bonded proton signals for systems **1a-b/2a-c** originate from the imines alone.



**Figure S5:** <sup>1</sup>H NMR spectra of imines **2a** (25 mM), **2b** (10 mM) and **2c** (10 mM) in CD<sub>2</sub>Cl<sub>2</sub> at 180 K and 600 MHz (concentrations are analogous to the samples of **1a-b/2a-c**). For **2b** and **2c**, broad peaks at 11 ppm are observed which indicate the presence of inter- and/or intramolecular hydrogen bonds. However, these signals were not observed in the samples with the catalysts.

### 5.3. $^1\text{H}$ NMR spectra for systems **1b/2a-c**

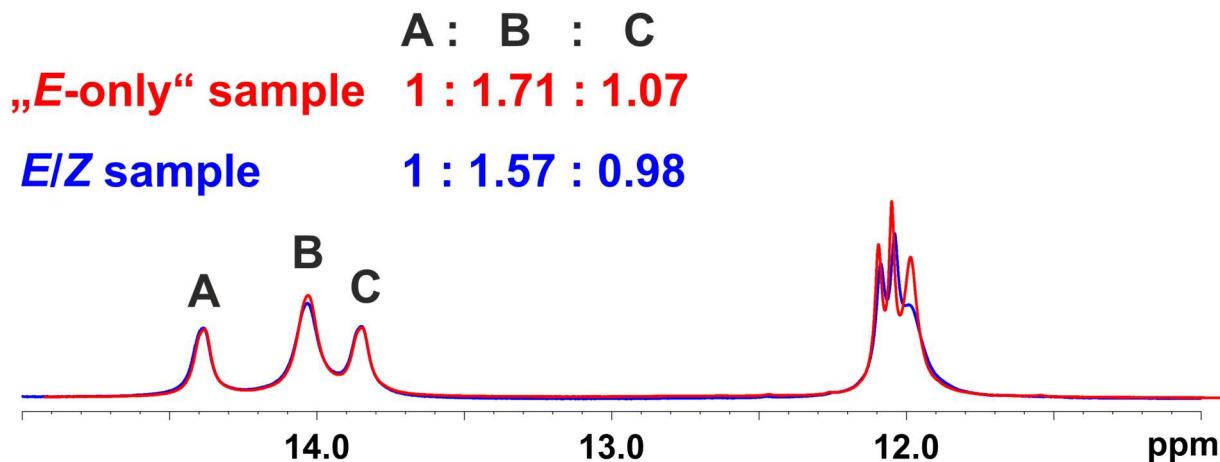
For samples containing catalyst **1b** and imines **2a-c**, the recorded  $^1\text{H}$  NMR spectra (see Figure S6) were similar to the ones with catalyst **1a** (see Figure 2). For imines **2b** and **2c** many hydrogen bonded proton signals were monitored, reflecting a broad structural space of the complexes. For imine **2a** however, 8 hydrogen bonded proton signal, reflecting 4 different complexes were observed. In contrast to **1a/2a** (see Figure 2), the set of  $\text{PO}^-$ ---H-N $^+$  hydrogen bonds as well as the set of PO---H-O hydrogen bonds have different chemical shifts within the set (2 H-bonds at 14-14.5 ppm, 2 H-bonds at 13 ppm for  $\text{PO}^-$ ---H-N $^+$ ; 2 H-bonds at 12.4 ppm, 2 H-bonds at 11.5 ppm for PO---H-O), which indicates a structural difference.



**Figure S6:** Section of the  $^1\text{H}$  NMR spectra of **1b/2a** (bottom), **1b/2b** (middle) and **1b/2c** (top) at a 1:1 ratio and a concentration of 10 mM in  $\text{CD}_2\text{Cl}_2$  at 600 MHz and 180 K. For **1b/2a**, 8 hydrogen bonds reflecting 4 different CPA/imine species were observed, while for **2b** and **2c** a multitude of different species was found.

#### 5.4. E/Z isomerization experiments

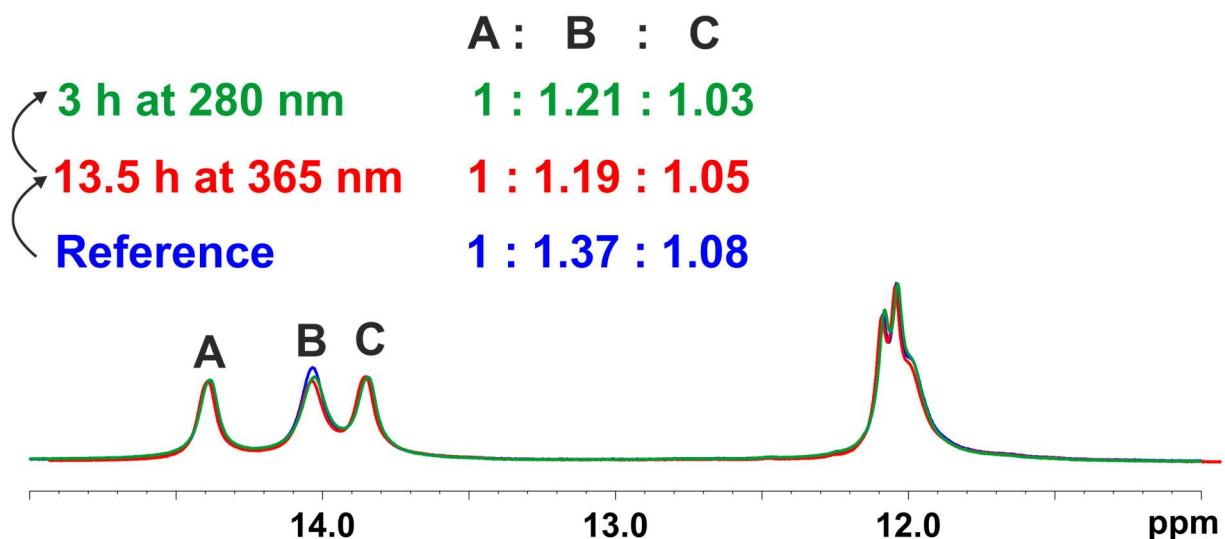
For the system **1a/2a** (see Figure 2), a second sample was prepared at low temperature (see Sample Preparation). In our previous work,<sup>9</sup> this method was applied to exclusively populate CPA/E-imine species. In our previous work,<sup>8,12</sup> the employed imines were predominantly present as the *E*-isomer, but in presence of a chiral phosphoric acid, CPA/Z-imine complexes were significantly populated (*E*:*Z* ratios up to approx. 50:50) due to a reduced steric repulsion within the binding pocket of the catalyst. However, at temperatures < -80 °C, the *E*-imine cannot isomerize. Hence, if the sample is prepared at low temperatures, it becomes possible to suppress CPA/Z-imine complexes. For **1a/2a**, the population of hydrogen bonded proton signals A, B and C (see Figure S7) are similar for the low-temperature (“*E*-only”) and room-temperature sample (“*E/Z*”). This indicates, that all 3 species feature an *E*-imine. The slight offset most likely originates in an insufficient equilibration of the “*E*-only” sample prior to the measurement.



**Figure S7:** Comparison of the <sup>1</sup>H NMR spectra of 2 samples of **1a/2a** which were prepared at low temperature (red) and at room temperature (blue) in CD<sub>2</sub>Cl<sub>2</sub> at 180 K and 600 MHz. No significant differences were observed, indicating that all 3 species feature an *E*-imine.

To further corroborate the assignment of all species as *E*-imine, photoisomerization experiments were carried out. In our previous work it was shown, that it is possible to increase the population of the CPA/Z-imine *via* illuminating the sample with an appropriate wavelength to induce a photoisomerization process.<sup>13</sup> Hence, **1a/2a** (1 : 1 ratio, 50 mM) was illuminated at 180 K first at a wavelength of 13.5 h and afterwards additionally for 3 h at 280 nm (see Figure S8). However, only minor changes in the integral ratios of the hydrogen bonded proton signals A, B and C were monitored, which could also originate in a change in relaxation times under

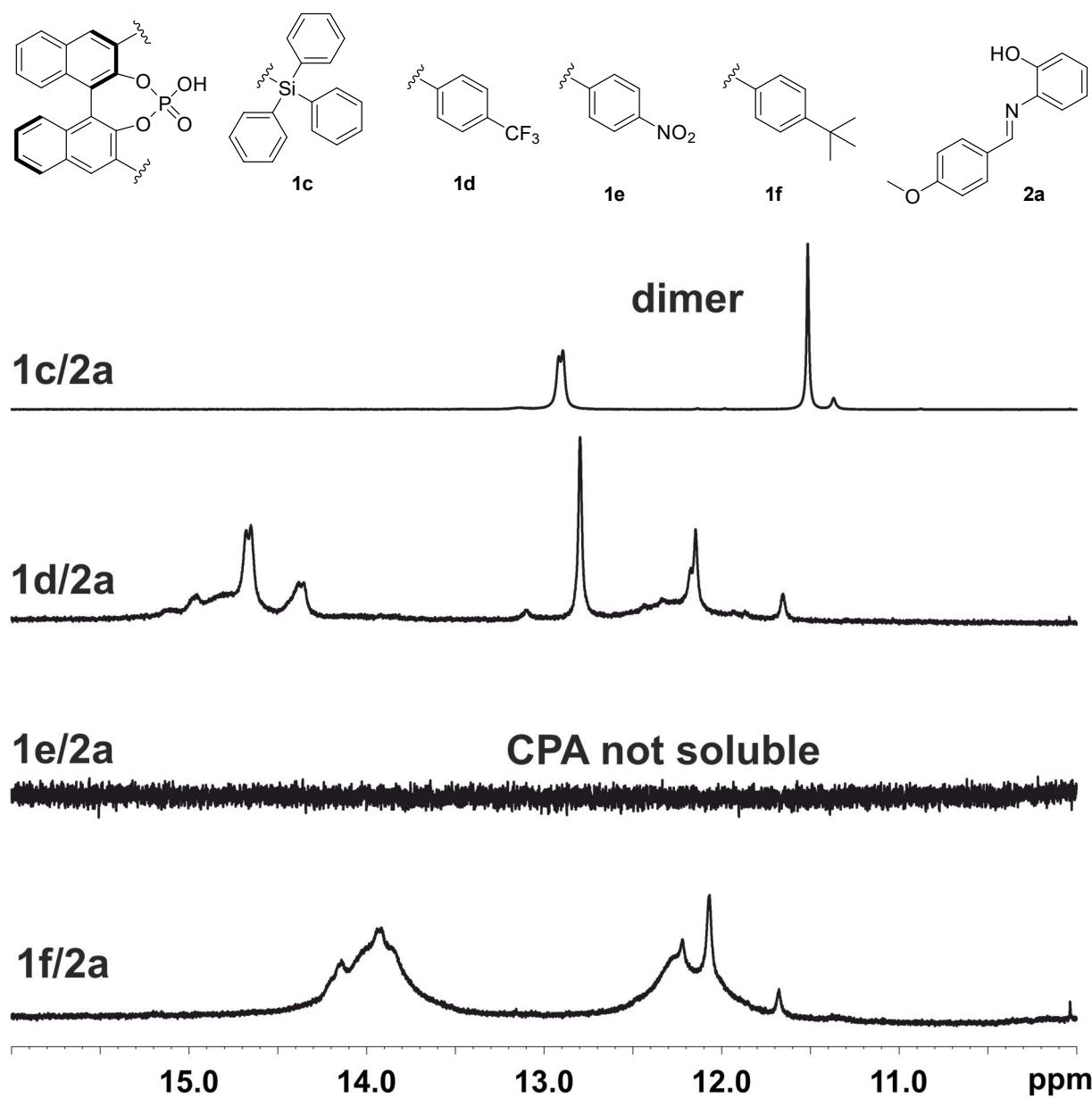
illumination. This indicates, that all three species feature the same imine configuration, which is most likely the more stable *E*.



**Figure S8:** Comparison of the hydrogen bonding section of the  $^1\text{H}$ -NMR spectra of **1a/2a** at 180 K and 600 MHz in  $\text{CD}_2\text{Cl}_2$ . The blue spectrum is the reference spectrum prior to illumination, the red spectrum was recorded after 13.5 h of continuous illumination at 365 nm and the green spectrum was recorded after following illumination at 280 nm for 3 h. Minor changes in the populations of species A, B and C were detected but unlikely originate in a change of the imine configuration.

## 6. Accessing CPA/imine monomers

In our previous research,<sup>12</sup> we identified the structure of [CPA/imine]<sub>2</sub> complexes featuring CPA **1b** and imines without additional hydrogen bond donor. In these structures, two stacked imines are nested within the binding pocket created by the 3,3'-substituents of the two catalyst molecules in a shifted phase to phase arrangement. Thereby, the 3,3'-substituents of the two catalyst molecules are close to each other. Initially we tried to modify the catalyst to hinder dimerization.

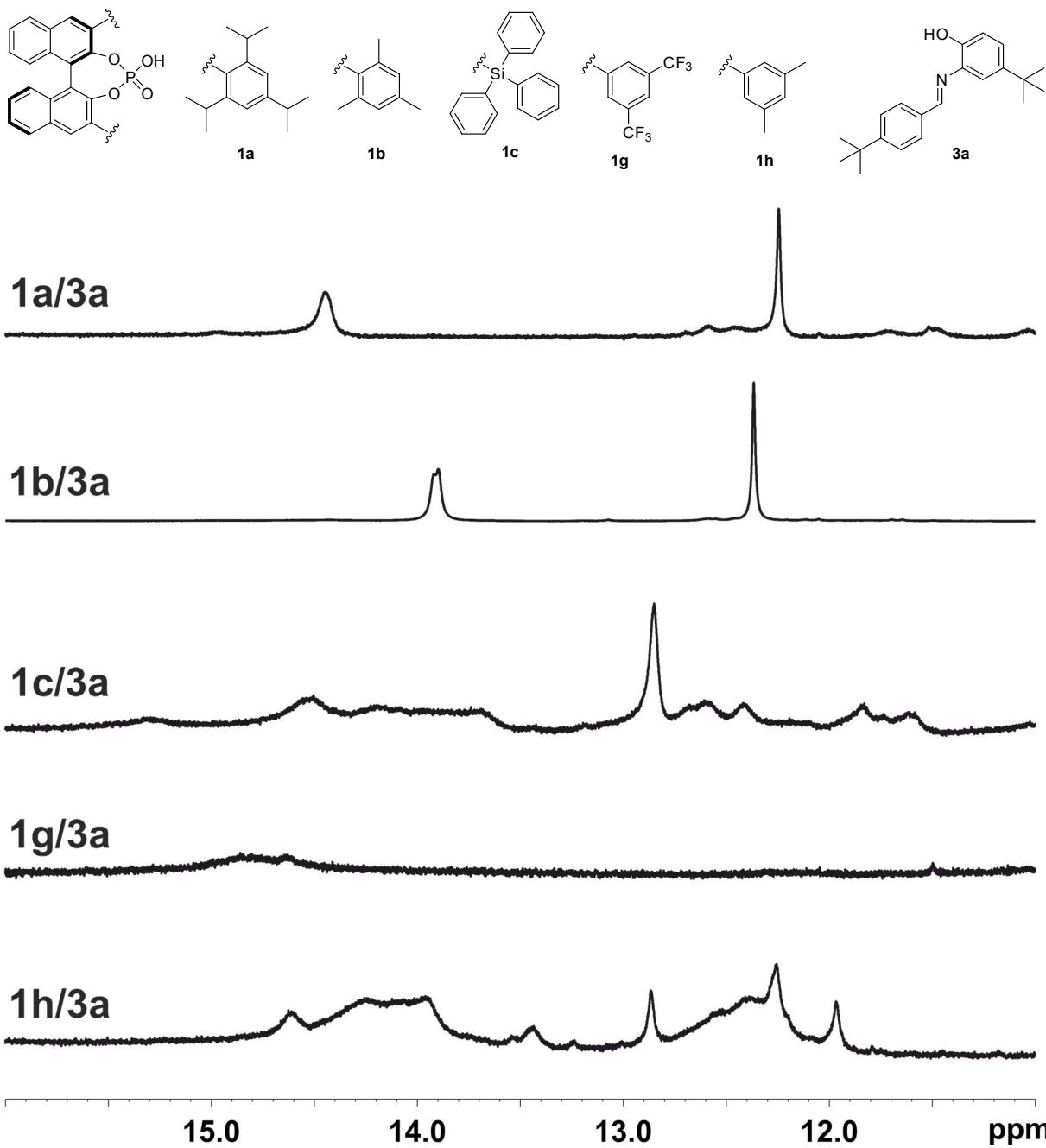


**Figure S9:** Section of the <sup>1</sup>H NMR spectra of **1c-f/2a** in CD<sub>2</sub>Cl<sub>2</sub> at a ratio of 1:1 and a concentration of 10 mM at a temperature of 180 K.

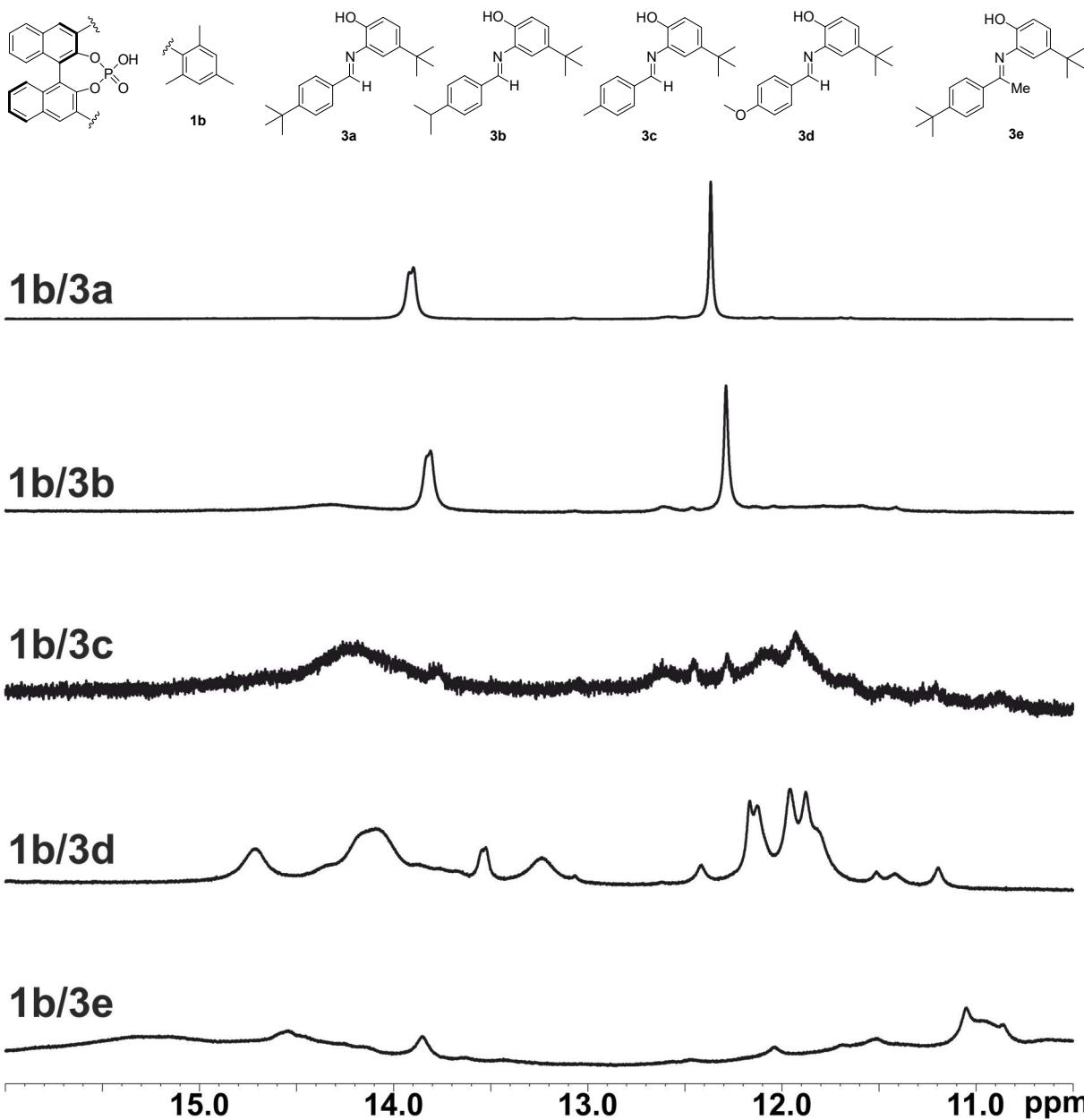
For the combination of **1c/2a** (see Figure S9, top), the steric bulk of the catalyst was increased to minimize the available space inside the binding pocket, so that two stacked imines would not fit in. The measured  $^1\text{H}$  NMR spectrum showed only two hydrogen bonds, reflecting one distinct species. However, DOSY measurements (see Table S5) showed that the monitored complex is still a dimer, giving a hydrodynamic radius of  $\sim 12.36 \text{ \AA}$ , which is close to the previously reported hydrodynamic radius of monodentate CPA/imine dimers.<sup>12</sup> For catalysts **1d-f**, the idea was to introduce electronic (**1d** and **1e**) or steric (**1f**) repulsion to the 3,3'-substituents. Catalyst **1e** was not soluble in  $\text{CD}_2\text{Cl}_2$  at low temperature and hence no hydrogen bonded proton signals are observed. For **1d/2a** and **1f/2a**, no clear and well defined hydrogen bonded proton NMR signals were observed or the spectra could not be analyzed in more detail due to signal overlap and line broadening.

In our previous research, we observed that installing sterically bulky groups such as *tert*-butyl groups on both sides of the imine can efficiently prevent dimerization of CPA/imine complexes.<sup>9</sup> Hence, imine **3a** was selected and screened with different catalysts. **1b/3a** gave a well defined spectrum with only one monomeric species (see Figure 3), thus showing that this approach can give the intended result. For **1a/3a**, a similar spectrum was recorded but with significantly broadened lines. Additionally, for **1a/3a** spontaneous gel formation – most likely by CPA catalyzed polymerization of **3a** – occurred after some time (the shown spectrum in Figure S10 was measured immediately after sample preparation). For **1c/3a**, **1g/3a** and **1h/3a**, only severely line broadened hydrogen bonded proton signals were found, highlighting the broad possible structural space of bidentate CPA/imine complexes.

After receiving a well resolved spectrum of monomeric **1b/3a**, we slightly changed the *para*-substituent or alpha-substituent to explore if the determined structure for **1b/3a** is retained if the imines are modulated. When changing the *tert*-butyl group to an *iso*-propyl group for **1b/3b** (see Figure S11), again a well resolved spectrum with one defined species was obtained (see Figure S3 for chemical shift assignment). However, when going to a methyl-substituent (**1b/3c**) or methoxy-substituent (**1b/3d**) or if the alpha-substituent is changed from hydrogen to methyl (**1b/3e**), again various broad hydrogen bonded proton signals are detected. This again highlights, that for N-(*ortho*-hydroxyaryl) substituted imines, a broad structural space is possible and that only for certain CPA/imine combinations, well defined complexes are obtained.



**Figure S10:** Section of the <sup>1</sup>H NMR spectra of **1a-c,g,h/3a** in CD<sub>2</sub>Cl<sub>2</sub> at a ratio of 1:1 and a concentration of 10 mM at a temperature of 180 K.

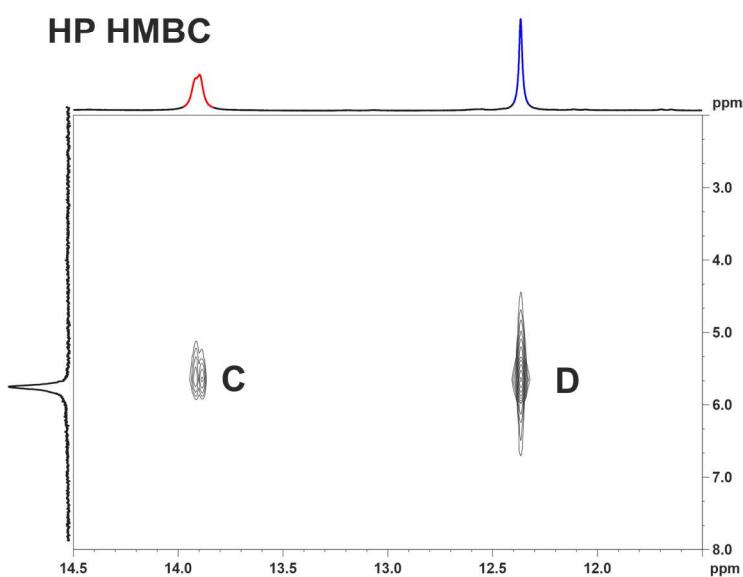
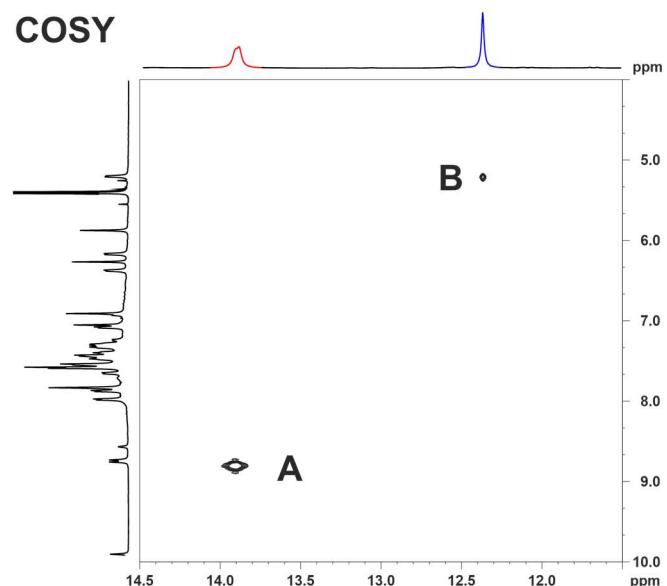
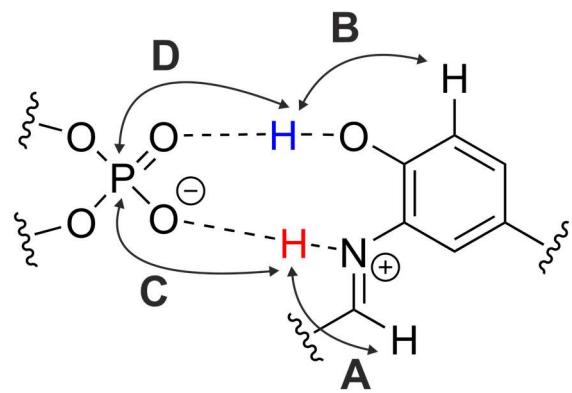


**Figure S11:** Section of the <sup>1</sup>H NMR spectra of **1b/3a-e** in CD<sub>2</sub>Cl<sub>2</sub> at a ratio of 1:1 and a concentration of 25 mM at a temperature of 180 K.

## 7. Hydrogen bond analysis

Due to their partial covalent character, hydrogen bonds can be characterized via the detection of scalar coupling between the involved nuclei.<sup>13,14</sup> For the system **1b/3a**, the full scalar coupling matrix between the two hydrogen bonded protons ( $\text{PO}^-$ ---H-N<sup>+</sup> in red and PO---H-O blue; see Figure S12) and the CPA as well as imine was detected *via* <sup>1</sup>H <sup>1</sup>H COSY and <sup>1</sup>H <sup>31</sup>P HMBC spectra. This clearly validates the assignment as hydrogen bonds and confirms the postulated binding motif featuring two hydrogen bonds as a rigid anchor.

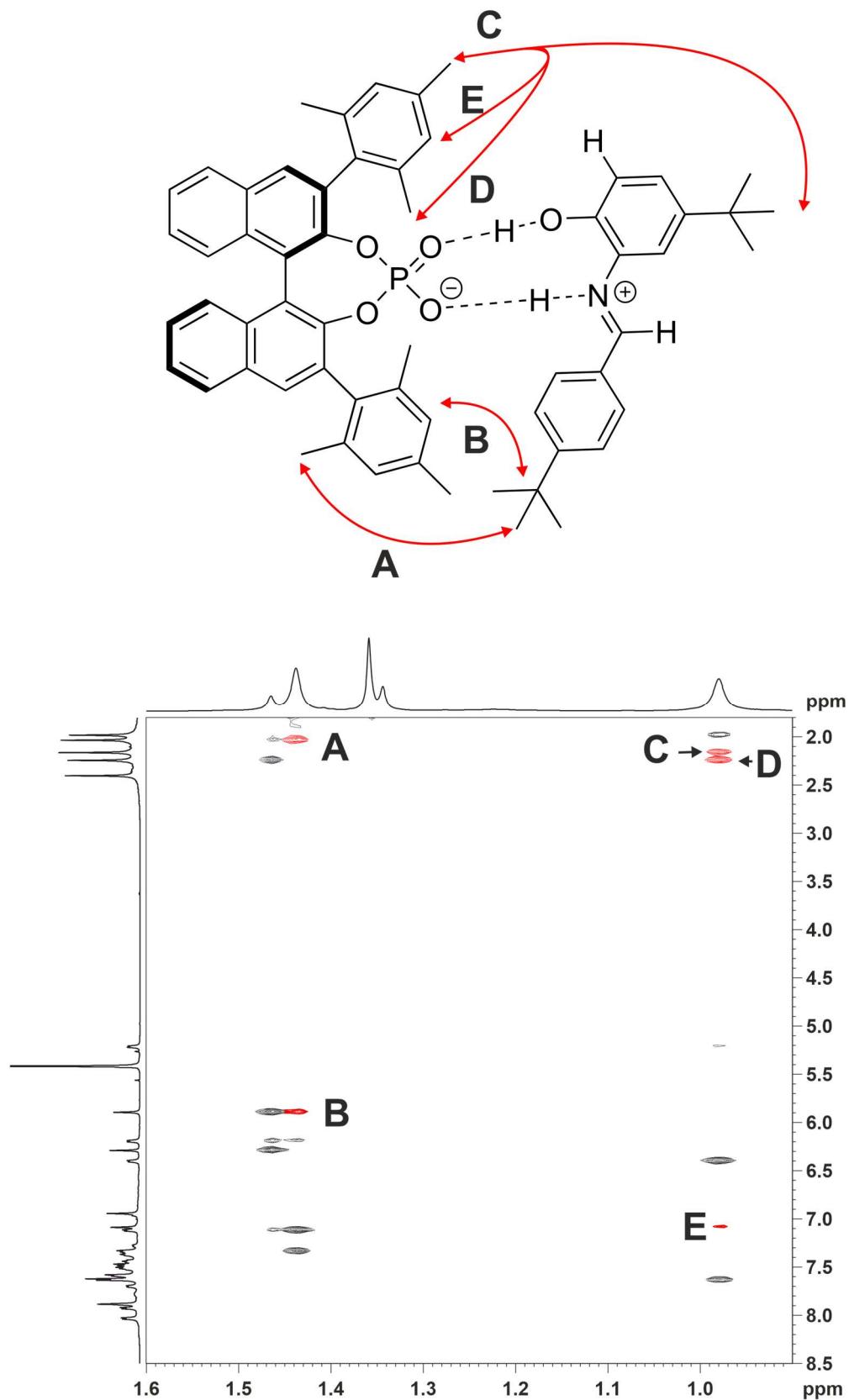
In our previous research with imines featuring no additional hydrogen bond donor, the CPA/imine complexes could be characterized as hydrogen bond assisted ion pairs by analysis of the <sup>1</sup>H and <sup>15</sup>N chemical shifts as well as  $J_{\text{NH}}$  coupling constants and comparison to reference systems with weak and very strong acids.<sup>13,14</sup> For CPA/imine complexes with imines **2** and **3**, we assume that the complexes are also present mainly as ion pairs, especially as the cooperativity effect of the second PO---H-O hydrogen bond is expected to better stabilize the resulting phosphate, which suggests an even stronger proton transfer on the imine substrate, resulting in a stronger ion pair character. This would be reflected by the <sup>1</sup>H<sub>NH</sub> coupling constant, which could not be accessed as imines **2** and **3** could not be <sup>15</sup>N labelled. However, the <sup>3</sup>H<sub>HH</sub> coupling between the hydrogen bonded proton and the  $\alpha$ -H of the imine (see e.g. Figure 12; doublet splitting was observed for all systems with  $\alpha$ -H imines) can only be present if the NH binding order is relatively close to 1, clearly indicating that the proton is strongly transferred on the imine, which results in an ion pair character. However, as there is still detectable magnetization transfer by scalar coupling to the CPA (see Figure 12 signal C), the proton transfer is not complete and the complex is best described as hydrogen bond assisted ion pair.



**Figure S12:** Excerpts of the  $^1\text{H}$   $^1\text{H}$  COSY and  $^1\text{H}$   $^{31}\text{P}$  HMBC spectra of **1b/3a** at a 1:1 ratio and a concentration of 25 mM at 180 K in  $\text{CD}_2\text{Cl}_2$ . The cross peaks A-D clearly show the full scalar coupling matrix of the hydrogen bonded protons.

## 8. NOE studies

In the computed structure model for the bidentate complex **1b/3a** (see Figure 4) the imine is located in between the two 3,3'-substituents of the catalyst. The performed NOESY study (see Figure S13) validated this structure model, as only NOE contacts between the imine and the two 3,3'-substituents were found and none to the BINOL backbone, which would be typical for monodentate CPA/imine complexes.<sup>8,12</sup> One site of the imine has only contact to one 3,3'-substituent (NOE cross peaks A and B or C-E), which could be resolved due to the signal splitting of the C<sub>2</sub>-symmetric catalyst in the CPA/imine complex.



**Figure S13:** Excerpt of the NOESY spectrum of **1b/3a** at a 1:1 ratio and a concentration of 25 mM at 180 K in  $\text{CD}_2\text{Cl}_2$ . The cross peaks A-E shown in red validate the theoretically computed structure model.

## 9. Computational Details

### 9.1 Molecular dynamics (MD) simulations

#### 9.1.1 Simulation details (**1a/2a**)

For the generation of the phosphate and iminium forcefields GAFF<sup>15</sup> parameters were used. For the iminium-ions the problem appears that the double bond atoms are not well described with the standard GAFF parameters. Due to prior investigations in our group, the parameters of “ce” and “na” were chosen as an approximation.

The parameters were assigned with antechamber<sup>16</sup>. For the point charges geometry optimizations with orca (versions 4.1.2/ 4.2.1)<sup>17</sup> at (B3LYP 6-311G\*\*) level were done and RESP-fits using Multiwfn<sup>18</sup> 3.6 with a Merz Kollman grid of 6 points/Å<sup>2</sup> or finer with at least 4 layers per atom were used. Constraints were set for equivalent charges on CH<sub>2</sub> and CH<sub>3</sub>. The conversion to a GROMACS topology was done with the tool acpype<sup>19</sup>.

The DCM force-field is derived of a density-corrected DCM (GAFF based) force field by Horinek<sup>20</sup> with additional virtual sites. (see appended force fields section 9.1.3)

The MD-runs were performed with a timestep of 2 fs at 200K (v-rescale thermostat, tau = 1.0 ps) at a concentration of 50mM in DCM, 1 bar (Parrinello-Rahman barostat with isotropic scaling, compressibility = 2.5e<sup>-5</sup> bar<sup>-1</sup>) with a cutoff of 1 nm with the GROMACS package [RMG-Q7] in versions ranging from 2018-2020 employing the verlet scheme.

PME was used for the coulomb interactions. A potential shift coulomb modifier was employed.

Dispersion corrections were added for energy and pressure.

Constraints for all bonds were used (LINCS, lincs-order = 4, number of iterations 2, verlet-buffer-tolerance = 0.0001 kJ/mol/ps).

For the first simulation all molecules were placed into the simulation box at random.

The obtained cluster-size distributions however do not correspond to the experimental distribution. This is due to the nature of the forcefields used. (Once associated to either a CPA-iminium monomer or dimer the ions do not diffuse away from one another on the simulation time scale. Whereas in the experiments dissociation happens as phosphoric acid and imine respectively even though nearly everything is incorporated in CPA-iminium dimers.)

While this leads to the “wrong” representation of aggregates in the system during normal simulation times, this does not mean that they cannot be studied:

For this we make a distinction between dimers where the iminium group and the alcohol group of one single imine form hydrogen bond with the same phosphate (non-bridging) and those where they form hydrogen bonds with two different phosphates (bridging dimers). That means the bridging dimers consist of two phosphoric acid residues interconnected via two iminium residues, whereas for non-bridging dimers one would assume stacking of the imines of two “monomer units” (each with their individual phosphate).

We expect that non-bridging dimers would have been readily seen in the initial simulation (compare to cluster size distributions tab S6). There only around 20% of “dimer” was detected. This corresponds to the identified bridging dimer and therefore we anticipate that bridging-dimers will correspond to the experimentally found dimers.

For the determination of free energy differences between the bridging dimers with four H-bonds each Thermodynamic Integration (TI)<sup>21</sup> is used (employing the lambda vector implementation of GROMACS with “vdw”, “coul” and “restraint”-lambdas, use of soft core potentials with sc\_alpha = 0.5, sc\_power = 1).

We achieve this by uncoupling one single iminium molecule and keeping the rest of the aggregate constant, which converges the dH/dlambda values faster than if the four molecules were to be coupled simultaneously. For this, simulations of at least 60 ns were employed. The derivative dH/dlambda was written out every 500 steps.

To fix the positions even in the uncoupled state distance constraints (umbrella-type, force constant of 1000 kJ/mol/nm<sup>2</sup>) set to the mean value of the unconstrained simulation are used along the HBs and coupled during the simulations.

Because the constraints are chosen differently (namely to the average from the unconstrained simulation) for all dimers they do not cancel one another and need to be removed at the end.

Therefore the TI procedure is given by:

- First one completely coupled system without constraints,
- then a constraining step for the completely interacting system,

- followed by coulomb decoupling in steps of delta lambda = 0.1,
- subsequent VdW decoupling in steps of delta lambda = 0.1,
- and finally the removal of the constraints.

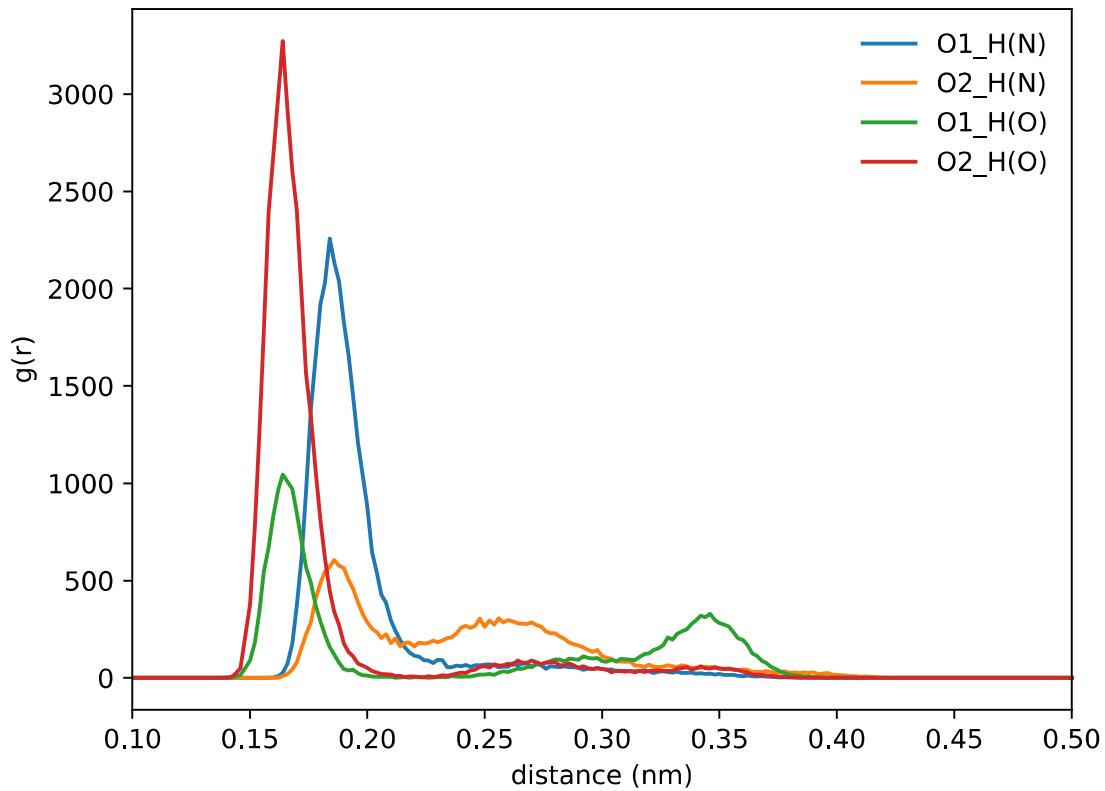
For the last part of the integration McLaurin integration (of Newton Cotes formula - used points 0.83 0.5 0.167) was employed.

### 9.1.2 Results (**1a/2a**)

#### *9.1.2.1 Simulation of randomly placed ions in DCM*

Because the question regarding the **1a/2a** system was ‘What might be hidden behind the three different H-bond situations?’ those were investigated. As a starting point the phosphate oxygen and iminium H(N) H(O) distances were therefore obviously of interest:

An initial evaluation of the radial distribution functions (RDFs) between phosphate oxygen (named O1 and O2) and the iminium hydrogen bonding atoms (both the alcohol and iminium group) of the box showed that while there is one main distance both for the iminium H(N) and for the alcohol H(O) towards the phosphorus atoms- the situation is a bit more complicated. (Additionally: Obviously the situation is not even completely converged, because O1 and O2 are not yet equivalent, which means additional species have been sampled, which did not interconvert on our timescale.)



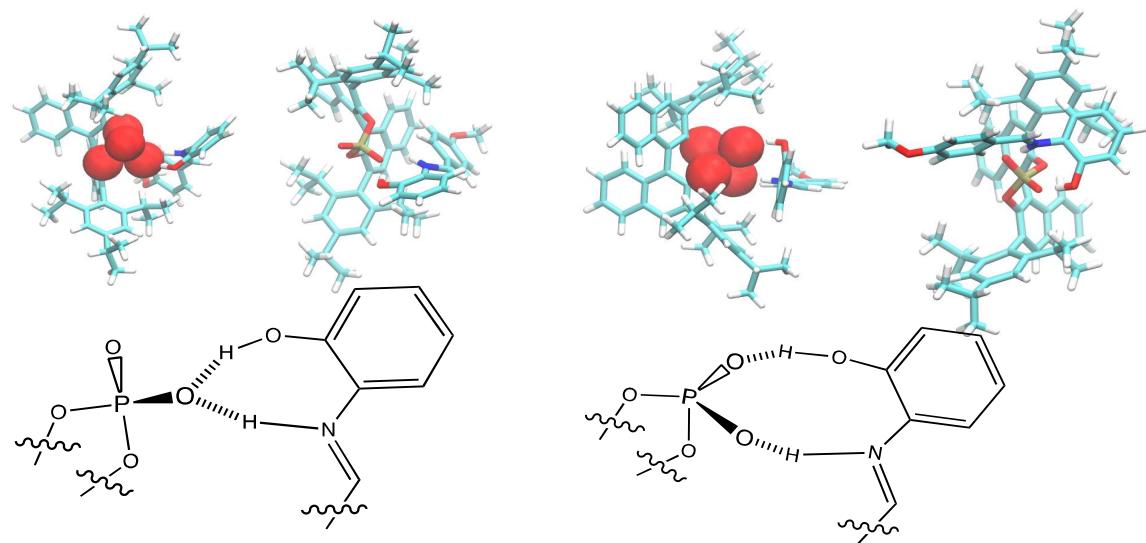
**Figure S14:** Distance RDF between phosphate oxygens and iminium H bonding atoms.

Therefore, the situations needed to be analyzed in more detail:

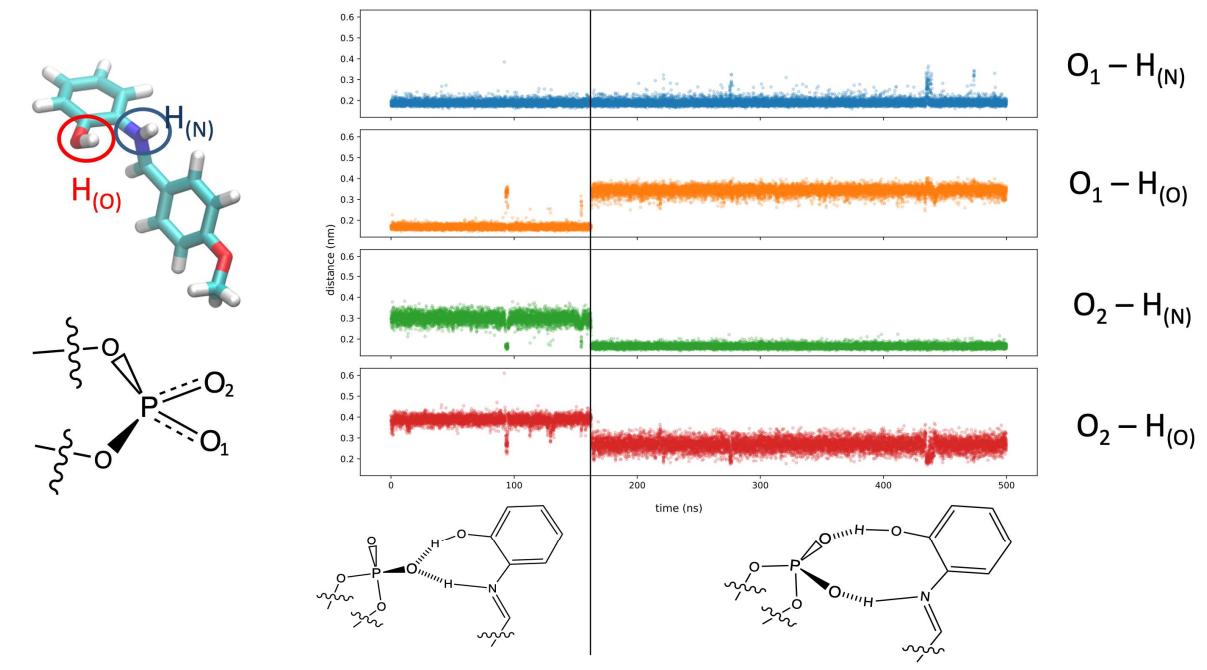
A simple cluster-size search (based of cutoff – excluding the solvent) showed that just inserting molecules into our simulation box did not reproduce the experimental dimers since our classical force fields cannot describe bond-breaking and formation which is necessary for the correct description of the dissociation process of both monomers and dimers.

The hydrogen-bonding situation differ between the observed monomers and dimers.

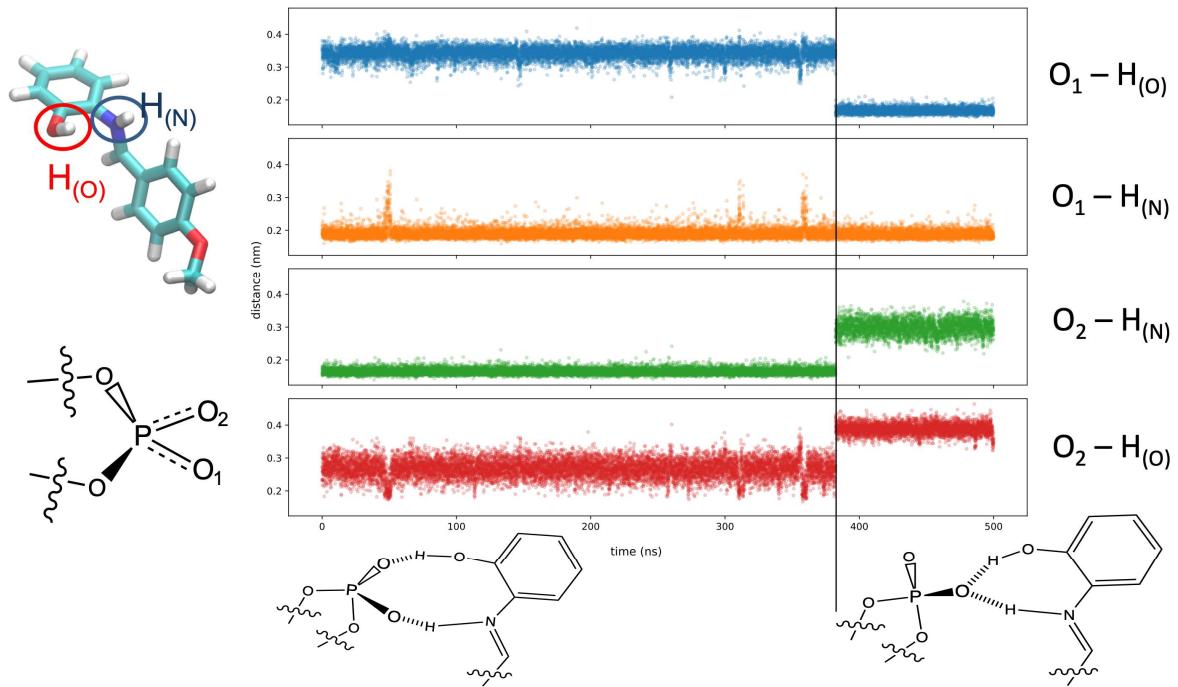
To elucidate this, first the monomer situation will be discussed: In case of the monomers two hydrogen bonds are formed between the CPA and imine. This can lead to the two different situations regarding the phosphate oxygen – namely one “bifurcated” bond towards one single oxygen or a single bond towards each of the CPA oxygens. (see fig. S15) Interestingly those interconvert quite slowly (rare events on the 100 ns timescale). This is not an equilibration issue because it happens in both directions (see figures S16/S17).



**Figure S15:** Different possible H-bond situations occurring for the monomers



**Figure S16:** Example of bond switching by the OH group

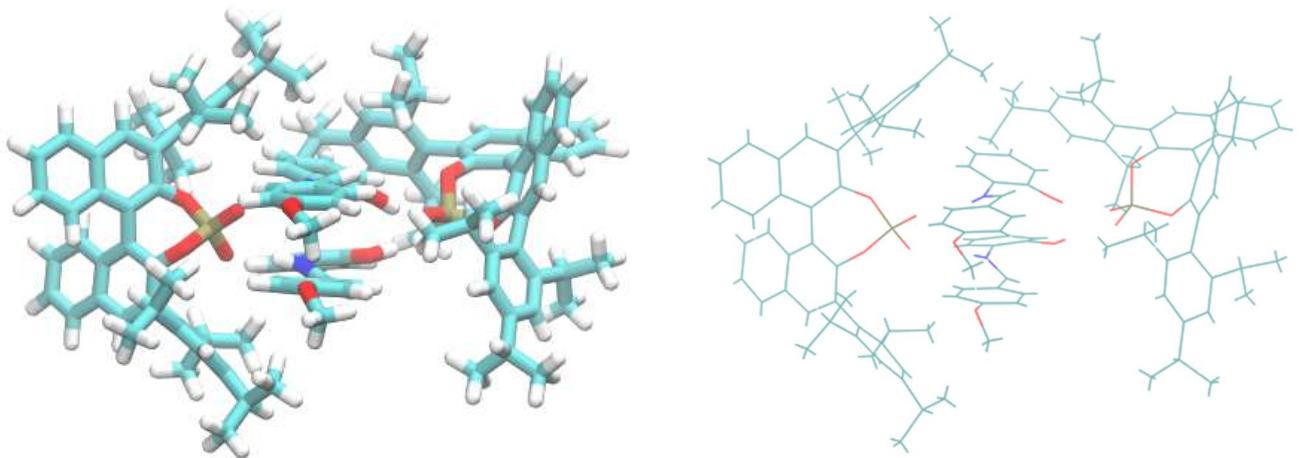


**Figure S17:** Example of bond switching by the OH group

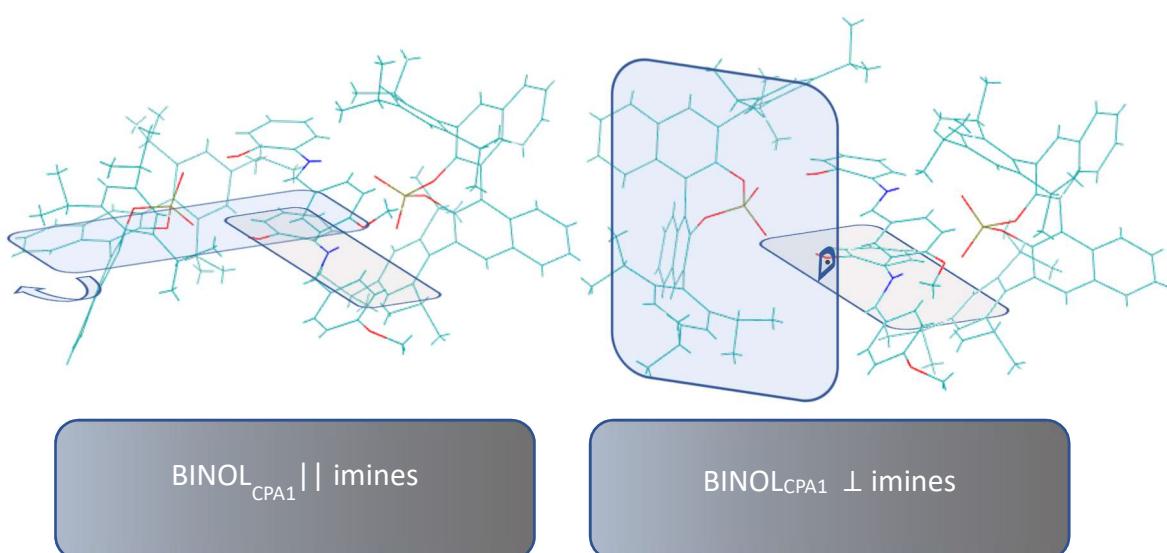
Opposed to the monomers the observed dimer is different. (see fig. S18) Not only is every individual iminium molecule forming an H-bond with two different CPA moieties; both imines are stacked on top of one another, leading to an asymmetry with regard to the positioning of alcohol and iminium groups. That in turn has another interesting consequence, which once more shows that a bidentate binding motif does not exclude orientational flexibility: We could observe two conformers (see fig. 19), which differ in the positioning of the CPA on the OH-side. The corresponding BINOL backbone can therefore be positioned either parallel or orthogonal to a plane given by the imines. This leads to different accessibilities of the substrate moiety.

**Table S6:** Clustersizes (in atoms - solvent for cluster search ignored) and their probabilities for the initial (randomly positioned ions) simulation broken down in the combination of cations and anions

cutoff	0.35				0.4			
	clustsize	probability (%)	cation	anion	clustsize	probability (%)	cation	anion
	142	75.22	1	1	142	74.09	1	1
	284	18.25	2	2	284	18.63	2	2
	426	4.54	3	3	426	4.96	3	3
	568	1.29	4	4	568	1.45	4	4
	710	0.55	5	5	710	0.67	5	5
	852	0.12	6	6	852	0.15	6	6
	994	0.04	7	7	994	0.05	7	7
cutoff	0.45				0.5			
	clustsize	probability (%)	cation	anion	clustsize	probability (%)	cation	anion
	142	72.57	1	1	142	70.29	1	1
	284	19.11	2	2	284	19.7	2	2
	426	5.52	3	3	426	6.37	3	3
	568	1.68	4	4	568	2.05	4	4
	710	0.83	5	5	710	1.09	5	5
	852	0.22	6	6	852	0.36	6	6
	994	0.07	7	7	994	0.13	7	7



**Figure S18:** Snapshot of the observed bridging dimer (both OH groups toward one single CPA, NH groups towards the second CPA residue)



**Figure S19:** Snapshots showing the conformational flexibility of the observed bridging dimer (flip of the CPA residue connected via OH bonds)

### 9.1.2.2 Simulation of bridging dimeric species – Thermodynamic integration (TI)

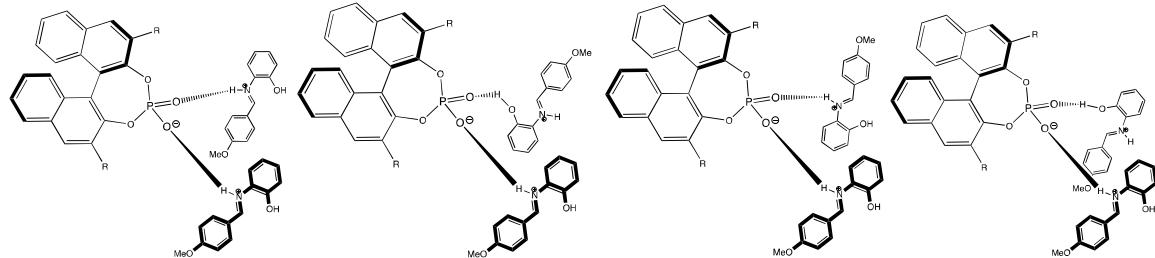
Due to the aforementioned equilibration issues TI<sup>21</sup> was chosen in order to obtain insight into the possible other bridging dimers. Those are characterized with respect to their respective functional groups' positions (in tab. S7).

**Table S7:** Summary of the possible bridging dimers

		iminium MeO groups	NH groups
<b><u>dimer 0</u></b>		same direction for both imines	towards one single CPA group (asymmetric)
<b><u>dimer 1</u></b>		same direction for both imines	towards two different CPA groups (symmetric)
<b><u>dimer 2</u></b>		different direction for both imines	towards two different CPA groups (symmetric)
<b><u>dimer 3</u></b>		different direction for both imines	towards one single CPA group (asymmetric)

**Table S8:** Summary of the Thermodynamic Integration (**1a** case)

	$-\Delta G_c$ [kJ/mol]	$-\Delta G_{\text{coul}}$ [kJ/mol]	$-\Delta G_{\text{vdw}}$ [kJ/mol]	$-\Delta G_{c2}$ [kJ/mol]	$\Delta\Delta G_{(i-0)}$ [kJ/mol]
<b>0</b>	-0.02	220.1	44.4	-21.3	0.0
<b>1</b>	-0.55	211.5	50.0	-18.9	1.0
<b>2</b>	-0.08	217.1	45.3	-19.8	0.6
<b>3</b>	-0.02	198.2	48.0	-17.0	14.1



**Figure S20:** Schematic sketch of bond situations in the dimers as described in table S7 (only one CPA shown) - left to right: I) “same direction for both MeOH” – “NH (OH) groups oriented towards one single CPA”; II) “different directions for MeOH groups” – “NH groups oriented towards different CPAs”; III) “different directions for MeOH groups” – “NH (OH) groups oriented towards one single CPA”, IV) “same direction for both MeOH” – “NH groups oriented towards different CPAs”

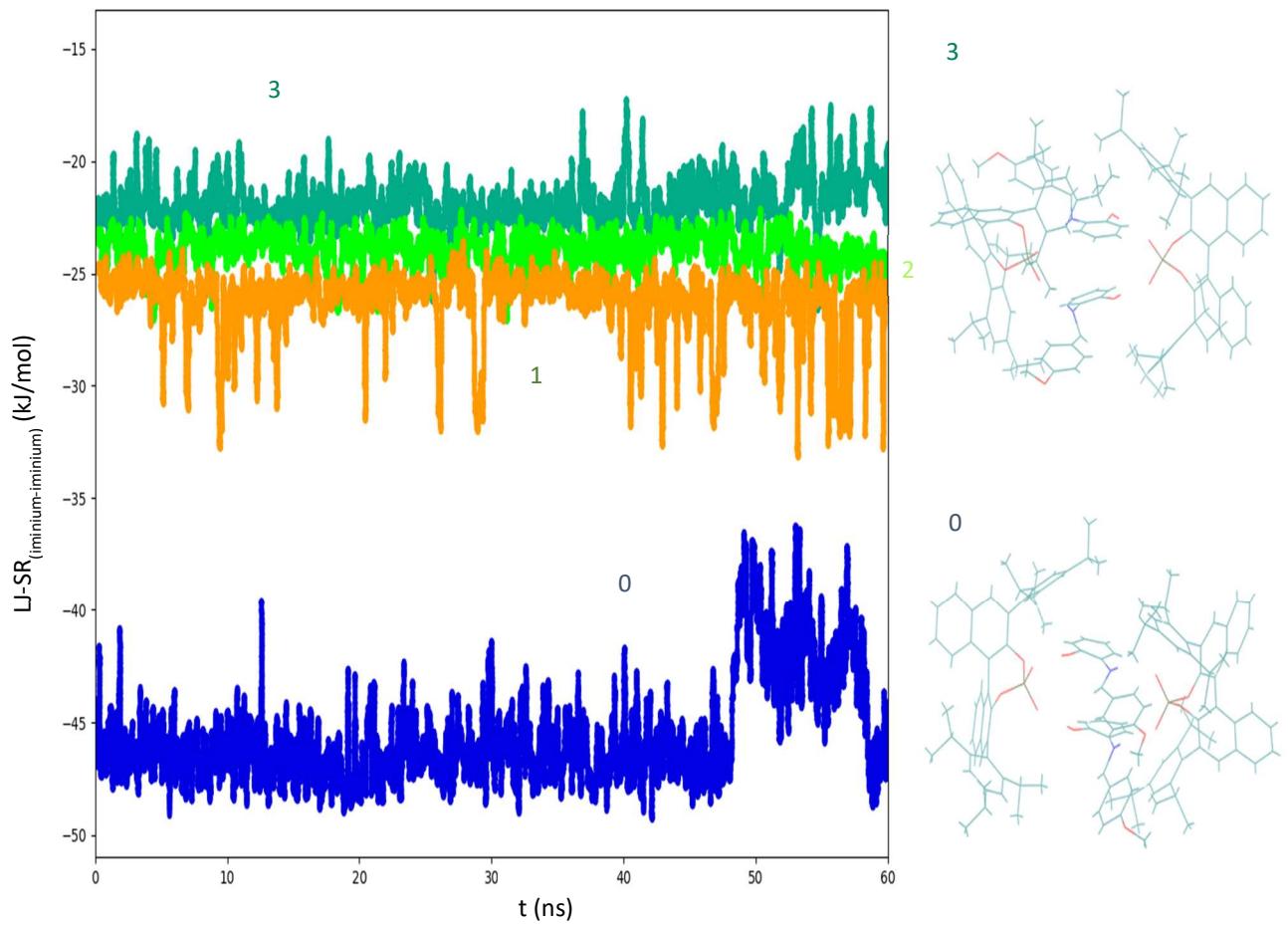
Table S8 gives an overview over the results from the TI. Here  $\Delta G_c$  denotes the difference in free Energy due to the constrained H-bonds (and  $\Delta G_{c2}$  to the released constraints).  $\Delta G_{\text{coul/vdw}}$  are due to the switching of coulomb or VdW interactions. ( $\Delta G_{\text{tot}}$  would correspond to the solvation free energy  $\Delta G_{\text{solv}}$  if not only one of the 2a molecules were coupled but the whole dimer.) The decisive term is  $\Delta\Delta G_{(i-0)}$  – even if it is not completely converged due to the influence of rare

events (and corresponding changes in the relative values). While the absolute value of  $\Delta\Delta G$  might not be completely converged no large qualitative differences can be expected. It clearly shows that “dimer 3” is definitively not accessible.

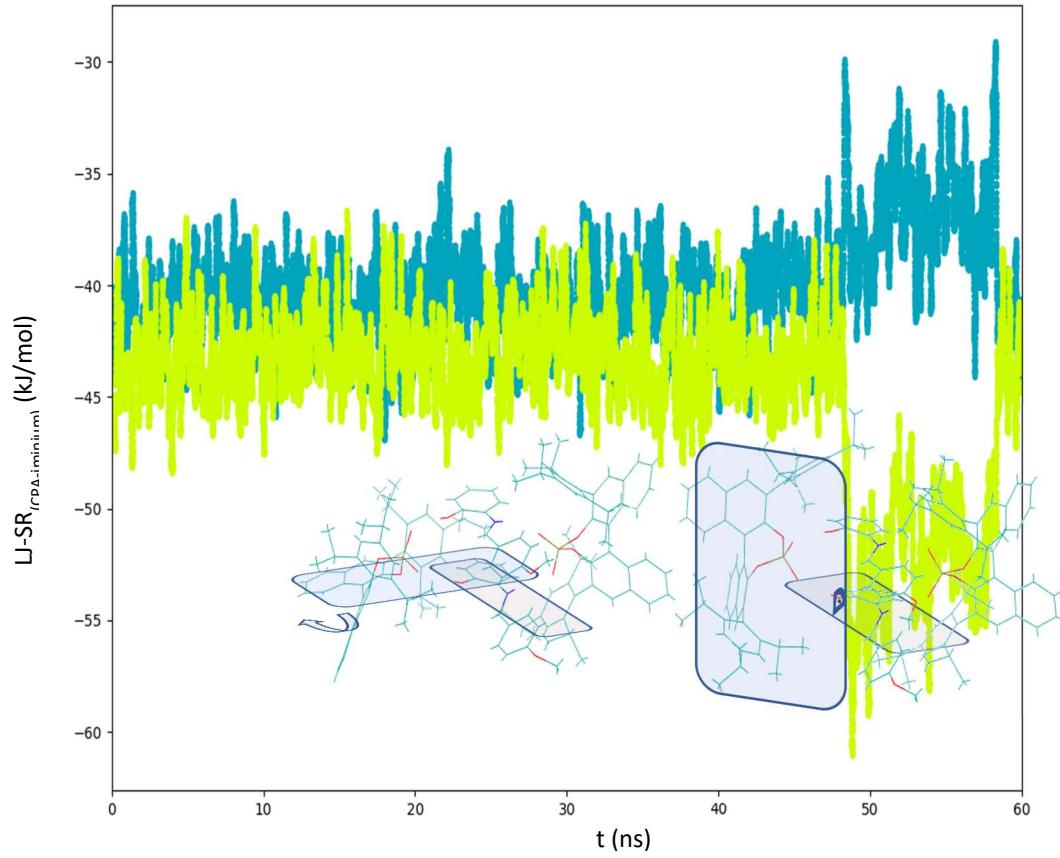
An investigation of the short range interactions sheds some light on the differences between the iminium dimers: The short range LJ interaction shows that while the symmetric dimers “1” and “2” show comparable LJ(iminium-iminium) stabilization, the asymmetric ones “0” and “3” show a major difference. – This is unsurprising because “0” shows an iminium stacking motif whereas the imines barely overlap in case of “3”. (see fig. S21) The apparent “loss” of LJ(iminium-iminium) of “0” is related to the aforementioned (see fig. S19) orientational change of the CPA residue on the alcohol side and is compensated by the LJ(CPA-iminium) interaction. (see fig. S22)

**Summarily: There is no indication of non-bridging dimers from the cluster size distributions and we identified three similarly stable bridging dimers (“0”, “1”, “2”). Those might correspond to the three experimentally different H-bond signals from H-NMR.**

**In addition, dimer “0” shows orientational flexibility on the OH side, which is possible due to LJ energy compensation switching between imines and CPA.**



**Figure S21:** Short range LJ(iminium-iminium) energies for the bridging imines (for the unconstrained TI run)



**Figure S22:** Short range  $LJ(CPA\text{-iminium})$  energies between CPA and each of the two imines individually for dimer “0” (for the unconstrained TI run)

### 9.1.3 Comparison with system (**1e/2a**)

The catalyst **1e** is poorly soluble in CH<sub>2</sub>Cl<sub>2</sub> (DCM), which prevented an extensive NMR study.

Still the question appeared, whether our model system with catalyst **1a** is comparable for small amounts dissolved in DCM (beneath the NMR detection limit) or whether different H-bond situations are dominating.

Keeping the iminium molecule **2a** identical and changing **1a** to the (R,R'=Ph-NO<sub>2</sub>)-CPA **1e** we investigated the differences.

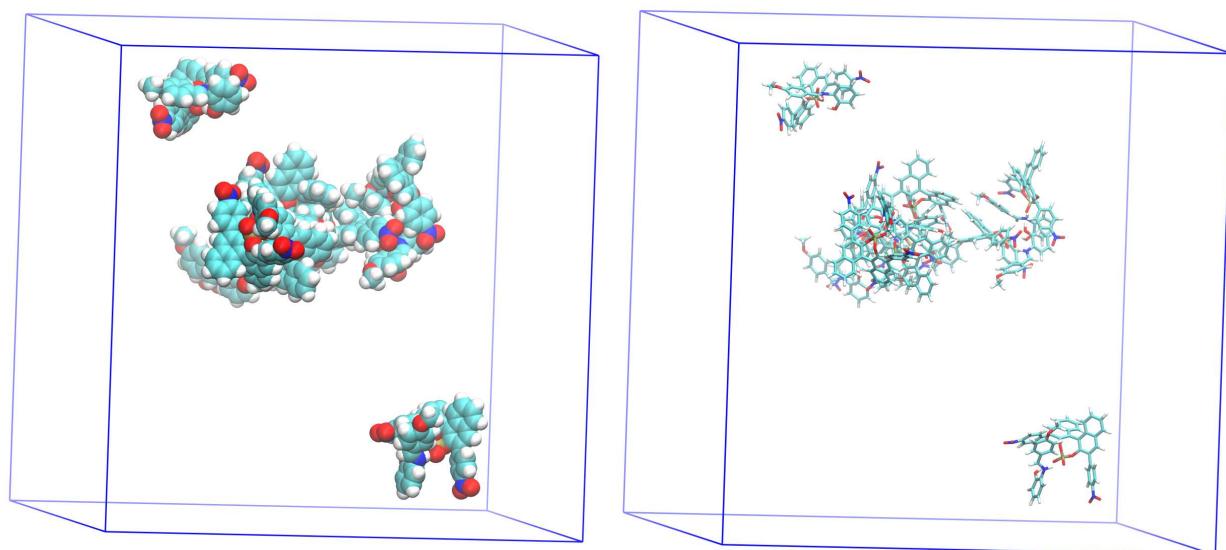
Two different approaches were employed when investigating the **1e** system:

- An unconstrained simulation to check the behavior of **1e/2a** in solution (as a check for non-bridged dimers created by stacking of iminium ions)
- Simulations constrained to the 4 bridging dimers analogous to the **1a/2a** system

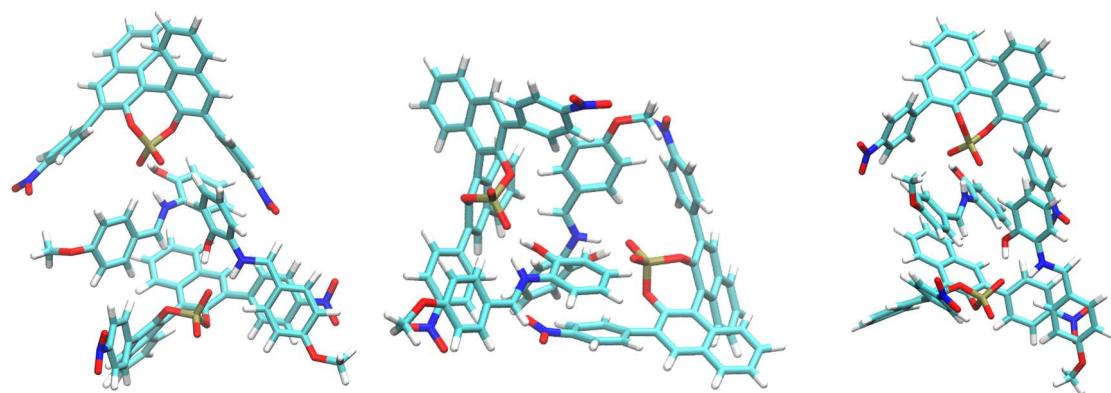
#### Unconstrained simulation

Starting with a NVT simulation of **1e/2a** (10 molecules each) ions an initial box was generated by solvating it in DCM (3300 molecules) for the NpT simulation, which was then run for >120 ns. We observed that the phase separation from DCM persists for the mayor part of the ions and we ended up in the situation where only few **1e/2a** species with the CPA bound via two H-bonds to a single iminium molecule in solution. For those no stacking of the iminium molecules was observed.

Still there appear persistent non-bridging species on the route towards the 1:1 species, where either the CPA backbone stacks with the iminium molecule or the NO<sub>2</sub> groups are part of the H-bonds. (examples see fig. S24)



**Figure S23:** **1e/2a** in DCM (solvent not shown) showing phase separation and single 1:1 **1e/2a** species in solution (left: VdW representation, right: stick representation)



**Figure S24:** Snapshots of **1e/2a** in DCM showing different non-bridging 2:2 species

These are not stable over the whole simulation time, so the question remains whether they are equilibrium species or only intermediary appearances during the equilibration towards the single 1:1 **1e/2a** species observed in the end.

### Constrained simulations

Similar conformations as above in the **1a/2a** case were enforced by a single set of umbrella constraints where the distances were taken from one of the constrained simulations above (in contrast to the previous system, where all distances had been taken from individual equilibrium runs. - Otherwise identical conditions as before were employed to the **1a/2a** case) assuming identical H-bond lengths for all conformers and avoiding (see fig. S24) the various other possible conformations.

The results of the thermodynamic integration ( $dH/d\lambda$  averaged over at least 30 ns) are given as follows (see table S9):

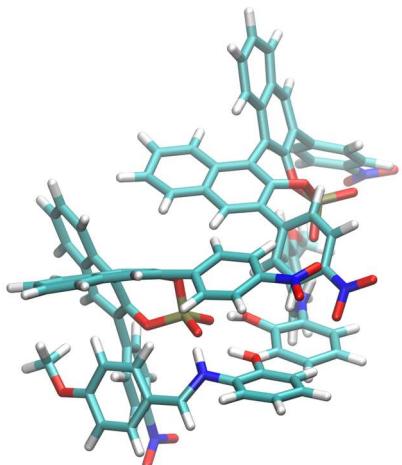
**Table S9:** Summary of the Thermodynamic Integration for the **1e** case

	$-\Delta G_c$ [kJ/mol]	$-\Delta G_{coul}$ [kJ/mol]	$-\Delta G_{vdw}$ [kJ/mol]	$-\Delta G_{c2}$ [kJ/mol]	$\Delta\Delta G_{(i-0)}$ [kJ/mol]
<b>0</b>	23	218	47	-29	0
<b>1</b>	13	210	47	-22	-10
<b>2</b>	16	205	41	-21	-17
<b>3</b>	34	207	43	-28	-2

The differences between the conformations are clearly larger than in our previous **1a** case.

Of exceptional note is that there is a large term associated with constraining the geometry to the desired conformers in the first place, which was negligible in the case of **1a** as they naturally occurred. That already means our system does not prefer the suggested conformations of **1a**. Therefore, this part of the coupling was not done in a single step (as in **1a**) but employed the same McLaurin steps as the uncoupling process.

Furthermore different 2:2 bridging complexes can be observed, in the fully coupled but unconstrained case. Here at least one of the NO<sub>2</sub> groups is involved in the H-bonds. This does not resemble the **1a** cases at all but rather show structures reminiscent of π-stacked structures.



**Figure S25:** Snapshot taken for a fully interacting (**1e** case) but non constrained 2:2 system reveals different H-bond situations opposed to **1a** case

A comparison of the constrained (bridging-dimers) systems **1a** to **1e** shows the fact that the two systems differ in their H-bonding situations. (**1a** yields 3 practically equally stable cases whereas **1e** shows strong differences).

It hints at the fact that for the **1e** case the ‘mono’-cluster (1:1 CPA/iminium) is decisive.

## 9.1.4 Force Field parameters

### 9.1.4.1 Combination rules

Combination rules '2' is used.

This corresponds to the use of LJ parameters  $\sigma$  (nm) and  $\epsilon$  (kJ/mol) with the combination rules of:  $\sigma_{ij} = (\sigma_i + \sigma_j)/2$  and  $\epsilon_{ij} = (\epsilon_i * \epsilon_j)^{(1/2)}$ . Pair generation is used both for coulomb and LJ with the respective multiplicative factors:

fudgeLJ=0.5, fudgeQQ= 0.833

### 9.1.4.2 Atomtypes

[ atomtypes ]

;name	bond_type	mass	charge	ptype	sigma	epsilon	Amb
c3s	c3s	0.00000	0.00000	A	3.39967e-01	4.57730e-01 ; 1.91	0.1094
cls	cls	0.00000	0.00000	A	3.30000e-01	1.20000e+00 ; DH scaled or correct density	
h2s	h2s	0.00000	0.00000	V	2.29317e-01	6.56888e-02 ; 1.29	0.0157
ca	ca	0.00000	0.00000	A	3.39967e-01	3.59824e-01 ; 1.91	0.0860
cp	cp	0.00000	0.00000	A	3.39967e-01	3.59824e-01 ; 1.91	0.0860
os	os	0.00000	0.00000	A	3.00001e-01	7.11280e-01 ; 1.68	0.1700
p5	p5	0.00000	0.00000	A	3.74177e-01	8.36800e-01 ; 2.10	0.2000
o	o	0.00000	0.00000	A	2.95992e-01	8.78640e-01 ; 1.66	0.2100
c3	c3	0.00000	0.00000	A	3.39967e-01	4.57730e-01 ; 1.91	0.1094
ha	ha	0.00000	0.00000	A	2.59964e-01	6.27600e-02 ; 1.46	0.0150
hc	hc	0.00000	0.00000	A	2.64953e-01	6.56888e-02 ; 1.49	0.0157
ce	ce	0.00000	0.00000	A	3.39967e-01	3.59824e-01 ; 1.91	0.0860
na	na	0.00000	0.00000	A	3.25000e-01	7.11280e-01 ; 1.82	0.1700
oh	oh	0.00000	0.00000	A	3.06647e-01	8.80314e-01 ; 1.72	0.2104
ho	ho	0.00000	0.00000	A	0.00000e+00	0.00000e+00 ; 0.00	0.0000
hn	hn	0.00000	0.00000	A	1.06908e-01	6.56888e-02 ; 0.60	0.0157

```
h4    h4      0.00000 0.00000 A   2.51055e-01 6.27600e-02 ; 1.41 0.0150
h1    h1      0.00000 0.00000 A   2.47135e-01 6.56888e-02 ; 1.39 0.0157
```

#### 9.1.4.3 Molecule definitions

##### 9.1.4.3.1 DCM

dcm.itp

[ moleculetype ]

;name nrexcl2

DCM 3

[ atoms ]

; nr type resi res atom cgnr charge mass ;  
1 c3s 1 DCM Cs 1 -0.137416 14.01000 ;  
2 cl1 1 DCM Cls 2 -0.117537 35.45000 ;  
3 h2s 1 DCM Hs 3 0.184786 0.00000 ;  
4 cl2 1 DCM Cls1 4 -0.115826 35.45000 ;  
5 h2s 1 DCM Hs1 5 0.185993 0.00000 ;

[ bonds ]

; ai aj funct r k  
1 2 1 1.7860e-01 2.3347e+05 ; C - cl2s  
1 4 1 1.7860e-01 2.3347e+05 ; C - cl2s1

[ angles ]

2 1 4 1 1.1103e+02 4.5380e+02 ; cl2s - C - cl2s1

[ virtual\_sites3 ]

```
; Site from      funct a     b     c
3   1   2   4   4   -0.2831048825030728 -0.2831048825030728 -3.154641146073173
5   1   2   4   4   -0.2831048825030728 -0.2831048825030728 3.154641146073173
```

[ exclusions ]

3 1 2 4 5

5 1 2 3 4

---

### 9.1.4.3.1 2a

[ moleculetype ]

```
;name      nrexcl
COU      3
```

[ atoms ]

```
1 ca  1 COU  C  1  -0.250798  12.01000 ; qtot 0.000
2 ca  1 COU  C1  2   0.526755  12.01000 ; qtot 0.000
3 ca  1 COU  C2  3  -0.315126  12.01000 ; qtot 0.000
4 ca  1 COU  C3  4  -0.054992  12.01000 ; qtot 0.000
5 ca  1 COU  C4  5  -0.060231  12.01000 ; qtot 0.000
6 ca  1 COU  C5  6  -0.132069  12.01000 ; qtot 0.000
7 ce  1 COU  C6  7   0.114329  12.01000 ; qtot 0.000
8 na  1 COU  N  8  -0.146003  14.01000 ; qtot 0.000
9 ca  1 COU  C7  9   0.086116  12.01000 ; qtot 0.000
10 ca  1 COU  C8 10   0.319648  12.01000 ; qtot 0.000
11 ca  1 COU  C9 11  -0.317838  12.01000 ; qtot 0.000
12 ca  1 COU  C10 12  -0.033055  12.01000 ; qtot 0.000
13 ca  1 COU  C11 13  -0.115345  12.01000 ; qtot 0.000
```

```

14 ca 1 COU C12 14 -0.251855 12.01000 ; qtot 0.000
15 oh 1 COU O 15 -0.516929 16.00000 ; qtot 0.000
16 ho 1 COU H 16 0.429366 1.00800 ; qtot 0.000
17 hn 1 COU H1 17 0.262125 1.00800 ; qtot 0.000
18 ha 1 COU H2 18 0.158792 1.00800 ; qtot 0.000
19 ha 1 COU H3 19 0.187615 1.00800 ; qtot 0.000
20 ha 1 COU H4 20 0.158076 1.00800 ; qtot 0.000
21 ha 1 COU H5 21 0.170400 1.00800 ; qtot 0.000
22 ha 1 COU H6 22 0.188612 1.00800 ; qtot 0.000
23 ha 1 COU H7 23 0.149913 1.00800 ; qtot 0.000
24 ha 1 COU H8 24 0.189841 1.00800 ; qtot 0.000
25 h4 1 COU H9 25 0.143911 1.00800 ; qtot 0.000
26 ha 1 COU H10 26 0.154738 1.00800 ; qtot 0.000
27 os 1 COU O1 27 -0.322577 16.00000 ; qtot 0.000
28 c3 1 COU C13 28 0.034780 12.01000 ; qtot 0.000
29 h1 1 COU H11 29 0.080600 1.00800 ; qtot 0.000
30 h1 1 COU H12 30 0.080600 1.00800 ; qtot 0.000
31 h1 1 COU H13 31 0.080600 1.00800 ; qtot 0.000

```

[ bonds ]

```

; ai aj funct r k
1 2 1 1.3984e-01 3.8585e+05 ; C - C1
1 6 1 1.3984e-01 3.8585e+05 ; C - C5
1 18 1 1.0860e-01 2.8937e+05 ; C - H2
2 3 1 1.3984e-01 3.8585e+05 ; C1 - C2
2 27 1 1.3696e-01 3.1514e+05 ; C1 - O1
3 4 1 1.3984e-01 3.8585e+05 ; C2 - C3
3 19 1 1.0860e-01 2.8937e+05 ; C2 - H3
4 5 1 1.3984e-01 3.8585e+05 ; C3 - C4
4 20 1 1.0860e-01 2.8937e+05 ; C3 - H4

```

```

5   6  1  1.3984e-01  3.8585e+05 ;  C4 - C5
5   7  1  1.4763e-01  3.0234e+05 ;  C4 - C6
6   21 1  1.0860e-01  2.8937e+05 ;  C5 - H5
7   8  1  1.4209e-01  3.1263e+05 ;  C6 - N
7   25 1  1.0916e-01  2.8267e+05 ;  C6 - H9
8   9  1  1.3840e-01  3.5187e+05 ;  N - C7
8   17 1  1.0100e-01  3.4175e+05 ;  N - H1
9   10 1  1.3984e-01  3.8585e+05 ;  C7 - C8
9   14 1  1.3984e-01  3.8585e+05 ;  C7 - C12
10  11 1  1.3984e-01  3.8585e+05 ;  C8 - C9
10  15 1  1.3637e-01  3.2133e+05 ;  C8 - O
11  12 1  1.3984e-01  3.8585e+05 ;  C9 - C10
11  22 1  1.0860e-01  2.8937e+05 ;  C9 - H6
12  13 1  1.3984e-01  3.8585e+05 ;  C10 - C11
12  23 1  1.0860e-01  2.8937e+05 ;  C10 - H7
13  14 1  1.3984e-01  3.8585e+05 ;  C11 - C12
13  26 1  1.0860e-01  2.8937e+05 ;  C11 - H10
14  24 1  1.0860e-01  2.8937e+05 ;  C12 - H8
15  16 1  9.7300e-02  3.1079e+05 ;  O - H
27  28 1  1.4316e-01  2.5824e+05 ;  O1 - C13
28  29 1  1.0969e-01  2.7665e+05 ;  C13 - H11
28  30 1  1.0969e-01  2.7665e+05 ;  C13 - H12
28  31 1  1.0969e-01  2.7665e+05 ;  C13 - H13

```

[ pairs ]

```

; ai  aj  funct
1   4   1;  C - C3
1   7   1;  C - C6
1   19  1;  C - H3

```

1 28 1; C - C13  
2 5 1; C1 - C4  
2 20 1; C1 - H4  
2 21 1; C1 - H5  
2 29 1; C1 - H11  
2 30 1; C1 - H12  
2 31 1; C1 - H13  
3 6 1; C2 - C5  
3 7 1; C2 - C6  
3 28 1; C2 - C13  
4 8 1; C3 - N  
4 21 1; C3 - H5  
4 25 1; C3 - H9  
4 27 1; C3 - O1  
5 9 1; C4 - C7  
5 17 1; C4 - H1  
5 19 1; C4 - H3  
6 8 1; C5 - N  
6 20 1; C5 - H4  
6 25 1; C5 - H9  
6 27 1; C5 - O1  
7 10 1; C6 - C8  
7 14 1; C6 - C12  
7 20 1; C6 - H4  
7 21 1; C6 - H5  
8 11 1; N - C9  
8 13 1; N - C11  
8 15 1; N - O  
8 24 1; N - H8

9 12 1; C7 - C10  
9 16 1; C7 - H  
9 22 1; C7 - H6  
9 25 1; C7 - H9  
9 26 1; C7 - H10  
10 13 1; C8 - C11  
10 17 1; C8 - H1  
10 23 1; C8 - H7  
10 24 1; C8 - H8  
11 14 1; C9 - C12  
11 16 1; C9 - H  
11 26 1; C9 - H10  
12 15 1; C10 - O  
12 24 1; C10 - H8  
13 22 1; C11 - H6  
14 15 1; C12 - O  
14 17 1; C12 - H1  
14 23 1; C12 - H7  
15 22 1; O - H6  
17 25 1; H1 - H9  
18 3 1; H2 - C2  
18 5 1; H2 - C4  
18 21 1; H2 - H5  
18 27 1; H2 - O1  
19 20 1; H3 - H4  
19 27 1; H3 - O1  
22 23 1; H6 - H7  
23 26 1; H7 - H10  
24 26 1; H8 - H10

[ angles ]

;	ai	aj	ak	funct	theta	cth	
1	2	3	1	1.2002e+02	5.5731e+02 ;	C - C1	- C2
1	2	27	1	1.1920e+02	5.8241e+02 ;	C - C1	- O1
1	6	5	1	1.2002e+02	5.5731e+02 ;	C - C5	- C4
1	6	21	1	1.1988e+02	4.0334e+02 ;	C - C5	- H5
2	1	6	1	1.2002e+02	5.5731e+02 ;	C1 - C	- C5
2	1	18	1	1.1988e+02	4.0334e+02 ;	C1 - C	- H2
2	3	4	1	1.2002e+02	5.5731e+02 ;	C1 - C2	- C3
2	3	19	1	1.1988e+02	4.0334e+02 ;	C1 - C2	- H3
2	27	28	1	1.1796e+02	5.2300e+02 ;	C1 - O1	- C13
3	2	27	1	1.1920e+02	5.8241e+02 ;	C2 - C1	- O1
3	4	5	1	1.2002e+02	5.5731e+02 ;	C2 - C3	- C4
3	4	20	1	1.1988e+02	4.0334e+02 ;	C2 - C3	- H4
4	3	19	1	1.1988e+02	4.0334e+02 ;	C3 - C2	- H3
4	5	6	1	1.2002e+02	5.5731e+02 ;	C3 - C4	- C5
4	5	7	1	1.2082e+02	5.3974e+02 ;	C3 - C4	- C6
5	4	20	1	1.1988e+02	4.0334e+02 ;	C4 - C3	- H4
5	6	21	1	1.1988e+02	4.0334e+02 ;	C4 - C5	- H5
5	7	8	1	1.1919e+02	5.7823e+02 ;	C4 - C6	- N
5	7	25	1	1.1699e+02	3.8995e+02 ;	C4 - C6	- H9
6	1	18	1	1.1988e+02	4.0334e+02 ;	C5 - C	- H2
6	5	7	1	1.2082e+02	5.3974e+02 ;	C5 - C4	- C6
7	8	9	1	1.2661e+02	5.2635e+02 ;	C6 - N	- C7
7	8	17	1	1.2554e+02	3.8995e+02 ;	C6 - N	- H1
8	7	25	1	1.1558e+02	4.2258e+02 ;	N - C6	- H9
8	9	10	1	1.1834e+02	5.7823e+02 ;	N - C7	- C8
8	9	14	1	1.1834e+02	5.7823e+02 ;	N - C7	- C12

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9   8   17   1   1.2554e+02   3.8995e+02 ;   C7 - N   - H1
9   10  11   1   1.2002e+02   5.5731e+02 ;   C7 - C8   - C9
9   10  15   1   1.1990e+02   5.8158e+02 ;   C7 - C8   - O
9   14  13   1   1.2002e+02   5.5731e+02 ;   C7 - C12  - C11
9   14  24   1   1.1988e+02   4.0334e+02 ;   C7 - C12  - H8
10  9   14   1   1.2002e+02   5.5731e+02 ;   C8 - C7   - C12
10  11  12   1   1.2002e+02   5.5731e+02 ;   C8 - C9   - C10
10  11  22   1   1.1988e+02   4.0334e+02 ;   C8 - C9   - H6
10  15  16   1   1.0858e+02   4.1003e+02 ;   C8 - O    - H
11  10  15   1   1.1990e+02   5.8158e+02 ;   C9 - C8   - O
11  12  13   1   1.2002e+02   5.5731e+02 ;   C9 - C10  - C11
11  12  23   1   1.1988e+02   4.0334e+02 ;   C9 - C10  - H7
12  11  22   1   1.1988e+02   4.0334e+02 ;   C10 - C9   - H6
12  13  14   1   1.2002e+02   5.5731e+02 ;   C10 - C11  - C12
12  13  26   1   1.1988e+02   4.0334e+02 ;   C10 - C11  - H10
13  12  23   1   1.1988e+02   4.0334e+02 ;   C11 - C10  - H7
13  14  24   1   1.1988e+02   4.0334e+02 ;   C11 - C12  - H8
14  13  26   1   1.1988e+02   4.0334e+02 ;   C12 - C11  - H10
27  28  29   1   1.0978e+02   4.2509e+02 ;   O1 - C13  - H11
27  28  30   1   1.0978e+02   4.2509e+02 ;   O1 - C13  - H12
27  28  31   1   1.0978e+02   4.2509e+02 ;   O1 - C13  - H13
29  28  30   1   1.0846e+02   3.2803e+02 ;   H11 - C13  - H12
29  28  31   1   1.0846e+02   3.2803e+02 ;   H11 - C13  - H13
30  28  31   1   1.0846e+02   3.2803e+02 ;   H12 - C13  - H13

```

[ dihedrals ] ; proper

; for gromacs 4.5 or higher, using funct 9

; i j k l func phase kd pn

1 2 3 4 9 180.00 15.16700 2; C- C1- C2- C3

1	2	3	19	9	180.00	15.16700	2;	C-	C1-	C2-	H3
1	2	27	28	9	180.00	3.76560	2;	C-	C1-	O1-	C13
1	6	5	4	9	180.00	15.16700	2;	C-	C5-	C4-	C3
1	6	5	7	9	180.00	15.16700	2;	C-	C5-	C4-	C6
2	1	6	5	9	180.00	15.16700	2;	C1-	C-	C5-	C4
2	1	6	21	9	180.00	15.16700	2;	C1-	C-	C5-	H5
2	3	4	5	9	180.00	15.16700	2;	C1-	C2-	C3-	C4
2	3	4	20	9	180.00	15.16700	2;	C1-	C2-	C3-	H4
2	27	28	29	9	0.00	1.60387	3;	C1-	O1-	C13-	H11
2	27	28	30	9	0.00	1.60387	3;	C1-	O1-	C13-	H12
2	27	28	31	9	0.00	1.60387	3;	C1-	O1-	C13-	H13
3	2	27	28	9	180.00	3.76560	2;	C2-	C1-	O1-	C13
3	4	5	6	9	180.00	15.16700	2;	C2-	C3-	C4-	C5
3	4	5	7	9	180.00	15.16700	2;	C2-	C3-	C4-	C6
4	3	2	27	9	180.00	15.16700	2;	C3-	C2-	C1-	O1
4	5	6	21	9	180.00	15.16700	2;	C3-	C4-	C5-	H5
4	5	7	8	9	180.00	2.92880	2;	C3-	C4-	C6-	N
4	5	7	25	9	180.00	2.92880	2;	C3-	C4-	C6-	H9
5	4	3	19	9	180.00	15.16700	2;	C4-	C3-	C2-	H3
5	7	8	9	9	180.00	7.11280	2;	C4-	C6-	N-	C7
5	7	8	17	9	180.00	7.11280	2;	C4-	C6-	N-	H1
6	1	2	3	9	180.00	15.16700	2;	C5-	C-	C1-	C2
6	1	2	27	9	180.00	15.16700	2;	C5-	C-	C1-	O1
6	5	4	20	9	180.00	15.16700	2;	C5-	C4-	C3-	H4
6	5	7	8	9	180.00	2.92880	2;	C5-	C4-	C6-	N
6	5	7	25	9	180.00	2.92880	2;	C5-	C4-	C6-	H9
7	5	4	20	9	180.00	15.16700	2;	C6-	C4-	C3-	H4
7	5	6	21	9	180.00	15.16700	2;	C6-	C4-	C5-	H5
7	8	9	10	9	180.00	1.25520	2;	C6-	N-	C7-	C8

7 8 9 14 9 180.00 1.25520 2; C6- N- C7- C12  
 8 9 10 11 9 180.00 15.16700 2; N- C7- C8- C9  
 8 9 10 15 9 180.00 15.16700 2; N- C7- C8- O  
 8 9 14 13 9 180.00 15.16700 2; N- C7- C12- C11  
 8 9 14 24 9 180.00 15.16700 2; N- C7- C12- H8  
 9 8 7 25 9 180.00 7.11280 2; C7- N- C6- H9  
 9 10 11 12 9 180.00 15.16700 2; C7- C8- C9- C10  
 9 10 11 22 9 180.00 15.16700 2; C7- C8- C9- H6  
 9 10 15 16 9 180.00 3.76560 2; C7- C8- O- H  
 9 14 13 12 9 180.00 15.16700 2; C7- C12- C11- C10  
 9 14 13 26 9 180.00 15.16700 2; C7- C12- C11- H10  
 10 9 8 17 9 180.00 1.25520 2; C8- C7- N- H1  
 10 9 14 13 9 180.00 15.16700 2; C8- C7- C12- C11  
 10 9 14 24 9 180.00 15.16700 2; C8- C7- C12- H8  
 10 11 12 13 9 180.00 15.16700 2; C8- C9- C10- C11  
 10 11 12 23 9 180.00 15.16700 2; C8- C9- C10- H7  
 11 10 9 14 9 180.00 15.16700 2; C9- C8- C7- C12  
 11 10 15 16 9 180.00 3.76560 2; C9- C8- O- H  
 11 12 13 14 9 180.00 15.16700 2; C9- C10- C11- C12  
 11 12 13 26 9 180.00 15.16700 2; C9- C10- C11- H10  
 12 11 10 15 9 180.00 15.16700 2; C10- C9- C8- O  
 12 13 14 24 9 180.00 15.16700 2; C10- C11- C12- H8  
 13 12 11 22 9 180.00 15.16700 2; C11- C10- C9- H6  
 14 9 8 17 9 180.00 1.25520 2; C12- C7- N- H1  
 14 9 10 15 9 180.00 15.16700 2; C12- C7- C8- O  
 14 13 12 23 9 180.00 15.16700 2; C12- C11- C10- H7  
 15 10 11 22 9 180.00 15.16700 2; O- C8- C9- H6  
 17 8 7 25 9 180.00 7.11280 2; H1- N- C6- H9  
 18 1 2 3 9 180.00 15.16700 2; H2- C- C1- C2

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18  1   2   27   9  180.00 15.16700 2;  H2-  C-  C1-  O1
18  1   6   5   9  180.00 15.16700 2;  H2-  C-  C5-  C4
18  1   6   21   9  180.00 15.16700 2;  H2-  C-  C5-  H5
19  3   2   27   9  180.00 15.16700 2;  H3-  C2-  C1-  O1
19  3   4   20   9  180.00 15.16700 2;  H3-  C2-  C3-  H4
22  11  12   23   9  180.00 15.16700 2;  H6-  C9-  C10-  H7
23  12  13   26   9  180.00 15.16700 2;  H7-  C10-  C11-  H10
24  14  13   26   9  180.00 15.16700 2;  H8-  C12-  C11-  H10

```

[ dihedrals ] ; impropers

; treated as propers in GROMACS to use correct AMBER analytical function

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; i   j   k   l func  phase  kd   pn
1   3   2   27   4  180.00 4.60240 2;  C-  C2-  C1-  O1
1   5   6   21   4  180.00 4.60240 2;  C-  C4-  C5-  H5
2   4   3   19   4  180.00 4.60240 2;  C1-  C3-  C2-  H3
3   5   4   20   4  180.00 4.60240 2;  C2-  C4-  C3-  H4
4   6   5   7   4  180.00 4.60240 2;  C3-  C5-  C4-  C6
5   25  7   8   4  180.00 4.60240 2;  C4-  H9-  C6-  N
9   7   8   17   4  180.00 4.60240 2;  C7-  C6-  N-  H1
9   11  10   15   4  180.00 4.60240 2;  C7-  C9-  C8-  O
9   13  14   24   4  180.00 4.60240 2;  C7-  C11-  C12-  H8
10  12  11   22   4  180.00 4.60240 2;  C8-  C10-  C9-  H6
10  14   9   8   4  180.00 4.60240 2;  C8-  C12-  C7-  N
11  13  12   23   4  180.00 4.60240 2;  C9-  C11-  C10-  H7
12  14  13   26   4  180.00 4.60240 2;  C10-  C12-  C11-  H10
18  1   6   2   4  180.00 4.60240 2;  H2-  C-  C5-  C1

```

### 9.1.4.3.3 1a

[ moleculetype ]

```
;name      nrexcl  
PIA      3
```

[ atoms ]

```
1 ca  1 PIA  C  1  -0.087898  12.01000 ;  
2 ca  1 PIA  C1  2  -0.201812  12.01000 ;  
3 ca  1 PIA  C2  3  -0.119114  12.01000 ;  
4 ca  1 PIA  C3  4   0.051137  12.01000 ;  
5 ca  1 PIA  C4  5   0.230956  12.01000 ;  
6 ca  1 PIA  C5  6  -0.289697  12.01000 ;  
7 cp  1 PIA  C6  7  -0.108763  12.01000 ;  
8 ca  1 PIA  C7  8   0.294717  12.01000 ;  
9 cp  1 PIA  C8  9  -0.078915  12.01000 ;  
10 ca  1 PIA  C9  10  -0.330488  12.01000 ;  
11 cp  1 PIA  C10 11  -0.112668  12.01000 ;  
12 ca  1 PIA  C11 12   0.054017  12.01000 ;  
13 ca  1 PIA  C12 13   0.230184  12.01000 ;  
14 ca  1 PIA  C13 14  -0.330135  12.01000 ;  
15 cp  1 PIA  C14 15  -0.078833  12.01000 ;  
16 ca  1 PIA  C15 16   0.295949  12.01000 ;  
17 ca  1 PIA  C16 17  -0.120041  12.01000 ;  
18 ca  1 PIA  C17 18  -0.201878  12.01000 ;  
19 ca  1 PIA  C18 19  -0.088877  12.01000 ;  
20 ca  1 PIA  C19 20  -0.289958  12.01000 ;  
21 os  1 PIA  O  21  -0.446130  16.00000 ;  
22 p5  1 PIA  P  22   1.316091  30.97000 ;
```

23 o 1 PIA O1 23 -0.751621 16.00000 ;  
 24 o 1 PIA O2 24 -0.751814 16.00000 ;  
 25 os 1 PIA O3 25 -0.4448087 16.00000 ;  
 26 cp 1 PIA C20 26 -0.102737 12.01000 ;  
 27 cp 1 PIA C21 27 -0.103767 12.01000 ;  
 28 ca 1 PIA C22 28 0.124283 12.01000 ;  
 29 ca 1 PIA C23 29 -0.422979 12.01000 ;  
 30 ca 1 PIA C24 30 0.198337 12.01000 ;  
 31 ca 1 PIA C25 31 -0.472431 12.01000 ;  
 32 ca 1 PIA C26 32 0.127149 12.01000 ;  
 33 ca 1 PIA C27 33 0.124660 12.01000 ;  
 34 ca 1 PIA C28 34 -0.467962 12.01000 ;  
 35 ca 1 PIA C29 35 0.194577 12.01000 ;  
 36 ca 1 PIA C30 36 -0.420254 12.01000 ;  
 37 ca 1 PIA C31 37 0.121976 12.01000 ;  
 38 c3 1 PIA C32 38 0.464838 12.01000 ;  
 39 c3 1 PIA C33 39 0.614021 12.01000 ;  
 40 c3 1 PIA C34 40 0.367455 12.01000 ;  
 41 c3 1 PIA C35 41 0.615426 12.01000 ;  
 42 c3 1 PIA C36 42 0.464726 12.01000 ;  
 43 c3 1 PIA C37 43 0.365226 12.01000 ;  
 44 ha 1 PIA H 44 0.103024 1.00800 ;  
 45 ha 1 PIA H1 45 0.126351 1.00800 ;  
 46 ha 1 PIA H2 46 0.114013 1.00800 ;  
 47 ha 1 PIA H3 47 0.143870 1.00800 ;  
 48 ha 1 PIA H4 48 0.152776 1.00800 ;  
 49 ha 1 PIA H5 49 0.152378 1.00800 ;  
 50 ha 1 PIA H6 50 0.113906 1.00800 ;  
 51 ha 1 PIA H7 51 0.126714 1.00800 ;

52	ha	1	PIA	H8	52	0.103501	1.00800 ;
53	ha	1	PIA	H9	53	0.144375	1.00800 ;
54	ha	1	PIA	H10	54	0.175565	1.00800 ;
55	ha	1	PIA	H11	55	0.161889	1.00800 ;
56	ha	1	PIA	H12	56	0.160612	1.00800 ;
57	ha	1	PIA	H13	57	0.175227	1.00800 ;
58	c3	1	PIA	C38	58	-0.311950	12.01000 ;
59	hc	1	PIA	H14	59	-0.010235	1.00800 ;
60	c3	1	PIA	C39	60	-0.673244	12.01000 ;
61	c3	1	PIA	C40	61	-0.315322	12.01000 ;
62	c3	1	PIA	C41	62	-0.828149	12.01000 ;
63	hc	1	PIA	H15	63	-0.023730	1.00800 ;
64	c3	1	PIA	C42	64	-0.430976	12.01000 ;
65	hc	1	PIA	H16	65	-0.058729	1.00800 ;
66	c3	1	PIA	C43	66	-0.241386	12.01000 ;
67	c3	1	PIA	C44	67	-0.827532	12.01000 ;
68	hc	1	PIA	H17	68	-0.024567	1.00800 ;
69	c3	1	PIA	C45	69	-0.317438	12.01000 ;
70	c3	1	PIA	C46	70	-0.674627	12.01000 ;
71	c3	1	PIA	C47	71	-0.311715	12.01000 ;
72	hc	1	PIA	H18	72	-0.009849	1.00800 ;
73	c3	1	PIA	C48	73	-0.239919	12.01000 ;
74	c3	1	PIA	C49	74	-0.429416	12.01000 ;
75	hc	1	PIA	H19	75	-0.058438	1.00800 ;
76	hc	1	PIA	H20	76	0.065650	1.00800 ;
77	hc	1	PIA	H21	77	0.065650	1.00800 ;
78	hc	1	PIA	H22	78	0.065650	1.00800 ;
79	hc	1	PIA	H23	79	0.138721	1.00800 ;
80	hc	1	PIA	H24	80	0.138721	1.00800 ;

81	hc	1	PIA	H25	81	0.138721	1.00800 ;
82	hc	1	PIA	H26	82	0.046179	1.00800 ;
83	hc	1	PIA	H27	83	0.046179	1.00800 ;
84	hc	1	PIA	H28	84	0.046179	1.00800 ;
85	hc	1	PIA	H29	85	0.085577	1.00800 ;
86	hc	1	PIA	H30	86	0.085577	1.00800 ;
87	hc	1	PIA	H31	87	0.085577	1.00800 ;
88	hc	1	PIA	H32	88	0.166897	1.00800 ;
89	hc	1	PIA	H33	89	0.166897	1.00800 ;
90	hc	1	PIA	H34	90	0.166897	1.00800 ;
91	hc	1	PIA	H35	91	0.064411	1.00800 ;
92	hc	1	PIA	H36	92	0.064411	1.00800 ;
93	hc	1	PIA	H37	93	0.064411	1.00800 ;
94	hc	1	PIA	H38	94	0.065673	1.00800 ;
95	hc	1	PIA	H39	95	0.065673	1.00800 ;
96	hc	1	PIA	H40	96	0.065673	1.00800 ;
97	hc	1	PIA	H41	97	0.139000	1.00800 ;
98	hc	1	PIA	H42	98	0.139000	1.00800 ;
99	hc	1	PIA	H43	99	0.139000	1.00800 ;
100	hc	1	PIA	H44	100	0.045808	1.00800 ;
101	hc	1	PIA	H45	101	0.045808	1.00800 ;
102	hc	1	PIA	H46	102	0.045808	1.00800 ;
103	hc	1	PIA	H47	103	0.085269	1.00800 ;
104	hc	1	PIA	H48	104	0.085269	1.00800 ;
105	hc	1	PIA	H49	105	0.085269	1.00800 ;
106	hc	1	PIA	H50	106	0.064991	1.00800 ;
107	hc	1	PIA	H51	107	0.064991	1.00800 ;
108	hc	1	PIA	H52	108	0.064991	1.00800 ;
109	hc	1	PIA	H53	109	0.166545	1.00800 ;

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110 hc 1 PIA H54 110 0.166545 1.00800 ;
111 hc 1 PIA H55 111 0.166545 1.00800 ;

[ bonds ]

; ai aj funct r k

1 2 1 1.3984e-01 3.8585e+05 ; C - C1
1 6 1 1.3984e-01 3.8585e+05 ; C - C5
1 44 1 1.0860e-01 2.8937e+05 ; C - H
2 3 1 1.3984e-01 3.8585e+05 ; C1 - C2
2 45 1 1.0860e-01 2.8937e+05 ; C1 - H1
3 4 1 1.3984e-01 3.8585e+05 ; C2 - C3
3 46 1 1.0860e-01 2.8937e+05 ; C2 - H2
4 5 1 1.3984e-01 3.8585e+05 ; C3 - C4
4 7 1 1.4058e-01 3.7673e+05 ; C3 - C6
5 6 1 1.3984e-01 3.8585e+05 ; C4 - C5
5 10 1 1.3984e-01 3.8585e+05 ; C4 - C9
6 47 1 1.0860e-01 2.8937e+05 ; C5 - H3
7 8 1 1.4058e-01 3.7673e+05 ; C6 - C7
7 11 1 1.4854e-01 2.9405e+05 ; C6 - C10
8 9 1 1.4058e-01 3.7673e+05 ; C7 - C8
8 21 1 1.3696e-01 3.1514e+05 ; C7 - O
9 10 1 1.4058e-01 3.7673e+05 ; C8 - C9
9 26 1 1.4854e-01 2.9405e+05 ; C8 - C20
10 48 1 1.0860e-01 2.8937e+05 ; C9 - H4
11 12 1 1.4058e-01 3.7673e+05 ; C10 - C11
11 16 1 1.4058e-01 3.7673e+05 ; C10 - C15
12 13 1 1.3984e-01 3.8585e+05 ; C11 - C12
12 17 1 1.3984e-01 3.8585e+05 ; C11 - C16
13 14 1 1.3984e-01 3.8585e+05 ; C12 - C13
13 20 1 1.3984e-01 3.8585e+05 ; C12 - C19

```

14	15	1	1.4058e-01	3.7673e+05 ;	C13 - C14
14	49	1	1.0860e-01	2.8937e+05 ;	C13 - H5
15	16	1	1.4058e-01	3.7673e+05 ;	C14 - C15
15	27	1	1.4854e-01	2.9405e+05 ;	C14 - C21
16	25	1	1.3696e-01	3.1514e+05 ;	C15 - O3
17	18	1	1.3984e-01	3.8585e+05 ;	C16 - C17
17	50	1	1.0860e-01	2.8937e+05 ;	C16 - H6
18	19	1	1.3984e-01	3.8585e+05 ;	C17 - C18
18	51	1	1.0860e-01	2.8937e+05 ;	C17 - H7
19	20	1	1.3984e-01	3.8585e+05 ;	C18 - C19
19	52	1	1.0860e-01	2.8937e+05 ;	C18 - H8
20	53	1	1.0860e-01	2.8937e+05 ;	C19 - H9
21	22	1	1.6147e-01	2.7665e+05 ;	O - P
22	23	1	1.4866e-01	4.0125e+05 ;	P - O1
22	24	1	1.4866e-01	4.0125e+05 ;	P - O2
22	25	1	1.6147e-01	2.7665e+05 ;	P - O3
26	33	1	1.4058e-01	3.7673e+05 ;	C20 - C27
26	37	1	1.4058e-01	3.7673e+05 ;	C20 - C31
27	28	1	1.4058e-01	3.7673e+05 ;	C21 - C22
27	32	1	1.4058e-01	3.7673e+05 ;	C21 - C26
28	29	1	1.3984e-01	3.8585e+05 ;	C22 - C23
28	42	1	1.5156e-01	2.6861e+05 ;	C22 - C36
29	30	1	1.3984e-01	3.8585e+05 ;	C23 - C24
29	54	1	1.0860e-01	2.8937e+05 ;	C23 - H10
30	31	1	1.3984e-01	3.8585e+05 ;	C24 - C25
30	43	1	1.5156e-01	2.6861e+05 ;	C24 - C37
31	32	1	1.3984e-01	3.8585e+05 ;	C25 - C26
31	55	1	1.0860e-01	2.8937e+05 ;	C25 - H11
32	41	1	1.5156e-01	2.6861e+05 ;	C26 - C35

33	34	1	1.3984e-01	3.8585e+05 ;	C27 - C28
33	39	1	1.5156e-01	2.6861e+05 ;	C27 - C33
34	35	1	1.3984e-01	3.8585e+05 ;	C28 - C29
34	56	1	1.0860e-01	2.8937e+05 ;	C28 - H12
35	36	1	1.3984e-01	3.8585e+05 ;	C29 - C30
35	40	1	1.5156e-01	2.6861e+05 ;	C29 - C34
36	37	1	1.3984e-01	3.8585e+05 ;	C30 - C31
36	57	1	1.0860e-01	2.8937e+05 ;	C30 - H13
37	38	1	1.5156e-01	2.6861e+05 ;	C31 - C32
38	58	1	1.5375e-01	2.5179e+05 ;	C32 - C38
38	59	1	1.0969e-01	2.7665e+05 ;	C32 - H14
38	60	1	1.5375e-01	2.5179e+05 ;	C32 - C39
39	61	1	1.5375e-01	2.5179e+05 ;	C33 - C40
39	62	1	1.5375e-01	2.5179e+05 ;	C33 - C41
39	63	1	1.0969e-01	2.7665e+05 ;	C33 - H15
40	64	1	1.5375e-01	2.5179e+05 ;	C34 - C42
40	65	1	1.0969e-01	2.7665e+05 ;	C34 - H16
40	66	1	1.5375e-01	2.5179e+05 ;	C34 - C43
41	67	1	1.5375e-01	2.5179e+05 ;	C35 - C44
41	68	1	1.0969e-01	2.7665e+05 ;	C35 - H17
41	69	1	1.5375e-01	2.5179e+05 ;	C35 - C45
42	70	1	1.5375e-01	2.5179e+05 ;	C36 - C46
42	71	1	1.5375e-01	2.5179e+05 ;	C36 - C47
42	72	1	1.0969e-01	2.7665e+05 ;	C36 - H18
43	73	1	1.5375e-01	2.5179e+05 ;	C37 - C48
43	74	1	1.5375e-01	2.5179e+05 ;	C37 - C49
43	75	1	1.0969e-01	2.7665e+05 ;	C37 - H19
58	76	1	1.0969e-01	2.7665e+05 ;	C38 - H20
58	77	1	1.0969e-01	2.7665e+05 ;	C38 - H21

58	78	1	1.0969e-01	2.7665e+05 ;	C38 - H22
60	79	1	1.0969e-01	2.7665e+05 ;	C39 - H23
60	80	1	1.0969e-01	2.7665e+05 ;	C39 - H24
60	81	1	1.0969e-01	2.7665e+05 ;	C39 - H25
61	91	1	1.0969e-01	2.7665e+05 ;	C40 - H35
61	92	1	1.0969e-01	2.7665e+05 ;	C40 - H36
61	93	1	1.0969e-01	2.7665e+05 ;	C40 - H37
62	88	1	1.0969e-01	2.7665e+05 ;	C41 - H32
62	89	1	1.0969e-01	2.7665e+05 ;	C41 - H33
62	90	1	1.0969e-01	2.7665e+05 ;	C41 - H34
64	85	1	1.0969e-01	2.7665e+05 ;	C42 - H29
64	86	1	1.0969e-01	2.7665e+05 ;	C42 - H30
64	87	1	1.0969e-01	2.7665e+05 ;	C42 - H31
66	82	1	1.0969e-01	2.7665e+05 ;	C43 - H26
66	83	1	1.0969e-01	2.7665e+05 ;	C43 - H27
66	84	1	1.0969e-01	2.7665e+05 ;	C43 - H28
67	109	1	1.0969e-01	2.7665e+05 ;	C44 - H53
67	110	1	1.0969e-01	2.7665e+05 ;	C44 - H54
67	111	1	1.0969e-01	2.7665e+05 ;	C44 - H55
69	106	1	1.0969e-01	2.7665e+05 ;	C45 - H50
69	107	1	1.0969e-01	2.7665e+05 ;	C45 - H51
69	108	1	1.0969e-01	2.7665e+05 ;	C45 - H52
70	97	1	1.0969e-01	2.7665e+05 ;	C46 - H41
70	98	1	1.0969e-01	2.7665e+05 ;	C46 - H42
70	99	1	1.0969e-01	2.7665e+05 ;	C46 - H43
71	94	1	1.0969e-01	2.7665e+05 ;	C47 - H38
71	95	1	1.0969e-01	2.7665e+05 ;	C47 - H39
71	96	1	1.0969e-01	2.7665e+05 ;	C47 - H40
73	100	1	1.0969e-01	2.7665e+05 ;	C48 - H44

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73 101 1 1.0969e-01 2.7665e+05 ; C48 - H45
73 102 1 1.0969e-01 2.7665e+05 ; C48 - H46
74 103 1 1.0969e-01 2.7665e+05 ; C49 - H47
74 104 1 1.0969e-01 2.7665e+05 ; C49 - H48
74 105 1 1.0969e-01 2.7665e+05 ; C49 - H49

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[ pairs ]

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; ai aj funct
1 4 1; C - C3
1 10 1; C - C9
1 46 1; C - H2
2 5 1; C1 - C4
2 7 1; C1 - C6
2 47 1; C1 - H3
3 8 1; C2 - C7
3 10 1; C2 - C9
3 11 1; C2 - C10
4 9 1; C3 - C8
4 12 1; C3 - C11
4 16 1; C3 - C15
4 21 1; C3 - O
4 45 1; C3 - H1
4 47 1; C3 - H3
4 48 1; C3 - H4
5 8 1; C4 - C7
5 11 1; C4 - C10
5 26 1; C4 - C20
5 46 1; C4 - H2
6 3 1; C5 - C2

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6 7 1; C5 - C6  
6 9 1; C5 - C8  
6 45 1; C5 - H1  
6 48 1; C5 - H4  
7 10 1; C6 - C9  
7 13 1; C6 - C12  
7 15 1; C6 - C14  
7 17 1; C6 - C16  
7 22 1; C6 - P  
7 25 1; C6 - O3  
7 26 1; C6 - C20  
7 46 1; C6 - H2  
8 12 1; C7 - C11  
8 16 1; C7 - C15  
8 23 1; C7 - O1  
8 24 1; C7 - O2  
8 25 1; C7 - O3  
8 33 1; C7 - C27  
8 37 1; C7 - C31  
8 48 1; C7 - H4  
9 11 1; C8 - C10  
9 22 1; C8 - P  
9 34 1; C8 - C28  
9 36 1; C8 - C30  
9 38 1; C8 - C32  
9 39 1; C8 - C33  
10 21 1; C9 - O  
10 33 1; C9 - C27  
10 37 1; C9 - C31

10 47 1; C9 - H3  
11 14 1; C10 - C13  
11 18 1; C10 - C17  
11 20 1; C10 - C19  
11 21 1; C10 - O  
11 22 1; C10 - P  
11 27 1; C10 - C21  
11 50 1; C10 - H6  
12 15 1; C11 - C14  
12 19 1; C11 - C18  
12 25 1; C11 - O3  
12 49 1; C11 - H5  
12 51 1; C11 - H7  
12 53 1; C11 - H9  
13 16 1; C12 - C15  
13 18 1; C12 - C17  
13 27 1; C12 - C21  
13 50 1; C12 - H6  
13 52 1; C12 - H8  
14 17 1; C13 - C16  
14 19 1; C13 - C18  
14 25 1; C13 - O3  
14 28 1; C13 - C22  
14 32 1; C13 - C26  
14 53 1; C13 - H9  
15 20 1; C14 - C19  
15 22 1; C14 - P  
15 29 1; C14 - C23  
15 31 1; C14 - C25

15 41 1; C14 - C35  
15 42 1; C14 - C36  
16 17 1; C15 - C16  
16 21 1; C15 - O  
16 23 1; C15 - O1  
16 24 1; C15 - O2  
16 28 1; C15 - C22  
16 32 1; C15 - C26  
16 49 1; C15 - H5  
17 20 1; C16 - C19  
17 52 1; C16 - H8  
18 53 1; C17 - H9  
19 50 1; C18 - H6  
20 49 1; C19 - H5  
20 51 1; C19 - H7  
21 26 1; O - C20  
25 27 1; O3 - C21  
26 35 1; C20 - C29  
26 48 1; C20 - H4  
26 56 1; C20 - H12  
26 57 1; C20 - H13  
26 58 1; C20 - C38  
26 59 1; C20 - H14  
26 60 1; C20 - C39  
26 61 1; C20 - C40  
26 62 1; C20 - C41  
26 63 1; C20 - H15  
27 30 1; C21 - C24  
27 49 1; C21 - H5

27 54 1; C21 - H10  
27 55 1; C21 - H11  
27 67 1; C21 - C44  
27 68 1; C21 - H17  
27 69 1; C21 - C45  
27 70 1; C21 - C46  
27 71 1; C21 - C47  
27 72 1; C21 - H18  
28 31 1; C22 - C25  
28 41 1; C22 - C35  
28 43 1; C22 - C37  
28 94 1; C22 - H38  
28 95 1; C22 - H39  
28 96 1; C22 - H40  
28 97 1; C22 - H41  
28 98 1; C22 - H42  
28 99 1; C22 - H43  
29 32 1; C23 - C26  
29 55 1; C23 - H11  
29 70 1; C23 - C46  
29 71 1; C23 - C47  
29 72 1; C23 - H18  
29 73 1; C23 - C48  
29 74 1; C23 - C49  
29 75 1; C23 - H19  
30 41 1; C24 - C35  
30 42 1; C24 - C36  
30 100 1; C24 - H44  
30 101 1; C24 - H45

30 102 1; C24 - H46  
30 103 1; C24 - H47  
30 104 1; C24 - H48  
30 105 1; C24 - H49  
31 54 1; C25 - H10  
31 67 1; C25 - C44  
31 68 1; C25 - H17  
31 69 1; C25 - C45  
31 73 1; C25 - C48  
31 74 1; C25 - C49  
31 75 1; C25 - H19  
32 42 1; C26 - C36  
32 43 1; C26 - C37  
32 106 1; C26 - H50  
32 107 1; C26 - H51  
32 108 1; C26 - H52  
32 109 1; C26 - H53  
32 110 1; C26 - H54  
32 111 1; C26 - H55  
33 36 1; C27 - C30  
33 38 1; C27 - C32  
33 40 1; C27 - C34  
33 88 1; C27 - H32  
33 89 1; C27 - H33  
33 90 1; C27 - H34  
33 91 1; C27 - H35  
33 92 1; C27 - H36  
33 93 1; C27 - H37  
34 37 1; C28 - C31

34 57 1; C28 - H13  
34 61 1; C28 - C40  
34 62 1; C28 - C41  
34 63 1; C28 - H15  
34 64 1; C28 - C42  
34 65 1; C28 - H16  
34 66 1; C28 - C43  
35 38 1; C29 - C32  
35 39 1; C29 - C33  
35 82 1; C29 - H26  
35 83 1; C29 - H27  
35 84 1; C29 - H28  
35 85 1; C29 - H29  
35 86 1; C29 - H30  
35 87 1; C29 - H31  
36 56 1; C30 - H12  
36 58 1; C30 - C38  
36 59 1; C30 - H14  
36 60 1; C30 - C39  
36 64 1; C30 - C42  
36 65 1; C30 - H16  
36 66 1; C30 - C43  
37 39 1; C31 - C33  
37 40 1; C31 - C34  
37 76 1; C31 - H20  
37 77 1; C31 - H21  
37 78 1; C31 - H22  
37 79 1; C31 - H23  
37 80 1; C31 - H24

37 81 1; C31 - H25  
38 57 1; C32 - H13  
39 56 1; C33 - H12  
40 56 1; C34 - H12  
40 57 1; C34 - H13  
41 55 1; C35 - H11  
42 54 1; C36 - H10  
43 54 1; C37 - H10  
43 55 1; C37 - H11  
44 3 1; H - C2  
44 5 1; H - C4  
44 45 1; H - H1  
44 47 1; H - H3  
45 46 1; H1 - H2  
50 51 1; H6 - H7  
51 52 1; H7 - H8  
52 53 1; H8 - H9  
58 79 1; C38 - H23  
58 80 1; C38 - H24  
58 81 1; C38 - H25  
59 76 1; H14 - H20  
59 77 1; H14 - H21  
59 78 1; H14 - H22  
59 79 1; H14 - H23  
59 80 1; H14 - H24  
59 81 1; H14 - H25  
60 76 1; C39 - H20  
60 77 1; C39 - H21  
60 78 1; C39 - H22

61 88 1; C40 - H32  
61 89 1; C40 - H33  
61 90 1; C40 - H34  
62 91 1; C41 - H35  
62 92 1; C41 - H36  
62 93 1; C41 - H37  
63 88 1; H15 - H32  
63 89 1; H15 - H33  
63 90 1; H15 - H34  
63 91 1; H15 - H35  
63 92 1; H15 - H36  
63 93 1; H15 - H37  
64 82 1; C42 - H26  
64 83 1; C42 - H27  
64 84 1; C42 - H28  
65 82 1; H16 - H26  
65 83 1; H16 - H27  
65 84 1; H16 - H28  
65 85 1; H16 - H29  
65 86 1; H16 - H30  
65 87 1; H16 - H31  
66 85 1; C43 - H29  
66 86 1; C43 - H30  
66 87 1; C43 - H31  
67 106 1; C44 - H50  
67 107 1; C44 - H51  
67 108 1; C44 - H52  
68 106 1; H17 - H50  
68 107 1; H17 - H51

68 108 1; H17 - H52  
68 109 1; H17 - H53  
68 110 1; H17 - H54  
68 111 1; H17 - H55  
69 109 1; C45 - H53  
69 110 1; C45 - H54  
69 111 1; C45 - H55  
70 94 1; C46 - H38  
70 95 1; C46 - H39  
70 96 1; C46 - H40  
71 97 1; C47 - H41  
71 98 1; C47 - H42  
71 99 1; C47 - H43  
72 94 1; H18 - H38  
72 95 1; H18 - H39  
72 96 1; H18 - H40  
72 97 1; H18 - H41  
72 98 1; H18 - H42  
72 99 1; H18 - H43  
73 103 1; C48 - H47  
73 104 1; C48 - H48  
73 105 1; C48 - H49  
74 100 1; C49 - H44  
74 101 1; C49 - H45  
74 102 1; C49 - H46  
75 100 1; H19 - H44  
75 101 1; H19 - H45  
75 102 1; H19 - H46  
75 103 1; H19 - H47

75 104 1; H19 - H48

75 105 1; H19 - H49

[ angles ]

; ai aj ak funct theta cth

1 2 3 1 1.2002e+02 5.5731e+02; C - C1 - C2

1 2 45 1 1.1988e+02 4.0334e+02; C - C1 - H1

1 6 5 1 1.2002e+02 5.5731e+02; C - C5 - C4

1 6 47 1 1.1988e+02 4.0334e+02; C - C5 - H3

2 1 6 1 1.2002e+02 5.5731e+02; C1 - C - C5

2 1 44 1 1.1988e+02 4.0334e+02; C1 - C - H

2 3 4 1 1.2002e+02 5.5731e+02; C1 - C2 - C3

2 3 46 1 1.1988e+02 4.0334e+02; C1 - C2 - H2

3 2 45 1 1.1988e+02 4.0334e+02; C2 - C1 - H1

3 4 5 1 1.2002e+02 5.5731e+02; C2 - C3 - C4

3 4 7 1 1.2069e+02 5.5480e+02; C2 - C3 - C6

4 3 46 1 1.1988e+02 4.0334e+02; C3 - C2 - H2

4 5 6 1 1.2002e+02 5.5731e+02; C3 - C4 - C5

4 5 10 1 1.2002e+02 5.5731e+02; C3 - C4 - C9

4 7 8 1 1.1838e+02 5.5815e+02; C3 - C6 - C7

4 7 11 1 1.2111e+02 5.3555e+02; C3 - C6 - C10

5 4 7 1 1.2069e+02 5.5480e+02; C4 - C3 - C6

5 6 47 1 1.1988e+02 4.0334e+02; C4 - C5 - H3

5 10 9 1 1.2069e+02 5.5480e+02; C4 - C9 - C8

5 10 48 1 1.1988e+02 4.0334e+02; C4 - C9 - H4

6 1 44 1 1.1988e+02 4.0334e+02; C5 - C - H

6 5 10 1 1.2002e+02 5.5731e+02; C5 - C4 - C9

7 8 9 1 1.1152e+02 5.7572e+02; C6 - C7 - C8

7 8 21 1 1.1691e+02 5.8660e+02; C6 - C7 - O

7	11	12	1	1.2111e+02	5.3555e+02 ;	C6 - C10	- C11
7	11	16	1	1.2111e+02	5.3555e+02 ;	C6 - C10	- C15
8	7	11	1	1.2111e+02	5.3555e+02 ;	C7 - C6	- C10
8	9	10	1	1.1838e+02	5.5815e+02 ;	C7 - C8	- C9
8	9	26	1	1.2111e+02	5.3555e+02 ;	C7 - C8	- C20
8	21	22	1	1.2318e+02	6.5019e+02 ;	C7 - O	- P
9	8	21	1	1.1691e+02	5.8660e+02 ;	C8 - C7	- O
9	10	48	1	1.1986e+02	4.0166e+02 ;	C8 - C9	- H4
9	26	33	1	1.2111e+02	5.3555e+02 ;	C8 - C20	- C27
9	26	37	1	1.2111e+02	5.3555e+02 ;	C8 - C20	- C31
10	9	26	1	1.2111e+02	5.3555e+02 ;	C9 - C8	- C20
11	12	13	1	1.2069e+02	5.5480e+02 ;	C10 - C11	- C12
11	12	17	1	1.2069e+02	5.5480e+02 ;	C10 - C11	- C16
11	16	15	1	1.1152e+02	5.7572e+02 ;	C10 - C15	- C14
11	16	25	1	1.1691e+02	5.8660e+02 ;	C10 - C15	- O3
12	11	16	1	1.1838e+02	5.5815e+02 ;	C11 - C10	- C15
12	13	14	1	1.2002e+02	5.5731e+02 ;	C11 - C12	- C13
12	13	20	1	1.2002e+02	5.5731e+02 ;	C11 - C12	- C19
12	17	18	1	1.2002e+02	5.5731e+02 ;	C11 - C16	- C17
12	17	50	1	1.1988e+02	4.0334e+02 ;	C11 - C16	- H6
13	12	17	1	1.2002e+02	5.5731e+02 ;	C12 - C11	- C16
13	14	15	1	1.2069e+02	5.5480e+02 ;	C12 - C13	- C14
13	14	49	1	1.1988e+02	4.0334e+02 ;	C12 - C13	- H5
13	20	19	1	1.2002e+02	5.5731e+02 ;	C12 - C19	- C18
13	20	53	1	1.1988e+02	4.0334e+02 ;	C12 - C19	- H9
14	13	20	1	1.2002e+02	5.5731e+02 ;	C13 - C12	- C19
14	15	16	1	1.1838e+02	5.5815e+02 ;	C13 - C14	- C15
14	15	27	1	1.2111e+02	5.3555e+02 ;	C13 - C14	- C21
15	14	49	1	1.1986e+02	4.0166e+02 ;	C14 - C13	- H5

15	16	25	1	1.1691e+02	5.8660e+02 ;	C14 - C15	- O3
15	27	28	1	1.2111e+02	5.3555e+02 ;	C14 - C21	- C22
15	27	32	1	1.2111e+02	5.3555e+02 ;	C14 - C21	- C26
16	15	27	1	1.2111e+02	5.3555e+02 ;	C15 - C14	- C21
16	25	22	1	1.2318e+02	6.5019e+02 ;	C15 - O3	- P
17	18	19	1	1.2002e+02	5.5731e+02 ;	C16 - C17	- C18
17	18	51	1	1.1988e+02	4.0334e+02 ;	C16 - C17	- H7
18	17	50	1	1.1988e+02	4.0334e+02 ;	C17 - C16	- H6
18	19	20	1	1.2002e+02	5.5731e+02 ;	C17 - C18	- C19
18	19	52	1	1.1988e+02	4.0334e+02 ;	C17 - C18	- H8
19	18	51	1	1.1988e+02	4.0334e+02 ;	C18 - C17	- H7
19	20	53	1	1.1988e+02	4.0334e+02 ;	C18 - C19	- H9
20	19	52	1	1.1988e+02	4.0334e+02 ;	C19 - C18	- H8
21	22	23	1	1.1546e+02	3.6736e+02 ;	O - P	- O1
21	22	24	1	1.1546e+02	3.6736e+02 ;	O - P	- O2
21	22	25	1	1.0184e+02	3.7656e+02 ;	O - P	- O3
23	22	24	1	1.1580e+02	3.8325e+02 ;	O1 - P	- O2
23	22	25	1	1.1546e+02	3.6736e+02 ;	O1 - P	- O3
24	22	25	1	1.1546e+02	3.6736e+02 ;	O2 - P	- O3
26	33	34	1	1.2069e+02	5.5480e+02 ;	C20 - C27	- C28
26	33	39	1	1.2063e+02	5.3053e+02 ;	C20 - C27	- C33
26	37	36	1	1.2069e+02	5.5480e+02 ;	C20 - C31	- C30
26	37	38	1	1.2063e+02	5.3053e+02 ;	C20 - C31	- C32
27	28	29	1	1.2069e+02	5.5480e+02 ;	C21 - C22	- C23
27	28	42	1	1.2063e+02	5.3053e+02 ;	C21 - C22	- C36
27	32	31	1	1.2069e+02	5.5480e+02 ;	C21 - C26	- C25
27	32	41	1	1.2063e+02	5.3053e+02 ;	C21 - C26	- C35
28	27	32	1	1.1838e+02	5.5815e+02 ;	C22 - C21	- C26
28	29	30	1	1.2002e+02	5.5731e+02 ;	C22 - C23	- C24

28	29	54	1	1.1988e+02	4.0334e+02 ;	C22 - C23	- H10
28	42	70	1	1.1207e+02	5.2802e+02 ;	C22 - C36	- C46
28	42	71	1	1.1207e+02	5.2802e+02 ;	C22 - C36	- C47
28	42	72	1	1.1047e+02	3.9162e+02 ;	C22 - C36	- H18
29	28	42	1	1.2077e+02	5.3137e+02 ;	C23 - C22	- C36
29	30	31	1	1.2002e+02	5.5731e+02 ;	C23 - C24	- C25
29	30	43	1	1.2077e+02	5.3137e+02 ;	C23 - C24	- C37
30	29	54	1	1.1988e+02	4.0334e+02 ;	C24 - C23	- H10
30	31	32	1	1.2002e+02	5.5731e+02 ;	C24 - C25	- C26
30	31	55	1	1.1988e+02	4.0334e+02 ;	C24 - C25	- H11
30	43	73	1	1.1207e+02	5.2802e+02 ;	C24 - C37	- C48
30	43	74	1	1.1207e+02	5.2802e+02 ;	C24 - C37	- C49
30	43	75	1	1.1047e+02	3.9162e+02 ;	C24 - C37	- H19
31	30	43	1	1.2077e+02	5.3137e+02 ;	C25 - C24	- C37
31	32	41	1	1.2077e+02	5.3137e+02 ;	C25 - C26	- C35
32	31	55	1	1.1988e+02	4.0334e+02 ;	C26 - C25	- H11
32	41	67	1	1.1207e+02	5.2802e+02 ;	C26 - C35	- C44
32	41	68	1	1.1047e+02	3.9162e+02 ;	C26 - C35	- H17
32	41	69	1	1.1207e+02	5.2802e+02 ;	C26 - C35	- C45
33	26	37	1	1.1838e+02	5.5815e+02 ;	C27 - C20	- C31
33	34	35	1	1.2002e+02	5.5731e+02 ;	C27 - C28	- C29
33	34	56	1	1.1988e+02	4.0334e+02 ;	C27 - C28	- H12
33	39	61	1	1.1207e+02	5.2802e+02 ;	C27 - C33	- C40
33	39	62	1	1.1207e+02	5.2802e+02 ;	C27 - C33	- C41
33	39	63	1	1.1047e+02	3.9162e+02 ;	C27 - C33	- H15
34	33	39	1	1.2077e+02	5.3137e+02 ;	C28 - C27	- C33
34	35	36	1	1.2002e+02	5.5731e+02 ;	C28 - C29	- C30
34	35	40	1	1.2077e+02	5.3137e+02 ;	C28 - C29	- C34
35	34	56	1	1.1988e+02	4.0334e+02 ;	C29 - C28	- H12

35	36	37	1	1.2002e+02	5.5731e+02 ;	C29 - C30	- C31
35	36	57	1	1.1988e+02	4.0334e+02 ;	C29 - C30	- H13
35	40	64	1	1.1207e+02	5.2802e+02 ;	C29 - C34	- C42
35	40	65	1	1.1047e+02	3.9162e+02 ;	C29 - C34	- H16
35	40	66	1	1.1207e+02	5.2802e+02 ;	C29 - C34	- C43
36	35	40	1	1.2077e+02	5.3137e+02 ;	C30 - C29	- C34
36	37	38	1	1.2077e+02	5.3137e+02 ;	C30 - C31	- C32
37	36	57	1	1.1988e+02	4.0334e+02 ;	C31 - C30	- H13
37	38	58	1	1.1207e+02	5.2802e+02 ;	C31 - C32	- C38
37	38	59	1	1.1047e+02	3.9162e+02 ;	C31 - C32	- H14
37	38	60	1	1.1207e+02	5.2802e+02 ;	C31 - C32	- C39
38	58	76	1	1.0980e+02	3.8744e+02 ;	C32 - C38	- H20
38	58	77	1	1.0980e+02	3.8744e+02 ;	C32 - C38	- H21
38	58	78	1	1.0980e+02	3.8744e+02 ;	C32 - C38	- H22
38	60	79	1	1.0980e+02	3.8744e+02 ;	C32 - C39	- H23
38	60	80	1	1.0980e+02	3.8744e+02 ;	C32 - C39	- H24
38	60	81	1	1.0980e+02	3.8744e+02 ;	C32 - C39	- H25
39	61	91	1	1.0980e+02	3.8744e+02 ;	C33 - C40	- H35
39	61	92	1	1.0980e+02	3.8744e+02 ;	C33 - C40	- H36
39	61	93	1	1.0980e+02	3.8744e+02 ;	C33 - C40	- H37
39	62	88	1	1.0980e+02	3.8744e+02 ;	C33 - C41	- H32
39	62	89	1	1.0980e+02	3.8744e+02 ;	C33 - C41	- H33
39	62	90	1	1.0980e+02	3.8744e+02 ;	C33 - C41	- H34
40	64	85	1	1.0980e+02	3.8744e+02 ;	C34 - C42	- H29
40	64	86	1	1.0980e+02	3.8744e+02 ;	C34 - C42	- H30
40	64	87	1	1.0980e+02	3.8744e+02 ;	C34 - C42	- H31
40	66	82	1	1.0980e+02	3.8744e+02 ;	C34 - C43	- H26
40	66	83	1	1.0980e+02	3.8744e+02 ;	C34 - C43	- H27
40	66	84	1	1.0980e+02	3.8744e+02 ;	C34 - C43	- H28

41	67	109	1	1.0980e+02	3.8744e+02 ;	C35 - C44	- H53
41	67	110	1	1.0980e+02	3.8744e+02 ;	C35 - C44	- H54
41	67	111	1	1.0980e+02	3.8744e+02 ;	C35 - C44	- H55
41	69	106	1	1.0980e+02	3.8744e+02 ;	C35 - C45	- H50
41	69	107	1	1.0980e+02	3.8744e+02 ;	C35 - C45	- H51
41	69	108	1	1.0980e+02	3.8744e+02 ;	C35 - C45	- H52
42	70	97	1	1.0980e+02	3.8744e+02 ;	C36 - C46	- H41
42	70	98	1	1.0980e+02	3.8744e+02 ;	C36 - C46	- H42
42	70	99	1	1.0980e+02	3.8744e+02 ;	C36 - C46	- H43
42	71	94	1	1.0980e+02	3.8744e+02 ;	C36 - C47	- H38
42	71	95	1	1.0980e+02	3.8744e+02 ;	C36 - C47	- H39
42	71	96	1	1.0980e+02	3.8744e+02 ;	C36 - C47	- H40
43	73	100	1	1.0980e+02	3.8744e+02 ;	C37 - C48	- H44
43	73	101	1	1.0980e+02	3.8744e+02 ;	C37 - C48	- H45
43	73	102	1	1.0980e+02	3.8744e+02 ;	C37 - C48	- H46
43	74	103	1	1.0980e+02	3.8744e+02 ;	C37 - C49	- H47
43	74	104	1	1.0980e+02	3.8744e+02 ;	C37 - C49	- H48
43	74	105	1	1.0980e+02	3.8744e+02 ;	C37 - C49	- H49
58	38	59	1	1.0980e+02	3.8744e+02 ;	C38 - C32	- H14
58	38	60	1	1.1151e+02	5.2635e+02 ;	C38 - C32	- C39
59	38	60	1	1.0980e+02	3.8744e+02 ;	H14 - C32	- C39
61	39	62	1	1.1151e+02	5.2635e+02 ;	C40 - C33	- C41
61	39	63	1	1.0980e+02	3.8744e+02 ;	C40 - C33	- H15
62	39	63	1	1.0980e+02	3.8744e+02 ;	C41 - C33	- H15
64	40	65	1	1.0980e+02	3.8744e+02 ;	C42 - C34	- H16
64	40	66	1	1.1151e+02	5.2635e+02 ;	C42 - C34	- C43
65	40	66	1	1.0980e+02	3.8744e+02 ;	H16 - C34	- C43
67	41	68	1	1.0980e+02	3.8744e+02 ;	C44 - C35	- H17
67	41	69	1	1.1151e+02	5.2635e+02 ;	C44 - C35	- C45

68	41	69	1	1.0980e+02	3.8744e+02 ;	H17 - C35	- C45
70	42	71	1	1.1151e+02	5.2635e+02 ;	C46 - C36	- C47
70	42	72	1	1.0980e+02	3.8744e+02 ;	C46 - C36	- H18
71	42	72	1	1.0980e+02	3.8744e+02 ;	C47 - C36	- H18
73	43	74	1	1.1151e+02	5.2635e+02 ;	C48 - C37	- C49
73	43	75	1	1.0980e+02	3.8744e+02 ;	C48 - C37	- H19
74	43	75	1	1.0980e+02	3.8744e+02 ;	C49 - C37	- H19
76	58	77	1	1.0758e+02	3.2970e+02 ;	H20 - C38	- H21
76	58	78	1	1.0758e+02	3.2970e+02 ;	H20 - C38	- H22
77	58	78	1	1.0758e+02	3.2970e+02 ;	H21 - C38	- H22
79	60	80	1	1.0758e+02	3.2970e+02 ;	H23 - C39	- H24
79	60	81	1	1.0758e+02	3.2970e+02 ;	H23 - C39	- H25
80	60	81	1	1.0758e+02	3.2970e+02 ;	H24 - C39	- H25
82	66	83	1	1.0758e+02	3.2970e+02 ;	H26 - C43	- H27
82	66	84	1	1.0758e+02	3.2970e+02 ;	H26 - C43	- H28
83	66	84	1	1.0758e+02	3.2970e+02 ;	H27 - C43	- H28
85	64	86	1	1.0758e+02	3.2970e+02 ;	H29 - C42	- H30
85	64	87	1	1.0758e+02	3.2970e+02 ;	H29 - C42	- H31
86	64	87	1	1.0758e+02	3.2970e+02 ;	H30 - C42	- H31
88	62	89	1	1.0758e+02	3.2970e+02 ;	H32 - C41	- H33
88	62	90	1	1.0758e+02	3.2970e+02 ;	H32 - C41	- H34
89	62	90	1	1.0758e+02	3.2970e+02 ;	H33 - C41	- H34
91	61	92	1	1.0758e+02	3.2970e+02 ;	H35 - C40	- H36
91	61	93	1	1.0758e+02	3.2970e+02 ;	H35 - C40	- H37
92	61	93	1	1.0758e+02	3.2970e+02 ;	H36 - C40	- H37
94	71	95	1	1.0758e+02	3.2970e+02 ;	H38 - C47	- H39
94	71	96	1	1.0758e+02	3.2970e+02 ;	H38 - C47	- H40
95	71	96	1	1.0758e+02	3.2970e+02 ;	H39 - C47	- H40
97	70	98	1	1.0758e+02	3.2970e+02 ;	H41 - C46	- H42

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97  70  99   1  1.0758e+02  3.2970e+02 ;  H41 - C46  - H43
98  70  99   1  1.0758e+02  3.2970e+02 ;  H42 - C46  - H43
100 73  101   1  1.0758e+02  3.2970e+02 ;  H44 - C48  - H45
100 73  102   1  1.0758e+02  3.2970e+02 ;  H44 - C48  - H46
101 73  102   1  1.0758e+02  3.2970e+02 ;  H45 - C48  - H46
103 74  104   1  1.0758e+02  3.2970e+02 ;  H47 - C49  - H48
103 74  105   1  1.0758e+02  3.2970e+02 ;  H47 - C49  - H49
104 74  105   1  1.0758e+02  3.2970e+02 ;  H48 - C49  - H49
106 69  107   1  1.0758e+02  3.2970e+02 ;  H50 - C45  - H51
106 69  108   1  1.0758e+02  3.2970e+02 ;  H50 - C45  - H52
107 69  108   1  1.0758e+02  3.2970e+02 ;  H51 - C45  - H52
109 67  110   1  1.0758e+02  3.2970e+02 ;  H53 - C44  - H54
109 67  111   1  1.0758e+02  3.2970e+02 ;  H53 - C44  - H55
110 67  111   1  1.0758e+02  3.2970e+02 ;  H54 - C44  - H55

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[ dihedrals ] ; propers

; for gromacs 4.5 or higher, using funct 9

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; i   j   k   l func phase kd   pn
1   2   3   4   9  180.00 15.16700 2;  C-  C1-  C2-  C3
1   2   3   46  9  180.00 15.16700 2;  C-  C1-  C2-  H2
1   6   5   4   9  180.00 15.16700 2;  C-  C5-  C4-  C3
1   6   5   10  9  180.00 15.16700 2;  C-  C5-  C4-  C9
2   1   6   5   9  180.00 15.16700 2;  C1-  C-  C5-  C4
2   1   6   47  9  180.00 15.16700 2;  C1-  C-  C5-  H3
2   3   4   5   9  180.00 15.16700 2;  C1-  C2-  C3-  C4
2   3   4   7   9  180.00 15.16700 2;  C1-  C2-  C3-  C6
3   4   5   6   9  180.00 15.16700 2;  C2-  C3-  C4-  C5
3   4   5   10  9  180.00 15.16700 2;  C2-  C3-  C4-  C9
3   4   7   8   9  180.00 15.16700 2;  C2-  C3-  C6-  C7

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3	4	7	11	9	180.00	15.16700	2 ;	C2-	C3-	C6-	C10
4	3	2	45	9	180.00	15.16700	2 ;	C3-	C2-	C1-	H1
4	5	6	47	9	180.00	15.16700	2 ;	C3-	C4-	C5-	H3
4	5	10	9	9	180.00	15.16700	2 ;	C3-	C4-	C9-	C8
4	5	10	48	9	180.00	15.16700	2 ;	C3-	C4-	C9-	H4
4	7	8	9	9	180.00	15.16700	2 ;	C3-	C6-	C7-	C8
4	7	8	21	9	180.00	15.16700	2 ;	C3-	C6-	C7-	O
4	7	11	12	9	180.00	4.18400	2 ;	C3-	C6-	C10-	C11
4	7	11	16	9	180.00	4.18400	2 ;	C3-	C6-	C10-	C15
5	4	3	46	9	180.00	15.16700	2 ;	C4-	C3-	C2-	H2
5	4	7	8	9	180.00	15.16700	2 ;	C4-	C3-	C6-	C7
5	4	7	11	9	180.00	15.16700	2 ;	C4-	C3-	C6-	C10
5	10	9	8	9	180.00	15.16700	2 ;	C4-	C9-	C8-	C7
5	10	9	26	9	180.00	15.16700	2 ;	C4-	C9-	C8-	C20
6	1	2	3	9	180.00	15.16700	2 ;	C5-	C-	C1-	C2
6	1	2	45	9	180.00	15.16700	2 ;	C5-	C-	C1-	H1
6	5	4	7	9	180.00	15.16700	2 ;	C5-	C4-	C3-	C6
6	5	10	9	9	180.00	15.16700	2 ;	C5-	C4-	C9-	C8
6	5	10	48	9	180.00	15.16700	2 ;	C5-	C4-	C9-	H4
7	4	3	46	9	180.00	15.16700	2 ;	C6-	C3-	C2-	H2
7	4	5	10	9	180.00	15.16700	2 ;	C6-	C3-	C4-	C9
7	8	9	10	9	180.00	15.16700	2 ;	C6-	C7-	C8-	C9
7	8	9	26	9	180.00	15.16700	2 ;	C6-	C7-	C8-	C20
7	8	21	22	9	180.00	3.76560	2 ;	C6-	C7-	O-	P
7	11	12	13	9	180.00	15.16700	2 ;	C6-	C10-	C11-	C12
7	11	12	17	9	180.00	15.16700	2 ;	C6-	C10-	C11-	C16
7	11	16	15	9	180.00	15.16700	2 ;	C6-	C10-	C15-	C14
7	11	16	25	9	180.00	15.16700	2 ;	C6-	C10-	C15-	O3
8	7	11	12	9	180.00	4.18400	2 ;	C7-	C6-	C10-	C11

8 7 11 16 9 180.00 4.18400 2; C7- C6- C10- C15  
 8 9 10 48 9 180.00 15.16700 2; C7- C8- C9- H4  
 8 9 26 33 9 180.00 4.18400 2; C7- C8- C20- C27  
 8 9 26 37 9 180.00 4.18400 2; C7- C8- C20- C31  
 8 21 22 23 9 0.00 0.41840 3; C7- O- P- O1  
 8 21 22 23 9 180.00 3.34720 2; C7- O- P- O1  
 8 21 22 24 9 0.00 0.41840 3; C7- O- P- O2  
 8 21 22 24 9 180.00 3.34720 2; C7- O- P- O2  
 8 21 22 25 9 0.00 3.34720 2; C7- O- P- O3  
 9 8 7 11 9 180.00 15.16700 2; C8- C7- C6- C10  
 9 8 21 22 9 180.00 3.76560 2; C8- C7- O- P  
 9 26 33 34 9 180.00 15.16700 2; C8- C20- C27- C28  
 9 26 33 39 9 180.00 15.16700 2; C8- C20- C27- C33  
 9 26 37 36 9 180.00 15.16700 2; C8- C20- C31- C30  
 9 26 37 38 9 180.00 15.16700 2; C8- C20- C31- C32  
 10 5 6 47 9 180.00 15.16700 2; C9- C4- C5- H3  
 10 9 8 21 9 180.00 15.16700 2; C9- C8- C7- O  
 10 9 26 33 9 180.00 4.18400 2; C9- C8- C20- C27  
 10 9 26 37 9 180.00 4.18400 2; C9- C8- C20- C31  
 11 7 8 21 9 180.00 15.16700 2; C10- C6- C7- O  
 11 12 13 14 9 180.00 15.16700 2; C10- C11- C12- C13  
 11 12 13 20 9 180.00 15.16700 2; C10- C11- C12- C19  
 11 12 17 18 9 180.00 15.16700 2; C10- C11- C16- C17  
 11 12 17 50 9 180.00 15.16700 2; C10- C11- C16- H6  
 11 16 15 14 9 180.00 15.16700 2; C10- C15- C14- C13  
 11 16 15 27 9 180.00 15.16700 2; C10- C15- C14- C21  
 11 16 25 22 9 180.00 3.76560 2; C10- C15- O3- P  
 12 11 16 15 9 180.00 15.16700 2; C11- C10- C15- C14  
 12 11 16 25 9 180.00 15.16700 2; C11- C10- C15- O3

12	13	14	15	9	180.00	15.16700	2 ;	C11-	C12-	C13-	C14
12	13	14	49	9	180.00	15.16700	2 ;	C11-	C12-	C13-	H5
12	13	20	19	9	180.00	15.16700	2 ;	C11-	C12-	C19-	C18
12	13	20	53	9	180.00	15.16700	2 ;	C11-	C12-	C19-	H9
12	17	18	19	9	180.00	15.16700	2 ;	C11-	C16-	C17-	C18
12	17	18	51	9	180.00	15.16700	2 ;	C11-	C16-	C17-	H7
13	12	11	16	9	180.00	15.16700	2 ;	C12-	C11-	C10-	C15
13	12	17	18	9	180.00	15.16700	2 ;	C12-	C11-	C16-	C17
13	12	17	50	9	180.00	15.16700	2 ;	C12-	C11-	C16-	H6
13	14	15	16	9	180.00	15.16700	2 ;	C12-	C13-	C14-	C15
13	14	15	27	9	180.00	15.16700	2 ;	C12-	C13-	C14-	C21
13	20	19	18	9	180.00	15.16700	2 ;	C12-	C19-	C18-	C17
13	20	19	52	9	180.00	15.16700	2 ;	C12-	C19-	C18-	H8
14	13	12	17	9	180.00	15.16700	2 ;	C13-	C12-	C11-	C16
14	13	20	19	9	180.00	15.16700	2 ;	C13-	C12-	C19-	C18
14	13	20	53	9	180.00	15.16700	2 ;	C13-	C12-	C19-	H9
14	15	16	25	9	180.00	15.16700	2 ;	C13-	C14-	C15-	O3
14	15	27	28	9	180.00	4.18400	2 ;	C13-	C14-	C21-	C22
14	15	27	32	9	180.00	4.18400	2 ;	C13-	C14-	C21-	C26
15	14	13	20	9	180.00	15.16700	2 ;	C14-	C13-	C12-	C19
15	16	25	22	9	180.00	3.76560	2 ;	C14-	C15-	O3-	P
15	27	28	29	9	180.00	15.16700	2 ;	C14-	C21-	C22-	C23
15	27	28	42	9	180.00	15.16700	2 ;	C14-	C21-	C22-	C36
15	27	32	31	9	180.00	15.16700	2 ;	C14-	C21-	C26-	C25
15	27	32	41	9	180.00	15.16700	2 ;	C14-	C21-	C26-	C35
16	11	12	17	9	180.00	15.16700	2 ;	C15-	C10-	C11-	C16
16	15	14	49	9	180.00	15.16700	2 ;	C15-	C14-	C13-	H5
16	15	27	28	9	180.00	4.18400	2 ;	C15-	C14-	C21-	C22
16	15	27	32	9	180.00	4.18400	2 ;	C15-	C14-	C21-	C26

16 25 22 21 9 0.00 3.34720 2; C15- O3- P- O  
 16 25 22 23 9 0.00 0.41840 3; C15- O3- P- O1  
 16 25 22 23 9 180.00 3.34720 2; C15- O3- P- O1  
 16 25 22 24 9 0.00 0.41840 3; C15- O3- P- O2  
 16 25 22 24 9 180.00 3.34720 2; C15- O3- P- O2  
 17 12 13 20 9 180.00 15.16700 2; C16- C11- C12- C19  
 17 18 19 20 9 180.00 15.16700 2; C16- C17- C18- C19  
 17 18 19 52 9 180.00 15.16700 2; C16- C17- C18- H8  
 18 19 20 53 9 180.00 15.16700 2; C17- C18- C19- H9  
 19 18 17 50 9 180.00 15.16700 2; C18- C17- C16- H6  
 20 13 14 49 9 180.00 15.16700 2; C19- C12- C13- H5  
 20 19 18 51 9 180.00 15.16700 2; C19- C18- C17- H7  
 21 8 9 26 9 180.00 15.16700 2; O- C7- C8- C20  
 25 16 15 27 9 180.00 15.16700 2; O3- C15- C14- C21  
 26 9 10 48 9 180.00 15.16700 2; C20- C8- C9- H4  
 26 33 34 35 9 180.00 15.16700 2; C20- C27- C28- C29  
 26 33 34 56 9 180.00 15.16700 2; C20- C27- C28- H12  
 26 33 39 61 9 0.00 0.00000 0; C20- C27- C33- C40  
 26 33 39 62 9 0.00 0.00000 0; C20- C27- C33- C41  
 26 33 39 63 9 0.00 0.00000 0; C20- C27- C33- H15  
 26 37 36 35 9 180.00 15.16700 2; C20- C31- C30- C29  
 26 37 36 57 9 180.00 15.16700 2; C20- C31- C30- H13  
 26 37 38 58 9 0.00 0.00000 0; C20- C31- C32- C38  
 26 37 38 59 9 0.00 0.00000 0; C20- C31- C32- H14  
 26 37 38 60 9 0.00 0.00000 0; C20- C31- C32- C39  
 27 15 14 49 9 180.00 15.16700 2; C21- C14- C13- H5  
 27 28 29 30 9 180.00 15.16700 2; C21- C22- C23- C24  
 27 28 29 54 9 180.00 15.16700 2; C21- C22- C23- H10  
 27 28 42 70 9 0.00 0.00000 0; C21- C22- C36- C46

27	28	42	71	9	0.00	0.00000	0 ;	C21-	C22-	C36-	C47
27	28	42	72	9	0.00	0.00000	0 ;	C21-	C22-	C36-	H18
27	32	31	30	9	180.00	15.16700	2 ;	C21-	C26-	C25-	C24
27	32	31	55	9	180.00	15.16700	2 ;	C21-	C26-	C25-	H11
27	32	41	67	9	0.00	0.00000	0 ;	C21-	C26-	C35-	C44
27	32	41	68	9	0.00	0.00000	0 ;	C21-	C26-	C35-	H17
27	32	41	69	9	0.00	0.00000	0 ;	C21-	C26-	C35-	C45
28	27	32	31	9	180.00	15.16700	2 ;	C22-	C21-	C26-	C25
28	27	32	41	9	180.00	15.16700	2 ;	C22-	C21-	C26-	C35
28	29	30	31	9	180.00	15.16700	2 ;	C22-	C23-	C24-	C25
28	29	30	43	9	180.00	15.16700	2 ;	C22-	C23-	C24-	C37
28	42	70	97	9	0.00	0.65084	3 ;	C22-	C36-	C46-	H41
28	42	70	98	9	0.00	0.65084	3 ;	C22-	C36-	C46-	H42
28	42	70	99	9	0.00	0.65084	3 ;	C22-	C36-	C46-	H43
28	42	71	94	9	0.00	0.65084	3 ;	C22-	C36-	C47-	H38
28	42	71	95	9	0.00	0.65084	3 ;	C22-	C36-	C47-	H39
28	42	71	96	9	0.00	0.65084	3 ;	C22-	C36-	C47-	H40
29	28	27	32	9	180.00	15.16700	2 ;	C23-	C22-	C21-	C26
29	28	42	70	9	0.00	0.00000	0 ;	C23-	C22-	C36-	C46
29	28	42	71	9	0.00	0.00000	0 ;	C23-	C22-	C36-	C47
29	28	42	72	9	0.00	0.00000	0 ;	C23-	C22-	C36-	H18
29	30	31	32	9	180.00	15.16700	2 ;	C23-	C24-	C25-	C26
29	30	31	55	9	180.00	15.16700	2 ;	C23-	C24-	C25-	H11
29	30	43	73	9	0.00	0.00000	0 ;	C23-	C24-	C37-	C48
29	30	43	74	9	0.00	0.00000	0 ;	C23-	C24-	C37-	C49
29	30	43	75	9	0.00	0.00000	0 ;	C23-	C24-	C37-	H19
30	29	28	42	9	180.00	15.16700	2 ;	C24-	C23-	C22-	C36
30	31	32	41	9	180.00	15.16700	2 ;	C24-	C25-	C26-	C35
30	43	73	100	9	0.00	0.65084	3 ;	C24-	C37-	C48-	H44

30	43	73	101	9	0.00	0.65084	3 ;	C24-	C37-	C48-	H45
30	43	73	102	9	0.00	0.65084	3 ;	C24-	C37-	C48-	H46
30	43	74	103	9	0.00	0.65084	3 ;	C24-	C37-	C49-	H47
30	43	74	104	9	0.00	0.65084	3 ;	C24-	C37-	C49-	H48
30	43	74	105	9	0.00	0.65084	3 ;	C24-	C37-	C49-	H49
31	30	29	54	9	180.00	15.16700	2 ;	C25-	C24-	C23-	H10
31	30	43	73	9	0.00	0.00000	0 ;	C25-	C24-	C37-	C48
31	30	43	74	9	0.00	0.00000	0 ;	C25-	C24-	C37-	C49
31	30	43	75	9	0.00	0.00000	0 ;	C25-	C24-	C37-	H19
31	32	41	67	9	0.00	0.00000	0 ;	C25-	C26-	C35-	C44
31	32	41	68	9	0.00	0.00000	0 ;	C25-	C26-	C35-	H17
31	32	41	69	9	0.00	0.00000	0 ;	C25-	C26-	C35-	C45
32	27	28	42	9	180.00	15.16700	2 ;	C26-	C21-	C22-	C36
32	31	30	43	9	180.00	15.16700	2 ;	C26-	C25-	C24-	C37
32	41	67	109	9	0.00	0.65084	3 ;	C26-	C35-	C44-	H53
32	41	67	110	9	0.00	0.65084	3 ;	C26-	C35-	C44-	H54
32	41	67	111	9	0.00	0.65084	3 ;	C26-	C35-	C44-	H55
32	41	69	106	9	0.00	0.65084	3 ;	C26-	C35-	C45-	H50
32	41	69	107	9	0.00	0.65084	3 ;	C26-	C35-	C45-	H51
32	41	69	108	9	0.00	0.65084	3 ;	C26-	C35-	C45-	H52
33	26	37	36	9	180.00	15.16700	2 ;	C27-	C20-	C31-	C30
33	26	37	38	9	180.00	15.16700	2 ;	C27-	C20-	C31-	C32
33	34	35	36	9	180.00	15.16700	2 ;	C27-	C28-	C29-	C30
33	34	35	40	9	180.00	15.16700	2 ;	C27-	C28-	C29-	C34
33	39	61	91	9	0.00	0.65084	3 ;	C27-	C33-	C40-	H35
33	39	61	92	9	0.00	0.65084	3 ;	C27-	C33-	C40-	H36
33	39	61	93	9	0.00	0.65084	3 ;	C27-	C33-	C40-	H37
33	39	62	88	9	0.00	0.65084	3 ;	C27-	C33-	C41-	H32
33	39	62	89	9	0.00	0.65084	3 ;	C27-	C33-	C41-	H33

33	39	62	90	9	0.00	0.65084	3 ;	C27-	C33-	C41-	H34
34	33	26	37	9	180.00	15.16700	2 ;	C28-	C27-	C20-	C31
34	33	39	61	9	0.00	0.00000	0 ;	C28-	C27-	C33-	C40
34	33	39	62	9	0.00	0.00000	0 ;	C28-	C27-	C33-	C41
34	33	39	63	9	0.00	0.00000	0 ;	C28-	C27-	C33-	H15
34	35	36	37	9	180.00	15.16700	2 ;	C28-	C29-	C30-	C31
34	35	36	57	9	180.00	15.16700	2 ;	C28-	C29-	C30-	H13
34	35	40	64	9	0.00	0.00000	0 ;	C28-	C29-	C34-	C42
34	35	40	65	9	0.00	0.00000	0 ;	C28-	C29-	C34-	H16
34	35	40	66	9	0.00	0.00000	0 ;	C28-	C29-	C34-	C43
35	34	33	39	9	180.00	15.16700	2 ;	C29-	C28-	C27-	C33
35	36	37	38	9	180.00	15.16700	2 ;	C29-	C30-	C31-	C32
35	40	64	85	9	0.00	0.65084	3 ;	C29-	C34-	C42-	H29
35	40	64	86	9	0.00	0.65084	3 ;	C29-	C34-	C42-	H30
35	40	64	87	9	0.00	0.65084	3 ;	C29-	C34-	C42-	H31
35	40	66	82	9	0.00	0.65084	3 ;	C29-	C34-	C43-	H26
35	40	66	83	9	0.00	0.65084	3 ;	C29-	C34-	C43-	H27
35	40	66	84	9	0.00	0.65084	3 ;	C29-	C34-	C43-	H28
36	35	34	56	9	180.00	15.16700	2 ;	C30-	C29-	C28-	H12
36	35	40	64	9	0.00	0.00000	0 ;	C30-	C29-	C34-	C42
36	35	40	65	9	0.00	0.00000	0 ;	C30-	C29-	C34-	H16
36	35	40	66	9	0.00	0.00000	0 ;	C30-	C29-	C34-	C43
36	37	38	58	9	0.00	0.00000	0 ;	C30-	C31-	C32-	C38
36	37	38	59	9	0.00	0.00000	0 ;	C30-	C31-	C32-	H14
36	37	38	60	9	0.00	0.00000	0 ;	C30-	C31-	C32-	C39
37	26	33	39	9	180.00	15.16700	2 ;	C31-	C20-	C27-	C33
37	36	35	40	9	180.00	15.16700	2 ;	C31-	C30-	C29-	C34
37	38	58	76	9	0.00	0.65084	3 ;	C31-	C32-	C38-	H20
37	38	58	77	9	0.00	0.65084	3 ;	C31-	C32-	C38-	H21

37	38	58	78	9	0.00	0.65084	3 ;	C31-	C32-	C38-	H22
37	38	60	79	9	0.00	0.65084	3 ;	C31-	C32-	C39-	H23
37	38	60	80	9	0.00	0.65084	3 ;	C31-	C32-	C39-	H24
37	38	60	81	9	0.00	0.65084	3 ;	C31-	C32-	C39-	H25
38	37	36	57	9	180.00	15.16700	2 ;	C32-	C31-	C30-	H13
39	33	34	56	9	180.00	15.16700	2 ;	C33-	C27-	C28-	H12
40	35	34	56	9	180.00	15.16700	2 ;	C34-	C29-	C28-	H12
40	35	36	57	9	180.00	15.16700	2 ;	C34-	C29-	C30-	H13
41	32	31	55	9	180.00	15.16700	2 ;	C35-	C26-	C25-	H11
42	28	29	54	9	180.00	15.16700	2 ;	C36-	C22-	C23-	H10
43	30	29	54	9	180.00	15.16700	2 ;	C37-	C24-	C23-	H10
43	30	31	55	9	180.00	15.16700	2 ;	C37-	C24-	C25-	H11
44	1	2	3	9	180.00	15.16700	2 ;	H-	C-	C1-	C2
44	1	2	45	9	180.00	15.16700	2 ;	H-	C-	C1-	H1
44	1	6	5	9	180.00	15.16700	2 ;	H-	C-	C5-	C4
44	1	6	47	9	180.00	15.16700	2 ;	H-	C-	C5-	H3
45	2	3	46	9	180.00	15.16700	2 ;	H1-	C1-	C2-	H2
50	17	18	51	9	180.00	15.16700	2 ;	H6-	C16-	C17-	H7
51	18	19	52	9	180.00	15.16700	2 ;	H7-	C17-	C18-	H8
52	19	20	53	9	180.00	15.16700	2 ;	H8-	C18-	C19-	H9
58	38	60	79	9	0.00	0.66944	3 ;	C38-	C32-	C39-	H23
58	38	60	80	9	0.00	0.66944	3 ;	C38-	C32-	C39-	H24
58	38	60	81	9	0.00	0.66944	3 ;	C38-	C32-	C39-	H25
59	38	58	76	9	0.00	0.62760	3 ;	H14-	C32-	C38-	H20
59	38	58	77	9	0.00	0.62760	3 ;	H14-	C32-	C38-	H21
59	38	58	78	9	0.00	0.62760	3 ;	H14-	C32-	C38-	H22
59	38	60	79	9	0.00	0.62760	3 ;	H14-	C32-	C39-	H23
59	38	60	80	9	0.00	0.62760	3 ;	H14-	C32-	C39-	H24
59	38	60	81	9	0.00	0.62760	3 ;	H14-	C32-	C39-	H25

60	38	58	76	9	0.00	0.66944	3 ;	C39-	C32-	C38-	H20
60	38	58	77	9	0.00	0.66944	3 ;	C39-	C32-	C38-	H21
60	38	58	78	9	0.00	0.66944	3 ;	C39-	C32-	C38-	H22
61	39	62	88	9	0.00	0.66944	3 ;	C40-	C33-	C41-	H32
61	39	62	89	9	0.00	0.66944	3 ;	C40-	C33-	C41-	H33
61	39	62	90	9	0.00	0.66944	3 ;	C40-	C33-	C41-	H34
62	39	61	91	9	0.00	0.66944	3 ;	C41-	C33-	C40-	H35
62	39	61	92	9	0.00	0.66944	3 ;	C41-	C33-	C40-	H36
62	39	61	93	9	0.00	0.66944	3 ;	C41-	C33-	C40-	H37
63	39	61	91	9	0.00	0.62760	3 ;	H15-	C33-	C40-	H35
63	39	61	92	9	0.00	0.62760	3 ;	H15-	C33-	C40-	H36
63	39	61	93	9	0.00	0.62760	3 ;	H15-	C33-	C40-	H37
63	39	62	88	9	0.00	0.62760	3 ;	H15-	C33-	C41-	H32
63	39	62	89	9	0.00	0.62760	3 ;	H15-	C33-	C41-	H33
63	39	62	90	9	0.00	0.62760	3 ;	H15-	C33-	C41-	H34
64	40	66	82	9	0.00	0.66944	3 ;	C42-	C34-	C43-	H26
64	40	66	83	9	0.00	0.66944	3 ;	C42-	C34-	C43-	H27
64	40	66	84	9	0.00	0.66944	3 ;	C42-	C34-	C43-	H28
65	40	64	85	9	0.00	0.62760	3 ;	H16-	C34-	C42-	H29
65	40	64	86	9	0.00	0.62760	3 ;	H16-	C34-	C42-	H30
65	40	64	87	9	0.00	0.62760	3 ;	H16-	C34-	C42-	H31
65	40	66	82	9	0.00	0.62760	3 ;	H16-	C34-	C43-	H26
65	40	66	83	9	0.00	0.62760	3 ;	H16-	C34-	C43-	H27
65	40	66	84	9	0.00	0.62760	3 ;	H16-	C34-	C43-	H28
66	40	64	85	9	0.00	0.66944	3 ;	C43-	C34-	C42-	H29
66	40	64	86	9	0.00	0.66944	3 ;	C43-	C34-	C42-	H30
66	40	64	87	9	0.00	0.66944	3 ;	C43-	C34-	C42-	H31
67	41	69	106	9	0.00	0.66944	3 ;	C44-	C35-	C45-	H50
67	41	69	107	9	0.00	0.66944	3 ;	C44-	C35-	C45-	H51

67	41	69	108	9	0.00	0.66944	3 ;	C44-	C35-	C45-	H52
68	41	67	109	9	0.00	0.62760	3 ;	H17-	C35-	C44-	H53
68	41	67	110	9	0.00	0.62760	3 ;	H17-	C35-	C44-	H54
68	41	67	111	9	0.00	0.62760	3 ;	H17-	C35-	C44-	H55
68	41	69	106	9	0.00	0.62760	3 ;	H17-	C35-	C45-	H50
68	41	69	107	9	0.00	0.62760	3 ;	H17-	C35-	C45-	H51
68	41	69	108	9	0.00	0.62760	3 ;	H17-	C35-	C45-	H52
69	41	67	109	9	0.00	0.66944	3 ;	C45-	C35-	C44-	H53
69	41	67	110	9	0.00	0.66944	3 ;	C45-	C35-	C44-	H54
69	41	67	111	9	0.00	0.66944	3 ;	C45-	C35-	C44-	H55
70	42	71	94	9	0.00	0.66944	3 ;	C46-	C36-	C47-	H38
70	42	71	95	9	0.00	0.66944	3 ;	C46-	C36-	C47-	H39
70	42	71	96	9	0.00	0.66944	3 ;	C46-	C36-	C47-	H40
71	42	70	97	9	0.00	0.66944	3 ;	C47-	C36-	C46-	H41
71	42	70	98	9	0.00	0.66944	3 ;	C47-	C36-	C46-	H42
71	42	70	99	9	0.00	0.66944	3 ;	C47-	C36-	C46-	H43
72	42	70	97	9	0.00	0.62760	3 ;	H18-	C36-	C46-	H41
72	42	70	98	9	0.00	0.62760	3 ;	H18-	C36-	C46-	H42
72	42	70	99	9	0.00	0.62760	3 ;	H18-	C36-	C46-	H43
72	42	71	94	9	0.00	0.62760	3 ;	H18-	C36-	C47-	H38
72	42	71	95	9	0.00	0.62760	3 ;	H18-	C36-	C47-	H39
72	42	71	96	9	0.00	0.62760	3 ;	H18-	C36-	C47-	H40
73	43	74	103	9	0.00	0.66944	3 ;	C48-	C37-	C49-	H47
73	43	74	104	9	0.00	0.66944	3 ;	C48-	C37-	C49-	H48
73	43	74	105	9	0.00	0.66944	3 ;	C48-	C37-	C49-	H49
74	43	73	100	9	0.00	0.66944	3 ;	C49-	C37-	C48-	H44
74	43	73	101	9	0.00	0.66944	3 ;	C49-	C37-	C48-	H45
74	43	73	102	9	0.00	0.66944	3 ;	C49-	C37-	C48-	H46
75	43	73	100	9	0.00	0.62760	3 ;	H19-	C37-	C48-	H44

```

75  43  73  101   9   0.00  0.62760  3;  H19- C37- C48- H45
75  43  73  102   9   0.00  0.62760  3;  H19- C37- C48- H46
75  43  74  103   9   0.00  0.62760  3;  H19- C37- C49- H47
75  43  74  104   9   0.00  0.62760  3;  H19- C37- C49- H48
75  43  74  105   9   0.00  0.62760  3;  H19- C37- C49- H49

```

[ dihedrals ] ; impropers

; treated as propers in GROMACS to use correct AMBER analytical function

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; i   j   k   l func phase kd   pn
1   3   2   45   4 180.00 4.60240 2;  C-  C2-  C1-  H1
1   5   6   47   4 180.00 4.60240 2;  C-  C4-  C5-  H3
2   4   3   46   4 180.00 4.60240 2;  C1-  C3-  C2-  H2
3   5   4   7    4 180.00 4.60240 2;  C2-  C4-  C3-  C6
4   6   5   10   4 180.00 4.60240 2;  C3-  C5-  C4-  C9
4   8   7   11   4 180.00 4.60240 2;  C3-  C7-  C6-  C10
5   9   10  48   4 180.00 4.60240 2;  C4-  C8-  C9-  H4
7   9   8   21   4 180.00 4.60240 2;  C6-  C8-  C7-  O
8   10  9   26   4 180.00 4.60240 2;  C7-  C9-  C8-  C20
11  15  16  25   4 180.00 4.60240 2;  C10- C14- C15- O3
12  14  13  20   4 180.00 4.60240 2;  C11- C13- C12- C19
12  16  11   7   4 180.00 4.60240 2;  C11- C15- C10- C6
12  18  17  50   4 180.00 4.60240 2;  C11- C17- C16- H6
13  15  14  49   4 180.00 4.60240 2;  C12- C14- C13- H5
13  17  12  11   4 180.00 4.60240 2;  C12- C16- C11- C10
13  19  20  53   4 180.00 4.60240 2;  C12- C18- C19- H9
14  16  15  27   4 180.00 4.60240 2;  C13- C15- C14- C21
17  19  18  51   4 180.00 4.60240 2;  C16- C18- C17- H7
18  20  19  52   4 180.00 4.60240 2;  C17- C19- C18- H8
28  30  29  54   4 180.00 4.60240 2;  C22- C24- C23- H10

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28 32 27 15 4 180.00 4.60240 2; C22- C26- C21- C14  
29 31 30 43 4 180.00 4.60240 2; C23- C25- C24- C37  
30 32 31 55 4 180.00 4.60240 2; C24- C26- C25- H11  
33 35 34 56 4 180.00 4.60240 2; C27- C29- C28- H12  
33 37 26 9 4 180.00 4.60240 2; C27- C31- C20- C8  
35 37 36 57 4 180.00 4.60240 2; C29- C31- C30- H13  
38 36 37 26 4 180.00 4.60240 2; C32- C30- C31- C20  
39 34 33 26 4 180.00 4.60240 2; C33- C28- C27- C20  
40 34 35 36 4 180.00 4.60240 2; C34- C28- C29- C30  
41 31 32 27 4 180.00 4.60240 2; C35- C25- C26- C21  
42 29 28 27 4 180.00 4.60240 2; C36- C23- C22- C21  
44 1 6 2 4 180.00 4.60240 2; H- C- C5- C1

#### 9.1.4.3.4 1e

[ moleculetype ]

;name nrexcl  
PIE 3

[ atoms ]

1	ca	1	PIE	C	1	-0.0938964813	12.01000 ; qtot 0.000
2	ca	1	PIE	C1	2	-0.1987372338	12.01000 ; qtot 0.000
3	ca	1	PIE	C2	3	-0.1103289115	12.01000 ; qtot 0.000
4	ca	1	PIE	C3	4	0.0972749636	12.01000 ; qtot 0.000
5	ca	1	PIE	C4	5	0.2190309538	12.01000 ; qtot 0.000
6	ca	1	PIE	C5	6	-0.2947057622	12.01000 ; qtot 0.000
7	cp	1	PIE	C6	7	-0.2119597284	12.01000 ; qtot 0.000
8	ca	1	PIE	C7	8	0.4629407522	12.01000 ; qtot 0.000
9	cp	1	PIE	C8	9	-0.2680034767	12.01000 ; qtot 0.000
10	ca	1	PIE	C9	10	-0.2455738072	12.01000 ; qtot 0.000
11	cp	1	PIE	C10	11	-0.1887614274	12.01000 ; qtot 0.000
12	ca	1	PIE	C11	12	0.0923454085	12.01000 ; qtot 0.000
13	ca	1	PIE	C12	13	0.2140803604	12.01000 ; qtot 0.000
14	ca	1	PIE	C13	14	-0.2400613588	12.01000 ; qtot 0.000
15	cp	1	PIE	C14	15	-0.2711414815	12.01000 ; qtot 0.000
16	ca	1	PIE	C15	16	0.4495551788	12.01000 ; qtot 0.000
17	ca	1	PIE	C16	17	-0.1108884328	12.01000 ; qtot 0.000
18	ca	1	PIE	C17	18	-0.1967234126	12.01000 ; qtot 0.000
19	ca	1	PIE	C18	19	-0.0950163073	12.01000 ; qtot 0.000
20	ca	1	PIE	C19	20	-0.2919810856	12.01000 ; qtot 0.000
21	os	1	PIE	O	21	-0.4823558839	16.00000 ; qtot 0.000
22	p5	1	PIE	P	22	1.2863663255	30.97000 ; qtot 0.000

23	o	1	PIE	O1	23	-0.7183497296	16.00000 ; qtot 0.000
24	o	1	PIE	O2	24	-0.7174534898	16.00000 ; qtot 0.000
25	os	1	PIE	O3	25	-0.4787697032	16.00000 ; qtot 0.000
26	cp	1	PIE	C20	26	0.3478384500	12.01000 ; qtot 0.000
27	cp	1	PIE	C21	27	0.3531595930	12.01000 ; qtot 0.000
28	ca	1	PIE	C22	28	-0.2196471945	12.01000 ; qtot 0.000
29	ca	1	PIE	C23	29	-0.1371283164	12.01000 ; qtot 0.000
30	ca	1	PIE	C24	30	-0.0298451034	12.01000 ; qtot 0.000
31	ca	1	PIE	C25	31	-0.1836030996	12.01000 ; qtot 0.000
32	ca	1	PIE	C26	32	-0.1813108660	12.01000 ; qtot 0.000
33	ca	1	PIE	C27	33	-0.1788081250	12.01000 ; qtot 0.000
34	ca	1	PIE	C28	34	-0.1810191238	12.01000 ; qtot 0.000
35	ca	1	PIE	C29	35	-0.0371258596	12.01000 ; qtot 0.000
36	ca	1	PIE	C30	36	-0.1332307102	12.01000 ; qtot 0.000
37	ca	1	PIE	C31	37	-0.2183995477	12.01000 ; qtot 0.000
38	ha	1	PIE	H	38	0.1080241742	1.00800 ; qtot 0.000
39	ha	1	PIE	H1	39	0.1306428676	1.00800 ; qtot 0.000
40	ha	1	PIE	H2	40	0.1084957706	1.00800 ; qtot 0.000
41	ha	1	PIE	H3	41	0.1574196784	1.00800 ; qtot 0.000
42	ha	1	PIE	H4	42	0.1007662540	1.00800 ; qtot 0.000
43	ha	1	PIE	H5	43	0.0994134081	1.00800 ; qtot 0.000
44	ha	1	PIE	H6	44	0.1083808311	1.00800 ; qtot 0.000
45	ha	1	PIE	H7	45	0.1302744844	1.00800 ; qtot 0.000
46	ha	1	PIE	H8	46	0.1079520057	1.00800 ; qtot 0.000
47	ha	1	PIE	H9	47	0.1570010539	1.00800 ; qtot 0.000
48	ha	1	PIE	H10	48	0.1550391029	1.00800 ; qtot 0.000
49	ha	1	PIE	H11	49	0.1491622784	1.00800 ; qtot 0.000
50	ha	1	PIE	H12	50	0.1491490094	1.00800 ; qtot 0.000
51	ha	1	PIE	H13	51	0.1546300859	1.00800 ; qtot 0.000

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52 ha 1 PIE H14 52 0.1671842564 1.00800 ; qtot 0.000
53 ha 1 PIE H15 53 0.1618629331 1.00800 ; qtot 0.000
54 no 1 PIE N 54 0.7752070158 14.01000 ; qtot 0.000
55 no 1 PIE N1 55 0.7798518044 14.01000 ; qtot 0.000
56 ha 1 PIE H16 56 0.1608407620 1.00800 ; qtot 0.000
57 ha 1 PIE H17 57 0.1659472384 1.00800 ; qtot 0.000
58 o 1 PIE O4 58 -0.4613937353 16.00000 ; qtot 0.000
59 o 1 PIE O5 59 -0.4548860852 16.00000 ; qtot 0.000
60 o 1 PIE O6 60 -0.4561546789 16.00000 ; qtot 0.000
61 o 1 PIE O7 61 -0.4625768413 16.00000 ; qtot 0.000

```

[ bonds ]

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; ai aj funct r k
1 2 1 1.3984e-01 3.8585e+05 ; C - C1
1 6 1 1.3984e-01 3.8585e+05 ; C - C5
1 38 1 1.0860e-01 2.8937e+05 ; C - H
2 3 1 1.3984e-01 3.8585e+05 ; C1 - C2
2 39 1 1.0860e-01 2.8937e+05 ; C1 - H1
3 4 1 1.3984e-01 3.8585e+05 ; C2 - C3
3 40 1 1.0860e-01 2.8937e+05 ; C2 - H2
4 5 1 1.3984e-01 3.8585e+05 ; C3 - C4
4 7 1 1.4058e-01 3.7673e+05 ; C3 - C6
5 6 1 1.3984e-01 3.8585e+05 ; C4 - C5
5 10 1 1.3984e-01 3.8585e+05 ; C4 - C9
6 41 1 1.0860e-01 2.8937e+05 ; C5 - H3
7 8 1 1.4058e-01 3.7673e+05 ; C6 - C7
7 11 1 1.4854e-01 2.9405e+05 ; C6 - C10
8 9 1 1.4058e-01 3.7673e+05 ; C7 - C8
8 21 1 1.3696e-01 3.1514e+05 ; C7 - O
9 10 1 1.4058e-01 3.7673e+05 ; C8 - C9

```

9 26 1 1.4854e-01 2.9405e+05 ; C8 - C20  
 10 42 1 1.0860e-01 2.8937e+05 ; C9 - H4  
 11 12 1 1.4058e-01 3.7673e+05 ; C10 - C11  
 11 16 1 1.4058e-01 3.7673e+05 ; C10 - C15  
 12 13 1 1.3984e-01 3.8585e+05 ; C11 - C12  
 12 17 1 1.3984e-01 3.8585e+05 ; C11 - C16  
 13 14 1 1.3984e-01 3.8585e+05 ; C12 - C13  
 13 20 1 1.3984e-01 3.8585e+05 ; C12 - C19  
 14 15 1 1.4058e-01 3.7673e+05 ; C13 - C14  
 14 43 1 1.0860e-01 2.8937e+05 ; C13 - H5  
 15 16 1 1.4058e-01 3.7673e+05 ; C14 - C15  
 15 27 1 1.4854e-01 2.9405e+05 ; C14 - C21  
 16 25 1 1.3696e-01 3.1514e+05 ; C15 - O3  
 17 18 1 1.3984e-01 3.8585e+05 ; C16 - C17  
 17 44 1 1.0860e-01 2.8937e+05 ; C16 - H6  
 18 19 1 1.3984e-01 3.8585e+05 ; C17 - C18  
 18 45 1 1.0860e-01 2.8937e+05 ; C17 - H7  
 19 20 1 1.3984e-01 3.8585e+05 ; C18 - C19  
 19 46 1 1.0860e-01 2.8937e+05 ; C18 - H8  
 20 47 1 1.0860e-01 2.8937e+05 ; C19 - H9  
 21 22 1 1.6147e-01 2.7665e+05 ; O - P  
 22 23 1 1.4866e-01 4.0125e+05 ; P - O1  
 22 24 1 1.4866e-01 4.0125e+05 ; P - O2  
 22 25 1 1.6147e-01 2.7665e+05 ; P - O3  
 26 33 1 1.4058e-01 3.7673e+05 ; C20 - C27  
 26 37 1 1.4058e-01 3.7673e+05 ; C20 - C31  
 27 28 1 1.4058e-01 3.7673e+05 ; C21 - C22  
 27 32 1 1.4058e-01 3.7673e+05 ; C21 - C26  
 28 29 1 1.3984e-01 3.8585e+05 ; C22 - C23

```

28  53  1  1.0860e-01  2.8937e+05 ;  C22 - H15
29  30  1  1.3984e-01  3.8585e+05 ;  C23 - C24
29  48  1  1.0860e-01  2.8937e+05 ;  C23 - H10
30  31  1  1.3984e-01  3.8585e+05 ;  C24 - C25
30  54  1  1.4689e-01  2.6920e+05 ;  C24 - N
31  32  1  1.3984e-01  3.8585e+05 ;  C25 - C26
31  49  1  1.0860e-01  2.8937e+05 ;  C25 - H11
32  52  1  1.0860e-01  2.8937e+05 ;  C26 - H14
33  34  1  1.3984e-01  3.8585e+05 ;  C27 - C28
33  57  1  1.0860e-01  2.8937e+05 ;  C27 - H17
34  35  1  1.3984e-01  3.8585e+05 ;  C28 - C29
34  50  1  1.0860e-01  2.8937e+05 ;  C28 - H12
35  36  1  1.3984e-01  3.8585e+05 ;  C29 - C30
35  55  1  1.4689e-01  2.6920e+05 ;  C29 - N1
36  37  1  1.3984e-01  3.8585e+05 ;  C30 - C31
36  51  1  1.0860e-01  2.8937e+05 ;  C30 - H13
37  56  1  1.0860e-01  2.8937e+05 ;  C31 - H16
54  58  1  1.2260e-01  6.2074e+05 ;  N - O4
54  59  1  1.2260e-01  6.2074e+05 ;  N - O5
55  60  1  1.2260e-01  6.2074e+05 ;  N1 - O6
55  61  1  1.2260e-01  6.2074e+05 ;  N1 - O7

```

[ pairs ]

```

; ai aj funct
1   4   1;   C - C3
1   10  1;   C - C9
1   40  1;   C - H2
2   5   1;   C1 - C4
2   7   1;   C1 - C6

```

2 41 1; C1 - H3  
3 8 1; C2 - C7  
3 10 1; C2 - C9  
3 11 1; C2 - C10  
4 9 1; C3 - C8  
4 12 1; C3 - C11  
4 16 1; C3 - C15  
4 21 1; C3 - O  
4 39 1; C3 - H1  
4 41 1; C3 - H3  
4 42 1; C3 - H4  
5 8 1; C4 - C7  
5 11 1; C4 - C10  
5 26 1; C4 - C20  
5 40 1; C4 - H2  
6 3 1; C5 - C2  
6 7 1; C5 - C6  
6 9 1; C5 - C8  
6 39 1; C5 - H1  
6 42 1; C5 - H4  
7 10 1; C6 - C9  
7 13 1; C6 - C12  
7 15 1; C6 - C14  
7 17 1; C6 - C16  
7 22 1; C6 - P  
7 25 1; C6 - O3  
7 26 1; C6 - C20  
7 40 1; C6 - H2  
8 12 1; C7 - C11

8 16 1; C7 - C15  
8 23 1; C7 - O1  
8 24 1; C7 - O2  
8 25 1; C7 - O3  
8 33 1; C7 - C27  
8 37 1; C7 - C31  
8 42 1; C7 - H4  
9 11 1; C8 - C10  
9 22 1; C8 - P  
9 34 1; C8 - C28  
9 36 1; C8 - C30  
9 56 1; C8 - H16  
9 57 1; C8 - H17  
10 21 1; C9 - O  
10 33 1; C9 - C27  
10 37 1; C9 - C31  
10 41 1; C9 - H3  
11 14 1; C10 - C13  
11 18 1; C10 - C17  
11 20 1; C10 - C19  
11 21 1; C10 - O  
11 22 1; C10 - P  
11 27 1; C10 - C21  
11 44 1; C10 - H6  
12 15 1; C11 - C14  
12 19 1; C11 - C18  
12 25 1; C11 - O3  
12 43 1; C11 - H5  
12 45 1; C11 - H7

12 47 1; C11 - H9  
13 16 1; C12 - C15  
13 18 1; C12 - C17  
13 27 1; C12 - C21  
13 44 1; C12 - H6  
13 46 1; C12 - H8  
14 17 1; C13 - C16  
14 19 1; C13 - C18  
14 25 1; C13 - O3  
14 28 1; C13 - C22  
14 32 1; C13 - C26  
14 47 1; C13 - H9  
15 20 1; C14 - C19  
15 22 1; C14 - P  
15 29 1; C14 - C23  
15 31 1; C14 - C25  
15 52 1; C14 - H14  
15 53 1; C14 - H15  
16 17 1; C15 - C16  
16 21 1; C15 - O  
16 23 1; C15 - O1  
16 24 1; C15 - O2  
16 28 1; C15 - C22  
16 32 1; C15 - C26  
16 43 1; C15 - H5  
17 20 1; C16 - C19  
17 46 1; C16 - H8  
18 47 1; C17 - H9  
19 44 1; C18 - H6

20 43 1; C19 - H5  
20 45 1; C19 - H7  
21 26 1; O - C20  
25 27 1; O3 - C21  
26 35 1; C20 - C29  
26 42 1; C20 - H4  
26 50 1; C20 - H12  
26 51 1; C20 - H13  
27 30 1; C21 - C24  
27 43 1; C21 - H5  
27 48 1; C21 - H10  
27 49 1; C21 - H11  
28 31 1; C22 - C25  
28 52 1; C22 - H14  
28 54 1; C22 - N  
29 32 1; C23 - C26  
29 49 1; C23 - H11  
29 58 1; C23 - O4  
29 59 1; C23 - O5  
30 52 1; C24 - H14  
30 53 1; C24 - H15  
31 48 1; C25 - H10  
31 58 1; C25 - O4  
31 59 1; C25 - O5  
32 53 1; C26 - H15  
32 54 1; C26 - N  
33 36 1; C27 - C30  
33 55 1; C27 - N1  
33 56 1; C27 - H16

34 37 1; C28 - C31  
34 51 1; C28 - H13  
34 60 1; C28 - O6  
34 61 1; C28 - O7  
35 56 1; C29 - H16  
35 57 1; C29 - H17  
36 50 1; C30 - H12  
36 60 1; C30 - O6  
36 61 1; C30 - O7  
37 55 1; C31 - N1  
37 57 1; C31 - H17  
38 3 1; H - C2  
38 5 1; H - C4  
38 39 1; H - H1  
38 41 1; H - H3  
39 40 1; H1 - H2  
44 45 1; H6 - H7  
45 46 1; H7 - H8  
46 47 1; H8 - H9  
48 53 1; H10 - H15  
48 54 1; H10 - N  
49 52 1; H11 - H14  
49 54 1; H11 - N  
50 55 1; H12 - N1  
50 57 1; H12 - H17  
51 55 1; H13 - N1  
51 56 1; H13 - H16

[ angles ]

	ai	aj	ak	funct	theta	cth
1	2	3	1	1.2002e+02	5.5731e+02 ;	C - C1 - C2
1	2	39	1	1.1988e+02	4.0334e+02 ;	C - C1 - H1
1	6	5	1	1.2002e+02	5.5731e+02 ;	C - C5 - C4
1	6	41	1	1.1988e+02	4.0334e+02 ;	C - C5 - H3
2	1	6	1	1.2002e+02	5.5731e+02 ;	C1 - C - C5
2	1	38	1	1.1988e+02	4.0334e+02 ;	C1 - C - H
2	3	4	1	1.2002e+02	5.5731e+02 ;	C1 - C2 - C3
2	3	40	1	1.1988e+02	4.0334e+02 ;	C1 - C2 - H2
3	2	39	1	1.1988e+02	4.0334e+02 ;	C2 - C1 - H1
3	4	5	1	1.2002e+02	5.5731e+02 ;	C2 - C3 - C4
3	4	7	1	1.2069e+02	5.5480e+02 ;	C2 - C3 - C6
4	3	40	1	1.1988e+02	4.0334e+02 ;	C3 - C2 - H2
4	5	6	1	1.2002e+02	5.5731e+02 ;	C3 - C4 - C5
4	5	10	1	1.2002e+02	5.5731e+02 ;	C3 - C4 - C9
4	7	8	1	1.1838e+02	5.5815e+02 ;	C3 - C6 - C7
4	7	11	1	1.2111e+02	5.3555e+02 ;	C3 - C6 - C10
5	4	7	1	1.2069e+02	5.5480e+02 ;	C4 - C3 - C6
5	6	41	1	1.1988e+02	4.0334e+02 ;	C4 - C5 - H3
5	10	9	1	1.2069e+02	5.5480e+02 ;	C4 - C9 - C8
5	10	42	1	1.1988e+02	4.0334e+02 ;	C4 - C9 - H4
6	1	38	1	1.1988e+02	4.0334e+02 ;	C5 - C - H
6	5	10	1	1.2002e+02	5.5731e+02 ;	C5 - C4 - C9
7	8	9	1	1.1152e+02	5.7572e+02 ;	C6 - C7 - C8
7	8	21	1	1.1691e+02	5.8660e+02 ;	C6 - C7 - O
7	11	12	1	1.2111e+02	5.3555e+02 ;	C6 - C10 - C11
7	11	16	1	1.2111e+02	5.3555e+02 ;	C6 - C10 - C15
8	7	11	1	1.2111e+02	5.3555e+02 ;	C7 - C6 - C10
8	9	10	1	1.1838e+02	5.5815e+02 ;	C7 - C8 - C9

8	9	26	1	1.2111e+02	5.3555e+02 ;	C7 - C8	- C20
8	21	22	1	1.2318e+02	6.5019e+02 ;	C7 - O	- P
9	8	21	1	1.1691e+02	5.8660e+02 ;	C8 - C7	- O
9	10	42	1	1.1986e+02	4.0166e+02 ;	C8 - C9	- H4
9	26	33	1	1.2111e+02	5.3555e+02 ;	C8 - C20	- C27
9	26	37	1	1.2111e+02	5.3555e+02 ;	C8 - C20	- C31
10	9	26	1	1.2111e+02	5.3555e+02 ;	C9 - C8	- C20
11	12	13	1	1.2069e+02	5.5480e+02 ;	C10 - C11	- C12
11	12	17	1	1.2069e+02	5.5480e+02 ;	C10 - C11	- C16
11	16	15	1	1.1152e+02	5.7572e+02 ;	C10 - C15	- C14
11	16	25	1	1.1691e+02	5.8660e+02 ;	C10 - C15	- O3
12	11	16	1	1.1838e+02	5.5815e+02 ;	C11 - C10	- C15
12	13	14	1	1.2002e+02	5.5731e+02 ;	C11 - C12	- C13
12	13	20	1	1.2002e+02	5.5731e+02 ;	C11 - C12	- C19
12	17	18	1	1.2002e+02	5.5731e+02 ;	C11 - C16	- C17
12	17	44	1	1.1988e+02	4.0334e+02 ;	C11 - C16	- H6
13	12	17	1	1.2002e+02	5.5731e+02 ;	C12 - C11	- C16
13	14	15	1	1.2069e+02	5.5480e+02 ;	C12 - C13	- C14
13	14	43	1	1.1988e+02	4.0334e+02 ;	C12 - C13	- H5
13	20	19	1	1.2002e+02	5.5731e+02 ;	C12 - C19	- C18
13	20	47	1	1.1988e+02	4.0334e+02 ;	C12 - C19	- H9
14	13	20	1	1.2002e+02	5.5731e+02 ;	C13 - C12	- C19
14	15	16	1	1.1838e+02	5.5815e+02 ;	C13 - C14	- C15
14	15	27	1	1.2111e+02	5.3555e+02 ;	C13 - C14	- C21
15	14	43	1	1.1986e+02	4.0166e+02 ;	C14 - C13	- H5
15	16	25	1	1.1691e+02	5.8660e+02 ;	C14 - C15	- O3
15	27	28	1	1.2111e+02	5.3555e+02 ;	C14 - C21	- C22
15	27	32	1	1.2111e+02	5.3555e+02 ;	C14 - C21	- C26
16	15	27	1	1.2111e+02	5.3555e+02 ;	C15 - C14	- C21

16	25	22	1	1.2318e+02	6.5019e+02 ;	C15 - O3	- P
17	18	19	1	1.2002e+02	5.5731e+02 ;	C16 - C17	- C18
17	18	45	1	1.1988e+02	4.0334e+02 ;	C16 - C17	- H7
18	17	44	1	1.1988e+02	4.0334e+02 ;	C17 - C16	- H6
18	19	20	1	1.2002e+02	5.5731e+02 ;	C17 - C18	- C19
18	19	46	1	1.1988e+02	4.0334e+02 ;	C17 - C18	- H8
19	18	45	1	1.1988e+02	4.0334e+02 ;	C18 - C17	- H7
19	20	47	1	1.1988e+02	4.0334e+02 ;	C18 - C19	- H9
20	19	46	1	1.1988e+02	4.0334e+02 ;	C19 - C18	- H8
21	22	23	1	1.1546e+02	3.6736e+02 ;	O - P	- O1
21	22	24	1	1.1546e+02	3.6736e+02 ;	O - P	- O2
21	22	25	1	1.0184e+02	3.7656e+02 ;	O - P	- O3
23	22	24	1	1.1580e+02	3.8325e+02 ;	O1 - P	- O2
23	22	25	1	1.1546e+02	3.6736e+02 ;	O1 - P	- O3
24	22	25	1	1.1546e+02	3.6736e+02 ;	O2 - P	- O3
26	33	34	1	1.2069e+02	5.5480e+02 ;	C20 - C27	- C28
26	33	57	1	1.1986e+02	4.0166e+02 ;	C20 - C27	- H17
26	37	36	1	1.2069e+02	5.5480e+02 ;	C20 - C31	- C30
26	37	56	1	1.1986e+02	4.0166e+02 ;	C20 - C31	- H16
27	28	29	1	1.2069e+02	5.5480e+02 ;	C21 - C22	- C23
27	28	53	1	1.1986e+02	4.0166e+02 ;	C21 - C22	- H15
27	32	31	1	1.2069e+02	5.5480e+02 ;	C21 - C26	- C25
27	32	52	1	1.1986e+02	4.0166e+02 ;	C21 - C26	- H14
28	27	32	1	1.1838e+02	5.5815e+02 ;	C22 - C21	- C26
28	29	30	1	1.2002e+02	5.5731e+02 ;	C22 - C23	- C24
28	29	48	1	1.1988e+02	4.0334e+02 ;	C22 - C23	- H10
29	28	53	1	1.1988e+02	4.0334e+02 ;	C23 - C22	- H15
29	30	31	1	1.2002e+02	5.5731e+02 ;	C23 - C24	- C25
29	30	54	1	1.1901e+02	5.5898e+02 ;	C23 - C24	- N

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30  29  48   1  1.1988e+02  4.0334e+02 ;  C24 - C23  - H10
30  31  32   1  1.2002e+02  5.5731e+02 ;  C24 - C25  - C26
30  31  49   1  1.1988e+02  4.0334e+02 ;  C24 - C25  - H11
30  54  58   1  1.1776e+02  5.7488e+02 ;  C24 - N   - O4
30  54  59   1  1.1776e+02  5.7488e+02 ;  C24 - N   - O5
31  30  54   1  1.1901e+02  5.5898e+02 ;  C25 - C24  - N
31  32  52   1  1.1988e+02  4.0334e+02 ;  C25 - C26  - H14
32  31  49   1  1.1988e+02  4.0334e+02 ;  C26 - C25  - H11
33  26  37   1  1.1838e+02  5.5815e+02 ;  C27 - C20  - C31
33  34  35   1  1.2002e+02  5.5731e+02 ;  C27 - C28  - C29
33  34  50   1  1.1988e+02  4.0334e+02 ;  C27 - C28  - H12
34  33  57   1  1.1988e+02  4.0334e+02 ;  C28 - C27  - H17
34  35  36   1  1.2002e+02  5.5731e+02 ;  C28 - C29  - C30
34  35  55   1  1.1901e+02  5.5898e+02 ;  C28 - C29  - N1
35  34  50   1  1.1988e+02  4.0334e+02 ;  C29 - C28  - H12
35  36  37   1  1.2002e+02  5.5731e+02 ;  C29 - C30  - C31
35  36  51   1  1.1988e+02  4.0334e+02 ;  C29 - C30  - H13
35  55  60   1  1.1776e+02  5.7488e+02 ;  C29 - N1  - O6
35  55  61   1  1.1776e+02  5.7488e+02 ;  C29 - N1  - O7
36  35  55   1  1.1901e+02  5.5898e+02 ;  C30 - C29  - N1
36  37  56   1  1.1988e+02  4.0334e+02 ;  C30 - C31  - H16
37  36  51   1  1.1988e+02  4.0334e+02 ;  C31 - C30  - H13
58  54  59   1  1.2508e+02  6.4183e+02 ;  O4 - N   - O5
60  55  61   1  1.2508e+02  6.4183e+02 ;  O6 - N1  - O7

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[ dihedrals ] ; proper

; for gromacs 4.5 or higher, using funct 9

; i j k l func phase kd pn

1 2 3 4 9 180.00 15.16700 2; C- C1- C2- C3

1	2	3	40	9	180.00	15.16700	2 ;	C-	C1-	C2-	H2
1	6	5	4	9	180.00	15.16700	2 ;	C-	C5-	C4-	C3
1	6	5	10	9	180.00	15.16700	2 ;	C-	C5-	C4-	C9
2	1	6	5	9	180.00	15.16700	2 ;	C1-	C-	C5-	C4
2	1	6	41	9	180.00	15.16700	2 ;	C1-	C-	C5-	H3
2	3	4	5	9	180.00	15.16700	2 ;	C1-	C2-	C3-	C4
2	3	4	7	9	180.00	15.16700	2 ;	C1-	C2-	C3-	C6
3	4	5	6	9	180.00	15.16700	2 ;	C2-	C3-	C4-	C5
3	4	5	10	9	180.00	15.16700	2 ;	C2-	C3-	C4-	C9
3	4	7	8	9	180.00	15.16700	2 ;	C2-	C3-	C6-	C7
3	4	7	11	9	180.00	15.16700	2 ;	C2-	C3-	C6-	C10
4	3	2	39	9	180.00	15.16700	2 ;	C3-	C2-	C1-	H1
4	5	6	41	9	180.00	15.16700	2 ;	C3-	C4-	C5-	H3
4	5	10	9	9	180.00	15.16700	2 ;	C3-	C4-	C9-	C8
4	5	10	42	9	180.00	15.16700	2 ;	C3-	C4-	C9-	H4
4	7	8	9	9	180.00	15.16700	2 ;	C3-	C6-	C7-	C8
4	7	8	21	9	180.00	15.16700	2 ;	C3-	C6-	C7-	O
4	7	11	12	9	180.00	4.18400	2 ;	C3-	C6-	C10-	C11
4	7	11	16	9	180.00	4.18400	2 ;	C3-	C6-	C10-	C15
5	4	3	40	9	180.00	15.16700	2 ;	C4-	C3-	C2-	H2
5	4	7	8	9	180.00	15.16700	2 ;	C4-	C3-	C6-	C7
5	4	7	11	9	180.00	15.16700	2 ;	C4-	C3-	C6-	C10
5	10	9	8	9	180.00	15.16700	2 ;	C4-	C9-	C8-	C7
5	10	9	26	9	180.00	15.16700	2 ;	C4-	C9-	C8-	C20
6	1	2	3	9	180.00	15.16700	2 ;	C5-	C-	C1-	C2
6	1	2	39	9	180.00	15.16700	2 ;	C5-	C-	C1-	H1
6	5	4	7	9	180.00	15.16700	2 ;	C5-	C4-	C3-	C6
6	5	10	9	9	180.00	15.16700	2 ;	C5-	C4-	C9-	C8
6	5	10	42	9	180.00	15.16700	2 ;	C5-	C4-	C9-	H4

7 4 3 40 9 180.00 15.16700 2; C6- C3- C2- H2  
 7 4 5 10 9 180.00 15.16700 2; C6- C3- C4- C9  
 7 8 9 10 9 180.00 15.16700 2; C6- C7- C8- C9  
 7 8 9 26 9 180.00 15.16700 2; C6- C7- C8- C20  
 7 8 21 22 9 180.00 3.76560 2; C6- C7- O- P  
 7 11 12 13 9 180.00 15.16700 2; C6- C10- C11- C12  
 7 11 12 17 9 180.00 15.16700 2; C6- C10- C11- C16  
 7 11 16 15 9 180.00 15.16700 2; C6- C10- C15- C14  
 7 11 16 25 9 180.00 15.16700 2; C6- C10- C15- O3  
 8 7 11 12 9 180.00 4.18400 2; C7- C6- C10- C11  
 8 7 11 16 9 180.00 4.18400 2; C7- C6- C10- C15  
 8 9 10 42 9 180.00 15.16700 2; C7- C8- C9- H4  
 8 9 26 33 9 180.00 4.18400 2; C7- C8- C20- C27  
 8 9 26 37 9 180.00 4.18400 2; C7- C8- C20- C31  
 8 21 22 23 9 0.00 0.41840 3; C7- O- P- O1  
 8 21 22 23 9 180.00 3.34720 2; C7- O- P- O1  
 8 21 22 24 9 0.00 0.41840 3; C7- O- P- O2  
 8 21 22 24 9 180.00 3.34720 2; C7- O- P- O2  
 8 21 22 25 9 0.00 3.34720 2; C7- O- P- O3  
 9 8 7 11 9 180.00 15.16700 2; C8- C7- C6- C10  
 9 8 21 22 9 180.00 3.76560 2; C8- C7- O- P  
 9 26 33 34 9 180.00 15.16700 2; C8- C20- C27- C28  
 9 26 33 57 9 180.00 15.16700 2; C8- C20- C27- H17  
 9 26 37 36 9 180.00 15.16700 2; C8- C20- C31- C30  
 9 26 37 56 9 180.00 15.16700 2; C8- C20- C31- H16  
 10 5 6 41 9 180.00 15.16700 2; C9- C4- C5- H3  
 10 9 8 21 9 180.00 15.16700 2; C9- C8- C7- O  
 10 9 26 33 9 180.00 4.18400 2; C9- C8- C20- C27  
 10 9 26 37 9 180.00 4.18400 2; C9- C8- C20- C31

11	7	8	21	9	180.00	15.16700	2;	C10-	C6-	C7-	O
11	12	13	14	9	180.00	15.16700	2;	C10-	C11-	C12-	C13
11	12	13	20	9	180.00	15.16700	2;	C10-	C11-	C12-	C19
11	12	17	18	9	180.00	15.16700	2;	C10-	C11-	C16-	C17
11	12	17	44	9	180.00	15.16700	2;	C10-	C11-	C16-	H6
11	16	15	14	9	180.00	15.16700	2;	C10-	C15-	C14-	C13
11	16	15	27	9	180.00	15.16700	2;	C10-	C15-	C14-	C21
11	16	25	22	9	180.00	3.76560	2;	C10-	C15-	O3-	P
12	11	16	15	9	180.00	15.16700	2;	C11-	C10-	C15-	C14
12	11	16	25	9	180.00	15.16700	2;	C11-	C10-	C15-	O3
12	13	14	15	9	180.00	15.16700	2;	C11-	C12-	C13-	C14
12	13	14	43	9	180.00	15.16700	2;	C11-	C12-	C13-	H5
12	13	20	19	9	180.00	15.16700	2;	C11-	C12-	C19-	C18
12	13	20	47	9	180.00	15.16700	2;	C11-	C12-	C19-	H9
12	17	18	19	9	180.00	15.16700	2;	C11-	C16-	C17-	C18
12	17	18	45	9	180.00	15.16700	2;	C11-	C16-	C17-	H7
13	12	11	16	9	180.00	15.16700	2;	C12-	C11-	C10-	C15
13	12	17	18	9	180.00	15.16700	2;	C12-	C11-	C16-	C17
13	12	17	44	9	180.00	15.16700	2;	C12-	C11-	C16-	H6
13	14	15	16	9	180.00	15.16700	2;	C12-	C13-	C14-	C15
13	14	15	27	9	180.00	15.16700	2;	C12-	C13-	C14-	C21
13	20	19	18	9	180.00	15.16700	2;	C12-	C19-	C18-	C17
13	20	19	46	9	180.00	15.16700	2;	C12-	C19-	C18-	H8
14	13	12	17	9	180.00	15.16700	2;	C13-	C12-	C11-	C16
14	13	20	19	9	180.00	15.16700	2;	C13-	C12-	C19-	C18
14	13	20	47	9	180.00	15.16700	2;	C13-	C12-	C19-	H9
14	15	16	25	9	180.00	15.16700	2;	C13-	C14-	C15-	O3
14	15	27	28	9	180.00	4.18400	2;	C13-	C14-	C21-	C22
14	15	27	32	9	180.00	4.18400	2;	C13-	C14-	C21-	C26

15	14	13	20	9	180.00	15.16700	2 ;	C14-	C13-	C12-	C19
15	16	25	22	9	180.00	3.76560	2 ;	C14-	C15-	O3-	P
15	27	28	29	9	180.00	15.16700	2 ;	C14-	C21-	C22-	C23
15	27	28	53	9	180.00	15.16700	2 ;	C14-	C21-	C22-	H15
15	27	32	31	9	180.00	15.16700	2 ;	C14-	C21-	C26-	C25
15	27	32	52	9	180.00	15.16700	2 ;	C14-	C21-	C26-	H14
16	11	12	17	9	180.00	15.16700	2 ;	C15-	C10-	C11-	C16
16	15	14	43	9	180.00	15.16700	2 ;	C15-	C14-	C13-	H5
16	15	27	28	9	180.00	4.18400	2 ;	C15-	C14-	C21-	C22
16	15	27	32	9	180.00	4.18400	2 ;	C15-	C14-	C21-	C26
16	25	22	21	9	0.00	3.34720	2 ;	C15-	O3-	P-	O
16	25	22	23	9	0.00	0.41840	3 ;	C15-	O3-	P-	O1
16	25	22	23	9	180.00	3.34720	2 ;	C15-	O3-	P-	O1
16	25	22	24	9	0.00	0.41840	3 ;	C15-	O3-	P-	O2
16	25	22	24	9	180.00	3.34720	2 ;	C15-	O3-	P-	O2
17	12	13	20	9	180.00	15.16700	2 ;	C16-	C11-	C12-	C19
17	18	19	20	9	180.00	15.16700	2 ;	C16-	C17-	C18-	C19
17	18	19	46	9	180.00	15.16700	2 ;	C16-	C17-	C18-	H8
18	19	20	47	9	180.00	15.16700	2 ;	C17-	C18-	C19-	H9
19	18	17	44	9	180.00	15.16700	2 ;	C18-	C17-	C16-	H6
20	13	14	43	9	180.00	15.16700	2 ;	C19-	C12-	C13-	H5
20	19	18	45	9	180.00	15.16700	2 ;	C19-	C18-	C17-	H7
21	8	9	26	9	180.00	15.16700	2 ;	O-	C7-	C8-	C20
25	16	15	27	9	180.00	15.16700	2 ;	O3-	C15-	C14-	C21
26	9	10	42	9	180.00	15.16700	2 ;	C20-	C8-	C9-	H4
26	33	34	35	9	180.00	15.16700	2 ;	C20-	C27-	C28-	C29
26	33	34	50	9	180.00	15.16700	2 ;	C20-	C27-	C28-	H12
26	37	36	35	9	180.00	15.16700	2 ;	C20-	C31-	C30-	C29
26	37	36	51	9	180.00	15.16700	2 ;	C20-	C31-	C30-	H13

27	15	14	43	9	180.00	15.16700	2 ;	C21-	C14-	C13-	H5
27	28	29	30	9	180.00	15.16700	2 ;	C21-	C22-	C23-	C24
27	28	29	48	9	180.00	15.16700	2 ;	C21-	C22-	C23-	H10
27	32	31	30	9	180.00	15.16700	2 ;	C21-	C26-	C25-	C24
27	32	31	49	9	180.00	15.16700	2 ;	C21-	C26-	C25-	H11
28	27	32	31	9	180.00	15.16700	2 ;	C22-	C21-	C26-	C25
28	27	32	52	9	180.00	15.16700	2 ;	C22-	C21-	C26-	H14
28	29	30	31	9	180.00	15.16700	2 ;	C22-	C23-	C24-	C25
28	29	30	54	9	180.00	15.16700	2 ;	C22-	C23-	C24-	N
29	28	27	32	9	180.00	15.16700	2 ;	C23-	C22-	C21-	C26
29	30	31	32	9	180.00	15.16700	2 ;	C23-	C24-	C25-	C26
29	30	31	49	9	180.00	15.16700	2 ;	C23-	C24-	C25-	H11
29	30	54	58	9	180.00	2.51040	2 ;	C23-	C24-	N-	O4
29	30	54	59	9	180.00	2.51040	2 ;	C23-	C24-	N-	O5
30	29	28	53	9	180.00	15.16700	2 ;	C24-	C23-	C22-	H15
30	31	32	52	9	180.00	15.16700	2 ;	C24-	C25-	C26-	H14
31	30	29	48	9	180.00	15.16700	2 ;	C25-	C24-	C23-	H10
31	30	54	58	9	180.00	2.51040	2 ;	C25-	C24-	N-	O4
31	30	54	59	9	180.00	2.51040	2 ;	C25-	C24-	N-	O5
32	27	28	53	9	180.00	15.16700	2 ;	C26-	C21-	C22-	H15
32	31	30	54	9	180.00	15.16700	2 ;	C26-	C25-	C24-	N
33	26	37	36	9	180.00	15.16700	2 ;	C27-	C20-	C31-	C30
33	26	37	56	9	180.00	15.16700	2 ;	C27-	C20-	C31-	H16
33	34	35	36	9	180.00	15.16700	2 ;	C27-	C28-	C29-	C30
33	34	35	55	9	180.00	15.16700	2 ;	C27-	C28-	C29-	N1
34	33	26	37	9	180.00	15.16700	2 ;	C28-	C27-	C20-	C31
34	35	36	37	9	180.00	15.16700	2 ;	C28-	C29-	C30-	C31
34	35	36	51	9	180.00	15.16700	2 ;	C28-	C29-	C30-	H13
34	35	55	60	9	180.00	2.51040	2 ;	C28-	C29-	N1-	O6

34 35 55 61 9 180.00 2.51040 2; C28- C29- N1- O7  
 35 34 33 57 9 180.00 15.16700 2; C29- C28- C27- H17  
 35 36 37 56 9 180.00 15.16700 2; C29- C30- C31- H16  
 36 35 34 50 9 180.00 15.16700 2; C30- C29- C28- H12  
 36 35 55 60 9 180.00 2.51040 2; C30- C29- N1- O6  
 36 35 55 61 9 180.00 2.51040 2; C30- C29- N1- O7  
 37 26 33 57 9 180.00 15.16700 2; C31- C20- C27- H17  
 37 36 35 55 9 180.00 15.16700 2; C31- C30- C29- N1  
 38 1 2 3 9 180.00 15.16700 2; H- C- C1- C2  
 38 1 2 39 9 180.00 15.16700 2; H- C- C1- H1  
 38 1 6 5 9 180.00 15.16700 2; H- C- C5- C4  
 38 1 6 41 9 180.00 15.16700 2; H- C- C5- H3  
 39 2 3 40 9 180.00 15.16700 2; H1- C1- C2- H2  
 44 17 18 45 9 180.00 15.16700 2; H6- C16- C17- H7  
 45 18 19 46 9 180.00 15.16700 2; H7- C17- C18- H8  
 46 19 20 47 9 180.00 15.16700 2; H8- C18- C19- H9  
 48 29 28 53 9 180.00 15.16700 2; H10- C23- C22- H15  
 48 29 30 54 9 180.00 15.16700 2; H10- C23- C24- N  
 49 31 30 54 9 180.00 15.16700 2; H11- C25- C24- N  
 49 31 32 52 9 180.00 15.16700 2; H11- C25- C26- H14  
 50 34 33 57 9 180.00 15.16700 2; H12- C28- C27- H17  
 50 34 35 55 9 180.00 15.16700 2; H12- C28- C29- N1  
 51 36 35 55 9 180.00 15.16700 2; H13- C30- C29- N1  
 51 36 37 56 9 180.00 15.16700 2; H13- C30- C31- H16

[ dihedrals ] ; impropers

; treated as propers in GROMACS to use correct AMBER analytical function

; i j k l func phase kd pn

1 3 2 39 4 180.00 4.60240 2; C- C2- C1- H1

1	5	6	41	4	180.00	4.60240	2;	C-	C4-	C5-	H3
2	4	3	40	4	180.00	4.60240	2;	C1-	C3-	C2-	H2
3	5	4	7	4	180.00	4.60240	2;	C2-	C4-	C3-	C6
4	6	5	10	4	180.00	4.60240	2;	C3-	C5-	C4-	C9
4	8	7	11	4	180.00	4.60240	2;	C3-	C7-	C6-	C10
5	9	10	42	4	180.00	4.60240	2;	C4-	C8-	C9-	H4
7	9	8	21	4	180.00	4.60240	2;	C6-	C8-	C7-	O
8	10	9	26	4	180.00	4.60240	2;	C7-	C9-	C8-	C20
11	15	16	25	4	180.00	4.60240	2;	C10-	C14-	C15-	O3
12	14	13	20	4	180.00	4.60240	2;	C11-	C13-	C12-	C19
12	16	11	7	4	180.00	4.60240	2;	C11-	C15-	C10-	C6
12	18	17	44	4	180.00	4.60240	2;	C11-	C17-	C16-	H6
13	15	14	43	4	180.00	4.60240	2;	C12-	C14-	C13-	H5
13	17	12	11	4	180.00	4.60240	2;	C12-	C16-	C11-	C10
13	19	20	47	4	180.00	4.60240	2;	C12-	C18-	C19-	H9
14	16	15	27	4	180.00	4.60240	2;	C13-	C15-	C14-	C21
17	19	18	45	4	180.00	4.60240	2;	C16-	C18-	C17-	H7
18	20	19	46	4	180.00	4.60240	2;	C17-	C19-	C18-	H8
28	30	29	48	4	180.00	4.60240	2;	C22-	C24-	C23-	H10
28	32	27	15	4	180.00	4.60240	2;	C22-	C26-	C21-	C14
29	27	28	53	4	180.00	4.60240	2;	C23-	C21-	C22-	H15
29	31	30	54	4	180.00	4.60240	2;	C23-	C25-	C24-	N
30	32	31	49	4	180.00	4.60240	2;	C24-	C26-	C25-	H11
30	58	54	59	4	180.00	4.60240	2;	C24-	O4-	N-	O5
31	27	32	52	4	180.00	4.60240	2;	C25-	C21-	C26-	H14
33	35	34	50	4	180.00	4.60240	2;	C27-	C29-	C28-	H12
33	37	26	9	4	180.00	4.60240	2;	C27-	C31-	C20-	C8
34	26	33	57	4	180.00	4.60240	2;	C28-	C20-	C27-	H17
34	36	35	55	4	180.00	4.60240	2;	C28-	C30-	C29-	N1

35 37 36 51 4 180.00 4.60240 2; C29- C31- C30- H13  
35 60 55 61 4 180.00 4.60240 2; C29- O6- N1- O7  
36 26 37 56 4 180.00 4.60240 2; C30- C20- C31- H16  
38 1 6 2 4 180.00 4.60240 2; H- C- C5- C1

## 9.2 Geometry optimization of 1b/3a

The structure of the complex **1b/3a** (see Figure 4) was optimized with GAUSSIAN<sup>22</sup> using TPSS,<sup>23</sup> the def2-SVP basis set<sup>24,25</sup> along with the D3 version of Grimme's dispersion with the D3 damping function<sup>26</sup> an „ultrafine“ integration grid, and the SMD<sup>27</sup> implicit solvent for CH<sub>2</sub>Cl<sub>2</sub> with an epsilon value of eps=16.2 to mimic the solvent properties at 180 K.<sup>8</sup> Considering this structure is similar to known structures,<sup>28</sup> no frequency analysis was performed.

The optimized structure is given below:

```
1\1\GINC-NODE0599\FOpt\RTPSSTPSS\def2SVP\C59H60N1O5P1\MAG24964\16-Apr-
2020\0\\#p tpsstpss/def2svp empiricaldispersion=gd3 int=ultrafine opt
scrf=(smd, solvent=ch2cl2, read)\\e1kombi\\0,1\C,6.1909568563,-1.78881
77742,-3.9331692683\C,5.8049083302,-2.8903039276,-3.1229283401\C,4.818
7376043,-2.7460262924,-2.1584308845\C,4.168264312,-1.4915503117,-1.952
1631679\C,4.5311424406,-0.3858467278,-2.8057439312\C,5.5599735257,-0.5
648323881,-3.7796771752\C,3.1430016725,-1.2956646087,-0.9583936278\C,2
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.7826144402\C,3.839876094,0.854299916,-2.6851727676\C,2.7384201326,-2.
3621176803,0.0023471007\C,3.6720009752,-2.9779563729,0.9111457956\C,3.
2168866205,-4.0547774622,1.7566607786\C,1.8521089617,-4.4586376561,1.7
06184883\C,0.9272897935,-3.8168565278,0.8920336916\C,1.3981338083,-2.7
592903387,0.056594759\C,5.0264933945,-2.5438912433,1.0408780459\C,5.89
51874027,-3.1653357346,1.9254884224\C,5.4555542803,-4.2529799699,2.727
347597\C,4.1407924284,-4.6823985808,2.6464571995\O,1.4457027832,0.1003
051973,-0.0092194836\P,-0.0291849237,-0.5925085014,-0.3837511933\O,-0.
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.9170908325\O,0.4752577034,-2.133131884,-0.7714015132\C,2.0194794579,2
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4, -2.3200765409 \C, -5.7907377186, -0.3461607137, -0.9553910356 \C, -4.75761  
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